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## Mineralogy and Density of Archean Volcanic Crust in the Mantle Transition Zone

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#### Abstract

The composition of Archean volcanic crust can be characterized by a higher Mg/Si ratio than modern mid-ocean ridge basalt (MORB), because of the higher degree melting from the warmer mantle in the Archean. Although modern MORB may become less dense than the surrounding mantle beneath the mantle transition zone (MTZ), the Mg-rich composition of Archean volcanic crust may result in the different density, and therefore different sinking behavior near the MTZ. In order to understand the compositional effect of Archean volcanic crust on the sinking behaviors and the scale of mantle mixing in the Archean, we investigated the mineralogy and density of Archean volcanic crust near the MTZ (470-910 km-depth). We conducted experiments at 19-34 GPa and 1,400-2,400 K using the laser-heated diamond anvil cell (LHDAC) combined with insitu X-ray diffraction (XRD). The in-situ XRD and the chemical analysis revealed that Archean volcanic crust forms garnet and ringwoodite (84 and 16 vol%), which gradually transforms to Brg and CaPv (82 and 18 vol%) at 23-25 GPa and 1,800 K. Our insitu XRD experiments allowed us to measure the volumes of stable phases and to estimate their densities at high pressure and temperature. The results suggest that Archean volcanic crust maintains greater density than the pyrolitic mantle in the Archean regardless of temperature at 20-34 GPa (570-850 km-depth), promoting further sinking

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into the deeper mantle in the Archean. We also considered the density of the subducting slab in the Archean. The density model showed that the subducting slab is still denser or at least equally dense as the surrounding pyrolitic mantle in the Archean. *Keywords:* Archean volcanic crust, Density, Mineralogy, Mantle transition zone, In-situ X-ray diffraction, High P-T experiments

#### 1 1. Introduction

Surface materials have been transported into the deep mantle and played vital roles in mantle geochemistry and dynamics for billions of years in the Earth's history (e.g., Van Hunen and Moyen, 2012; Laurent et al., 2014; Tang et al., 2016; Condie, 2018). Seismic tomography images have shown stagnation of the slabs beneath the mantle transition zone (MTZ) in the Western Pacific (e.g., van der Hilst and Seno, 1993; Fukao et al., 2009; Li and McNamara, 2013; King et al., 2015). Indeed, sinking of the slabs into the deep mantle can be challenged beneath the MTZ at 670–760 km-depth (Christensen, 1997; Ogawa, 2003; Davies, 2008; Klein et al., 2017) where the density of the surrounding mantle increases dramatically because of the post-spinel transition (Hirose, 2002; Ye et al., 2014; Ishii et al., 2018).

While the density of the modern subductig slab has been well documented at high 12 pressures and high temperatures (Hirose et al., 1999; Irifune and Ringwood, 1993; 13 Ringwood and Irifune, 1988), the density of the early Earth's materials is not the case. 14 In the Archean, the mantle potential temperature was likely much greater than present 15 day by approximately 200 K (Korenaga, 2008; Herzberg et al., 2010). The high tem-16 perature results in the more extensive partial melting in the Archean, resulting in two 17 important differences for Archean volcanic crust. First, Archean volcanic crust forms with a much greater MgO content than present-day mid ocean ridge basalt (MORB; 19 Table 1) (Herzberg et al., 2007, 2010). The different composition of Archean volcanic 20 crust would lead to different mineralogy and density at high pressure-temperature (P-21 T) conditions in the mantle. Second, the thickness of the crust should be much greater 22 in the Archean (up to 45 km-thick) than in the modern day (7 km-thick; Herzberg et al., 23 2010; Herzberg and Rudnick, 2012). With the large thickness, Archean volcanic crust 24

takes a large portion of the subducting slab in the Archean, and therefore plays a more
 important role in the sinking behavior of the subducting slab.

Archean volcanic crust could have been transported into the mantle through a range 27 of different processes: delamination (e.g., Bédard, 2006; Johnson et al., 2014), and 28 subduction (e.g., Laurent et al., 2014; Tang et al., 2016; Condie, 2018). For the case 29 of subduction, timing for the onset of modern-style plate tectonics is still under debate. 30 Multiple lines of geochemical evidence suggest that modern-style global subduction 3. would have occurred since 2.0–3.5 Ga (Laurent et al., 2014; Tang et al., 2016; Condie, 32 2018). Subduction style in the Archean could have been episodic (Van Hunen and 33 Moyen, 2012). Geodynamic modeling has shown that the hotter mantle in the Archean 34 results in weak subducting slabs, leading to frequent break-off of the slabs (Van Hunen 35 and Moyen, 2012). 36

If Archean volcanic crust remains as a heterogeneity in the present-day mantle owing to the slow chemical diffusion at high pressure (Holzapfel et al., 2005; Stixrude and Lithgow-Bertelloni, 2012; Xu et al., 2011; Ito and Toriumi, 2010), the behavior of such heterogeneity is also important for understanding the origin of the seismic heterogeneities found in the mantle (e.g. Fukao et al., 2009; Fukao and Obayashi, 2013; Wu et al., 2019).

It is also of interest if Archean volcanic crust would have contributed to the rise of 43 atmospheric O2 at approximately 2.4 Ga (the Great Oxidation Event or GOE) (Kast-44 ing, 2013). The MgO-rich composition might have resulted in the more extensive  $H_2$ 45 production through serpentinization than in the modern day (Kasting, 2013). The com-46 positional change in the volcanic crust during the mantle cooling might have reduced 47 such H<sub>2</sub> production, leading to the accumulation of O<sub>2</sub> in the atmosphere. In addi-48 tion, the recent measurements of the V partitioning in ancient volcanic rocks suggested 49 that the reducing Archean upper mantle might have undergone the secular oxidation 50 between 3.5 and 1.9 Ga (Nicklas et al., 2019). Possible mechanisms for the oxidation 5 processes include not only the compositional change of the volcanic crust recycled into 52 the mantle, but also homogenization of the redox state of the mantle by effective man-53 tle mixing (Nicklas et al., 2019; Aulbach and Stagno, 2016). The style of the mantle 54 convection in the Archean could have been also affected by the changes in the tectonic 55

mode. It is possible that the subducting slabs might have facilitated the whole mantle mixing with the primordial oxidizing materials stored in the lower mantle (Nicklas
et al., 2019; Aulbach and Stagno, 2016; Andrault et al., 2018a).

The composition of the Archean materials sunken to the mantle has been modeled 59 with komatiitic compositions (e.g. Nishihara and Takahashi, 2001; Klein et al., 2017). 60 Nishihara and Takahashi (2001) conducted multi-anvil experiments and calculated the 61 density of komatiite with thermodynamic parameters and the calculated zero-pressure 62 densities of the synthesized phases. They showed that the density of komatiite is greater 63 than the PREM density in the MTZ up to 23 GPa. The density of the surrounding man-64 tle in the Archean is likely lower than the PREM density (i.e., present-day mantle den-65 sity) because of the higher temperature. Therefore, the density contrast between their 66 komatiite and the Archean mantle may be even greater. However, their komatiite com-67 position may not be representative Archean volcanic crust formed by partial melting 68 of the mantle, because of the higher MgO content (32 wt%) and the lower Al<sub>2</sub>O<sub>3</sub> con-69 tent (4 wt%) compared to non-arc basaltic compositions in the Archean (23 and 9 wt%, 70 respectively; Herzberg et al., 2010; Johnson et al., 2014, ; Table 1). Furthermore, the 71 density was not directly measured at high P-T. 72

Klein et al. (2017) has recently studied the density of komatiitic composition at depths of the MTZ and the topmost lower mantle using a thermodynamic modelling program, Perple\_X (Connolly, 2009). They used a less MgO-rich komatiitic composition (26 wt% of MgO) for Archean volcanic crust (Herzberg et al., 2010). The model predicted that komatiite is denser than the pyrolitic Archean mantle in the MTZ and the topmost lower mantle, supporting the likelihood of the sinking of the materials into the lower mantle.

However, unlike other compositions, such as modern MORB and pyrolite which
have been cross-examined by theoretical calculations (Xu et al., 2008; Stixrude and
Lithgow-Bertelloni, 2012) and in-situ experiments (Ricolleau et al., 2010; Hirose et al.,
2005), the density of the Archean volcanic crustal composition has not been directly
measured by in-situ experiments. Also, it would be worthwhile to explore differences in
mineralogy and density between komatiite and modeled compositions (Johnson et al.,
2014) for Archean volcanic crust. More importantly, no study has explored the com-

<sup>87</sup> bined density of the crust and harzburgitic layer for the Archean subducting slabs. The
<sup>88</sup> lack of such consideration limits our understanding of the sinking behaviors of the
<sup>89</sup> Archean subducting slabs.

As mentioned above, Archean volcanic crust is different from modern MORB in two aspects: 1) much more MgO-rich composition and 2) greater thickness (Korenaga, 2008; Herzberg et al., 2010). Because of these differences, Archean volcanic crust would have played a different role in the sinking behaviors of the Archean subducting slabs than modern MORB does. It would be particularly important to address a question of how sinking behaviors of subducting slabs may have evolved over time as Earth's mantle cools.

Here, we report experimental results on the composition for Archean volcanic crust at pressures of 19–34 GPa and temperatures of 1,400–2,400 K. Our in-situ X-ray diffraction provides direct measurements of the unit-cell volumes of the stable phases, allowing us to estimate the densities at high P-T. We also explored the effects of harzburgite layer using Perple\_X, for the case where Archean volcanic crust is transported to the MTZ through subduction style.

103 [Table 1]

#### **104 2. Experimental Methods**

#### 105 2.1. Starting material

We synthesized a homogeneous glass using the laser-levitation method (Tangeman 106 et al., 2001) under a reducing condition (1:1 ratio of the gas mixture  $CO:CO_2$ ). The 107 chemical composition of the starting material was obtained from the partial melting 108 model by Johnson et al. (2014) (Table 1). The model assumes that the MgO and FeO 109 contents of the crustal composition linearly increase with an increase in the mantle 110 potential temperature  $(T_P)$  based on the reported chemical compositions of a range of 111 non-arc volcanic crustal compositions from 3.5 Ga to the present day (Johnson et al., 112 2014; Herzberg et al., 2007, 2010; Palin and White, 2016). From this model, we chose 113 the composition for Archean volcanic crust (2.5–3.0 Ga; hereafter B30) that might have 114

formed at a  $T_P$  of ~1,900 K. We confirmed successful synthesis of the starting material in an electron probe micro analyzer (JEOL JXA-8530F EPMA) combined with wavelength-dispersive X-ray spectrometer (WDS) at Arizona State University (ASU) (Table 1). We did not measure the Fe<sup>3+</sup>/ $\Sigma$ Fe in the starting material.

#### 119 2.2. Laser-heated diamond-anvil cell experiments

The powdered glass was mixed with ~10 wt% of Au powder (grain size of  $1-3 \mu m$ ), 120 which served as a pressure standard and a laser absorber in the laser-heated diamond-12 anvil cell (LHDAC) experiments. The mixture was compressed to thin foils with a 122 size of  $100-200\,\mu\text{m}$  and a thickness of  $\sim 10\,\mu\text{m}$ . Then, the foil was loaded into a hole 123 in a pre-indented rhenium gasket with a micro-manipulator (Microsupport Axis Pro 124 SS) at ASU. We placed four spacers ( $<10 \,\mu m$  sized pure sample particle) between the 125 sample foil and each side of the diamond culets for the gas or liquid medium loading. 126 We loaded Ne and Ar in the COMPRES-GSECARS high-pressure gas loading system 127 (Rivers et al., 2008) or the cryogenic liquid loading system at ASU. For the solid 128 media (NaCl and KCl), powder was dried at 120°C for 4 hours before loading. The 129 solid media were prepared as foils and placed below and above the sample foil in the 130 sample chamber. We compressed the samples in the symmetric-type diamond-anvil 13 cell. The samples were compressed with diamond anvils (400  $\mu$ m-sized culet). 132

We have conducted LHDAC experiments combined with in-situ X-ray diffraction 133 (XRD) at sector 13-IDD of the GeoSoilEnviroConsortium (GSECARS) in the Ad-134 vanced Photon Source (APS) (Prakapenka et al., 2008) (Tables S1 and S2). The pres-135 sure and temperature ranges of the experiments were 19-34 GPa and 1,400-2,400 K, 136 respectively. For run #4430 only, we synthesized the sample in LHDAC at ASU and 137 then measured diffraction patterns at high pressure and room temperature in DAC at 138 sector 12.2.2 in the Advanced Light Source (ALS) (Table S1). The wavelengths of the 139 monochromatic X-ray beams were 0.3344, 0.4133, and 0.4959 Å for GSECARS and 140 0.4971 Å for ALS. For in-situ experiments, we focused the laser beams on both sides 14 of the sample and aligned the laser beams coaxially with the X-ray beam by visually 142 observing X-ray fluorescence, in order to measure XRD patterns at the center of the 143 laser heating spot. Typical beam diameters for the X-ray focus and the laser hotspot on 144

the sample in LHDAC were ~5 and  $20\,\mu$ m, respectively. We fit the thermal radiation spectra measured from both sides of the sample to Planck gray-body equation for the estimation of temperature. We attached a gas membrane to the LHDAC for fine control of pressure during in-situ laser heating (Letoullec et al., 1988), except for run #218 where the pressure was increased manually to the target pressure.

We measured diffraction images using a MarCCD or a Dectris Pilatus detector at 150 GSECARS. For run #4430, diffraction images were collected with a Mar345 detec-15 tor at ALS. The diffraction images were integrated to 1D diffraction patterns with 152 the calibration parameters obtained from the LaB<sub>6</sub> standard in the DIOPTAS software 153 (Prescher and Prakapenka, 2015). Diffraction images were collected for 5–10 seconds 154 to ensure enough quality for data analysis. For run #4430 at ALS, we exposed the de-155 tector for 120-300 seconds. The collected diffraction patterns were analyzed and fitted 156 with a pseudo-Voigt profile function to obtain the peak positions in the PeakPo soft-157 ware (Shim, 2017a). We obtained unit-cell parameters using the PeakPo and UnitCell 158 softwares (Holland and Redfern, 1997). We determined pressures with the equation 159 of state (EOS) of Au (Dorogokupets and Dewaele, 2007) using the Pytheos software 160 (Shim, 2017b). The pressures at high temperatures were further corrected according to 16 Ye et al. (2017). The unit-cell volume of Au was calculated from the 2–5 diffraction 162 lines for majority of the data points. 163

We heated multiple spots in each sample and avoided areas that are close to the Re gasket. For each heating spot, an area with a diameter of  $20\,\mu$ m was heated for up to 11 minutes (Table S1). The locations of heating spots were chosen with an enough distance up to  $30\,\mu$ m from the previously heated areas in order to start a new heating cycle with unheated fresh areas.

In-situ XRD patterns showed that our glass sample transformed into crystalline phases within 5–60 seconds depending on temperature. Once formed, the XRD peak intensities of crystalline phases gradually grow over time without significant changes in the further heating. For most runs, the laser power was first increased at a previously heated spot until a target temperature is reached, and then instantly translated to an unheated amorphous area such that the new area can reach target temperature directly. This heating method helps preventing kinetic effects because the amorphous starting material does not experience low temperatures during gradual increase in laser power

<sup>177</sup> in typical laser heating experiments.

178 [Fig. 1]

#### 179 2.3. Chemical analysis

We recovered the synthesized samples from runs #218 and #4430 in a micro-180 manipulator. The samples were processed for the STEM measurements in a focused ion 18 beam (FIB; FEI Nova 200 and FEI Helios) at ASU and University of Arizona (UofA), 182 respectively. We extracted the center part of the heated area and mounted it on a Cu 183 or Mo grid. The composition of the sample was analyzed in the aberration-corrected 184 scanning transmission electron microscope (STEM): JEOL ARM200F at ASU and Hi-185 tachi HF-5000 at UofA combined with energy-dispersive X-ray spectroscopy (EDS). 186 A 120 and 200 keV of acceleration voltages were used at ASU and UofA, respectively. 187 We measured standard spectra from the starting glass material for the data analysis. We 188 performed a 2D chemical mapping for run #4430 and then estimated volume fractions 189 of phases from the area fraction (Fig. 1). We also performed a point analysis with ex-190 posure time of 20–40 seconds depending on the beam sensitivity at a target grain. The 191 collected EDS spectra were fitted using a python code LMFIT (Newville et al., 2016). 192 [Fig. 2] 193

#### **3. Experimental results**

<sup>195</sup> We have found three distinct P-T fields with different mineralogy in our exper-<sup>196</sup> iments (Fig. 2): garnet-dominant (Gt-dominant), garnet-bridgmanite-transition (Gt-<sup>197</sup> Brg-transition), and bridgmanite-dominant (Brg-dominant) P-T fields (Fig. 3).

198 [Fig. 3]

#### 199 3.1. Garnet-dominant P–T field

Between 19 and 25 GPa, we observed Gt and ringwoodite (Rw) as stable phases (Fig. 2a). The diffraction intensities indicates that Gt represents the major phase of the mineralogy (Fig. 3). We employed the Rietveld refinement using GSAS-II (Toby and Von Dreele, 2013) and obtained 86.5 and 14.5 vol% for Gt and Rw, respectively, which is supported by our STEM observations (Table 2). Fig. 1 shows a high-angle annular dark-field (HDADF) image of the sample from run #4430 which was synthesized at 19.5 GPa (pressure measured at 300 K) and 1,900 K in LHDAC. The chemical maps of the image shows that Gt exists as matrix that surrounds the Rw grains. From the STEM-EDS images, we obtained 85.3 and 14.7 vol% for the area proportions of Gt and Rw, respectively, consistent with our mass-balance calculation, which yielded 82.7 and 17.3 vol% of Gt and Rw, respectively (Table 2).

Fe<sup>3+</sup>/ $\Sigma$ Fe was not measured for the recovered sample in this study. Because the starting glass was synthesized in the reducing condition, we expect the dominant valence state of Fe is Fe<sup>2+</sup>. If Fe<sup>3+</sup>/ $\Sigma$ Fe was large in the synthesized samples, the phase proportion of Gt would be even larger because Gt can accommodate the large amount of Fe<sup>3+</sup> in the crystal structure, unlike Rw (McCammon and Ross, 2003).

The modern MORB mineralogy features Gt as the most abundant component (60-216 90 vol%) (Irifune and Ringwood, 1993; Litasov and Ohtani, 2005) as does B30 min-217 eralogy (86 vol%; Table 2). However, stable minor phases are significantly different 218 between these two compositions. The (Mg+Fe+Ca)/Si ratio of B30 is higher than 1 219 (~1.15) and therefore excess MgO and FeO can form Rw in addition to Gt as ob-220 served in our experiments. On the other hand, in the modern MORB composition the 22 (Mg+Fe+Ca)/Si ratio is low enough (0.63) to form stishovite (Stv) from excess silica 222 content (Gale et al., 2013). In addition, the greater Al content (15.5 wt%) in modern 223 MORB adds Al-rich phase to the mineralogy, whereas nearly all Al is incorporated in 224 Gt in B30 (8.6 wt%). The mineralogy of the modern MORB composition consists of 225 Gt, Stv, CaPv, and Al-phase, in an order of their abundances, at 15-26 GPa and 1,473-226

227 2,573 K (Irifune et al., 1986; Irifune and Ringwood, 1993; Hirose and Fei, 2002).

At a similar pressure range, pyrolite has Rw (or Wd), Gt, CaPv, and akimotoite (Aki) in an order of abundance (Hirose, 2002; Ye et al., 2014; Ishii et al., 2018). Pyrolite has a much higher (Mg+Fe+Ca)/Si ratio (1.49) than modern MORB and B30 (Table 1), making Rw (or Wd) the most abundant phase in the mineralogy. As shown in Table 2, Gt represents 84 vol% in B30, whereas Rw represents 60 vol% in pyrolite (Irifune et al., 2010; Ishii et al., 2018).

The chemical composition of Gt is significantly different among B30, modern

MORB, and pyrolite (Table S2). Compared to B30, Gt in modern MORB has twice more Al<sub>2</sub>O<sub>3</sub> (8.6 and 20.6 wt%, respectively) and half as many MgO (20.9 and 12.3 wt%, respectively) as B30 has (Hirose and Fei, 2002). Gt in pyrolite has nearly half as much FeO (5.0 and 11.8 wt%, respectively) and CaO (4.6 and 9.0 wt%) as B30 does, but slightly more MgO than in B30 (Hirose, 2002). These compositional differences are important to consider for the density and velocity profiles.

241 [Table 2]

#### 242 3.2. Garnet-bridgmanite-transition P–T field

At intermediate pressures in our experiments, we observed the Gt-Brg-transition 243 P-T field in B30 (Fig. 3). Above 23 GPa, Rw disappears and Brg appears, suggesting 244 the occurrence of the post-spinel transition where Rw decomposes into Brg and fer-245 ropericlase (Fp). While the diffraction peak intensities show that Gt is still the major 246 phase in the Gt+Brg mineralogy (Fig. 2), we could not resolve Fp in XRD patterns 247 because of the peak overlap with Ar, Gt, or Brg peaks. Furthermore, the peak inten-248 sities of Fp is expected to be small because of the possible reduction process in the 249 amount of Fp. Brg is known to have high capacity for Fe<sup>3+</sup> through the charge dispro-250 portionation reaction: 3FeO to 2Fe<sub>2</sub>O<sub>3</sub> and Fe metal (Frost et al., 2004). If a significant 25 fraction of Fe is  $Fe^{3+}$ , the more Brg can form by the coupled-substitution of  $Mg^{2+}$  + 252  $Si^{4+}$  with  $Fe^{3+} + Al^{3+}$ , and therefore reduce the available FeO for Fp in the system 253 (e.g., McCammon, 1997; Andrault et al., 2001). We estimated  $Fe^{3+}/\Sigma$ Fe in Brg based 254 on the measured compositions by assuming the charge-balanced substitutions of Fe<sup>3+</sup> 255 and Al<sup>3+</sup>. The estimation yields approximately 0.57 for Fe<sup>3+</sup>/ $\Sigma$ Fe, which is consistent 256 with previous reports (e.g. Shim et al., 2017). However, we note that this estimation 257 method for  $Fe^{3+}/\Sigma Fe$  is very sensitive to the Si# in the stoichiometric composition 258 of Brg, and therefore the large uncertainties are involved given the error range of the 259 EDS quantitative analysis (~10%). We did not observe Fe metal in either STEM ob-260 servations or XRD patterns, perhaps because of the expected small amount of Fe metal 26 (Section 3.3). Although the limited number of data points increases the uncertainties, 262 the P-T slope of Brg-in boundary was roughly estimated to be  $-0.8 \pm 0.3$  MPa/K, in 263 a good agreement with the Clapeyron slope of the post-spinel transition from previous 264

265 studies (e.g. Katsura et al., 2003; Litasov et al., 2005; Ishii et al., 2011, 2018).

At higher pressures above 1,600 K, CaPv was observed alongside Gt and Brg. The 266 diffraction peak intensities of Brg are in general stronger in the Gt+Brg+CaPv mineral-26 ogy than in the Gt+Brg mineralogy. These observations suggest the post-Gt transition 268 where Gt decomposes into CaPv and Brg. We observed the two phase boundaries 269 for the post-Gt transition with 1.7 GPa of the pressure interval: the CaPv-in bound-270 ary and Gt-out boundary. The P-T slopes of these boundaries were determined to be 27  $3.5 \pm 0.5$  MPa/K, which is in the middle of the range of the Clapeyron slopes of the 272 post-Gt transition in previous studies (Hirose et al., 1999; Kubo and Akaogi, 2000; 273 Oguri et al., 2000; Hirose, 2002; Litasov et al., 2004; Ishii et al., 2018). Note that the 274 uncertainties of the P-T slopes may be large due to the limited number of data points 275 across the boundaries. The gradual post-Gt transition has been reported by earlier ex-276 periments of Al-bearing Gt (Irifune et al., 1996) up to 2.1 GPa of an interval at 1,773 K 277 for 3–11 mol% of Al<sub>2</sub>O<sub>3</sub>. Gt in B30 contains 4.8 mol% of Al<sub>2</sub>O<sub>3</sub> which is estimated 278 to yield 1.1 GPa interval for the post-Gt transition according to Irifune et al. (1996). 279 However, Gt in B30 also contains a significant amount of FeO (11.8 wt%; Table 1), the 280 effect of which on the pressure interval of the post-Gt transition is not well known. 28

The opposite signs of P-T slopes of the post-spinel and post-Gt transitions ( $-0.8\pm0.3$ 282 and 3.5±0.5 MPa/K, respectively) results in the wider stability of the Gt-Brg-transition 283 P-T field with an increase in temperature above 1,600 K. Below 1,600 K, we do not 284 have sufficient number of data points due to the large kinetic effects at lower tempera-285 tures. One run at 22.6 GPa and 1,570 K (the green diamond in Fig. 3) shows the signs 286 of Brg 112, 200, and 220 lines and strong CaPv 110 line in the XRD pattern, alongside 287 the strong Gt and Rw lines. These observations are in line with CaPv-in boundary ob-288 served from the higher temperatures, suggesting that the post-Gt transition may occur 289 at shallower depths than the post-spinel transition at lower temperatures. 290

The most noticeable difference in mineralogy between modern MORB and B30 at the intermediate pressure range is the depths of Brg-in and Gt-out boundaries (Fig. 3b). For example, at 1,700 K Brg-in boundary appears as the post-spinel transition at a lower pressure of 23 GPa in B30 than the post-Gt transition at 25.8 GPa in modern MORB (Hirose et al., 1999). The Si-rich composition of modern MORB does not form Rw in the mineralogy, resulting in the absence of the post-spinel transition, which can occur at the shallower depth than the post-Gt transition. Then, Gt completely transformed to Brg at 24.8 GPa in B30, whereas Gt persists until 26.8 GPa in modern MORB (60– 70 vol%) (Hirose et al., 1999; Irifune and Ringwood, 1993; Litasov and Ohtani, 2005). The difference in the depth and the P-T slope of Gt-out boundary between B30 and modern MORB is likely caused by the compositional difference in Gt (Table S2).

<sup>302</sup> Pyrolite has similar mineralogy as B30 in the Gt-Brg-transition P-T field. In py-<sup>303</sup> rolitic composition, Gt and Brg coexist as major components (Hirose, 2002; Ishii et al., <sup>304</sup> 2018) at 22–26 GPa. Also, pyrolite exhibits the negative slope for Brg-in boundary be-<sup>305</sup> cause of the post-spinel transition. Brg-in boundary appears at a nearly same pressure <sup>306</sup> in B30 and pyrolite: 22.8 and 22.6 GPa at 1,900 K, respectively (Ishii et al., 2018). On <sup>307</sup> the other hand, the Gt-out boundary appears at a higher pressure in B30 than in pyrolite <sup>308</sup> by 1.1 GPa: 25.5 and 24.4 GPa at 1,900 K, respectively (Ishii et al., 2018).

#### 309 3.3. Bridgmanite-dominant P–T field

At pressures above 25 GPa, the mineralogy of B30 becomes Brg and CaPv (Fig. 3). 310 The diffraction intensities show that Brg is the major phase in the Brg+CaPv system 311 (Fig. 2). The (Mg+Fe+Ca)/Si ratio of B30 is higher than 1 (1.15), the excess MgO 312 and FeO can stabilize Fp together with Brg and CaPv, if the majority of Fe is Fe<sup>2+</sup>. 313 However, we could not identify Fp peaks in the diffraction patterns because of the ex-314 pected peak overlap with the Brg peaks. Fp was not observed in TEM analysis either. 315 The amount of Fp might have been too small to be detected because the dispropotion-316 ation reaction of Brg (Section 3.2; Frost et al., 2004) can reduce the available number 317 of Fe<sup>2+</sup> for Fp. Although the heating duration was adequately controlled in our ex-318 periments within 8 mins (Table S1), we do not rule out the possibility that the Soret 319 diffusion caused the Fe loss in the heating center (Sinmyo and Hirose, 2010). Some 320 amount of FeO or Fe metal might have migrated away from the heating center to the 32 low-temperature regions. The calculated phase proportions based on the measured 322 compositions by TEM-EDS (Table 2) yield only 1±1 vol% of Fp. For Brg and CaPv, 323 81.6 and 18.4 vol% were obtained by the mass-balance calculations, consistent well 324 with 82.2 and 17.8 vol% from Rietveld refinements within 1% of difference (Table 2). 325

As described in Section 3.1, the starting material was synthesized under a reducing condition. Even if  $Fe^{3+}/\Sigma Fe$  was larger in the synthesis products, the phase proportion of Brg would be still large because Brg is the only phase that can incorporate  $Fe^{3+}$  in the mineralogy in the Brg-dominant P-T field.

Modern MORB contains much lower content of Brg than B30 at P-T conditions of the lower-mantle. Modern MORB contains 30, 27, 22, and 21 vol% of CaPv, aluminous phase, Stv, and Brg, respectively at 31 GPa (Ricolleau et al., 2010). Pyrolite and B30 are much alike in that Brg is abundant (75, 18, and 7 vol% for Brg, Fp, and CaPv, respectively, at above 27 GPa for pyrolite; Irifune et al. (2010); Ishii et al. (2018)), but different in that B30 contains very small or no Fp in the mineralogy.

336 [Fig. 4]

#### 337 **4. Density profiles**

#### 338 4.1. Density profile of Archean volcanic crust

We estimated the density profiles of Archean volcanic crust (B30) along the low 339 and equilibrated temperature profiles in the Archean (hereafter referred to as AL and 340 AE, respectively). The AL temperature was obtained from the temperature model of 34 the subducting slab in Klein et al. (2017). For the AE temperature, we assumed the 342 same temperature as the surrounding mantle in the Archean, which was shifted from 343 the modern mantle geotherm (Brown and Shankland, 1981) by 200 K (Korenaga, 2008; 344 Herzberg et al., 2010). Fig. S2 compares the geotherm models with the adiabatic tem-345 perature gradients. 346

Our in-situ XRD experiments provided the direct measurements of the unit-cell 347 volumes of the stable phases at high P-T. We considered Gt, Rw, Brg, and CaPv at 348 20-34 GPa. For the Gt+Rw mineralogy, the chemical compositions and phase pro-349 portions of Gt and Rw were obtained from our STEM-EDS measurements, which were 350 assumed to remain unchanged within the same mineralogy field (Table 2). At pressures 351 above Brg-in boundary and below CaPv-in boundary, we took Fp into account the min-352 eralogy (Gt+Brg+Fp) and density calculations. The measured Rw composition was 353 used to model the compositions and proportions of Brg and Fp, which form through 354

the post-spinel transition. The unit-cell volumes of Fp were calculated by the equa-355 tion of state from Speziale et al. (2007). At pressures above CaPv-in boundary (for the 356 Gt+Brg+CaPv mineralogy), the compositions of Brg and CaPv were obtained from our STEM-EDS measurements and assumed to remain unchanged (Table 2). Their propor-358 tions were obtained by mass-balance calculations based on the gradual transformation 359 of Gt into Brg and CaPv. At pressures above Gt-out boundary (for the Brg+CaPv min-360 eralogy), the compositions and proportions of Brg and CaPv were assumed to remain 36 unchanged (Table 2). More information on the density profile model can be found in 362 Text S1. 363

We also calculated the density of pyrolite in the Archean for comparisons with the B30 density (See Section 4.3 for the method). Fig. 4 shows that the B30 density is greater than the pyrolite density in the Archean at most depths between 20–34 GPa (570 and 900 km). The pyrolite density shows a sharp increase at 23–23.5 GPa because of the post-spinel transition, followed by the gradual increase, contributed by the post-Gt transition that occurs in the larger pressure interval (22.5-28 GPa) at the elevated mantle temperature in the Archean.

The similar patterns of the phase transitions are observed in B30. The B30 density 37 along the AE temperature markedly increases over a pressure interval of 25-26 GPa 372 (Fig. 4), because of the phase transformation of Gt and Rw into Brg (Section 3.2). 373 The post-spinel transition in B30 at 23 GPa may not increase the density as much as 374 in pyrolite because the lower proportions of Rw (16 and 60 vol%, respectively; Irifune 375 et al. (2010); Ishii et al. (2018)) forms the limited amount of Brg. The B30 density 376 becomes slightly lower than the pyrolite density at 24-25 GPa. However, the Gt-out 377 boundary of the post-Gt transition is at shallower depth in B30 than in pyrolite (26 378 and 28 GPa, respectively) because of the narrower transition thickness (1.7 GPa of the 379 pressure interval) in B30. As a result, B30 becomes denser than pyrolite by  $0.1 \text{ g/cm}^3$ 380 above 25 GPa in the AE temperature (Fig. 6a). 38

In the AL temperature, the B30 density is always greater than the pyrolite density at 20–34 GPa (Fig. 4). The B30 density notably increases at 24–25 GPa because of the post-Gt transition with the narrow pressure interval (1.7 GPa). The density along the AL temperature is greater than along the AE temperature, because of the thermal expansion. On the other hand, the B30 density along the colder temperatures was not investigated in this study due to the limited number of data points at low temperatures. However, we speculate that the B30 density along the colder temperatures should be even greater than the densities in the AE and AL temperatures because the depths of phase transition into Brg will be shallower and the volumes of the phases will be smaller with lower temperatures.

392 [Fig. 5]

#### <sup>393</sup> 4.2. Comparisons of the B30 density between experiments and Perple\_X

To examine possible differences between experiments and theoretical calculations, 394 we used Perple\_X (Connolly, 2009) to calculate the densities of B30 for the AL and 395 AE temperatures and compared with our experiments. In Perple\_X, we used the Mie-Grüneisun formulation from Stixrude and Lithgow-Bertelloni (2005) and the thermo-397 dynamic data of mineral phases from Stixrude and Lithgow-Bertelloni (2011), which 398 has been widely used for multi-phase assemblages in the Earth's mantle conditions 399 (e.g., Nakagawa et al., 2010; Stixrude and Lithgow-Bertelloni, 2012; Klein et al., 2017). 400 The B30 density appears to be in a good agreement between the experiments and 40 the Perple\_X model except for 24.5–28 GPa, where the density from the experiments 402 becomes much greater (Fig. 4). The transition thickness of the post-Gt transition is 403 narrower in the experiments than in the Perple\_X model (1.7 and 5 GPa, respectively). 404 The B30 density above 28 GPa in the exeptiments is in a good agreement with that in 405 the Perple\_X model. 406

Brg-in boundary is consistent with each other within 1% (23.6 and 23.4 GPa, re-407 spectively), whereas the Gt-out boundary is considerably at higher pressure in the Per-408 ple\_X model than in our experiments (28.1 and 24.8 GPa at 1,700 K, respectively). The 409 discrepancy results in the greater density in the experiments than in the Perple\_X model 410 at 24.8-28 GPa (Fig. 4). However, the Clapeyron slopes of the post-spinel and post-Gt 41 transitions are in an excellent agreement between the experiments and Perple\_X cal-412 culations. The experiments yielded -0.8±0.3 and 3.5±0.5 MPa/K, whereas Perple\_X 413 predicted -0.81 MPa/K and 3.1 MPa/K for the post-spinel transition and the post-Gt 414 transition, respectively. 415

On the other hand, the predicted mineralogy by Perple\_X was somewhat different 416 from the experiments (Fig. 5). While the major phases, such as, Gt and Brg are the 417 same with different proportions between the experiments and the Perple\_X model, the 418 minor phases are additionally predicted from the Perple\_X model. For example, in the 419 AL temperature, the Perple\_X mineralogy model shows Stv up to 3 vol% together with 420 either Rw or Fp at 21–25 GPa, unlike our experiments (Fig. 5). In the AL and AE tem-421 peratures, the Perple\_X mineralogy model predicted 10 vol% of calcium-ferrite phase 422 above 26 GPa, which was not observed from the experiments. In addition, the Perple\_X 423 mineralogy model shows Fp above 24 GPa up to 5 vol%, which was not observed from 424 the experiments either. This is likely caused by the lack of consideration of  $Fe^{3+}$  in 425 Brg from the Perple\_X model, which ignores a reduced amount of Fe<sup>2+</sup> through the 426 charge disproportionation reaction (Section 3.2; Frost et al., 2004). Our stoichiometric 427 estimation of Brg from the experiments yields  $Fe^{3+}/\Sigma Fe=57\%$  (Section 3.2), which is 428 in a good agreement with the reported values (Frost et al., 2004; Shim et al., 2017). 429

#### 430 [Fig. 6]

#### 431 4.3. Density difference between the volcanic crust and pyrolite

Fig. 6a,b shows the density difference of B30 and modern MORB with respect to 432 pyrolite (the surrounding mantle) in the Archean and modern day, respectively. The 433 B30 density is directly from our experiments (Section 4.1). The densities of mod-434 ern MORB and pyrolite were calculated from Perple\_X (Section 4.2), which are in an 435 agreement with previous experiments (Irifune and Ringwood, 1993; Hirose, 2002; Irifune et al., 2010; Ishii et al., 2018). The low temperature profiles for modern MORB 437 (hereafter referred to as ML) were obtained from the temperature model of the subduct-438 ing model (Klein et al., 2017). The equilibrated temperature for modern MORB (ME) 439 was assumed to be same as the surrounding mantle (Brown and Shankland, 1981). 440

At 23–23.5 GPa, because of the post-spinel transition in pyrolite, the density difference between B30 and pyrolite drops by less than 0.1 g/cm<sup>3</sup>. The drop can be seen in modern MORB but with a larger magnitude, 0.2 g/cm<sup>3</sup>. The proportions of Rw in pyrolite is much smaller in the Archean than in the modern day (30 and 60 vol%, respectively; Fig.S1), therefore the effect of the post-spinel transition on the density

increase is smaller in the Archean. The post-Gt transition in B30 and modern MORB 446 increases the density difference, but occurs at the different depths. The increase in the 447 density difference occurs at the lower pressures of 24.5-26 GPa in B30 than at 25-29 GPa in modern MORB, depending on the temperature. If thermally equilibrated ллс with the surrounding mantle (i.e. AE or ME), the density difference of B30 becomes 450 negative only at 24–25 GPa (up to -0.03 g/cm<sup>3</sup>), whereas that of modern MORB does 451 at 23.5–29 GPa (up to  $-0.09 \text{ g/cm}^3$ ). At the deeper depths, B30 maintains the positive 452 density difference by 0.1–0.2 g/cm<sup>3</sup> at the shallow depth than modern MORB does (26 453 and 29 GPa, respectively). 454

#### 455 4.4. Density difference between subducting slabs and pyrolite

We calculated the density of the subducting slab by combining the densities of the crust and harzburgite. The thicknesses of crust and harzburgite are 7 and 32 km for the modern day and 40 and 74 km for the Archean. The detailed descriptions about the density calculations for the subducting slab can be found in Text S2. The density difference of the subducting slab with respect to pyrolite is shown as dashed lines in Fig. 6.

In the Archean, subducting slabs are equal to or denser than pyrolite at most pres-462 sures between 20–34 GPa (Fig. 6a). For the AL temperature, because of the post-spinel 463 transition in pyrolite, the density difference drops to  $-0.01 \text{ g/cm}^3$  at 23.3 GPa. For 464 the AE temperature, the drop in the density difference does not appear because of the 465 shallower depth of the post-spinel transition in Archean harzburgite at the higher slab temperature (Text S2). Then, the cold slab maintains the greater density than pyrolite 467 above 28 GPa, while the thermally equilibrated slab becomes equally dense as pyrolite 468 at 29 GPa. On the other hand, the colder slabs are expected to be even denser than the 469 AL and AE slab cases because of the reduced volume at lower temperatures, therefore 470 the greater density difference with respect to pyrolite. 47

In the modern mantle, the density difference for the ML temperature is always positive at 20–34 GPa (Fig. 6b). In contrast to the AL temperature, the ML temperature does not show the large decrease in density difference at 23.3 GPa although the postspinel transition occurs at deeper depth in harzburgite of the slab than in pyrolite due

to the negative sign of the Clapeyron slope (Fig. S3). In the calculated model, the 476 temperature contrast of the slab along ML temperature with respect to pyrolite is not 477 large enough (~300 K; Text S2) to generate the noticeable difference in the depth of the 478 post-spinel transition between the slab and pyrolite. For even colder slabs in the modern 479 day, the marked decrease in the density difference is expected because of the larger 480 temperature contrast with respect to pyrolite, which will delay the post-spinel transition 481 in the slabs. For the ME temperature, the density difference becomes negative in 27.5-482 29.6 GPa because of the persistent Gt in modern MORB at the higher slab temperature. 483 Above 30 GPa, the density difference for the ME slab becomes -0.01 g/cm<sup>3</sup>. 484

These results suggest that the contribution from the crustal density is much greater in the Archean than in modern day because of the greater thickness of the crust in the Archean slab (40 and 7 km-thick, respectively).

#### 488 5. Discussion

#### 5.1. The density behaviors of the volcanic crust and subducting slab in the Archean

We found that Archean volcanic crust (B30) is denser than the pyrolitic mantle at 490 the most depths of the lower MTZ and the shallow lower mantle regardless of the ther-491 mal state of the crust, unlike modern MORB (Fig. 6). The different density behaviors 492 of the crust between in the Archean and in the modern day can be attributed to (1) the 493 effect of the Gt composition on the depth of the post-Gt transition in the crust and (2) 494 the temperature effect on the mineralogy of the pyrolitic mantle. First, both Archean 495 volcanic crust and modern MORB have Gt as the most abundant phase (~80 vol%) in 496 the mineralogy. The composition of Gt differs a lot between the two crusts (Table S2), 497 which may affect the depth of the post-Gt transition. For example, the Gt composi-498 tion in Archean volcanic crust exhibits much smaller Al<sub>2</sub>O<sub>3</sub> content than in modern 499 MORB (4.8 and 13.2 mol%, respectively; Table S2). The previous experiments (Ir-500 ifune et al., 1996) suggested that the Al<sub>2</sub>O<sub>3</sub> content may increase the stability of Gt 50° at higher pressure, which is consistent with our observations that Gt-out boundary in 502 Archean volcanic crust is at the shallower depth by 25-50 km-depth (1–2 GPa; Fig.3) 503 than that in the modern day. Second, the mineralogy of the pyrolitic mantle may be 504

different depending on temperature. Pyrolite shows only 30 vol% of Rw in the hotter Archean, whereas 60 vol% of Rw in the cooler modern day (Fig. S1). The larger Rw proportions in modern-day pyrolite results in the greater increase in the pyrolite density at 660 km-depth through the post-spinel transition. Therefore, the impact on the decrease in the density difference of the crust with respect to the pyrolitic mantle might be much larger in the modern day than in the Archean (Fig. 6).

Our density models of the subducting slab suggest that the effects on the net density of the slab from the less dense harzburgite are smaller in the Archean than in the modern day (Fig. 6). Rather, the greater density of Archean volcanic crust is more effective to the net density of the slab owing to its larger portion in the slab (35 and 18% for Archean volcanic crust and modern MORB, respectively; Table 1). As a result, the density contrast between the subducting slab and the pyrolitic mantle is larger in the Archean than in the modern day (Fig. 6).

#### 518 5.2. Previous studies on candidates for the subducted Archean volcanic rocks

The exact composition of Archean volcanic crust remains uncertain (Herzberg 519 et al., 2010; Herzberg and Rudnick, 2012). Therefore, it is worthwhile to consider the 520 relevant compositions. The mineralogy and density of komatiite in the MTZ and the 52 topmost lower mantle have been documented by previous studies. Earlier experiments 522 by Nishihara and Takahashi (2001) studied Al-depleted komatiite with (Mg+Fe+Ca)/Si 523 = 1.38 up to 23 GPa. They observed Gt and Rw at 19 GPa, consistent with our experi-524 ments on B30 (Section 3.1). The Brg-in boundary is slightly shallower in Al-depleted 525 komatiite by 0.3 GPa than in B30, although their pressure calibration had a large uncer-526 tainty at high pressure (Nishihara and Takahashi, 2001). At 21 GPa, Al-depleted ko-527 matiite has a much larger proportion of Rw than B30 (40 and 14 vol%, repectively) be-528 cause of the higher (Mg+Fe+Ca)/Si ratio. At 23 GPa, the stable phases of Al-depleted 529 komatiite are Gt + Brg + CaPv + Fp, consistent with B30 except that we did not find 530 Fp (Section 3.2) because of the lower MgO content in B30 (32.1 and 22.8 wt%, respec-531 tively). They estimated the density of Al-depleted komatiite at high P-T through ther-532 modynamic calculations based on the calculated zero-pressure densities of the phases 533 (Brown and Shankland, 1981). Their results showed the greater density of Al-depleted 534

komatiite than PREM density by  $0.02-0.44 \text{ g/cm}^3$  in the upper mantle and the MTZ.

Klein et al. (2017) proposed a model with a komatiitic composition with (Mg+Fe+Ca)/Si 536 = 1.12 that has less MgO than the one used in Nishihara and Takahashi (2001), but 537 similar to B30 in the present study (Table 1). In the thermally equilibrated slab, Gt 538 takes up to 92 vol% with minor CaPv and Fp at 20-23 GPa, but without Rw. Klein 539 et al. (2017) predicted the greater density of komatiite than the pyrolitic mantle in the 540 Archean except when thermally equilibrated. With the equilibrated temperature, the 54 komatiite density becomes smaller than pyrolite at 24-26 GPa, whereas the B30 den-542 sity is always greater regardless of the slab temperature. 543

The general conclusions drawn from the previous studies on komatiite (Nishihara and Takahashi, 2001; Klein et al., 2017) concur with this study on B30 in that the greater density of Archean volcanic crust likely facilitates sinking of slabs into the lower mantle.

#### 548 5.3. The sinking style of Archean volcanic crust and the scale of the mantle mixing

Whether modern-style subduction occurred in the Archean is still under debate (e.g. 549 Van Hunen and Moyen, 2012; Tang et al., 2016; Condie, 2018). Other scenarios for 550 the material transportation into the deep interior in the Archean include the episodic-55 style subduction (Van Hunen and Moyen, 2012), delamination (Johnson et al., 2014) 552 and shallow subduction (Foley et al., 2003). Van Hunen and Moyen (2012) showed 553 that the high temperature of the mantle would have weakened the harzburgite layer 554 of subducting slabs in the Archean, which would lead to the slab breaking. The fre-555 quent break-off of the slabs could have resulted in the episodic-style subduction in the 556 Archean (Van Hunen and Moyen, 2012). The fragments of the slabs would lose some 557 of the driving forces of sinking, such as ridge-push and slab-pull, but we speculate that 558 the slab fragments would keep sinking because of the greater density of the slabs than 559 that of the surrounding mantle in the Archean. Johnson et al. (2014) suggested that 560 the base of Archean volcanic crust could have been gravitationally unstable due to the 561 local thickening, and eventually delaminated by Rayleigh-Taylor instabilities. The de-562 laminated crust would sink because of the greater density than that of the surrounding 563 mantle (Klein et al., 2017). In addition, Foley et al. (2003) proposed that Archean crust 564

would have been recycled to the upper mantle through shallow subduction in the early 565 Archean. The subducting slabs may become stagnant in the upper mantle (Foley et al., 566 2003). However, it is also possible that the stagnant slabs in the upper mantle break off 567 due to the hot mantle in the Archean (Van Hunen and Moyen, 2012) and sink into the 568 MTZ if the broken slabs are negatively buoyant in the upper mantle (Klein et al., 2017). 569 In any scenario, we speculate that in the Archean, the slabs or crusts would arrive to 570 the 670 km-discontinuity. Even though the slabs or crusts stagnate near the 670 km-57 discontinuity, they would eventually descend into the lower mantle by Rayleigh-Taylor 572 instabilities. 573

Once modern-style global subduction had initiated, the volume of the material exchange between the MTZ and lower mantle might have increased. If the subduction transported more materials into the lower mantle with modern-style global subduction, the whole-mantle mixing might have become more efficient. In this case, the mantle mixing with the redox heterogeneities from the lower mantle might have oxidized the upper mantle, eventually leading to the rise of  $O_2$  in the atmosphere toward the GOE (Nicklas et al., 2019; Aulbach and Stagno, 2016; Andrault et al., 2018a).

#### 581 5.4. The time-dependent changes in the scale of material circulations in the mantle

Since the onset of modern-style global subduction in the Archean (e.g., Laurent 582 et al., 2014; Tang et al., 2016; Condie, 2018), the composition of the subducted volcanic 583 crust have been changed with time. In this view, modern MORB and Archean volcanic 584 crust can represent the two end-members of the crustal compositions (Herzberg et al., 585 2010; Johnson et al., 2014). As Earth's mantle cools, the composition of the volcanic 586 crust evolves toward the lower (Mg+Fe+Ca)/Si and more Al, and so does the compo-587 sition of Gt. The change in the Gt composition might have deepen the depth of the 588 post-Gt transition with time. As a result, Gt becomes more stable at a deeper depth, 589 delaying the appearance of Brg, the high-density phase. Also, the thickness of the crust 590 in the slab becomes smaller, and therefore the density contribution from the crust be-59 comes smaller. At the same time, the contribution from the harzburgitic layer becomes 592 greater for the net density of the subducting slab, if slab remains mechanically inte-593 grated. As shown in Fig. 6, the subducting slab may have become less denser than the 594

<sup>595</sup> surrounding mantle over time.

Today, seismic tomography images show different states of the modern subducting 596 slabs right beneath the MTZ. Slab stagnation is observed beneath the Western Pacific, 59 whereas subducting slabs seem to directly penetrate into the lower mantle beneath the 598 Central and South American subduction zones (e.g., van der Hilst and Seno, 1993; 599 Fukao et al., 2009; Li and McNamara, 2013; King et al., 2015). As discussed above, the 600 change in the composition and thickness of the crust over time might have reduced the 60' positive density contrast of the subducting slabs, which might partly contribute to the 602 slab stagnation. What we observe from the seismic tomography may be a snapshot of 603 the time-dependent transition of the style of the material circulations between the MTZ 604 and the lower mantle. Therefore, the scale of the mantle mixing may have changed 605 from single-layered since Archean to the two-layered in the future (Klein et al., 2017). 606 [Fig. 7] 607

#### 5.5. Archean materials in the present-day mantle

The slow diffusion in the lower mantle (Holzapfel et al., 2005) may have allowed 609 Archean volcanic crust to survive until the present day. Such materials can contribute 610 to the seismic heterogeneities observed in the present-day mantle (Frost et al., 2017; 61 Wu et al., 2019). Particularly, a recent study raised a possibility of an accumulation of 612 ancient oceanic crust at 660 km-depth for the cause of small-scale seismic structures 613 (Wu et al., 2019). In order to evaluate the effect of density on the buoyancy of Archean 614 volcanic crust and Archean subducting slab in the pyrolitic modern mantle, we cal-615 culated the densities of B30 and Archean harzburgite. The temperature for the crust 616 and slab was assumed to be the same as the surrounding mantle, therefore thermally 617 equilibrated (Brown and Shankland, 1981). The B30 density was obtained based on 618 the experimental data (Fig. 3), while those of Archean harzburgite and modern pyrolite 619 were calculated using Perple\_X (Connolly, 2009) and the data set from Stixrude and 620 Lithgow-Bertelloni (2011). For Archean subducting slab, we assumed thicknesses of 62 40 and 74 km-thick for the crust and harzburgitic layer, respectively (Table 1). 622

We found that B30 remains denser than the pyrolitic mantle by  $0.1 \text{ g/cm}^3$  throughout at most depths between 580 and 900 km-depth (Fig. 7). If detached, the density

of Archean volcanic crust in the present-day mantle may prefer sinking into the lower 625 mantle to staying at 660 km-depth. In this case, the possible ancient oceanic crust at 626 660 km-depth that Wu et al. (2019) suggested is unlikely originated from Archean, but 62 more likely sometime close to the present day. For the case of the integrated slab, the 628 density of Archean slab appears to be the same as the modern mantle density at most 629 depths between 580 and 900 km-depth because the lower density of Archean harzbur-630 gite cancels out the greater density of B30 (Fig. 7). Therefore, the density effect on the 63 buoyancy of Archean slab in the modern mantle at these depths may be negligible. 632

#### <sup>633</sup> 5.6. Improving mineralogy modeling for Earth-like planets

From the stellar elemental abundance, we can infer the bulk chemical compositions 634 of the exoplanets around the stars (Bond et al., 2010; Young et al., 2014; Unterborn and 635 Panero, 2017). Such studies have found a surprisingly large range of Mg/Si ratios for 636 the compositions. For the rocky exoplanets, Mg and Si are often assumed to exclu-637 sively compose the mantle minerals. Therefore, in order to model the dynamics and 638 structures of the interior of the rocky exoplanets, it is important to have capability to 639 predict mineralogy from a range of different compositions. Because of the large param-640 eteral space and the paucity of experimental data for such a large range of Mg/Si ratio, 641 thermodynamics codes, such as Perple\_X, have become important tools to model the 642 mineralogy of the exoplanetary mantle (e.g. Dorn et al., 2015; Unterborn et al., 2017; 643 Vance et al., 2018). 644

Such thermodynamics codes have successfully reproduced the phase relations ob-645 served in experiments for Earth's mantle-related compositions, such as pyrolite and 646 MORB, as also demonstrated in our modeling. Besides these two compositions, there 647 are little data at sufficiently high pressure to examine the predicted models through 648 comparison with experimental data. While the Mg/Si ratio of MORB (0.23; Gale et al., 649 2013) is too low to be within the possible range of the Mg/Si ratio for Earth-like ex-650 oplanets (e.g. 0.8–2 based on the stellar elemental abundances; Young et al., 2014), 651 that of pyrolite (e.g. Mg/Si=1.25; McDonough and Sun, 1995) can be useful to infer 652 the mantle mineralogy where Mg out-numbers Si (Mg/Si>1). Our new data on the B30 653 composition presented in this study provide the mineralogy with the sufficiently low 654

<sup>655</sup> Mg/Si ratio (0.72), which is an important opportunity to test the modeling capability <sup>656</sup> for the large range of Mg/Si ratio for the possible mantle compositions.

Section 4.2 compares the mineralogy and the density between our experiments and 65 Perple\_X model. While the density remains similar to each other at 20-34 GPa (Fig. 4), 658 the minor phases were additionally found in Perple\_X model, partly because of the lack 659 of consideration of Fe<sup>3+</sup> in the data set (Stixrude and Lithgow-Bertelloni, 2011). This 660 difference might be the reason for the different chemical compositions of the major 66' phase, such as Gt, which may in turn result in the different depths of the phase tran-662 sition (24.8 and 28.1 GPa for the Gt-out boundary in our experiments and Perple\_X 663 model, respectively). However, because the density and the major phase are similar be-664 tween the experiments and Perple\_X model for the other pressure ranges (20-24 GPa 665 and 28-34 GPa), we speculate these similarity may continue for the shallower mantle 666 and the deeper mantle. Gt is known to be stable throughout the upper mantle and the 667 MTZ (Irifune et al., 1996), while the Rw composition can exist in other forms of poly-668 morphs, such as olivine and wadsleyite (Nishihara and Takahashi, 2001). In the lower 669 mantle, Brg and CaPv should persist up to ~100 GPa because of the large stability of 670 the perovskite structure (Irifune et al., 2010). Therefore, in spite of some differences in 67 the mineralogy and phase transition behaviors at 25–28 GPa, the use of Perple\_X with 672 the data set from Stixrude and Lithgow-Bertelloni (2011) can be effective for modelling 673 the mantle mineralogy and density of Earth-like planets with the low Mg/Si ratio. 674

#### 675 6. Conclusion

We have conducted experiments to investigate stable phases and density of Archean 676 volcanic crust in the pyrolitic mantle transition zone and lower mantle. Our experi-677 ments showed that Archean volcanic crust is denser than the pyrolitic mantle regard-678 less of temperature, increasing the likelihood of slab penetration across the lower MTZ 679 where modern MORB may experience the slab stagnation beneath some subduction 680 zones. We also modelled the density of the subducting slab in a combination of the 68' densities of the crust and the harzburgitic layer, with respect to the pyrolitic mantle 682 for both the Archean and modern day. The models predicted that the positive density 683

contrast between the subducting slab and the pyrolitic mantle is larger in the Archean
than in the modern day. Our results suggest that the density and the large thickness of
Archean volcanic crust would have promoted the further sinking into the lower mantle
in any tectonic styles. Therefore, the mantle mixing between the MTZ and the lower
mantle might have been efficient in the Archean.

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Figure 1: A high-angle annular dark-field image in the different region with chemical maps of B30 in LH-DAC. The sample was synthesized at 19.5 GPa and 1,900 K. The brighter color in each chemical map represents the more concentration of the corresponding element. The chemical compositions and the proportions of individual phases are provided in Table 2. Gt: garnet; Rw: ringwoodite.



Figure 2: In-situ X-ray diffraction (XRD) patterns of B30 at high P-T in laser-heated diamond anvil cell (LHDAC). X-ray energy was 37 keV. We provided the Miller indices of the major lines. Gt: garnet; Rw: ringwoodite; CaPv: CaSiO<sub>3</sub> perovskite; Brg: bridgmanite; Au: gold (pressure standard); Ar: argon (medium); Ne: Neon (medium); and \*: unidentified peaks.



Figure 3: Phase changes in the B30 composition at high pressure and temperature. The light gray line represents the Archean mantle geotherm, which was shifted by 200 K from the modern mantle geotherm (the dashed gray line; Brown and Shankland, 1981). The gray line represent the temperature profile of the subducting slab after Klein et al. (2017). Top: Each symbol represents one heating cycle in the LHDAC experiments. The dashed and dot-dashed lines indicate the post-spinel and post-Gt transition boundaries determined in this study, respectively. Bottom: The major phases in the distinctive mineralogy fields. The black solid lines are Brg-in and Gt-out boundaries determined in this study. The Brg-in log of the post-spinel transition in pyrolite (Hirose, 2002) is plotted as the dashed line. The Brg-in (left) and Gt-out (right) boundaries in modern MORB (Hirose et al., 1999) are plotted as the dot-dashed lines. The mantle solidus is obtained from Andrault et al. (2018b). Gt: garnet; Rw: ringwoodite; CaPv: calcium perovskite; Brg: bridgmanite.



Figure 4: Density of B30 along different temperature profiles in the Archean. The orange, and red curves represent low (AL), and equilibrated (AE) temperature profiles in the Archean, respectively. The solid and dashed lines indicate densities determined by experiments and by Perple\_X (PX; Connolly, 2009), respectively. The circles along the solid lines represent the data points obtained from our experiments. Information on the temperature profiles can be found in Table S3.



Figure 5: The mineralogy for pyrolite, B30, MORB in vol% in the MTZ and the topmost lower mantle. The mineralogy of B30 is from the experimental results in this study. Mineralogies of pyrolite, MORB, and B30\_PX were calculated using Perple\_X (Connolly, 2009) and the data set from Stixrude and Lithgow-Bertelloni (2011). Gt: garnet; Rw: ringwoodite; Brg: bridgmanite; CaPv: CaSiO<sub>3</sub> perovskite; Fp: ferropericlase; St: stishovite; CF: calcium ferrite; Aki: Akimotoite.



Figure 6: Density differences of the crusts and subducting slabs with respect to the pyrolitic mantle: (a) B30 and subducting slab in the Archean and (b) MORB and subducting slab in the modern day. The open circles in (a) represent the experimental data points from this study. The vertical gray bars indicate the depth of the post-spinel transition in the pyrolitic mantle. The depths of the negative density contrast are shaded. For subducting slabs, the thicknesses of crust and harzburgite are 40 and 74 km for the Archean, and 7 and 32 km for the modern day, respectively (Text S2). Information on the temperature profiles can be found in Table S3.



Figure 7: The density differences of B30 and Archean subducting slab with respect to the pyrolitic mantle in the modern day. The open circles represent our experimental data. The temperatures of B30 and Archean harzburgite were assumed to be equilibrated to modern mantle (Brown and Shankland, 1981)). The depth of the post-spinel transition in pyrolite is marked as the vertical grey bar. The densities of Archean harzburgite and pyrolite were calculated using Perple\_X (Connolly, 2009) and the data set from Stixrude and Lithgow-Bertelloni (2011).

wt%	Pyrolite <sup>a</sup>	B30 <sup>b</sup>	Modern	Archean	Modern	Archean
			MORB <sup>c</sup>	residue <sup>b</sup>	residue <sup>d</sup>	crust <sup>e</sup>
SiO <sub>2</sub>	45.1	47.0	51.8	43.8	43.4	47.6
$Al_2O_3$	4.8	7.5	15.5	2.8	1.3	7.1
FeO <sup>†</sup>	8.4	12.7	10.1	6.4	8.6	12.3
MgO	37.7	22.8	8.0	46.3	45.7	22.9
CaO	3.5	8.8	11.7	0.8	0.9	8.3
Na <sub>2</sub> O	0.4	1.3	2.1	0	0	1.7
Mg# <sup>‡</sup>	88.9	76.2	58.5	92.8	90.4	76.8
(Mg+Fe+Ca)/Si	1.49	1.15	0.63	1.72	1.76	1.12
Thickness (km)§	-	40	7	74	32	-

Table 1: Chemical compositions of pyrolite, basaltic crust, complementary residues, and Archean volcanic crust. After excluding minor elements, we normalized each composition to 100%.

<sup>*a*</sup> Composition obtained from McDonough and Sun (1995)

<sup>b</sup> Modeled compositions obtained from Johnson et al. (2014). B30 was used for experiments in this study.

<sup>*c*</sup> Composition obtained from Gale et al. (2013)

<sup>d</sup> Composition obtained from Baker and Beckett (1999)

<sup>*e*</sup> Oceanic plateau basalt 3.3 Ga; Schulz (1982)

† All Fe in B30 is assumed to exist as FeO.

Mg = Mg/(Mg + Fe)

§ Thickness of a layer in the subducting slab considered in this study.

	19.5 GPa <sup>†</sup> , 1,900 K		31 GPa	a, 2,000 K
wt%	Gt	Rw	Brg	CaPv
SiO <sub>2</sub>	49.7	41.4	47.9	48.1
$Al_2O_3$	8.6	2.7	9.0	1.1
FeO <sup>a</sup>	11.8	23.8	13.1	4.6
MgO	20.9	29.2	26.3	3.4
CaO	9.0	2.8	3.7	42.8
Proportions (vol%)				
Mass balance	82.7	17.3	81.6	18.4
EDS mapping	85.3	14.7		
Rietveld	85.5	14.5	82.2	17.8
Average <sup>b</sup> (vol%)	84.5	15.5	81.9	18.1

Table 2: Chemical compositions of the recovered samples from the Gt-dominant P-T field and the Brg-dominant P-T field. The compositions were normalized to be 100% in total.

<sup>*a*</sup>All Fe is assumed to be Fe<sup>2+</sup>. <sup>*b*</sup>Averaged proportions from the two different methods. <sup>†</sup>Pressure was measured at 300 K.

## Supplementary material for "Mineralogy and Density of Archean Volcanic Crust in the Mantle Transition Zone"

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#### Contents of this file

- 1. Text S1 to S2  $\,$
- 2. Figures S1 to S3
- 3. Tables S1 to S3

#### Introduction

This supplementary material provides the texts, figures and tables to support the main article. The texts includes the further explanations for the density models used in this study and the descriptions of the harzburgite density. The figures include the phase diagrams, the plot of temperature gradients, and the harzburgite density. The tables provide details of experimental conditions and results.

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#### <sup>964</sup> Text S1. The density model for the Gt-Brg transition P-T field

In our model for density calculation in Gt-Brg transition P-T field, we as-965 sume that Rw transforms univariantly to Brg + Fp at the Brg-in boundary, and 966 that Gt transforms gradually to Brg + CaPv over a pressure interval. We as-967 sume that Fp exists together with Gt and Brg although we did not observe Fp in 968 XRD patterns. The phase proportions for the Gt+Brg+Fp mineralogy were de-969 termined by converting Rw proportion to Brg and Fp, whose compositions were 970 passed down from the measured Rw composition (run #4430; Table 2). We cal-971 culated the molar volume of Fp using the equation of state of  $Mg_{0.83}Fe_{0.17}$  from 972 Speziale et al. (2007). For the Gt+Brg+CaPv mineralogy, the gradual trans-973 formation of Gt to Brg and CaPv was considered along the slab temperatures 974 between CaPv-in boundary and Gt-out boundary. At CaPv-in boundary, the 975 number of mols of Gt is  $X_{Gt}=1$ , which decreases linearly by breaking down into 976 Brg and CaPv until Gt is all consumed  $(X_{Gt}=0)$  at Gt-out boundary. With the 977 fixed chemical composition of Gt (Table 2), we estimated the number of mols 978 of Brg  $(X_{Brg})$  and CaPv  $(X_{CaPv})$  formed by the breakdown of Gt  $(1 \mod_{Gt} =$ 979  $3.3 \operatorname{mols}_{Brg} + 0.7 \operatorname{mols}_{CaPv}$ ). On the other hand, we assumed that Fp does not 980 exist in the Gt+Brg+CaPv mineralogy because of the disproportionation reac-981 tion of Brg (Frost et al., 2004). In this way, we predicted the phase proportions 982 at the Gt-out boundary to be 85.7, and 14.3 mol% for Brg and CaPv, respec-983 tively, consistent well with the phase proportions determined by mass balance 984 and Rietveld refinement for the Brg-dominant mineralogy (run #218; Table 2). 985

# Text S2. Density difference of harzburgite in the Archean and modern day

Ringwood (1982) proposed a model for the petrological structure of the 988 oceanic lithosphere that is composed mainly of the basaltic crust, harzburgitic, 989 and lherzolitic (depleted pyrolite) lithospheres. The density of the lherzolitic 990 lithosphere should be greater than or at least the same as the surrounding 991 mantle because of the same chemical composition but in the lower temperature 992 or the equilibrated temperature, respectively. Therefore, for the case of the 993 subducting slab the chemical effects on the density of the subducting slab is 994 primarily governed by the harzburgitic lithosphere as well as the crust (Klein 995 et al., 2017; Irifune and Ringwood, 1987). 996

When the volcanic crust forms from partial melting of the mantle, it will create complementary residue (harzburgite) beneath the crustal layer. Therefore, the crust and complementary harzburgite have compositional relations with respect to the undepleted mantle (pyrolite). Such relations allow us to estimate the thickness of harzburgitic layer. If we assume their volumes are linearly proportional to their thicknesses, we can describe an equation as follows (Xu et al., 2008):

$$H_{Crust}X_{Crust} + H_{Harz}X_{Harz} = (H_{Crust} + H_{Harz})X_{Pyro}$$
(S1)

where  $X_{Crust}$ ,  $X_{Harz}$ , and  $X_{Pyro}$  are the compositions of the crust, complementary harzburgite, and pyrolite, respectively and  $H_{Crust}$  and  $H_{Harz}$  are the thicknesses of the crust and complementary harzburgite, respectively. With the given thickness of the crust and the compositions of the crust and complementary harzburgite, we can determine the thickness of the complementary harzburgite using the following equation:

$$H_{Harz} = H_{Crust} \frac{X_{Pyro} - X_{Crust}}{X_{Harz} - X_{Pyro}}$$
(S2)

In our study, we used thicknesses of 7 and 40 km for the modern and Archean crusts (i.e. modern MORB and B30), respectively, which yield thicknesses of

32 and 74 km for the modern and Archean complementary harzburgites, re-1012 spectively (Table 1). We used Perple\_X to calculate the density of Archean 1013 harzburgite. Although Archean harzburgite has not been cross-examined by 1014 both the experiments and Perple\_X, unlike the modern harzburgite (Irifune and 1015 Ringwood, 1987; Stixrude and Lithgow-Bertelloni, 2012; Xu et al., 2008), its 1016 composition remains very similar to modern harzburgite composition (Table 1). 1017 Therefore, Archean harzburgite is expected to yield nearly the same mineralogy 1018 as modern harzburgite (Fig. S1). 1019

The density difference between harzburgite and pyrolite in the Archean and 1020 modern day is shown in Fig. S3. The calculated mineralogy of harzburgite is 1021 nearly the same in between the Archean and modern day (Fig. S1) because 1022 of the similar chemical compositions (Table 1). At 20-23.5 GPa, Rw is dom-1023 inant (80–90 vol%), while Gt represents 10–19 vol% in harzburgite, depending 1024 on temperature. Above 23.5 GPa, Brg becomes the most abundant phase and 1025 its proportion increases by the post-Gt transition (70-75 vol%) up to 25.9 and 1026 27.5 GPa, for both Archean and modern harzburgites, respectively (Fig. S1). 1027

The density difference of Archean harzburgite with respect to pyrolite changes 1028 with the temperature profile. The AL and AE temperatures have lower den-1029 sities than pyrolite (Fig. S3) except for 23–28 GPa. For the AL temperature, 1030 the density difference decreases at  $\sim 23.3$  GPa due to the post-spinel transition 1031 in pyrolite, followed by a sharp increase at  $\sim 24 \text{ GPa}$  because of the post-spinel 1032 transition in Archean harzburgite. For the AE temperature, the density differ-1033 ence abrubtly increases at 23 GPa, because of the post-spinel transition which 1034 occurs at the same depth in Archean harzburgite and pyrolite. Archean harzbur-1035 gite forms the larger amount of Brg at 23 GPa than pyrolite (53 and 23.54 vol%, 1036 respectively) because of the larger amount of Rw (76 and 34 vol%, respectively), 1037 leading the steep increase without the sudden drop in density difference. 1038

Above 24 GPa, Archean harzburgite shows a gradual decrease in the density difference because of the gradual post-Gt transition in pyrolite until 28.5 GPa. Above 28.5 GPa, Archean harzburgite becomes less dense than pyrolite up to 34 GPa for the AL and AE temperatures, respectively.

Compared to Archean harzburgite, modern harzburgite shows very similar 1043 trend in the density difference, but with the systemically greater density. The 1044 greater density of modern harzburgite may largely be the lower temperature in 1045 the modern day than in the Archean. At 23.8 GPa, both the ML and ME tem-1046 peratures show the marginal decrease  $(0.01-0.02 \text{ g/cm}^3)$  in the density difference 1047 of modern harzburgite. Although the transition depth of modern harzburgite 1048 becomes deeper with the lower temperature (ML) because of the negative sign 1049 of the Clapeyron slope of the post-spinel transition, the difference in the depth 1050 of the post-spinel transition between the slab and pyrolite is relatively small. 1051 This is because the temperature contrast between the slab along the ML tem-1052 perature and pyrolite is smaller in the modern day than in the Archean ( $\sim 300$ 1053 and  $\sim 400 \,\mathrm{K}$ , respectively). Therefore, the difference in the depth of the post-1054 spinel transition with respect to pyrolite becomes smaller in the modern day. 1055 Modern harzburgite along the ML temperature does not show the sudden drop 1056 of the density difference at 23.8 GPa, unlike Archean harzburgite with the AL 1057 temperature. 1058

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Figure S1: Modal proportions (vol%) of (clockwise, from top left) modern MORB, B30, pyrolite in the Archean, Archean harzburgitic lithosphere, modern harzburgitic lithosphere, and pyrolite in the modern day. Phase equilibria were calculated using Perple\_X (Connolly, 2009) with thermodynamic data (after Stixrude and Lithgow-Bertelloni, 2011) except for B30 (Archean crust). For B30, experimental data in the present study were used at 20–35 GPa and extrapolated based on Perple\_X model for 10–20 GPa. We used the temperature model of the subducting slab from Klein et al. (2017). Modern (Brown and Shankland, 1981) and Archean geotherms (elevated by 200 K with respect to modern geotherm) were used for pyrolite, respectively.



Figure S2: Temperature gradients of subducting slabs (blue) and pyrolite (red), and mantle adiabatic gradients (black). Solid and dashed lines represent the modern day and Archean. Temperature gradients of subducting slabs are from Klein et al. (2017). 0.4 K/km is used to plot the mantle adiabats.



Figure S3: The density differences between harzburgite and the pyrolitic surrounding mantle in (a) the Archean and (b) the modern day. The densities were calculated using Perple\_X (Connolly, 2009) with the thermodynamic data set (Stixrude and Lithgow-Bertelloni, 2011). The depths of the post-spinel transition in pyrolite is marked by the vertical grey bars. Information on the temperature profiles can be found in Table S3.

Sample#	Bun#	Modium	Dur.	Au V	$\mathbf{P}^{a}$	$P_{-}err$	Т	T_err	Phase
Sample#	nun#	meulum	(min)	$(Å^3)$	(GPa)	(GPa)	(K)	(K)	assemblage
218	7	Ne	4.5	61.66	33.62	0.75	2356	100	Brg+CaPv
	13	Ne	5	61.91	32.54	0.75	2382	100	Brg+CaPv
	15	Ne	5	61.65	31.06	0.75	2012	100	Brg+CaPv
812	107	Ar	3.5	62.26	29.13	0.75	2160	100	Brg+CaPv
	76	Ar	2.5	62.51	24.67	0.76	1719	100	Brg+CaPv
	122	Ar	5	61.88	28.06	0.75	1766	100	Brg+CaPv
	102	Ar	3	62.18	28.87	1.09	2072	145	Brg+CaPv
914	96	Ar		62.74	25.98	0.76	2040	100	Brg+CaPv
	114	Ar	4	62.64	25.68	0.76	1938	100	Brg+CaPv
	16	Ar	2.5	65.07	13.09	0.78	1592	100	Gt+Rw
1415	54	Ar	3	63.61	21.20	0.76	1906	100	Gt+Rw
	137	Ar	3	63.15	21.31	0.76	1663	100	Gt+Rw
	38	$\mathbf{Ar}$	3.5	62.70	22.64	0.76	1573	100	$Gt+Rw+CaPv+(Brg)^{c}$
1514	93	$\operatorname{Ar}$	3	63.11	23.79	1.64	1966	215	Gt+Brg+CaPv
	44	$\operatorname{Ar}$	3.5	62.68	25.72	0.76	1966	100	Gt+Brg+CaPv
1615	82	$\mathbf{Ar}$	2	63.07	24.25	0.76	2004	100	Gt+Brg
	86	$\mathbf{Ar}$	5	62.89	23.44	0.76	1791	100	Gt+Brg
	97	$\operatorname{Ar}$	4	62.61	24.89	0.76	1816	100	$Gt+Brg+(CaPv)^{c}$
	41	Ar	6	63.05	24.06	0.76	1966	100	Gt+Brg+CaPv
	$\overline{46}$	Ar	Ă	63.15	24.92	0.76	2140	100	Gt+Brg+CaPv
	$\overline{63}$	Ar	5.5	62.82	23.66	0.76	1777	100	Gt+Brg+CaPv
	$\tilde{76}$	Ār	6.5	62.78	24.69	0.76	1889	100	Gt+Brg+CaPv
	51	$\operatorname{Ar}$	3	62.99	23.18	0.76	1820	100	Gt+Brg+CaPv
1912	$\tilde{28}$	NaCl	2 sec.	60.70	34.16	0.74	1737	100	Brg+CaPv
	<u>9</u> 1	NaCl	2 sec.	61.21	33.28	0.75	2003	100	Brg+CaPv
2115	127	Ar	2	61.99	29.40	0.75	$\bar{2}014$	100	Brg+CaPv
2715	167	$\operatorname{Ar}$	1.5	62.94	26.65	0.76	2247	100	Gt+Brg+CaPv
	189	Ār	3	62.72	$\bar{25.08}$	0.76	1904	100	Gt+Brg+CaPv
	196	$\operatorname{Ar}$	$\overline{5}$	62.54	25.88	0.76	1899	100	Gt+Brg+CaPv
	20	$\operatorname{Ar}$	$\overline{5}$	63.82	18.90	0.77	1718	100	Gt + Rw
	81	$\operatorname{Ar}$	4	63.21	22.56	0.76	1834	100	Gt+Rw
	86	$\mathbf{Ar}$	1	63.56	20.75	0.76	1822	100	Gt+Rw
	89	$\mathbf{Ar}$	2	63.92	20.24	0.83	1947	108	Gt+Rw
	95	$\mathbf{Ar}$	1.5	63.36	18.98	0.76	1476	100	Gt+Rw
	99	$\operatorname{Ar}$	2	62.91	21.09	0.76	1496	100	Gt+Rw
	114	Ar	3.5	63.38	20.92	0.76	1746	100	Gt+Rw
	141	Ar	2	63.50	20.04	0.76	1695	100	Gt+Rw
4530	98	KCl	4	63.38	23.77	1.25	2120	164	$\operatorname{Gt+Brg}$
	91	KCl	5.5	62.81	25.83	1.31	2058	173	Gt+Brg+CaPv
	119	KCl	5	62.59	24.24	1.06	1715	140	Gt+Brg+CaPv
	25	KCl	5	62.83	23.18	0.83	1724	110	Gt+Brg+CaPv
	38	KCl	7.5	63.14	24.89	1.22	2126	161	Gt+Brg+CaPv
$4430^{b}$	-	Ar	NA	62.02	19.52	NA	1900	150	$\operatorname{Gt+Rw}$
a D	1	1	1 4	1 /1			1.D	1	0007) 1 1 1

Table S1: Run table for in situ experiments of B30.

	$20.5{ m GPa}$ 2,073 K	$23{ m GPa}$ 1,900 K	$24{ m GPa}2,023{ m K}$
Wt%	$Pyrolite^{a}$	$B30^{b}$	$MORB^{c}$
$SiO_2$	50.8	49.7	40.7
$TiO_2$	0	0	0.7
$Al_2O_3$	10.6	8.6	20.6
FeO	5.0	11.8	14.4
MnO	0	0	0.2
MgO	28.0	20.9	12.3
CaO	4.6	9.0	8.1
$Na_2O$	0.3	NA	2
$K_2O$	0.2	0	0
$Cr_2O_3$	0.7	0	0.1
Total	100.1	99.3	99.1

Table S2: Garnet compositions of pyrolite, B30, and modern MORB.

 $\frac{a}{a} \text{ Hirose (2002)}$   $\frac{b}{c} \text{ This study}$   $\frac{b}{c} \text{ Hirose and Fei (2002)}$ 

Mineralogy	Archean	Modern
Crust	EXP	PX
Harzburgite	$\mathbf{PX}$	$\mathbf{PX}$
Pyrolite	$\mathbf{PX}$	$\mathbf{PX}$
Temperature profile	Archean	Modern
Low	$K17^{a}$	$K17^{a}$
Equilibrated	$B81 + 200  K^c$	$B81^b$

Table S3: Methods used for the density calculation in this study. Exp: from our experiments; PX: from Perple\_X calculation.

<sup>a</sup> Klein et al. (2017) <sup>b</sup> Brown and Shankland (1981) <sup>c</sup> 200 K added to the modern mantle geotherm.