

UC Davis

UC Davis Previously Published Works

Title

The application of metabolomics to ascertain the significance of prolonged maturation in the production of lager-style beers

Permalink

<https://escholarship.org/uc/item/9kn0d8hp>

Journal

Journal of the Institute of Brewing, 125(2)

ISSN

0046-9750

Authors

Metrulas, Laura K
McNeil, Christopher
Slupsky, Carolyn M
[et al.](#)

Publication Date

2019

DOI

10.1002/jib.557

Peer reviewed

1 **The application of metabolomics to ascertain the significance of prolonged maturation in**
2 **the production of lager-style beers.**

3
4 Laura K. Metrulas, Christopher McNeil, Carolyn M. Slupsky and Charles W. Bamforth*

5
6 Department of Food Science & Technology, University of California, Davis,
7 CA, 95616-8598, USA

8
9
10 *Correspondence to: Charles W. Bamforth, Department of Food Science &
11 Technology, University of California, Davis, CA 95616-8598, USA.

12 E-mail: cwbamforth@ucdavis.edu

13
14 **Abstract**

15
16 NMR focused metabolomic analysis has been employed to ascertain the extent to which a
17 diversity of non-volatile substances change in level during maturation and storage on a pilot and
18 commercial scale. No substantive changes were observed, leading to the conclusion that once
19 materials such as vicinal diketones and acetaldehyde have been dealt with, there is no merit in
20 prolonged storage of beer.

21
22 **Key words:** Beer, lagering, metabolomics, non-volatile substances, NMR, storage

23
24
25 **Introduction**

26
27 There is a widespread acceptance of the dogma that certain beers, notably lagers, require a
28 degree of storage to effect their maturation (1). This practice presumably originated from the
29 days when beers were perforce held through the summer months (lagered) as the brewing of
30 new batches of beer was forbidden (2).

31
32 It is helpful to divide maturation into two separate requirements: the physical stabilization of beer
33 and the refinement of flavor.

34
35 In terms of the former, which is frequently referred to as colloidal stabilization, we have long
36 since arrived at a situation wherein there is clearly no necessity for prolonged treatment times.
37 The availability of palliative treatments to remove haze-forming polypeptides, polyphenols,
38 polysaccharides etc., together with fining agents, centrifuges, filters as well as a recognition that
39 it is the lowness of the temperature that is more important than lengthy holding time that is the
40 more relevant in cold conditioning all render extended processing irrelevant (3 ,4).

41
42 It is the matter of flavor maturation that remains controversial, with polar opinions ranging from
43 those who insist that it is essential that lager-style beers have a prolonged aging period in the
44 cellar through to those adamant that such periods can be very brief, especially as another of the

45 original purposes of lagering was an increase in carbonation, which of course can nowadays be
46 effected in short order (5).

47
48 Two of the key volatile substances that historically were removed in lagering are the vicinal
49 diketones (VDK, 6) and acetaldehyde (7). However, the scientific understanding of the origins
50 and control of these substances is now thoroughly appreciated. VDK can be dealt with
51 effectively by careful attention to primary fermentation conditions and even for those insistent
52 that more needs to be done, there is a range of options to accelerate the removal of these
53 molecules (8). Effective removal of acetaldehyde is even more straightforward.

54
55 The question is begged, then, if there are any other chemical entities that change in their levels,
56 either increase or decrease, in maturation, thereby benefitting the flavor of beer. The only paper
57 that the authors have located that dwells on this issue was by Masschelein (9). In it, he claims
58 that amino acids, peptides, nucleotides and organic acids as well as inorganic phosphates are
59 released by yeast when left in contact with the beer and he indicates that this is a desirable
60 occurrence.

61
62 Here we have applied the tool of metabolomics to investigate whether any significant changes
63 occur during the ageing of a lager, beyond the matter of removing vicinal diketones and
64 acetaldehyde.

65

66 **Materials and Methods**

67

68 **Brewing and sampling aged beer**

69

70 The experimental beer was brewed on a 176 L automated system. Salts (5 g magnesium
71 sulfate, 5 g Calcium sulfate, and 20 g calcium chloride) were added to 98.4 L of 57°C strike
72 water in the mash tank. Thirty-five kg of pilsner malt was milled on a two-roll mill and added to
73 the mash tank. The mash was then topped off with 6.8 L of water. Mechanical agitation was
74 used during mash heating. After the grist was added, the mash was held at 55°C for 10
75 minutes, it was then raised to 60°C for 10 minutes, 65°C for 30 minutes, and 76°C for 10
76 minutes. The mash was then transferred to the lauter tun, where it was vorlaufed for 10 minutes.
77 The wort was then transferred to the kettle. Sparging was with water (74.3°C) for 74 minutes
78 until kettle full (212 L) was achieved. Once the wort level in the kettle reached 45 L, the lower
79 internal calandria was initiated, followed by the upper calandria when the wort level reached 113
80 L. The wort was boiled for 90 minutes. Magnum hop pellets (70 g) were added 30 minutes
81 after the start of the boil. This was followed by an addition of 340 g Kazbec hop pellets after 60
82 minutes. After 85 minutes, 17 g Protofloc and 15 g Yeast nutrients were added. Kazbec hop
83 pellets (100 g) were placed in the whirlpool prior to the transfer of wort. After the boil concluded,
84 177.5 L of wort was transferred to the whirlpool. The wort was then cooled through a heat
85 exchanger to 20.1°C and transferred to a cylindroconical fermenter. The beer was pitched with
86 BSI Czech Lager Yeast at a rate of 2×10^6 cells/mL/°Plato. The fermentation was carried out at
87 10°C. It took 6 days to reach 6°Plato at which point the beer was allowed to free rise up to 15°C
88 for a vicinal-diketone rest. After 11 days, the beer was cold crashed for three days, which

89 brought the temperature down to 1.4°C. At this point, the beer was transferred into six 18 L
90 kegs. Three of the kegs were then filtered into clean 18 L kegs using two polysponge cartridge
91 type filters in series. The first filter was a 3-micron super high efficiency 1D by BevBright
92 followed by an absolute rated 10" sterile 0.45 µm BevBright filter. The kegs of the three filtered
93 beers were stored at -1°C for one month. This filtered beer was considered finished beer.
94 Carbon dioxide was used to pressurize the filtered beer for the purpose of forced carbonation as
95 well as for sample acquisition. The three unfiltered beers were conditioned at 2°C for one
96 month. During the conditioning stage, forced carbonation was not desired; so nitrogen was
97 used to pressurize the tank in order to retrieve samples from the keg. Samples were taken
98 daily for the first week and then once a week for the following three weeks. All samples were
99 immediately placed in a -20°C freezer and stored there until sample preparation for NMR
100 analysis. Metabolites in frozen samples are assumed to be stable at -20°C for short term
101 storage and limited freeze-thaw cycles (10).

102
103 A commercial pilsner-style lager (4.9% ABV; starting yeast count 1.5×10^6 cells/mL/°Plato) was
104 stored at 13.7°C for the first 14 days and then was lowered to -1.6°C during the following 16
105 days in tanks that were 8.2 m high, 2.9 m wide and with a cone angle of 70°. Samples were
106 collected at nine time points during the 30 days of maturation.

107 108 **Sample handling and NMR spectroscopy**

109
110 Frozen samples were de-frosted at room temperature. Once liquid, the samples were placed in
111 Amicon Ultra-0.5 Centrifugal Filter Units with Ultracel-3 membranes, which were previously
112 cleaned with deionized water. An internal standard containing 5 mmol/L of DSS-*d*₆ (3-
113 (trimethylsilyl)-1-propanesulfonic acid-*d*₆) and 0.2% NaN₃ (to prevent bacterial growth) in 99.8%
114 D₂O (for instrument locking) was added to each sample in a ratio of 1:10. Following this step,
115 the samples were adjusted to a pH of 6.8± 0.1, and NMR spectra were acquired using a Bruker
116 Avance 600 MHz NMR spectrometer following the method laid out by Slupsky et al. (11).
117 Metabolites were assigned using Chenomx NMRSuite Profiler v8.31 as described elsewhere
118 (12). The compounds found in the Chenomx library have been verified against known
119 concentrations of pure compounds and are shown to produce accurate and reproducible results
120 (11, 12).

121 122 **Statistical Analysis**

123
124 Statistical analysis of variance (ANOVA) and correlation analyses were performed using R (R
125 Development Core Team, 2014; [http://www/RXproject.org](http://www.RXproject.org)). Regression analysis was performed
126 using GraphPad Prism. Principal component analysis was performed using Umetrics SIMCA
127 13.0.3.

128
129
130
131
132

133 Results and Discussion

134
135

136 Comparing unfiltered beer during maturation to filtered beer during storage on the pilot 137 scale

138

139 To examine the impact of yeast on beer during the maturation stage of beer production after
140 fermentation, a lager brewed on a pilot system was divided into two streams: filtered and
141 unfiltered. While the filtered beer was sent through two cartridge filters in series, the unfiltered
142 beer was simply racked off the bulk of the yeast. Samples were taken at multiple time points for
143 both treatments and the concentrations of the metabolites present in those samples are shown
144 in Table 1. Figure 1 shows a PCA of the metabolite concentrations obtained from samples
145 collected from three replicate kegs for the filtered and unfiltered lagers. Two samples were
146 outliers. These samples corresponded to first sample taken on two separate days from the
147 same keg (Figure 1A). Examination of the metabolite concentrations from these two samples
148 (collected on days 2 and 3) revealed that the concentrations were approximately 1/3 smaller
149 than the two replicate samples taken on the same day. Therefore, these samples were
150 considered outliers and removed from further analysis. Comparison of filtered and unfiltered
151 beers (Figure 1B) revealed no clustering based on filtering. Moreover, no clustering was
152 observed based on day of sample collection. Additionally, comparison of the samples taken at
153 day two from the filtered beer (light open circles) and samples taken at day 30 from the
154 unfiltered beer (black closed circles), which represent beer conditioned on yeast, tend to cluster
155 in the middle of the PCA.

156
157

158 Analysis of variance (ANOVA) was performed on the measured metabolites to determine if
159 there were significant ($p < 0.05$) differences between treatments. When comparing the average
160 across kegs, fructose and ethanolamine were the only metabolites that differed significantly
161 between the two treatments (Table 1). After multiple comparisons correction (with false
162 discovery rate set at 5%), neither of these metabolites were significantly different. In both the
163 filtered and unfiltered beer, fructose appeared to increase between each time point; however,
164 regression analysis revealed a non significant increase. Ethanolamine remained consistent
165 throughout the maturation.

166
167

168 It has been suggested that yeast autolysis could be a main factor in the increased mouthfeel
169 described in beer aged in the presence of yeast. A number of autolysis products including
170 amino acids, amino acid derivatives, nucleosides, and nucleoside derivatives were measured in
171 this study. Interestingly, all of the amino acids and their derivatives were not different between
172 the filtered and the unfiltered beer.

173

174 Previous research also found an increase in concentration of nucleosides, specifically, cytidine,
175 uridine, guanosine, and adenosine during induced yeast autolysis (13). These metabolites were
176 not different between the filtered and unfiltered beers.

177
178 Overall, the data collected does not suggest that yeast are undergoing autolysis or that lagers
179 conditioned in the presences of yeast are markedly different from lagers conditioned without
180 yeast or even lagers without prolonged maturation of any type.

181 182 **Metabolomic changes during the maturation of a commercial lager**

183
184 To examine the metabolomic trends during maturation with yeast on a commercial scale,
185 samples were taken at nine time points during the maturation stage of a commercially produced
186 lager (Table 2).

187
188 No significant changes in concentration were observed during the maturation of the commercial
189 lager. For the vast majority of the metabolites measured during commercial maturation, non-
190 significant fluctuations in concentration occurred. The concentrations appeared to fluctuate less
191 after 23 and 30 days of maturation, but this may be misleading since the measurements are
192 more spread out temporally. With this in mind, it is difficult to assign a weight to the importance
193 of these small changes in metabolite concentrations.

194 195 196 **General discussion and conclusions**

197
198 It has become a part of received wisdom that lager-style beers should be stored post-
199 fermentation, although the rationale for this is less than clear (14, 15, 16, 17). Perusal of the
200 justification for the lagering process highlight the need to carbonate, to cold-stabilize and, in
201 respect of flavor, to deal with vicinal diketones, hydrogen sulphide and acetaldehyde. The
202 simple reality is that all of these requirements can be achieved without prolonged beer storage,
203 as was mentioned into the Introduction to this paper. Thus we are left with some nefarious
204 mention of a lager being brought to some superior state of aroma and taste balance in a storage
205 period. As stated above, the only paper to firmly refer to changes in the level of non-volatile
206 materials derived from yeast is that of Masschelein (9). As reported in the present paper, we
207 have been unable to show that there is a convincing change in the level of any flavour-relevant
208 substance in maturation of pilot scale and commercial brews. The present paper has not
209 dwelled on volatile substances, however it is amply documented that the key entities such as
210 the esters, sulphur-containing molecules, vicinal diketones, carbonyl substances (such as
211 acetaldehyde) etc should be controllable by competent fermentation and upstream process
212 practices (18). For example there are those that say that lagering is necessary to remove
213 undeirable sulphidic character, e.g. that arising from hydrogen sulphide. However ensuring
214 vigorous fermentation causes this substance to be purged with the fermentation gases (19). In
215 just the same way entities like diacetyl (6) and acetaldehyde (18) can be eliminated in the
216 fermenter and without recourse to lengthy storage periods.

217
218 There is of course no question that the flavor of beer changes with time (flavor instability), an
219 occurrence that is undesirable for most beers but potentially favorable for more alcoholic brews
220 (19). However this is a very different matter from the maturation of beer in the brewery.

221

222 The present authors contend that whilst there may be a need for some brewers to address
223 matters like diacetyl, H₂S, acetaldehyde and perhaps a few other volatile substances post
224 primary fermentation it is simply a reflection of them not having sought to, or succeeded in,
225 dealing with them earlier.

226

227 The authors suggest that perhaps there is but one area worthy of further investigation in the
228 context of flavour maturation and that would be in respect of polyphenols. Are there changes in
229 the polymerization of such materials which influences the character of beer? We also suggest
230 the need for authoritative organoleptic investigations such that the changes (if any) that occur
231 during lagering might be legitimately identified.

232

233 However, in respect of the current study, the overall lack of trends differentiating beer matured
234 in the presence of yeast from beer conditioned without yeast, suggests that prolonged contact
235 with yeast is a nonessential step in lager production in respect of non-volatile compounds. Our
236 conclusions concur with those drawn by Rennie and Wilson (20).

237

238 **Acknowledgments**

239

240 Kyle Simon is thanked for his assistance with the commercial trial.

241

242

243 **References**

244

245

- 246 1. Bamforth, C. (2002) Great Brewing Debates: Part 4 Does Beer Get Better With Ageing?
247 *Brew. Guard.*, 131 (10), 26-28
- 248 2. Hornsey, I.S. (2003) A History of Beer and Brewing. Royal Society of Chemistry, London
- 249 3. Bamforth, C.W. (1999) Beer Haze. *J. Am. Soc. Brew. Chem.*, 57, 81-90. DOI:
250 <https://doi.org/10.1094/ASBCJ-57-0081>
- 251 4. Miedl, M. and Bamforth, C.W. (2004) The relative importance of temperature and time in
252 the cold conditioning of beer. *J. Am. Soc. Brew. Chem*, 62, 75-78 DOI: DOI: 10.1094/ASBCJ-
253 62-0075
- 254 5. Speers, R.A., and Macintosh, A.J. (2013) Carbon Dioxide Solubility in Beer. *J. Am. Soc.*
255 *Brew. Chem.* 71, 242-247. DOI: <https://doi.org/10.1094/ASBCJ-2013-1008-01>
- 256 6. Inoue, T. (2008) Diacetyl in Fermented Foods and Beverages. American Society of
257 Brewing Chemists, St Paul MN.
- 258 7. Shen, N., Wang, J., Liu, C., Li, Y., and Li, Q. (2014) Domesticating brewing yeast for
259 decreasing acetaldehyde production and improving beer flavor stability. *Eur. Food Res. Tech.*
260 238, 347–355 DOI: <https://doi.org/10.1007/s00217-014-2169-0>
- 261 8. Virkajärvi, I. (2006) Accelerated processing of beer, pages 254-274, in *Brewing: New*
262 *Technologies* (Bamforth, C. Ed), Woodhead Publishing
- 263 9. Masschelein, C.A. (1986) Centenary Review: The Biochemistry of Maturation. *J. Inst.*
264 *Brew*, 92, 213-219 DOI: <https://doi.org/10.1002/j.2050-0416.1986.tb04403.x>

265 10. Pinto, J., Dominques, M.R.M., Galhano, E., Pita, C., Almeida M.d.C., Carreirade I.M.,
266 Gil, A.M. (2014). Human plasma stability during handling and storage: impact on NMR
267 metabolomics. *Analyst*. 139, 1168-1177 DOI: 10.1039/c3an02188b
268 11. Slupsky, C.M., Rankin, K.N., Wagner, J., Fu, H., Chang, D., Weljie, A.M., Saude, E.J.,
269 Lix, B., Adamko, D.J., and Shah, S. (2007). Investigations of the effects of gender, diurnal
270 variation, and age in human urinary metabolomic profiles. *Anal. Chem.* 79, 6995-7004 DOI:
271 10.1021/ac0708588
272 12. Smilowitz, J.T., O'Sullivan, A., Barile, D., German, B.J., Lonnerdal, B., and Slupsky,
273 C.M. (2013). The human milk metabolome reveals diverse oligosaccharide profiles. *J. Nut.* 143,
274 1709-1718 DOI: <https://doi.org/10.3945/jn.113.178772>
275 12.
276 13. Charpentier, C., Aussenac, J., Charpentier, M., Prome, J.C., Duteurtre, B., and Feuillat,
277 M. (2005). Release of nucleotides and nucleosides during yeast autolysis: kinetics and potential
278 impact on flavor. *J. Ag. Food Chem.* 53, 3000-3007 DOI: 10.1021/jf040334y
279 14. Briggs, D.E., Boulton, C.A., Brookes, P.A., and Stevens, R. (2004). *Brewing: Science and*
280 *Practice*. Woodhead Publishing.
281 15. De Clerck, J. (1957). *A Textbook of Brewing, Volume One*. Chapman and Hall.
282 16. Patino, H. (1999). Overview of Cellar Operations, pages 299-326, in *The Practical Brewer*
283 (McCabe, J.T. Ed), Master Brewers Association of the Americas
284 17. Munroe, J.H. (2006) Aging and Finishing, in *Handbook of Brewing*, (Priest, F.G. and
285 Stewart, G.G. Eds), CRC Taylor and Francis
286 18. Macdonald, J., Reeve, P.T.V., Ruddlesden, J.D., and White, F.H. (1984) Current
287 Approaches to Brewery Fermentations, Pges 47-198, in *Progress in Industrial Microbiology*,
288 volume 19, (Bushell, M.E., ed), Elsevier.
289 19. Bamforth, C.W. (2017) *Practical Guides for Beer Quality: Freshness*. ASBC Handbook
290 Series. American Society of Brewing Chemists.
291 20. Rennie, H., and Wilson, R.J.H. (1977) The influence of conditioning on lager beer quality. *J.*
292 *Inst. Brew.* 83, 20-24 DOI: <https://doi.org/10.1002/j.2050-0416.1975.tb03785.x>
293
294
295
296
297

298 Table 1. The average (n=3) concentration (μM) and standard error of metabolites sampled from
 299 three kegs, sourced from one fermentation, during the maturation (2°C) of unfiltered beer and
 300 the average (n=3) concentration (μM) and standard error of metabolites sampled from three
 301 kegs, sourced from the same fermentation as the unfiltered beer, during cold storage (-1°C) of
 302 the beer post filtration. Both beers were sampled at four time points over 30 days.
 303

Metabolite	Unfiltered Beer					Filtered Beer				
	Time (Day)				Average Standard Error	Time (Day)				Average Standard Error
	2*	3*	8	30		2	3	8	30	
<i>Sugars</i>										
1,6-Anhydro- β -D-glucose	75	86	80	80	7	93	76	80	71	31
Fructose	496	639	529	599	58	510	522	513	557	213
Gentiobiose	376	515	350	392	62	512	466	395	376	120
Glucose	1227	1345	1103	1236	224	1250	1436	1243	1145	492
Isomaltose	1154	1276	907	1049	246	1211	1129	1082	1129	365
Isomaltotriose	209	259	244	275	29	249	248	274	275	89
Maltose	7082	8708	7006	6873	725	8478	7016	7805	6544	2525
Melibiose	222	240	285	244	53	258	288	289	316	98
Xylose	356	361	319	392	76	390	407	377	350	138
<i>Amino Acids and Derivatives</i>										
4-Aminobutyrate	606	688	593	633	81	716	697	647	631	193
Alanine	1675	1894	1584	1714	201	1898	1889	1673	1703	552
Asparagine	321	360	310	311	40	357	346	317	322	100
Aspartate	325	375	315	331	36	366	368	332	338	104
Betaine	766	863	735	790	97	833	874	782	804	264
Glutamate	527	641	523	603	79	649	523	463	636	251
Glutamine	113	131	107	153	22	133	129	122	108	46
Histidine	239	269	235	244	27	283	270	249	258	81
Isoleucine	281	313	266	287	30	335	314	270	284	85
Leucine	366	437	412	430	41	456	431	427	428	129
Lysine	217	226	206	207	28	237	225	216	219	74
Methionine	58	61	55	59	9	67	60	58	57	19
Phenylalanine	497	564	474	514	67	567	549	492	506	151
Proline	3308	3706	3229	3542	422	3899	3853	3646	3263	1084
Pyroglutamate	1153	1335	1119	1170	122	1276	1299	1138	1150	379
Threonine	123	138	77	114	27	108	100	96	99	34
Tryptophan	161	181	154	163	20	184	182	160	163	50
Tyrosine	494	560	446	485	53	492	529	464	461	158
Valine	713	807	676	709	77	818	789	716	717	224
<i>Nucleotides and derivatives</i>										
2'-Deoxyadenosine	178	202	165	183	18	201	199	180	186	58
2'-Deoxyguanosine	12	13	11	11	1	12	12	12	11	3

Adenine	7	9	7	8	2	7	7	6	7	4
Adenosine	101	114	96	104	13	115	112	102	103	31
Cytidine	150	169	144	152	16	171	170	151	153	48
Cytosine	7	7	8	7	1	8	8	7	7	2
Guanosine	222	251	258	254	11	255	246	250	244	69
Hypoxanthine	19	20	15	18	2	19	20	17	17	6
Inosine	30	33	29	30	4	35	32	31	32	10
Oxypurinol	19	17	20	17	2	28	23	18	25	11
Thymidine	55	64	53	59	7	65	65	56	58	18
Uracil	45	47	42	42	6	51	48	44	44	13
Uridine	249	277	234	252	30	283	281	248	254	77
<i>Energy related metabolites</i>										
2-Methylglutarate	22	22	20	21	5	23	20	20	22	7
2-Oxoglutarate	36	37	32	35	7	42	39	35	38	13
Ethanol	661510	750148	630240	691894	102555	797279	746897	648956	671466	222031
Fumarate	36	41	34	37	4	41	40	36	37	11
Lactate	1033	1038	973	984	128	1119	1187	987	976	322
Malate	177	151	143	168	19	165	163	155	164	54
Pyruvate	395	442	369	393	52	467	434	390	411	132
Succinate	462	515	445	481	46	531	511	466	479	144
trans-Aconitate	16	19	15	18	2	19	19	17	18	6
<i>Fatty acid associated metabolites</i>										
Acetate	1222	1369	1152	1232	163	1359	1369	1184	1203	390
Acetoacetate	17	18	14	17	3	17	16	15	17	6
Choline	618	708	583	637	80	686	709	621	624	210
Ethanolamine	112	129	107	115	13	149	152	135	136	51
Glycero-3-phosphocholine	531	584	491	538	58	598	587	527	549	173
Glycerol	10357	11321	10188	11089	1099	12016	11834	10517	10752	3323
O-Phosphocholine	13	15	14	13	1	14	14	16	17	6
<i>Vitamins</i>										
4-Pyroxidate	18	14	13	14	3	15	17	14	14	5
Nicotinate	19	22	19	20	3	23	22	20	20	7
Pyroxidine	17	19	16	18	2	20	20	17	17	5
<i>Plant associated metabolites</i>										
Ferulate	11	10	10	11	1	13	11	11	11	4
Trigonelline	20	25	19	21	2	24	24	20	21	7
<i>Miscellaneous metabolites</i>										
Acetoin	6	7	5	6	0	7	7	6	7	2
Formate	68	79	66	71	7	78	76	68	71	22
Methanol	38	40	33	37	6	43	42	36	36	12
Propylene glycol	882	1039	688	778	288	833	739	800	618	270
3-Hydroxyisobutyrate	24	26	22	21	4	27	28	28	23	10

Critonellol	58	56	51	55	8	60	56	55	52	18
Dimethyl sulfone	5	6	3	5	1	4	4	4	4	2
o- Cresol	8	8	4	8	2	7	6	5	6	4
Theophylline	4	4	4	4	0	4	4	4	4	2
*Average of 2 replicates.										

304

305

306 Table 2. Concentration (μM) of metabolites measured during the maturation process of a commercial
 307 lager. The beer was stored at 13.7°C for the first 14 days and then was lowered to -1.6°C during the
 308 following 16 days. Samples were collected at 9 time points during the 30 days of maturation

Metabolite	Time (Days)									% Net Change	Pearson's r	P-Value
	0	1	2	3	4	5	6	23	30			
<i>Sugars</i>												
1,6-Anhydro- β -D-glucose	211	189	208	206	170	197	189	125	210	0	-0.251	0.515
Fructose	1528	889	917	739	676	916	901	479	712	-53	-0.562	0.115
Gentiobiose	425	316	333	309	278	294	290	176	347	-18	-0.327	0.390
Glucose	3017	189	223	237	214	221	195	141	177	-94	-0.375	0.320
Isomaltose	993	749	986	957	644	921	995	484	1042	5	-0.058	0.882
Isomaltotriose	160	117	150	133	109	121	123	101	132	-18	-0.328	0.389
Maltose	12069	9115	6949	6118	5884	9107	6110	3566	5395	-55	-0.641	0.063
Melibiose	285	364	420	377	338	396	413	197	282	-1	-0.524	0.148
Xylose	715	684	898	863	673	867	882	394	688	-4	-0.424	0.256
<i>Amino Acids and Derivatives</i>												
4-Aminobutyrate	754	771	855	829	705	830	829	473	808	7	-0.290	0.450
Alanine	504	476	737	712	603	718	713	425	732	45	0.149	0.702
Asparagine	73	34	39	40	28	38	33	25	48	-34	-0.184	0.635
Aspartate	65	45	31	20	18	24	38	31	39	-40	-0.140	0.720
Betaine	919	952	1037	1008	883	1025	1003	578	966	5	-0.342	0.367
Glutamate	236	143	237	209	137	189	115	103	184	-22	-0.336	0.376
Glutamine	69	59	79	65	63	136	98	62	88	28	0.165	0.671
Histidine	90	78	98	95	85	102	98	62	103	14	0.021	0.956
Isoleucine	64	49	73	89	57	66	52	44	69	8	-0.110	0.778
Leucine	130	88	110	189	78	75	75	65	89	-32	-0.376	0.319
Lysine	105	125	114	103	109	115	97	68	105	0	-0.483	0.188
Methionine	20	15	17	17	14	19	19	11	20	0	-0.029	0.942
Phenylalanine	118	96	145	143	129	136	140	87	153	30	0.160	0.682
Proline	2828	2788	3313	3217	2625	3333	3186	1999	3120	10	-0.189	0.626
Pyroglutamate	1127	964	1294	1160	907	1154	1137	806	1055	-6	-0.346	0.362
Threonine	38	21	56	56	44	39	42	20	27	-29	-0.466	0.206
Tryptophan	86	85	105	100	89	105	99	63	103	20	-0.068	0.862
Tyrosine	209	196	287	277	231	268	286	184	286	37	0.186	0.633
Valine	189	185	268	281	245	305	287	170	287	52	0.187	0.630
<i>Nucleotides and derivatives</i>												
2'-Deoxyadenosine	148	157	163	167	139	155	172	94	159	7	-0.290	0.4487
2'-Deoxyguanosine	12	12	9	13	9	12	11	6	10	-17	-0.420	0.2597
Adenine	12	6	8	5	4	8	3	3	4	-67	-0.556	0.120
Adenosine	46	48	53	51	42	53	50	29	51	11	-0.236	0.541

Cytidine	107	104	116	109	96	113	105	67	106	-1	-0.382	0.311
Cytosine	3	5	5	4	3	4	3	3	5	67	0.204	0.599
Guanosine	202	202	216	217	188	226	212	143	210	4	-0.300	0.434
Hypoxanthine	6	6	6	3	6	4	7	4	8	33	0.352	0.353
Inosine	58	60	71	65	57	63	66	38	65	12	-0.244	0.526
Oxypurinol	36	35	42	30	30	40	32	28	31	-14	-0.454	0.220
Thymidine	66	64	74	69	61	73	72	41	70	6	-0.274	0.476
Uracil	13	8	11	10	6	7	7	8	12	-8	0.144	0.711
Uridine	263	280	317	301	272	314	304	183	297	13	-0.242	0.530
<i>Energy related metabolites</i>												
2-Methylglutarate	11	11	14	11	11	13	7	11	12	9	0.024	0.950
2-Oxoglutarate	26	31	34	40	30	31	30	28	31	19	-0.110	0.777
Ethanol	532122	579855	622805	643788	564022	656840	667088	363097	614220	15	-0.213	0.582
Fumarate	47	47	53	51	44	52	51	27	49	4	-0.340	0.370
Lactate	914	876	843	784	718	946	862	740	853	-7	-0.205	0.596
Malate	151	171	228	202	141	196	176	125	119	-21	-0.628	0.070
Pyruvate	837	878	929	913	795	905	889	476	813	-3	-0.486	0.185
Succinate	393	399	457	405	353	443	445	278	405	3	-0.305	0.425
trans-Aconitate	14	15	18	15	15	17	18	12	17	21	-0.047	0.904
<i>Fatty acid associated metabolites</i>												
Acetate	292	173	560	549	469	567	566	370	632	116	0.449	0.225
Acetoacetate	11	14	14	18	12	16	17	11	16	45	0.116	0.766
Choline	974	973	1107	1074	915	1066	1067	604	1033	6	-0.302	0.430
Ethanolamine	124	146	100	102	104	177	174	93	189	52	0.390	0.300
Glycero-3-phosphocholine	20	22	21	15	14	16	21	10	22	10	-0.089	0.820
Glycerol	9960	11843	12565	11885	10346	11625	11681	7465	11171	12	-0.339	0.372
O-Phosphocholine	270	291	295	290	262	297	294	175	286	6	-0.331	0.384
<i>Vitamins</i>												
4-Pyroxidate	6	8	7	11	8	9	8	6	9	50	0.134	0.731
Nicotinate	5	3	4	4	5	4	4	3	4	-20	0.126	0.746
Pyroxidine	12	13	16	11	11	15	14	6	16	33	0.023	0.954
<i>Plant associated metabolites</i>												
Ferulate	8	13	9	11	10	13	11	10	13	63	0.410	0.273
Trigonelline	19	20	21	21	18	23	25	13	19	0	-0.303	0.428
<i>Miscellaneous metabolites</i>												
Acetoin	12	15	17	15	14	17	16	9	14	17	-0.332	0.383
Formate	49	30	38	38	30	34	35	20	32	-35	-0.514	0.157
Methanol	47	51	51	56	45	54	51	30	46	-2	-0.488	0.182
Propylene glycol	917	464	532	456	625	732	700	468	774	-16	0.104	0.789
3-Hydroxyisobutyrate	45	27	37	57	35	31	31	21	25	-44	-0.560	0.117
Critonellol	58	57	68	62	39	54	40	39	47	-19	-0.536	0.137
Dimethyl sulfone	12	14	15	15	13	13	14	8	13	8	-0.365	0.334

o- Cresol	10	9	14	11	11	12	12	7	15	50	0.251	0.515
Theophylline	3	3	6	3	4	4	3	2	3	0	-0.288	0.452

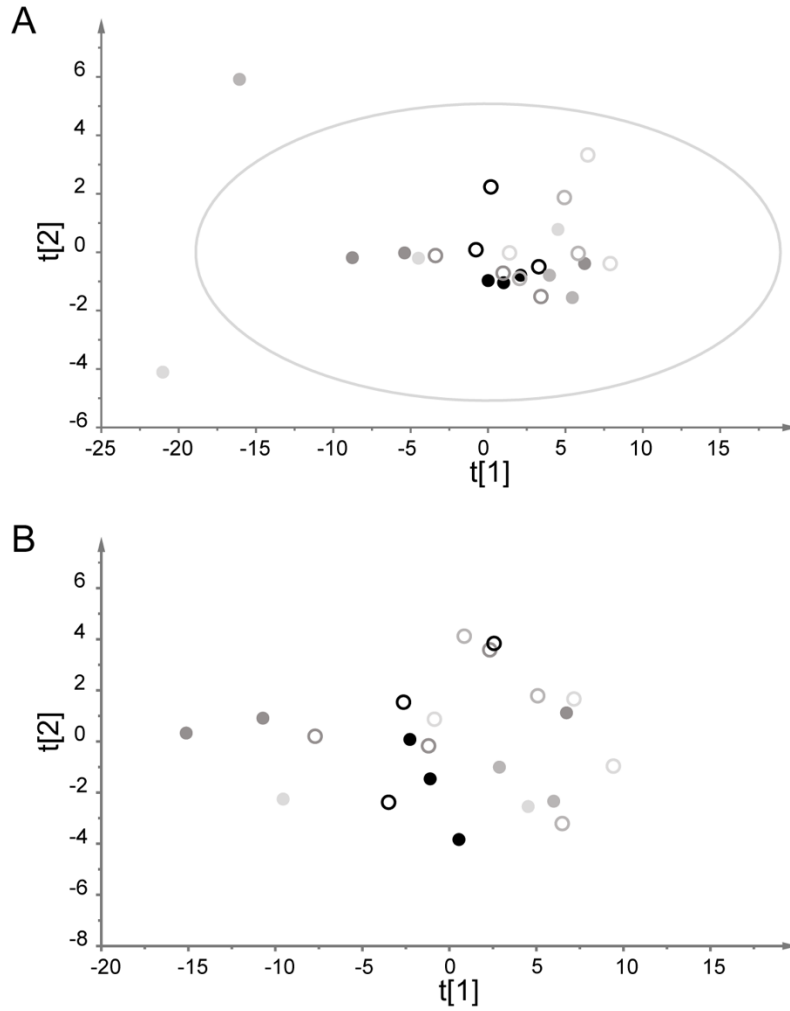
309

310

311

312

313 Fig 1. PCA of filtered and unfiltered beer at various time points during maturation. (A) All samples. The
314 ellipse represents Hotellings T2. (B) Removal of two outliers (outside of the Hotellings T2 limit).
315 Unfiltered beer closed circles; filtered beer, open circles. Day 2 (light grey), Day 3 (medium grey), Day 8
316 (dark grey), Day 30 (black).
317



318