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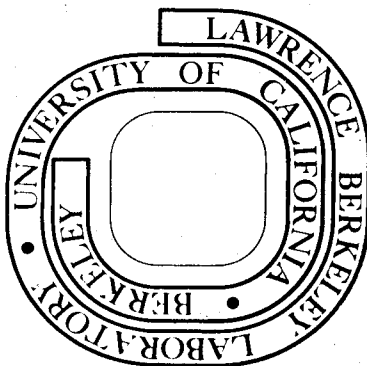
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Relativistic Effects in the UV Photoelectron  
Spectra of Group VI Diatomic Molecules

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ABSTRACT

The anomalous  ${}^2\Pi_{1/2}/{}^2\Pi_{3/2}$  intensity ratios observed recently in the photoelectron spectra of  $S_2$ ,  $Se_2$ , and  $Te_2$  are shown to arise from spin-orbit coupling in the ground state. An intermediate coupling calculation gives good agreement with all observed intensity ratios and predicts a ratio of  $\sim 35$  in  $Po_2$ .

-1-

The HeI photoelectron spectra of the group VI homonuclear diatomics,  $O_2$ ,<sup>1,2</sup>  $S_2$ ,<sup>3,4</sup>  $Se_2$ ,<sup>5</sup> and  $Te_2$ <sup>4,5</sup> have all been reported. In every case, the spin-orbit (s-o) components of the  $X^2\Pi_g$  ionic state were resolved. The branching ratio of  ${}^2\Pi_{1/2}/{}^2\Pi_{3/2}$  (the subscript g is dropped for simplicity) was found to be increasingly larger than 1, the statistical ratio, with increasing molecular weight (Table I), and for  $Te_2$ , the ratio reached a dramatic value of 10. Berkowitz and coworker<sup>4,5</sup> attempted to explain this intensity variation as follows. During the photoionization the allowed transitions  $\Delta\Omega = 0$  were postulated to be strongly preferred to  $\Delta\Omega = \pm 1$ , as was observed in the absorption spectra of  $Se_2$ <sup>6</sup> and  $TeO$ <sup>7</sup> in the visible and near UV region. In addition, the continuum photoelectrons resulting from photoionization were postulated to have a strong preference for the  $\epsilon p\sigma$ , rather than  $\epsilon p\pi$ , channel. These additional constraints, together with the ground state of the heavier diatomic chalcogens being more appropriately described as  $XO_g^+$ , rather than  $X^3\Sigma_g^-$  as a result of Hund's coupling case (c) would account for the greater intensity of the  ${}^2\Pi_{1/2}$  peak in  $Te_2^+$ , relative to  ${}^2\Pi_{3/2}$ . The smaller  ${}^2\Pi_{1/2}/{}^2\Pi_{3/2}$  ratios in lighter diatomics than  $Te_2$  were attributed to contributions from the  $X1_g$  state, populated thermally because of smaller  $X1_g - XO_g^+$  splitting.

We note however, the above explanation is unsatisfactory if applied to other heavy linear molecules, where  $\Omega$  is more nearly a good quantum number in the ions and neutral molecules. The same arguments<sup>4,5</sup> would then predict intensity ratios  ${}^2\Pi_{1/2}/{}^2\Pi_{3/2}$  much larger than unity in photoionization, because  $O^+$  is normally the ground state. This, however, is not the case; for example, in  $I_2$ <sup>8</sup> and  $Bi_2$ <sup>9</sup> the ratio is close to 1.

Furthermore, to draw an analogy between photoionization and photoabsorption to a valence state involving different MO's might be invalid. In this communication, we offer an alternative explanation for the observed  ${}^2\Pi_{1/2}/{}^2\Pi_{3/2}$  ratios in dichalcogens. Our approach requires no assumptions about the dynamics of photoionization, but the effects of intermediate coupling on the  $\pi_{1/2} - \pi_{3/2}$  composition of the ground state are taken into account.

Recently, we reported relativistic effects in the photoelectron spectrum of Pb vapor,<sup>10</sup> which caused the observed branching ratio of  ${}^2P_{1/2}/{}^2P_{3/2}$  in  $Pb^+$  to be 14.1 as opposed to the statistical value of 0.5 expected in L-S coupling. This is a consequence of the ground state of Pb being best approximated in terms of j-j coupling as  $(p_{1/2})^2$  with a small configuration interaction (CI) admixture of  $(p_{3/2})^2$ ; i.e.,  $\Psi = a(p_{1/2})_0^2 + b(p_{3/2})_0^2$ ; with  $a^2/b^2 \sim 14$ . We believe a similar phenomenon happens with the heavier Group VI diatomics, except that here we have the molecular analogue. As the s-o coupling gets stronger, it becomes more appropriate to describe the coupling of angular momenta in diatomics in Hund's case c rather than case b (or a) coupling.<sup>11,12</sup> If only the electronic states are considered, this is identical to using  $\omega-\omega$  rather than  $\Lambda-\Sigma$  coupling,<sup>12</sup> the molecular equivalents of j-j and L-S couplings respectively. In  $\omega-\omega$  coupling, the  $\vec{L} \cdot \vec{S}$  interaction splits the  $\pi$  molecular orbital (MO) into  $\omega = 1/2$  and  $3/2$  orbitals; with the former being lower in energy. For Group VI diatomics with the  $\pi^2$  configuration, this gives rise to three configurations and four states; i.e.,  $(1/2, 1/2)_0$ ,  $(1/2, 3/2)_{1,2}$ , and  $(3/2, 3/2)_0$  (where the subscripts denote  $\Omega$ , the axial projection of the total electronic angular momentum). The ground state of  $\pi^2$  ( $\Omega = 0$ ) can then be approximated in  $\omega-\omega$  coupling by a simple CI

expansion

$$\Psi(X0_g^+) = a(1/2, 1/2)_0 + b(3/2, 3/2)_0. \quad (1)$$

If we neglect the difference in the  $\pi_{1/2}$  and  $\pi_{3/2}$  photoionization cross sections (which in any event should be small), the intensity ratio  $^2\Pi_{1/2}/^2\Pi_{3/2}$  is given by  $a^2/b^2$ . The relative weights of  $a$  and  $b$  depend on the strength of the  $\vec{l} \cdot \vec{s}$  interaction. In terms of  $\Lambda$ - $\Sigma$  coupling, eq.(1) corresponds to a mixture of  $^3\Sigma_0^-$  and  $^1\Sigma_0^+$  states. In the absence of s-o coupling, the ground state is (see below)

$$\Psi^{\circ}(X^3\Sigma_{g,0}^- \text{ or } X0_g^+) = \frac{1}{\sqrt{2}} (1/2, 1/2)_0 + \frac{1}{\sqrt{2}} (3/2, 3/2)_0, \quad (2)$$

and the  $^2\Pi_{1/2}/^2\Pi_{3/2}$  ratio is 1. By contrast, if the s-o coupling is large, then the ground state will be predominantly  $(1/2, 1/2)_0$  and the ratio will be much greater than 1, or  $X^2\Pi_{3/2}$  may hardly be detected at all. If the splitting<sup>13</sup> between  $X^3\Sigma_0^-$  (or  $0^+$ ) and  $X^3\Sigma_1$  (or 1) is not too large for the  $X1_g$  state to be populated thermally, but too small to be resolved, then the experimental ratio will be altered by contribution from  $X1_g$ , which would give a  $^2\Pi_{1/2}/^2\Pi_{3/2}$  ratio of unity.

In the following, we present a simple calculation by which the CI wave functions (eq.(1)) of Group VI diatomics can be estimated, to obtain the  $^2\Pi_{1/2}/^2\Pi_{3/2}$  ratio theoretically. Because relativistic molecular wave functions are not available, we shall instead expand the nonrelativistic CI function in terms of  $\Lambda$ - $\Sigma$  basis functions. Thus the ground state can be expressed as

$$\Psi(X^3\Sigma_{g,0}^- \text{ or } X0_g^+) = C_1\Psi^{\circ}(^3\Sigma_0^-) + C_2\Psi^{\circ}(^1\Sigma_0^+), \quad (3)$$

where

$$\Psi^{\circ}(^3\Sigma_0^-) = A \left\{ \pi_g^+(1) \pi_g^-(2) (1/\sqrt{2}) [\alpha(1)\beta(2) + \beta(1)\alpha(2)] \times [\text{paired electrons}] \right\}$$

and

$$\Psi^{\circ}(^1\Sigma_0^+) = A \left\{ \pi_g^+(1) \pi_g^-(2) (1/\sqrt{2}) [\alpha(1)\beta(2) - \beta(1)\alpha(2)] \times [\text{paired electrons}] \right\}. \quad (4)$$

Approximating the s-o Hamiltonian by  $H_{s-o} = \sum_i \zeta_i \vec{l}_i \cdot \vec{s}_i$ ,<sup>14</sup> where  $\zeta_i$  is the atomic s-o coupling constant, we then obtain the matrix elements of  $H = H_0 + H_{s-o}$  as follows:

	$^3\Sigma_0^-$	$^1\Sigma_0^+$
$^3\Sigma_0^-$	$E_0(^3\Sigma_0^-)$	$\zeta_{np}$
$^1\Sigma_0^+$	$\zeta_{np}$	$E_0(^1\Sigma_0^+)$

where  $H_0$  is the electrostatic Hamiltonian. The expressions for  $E_0$ 's can be readily obtained from eq.(4), and of relevance here is the difference  $E_0(^1\Sigma_0^+) - E_0(^3\Sigma_0^-) = 2K$ , i.e., two times the exchange integral between  $\pi_g^+$  and  $\pi_g^-$  MO's. Referenced to  $E_0(^3\Sigma_0^-)$ , the solutions of the corresponding secular equation are  $E = K \pm \sqrt{K^2 + \zeta_{np}^2}$  and  $C_2/C_1 = E/\zeta_{np}$ . In extreme  $\omega$ - $\omega$  coupling, i.e.,  $K = 0$ , we have  $E = \pm \zeta_{np}$  and  $C_2/C_1 = \pm 1$ , eq.(3) gives the wave functions of  $(1/2, 1/2)_0$  and  $(3/2, 3/2)_0$  respectively. From this, we get eq.(2) and

$$\Psi^{\circ}(^1\Sigma_0^+) = -\frac{1}{\sqrt{2}}(1/2, 1/2)_0 + \frac{1}{\sqrt{2}}(3/2, 3/2)_0, \quad (5)$$

by a simple transformation. Thus, a and b in eq.(1) can be expressed in terms of  $C_1$  and  $C_2$  using eqs.(1),(2),(3), and (5), giving  $a/b = (C_1 - C_2)/(C_1 + C_2)$ . In calculating a/b, we estimated E in two different ways. One



approach was to calculate  $E$  directly from  $K$  and  $\zeta_{np}$ , where  $K$  was evaluated by approximating  $\pi_g^\pm$  as linear combinations of Clementi and Roetti's double zeta valence of  $np$  atomic orbitals.<sup>15</sup> Alternatively,  $E$  was set equal to the splitting,<sup>13</sup> i.e.,  $2\lambda_0$ , between  $X^3\Sigma_0^-$  and  $X^3\Sigma_1$  states, since this triplet splitting is predominantly due to the  $s$ - $o$  interaction discussed above,<sup>16</sup> except perhaps in  $O_2$ .

In Table I, the results of the calculation are presented and compared with experiments, and  $Po_2$  is also included for completeness. It is evident that the calculated  $K$  grossly overestimates the splitting between  $^3\Sigma_g^-$  and  $^1\Sigma_g^+$ , since it is 2.66 in  $O_2$  compared to 1.64 eV by experiment.<sup>11</sup> Therefore, it seems better to use the triplet splitting  $2\lambda_0$  for  $E$ , and in fact, by doing this the calculation is brought closer to experiment for  $S_2$  and  $Se_2$ . In  $Te_2$ , the probable cause of the disparity between calculation and experiment lies mainly in the uncertainty in the value of  $\zeta_{np}$  or molecular  $s$ - $o$  splitting. It therefore seems more appropriate to describe  $Te$  in the intermediate coupling scheme and extract  $\zeta_{np}$  accordingly. When this  $\zeta_{np}$  is used, the agreement with experiment is impressive, although  $\zeta_{np}$  thus obtained for  $Te$  is actually larger than the  $s$ - $o$  coupling constant in  $Te_2^+$  (which is 0.47 eV).<sup>4</sup> In any event, considering the inherent approximations of the calculation, the agreement between experiments and calculations are generally very good; and this supports our interpretation. It is predicted that in  $Po_2$  the  $^2\Pi_{1/2}/^2\Pi_{3/2}$  ratio will be even larger, to the extent the  $^2\Pi_{3/2}$  state may be difficult to detect.

In the light of our discussion, it is evident that relativistic effects will have a similar influence on the intensity ratio of the  $s$ - $o$  components in every molecule with open-shell,  $\pi^2$  configuration and

appreciable s-o interactions. This includes, for example, the hetero-nuclear Group VI, all heavy Group IV, and Group V-VII diatomic molecules. Furthermore, the present interpretation predicts, in the absence of cross section difference,  ${}^2\Pi_{1/2}/{}^2\Pi_{3/2}$  ratios of 1 for  $I_2$ ,  $Bi_2$ , and for other closed shell molecules, as corroborated by experiment.<sup>8,9</sup>

The present case of intermediate coupling can be conveniently described pictorially. Fig. 1 shows the dependence of the intermediate coupling or  $a^2/b^2$  upon the relative strengths of  $\vec{l}\cdot\vec{s}$  interaction and exchange integral. Although the figure is drawn for dichalcogens, it can be generally applied to other open shell diatomics to estimate the  ${}^2\Pi_{1/2}/{}^2\Pi_{3/2}$  ratio as long as the s-o and exchange splittings are known.

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Table I. Calculated and Experimental Branching Ratios of  ${}^2\Pi_{g,1/2}/{}^2\Pi_{g,3/2}$  of Group VI Diatomics.

Molecule	$\zeta_{np}$ (eV) <sup>a</sup>	K(eV)	$E(XO_g^+)$ (eV) <sup>b</sup>	$C_2/C_1$	$a^2/b^2$	${}^2\Pi_{g,1/2}/{}^2\Pi_{g,3/2}$ <sup>c</sup>	
O <sub>2</sub>	0.0187	1.331	$-1.31 \times 10^{-4}$	$-7.02 \times 10^{-3}$	1.03	1.01	$\sim 1.0^d$
	0.0187	—	$-4.92 \times 10^{-4*}$	$-2.63 \times 10^{-2}$	1.11	1.04	
S <sub>2</sub>	0.0474	0.855	$-1.31 \times 10^{-3}$	$-2.77 \times 10^{-2}$	1.12	1.04	$1.2^e$
	0.0474	—	$-2.94 \times 10^{-3*}$	$-6.19 \times 10^{-2}$	1.28	1.10	
Se <sub>2</sub>	0.210	0.779	$-2.78 \times 10^{-2}$	-0.132	1.70	1.38	$\sim 2^f$
	0.240	0.779	$-3.61 \times 10^{-2}$	-0.151	1.84	1.43	
	0.210	—	$-4.55 \times 10^{-2*}$	-0.216	2.41	1.74	
	0.240	—	$-4.55 \times 10^{-2*}$	-0.190	2.16	1.61	
Te <sub>2</sub>	0.389	0.677	-0.104	-0.267	2.99	2.99	$\sim 10^g$
	0.530	0.677	-0.183	-0.345	4.21	4.21	
	0.389	—	-0.276*	-0.711	35.0	35.0	
	0.530	—	-0.276*	-0.521	10.1	10.1	
Po <sub>2</sub> <sup>g</sup>	1.688	0.586	-1.20	-0.711	35.2	35.2	—

(Continued)

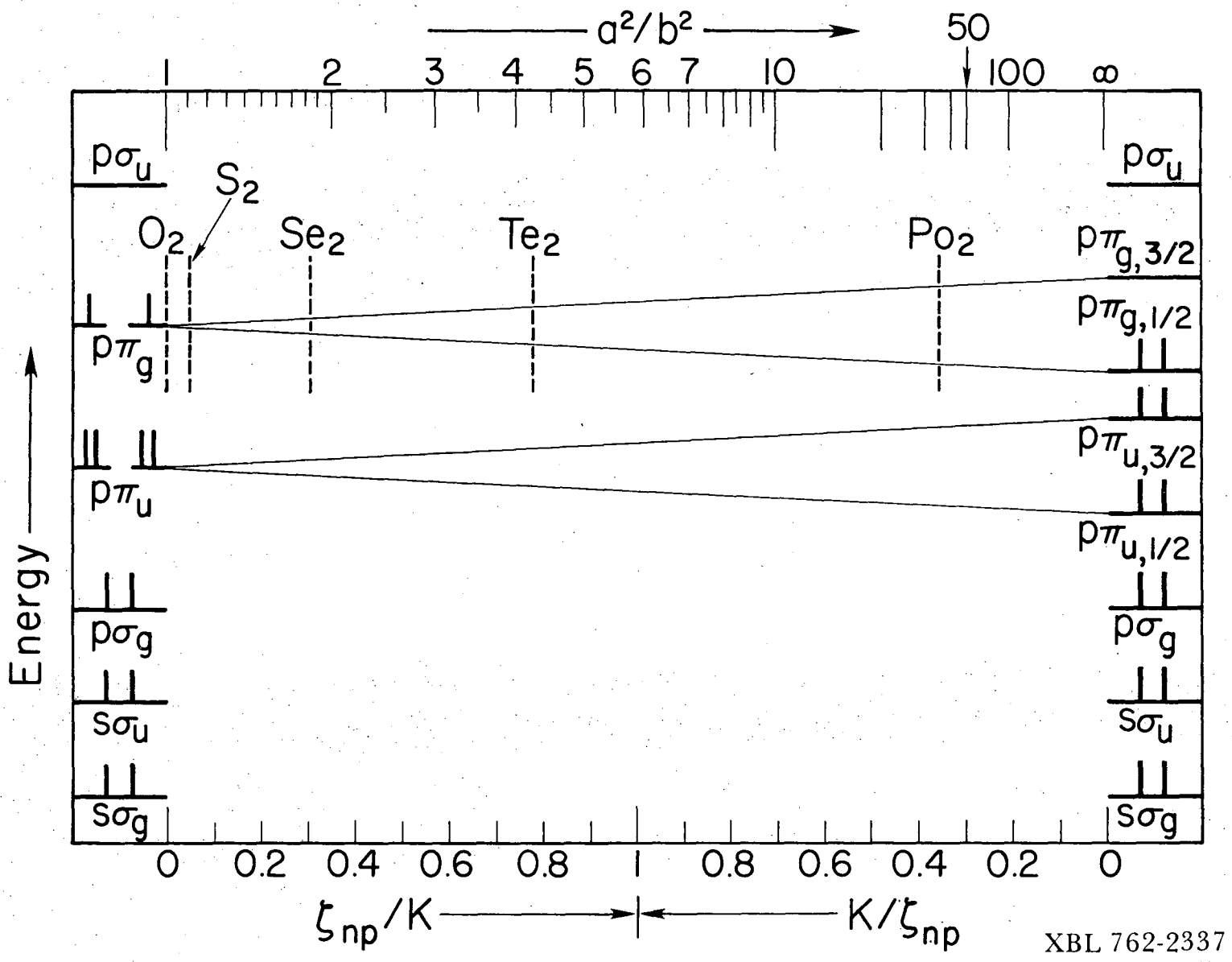
Table I. Calculated and Experimental Branching Ratios of  ${}^2\Pi_{g,1/2}/{}^2\Pi_{g,3/2}$  of Group VI Diatomics. (Continued)

- a) Atomic s-o coupling constants estimated from optical data (Ref. 17); for  $O_2$ ,  $S_2$ , and the first  $\zeta_{np}$  value of  $Se_2$  and  $Te_2$ ,  $\zeta_{np} = 2/3({}^3P_0 - {}^3P_2)$ , while the second  $\zeta_{np}$  value of  $Se_2$  and  $Te_2$ , and that for  $Po_2$  were obtained from extrapolation using  $\zeta_{np}$  values of their respective neighboring atoms in the Periodic Table as given in Ref. 18. These latter  $\zeta_{np}$ 's were obtained in the intermediate coupling scheme, which are more appropriate for the heavier atoms.
- b) The energy given here is referenced to the unperturbed  $X^3\Sigma_g^-$ . The E values are from calculation (see text) and those denoted by stars are the splittings  $2\lambda_0$  in  $X^3\Sigma_g^-$  given in Ref. 13.
- c) First column are calculated ratios by weighting  $a^2/b^2$  appropriately by accounting for the thermal population of  $1_g$  states (see text), i.e.,  $1_g/0_g^+ = 2\exp[E(0_g^+)/kT]$ , and second column are experimental results. No such correction is necessary for  $Te_2$  and  $Po_2$ .
- d) Ref. 1 gives  $\sim 0.95$  and Ref. 2  $\sim 1.0$  as estimated by present authors.
- e) Estimated from Ref. 3 (Ref. 5 gives this estimated value as  $\sim 1.5$ ).
- f) Ref. 4 and 5.
- g) The MO of  $Te_2$  with principal quantum number of atomic functions raised correspondingly by 1 was used for  $Po_2$  to calculate K.

Figure Caption

Fig. 1 Schematic correlation between  $\Lambda$ - $\Sigma$  (extreme left), intermediate and  $\omega$ - $\omega$  coupling (extreme right) showing the dependence of  $a^2/b^2$  upon the relative strengths of  $\vec{\ell} \cdot \vec{s}$  ( $\zeta_{np}$ ) and exchange (K) interactions. The coupling cases of the individual dichalcogens are indicated explicitly with electrons denoted by bars.





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Fig. 1

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