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THE ISOTOPES OF AMERICIUM

K. Street, Jr., A. Ghiorso, and G. T. Seaborg

April 11, 1950

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THE ISOTOPES OF AMERICIUM

K. Street, Jr., A. Ghiorso, and G. T. Seaborg Radiation Laboratory and Department of Chemistry University of California, Berkeley, California April 11, 1950

ABSTRACT

Three new americium activities $(Am^{238?}, Am^{243}, and Am^{244?}, the latter two formed by n, <math>\gamma$ reactions) are described and some additional information is given on previously reported americium isotopes.

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THE ISOTOPES OF AMERICIUM

DECLASSIFIED K. Street, Jr., A. Ghiorso, and G. T. Seaborg Radiation Laboratory and Department of Chemistry University of California, Berkeley, California April 11, 1950

Intermittent investigation of the isotopes of americium during the past two years has resulted in the production of three new americium activities and has also yielded some additional information about previously $^{\perp}$ reported activities. Since the work on americium is still in progress, this letter will give only a brief report of the new results obtained and a detailed description of the experiments will be deferred until a later date.

Am^{238?} .-- Bombardment of Pu²³⁹ with 50-Mev deuterons in the 184-inch cyclotron results in the production of a new americium activity of ca. 1.2 hours half-life in addition to the previously reported 12- and 50-hour activities. Differential counting with beryllium and lead absorbers indicate that the decay is accompanied by the emission of conversion electrons and x-rays. The same activity is produced in good yield with 17-Mev deuterons in the 60-inch cyclotron, thus its mass number is probably greater than 237, since there is good evidence^{2,3} that even with the heavy elements the yield of the (d,4n) reaction is very small at 17 Mev. In view of its half-life, radiation characteristics, and method of formation, this activity is probably an orbital electron capturing isotope and is best assigned to Am^{238} .

^LG. T. Seaborg, R. A. James, and L. O. Morgan, National Nuclear Energy Series, Plutonium Project Record, Vol. 14B, "The Transuranium Elements: Research Papers," Paper 22.2 (McGraw-Hill Book Co., Inc., New York, 1949).

 2 R. A. James, A. E. Florin, H. H. Hopkins, Jr., and A. Ghiorso, <u>ibid</u>., Paper No. 22.8.

³L. B. Magnusson, S. G. Thompson, and G. T. Seaborg, Phys. Rev. (May 15, 1950 issue).

 $\underline{Am^{239}}$.-- The l2-hour orbital electron capturing $\underline{Am^{239}}$ has been previously reported to decay by alpha-particle emission of unknown energy to the extent of <u>ca</u>. 0.1 percent.¹ A redetermination of the alpha branching ratio yielded the better value <u>ca</u>. 0.01 percent. The energy of the alpha-particle has been determined by means of a multi-channel pulse analyzer to be 5.77 \pm 0.05 Mev.

 \underline{Am}^{240} .-- The 50-hour electron capturing americium activity was tentatively assigned¹ to \underline{Am}^{240} . This assignment has been confirmed. The 50-hour activity is produced in good yield from the bombardment of \underline{Pu}^{239} with 10-Mev deuterons and is not observed in the bombardment of \underline{Pu}^{239} with 9-Mev protons (60-inch cyclotron) in which \underline{Am}^{239} is produced in good yield. Since the yield of the (d,3n) reaction is low at 10 Mev and since (p,n) should be the principle reaction producing americium with 9-Mev protons, the 50-hour activity is best assigned to \underline{Am}^{240} . No alpha-particles associated with the 50-hour americium have been observed. This is consistent with its assignment to \underline{Am}^{240} since the systematics of alpha-radioactivity⁴ predict for this odd-odd nucleus a partial half-life for alpha-decay of some 10^3 years.

 $\underline{Am^{241}}$.-- The production of this isotope in milligram amounts by the neutron irradiation of plutonium⁵ makes it possible to study the higher mass isotopes of americium produced by (n, Υ) reactions and the results of such experiments are reported in the following paragraphs. Recent specific activity measurements by B. B. Cunningham, S. G. Thompson, and H. R. Lohr⁶ give a somewhat shorter

4I. Perlman, A. Ghiorso, and G. T. Seaborg, Phys. Rev. 77, 26 (1950).

⁵A. Ghiorso, R. A. James, L. O. Morgan, and G. T. Seaborg, Phys. Rev., in press.

⁶B. B. Cunningham, S. G. Thompson, and H. R. Lohr, unpublished work.

half-life for Am^{241} than that previously reported.⁷ The new half-life is 475 years.

 \underline{Am}^{242} .-- In addition to the 16-hour $\underline{Am}^{242}\underline{m}$, a long-lived \underline{Am}^{242} produced by the (n,Y) reaction on \underline{Am}^{241} has been reported;¹ alpha-decay of this isotope was demonstrated by separating the Np²³⁸ daughter and 0.5-Mev beta-particles in the americium fraction were also ascribed to \underline{Am}^{242} . The fact that the observed beta-particles do belong to \underline{Am}^{242} has now been demonstrated by observing the growth of \underline{Cm}^{242} (half-life 162 days) in the purified, pile neutron-bombarded americium in an amount corresponding to the number of beta-particles observed. Mass spectrographic analysis of this americium shows \underline{Am}^{242} to be present to the extent of <u>ca</u>. 0.5 percent. This analysis together with the results from the above \underline{Cm}^{242} growth experiment and the yield from a Np²³⁸ extraction allow both the alpha- and beta-decay half-lives of \underline{Am}^{242} to be estimated. The partial half-life obtained for beta-particle emission is roughly 10² years and that for alpha-particle emission is <u>ca</u>. 10⁴ years. The (n,Y) cross section for the formation of \underline{Am}^{242} is very roughly 10⁻²² cm² but this may be off by a factor of several due to the many uncertainties involved, especially the neutron flux.

 $\underline{Am^{243}}$.-- Neptunium separations from the americium fraction of an irradiation of $\underline{Am^{241}}$ with pile neutrons show the presence of equilibrium amounts of both Np^{238} and Np^{239} . The presence of Np^{239} proves the existence of the new isotope $\underline{Am^{243}}$ which is produced by two successive neutron capture processes in $\underline{Am^{241}}$ and decays by alpha-particle emission to Np^{239} . Mass spectrographic analysis of the americium of this bombardment showed $\underline{Am^{243}}$ present to the extent of <u>ca</u>. 0.5 percent. This together with the yield of Np^{239} determined in the chemical

7B. B. Cunningham, National Nuclear Energy Series, Plutonium Project Record, Vol. 14B, "The Transuranium Elements: Research Papers," Paper No. 19.2 (McGraw-Hill Book Co., Inc., New York, 1949).

extraction experiments gives a partial half-life for alpha-particle emission for Am^{243} of roughly 10⁴ years. Alpha-particle pulse analysis of americium from another irradiation (containing approximately 10 percent Am^{243}) showed the alpha-particle energy to be 5.21 \pm 0.03 Mev. This energy and half-life indicate that alpha-emission in this odd-even nucleus is not prohibited,⁴ but like the case of the analogous nucleus Am²⁴¹, the measured alpha-particle energy may not represent the ground state transition. If one takes 5.84 Mev^{8} for the alpha-particle energy of the ground state transition of Cm^{243} and closes a decay cycle using the measured alpha-particle energy of Am²⁴³ and 0.7 Mev for the beta-decay disintegration energy⁹ of Np²³⁹, one finds that Am²⁴³ could be unstable with respect to beta-decay by ca. 0.05 Mev. However, due to the uncertainties in the above energies (especially Np²³⁹) it is entirely possible that Am^{243} is beta-stable; the alpha-particle energy of Cm^{243} is also very uncertain and it is not known whether it corresponds to the ground state transition. In order to test this point curium was separated from an aged sample containing Am^{243} and no detectable Cm^{243} was found; this experiment places a lower limit of about 10^3 years on the partial half-life of Am²⁴³ for negative beta-emission. The cross section for the reaction $Am^{242}(n,\gamma)Am^{243}$ seems to be large, of the order of some 4×10^{-21} cm², a value which is subject to large error because of the uncertainty in the neutron flux.

<u>Am^{244?}</u>.-- Irradiation of americium containing approximately 10 percent of the isotope Am²⁴³ with thermal neutrons in the uranium-heavy water pile at the Argonne National Laboratory produced a new americium activity of <u>ca</u>. 25-min. half-life at a yield corresponding to a cross section of roughly $1/2 \ge 10^{-22}$ cm².

⁸K. Street, Jr., S. G. Thompson, and A. Ghiorso (to be published).

⁹G. T. Seaborg and I. Perlman, Revs. Mod. Phys. <u>20</u>, 585 (1948).

This activity is probably due to the beta-emitting isotope Am^{244} , formed by an (n, γ) reaction.

We would like to thank Dr. R. A. James and Dr. S. G. Thompson for their assistance in many of the experiments, and Mr. F. L. Reynolds for the mass spectrographic analysis of the americium samples. It is a pleasure to thank T. M. Putnam, Jr., G. B. Rossi, J. T. Vale, and the crews of the 60-inch and 184-inch cyclotrons for their cooperation in the bombardments with these machines, and also W. H. Zinn and the operating group of the Argonne Laboratory pile for their cooperation and help in neutron irradiation experiments.

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