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Nickel-Catalyzed Cross-Coupling Reactions: Stereospecific Arylations, Formation of 2-PyridylZinc Reagents & Directed Hydroarylation of Alkynes

DISSERTATION

submitted in partial satisfaction of the requirements for the degree of

DOCTOR OF PHILOSOPHY

in Chemistry

by

Luke Edward Hanna

Dissertation Committee: Professor Elizabeth R. Jarvo, Chair Professor Vy M. Dong Professor Christopher D. Vanderwal

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DEDICATION

То

my parents, family and friends

in recognition of their inspiring character, love and support

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ACKNOWLEDGMENTS

I thank the American Chemical Society for permission to include Chapter 1 which was originally published in the Journal of the American Chemical Society. I also thank John Wiley and Sons for permission to include portions of Chapter 3, which were originally published in Angewandte Chemie.

I thank Dr. Peter De lijser for his great mentorship as my undergraduate advisor and for being a good friend throughout my undergraduate and graduate years. I would also like to thank all of the past and present members of the Jarvo lab; Ruben Martinez, Peg Green, Ivelina Yonava, Hanna Wisniewska, Elizabeth Swift, Buck Taylor, Mike Shaghafi, Ben Kohn, Michael Harris, Emily Tollefson, Charlotte Osborne, Aaron Johnson, Mikhail Konev, Lucas Erickson, David Dawson, Thom Endean, Erika Lucas, and Kenji Domon for their great friendship, patience, willingness to listen and many great discussions. I would also like to specifically thank David Dawson for being an awesome baymate with whom I have discussed almost every topic and have enjoyed working alongside. Additionally, my former baymate Peg Green was an outstanding mentor and friend. My co-authors Michael Harris, Emily Tollefson, Peg Green, Mikhail Konev deserves much recognition for making the projects we worked on both scholarly and successful. I would also like to acknowledge Sarah Nainar for being close friend and source of support. Additionally, I thank Erin Campbel, whom I have known since we were kids for being a great sounding board and brother over the years.

I thank Professor Liz Jarvo for all her patience, guidance and mentorship during my time as a student in her research group. I have learned a great deal about organometallic chemistry during my tenure in the Jarvo group, which has been an outstanding setting for me to grow as a person and a scientist. The trust, respect and calm professional temperament that Professor Jarvo shows her students are what make the Jarvo group an excellent environment to do and learn organic chemistry.

I would like to thank the DOE for funding (GAANN PA200A120070 to L.E.H.), and Frontier Scientific for generous donations of boronic acids, Dr. Joseph Ziller and Dr. John Greaves are acknowledged for X-ray crystallographic and mass spectrometry data, respectively. I would also like to acknowledge professors David Van Vranken, Andy Borovik, Christopher Vanderwal, Vy Dong, Jennifer Prescher, Larry Overman and James Nowick, for their passion for education and for being great sources of inspiration.

Finally, I would like to thank my family. Particularly my mom and dad who against many odds immigrated to this great country in the 1980's and struggled to give me a good life and bright future; I will forever be indebted to them for their hard work, honesty, and love.

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Hanna, L. E.; Jarvo E. R. "Selective Cross-Electrophile Coupling By Dual Catalysis" Angew. Chem. Int. Ed. 2015, 54, 15618.**

Konev, M. O.; <u>Hanna, L. E.</u>; Jarvo E. R. "Nickel-Catalyzed Reductive Cross-Electrophile Coupling Reaction of Primary and Secondary Benzylic Esters With Aryl Halides" *Angew. Chem. Int. Ed.* **2016**, *55*, 6730.

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ABSTRACT OF THE DISSERTATION

Nickel-Catalyzed Cross-Coupling Reactions: Stereospecific Arylations, Formation of 2-PyridylZinc Reagents & Directed Hydroarylation of Alkynes

by

Luke E. Hanna Doctor of Philosophy in Chemistry University of California, Irvine, 2016 Professor Elizabeth R. Jarvo, Chair

Cross-coupling technology has become an indispensable tool for the rapid and efficient synthesis of complex molecules. Over the past few decades a foundational understanding of organometallic chemistry has been laid using palladium and other precious metals. Recent research on first row base metal catalysts such as nickel, cobalt and iron has uncovered new and complementary modes of reactivity compared to their more well-studied precious metal counterparts. While nickel sits one row above palladium on the periodic table, ongoing research has illustrated that nickel possesses a unique reactivity profile. Thus, while nickel is commonly thought of as a cheaper alternative to palladium, research in the field of nickel catalysis has demonstrated far more potential than this. The unique propensity of nickel to undergo single electron chemistry as well as its ability to break strong carbon oxygen bonds make research into nickel reactivity an immensely beneficial endeavor to the fields of inorganic, organometallic and synthetic organic chemistry.

Chapter 1 describes the development of a stereospecific Suzuki coupling of benzylic carbamates and pivalates with aryl- and heteroarylboronic esters. The reaction proceeds with selective inversion or retention at the electrophilic carbon, depending on the identity of the ligand used. Tricyclohexylphosphine ligand provides products with retention of configuration at the electrophilic carbon, while an N-heterocyclic carbene ligand SIMes provides products with inversion.

Chapter 2 discusses the development of a regio- and stereoselective nickel-catalyzed hydroarylation of alkynes using propargylic carbamates as directing groups. The reaction proceeds under mild reaction conditions using arylboronic acids in the absence of base. A range of heterocycles and functional groups are tolerated under the reaction conditions. Additionally, the method is applied to the synthesis of tamoxifen.

Chapter 3 details a nickel-catalyzed cross-electrophile coupling reaction of benzylic esters and aryl halides. Both inter- and intramolecular variants proceed under mild reaction conditions. A range of heterocycles and functional groups are tolerated under the reaction conditions. Additionally, the first example of a stereospecific cross-electrophile coupling of a secondary benzylic ester is described.

Chapter 4 presents secondary benzylzinc reagents generated from 2-pyridylcarbinols using a nickel catalyst and diethylzinc. Substrates are activated in situ using a chlorophosphate reagent. Quenching the organozinc reagents allows for facile deoxygenation of 2-pyridylcarbinols in a one-pot reaction with straightforward incorporation of a deuterium label from deuteromethanol. An intramolecular conjugate addition of a secondary benzylzinc reagent with an α , β -unsaturated ester is also demonstrated.

Chapter 1

Nickel-Catalyzed Stereospecific Suzuki Cross-Coupling: A Novel Approach to Optically Enriched Triarylmethanes

1.1 Introduction

Triarylmethanes are a diverse class of compounds, from their early uses as precursors to dyes, their utility has since expanded to include a range of chemical and biological applications (Figure 1.1).¹ For example, triarylmethanes have been used as compounds for live cell imaging, selective sensors for metal ions, anticancer and antitubercular agents, potassium ion channel blockers, and even have applications to material science.^{1,2} Hence, the synthesis of triarylmethanes is a rich and growing area of research. Given the wide ranging applications of triarylmethanes, many methods have emerged to access racemic mixtures;³ however, methods that address their asymmetric synthesis have notably been limited.⁴ In this Chapter, an account of the development of a stereospecific nickel-catalyzed Suzuki cross-coupling that can access a library of enantioenriched triarylmethanes is described.

¹ (a) Nambo M.; Crudden C. M. ACS Catal. 2015, 5, 4734. (b) Mondal, S.; Panda, G. RSC Adv., 2014, 4, 28317.

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² Physical properties of triarylmethanes: (a) Breslow, R.; Chu, W. J. Am. Chem. Soc. **1973**, 95, 411. (b) Finocchiaro, P.; Gust, D.; Mislow, K. J. Am. Chem. Soc. **1974**, 96, 3198. (c) Duxbury, D. F. Chem. Rev. **1993**, 93, 381.

³ (a) Zhang, J.; Bellomo, A.; Creamer, A. D.; Dreher, S. D.; Walsh, P. J. J. Am. Chem. Soc. **2012**, 134, 13765. (b) Yu, J.-Y.; Kuwano, R. Org. Lett. **2008**, 10, 973. (c) Molander, G.A.; Elia, M. D. J. Org. Chem. **2006**, 71, 9198. (d) Li, Y.-Z.; Li, B.-J.; Lu, X.-Y.; Lin, S.; Shi, Z.-J. Angew. Chem. Int. Ed. **2009**, 48, 3817. (e) Iovel, I.; Mertins, K.; Kischel, J.; Zapf, A.; Beller, M. Angew. Chem. Int. Ed. **2005**, 44, 3913. (f) For a representative Freidel–Crafts strategy, see: Esquivias, J.; Arrayás, R. G.; Carretero, J. C. Angew. Chem. Int. Ed. **2006**, 45, 629.

⁴ (a) Shi, B.-F.; Maugel, N.; Zhang, Y.-H.; Yu, J.-Q. Angew. Chem. Int. Ed. **2008**, 47, 4882. (b) Sun, F.-L.; Zheng, X.-J.; Gu, Q.; He, Q.-L.; You, S.-L. Eur. J. Org. Chem. **2010**, 47.

Figure 1.1. Examples of biologically active triarylmethanes



The mechanisms of alkyl cross-coupling reactions are hardwired with implications for the stereochemical outcome at the reactive centers.⁵ Simple changes to the reaction conditions do not typically perturb the inherent bias for racemization, retention, or inversion at the reactive centers. For example, palladium-catalyzed reactions of alkyl electrophiles are typically stereospecific and proceed with inversion at the stereogenic center,^{6,7} while nickel-catalyzed reactions of alkyl halides proceed with racemization at the electrophilic carbon⁸ and judicious use of a chiral catalyst permits stereoconvergent reactions.⁹ Overcoming the intrinsic preference, such that a reaction that typically proceeds with inversion at the stereogenic center can proceed with retention is quite unusual, and requires a significant change to the mechanism of the transformation. For stereospecific reactions, special cases using α -chiral *transmetallating agents* have been reported where modification of reaction conditions or substrate structure can effect a switch in the sense of

⁵ Jana, R.; Pathak, T. P.; Sigman, M. S. Chem. Rev. **2011**, 111, 1417.

⁶ (a) Lau, K. S. Y.; Fries, R. W.; Stille, J. K. *J. Am. Chem. Soc.* **1974**, *96*, 4983. (b) Netherton, M. R.; Fu, G. C. *Angew. Chem. Int. Ed.* **2002**, *41*, 3910. (c) Legros, J.-Y.; Toffano, M.; Fiaud, J.-C. *Tetrahedron* **1995**, *51*, 3235. (d) Rodriquez, N.; de Arellano, C. R.; Asensio, G.; Medio-Simon, M. *Chem. Eur. J.* **2007**, *13*, 4223. (e) Lopez-Perez, A.; Adrio, J.; Carretero, J. C. *Org. Lett.* **2009**, *11*, 5514. (f) He, A.; Falck, J. R. *J. Am. Chem. Soc.* **2010**, *132*, 2524. (g) Rudolph, A.; Rackelmann, N.; Lautens, M. *Angew. Chem. Int. Ed.* **2007**, *46*, 1485.

⁷ Pd-catalyzed *allylic* substitutions can occur with inversion or retention, depending on the nucleophile. See: Trost, B. M.; VanVranken, D. L. *Chem. Rev.* **1996**, *96*, 395.

⁸ Stille, J. K.; Cowell, A. B. J. Organomet. Chem. 1977, 124, 253.

⁹ (a) Saito, B.; Fu, G.C. J. Am. Chem. Soc. **2008**, 130, 6694. (b) Wilsily, A.; Tramutola, F.; Owston, N. A.; Fu, G. C. J. Am. Chem. Soc. **2012** 134, 5794, and cited therein. (c) Glorius, F. Angew. Chem. Int. Ed. **2008**, 47, 8347.

absolute configuration.¹⁰ Transmetallation typically occurs with retention at the stereogenic center;^{11,12} select examples that proceed with inversion have been reported.¹³ In seminal contributions, Hiyama demonstrated that palladium-catalyzed couplings of alkylsilanes could proceed with retention or inversion, depending on the reaction conditions.^{14,15} Recently, the Suginome group has developed stereodivergent reactions of α -(acetylamino)benzylboronic esters that are controlled by choice of additive to afford, selectively, either retention or inversion (Scheme 1.1a).

Scheme 1.1. Control of product stereochemistry in stereospecific reactions



¹⁰ For a discussion, see: Molander, G. A.; Wisniewski, S. R. J. Am. Chem. Soc. 2012, 134, 16856.

¹¹ For labeling studies, see: (a) Bock, P. L.; Boschetto, D. J.; Rasmussen, J. R.; Demers, J. P.; Whitesides, G. M. *J. Am. Chem. Soc.* **1974**, *96*, 2814. (b) Ridgway, B. H.; Woerpel, K. A. *J. Org. Chem.* **1998**, *63*, 458. (c) Matos, K.; Soderquist, J. A. *J. Org. Chem.* **1998**, *63*, 461. (d) Taylor, B. L. H.; Jarvo, E. R. *J. Org. Chem.* **2011**, *76*, 7573.

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¹⁴ (a) Hatanaka, Y.; Hiyama, T. J. Am. Chem. Soc. **1990**, 112, 7793. (b) Hiyama, T. J. Organomet. Chem. **2002**, 653, 58.

¹⁵ For enantiodivergent reactions of alkyllithium reagents: Stymiest, J. L.; Bagutski, V.; French, R. M.; Aggarwal, V. M. *Nature*, **2008**, *456*, 778.

In this Chapter, we demonstrate catalyst control of the stereochemical course with respect to the *electrophilic* partner in a cross-coupling reaction. Stereospecific nickel-catalyzed crosscoupling reactions of benzylic alcohol derivatives typically proceed with inversion at the electrophilic carbon.^{16,17} Nickel-catalyzed cross-coupling of benzylic esters where the achiral ligand structure dictates if the reaction proceeds with retention or inversion (Scheme 1.1b). Use of SIMes, an N-heterocyclic carbene (NHC) ligand, affords inversion, while PCy₃ gives retention. To the best of our knowledge, these results constitute the first cross-coupling reactions of alkyl electrophiles that proceed via two distinct stereospecific mechanistic pathways to provide either retention or inversion at the electrophilic carbon.

In previous work, we established the synthesis of enantioenriched triarylmethanes by stereospecific nickel-catalyzed cross-coupling of ethers with aryl Grignard reagents.^{16b} As part of our ongoing interest in developing nickel-catalyzed stereospecific reactions of alkyl electrophiles, we chose to examine cross-coupling reactions of arylboronic esters for triarylmethane synthesis. The functional group tolerance and availability of a wide range of boronic esters makes them attractive coupling partners.

1.2 Development of a Stereospecific Cross-Coupling Reaction of Arylboronic Esters

We began by examining a range of benzylic alcohol derivatives (Table 1.1). Our initial reaction conditions resulted in a modest conversion of carbonate (*S*)-1.4 and low enantiospecificity (es) (entry 1).¹⁸ To our surprise, in contrast to the Kumada coupling, the product, (*R*)-1.6, results

¹⁶ (a) Taylor, B. L. H.; Swift, E. C.; Waetzig, J. D.; Jarvo, E. R. J. Am. Chem. Soc. 2011, 133, 389. (b) Taylor, B. L. H.; Harris, M. R.; Jarvo, E. R. Angew. Chem. Int. Ed. 2012, 51, 7790. (c) Greene, M. A.; Yonova, I. M.; Williams, F. J.; Jarvo, E. R. Org. Lett. 2012, 14, 4293. (d) For a review, see: Taylor, B. L. H.; Jarvo, E. R. Synlett, 2011, 19, 2761.
¹⁷ For recent studies of the stereochemical course of nickel-catalyzed reactions of epoxides and aziridines, see: (a) Beaver, M. G.; Jamison, T. F. Org. Lett. 2011, 13, 4140. (b) Sylvester, K. T.; Wu, K.; Doyle, A. G. J. Am. Chem. Soc. 2012, 134, 9541. (c) Lin, B. L.; Clough, C. R.; Hillhouse, G. L. J. Am. Chem. Soc. 2002, 124, 2890.
¹⁸ es: Denmark, S. E.; Vogler, T. Chem.-Eur. J. 2009, 15, 11737.

from *retention* at the electrophilic carbon. An improvement to 43% es was observed when the solvent was changed from toluene to THF (entry 2). Alcohol additives further improved the yield and stereochemical fidelity of the reaction, with *n*-BuOH providing the highest es, 87% (entry 4). More sterically encumbered alcohols provided more modest improvements, while water and the electron-deficient alcohol trifluoroethanol proved detrimental to the reaction (entries 3, 5, and 7). The enantiospecificity of the reaction showed a marked dependence on the identity of the leaving group. While the use of pivalate (*S*)-1.8 in the cross-coupling reaction resulted in lower enantiomeric excess of the product (entry 8), the benzoate and carbamate derivatives (*S*)-1.9 and (*S*)-1.4 showed a significant increase in product ee, providing 91 and 95% es, respectively (entries 8, 10, and 12). An additional small improvement in yield and es resulted from using a 1:1 mixture of THF/toluene as the solvent (c.f. entries 12 and 15).



		Ċ,	OR	Ph + P-MeO (2	Ne Me B C ₆ H ₄ equiv)		Ni(cod) ₂ (Lig <i>t</i> -BuOK additive rt,	(10 m and (2 eq (3 eq 20 h	ol %) → uiv) uiv)	(<i>R</i>)-1	OMe Ph	or (S)-1	.6	e] ⊃h	
Entry	R	ligand ^a	solvent	additive 9	% yield ^b	es ^c	retention/ inversion	Entry	R	ligand ^a	solvent	additive	% yield ^b	es ^c	retention/ inversion
1	.	PCy ₃	PhMe	none	46	7	retention	10	$\frac{1}{2}$	PCy ₃	THF	<i>n</i> -BuOH	57	91	retention
2	メヘ _{Ot-B}	u _{PCy3}	THF	none	53	43	retention	¹¹ (\$ S)-1.9	SIMes	THF	<i>n</i> -BuOH	83	>99	inversion
3	• •	PCy ₃	THF	H ₂ O	74	10	retention	12	0	PCy ₃	THF	<i>n</i> -BuOH	62	 95	retention
4		PCy ₃	THF	<i>n</i> -BuOH	76	87	retention	13	$\frac{1}{2}$	PCy ₃	THF/PhMe	none	67	35	retention
5		PCy ₃	THF	<i>i</i> -PrOH	46	78	retention	14	(5)-1.4	SIMes	THF/PhMe	none	82	92	inversion
6		PCy ₃	THF	<i>t</i> -BuOH	55	43	retention	15	(0)	PCy ₃	THF/PhMe	<i>n</i> -BuOH	88	99	retention
7		PCy ₃	THF	F ₃ CCH ₂ OH	< 5	na	retention	16		SIMes	THF/PhMe	<i>n</i> -BuOH	84	99	inversion
8	о Ъ	PCy ₃	THF	<i>n</i> -BuOH	53	76	retention	^a PCy	3 (20 mol %	b), SIMes ((11 mol %).	^b Isolated	yield aft	er col	umn
9	メン _{t-Bu} (S)-1.8	SIMes	THF	<i>п</i> -ВиОН	60	77	inversion	chror	matography	. ^c Enantios	pecificity (e	s)=ee _{produ}	_{ict} /ee _{start}	ing mat	_{erial} x100%.

We examined other ligands¹⁹ under the reaction conditions and found that the NHC ligand SIMes²⁰ afforded comparable yields and enantiospecificity of **1.6**, however, the major product was the (*S*)-enantiomer, resulting from *inversion* at the electrophilic carbon.²¹ Catalyst-control of the stereochemical outcome of the reaction was consistent across the range of esters and carbamates that we examined: PCy₃ and SIMes reliably afforded opposite enantiomers of product (entries 8–11, 15 and 16).²² Under the optimal reaction conditions, addition of *n*-BuOH was found to improve stereochemical fidelity when using either ligand (c.f. entries 13–16).

1.3 Investigation of Scope in Aryl Boronic Ester

Having optimized reaction conditions for the stereospecific synthesis of either enantiomer of product, we turned our attention to the scope of the reaction with respect to the boronic ester (Table 1.2). Electron-donating and withdrawing substituents on the arylboronic ester are well tolerated under the reaction conditions (entries 1–6). Reaction conditions are mild and allow for broad functional group tolerance. Boronic esters containing ketone, free alcohol and carbamate functional groups all undergo cross-coupling in good yield and es (entries 9–13). Heterocyclic boronic esters including pyrimidine, furan, and indole undergo smooth cross-coupling (entries 14– 19). The reaction conditions developed for the formation of either enantiomer of **1.6** are general across the range of boronic esters that we examined: of 20 examples, 18 provide high es. Therefore, by choosing the appropriate ligand, PCy₃ or SIMes, either enantiomer of a given product can be obtained from the same enantiomer of starting material.

¹⁹ For results with other ligands, see the Experimental Details (table 1.4)

²⁰ SIMes = 1,3-Bis(2,6-diisopropylphenyl)-4,5-di-hydroimidazoliumtetrafluoroborate

²¹ Comparison of NHC to PR₃: (a)Clavier, H.; Nolan, S. P. Chem. Commun. 2010, 46, 841. (b)Dorta, R., Stevens, E.

D., Scott, N. M., Costabile, C., Cavallo, L., Hoff, C. D., Nolan, S. P. J. Am. Chem. Soc. **2005**, *127,2485*. (c)Kelly, R. A., III; Clavier, H.; Giudice, S.; Scott, N. M.; Stevens, E. D.; Bordner, J.; Samardjiev, I.; Hoff, C. D.; Cavallo, L.; Nolan, S. P. Organometallics **2008**, *27*, 202.

 $^{^{22}}$ Changing PCy₃ loading from 20 mol % to 11 mol % does not affect the stereochemical outcome; see the Experimental Details.

Table 1.2. Investigation of Scope with respect to Aryl Boronic Ester^a



^aAll data are average of two experiments unless otherwise indicated. ^bPCy3 (20 mol %), SIMes (11 mol %).

^oIsolated yield after column chromatography. ^dDetermined by chiral SFC chromatography. ^eData obtained from a single experiment.

1.4 Stereochemical Course of the Cross-Coupling Reaction

To demonstrate the stereochemical course of the reaction with either catalyst, we utilized X-ray crystallographic analysis. We first showed that the reaction proceeds with inversion when SIMes is used as the ligand by comparison of the optical rotation of the triarylmethane (R)-1.12 with the literature value (Scheme 1.2a).^{23,24} This product corresponds to net inversion at the benzylic carbon during the Suzuki–Miyaura cross-coupling reaction. Previous work in our group demonstrated that cross-coupling of aryl Grignard reagents and benzylic ethers also results in inversion at the benzylic carbon and assigned the absolute configuration of (R)-1.12 based on X-ray crystallographic analysis. The stereochemical course of the Suzuki–Miyaura cross-coupling

²³ For optical rotation data for **1.10**, see: (a) Yamamoto, Y.; Kurihara, K.; Miyaura, N. Angew. Chem. Int. Ed. **2009**, 48, 4414. (b) Shannon, J.; Bernier, D.; Rawson, D.; Woodward, S. Chem. Commun. **2007**, 3945. (c) Tjosaas, F.; Anthonsen, T.; Jacobsen, E. E. ARKIVOC **2008**, (6), 8190.

²⁴ For characterization data for **1.12** including optical rotation and X-ray crystallographic data, see: Taylor, B. L. H.; Harris, M. R.; Jarvo, E. R. *Angew. Chem., Int. Ed.* **2012**, *51*, 7790.

reaction when PCy_3 is used as the ligand provides the opposite enantiomer. Based on the absolute configuration of **(S)-1.14**, the reaction proceeds with retention at the benzylic carbon (Scheme 1.2c). *Scheme 1.2.* Demonstration of stereochemical course of cross-coupling reaction



A mechanistic model for obtaining either product of retention or inversion is shown in Scheme 1.3b. We propose that when PCy₃ ligand is employed, the carbamate leaving group coordinates to the nickel catalyst. This coordination event directs the oxidative addition of the catalyst to the substrate, resulting in oxidative addition with retention. Transmetallation followed by reductive elimination with retention leads to formation of product with overall retention. In contrast, when SIMes is used as the ligand, coordination of the carbamate to the catalyst prior to oxidative addition does not occur. Without precoordination of the nickel catalyst to the leaving group, oxidative addition proceeds with inversion, ultimately affording triarylmethane products with overall inversion. *Scheme 1.3.* Current mechanistic model for stereodivergent pathways using PCy₃ or SIMes a) Catalytic cycle for Ni-SiMes catalyst; oxidative addition with *inversion* of configuration



b) Catalytic cycle for Ni-PCy₃ catalyst; oxidative addition with *retention* of configuration



1.5 Scope of the Oxidative Addition Partner

We set as our goal the cross-coupling of oxidative addition partners that do not include a naphthalene moiety. Electrophiles that only have simple benzene rings are typically less reactive in cross-coupling reactions,^{16c} and were not competent for triarylmethane synthesis via Kumada coupling.^{16b} Indeed, neither the corresponding carbamates nor the use of PCy₃ as ligand provide acceptable yields of product. However, benzhydryl pivalates undergo smooth cross-coupling under our optimized reaction conditions when SIMes is used as the ligand (Table 1.3). Efficient cross-coupling was achieved for pivalates with a range of arylboronic esters, including an indoleboronic ester (entries 1–3). Heterocyclic functionality was also tolerated on the electrophile: furan- and benzodioxane- substituted pivalates couple in good yield and excellent es (entries 4 and 5).

R	r-Bu OOO Ph	Me Me Me Ni(cod) ₂ SIMes (<i>n</i> -BuOH (2 equiv.) THF:PhMe	(10 mol 11 mol ⁰ (3 equi (2 equi (1:1), rt	%) %) V) V) R , 24 h		Ar Ph
Entry	R'	Ar	yield (%) ^b	SM ee (%) ^c	product ee (%) ^c	es (%)
1	Ph	<i>p</i> -MeOC ₆ H ₄	85	96	84	88
2	Ph	<i>p</i> -(Me₂N)C ₆ H₄	75	82	79	96
3	Ph		66	96	96	>99
4	J.	z,	80	93	87	94
5		5	60	93	93	99

Table 1.3. Scope of Substrates Bearing Non-extended Arenes^a

^aAll data are average of two experiments. ^bIsolated yield after column chromatography. ^cDetermined by chiral SFC chromatography.

1.6 Conclusions

In summary, we have developed a nickel-catalyzed Suzuki-Miyaura cross-coupling reaction for the synthesis of enantioenriched triarylmethanes. Reactions proceed with high stereochemical fidelity. Achiral ligand identity controls if the reaction proceeds with inversion or retention at the electrophilic carbon; therefore either enantiomer of product can be formed from a single enantiomer of starting material. This method expands the range of triarylmethanes that may be prepared in enantioenriched form, as simple benzhydryl pivalates and a variety of functionalized arylboronic esters, including heterocyclic compounds, can be used in the reaction.

1.7 Experimental Details

General Procedures

All reactions were carried out under an atmosphere of N2, or Ar when noted. All glassware was oven- or flame-dried prior to use. Tetrahydrofuran (THF), diethyl ether (Et₂O), dichloromethane (CH₂Cl₂), and toluene (PhMe) were degassed with Ar and then passed through two 4 x 36 inch columns of anhydrous neutral A-2 alumina (8 x 14 mesh; LaRoche Chemicals; activated under a flow of argon at 350 °C for 12 h) to remove H₂O. All other solvents utilized were purchased "anhydrous" commercially, or purified as described. ¹H NMR spectra were recorded on Bruker DRX-400 (400 MHz ¹H, 100 MHz ¹³C, 376.5 MHz ¹⁹F), GN-500 (500 MHz ¹H, 125.7 MHz ¹³C), or CRYO-500 (500 MHz ¹H, 125.7 MHz ¹³C) spectrometers. Proton chemical shifts are reported in ppm (δ) relative to internal tetramethylsilane (TMS, δ 0.00). Data are reported as follows: chemical shift (multiplicity [singlet (s), broad singlet (br s), doublet (d), doublet of doublets (dd), triplet (t), doublet of triplets (dt), doublet of doublet of triplets (ddt), triplet of triplets (tt), quartet (q), multiplet (m)], coupling constants [Hz], integration). Carbon chemical shifts are reported in ppm (δ) relative to TMS with the respective solvent resonance as the internal standard (CDCl₃, δ 77.16 ppm). Unless otherwise indicated, NMR data were collected at 25 °C. Infrared (IR) spectra were obtained on a Perkin-Elmer Spectrum 1000 FT-IR Systems and are reported in terms of frequency of absorption (cm⁻¹). Analytical thin-layer chromatography (TLC) was performed using Silica Gel 60 F₂₅₄ precoated plates (0.25 mm thickness). Visualization was accomplished by irradiation with a UV lamp and/or staining with KMnO4, ceric ammonium molybdate (CAM), or p-anisaldehyde (PAA) solutions. Flash chromatography was performed using Silica Gel 60 (170-400 mesh) from Fisher Scientific. Melting points (m.p.) were obtained using a Mel-Temp melting point apparatus and are uncorrected. Optical rotations were measured on a Rudolph Research

Analytical Autopol IV Automatic Polarimeter. SFC determinations of enantiopurity were performed on a Berger Analytical instrument using a DaicelTM Chiralpak® column (OD-H, OJ-H, or AD-H; 100 bar, 50 °C, 215 nm). High resolution mass spectrometry was performed by the University of California, Irvine Mass Spectrometry Center. Boronic esters were prepared from the corresponding boronic acids and 2,2-dimethylpropane-1,3-diol.²⁵ Boronic acids were generously donated from Frontier, stored at 4 °C, and used as received. 1,8-bis(1,5-cyclooctadiene)nickel was purchased from Strem, stored in a glovebox freezer (-20 °C) under an atmosphere of N₂, and used as received. Tricyclohexylphosphine (PCy₃), (1,3-Bis(2,6-diisopropylphenyl)-4,5dihydroimidazoliumtetrafluoroborate (SIMes), and tris(di-benzylideneacetone)dipalladium (Pd₂(dba)₃) were purchased from Strem, stored in a glovebox, and used as received. All other reagents were purchased commercially and used as received.

²⁵ Tivola, B. P.; Deagostino, A.; Prandi, C.; Venturello, P. Org. Lett. 2002, 4, 1275.

Synthesis and Characterization of Substrates

A. Representative procedure for synthesis of diarylmethyl alcohols.



Rac-1.15 In a flame-dried round-bottom flask, to a solution of biphenyl-4-carboxaldehyde (1.04 g, 5.68 mmol, 1.00 equiv) in THF (10 mL) was added phenylmagnesium bromide (0.71 M in THF, 12 mL, 8.5 mmol, 1.5 equiv). After stirring at room temperature for 4 h, saturated ammonium chloride (10 mL) was added and the reaction was extracted with EtOAc (3 x 5 mL). The combined organic layers were washed with brine (3 x 5 mL), dried over MgSO₄, and concentrated in vacuo to afford *rac*-1.15 as a white solid (1.2 g, 4.7 mmol, 55%). Analytical data are consistent with the values listed for (*S*)-1.15 (vide infra).



Rac-1.10 Using the representative procedure A outlined above, the following amounts of reagents were used: 2-naphthaldehyde (6.24 g, 40.0 mmol, 1.00 equiv), phenylmagnesium bromide (58 mL, 0.83 M in THF, 48 mmol, 1.2 equiv), and THF (25 mL). The reaction mixture was purified by silica gel flash column chromatography (5–20% EtOAc/hexanes) to afford the product as a white solid (6.74 g, 28.7 mmol, 72%). Analytical data are consistent with the values listed for (*S*)-1.10 (vide infra).



Rac-1.16 Using the representative procedure A outlined above, the following amounts of reagents were used: 4-bromobenzaldehyde (1.85 g, 10.0 mmol, 1.00 equiv), phenylmagnesium bromide (7.0 mL, 1.7 M in THF, 12 mmol, 1.2 equiv), and THF (10 mL). The crude reaction mixture was purified by flash chromatography (5–20% EtOAc/hexanes) to afford the product as a white solid (1.92 g, 7.29 mmol, 73%). Analytical data is consistent with the values listed below for *(S)-1.16*.

B. Representative procedure for enantioselective synthesis of diarylmethyl alcohols by asymmetric arylation.



Enantioenriched alcohols were prepared according to a modified procedure of Braga and coworkers.²⁶

(*S*)-1.15 To a solution of phenylboronic acid (0.732 g, 6.00 mmol, 2.40 equiv) in toluene (10 mL) was added diethylzinc (18 mL, 18 mmol, 1.0 M in toluene, 7.2 equiv), and the solution was allowed to stir at 60 °C for 12 h. Upon cooling to room temperature, (*S*)-(1-tritylaziridin-2-yl)diphenylmethanol (0.084 g, 0.06 mmol, 0.01 equiv) was added as a solution in toluene (5 mL) and the reaction mixture was allowed to stir for 10 minutes before the addition of a solution of biphenyl-4-carboxaldehyde (0.456 g, 2.50 mmol, 1.00 equiv) in toluene (5 mL). After stirring 12 h at room temperature, 1 N hydrochloric acid (10 mL) was added and the product was extracted

²⁶ Braga, A. R.; Paixao, M. W.; Westeman, B.; Schneider, P. H.; Wessjohan, L.A. J. Org. Chem. 2008, 73, 2879.

with EtOAc (3 x 10 mL). The combined organics were washed with brine (10 mL), dried over MgSO₄, and concentrated in vacuo. The product was purified by flash column chromatography (0–1% EtOAc/benzene) and then recrystallized from hexanes and EtOAc to upgrade the ee (0.488 g, 1.85 mmol, 75% yield, 96% ee). **TLC R**_f = 0.2 (benzene); **m.p.** = 90–92 °C; ¹**H NMR** (400 MHz, CDCl₃) δ 7.55 (m, 4H), 7.40 (m, 6H), 7.32 (m, 4H), 5.81 (s, 1H), 2.32 (d, *J* = 2.8, 1H); ¹³**C NMR** δ (100 MHz, CDCl₃) δ 143.8, 142.9, 140.9, 140.6, 128.9, 128.7, 127.8, 127.4, 127.38, 127.2, 127.1, 126.7, 76.1; **IR** (neat) 3361, 3029, 1408, 1006, 763 cm⁻¹; **HRMS** (TOF MS ES+) *m/z* calcd for C₁₉H₁₆O (M + Na)⁺ 283.1099, found 283.1110; **[a]**²³**p** +4.72 (*c* 1.10, CHCl₃); **SFC** analysis (AD-H, 15% IPA, 3 mL/min) indicated 96% ee: t_R (major) = 18.9 minutes, t_R (minor) = 20.5 minutes.



(*S*)-1.10 Using the representative procedure B outlined above, the following amounts of reagents were used: phenylboronic acid (0.732 g, 6.00 mmol, 2.4 equiv), diethylzinc (18 mL, 18 mmol, 1.0 M in toluene), (*S*)-diphenyl(1-tritylaziridin-2-yl)methanol (116 mg, 0.250 mmol, 0.100 equiv), and 2-naphthaldhyde (0.390 g, 2.50 mmol, 1.00 equiv). The product was purified by flash chromatography (10–20% EtOAc/hexanes) to afford the product as a white solid (0.608 g, 2.59 mmol, 93%, 89% ee). The product was then recrystallized from hexanes to upgrade the ee (99% ee). Analytical data is consistent with literature values.**Error! Bookmark not defined.** ¹H NMR (400 MHz, CDCl₃) δ 7.91 (s, 1H), 7.82 (dt, *J* = 9.2, 2.6 Hz, 2H), 7.80 (d, *J* = 8.2 Hz, 1H), 7.49–7.40 (m, 5H), 7.35 (t, *J* = 7.0 Hz, 2H), 7.29 (dt, *J* = 7.4, 1.5 Hz, 1H), 6.02 (d, *J* = 3.5 Hz, 1H), 2.29 (d, *J* = 3.5 Hz, 1H); [**a**]²³**b** +7.8 (*c* 0.92, CHCl₃), literature [**a**]²⁰**b** +11.2 (*c* 0.83, CHCl₃); **SFC** analysis (OD-H, 20% 2-propanol, 3 mL/min) indicated >99% ee: t_R (major) = 6.4 min, t_R (minor) = 7.3 min.



(*S*)-1.16 Using the representative procedure B outlined above, the following amounts of reagents were used: phenylboronic acid (0.732 g, 6.00 mmol, 2.4 equiv), diethylzinc (18 mL, 18 mmol, 1.0 M in toluene), (*S*)-(1-tritylaziridin-2-yl)diphenylmethanol (116 mg, 0.250 mmol, 0.100 equiv), and 4-bromobenzaldehyde (0.463 g, 2.50 mmol, 1.00 equiv). The product was purified by flash chromatography (10–20% EtOAc/hexane) to afford the product as a white solid (0.608 g, 2.31 mmol, 93%, 92% ee). The product was then recrystallized from hexanes to yield higher enantiopurity (96% ee). Analytical data is consistent with literature values.²⁷ ¹H NMR (500 MHz, CDCl₃) δ 7.44 (d, *J* = 8.4 Hz, 2H), 7.36–7.30 (m, 4H), 7.29–7.25 (m, 1H), 7.23 (d, *J* = 8.4 Hz, 2H), 5.76 (d, *J* = 3.3 Hz, 1H), 2.34 (d, *J* = 3.3 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 143.5, 142.8, 131.7, 128.8, 128.3, 128.0, 126.6, 121.5, 75.8; [**a**]²³**b** +17.5 (*c* 1.65, CHCl₃); **SFC** analysis (AD-H, 10% IPA, 2.5 mL/min) indicated 96% ee: t_R (major) = 10.4 minutes, t_R (minor) = 9.8 minutes. *C. Representative procedure for the Suzuki cross-coupling of aryl bromide* (**S**)-1.17 with aryl

boronic acids.



(S)-1.17 The product was prepared according to a modified procedure by Fu and co-workers.²⁸ Tris(dibenzylideneacetone)dipalladium (55 mg, 0.060 mmol, 0.030 equiv) and tricyclohexylphosphine (39 mg, 0.14 mmol, 0.070 equiv) were weighed out into a flame dried two neck, round bottom flask inside a glovebox. The flask was fitted with septa, removed from the

²⁷ Wu, X.; Liu, X.; Zhao, G. Tetrahedron: Asymmetry 2005, 16, 2299.

²⁸ Kudo, N.; Perseghini, M.; Fu, G. C. Angew. Chem. Int. Ed. 2006, 45, 1282.

glovebox, and 3-furanboronic acid (0.262 g, 2.20 mmol, 1.10 equiv), (*S*)-1.16 (0.526 g, 2.00 mmol, 1.00 equiv), aqueous potassium phosphate (2.7 mL, 3.4 mmol, 1.3 M in H₂O, 1.7 equiv) and dioxane (6 mL) were added. The reaction flask was fitted with a reflux condenser and heated to 95 °C for 16 h. After cooling, the solvent was removed under reduced pressure. The resultant residue was purified by flash column chromatography (10–20% EtOAc/hexane) to afford (*S*)-1.17 as a yellow solid (0.437 g, 1.75 mmol, 87%, 97% ee). TLC $\mathbf{R}_{\mathbf{f}} = 0.2$ (4:1 hexane/EtOAc); m.p. = 97–99 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.70 (s, 1H), 7.45 (s, 2H), 7.43 (s, 1H), 7.37 (t, *J* = 8.2 Hz, 3H), 7.33 (t, *J* = 7.3 Hz, 3H), 7.27 (d, *J* = 7.4 Hz, 1H), 6.67 (s, 1H), 5.82 (s, 1H), 2.32 (s, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 143.9, 143.8, 142.7, 138.6, 131.8, 128.7, 127.8, 127.1, 126.7, 126.2, 126.1, 108.9, 76.1; IR (neat) 3279, 1160, 1012, 780, 699 cm⁻¹; HRMS (TOF MS ES+) *m* / *z* calcd for C₁₇H₁₄O₂ (M + Na)⁺ 273.0891, found 273.0883; [*a*]²⁹D –37.3 (*c* 1.00, CHCl₃); SFC analysis (OD-H, 13% IPA, 2.5 mL/min) indicated 97% ee: t_R (major) = 12.9 minutes, t_R (minor) = 14.7 minutes.



(*S*)-1.18 Using representative procedure C outlined above, the following amounts of reagents were used: tris(dibenzylideneacetone)dipalladium (28 mg, 0.030 mmol, 0.030 equiv), tricyclohexylphosphine (20 mg, 0.07 mmol, 0.070 equiv), 1,4-benzodioxane-6-boronic acid (0.198 g, 1.10 mmol, 1.10 equiv), (*S*)-1.16 (0.263 g, 1.00 mmol, 1.00 equiv), aqueous potassium phosphate (1.4 mL, 1.7 mmol, 1.3 M in H₂O, 1.7 equiv) and dioxane (3 mL). The product was purified by flash column chromatography (10–30% EtOAc/hexane) to afford (*S*)-1.18 as a brown solid (0.296 g, 0.929 mmol, 93%, 96% ee). TLC $\mathbf{R}_{\mathbf{f}} = 0.2$ (4:1 hexane/EtOAc); TLC $\mathbf{R}_{\mathbf{f}} = 0.3$ (30% EtOAc/hexanes); **m.p.** = 108–110 °C; ¹**H NMR** (500 MHz, CDCl₃) δ 7.48 (d, *J* = 7.7 Hz,

2H), 7.40 (q, J = 7.8 Hz, 4H), 7.34 (t, J = 7.2 Hz, 2H), 7.27 (d, J = 7.3 Hz, 1H), 7.08 (d, J = 1.9 Hz, 1H), 7.04 (dd, J = 8.5, 2.1 Hz, 1H), 6.90 (d, J = 8.2 Hz, 1H), 5.85 (s, 1H), 4.36 (s, 4H), 2.35 (s, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 143.9, 143.8, 143.3, 142.5, 140.0, 134.5, 128.7, 127.7, 127.1, 127.0, 126.6, 120.2, 117.7, 115.9, 76.2, 64.6, 64.5; **IR** (neat) 3550, 1494, 1304, 1284, 1070 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₁H₁₈O₃ (M + Na)⁺ 341.1154, found 341.1147; **[a]²⁹b** +3.1 (*c* 1.04, CHCl₃); **SFC** analysis (AD-H, 14% IPA, 2.5 mL/min) indicated 96%: t_R (major) = 6.9 minutes, t_R (minor) = 8.8 minutes.

Preparation of protected carbinols.



(*S*)-1.4 The product was prepared according to a modified procedure by Zhang and co-workers.²⁹ To a suspension of NaH (0.153 g, 6.37 mmol, 1.80 equiv) in DMF (3 mL) was added a solution of (*S*)-1.10 (0.823 g, 3.54 mmol, 1.00 equiv) in DMF (2 mL) at 0 °C. The mixture was stirred for 1 h before addition of neat 1-pyrollidinecarbonyl chloride (0.41 mL, 3.7 mmol, 1.1 equiv) at room temperature. After stirring for 3 h, the reaction was quenched with saturated aqueous ammonium chloride (6 mL), and the product was extracted with methylene chloride (3 x 10 mL). The combined organics were washed with brine (5 mL), dried over Na₂SO₄, and concentrated in vacuo. The product was purified by flash column chromatography (20% EtOAc/hexane) to afford (*S*)-1.4 as a white solid (0.963 g, 2.91 mmol, 83%, 94% ee): TLC R_f = 0.2 (20% EtOAc/hexanes); m.p. = 151–153 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.84 (s, 1H), 7.82–7.77 (m, 3H), 7.47–7.44 (m, 3H),

²⁹ DeKorver, K. A.; Johnson, W. L.; Zhang, Y.; Hsung, R. P.; Dai, H.; Deng, J.; Lohse, A. G.; Zhang, Y. S. J. Org. Chem. **2011**, *76*, 5092.
7.41 (d, J = 7.3 Hz, 2H), 7.33 (t, J = 7.3 Hz, 2H), 7.27 (d, J = 7.4 Hz, 1H), 7.00 (s, 1H), 3.55 (t, J = 6.7 Hz, 2H), 3.40 (t, J = 6.7 Hz, 2H), 1.90 (dt, J = 13.3, 6.7 Hz, 2H), 1.84 (dt, J = 13.3, 6.7 Hz, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 154.2, 141.4 138.7, 133.2, 133.0, 128.5, 128.4 128.3, 127.8, 127.7, 127.2, 126.3, 126.2, 126.1, 125.2, 77.4, 46.4, 46.0, 25.9, 25.0; IR (neat) 1690, 1412, 1102, 828, 765 cm⁻¹; HRMS (TOF MS ES+) m / z calcd for C₂₂H₂₁NO₂ (M + Na)⁺ 354.1470, found 354.1463; [α]²⁹D +45.9 (c 1.15, CHCl₃); SFC analysis (OD-H, 18% IPA, 2.5 mL/min) indicated 93% ee: t_R (major) = 7.1 minutes, t_R (minor) = 6.6 minutes.



(*S*)-1.9 The product was prepared according to a modified procedure by Hassner and co-workers.³⁰ To a 25 mL round bottom flask was added alcohol (*S*)-1.10 (0.175 g, 0.750 mmol, 1.00 equiv), and 4-(dimethylamino)pyridine (9.0 mg, 0.075 mmol, 0.10 equiv). The flask was evacuated and backfilled with nitrogen before addition of methylene chloride (6 mL), triethylamine (0.48 mL, 4.5 mmol, 6.0 equiv), and benzoyl chloride (0.18 mL, 1.5 mmol, 2.0 equiv). After stirring for 8 h, the reaction was quenched with 1 M HCl (6 mL), and the product was extracted with methylene chloride (3 x 10 mL). The combined organics were washed with brine (10 mL), dried over MgSO4, and concentrated in vacuo. The product was purified by flash column chromatography (5–10% EtOAc/hexane) to afford (*S*)-1.9 as a white solid (0.177 g, 0.523 mmol, 70%, 89% ee): TLC R_f = 0.4 (10% EtOAc/hexanes); m.p. = 91–93 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.26 (d, *J* = 7.9 Hz, 2H), 8.00 (s, 1H), 7.91–7.90 (m, 3H), 7.66 (t, *J* = 7.2 Hz, 1H), 7.61–7.53 (m, 6H), 7.45 (t, *J* = 7.1

³⁰ Basel, Y.; Hassner, A. J. Org. Chem. 2000, 65, 6368.

Hz, 2H), 7.39 (d, J = 7.2 Hz, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 165.7, 140.3, 137.7, 133.3, 133.2, 133.1, 130.3, 130.0, 128.7, 128.6, 128.3, 128.2, 127.8, 127.4, 126.44, 126.40, 126.3, 125.1, 77.7; **IR** (neat) 1712, 1259, 1108, 732, 700 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₄H₁₈O₂ (M + Na)⁺ 361.1205, found 361.1201; **[a]**²⁹D +10.0 (*c* 0.99, CHCl₃); **SFC analysis** (OD-H, 10.0% IPA, 2.5 mL/min) indicated 89% ee: t_R (major) = 6.5 minutes, t_R (minor) = 6.3 minutes.



(S)-1.7 The product was prepared according to a modified procedure by Hassner and co-workers.³⁰ To a 25 mL round bottom flask was added alcohol (S)-1.10 (0.234 g, 1.00 mmol, 1.00 equiv), and 4-(dimethylamino)pyridine (12 mg, 0.010 mmol, 0.10 equiv). The flask was evacuated and backfilled with nitrogen before addition of methylene chloride (8 mL), triethylamine (0.10 mL, 1.2 mmol, 1.2 equiv), and di-tert-butyl dicarbonate (0.228 g, 1.05 mmol, 1.05 equiv). After stirring for 8 h, the reaction was quenched with 1 M HCl (6 mL), and the product was extracted with methylene chloride (3 x 10 mL). The combined organics were washed with brine (10 mL), dried over MgSO₄, and concentrated in vacuo. The product was purified by flash column chromatography (5–10% EtOAc/hexane) to afford (S)-1.7 as a white solid (0.284 g, 0.849 mmol, 85%, 88% ee): TLC $\mathbf{R}_{f} = 0.4$ (9:1 hexane/EtOAc); m.p. = 90–92 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.87 (s, 1H), 7.83 (d, J = 7.0 Hz, 1H), 7.79 (d, J = 8.2 Hz, 2H), 7.48–7.45 (m, 2H), 7.43 (s, 1H), 7.41 (d, J = 8.3 Hz, 2H), 7.33 (t, J = 7.3 Hz, 2H), 7.27 (d, J = 7.7 Hz, 1H), 1.47 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 153.1, 140.1, 137.6, 133.2, 133.1, 128.7, 128.5, 128.3, 128.1, 127.8, 127.2, 126.4, 126.3, 126.0, 125.0, 82.7, 80.0, 27.9; **IR** (neat) 1742, 1270, 1251, 1150, 1081 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₂H₂₂O₃ (M + Na)⁺ 357.1467, found 357.1467; $[\alpha]^{29}D$ –19.3 (c 0.90, CHCl₃); SFC analysis (AD-H, 5% IPA, 3.0 mL/min) indicated 88% ee: t_R (major) = 5.6

minutes, t_R (minor) = 6.1 minutes.



(S)-1.8 The product was prepared according to a modified procedure by Hassner and co-workers.³⁰ To a 25 mL round bottom flask was added alcohol (S)-1.10 (0.281 g, 1.20 mmol, 1.20 equiv), and 4-(dimethylamino)pyridine (15 mg, 0.012 mmol, 0.10 equiv). The flask was evacuated and backfilled with nitrogen before addition of methylene chloride (8 mL), triethylamine (0.19 mL, 2.6 mmol, 2.2 equiv), and trimethylacetyl chloride (0.160 mL, 1.26 mmol, 1.05 equiv). After stirring for 8 h, the reaction was quenched with 1M HCl (6 mL), and the product was extracted with methylene chloride (3 x 10 mL). The combined organics were washed with brine (10 mL), dried over MgSO₄, and concentrated in vacuo. The product was purified by flash column chromatography (5-10% EtOAc/hexane) to afford (S)-1.8 as a white solid (0.334 g, 1.05 mmol, 88%, 82% ee): TLC $R_f = 0.5$ (10% EtOAc/hexanes); m.p. = 80–83 °C; ¹H NMR (500 MHz, $CDCl_3$) δ 7.82 (s, 2H), 7.79 (d, J = 8.8 Hz, 2H), 7.48–7.45 (m, 2H), 7.42 (d, J = 8.3 Hz, 1H), 7.38 $(d, J = 7.6 \text{ Hz}, 2\text{H}), 7.33 (t, J = 7.1 \text{ Hz}, 2\text{H}), 7.28 (d, J = 7.6 \text{ Hz}, 1\text{H}), 1.27 (s, 9\text{H}); {}^{13}C \text{ NMR} (125 \text{ Hz}, 120 \text{ Hz})$ MHz, CDCl₃) & 177.4, 140.6, 138.0, 133.2, 133.0, 128.6, 128.5, 128.3, 127.9, 127.8, 127.1, 126.4, 126.3, 126.1, 125.0, 76.8, 39.1, 27.3; **IR** (neat) 1721, 1276, 1148, 1123, 823 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₂H₂₂O₂ (M + Na)⁺ 341.1518, found 341.1526; $[\alpha]^{29}$ D -37.4 (c 1.18, CHCl₃); SFC analysis (AD-H, 5.0% IPA, 3.0 mL/min) indicated 82% ee: t_R (major) = 6.8 minutes, t_R (minor) = 7.1 minutes.



(S)-1.19 The product was prepared according to a modified procedure by Zhang and co-workers.²⁹ NaH (500 mg, 20.8 mmol, 4.00 equiv) was suspended in 40 mL of dry DMF and cooled to 0 °C. To this solution, alcohol (S)-1.15 (1.28 g, 4.92 mmol, 1.00 equiv) in dry DMF (10 mL) was added dropwise. The mixture was allowed to stir at 0 °C for 30 minutes after which pivaloyl chloride (4.3 mL, 35 mmol, 7.0 equiv) was added dropwise. The reaction was stirred at 0 °C for 1.5 hours then warmed to room temperature and stirred for 22 hours. The reaction was quenched by consecutive addition of water (5 x 2 mL) and stirring for 3 minutes. The reaction was diluted with more water (10 mL) and the organics were extracted with EtOAc (3 x 30 mL). The combined organic layers were dried over MgSO₄ and the solvent was removed under reduced pressure. The crude was purified by flash chromatography $(0-1\% \text{ Et}_2\text{O}/\text{petroleum ether})$ yielding (S)-1.19 as a white solid (1.56 g, 4.53 mmol, 92%). TLC $\mathbf{R}_{f} = 0.4$ (10% Et₂O:petroleum ether); m.p. = 109– 110 °C; ¹**H NMR** (400 MHz, CDCl₃) δ 7.54 (dd, J = 6.1, 1.9 Hz, 4H), 7.35 (m, 10H), 6.87 (s, 1H), 1.27 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.4, 140.8, 140.6, 139.8, 128.9, 128.7, 127.9, 127.8, 127.5, 127.40, 127.38, 127.2, 127.0, 76.5, 39.1, 27.3; **IR** (neat) 3029, 2974, 1722, 1275, 1138 cm⁻¹ ¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₄H₂₄O₂ (M + Na)⁺ 367.1674, found 367.1681; $[\alpha]^{23}$ D – 23.6 (c 1.09, CHCl₃); SFC analysis (AD-H, 10% IPA, 3 mL/min) indicated 96% ee: t_R (minor) = 4.0 minutes, t_R (major) = 6.4 minutes.



(S)-1.20 The product was prepared according to a modified procedure by Zhang and co-workers.²⁹ To a suspension of NaH (72 mg, 3.0 mmol, 2.0 equiv) in DMF (3 mL) was added a solution of (S)-1.17 (0.374 g, 1.50 mmol, 1.00 equiv) in DMF (2 mL) at 0 °C. The mixture was stirred for 1 h before addition of neat trimethylacetyl chloride (0.200 mL, 1.60 mmol, 1.05 equiv) at room temperature. After stirring for 3 h, the reaction was quenched with saturated aqueous ammonium chloride (6 mL), and the product was extracted with methylene chloride (3 x 10 mL). The combined organics were washed with brine (5 mL), dried over Na₂SO₄, and concentrated in vacuo. The product was purified by flash column chromatography (30% Et₂O/hexane) to afford (S)-1.20 as a pale yellow solid (0.427 g, 1.28 mmol, 85%, 93% ee): TLC $R_f = 0.2$ (4:1 hexane/Et₂O); m.p. = 105–108 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.70 (s, 1H), 7.45 (s, 2H), 7.43 (s, 1H), 7.35 (s, 4H), 7.33 (s, 2H), 7.29–7.26 (m, 1H), 1.26 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 177.4, 143.8, 140.6, 139.5, 138.7, 132.1, 128.6, 127.9, 127.6, 127.0, 126.2, 126.1, 108.9, 76.5, 39.0, 27.3; IR (neat) 1724, 1159, 1138, 757, 699 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₂H₂₂O₃ (M + Na)⁺ 357.1467, found 357.1475; [α]²⁹p -26.0 (c 1.25, CHCl₃); SFC analysis (OJ-H, 8% IPA, 2.0 mL/min) indicated 93% ee: t_R (major) = 3.6 minutes, t_R (minor) = 4.2 minutes.



(*S*)-1.21 The product was prepared according to a modified procedure by Zhang and co-workers.²⁹ To a suspension of NaH (35 mg, 1.4 mmol, 1.8 equiv) in DMF (3 mL) was added a solution of

(S)-1.18 (0.254 g, 0.800 mmol, 1.00 equiv) in DMF (2 mL) at 0 °C. The mixture was stirred for 1 h before addition of neat trimethylacetyl chloride (0.103 mL, 0.840 mmol, 1.05 equiv) at room temperature. After stirring for 3 h, the reaction was quenched with saturated aqueous ammonium chloride (6 mL), and the product was extracted with methylene chloride (3 x 10 mL). The combined organics were washed with brine (5 mL), dried over Na₂SO₄, and concentrated in vacuo. The product was purified by flash column chromatography (30% Et₂O/hexanes) to afford (*S*)-1.21 as a tan solid (0.232 g, 0.576 mmol, 73%, 94% ee): TLC R_f = 0.1 (20% Et₂O/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.48 (d, *J* = 8.2 Hz, 2H), 7.37–7.32 (m, 6H), 7.28 (d, *J* = 7.2 Hz, 1H), 7.08 (s, 1H), 7.04 (d, *J* = 8.4 Hz, 1H), 6.90 (d, *J* = 8.6 Hz, 1H), 6.84 (s, 1H), 4.27 (s, 4H), 1.27 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 177.4, 143.8, 143.4, 140.7, 140.2, 139.3, 134.4, 128.7, 127.9, 127.4, 127.0, 126.9, 120.2, 117.7, 115.9, 76.5, 64.6, 64.5, 39.0, 27.3; IR (neat) 1723, 1494, 1309, 1147, 1068 cm⁻¹; HRMS (TOF MS ES+) *m* / *z* calcd for C₂₆H₂₆O₄ (M + Na)⁺ 425.1729, found 425.1715; [*a*]²⁹**b** –20.3 (c 0.96, CHCl₃); SFC analysis (OD-H, 30% MeOH, 2.5 mL/min) indicated 94% ee: t_R (major) = 6.9 minutes, t_R (minor) = 8.8 minutes.

Procedures for Cross-Coupling Reactions

A. Procedure for the synthesis of products with retention (Table 2).



(*R*)-1.6 To a flame dried vial in a glovebox was added 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), tricyclohexylphosphine (11 mg, 0.040 mmol, 0.20 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μ L, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-methoxyphenyl)-1,3,2-dioxaborinane (88 mg, 0.40 mmol, 2.0 equiv), (*S*)-1.4 (66

mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The reaction was stirred for 24 hours before removing the vial from the glovebox, opening to atmosphere, and running through a silica gel plug (1:1 Et₂O:hexane). The combined organics were concentrated in vacuo, internal standard (PhTMS, 0.20 mmol) was added and ¹H NMR yield was collected. The product was purified by flash chromatography (1–3% Et₂O/pentane) to afford (*R*)-1.6 as a colorless oil. First run: (56.0 mg, 0.173 mmol, 86%, 93% ee). Second run: (56.4 mg, 0.174 mmol, 87%, 93% ee). Analytical data is consistent with literature values:^{13b} ¹H NMR (500 MHz, CDCl₃) δ 7.81– 7.76 (m, 1H), 7.74 (d, *J* = 8.6 Hz, 1H), 7.72–7.67 (m, 1H), 7.46 (s, 1H), 7.42 (dt, *J* = 9.5, 3.2 Hz, 2H), 7.29 (t, *J* = 7.2 Hz, 3H), 7.24–7.19 (m, 1H), 7.15 (d, *J* = 7.2 Hz, 2H), 7.07 (d, *J* = 8.4 Hz, 2H), 6.83 (d, *J* = 8.8 Hz, 2H), 5.65 (s, 1H), 3.78 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 158.2, 144.1, 142.0, 136.0, 133.5, 132.2, 130.6, 129.6, 128.4, 128.2, 128.0, 127.9, 127.8, 127.6, 126.4, 126.1, 125.7, 113.8, 56.2, 55.3; [*a*]²³**b** –0.77 (*c* 2.70, CHCl₃); SFC analysis (AD-H, 15% IPA, 2.5 mL/min) indicated 93% ee: t_R (major) = 13.9 minutes, t_R (minor) = 13.2 minutes.

B. Procedure for the synthesis of products with inversion of configuration.



(S)-1.6 To a flame dried vial in a glovebox was added 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-Bis(2,4,6-trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.27 mg, 0.0210 mmol, 0.105 equiv), potassium tert-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 µL, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-methoxyphenyl)-1,3,2-dioxaborinane (88 mg, 0.40 mmol, 2.0 equiv), (S)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The reaction was stirred for 24 hours before removing the vial from the glovebox, opening to atmosphere, and running through a silica gel plug (1:1 Et₂O:hexane). The combined organics were concentrated in vacuo, internal standard (PhTMS, 0.20 mmol) was added and ¹H NMR yield was collected. The product was purified by flash chromatography (1-3% Et₂O/pentane) to afford (S)-1.6 as a colorless oil. First run: (53.2 mg, 0.164 mmol, 82%, 93% ee). Second run: (56.0 mg, 0.173 mmol, 86%, 93% ee). Analytical data is consistent with the values listed above for (R)-1.6 $[\alpha]^{23}$ _D +2.1 (c 2.70, CHCl₃); SFC analysis (AD-H, 15% IPA, 2.5 mL/min) indicated 90% ee: t_R (major) = 13.2 minutes, t_R (minor) = 13.9 minutes.

Characterization Data for Products



(*R*)-1.22 Using representative procedure A outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), tricyclohexylphosphine (11 mg, 0.040 mmol, 0.20 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-

butanol (54 µL, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-dimethylaminophenyl)-1,3,2dioxaborinane (83 mg, 0.40 mmol, 2.0 equiv), (*S*)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash chromatography (1–3% Et₂O/pentane) to afford the product as a colorless oil. First run: (49.8 mg, 0.148 mmol, 80%, 90% ee). Second run: (52.6 mg, 0.167 mmol, 84%, 90% ee). Analytical data is consistent with literature values:^{13b} ¹**H** NMR (500 MHz, CDCl₃) δ 7.80–7.75 (m, 1H), 7.73 (d, *J* = 8.7 Hz, 1H), 7.71–7.67 (m, 1H), 7.48 (s, 1H), 7.40 (dt, *J* = 9.4, 3.3 Hz, 2H), 7.31 (d, *J* = 8.4 Hz, 1H), 7.27 (t, *J* = 7.6 Hz, 2H), 7.19 (t, *J* = 7.5 Hz, 1H), 7.16 (d, *J* = 7.5 Hz, 2H), 7.01 (d, *J* = 8.8 Hz, 2H), 6.67 (d, *J* = 8.8 Hz, 2H), 5.61 (s, 1H), 2.90 (s, 6H); ¹³C NMR (125 MHz, CDCl₃) δ 149.2, 144.6, 142.5, 133.6, 132.2, 131.8, 130.3, 129.6, 128.4 (2C), 128.0, 127.8, 127.7, 127.6, 126.3, 126.0, 125.6, 112.6, 56.2, 40.8; **IR** (neat) 3054, 3023, 2879, 1612, 1350 cm⁻¹; [*a*]²³**b** –9.43 (*c* 2.28, CHCl₃); **SFC** analysis (AD-H, 20% MeOH, 3 mL/min) indicated 92% ee: t_R (major) = 4.2 min, t_R (minor) = 4.8 min.



(*S*)-1.22 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-Bis(2,4,6-trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.27 mg, 0.0210 mmol, 0.105 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μ L, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-dimethylaminophenyl)-1,3,2-dioxaborinane (83 mg, 0.40 mmol, 2.0 equiv), (*S*)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product

was purified by flash chromatography (1–3% Et₂O/pentane) to afford the product as a colorless oil. First run: (42.1 mg, 0.125 mmol 62%, 92% ee). Second run: (53.6 mg, 79%, 0.159 mmol, 92% ee). Analytical data is consistent with the values listed above for (*R*)-1.22 [*a*]²³D +8.0 (*c* 1.00, CHCl₃); SFC analysis (AD-H, 20% MeOH, 3 mL/min) indicated 92% ee: t_R (major) = 3.9 min, t_R (minor) = 4.6 min.



(*R*)-1.23 Using representative procedure A outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), tricyclohexylphosphine (11 mg, 0.040 mmol, 0.2 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μ L, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-fluorophenyl)-1,3,2-dioxaborinane (83 mg, 0.40 mmol, 2.0 equiv), (*S*)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash chromatography (1–3% Et₂O/pentane) to afford the product as a colorless oil. First run: (49.8 mg, 0.159 mmol, 80%, 90% ee). Second run: (52.6 mg, 0.168 mmol, 84%, 90% ee). Analytical data is consistent with literature values:^{13b} 1H NMR (500 MHz, CDCl₃) δ 7.81–7.77 (m, 1H), 7.75 (d, *J* = 8.6 Hz, 1H), 7.71–7.67 (m, 1H), 7.45–7.39 (m, 3H), 7.31–7.21 (m, 4H), 7.15–7.07 (m, 4H), 6.97 (t, *J* = 8.8 Hz, 2H), 5.70 (s, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 161.6 (d, *J* = 245 Hz), 143.6, 141.4, 139.5 (d, *J* = 3 Hz), 133.5, 132.3, 131.1 (d, *J* = 8 Hz), 129.6, 128.6, 128.1, 128.05, 127.99, 127.8, 127.7, 126.7, 126.2, 125.9, 115.3 (d, *J* = 21 Hz), 56.3; [a]²³n +4.5 (*c* 4.47, CHCl₃); SFC analysis (OJ-H, 12% IPA, 2.5 mL/min) indicated 90% ee: t_R (major) = 9.4 minutes, t_R (minor) = 8.7 minutes.



(S)-1.23 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-Bis(2,4,6trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.27 mg, 0.0210 mmol, 0.105 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μL, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-fluorophenyl)-1,3,2-dioxaborinane (83 mg, 0.40 mmol, 2.0 equiv), (S)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash chromatography $(1-3\% \text{ Et}_2\text{O}/\text{pentane})$ to afford the product as a colorless oil. First run: (49.8 mg, 0.159 mmol, 80%, 88% ee). Second run: (50.0 mg, 0.168 mmol, 84%, 88% ee). Analytical data is consistent with literature values:^{13b} ¹H NMR (500 MHz, CDCl₃) & 7.81–7.77 (m, 1H), 7.75 (d, J = 8.6 Hz, 1H), 7.71–7.67 (m, 1H), 7.45–7.39 (m, 3H), 7.31–7.21 (m, 4H), 7.15– 7.07 (m, 4H), 6.97 (t, J = 8.8 Hz, 2H), 5.70 (s, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 161.6 (d, J =245 Hz), 143.6, 141.4, 139.5 (d, J = 3 Hz), 133.5, 132.3, 131.1 (d, J = 8 Hz), 129.6, 128.6, 128.1, 128.05, 127.99, 127.8, 127.7, 126.7, 126.2, 125.9, 115.3 (d, J = 21 Hz), 56.3; $[\alpha]^{23}D = -3.6$ (c 4.10, CHCl₃); SFC analysis (OJ-H, 12% IPA, 2.5 mL/min) indicated 88% ee: t_R (major) = 9.7 minutes, t_R (minor) = 10.6 minutes.



(*R*)-1.24 Using representative procedure A outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv),

tricyclohexylphosphine (11 mg, 0.040 mmol, 0.2 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 µL, 0.60 mmol, 3.00 equiv), 5,5-dimethyl-2-(4-trifluoromethylphenyl)-1,3,2-dioxaborinane (103 mg, 0.400 mmol, 2.00 equiv), (*S*)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (1% Et₂O/pentane) to afford the product as a colorless oil. First run: (64.4 mg, 0.178 mmol, 89%, 57% ee). Second run: (62.1 mg, 0.172 mmol, 86%, 57% ee). **TLC R**_f = 0.4 (pentane); ¹H NMR (500 MHz, CDCl₃) δ 7.85–7.79 (m, 1H), 7.77 (d, *J* = 8.6 Hz, 1H), 7.74–7.68 (m, 1H), 7.55 (d, *J* = 8.1 Hz, 2H), 7.49–7.41 (m, 3H), 7.32 (t, *J* = 7.4 Hz, 2H), 7.29–7.22 (m, 4H), 7.14 (d, *J* = 7.6 Hz, 2H), 5.75 (s, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 148.0, 142.9, 140.6, 133.5, 132.4, 130.0, 129.6, 128.9 (q, *J* = 32.4 Hz), 128.7, 128.3, 128.02, 128.00, 127.9, 127.7, 126.9, 126.4, 126.1, 125.5 (q, *J* = 3.7 Hz), 124.4 (q, *J* = 271.9 Hz), 56.9; IR (neat) 3057, 1600, 1323, 1119 cm⁻¹; HRMS (TOF MS CI+) *m* / *z* calcd for C₁₅H₁₆O (M)⁺ 362.1282, found 362.1273; [*a*]²³**b** +4.84 (*c* 0.915, CHCl₃); SFC analysis (AD-H, 5% IPA, 2.5 mL/min) indicated 57% ee: t_R (major) = 7.7 minutes, t_R (minor) = 7.0 minutes.



(*S*)-1.24 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-bis(2,4,6-trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.27 mg, 0.0210 mmol, 0.105 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μ L, 0.60 mmol, 3.00 equiv), 5,5-dimethyl-2-(4-trifluoromethylphenyl)-1,3,2-dioxaborinane (103 mg, 0.400 mmol, 2.00 equiv), (*S*)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product

was purified by flash column chromatography (1% Et₂O/pentane) to afford the product as a colorless oil. First run: (52.8 mg, 0.146 mmol, 73%, 91% ee). Second run: (48.6 mg, 0.134 mmol, 67%, 90% ee). **TLC R**_f = 0.4 (pentane); ¹**H NMR** (500 MHz, CDCl₃) δ 7.85–7.79 (m, 1H), 7.77 (d, *J* = 8.6 Hz, 1H), 7.74–7.68 (m, 1H), 7.55 (d, *J* = 8.1 Hz, 2H), 7.49–7.41 (m, 3H), 7.32 (t, *J* = 7.4 Hz, 2H), 7.29–7.22 (m, 4H), 7.14 (d, *J* = 7.6 Hz, 2H), 5.75 (s, 1H); ¹³**C NMR** (125 MHz, CDCl₃) δ 148.0, 142.9, 140.6, 133.5, 132.4, 130.0, 129.6, 128.9 (q, *J* = 32.4 Hz), 128.7, 128.3, 128.02, 128.00, 127.9, 127.7, 126.9, 126.4, 126.1, 125.5 (q, *J* = 3.7 Hz), 124.4 (q, *J* = 271.9 Hz), 56.9; **IR** (neat) 3057, 1600, 1323, 1119 cm⁻¹; **HRMS** (TOF MS CI+) *m* / *z* calcd for C₁₅H₁₆O (M)⁺ 362.1282, found 362.1273; **[a]**²³**p** –16.5 (*c* 1.00, CHCl₃); **SFC** analysis (AD-H, 5% IPA, 2.5 mL/min) indicated 89% ee: t_R (major) = 6.6 minutes, t_R (minor) = 7.3 minutes.



(*R*)-1.25 Using representative procedure A outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), tricyclohexylphosphine (11 mg, 0.040 mmol, 0.20 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μ L, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-acetylphenyl)-1,3,2-dioxaborinane (93 mg, 0.40 mmol, 2.0 equiv), (*S*)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (10% EtOAc/hexanes) to afford the product as an amorphous white solid. First run: (50.8 mg, 0.151 mmol, 76%, 89% ee). Second run: (51.0 mg, 0.152 mmol, 76%, 89% ee). TLC R_f = 0.4 (20% EtOAc/hexanes); ¹H

NMR (500 MHz, CDCl₃) δ 7.80 (d, J = 8.4 Hz, 2H), 7.73–7.69 (m, 1H), 7.67 (d, J = 8.6 Hz, 1H), 7.63–7.57 (m, 1H), 7.40–7.30 (m, 3H), 7.21 (q, J = 7.7 Hz, 2H), 7.19–7.10 (m, 4H), 7.05 (d, J = 7.5 Hz, 2H), 5.65 (s, 1H), 2.47 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 197.9, 149.4, 142.9, 140.7, 135.6, 133.5, 132.3, 129.9, 129.6, 128.7, 128.6, 128.3, 127.97, 127.95, 127.9, 127.7, 126.9, 126.3, 126.0, 57.0, 26.7; **IR** (neat) 3055, 2923, 1679, 1600, 1506 cm⁻¹; **HRMS** (TOF MS CI+) m / z calcd for C₂₅H₂₀O (M)⁺ 336.1514, found 3316.1514; **[a]**²³D –17.2 (c 2.3, CHCl₃); **SFC** analysis (OD-H, 20% IPA, 3.0 mL/min) indicated 89% ee: t_R (major) = 6.3 minutes, t_R (minor) = 5.9 minutes.



(*S*)-1.25 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-Bis(2,4,6-trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.67 mg, 0.0220 mmol, 0.11 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μL, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-acetylphenyl)-1,3,2-dioxaborinane (93 mg, 0.40 mmol, 2.0 equiv), (*S*)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (10% EtOAc/hexanes) to afford the product as an amorphous white solid. First run: (66.5 mg, 0.198 mmol, 99%, 97% ee). Second run: (66.0 mg, 0.196 mmol, 98%, 97% ee). TLC $\mathbf{R}_{\mathbf{f}} = 0.4$ (20% EtOAc/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.80 (d, *J* = 8.4 Hz, 2H), 7.73–7.69 (m, 1H), 7.67 (d, *J* = 8.6 Hz, 1H), 7.63–7.57 (m, 1H), 7.40–7.30 (m, 3H), 7.21 (q, *J* = 7.7 Hz, 2H), 7.19–7.10 (m, 4H), 7.05 (d, *J* = 7.5 Hz, 2H), 5.65 (s, 1H), 2.47 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 197.9, 149.4, 142.9, 140.7, 135.6, 133.5, 132.3, 129.9, 129.6, 128.7, 128.6, 128.3, 127.97, 127.95, 127.9, 127.7, 126.9, 126.3, 126.0, 57.0, 26.7; **IR** (neat) 3055,

2923, 1679, 1600, 1506 cm⁻¹; **HRMS** (TOF MS CI+) m / z calcd for C₂₅H₂₀O (M)⁺ 336.1514, found 3316.1514; $[\alpha]^{29}_{D}$ +5.05 (c 1.01, CHCl₃); **SFC** analysis (OD-H, 20% IPA, 3.0 mL/min) indicated 97% ee: t_R (major) = 5.9 minutes, t_R (minor) = 6.5 minutes.



(R)-1.26 Using representative procedure A outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), tricyclohexylphosphine (11 mg, 0.040 mmol, 0.20 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1butanol (54 µL, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-hydroxymethylphenyl)-1,3,2dioxaborinane (88 mg, 0.40 mmol, 2.0 equiv), (S)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (10% EtOAc/hexane) to afford the product as an oil. First run: (51.0 mg, 0.157 mmol, 79%, 82% ee). Second run: (50.0 mg, 0.154 mmol, 77%, 81% ee). TLC $R_f = 0.2$ (20%) EtOAc/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.82–7.76 (m, 1H), 7.74 (d, J = 8.7 Hz, 1H), 7.71-7.66 (m, 1H), 7.46 (s, 1H), 7.41 (dt, J = 9.5, 4.4 Hz, 2H), 7.34-7.24 (m, 5H), 7.24-7.19 (m, 1H), 7.41 (m, 2H), 7.411H), 7.15 (d, J = 8.4 Hz, 4H), 5.69 (s, 1H), 4.64 (d, J = 4 Hz, 2H), 1.77 (t, J = 4.5 Hz, 1H); ¹³C **NMR** (125 MHz, CDCl₃) & 143.7, 143.3, 141.5, 139.1, 133.5, 132.3, 129.9, 129.6, 128.5, 128.1, 128.02, 127.96, 127.9, 127.7, 127.3, 126.6, 126.1, 125.8, 65.2, 56.8; IR (neat) 3330 (br), 2953, 1600, 1506 cm⁻¹; HRMS (TOF MS CI+) m / z calcd for C₂₄H₁₈O (M – 2H)⁺ 322.1358, found 322.1364; [a]²³D-18.3 (c 1.66, CHCl₃); SFC analysis (AD-H, 30% MeOH, 2.5 mL/min) indicated 89% ee: t_R (major) = 4.3 minutes, t_R (minor) = 6.1 minutes.



(R)-1.27 Using representative procedure A outlined above, the following amounts of reagents 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 were used: equiv), tricyclohexylphosphine (11 mg, 0.040 mmol, 0.2 equiv), potassium tert-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 µL, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-{[(tertbutoxycarbonyl)amino]methyl}phenyl)-1,3,2-dioxaborinane (128 mg, 0.400 mmol, 2.00 equiv), (S)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (15-25% Et₂O/hexanes) to afford the product as a white solid. First run: (71.0 mg, 0.168 mmol, 84%, 92% ee). Second run: (70.5 mg, 0.166 mmol, 83%, 89% ee). TLC $\mathbf{R}_{f} = 0.3$ (20% EtOAc/hexanes); m.p. = 57 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.81–7.76 (m, 1H), 7.74 (d, J = 8.6 Hz, 1H), 7.71–7.66 (m, 1H), 7.45 (s, 1H), 7.41 (dt, J = 9.5, 1.0 Hz, 2H), 7.28 (t, J = 7.4 Hz, 3H), 7.22 (d, J = 7.6 Hz, 1H), 7.19 (d, J = 8.0 Hz, 2H), 7.14 (d, J= 7.5 Hz, 2H), 7.11 (d, J = 8.0 Hz, 2H), 5.67 (s, 1H), 4.83 (br s, 1H), 4.29 (d, J = 5.1 Hz, 2H), 1.45 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 156.0, 143.7, 142.9, 141.5, 137.1, 133.5, 132.2, 129.9, 129.6, 128.5, 128.1, 128.01, 127.95, 127.8, 127.64, 127.59, 126.5, 126.1, 125.8, 79.5, 56.7, 44.4, 28.5; IR (neat) 3346, 2876, 1698, 1600, 1365 cm⁻¹; HRMS (TOF MS ES+) m / z calcd for $C_{29}H_{29}O_2N (M + Na)^+ 446.2096$, found 446.2078; $[\alpha]^{23}D - 14.3 (c 4.4, CHCl_3)$; SFC analysis (AS-H, 20% MeOH, 2.5 mL/min) indicated 92% ee: t_R (major) = 4.3 minutes, t_R (minor) = 4.7 minutes.



(S)-1.27 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-Bis(2,4,6trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.27 mg, 0.0210 mmol, 0.105 equiv), potassium tert-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 µL, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-{[(tert-butoxycarbonyl)amino]methyl}phenyl)-1,3,2-dioxaborinane (128 mg, 0.400 mmol, 2.00 equiv), (S)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (15-25%) Et₂O/hexanes) to afford the product as a white solid. First run: (84.0 mg, 0.198 mmol, 99%, 96% ee). Second run: (75.4 mg, 0.178 mmol, 89%, 94% ee). TLC $R_f = 0.3$ (20% EtOAc/hexanes); **m.p.** = 57 °C; ¹**H** NMR (500 MHz, CDCl₃) δ 7.81–7.76 (m, 1H), 7.74 (d, J = 8.6 Hz, 1H), 7.71– 7.66 (m, 1H), 7.45 (s, 1H), 7.41 (dt, J = 9.5, 1.0 Hz, 2H), 7.28 (t, J = 7.4 Hz, 3H), 7.22 (d, J = 7.6Hz, 1H), 7.19 (d, J = 8.0 Hz, 2H), 7.14 (d, J = 7.5 Hz, 2H), 7.11 (d, J = 8.0 Hz, 2H), 5.67 (s, 1H), 4.83 (br s, 1H), 4.29 (d, J = 5.1 Hz, 2H), 1.45 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 156.0, 143.7, 142.9, 141.5, 137.1, 133.5, 132.2, 129.9, 129.6, 128.5, 128.1, 128.01, 127.95, 127.8, 127.64, 127.59, 126.5, 126.1, 125.8, 79.5, 56.7, 44.4, 28.5; **IR** (neat) 3346, 2876, 1698, 1600, 1365 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₉H₂₉O₂N (M + Na)⁺ 446.2096, found 446.2078; $[\alpha]^{29}$ D +22.1 (c 1.01, CHCl₃); SFC analysis (AS-H, 20% MeOH, 2.5 mL/min) indicated 96% ee: t_R $(major) = 4.5 minutes, t_R (minor) = 4.3 minutes.$



(S)-1.28 Using representative procedure A outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), tricyclohexylphosphine (11 mg, 0.040 mmol, 0.20 equiv), potassium tert-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 µL, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-((dimethylamino)-5pyrimidinylphenyl)-1,3,2-dioxaborinane (94 mg, 0.40 mmol, 2.0 equiv), (S)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (5% EtOAc/benzene) to afford the product as a white solid. First run: (58.2 mg, 0.171 mmol, 86%, 89% ee). Second run: (58.6 mg, 0.173 mmol, 86%, 89% ee). TLC $\mathbf{R}_{f} = 0.5$ (5% EtOAc/benzene); m.p. = 45–47 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.04 (s, 2H), 7.73-7.67 (m, 1H), 7.66 (d, J = 8.6 Hz, 1H), 7.64-7.59 (m, 1H), 7.40 (s, 1H), 7.33 (dt, J = 9.5, 3.5Hz, 2H), 7.23–7.16 (m, 3H), 7.15–7.10 (m, 1H), 7.06 (d, J = 7.6 Hz, 2H), 2.39 (s, 1H), 3.08 (s, 6H); ¹³C NMR (125 MHz, CDCl₃) δ 161.3, 158.5, 142.9, 140.7, 133.5, 132.3, 129.8, 128.7, 128.3, 127.9, 127.67, 127.65, 127.6, 126.8, 126.3, 125.9, 123.6, 51.7, 37.2; IR (neat) 3054, 3023, 2861, 1599, 1531 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₃H₂₁N₃ (M + H)⁺ 340.1814, found 340.1819; [a]²³ D +15.7 (c 2.51, CHCl₃); SFC analysis (AD-H, 30% MeOH, 2.5 mL/min) indicated 89% ee: t_R (major) = 4.6 minutes, t_R (minor) = 6.4 minutes.



(S)-1.28 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-Bis(2,4,6trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.27 mg, 0.0210 mmol, 0.105 equiv), potassium tert-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 µL, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-((dimethylamino)-5-pyrimidinylphenyl)-1,3,2-dioxaborinane (94 mg, 0.40 mmol, 2.0 equiv), (S)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (5% EtOAc/benzene) to afford the product as a white solid (50.8 mg, 0.150 mmol, 75%, 92% ee). TLC R_f = 0.5 (5% EtOAc/benzene); **m.p.** = 45–47 °C; ¹**H** NMR (500 MHz, CDCl₃) δ 8.04 (s, 2H), 7.73–7.67 (m, 1H), 7.66 (d, J = 8.6 Hz, 1H), 7.64–7.59 (m, 1H), 7.40 (s, 1H), 7.33 (dt, J = 9.5, 3.5 Hz, 2H), 7.23–7.16 (m, 3H), 7.15– 7.10 (m, 1H), 7.06 (d, J = 7.6 Hz, 2H), 2.39 (s, 1H), 3.08 (s, 6H); ¹³C NMR (125 MHz, CDCl₃) δ 161.3, 158.5, 142.9, 140.7, 133.5, 132.3, 129.8, 128.7, 128.3, 127.9, 127.67, 127.65, 127.6, 126.8, 126.3, 125.9, 123.6, 51.7, 37.2; **IR** (neat) 3054, 3023, 2861, 1599, 1531 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₃H₂₁N₃ (M + H)⁺ 340.1814, found 340.1819; $[\alpha]^{29}D$ –13.2 (*c* 0.675, CHCl₃); SFC analysis (AD-H, 30% MeOH, 2.5 mL/min) indicated 92% ee: t_R (major) = 6.1 minutes, t_R (minor) = 4.5 minutes.



(*S*)-1.29 Using representative procedure A outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), tricyclohexylphosphine (11 mg, 0.040 mmol, 0.20 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μ L, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(3-furanyl)-1,3,2-dioxaborinane (72 mg, 0.40 mmol, 2.0 equiv), (*S*)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene

(1 mL). The product was purified by flash column chromatography (0.5–1% Et₂O/pentane) to afford the product as a white solid. First run: (45.8 mg, 0.161 mmol, 80%, 94% ee). Second run: (44.0 mg, 0.155 mmol, 78%, 94% ee). **TLC** $\mathbf{R}_{\mathbf{f}} = 0.5$ (1% Et₂O/pentane); **m.p.** = 65–67 °C; ¹**H NMR** (500 MHz, CDCl₃) δ 7.83–7.69 (m, 3H), 7.60 (s, 1H), 7.47–7.37 (m, 3H), 7.34 (dd, J = 8.6, 1 Hz, 1H), 7.31–7.16 (m, 5H), 6.97 (s, 1H), 6.26 (s, 1H), 5.42 (s, 1H); ¹³**C NMR** (125 MHz, CDCl₃) δ 143.4, 143.3, 141.3, 141.1, 133.5, 132.4, 129.0, 128.6, 128.3, 128.1, 128.0, 127.7, 127.6, 127.1, 126.7, 126.2, 125.8, 111.6, 48.3; **IR** (neat) 3145, 3024, 1599, 1492 cm⁻¹; **HRMS** (TOF MS CI+) m / z calcd for C₂₁H₁₆O (M)⁺ 284.1201, found 284.1203; $[\alpha]^{23}\mathbf{p}$ +22.3 (c 1.67, CHCl₃); **SFC** analysis (AD-H, 5% IPA, 2.5 mL/min) indicated 94% ee: t_R (major) = 12.2 minutes, t_R (minor) = 11.3 minutes.



(*R*)-1.29 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-Bis(2,4,6-trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.27 mg, 0.0210 mmol, 0.105 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μ L, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(3-furanyl)-1,3,2-dioxaborinane (72 mg, 0.40 mmol, 2.0 equiv), (*S*)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (0.5–1% Et₂O/pentane) to afford the product as a white solid. First run: (35.5 mg, 0.125 mmol, 62.5 %, 82% ee). Second run: (38.7 mg, 0.136 mmol, 68%, 84% ee). TLC R_f = 0.5 (1% Et₂O/pentane); m.p. = 65–67 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.83–7.69 (m, 3H), 7.60 (s, 1H), 7.47–7.37 (m, 3H), 7.34 (dd, *J* = 8.6, 1 Hz, 1H), 7.31–7.16 (m, 5H), 6.97 (s, 1H), 6.26 (s, 1H), 5.42 (s, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 143.4, 143.3, 141.3, 141.1, 133.5,

132.4, 129.0, 128.6, 128.3, 128.1, 128.0, 127.7, 127.6, 127.1, 126.7, 126.2, 125.8, 111.6, 48.3; **IR** (neat) 3145, 3024, 1599, 1492 cm⁻¹; **HRMS** (TOF MS CI+) m/z calcd for C₂₁H₁₆O (M)⁺ 284.1201, found 284.1203; $[\alpha]^{29}$ D –22.0 (*c* 1.00, CHCl₃); **SFC** analysis (AD-H, 5% IPA, 2.5 mL/min) indicated 84% ee: t_R (major) = 12.2 minutes, t_R (minor) = 13.4 minutes.



(S)-1.14 Using representative procedure A outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), tricyclohexylphosphine (11 mg, 0.040 mmol, 0.20 equiv), potassium tert-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1butanol (54 µL, 0.60 mmol, 3.0 equiv), 5-(5,5-dimethyl-1,3,2-dioxaborinan-2-yl)-1-methyl-1Hindole (97 mg, 0.40 mmol, 2.0 equiv), (S)-1.4 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (5–20% Et₂O/hexane, 0.5% TEA) to afford the product as a white solid. First run: (63.4 mg, 0.182 mmol, 91%, 92% ee). Second run: (61.4 mg, 0.178 mmol, 89%, 93% ee). TLC $R_f = 0.3$ (20%) Et₂O/hexane); m.p. = 49–52 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.80–7.75 (m, 1H), 7.73 (d, J = 8.6 Hz, 1H), 7.70–7.64 (m, 1H), 7.50 (s, 1H), 7.40 (dt, J = 9.3, 4.9 Hz, 2H), 7.37–7.31 (m, 2H), 7.27 (t, J = 7.3 Hz, 2H), 7.25–7.16 (m, 4H), 7.07 (dd, J = 8.7, 1.0 Hz, 1H), 6.98 (d, J = 2.9 Hz, 1H), 6.36 (d, J = 2.9 Hz, 1H), 5.83 (s, 1H), 3.72 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 144.7, 142.6, 135.6, 134.9, 133.5, 132.2, 129.8, 129.2, 128.55, 128.52, 128.4, 128.0, 127.9, 127.8, 127.6, 126.3, 126.0, 125.6, 123.9, 121.7, 109.2, 101.1, 57.1, 33.0; **IR** (neat) 3022, 2884, 1599, 1489 cm⁻ ¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₆H₂₁N (M + Na)⁺ 370.1572, found 370.1576; $[\alpha]^{23}$ D – 10.8 (c 1.00, CHCl₃); SFC analysis (AD-H, 20% MeOH, 2.5 mL/min) indicated 93% ee: t_R (major) = 8.1 minutes, t_R (minor) = 9.0 minutes. Single crystals suitable for X-ray crystallographic analysis were grown by slow diffusion of hexane into a solution of **(S)-1.14** in benzene at 4 °C.



(*R*)-1.14 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-bis(2,4,6-trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.3 mg, 0.021 mmol, 0.11 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μ L, 0.60 mmol, 3.0 equiv), 5-(5,5-dimethyl-1,3,2-dioxaborinan-2-yl)-1-methyl-1*H*-indole (97 mg, 0.40 mmol, 2.0 equiv), (*S*)-2.1 (66 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography (5–20% Et₂O/hexanes, 0.5% TEA) to afford the product as a white solid. First run: (41.5 mg, 0.119 mmol, 57%, 92% ee). Second run: (57.0 mg, 0.164 mmol, 82%, 92% ee). Analytical data is consistent with the values listed above for (*S*)-1.4 [*a*]²³**p** +6.0 (*c* 0.9, CHCl₃); **SFC** analysis (AD-H, 20% MeOH, 2.5 mL/min) indicated 93% ee: t_R (major) = 8.9 minutes, t_R (minor) = 8.1 minutes.



(*R*)-1.30 Using representative procedure B above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-bis(2,4,6-trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.3 mg, 0.0210 mmol, 0.11 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μ L, 0.60 mmol, 3.0 equiv), 5,5-dimethyl-2-(4-methoxyphenyl)-1,3,2-dioxaborinane (88 mg, 0.40 mmol, 2.0 equiv), (*S*)-1.19 (69 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was

purified by flash column chromatography (1–3% Et₂O/hexanes) to afford the product as a colorless oil. First run: (54.8 mg, 0.156 mmol, 78%, 81% ee). Second run: (55.8 mg, 0.159 mmol, 80%, 81% ee). **TLC R**_f = 0.4 (10% Et₂O/Hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.56 (d, J = 7.2, 2H), 7.51 (d, J = 7.1, 2H), 7.42 (t, J = 7.1, 2H), 7.29 (m, 3H), 7.19 (m, 5H), 7.06 (d, J = 7.8, 2H), 6.83 (d, J = 8.2, 2H) 5.53 (s, 1H), 3.78 (s, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 158.2, 144.3, 143.5, 141.0, 139.2, 136.2, 130.5, 129.9, 129.5, 128.9, 128.5, 128.3, 127.15, 127.14, 126.4, 113.9, 55.9, 55.4; **IR** (neat) 3020, 2996, 1508, 1244, 1030 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₆H₂₂O (M + Na)⁺ 350.1671, found 367.1679; **[α]²³**D +1.2 (*c* 1.01, CHCl₃), **SFC** analysis (AD-H, 10% MeOH, 2.5 mL/min) indicated 84% ee: t_R (minor) = 21.5 minutes, t_R (major) = 19.8 minutes.



(*R*)-1.31 Using representative procedure B above, the following amounts and reagents: 1,8bis(1,5-cyclooctadiene)nickel (8.3 mg, 0.030 mmol, 0.10 equiv), 1,3-bis(2,4,6-trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (12 mg, 0.030 mmol, 0.10 equiv), potassium *tert*butoxide (64 mg, 0.60 mmol, 2.0 equiv), 4-(5,5-dimethyl-1,3,2-dioxaborinan-2-yl)-*N*,*N*dimethylaniline (134 mg, 0.600 mmol, 2.00 equiv), (*S*)-1.19 (103 mg, 0.300 mmol, 1.00 equiv) and 1-butanol (54 μ L, 0.90 mmol, 3.00 equiv). Purified by flash column chromatography (0–10% Et₂O/hexanes) to afford (*R*)-1.32, as a light yellow oil (55 mg, 0.15 mmol, 75%). TLC **R**_f = 0.3 (10% Et₂O/Hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.57 (d, *J* = 7.2, 2H), 7.49 (d, *J* = 8.2, 2H), 7.41 (t, *J* = 7.7, 2H), 7.29 (m, 3H), 7.27 (s, 1H) 7.21 (m, 5H), 7.02 (dd, *J* = 8.6, 2H) 6.67 (d, *J* = 8.9, 2H), 6.4 (d, *J* = 2.8, 1H), 5.5 (s, 1H), 2.9 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 149.2, 144.7, 143.9, 141.1, 139.0, 131.9, 130.2, 129.9, 129.5, 128.8, 128.4, 127.20, 127.15, 127.1, 126.3, 112.7, 55.7, 40.8; **IR** (neat) 3024, 2841, 2360, 1613, 1485, 1347, 763 cm⁻¹; **HRMS** (TOF MS EI+) m/z calcd for C₂₇H₂₅O (M + Na)⁺ 364.2065, found 364.2061; **[\alpha]**²³D –2.9 (c 1.07, CHCl₃); **SFC** analysis (AD-H, 16% MeOH, 3.0 mL/min) indicated 79% ee: t_R (minor) = 26.3 minutes, t_R (major) = 11.6 minutes.



(S)-1.32 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-bis(2,4,6trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.67 mg, 0.022 mmol, 0.11 equiv), potassium tert-butoxide (45 mg, 0.4 mmol, 2.0 equiv), tert-butyl 4-(5,5-dimethyl-1,3,2dioxaborinan-2-yl)benzylcarbamate (127.3 mg, 0.200 mmol, 2.00 equiv), (S)-1.19 (68.9 mg, 0.200 mmol, 1.00 equiv) and 1-butanol (54 µL, 0.60 mmol, 3.0 equiv). Purified by flash column chromatography (0-15 % EtOAc/Hexane) to afford to afford the desired triarylmethane as a clear colorless oil (48.5 mg, 54%); TLC $\mathbf{R}_{\mathbf{f}} = 0.1$ (9:1 Hexanes:EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 7.57 (dd, *J* = 1.5, *J* = 8.4, 2H), 7.51 (d, *J* = 8.3, 2H), 7.39 (t, *J* = 7.5, 2H), 7.29 (m, 3H), 7.25 (m, 3H), 7.15 (m, 6H), 5.56 (s, 1H), 4.82 (s, 1H), 4.30 (d, J = 5.3, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 156.0, 143.9, 143.13, 143.05, 140.9, 139.3, 137.1, 129.9, 129.8, 129.5, 128.9, 128.5, 127.6, 127.3, 127.16, 127.14, 126.5, 56.3, 44.3, 28.5; **IR** (neat) 3294, 3028, 1695, 1486, 1316, 1016, 757 cm⁻¹; **HRMS** (TOF MS EI+) m / z calcd for C₃₁H₃₁NO₂ [M+Na]⁺ 472.2253, found 472.2261. $[\alpha]^{29}$ –8.2 SFC analysis (AS-H, 20% MeOH, 2.5 mL/min) indicated 92% ee: t_R (major) = 6.86 minutes, t_R (minor) = 7.47 minutes.



(R)-1.33 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-bis(2,4,6trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.27 mg, 0.0210 mmol, 0.105 equiv), potassium tert-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 µL, 0.60 mmol, 3.0 equiv), 5-(5,5-dimethyl-1,3,2-dioxaborinan-2-yl)-1-methyl-1*H*-indole (97 mg, 0.40 mmol, 2.0 equiv), (S)-1.19 (69 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified by flash column chromatography $(5-20\% \text{ Et}_2\text{O}/\text{hexane}, 0.5\% \text{ TEA})$ to afford the product as a white solid. First run: (61.0 mg, 0.163mmol, 82%, 96% ee). Second run: (37.8 mg, 0.101 mmol, 51%, 96% ee). TLC $R_f = 0.3$ (10% EtOAc/hexanes); m.p. = 54–55 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.56 (d, J = 7.4, 2H), 7.49 (d, J = 8.2, 2H), 7.37 (t, J = 7.4, 2H), 7.27 (m, 3H), 7.26 (s, 1H) 7.22 (m, 9H), 7.07 (dd, J = 1.3, 8.6, 1H) 6.87 (d, J = 3.0, 1H), 6.37 (d, J = 2.8, 1H), 5.70 (s, 1H), 3.73 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 144.9, 144.2, 141.1, 138.9, 135.6, 135.1, 130.1, 129.7, 129.2, 128.8, 128.6, 128.4, 127.17, 127.15, 127.0, 126.2, 123.8, 121.6, 109.2, 101.1, 56.7, 32.9; IR (neat) 3025, 2360, 1486, 1449, 1246, 1006, 760 cm⁻¹; HRMS submitted; $[\alpha]^{23}$ D – 5.2; SFC analysis (AD-H, 25% MeOH, 2.5 mL/min) indicated 96% ee: t_R (major) = 11.3 minutes, t_R (minor) = 16.1 minutes.



(R)-1.34 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-Bis(2,4,6trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.27 mg, 0.0210 mmol, 0.105 equiv), potassium tert-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 µL, 0.60 mmol, 3.0 equiv), 5-(5,5-dimethyl-1,3,2-dioxaborinan-2-yl)-1-methyl-1*H*-indole (97 mg, 0.40 mmol, 2.0 equiv), (S)-1.20 (67 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified twice by flash column chromatography (6% Et₂O/hexane and then 60% benzene/pentane) to afford the desired triarylmethane as a white solid. First run: (58.5 mg, 0.161 mmol, 80%, 87%) ee). Second run: (58.5 mg, 0.161 mmol, 80%, 87% ee). TLC $R_f = 0.6$ (40% pentane/benzene); **m.p.** = 149–151 °C; ¹**H** NMR (500 MHz, CDCl₃) δ 7.59 (s, 1H), 7.34 (s, 1H), 7.29 (d, *J* = 8.3 Hz, 2H), 7.23 (s,1H), 7.18 (t, J = 7.4 Hz, 2H), 7.15–7.09 (m, 2H), 7.07 (t, J = 7.4 Hz, 4H), 6.95 (d, J = 8.6 Hz, 1H), 6.90 (d, J = 2.8 Hz, 1H), 6.57 (s, 1H), 6.29 (d, J = 2.8 Hz, 1H), 5.58 (s, 1H), 3.63 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 144.9, 143.9, 141.7, 138.5, 135.6, 135.0, 130.3, 130.1, 129.7, 129.2, 128.5, 128.3, 126.4, 126.2, 125.8, 123.7, 121.5, 109.2, 109.0, 101.1, 56.7, 33.0; **IR** (neat) 3145, 3024, 1599, 1492 cm⁻¹; [a]²³D -3.1 (c 2.24, CHCl₃); HRMS submitted; SFC analysis (OJ-H, 30% MeOH, 3.0 mL/min) indicated 87% ee: t_R (major) = 23.6 minutes, t_R (minor) = 26.2 minutes.



(R)-1.35 Using representative procedure B outlined above, the following amounts of reagents were used: 1,8-bis(1,5-cyclooctadiene)nickel (5.5 mg, 0.020 mmol, 0.10 equiv), 1,3-bis(2,4,6trimethylphenyl)-4,5-dihydroimidazolium tetrafluoroborate (8.27 mg, 0.0210 mmol, 0.105 equiv), potassium *tert*-butoxide (45 mg, 0.40 mmol, 2.0 equiv), 1-butanol (54 μL, 0.60 mmol, 3.0 equiv), 5-(5,5-dimethyl-1,3,2-dioxaborinan-2-yl)-1-methyl-1*H*-indole (97 mg, 0.40 mmol, 2.0 equiv), (S)-**1.21** (67 mg, 0.20 mmol, 1.0 equiv), tetrahydrofuran (1 mL) and toluene (1 mL). The product was purified twice by flash column chromatography (6% EtOAc/hexanes and then 70% benzene/pentane) to afford the desired triarylmethane as a white solid. First run: (51.4 mg, 0.119 mmol, 60%, 93% ee). Second run: (52.0 mg, 0.121 mmol, 60%, 93% ee). TLC $R_f = 0.3$ (40%) pentane/benzene); m.p. = 81-84 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.43 (s, J = 8.0 2H), 7.33 (s, 1H), 7.29 (t, J = 7.4 Hz, 2H), 7.24 (s,1H), 7.22–7.14 (m, 5H), 7.10 (s, 1H), 7.06 (dt, J = 8.2, 2.0Hz, 2H), 7.01 (d, *J* = 2.7 Hz, 1H), 6.90 (d, *J* = 8.2 Hz, 1H), 6.39 (d, *J* = 2.5 Hz, 1H), 5.69 (s, 1H), 4.27 (s, 4H), 3.76 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 144.9, 143.8, 143.7, 143.1, 138.4, 135.6, 135.1, 134.8, 130.0, 129.7, 129.2, 128.6, 128.4, 126.6, 126.2, 123.8, 121.6, 120.2, 117.6, 115.8, 109.2, 101.1, 64.58, 64.56, 56.7, 33.0; **IR** (neat) 2916, 1586, 1513, 1449 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₃₀H₂₅O₂N (M + Na)⁺ 454.1783, found 454.1772; $[\alpha]^{23}$ D -8.2 (c 2.36, CHCl₃); SFC analysis (OD-H, 25% IPA, 2.5 mL/min) indicated 93% ee: t_R (major) = 23.6 minutes, t_R (minor) = 26.2 minutes.



(R)-1.12 Prepared according to general procedure B using the following amounts and reagents: 0.10 $Ni(cod)_2$ (2.8)mg, 0.010 mmol, equiv), 1,3-bis(2,4,6-trimethylphenyl)-4,5dihydroimidazolium tetrafluoroborate (3.9 mg, 0.010 mmol, 0.10 equiv), potassium tert-butoxide (22 mg, 0.20 mmol, 2.0 equiv), 5,5-dimethyl-2-(thiophen-2-yl)-1,3,2-dioxaborinane (39.2 mg, 0.200 mmol, 2.00 equiv), (S)-1.4 (33.1 mg, 0.100 mmol, 1.00 equiv) and 1-butanol (27 µL, 0.30 mmol, 3.0 equiv). Purified by flash column chromatography (0-5% Et₂O/hexanes) to afford the desired triarylmethane as a yellow solid (11.4 mg, 0.0379 mmol, 38%). Analytical data is consistent with literature values.^{13b} ¹H NMR (400 MHz, CDCl₃) δ 7.84–7.72 (m, 3H), 7.61 (s, 1H), 7.37–7.41 (m, 2H), 7.37 (dd, J = 1.6, 8.5), 7.34–7.20 (m, 6H), 6.95 (t, J = 4.3), 6.73 (d, J = 4.3) 3.3), 5.84 (s, 1H); ¹³C NMR δ (125 MHz, CDCl₃) δ 147.8, 143.7, 141.4, 133.5, 132.4, 129.1, 128.6, 128.2, 128.1, 127.7, 127.5, 127.4, 126.9, 126.8, 126.7, 126.2, 125.9, 124.8, 52.3 [α]²⁵D -9.3 (*c* 0.57, CHCl₃).

Tables of results using alternative ligands and bases.

Other ligands and bases were tested under reaction conditions similar to Table 1.1.

Representative examples are shown below.

			OMe
($\begin{array}{c c} & Me & Me \\ & OBoc \\ & Ph_{+}O_{-B}O \\ & & & & \\ & & & \\ & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & $	Ph
	Entry	ligand	yield ^a
	1	DPEphos (Bis[(2-diphenylphosphino)phenyl] ether)	< 5%
	2	Cy-DPEphos (Bis[(2-dicyclohexylphosphino)phenyl] ether)	< 5%
	3	DPPO (1,8-bis(diphenylphosphino)octane)	13%
	4	PPh ₃ (triphenylphosphine)	22%
	5	P(t-Bu) ₃ tri- <i>tert</i> -butylphosphine	< 5%
	6	XPhos (2-Dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl)	< 5%
	7	SPhos (2-Dicyclohexylphosphino-2',6'-dimethoxybiphenyl)	< 5%
	8	SIPr-HBF ₄ (1,3-Bis(2,6-diisopropylphenyl)- 4,5-dihydroimidazolium tetrafluoroborate	31%
	9	tricyclohexylphosphine	86%
	10	PCy ₃ tricylohexylphosphine (11 mol %)	83%
	11	SIMes-HBF ₄ 1,3-bis(2,4,6-trimethylphenyl)- 4,5-dihydroimidazolium tetrafluoroborate	84%
	12	None	< 5%

Table 1.4. Examination of additional ligands in the cross-coupling reaction

^aDetermined by ¹H NMR analysis using an internal standard (PhSiMe₃).














































































	mucx	Hame	Otant	Time	Ena	Ter Onser	Quantity	rieigin	7000	7000
			[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
	1	UNKNOWN	3.70	4.03	4.41	0.00	96.61	337.3	69.5	96.611
	2	UNKNOWN	4.43	4.66	4.98	0.00	3.39	10.1	2.4	3.389
	Total						100.00	347.4	71.9	100.000














Total			100.00	227.3	115.7	100.000















muex	Name	Start	Time	Ena	RT Uliset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	11.31	11.76	12.51	0.00	97.87	252.3	70.7	97.873
2	UNKNOWN	14.45	16.15	17.00	0.00	2.13	2.3	1.5	2.127
Total						100.00	254.6	72.2	100.000



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	23.55	24.74	25.88	0.00	93.55	103.0	57.7	93.545
2	UNKNOWN	26.15	26.89	27.58	0.00	6.45	6.2	4.0	6.455
Total						100.00	109.2	61.7	100.000





SFC traces (complete SFC data can be found in section X):

Chapter 2

Propargylic Carbamates Facilitate Regio- and Stereoselective Nickel-Catalyzed Hydroarylation of Alkynes

2.1 Introduction

Functionalized alkenes are present in pharmaceuticals¹ and natural products, and are commonly used as building blocks in synthesis² and polymer chemistry³ (Figure 2.1).⁴ Hydroarylation reactions of alkynes offer an alternative manifold for the facile synthesis of substituted alkenes. The major challenges associated with hydroarylation of alkynes has primarily been control of regioisomeric ratios and control of E/Z stereochemistry of the resulting olefin without relying on steric or electronic biases in the substrate.⁵ For example, early work from Hayashi demonstrated that Rh(I) catalysts cleanly furnished hydroarylated products using symmetrical alkynes and arylboronic acids, however, using unsymmetrical starting materials that lacked strong electronic or steric bias resulted in mixtures of regioisomers (Scheme 2.1a).⁶ The Lautens group judiciously addressed this problem with the use of pyridine or alcohol directing

¹ (a) Liu, X.; Shimizu, M.; Hiyama, T. *Angew. Chem. Int. Ed.* **2004**, *43*, 879. (b) Levenson, A. S.; Jordan, V. C. *Eur. J. Cancer*, **1999**, *35*, 1628. (c) Prasit, P.; Wang, Z.; Brideau, C.; Chan, C. C.; Charleson, S.; Cromlish, W.; Ethier, D.; Evans, J. F.; Ford-Hutchinson, A. W.; Gauthier, J. Y.; Gordon, R.; Guay, J.; Gresser, M.; Kargman, S.; Kennedy, B.; Leblanc, Y.; Leger, S.; Mancini, J.; O' 'Neill, G. P.; Ouellet, M.; Percival, M. D.; Perrier, H.; Riendeau, D.; Rodger, I.; Tagari, P.; Therien, M.; Vickers, P.; Wong, E.; Xu, L. J.; Young, R. N.; Zamboni, R.; Boyce, S.; Rupniak, N.; Forrest, M.; Visco, D.; Patrick, D. *Bioorg. Med. Chem. Lett.* **1999**, *9*, 1773. (d) Fallis, A. G.; Forgione, P. *Tetrahedron*, **2001**, *28*, 5899.

²For examples of alkene utility in synthesis: (a) Hoffmann, N. *Chem. Rev.* **2008**, *108*, 1052. (b) Sato, F.; Urabe, H.; Okamoto, S. *Chem. Rev.* **2000**, *100*, 2835. (c) Nicolaou, K.C.; Snyder, S. A.; Montagnon, T.; Vassilikogiannakis, G. *Angew. Chem. Int. Ed.* **2002**, *41*, 1668.

³ Hall, H. K. Angew. Chem., Int. Ed. 1983, 22, 440.

⁴ (a) Example of natural products: Williams, R.B.; Norris, A.; Slebodnick, C.; Merola, J.; Miller, J. S.; Andriantsiferana, R.; Rasamison, V. E.; Kingston, D. G. I. *J. Nat. Prod.* **2005**, *68*, 1371. (b) Lin, C. M.; Ho, H. H.; Pettit, G. R.; Hamel, E. *Biochemistry* **1989**, *28*, 6984.

⁵ for examples using electronic biased substrates (a) Lin, P.-S.; Jeganmohan, M.; Cheng, C.-H. *Chem. - Eur. J.* **2008**, 14, 11296. (b) Bai, Y.; Yin, J.; Kong, W.; Mao, M.; Zhu G. *Chem. Commun.*, **2013**,49, 7650. (c) Cui, W.; Yin, J.; Zheng, R.; Cheng, C.; Bai, Y.; Zhu, G. J. Org. Chem. 2014, 79, 3487. Also see reference 6 for examples of steric and electronic bias

⁶(a)Hayashi, T.; Inoue, K.; Taniguchi, N.; Ogasawara, M. J. Am. Chem. Soc. **2001**, 123, 9918. (b)Zhang, W.; Liu, M.; Wu, H.; Ding, J.; Cheng, J. Tetrahedron Lett. **2008**, 49, 5214. (c) Genin, E.; Michelet, V.; Genêt, J.-P. Tetrahedron Lett. **2004**, 45, 4157.

groups that could deliver hydroarylated products with excellent regio- and stereocontrol (Scheme 2.1b).⁷ Efforts to employ group 10 metals such as palladium and nickel in hydroarylation reactions have been explored with low regio- and sterocontrol (Scheme 2.1d).⁸ However, Engle and co-workers have recently reported the use of pendant directing groups that address these obstacles with great success in palladium-catalyzed reactions (Scheme 2.1c).⁹ To date, hydroarylation reactions that use nickel catalysts suffer from both regio- and stereocontrol of the resulting olefin. (Scheme 2d).

Figure 2.1. Examples of pharmaceutical agents and natural products bearing alkenes



Scheme 2.1. Summary of transition metal catalyzed hydroarylations using boronic acids



⁷ (a) Panteleev, J.; Huang, R. Y.; Lui, E. K. J.; Lautens, M. Org. Lett. **2011**, 13, 5314. (b) Lautens, M; Yoshida, M. Org. Lett. **2002**, 4, 123.

⁸(a)Shirakawa, E.; Takahashi, G.; Tsuchimoto, T.; Kawakami, Y. *Chem. Commun.* **2001**, 2688 (b) Hartwig, J. F. *Science* **2011**, 333, 1423. (c) Oh, C. H.; Jung, H. H.; Kim, K. S.; Kim, N. *Angew. Chem., Int. Ed.* **2003**, 42, 805. (d) Xu, X.; Chen, J.; Gao, W.; Wu, H.; Ding, J.; Su, W. *Tetrahedron* **2010**, 66, 2433

⁹ Liu, Z.; Derosa, J.; Engle, K. M. J. Am. Chem. Soc., 2016, 138, 13076.

Moving forward, efforts in our group have focused on the use of base metals and propargylic directing groups to address this challenge. Previous work in our group has demonstrated that carbamates can serve as directing groups for nickel catalysts in Suzuki coupling reactions. Thus, we hypothesized that propargylic carbamates could also serve as directing groups for hydroarylation reactions as well. Herein, we report a regio- and stereoselective hydroarylation of alkynes under mild conditions (Scheme 2.1e).

2.2 Development of Nickel-Catalyzed Directed Hydroarylation

We began our investigation by investigating a range of propargylic directing groups including pivalates, carbonates and carbamates. Using pivalates and carbonates as directing groups led to undesired formation of allene **2.19** while still forming small amounts of desired product. A mixture of byproducts and only 10% hydroarylated product was observed when free secondary propargyl alcohols were used. Interestingly, nearly identical results were observed without the alcohol or directing groups present (Table 2.1, entries 4 and 5). Employing bulky carbamates such as diisopropyl carbamate led to a severe decrease in product formation. However, moving to sterically smaller alkyl groups such as pyrrolidine carbamate led to an increase in product formation, with a dimethyl carbamate giving the highest yields. A range of ligands were investigated during the optimization. The reaction performs well with monodentate phosphine ligands. During further studies of ligand effects, it was uncovered that catalysts with sterically bulky Buchwald type ligands such as TrixiePhos were uniquely capable of suppressing the formation of allene, while still furnishing the desired product in good yields.

Interestingly, the reaction performs much better in the absence of bases such as *t*-BuOK and K₃PO₄, commonly used bases for transmetallation of Suzuki reagents. Furthermore, the addition of just one equivalent of LiCl to the reaction mixture led to a complete shutdown of reactivity and

recovery of only starting material (Table 2.1, entry 12). Additionally, during an investigation of solvents, it was found that the use of strongly coordinating solvents such as DMF also shut down reactivity (Table 2.1, entry 11).

	Ni(cod) ₂ (10 mol %) TrixiePhos (10 mol %)	Рһ н. 人 "О.	. ⊿NMe₂ Ph.	
Ph	Et PhB(OH) ₂ (2.5 equiv) THF (0.05M), rt, 24 h 2.17	Ph Et 2.18	0 + I	^{>h} 2.19
entry	variation from standard conditions	recovored 2.17	(%) ^a 2.18 (%) ^a	2.19 (%) ^a
1	none	<2	70	<2
2	2.17a instead of 2.17	57	16	14
3	2.17b instead of 2.17	81	8	<2
4	2.17c instead of 2.17	73	20	<2
5	2.20 instead of 2.17	57	19	<2
6	no nickel	100	0	0
7	no ligand	8	55	15
8	SPhos instead of TrixiePhos	16	63	2
9	BrettPhos instead of TrixiePhos	11	50	9
10	CPhos instead of TrixiePhos	12	50	12
11	DMF instead of THF	86	13	7
12	1 equivalent of LiCI	100	0	0
13	1 equivalent of K_3PO_4	10	40	6
14	PhBpin instead of PhB(OH) ₂	78	8	4
15	(PhBO) ₃ instead of PhB(OH) ₂	65	19	6
	N(Ph) ₂ NH ₂	QН		Ĥ
C				\downarrow_{Ma}
2	Et Et PI	h 2.17c	Ph 2.	20
Ph	2.17a _{Ph} 2.17b			

Table 2.1. Optimization of hydroarylation reaction conditions

2.3 Scope of Substrates for Hydroarylation

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Having identified suitable reaction conditions for model substrate 2.17, we turned our attention to interrogating a broad scope of internal alkynes (Table 2.2). Substrates bearing electron-deficient arenes provided the highest yields of hydroarylated product at room temperature while electron-rich arenes required heating (Table 2.2, entry 2.22 and 2.23). Ketone and ester substituents were well tolerated, however, free aldehydes were not. Gratifyingly, this oxidation state can still be incorporated as an acetal, although heating is required. The presence of aryl fluorides and chlorides did not produce the corresponding Suzuki-coupled products, allowing for

^aYied determined by ¹H NMR spectroscopy using PhTMS as internal standard.

further orthogonal functionalization (Table 2.2, entry **2.27** and **2.28**). An X-ray crystallographic structure of trifluoromethyl-substituted **2.29** was obtained, supporting the completely regioselective *cis*-hydroarylation. Benzyl-protected alcohols on the alkyl moiety of the starting materials were carried through the reaction untouched. Citronellal-derived substrate **2.31** shows that alkenes do not shut down reactivity and can be manipulated in downstream reactions. Surprisingly, sp²-hybridized side chains yielding both allylic and benzylic-activated carbamates did not undergo Tsuji-Trost type cross-coupling and provided products in high yields (entry **2.32**).





yield is that of isolated compound after column chromatography.

2.4 Scope of ArylBoronic Acid for Hydroarylation

Next, we turned our attention to investigating the scope of arylboronic acids (Table 2.3). Again, a similar trend in sensitivity to electronics was observed and electron-deficient arylboronic acids provided the highest yields at room temperature while electron-rich arylboronic acids required heating. A variety of functional groups were well-tolerated, allowing for the incorporation of phenyl TMS, acetoxy, ether, thioether, and trifluoromethyl moieties (Table 2.3, entries **2.32-2.34 & 2.43**). Ketones and Boc-protected amines were also well tolerated and provided hydroarylated product in good yields. Chloro- and fluorophenyl products, such as **2.41** and **2.42** did not participate in either hydrodehalogenation or cross-coupling. Furthermore, boronic acids containing heterocycles such as furan and thiophene underwent hydroarylation smoothly at elevated temperatures while benzofuran and indoleboronic acids could be coupled at room temperature. (Table 2.3, entries **2.38** and **2.39**).



2.5 Synthesis of Tamoxifen

Functionalized allylic alcohols are common synthetic building blocks en route to more complex molecular scaffolds. Allylic carbamates can be used in Tsuji-Trost chemistry as well as copper-mediated substitutions allowing for a range of back-to-back transition-metal-catalyzed reactions following hydroarylation. Scheme 2.2 demonstrates one of the many synthetic uses of these products. **2.47** can be converted to tamoxifen in two steps via a cuprate-mediated allylic substitution of the allylic carbamate followed by isomerization of the double bond to form the conjugated target **2.13**. As previously stated, these alkenes are known to be biologically active, and have been described to have anti-cancer activity. Hence, the herein described method can serve as a useful platform to generate a library of tamoxifen-like structures for SAR studies.



2.6 Mechanistic Studies and Deuterium Labeling

Lastly, we were interested gaining insight into the mechanism of the reaction by determining the origin of the hydrogen atom in our hydroarylation reaction. Initial attempts to replace phenylboronic acid with phenyl boroxine and deuterium oxide resulted in low yields of product (17-30%) with 99% deuterium incorporation. However, using PhB(OD)₂ led to a 68% yield of product **2.48** with 88% deuterium incorporation (Equation 2.3). These results led us to definitively conclude that the acidic protons in the boronic acid were the source of hydrogen in the hydroarylation. These results are consistent with a mechanism that involves a directed hydrometalation to afford a vinylnickelintermediate, which can then undergo transmetallation and reductive elimination to furnish the desired product. This is also consistent with the previously proposed hydoarylation reactions that do not require base.^{8c}





2.7 Conclusion

In conclusion, we have developed a regio- and stereoselective nickel-catalyzed hydroarylation of alkynes using a propargyl carbamate as a directing group. The reaction is tolerant of a range of functional groups and heterocycles. Furthermore, the synthesis of tamoxifen can be completed in two, operationally simple, steps in good overall yield.

2.8 Experimental Details

General Procedures

All hydroarylations were carried out under an atmosphere of N₂, or Ar when noted. All glassware was oven- or flame-dried prior to use. Tetrahydrofuran (THF), diethyl ether (Et_2O), dichloromethane (CH₂Cl₂), and dimethylacetamide (DMA) were degassed with Ar and then passed through two 4 x 36 inch columns of anhydrous neutral A-2 alumina (8 x 14 mesh; LaRoche Chemicals; activated under a flow of argon at 350 °C for 12 h) to remove H₂O. All other solvents utilized were purchased "anhydrous" commercially, or purified as described. ¹H NMR spectra were recorded on Bruker DRX-400 (400 MHz ¹H, 100 MHz ¹³C, 376.5 MHz ¹⁹F) or CRYO-500 (500 MHz ¹H, 125.7 MHz ¹³C) spectrometers. Proton chemical shifts are reported in ppm (δ) relative to internal tetramethylsilane (TMS, $\delta 0.00$). Data are reported as follows: chemical shift (multiplicity [singlet (s), broad singlet (br s), doublet (d), doublet of doublets (dd), triplet (t), doublet of triplets (dt), doublet of doublet of triplets (ddt), triplet of triplets (tt), quartet (q), quintet (quin), apparent doublet (ad), apparent triplet (at), multiplet (m)], coupling constants [Hz], integration). Carbon chemical shifts are reported in ppm (δ) relative to TMS with the respective solvent resonance as the internal standard (CDCl₃, δ 77.16 ppm). Unless otherwise indicated, NMR data were collected at 25 °C. Infrared (IR) spectra were obtained on a Thermo Scientific Nicolet iS5 spectrometer with an iD5 ATR tip (neat) and are reported in terms of frequency of absorption (cm⁻¹). Analytical thin-layer chromatography (TLC) was performed using Silica Gel 60 F₂₅₄ precoated plates (0.25 mm thickness). Visualization was accomplished by irradiation with a UV lamp and/or staining with KMnO₄, ceric ammonium molybdate (CAM), or *p*-anisaldehyde (PAA) solutions. Flash chromatography was performed using Silica Gel 60 (170-400 mesh) from Fisher Scientific. Melting points (m.p.) were obtained using a Mel-Temp melting point apparatus and are uncorrected. Optical rotations were measured on a Rudolph Research Analytical Autopol IV Automatic Polarimeter. High resolution mass spectrometry was performed by the University

of California, Irvine Mass Spectrometry Center. $Bis(cyclooctadiene)nickel Ni(cod)_2$ complex was purchased from Strem, stored in a glovebox under an atmosphere of N₂, and used as received. All other reagents were purchased commercially and used as received.

GENERAL METHODS FOR STARTING MATERIAL SYNTHESIS

The general methods for starting material synthesis will be used throughout the rest of the SI. In each instance a general method is used, it is specified by letter (A, B, etc.) and the exact amounts of reagents used for each reaction are listed for the specific compounds synthesized.

METHOD A: SONOGASHIRA COUPLING, CARBAMATE INSTALLATION



SONOGASHIRA COUPLING:

The products were prepared according to a modified procedure reported by Lautens.¹⁰ To 100 mL round bottom flask equipped with a stir bar was added PdCl₂(PPh₃)₂ (5 mol%), CuI (10 mol%), 1-butyne-3-ol (1.1 equiv), aryl iodide (1.0 equiv), and triethyl amine (25 mL). The reaction was stirred at room temperature for 4 hours. The reaction mixture was then filtered through silica to remove palladium and copper, and the silica was washed with diethyl ether. The solvents were removed under reduced pressure to provide the crude product as a dark yellow-brown oil. The crude mixtures were carried forward to carbamate installation without further purification.

CARBAMATE INSTALLATION:

¹⁰ Panteleev, J.; Huang, R. Y.; Lui, E. K. J.; Lautens, M. Org. Lett. 2011, 13, 5314

To a 100 mL round bottom flask was added NaH (1.1 equiv) and THF (30 mL). The suspension was cooled to 0 °C in an ice bath and a solution of the alcohol (1 M in THF) was added dropwise over 15 minutes. The ice bath was then removed and the mixture was allowed to warm to room temperature before dimethylcarbamoyl chloride (1.15 equiv) was added. The reaction was stirred overnight at room temperature for 15 hours after which the mixture was quenched with saturated aqueous ammonium chloride, extracted with ethyl ether, dried over MgSO₄, and concentrated under reduced pressure. The product was purified by flash column chromatography.

METHOD B: COREY-FUCHS, 1, 2 ADDITION, CARBAMATE INSTALLATION



COREY-FUCHS:

The product was prepared according to a modified procedure reported by Hoppe.¹¹ To a 250 mL round bottom flask was added PPh₃ (40 mmol, 4 equiv) and dichloromethane (35 mL) and was cooled to 0 °C, after which CBr₄ (20 mmol, 2 equiv) was added as a solution in dichloromethane (15 mL). The aryl aldehyde (10 mmol, 1 equiv) was then added as a solution in dichloromethane (10 mL) dropwise over the course of 10 minutes. The reaction was stirred for 30 min at 0 °C before the solvent was removed and the crude yellow oil was purified by column chromatography.

1, 2 ADDITION, CARBAMATE INSTALLATION:

¹¹ Oestreich, M; Fröhlich, R.; Hoppe, D. J. Org. Chem. 1999, 64, 8616

To a 250 mL round bottom flask was added *gem*-dibromo styrene and THF (50 mL). The solution was cooled to -78 °C in a dry ice and acetone bath after which *n*-BuLi (1.2 equiv) was added dropwise over the course for 20 min. The bath was then removed and the reaction was allowed to warm to room temperature and stirred for 30 minutes before it was cooled back down to -78 °C, and a solution of propionaldehyde (1M in THF) was added dropwise over the course of 10 minutes. The reaction was allowed to warm to room temperature and stirred for a solution of propionaldehyde (1M in THF) was added dropwise over the course of 10 minutes. The reaction was allowed to warm to room temperature and stirred for 4 hours before being quenched with dimethylcarbamoyl chloride (1.15 equiv). The resulting mixture was stirred overnight at room temperature for 15 hours before being quenched with saturated aqueous ammonium chloride and extracted with ethyl ether, dried over MgSO₄, and concentrated under reduced pressure. The product was purified by flash column chromatography.

METHOD C: 1, 2 ADDITION, CARBAMATE INSTALLATION



To a 250 mL round bottom flask was added phenyl acetylene (1 equiv) and THF (50 mL). The solution was cooled to -78 °C in a dry ice and acetone bath after which *n*-BuLi (1.2 equiv) was added dropwise over the course for 20 min. The bath was then removed and the reaction was allowed to warm to room temperature and stirred for 30 minutes before it was cooled back down to -78 °C, and a solution of propionaldehyde (1M in THF) was added dropwise over the course of 10 minutes. The reaction was allowed to warm to room temperature (1.15 equiv). The resulting mixture was stirred overnight at room temperature for 15 hours before being quenched with saturated aqueous

ammonium chloride and extracted with ethyl ether, dried over MgSO₄, and concentrated under reduced pressure. The product was purified by flash column chromatography.

METHOD D: NICKEL CATALYZED HYDROARYLATION



In a glove box, a flame dried 7 mL dram vial was charged with Ni(cod)₂ (5.5 mg, 0.02 mmol, 0.10 equiv), TrixiePhos (8.0 mg, 0.02 mg, 0.10 equiv), arylboronic acid (0.5 mmol, 2.5 equiv), and the alkyne (0.2 mmol, 1.0 equiv) and dissolved in THF (4 mL). The reaction was then stirred for 24 hours at 24 $^{\circ}$ C or 85 $^{\circ}$ C depending on the substrate and arylboronic acid used. The crude reaction mixture was then filtered through a pad of silica and eluted with Et₂O to remove the catalyst and then concentrated under reduced pressure. The crude mixture was purified by flash column chromatography to isolate pure product.

CHARACTERIZATION FOR STARTING MATERIALS



2.49. The product was prepared according to general method C using propionaldehyde (1.9 mL, 26.8 mmol), phenylacetylene (2.9 mL, 26.4 mmol) *n*-BuLi (11.4 mL, 28.5 mmol) and dimethylcarbamoyl chloride (2.5 mL, 27.5 mmol). The product was purified by flash column chromatography (20% Et₂O/pentane) to afford the title compound as a colorless oil (4.4 g, 19 mmol, 72% yield). **TLC R**_f = 0.4 (12% EtOA/hexane); ¹**H NMR** (500 MHz, CDCl₃) δ 7.49–7.41

(m, 2H), 7.33–7.28 (m, 3H), 5.5 (t, J = 6.48 Hz, 1 H), 2.94 (s, 6H), 1.89 (quintet J = 7.5 Hz, 2H) 1.08 (t, J = 7.3 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 156.0, 132.2, 128.7, 128.5, 122.9, 87.6, 85.2, 66.9, 28.9, 9.8; **IR** (neat) 2970, 2935, 1702, 1489, 1128, 1392 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₁₄H₁₇NO₂Na (M + Na)⁺ 254.1157, found 254.1151



2.50. The product was prepared according to general method C using citronellal (2.0 mL, 11.0 mmol), phenylacentylene (1.1 mL, 10 mmol,) *n*-BuLi (4.4 mL, 11 mmol) and dimethylcarbamoyl chloride (1.0 ml, 11 mmol). The product was purified by flash column chromatography (20% Et₂O/pentane) to afford the title compound as a colorless oil (3.3 g, 10 mmol, 92% yield). **TLC R**_f = 0.4 (20% Et₂O/pentane); ¹**H NMR** (500 MHz, CDCl₃) δ 7.46–7.40 (m, 2H), 7.31–7.27 (m, 3H), 5.65–5.58 (m, 1H), 5.14–5.07 (m, 1H), 2.94 (s, 6H) 2.08–1.84, (m, 3H), 1.83–1.65 (m, 5H), 1.60 (s, 3H), 1.47–1.37 (m, 1H), 1.29–1.18 (m, 1H), 1.00–0.96 (m, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 156.1, 156.0, 132.21, 132.19, 131.66, 131.65, 128.68, 128.66, 128.49, 128.48, 124.91, 124.90, 123.0, 88.3, 88.0, 85.3, 85.0, 64.9, 64.4, 42.8, 42.5, 37.4, 37.2, 29.7, 29.4, 26.0, 25.66, 25.62, 19.97, 19.89, 18.0; **IR** (neat) 2955, 2924, 1705, 1489, 1392, 1178 cm⁻¹; **HRMS** (TOF MS ES+) *m* /*z* calcd for C₂₁H₂₉NO₂Na (M + Na)⁺ 350.2096, found 350.2083



2.51. The product was prepared according to general method C using phenyl acetylene (0.45 mL, 4.1 mmol), *n*-BuLi (1.9 mL, 4.6 mmol), 3-(benzyloxy)propanal (660 mg, 4.0 mmol), and dimethyl carbamoyl chloride (1.1 mL, 11 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a colorless oil (1.1 g, 3.4 mmol, 85% yield). **TLC R**_f = 0.3 (15% EtOAc/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.43–7.38 (m, 2H), 7.36–7.23 (m, 8H), 5.73 (t, *J* = 6.7 Hz, 1H), 4.53 (s, 2H), 3.74–3.62 (m 2H), 2.93 (s, 3H), 2.89 (s, 3H), 2.28–2.12 (m, 2H); ¹³**C NMR** (125 MHz, CDCl₃) δ 155.8, 138.6, 132.2, 128.8, 128.7, 128.5, 128.0, 127.8, 122.8, 87.5, 85.4, 73.3, 66.4, 63.4, 35.9; **IR** (neat) 2931, 2863, 1699, 1490, 1395, 1184 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₁H₂₃NO₃Na (M + Na)⁺ 360.1576, found 360.1564



2.52. The product was prepared according to general method C using phenyl acetylene (1.1 mL 10 mmol) *n*-BuLi (4.8 mL, 12 mmol), *p*-Tolualdehyde (1.2 mL, 2.0 mmol), and dimethyl carbamoyl chloride (0.41mL, 4.4 mmol). The product was purified by flash column chromatography (20% Et₂O/pentanes) to afford the title compound as a colorless oil (1.14 g, 3.38mmol, 85% yield). **TLC R**_f = 0.3 (20% Et₂O/pentane); ¹**H NMR** (500 MHz, CDCl₃) δ 7.52–7.46, (m, 4H), 7.34–7.27 (m, 3H), 7.22–7.18 (m, 2H), 6.63 (s, 1H), 2.94 (s, 6H), 2.36 (s, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 155.8, 138.9, 135.5, 132.2, 129.6, 128.9, 128.5, 127.9, 122.8, 87.0, 86.8, 67.2, 21.6; **IR** (neat) 2929, 1697, 1490, 1394, 1175 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₁₉H₁₉NO₂Na (M + Na)⁺ 316.1313, found 316.1313



2.53. The product was prepared according to general method A using PdCl₂(PPh₃)₂ (90 mg), CuI (45 mg) 1-butyne-3-ol (0.54 mL, 7.0 mmol), aryl iodide (0.78 mL, 7.0 mmol), trielthyl amine (30 mL), dimethylcarbamoyl chloride (0.71 mL, 7.7 mmol), NaH (200 mg, 8.4 mmol), THF (25 mL) The product was purified by flash column chromatography (20% Et₂O/pentanes) to afford the title compound as a colorless oil (1.3 g, 6.0 mmol, 86% yield). **TLC R**_f = 0.4 (12% EtOA/hexane); ¹**H NMR** (500 MHz, CDCl₃) δ 7.46–7.42 (m ,2H), 7.32–7.25 (m, 3H), 5.65 (q, *J* = 6.6 Hz, 1H), 2.93 (s, 6H), 1.59 (d, *J* = 6.8 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 155.8, 132.1, 128.7, 128.5, 122.8, 88.6, 84.4, 61.9, 20.2; **IR** (neat) 2986, 2935, 1699, 1489, 1392, 1177 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₁₃H₁₅NO₂Na (M + Na)⁺ 240.1001, found 240.0989



2.54 The product was prepared according to general method A using $PdCl_2(PPh_3)_2$ (350 mg, 0.5 mmol), CuI (190 mg, 1.0 mmol) 1-butyne-3-ol (0.83 mL, 11 mmol), aryl bromide (2.3g, 10 mmol), trielthyl amine (30 mL), dimethylcarbamoyl chloride (1.0 mL, 11 mmol), NaH (290 mg, 12 mmol), THF (25 mL) The product was purified by flash column chromatography (12% EtOAc/hexanes) to afford the title compound as a pale yellow oil (2.3 g, 8.0 mmol, 80% yield). **TLC R**_f = 0.2 (12%)

EtOAc/Hexanes); ¹**H** NMR (500 MHz, CDCl₃) δ 7.47–7.39 (m, 4H), 5.79 (s,1H), 5.64 (q, J = 6.7 Hz, 1H), 4.14–3.99 (m, 4H), 2.94 (s, 6H), 1.58 (d, 3, J = 6.7 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 155.9, 138.4, 132.2, 126.7, 123.7, 103.6, 89.2, 84.1, 65.6, 62.0, 22.3; **IR** (neat) 2985, 2886, 1699, 1391, 1176, 1079 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₁₆H₁₉NO₄Na (M + Na)⁺ 312.1212, found 312.1197



2.55 The product was prepared according to general method A using PdCl₂(PPh₃)₂ (250 mg, 0.35 mmol), CuI (130 mg, 0.70 mmol) 1-butyne-3-ol (0.58 mL, 7.7 mmol), aryl iodide (0.82 mL, 7.0 mmol), trielthyl amine (30 mL), dimethylcarbamoyl chloride (0.71 mL, 7.7 mmol), NaH (200 mg, 8.2 mmol), THF (25 mL) The product was purified by flash column chromatography (12% EtOAc/hexanes) to afford the title compound as a pale yellow oil (1.9 g, 8.2 mmol, 82% yield). **TLC R**_f = 0.4 (12% EtOAc/Hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.29–7.20 (m, 2H), 7.16–7.12 (m,1H), 7.05–6.98 (m, 1H), 5.63 (q, *J* = 6.8 Hz, 1H), 2.94 (s, 6H), 1.58 (d, *J* = 6.8 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 162.5 (d, *J* = 246.0 Hz) 155.8, 130.1 (d, *J* = 8.8 Hz) 128.1 (d, *J* = 3.2 Hz) 124.7 (d, *J* = 9.7 Hz), 119.0 (d, *J* = 22.7 Hz), 116.1 (d, *J* = 21.4 Hz), 89.7, 83.2, 61.9, 22.2; **IR** (neat) 2936, 1701, 1579, 1486, 1392, 1170 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₁₃H₁₄FNO₂Na (M + Na)⁺ 258.0906, found 258.0905



2.56. The product was prepared according to general method A using PdCl₂(PPh₃)₂ (180 mg, 0.25 mmol), CuI (95 mg, 0.50 mmol) 1-butyne-3-ol (0.43 mL, 5.5 mmol), aryl iodide (1.2 g, 5.0 mmol), trielthyl amine (30 mL), dimethylcarbamoyl chloride (0.51 mL, 5.5 mmol), NaH (140 mg, 6 mmol), THF (25 mL) The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (0.86 g, 3.3 mmol, 66% yield). **TLC R**_f = 0.4 (12% EtOAc/Hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.89 (d, *J* = 8.4 Hz, 2H), 7.52 (d, *J* = 8.4 Hz, 2H), 5.65 (q, *J* = 6.7 Hz, 1H), 2.95 (s, 6H), 2.60 (s, 3H), 1.60 (d, *J* = 6.9 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 197.7, 155.8, 136.7, 132.3, 128.4, 127.8, 92.0, 83.6, 61.8, 27.0, 22.1; **IR** (neat) 2987, 2936, 1701, 1682, 1392, 1260, 1176 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₁₅H₁₇NO₃Na (M + Na)⁺ 282.1106, found 282.1102



2.57. The product was prepared according to general method A using $PdCl_2(PPh_3)_2$ (180 mg, 0.25 mmol), CuI (95 mg, 0.50 mmol) 1-butyne-3-ol (0.43 mL, 5.5 mmol), aryl iodide (1.2 g, 5.0 mmol), trielthyl amine (30 mL), dimethylcarbamoyl chloride (0.51 mL, 5.5 mmol), NaH (140 mg, 6 mmol), THF (25 mL) The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (0.89 g, 3.6 mmol, 71% yield). **TLC R**_f = 0.4 (12% EtOAc/Hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.37 (d, *J* = 8.6 Hz, 2H), 7.27 (d, *J* = 8.4 Hz, 2H), 5.62 (q, *J* = 6.7 Hz, 1H), 2.94 (s, 6H), 1.58 (d, *J* = 6.6 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 155.8, 134.8, 133.4, 128.9, 121.4, 89.8, 83.3, 61.9, 22.2; **IR** (neat) 2987,

2934, 1701, 1488, 1391, 1177 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₁₃H₁₄ClNO₂Na (M + Na)⁺ 274.0611, found 274.0617



2.58. The product was prepared according to general method A using PdCl₂(PPh₃)₂ (180 mg, 0.25 mmol), CuI (95 mg, 0.50 mmol) 1-butyne-3-ol (0.43 mL, 5.5 mmol), aryl iodide (1.3 g, 5.0 mmol), trielthyl amine (30 mL), dimethylcarbamoyl chloride (0.51 mL, 5.5 mmol), NaH (140 mg, 6 mmol), THF (25 mL) The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (1.0 g, 3.8 mmol, 75% yield). **TLC R**_f = 0.3 (15% EtOAc/Hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.97 (d, *J* = 8.6 Hz, 2H), 7.50 (d, *J* 8.6 = Hz, 2H), 5.64 (q, *J* = 6.7 Hz, 1H), 3.92 (s, 3H), 2.95 (s, 6H) 1.59 (d, *J* = 6.7 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 166.8, 155.8, 132.1, 130.0, 129.7, 127.6, 91.7, 83.7, 61.9, 52.6, 22.1; **IR** (neat) 2988, 2936, 1701, 1393, 1272, 1174, 1085 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₁₅H₁₇NO₄Na (M + Na)⁺298.1055, found 298.1047



2.59. The product was prepared according to general method A using $PdCl_2(PPh_3)_2$ (180 mg, 0.25 mmol), CuI (95 mg, 0.50 mmol) 1-butyne-3-ol (0.43 mL, 5.5 mmol), aryl iodide (0.73 mL, 5.0 mmol), trielthyl amine (30 mL), dimethylcarbamoyl chloride (0.51 mL, 5.5 mmol), NaH (140 mg, 6 mmol), THF (25 mL) The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (1.0 g, 3.7 mmol, 73% yield). **TLC R_f** = 0.4 (15% EtOAc/Hexanes);¹**H NMR** (500 MHz, CDCl₃) δ 7.58–7.52 (m, 4H), 5.64 (q,

J 7.0 = Hz, 1H), 2.95 (s, 6H), 1.59 (d, J = 6.7 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 155.8, 132.4, 130.5 (q, J = 32.8 Hz), 126.7 (q, J = 1.8 Hz), 124.1 (q, J = 271.9 Hz) 125.5 (q, J = 3.7 Hz), 91.2, 83.1, 61.8, 22.0; **IR** (neat) 2938, 1703, 1394, 1319, 1165, 1064 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₁₄H₁₄F₃NO₂Na (M + Na)⁺ 308.0874, found 308.0863



2.60. The product was prepared according to general method B using *p*-methoxy benzaldehyde (1.2 mL 10 mmol) CBr₄ (6.6 g, 20 mmol), PPh₃ (10.5 g 40.0 mmol), *n*-BuLi (8.4 mL, 21 mmol) propionaldehyde (0.72 mL, 10 mmol) and dimethyl carbamoyl chloride (1.0 mL, 11 mmol). The product was purified by flash column chromatography (12 % EtOAc/Hex) to afford the title compound as a pale yellow oil (1.2 g, 4.5 mmol, 45% yield). **TLC R**_f = 0.3 (12% EtOAc/Hexanes); **¹H NMR** (500 MHz, CDCl₃) δ 7.39 (d, *J* = 8.8 Hz, 2H), 6.82 (d, *J* = 8.7, Hz, 2H), 5.50 (t, *J* = 6.5 Hz, 1H), 3.80 (s, 3H), 2.94 (s, 6H), 1.87 (quintet *J* = 7.2 Hz, 2H), 1.07 (t, *J* = 7.3 Hz, 3H); ¹³C **NMR** (125 MHz, CDCl₃) δ 160.0, 156.1, 133.7, 115.1, 114.1, 86.3, 85.1, 67.0, 55.6, 29.0, 9.8; **IR** (neat) 2962, 2930, 1698, 1509, 1248, 1176 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₁₅H₁₉NO₃Na (M + Na)⁺ 284.1263, found 284.1265



2.61. The product was prepared according to general method A using PdCl₂(PPh₃)₂ (180 mg, 0.25 mmol), CuI (95 mg, 0.50 mmol) 1-butyne-3-ol (0.43 mL, 5.5 mmol), aryl iodide (0.73 mL, 5.0 mmol), trielthyl amine (30 mL), dimethylcarbamoyl chloride (0.51 mL, 5.5 mmol), NaH (140 mg,

6 mmol), THF (25 mL) The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (1.0 g, 3.7 mmol, 73% yield). **TLC R_f** = 0.3 (12% EtOAc/Hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.20 (t, *J* = 7.7 Hz, 1H), 7.06–7.02 (m, 1H), 6.99–6.95 (m, 1H), 6.89–6.84 (m, 1H), 5.64 (q, *J* = 6.9 Hz, 1H), 3.79 (s, 3H), 2.94 (s, 6H), 1.58 (d, *J* = 6.7 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 159.5, 155.9, 129.6, 124.7, 123.8, 116.9, 115.5, 88.5, 84.4, 62.0, 55.6, 22.3; **IR** (neat) 2936, 1700, 1574, 1392, 1173 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₁₃H₁₇NO₃Na (M + Na)⁺ 270.1106, found 270.1097

CHARACTERIZATION OF PRODUCTS



2.21 The product was prepared according to general method D using alkyne **2.49** (46.3 mg, 0.2 mmol), phenyl boronic acid (60 mg, 0.5 mmol), Ni(cod)₂ (5.5 mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (43.3 mg, 70% yield). **TLC R**_f = 0.4 (15% EtOAc/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.51–7.46 (m, 2H), 7.43–7.23 (m, 8H), 6.71 (s, 1H), 5.84 (t, *J* = 6.85 Hz, 1H), 2.88 (s, 3H), 2.7 (s, 3H), 1.85 (m, 1H), 1.65 (m, 3H), 0.82 (t, *J* = 7.83 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) 156.3, 141.9, 141.4, 137.3, 132.6, 129.2, 128.9, 128.6, 128.2, 127.5, 127.3, 75.6, 27.5, 10.5; **IR** (neat) 2968, 2934, 1699, 1598, 1443 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₂₀H₂₃NO₂Na (M + Na)⁺ 332.1627, found 332.1611



2.26. The product was prepared according to general method D using alkyne **2.54** (57.9 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5 mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (41.9 mg, 57% yield). **TLC R**_f = 0.3 (25% EtOAc/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.51–7.45 (m, 4H), 7.41–7.28 (m, 5H), 6.64 (s, 1H), 6.02 (q, *J* = 6.9 Hz, 1H), 5.83 (s, 1H), 4.16–4.01 (m, 4H), 2.86 (s, 3H), 2.70 (s, 3H), 1.39 (d, *J* = 6.8 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) 156.1, 143.4, 141.0, 138.1, 137.0, 131.0, 129.2, 128.9, 128.2, 127.6, 126.8, 103.9, 70.6, 65.7, 65.6, 30.1, 20.8; **IR** (neat) 2926, 1698, 1442, 1390, 1183, 1080 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₂₂H₂₅NO₄Na (M + Na)⁺ 390.1681, found 390.1668



2.27 The product was prepared according to general method D using alkyne **2.55** (47.1 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (54.5 mg, 87% yield). TLC $\mathbf{R}_{\mathbf{f}} = 0.3$ (25% EtOAc/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.48–7.44 (m, 2H), 7.39–7.29 (m, 4H), 7.17–7.13 (m, 1H), 7.10–7.05
(m, 1H), 6.93 (dt, J = 8.4, J = 2.4 Hz, 1H), 6.58 (s, 1H), 5.99 (q, J = 6.8 Hz, 1H), 2.87 (s, 3H), 2.71 (s, 3H), 1.41 (d, J = 6.8 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 162.9 (d, J = 245.5 Hz) 156.0, 144.2, 140.8, 139.4 (d, J = 7.9 Hz), 130.1 (d, J = 8.3 Hz), 130.0 (d, J = 1.8 Hz), 128.9, 128.3, 127.7, 124.9, (d, J = 3.2 Hz), 116.1 (d, J = 21.7 Hz), 114.3 (d, J = 21.3 Hz), 70.4, 20.8; **IR** (neat) 2985, 2933, 1698, 1489, 1391, 1180 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₁₉H₂₀FNO₂Na (M + Na)⁺ 336.1376, found 336.1362



2.29. The product was prepared according to general method D using alkyne **2.59** (57.1 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (60.3mg, 83% yield). **TLC R**_f = 0.4 (10% EtOAc/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.65–7.61 (m, 2H), 7.51–7.46 (m, 4H), 7.40–7.31 (m, 3H), 6.63 (s, 1H), 5.95 (q, *J* = 6.7 Hz, 1H), 2.87 (s, 3H), 2.68 (s, 3H), 1.40 (d, *J*=6.7 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 156.0, 144.9, 140.9 (q, *J* = 1.4 Hz), 140.6, 129.7, 129.4, 129.3 (q, *J* = 132.4 Hz), 128.8, 128.4, 127.9, 125.6 (q, *J* = 3.7 Hz), 124.5 (q, *J* = 271.9), 70.4, 20.8; **IR** (neat) 2980, 2934, 1700, 1615, 1393, 1323, 1121 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₀H₂₀F₃NO₂Na (M + Na)⁺ 386.1344, found 386.1335



2.31. The product was prepared according to general method D using alkyne **2.50** (65.5 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (53.5 mg, 66% yield). TLC $R_f = 0.6$ (15% EtOAc/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.50–7.44 (m, 2H), 7.44–7.40 (m, 2H), 7.38–7.32 (m, 4H), 7.31–7.27 (m, 1H), 7.27-7.22 (m, 1H), major diastereomer 6.69 (s, 0.52H), minor diastereomer 6.64 (s, 0.46H), 6.06-5.99 (m, 1H), 4.99-4.91 (m, 1H), major diastereomer 2.92- 2.80 (m, 3H) minor diastereomer 2.77-2.58 (m, 3H), 1.97–1.72 (m, 2H), 1.69–1.59 (m, 4H), 1.54–1.49 (m, 3H), 1.49– 133 (m, 2H), 1.22–0.90 (m, 2H), minor diastereomer 0.73 (d, J = 6.6 Hz, 1.3H) major diastereomer 0.68 (d, J = 6.7 Hz, 1.59H); ¹³C NMR (125 MHz, CDCl₃) δ 156.3, 156.2, 143.1, 142.4, 141.28, 141.24, 137.4, 137.3, 132.3, 131.33, 131.28, 131.26, 129.10, 129.09, 128.90, 128.89, 128.6, 128.5, 128.19, 128.16, 127.48, 127.45, 127.3, 127.2, 125.03, 125.01, 73.2, 72.7, 41.9, 41.3, 37.5, 37.0, 29.6, 29.4, 26.01, 26.00, 25.7, 25.4, 19.9, 19.4, 17.9; **IR** (neat) 3022, 2959, 1700, 1493, 1392, 1186 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₇H₃₅NO₂Na (M + Na)⁺ 428.2566, found 428.2574



2.30. The product was prepared according to general method D using alkyne **2.51** (67.5 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and

concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (58.2 mg, 70% yield). **TLC R**_f = 0.6 (20% EtOAc/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.48–7.40 (m, 4H), 7.36–7.19 (m, 11H), 6.67 (s, 1H), 6.16 (dd, *J*=8.9 Hz, *J* = 5.3 Hz 1H), 4.35 (s, 2H), 3.50–3.40 (m, 2H), 2.84 (s, 3H), 2.64 (s, 3H), 2.22–2.12 (m, 1H), 2.06, 1.97 (m, 1H); ¹³**C NMR** (125 MHz, CDCl₃) δ 156.1, 141.9, 141.2, 138.7, 137.1, 132.3, 129.3, 128.9, 128.6, 128.5, 128.2, 127.9, 127.7, 127.5, 127.3, 73.1, 71.6, 67.0, 35.0; **IR** (neat) 3056, 2927, 1699, 1494, 1391, 1179 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₇H₂₉NO₃Na (M + Na)⁺ 438.2045, found 438.2048



2.47 The product was prepared according to general method D using alkyne **2.53** (43.5 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (41.9 mg, 71% yield). **TLC R**_f = 0.4 (15% EtOAc/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.51–7.47 (m, 2H), 7.40–7.24 (m, 7H), 6.65 (s, 1H), 6.05 (q, *J* = 6.6 Hz, 1H), 2.87 (s, 3H), 2.72 (s, 3H), 1.40 (d, *J* = 6.7 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 156.1, 142.9, 141.2, 137.1, 131.5, 129.2, 129.0, 128.7, 128.2, 127.5, 127.4, 70.5, 20.8; **IR** (neat) 3022, 2932, 1698, 1491,m 1392, 1185 cm⁻¹; **HRMS** (TOF MS ES+) *m* /*z* calcd for C₁₉H₂₁NO₂Na (M + Na)⁺ 318.1470, found 318.1469



2.32 The product was prepared according to general method D using alkyne **2.52** (58.7 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (60.2 mg, 81% yield). **TLC R**_f = 0.4 (15% EtOAc/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.52 (d, J = 8.1 Hz, 2H), 7.45–7.38 (m, 4H), 7.34–7.28 (m, 6H), 7.22 (s, 1H), 7.19 (d, *J* = 7.7 Hz, 2H), 7.05, (s, 1H), 2.92 (s, 3H), 2.68 (s, 3H), 2.40 (s, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 156.1, 140.5, 140.4, 137.4, 137.03, 136.94, 132.4, 129.4, 129.2, 128.78, 128.76, 128.1, 127.6, 127.5, 126.7, 74.6, 21.5; **IR** (neat) 3023, 2923, 1703, 1492, 1389, 1173 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₅H₂₅NO₂Na (M + Na)⁺ 394.1783, found 394.1784



2.28 The product was prepared according to general method D using alkyne **2.57** (50.3 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (52.1 mg, 79% yield). TLC $\mathbf{R}_{f} = 0.5$ (15% EtOAc/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.48–7.44 (m, 2H), 7.38–7.28 (m, 7H), 6.57 (s, 1H), 5.99 (q, *J* = 6.7 Hz, 1H) 2.87 (s, 3H), 2.72 (s, 3H), 1.38 (d, *J* = 6.7 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 156.0, 143.7, 140.8, 135.6, 133.3, 130.5, 130.1, 128.85, 128.82, 128.3, 127.7, 70.3, 20.7; IR (neat) 2977,

2930, 1698, 1490, 1391, 1183 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₁₉H₂₀ClNO₂Na (M + Na)⁺ 352.1080, found 352.1077



2.24. The product was prepared according to general method D using alkyne **2.56** (51.9 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (50.6mg, 75% yield). **TLC R**_f = 0.3 (25% EtOAc/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 8.00–7.95 (m, 2H), 7.51–7.45 (m, 4H), 7.40–7.31 (m, 3H), 6.66 (s, 1H), 6.01 (q, *J* = 6.7 Hz, 1H), 2.87 (s, 3H), 2.72 (s, 3H), 2.61 (s, 3H), 1.40 (d, *J* = 6.7 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 156.0, 144.8, 142.1, 140.6, 135.9, 130.3, 129.4, 128.83, 128.76, 128.3, 127.8, 70.3, 27.0, 20.7; **IR** (neat) 2979, 2931, 1699, 1682, 1267, 1182 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₂₁H₂₃NO₃Na (M + Na)⁺ 360.1576, found 360.1572



2.25 The product was prepared according to general method D using alkyne **2.58** (55.1 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title

compound as a pale yellow oil (50.9mg, 72% yield. **TLC** $\mathbf{R}_{\mathbf{f}} = 0.3$ (15% EtOAc/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 8.06–8.03 (m, 2H), 7.50–7.46 (m, 2H), 7.46–7.42 (m, 2H), 7.39–7.31 (M, 3H), 6.65 (s, 1H), 5.99 (q, J = 6.7 Hz, 1H), 3.92 (s, 3H), 2.86 (s, 3H), 2.69 (s, 3H), 1.43 (d, J = 6.7Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 167.2, 156.0, 144.7, 142.0, 140.7, 130.3, 129.9, 129.2, 128.9, 128.8, 128.3, 127.8, 70.4, 52.4, 20.7; **IR** (neat) 2929, 1718, 1698, 1391, 1275, 1178 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₁H₂₃NO₄Na (M + Na)⁺ 376.1525, found 376.1522



2.22 The product was prepared according to general method D using alkyne **2.61** (49.5 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (35.1 mg, 54% yield). TLC $\mathbf{R}_{f} = 0.3$ (12% EtOAc/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.50–7.46 (m, 2H), 7.39–7.26 (m, 4H), 6.98–6.94 (m, 1H), 6.93–6.90 (m, 1H), 6.85–6.80 (m, 1H), 6.63 (s, 1H), 6.06 (q, *J* = 6.4 Hz 1H), 3.83 (s, 3H), 2.87 (s, 3H) 2.74 (s, 3H), 1.40 (d, *J* = 6.6 Hz 3H); ¹³C NMR (125 MHz, CDCl₃) δ 159.8, 143.1, 141.1, 138.5, 131.5, 129.7, 128.9, 128.2, 127.6, 121.6, 114.4, 113.4, 70.6, 55.5, 20.8; IR (neat) 2928, 1697, 1488, 1391, 1268, 1182 cm⁻¹; HRMS (TOF MS ES+) *m*/*z* calcd for C₂₀H₂₃NO₃Na (M + Na)⁺ 348.1576, found 348.1577



2.23 The product was prepared according to general method D using alkyne **2.61** (49.5 mg, 0.2 mmol), phenyl boronic acid (61 mg, 0.5 mmol), Ni(cod)₂ (5.5 mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (32.5 mg, 50% yield). **TLC R**_f = 0.3 (12% EtOAc/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.50–7.46 (m, 2H), 7.39–7.26 (m, 5H), 6.94–6.88 (m, 2H), 6.65 (S, 1H), 5.89 (t, *J* = 7.34 Hz, 1H), 3.82 (s, 3H), 2.89 (s, 3H), 2.78 (s, 3H), 1.84 (m, 1H), 1.64 (m, 1H), 0.82 (t, *J* = 7.58 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 158.9, 141.5, 140.5, 132.5, 130.5, 129.7, 128.9, 128.2, 127.3, 114.1, 75.5, 55.6, 27.4, 10.5; **IR** (neat) 2929, 1698, 1573, 1442, 1298, 1033 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₂₁H₂₅NO₃Na (M + Na)⁺ 362.1732, found 362.1730



2.38 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 2-benzofuranylboronic acid (81 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (51.0 mg, 73% yield). TLC $\mathbf{R}_{\mathbf{f}} = 0.3$ (15% EtOAc/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.57–7.50 (m, 4H), 7.47 (d, J = 8.2 Hz, 1H), 7.40 (t, J = 7.7 Hz, 2H),

7.31–7.23 (m, 2H), 7.20 (t, J = 7.6 Hz, 1H), 6.96 (s, 1H), 6.03 (dd, J = 8.6, J = 6.4 Hz, 1H), 2.93 (s, 3H), 2.88 (s, 3H), 2.18–2.08 (m, 1H), 1.96–1.86 (m, 1H), 0.90 (t, J = 7.5 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 156.3, 154.4, 154.3, 136.6, 130.8, 130.5, 129.5, 129.4, 128.8, 127.8, 124.8, 123.1, 121.2, 111.2, 105.3, 74.0, 28.1, 10.8; **IR** (neat) 3057, 2967, 1699, 1451, 1390, 1181 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₂H₂₃NO₃Na (M + Na)⁺ 372.1576, found 372.1582



2.43 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 4-(Trifluoromethyl) phenyl boronic acid (95.0 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (62.6 mg, 83% yield). **TLC R**r = 0.4 (15% EtOAc/Hex); ¹**H NMR** (500 MHz, CDCl₃) δ 7.63–7.58 (m, 4H), 7.43–7.36, (m, 4H), 7.31–7.26 (m, 1H), 6.73 (s, 1H), 5.85 (t, *J* = 7.3 Hz, 1H), 2.89 (s, 3H), 2.73 (s, 3H), 1.88–1.80 (m, 1H), 1.68–1.59 (m,1H), 0.83 (t, *J* = 7.5 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 156.2, 145.1 (q, *J* = 1.4 Hz), 140.9, 136.7, 133.7, 129.6 (q, *J* = 32.4 Hz)129.23, 129.16, 128.7, 127.7, 125.2 (q, *J* = 3.7 Hz), 124.5 (q, *J* = 127.4 Hz) 75.4, 27.6, 10.5; **IR** (neat) 2969, 2934, 1699,1322, 1164, 1122 cm⁻¹; **HRMS** (TOF MS ES+) *m/z* calcd for C₂₁H₂₂F₃NO₂Na (M + Na)⁺ 400.1500, found 400.1489



2.41 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 3-chlorophenylboronic acid (78 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (59.8 mg, 87% yield). **TLC R**_f = 0.4 (15% EtOAc/Hex); ¹**H NMR** (500 MHz,CDCl₃) δ 7.51–7.49 (m, 1H), 7.42–7.34 (m, 5H), 7.31–7.26 (m, 3H), 6.71 (s, 1H), 5.83 (dd, *J* = 8.1, *J* = 6.7, Hz, 1H), 2.89 (s, 3H), 2.76 (s, 3H), 1.88–1.78 (m, 1H), 1.67–1.58 (m, 1H), 0.82 (t, *J* = 7.3 Hz, 3H); ¹³**C NMR** (125 MHz,CDCl₃) δ 156.2, 143.1, 140.7, 136.8, 134.0, 133.3, 129.5, 129.2, 129.1, 128.7, 127.59, 127.58, 127.1, 75.3, 27.5, 10.5; **IR** (neat) 2968, 2933, 1699, 1472, 1390, 1182 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₀H₂₂ClNO₂Na (M + Na)⁺, 366.1237 found 366.1227



2.40 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 4-(tert-butoxycarbonylaminomethyl)phenylboronic acid (126 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a

silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (57.0 mg, 65% yield). **TLC R**_f = 0.3 (25% EtOAc/Hex); ¹H NMR (500 MHz,CDCl₃) δ 7.46 (d, *J* = 8.2 Hz, 2H), 7.42–7.33 (m, 4H), 7.29–7.23 (m, 3H), 6.69 (s, 1H), 5.83 (t, *J* = 7.2 Hz, 1H), 4.88 (s, 1H), 4.34 (d, *J* = 5.5 Hz, 2H), 2.88 (s, 3H), 2.74 (s, 3H), 1.88–1.78 (m, 1H), 1.66–1.58 (m, 1H), 1.48 (s, 9H), 0.80 (t, *J* = 7.5 Hz, 3H); ¹³C NMR (125 MHz,CDCl₃) δ 156.3, 141.5, 140.4, 137.2, 132.6, 129.18, 129.11, 128.6, 127.3, 75.6, 44.7, 28.8, 27.5, 10.5; **IR** (neat) 3339, 2970, 2932, 1699, 1507, 1175 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₂₆H₃₄N₂O₄Na (M + Na)⁺, 461.2416 found 461.2411



2.33 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 3-benzyloxyphenylboronic acid (114 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (48.2 mg, 58% yield). **TLC R**_f = 0.3 (25% EtOAc/Hex); ¹**H NMR** (500 MHz,CDCl₃) δ 7.47–7.43 (m, 2H), 7.41–7.31 (m, 6H), 7.28–7.23 (m, 2H), 7.15–7.13 (m, 1H), 7.12–7.09 (m, 1H), 6.93 (ddd, *J* = 8.3, *J* = 2.6 *J* = 0.7 Hz, 1H), 6.72 (s, 1H), 5.82 (dd, *J* = 7.8, *J* = 6.8 Hz, 1H), 5.09 (s, 2H), 2.89 (s, 3H), 2.73 (s, 3H), 1.89–1.79 (m, 1H), 1.70–1.60 (m, 1H), 0.81 (t, *J* = 7.5, 3H); ¹³C **NMR** (125 MHz,CDCl₃) δ 158.7, 156.3, 142.8, 141.7, 137.4, 137.2, 132.7, 129.20, 129.18, 128.9, 128.6, 128.3, 127.8, 127.4, 121.7, 115.7, 113.8, 75.6, 70.3, 27.5, 10.6; **IR** (neat) 2967, 2930, 1698, 1574, 1391, 1182 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₂₇H₂₉NO₃Na (M + Na)⁺, 438.2045 found 438.2029



2.39 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 1-Methyl-5-indolylboronic acid (87 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (44.9 mg, 62% yield). **TLC R**_f = 0.2 (15% EtOAc/Hex); ¹**H NMR** (500 MHz,CDCl₃) δ 7.75–7.74 (m, 1H), 7.44–7.34 (m, 5H), 7.31–7.28 (m, 1H), 7.26–7.22 (m, 1H), 7.06 (d, *J* = 3.1 Hz, 1H), 6.74 (s, 1H), 6.49 (dd, *J* = 3.2 *J* = 0.7 Hz, 1H), 5.88 (t, *J* = 7.3 Hz, 1H), 3.80 (s, 3H), 2.89 (s, 3H), 2.74 (s, 3H), 1.91–1.81 (m, 1H), 1.73–1.64 (m, 1H), 0.81 (t, *J* = 7.5 Hz, 3H); ¹³**C NMR** (125 MHz,CDCl₃) δ 156.5, 142.7, 137.9, 136.4, 132.7, 132.0, 129.5, 129.2, 128.5, 127.0, 123.0, 121.2, 108.8, 101.4, 76.2, 33.2, 27.6, 10.6; **IR** (neat) 2964, 2931, 2695, 1488, 1391, 1186 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₃H₂₆N₂O₂Na (M + Na)⁺, 385.1892 found 385.1900



2.32 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 4-trimethylsilylphenylboronic acid (97 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure

Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (59.5 mg, 78% yield). **TLC** $\mathbf{R}_{\mathbf{f}} = 0.5$ (15% EtOAc/Hex); ¹**H NMR** (500 MHz,CDCl₃) δ 7.52–7.46 (m, 4H), 7.41–7.34 (m, 4H), 7.28–7.23 (m, 1H), 6.73 (s, 1H), 5.86 (dd, J = 7.7 J = 6.7 Hz, 1H), 2.89 (s, 3H), 2.75 (s, 3H), 1.89–1.80 (m,1H), 1.70–1.60 (m, 1H), 0.81 (t, J = 7.5 Hz, 3H), 0.29 (s, 9H); ¹³C **NMR** (125 MHz,CDCl₃) δ 156.4, 141.76, 141.74, 139.5, 137.4, 133.3, 132.8, 129.2, 128.6, 128.2, 127.3, 75.6, 27.5, 10.5, –0.7; **IR** (neat) 2955, 1703, 1460, 1392, 1248, 1183 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₃H₃₁NO₂SiNa (M + Na)⁺, 404.2022 found 404.2018



2.36 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 3-thienylboronic acid (64 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (47.3mg, 75% yield). **TLC R**_f = 0.4 (15% EtOAc/Hex); ¹**H NMR** (500 MHz,CDCl₃) δ 7.45–7.34 (m, 4H), 7.33–7.28 (m, 2H), 7.28–7.24 (m, 2H), 6.95 (s, 1H), 5.89 (dd, *J* = 8.1, *J* = 6.9 Hz, 1H), 2.88 (s, 3H), 2.85 (s, 3H), 1.98–1.87 (m, 1H), 1.77–1.68 (m, 1H), 0.83 (t, *J* = 7.5 Hz, 3H); ¹³**C NMR** (125 MHz,CDCl₃) δ 156.3, 140.8, 137.2, 136.3, 131.1, 129.2, 128.7, 128.0, 127.4, 125.0, 122.5, 75.3, 27.7, 10.6; **IR** (neat) 2969, 1702, 1491, 1390, 1183 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₁₈H₂₁NO₂SNa (M + Na)⁺, 338.1191 found 338.1184



2.34 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 3-methylthiophenylboronic acid (84 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (49.1mg, 69% yield). TLC $\mathbf{R}_{\mathbf{f}} = 0.4$ (15% EtOAc/Hex); ¹H NMR (500 MHz,CDCl₃) δ 7.42–7.35 (m, 5H), 7.30–7.24 (m, 3H), 7.20 (dt, J = 7.1, J = 1.8 Hz, 1H) 6.70 (s, 1H), 5.83 (dd, J = 8.0, J = 6.7 Hz, 1H), 2.89 (s, 3H), 2.75 (s, 3H), 2.51 (s, 3H), 1.88–1.79 (m, 1H), 1.69–1.60 (m, 1H), 0.82 (t, J = 7.5 Hz, 3H); ¹³C NMR (125 MHz,CDCl₃) δ 156.3, 142.0, 141.5, 138.2, 137.1, 132.9, 129.2, 128.7, 128.6, 127.4, 127.2, 125.8, 125.6, 75.5, 27.5, 16.2, 10.5; **IR** (neat) 2966, 1698, 1584, 1443, 1390, 1182 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₁H₂₅NO₂SNa (M + Na)⁺, 378.1504 found 378.1510



2.37 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 3-furanboronic acid (56 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (34.1 mg, 57% yield). **TLC R**_f = 0.3 (15% EtOAc/Hex); ¹H **NMR** (500 MHz,CDCl₃) δ 7.67 (s, 1H), 7.43–7.40 (m, 2H), 7.39–7.34 (m, 2H), 7.27–7.23 (m, 2H), 6.88

(s, 1H), 6.65–6.63 (m, 1H), 5.87 (dd, J = 8.3, J = 6.6 Hz, 1H), 2.90 (s, 6H), 2.0–1.89 (m, 1H), 1.78–1.69 (m, 1H), 0.84 (t, J = 7.6 Hz, 3H); ¹³C NMR (125 MHz,CDCl₃) δ 156.3, 142.8, 140.5, 137.1, 132.6, 129.2, 129.1, 128.7, 127.3, 124.2, 109.8, 75.2, 27.7, 10.6; **IR** (neat) 2971, 1698, 1491, 1394, 1264, 1188 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₁₈H₂₁NO₃Na (M + Na)⁺, 322.1419 found 322.1426



2.44 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 4-acetylphenylboronic acid (82 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (56.2 mg, 80% yield). TLC $\mathbf{R}_{\mathbf{f}}$ = 0.2 (12% EtOAc/Hex); ¹H NMR (500 MHz,CDCl₃) δ 7.95 (d, *J* = 8.2 Hz, 2H) 7.60 (d, *J* = 8.3 Hz, 2H), 7.43–7.35 (m, 4H), 7.31–7.27 (m, 1H), 6.77 (s, 1H), 5.87 (dd, *J* = 7.7 , *J* =6.8 Hz, 1H), 2.88 (s, 3H), 274 (s, 3H), 2.63 (s, 3H), 1.93–1.80 (m, 1H) 1.71–1.59 (m, 1H) 0.82 (t, *J* = 7.4 Hz, 3H); ¹³C NMR 198.2, 156.2, 146.4, 141.2, 136.8, 136.2, 133.7, 129.2, 129.1, 128.7, 128.4, 127.7, 75.4, 27.7, 27.0, 10.5; IR (neat) 2963, 2925, 1698, 1681, 1602, 1265, 1180 cm⁻¹; HRMS (TOF MS ES+) *m* / *z* calcd for C₂₂H₂₅NO₃Na (M + Na)⁺, 374.1732 found 374.1729



2.44 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 2-fluorophenylboronic acid (70 mg, 0.5 mmol), Ni(cod)₂ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (47.1 mg, 72% yield). **TLC R**_f = 0.4 (12% EtOAc/Hex); ¹**H NMR** (500 MHz,CDCl₃) δ 7.41–7.43 (m, 2H), 7.40–7.35 (m, 2H), 7.33–7.26 (m, 3H), 7.14–7.06 (m, 2H), 6.60 (s, 1H), 5.87 (dd, *J* = 8.4 , *J* = 6.0 Hz, 1H), 2.89 (s, 3H), 2.68 (s, 3H), 1.81–1.58 (m, 2H), 0.85 (t, *J* = 7.4 Hz, 3H); ¹³**C NMR** (125 MHz,CDCl₃) δ 160.6 (d, *J* = 245.5 Hz), 156.4, 134.1, (*J* = 1.4 Hz), 132.3 (*J* = 3.7 Hz), 131.9, 129.3, 129.21, 129.18, 128.7, 127.5, 123.6 (d, *J* = 3.7), 115.7 (d, *J* = 23.1 Hz), 75.3, 27.3, 10.5; **IR** (neat) 2931, 1698, 1487, 1391, 1182 cm⁻¹; **HRMS** (TOF MS CI+) *m*/*z* calcd for C₂₀H₂₂FNO₂ (M)⁺, 327.1635 found 327.1630



2.35 The product was prepared according to general method D using alkyne **2.17** (46.3 mg, 0.2 mmol), 4-acetoxyphenylboronic acid (90 mg, 0.5 mmol), $Ni(cod)_2$ (5.5mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title

compound as a pale yellow oil (47.0 mg, 64% yield). **TLC** $\mathbf{R}_{\mathbf{f}} = 0.3$ (12% EtOAc/Hex); ¹**H NMR** (500 MHz,CDCl₃) δ 7.51–7.47 (m, 2H), 7.41–7.33 (m ,4H), 7.28–7.23 (m, 1H), 7.10–7.05 (m, 2H), 6.70 (s, 1H), 5.83 (dd, J = 8.0, J = 6.6 Hz, 1H), 2.87 (s, 3H), 2.72 (s, 3H), 2.32 (s, 3H), 1.91–1.79 (m, 1H), 1.71–1.60 (m ,1H), 0.82 (t, J = 7.4 Hz, 3H); ¹³**C NMR** (125 MHz,CDCl₃) δ 169.8, 156.3, 150.2, 141.1, 139.0, 137.2, 132.8, 129.9, 129.2, 128.6, 127.4, 121.3, 75.6, 27.5, 21.5, 10.6; **IR** (neat) 2926, 1760, 1698, 1503, 1186, 1166 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₂H₂₅NO₄Na (M + Na)⁺, 390.1681 found 390.1684

DEUTERIUM LABELING EXPERIMENTS



2.48 The product was prepared according to general method D using alkyne **2.17** (43.5 mg, 0.2 mmol), phenyl boronic acid-D₂ **2.62** (62 mg, 0.5 mmol), Ni(cod)₂ (5.5 mg, 0.02 mmol), TrixiePhos (8.0 mg, 0.02 mmol). The reaction mixture was eluted through a silica plug (with pure Et₂O) and concentrated under reduced pressure. Purified by flash column chromatography to afford the title compound as a pale yellow oil (42.2 mg, 68% yield). TLC $\mathbf{R_f} = 0.4$ (15% EtOAc/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.51–7.46 (m, 2H), 7.43–7.23 (m, 8H), 6.71 (s, (12% H incorporation,

1H), 5.84 (t, J = 6.85 Hz, 1H), 2.88 (s, 3H), 2.7 (s, 3H), 1.85 (m, 1H), 1.65 (m, 3H), 0.82 (t, J = 7.83 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) 156.3, 141.9, 141.4, 137.3, 132.6, 129.2, 128.9, 128.6, 128.2, 127.5, 127.3, 75.6, 27.5, 10.5; **IR** (neat) 2966, 2932, 1698, 1493, 1390, 1179 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₀H₂₂DNO₂Na (M + Na)⁺ 333.1689, found 333.1677.

SYNTHESIS OF TAMOXIFEN



2.63. The Grignard reagent was prepared by adding to Taillefer's method 25 mL round bottom flask: aryl bromide (0.80 g, 3.3 mmol), magnesium turnings (160 mg, 6.6 mmol, 2 equiv), THF (3 mL), and 1,2-dibromoethane (70 μ L). The mixture was stirred at room temperature for 3 hours, resulting in a solution of ~ 0.55 M of Grignard reagent. To a 7 dram vial was added CuI (76 mg, 0.40 mmol) and Grignard reagent (1.5 mL, 0.80 mmol) at 0 °C, and the mixture was stirred for 30 minutes. The mixture was then cooled to -25 °C and a solution of allylic carbamate (30 mg, 0.10

mmol, 0.1 M in THF) was added dropwise over 40 minutes. The mixture was stirred for 6 hours at –25 °C then quenched with saturated aqueous ammonium chloride, extracted with EtOAc, dried over Mg₂SO₄ and concentrated under reduced pressure. The product was purified by reverse-phase HPLC with H₂O/CH₃CN (ZORBAX SB-C18 column, gradient elution of 40–70% CH3CN w/ 0.1% TFA) to afford the compound as a colorless oil (24.1 mg, 65% yield). **TLC R**_f = 0.3 (15% EtOAc/hexanes w/ 5% NEt₃); ¹**H NMR** (500 MHz, CDCl₃) δ 7.37–7.28 (m, 5H), 7.27–7.16 (m, 7H), 6.78 (d, *J* = 8.7 Hz, 2H), 5.24 (q, *J* = 7.0 Hz, 1H), 5.08 (s, 1H), 4.30 (t, *J* = 3.8 Hz, 2H), 3.53 (t, *J* = 3.9 Hz, 2H), 2.99 (s, 6H), 1.58 (d, *J* = 6.9 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 155.7, 144.2, 142.9, 141.8, 136.8, 131.2, 129.8, 129.1, 128.5, 128.3, 126.8, 126.6, 126.5, 114.3, 62.6, 58.9, 57.2, 44.3, 15.3; **IR** (neat) 3027, 1609, 1509, 1238, 1175⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₆H₂₉NOH (M + H)⁺ 372.2327, found 372.2328



2.13 The product was prepared according to a modified procedure reported by Taillefer.¹² **2.63** (9.1 mg, 0.024 mmol) was dissolved in dry DMSO (0.8 mL) and KO*t*Bu (14 mg, 0.12 mmol, 5 equiv) was added. The mixture was then heated at 50 °C for 5 h. The crude reaction mixture was then filtered through a pad of silica and eluted with Et₂O then concentrated under reduced pressure. The product was purified by flash column chromatography to afford the title compound as a colorless oil (7.6 mg, 84% yield, 1:1 *E:Z*). ¹**H NMR** (500 MHz, CDCl₃) δ 7.38–7.33 (m, 2H),

¹² Danoun, G.; Tlili, A.; Monnier, F.; Taillefer, M. Angew. Chem. Int. Ed. 2012, 51, 12815

7.30–7.23 (m, 4H), 7.21–7.08 (m, 12H), 7.02–6.97 (m, 2H), 6.93–6.87 (m, 4H), 6.80–6.75 (m, 2H), 6.59–6.55 (m, 2H), 4.10 (t, J = 6.1 Hz, 1H), 3.94 (t, J = 6.0 Hz, 1H), 2.76 (t, J = 5.8 Hz, 1H), 2.65 (t, J = 5.8 Hz, 1H), 2.52 (q, J = 7.4 Hz, 1H), 2.47 (q, J = 7.4 Hz, 1H), 2.36 (s, 6H), 2.29 (s, 6H), 0.94 (q, J = 8.6 Hz, 6H).






















































































Chapter 3

Intramolecular Nickel-Catalyzed Reductive Cross-Electrophile Coupling Reactions of Benzylic Esters with Aryl Halides

3.1 Introduction

Nickel-catalyzed reductive cross-electrophile coupling reactions have recently undergone rapid advances in synthetic utility and mechanistic understanding.¹ Their mild reaction conditions can provide advantages to traditional cross-coupling reactions. For example, reductive coupling reactions offer an attractive strategy toward intramolecular cyclization reactions because they do not require installation of both electrophilic and organometallic functional groups into the starting material.² We envisioned a cross-electrophile reductive coupling reaction would provide straightforward synthesis of 1-arylindanes and tetralins, common motifs in natural products and pharmaceutical agents (Scheme 3.1).^{3,4} While intermolecular reductive coupling reactions have undergone rapid development in recent years,^{1,5} few intramolecular variants have been reported. The Peng group disclosed a stoichiometric nickel-catalyzed reductive coupling reaction to access nitrogen- and oxygen-containing heterocycles (Scheme 3.2a).⁶ In 2014, Gong and coworkers

¹(a) Knappke, E. I. C.; Grupe, G.; Gartner, D.; Corpet, M.; Gosmini, C.; Wangelin, A. J. V. *Chem. Eur. J.* **2014**, *20*, 6828. (b) Weix, D. *Acc. Chem. Res.* **2015**, *48*, 1767. (c) Moragas, T.; Correa, A.; Martin, R. *Chem. Eur. J.* **2014**, *20*, 8242. (d) Gu, J.; Wang, X.; Xue, W.; Gong, H. *Org. Chem. Front.* **2015**, *2*, 1141.

² Traditional cross-coupling reactions have had a transformative impact on synthesis. For reviews, see: (a) Bulger, P. G.; Sarlah, D.; Nicolaou, K. C. *Angew. Chem. Int. Ed.* **2005**, *44*, 4442. (b) Chemler, S. R.; Trauner, D.; Danishefsky, S. J. *Angew. Chem. Int. Ed.* **2001**, *40*, 4544.

³For lead references in alternative methods for synthesis of indanes and tetralins, see: (a) Parham, W. E.; Bradsher, C. K.; Reames, D. C. *J. Org. Chem.* **1981**, *46*, 4804. (b) Miyaura, N.; Ishiyama, T.; Sasaki, H.; Ishikawa, M.; Satoh, M.; Suzuki, A. *J. Am. Chem. Soc.* **1989**, *111*, 314. (c) Bailey, W. F.; Longstaff, S. C. *J. Org. Chem.* **1998**, *63*, 432. (d) Segura, J. L.; Martin, N. *Chem. Rev.* **1999**, *99*, 3199. (f) Begouin, J-M.; Capitta, F.; Wu, X.; Niggemann, M. *Org. Lett*, **2013**, *15*, 1370. (g) Wang, Y.-M.; Bruno, N. C.; Placeres, A. L.; Zhu, S.; Buchwald, S. L. *J. Am. Chem.*

Soc. 2015, 137, 10524.

⁴For representative nickel-catalyzed reactions that generate indanes, see: (a) Deng, R.; Sun, L.; Li, Z. *Org. Lett.*, **2007**, *9*, 5207. (b) Watson, M. P.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2008**, *130*, 12594. (c) Arp, F. O.; Fu, G. C. *J. Am. Chem. Soc.* **2005**, *127*, 10482.

⁵For lead references, see: (a) Kadunce, N. T.; Reisman S. E. *J. Am. Chem. Soc.* **2015**, *137*, 10480. (b) Ackerman, L. K. G.; Lovell, M. M.; Weix, D. J. *Nature* **2015**, *524*, 454.

⁶ Yan, C. S.; Peng, Y.; Xu, X. B.; Wang, Y. W. Chem. Eur. J. 2012, 18, 6039.

reported a catalytic intramolecular cyclization of dihaloalkanes to access 5- and 6-membered rings (Scheme 3.2b).⁷ Recently, much interest has focused on the use of C–O electrophiles in reductive coupling reactions.^{8,9} Our laboratory has reported a reductive ring-contraction of 4- chlorotetrahydropyrans to generate cyclopropanes (Scheme 3.2c).¹⁰ To further expand the scope of intramolecular cross-electrophile coupling reactions, we targeted cyclization reactions of benzylic esters with aryl halides to afford valuable indanes and tetralins (Scheme 3.2d). These reactions would also complement recent efforts to develop intermolecular reductive coupling reactions to include alcohol derivatives.¹¹ In this chapter we report the intramolecular reductive cross-electrophile coupling reactions of benzylic pivalates with aryl halides and provide evidence for a stereospecific cyclization reaction.





⁷ Xue, W.; Xu, H.; Liang, Z.; Qian, Q.; Gong, H. Org. Lett. 2014, 16, 4984.

⁸ (a) Weix, D. Acc. Chem. Res. **2015**, 48, 1767. (b) Moragas, T.; Correa, A.; Martin, R. Chem. Eur. J. **2014**, 20, 8242. (c) Gu, J.; Wang, X.; Xue, W.; Gong, H. Org. Chem. Front. **2015**, 2, 1141.

⁹For intermolecular reductive coupling reactions of benzylic alcohol derivatives, see: (a) Correa, A.; Leon, T.; Martin, R. J. Am. Chem. Soc. **2014**, 136, 1062. (b) Ackerman, L. K. G.; Anka-Lufford, L. L.; Naodovic, M.; Weix, D. J. Chem. Sci. **2015**, 6, 1115.

¹⁰Tollefson, E. J.; Erickson, L. W.; Jarvo, E. R. J. Am. Chem. Soc. **2015**, 137, 9760.

¹¹ For intermolecular reductive coupling of allylic esters with halides, see: Anka-Lufford, L. L.; Prinsell, M. R.; Weix, D. J. *J. Org. Chem.* **2012**, *77*, 9989. (b) Cui, X.; Wang, S.; Zhang, Y.; Deng, W.; Qian, Q.; Gong, H. Org. Biomol. Chem. **2013**, *11*, 3094.



Scheme 3.2. Nickel-Catalyzed Intramolecular Reductive Cross-Electrophile Coupling Reactions

3.2 Development of Nickel-Catalyzed Reductive Cross-Electrophile Coupling

We designed secondary benzylic pivalate **3.9** as a model substrate, based on our previous work in cross-coupling reactions of benzylic electrophiles.¹² These substrates are easily prepared by lithiation of an arene and addition into the corresponding bromophenyl aldehyde. Additionally, benzylic pivalate esters are less reactive than their halide counterparts. Reaction of **3.9** in the presence of catalytic quantities of NiBr₂-glyme, bathophenanthroline (BPhen), and Zn⁰ provided desired product **3.10** in excellent yield with negligible yields of hydrodehalogenation (Table 3.1, entry 1). Less sterically encumbered esters, such as acetate, provided lower conversion under the reaction conditions (entry 2). Alternative reducing agents such as Mn^0 also led to lower yields (entry 3). Utilizing phosphine and other aromatic nitrogen-containing ligands, such as bipy or pybox, resulted in a dramatic decrease in product formation (entries 4–7). In the absence of a ligand

¹² Tollefson, E. J.; Hanna, L. E.; Jarvo, E. R. *Acc. Chem. Res.* **2015**, *48*, 2344. For a related transformation, see: Zhou, Q.; Srinivas, H. D.; Dasgupta, S.; Watson, M. P. J. Am. Chem. Soc. **2013**, *135*, 3307.

or a nickel catalyst, the desired cyclization does not occur (entries 8 and 9). Additives known to promote reactivity in other reductive cross-electrophile coupling reactions were also examined.¹³ The addition of either pyridine or NaI favored hydrodehalogenation (entries 10 and 11).

nap	OPiv Br 3.9 NiBr ₂ -glyme (10 mol%) BPhen (15 mol%) Zn ⁰ (3 equiv.) DMA, 24 °C, 16 h Standard Conditions	nap 3.10	• nap 3.11	H
Entry	Variation From Standard Conditions	Recovered 3.9 (%) ^[a]	3.10 (%) ^[a]	3.11 (%) ^[a]
1	none	<2	90	<2
2	Ac instead of Piv	59	9	13
3	Mn instead of Zn	80	<2	<2
4	dppf instead of BPhen	26	25	49
5	bipy instead of BPhen	20	42	25
6	terpyridine instead of BPhen	86	<2	3
7	pybox instead of BPhen	98	<2	<2
8	no ligand	100	<2	<2
9	no NiBr ₂ -glyme	93	<2	<2
10	pyridine (40 mol%)	25	4	61
11	Nal (25 mol%)	<2	25	68
12	NiCl ₂ -glyme instead of NiBr ₂ -glyme	12	79	4
13	DMF instead of DMA	<2	44	35

Table 3.1. Optimization of reductive coupling reaction conditions

^[a]Determined by ¹H NMR spectroscopy using PhTMS as internal standard.

3.3 Scope of Intramolecular Reductive Coupling

Having established conditions for the cyclization of model substrate **3.9**, we set out to investigate the scope of the transformation (Table 3.2). Cyclization of a series of naphthylic esters provides tetralins **3.12** and **3.14**, which correspond to the core of bicunningines A and B,¹⁴ and tetracyclic indanes **3.10**, **3.13**, and **3.15**. Benzofuran and benzothiophene moieties were well-tolerated, providing good yields for both indanes and tetralins (**3.16–18**, **3.20–21**). Additionally, substrates containing N-heterocycles were found to undergo the desired cyclization. Pyridine-

¹³ (a) Everson, D. A.; Shrestha, R.; Weix, D. J. J. Am. Chem. Soc. **2010**, 132, 920. (b) Cherney, A. H.; Reisman, S. E. J. Am. Chem. Soc. **2014**, 136, 14365.

¹⁴ Hou, X.-F.; Yao, S.; Mandi, A.; Kurtan, T.; Tang, C.-P.; Ke, C.-Q.; Li, X.-Q.; Ye, Y. Org. Lett. 2012, 14, 460.

substituted indane **3.19** can be synthesized in 65% yield and *N*-tosylindoles **3.22–24** were obtained in good yields. Substrates containing methoxy-, fluoro-, silyl- and ester substituents were also well-tolerated under the reaction conditions. Notably, these cyclization reactions proceed smoothly without the aid of a Thorpe-Ingold effect.¹⁵





^[a]Reaction run with 15 mol% NiBr₂-glyme and 45 °C.^[b]Reaction run at 45 °C. ^[c]Both sm and product are 1:1 dr.

3.4 Scope of Enantiospecific Intramolecular Reductive Coupling

Finally, we sought to determine whether the intramolecular cyclization could proceed in an enantiospecific fashion. While several examples of stereoconvergent reductive coupling reactions have been reported,¹⁶ to the best of our knowledge, there is only one example of an enantiospecific reductive coupling reaction.⁸ Subjecting (*R*)-3.9 to the optimized conditions afforded (*S*)-3.10 in 90% yield in 88% enantiomeric excess with 92% enantiospecificity (Table

¹⁵ Beesley, R. M.; Ingold, C. K.; Thorpe, J. F. J. Chem. Soc. Trans. 1915, 107, 1080.

¹⁶ Cui, X.; Wang, S.; Zhang, Y.; Deng, W.; Qian, Q.; Gong, H. Org. Biomol. Chem. **2013**, *11*, 3094. Cherney, A. H.; Kadunce, N. T.; Reisman, S. E. J. Am. Chem. Soc. **2013**, *135*, 7442. Cherney, A. H.; Reisman, S. E. J. Am. Chem. Soc. **2014**, *136*, 14365. Kadunce, N. T.; Reisman S. E. J. Am. Chem. Soc. **2015**, *137*, 10480. Ackerman, L. K. G.; Anka-Lufford, L. L.; Naodovic, M.; Weix, D. J. Chem. Sci. **2015**, *6*, 1115.

3.3). Based on comparison of (*S*)-3.10 to literature values, ^{17,18} the reductive cross-electrophile coupling reaction proceeds with inversion at the benzylic center. Indane 3.13 was also formed with high enantiospecificy, as was tetralin 3.12. Notably, all three of these substrates contain the naphthylic ether moiety which we hypothesize is prone to rapid and stereospecific oxidative addition reactions. ^[12] Substrates wherein the ester is activated by a heterocycle such as benzofuran, benzothiophene, or indole provided lower enantiospecificity.¹⁹ This change in stereoselectivity likely correlates to a change in the reaction mechanism. Investigation of the mechanistic details is ongoing.



(S)-3.13

69% yield

96% es^a

Table 3.3. Nickel-catalyzed stereospecific reductive cyclization

a) NiBr₂-glyme (10 mol%), rt. b) Starting material 23% ee c) NiBr₂-glyme (15 mol%), 45 $^\circ\text{C}$ **3.5 Conclusion**

In summary, the intramolecular reductive cyclization for the synthesis of indanes and tetralins has been developed. The reactions are tolerant of a variety of heterocycles and functional groups. We have also demonstrated stereospecific cross-electrophile coupling reactions of benzylic esters for synthesis of enantioenriched 1-arylindanes and tetralins.

(S)-3.12

62% yield

83% es^a

(S)-**3.18**

60% yield

50% es^c

(S)-3.20

75% yield

0% es^a

(S)-3.16

69% yield

28% es^a

(S)-3.10

90% yield

92% es^a

(S)-3.23

75% yield

0% es^{a,b}

¹⁷ (a) Yu, N-U.; Xu, M-H. J. Org. Chem. **2013**, 78, 2736. (b) Yue, G.; Lei, K.; Hirao, H.; Zhou, J. Angew. Chem. Int. Ed. **2015**, 54, 6531.

¹⁸ For complete details, see the experimental details.

¹⁹ When enantioenriched starting materials were employed, products **3.16**, **3.18**, **3.20**, and **3.23** were formed with <50% es. For details, see the experimental details.

3.6 Experimental Details

General Procedures

All reactions were carried out under an atmosphere of N₂, or Ar when noted. All glassware was oven- or flame-dried prior to use. Tetrahydrofuran (THF), diethyl ether (Et₂O), dichloromethane (CH_2Cl_2) , and dimethylacetamide (DMA) were degassed with Ar and then passed through two 4 x 36 inch columns of anhydrous neutral A-2 alumina (8 x 14 mesh; LaRoche Chemicals; activated under a flow of argon at 350 °C for 12 h) to remove H₂O. All other solvents utilized were purchased "anhydrous" commercially, or purified as described. ¹H NMR spectra were recorded on Bruker DRX-400 (400 MHz ¹H, 100 MHz ¹³C, 376.5 MHz ¹⁹F) or CRYO-500 (500 MHz ¹H, 125.7 MHz ¹³C) spectrometers. Proton chemical shifts are reported in ppm (δ) relative to internal tetramethylsilane (TMS, $\delta 0.00$). Data are reported as follows: chemical shift (multiplicity [singlet (s), broad singlet (br s), doublet (d), doublet of doublets (dd), triplet (t), doublet of triplets (dt), doublet of doublet of triplets (ddt), triplet of triplets (tt), quartet (q), quintet (quin), apparent doublet (ad), apparent triplet (at), multiplet (m)], coupling constants [Hz], integration). Carbon chemical shifts are reported in ppm (δ) relative to TMS with the respective solvent resonance as the internal standard (CDCl₃, δ 77.16 ppm). Unless otherwise indicated, NMR data were collected at 25 °C. Infrared (IR) spectra were obtained on a Thermo Scientific Nicolet iS5 spectrometer with an iD5 ATR tip (neat) and are reported in terms of frequency of absorption (cm⁻¹). Analytical thin-layer chromatography (TLC) was performed using Silica Gel 60 F₂₅₄ precoated plates (0.25 mm thickness). Visualization was accomplished by irradiation with a UV lamp and/or staining with KMnO₄, ceric ammonium molybdate (CAM), or *p*-anisaldehyde (PAA) solutions. Flash chromatography was performed using Silica Gel 60 (170-400 mesh) from Fisher Scientific. Melting points (m.p.) were obtained using a Mel-Temp melting point apparatus and are uncorrected. Optical rotations were measured on a Rudolph Research Analytical Autopol IV Automatic Polarimeter. SFC determinations of enantiopurity were performed on a Berger Analytical instrument using a DaicelTM Chiralpak® column (OD-H, OJ-H, or AD-H; 100 bar, 50 °C, 215 nm). High resolution mass spectrometry was performed by the University of California, Irvine Mass Spectrometry Center. Nickel(II) bromide ethylene glycol dimethyl ether complex was purchased from Aldrich, stored in a glovebox under an atmosphere of N₂, and used as received. Zinc powder (100 Mesh) was purchased from Alfa Aesar and used as received. All other reagents

were purchased commercially and used as received.

General methods for starting material synthesis

The general methods for starting material synthesis will be used throughout the rest of the SI. In each instance a general method is used, it is specified by letter (A, B, etc.) and the exact amounts of reagents used for each reaction are listed for the specific compounds synthesized.

METHOD A: ALDOL CONDENSATION, SODIUM BOROHYDRIDE REDUCTION, AND DIIMIDE REDUCTION



The products were prepared according to a modified procedure reported by Franzen.²⁰ To a 250 mL round bottom flask equipped with a stir bar was added ketone (1.0 equiv), aldehyde (1.0 equiv), KOH (2.0 equiv), and MeOH (30 mL). The reaction was stirred overnight at room temperature. The resulting solid was filtered, washed with water, dried by vacuum filtration, and taken on to

²⁰ Wang, Y.; Franzen, R. Synlett., **2012**, 23, 925.

the next step without further purification. The unpurified chalcone was taken up in MeOH (50 mL) and NaBH₄ (1.2 equiv) was added in two portions over 10 minutes. Upon complete reaction of starting material, as judged by TLC, the mixture was quenched with careful addition of saturated NH₄Cl (15 mL). EtOAc (50 mL) was added, the layers were separated, and the aqueous layer was extracted with EtOAc (3 x 30 mL). The organic extract was washed with brine, dried over MgSO₄, and concentrated under reduced pressure and the unpurified allylic alcohol was taken on to the next step without further purification. The unpurified residue was dissolved in THF (100 mL) and tosylhydrazide (4.0 equiv) and NaOAc•3H₂O (4.0 equiv) were added. The reaction vessel was equipped with a reflux condenser and the mixture was heated at reflux for 24 hours. The mixture was cooled to room temperature and quenched with H₂O (30 mL), and the aqueous layer was extracted with EtOAc (3 x 40 mL). The organic extract was washed with brine, dried over MgSO₄, and concentrated under reduced pressure. The product was then purified by flash column chromatography.

METHOD B: PIVALATION OF BENZYLIC ALCOHOLS

$$\begin{array}{c} \mathsf{PivCl} \\ \mathsf{OH} \quad \begin{array}{c} \mathsf{DMAP} \\ \mathsf{DMAP} \\ \mathsf{Ar} \\ \mathsf{R} \\ \mathsf{CH}_2\mathsf{Cl}_2 \\ \mathsf{Ar} \\ \mathsf{R} \\ \mathsf{R} \\ \mathsf{R} \end{array}$$

The product was prepared according to a modified procedure reported by Martin.²¹ To a solution of benzylic alcohol (1.0 equiv) in CH₂Cl₂ (20 mL) was added pivaloyl chloride (1.1 equiv) and dimethylaminopyridine (1.1 equiv). The reaction was allowed to stir overnight at room temperature. The reaction was quenched with water (10 mL) and extracted with CH₂Cl₂ (3 x 10 mL). The organic extract was washed with brine, dried over MgSO₄, and concentrated under reduced pressure. The product was purified by flash column chromatography.

²¹ Zarate, C.; Martin, R. J. Am. Chem. Soc., 2014, 136, 2236

METHOD C: HECK REACTION FOR THE PREPARATION OF 2-BROMOPHENYLALDEHYDES



The product was prepared according to a modified procedure reported by Tietze.²² A flame-dried 100 mL round bottom flask equipped with a stir bar and septum was charged with TBACl (1.0 equiv), NaHCO₃ (2.5 equiv), Pd(OAc)₂ (0.05 equiv), and dry DMF (30 mL). The mixture was stirred for 5 minutes followed by simultaneous addition of 1,2-iodobromobenzene (1.0 equiv), and alkenol (1.5 equiv). The reaction mixture was heated to 40 °C and stirred for 30 hours. The reaction was slowly quenched with saturated NH₄Cl (25 mL) and extracted with EtOAc (3 x 30 mL). The organic extract was washed with brine, dried over MgSO₄, and concentrated under reduced pressure. The product was purified by flash column chromatography.

METHOD D: ARENE LITHIATION/LITHIUM-HALOGEN EXCHANGE, THEN ADDITION INTO ALDEHYDE

$$Ar^{-Li} + H^{+} H^{+} H^{-} H^{-78 \circ C - rt} Ar^{-16 \circ C - rt} H^{+} H^{+} H^{+} H^{-} H^{-}$$

The product was prepared according to a modified procedure reported by O'Doherty.²³ A flamedried 25 mL round bottom flask equipped with a stir bar and septum was charged with Et₂O (6 mL) and arene (1.3 equiv) and cooled to -78 °C. *n*-BuLi (2.5 M in hexanes, 1.4 equiv) was added dropwise over 10 minutes. The mixture was stirred cold for 30 minutes and then allowed to warm to room temperature and stir for an additional 20 minutes. A separate flask of requisite aldehyde

²² Tietze, L. F.; Hungerland, T.; Dufert, A.; Objartel, I.; Stalke, D. Chem. Eur. J., 2012, 18, 3286

²³ Guo, H.; O'Doherty, G.A. Org. Lett. 2006, 8, 1609
(1.0 equiv) dissolved in Et₂O (10 mL) was cooled to -78 °C. The solution of aryl lithium was added dropwise over 30 minutes to the aldehyde. The reaction was warmed to room temperature overnight. The mixture was quenched with saturated NH₄Cl (15 mL). Layers separated and the aqueous layer was extracted with EtOAc (3 x 25 mL). The organic extract was washed with brine, dried over MgSO₄, and concentrated under reduced pressure. The product was purified by flash column chromatography.

SYNTHESIS AND CHARACTERIZATION OF STARTING MATERIALS FOR TABLE 3.2



Scheme 3.3: Synthesis of starting material for formation of *rac*-3.10 (Table 3.2)

rac-**3.25**The product was prepared according to general method A using 2-acetonaphthone (3.4 g, 20 mmol), 2-bromobenzaldehyde (2.3 mL, 20 mmol), and KOH (2.2 g, 40 mmol). Then NaBH₄ (1.0 g, 25 mmol), tosylhydrazide (15 g, 80 mmol) and NaOAc•3H₂O (11 g, 80 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (6.2 g, 18 mmol, 91% over 3 steps). TLC $\mathbf{R}_{\mathbf{f}} = 0.6$ (10% EtOAc/hexanes); ¹H **NMR** (400 MHz, CDCl₃) δ 7.86–7.78 (m, 4H), 7.54–7.43 (m, 4H), 7.24–7.18 (m, 2H), 7.04 (ddd, J = 8.1, 6.2, 3.3 Hz, 1H), 4.89 (t, J = 6.4 Hz, 1H), 2.97–2.76 (m, 2H), 2.22–2.09 (m, 2H), 2.05 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 141.9, 141.3, 133.5, 133.3, 133.1, 130.7, 128.6, 128.2, 127.92, 127.90, 127.7, 126.4, 126.1, 124.9, 124.7, 124.3, 74.3, 39.0, 32.8; **IR** (neat) 3365, 3054,

2929, 1601, 1470, 1021 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₁₉H₁₇BrONa (M + Na)⁺ 363.0360, found 363.0368.

rac-3.9. The product was prepared according to general method B using *rac-3.25* (1.7 g, 5.0 mmol), pivaloyl chloride (0.68 mL, 5.5 mmol, 1.1 equiv) and dimethylaminopyridine (0.67 g, 5.5 mmol, 1.1 equiv). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title compound as a colorless oil (1.7 g, 4.1 mmol, 82%). **TLC R**_f = 0.7 (5% EtOAc/hexanes); ¹H **NMR** (400 MHz, CDCl₃) δ 7.86–7.74 (m, 4H), δ 7.52–7.39 (m, 4H), δ 7.22–7.11 (m, 2H), 7.00 (t, *J* = 7.0 Hz, 1H), 5.95 (t, *J* = 6.5 Hz, 1H), 2.94–2.68 (m, 2H), 2.36–2.11 (m, 2H), 1.26 (s, 9H); ¹³C **NMR** (100 MHz, CDCl₃) δ 177.9, 140.9, 138.2, 133.5, 133.3, 133.2, 130.6, 128.7, 128.3, 128.1, 128.0, 127.8, 126.5, 126.3, 125.8, 124.6, 124.3, 75.5, 39.2, 36.9, 32.8, 27.5; **IR** (neat) 3055, 2969, 1725, 1471, 1279, 1147, 1025 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₄H₂₅BrO₂Na (M + Na)⁺ 447.0936, found 447.0926.

Synthesis of building blocks 3.26 and 3.27.

Aldehydes **3.26** and **3.27** were employed as building blocks in the synthesis of the starting materials for compounds **3.12 and 3.13**, **3.18-21** (Table 3.2).



3.26 The product was prepared according to general method C using TBACl (5.6 g, 20.0 mmol), NaHCO₃ (4.2 g, 50 mmol), Pd(OAc)₂ (0.11 g, 0.5 mmol), iodobromobenzene (2.6 mL, 20 mmol, 1.0 equiv), and allyl alcohol (2.1 mL, 30 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (2.4 g, 10 mmol, 52%). Analytical data is consistent with literature values.³ ¹H NMR (400 MHz, CDCl₃) δ

9.84 (s, 1H), 7.54 (d, *J* = 7.9 Hz, 1H), 7.28–7.20 (m, 2H), 7.08 (q, *J* = 4.1 Hz, 1H), 3.07 (t, *J* = 7.4 Hz, 2H), 2.81 (t, *J* = 7.4 Hz, 2H).



3.27 The product was prepared according to general method C using TBABr (6.5 g, 20.0 mmol), NaHCO₃ (4.2 g, 50 mmol), Pd(OAc)₂ (0.11 g, 0.5 mmol), iodobromobenzene (2.6 mL, 20 mmol, 1.0 equiv), and 3-buten-1-ol (2.6 mL, 30 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (2.4 g, 10 mmol, 52%). Analytical data is consistent with literature values.²⁴ ¹**H** NMR (400 MHz, CDCl₃) δ 9.79 (s, 1H), 7.53 (d, *J* = 7.9 Hz, 1H), 7.25–7.18 (m, 2H), 7.07 (dd, *J* = 7.9, 2.2 Hz, 1H), 2.79 (dt *J* = 7.5 Hz, 2H), 2.50 (td, *J* = 7.4, 1.5 Hz, 2H), 1.97 (d, *J* = 7.5 Hz, 2H).

Scheme 3.4: Synthesis of starting material for formation of 3.12 (Table 3.2)



3.28 The product was prepared according to general method D using 2-bromonaphthalene (1.25 g, 6.00 mmol), *n*-BuLi (2.6 mL, 2.5 M in hexane, 6.6 mmol), and **3.27** (1.36 g, 6.00 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (1.89 g, 5.31 mmol, 85%).

²⁴ Gibson, S. E.; Jones, J. O.; McCague R.; Tozer, M. J.; Whitcombe, N. J. Synlett, 1999, 954

3.29 The product was prepared according to general method B using *rac*-**3.28** (0.18 g, 0.50 mmol), pivaloyl chloride (0.080 mL, 0.60 mmol) and dimethylaminopyridine (0.80 g, 0.60 mmol). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title compound as a colorless oil (0.18 g, 0.40 mmol, 80%). **TLC R**f = 0.4 (12% Et₂O/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.85–7.78 (m, 3H), 7.76 (s, 1H), 7.52–7.41 (m, 4H), 7.19 (td, *J* = 7.2, 1.2 Hz, 1H), 7.14 (dd, *J* = 7.6, 1.27 Hz, 1H), 7.03 (td, *J* = 7.6, 1.9 Hz, 1H), 5.91 (dd, *J* = 8.0, 5.6 Hz, 1H), 2.76 (t, *J* = 7.9 Hz, 2H), 2.07 (dddd, *J* = 13.4, 10.2, 8.0, 5.1 Hz, 1H), 1.93 (ddt, *J* = 13.8, 11.2, 5.6 Hz, 1H), 1.78–1.58 (m, 2H), 1.22 (s, 9H). ¹³**C NMR** (125 MHz, CDCl₃) δ 177.8, 141.4, 138.5, 133.3, 133.2, 133.0, 130.5, 128.5, 128.2, 127.8, 127.8, 127.6, 126.3, 126.1, 125.6, 124.6, 124.3, 75.6, 39.0, 36.3, 36.0, 27.4, 26.0; **IR** (neat) 2971, 2866, 1726, 1500, 1279, 1148 cm⁻¹; **HRMS** (TOF MS CI+) *m*/*z* calcd for C₂₅H₂₇BrO₂ (M)⁺ 438.1194, found 438.1187.

Scheme 3.5: Synthesis of starting material for formation of 3.13 (Table 3.2)



rac-3.30 The product was prepared according to general method D using 2-bromo-6-(*tert*-butyldimethylsiloxy) naphthalene (2.02 g, 6.00 mmol), *n*-BuLi (2.6 mL, 2.5 M in hexane, 6.6 mmol), and **3.26** (1.4 g, 6.0 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (1.9 g, 4.0 mmol, 67%).

rac-**3.31** The product was prepared according to general method B using *rac*-**3.30** (0.47 g, 1.0 mmol), pivaloyl chloride (0.13 mL, 1.1 mmol) and dimethylaminopyridine (0.13 g, 1.1 mmol). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title

compound as a colorless oil (0.45 g, 0.81 mmol, 81%). **TLC** $\mathbf{R}_{f} = 0.5$ (5% EtOAc/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.73–7.66 (m, 3H), 7.50 (dd, J = 8.0, 1.1 Hz, 1H), 7.42 (dd, J = 8.5, 1.7 Hz, 1H), 7.22–7.13 (m, 3H), 7.07 (dd, J = 8.8, 2.4 Hz, 1H), 7.02 (ddd, J = 8.0, 6.9, 2.3 Hz, 1H), 5.91 (dd, J = 8.3, 5.3 Hz, 1H), 2.86 (ddd, J = 14.0, 11.2, 5.0 Hz, 2H), 2.73 (ddd, J = 13.5, 10.7, 5.7 Hz, 2H), 2.35–2.10 (m, 2H), 1.25 (s, 9H), 1.01 (s, 9H), 0.24 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 177.8, 153.9, 140.9, 136.1, 134.5, 133.0, 130.5, 129.6, 129.1, 127.9, 127.7, 127.3, 125.5, 124.54, 124.50, 122.5, 114.9, 75.5, 39.1, 36.7, 32.7, 27.4, 25.9, 18.4, -4.2; IR (neat) 2956, 2930, 1728, 1605, 1505, 1261, 1152 cm⁻¹; HRMS (TOF MS CI+) m / z calcd for C₃₀H₃₉BrO₃Si (M + Na)⁺ 554.1852, found 554.1850.

Scheme 3.6: Synthesis of starting material for formation of 3.14 (Table 3.2)



3.32. The product was prepared according to general method D using 2-bromo-6-(*tert*-butyldimethylsiloxy) naphthalene (1.01 g, 3.00 mmol), *n*-BuLi (1.3 mL, 2.5 M in hexane, 3.3 mmol), and **3.27** (0.68 g, 3.0 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (1.4 g, 2.9 mmol, 98%).

3.33 The product was prepared according to general method B using **3.32** (0.75 g, 1.5 mmol), pivaloyl chloride (0.22 mL, 1.8 mmol) and dimethylaminopyridine (0.22 g, 1.8 mmol). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title

compound as a colorless oil (0.33 g, 0.77 mmol, 96%). **TLC** $\mathbf{R}_{\mathbf{f}} = 0.5$ (5% EtOAc/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.70–7.64 (m, 3H), 7.49 (dd, J = 8.0, 1.3 Hz, 1H), 7.38 (dd, J = 8.6, 1.8 Hz, 1H), 7.19–7.11 (m, 3H), 7.06 (dd, J = 8.9, 2.5 Hz, 1H), 7.00 (ddd, J = 8.0, 7.0, 2.1 Hz, 1H), 5.88 (dd, J = 7.8, 5.6 Hz, 1H), 2.75 (t, J = 7.9 Hz, 2H), 2.11–1.85 (m, 2H), 1.80–1.56 (m, 2H), 1.21 (s, 9H), 1.01 (s, 9H), 0.23 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 177.9, 153.9, 141.5, 136.5, 134.5, 133.1, 130.6, 129.7, 129.2, 127.8, 127.7, 127.3, 125.5, 124.7, 122.6, 115.1, 75.8, 39.1, 36.4, 36.1, 27.5, 26.1, 26.0, 18.5, -4.04; **IR** (neat) 2954, 2929, 1726, 1605, 1479, 1260, 1150 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₃₁H₄₁BrO₃SiNa (M + Na)⁺ 591.1906, found 591.1910.





3.34 The product was prepared according to a modified procedure reported by Liu.²⁵ A 100 mL round bottom flask charged with LiCl (0.50 g, 10 mmol, 1.0 equiv) and a Teflon stir bar was flame-dried under vacuum. Once cooled to room temperature, the flask was removed from vacuum and charged with MeCN (50 mL), 2-bromobenzaldehyde (1.2 mL, 10 mmol, 1.0 equiv), trimethyl phosphonoacetate (2.00 mL, 12.5 mmol, 1.25 equiv), and triethylamine (1.4 mL, 10 mmol, 1.0 equiv), then sealed and stirred at 24 °C for 12 hr. The reaction mixture was passed through a cake

²⁵ Pan, X.; Liu, Z. Tetrahedron, 2014, 70, 4602

of silica and the solvent was removed under reduced pressure affording a clear, colorless oil that was taken directly to the next step in the synthetic sequence.

A 100 mL round bottom flask containing methyl (*E*)-3-(2-bromophenyl)-acrylate was charged with 2-acetonaphthone (1.7 g, 10 mmol, 1.0 equiv), dissolved in THF (40 mL) and cooled to $-5 \,^{\circ}$ C in a brine-ice bath. Following this, potassium *tert*-butoxide (1.2 g, 10 mmol, 1.0 equiv) was added and the reaction was stirred at $-5 \,^{\circ}$ C to 0 $^{\circ}$ C for 3 hrs. The reaction was then quenched with saturated NH₄Cl (20 mL) and extracted with Et₂O (45 mL). The ether layer was then washed with brine and dried over Na₂SO₄, filtered and concentrated under vacuum. The unpurified product was purified by flash column chromatography (gradient: hexanes to 15% Et₂O/hexanes). The fractions containing product were concentrated in a 100 mL round bottom flask, dissolved in MeOH (45 mL), cooled to 0 $^{\circ}$ C and NaBH₄ (0.45 g, 12 mmol 1.2 equiv) was added. The reaction was allowed to warm to room temperature over the course of 30 min and stirred for an additional 2 hr. The unpurified mixture was then concentrated under reduced pressure and the unpurified oil was passed through a cake of silica with ethyl ether to afford the title compound as a pale yellow oil (1.4 g, 3.3 mmol, 33% over 3 steps).

3.35 The product was prepared according to general method B using **3.34** (1.3 g, 3.3 mmol), pivaloyl chloride (0.45 mL, 3.7 mmol) and dimethylaminopyridine (0.46 g, 3.8 mmol). The product was purified by flash column chromatography (10% Et₂O/hexanes) to afford the title compound as a colorless oil (1.6 g, 3.2 mmol, 99%, 1:1 dr). **TLC R**_f = 0.35 (12% EtOAc/hexanes); ¹**H NMR** (500 MHz, CDCl₃) δ 7.84–7.76 (m, 3H), 7.63 (s, 1H), 7.55 (d, *J* = 8.3 Hz, 1H), 7.49– 7.43 (m, 2H), 7.40 (dd, *J* = 8.4, 1.8 Hz, 1H), 7.35–7.28 (m, 2H), 7.11–7.06 (m, 1H), 5.79 (t, *J* = 7.3 Hz, 1H), 3.68 (quint, *J* = 7.2 Hz, 1H), 3.56 (s, 3H), 2.68 (dd, *J* = 15.4, 7.2 Hz, 1H), 2.61 (dd, *J* = 15.5, 7.2 Hz, 1H), 2.49 (dt, J = 15.4, 8.2 Hz, 1H), 2.26–2.17 (m, 1H), 1.20 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 177.8, 172.0, 141.9, 137.5, 133.4, 133.27, 133.24, 128.6, 128.4, 128.2, 128.0, 127.8, 126.3, 126.2, 126.1, 124.8, 124.1, 74.0, 51.8, 40.8, 39.7, 38.9, 37.2, 27.2; **IR** (neat) 3056, 2972, 1728 (b), 1600, 1471, 1278, 1148, 1021 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₇H₂₉BrO₄Na (M + Na)⁺ 519.1147, found 519.1132.

Scheme 3.8: Synthesis of starting material for formation of 3.16 (Table 3.2)



3.36 The product was prepared according to general method A using 2-benzofuran methyl ketone (2.9 g, 18 mmol), 2-bromobenzaldehyde (2.1 mL, 18 mmol), and KOH (2.0 g, 36 mmol). Then NaBH₄ (0.76 g, 20 mmol), tosylhydrazide (7.5 g, 40 mmol) and NaOAc•3H₂O (5.4 g, 40 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (2.1 g, 6.3 mmol, 35% over 3 steps).

3.37 The product was prepared according to general method B using **3.36** (0.66 g, 2.0 mmol), pivaloyl chloride (0.27 mL, 2.2 mmol) and dimethylaminopyridine (0.27 g, 2.2 mmol). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title compound as a colorless oil (0.59 g, 1.4 mmol, 70%). **TLC R**_f = 0.5 (5% EtOAc/hexanes); ¹H **NMR** (400 MHz, CDCl₃) δ 7.51 (d, *J* = 8.3 Hz, 1H), 7.47 (d, *J* = 8.3 Hz, 1H), 7.44 (d, *J* = 8.2 Hz, 1H), 7.27–7.12 (m, 4H), 7.02–6.94 (m, 1H), 6.67 (s, 1H), 6.04 (t, *J* = 6.8 Hz, 1H), 2.90–2.72 (m, 2H), 2.41–2.27 (m, 2H), 1.25 (s, 9H); ¹³C **NMR** (100 MHz, CDCl₃) δ 177.8, 155.6, 155.2, 140.6,

133.2, 130.7, 129.8, 128.3, 127.9, 124.74, 124.68, 123.2, 121.5, 111.6, 104.9, 68.8, 39.3, 33.3, 32.4, 27.5; **IR** (neat) 2970, 1729, 1453, 1278, 1142 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₂H₂₃BrO₃Na (M + Na)⁺ 437.0728, found 437.0725.

Scheme 3.9: Synthesis of starting material for formation of 3.17 (Table 3.2)



3.38 The product was prepared according to general method A using 2-acyl-5-methoxybenzofuran (2.0 g, 10 mmol), 2-bromobenzaldehyde (1.2 mL, 10 mmol), and KOH (1.2 g, 21 mmol). Then NaBH₄ (0.95 g, 25 mmol), tosylhydrazide (5.4 g, 29 mmol) and NaOAc•3H₂O (3.9 g, 29 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (2.0 g, 5.5 mmol, 55% over 3 steps).

3.39 The product was prepared according to general method B using **3.38** (0.50 g, 1.6 mmol), pivaloyl chloride (0.19 mL, 1.6 mmol) and dimethylaminopyridine (0.19 g, 1.6 mmol). The product was purified by flash column chromatography (12% EtOAc/hexanes) to afford the title compound as a colorless oil (0.52 g, 1.2 mmol, 75%). **TLC R**_f = 0.4 (12% EtOAc/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.49 (d, *J* = 8.1 Hz, 1H), 7.33 (d, *J* = 8.9 Hz, 1H), 7.20–7.16 (m, 2H), 7.01 (ddd, *J* = 7.9, 6.0, 3.2 Hz, 1H), 6.99 (d, *J* = 2.6 Hz, 1H), 6.87 (dd, *J* = 8.9, 2.6 Hz, 1H), 6.62 (t, *J* = 0.6 Hz, 1H), 6.00 (t, *J* = 6.6 Hz, 1H), 3.78 (s, 3H), 2.90–2.72 (m, 2H), 2.41–2.26 (m, 2H), 1.25 (s, 9H); ¹³**C NMR** (100 MHz, CDCl₃) δ 177.8, 156.34, 156.32, 150.1, 140.6, 133.2, 130.7,

128.8, 128.2, 127.9, 124.6, 113.5, 112.0, 105.0, 103.9, 68.8, 56.1, 39.2, 33.2, 32.3, 27.4; **IR** (neat) 2968, 1729, 1617, 1476, 1204, 1192 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₂H₂₅BrO₄Na (M + Na)⁺ 467.0834, found 467.0832.

Scheme 3.10: Synthesis of starting material for formation of 3.18 (Table 3.2)



3.40 The product was prepared according to general method D using benzofuran (0.66 mL, 6.0 mmol), *n*-BuLi (2.6 mL, 2.5 M in hexane, 6.6 mmol), and **3.27** (1.0 g, 4.5 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (1.4 g, 3.8 mmol, 84%).

3.41 The product was prepared according to general method B using **3.40** (0.28 g, 0.80 mmol), pivaloyl chloride (0.11 mL, 0.90 mmol) and dimethylaminopyridine (0.11 g, 0.90 mmol). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title compound as a colorless oil (0.33 g, 0.77 mmol, 96%). **TLC R**_f = 0.5 (5% EtOAc/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.51 (d, *J* = 7.3 Hz, 1H), 7.48 (d, *J* = 8.1 Hz, 1H), 7.44 (d, *J* = 8.2 Hz, 1H), 7.24 (t, *J* = 8.0 Hz, 1H), 7.21–7.11 (m, 3H), 7.00 (td, *J* = 7.8, 1.8 Hz, 1H), 6.63 (s, 1H), 6.01 (t, *J* = 6.8 Hz, 1H), 2.77 (t, *J* = 7.8 Hz, 2H), 2.18–2.03 (m, 2H), 1.80–1.60 (m, 2H), 1.21 (s, 9H); ¹³**C NMR** (100 MHz, CDCl₃) δ 177.8, 155.9, 155.1, 141.4, 133.1, 130.6, 128.2, 128.0, 127.7, 124.7, 124.6, 123.1, 121.4, 111.6, 104.6, 69.0, 39.2, 36.0, 32.7, 27.4, 25.7; **IR** (neat) 3057, 2969,

2869, 1728, 1471, 1454, 1278, 1254, 1143 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₃H₂₅BrO₃Na (M + Na)⁺ 451.0885, found 451.0891.



Scheme 3.11: Synthesis of starting material for formation of 3.19 (Table 3.2)

3.42 The product was prepared according to general method D using 3-bromopyridine (0.39 g, 4.0 mmol), *n*-BuLi (1.7 mL, 2.5 M in hexane, 4.4 mmol), and **3.26** (0.85 g, 4.0 mmol) and -78 °C. The product was purified by flash column chromatography (1% NEt₃ in EtOAc) to afford the title compound as a yellow oil (0.72 g, 2.5 mmol, 62%).

3.43 The product was prepared according to general method B using **3.42** (0.58 g, 1.5 mmol), pivaloyl chloride (0.22 mL, 1.8 mmol) and dimethylaminopyridine (0.22 g, 1.8 mmol). The product was purified by flash column chromatography (50% EtOAc/hexanes) to afford the title compound as a colorless oil (0.40 g, 1.1 mmol, 70%). **TLC R**_f = 0.2 (12% EtOAc/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 8.63 (d, *J* = 1.4 Hz, 1H), 8.54 (dd, *J* = 4.9, 1.2 Hz, 1H), 7.65 (dt, *J* = 8.2, 1.8 Hz, 1H), 7.51 (d, *J* = 8.0 Hz, 1H), 7.27 (dd, *J* = 7.3, 4.8 Hz, 1H), 7.22 (t, *J* = 7.4 Hz, 1H), 7.16 (dd, *J* = 7.5, 1.5 Hz, 1H), 7.05 (td, *J* = 7.8, 1.5 Hz, 1H), 5.80 (dd, *J* = 8.2, 5.1 Hz, 1H), 2.89–2.68 (m, 2H), 2.30–2.05 (m, 2H), 1.25 (s, 9H); ¹³**C NMR** (100 MHz, CDCl₃) δ 177.7, 149.5, 148.3, 140.4, 136.4, 134.1, 133.2, 130.5, 128.2, 127.9, 124.5, 123.6, 73.2, 39.1, 36.6, 32.5, 27.4; **IR** (neat) 3055, 2970, 1728, 1471, 1279, 1146 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₁₉H₂₂BrNO₂Na (M + Na)⁺ 398.0732, found 398.0746.

Scheme 3.12: Synthesis of starting material for formation of 3.20 (Table 3.2)



3.44 The product was prepared according to general method D using benzothiophene (0.54 g, 4.0 mmol), *n*-BuLi (1.7 mL, 2.5 M in hexane, 4.4 mmol), and **3.26** (0.85 g, 4.0 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (1.1 g, 3.2 mmol, 79%).

3.45 The product was prepared according to general method B using **3.44** (0.21 g, 0.60 mmol), pivaloyl chloride (0.08 mL, 0.7 mmol) and dimethylaminopyridine (0.080 g, 0.70 mmol). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title compound as a colorless oil (0.21 g, 0.48 mmol, 79%). **TLC R**_f = 0.5 (5% EtOAc/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.84 (d, *J* = 7.7 Hz, 1H), 7.77 (d, *J* = 7.4 Hz, 1H), 7.57 (d, *J* = 8.0 Hz, 1H), 7.40–7.30 (m, 3H), 7.28–7.20 (m, 2H), 7.01 (dd, *J* = 8.1, 2.2 Hz, 1H), 6.21 (dd, *J* = 7.6, 5.7 Hz, 1H), 2.99–2.79 (m, 2H), 2.46–2.27 (m, 2H), 1.32 (s, 9H); ¹³**C NMR** (100 MHz, CDCl₃) δ 177.7, 144.4, 140.5, 139.7, 139.5, 133.2, 130.6, 129.8, 128.2, 127.9, 124.7, 124.6, 124.0, 122.7, 122.3, 71.4, 39.2, 36.6, 32.6, 27.4; **IR** (neat) 3058, 2970, 1728, 1477, 1457, 1277, 1140, 1028 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₂₂H₂₃BrO₂SNa (M + Na)⁺ 453.0500, found 453.0512.

Scheme 3.13: Synthesis of starting material for formation of 2.21 (Table 3.2)



3.46 The product was prepared according to general method D using benzothiophene (0.34 g, 2.5 mmol), *n*-BuLi (1.1 mL, 2.5 M in hexane, 2.2 mmol), and **3.27** (0.45 g, 2.0 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (0.72 g, 2.0 mmol, 99%).

3.47 The product was prepared according to general method B using **3.46** (0.72 g, 2.0 mmol), pivaloyl chloride (0.27 mL, 2.3 mmol) and dimethylaminopyridine (0.27 g, 2.3 mmol). The product was purified by flash column chromatography (12% Et₂O/hexanes) to afford the title compound as a colorless oil (0.46 g, 1.0 mmol, 50%). **TLC R**_f = 0.6 (12% Et₂O/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.81–7.77 (m, 1H), 7.73–7.69 (m, 1H), 7.51 (dd, *J* = 7.9, 1.2 Hz, 1H), 7.35–7.27 (m, 2H), 7.23–7.15 (m, 3H), 7.04 (ddd, *J* = 7.9, 7.0, 2.1 Hz, 1H), 6.12 (dd, *J* = 7.6, 6.0 Hz, 1H), 2.78 (t, *J* = 7.8 Hz, 2H), 2.17–1.96 (m, 2H), 1.82–1.61 (m, 2H), 1.22 (s, 9H); ¹³**C NMR** (125 MHz, CDCl₃) δ 177.6, 144.5, 141.1, 139.4, 139.3, 132.9, 130.4, 127.7, 127.5, 124.5, 124.40, 124.35, 123.7, 122.4, 121.8, 71.4, 38.9, 35.9, 35.8, 27.2, 25.7; **IR** (neat) 3054, 2972, 1725, 1438, 1264, 1146 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₃H₂₅BrO₂SNa (M + Na)⁺ 467.0656, found 467.0652.

Scheme 3.14: Synthesis of starting material for formation of 3.22 (Table 3.2)



3.48 The product was prepared according to general method A using 3-acetylindole (1.6 g, 10 mmol), 2-bromobenzaldehyde (1.9 g, 10 mmol), and KOH (1.2 g, 22 mmol). The indolyl chalcone was then tosylated according to a modified procedure reported by Carreira. A using tosyl chloride (2.2 g, 12 mmol), TBABr (0.34 g, 0.12 mmol), and NaOH (30% w/v) and benzene.²⁶ The chalcone was then reduced by NaBH₄ (0.38 g, 10 mmol), and tosylhydrazide (7.5 g, 40 mmol) and NaOAc•3H₂O (5.5 g, 40 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (3.6 g, 7.4 mmol, 74% over 4 steps).

3.49 The product was prepared according to general method B using **3.48** (0.78 g, 1.6 mmol), pivaloyl chloride (0.25 mL, 2.0 mmol) and dimethylaminopyridine (0.25 g, 2.0 mmol). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title product as a colorless oil (0.86 g, 1.5 mmol, 94%). **TLC R**_f = 0.6 (15% EtOAc/hexanes); ¹H **NMR** (400 MHz, CDCl₃) δ 8.00 (d, J = 8.4 Hz, 1H), 7.71 (d, J = 8.2 Hz, 2H), 7.62 (d, J = 6.5 Hz, 2H), 7.43 (d, J = 7.8 Hz, 1H), 7.29 (t, J = 7.9 Hz, 1H), 7.20 (t, J = 7.7 Hz, 1H), 7.17–7.06 (m, 4H), 6.96 (td, J = 7.7, 1.7 Hz, 1H), 6.14 (t, J = 6.6 Hz, 1H), 2.84–2.64 (m, 2H), 2.39–2.19 (m, 2H), 2.17 (s, 3H), 1.21 (s, 9H); ¹³C **NMR** (100 MHz, CDCl₃) δ 177.8, 145.3, 140.6, 135.7, 135.3, 133.1, 130.6,

²⁶ Waser, J.; Gaspar, B.; Nambu, H.; Carreira, E. M. J. Am. Chem. Soc., 2006, 128, 11693.

130.2, 129.1, 128.2, 127.9, 127.0, 125.3, 124.6, 124.3, 123.7, 122.3, 120.6, 114.2, 68.9, 39.2, 35.1, 32.7, 27.5, 21.7; **IR** (neat) 3053, 2972, 1725, 1596, 1446, 1370, 1174, 1152, 1122 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₂₉H₃₀BrNO₄SNa (M + Na)⁺ 590.0977, found 590.0964.

Scheme 3.15: Synthesis of starting material for formation of 3.23 (Table 3.2)



3.50 The product was prepared according to general method A using 3-acetylindole (1.6 g, 10 mmol), 2-bromo-5-fluorobenzaldehyde (2.0 g, 10 mmol), and KOH (1.4 g, 23 mmol). The indolyl chalcone was then tosylated according to a modified procedure reported by Carreira using tosyl chloride (2.2 g, 12 mmol), TBABr (0.34 g, 0.12 mmol), and NaOH (30% w/v) and benzene. The chalcone was then reduced by NaBH₄ (0.40 g, 10 mmol), and tosylhydrazide (7.5 g, 40 mmol) and NaOAc•3H₂O (5.5 g, 40 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (3.6 g, 7.4 mmol, 64% over 4 steps).

3.51 The product was prepared according to general method B using **3.50** (0.75 g, 1.5 mmol), pivaloyl chloride (0.25 mL, 2.0 mmol) and dimethylaminopyridine (0.24 g, 2.0 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title product as a yellow oil (0.66 g, 1.1 mmol, 73%). **TLC R**_f = 0.6 (15% EtOAc/hexanes); ¹H **NMR** (400 MHz, CDCl₃) δ 7.98 (d, *J* = 8.3 Hz, 1H), 7.74 (d, *J* = 8.3 Hz, 1H), 7.61 (d, *J* = 8.0 Hz, 1H), 7.69 (s, 1H), 7.42 (dd, *J* = 8.7, 6.3 Hz, 1H), 7.32 (t, *J* = 7.7 Hz, 1H), 7.23 (t, *J* = 7.6 Hz, 1H), 7.18 (d, *J* = 8.1 Hz, 2H), 6.85 (dd, *J* = 9.3, 3.0 Hz, 1H), 6.77 (td, *J* = 8.3, 3.0 Hz, 1H), 6.11 (t, *J* = 6.6

Hz, 1H), 2.80–2.61 (m, 2H), 2.36–2.17 (m, 5H), 1.21 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.8, 162.1 (d, *J* = 247 Hz), 145.3, 142.7 (d, *J* = 7 Hz), 135.7, 135.3, 134.2 (d, *J* = 8 Hz), 130.1, 129.0, 127.0, 125.2, 124.2, 123.6, 122.0, 120.5, 118.6 (d, *J* = 3 Hz), 117.3 (d, *J* = 22 Hz), 115.3 (d, *J* = 22 Hz), 114.1, 68.8, 39.2, 34.7, 32.7, 27.4, 21.8; **IR** (neat) 3068, 2972, 1726, 1598, 1370, 1174, 1121 cm⁻¹; **HRMS** (TOF MS ES+) *m* /*z* calcd for C₂₉H₂₉BrFNO₄SNa (M + Na)⁺ 608.0883, found 608.0860.

Scheme 3.16: Synthesis of starting material for formation of 3.24 (Table 3.2)



3.52 The product was prepared according to general method A using 3-acetylindole (1.6 g, 10 mmol), 2-bromo-5-fluorobenzaldehyde (2.3 g, 10 mmol), and KOH (1.4 g, 23 mmol). The indolyl chalcone was then tosylated according to a modified procedure reported by Carreira using tosyl chloride (2.2 g, 12 mmol), TBABr (0.34 g, 0.12 mmol), and NaOH (30% w/v) and benzene. The chalcone was then reduced by NaBH₄ (0.40 g, 10 mmol), and tosylhydrazide (7.5 g, 40 mmol) and NaOAc•3H₂O (5.5 g, 40 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a yellow oil (3.6 g, 7.4 mmol, 11% over 4 steps).

3.53 The product was prepared according to general method B using **3.52** (0.60 g, 1.1 mmol), pivaloyl chloride (0.16 mL, 1.3 mmol) and dimethylaminopyridine (0.16 g, 1.3 mmol). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title product as a yellow oil (0.25 g, 0.41 mmol, 36%). TLC $\mathbf{R}_{\mathbf{f}} = 0.5$ (15% EtOAc/hexanes); ¹H NMR

(400 MHz, CDCl₃) δ 7.97 (d, J = 8.2 Hz, 1H), 7.73 (d, J = 8.4 Hz, 2H), 7.61 (d, J = 7.6 Hz, 1H), 7.58 (s, 1H), 7.30 (td, J = 8.4, 1.3 Hz, 1H), 7.22 (td, J = 8.1, 1.2 Hz, 1H), 7.17 (d, J = 8.0 Hz, 2H), 6.93 (s, 1H), 6.59 (s, 1H), 6.08 (t, J = 6.6 Hz, 1H), 5.89 (s, 2H), 2.73–2.54 (m, 2H), 2.32–2.11 (m, 5H), 1.20 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.8, 147.7, 147.1, 145.3, 135.7, 135.3, 133.5, 130.1, 129.0, 127.0, 125.2, 124.2, 123.6, 122.2, 120.6, 114.4, 114.1, 113.0, 110.1, 101.9, 68.8, 39.2, 35.2, 32.6, 27.4, 21.8; **IR** (neat) 2971, 1724, 1476, 1369, 1174, 1121 cm⁻¹; **HRMS** (TOF MS ES+) m/z calcd for C₃₀H₃₀BrNO₆SNa (M + Na)⁺ 634.0875, found 634.0860.

GENERAL PROCEDURES FOR REDUCTIVE CROSS-ELECTROPHILE COUPLING REACTIONS

METHOD E: INTRAMOLECULAR REDUCTIVE COUPLING REACTIONS



In a glovebox, a flame-dried 7 mL vial equipped with a stir bar was charged with NiBr₂•glyme (10 or 15 mol %), bathophenanthroline (15 mol %), Zn^0 (3 equiv), DMA (0.60 mL), and substrate (1.00 equiv). The reaction was stirred for 20 h before removing the vial from the glovebox. The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography.

CHARACTERIZATION OF PRODUCTS IN TABLE 3.2



rac-**3.10**. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and *rac*-**3.9** (85.1 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% hexanes) to afford the title compound as a colorless oil (44.0 mg, 0.180 mmol, 90%). **TLC R**_f = 0.3 (100% hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.80–7.71 (m, 3H), 7.63 (s, 1H), 7.44–7.36 (m, 2H), 7.30 (d, *J* = 7.5 Hz, 1H), 7.26 (dd, *J* = 8.4, 1.6 Hz, 1H), 7.19 (t, *J* = 7.3 Hz, 1H), 7.11 (t, *J* = 8.1 Hz, 1H), 6.95 (d, *J* = 7.6 Hz, 1H), 4.48 (t, *J* = 8.3 Hz, 1H), 3.12–2.92 (m, 2H), 2.65–2.55 (m, 1H), 2.19–2.07 (m, 1H); ¹³**C NMR** (100 MHz, CDCl₃) δ 147.1, 144.7, 143.1, 133.9, 132.7, 128.6, 127.98, 127.96, 126.99, 126.95, 126.82, 126.78, 126.3, 125.7, 125.4, 124.8, 52.1, 36.8, 32.3; **IR** (neat) 3050, 3018, 2937, 1599, 1506, 1477, 1457 cm⁻¹; **HRMS** (TOF MS CI+) *m*/*z* calcd for C₁₉H₁₆ (M)⁺ 224.1252, found 224.1257.



rac-3.12. The product was prepared according to general method E using NiBr₂•glyme (3.1 mg, 0.010 mmol, 10 mol %), bathophenanthroline (5.0 mg, 0.015 mmol, 15 mol %), Zn^0 (18.8 mg, 0.300 mmol, 3 equiv), DMA (0.40 mL), and **3.29** (44.0 mg, 0.100 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The

product was purified by flash column chromatography (10% EtOAc/hexanes) to afford the title compound as a colorless oil (16 mg, 0.062 mmol, 62%). **TLC** $\mathbf{R_f} = 0.3$ (hexanes); ¹H **NMR** (500 MHz, CDCl₃) δ 7.84–7.73 (m, 3H), 7.53 (s, 1H), 7.49–7.41 (m, 2H), 7.29–7.23 (m, 1H), 7.21–7.11 (m, 2H), 7.02 (t, J = 7.4 Hz, 1H), 6.86 (d, J = 7.8 Hz, 1H), 4.28 (t, J = 7.0 Hz, 1H), 3.02–2.84 (m, 2H), 2.27–2.18 (m, 1H), 2.03–1.90 (m, 2H), 1.86–1.75 (m, 1H); ¹³C **NMR** (125 MHz, CDCl₃) δ 145.1, 139.4, 137.8, 133.6, 132.3, 130.5, 129.2, 128.1, 127.79, 127.77, 127.5, 127.4, 126.2, 126.1, 125.9, 125.5, 46.0, 33.3, 30.0, 21.4; **IR** (neat) 3053, 2925, 2854, 1599, 1448 cm⁻¹; **HRMS** (TOF MS CI+) m/z calcd for C₂₀H₁₈ (M)⁺ 258.1408, found 258.1415.



rac-**3.13**. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.31** (111 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (10% EtOAc/hexanes) to afford the title compound as a colorless oil (51.7 mg, 0.138 mmol, 69%). **TLC R**_f = 0.3 (hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.64 (dd, *J* = 8.6, 6.8 Hz, 2H), 7.57 (s, 1H), 7.31 (d, *J* = 7.5 Hz, 1H), 7.23–7.10 (m, 4H), 7.05 (dd, *J* = 8.8, 2.5 Hz, 1H), 6.97 (d, *J* = 7.5 Hz, 1H), 4.47 (t, *J* = 8.4 Hz, 1H), 3.13–2.93 (m, 2H), 2.66–2.57 (m, 1H), 2.13 (dq, *J* = 12.7, 8.8 Hz, 1H), 1.02 (s, 9H), 0.23 (s, 6H); ¹³**C NMR** (125 MHz, CDCl₃) δ 153.3, 147.1, 144.6, 140.8, 133.6, 129.5, 129.2, 127.2, 127.1 126.8, 126.6, 126.4, 125.2, 124.6, 122.4, 115.0, 51.8, 36.7, 32.1, 25.9, 18.5, –4.1; **IR** (neat) 2954, 2928,

1602, 1478, 1258 cm⁻¹; **HRMS** (TOF MS CI+) m / z calcd for C₂₅H₃₀OSi (M)⁺ 374.2066, found 374.2054.



3.14. The product was prepared according to general method E using NiBr₂•glyme (3.1 mg, 0.010 mmol, 5.0 mol %), bathophenanthroline (5.0 mg, 0.015 mmol, 7.5 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and 3.33 (113.9 mg, 0.200 mmol, 1.00 equiv). After 8 hours, the reaction vial was taken out of the oil bath. In the glovebox, an additional amount of NiBr₂•glyme (3.1 mg, 0.010 mmol, 5 mol %) and bathophenanthroline (5.0 mg, 0.015 mmol, 7.5 mol %) was added to the reaction mixture. The reaction vial was put back into the oil bath for the remainder of the reaction time. The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% pentane) to afford the title compound as a colorless oil (29.8 mg, 0.120 mmol, 58%). TLC $\mathbf{R}_{\mathbf{f}} = 0.2$ (pentane); ¹H NMR (400 MHz, CDCl₃) δ 7.61 (dd, J = 8.6, 6.9 Hz, 2H), 7.45 (s, 1H), 7.18 (dd, J = 8.6, 1.8 Hz, 1H), 7.16–7.10 (m, 3H), 7.04 (dd, J = 8.8, 2.4 Hz, 1H), 7.04–6.99 (m, 1H), 6.87 (d, J = 7.7 Hz, 1H), 4.23 (t, J = 6.7 Hz, 1H), 3.02–2.82 (m, 2H), 2.26–2.15 (m, 1H), 2.00–1.88 (m, 2H), 1.84–1.73 (m, 1H), 1.01 (s, 9H), 0.23 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) & 153.2, 142.8, 139.5, 137.7, 133.3, 130.4, 129.3, 129.04, 129.02, 127.6, 127.2, 126.8, 126.0, 125.7, 122.1, 114.8, 45.7, 33.2, 29.9, 25.8, 21.2, 18.3, -4.3; **IR** (neat) 3053, 2928, 1633, 1602, 1478, 1259 cm⁻¹; **HRMS** (TOF MS CI+) m/z calcd for C₂₆H₃₂OSi (M)⁺ 388.2222, found 388.2227.



3.15. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.35** (99.5 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (10% EtOAc/hexanes) to afford the titlecompound as a colorless oil (29.7 mg, 0.094 mmol, 47%, 1:1 dr). **TLC R**_f = 0.6 (10% EtOAc in hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.82–7.73 (m, 3H), 7.60–7.57 (m, 1H), 7.47–7.40 (m, 2H), 7.30 (d, *J* = 7.5 Hz, 1H), 7.27–7.22 (m, 2H), 7.18 (tdd, *J* = 7.5, 1.5, 0.5 Hz, 1H), 7.01 (d, *J* = 7.5 Hz, 1H), 4.62 (t, *J* = 7.7 Hz, 1H), 3.87–3.78 (m, 1H), 3.72 (s, 3H), 2.76 (dd, *J* = 15.5, 6.1 Hz, 1H), 2.57 (dd, *J* = 15.4, 8.9 Hz, 1H), 2.51–2.36 (m, 2H); ¹³**C NMR** (100 MHz, CDCl₃) δ 173.3, 146.4, 146.3, 142.6, 133.7, 132.6, 128.5, 127.86, 127.85, 127.6, 127.4, 126.7, 126.5, 126.2, 125.7, 124.1, 123.1, 51.9, 49.9, 42.5, 40.6, 40.2; **IR** (neat) 3051, 2949, 1734, 1599, 1435, 1250, 1164 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₂₂H₂₀O₂Na (M + Na)⁺ 339.1361, found 339.1367.



3.16. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.37** (83.1 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (neat Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% hexanes) to afford the title

compound as a colorless oil (35.1 mg, 0.150 mmol, 75%). **TLC R**_f = 0.7 (5% EtOAc/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.54–7.50 (m, 1H), 7.48–7.44 (m, 1H), 7.36–7.29 (m, 2H), 7.27–7.19 (m, 4H), 6.43 (s, 1H), 4.61 (t, *J* = 7.7 Hz, 1H), 3.19–2.98 (m, 2H), 2.66–2.37 (m, 2H); ¹³**C NMR** (100 MHz, CDCl₃) δ 161.1, 155.2, 144.2, 143.5, 129.0, 127.5, 126.7, 125.1, 124.9, 123.6, 122.7, 120.7, 111.2, 102.4, 45.1, 32.4, 31.8; **IR** (neat) 3066, 3022, 2942, 1597, 1583, 1453, 1252 cm⁻¹; **HRMS** (TOF MS CI+) *m*/*z* calcd for C₁₇H₁₄ONH₄ (M + NH₄)⁺ 252.1388, found 252.1386.



3.17. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.39** (89.0 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% hexanes) to afford the title compound as a colorless oil (41.2 mg, 0.156 mmol, 79%). **TLC R**_f = 0.3 (6% Et₂O/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.31–7.26 (m, 2H), 7.25–7.14 (m, 3H), 6.95 (d, *J* = 2.6 Hz, 1H), 6.81 (dd, *J* = 8.9, 2.6 Hz, 1H), 6.34–6.32 (m, 1H), 4.53 (t, *J* = 7.7 Hz, 1H), 3.82 (s, 3H), 3.14–2.93 (m, 2H), 2.61–2.51 (m, 1H) 2.41–2.30 (m, 1H); ¹³**C NMR** (100 MHz, CDCl₃) δ 162.0, 156.0, 150.2, 144.2, 143.5, 129.5, 127.5, 126.7, 125.0, 124.9, 112.0, 111.5, 103.5, 102.6, 56.2, 45.1, 32.3, 31.8; **IR** (neat) 3068, 2929, 1615, 1599, 1474, 1202, 1030 cm⁻¹; **HRMS** (TOF MS CI+) *m* / *z* calcd for C₁₈H₁₆O₂NH₄ (M + NH₄)⁺ 282.1494, found 282.1494.



3.18. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.41** (85.9 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% hexanes) to afford the title compound as a colorless oil (29.8 mg, 0.120 mmol, 60%). **TLC R**_f = 0.8 (5% EtOAc/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.44 (t, *J* = 8.3 Hz, 2H), 7.23 (td, *J* = 7.7, 1.4 Hz, 1H), 7.20–7.09 (m, 5H), 6.19 (s, 1H), 4.33 (t, *J* = 5.7 Hz, 1H), 2.91–2.78 (m, 2H), 2.35–2.08 (m, 2H), 1.95–1.75 (m, 2H); ¹³**C NMR** (100 MHz, CDCl₃) δ 162.6, 155.5, 137.6, 136.0, 130.1, 129.6, 128.9, 126.9, 125.9, 123.5, 122.7, 120.6, 111.2, 104.3, 39.9, 29.6, 28.7, 20.5; **IR** (neat) 3064, 2926, 2858, 1453, 1253 cm⁻¹; **HRMS** (TOF MS CI+) *m*/*z* calcd for C₁₈H₁₆OH (M + H)⁺ 249.1279, found 249.1276.



3.19. The product was prepared according to general method E using NiBr₂•glyme (9.2 mg, 0.030 mmol, 15 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.43** (75.3 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% hexanes) to afford the title compound as a colorless oil (25.4 mg, 0.130 mmol, 65%). **TLC R**_f = 0.1 (12% EtOAc: hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 8.52 (d, *J* = 1.8 Hz, 1H), 8.50 (dd, *J* = 4.9, 1.7 Hz, 1H), 7.45 (dt, *J* = 7.9, 2.0 Hz, 1H), 7.32 (d, *J* = 7.4 Hz, 1H), 7.25–7.20 (m, 2H), 7.16 (t, *J* = 7.4 Hz, 1H), 6.94 (d,

J = 7.6 Hz, 1H), 4.37 (t, J = 8.2 Hz, 1H), 3.13–2.95 (m, 2H), 2.63 (dtd, J = 12.8, 7.9, 7.9, 3.9 Hz, 1H), 2.08 (dq, J = 12.7, 8.7 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 150.0, 148.0, 145.7, 144.3, 140.8, 135.3, 127.0, 126.6, 124.7, 124.6, 123.6, 49.0, 36.5, 31.9; IR (neat) 3020, 2925, 1574, 1478, 1423 cm⁻¹; HRMS (TOF MS CI+) m/z calcd for C₁₄H₁₃NH (M + H)⁺ 196.1126, found 196.1119.



3.20. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.45** (86.3 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% pentane) to afford the title compound as a colorless oil (37.6 mg, 0.150 mmol, 75%). **TLC R**_f = 0.5 (pentane); ¹**H NMR** (400 MHz, CDCl₃) δ 7.74 (d, *J* = 7.8 Hz, 1H), 7.65 (d, *J* = 8.0 Hz, 1H), 7.32–7.14 (m, 6H), 7.06 (s, 1H), 4.68 (t, *J* = 7.8 Hz, 1H), 3.14–2.90 (m, 2H), 2.66 (dtd, *J* = 12.6, 7.9, 4.6 Hz, 1H), 2.25 (dq, *J* = 12.7, 7.8 Hz, 1H); ¹³**C NMR** (100 MHz, CDCl₃) δ 150.2, 145.6, 143.9, 140.2, 139.7, 127.4, 126.8, 125.2, 124.8, 124.4, 123.9, 123.2, 122.5, 120.9, 47.4, 36.6, 31.9; **IR** (neat) 3060, 2922, 1567, 1475, 1456, 1133 cm⁻¹; **HRMS** (TOF MS CI+) *m* / *z* calcd for C₁₇H₁₄OH (M + H)⁺ 251.0894, found 251.0894.



3.21. The product was prepared according to general method E using NiBr₂•glyme (9.2 mg, 0.030 mmol, 15 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.47** (89.0 mg, 0.200 mmol, 1.00 equiv). The reaction

mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% hexanes) to afford the title compound as a colorless oil (29.1 mg, 0.110 mmol, 55%). **TLC R**_f = 0.5 (hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.74 (d, *J* = 7.8 Hz, 1H), 7.63 (d, *J* = 7.6 Hz, 1H), 7.32–7.27 (m, 1H), 7.25–7.08 (m, 5H), 6.87 (s, 1H), 4.46 (t, *J* = 5.9 Hz, 1H), 2.96–2.79 (m, 2H), 2.29–2.07 (m, 2H), 2.00–1.75 (m, 2H); ¹³**C NMR** (125 MHz, CDCl₃) δ 152.0, 139.8, 139.6, 137.9, 137.1, 130.2, 129.3, 126.6, 125.8, 124.1, 123.6, 123.0, 122.3, 122.1, 41.2, 32.8, 29.4, 20.4; **IR** (neat) 3060, 2915, 1456, 1435, 1308, 1129 cm⁻¹; **HRMS** (TOF MS CI+) *m*/*z* calcd for C₁₈H₁₆S (M)⁺ 264.0973, found 264.0966.



3.22. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.49** (114 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title compound as a colorless oil (59.4 mg, 0.153 mmol, 75%). **TLC R**_f = 0.4 (5% EtOAc/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.98 (d, *J* = 8.4 Hz, 1H), 7.72 (d, *J* = 8.4 Hz, 2H), 7.34–7.26 (m, 3H), 7.25–7.09 (m, 6H), 7.01 (d, *J* = 7.6 Hz, 1H), 4.53 (t, *J* = 8.1 Hz, 1H), 3.08–2.92 (m, 2H), 2.55 (dtd, *J* = 12.4, 7.8, 4.6 Hz, 1H), 2.32 (s, 3H), 2.12 (dq, *J* = 12.6, 8.4 Hz, 1H); ¹³**C NMR** (100 MHz, CDCl₃) δ 144.99, 144.98, 144.2, 136.0, 135.5, 130.6, 130.0, 127.2, 127.0, 126.7, 126.4, 124.90, 124.86, 124.8, 123.3, 123.2, 120.4, 114.1, 42.3, 34.2, 31.9, 21.8; **IR** (neat) 3065, 2926, 2850, 1596, 1446, 1368, 1171 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₂₄H₂₁NO₂SNa (M + Na)⁺ 410.1191, found 410.1193.



3.23. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.51** (117.3 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title compound as a colorless oil (68.1 mg, 0.168 mmol, 84%). **TLC R**_f = 0.4 (5% EtOAc/hexanes); ¹**H NMR** (400 MHz, CDCl₃) δ 7.98 (d, *J* = 8.6 Hz, 1H), 7.73 (d, *J* = 8.4 Hz, 2H), 7.32–7.27 (m, 2H), 7.23–7.20 (m, 3H), 7.16 (td, *J* = 7.7, 0.9 Hz, 1H), 7.01–6.97 (m, 1H), 6.92 (dd, *J* = 8.3, 5.4 Hz, 1H), 6.81 (td, *J* = 9.0, 2.3 Hz, 1H), 4.49 (t, *J* = 8.1 Hz, 1H), 3.05–2.91 (m, 2H), 2.57 (dtd, *J* = 12.6, 8.0, 4.7 Hz, 1H), 2.35 (s, 3H), 2.17 (dq, *J* = 12.7, 8.3 Hz, 1H); ¹³**C NMR** (100 MHz, CDCl₃) δ 162.3 (d, *J* = 243 Hz), 146.2 (d, *J* = 8 Hz), 144.8, 140.2 (d, *J* = 2 Hz), 135.8, 135.3, 130.2, 129.9, 126.8, 125.5 (d, *J* = 9 Hz), 126.0, 124.7, 123.1, 123.0, 120.1 113.9, 113.4 (d, *J* = 22 Hz), 111.7 (d, *J* = 22 Hz), 41.4, 34.4, 31.7, 21.6; **IR** (neat) 2923, 1595, 1482, 1368, 1172, 1121 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₂₄H₂₀FNO₂SNa (M + Na)⁺ 428.1096, found 428.1081.



3.24. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and **3.53** (122.5 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title

compound as a colorless oil (38.0 mg, 0.088 mmol, 44%). **TLC** $\mathbf{R}_{f} = 0.5$ (15% EtOAc/hexanes); ¹H NMR (400 MHz, CDCl₃) δ 7.98 (d, J = 8.3 Hz, 1H), 7.72 (d, J = 8.3 Hz, 2H), 7.35–7.26 (m, 2H), 7.24–7.12 (m, 4H), 6.76 (s, 1H), 6.43 (s, 1H), 5.92 (d, J = 8.3 Hz, 2H), 4.43 (t, J = 7.7 Hz, 1H), 2.97–2.81 (m, 2H), 2.60–2.48 (m, 1H), 2.33 (s, 3H), 2.13 (dq, J = 12.7, 7.9 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 147.3, 146.9, 145.0, 137.7, 137.0, 136.1, 135.5, 130.5, 130.0, 127.0, 126.6, 124.9, 123.3, 123.2, 120.4, 114.1, 105.33, 105.31, 101.2, 42.2, 34.6, 31.7, 21.8; **IR** (neat) 2926, 1596, 1474, 1367, 1172, 1120, 1021 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₂₅H₂₁NO₄SNa (M + Na)⁺ 454.1089, found 454.1085.

VIII. SYNTHESIS AND CHARACTERIZATION OF ENANTIOENRICHED ESTERS FOR TABLE 4 AND STEREOSPECIFIC INTRAMOLECULAR REDUCTIVE CROSS-ELECTROPHILE COUPLING REACTION

METHOD G: MANGANESE DIOXIDE OXIDATION OF BENZYLIC ALCOHOLS

$$Ar \xrightarrow{OH} Br \xrightarrow{MnO_2} Ar \xrightarrow{O} Br$$

The product was prepared according to a modified procedure reported by Wipf.²⁷ To a solution of *rac*-benzylic alcohol (1.0 equiv) in CH_2Cl_2 (30 mL) was added in a single portion MnO_2 (8 equiv). The reaction was allowed to stir overnight at room temperature. The resulting slurry was filtered through celite, and the celite was washed with CH_2Cl_2 . Solvent was removed under reduced pressure to afford the pure title compound.

²⁷ Wipf, P.; Xu, W. J. Org. Chem. 1996, 61, 6556

METHOD H: CBS REDUCTION OF BENZYLIC KETONES

Ar
$$(S)$$
-CBS Catecholborane (S) -CBS (T) (T)

The product was prepared according to a modified procedure reported by Okamura.²⁸ In a glovebox, (*S*)-Me-CBS (0.100 equiv) was added to a 50 mL flame-dried round bottom flask equipped with a stir bar. The flask was capped with a septum and removed from the box. Benzylic ketone (1.0 equiv) was added to the flask as a solution in PhMe (20 mL). The reaction was then cooled to -78 °C and catecholborane (2.0 equiv) was added dropwise. After stirring for 24 h at -78 °C, the reaction was warmed to ambient temperature and quenched with water. Saturated NaHCO₃ (15 mL) was added to the reaction flask and the mixture was extracted with EtOAc (3 x 30 mL). The combined organics were washed with brine, dried over MgSO4, and concentrated under reduced pressure. The product was purified by flash column chromatography to afford the title compound.





²⁸ Okamoto, K.; Hayashi, T. Org. Lett. 2007, 9, 5067

3.54. The product was prepared according to general procedure G using *rac*-**3.25** (1.32 g, 3.86 mmol, 1.0 equiv), MnO₂ (2.08 g, 24.0 mmol, 6.22 equiv). The product was purified by flash column chromatography to afford the pure title compound as a yellow oil (0.962 g, 2.84 mmol, 74%). **TLC R**_f = 0.6 (10% EtOAc/hexanes); ¹H **NMR** (400 MHz, CDCl₃) δ 8.47 (s, 1H), 8.04 (dd, *J* = 8.7, 1.8 Hz, 1H), 7.92 (d, *J* = 8.0 Hz, 1H), 7.86 (t, *J* = 8.6 Hz, 2H), 7.61–7.50 (m, 3H), 7.34 (dd, *J* = 7.6, 1.6 Hz, 1H), 7.27–7.21 (m, 1H), 7.08 (td, *J* = 7.8, 1.8 Hz, 1H), 3.44 (t, *J* = 8.3 Hz, 2H), 3.24 (t, *J* = 8.2 Hz, 2H); ¹³C **NMR** (100 MHz, CDCl₃) δ 199.1, 140.9, 135.8, 134.3, 133.2, 132.8, 131.1, 130.0, 129.8, 128.7 (2C), 128.3, 128.0, 127.9, 127.0, 124.6, 124.1, 39.0, 31.2; **IR** (neat) 3057, 2928, 1677, 1626, 1469, 1182, 1122 cm⁻¹; **HRMS** (TOF MS ES+) *m*/*z* calcd for C₁₉H₁₅BrONa (M + Na)⁺ 361.0204, found 361.0218.

(*R*)-**3.25**. The product was prepared according to a general procedure H using (*S*)-Me-CBS (98 mg, 0.28 mmol, 0.100 equiv), **3.54** (0.96 g, 2.8 mmol, 1.0 equiv), and catecholborane (0.59 mL, 5.5 mmol, 2.0 equiv). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (0.91 g, 2.6 mmol, 94%, 96% ee). Analytical data is consistent with the values listed for **3.25** (vide supra).

(*R*)- **3.9.** The product was prepared according to general method B using (*R*)-**3.25** (0.60 g, 1.7 mmol, 1.0 equiv), pivaloyl chloride (0.24 mL, 1.9 mmol, 1.1 equiv) and dimethylaminopyridine (0.24 g, 1.9 mmol, 1.1 equiv). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title compound as a colorless oil (0.49 g, 1.1 mmol, 66%). Analytical data is consistent with the values listed for *rac*-**9** (vide supra). $[\alpha]^{28}_{D}$ +39 (c 1.7, CHCl₃); SFC

analysis (OD-H, 6% IPA, 2.5 mL/min) indicated 96% ee: t_R (major) = 13.6 minutes, t_R (minor) = 14.9 minutes.

(*S*)- **3.10**. The product was prepared according to general method E using NiBr₂•glyme (6.2 mg, 0.020 mmol, 10 mol %), bathophenanthroline (10.0 mg, 0.030 mmol, 15 mol %), Zn⁰ (39.6 mg, 0.600 mmol, 3 equiv), DMA (0.60 mL), and (*R*)- **3.9** (85.1 mg, 0.200 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% hexanes) to afford the title compound as a colorless oil (44.0 mg, 0.180 mmol, 90%, 88% ee). Analytical data is consistent with the values listed for *rac*-**3.10** (vide supra); $[\alpha]^{28}_{D}$ +10 (c 0.9, CHCl₃); **SFC** analysis (OD-H, 6.0% IPA, 2.5 mL/min) indicated 88% ee: t_R (major) = 11.3 minutes, t_R (minor) = 10.5 minutes.

Scheme 3.18: Synthesis of starting material and formation of (S)- 3.13 (Table 3.3)



3.55. The product was prepared according to general procedure G using *rac*-**3.32** (0.94 g, 2.0 mmol, 1.0 equiv), MnO₂ (1.8 g, 20.0 mmol, 10 equiv). The product was purified by flash column chromatography to afford the pure title compound as a yellow oil (0.56 g, 1.2 mmol, 60%). **TLC** $\mathbf{R}_{f} = 0.4$ (5% EtOAc/hexanes); ¹H NMR (400 MHz, CDCl₃) δ 8.38 (s, 1H), 7.98 (dd, J = 8.6, 1.6 Hz, 1H), 7.78 (d, J = 9.0 Hz, 1H), 7.68 (d, J = 8.8 Hz, 1H), 7.51 (dd, J = 8.1, 1.0 Hz, 1H), 7.30

(dd, J = 7.7, 1.6 Hz, 1H), 7.22–7.17 (m, 2H), 7.10 (dd, J = 8.9, 2.6 Hz, 1H), 7.02 (td, J = 7.6, 1.6 Hz, 1H), 3.37 (t, J = 7.9 Hz, 2H), 3.20 (t, J = 7.8 Hz, 2H), 1.00 (s, 9H), 0.25 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 198.5, 156.0, 140.8, 137.3, 133.0, 132.4, 131.4, 130.9, 129.8, 128.2, 128.1, 127.8, 127.2, 124.5, 124.4, 123.1, 115.0, 38.6, 31.1, 25.8, 18.4, -4.1; **IR** (neat) 3056, 2954, 1680, 1598, 1467, 1256 cm⁻¹; **HRMS** (TOF MS CI+) m/z calcd for C₂₅H₂₉BrO₂Si (M)⁺ 468.1120, found 468.1132.

(*R*)-3.32. The product was prepared according to a general procedure H using (*S*)-Me-CBS (66 mg, 0.24 mmol, 0.200 equiv), 3.55 (0.56 g, 1.2 mmol, 1.0 equiv), and catecholborane (0.26 mL, 2.4 mmol, 2.0 equiv). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (0.42 g, 0.90 mmol, 75%). Analytical data is consistent with the values listed for 3.32 (vide supra).

(*R*)-**3.33** The product was prepared according to general method B using (*R*)-**3.32** (0.40 g, 0.85 mmol, 1.0 equiv), pivaloyl chloride (0.12 mL, 0.98 mmol, 1.1 equiv) and dimethylaminopyridine (0.14 g, 1.0 mmol, 1.1 equiv). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title compound as a colorless oil (0.23 g, 0.41 mmol, 52%). Analytical data is consistent with the values listed for *rac*-**3.33** (vide supra). $[\alpha]^{27.5}_{D}$ +67 (c 3.7, CHCl₃); **SFC** analysis (OD-H, 8% IPA, 2.5 mL/min) indicated 96% ee: t_R (major) = 8.1 minutes, t_R (minor) = 8.9 minutes.

(*S*)- **3.13**. The product was prepared according to general method E using NiBr₂•glyme (3.1 mg, 0.010 mmol, 10 mol %), bathophenanthroline (5.0 mg, 0.015 mmol, 15 mol %), Zn⁰ (18.6 mg,

0.300 mmol, 3 equiv), DMA (0.40 mL), and (*R*)-**3.33** (55.5 mg, 0.100 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% hexanes) to afford the title compound as a colorless oil (26 mg, 0.069 mmol, 69%, 92% ee). Analytical data is consistent with the values listed for *rac*-**3.13** (vide supra); $[\alpha]^{26.6}_{D}$ +91 (c 0.55, CHCl₃); **SFC** analysis (AD-H, 7.0% IPA, 2.5 mL/min) indicated 92% ee: t_R (major) = 7.1 minutes, t_R (minor) = 8.6 minutes.

Scheme 3.19: Synthesis of starting material and formation of (S)- 3.12 (Table 3.3)



3.56. The product was prepared according to general procedure G using *rac*-**3.28** (1.79 g, 5.03 mmol, 1.0 equiv), MnO₂ (4.4 g, 51 mmol, 10 equiv). The product was purified by flash column chromatography to afford the pure title compound as a yellow oil (1.22 g, 3.69 mmol, 73%). **TLC** $\mathbf{R}_{\mathbf{f}} = 0.5$ (12% Et₂O/hexanes); ¹H NMR (400 MHz, CDCl₃) δ 8.45 (s, 1H), 8.03 (dd, J = 8.6, 1.6 Hz, 1H), 7.97–7.93 (m, 1H), 7.91–7.85 (m, 1H), 7.58 (dddd, J = 18.5, 7.7, 6.6, 1.3 Hz, 3H), 7.29–7.22 (m, 3H), 7.07 (ddd, J = 8.0, 7.0, 2.1 Hz, 1H), 3.17 (t, J = 7.3 Hz, 2H), 2.90 (t, J = 7.7 Hz, 2H), 2.20–2.11 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 200.0, 141.2, 135.7, 134.4, 133.0, 132.7, 130.7, 129.86, 129.74, 128.62, 128.58, 128.0, 127.9, 127.7, 126.9, 124.8, 124.1, 38.0, 35.6, 24.7; **IR**

(neat) 2939, 2893, 1679, 1620, 1438, 1165 cm⁻¹; **HRMS** (TOF MS CI+) m / z calcd for C₂₀H₁₇BrOH (M + H)⁺ 353.0541, found 353.0442.

(*R*)-**3.28**. The product was prepared according to a general procedure H using (*S*)-Me-CBS (67 mg, 0.29 mmol, 0.100 equiv), **3.56** (1.0 g, 2.9 mmol, 1.0 equiv), and catecholborane (0.62 mL, 5.8 mmol, 2.0 equiv). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (0.83 g, 2.3 mmol, 80%). Analytical data is consistent with the values listed for **3.28** (vide supra).

(*R*)-**3.29.** The product was prepared according to general method B using (*R*)-**3.28** (0.728 g, 2.05 mmol, 1.0 equiv), pivaloyl chloride (0.30 mL, 2.4 mmol, 1.2 equiv) and dimethylaminopyridine (0.33 g, 2.7 mmol, 1.3 equiv). The product was purified by flash column chromatography (5% EtOAc/hexanes) to afford the title compound as a colorless oil (0.23 g, 0.41 mmol, 52%). Analytical data is consistent with the values listed for *rac*-**3.29** (vide supra). [α]^{27.6}_D +84 (c 2.1, CHCl₃); **SFC** analysis (OD-H, 6% IPA, 2.5 mL/min) indicated 94% ee: t_R (major) = 18.2 minutes, t_R (minor) = 21.6 minutes.

(*S*)- **3.12**. The product was prepared according to general method E using NiBr₂•glyme (3.1 mg, 0.010 mmol, 10 mol %), bathophenanthroline (5.0 mg, 0.015 mmol, 15 mol %), Zn⁰ (18.6 mg, 0.300 mmol, 3 equiv), DMA (0.40 mL), and (*R*)-**3.29** (44.0 mg, 0.100 mmol, 1.00 equiv). The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (100% hexanes) to afford the title compound as a colorless oil (16 mg, 0.062 mmol, 62%, 78% ee). Analytical data is consistent

with the values listed for *rac*-3.12 (vide supra); $[\alpha]^{27.6}{}_{D}$ +43 (c 0.66, CHCl₃); SFC analysis (OD-H, 6.0% IPA, 2.5 mL/min) indicated 78% ee: t_R (major) = 12.5 minutes, t_R (minor) = 16.8 minutes.

STEREOCHEMICAL PROOF

Enantioenriched alcohol (*R*)-**3.25** was prepared by enantioselective CBS reduction (vida supra). Absolute configuration of (*R*)-**3.25** and (*R*)-**3.9** were assigned based on the accepted model for selectivity in CBS reductions.²⁹

The absolute configuration of the enantioenriched indane **3.10** was assigned by derivatization of known enantioenriched indanone (*S*)-**3.58** to indane (*S*)-**3.10**. Enantioenriched (*S*)-**3.58** was prepared by an asymmetric reductive Heck reaction as reported by Zhou. The stereochemistry was verified by comparison of the optical rotation to the literature value. Reduction to indane (*S*)-**3.10** and subsequent comparison of the optical rotation and SFC data matched that of (*S*)-**3.10** synthesized by the reductive cross-electrophile coupling reaction. This product corresponds to net inversion at the benzylic center in the reductive cross-electrophile coupling reaction.

²⁹ Corey, E. J.; Helal, C. J. Angew. Chem., Int. Ed. 1998, 37, 1986

Scheme 3.20: Stereochemical course of the reductive cross-electrophile coupling reaction



(Table 3.3).

3.57. The product was prepared according to a modified procedure reported by Zhou.³⁰ To a 100 mL round bottom flask equipped with a stir bar was added 2'-bromoacetophenone (0.67 mL, 5.0 mmol, 1.0 equiv), naphthaldehyde (0.78 g, 5.0 mmol, 1.0 equiv), KOH (0.62 g, 11 mmol, 2.2 equiv), and MeOH/H₂O (15:15 mL). The reaction was stirred overnight at room temperature. The resulting solid was filtered, washed with MeOH/H₂O, and dried by vacuum filtration to afford the title compound as a solid (1.3 g, 4.5 mmol, 90 % yield).

(S)-3.58. The product was prepared according to a modified procedure reported by Zhou.¹⁵ In a glovebox, a flame-dried 7 mL vial equipped with a stir bar was charged with $Pd(dba)_2$ (14 mg,

³⁰ Yue, G.; Lei, K.; Hirao, H.; Zhou, J. Angew. Chem. Int. Ed. 2015, 54, 6531

0.025 mmol, 10 mol %), (*S*)- Tol-SDP (18 mg, 0.03 mmol, 12 mol %), benzoic acid (30 mg, 0.25 mmol, 1.0 equiv) and degassed ethylene glycol (1.25 mL). After stirring for 10 min, N-diisopropylethylamine (130 μ L, 1.5 mmol, 3.0 equiv) and **3.57** (85 mg, 0.25 mmol, 1.0 equiv) were added and the mixture was taken out of the glove box. The reaction was stirred for 24 h in a pre-warmed oil bath at 100 °C then allowed to cool to room temperature. The reaction mixture was eluted through a silica plug (with Et₂O) and concentrated under reduced pressure. The product was purified by flash column chromatography (0%-10% Et₂O/hexanes) to afford the title compound as a colorless oil (27 mg, 0.13 mmol, 50% yield). Analytical data is consistent with literature values.³¹ **¹H NMR** (400 MHz, CDCl₃) δ 7.87 (d, *J* = 7.9 Hz, 1H), 7.84–7.77 (m, 3H), 7.67 (s, 1H), 7.08 (t, *J* = 7.5 Hz, 1H), 7.50–7.43 (m, 3H), 7.30 (d, *J* = 7.8 Hz, 1H), 7.15 (dd, *J* = 8.4, 1.6 Hz, 1H), 4.77 (dd, *J* = 8.2, 3.7 Hz, 1H), 3.31 (dd, *J* = 19.3, 8.1 Hz, 1H), 2.79 (dd, *J* = 19.3, 3.9 Hz, 1H).

(*S*)- **3.10**. A 20 ml scintillation vial was charged with **3.58** (27 mg, 0.13 mmol, 1.0 equiv), dissolved in TFA (0.5 ml) and cooled to 0 °C in an ice bath. Et₃SiH (1.0 ml) was added dropwise over the course of 20 min and the reaction was allowed to stir for 3 h at 0 °C. The reaction was then allowed to warm to room temperature, was concentrated under reduced pressure, and the residue was taken up and purified by flash column chromatography (100% hexanes) to afford the title compound as a colorless oil (9.2 mg, 0.038 mmol, 36% yield, 79% ee). Analytical data is consistent with the values listed for (*S*)-**10** (vide supra). $[\alpha]^{27.4}_{D}$ +7 (c 0.46, CHCl₃); **SFC** analysis (OD-H, 6.0% IPA, 2.5 mL/min) indicated 79% ee: t_R (major) = 11.4 minutes, t_R (minor) = 10.6 minutes.

³¹ Yu, N-U.; Xu, M-H. J. Org. Chem., 2013, 78, 2736.




































































Method Name:MOK-II-266rac Run Name:LEH-5-RAC-Np-Piv-Br-22 Date:5/10/2015 Time:9:53:59 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	12.02	12.81	13.95	0.00	51.51	209.7	88.4	51.507
2	UNKNOWN	13.95	14.86	15.79	0.00	48.49	179.2	83.2	48.493
Total						100.00	388.8	171.5	100.000

Method Name:MOK-II-266rac Run Name:MOK-2-Np-Piv-Br-ee-22

Date:5/10/2015 Time:9:43:48 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	12.25	12.74	13.59	0.00	97.90	2143.7	1166.3	97.901
2	UNKNOWN	14.45	14.90	15.62	0.00	2.10	51.7	25.0	2.099
Total						100.00	2195.5	1191.3	100.000
Method Name:MOK-II-266rac Run Name:LEH-6-080-D3 Date:11/24/2015 Time:7:09:09 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	17.99	18.53	19.68	0.00	49.85	678.1	395.5	49.846
2	UNKNOWN	20.86	21.48	22.75	0.00	50.15	564.1	397.9	50.154
Total						100.00	1242.2	793.3	100.000

Method Name:MOK-II-266rac Run Name:LEH-5-084-A3 Date:11/24/2015 Time:7:13:20 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	17.57	18.20	19.67	0.00	96.84	138.8	94.2	96.842
2	UNKNOWN	21.01	21.63	22.76	0.00	3.16	5.9	3.1	3.158
Total						100.00	144.7	97.3	100.000

Date:11/24/2015 Time:7:00:17 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	7.80	8.20	8.58	0.00	50.65	813.6	216.6	50.651
2	UNKNOWN	8.58	8.87	9.48	0.00	49.35	730.6	211.1	49.349
Total						100.00	1544.2	427.7	100.000

Date:11/24/2015 Time:7:01:31 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	7.69	8.12	8.65	0.00	97.92	2231.6	705.5	97.918
2	UNKNOWN	8.65	8.86	9.54	0.00	2.08	45.6	15.0	2.082
Total						100.00	2277.2	720.5	100.000

Method Name:MOK-II-266rac Run Name:MOK-2-266rac3 Date:9/7/2015 Time:3:04:42 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	10.21	10.54	11.04	0.00	49.76	1237.1	347.4	49.764
2	UNKNOWN	11.04	11.35	11.97	0.00	50.24	1143.6	350.7	50.236
Total						100.00	2380.8	698.2	100.000

Method Name:MOK-II-266rac Run Name:LEH-5-151-pub3

Date:9/7/2015 Time:3:03:27 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
2	UNKNOWN	10.23	10.50	10.87	0.00	6.12	174.1	48.8	6.121
1	UNKNOWN	10.96	11.30	11.98	0.00	93.88	2091.5	747.8	93.879
Total						100.00	2265.6	796.6	100.000

Method Name:MOK-266-NAP-6-Pdt Run Name:LEH-6-083-F1 Date:11/24/2015 Time:7:04:23 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	12.21	12.71	13.45	0.00	51.53	316.7	131.5	51.529
2	UNKNOWN	16.40	17.02	17.84	0.00	48.47	228.9	123.7	48.471
Total						100.00	545.7	255.3	100.000

Method Name:MOK-266-NAP-6-Pdt Run Name:LEH-6-085-A1

Date:11/24/2015 Time:7:06:01 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	12.02	12.53	13.33	0.00	88.75	2280.2	1130.7	88.750
2	UNKNOWN	16.35	16.81	17.57	0.00	11.25	273.6	143.3	11.250
Total						100.00	2553.9	1274.1	100.000

Method Name:MOK-IV-ractbsprodMETH Run Name:MOK-IV-otbsracprod1

Date:11/24/2015 Time:6:58:37 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	6.44	6.88	7.40	0.00	49.85	1904.4	421.7	49.849
2	UNKNOWN	7.88	8.19	8.71	0.00	50.15	1503.7	424.3	50.151
Total						100.00	3408.1	846.0	100.000

Method Name:MOK-IV-ractbsprodMETH Run Name:MOK-IV-otbseeprod2 Date:11/24/2015 Time:6:57:22 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	6.76	7.13	7.53	0.00	95.85	676.0	138.3	95.849
2	UNKNOWN	8.29	8.61	9.00	0.00	4.15	22.6	6.0	4.151
Total						100.00	698.6	144.3	100.000

Method Name:MOK-II-266rac Run Name:LEH-5-036-realeezhou3

Date:9/7/2015 Time:2:49:52 PM



Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[µV]	[µV.Min]	[%]
1	UNKNOWN	10.33	10.57	10.96	0.00	10.54	58.2	15.8	10.536
2	UNKNOWN	11.03	11.36	11.93	0.00	89.46	446.0	134.4	89.464
Total						100.00	504.2	150.3	100.000

Chapter 4

Nickel-Catalyzed Generation of 2-PyridylZinc Reagents:

Applications to Deoxygenation and Intramolecular Conjugate Additions

4.1 Introduction

Functionalized organozinc reagents are critical for the construction of natural products and other complex organic molecules.^{1,2} Their tendency to undergo selective reactions with transition metals enables the construction of new carbon–carbon bonds under mild reaction conditions. Methods to prepare functionalized organozincs are diverse and are an important topic of continuing research.³ However, the preparation of benzylzinc reagents present particular challenges. While benzylzinc reagents can be prepared by insertion of elemental zinc with primary benzylic halides (Scheme 4.1a), secondary benzylic halides frequently provide low yields due to formation of stable radicals that undergo competitive Wurtz-type coupling. Exchange of alkylboron reagents with a zinc (II) species provides an alternative method for the preparation of secondary benzylzinc reagents, but relies on synthesis of the requisite alkylboron reagent (Scheme 4.1b).⁴ Fragmentation of bulky homobenzylic zinc alkoxides is a creative approach that circumvents formation of benzylzic radicals (Scheme 4.1c).⁵

¹ (a) G.S. Silverman, in: P.E. Rakita (Ed.), Handbook of Grignard reagents, Marcel Dekker, New York, 1996. (b) Knochel, P.; Leuser, H.; Gong, L.-Z.; Perrone, S.; Kneisel, F. F. Handbook of Functionalized Organometallics: Applications in Synthesis; Knochel, P., Ed.; Wiley: Weinheim, 2005; p 251.

² (a) Knochel, P.; Leuser, H.; Gong, L.-Z.; Perrone, S.; Kneisel, F. F. Polyfunctional Zinc Organometallics for Organic Synthesis. In *Handbook of Functionalized Organometallics: Applications in Synthesis*; Knochel, P., Ed.; Wiley: Weinheim, 2005; p 251. For specific examples in total synthesis, see: (b) Fürstner, A.; Weintritt, H. *J. Am. Chem. Soc.* **1998**, *120*, 2817. (c) Masaki, H.; Maeyama, J.; Kamada, K.; Esumi, T.; Iwabuchi, Y.; Hatakeyama, S. *J. Am. Chem. Soc. Soc.* **2000**, *122*, 5216. (d) Aoyagi, S.; Hirashima, S.; Saito, K.; Kibayashi, C. *J. Org. Chem.* **2002**, *67*, 5517.

³ (a) Klatt, T.; Markiewicz, J. T.; Sämann, C.; Knochel, P. *J. Org. Chem.* **2014**, *79*, 4253. (b) Stathakis, C. I.; Manolikakes, S. M.; Knochel, P. *Org. Lett.* **2013**, *15*, 1302. (c) Colombe, J. R.; Bernhardt, S.; Stathakis, C.; Buchwald, S. L.; Knochel, P. **2013**, *15*, 5754. (d) Quinio, P.; François, C.; Cuesta, A. E.; Steib, A. K.; Achrainer, F.; Zipse, H.; Karaghiosoff, K.; Knochel, P. *Org. Lett.* **2015**, *17*, 1010.

⁴ (a) Langer, F.; Knochel, P. *Tetrahedron Lett.* **1995**, *36*, 4591. (b) Hupe, E.; Calaza, M. I.; Knochel, P. J. Organomet. Chem. **2003**, *680*, 136.

⁵ Piazza, C.; Millot, N.; Knochel, P. J. Organomet. Chem. 2001, 624, 88.



Scheme 4.1 Summary of methods for preparation of benzylic zinc reagents

Transition-metal catalysis offers alternative strategies for formation of organometallic reagents.^{6,7,8} Formation of benzylnickel complexes can be accomplished by oxidative addition of nickel catalysts with benzylic electrophiles.⁹ We reasoned that, with catalytic quantities of a nickel catalyst and in the presence of a super-stoichiometric organozinc reagent such as diethylzinc, transmetallation would generate the desired organozinc reagent. The Shi group has reported a related synthesis of primary benzylic Grignard reagents from benzylic alcohols.¹⁰ In this chapter, we report nickel-catalyzed formation of secondary benzylzinc reagents from readily accessible 2pyridylcarbinols (Scheme 4.1d). Activation the alcohol is achieved of using diethylchlorophosphate;¹¹ quenching with methanol provides the products of deoxygenation in a one-pot reaction. We propose that the pyridyl substituent serves as a directing group and stabilizes

⁶ (a) Ishiyama, T.; Murata, M.; Miyaura, N. J. Org. Chem. **1995**, 60, 7508. (b) Rosen, B. M.; Huang, C.; Percec, V. Org. Lett. **2008**, 10, 2597. (c) Ishiyama, T.; Takagi, J.; Ishida, K.; Miyaura, N.; Anastasi, N. R.; Hartwig, J. F. J. Am Chem. Soc. **2002**, 124, 390.

⁷ Gao, F.; Hoveyda, A. H. J. Am. Chem. Soc. **2010**, 132, 10961.

⁸ Dao, H. T.; Baran, P. S. Angew. Chem. Int. Ed. 2014, 53, 14382.

⁹ (a) Wisniewska, H. M.; Swift, E. C.; Jarvo, E. R. J. Am. Chem. Soc. **2013**, 135, 9083. (b) Do, H-Q.; Chandrashekar, E. R. R.; Fu, G. C. J. Am. Chem. Soc. **2013**, 135, 16288

¹⁰ Yu, D-G.; Wang, X.; Zhu, R-Y.; Luo, S.; Zhang, X-B.; Wang, B-Q.; Wang, L.; Shi, Z-J. J. Am. Chem. Soc. **2012**, 134, 14638.

¹¹ Phosphate esters generated in situ are effective electrophiles in nickel-catalyzed transformations, see: (a) Ackerman,

L. K. G.; Anka-Lufford, L. L.; Naodovic, M.; Weix, D. J. Chem. Sci. 2015, 6, 1115.

the benzylzinc reagent.^{12,13} This method provides reduction of hydroxymethylpyridines with incorporation of deuterium in the benzylic position, which can be useful probing metabolic pathways.

4.2 Development of Nickel-Catalyzed Formation of 2-Pyridyl Benzylzinc Reagents

We examined 2-pyridyl carbinol **4.1** as our model substrate. Reactions were performed employing diethylchlorophosphate to activate the alcohol, a series of nickel catalysts, and diethylzinc as transmetallating agent. We found that at room temperature, Ni(dppe)Cl₂ furnishes the desired deoxygenated product **4.2** in high yield (Table 4.1, entry1). The data in Table 4.1 illustrate how changes in the reaction parameters affect the transformation. When the nickel catalyst or diethylchlorophosphate are omitted, the reaction does not afford desired product (entries 2–4). These results are consistent with a mechanism of benzylic alcohol activation prior to reaction with a nickel catalyst. The use of a non-polar solvent or ZnMe₂ instead of ZnEt₂ does not significantly affect the efficiency of the reaction (entries 5 and 6). A decreased yield of **4.2** was observed when catalysts ligated by other bidentate or monodentate phosphine ligands were used in the reaction (entries 7-10). Under optimized conditions, the reaction scales well, providing desired product in high yield when run on gram scale (entry 11).

¹² General pyridyl directing groups: (a) Reviews on removable pyridine directing groups (a) Itami, K.; Mitsudo, K.; Nokami, T.; Kamei, T.; Koike, T.; Yoshida, J-I. J. Organomet. Chem. 2002, 653, 105. (b) Itami, K.; Yoshida, J-I Synlett. 2006, 2, 157. Direcred reactions by Chatani/Murai (a) Chatani, N.; Inoue, S.; Yokata, K.; Tatamidani, H.; Fukumoto, Y. Pure. Appl. Chem. 2010, 82, 1443. (b) Chatani, N.; Ie, Y.; Kakiuchi, F.; Murai, S. J. Am. Chem. Soc. 1999, 121, 8645. (c) Tatamidani, H.; Yokata, K.; Kakiuchi, F.; Chatani, N. J. Org. Chem. 2004, 69, 5615. More recent publications: (a) Wang, J.; Chen, W.; Zuo, S.; Liu, L.; Zhang, X.; Wang, J. Angew. Chem., Int. Ed. 2012, 51, 12334. (b) Lei, Z-Q.; Pan, F.; Li, H.; Li, Y.; Zhang, X-S.; Chen, K.; Wang, X.; Li, Y-X.; Sun, J.; Shi, Z-J. J. Am. Chem. Soc. 2015, 137, 5012. Pyridine is a privileged directing group in C–H bond activation, see: (a) Suggs, J. W. J. Am. Chem. Soc. 1979, 101, 489. (b) Lim, Y–G.; Kang, J–B.; Kim, Y. H. J. Chem. Soc., Perkin Trans. 1996, 1, 2201. (c) Jun, C–H.; Hwang, D–C.; Na, S–J. Chem. Commun. 1998, 1405. (d) Chatani, N.; Ishii, Y.; Kakiuchi F.; Murai, S. J. Org. Chem. 1998, 63, 5129. (e) Chatani, N.; Asaumi, T.; Ikeda, T.; Yorimitsu, S.; Ishii, Y.; Kakuichi, F.; Murai, S. J. Am. Chem. Soc. 2000, 122, 12882.

 ¹³ Bioactive references: (a) Shi, N.; Lu, C.; Ho, C. C.; Shen, Y. *Rec. Nat. Prod.* 2013, 7, 1. (b) Fürstner, A.; Leitner, A.; Méndez, M.; Krause, H. *J. Am. Chem. Soc.* 2002, *124*, 13856. (c) Kubota, T.; Ishiguro Y.; Yamamoto, S.; Fromont, J.; Kobayashi, J. *Heterocycles* 2010, *80*, 1407. (d) Fischer, D. F.; Sarpong, R. *J. Am. Chem. Soc.* 2010, *132*, 5926. Actos (Takeda). Ligand references: (a) Chelucci, G.; Thummel, R. P. *Chem. Rev.* 2002, *102*, 3129.



Table 4.1. Effect of Reaction Parameters on the Efficiency of Deoxygenation of 4.1

4.3 Exploration of Scope of Deoxygenation of 2-pyridyl Alcohols

Having developed robust conditions for the deoxygenation of **4.1**, we turned our attention to investigating the source of "H" found in the product. We proposed that **4.2** is formed from the methanol quench of the benzylzinc reagent generated in situ. To test our hypothesis, we performed an isotopic labeling study with methanol-d4. Upon quenching the deoxygenation reaction of **4.3** with CD₃OD, we observed high yield and high deuterium incorporation of the corresponding product **4.4** (eq 4.1). The result of the isotopic labeling experiment provides strong support for our proposed hypothesis that the reaction proceeds by formation of a benzylzinc reagent.

Equation 4.1. Deuterium incorporation via quenching with deuteromethanol



Next, we examined the formation of alkylzinc reagents from a range of benzylic alcohols (Table 4.2). Secondary benzylic alcohols bearing alkyl chains are well tolerated under the reaction conditions (entries **4.5**, and **4.8**). Substrates containing acetal and alkene functionalities provide

deoxygenated products in good yield (**4.6** and **4.11**). The high yield of **4.6** demonstrates excellent orthogonality to hydroboration transmetallation procedures often used to prepare secondary benzylic zinc reagents.¹⁴ Of particular note, is the formation of byproduct **4.12**, isolated from reactions desired to produce only **4.11**. We hypothesize this by product is the result of a metalloene reaction resulting from the proposed benzylic reagent which has been previously described.¹⁵

Table 4.2 Scope of Deoxygenation



A mechanism consistent with the formation of the proposed benzylic zinc reagents is outlined in Figure 1. The active catalyst is formed by reduction of Ni(dppe)Cl₂ with ZnEt₂ to give the reduced Ni(0)(dppe) species. The Ni(0) catalyst then undergoes oxidative addition into the phosphorylated alcohol to form a benzylic Ni(II) intermediate. Transmetallation results in formation of the benzylic zinc reagent and an ethylnickel complex. β -Hydride elimination followed by loss of ethylene yields a nickel hydride intermediate. This intermediate undergoes a

¹⁴ Hupe, E.; Calaza, M. I.; Knochel, P. J. Organomet. Chem. 2003, 680, 136.

¹⁵ Meyer, C.; Marek, I.; Courtemanche, G.; Normant, J.-F. J. Org. Chem. 1995, 60, 863.

final transmetallation of an ethyl group from zinc followed by reductive elimination to form ethane and regenerate the Ni(0) catalyst.





To further demonstrate the utility of this methodology, we coupled the formation of a benzylic zinc reagent to an intramolecular conjugate addition (Scheme 4.2). We anticipated that upon forming the corresponding benzylzinc reagent from **4.14** or **4.16**, cyclization should occur to afford cyclopentane **4.15** or cyclohexane **4.17** (Scheme 4.2 a and b respectively). 2-Pyridyl carbinol **4.14** and **4.16** bearing a pendant α , β -unsaturated ester underwent smooth cyclization to produce the desired product in modest yield and 4.5:1 diastereomeric ratio. Upon investigation of smaller ring sized such as 3-membered rings yielded no cyclization product, this was also the case when we attempted to produce 7-membered rings using this methodology.





4.4 Conclusion

In summary, we have developed a concise route to secondary organozinc reagents directly from benzylic alcohols. We have successfully applied this methodology to the formal hydrogenolysis of a range of 2-pyridyl carbinols. The reaction proceeds in high yield and with straightforward incorporation of deuterium from deuteromethanol. The reaction has been applied to an intramolecular 1,4-addition. Further exploration of the reactivity of these compounds and elucidation of mechanistic details are underway.

4.5 Experimental Details

GENERAL PROCEDURES

All reactions were carried out under an atmosphere of N₂, or Ar when noted. All glassware was oven- or flame-dried prior to use. Tetrahydrofuran (THF), diethyl ether (Et₂O), dichloromethane (CH₂Cl₂), and toluene (PhMe) were degassed with Ar and then passed through two 4 x 36 inch columns of anhydrous neutral A-2 alumina (8 x 14 mesh; LaRoche Chemicals; activated under a flow of argon at 350 °C for 12 h) to remove H₂O. All other solvents utilized were purchased "anhydrous" commercially, or purified as described. ¹H NMR spectra were recorded on Bruker DRX-400 (400 MHz ¹H, 100 MHz ¹³C, 376.5 MHz ¹⁹F), GN-500 (500 MHz ¹H, 125.7 MHz ¹³C), or CRYO-500 (500 MHz ¹H, 125.7 MHz ¹³C) spectrometers. Proton chemical shifts are reported in ppm (δ) relative to internal tetramethylsilane (TMS, δ 0.00). Data are reported as follows: chemical shift (multiplicity [singlet (s), broad singlet (br s), doublet (d), doublet of doublets (dd), triplet (t), doublet of triplets (dt), doublet of doublet of triplets (ddt), triplet of triplets (tt), quartet (q), quintet (quin), apparent doublet (ad), apparent triplet (at), multiplet (m)], coupling constants [Hz], integration). Carbon chemical shifts are reported in ppm (δ) relative to TMS with the respective solvent resonance as the internal standard (CDCl₃, δ 77.16 ppm). Unless otherwise indicated, NMR data were collected at 25 °C. Infrared (IR) spectra were obtained on a Thermo Scientific Nicolet iS5 spectrometer with an iD5 ATR tip (neat) and are reported in terms of frequency of absorption (cm⁻¹). Analytical thin-layer chromatography (TLC) was performed using Silica Gel 60 F₂₅₄ precoated plates (0.25 mm thickness). Visualization was accomplished by irradiation with a UV lamp and/or staining with KMnO₄, ceric ammonium molybdate (CAM), or *p*-anisaldehyde (PAA) solutions. Flash chromatography was performed using Silica Gel 60 (170-400 mesh) from Fisher Scientific. Melting points (m.p.) were obtained using a Mel-Temp melting

point apparatus and are uncorrected. High resolution mass spectrometry was performed by the University of California, Irvine Mass Spectrometry Center. [1,2-Bis(diphenylphosphino)ethane]dichloronickel(II) was purchased from Strem, stored in a glovebox under an atmosphere of N₂, and used as received. Diethylzinc (ZnEt₂) and diethyl chlorophosphate were purchased from Sigma and used as received. 1-isoquinolinecarboxaldehyde was prepared from selenium (IV) oxide oxidation of 1-methylisoquinoline by a procedure reported by Long.ⁱ All other reagents were purchased commercially and used as received.

SYNTHESIS AND CHARACTERIZATION OF ALL SUBSTRATES

Synthesis of Benzylic Alcohols

General Procedure A. Grignard addition to aldehydes.



4.18. In a flame-dried round-bottom flask, to a solution of 2-pyridinecarboxaldehyde (1.43 mL, 15.0 mmol, 1.00 equiv) in THF (30 mL) was added at 0 °C (3-phenylpropyl)magnesium bromide (1.6 M in THF, 10 mL, 16 mmol, 1.1 equiv). After stirring at room temperature for 1 h, saturated ammonium chloride (25 mL) was added at 0 °C and the reaction was extracted with EtOAc (3 x 25 mL). The combined organic layers were washed with brine (1 x 40 mL), dried over Na₂SO₄, and concentrated in vacuo. The product was purified by column flash chromatography (25–40% EtOAc/hexanes) to afford the title compound as a yellow solid (1.70 g, 7.50 mmol, 50%). **TLC R**r = 0.16 (20% EtOAc/hexanes, UV active); **m.p.** = 58–60 °C; ¹**H NMR** (500 MHz, CDCl₃) δ 8.53 (d, *J* = 4.8 Hz, 1H), 7.66 (td, *J* = 7.8, 1.8 Hz, 1H), 7.29–7.23 (m, 2H), 7.22–7.13 (m, 5H), 4.75 (s, 1H), 4.19 (s, 1H), 2.71–2.59 (m, 2H), 1.91–1.82 (m, 1H), 1.80–1.67 (m, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 161.9, 148.2, 142.4, 136.7, 128.5, 128.3, 125.8, 122.3, 120.4, 72.5, 38.2, 35.8, 26.9; **IR**

(neat) 3192, 2925, 2861, 1594, 1492, 1452 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₁₅H₁₇NONa (M + Na)⁺ 227.1310, found 227.1315.



4.19 Prepared according to a procedure reported by Braga¹⁶, the following amounts of reagents were used: 2-pyridinecarboxaldehyde (0.38 mL, 4.00 mmol, 1.00 equiv), ethylmagnesium bromide (2.4 M in Et₂O, 2.0 mL, 4.8 mmol, 1.2 equiv), and THF (15 mL). Analytical data is consistent with literature values.¹⁷ ¹H NMR (400 MHz, CDCl₃) δ 8.55 (dt, *J* = 4.9, 1.3 Hz, 1H), 7.68 (td, *J* = 7.6, 1.8 Hz, 1H), 7.25 (dd, *J* = 7.8 Hz, 0.5 Hz, 1H), 7.20 (ddd, *J* = 7.6, 4.9, 0.5 Hz, 1H), 4.73–4.66 (m, 1H), 4.17 (d, *J* = 5.5 Hz, 1H), 1.95–1.84 (m,



4.20 Using representative procedure A outlined above, the following amounts of reagents were used: 2-pyridinecarboxaldehyde (0.95 mL, 10 mmol, 1.0 equiv), hept-6-en-1-ylmagnesium bromide (1.2 M in THF, 9.2 mL, 11 mmol, 1.1 equiv), and THF (10 mL). The product was purified by column flash chromatography (25–40% EtOAc/hexanes) to afford the title compound as a yellow oil (1.31 g, 6.38 mmol, 64%). Analytical data is consistent with literature values.¹⁸ **TLC R**_f = 0.1 (50% EtOAc/hexanes, UV active); ¹**H NMR** (500 MHz, CDCl₃) δ 8.55–8.50 (m, 1H), 7.67 (td, *J* = 7.8, 1.8 Hz, 1H), 7.29 (d, *J* = 7.8 Hz, 1H), 7.21–7.15 (m, 1H), 4.93–4.88 (m, 1H), 4.83–4.75 (m, 1H), 4.33 (d, *J* = 5.4 Hz, 1H), 3.99–3.80 (m, 4H), 2.07–1.95 (m, 1H), 1.87–1.74 (m,

¹⁶ Braga, A. L.; Paixao, M. W.; Ludtke, D. S.; Silveira, C. C.; Rodrigues, O. E. D. Org. Lett. 2003, 5, 2365.

¹⁷ Moody, C. J.; Morfitt, C. N. Synthesis **1998**, 7, 1039.

¹⁸ Gebert, A.; Barth, M.; Linden, A.; Widmer, U.; Heimgartner, H. Helv. Chim. Acta 2012, 95, 737.

3H); ¹³C NMR (125 MHz, CDCl₃) δ 161.9, 148.3, 136.7, 122.3, 120.4, 104.4, 72.38, 64.9, 32.5, 29.5.



4.21 Using representative procedure A outlined above, the following amounts of reagents were used: 2-pyridinecarboxaldehyde (0.95 mL, 10 mmol, 1.0 equiv), hept-6-en-1-ylmagnesium bromide (1.2 M in THF, 9.2 mL, 11 mmol, 1.1 equiv), and THF (10 mL). The product was purified by column flash chromatography (25–40% EtOAc/hexanes) to afford the title compound as a red oil (1.31 g, 6.38 mmol, 64%). **TLC R**_f = 0.4 (40% EtOAc/hexanes, UV active); ¹**H NMR** (500 MHz, CDCl₃) δ 8.53 (d, *J* = 4.7 Hz 1H), 7.67 (t, *J* = 7.6 Hz, 1H), 7.26 (d, *J* = 7.6 Hz, 1H), 7.19 (t, *J* = 5.5 Hz, 1H), 5.79 (ddt, *J* = 17.2, 10.2, 6.9 Hz, 1H), 4.98 (d, *J* = 17.2 Hz, 1H), 4.92 (d, *J* = 10.2 Hz, 1H), 4.73 (dd, *J* = 7.7, 4.4 Hz, 1H), 4.25 (br s, 1H), 2.03 (dd, *J* = 10.2, 6.9 Hz, 2H), 1.87–1.77 (m, 1H) 1.74–1.65 (m, 1H), 1.50–1.24 (m, 6H); ¹³**C NMR** (125 MHz, CDCl₃) δ 162.4, 148.3, 139.2, 136.7, 122.3, 120.4, 114.3, 72.9, 38.7, 33.8, 29.2, 28.9, 25.2; **IR** (neat) 3260, 2926, 1594 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₁₃H₁₉ONNa (M + Na)⁺ 228.1364, found 228.1364.



4.22 Using representative procedure A outlined above, the following amounts of reagents were used: 2-pyridinecarboxaldehyde (0.95 mL, 10 mmol, 1.0 equiv) (3-methoxyphenyl)magnesium bromide (0.92 M in THF, 13 ml, 12 mmol, 1.2 equiv), and THF (20 mL). The product was purified by column flash chromatography (20–40% EtOAc/hexanes) to afford the title compound as a yellow oil (0.62 g, 3.1 mmol, 62%). Analytical data is consistent with literature values.⁶ TLC **R**f

= 0.2 (20% EtOAc/hexanes, UV active); ¹**H NMR** (500 MHz, CDCl₃) δ 8.56 (d, *J* = 4.4 Hz, 1H), 7.62 (td, *J* = 7.5, 1.3 Hz, 1H), 7.29–7.14 (m, 3H), 6.96 (d, *J* = 6.7 Hz, 1H), 6.93 (s, 1H), 6.81 (d, *J* = 7.2 Hz, 1H), 5.27 (d, *J* = 3.6 Hz, 1H), 5.29 (d, *J* = 4.3 Hz, 1H); ¹³**C NMR** (125 MHz, CDCl₃) δ 160.6, 159.8, 147.8, 144.8, 136.9, 129.6, 122.5, 121.4, 119.5, 113.5, 112.5, 74.9, 55.3.



4.23 Using representative procedure A outlined above, the following amounts of reagents were used: 2-pyridinecarboxaldehyde (0.36 mL, 3.8 mmol, 1.1 equiv) (4-fluorophenyl)magnesium bromide) (0.27 M in THF, 15 ml, 4.0 mmol, 1.2 equiv), and THF (20 mL). The product was purified by column flash chromatography (20–40% EtOAc/hexanes) to afford the title compound as a yellow oil. (0.60 g, 3.0 mmol, 78%). Analytical data is consistent with literature values.¹⁹ **TLC** $\mathbf{R}_{\mathbf{f}} = 0.2$ (20% EtOAc/hexanes, UV active); ¹**H NMR** (500 MHz, CDCl₃) δ 8.56 (d, J = 4.7 Hz, 1H), 7.62 (dt, J = 7.7, 1.8 Hz, 1H), 7.36–7.30 (m, 2H), 7.23–7.17 (m, 1H), 7.11 (d, J = 7.9 Hz, 1H), 7.05–6.97 (m, 2H), 5.73 (s, 1H), 5.31 (s, 1H); ¹³**C NMR** (125 MHz, CDCl₃) δ 163.4, 161.5, 160.6, 147.9, 139.1 (d, $J_{CF} = 2.77$ Hz), 137.0, 128.9 (d, $J_{CF} = 8.3$ Hz), 122.6 (d, $J_{CF} = 161.8$ Hz), 115.6 (d, $J_{CF} = 21.5$ Hz), 74.31.



4.24 Using representative procedure A outlined above, the following amounts of reagents were used: 2-pyridinecarboxaldehyde (0.95 mL, 10 mmol, 1.0 equiv) (3-(trifluoromethyl)phenyl)magnesium bromide (0.50 M in THF, 20 ml, 10 mmol, 1.2 equiv), and

¹⁹ Kamitani, M.; Ito, M.; Itazaki, M.; Nakazawa, H. Chem. Commun. 2014, 50, 7941.

THF (20 mL). The product was purified by column flash chromatography (20–40% EtOAc/hexanes) to afford the title compound as a yellow oil (0.60 g, 3.0 mmol, 78%). Analytical data is consistent with literature values.²⁰ **TLC** $\mathbf{R}_{\mathbf{f}} = 0.2$ (20% EtOAc/hexanes, UV active); ¹H **NMR** (500 MHz, CDCl₃) δ 8.57 (d, J = 5.0 Hz, 1H), 7.69–7.62 (m, 2H), 7.57 (d, J = 7.8 Hz, 1H), 7.53 (d, J = 7.6 Hz, 1H), 7.45 (t, J = 8.0 Hz, 1H), 7.25–7.20 (m, 1H), 7.14 (d, J = 8.0 Hz, 1H), 5.80 (s, 1H), 5.44 (s, 1H); ¹³C **NMR** (125 MHz, CDCl₃) δ 159.9, 148.0, 144.2, 137.1, 130.8 (q J = 32.4 Hz), 130.4, 129.0, 124.8, 124.7 (q, J = 4.16 Hz), 123.8 (q, J = 3.70 Hz), 122.8, 121.2, 74.4.

Synthesis of Substrate for Scheme 4.2



4.25 Using representative procedure A outlined above, the following amounts of reagents were used: 2-pyridinecarboxaldehyde (0.95 mL, 10 mmol, 1.0 equiv), pent-4-en-1-ylmagnesium bromide (1.1 M in THF, 10 mL, 11 mmol, 1.1 equiv), and THF (15 mL). The product was purified by column flash chromatography (20–40% EtOAc/hexanes) to afford the title compound as a yellow oil (0.784 g, 4.40 mmol, 44%). **TLC R**_f = 0.4 (40% EtOAc/hexanes, UV active); ¹H **NMR** (500 MHz, CDCl₃) δ 8.53 (d, *J* = 4.5 Hz, 1H), 7.68 (td, *J* = 7.6, 1.5 Hz, 1H), 7.25 (d, *J* = 8.1 Hz, 1H), 7.19 (dd, *J* = 6.9, 5.1 Hz, 1H), 5.79 (ddt, *J* = 17.1, 10.3, 6.7 Hz, 1H), 4.99 (dd, *J* = 17.1, 1.6 Hz, 1H), 4.94 (d, *J* = 10.3 Hz, 1H), 4.80–4.78 (m, 1H), 4.25 (s, 1H), 2.16–2.02 (m, 2H), 1.89–1.78

²⁰ Agai, B.; Proszenyak, A.; Tarkanyi, G.; Vida, L.; Faigl, F. Eur. J. Org. Chem. 2004, 3623.

(m, 1H), 1.74–1.62 (m, 1H), 1.58–1.47 (m, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 162.3, 148.3, 138.8, 136.8, 122.4, 120.4, 114.7, 72.7, 38.1, 33.8, 24.6; **IR** (neat) 3260, 2936, 1594 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₁₁H₁₅ONNa (M + Na)⁺ 200.1051, found 200.1056.



4.14 The title compound was prepared according to a modified procedure reported by Grubbs.²¹ In a glovebox, a flame-dried bomb flask was charged with a stir bar, SI-10 (0.71 g, 4.0 mmol, 1.0 equiv), and Hoveyda-Grubbs Catalyst 2nd Generation (178 mg, 0.280 mmol, 0.0700 equiv). The flask was removed from the glovebox, and anhydrous CH₂Cl₂ (50 mL) and ethyl acrylate (2.2 mL, 20 mmol, 5.0 equiv) were added. The flask was sealed and heated to reflux over 24 h. The flask was then cooled to ambient temperature, and the solvent was removed in vacuo. The residue was purified by flash column chromatography to afford the title compound as a pale yellow oil (125 mg, 0.501 mmol, 13%, 13:1 E:Z). TLC $R_f = 0.1$ (40% EtOAc/hexanes, UV active); ¹H NMR (500 MHz, CDCl₃) δ 8.53 (d, J = 4.4 Hz, 1H), 7.68 (td, J = 7.6, 1.5 Hz, 1H), 7.26 (d, J = 7.7 Hz, 1H), 7.20 (dd, J = 6.9, 5. Hz, 1H), 6.93 (dt, J = 15.5, 7.1 Hz, 1H), 5.80 (dt, J = 15.5, 1.5 Hz, 1H), 4.75 (dd, J = 7.2, 4.3 Hz, 1H), 4.33 (br s, 1H), 4.16 (q, J = 7.2 Hz, 2H), 2.28–2.18 (m, 2H), 1.90–1.81 (m, 1H), 1.75–1.66 (m, 1H), 1.64–1.55 (m, 2H), 1.28 (t, J = 7.2 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) & 166.8, 162.0, 148.9, 148.3, 136.8, 122.4, 121.7, 120.4, 72.5, 60.2, 38.0, 32.1, 23.7, 14.3; **IR** (neat) 2938, 1730 cm⁻¹; **HRMS** (TOF MS ES+) m / z calcd for C₁₄H₁₉NO₃Na (M + Na)⁺ 272.1263, found 272.1271.

²¹ Chatterjee, A. K; Choi, T-L.; Sanders, D. P.; Grubbs, R. H. J. Am. Chem. Soc. 2003, 125, 11360.

NICKEL CATALYZED FORMATION OF ALKYL ZINC REAGENTS AND DEOXYGENATION OF

BENZYLIC CARBINOLS

Synthesis of Products for table 4.2

General Procedure A. Deoxygenation of benzylic carbinols



In a glovebox, a flame-dried 7 mL vial equipped with a stir bar was charged with NiCl₂(dppe) (11 mg, 0.020 mmol, 0.10 equiv), alcohol 1 (45 mg, 0.20 mmol, 1.0 equiv) and diethyl chlorophosphate (35 µl, 0.24 mmol, 1.2 equiv) and suspended in THF (1.6 mL). The reaction vial was capped with a screw-cap, fitted with a septum and removed from the glove box. The reaction was placed under N₂, and stirred at which point diethyl zinc (1 M in PhMe, 0.5 mL, 0.5 mmol, 2.5 equiv) was added at room temperature resulting in an immediate color change from slightly orange to transparent yellow. The reaction was then sealed with parafilm and allowed to stir for 18 hours at 24 °C after which the reaction typically is dark brown or black. The reaction was quenched with MeOH and filtered through a plug of silica gel (100% EtOAc). The solvent was removed under reduced vacuum and the crude purified by flash column chromatography (12% EtOAc, 1% triethyl amine in hexanes) to afford the title compound as a pale yellow oil (29 mg, 0.13 mmol, 69%). TLC $R_f = 0.3$ (20% EtOAc/hexanes, UV active); ¹H NMR (500 MHz, CDCl₃) δ 8.52 (ddd, J = 4.9, 1.9, 1.0 Hz, 1H), 7.56 (td, J = 2.0, 7.7 Hz, 1H), 7.29–7.23 (m, 2H), 7.19–7.14 (m, 3H), 7.13–7.06 (m, 2H), 2.81 (t, J = 7.3 Hz, 2H), 2.65 (t, J = 7.7 Hz, 2H), 1.83–1.74 (m, 2H) 1.74–1.64 (m, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 162.2, 149.3, 142.6, 136.3, 128.5, 128.3, 125.7, 122.7, 120.9, 38.3, 35.8, 31.3, 29.6; IR (neat) 3025, 2926, 2855, 2359, 2341, 1598 cm⁻¹; HRMS (TOF MS CI+) m / z calcd for C₁₆H₁₈ONH₄ (M + NH₄)⁺ 211.1361, found 211.1351.

4.8 Using representative procedure B outlined above, the following amounts of reagents were used: NiCl₂(dppe) (11 mg, 0.020 mmol, 0.10 equiv), alcohol **SI-1** (28 mg, 0.20 mmol, 1.0 equiv), diethyl chlorophosphate (35 \Box 1, 0.24 mmol, 1.2 equiv), THF (1.6 mL) and diethyl zinc (1.0 M in PhMe, 0.50 mL, 0.50 mmol, 2.5 equiv). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (16 mg, 0.13 mmol, 67%). Analytical data is consistent with literature values.ⁱⁱ **TLC R**f = 0.3 (20% EtOAc/hexanes, UV active); ¹**H NMR** (500 MHz, CDCl₃) δ 8.53 (d, *J* = 4.4 Hz, 1H), 7.58 (t, *J* = 7.7 Hz, 1H), 7.14 (d, *J* = 7.7 Hz, 1H), 7.09 (t, *J* = 6.5 Hz, 1H), 2.77 (t, *J* = 7.8 Hz, 2H), 1.76 (sex, *J* = 7.7 Hz, 2H), 0.97 (t, *J* = 7.7 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 162.4, 149.3, 136.3, 122.9, 121.0, 40.5, 23.2, 14.0.



4.9 Using representative procedure B outlined above, the following amounts of reagents were used: NiCl₂(dppe) (11 mg, 0.020 mmol, 0.10 equiv), alcohol **3** (42 mg, 0.20 mmol, 1.0 equiv), diethyl chlorophosphate (35 µl, 0.24 mmol, 1.2 equiv), THF (1.6 mL) and diethyl zinc (1.0 M in PhMe, 0.50 mL, 0.50 mmol, 2.5 equiv). The product was purified by flash column chromatography (12% EtOAc/hexanes) to afford the title compound as a pale yellow oil (31 mg, 0.15 mmol, 74 %). Analytical data are consistent with literature values.²² **TLC R**_f = 0.1 (20% EtOAc/hexanes, UV active); ¹**H NMR** (500 MHz, CDCl₃) δ 8.52 (d, *J* = 4.9 Hz, 1H), 7.57 (td, *J* = 7.6, 1.6 Hz, 1H),

²² Kitbunnadaj, R.; Zuiderveld, O.P.; Christophe, B.; Hulscher, S.; Menge, W.M.P.B.; Gelens, E.; Snip, E.; Bakker, R.A.; Celanire, S.; Gillard, M.; Talaga, P.; Timmerman, H.; Leurs, R. *J. Med. Chem.* **2004**, *47*, 2414.

7.14 (d, *J* = 7.8 Hz, 1H), 7.10 (dd, *J* = 7.6, 5.3 Hz, 1H), 4.89 (t, *J* = 6 Hz, 1H), 4.00–3.79 (m, 4H), 2.84 (t, *J* = 7.7 Hz, 2H), 1.92–1.82 (m, 2H), 1.76–1.68 (m, 2H); ¹³**C NMR** (125 MHz, CDCl₃) δ 161.8, 149.3, 136.3, 122.8, 121.1, 104.5, 64.9, 38.1, 33.4, 24.2.



4.4 Using representative procedure B outlined above, the following amounts of reagents were used: NiCl₂(dppe) (11 mg, 0.020 mmol, 0.10 equiv), alcohol **3** (42 mg, 0.20 mmol, 1.0 equiv), diethyl chlorophosphate (35 µl, 0.24 mmol, 1.2 equiv), THF (1.6 mL) and diethyl zinc (1.0 M in PhMe, 0.50 mL, 0.50 mmol, 2.5 equiv). The reaction was quenched with CD₃OD from a sealed ampoule (1 mL) and the product was purified by flash column chromatography (12 % EtOAc, 1% triethyl amine, in hexane) to afford the title compound as a pale yellow oil (30 mg, 0.15 mmol, 78%). **TLC R**_f = 0.1 (20% EtOAc/hexanes, UV active); ¹**H NMR** (500 MHz, CDCl₃) δ 8.52 (d, *J* = 4.6 Hz, 1H), 7.58 (td, *J* = 7.7, 1.9 Hz, 1H), 7.15 (d, *J* = 7.7 Hz, 1H), 7.11–7.08 (m, 1H), 4.89 (t, *J* = 4.7 Hz, 1H), 4.01–3.79 (m, 4H), 2.87–2.79 (m, 1H), 1.91–183 (m, 2H), 1.76–1.68 (m, 2H); ²**H NMR** (500 MHz, CHCl₃) 3.10 (s, 1D) ¹³**C NMR** (125 MHz, CDCl₃) δ 161.8, 149.3, 136.3, 122.8, 121.1, 104.5, 64.9, 38.1, 33.4, 24.2; **IR** (neat) 2922, 2874, 1591, 1567, 1473, 1410 cm⁻¹; **HRMS** (TOF MS CI+) *m* / *z* calcd for C₁₆H₁₈ONH₄ (M + NH₄)⁺ 195.1244, found 195.1251.



4.6 Using representative procedure B outlined above, the following amounts of reagents were used: NiCl₂(dppe) (11 mg, 0.020 mmol, 0.10 equiv), alcohol **SI-3** (41 mg, 0.20 mmol, 1.0 equiv), diethyl chlorophosphate (35 μl, 0.24 mmol, 1.2 equiv), THF (1.6 mL) and diethyl zinc (1.0 M in PhMe, 0.50 mL, 0.50 mmol, 2.5 equiv). The product was purified by flash column chromatography (15% EtOAc/hexanes) to afford the title compound as a pale yellow oil (29 mg, 0.15 mmol, 77%). **TLC R**_f = 0.2 (20% EtOAc/hexanes, UV active); ¹**H NMR** (500 MHz, CDCl₃) δ 8.52 (d, *J* = 4.5 Hz, 1H), 7.58 (td, *J* = 7.6, 1.5 Hz, 1H), 7.14 (d, *J* = 7.6 Hz, 1H), 7.09 (dd, *J* = 7.3, 5.5 Hz, 1H), 5.80 (ddt, *J* = 17.1, 10.4, 3.3 Hz, 1H), 4.98 (dd, *J* = 17.1, 1.1 Hz, 1H), 4.92 (d, *J* = 10.4 Hz, 1H), 2.78 (t, *J* = 7.8 Hz, 2H), 2.03 (dd, *J* = 13.5, 6.5 Hz, 2H), 1.72 (quint, *J* = 7.6 Hz, 2H), 1.44–1.30 (m, 6H); ¹³**C NMR** (125 MHz, CDCl₃) δ 162.2, 149.3, 139.3, 136.3, 122.8, 121.0, 114.3, 38.6, 33.9, 30.0, 29.6, 29.1, 28.9; **IR** (neat) 3075, 2925, 1589 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₁₃H₁₉NNa (M + Na)⁺ 212.1415, found 212.1406.



4.13 Using representative procedure B outlined above, the following amounts of reagents were used: NiCl₂(dppe) (11 mg, 0.020 mmol, 0.10 equiv), alcohol **SI-7** (43 mg, 0.20 mmol, 1.0 equiv), diethyl chlorophosphate (35 µl, 0.24 mmol, 1.2 equiv), THF (1.6 mL) and diethyl zinc (1.0 M in PhMe, 0.50 mL, 0.50 mmol, 2.5 equiv). The product was purified by flash column chromatography (20 % EtOAc, 10% DCM, in hexanes) to afford the title compound as a pale yellow oil (21.5 mg, 0.120 mmol, 54%). Analytical data are consistent with literature values.²³ **TLC R**_f = 0.3 (20% EtOAc/hexanes, UV active); ¹H **NMR** (500 MHz, CDCl₃) δ 8.54 (d, *J* = 4.3 Hz, 1H), 7.57 (td, *J* = 7.6, 1.7, 1H), 7.21 (t, *J* = 7.9 Hz, 1H), 7.12 (s, 1H), 7.10 (t, *J* = 3.2 Hz, 1H), 6.85 (d, *J* = 7.4 Hz, 1H), 6.81 (s, 1H), 6.76 (dd, *J* = 8.4, 2.4 Hz, 1H), 4.13 (s, 2H), 3.77 (s, 3H),

²³ Chen, X.; Zhou, L.; Li, Y.; Xie, T.; Zhou, S. J. Org. Chem. 2014, 79, 230.

¹³C NMR (125 MHz, CDCl₃) δ 160.9, 159.8, 149.4, 141.1, 136.6, 129.6, 123.2, 121.5, 121.3, 114.9, 111.9, 55.2, 44.8.

4.7 Using representative procedure B outlined above, the following amounts of reagents were used: NiCl₂(dppe) (11 mg, 0.020 mmol, 0.10 equiv), alcohol **SI-8** (43 mg, 0.20 mmol, 1.0 equiv), diethyl chlorophosphate (35 µl, 0.24 mmol, 1.2 equiv), THF (1.6 mL) and diethyl zinc (1.0 M in PhMe, 0.50 mL, 0.50 mmol, 2.5 equiv). The crude product was pre-absorbed to a minimal amount of silica and purified by flash column chromatography (12 % EtOAc, 1% triethyl amine, in hexane) to afford the title compound as a pale yellow oil (20.3 mg, 0.120 mmol, 60%). Analytical data are consistent with literature values.²⁴ **TLC R**_f = 0.3 (20% EtOAc/hexanes, UV active); ¹**H NMR** (500 MHz, CDCl₃) δ 8.55 (ddd, *J* = 4.9, 1.9, 1.0 Hz, 1H), 7.58 (td, *J* = 7.6, 1.9 Hz, 1H), 7.24–7.19 (m, 2H), 7.14–7.07 (m, 2H), 7.01–6.95 (m, 2H), 4.12 (s, 2H); ¹³**C NMR** (125 MHz, CDCl₃) δ 162.5 (d, *J*_{CF} = 244.14 Hz), 160.6, 149.4, 136.6, 135.1 (d, J_{CF} = 3.24 Hz), 130.5 (d, J_{CF} = 7.87 Hz), 123.0, 121.3, 115.3 (d, J_{CF} = 21.2 Hz), 43.8.



4.10 Using representative procedure B outlined above, the following amounts of reagents were used: NiCl₂(dppe) (11 mg, 0.020 mmol, 0.10 equiv), alcohol **SI-9** (51 mg, 0.20 mmol, 1.0 equiv),

²⁴ De Houwer, J.; Tehrani, K. A.; Maes, B. U. W. Angew. Chem., Int. Ed. 2012, 51, 2745.

diethyl chlorophosphate (35 µl, 0.24 mmol, 1.2 equiv), THF (1.6 ML) and diethyl zinc (1.0 M in PhMe, 0.50 mL, 0.50 mmol, 2.5 equiv). The product was purified by flash column chromatography (30–50% EtOAc/hexanes) to afford the title compound as a pale yellow oil (29.4 mg, 0.124 mmol, 62%). Analytical data are consistent with literature values¹⁰ TLC $\mathbf{R}_{\mathbf{f}} = 0.2$ (10% EtOAc/hexanes, UV active); ¹H NMR (500 MHz, CDCl₃) δ 8.56 (d, *J* = 4.9 Hz, 1H), 7.61 (dt, *J* = 2.0, 7.6 Hz, 1H), 7.53 (s, 1H), 7.50–7.38 (m, 3H), 7.17–7.09 (m, 2H), 4.21 (s, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 159.8, 149.6, 140.4, 136.8, 132.5, 128.9, 125.8, 125.7, 123.4, 123.3, 123.2, 121.6, 44.3.

Intramolecular 1,4-addition (Scheme 4.2)



4.15 Using representative procedure B outlined above, the following amounts of reagents were used: NiCl₂(dppe) (11 mg, 0.020 mmol, 0.10 equiv), alcohol **16** (50 mg, 0.20 mmol, 1.0 equiv), diethyl chlorophosphate (35 μ l, 0.24 mmol, 1.2 equiv), THF (1.6 mL) and diethyl zinc (1.0 M in PhMe, 0.50 mL, 0.50 mmol, 2.5 equiv). The product was purified by flash column chromatography (20% EtOAc/hexanes) to afford the title compound as a mixture 2.5:1 mixture of diastereomers (37 mg, 0.16 mmol, 80%). H_c was integrated to determine the ratio of diastereomers. Flash column chromatography (5–10% EtOAc/hexanes) was performed a second time to isolate analytically pure cis diastereomer (19 mg, 0.081 mmol, 40%) and a 1:1 mixture of diastereomers (16 mg, 0.07 mmol, 34%). The cis diastereomer was assigned based on the nOe correlation shown.



TLC R_f = 0.1 (10% EtOAc/hexanes, UV active); ¹**H NMR** (500 MHz, CDCl₃) δ 8.55 (d, *J* = 4.7 Hz, 1H), 7.56 (td, *J* = 7.7, 2.0 Hz, 1H), 7.12–7.07 (m, 2H), 4.00 (q, *J* = 7.2 Hz, 2H), 3.43 (q, *J* = 8.0 Hz, 1H), 2.79–2.71 (m, 1H), 2.14–2.04 (m, 2H), 2.02–1.90 (m, 4H), 1.75–1.67 (m, 1H), 1.63–1.54 (m, 1H), 1.17 (t, *J* = 7.2 Hz, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 173.6, 163.2, 149.2, 136.0, 123.5, 121.2, 60.2, 49.8, 40.8, 36.2, 31.9, 30.2, 24.2, 14.3; **IR** (neat) 2935, 1716 cm⁻¹; **HRMS** (TOF MS ES+) *m* / *z* calcd for C₁₄H₁₉NO₂Na (M + Na)⁺ 256.1313, found 256.1316.

ⁱ Cao, B.; Wang, Y.; Ding, K.; Neamati, N.; Long, Y-Q. Org. Biomol. Chem. 2012, 10, 1239.

ⁱⁱ Groenhagen, U.; Maczka, M.; Dickschat, J. S.; Schulz, S. Beilstein J. Org. Chem. 2014, 10, 1421.


























