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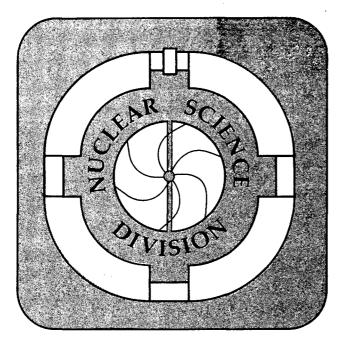
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Search for Supermassive Cahn-Glashow Particles in Lead

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Abstract

A radiochemical search has been made for negatively charged massive elementary particles (X^-) bound to lead nuclei by utilizing the fact that such systems would have the nuclear properties of lead but the chemical properties of thallium. We looked for evidence of Pb X^- by searching for gamma rays emitted following the beta decay of $206 \text{Bi}X^-$ produced by Pb X^- (p,xn) reactions. No evidence of such decays was observed and a limit has been established on the possible concentration of X^- particles in lead of < 1.5 x 10⁻¹³ per nucleon.

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PACS numbers: 14.80.Pb, 23.20.-g, 27.80.+w

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I. INTRODUCTION

It has been suggested that massive, long-lived, electrically charged elementary particles (X^{\pm}) may have been produced in the Big Bang and have survived to the present day in concentrations small enough to have escaped detection.¹ If, as has been suggested, such particles have only electromagnetic interactions with ordinary matter, then an X^+ particle would eventually capture an electron and then behave as an anomalously heavy isotope of hydrogen. An X^- particle would eventually become bound to an atomic nucleus and thereby reduce the effective nuclear charge by one unit. Thus, the chemical properties of an atom of $(Z,N)X^{-}$ would be those of an ordinary atom of (Z-1,N), while its nuclear properties would be nearly the same as those of the nucleus (Z,N) (see Glashow and co-workers ^{1,2}). Estimates of the numbers of such massive elementary particles that were produced in the Big Bang depend sensitively on their assumed properties. If these objects are hadrons, then their abundances have been estimated to be in the range of 10^{-12} to 10^{-10} per nucleon.^{3,4} If they are leptons. however, their concentration may be as high as 10⁻⁵ per nucleon.⁴

Several previous searches have looked for anomalously heavy isotopes of hydrogen, which could correspond either to X^+ particles or to He $X^$ bound states. ^{5,6,7} These experiments have established that for masses < 1 TeV, the concentration of such objects in terrestrial water is less than 10^{-29} per nucleon.⁷ While this limit appears to rule out the existence of X^+ particles in anywhere near their predicted abundance, it does not necessarily preclude a much larger abundance of X^- 's. Such an asymmetry might have been produced by the same type of process that led

to the observed excess of baryons over antibaryons in the universe. Searches in beryllium, boron, carbon, nitrogen, fluorine, magnesium, and curium have established upper limits on the possible concentrations of $X^$ particles in these elements that range from 10^{-8} to 10^{-20} per nucleon.⁸⁻¹² However, as discussed by Turkevich, Wielgoz, and Economou,⁹ nucleosynthesis arguments suggest that these elements may not have been the most favorable places in which to look for X^- particles.

Calculations by several authors^{9,13,14,15} have shown that during charged-particle-induced nucleosynthesis, X^- 's will be preferentially concentrated in the heaviest element produced. Since iron is the element with the highest nuclear charge produced through such fusion reactions, an enhanced abundance of X^- 's in iron was suggested by some of these investigators. A recent search by Norman, Gazes, and Bennett¹³, however, found no evidence for X^- particles in iron and established a limit of < 1.2 x 10^{-12} per nucleon.

Nucleosynthesis of the elements more massive than iron occurs via the s-(slow) and r-(rapid) processes in which successive neutron captures are interspersed with beta decays. Neutron-capture cross sections should not be affected by the presence of an X^- in a nucleus. However, as a result of the variation in the Coulomb binding energy with nuclear charge, the presence of X^- particles in nuclei alters the relative masses of all members of a given isobar. This changes the energies available in beta decay and, hence, the beta decay half lives. The beta-minus decay rates of nuclei on the neutron-rich side of the stability line will be increased by this effect. Since lead is a major product of such neutron-capture nucleosynthesis, an enhanced concentration of X^- particles in lead has been suggested.¹

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We have performed a radiochemical search for PbX^{-} systems, whose chemical properties would be those of thallium. As in all previous searches of this type, the critical assumption we made is that X^{-} particles have only electromagnetic interactions with nuclei. We calculated the Coulomb binding energy of an X^{-} particle to a nucleus using the expression derived by Cahn and Glashow¹ for spherical nuclei with uniform charge distributions. This calculation shows that an X^{-} particle is bound by 302 keV more energy to a bismuth nucleus than to a lead nucleus. Thus, the energy available for the nuclear beta decay of 206 BiX⁻ is 302 keV less than that for the decay of ²⁰⁶Bi. The two principal decay modes of ²⁰⁶Bi to the levels at 3.403- and 3.279-MeV excitation energy in ²⁰⁶Pb have log ft values of of 6.5 and 6.9, respectively.¹⁶ We assume that the log *ft* values for the beta decays of $^{206}BiX^{-}$ are the same as those for ^{206}Bi , because nuclear matrix elements should not be affected by the presence of an X^{-} . An X^{-} bound to a nucleus could also shift the energies of the excited states of that nucleus. As a result of the collective nature of the excited states of ²⁰⁶Pb, such shifts in the positions of these levels are expected to be quite small¹⁷ and are neglected in our analysis. The known¹⁶ major decay modes of 206 Bi and the results of our calculations for the 206 BiX⁻ decay scheme are shown in Fig. 1. We find that the beta decay half-life of $^{206}BiX^{-1}$ is approximately 64 days. The primary gamma ray from $^{206}BiX^{-}$ decay will be essentially the same as that from ²⁰⁶Bi decay.

II. EXPERIMENTAL PROCEDURE

We looked for evidence of PbX⁻ by searching for characteristic gamma rays emitted following the beta decay of 206 BiX⁻. We attempted to produce

 206 BiX⁻ via the PbX⁻(p,xn) reaction. A thick disk containing approximately 23 grams of 99.9999% pure thallium metal (which could contain PbX^{-} bound systems) was bombarded for 4.5 hours with a 3 microAmpere beam of 25-MeV protons from Lawrence Berkeley Laboratory's 88-Inch Cyclotron. The bombarding energy was chosen to be just below the threshold for scattering an X^{-} particle out of a lead nucleus. After the sample was allowed to cool for approximately one month, the lead fraction, which would contain any ${}^{206}\text{Bi}X^{-}$ present, was separated from the target. To do so, the target was first dissolved in 12 M H₂SO₄. In order to determine the lead recovery efficiency, a small amount of ²¹²Pb tracer was added to the The lead fraction was precipitated as PbSO₄ and subsequently solution. dissolved in a mixture of 6 M NH₄COOCH₃ and 6 M CH₃COOH. Na₂CrO₄ was then added to precipitate PbCrO_{Δ}. This precipitate was dissolved in a mixture of 6 M HCl and 3% H₂O₂ and sorbed on a 5 mm x 100 mm column of AG1-X10 anion- exchange resin. The lead fraction was eluted with 6 M HCI and then evaporated to dryness. 6 M NH₄COOCH₃ and 6 M CH₃COOH were added and the solution was brought to the boiling point. Na₂CrO₄ was again added to reprecipitate PbCrO₄ which was filtered and then mounted on a planchet for counting. From measuring the ²¹²Pb activity before and after the chemical separation, we determined that the lead recovery efficiency was 96%.

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Approximately two months after the proton bombardment, the lead fraction was counted for a period of 24 hours at a distance of 1 cm from the front face of a 100-cm³ intrinsic germanium detector. The detector was shielded on all sides with 10 cm of low activity lead. The detector efficiency was measured in the same geometry using standard calibrated gamma sources. The photopeak efficiency at 803 keV was found to be 2.46 x 10^{-2} .

III. RESULTS AND DISCUSSION

The gamma-ray spectrum below 1 MeV observed in counting this lead fraction is shown in Fig. 2. The major peaks seen in this spectrum are due to the decay of ²⁰³Pb, which is produced by the ²⁰³Tl(p,n) reaction and was obviously recovered in the lead chemistry, and the decay of ²⁰²Tl, which is produced by the ²⁰³Tl(p,pn) reaction and was not completely removed by the chemistry. All of the observed peaks could be assigned to the decays of known nuclei. From the measured widths of known gamma-ray peaks and from the measured background counting rate near the expected ²⁰⁶BiX⁻ photopeak position, a 1_o upper limit of 17 counts was established for the ²⁰⁶BiX⁻ decay gamma ray. It should be noted that because there are no unidentified peaks between 525 and 1000 keV, this limit is not very sensitive to the exact position chosen for the ²⁰⁶BiX⁻ decay gamma ray. From this limit and the measured efficiencies, we placed a limit on the number of ²⁰⁶BiX⁻ nuclei present in the irradiated sample of <1.3 x 10⁵.

In order to convert this limit on the number of ${}^{206}\text{Bi}X^-$ nuclei into a limit on the number of PbX⁻ nuclei initially present in the thallium target, we needed to determine the thick-target yield for the Pb(p,xn) ${}^{206}\text{Bi}$ reaction. To do so, we bombarded a thick disk of 99.95% pure lead metal of natural isotopic abundance with a 115 nanoAmpere beam of 25-MeV protons. After bombardment, the sample was counted with a 100-cm³ germanium detector. From the measured yield of ${}^{206}\text{Bi}$ gamma rays, we determined that the ${}^{206}\text{Bi}$ yield per incident 25-MeV proton on a thick target of natural isotopic composition lead is $(1.0 + 0.2) \times 10^{-3}$. The presence of a X⁻ particle

in a lead nucleus should increase this yield as a result of the lowering of both the Coulomb barrier and the reaction threshold energy. However, in order to establish a conservative limit, we have assumed that this yield is unaffected by the presence of a X^- .

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We then established an upper limit on the concentration of PbX⁻ nuclei in thallium. Using the fact that the solar-system abundance of lead is 13.7 times that of thallium¹⁸, we find that the concentration of X⁻ particles in lead is < 1.5 x 10⁻¹³ per nucleon. This limit is well below the level expected if X⁻'s were distributed uniformly in all matter. This is in spite of the enhanced concentration of X⁻ 's in lead suggested by nucleosynthesis arguments. As in all previous searches of this type, the present search was carried out on a terrestrial sample of unknown geochemical history. While the possibility of significant mass fractionation cannot be completely ruled out, the studies of Jørgensen¹⁹ and Bigeleisen²⁰ indicate that such effects should not have greatly depleted the abundance of PbX⁻ in terrestrial thallium. On the basis of the possible mass fractionation during our laboratory chemical processing of the sample, the present limit should be applicable for X⁻ masses up to about 100 TeV.⁹

IV. ACKNOWLEDGEMENTS

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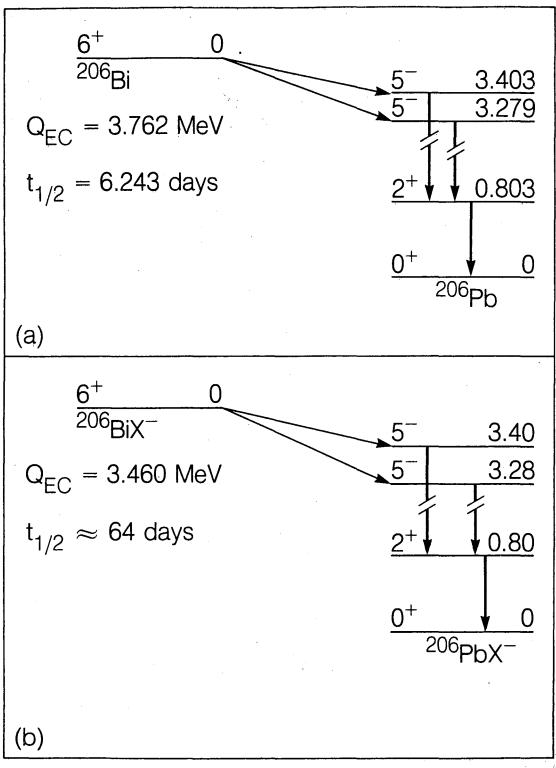
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FIGURE CAPTIONS

- 1. a) Known principal decay modes of ²⁰⁶Bi.
 - b) Calculated decay modes of ²⁰⁶BiX⁻. Note that the spins shown for the levels in ²⁰⁶BiX⁻ and ²⁰⁶PbX⁻ neglect the possible spin of the X⁻ particle.
- 2. Gamma-ray spectrum below 1 MeV observed in 24 hours of counting the lead fraction extracted from the proton-irradiated thallium sample. The expected position of the ²⁰⁶BiX⁻ gamma ray (which is essentially the same as that for ²⁰⁶Bi) is indicated by an arrow. Unlabeled peaks are due to background activities. All gamma-ray energies are given in keV.



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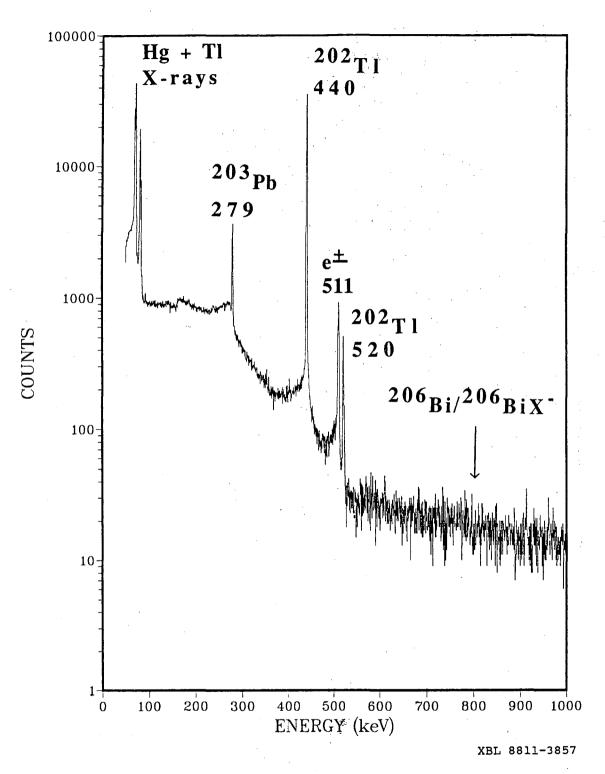


Fig. 2

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