

# Lawrence Berkeley National Laboratory

## Recent Work

### Title

LANTHANUM HEXABORIDE (LaB<sub>6</sub>) RESISTIVITY MEASUREMENT

### Permalink

<https://escholarship.org/uc/item/7qf350jv>

### Author

Williams, M.D.

### Publication Date

1987-02-01



# Lawrence Berkeley Laboratory

UNIVERSITY OF CALIFORNIA

## Accelerator & Fusion Research Division

APPLIED PHYSICS LETTERS  
FEBRUARY 22 1987  
APPLIED PHYSICS LETTERS

Submitted to Applied Physics Letters

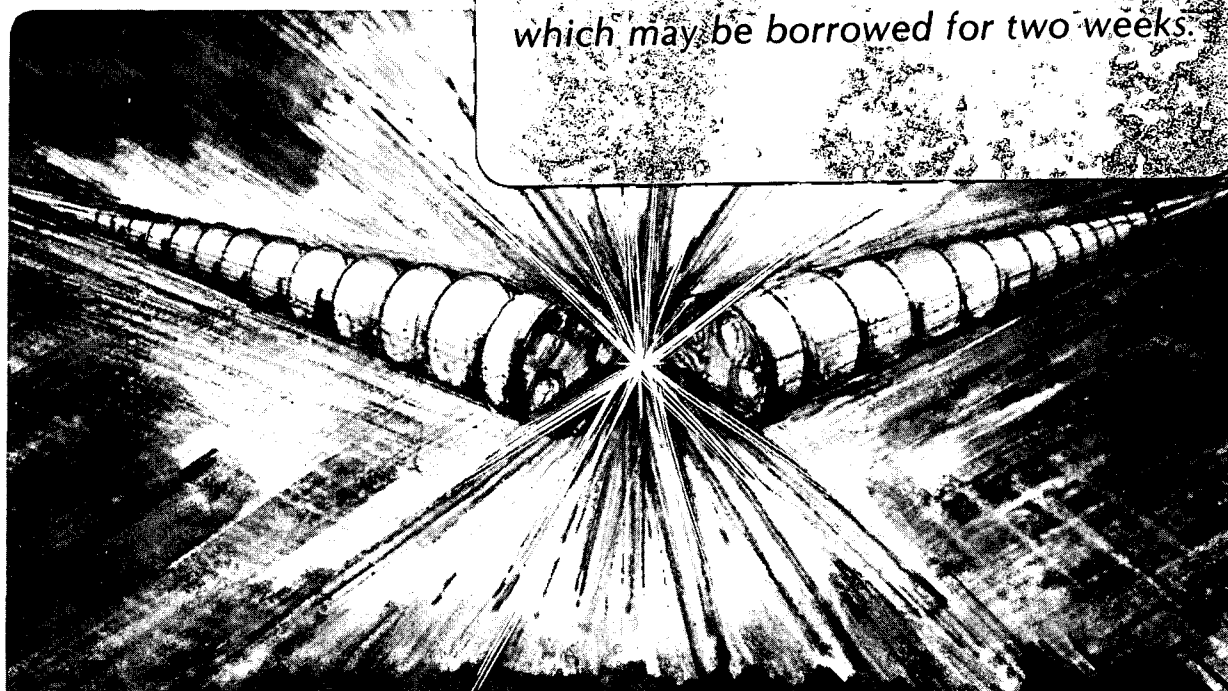
### LANTHANUM HEXABORIDE ( $\text{LaB}_6$ ) RESISTIVITY MEASUREMENT

M.D. Williams, L.T. Jackson, D.O. Kippenhan,  
K.N. Leung, and M.K. West

February 1987

**TWO-WEEK LOAN COPY**

*This is a Library Circulating Copy  
which may be borrowed for two weeks.*



LBL-22931  
c.2

## **DISCLAIMER**

This document was prepared as an account of work sponsored by the United States Government. While this document is believed to contain correct information, neither the United States Government nor any agency thereof, nor the Regents of the University of California, nor any of their employees, makes any warranty, express or implied, or assumes any legal responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by its trade name, trademark, manufacturer, or otherwise, does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof, or the Regents of the University of California. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof or the Regents of the University of California.

LANTHANUM HEXABORIDE ( $\text{LaB}_6$ ) RESISTIVITY MEASUREMENT\*

M. D. Williams, L. T. Jackson, D. O. Kippenhan,  
K. N. Leung, and M. K. West

Lawrence Berkeley Laboratory  
University of California  
Berkeley, CA 94720

and

C. K. Crawford

Kimball Physics Inc.  
Wilton, NH 03086

ABSTRACT

In the development of high power free electron lasers, intense electron beams are required. Large area, directly-heated lanthanum hexaboride cathodes have been proposed as the electron emitter. To aid in the design of the cathode, the resistivity of lanthanum hexaboride as a function of material density and temperature has been measured.

\* Supported by Lawrence Livermore National Laboratory under contract No. W-7405-ENG-48, and Lawrence Berkeley Laboratory under contract No. DE-AC03-76SF00098.

It has been known for some time that lanthanum hexaboride ( $\text{LaB}_6$ ) is a good material for use as an electron emitter.<sup>1,2,3</sup> It has also been demonstrated that directly-heated  $\text{LaB}_6$  cathodes, either in the form of filaments or in a coaxial geometry can be operated successfully in a plasma source.<sup>4,5,6</sup> In some applications, such as high power free electron lasers, large area cathodes are required. A directly-heated  $\text{LaB}_6$  electron emitter has been proposed for the generation of intense electron beams.<sup>7</sup> By balancing the ohmic heating with the cooling power; the proper shape of the cathode can be computed. The resistivity of  $\text{LaB}_6$  is an important parameter in the computation. However, there is little available data on the resistivity of  $\text{LaB}_6$  as a function of temperature and material density. To aid in the design of the proposed cathode, measurements of the resistivity of various densities of  $\text{LaB}_6$  at different temperatures have been performed. This paper describes the apparatus, test procedures and the results of these measurements.

The manufacture of solid lanthanum hexaboride is normally accomplished by sintering of  $\text{LaB}_6$  powder under various temperatures and pressures to obtain the desired material density.<sup>1,8</sup> In principle, the maximum obtainable density is  $4.7 \text{ g/cm}^3$ . Densities ranging from 60% to 95% of this value are readily available. Higher densities can be produced, but at a higher manufacturing cost.  $\text{LaB}_6$  material with densities lower than 60% is quite soft and structurally weak, and therefore is not suitable for cathode use. For high densities,  $\text{LaB}_6$  has ceramic-like properties in hardness and requires special tooling and techniques for machining. Most data available in the literature correspond to densities of 80 to 85%, which is the most common off-the-shelf material available.<sup>1,2,3,9</sup>

The experimental set-up is shown in Figure 1. The test apparatus was installed in a vacuum chamber equipped with a window which provided a full view of the  $\text{LaB}_6$  sample. An optical pyrometer (wavelength 0.65 microns) was used to measure sample temperature. The window transmission was separately checked with a reference light, and a transmission loss of approximately 2% was determined. An emissivity of 0.7 was used for correction of the observed temperature.<sup>9</sup>

The  $\text{LaB}_6$  samples<sup>10</sup> were cylindrical rods of approximately 0.63 cm diameter and 5 cm length. Material density of the tested samples varied from 60% to 95%. The test fixture was designed to hold the rods in compression between the current supplying contacts. The difference in voltage between two points on the  $\text{LaB}_6$  rod was measured with two voltage sensing contacts (typically 2.3 cm apart). The central, most uniform temperature section of the  $\text{LaB}_6$  rod was chosen for the voltage measurements; temperature variations along the section were in the range of 2%. To avoid reactions with the  $\text{LaB}_6$ , rhenium was used for all contact surfaces.

Each  $\text{LaB}_6$  sample was in turn installed in the assembly. Current and voltage readings were first recorded at a very low current to establish the resistivity,  $\rho$ , at ambient temperature. Results of these measurements are represented by the five data points shown in the lower part of the curves in Fig. 2.

The applied heater current was then increased gradually until the  $\text{LaB}_6$  temperature came within the pyrometer's range ( $\sim 800^\circ\text{C}$ ). Voltage, current and temperature readings were recorded after allowing the sample to come to equilibrium. All  $\text{LaB}_6$  samples were tested several times at various temperatures to assure repeatability. It was found that all test runs were in close agreement.

Figure 2 is a plot of the measured resistivity,  $\rho$ , for six different  $\text{LaB}_6$  densities as a function of temperature. It can be seen that  $\rho$  increases approximately linearly with temperature. For a given temperature,  $\rho$  decreases as the material density increases. Note that there is nearly a factor of four difference in resistivity between the 60% and the 95% material.

For directly heated cathodes, the cathode resistance drops as size is scaled up. Thus to keep cathode impedances matched to conveniently available power supplies, it is advantageous to use the lower density materials for larger cathodes.

Looking at Fig. 2, it can be observed that straight lines through the data points approximately intersect at the origin. Thus an empirical relationship between the resistivity, the density, and the temperature can be written in the form:

$$\rho = \rho' \cdot T = A \cdot D^n \cdot T \quad (1)$$

where A and n are constants, D is the material density fraction (defined as the ratio of the measured density over the theoretical density), and  $\rho'$  is the resistivity coefficient. The resistivity coefficient  $\rho'$  for each density is just the slope of the corresponding curve.

Figure 3 shows a logarithmic plot of  $\rho'$  versus D. Because the data points lie roughly on a straight line, the  $D^n$  power law is a good approximation. Since  $\ln \rho' = \ln A + n \ln D$ , the constants n and A can be obtained from the slope, and from the value of the curve at  $D = 1$ , respectively. Hence  $n = -8/3$ , and  $A = 4.1 \times 10^{-8} \Omega \text{ cm K}^{-1}$ . Thus, a good approximation for the resistivity of  $\text{LaB}_6$  for different material densities at various temperatures can be written as:

$$\rho(\Omega \text{ cm}) = 4.1 \times 10^{-8} (\Omega \text{ cm K}^{-1}) \cdot D^{-8/3} \cdot T(\text{K}) \quad (2)$$

Experiments are now planned to determine other  $\text{LaB}_6$  parameters such as coefficient of thermal expansion and emission density as a function of material density and temperature.

This work is supported by Lawrence Livermore National Laboratory under contract No. W-7405-ENG-48, and Lawrence Berkeley Laboratory under contract No. DE-AC03-76SF00098.

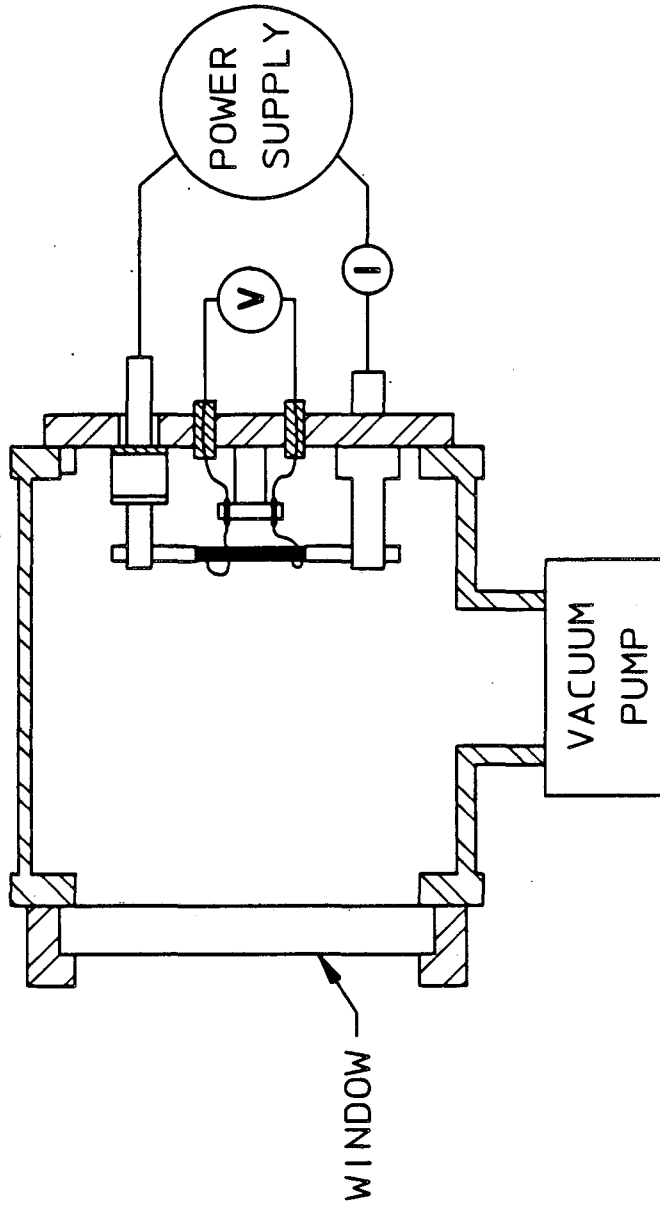


## References

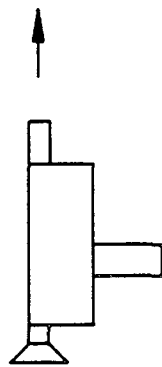
1. J. M. Lafferty, J. Appl. Phys. 22, 299 (1951).
2. H. Ahmed and A. N. Broers, J. Appl. Phys. 43, 2185 (1972).
3. W. H. Kohl, Handbook of Materials and Techniques for Vacuum Devices, (Reinhold, New York, 1967).
4. K. N. Leung, P. A. Pincosy, and K. W. Ehlers, Rev. Sci. Instrum. 55, 1064 (1984).
5. P. A. Pincosy, K. N. Leung, Rev. Sci. Instrum. 56, 655 (1985).
6. K. N. Leung, D. Moussa, and S. B. Wilde, Rev. Sci. Instrum. 57, 1274 (1986).
7. K. Gordon, D. Kippenhan, K. N. Leung, D. Moore, D. Moussa, S. B. Wilde and M. D. Williams, Bull. Am. Phys. Soc. 31, 1507 (1986).
8. J. M. Lafferty, Phys. Rev. 79, 1012 (1950).
9. G. V. Samsonov, High-Temperature Compounds of Rare Earth Metals with Nonmetals (Consultants Bureau, New York, 1965).
10. LaB<sub>6</sub> material obtained from Cerac, Inc. Milwaukee, WI 53201.

### Figure Caption

- Fig. 1 Apparatus for measuring the resistivity of  $\text{LaB}_6$  samples.
- Fig. 2. Resistivity of  $\text{LaB}_6$  as a function of temperature, for material density fractions ranging from .60 to .95.
- Fig. 3 Resistivity coefficient of  $\text{LaB}_6$  as a function of density fraction.

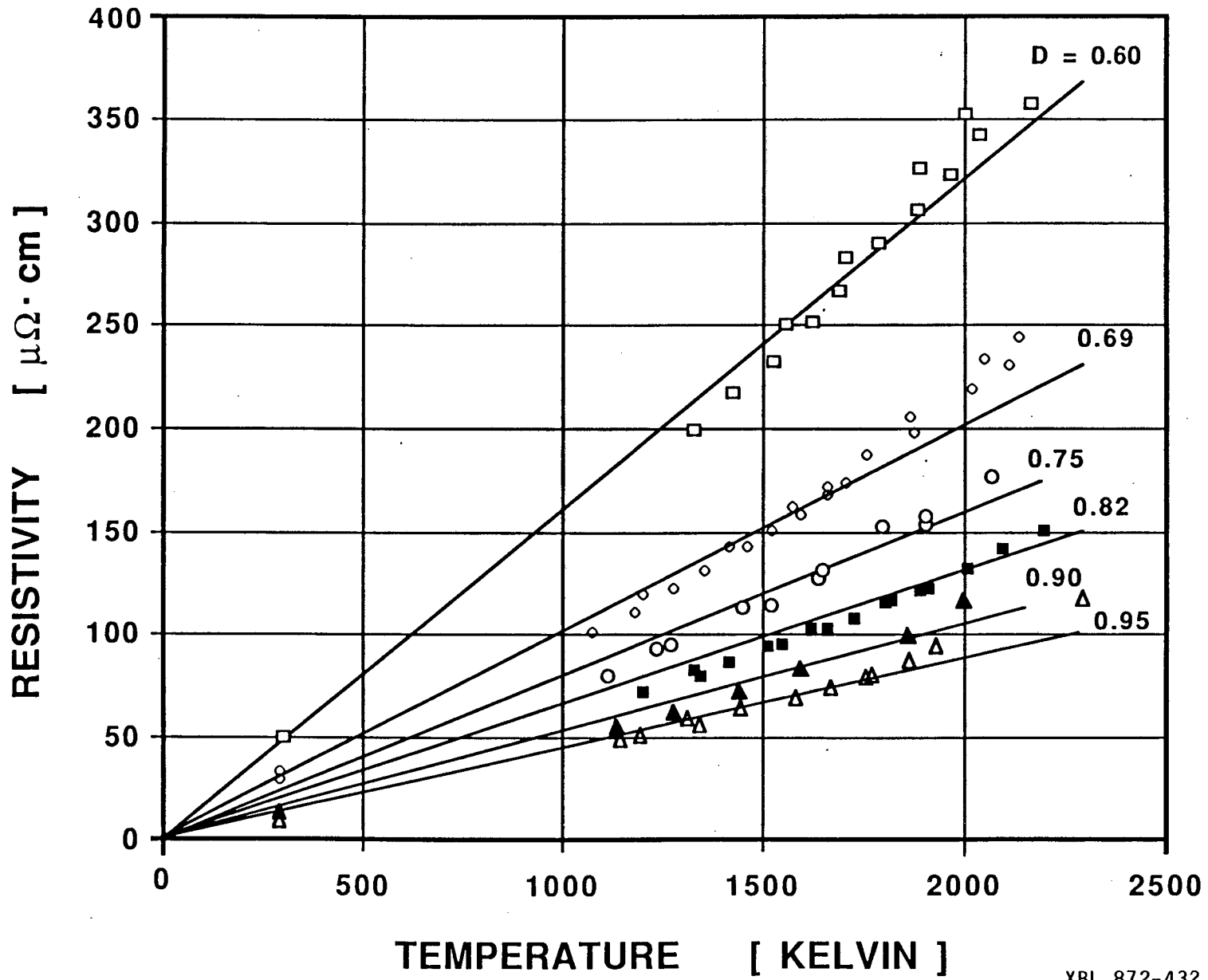


XBL 872-430



OPTICAL  
PYROMETER  
 $\lambda = 0.65$  MICRON

Fig. 1

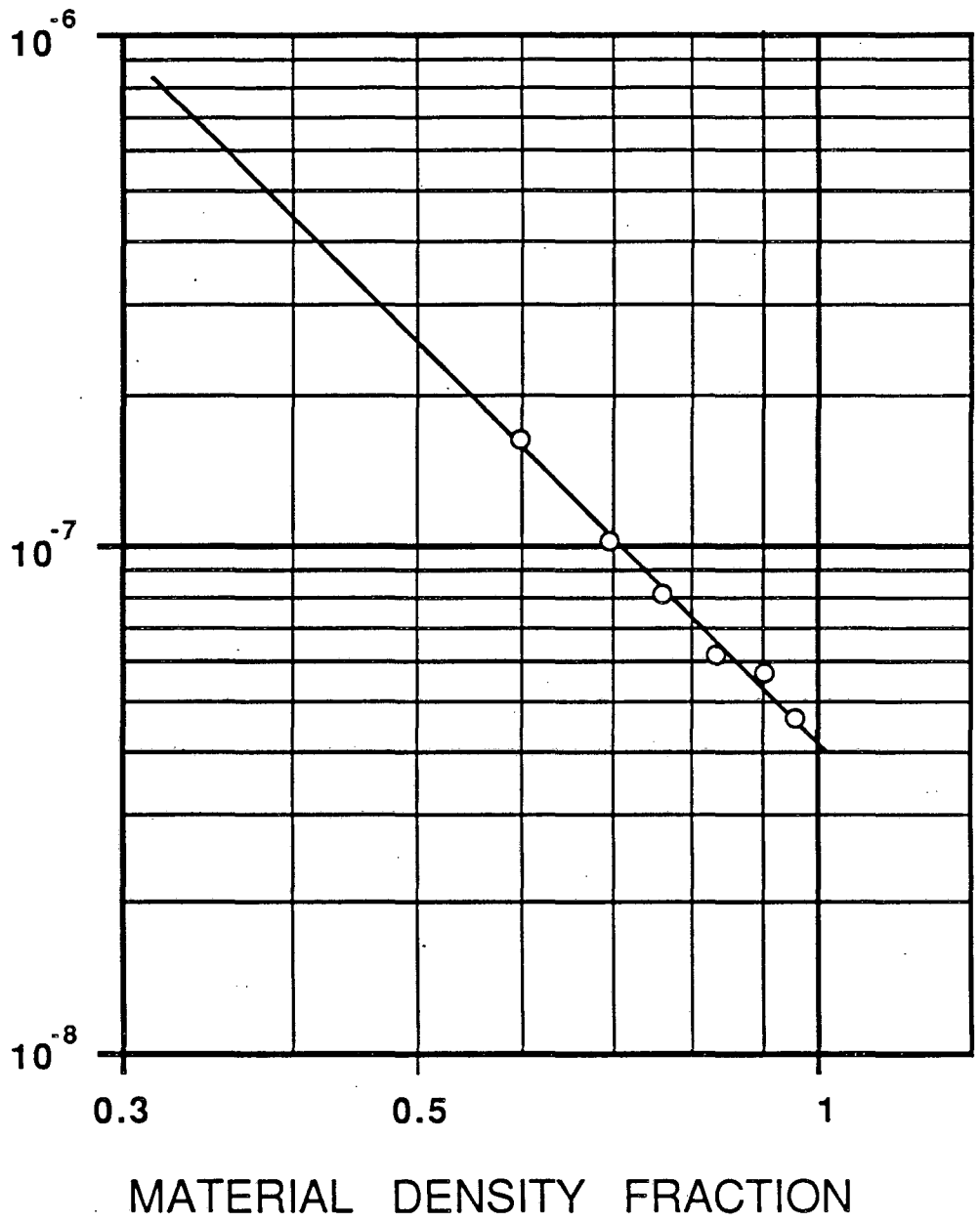


XBL 872-432

Fig. 2

COEFFICIENT OF RESISTIVITY

[ OHMS - CM - K<sup>-1</sup> ]



XBL 872-431

Fig. 3

This report was done with support from the Department of Energy. Any conclusions or opinions expressed in this report represent solely those of the author(s) and not necessarily those of The Regents of the University of California, the Lawrence Berkeley Laboratory or the Department of Energy.

Reference to a company or product name does not imply approval or recommendation of the product by the University of California or the U.S. Department of Energy to the exclusion of others that may be suitable.

*LAWRENCE BERKELEY LABORATORY  
TECHNICAL INFORMATION DEPARTMENT  
UNIVERSITY OF CALIFORNIA  
BERKELEY, CALIFORNIA 94720*