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MONTHLY PROGRESS REPORT FOR OCTOBER - THE PARTITIONING OF MAJOR, MINOR AND TRACE ELEMENTS DURING SIMULATED IN SITU OIL SHALE RETORTING

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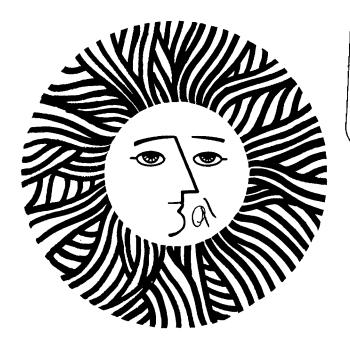
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November 9, 1979

TO: Art Hartstein

FROM: Phyllis Fox and Richard Fish

RE: Monthly Progress Report for October

The Partitioning of Major, Minor and Trace Elements

during Simulated In Situ Oil Shale Retorting

LBID-135

Navy Core Project

Characterization studies of two cores from the Naval Oil Shale Reserve No. 1 are being carried out in a joint program with LETC. These cores have been composited (280 samples) vertically based on the retorting technology likely to be used in each zone, pulverized and characterized chemically. X-ray diffraction analyses, Fischer Assays and major, minor and trace element analyses have been completed on both cores. During the past month, these data were reduced, quality control checks performed and work continued to develop statistical analysis packages and plotting routines to analyze the resulting data.

Mass Balance Studies

Elemental mass balance studies of 23 runs of the LETC controlledstate retort and LLL's 125-kg and 6000-kg retort for 52 major, minor
and trace elements have been completed and work is in progress to
analyze and interpret the resulting data. Experiments have been initiated
to determine the origin of As, Se and other trace elements in in-situ
retort water. It is hypothesized that these elements are incorporated
into the retort water by solubilization from raw shale and shale oil.

Weighted quantities of raw oil shale and of shale oil are being mixed with fixed volumes of $\mathrm{NH_4HCO_3}$ solution for time periods and at temperatures which simulate in situ retorting conditions. The aqueous phase will be separated from the solids and oil and analyzed for trace elements. Organometallic Speciation Studies

Previous work revealed that a significant fraction of the As, Se and Zn exists in the organic fraction of oil shale by-product waters. Work is in progress to identify the chemical forms of these elements using high performance liquid chromatography (HPLC), gas chromatography (GC), mass spectrometry (MS), Zeeman atomic absorption spectroscopy (ZAA) and other techniques.

Work this month focused on the identification and quantification of arsenic compounds in Heater Treater Water (HTW) from Room 6 of Occidental's Logan Wash site. We believe these compounds are present in the methylene chloride extracts of both the acidic (v2 ppm) and basic (v0.5 ppm) components of HTW samples. The HTW samples were extracted with methylene chloride at pH 2.0 and 9.2 and eluted on a florisil column with hexane, hexane-methylene chloride, methylene chloride and methylene chloride-methanol. HPLC was used to identify fractions with products, and these were then analyzed by capillary gas chromatography in combination with a quadrapole mass spectrometer in the electron impact mode (GC-EIMS). The results of these analyses for the pH 9.2 extracts are summarized in Tables 1 and 2 and Figures 1 and 2. The hexane eluates provided the long chain alkenes and alkanes and the hexane-methylene chloride (50/50) eluates provided the heterocyclic aromatic bases. Carboxylic acids have been identified in the pH 2.0 extract and GC-EIMS analysis is in progress.

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Work was initiated on another separation technique, the combination of HPLC with ZAA. Strong anion and cation exchange chromatography columns are being used to separate known arsenates, arsenites, alkylarsonic, arylarsonic and arsenic acids. The As content of these eluates will be determined by ZAA. If this technique is successful, it will be extended to Zn, Se and other toxic trace elements.

Work was also initiated to characterize toxic trace metals in retort waters by chelation followed by analysis using capillary gas chromatography in combination with chemical ionization (CI) and electron impact mass spectrometery (EI-MS). The ligand-1,1,1,2,2,3,3-heptafluoro-7,7-dimethyl-4,6-octanedione (HFOD), was used to remove inorganic and organometallic compounds from the HTW samples by refluxing in 50% ethanol and extraction into hexane. We are presently examining pure compounds of FOD by CI and EI-MS to facilitate identification of some metal species suspected as being present in retort waters.

Table 1. Alkanes and Alkenes found in pH 9.2 Methylene Chloride Extract of Heater Treater Waters by ${\tt GC-EIMS}^a$

Scan Number ^b	Formula	Compound ^C
1359	с ₁₄ н ₂₈	1-Tetradecene
1442	^C 14 ^H 28	Octylcyclohexane
1662	с ₁₆ ^н 32	1-Hexadecene
1933	с ₁₈ н ₃₆	1-Octadecene
2179	с ₁₆ н ₃₄	Hexadecane

- a) Finnigan 4023 equipped with a 30m \times 0.25 mm glass capillary column coated with OV 101. Programmed from 40-250° at 5°/min.
- b) See Figure 1 for scan number (MS) assignment of a particular peak for the GC analysis.
- c) From the file of known compounds and the best fit for the mass spectrum.

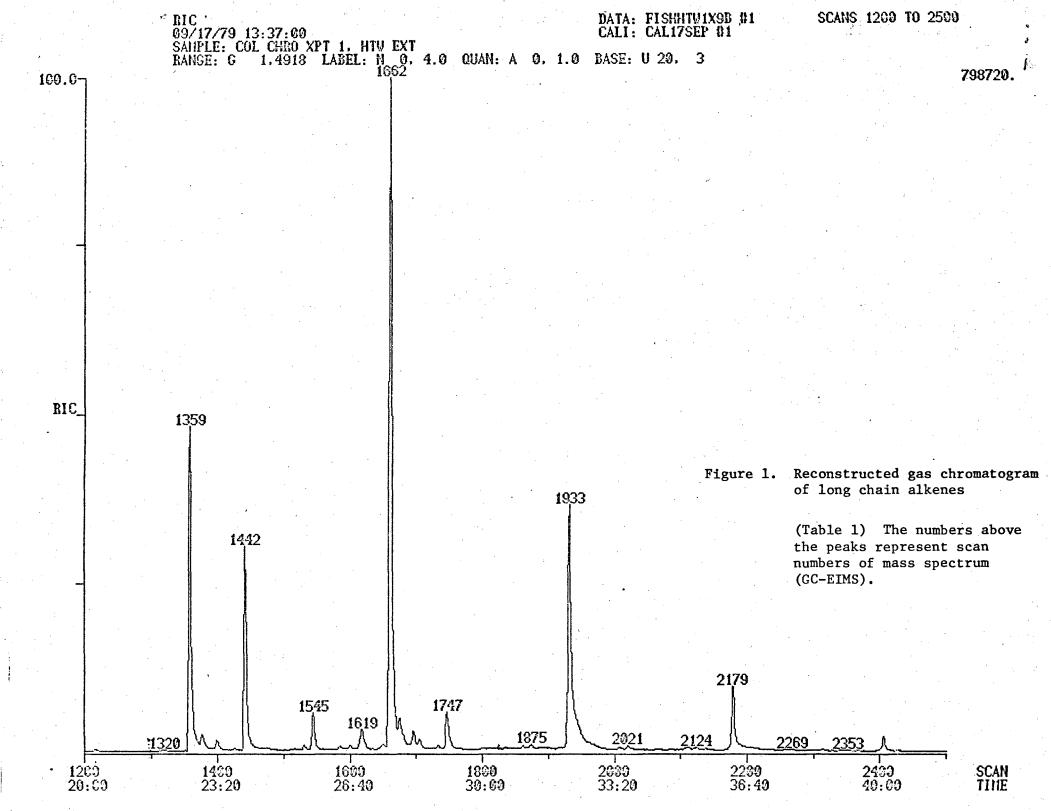
Table 2. Substituted Pyridines and Quinolines found in pH 9.2 Methylene Chloride Extract of Heater Treater Water by $GC\text{-}EIMS^a$

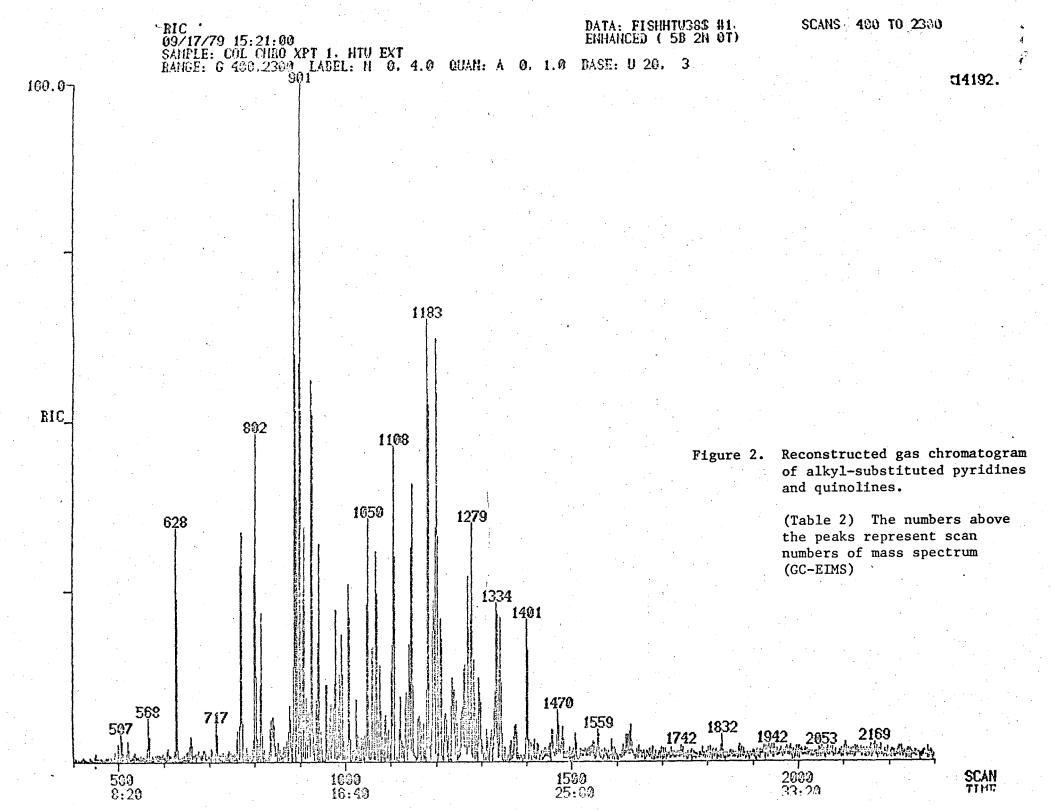
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Scan Number ^b	Formula	Compound ^C
628	C8H11N	2,3,5-trimethylpyridine
772	-	Not assigned accurately
802	c ₁₈ H ₁₃ ON	2-pentan-1-onepyridine
815	-	Not assigned accurately
890	· · · · · · · · · · · · · · · · · · ·	Not assigned accurately
901	C ₈ H ₁₇ N	1-ethy1-2-methy1piperidine
926	с ₉ н ₁₉ N	3-methyl-1,2-diisopropyl- aziridine
1050	<u>-</u>	Not assigned accurately
1108		Not assigned accurately
1183	с ₁₀ н ₉ N	2-methylquinoline
1202	с ₉ н ₁₉ n	Not assigned accurately
1279	с ₇ н ₁₂ о	Not assigned accurately
1401	$c_{11}^{H}_{11}^{N}$	2,4-dimethylquinoline

a) Finnigan 4023 equipped with a 30m x 0.25 mm glass capillary column coated with OV101. Programmed from 40-250° at 5°/min.

b) See Figure 2 for scan number (MS) assignment of a particular peak for the GC analysis.

c) From data base of known compounds.





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