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A Growth-Based, High-Throughput Selection Platform Enables Remodeling of 4-Hydroxybenzoate Hydroxylase Active Site

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Abstract

We report an aerobic, growth-based selection platform founded on NADP(H) redox balance restoration in *Escherichia coli*, and we demonstrate its application in the high-throughput evolution of an oxygenase. A single round of selection followed by a facile growth assay enabled *Pseudomonas aeruginosa* 4-hydroxybenzoate hydroxylase (PobA) to efficiently hydroxylate both 4-hydroxybenzoic acid (4-HBA) and 3,4-dihydroxybenzoic acid (3,4-DHBA), two consecutive steps in gallic acid biosynthesis. Structural modeling suggests precise reorganization of active site

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Author Contributions

S.M., D.A., and H.L. designed the experiments. S.M., D.A., L.Z., and A.P.P. performed the experiments and analyzed the results. E.K. performed Rosetta modeling. All authors wrote the manuscript.

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acscatal.0c01892>.

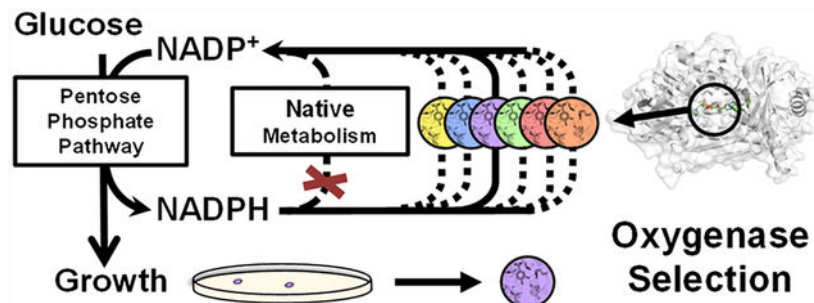
Experimental methods, plasmids and strains used in this study (Table S1); detailed analysis of the sequences of PobA variants obtained from selection (Table S2); growth phenotypes of *E. coli* strains with modified NADPH metabolism (Figure S1); active site volumes of wild type, DA015, and DA016 PobA (Figure S2); and model of PobA wild type with 3,4-DHBA bound (Figure S3) (PDF)

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hydrogen bond network, which is difficult to obtain without deep navigation of combinatorial sequence space. We envision universal application of this selection platform in engineering NADPH-dependent oxidoreductases.

Graphical Abstract



Keywords

4-hydroxybenzoate hydroxylase; NADPH-dependent monooxygenase; 3,4-dihydroxybenzoic acid; redox balance; directed evolution

INTRODUCTION

Oxygenases direct the strong oxidative power in molecular oxygen to activate inert C–H bonds with high regiospecificity and stereospecificity, which is difficult in traditional chemical synthesis.¹ This makes them valuable catalysts for the production of industrially important and medicinal molecules such as ϵ -caprolactone,² omeprazole,³ and artemisinin.⁴ However, engineering oxygenases is challenging, because of their highly complex catalytic mechanisms, which can obscure structure–function relationships.^{5,6}

Recently, a “click” chemistry-based high-throughput mass spectrometric assay was established to facilitate P450 enzyme discovery.⁷ However, throughput remained insufficient (~1000 variants screened for a protein library with a theoretical library size of 400) for libraries routinely generated in directed evolution (10^4 – 10^6 , a theoretical library size). To improve throughput, techniques using single cell or single enzyme analysis such as Fluorescence Activated Cell Sorting (FACS)⁸ and microfluidic assays⁹ have been developed. However, these methods are often limited by a need for colorimetric reactions,¹⁰ substrate labeling,⁷ or indirect activity readouts such as GFP fluorescence.¹¹ In addition, these platforms often require extensive expertise and costly equipment.

We sought to address these limitations by developing a facile and universal selection scheme that employs *E. coli* growth as a simple readout. Specifically, we created an *E. coli* strain with overly high levels of NADPH/NADP⁺, which only grows on glucose in aerobic conditions when the desired oxygenase activity is present to recycle NADPH. This selection scheme is inspired by Nature: previously, natural evolution was observed under high NADPH stress, which enabled *E. coli*'s respiratory chain to oxidize NADPH using oxygen.^{12,13} Similar growth-based selection platforms based on redox balance principles have been

developed previously.^{14–16} However, the metabolic pathway designs in these selection strains limited their functionality to anaerobic conditions. Therefore, they are not compatible with engineering oxygenases, which require oxygen for catalysis.

We applied this selection platform for directed evolution of an NADPH-dependent monooxygenase *Pseudomonas aeruginosa* 4-hydroxybenzoate hydroxylase (PobA) to engineer high activity for a non-native substrate, 3,4-dihydroxybenzoic acid (3,4-DHBA),^{6,17} which is the precursor of a natural product antioxidant, gallic acid (GA). Through a single round of selection of $\sim 3 \times 10^5$ variants from a four-residue, site-saturated mutagenesis library (theoretical library size of $20^4 = 1.6 \times 10^5$), we obtained variants with roughly 8-fold improved apparent catalytic efficiency (k_{cat}/K_M) for 3,4-DHBA, compared to the wild type.⁶ Importantly, a subsequent screen using the same selection platform allowed rapid identification of variants which also retained $\sim 50\%$ catalytic efficiency with the native substrate 4-hydroxybenzoic acid (4-HBA), compared to the wild type. While the two consecutive hydroxylations of 4-HBA and 3,4-DHBA are required for GA biosynthesis, enzyme catalysts with dual activity have been difficult to obtain through rational design.^{6,17} Notably, Rosetta modeling suggests that the selected PobA variants form intricate hydrogen bonding networks at the active site to recognize 3,4-DHBA and coordinate catalysis, which is difficult to recapitulate in engineered enzymes without extensive sampling of sequence space.

RESULTS AND DISCUSSION

Development and Validation of the Selection Platform.

The selection process relies on an engineered *E. coli* strain (MX203) with a growth deficiency linked to NADPH/NADP⁺ imbalance. The perturbed redox state results from deletion of central metabolism genes, glucose-6-phosphate isomerase *pgi* and phosphogluconate dehydratase *edd*; a critical rebalancing tool, soluble pyridine nucleotide transhydrogenase *udhA*; and a significant sink for reduced nicotinamide cofactors, NAD(P)H:quinone oxidoreductase *qor* (Figure 1A).¹² These disruptions cause MX203 to exhibit poor growth in glucose minimal media (Figure S1 in the Supporting Information), and we sought to restore growth through heterologous expression of NADPH-consuming enzymes. We employed water-producing oxidases, *Lb* NOX and *TP*NOX¹⁸ (on plasmids pLS101 and pLS102, respectively) (Figure 1B), derived from *Lactobacillus brevis*. When introduced into selection strain MX203, the NADPH-specific *TP*NOX restored growth, while the NADH-specific *Lb* NOX did not (Figure 1C). Desirable growth behavior demonstrated in both liquid and solid media provides flexibility in selection.

Construction and Selection of PobA Library.

Gallic acid has broad application in food and pharmaceutical industries¹⁹ and as the biosynthetic precursor of a platform chemical, pyrogallol.²⁰ Biosynthesis of gallic acid from simple carbon sources is hindered by the absence of a natural enzyme that can catalyze two consecutive hydroxylation steps, namely, from 4-hydroxybenzoic acid (4-HBA) to 3,4-DHBA, and subsequently to gallic acid (Figure 2A). The NADPH-dependent 4-HBA hydroxylase from *P. aeruginosa* (*Pa* PobA) natively catalyzes the first step, and previous

work engineered additional activity for the second step.^{6,17} While promising variants obtained from rational design, namely, Y385F, Y385F/T294A, and Y385F/L199 V, enabled 3,4-DHBA hydroxylation,^{6,17,20} they also had severely decreased native activity for the first hydroxylation step.^{6,17} We hypothesized that achieving high activity with both 4-HBA and 3,4-DHBA would require the active site to flexibly accommodate the 3-hydroxyl of 3,4-DHBA away from the C4a of flavin adenine dinucleotide (FAD) (Figure 2B). We inferred this mode of substrate binding would be more catalytically competent because oxygen transfer must occur between the flavin-C4a-hydroperoxide intermediate and a position on the aromatic ring with excess electron density, C5.²¹ Guided by this hypothesis, we targeted three residues within 4 Å of the 3-hydroxyl for site-saturated mutagenesis (V47, L199, Y385 in 2B). Position T294 was selected for its proximity to the 4-hydroxyl of 4-HBA and potential role in the hydrogen bonding network of the active site.¹⁷

We simultaneously randomized residues 47, 199, 294, and 385 using NNK degenerate codons (Figure 2B). The resulting DNA library (pDA008) was transformed to selection strain MX203, yielding $\sim 3 \times 10^5$ independent transformants (see the Methods section in the Supporting Information), which was sufficient to cover 1.8-fold of the theoretical library size of $20^4 = 1.6 \times 10^5$. Selection was performed on agar plates with 2 g/L D-glucose in M9 minimal medium and 1 g/L 3,4-DHBA, at 30 °C for 60 h. MX203 cells harboring wild-type *Pa* PobA served as a negative control. After incubation, ~ 600 colonies had formed on selection plates.

Characterization of PobA Variants Obtained from Selection.

We picked eight colonies with robust growth and extracted the mutant *Pa* PobA plasmids. Sequencing the eight candidates revealed six unique residue combinations (Table S2). Repetitive hits found within eight samples suggests that only a small fraction of the variants in the library were able to alleviate redox imbalance. Discussion on observed consensus is included in the Supporting Information.

We next examined growth restoration by retransforming the plasmids into selection strain MX203 and characterizing growth with 3,4-DHBA or 4-HBA as a substrate (Figure 2C). While the negative control, wild-type PobA, showed little to no growth with 3,4-DHBA, all six unique variants exhibited robust growth on this non-native substrate.

Four of the six unique variants were further characterized using purified proteins to inspect the mechanisms underlying their distinct growth restoration patterns on the two substrates. All four variants demonstrated improved 3,4-DHBA activity as measured in vitro by enzyme assay (Figure 3), indicating that growth was coupled to 3,4-DHBA activity. Interestingly, DA014, which struggled to grow on plates containing 4-HBA, exhibited high specific activities for 4-HBA (Figure 3). As noted in previous work, this discrepancy might arise from differences between measured in vitro specific activity and in vivo activity.¹⁴ For example, while the specific activity assays were performed with 1 mM substrates, the in vivo concentrations may be lower.

While DA013 and DA014 exhibited the most significant improvement, compared to wild type for 3,4-DHBA specific activity (Figure 3), we chose to further characterize DA015

(L199R-T294C-Y385M) and DA016 (V47I-L199N-T294A-Y385I) for their robust growth with both 4-HBA and 3,4-DHBA (Figure 2C, Tables 1 and 2). As shown in Table 1, the best variant DA015 showed 7.9-fold improved apparent catalytic efficiency ($k_{\text{cat}}/K_{\text{M}}$), reaching $\sim 0.062 \mu\text{M}^{-1} \text{s}^{-1}$ for 3,4-DHBA, compared to the wild type ($\sim 0.0078 \mu\text{M}^{-1} \text{s}^{-1}$, as reported by Moriwaki et al.⁶). Notably, the 4-HBA hydroxylating activity of DA015 reached $\sim 50\%$ of that of the wild-type enzyme, which is ~ 4 -fold higher, compared to previously engineered variants, Y385F/T294A¹⁷ and Y385F/L199 V.⁶ DA016 has a high apparent k_{cat} for 3,4-DHBA and 4-HBA ($\sim 4 \text{s}^{-1}$ for both), which is over 50% of wild-type Poba's apparent k_{cat} value ($\sim 7.6 \text{s}^{-1}$) toward the native substrate 4-HBA. This high turnover rate may be instrumental for in vitro catalysis.

Because the variants are selected under the physiological concentration of NADPH, we hypothesize that the apparent catalytic efficiency ($k_{\text{cat}}/K_{\text{M}}$) of NADPH for DA015 and DA016 would be comparable to that of the wild-type enzyme, which has been shown to function in *E. coli* for gallic acid production in vivo. This is indeed the case, as shown in Table 2. Notably, both variants have lower apparent K_{M} (ranging from $47 \mu\text{M}$ to $120 \mu\text{M}$) for NADPH when using either 4-HBA or 3,4-DHBA as the substrate, compared to that of the wild-type enzyme using the native substrate 4-HBA ($\sim 250 \mu\text{M}$). These results highlight the potential of the growth-based selection to directly discover enzyme variants that are compatible with in vivo biotransformation.

Mechanistic Study of the Poba Variants by Rosetta Modeling.

Molecular modeling has been established as an important tool in understanding the effect of mutations on catalytic function.^{22,23} We modeled the 3,4-DHBA binding pose with wild-type, DA015, and DA016 to identify structural features contributing to enhanced activity over the wild-type and previously reported variants. Wild-type Poba crystal structure with 4-HBA bound (PDB: 1IUW) indicates 4-HBA binding mediated by a densely packed hydrogen bond network (Figure 4A). Docking of 3,4-DHBA into wild-type suggests that the ligand is unable to bind in a catalytically active pose with the reactive carbon facing the FAD, because of the lack of favorable polar contacts in the interior of the pocket for the 3-hydroxyl to interact with (see Figure S3 in the Supporting Information). Previously reported Poba variants contained rationally designed mutations to disrupt hydrogen-bond networks, aiming to increase backbone flexibility and substrate promiscuity;^{17,24} additional mechanistic studies determined that, for 3,4-DHBA activity, individual L199 and Y385 replacement with hydrophilic residues reduces the relative activity and suggested that the formation of additional hydrogen bonds with Y201, Y385, or the substrate would be detrimental for the reaction.⁶ In contrast, variants obtained in this study are suggested to feature a remodeled hydrogen-bond network tailored for 3,4-DHBA (see Figures 4B and 4C).

Specifically, in the DA015 (L199R-T294C-Y385M) model, the L199R guanidinium extends to form two novel contacts: one to the Y201 hydroxyl and another to the 3,4-DHBA 3-hydroxyl. T294C eliminates a hydrogen bond to T347, allowing another new hydrogen bond to form between the backbone carbonyl to the 3,4-DHBA 4-hydroxyl (see Figure 4B). In the DA016 (V47I-L199N-T294A-Y385I) model, the L199N amide acts as a bridge stabilizing

the ligand 3-hydroxyl and the S212 hydroxyl associated with the ligand carboxylic acid; V47I packs against and reinforces L199N; T294A, similarly to T294C, breaks a hydrogen bond to T347 and allows expanded loop mobility for improved backbone hydrogen bonding to the ligand 4-hydroxyl (Figure 4C). Comparison to wild-type structure shows a binding pocket volume increase from 55 Å³ to 69 Å³ in DA015 and 72 Å³ in DA016 (see Figure S2). The additional space accommodates the extra hydroxyl in 3,4-DHBA binding.

The discovered mutations are predicted to restructure the central hydrogen bonding array for a more-productive 3,4-DHBA binding orientation. Rational design to achieve the same effect is challenging, because of the multiple, simultaneous substitutions required to reach this functionality. Designing hydrogen-bonding networks is especially difficult, because of their high cooperativity and sensitivity to small perturbations in length and angle between hydrogen-bonding partners. For instance, singular L199 mutation to a polar residue contacting the ligand 3-hydroxyl would likely be ineffective without compensatory substitutions at V47 and Y385 relieving steric clash and reinforcing optimal hydrophobic packing. This highlights the effectiveness of our high-throughput selection approach in comprehensively exploring the sequence space and discovering highly synergistic mutations.

CONCLUSION

In summary, we established a high throughput screening platform with potential applications for directed evolution of various NADPH-dependent oxygenases. This aerobic in vivo system demonstrated strict cofactor dependence and simple growth-based selection. We successfully engineered PcbA for enhanced activity toward the non-native substrate 3,4-DHBA. In one round of selection, we identified a variant, DA015, with greatly improved activity for both 4-HBA and 3,4-DHBA, compared to variants previously designed with restricted exploration of the available sequence space. Rosetta modeling suggests that enhanced activity is attributed to concerted changes in binding pocket volume, shape complementarity, and, more importantly, formation of a fully connected hydrogen-bond network in the active site. Future work is needed to overcome some of the limitations of this selection platform. For example, the target substrates may have cellular toxicity or low solubility. Therefore, a two-phase culturing system, which combines an organic solvent phase with the culture medium, may be needed.²⁵ Because NADPH provides reducing power for a variety of oxygen-dependent industrially important enzyme classes, such as Baeyer–Villiger monooxygenases and cytochrome P450s, we envision applications of this platform for engineering a broad range of catalysts.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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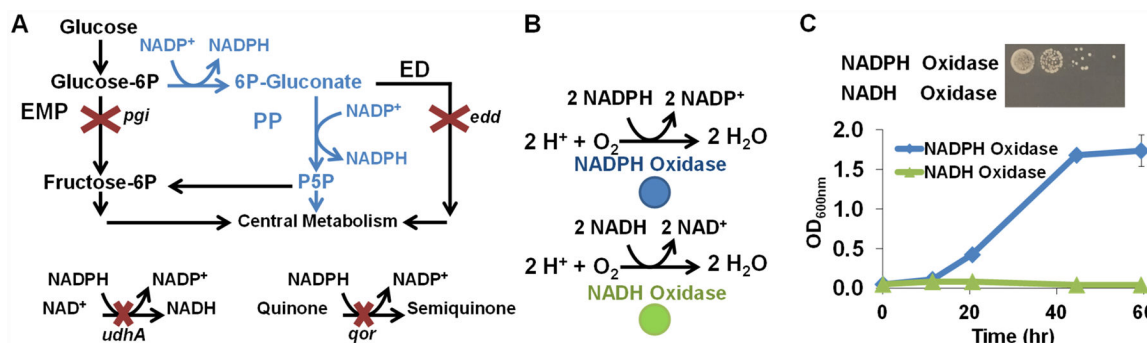


Figure 1. Development and validation of the selection strain with high NADPH accumulation. (A) Central metabolism redirection through pentose phosphate (PP, blue) pathway and disruption of Embden-Meyerhof-Parnas (EMP) and Entner Doudoroff (ED) pathways increased NADP⁺ reduction. Deletion of NADPH reoxidation transhydrogenase *udhA* and quinone oxidoreductase *qor* reduce NADPH sinks. (B) Complementary oxidases use reduced cofactors, NADPH (blue) or NADH (green), and oxygen to generate water. (C) In M9 minimal glucose media, growth restoration was achieved by heterologous expression of the NADPH-specific oxidase but not the NADH-specific oxidase in both solid and liquid media.

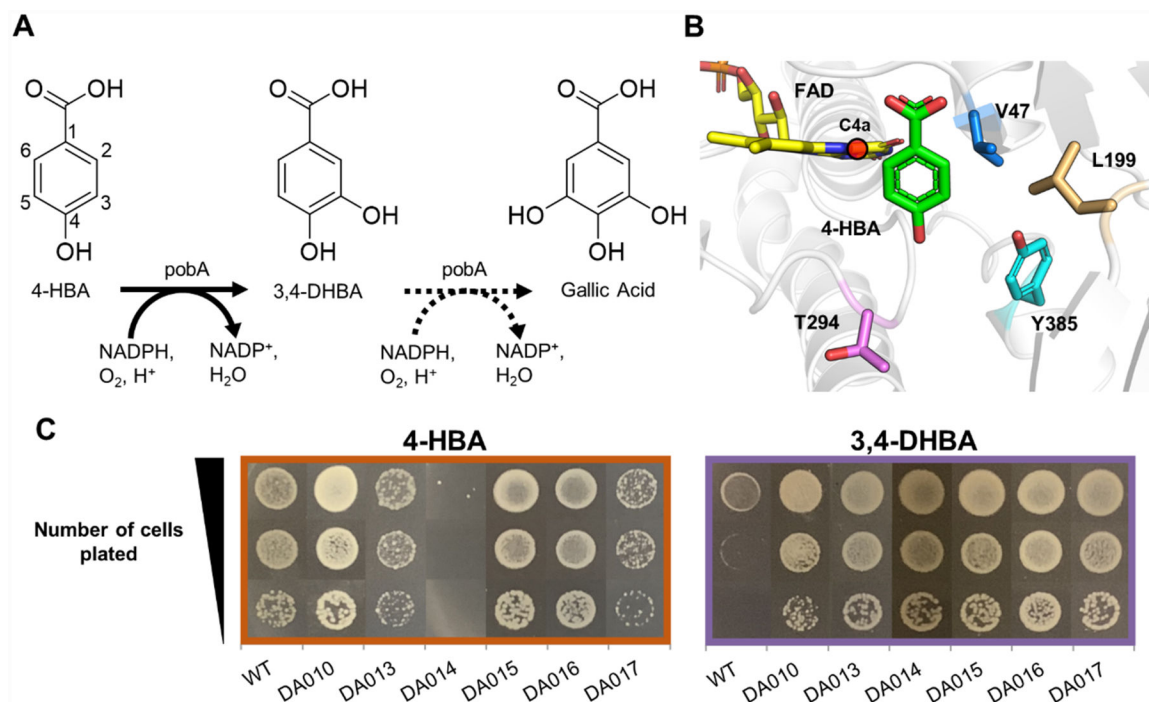


Figure 2.

Construction and selection of the PobA library for 3,4-DHBA utilization. (A) Biosynthetic pathway of gallic acid involves two sequential hydroxylation steps. (B) Wild-type *Pa* PobA crystal structure with 4-HBA bound (PDB: 1IUW). Residues highlighted were targeted for site-saturated mutagenesis. The reactive FAD-C4a is highlighted by red sphere. (C) Growth restoration phenotype for wild-type PobA and variants identified in selection, with 4-HBA or 3,4-DHBA supplementation.

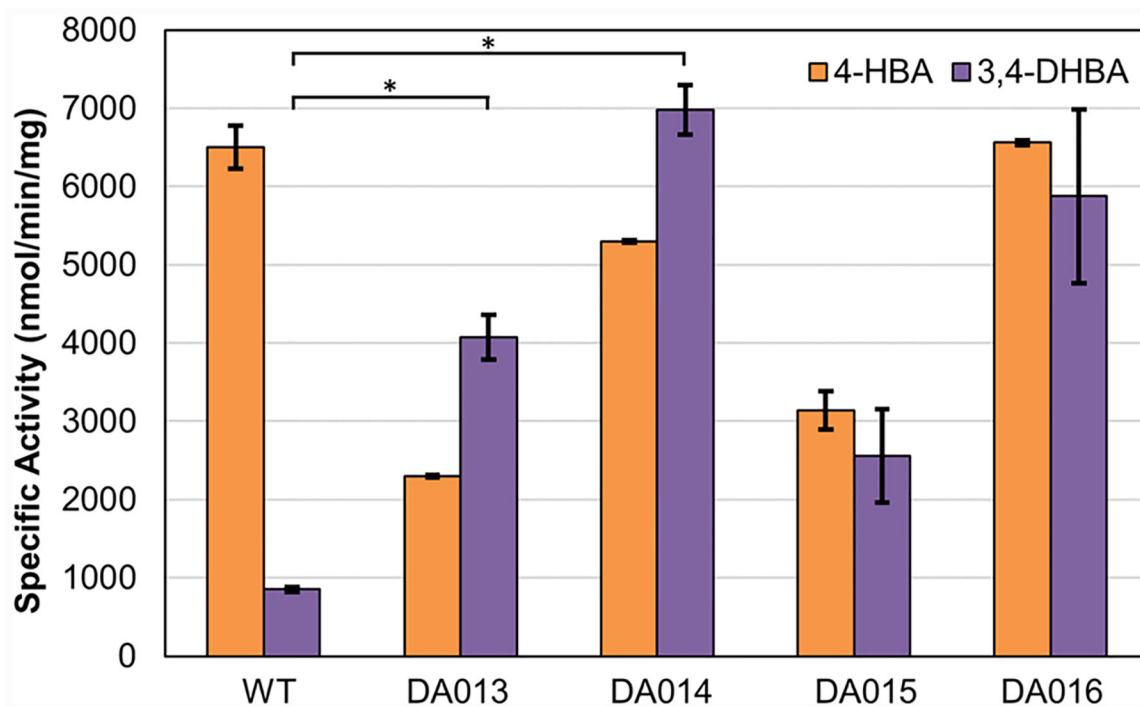


Figure 3.

In vitro specific activity of four PobA variants identified in growth-based selection. All variants exhibited improved activity toward 3,4-DHBA. Detailed assay conditions are described in the Methods section in the Supporting Information. Statistical significance indicated by an asterisk was determined by two-tailed *t*-tests ($P < 0.05$).

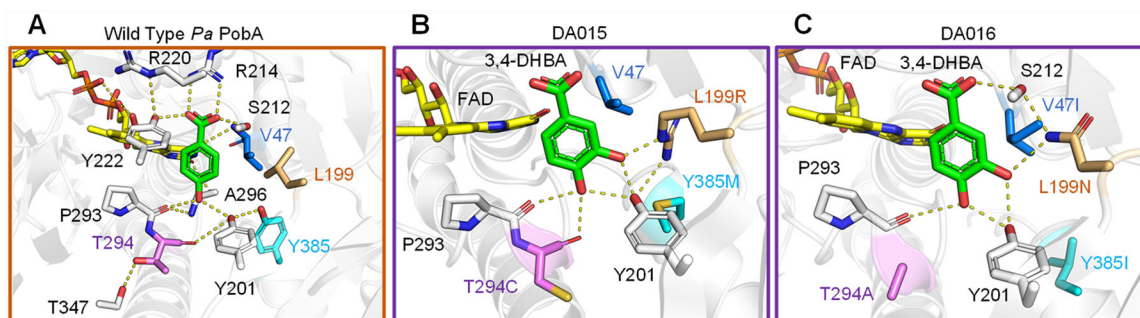


Figure 4.

4-HBA and 3,4-DHBA binding poses. (A) In wild-type PobA (PDB: 1IUW), the productive binding mode is described by the hydrogen-bonding networks stabilizing 4-HBA and positioning the 3-carbon toward FAD for hydroxylation. (B) In the DA015 model, L199R supports Y201 and forms a new contact to the ligand 3-hydroxyl. Y385M makes space, no substitution occurs at V47, which maintains close hydrophobic packing against L199R, and T294C loosens the helix for increased flexibility and improved backbone hydrogen bonding to the 4-hydroxyl. (C) In the DA016 model, L199N forms interactions stabilizing S212 and to the ligand 3-hydroxyl. Y385I creates space, V47I braces L199N to minimize side-chain mobility, and T294A allows P293 to move closer to 3,4-DHBA. Importantly, the mutations orient 3,4-DHBA such that the 5-carbon is optimally exposed to FAD for hydroxylation in both variants. Detailed methods of Rosetta modeling are included in the Supporting Information.

Table 1.

Apparent Kinetic Parameters for Substrates

variant	4-HBA				3,4-DHBA			
	K_M^a (μM)	k_{cat}^a (s^{-1})	k_{cat}/K_M^a ($\mu\text{M}^{-1}\text{s}^{-1}$)	K_M^a (μM)	k_{cat}^a (s^{-1})	k_{cat}/K_M^a ($\mu\text{M}^{-1}\text{s}^{-1}$)	k_{cat}/K_M^a ($\mu\text{M}^{-1}\text{s}^{-1}$)	
this study								
wild-type	48 ± 7	7.6 ± 0.1	0.16 ± 0.03	–	–	–	–	
DA015 ^b	24 ± 10	1.9 ± 0.2	0.079 ± 0.04	48 ± 10	3.0 ± 0.2	0.062 ± 0.01	0.062 ± 0.01	
DA016 ^c	120 ± 10	4.0 ± 0.1	0.034 ± 0.001	120 ± 10	4.1 ± 0.2	0.033 ± 0.004	0.033 ± 0.004	
Chen et al. ¹⁷	Y385F/T294A	48 ± 3	0.90 ± 0.03	160 ± 30	1.7 ± 0.2	0.012 ± 0.003	0.012 ± 0.003	
Moriwaki et al. ⁶	Y385F/L199V	22 ± 5	0.45 ± 0.02	52 ± 16	4.4 ± 0.4	0.085 ± 0.02	0.085 ± 0.02	
wild-type		39 ± 6	8.9 ± 0.5	42 ± 17	0.33 ± 0.04	0.0078 ± 0.002	0.0078 ± 0.002	

^a Apparent kinetic parameters shown for constant NADPH concentration of 1 mM. Data by Chen et al.¹⁷ and Moriwaki et al.⁶ is presented as it appeared in the references.

^b DA015, L199R-T294C-Y385M.

^c DA016, V47I-L199N-T294A-Y385I.

Table 2.

Apparent Kinetic Parameters for NADPH

variant	substrate	K_M^a (μM)	k_{cat}^a (s^{-1})	k_{cat}/K_M^a ($\mu\text{M}^{-1} \text{s}^{-1}$)
wild-type	4-HBA	250 \pm 40	7.8 \pm 0.8	0.031 \pm 0.006
DA015 ^b	4-HBA	47 \pm 7	2.1 \pm 0.2	0.044 \pm 0.008
DA016 ^c	4-HBA	140 \pm 40	4.4 \pm 0.8	0.032 \pm 0.01
wild-type	3,4-DHBA	–	–	–
DA015 ^b	3,4-DHBA	76 \pm 10	2.5 \pm 0.2	0.033 \pm 0.006
DA016 ^c	3,4-DHBA	120 \pm 4	4.1 \pm 0.1	0.033 \pm 0.001

^a Apparent kinetic parameters shown for constant substrate concentration of 0.5 mM.

^b DA015, L199R-T294C-Y385M.

^c DA016, V47I-L199N-T294A-Y385I.