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Winter and summer nitrous oxide and nitrogen oxides fluxes from a seasonally snow-covered subalpine meadow at Niwot Ridge, Colorado

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Abstract The soil emission rates (fluxes) of nitrous oxide (N_2O) and nitrogen oxides ($\text{NO} + \text{NO}_2 = \text{NO}_x$) through a seasonal snowpack were determined by a flux gradient method from near-continuous 2-year measurements using an automated system for sampling interstitial air at various heights within the snowpack from a subalpine site at Niwot Ridge,

Colorado. The winter seasonal-averaged N_2O fluxes of $0.047\text{--}0.069 \text{ nmol m}^{-2} \text{ s}^{-1}$ were ~ 15 times higher than observed NO_x fluxes of $0.0030\text{--}0.0067 \text{ nmol m}^{-2} \text{ s}^{-1}$. During spring N_2O emissions first peaked and then dropped sharply as the soil water content increased from the release of snowpack meltwater, while other gases, including NO_x and CO_2 did not show this behavior. To compare and contrast the winter fluxes with snow-free conditions, N_2O fluxes were also measured at the same site in the summers of 2006 and 2007 using a closed soil chamber method. Summer N_2O fluxes followed a decreasing trend during the dry-out period after snowmelt, interrupted by higher values related to precipitation events. These peaks were up to 2–3 times higher than the background summer levels. The integrated N_2O -N loss over the summer period was calculated to be $1.1\text{--}2.4 \text{ kg N ha}^{-1}$, compared to $\sim 0.24\text{--}0.34 \text{ kg N ha}^{-1}$ for the winter season. These wintertime N_2O fluxes from subniveal soil are generally higher than the few previously published data. These results are of the same order of magnitude as data from more productive ecosystems such as fertilized grasslands and high-N-cycling forests, most likely because of a combination of the relatively well-developed soils and the fact that subnivean biogeochemical processes are promoted by the deep, insulating snowpack. Hence, microbially mediated oxidized nitrogen emissions occurring during the winter can be a significant part of the N-cycle in seasonally snow-covered subalpine ecosystems.

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Introduction

Nitrous oxide (N_2O) and nitrogen oxides ($NO + N_2O = NO_x$) are oxidized nitrogen (N) compounds produced by soils from microbially mediated processes. In recent decades the atmospheric concentration of N_2O has increased due to anthropogenic activity, becoming a concern with regard to global warming (IPCC 2001). N_2O emissions from soils are a significant part of the global N cycle, contributing 58% of the total global N_2O -N emission rate on a yearly basis; with more than half of the N_2O emitted from non-agricultural ecosystems (IPCC 2001). In contrast, the main sources of NO_x are combustion processes, while soils are responsible for between 10 and 23% of the total NO_x emissions, with slightly less than half of this fraction emitted by agricultural, fertilized soils (Delmas et al. 1997). NO_x emissions can play a significant role in the tropospheric oxidation chemistry, and affect the formation and increase in tropospheric ozone, and thereby, similarly as N_2O , contributing to the greenhouse gas effect.

N gaseous fluxes from soils have been well characterized during the growing season (Bowden 1986; Dunfield et al. 1995; Smith et al. 1998). In contrast, over-winter emission rates in seasonally snow-covered catchments have been poorly assessed, although there is evidence that microbial activity and consequent emission of gaseous C and N continue in the relatively warm conditions that may occur in soils under a continuous snowpack (Sommerfeld et al. 1993; Brooks et al. 1996; Schürmann et al. 2002). Most over-winter investigations of emissions of N gases have been conducted in agricultural (van Bochove et al. 1996; Mosier et al. 1997; Maljanen et al. 2007) and organic soils (Alm et al. 1999; Maljanen et al. 2003) or lower-elevation forested areas (Groffman et al. 2006), with relatively few studies conducted in high-elevation mountain ecosystems (Brooks et al. 1996). While substantial work has been done on wintertime oxidized carbon (i.e.,

CO_2) emissions in such ecosystems (Monson et al. 2006a, b; Jones et al. 1999; Liptzin et al. 2009), it is poorly known how these findings relate to oxidized N compounds (Schürmann et al. 2002). Moreover, it is not well established how changes in climate, and changes in the insulating snow cover affect the fluxes of oxidized N compounds in mountainous, non-agricultural soils.

Gaseous soil fluxes have been commonly measured by the chamber technique (e.g., Martikainen et al. 1993; Alm et al. 1999). During winter chamber measurements may cause sampling problems by changing the snow properties and soil insulation inside the enclosure from the experiment. Additionally, chamber measurements are time consuming and provide a discontinuous dataset. A further concern is how deep chamber walls should be placed in a porous media, such as snow, to realize a closed chamber design.

Here we conducted continuous sampling of air from inside a deep snowpack and determined vertical N_2O and NO_x concentration profiles in the snow-covered seasons of 2006 (late fall 2005 to late spring 2006, hereafter referred to as winter 2006) and 2007 (late fall 2006 to late spring 2007, hereafter referred to as winter 2007) in a subalpine meadow soil in the Colorado Rocky Mountains. We then applied Fick's law to calculate diffusional fluxes from these gradient data. These winter measurements are contrasted to the snow-free period (i.e., summers 2006 and 2007) when N_2O fluxes were measured approximately weekly using a closed soil chamber technique. Moreover, concomitant measurements of CO_2 efflux during the same experiments (Liptzin et al. 2009) allow us to evaluate the relationships among these gases.

The resulting data are used to:

- (1) Estimate the N_2O flux during the snow-covered season and compare it to the summer flux, to assess an annual budget;
- (2) Evaluate total oxidized N emissions and test if NO_x winter fluxes are a significant loss of N for this ecosystem;
- (3) Evaluate the seasonal and inter-year variability of N_2O and NO_x fluxes;
- (4) Investigate environmental parameters controlling N_2O and NO_x export from this ecosystem during both winter and summer periods.

Materials and methods

Description of the study site

This research was conducted on Niwot Ridge, Colorado (40°02'52"N; 105°34'15"W, 3,345 m a.s.l.). This site is a subalpine open meadow surrounded by ribbon forest, on a 10°SW slope.

Soil is classified as a mixed Typic Humicryept, sandy-skeletal in texture. Soil pH ranges from 4.6 to 5.0 (Fisk and Schmidt 1995). The organic carbon in the A horizon ranges from 150 to 190 g kg⁻¹ and N ranges from 11 to 22 g kg⁻¹ (Burns 1980; Brooks et al. 1995). The climate is characterized by a long (6–7 months), cold winter and a short, cool growing season. Mean annual temperature is -3.7°C, annual precipitation is 1,050 mm, 80% of which falls as snow (Caine 1995).

Snowpack gradient sampling method

The snowpack flux experiments were performed using a multi-level gradient tower. This tower was installed at the site before the onset of the snow accumulation season and allowed to be covered by the precipitating snow without any disturbance throughout the snow-covered time period. In winter 2006, the snow tower consisted of five paired inlets at 0, 30, 60, 90 and 245 cm above the soil surface. In 2007, three new inlet heights were added resulting in paired inlets located at 0, 10, 30, 60, 90, 120, 150, and 245 cm above the soil surface, with the latter inlet remaining above the maximum snowpack height throughout the winter season and representing ambient atmospheric conditions.

Sampling lines were all perfluoroalkoxy (PFA) Teflon[®], inner diameter of 3.9 mm and outer diameter of 6.4 mm (Parker Hannifin, Cleveland, Ohio, USA), with equal lengths of 18 m. The inlets were fitted with 25-mm Acrodisc[®] hydrophobic polytetrafluoroethylene (PTFE) syringe filters (Pall Life Sciences, Ann Arbor, Michigan, USA) to prevent debris from entering the sampling line. Prior to field installation inlet filters and sampling lines were conditioned for >24 h in a ~1 l flow of 200–300 ppbv of ozone in room air. Sampling was done by drawing air sequentially from each inlet height for 10 min, resulting in an 80-min cycle time. Air delivery from the snow inlets was provided by the

combined flow rates of the pumps in the nitrogen oxides and ozone analyzers (~2.8 L min⁻¹), with the N₂O gas chromatography (GC) system pulling another small fraction (45 mL min⁻¹) from this manifold. Seok et al. (2009) and Bowling et al. (2009) report that active pumping of the air from the snowpack at this rate did not result in any significant artifact from the ventilation induced by the air sampling flow, because collected gas volumes were relatively small and the time lag between subsequent samplings was long enough to re-equilibrate gas levels in the snowpack.

N₂O data

For the 2006 winter, vertical profile data are available from day of year (DOY) 30 to 49, 52 to 82, 85 to 108, 129 to 136, from four inlet heights sampling interstitial air in the snowpack (i.e., 90, 60, 30 and 0 cm), and one inlet sampling atmospheric air (245 cm), summing a total of 86 days sampled and ~1,235 profiles. For the 2007 winter, vertical profile data are available from DOY 29 to 47, 51 to 61, 93 to 101, 112 to 114, 116 to 130, 137 to 149, from seven inlets in the snowpack (i.e. 150, 120, 90, 60, 30, 10, 0 cm), and one inlet sampling atmospheric air (245 cm), summing a total of 68 days sampled and ~1,000 profiles.

NO_x data

For the 2006 winter, a dataset of vertical profile concentrations is available from DOY (2006) -46 to 1, 30 to 49, 52 to 82, 85 to 136, from four different inlets sampling interstitial air in the snowpack (i.e., 90, 60, 30 and 0 cm), and one inlet sampling atmospheric air (245 cm), summing a total of 148 days and approximately 2,500 profiles sampled. In winter 2007, data are available from DOY (2007) -32 to 82, 86 to 97, 99 to 101, 112 to 114, 116 to 130, from seven inlets in the snowpack (i.e., 150, 120, 90, 60, 30, 10, 0 cm), and one inlet sampling atmospheric air (245 cm), summing a total of 161 days and approximately 2,700 profiles sampled.

All instruments were fully automated, remained at the site, and were operated throughout the winters of 2006 and 2007. Site visits occurred typically once a week. Gaps in both the NO_x and N₂O records resulted from occasional technical problems in the data logging or instrument operation.

N₂O measurement method

N₂O was measured by a gas chromatograph (GC) equipped with an electron capture detector (ECD) (Shimadzu GC-8AIE, Shimadzu Scientific Instruments, Columbia, Maryland, USA). Air samples drawn from the snowpack first passed through a scrubber (1.59 cm i.d., 1.90 cm o.d., 15.24 cm length) filled with 20–30 g Ascarite (20–30 mesh) to remove the elevated and highly variable CO₂ in the air sampled from the snowpack tower. The scrubber was refilled with fresh material every 7–14 days. A 9 ml-sample loop, connected to an electrically actuated 2-position, 10-port gas switching valve (Model E60, VICI, Huston, Texas, USA) was used for sample injection onto the GC column, which consisted of two sections of 3.18 mm o.d. and 3.67 m length stainless steel tubing packed with Porapak Q (Supleco, Sigma-Aldrich, St. Louis, Missouri, USA). The plumbing of the system allowed for the timed backflushing of the first column in order to avoid elution of water onto the second column and ECD detector. The carrier gas flow was 45 mL min⁻¹ of a 95% argon, 5% methane mixture (AirGas Intermountain, Boulder, Colorado, USA). The GC oven was operated isothermally at 75°C, and the detector temperature was 330°C. The system was fully automated and computer-controlled. Every night a multipoint calibration was done to determine the ECD response and for tracking instrument drifts. Calibration standards were prepared gravimetrically at 320, 408, 485 ppbv (estimated uncertainty 0.1%). The sample loop pressure was recorded for each sample and calibration run, and determined peak areas were corrected for the ~0.9% variability (1σ) seen in the injection loop pressure during collection of samples from the different inlets and the standards. A first order regression equation was fit through the calibration data and used for quantification of N₂O in samples. As N₂O concentration in the snowpack air samples during the peak of the snowpack height exceeded the calibration range by up to 200%, increasingly larger uncertainties in the N₂O quantification are expected at these higher concentrations. After conclusion of the 2007 experiment we discovered a diurnal cycle/shift in the retention time and signal area of the N₂O peak. This effect was traced to diurnal changes in the carrier gas cylinder temperature and regulator pressure (which were placed outside the underground laboratory,

partially exposed to solar radiation). A series of further standard experiments was conducted to determine an algorithm that allowed applying a peak area correction based on the recorded retention time ($r^2 = 0.98$, $p < 0.001$, $n = 144$, data not shown).

The precision of the GC measurement was estimated as ~2% (~6 ppbv), resulting in a sensitivity for the lowest measurable N₂O gradient of ~10 ppbv. In addition, the sensitivity in flux determination is a function of other parameters that go into the diffusion calculation, and these in turn depend on snowpack physical properties. Based on the range of conditions encountered in both winters, the sensitivity (lowest measurable flux) was estimated to be on the order of 0.012 nmol m⁻² s⁻¹.

NO/NO₂ measurement method

NO_x concentrations were measured using a commercial chemiluminescence NO_x analyzer (Model 42CTL, Thermo Environmental Instruments, Franklin, Massachusetts, USA), with automated, multipoint calibrations performed nightly in a similar manner as for N₂O. The NO_x analyzer sensitivity is given as 50 pptv. Here, in order to estimate the lowest measurable NO_x flux we used a more conservative value of ~1 ppbv for the lowest detectable gradients (from two height measurements, with data acquisition over 10 min). The lowest discernable flux was calculated as for N₂O, and was found to be ~0.001 nmol m⁻² s⁻¹.

Environmental parameters

Air, snow and soil temperatures, atmospheric pressure, precipitation and a set of environmental parameters were recorded during both years of investigation. Volumetric soil moisture was determined for the 2007 winter at an integrated depth of 0–30 cm, using four CS616-L Water Content Reflectometers with 30-cm long probes installed vertically into the soil in a 1-m radius from the snow tower (Campbell Scientific, Logan, Utah, USA), with the average of the four sensors used for all further data applications. Temperatures at each of the gas sampling inlets were measured using type-E thermocouples (Omega Engineering, Inc., Stamford, Connecticut, USA) and were used in Eq. 2. The 0-cm-inlet level measurement was also used as the estimate of the temperature at the

snow–soil interface. Each thermocouple was covered by white shrink tubing to reduce radiation artifacts. Barometric pressure included in Eq. 2 was measured using a CS105 Vaisala PTB101B Barometer (Campbell Scientific, Logan, Utah, USA) at a meteorological tower 10 m away from the snow flux tower.

Winter flux measurements and calculations

Winter fluxes were measured using the vertical N₂O concentration profiles based on Fick's law,

$$J_{\text{N}_2\text{O}} = -D_{\text{N}_2\text{O}}(dC_{\text{N}_2\text{O}}/dz) \quad (1)$$

where $(dC_{\text{N}_2\text{O}}/dz)$ is the concentration gradient ($\text{nmol m}^{-3} \text{m}^{-1}$) and $D_{\text{N}_2\text{O}}$ is the gas diffusivity ($\text{m}^2 \text{s}^{-1}$), calculated as follows:

$$D_{\text{N}_2\text{O}} = \Phi \tau D_{\text{STP}}(P_0/P)(T/T_0)^\alpha \quad (2)$$

where $D_{\text{STP}} = 0.1436 \times 10^{-4} \text{m}^2 \text{s}^{-1}$ (Massman 1998) is the gas diffusivity at standard temperature and pressure, $\alpha = 1.81$ (Massman 1998), τ is the snowpack tortuosity, and Φ is the snowpack porosity. The same procedure was used to calculate NO_x fluxes, assuming a diffusivity D_{STP} of $0.1361 \times 10^{-4} \text{m}^2 \text{s}^{-1}$ and $0.1802 \times 10^{-4} \text{m}^2 \text{s}^{-1}$ for NO₂ and NO, respectively (Massman 1998).

The porosity and the tortuosity of the snowpack were both estimated from the snow density:

$$\Phi = 1 - (\rho_{\text{snow}}/\rho_{\text{ice}}) \quad (3)$$

$$\tau = \Phi^{1/3} \quad (4)$$

where ρ_{snow} is the snow density (kg m^{-3}) and ρ_{ice} is the density of ice (917kg m^{-3}).

Snow density was measured at 10-cm height intervals in the snowpack ~bi-weekly by digging a snowpit about ~30 m away from the snow flux tower at a location with similar vegetation and slope characteristics, following the protocol of Williams et al. (1996). During the 2-year experiment, corrected $D_{\text{N}_2\text{O}}$ values ranged between 0.072×10^{-4} and $0.140 \times 10^{-4} \text{m}^2 \text{s}^{-1}$.

Summer flux measurements and calculations

Summer fluxes were determined using gas measurements from five closed, dark, vented chambers (i.d. 0.37 m i.d., 0.20 m height, 0.110 m² area, 0.022 m³ volume). The chambers were located

approximately 5 m away from the winter snow flux tower, and within 30 cm of each other. At each sampling date the order of measurement was randomly assigned in order to avoid systematic over or under-estimation of fluxes due to the diurnal flux fluctuation. Air in the headspace of each chamber was typically sampled every 10 min over 40–70 min of measurements, using a polypropylene 20 mL syringe. The lid of the chamber was equipped with a narrow tube for pressure equilibration (Welles et al. 2001). Fluxes were calculated by integrating N₂O concentrations over time, using a Michaelis–Menten fitting curve (Welles et al. 2001), following the equation:

$$J_{\text{N}_2\text{O}} = V/A \times \rho \times (dC_{\text{N}_2\text{O}}/dt) \quad (5)$$

where $J_{\text{N}_2\text{O}}$ is the gas flux, V and A are the volume and the surface area of the chamber, respectively, ρ (mol m^{-3}) is the molar concentration of air corrected for ambient pressure and temperature, and $(dC_{\text{N}_2\text{O}}/dt)$ is the concentration change with time ($\text{nmol mol}^{-1} \text{s}^{-1}$), which is obtained from:

$$dC_{\text{N}_2\text{O}}/dt = k \times (C_S - C_0) \quad (6)$$

where C_0 is the ambient N₂O concentration (nmol mol^{-1}), k (s^{-1}). C_S (nmol mol^{-1}) (the estimated concentration of N₂O in soil) was calculated from an exponential fitting curve using the simplified model proposed by Welles et al. (2001). We used this approach to integrate fluxes because it appears more physically correct compared to traditional linear fitting (Welles et al. 2001).

In summer 2006, chamber measurements were carried out every 7–15 days (eight sampling days total), while in summer 2007 measurements occurred approximately twice a week (17 days of experiments). NO_x emissions were not measured from the chamber experiments.

Soil analyses

Topsoil (0–10 cm) was sampled weekly during summer 2007 in a 10 m² subplot ~10 m away from the snow flux tower. Per each date a representative sample of the whole plot was generated from 5 subsamples. Soil samples were processed within 24 h of returning from the field. Fresh soils were sieved and homogenized using a 2-mm sieve. The 5 subsamples were then extracted with 0.5 M K₂SO₄

(1:5 weight:volume) by shaking for 1 h. Samples were then filtered through Whatman #1 filter papers. Before extraction, each sample was divided in two subsamples, one was extracted and one fumigated for 18 h with chloroform and subsequently extracted as explained before. Extractable ammonium (NH_4^+) and nitrate (NO_3^-) were determined using a Spectrophotometric Analyzer (OI Analytical Flow Solution IV). NH_4^+ determination was based on Berthelot reaction, while nitrate was reduced to nitrite on a cadmium column, and the resulting ($\text{NO}_3^- + \text{NO}_2^-$) were determined by diazotizing with sulfanilamide, followed by coupling with N-(1-naphthyl) ethylenediamine dihydrochloride. Total dissolved nitrogen (TDN) was determined as nitrate after digestion and oxidation with alkaline potassium persulfate. Dissolved organic nitrogen (DON) was calculated by the difference between TDN and inorganic N (i.e., NH_4^+ and NO_3^-). The microbial N ($N_{\text{microbial}}$) was calculated by difference between the TDN of the fumigated samples and the TDN of the control samples. The resulting data were then corrected by a recovery factor of 0.54 (Brookes et al. 1985).

Statistical analyses

For the winter experiment, each 80-min cycle gradient data were used to calculate “hourly” fluxes (17 per day) from all combination of inlets. Winter daily means were then obtained by averaging the 17 calculated fluxes from all combination of inlets (Fig. 3). The daily coefficients of variation (CVs, calculated as the daily standard deviation divided by the daily mean \times 100) averaged for the whole winter were 30 and 25% for NO_x , and N_2O , respectively, in 2006. In winter 2007 CVs were 19 and 25% for NO_x , and N_2O , respectively. In summer daily means were obtained by averaging the 5 flux measurements obtained by the 5 chambers. The daily CVs, averaged over the whole season, were \sim 64%, and 139% for 2006, and 2007, respectively.

Correlations between gas fluxes and selected parameters were carried out using Spearman or Pearson correlation, depending on homogeneity of variances (Levene test). Inter-year variability was tested with one-way ANOVA using daily averages as variables, year as treatment. All statistical analyses were carried out with the software SPSS 12.0 for Windows.

Advection correction

Seok et al. (2009) show how advection from wind pumping reduces not just the gas concentrations in the snowpack, but also gas gradients and consequently fluxes calculated by Eq. 1. Their findings show that at high wind speeds gas transport in the snow is not solely driven by molecular diffusion but that advection can play a substantial role in determining the gas gradients through the snowpack. From a careful analysis of the dependency of calculated CO_2 fluxes on wind speed, it was concluded that, for the wind and snow conditions encountered during a 3-week interval in the middle of the winter, data derived from the diffusion approach underestimated actual fluxes by 36%. The wind conditions encountered during that period appeared to be reasonably representative of the conditions during the 2006 and 2007 winters. Throughout this paper we report fluxes derived from the simplified diffusion approach, in order to be consistent with previous literature and with Liptzin et al. (2009) who used the same approach for CO_2 flux. However, in Table 3, we present the larger, corrected flux results for the seasonal and yearly average values, which were derived applying a correction factor of 1.56 as suggested by Seok et al. (2009).

Results and discussion

Soil nitrogen pools

Pools of soil nitrogen were measured over time during summer 2007 (Table 1). The average nitrate content of $1.3 \text{ mg N kg dry soil}^{-1}$ and average ammonium content of $\sim 14 \text{ mg N kg dry soil}^{-1}$ are similar to those reported by Williams et al. (1996) and Neff et al. (1994) for the growing season at a nearby dry alpine meadow at Niwot Ridge. Compared to a subalpine meadow soil (Cryorthent) in the European Alps (Freppaz et al. 2007), ammonium content at NWT is similar and nitrate is one order of magnitude lower. The microbial biomass ($N_{\text{microbial}}$) of $102 \text{ mg N kg dry soil}^{-1}$ is similar to the microbial biomass of $110 \text{ mg N kg dry soil}^{-1}$ reported by Fisk and Schmidt (1995) for the Saddle site at Niwot Ridge. Contrastingly, DON of about $50 \text{ mg N kg dry soil}^{-1}$ was about four times higher than what was

Table 1 Soil N content in summer 2007 at Niwot Ridge (mg kg d.w.⁻¹)

Day of year 2007	NH ₄ ⁺	NO ₃ ⁻ + NO ₂ ⁻	IN ^a	DON	TDN	N _{microbial}
207	18.4	1.8	20.1	53.5	73.7	59.8
214	14.4	0.8	15.1	47.5	62.6	106.3
223	12.1	0.8	12.9	45.3	58.2	80.4
236	8.9	0.5	9.4	63.4	72.8	118.6
242	12.8	0.8	13.6	44.1	57.7	95.0
249	11.9	1.7	13.6	33.1	46.7	91.2
257	13.2	1.1	14.4	51.4	65.7	100.3
264	15.4	1.9	17.2	51.2	68.5	127.0
276	20.6	2.2	22.8	63.5	86.3	140.6
Seasonal mean (SE)	14.2 (3.5)	1.3 (0.6)	15.5 (4.0)	50.3 (9.5)	65.8 (11.4)	102.1 (24.6)

^a Abbreviations: *IN* inorganic nitrogen, *DON* dissolved organic nitrogen, *TDN* total dissolved nitrogen

reported by Fisk and Schmidt (1995). These values are indicative of a fast turnover of nitrogen at this site.

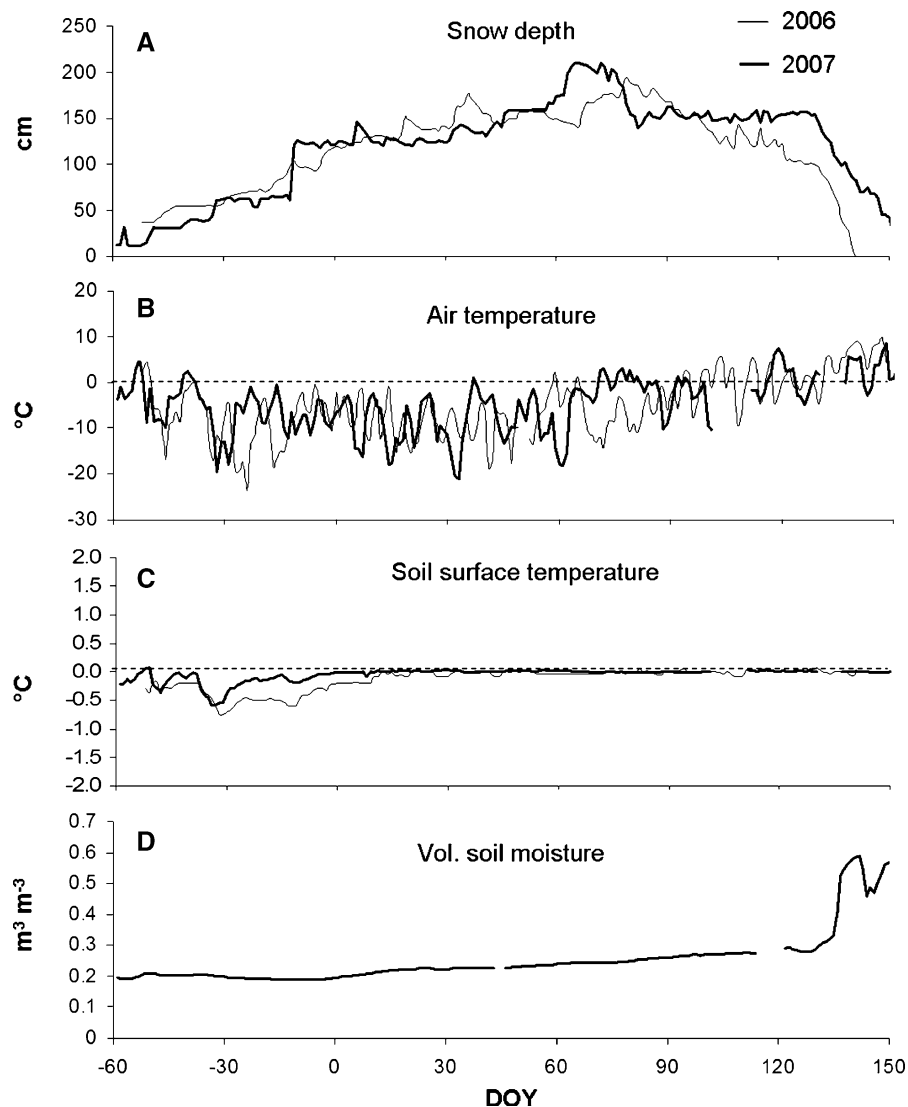
Winter results

Detailed description of meteorological patterns during the winter seasons 2006 and 2007 is provided in Seok et al. (2009). Here we present selected parameters useful to discuss the gaseous N fluxes. A continuous snowpack developed between the end of October and early November in both years (Fig. 1a). Maximum snow depth was reached on DOY 79 in 2006 (194 cm) and DOY 65 in 2007 (220 cm). The average air temperature during the snow-covered period was approximately -5°C for both years (Fig. 1b). Air temperature followed approximately the same trend during both winters, with minimum values of about -22°C in mid-winter and an increasing trend in late winter and early spring. Soil surface temperature increased to $\sim 0^{\circ}\text{C}$ in both years when the snowpack reached a depth of approximately 120 cm (Fig. 1a, c). In winter 2007, when volumetric soil moisture was recorded, there was a small but steady increase in soil moisture starting soon after DOY 1, i.e., coinciding with soil surface temperature warming up to $\sim 0^{\circ}\text{C}$. The gradual increase of liquid water may originate from the snowmelt at the bottom of the snowpack (Feng et al. 2001) or the transport of water vapor from the snowpack to soil because of temperature gradients (Sommerfeld et al. 1996). The gradual increase can also be explained as melting of frost within the first centimeter of soil. Coincident with the main flush of snowmelt starting in early

May, soil volumetric water content increased from about 25% on DOY 133 to about 60% by volume on DOY 143 (Fig. 1d).

Concentrations of N₂O within the snowpack were highest near the ground and decreased with distance from the soil surface (Fig. 2). Concentrations in the snowpack usually linearly decreased with increasing sampling height. Maximum concentrations of N₂O within the snowpack were near 1,050 ppbv in both years. The multiple inlets allowed us to calculate for each sampling cycle fluxes from nine different combinations of gradient heights in 2006 and from twenty-seven combinations for 2007 (Fig. 3). A detailed analysis on the dependence of concentration gradients on environmental and physical conditions of the snowpack is presented in Seok et al. (2009). Briefly, this work demonstrated that the gradients calculated from within the snowpack were consistently linear, but that a departure from this behavior was observed at the snow–soil, and at the snow–atmosphere interface. As with concentrations, N₂O fluxes calculated between inlets that were close to the snow surface, or close to the soil, deviated more (up to 60%) from the average flux than fluxes calculated closer to the center of the snowpack (Fig. 3). The correlation between fluxes from nearby inlets within the snowpack was consistently the highest. To illustrate this behavior, a simple linear regression was applied for fluxes calculated between the 30 and 60-cm inlets and the 60 and 90-cm inlets; results show a correlation with an R^2 of 0.561 ($n = 1,235$, $p \ll 0.001$) and a slope value near one (equation: $y = 0.953x$) (Fig. 4). The fact that lower correlations were obtained for fluxes derived from concentration

Fig. 1 Seasonal course of selected parameters for winters 2006 and 2007. **A** Snow depth, **B** Air temperature, **C** Soil surface temperature, **D** Soil moisture



gradients at the snow–atmosphere interface or from measurements near the soil–snow interface or both, and that better correlations were found between fluxes calculated within the snowpack implies that in the center of the snowpack gas transport is well approximated by the diffusion law, without any obvious deviation because of physical barriers such as ice lenses.

For further analyses and interpretations presented in this paper we used the N_2O flux calculated from inlets at heights of 30 and 60 cm (Fig. 5). These data were found to generally be in close agreement to the median flux values that were derived from the whole

array of flux values calculated from different inlet combinations (Seok et al. 2009). Furthermore, measurements from these heights represented the largest available dataset. For 2006, at the beginning of the winter season fluxes first gradually increased from about 0.020 to $0.090 \text{ nmol m}^{-2} \text{ s}^{-1}$. Fluxes then averaged around $0.080 \text{ nmol m}^{-2} \text{ s}^{-1}$, and decreased again during the snowmelt period. In winter 2007, the maximum flux of N_2O was measured at the initiation of the snowmelt, on DOY 128 (Fig. 5). This peak coincided with the rise in air temperature to above 0°C , which drove the sharpest snowmelt event of the season (Fig. 1a, b). Shortly after this event, the N_2O

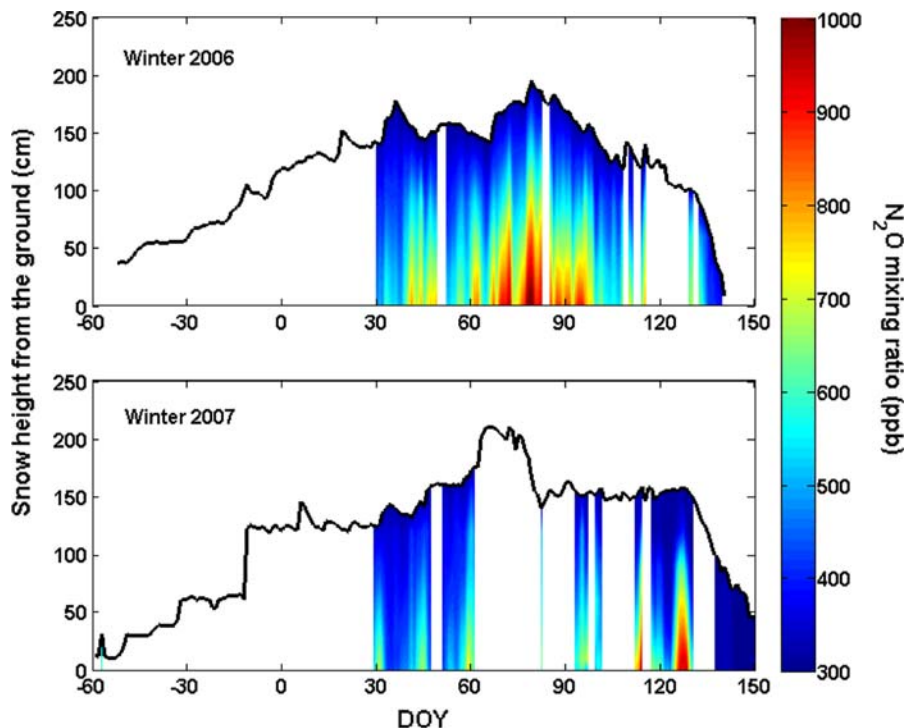


Fig. 2 Contour plots showing N_2O snowpack interstitial air concentration during the 2006 and 2007 winters. The *black line* depicts the height of the snowpack

production rapidly decreased to very low values for the rest of the season (i.e., between DOY 140 and 150). Unfortunately, a lack of data during this phase in the winter 2006 record prohibits us from further evaluating the occurrence of such low fluxes during that season (Fig. 5a). The seasonal average N_2O flux was $0.069 \text{ nmol m}^{-2} \text{ s}^{-1}$ for 2006, and significantly lower in 2007 at $0.047 \text{ nmol m}^{-2} \text{ s}^{-1}$ ($p < 0.001$).

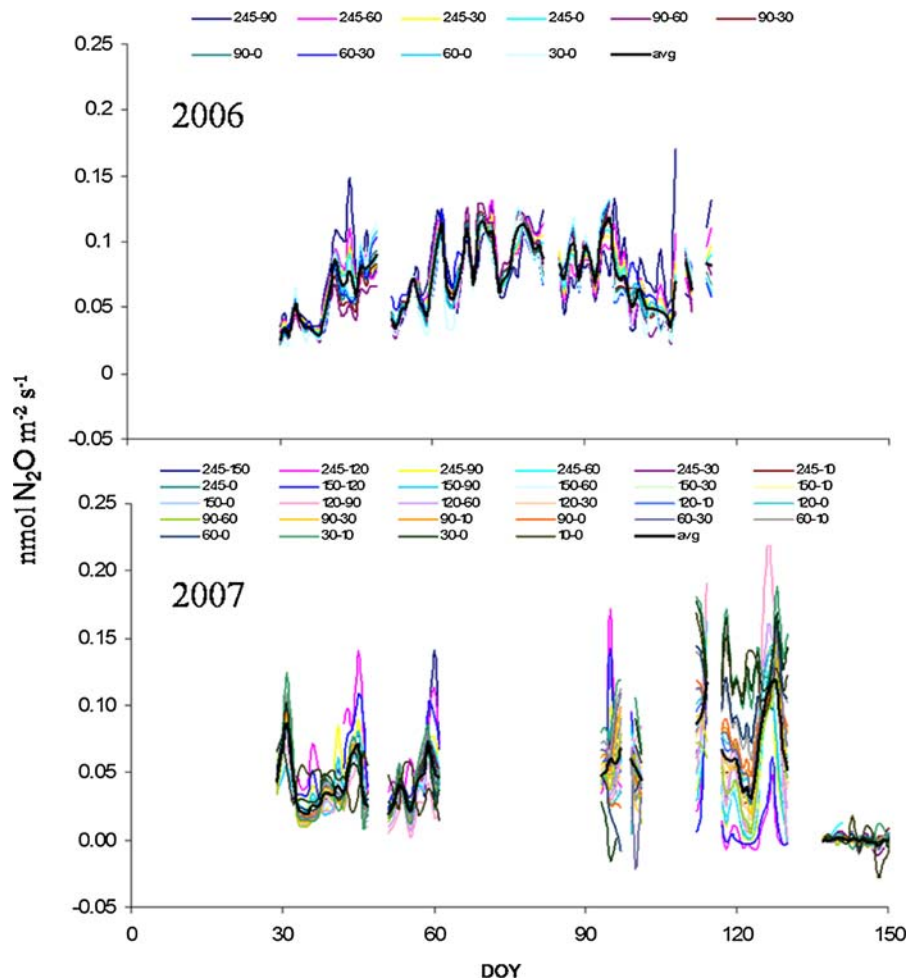
The behavior of NO_x concentrations and fluxes was a bit more complex than for N_2O . At the bottom of the snowpack only NO was observed, but with increasing height an increasing portion of NO_x was in the form of NO_2 . This feature was interpreted as a portion of NO converting to NO_2 , as NO diffuses from the soil towards the snowpack surface (Helmig et al. 2009). NO_x fluxes were on the order of 0.003 – $0.007 \text{ nmol m}^{-2} \text{ s}^{-1}$, with a slightly decreasing trend towards the end of the winter season, making the NO_x contribution on the order of a fifteenth (on a molar basis) of N_2O fluxes (Fig. 5). The average NO_x flux during the snow-covered season was significantly higher in 2006 than in 2007 ($p \ll 0.001$; Table 2).

The statistically higher fluxes of both N_2O and NO_x in winter 2006 than in 2007, without any obvious difference in surface soil temperature, snowfall timing and depth between the 2 years, suggest that factors other than the parameters recorded here may influence the N gaseous production under the snow. These factors may include soil moisture (not measured in winter 2006) and the soil nutrient status at the end of the growing season.

Summer results

Summer N_2O fluxes were studied during the day, typically between 8 am and 4 pm. Due to the chamber sampling technique it was not possible to perform night-time measurements, when fluxes are expected to be lower (e.g., Mosier et al. 1997), especially during periods with wide variation in temperature or during rainfalls (Maljanen et al. 2002). Daily fluxes calculated from our dataset are therefore likely to over-estimate the real 24-h average flux. Summer daily fluxes (Fig. 6) ranging from 0.2 to 1.8

Fig. 3 Seasonal course of daily mean N_2O fluxes in winters 2006 and 2007, as derived from the available inlet combinations. The solid black line depicts the average of all data



$\text{nmol m}^{-2} \text{s}^{-1}$ for 2006 and from -0.1 to $1.1 \text{ nmol m}^{-2} \text{s}^{-1}$ for 2007, were about one order of magnitude greater than winter fluxes. Summer fluxes of N_2O at our site are comparable to the rates reported by Neff et al. (1994) for Niwot Ridge alpine tundra and by Sommerfeld et al. (1993) at a Wyoming subalpine meadow. The summer 2006 N_2O efflux followed a decreasing trend during the dry-out period. Similarly in 2007, there was a general trend of decreasing flux over time, with two peaks following high precipitation events interrupting this pattern. The average N_2O flux of summer 2006 was not statistically different from the summer 2007 flux.

Seasonal (NO_x) and annual (N_2O) budgets

We integrated the fluxes over the whole season by taking the seasonal mean flux, calculated from the

daily averages, and multiplied it by the length (in days) of the snow-covered period (200 and 213 days for 2006 and 2007, respectively) and of the snow-free period (158 and 165 days for 2006 and 2007, respectively) to derive a winter and summer budget, respectively. The length of the seasons were calculated based on the first consistent snowfall (i.e., after that event the soil was permanently snow-covered until snowmelt) and by the end of the snowmelt. By integrating the average daily fluxes, we estimated the individual and total ($\text{N}_2\text{O} + \text{NO}_x$) loss for this ecosystem for both winters (Table 2). N_2O -N winter losses were 0.34 and $0.24 \text{ kg N ha}^{-1}$ for 2006 and 2007, 30–50 times (on a N atom basis) that of NO_x -N winter fluxes at 0.011 and $0.0051 \text{ kg N ha}^{-1}$ for 2006 and 2007.

The estimate of seasonal budgets from a dataset leads to uncertainties, and this is especially the case

Fig. 4 Scatter plot between 90–60 cm and 60–30 cm fluxes ($\text{nmol m}^{-2} \text{s}^{-1}$), winter 06. The 1:1 line and the result for the regression equation through all data points are shown

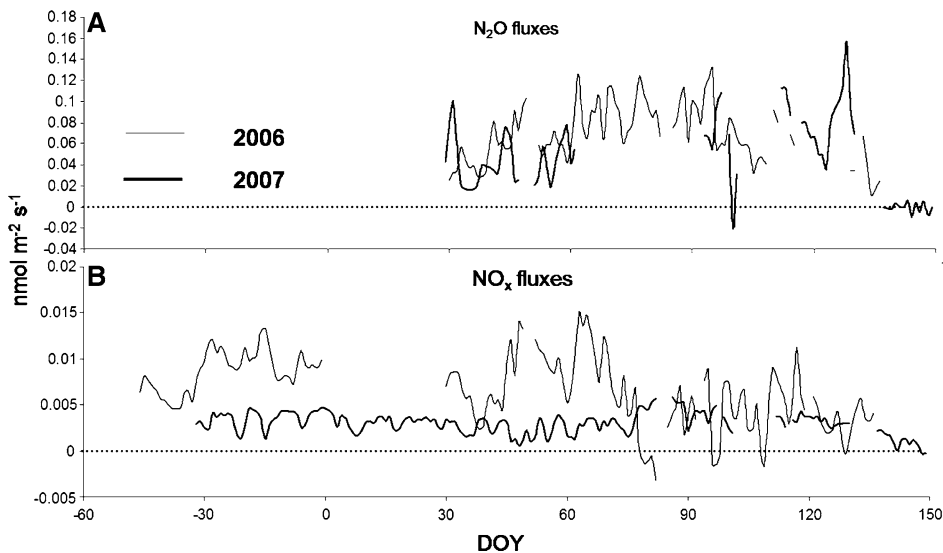
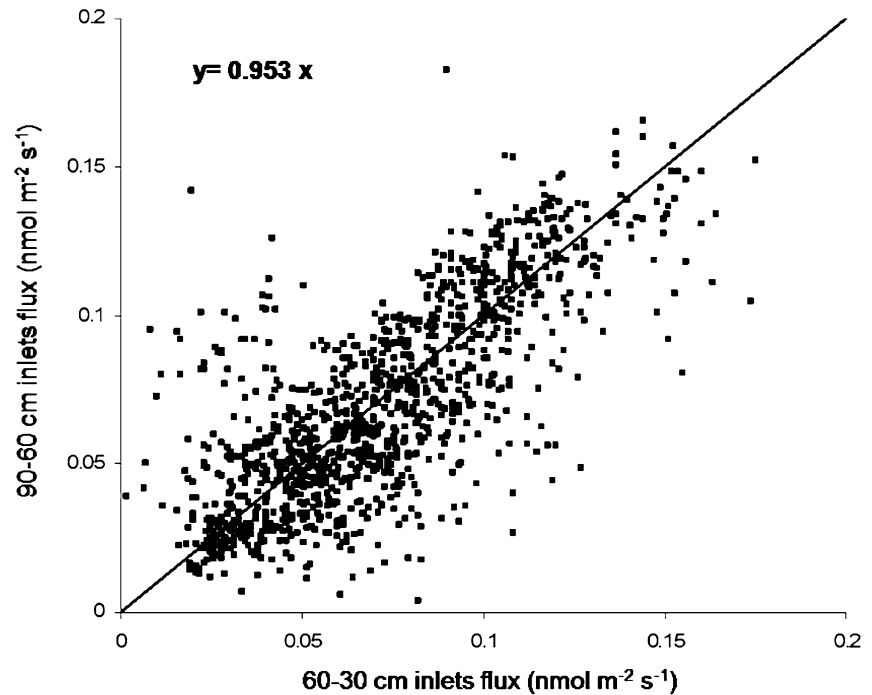


Fig. 5 Seasonal course of mean daily gas fluxes for winters 2006 and 2007. **a** N_2O fluxes (from 60 to 30 cm inlets), **b** NO_x fluxes (60–30 cm inlets)

for data from occasional, non-continuous sampling such as in our summer experiments. Furthermore, about half of our N_2O winter fluxes were filled in by extrapolation (while the NO_x data set was more continuous), and winter N_2O fluxes were not determined for the November to January period. In early

winter the soil surface temperature is relatively low (compared to the mid winter), which suggests lower N_2O fluxes. Since we applied an estimate based on the mid to late winter data for that period, this approach may have resulted in an over-estimation of the cumulative winter flux.

Table 2 Seasonal and annual N-N₂O and N-NO_x loss during years 2006 and 2007

	Source of N	Seasonal mean (SE)		% of the seasonal gaseous N loss	% of the annual N-N ₂ O loss
		nmol m ⁻² s ⁻¹	kg N ha ⁻¹ season		
Winter 06	N ₂ O	0.069 (0.0033)	0.34 (0.015)	97	12
	NO _x	0.0067 (0.00031)	0.011 (0.0012)	3	
Summer 06	N ₂ O	0.61 (0.18)	2.4 (0.72)	NE	88
	NO _x	NM ^a	NM		
Winter 07	N ₂ O	0.047 (0.0045)	0.24 (0.023)	98	18
	NO _x	0.0030 (0.000094)	0.0051 (0.00016)	2	
Summer 07	N ₂ O	0.31 (0.079)	1.1 (0.29)	NE	82
	NO _x	NM	NM		

The length of the seasons (days) was established according to onset of the snow cover and the end of snowmelt (see text). Winter 2006: 200 days; winter 2007: 213 days; summer 2006: 158 days; summer 2007: 165 days

^a Abbreviations: *NM* not measured, *NE* not estimated

Our experiment did not include summer measurements of NO_x, which limits our ability to assess the total annual oxidized N balance. Comparing the N₂O-N loss between winter and following summer we found that winter fluxes account for ~12 and 18% of the total N₂O emitted from this subalpine soil on a yearly basis. When the wind pumping correction is considered the contribution of the winter N₂O flux increases to 19–28% of the total annual N₂O flux.

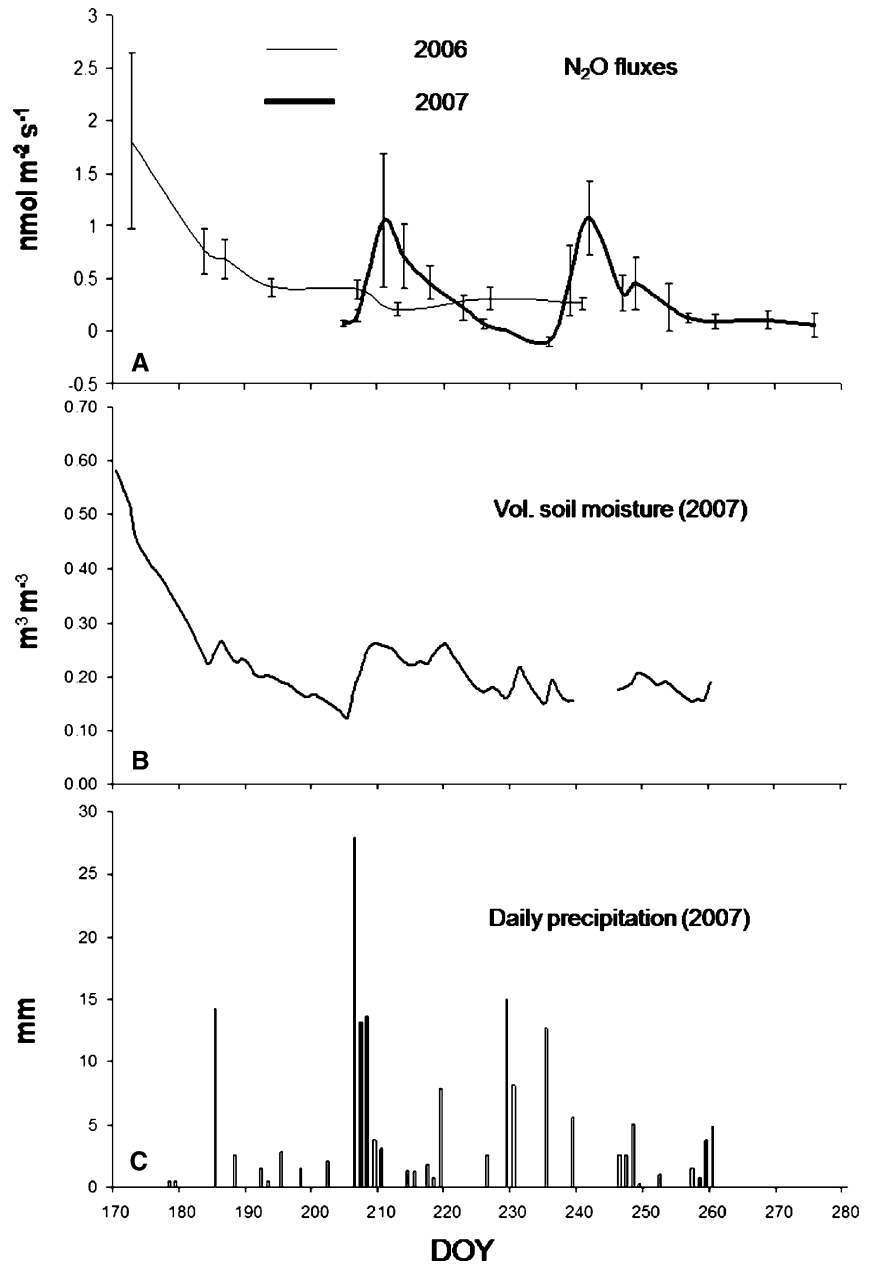
Comparison with other ecosystems

The 0.24–0.34 kg N ha⁻¹ of N₂O-N flux from our continuous winter measurements at this subalpine meadow is of similar order as the 0.17–0.23 kg N ha⁻¹ that Brooks et al. (1997) report for a much more limited data set from alpine tundra on Niwot Ridge (Table 3). Both the emission rate and the snow depth at our study site were comparable to that of Brooks et al. (1997). Perhaps most importantly, in both studies the soil temperature was about 0°C throughout much of the winter because of early-accumulating and deep snowpacks. N₂O winter fluxes at NWT are in the high range of data reported for forested ecosystems; e.g., our emission rates are comparable to the ones reported by Groffman et al. (2006) and van Bochove et al. (2000) for yellow birch and sugar maple forests (Table 3), which are ecosystems dominated by species associated with high rates of N cycling (Lovett et al. 2004). Even more surprisingly, N₂O winter fluxes at NWT are two to threefold higher than for fertilized grasslands in Colorado (Mosier et al. 1997). The

relatively well developed soils and the deep snowpack results in prolonged, relatively warm and stable soil temperatures, which may allow the formation of a cold-adapted microbial community (Schadt et al. 2003). The microbial community under the snowpack is phylogenetically and physiologically distinct from that developing during the growing season (Schadt et al. 2003; Monson et al. 2006a, b), and such community can reach its maximum biomass during winter (Brooks et al. 1996; Lipson et al. 2000). This active microbial community has the potential to mineralize nitrogen at high rates (Brooks et al. 1996), providing nitrate supply for denitrification, thus explaining the high emission rates recorded in this study. Supporting our findings, Schürmann et al. (2002) reported a N₂O efflux as high as 0.5 nmol m⁻² s⁻¹ for a fertilized alpine grassland, suggesting that when deep, insulating snowpacks are overlying N-enriched soils, the potential for the N₂O-loss in winter is much higher than during the growing season.

Two previous measurements of NO_x flux from snow-covered soils allow comparison with our data (Maggiotto and Wagner-Riddle 2001; Ludwig et al. 2001). Compared to these reports, NWT NO_x subnivean production is about two-orders of magnitude lower. Moreover, Ludwig et al. (2001) reviewed the current literature on NO emissions from soils, with only two of over 200 studies from tundra and subalpine ecosystems. Most of the NO_x emitted through the snowpack at NWT was in the form of NO, making it possible to compare our winter fluxes

Fig. 6 Seasonal course of daily N_2O fluxes for summer 2006 and 2007 (A, error bars are 1SE), volumetric soil moisture for 2007 (B) and daily precipitation for 2007 (C)



of NO_x with these published data. NWT NO_x flux is roughly comparable to the values reported by Ludwig et al. (2001) for the snow-free season of tundra and subalpine ecosystems, thus confirming that alpine and subalpine snow-covered soils are comparatively weak sources of NO_x , but also suggesting that winter production could be as high as the snow-free season production in this ecosystem.

Controls on N gas fluxes: summer period

During summer, the flux of N_2O was strongly regulated by precipitation, as shown by several investigators for other ecosystems (Binstock 1984; Sexstone et al. 1985; Neff et al. 1994; Dunfield et al. 1995; Smith et al. 1998; Schürmann et al. 2002). For both summers, there was a general trend of

Table 3 Comparison of studies that reported N₂O fluxes in the snow-covered season in different ecosystems

Reference	Site description (abbreviation)	Mean annual T (°C)	N ₂ O Flux (nmol m ⁻² s ⁻¹)	kg N ha ⁻¹ season ⁻¹ (for winter)	% contribution to the year total
This study	Subalpine meadow (NWT) ^a	-3.7	0.047–0.069	0.24–0.34	12–18%
	Corrected for wind pumping effect		0.073–0.11	0.38–0.53	19–28%
Brooks et al. (1997)	Alpine Tundra (NWT1)	-3.7	0.016–0.046	0.17–0.23	n.e.
Brooks et al. (1996)	Alpine tundra (NWT2)	-3.7	0.026–0.031	NE	NE
Sommerfeld et al. (1993)	Subalpine meadow (WY)	-1.0	0.006–0.01	NE	NE
Schürmann et al. (2002)	Alpine acidic and neutral pastures (SA)	0.0	0.03–0.50	0.01–3.47	5–85%
Zhu et al. (2005)	Antarctic tundra soil (MA)	-20.5	0.008–0.013	NE	NE
van Bochove et al. (2000)	Forested soil (QU1)	4.2	0.033*	0.05	NE
van Bochove et al. (1996)	Forested soil (QU2)	4.2	<0.023	NE	NE
Groffman et al. (2006)	Forested soil (Sugar maple and yellow birch) (NH)	4.5	0.062–0.087		
Kim and Tanaka (2002)	Grassland (HO)	7.5	0–0.004	0.005	10%
Maggiotto and Wagner-Riddle (2001)	Ryegrass fields (ON) (control, unfertilized plots)	6.7	0–0.09**	-0.07	n.e.
Mosier et al. (1997)	Colorado fertilized grasslands (CO)	9.3	0–0.02	NE	NE
Maljanen et al. (2003)	Organic soil (FI)	2.2	0–5.7	NE	NE

Listed data represent the range of the two seasonal mean results for the 2006 and 2007 experiments. For this study, the seasonal and yearly average values, corrected for the wind pumping effect (Seok et al. 2009) are reported, together with the uncorrected data

^a Abbreviations: *NE* not estimated, *NWT* Niwot Ridge (Colorado), *WY* Wyoming, *SA* Swiss Alps *QU* Quebec (Canada) *NH* New Hampshire, *HO* Hokkaido (Japan), *FI* Finland, *MA* Maritime Antarctica, *CO* Colorado, *ON* Ontario (Canada)

* Extrapolated from monthly data

** Range extrapolated from graphs

decreasing soil water content after snowmelt associated with a decrease in the flux of N₂O (Fig. 6). The higher frequency of summer sampling in 2007 was able to capture flux peaks after precipitation events that were two- to threefold higher than background fluxes (Fig. 6). These results are similar to those reported by other researchers who have shown increases in N₂O flux that usually occur within a few hours to days after precipitation events (Binstock 1984; Sexstone et al. 1985; Neff et al. 1994).

The N pool in soil is expected to play a role in regulating the N₂O emissions. Both organic and inorganic nitrogen tend to accumulate in soil in autumn (Table 2). No direct relationship between DIN and N₂O fluxes was found and perhaps more surprising, summer fluxes of N₂O were found to be inversely correlated with the DON content of the soil (Fig. 7). Figure 7 shows that as N₂O fluxes decrease from about 1 nmol m⁻² s⁻¹ to <0.1 nmol m⁻² s⁻¹, DON doubles from near 30 mg N kg dry soil⁻¹ to over 60 mg N kg dry soil⁻¹ ($r = -0.633$, $n = 8$).

These results suggest that summer N₂O efflux decreases as DON is produced. Both plants and microbes may sequester inorganic nitrogen thus reducing available DIN for denitrification process and in turn release N in organic forms through exudates or cell lyses, respectively, therefore increasing the pool of DON in soil. The lack of correlation between N₂O fluxes and DIN pool could suggest that the latter undergoes fast transformations in soil (i.e., plant uptake and microbial immobilization) that makes it difficult to track the N₂O behavior with respect to such labile pool.

Summarizing, these interactions among soil moisture, soil temperature, soil pools of DON and DIN, and N₂O flux are intriguing but remain partially unexplained, due to the fact that the ecological factors controlling N gaseous efflux are highly interactive. As a result, the response of N₂O flux to soil N appears to be less linear than what was reported in earlier studies (e.g., Skiba et al. 1997; Skiba and Smith 2000; Groffman et al. 2000).

Fig. 7 Correlation between daily average N_2O fluxes and dissolved organic nitrogen in topsoil (0–10 cm) during summer 2007 (Spearman r : -0.633 , $p < 0.05$). Error bars show the standard error

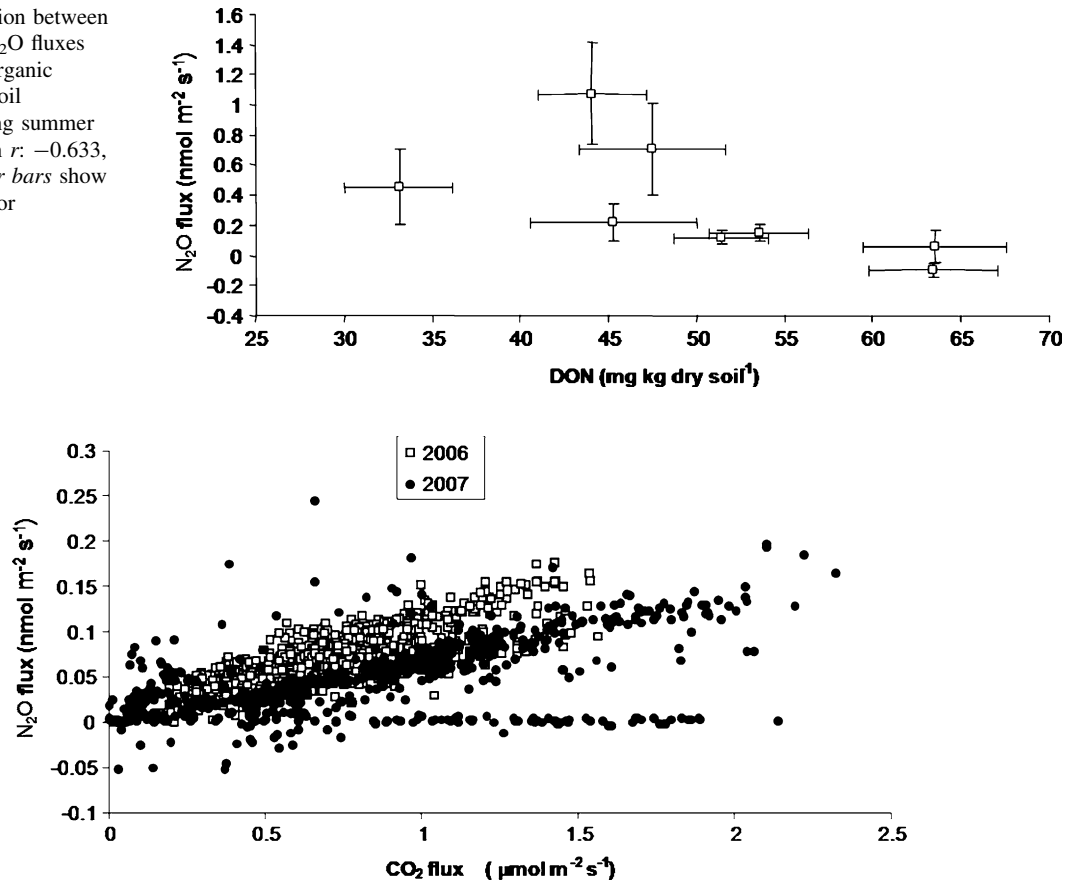


Fig. 8 Correlation between CO_2 and N_2O fluxes (from 60 to 30 cm inlets) for winter season 2006 and 2007

Controls on N gas fluxes: winter and snowmelt period

We evaluated whether the winter N gaseous flux may be related to CO_2 production by comparing the fluxes of both gases (Fig. 8). The CO_2 fluxes were measured at the same site and are those reported by Liptzin et al. (2009). There was a strong correlation between the N_2O and CO_2 fluxes (Spearman $r = 0.806$, $p < 0.01$). However, note that for a portion of the 2007 data set there was little if any flux of N_2O when fluxes of CO_2 remained elevated (Fig. 8). These data resulted from a time period of low fluxes of N_2O in 2007 during the later stages of snowmelt runoff, i.e., between DOY 140 and 150 (Figs. 3, 5). The high correlation between the two gases during the earlier part of the winter suggests a common source, i.e., that microbially mediated processes are the primary control on both CO_2 and N_2O production. This

relationship implies that CO_2 production may be primarily from microbial respiration rather than root respiration. A statistically significant correlation was also found between the CO_2 and NO_x fluxes (Spearman $r = 0.325$, $p < 0.01$), and NO_x and N_2O (Spearman $r = 0.342$, $p < 0.01$), corroborating the hypothesis that the same environmental parameters are controlling the emission of the three gases.

An ongoing question with respect to trace gas flux through snow is what processes determine the observed flux increases during snowmelt runoff that other investigators have reported during this period (Brooks et al. 1997; Mast et al. 1998; Monson et al. 2006a, b; Sommerfeld et al. 1996). In fact, increases in soil moisture and nitrate concentration during spring snowmelt have been associated with increases in the amount of N_2O measured (Brooks et al. 1997; Maggiotto and Wagner-Riddle 2001; Maljanen et al. 2007). Consistent with such reports, in winter 2007

we measured the maximum flux of N_2O through the snowpack at the initiation of snowmelt. Another mechanism which has been suggested for high N_2O emission peaks at snowmelt is the release of stored N_2O from frozen soils (Maljanen et al. 2007); however, the maximum emission recorded in this study occurred when surface soil temperature was close to 0°C . The fact that the concomitantly determined CO_2 flux increased gradually (Liptzin et al. 2009), without showing the rapid increase seen in the N_2O emission, suggests that biological, rather than physical conditions (which would be expected to exert similar controls on both gases) were the driving factor in the N_2O behavior.

For the remainder of the snow-covered season (i.e., between DOY 140 and 150 of 2007, Figs. 3 and 5) the production of N_2O decreased to lower values. This sharp decrease in N_2O production after the initiation of snowmelt in 2007 is an interesting and unexpected pattern. This drop in N_2O fluxes causes a remarkable departure from the relationship between fluxes of N_2O and CO_2 (Fig. 8). First, we suspected an artifact in our experiment be responsible for this behavior. Therefore, we carefully re-investigated these data but could not identify any analytical or data processing errors to explain the low amounts of N_2O flux. Moreover, there was not a similar reduction in our concurrent measurements of the efflux of CO_2 . This further confirms that an experimental artifact is unlikely, and that sampling problems do not seem adequate to explain this observed decrease in the flux of N_2O . One explanation for the sharp decrease in N_2O production is that during this period N_2O production was inhibited by possibly N_2O being completely reduced to N_2 . High amounts of soil moisture have been shown to promote the reduction of nitrate to N_2 , both because of the lower redox potential and the reduced diffusivity of N_2O in the water-saturated soil, which allows time for the reduction to occur (Chapuis-Lardy et al. 2007). Note that the decrease in N_2O production after the initiation of snowmelt that we report occurred concurrently with a sharp increase in soil moisture (Fig. 1).

Nutrient limitation (e.g., some combination of carbon and inorganic nitrogen) is another possible reason for this reduction in the measured flux of N_2O during the later stages of snowmelt. Lipson et al. (2000) have shown that the decline in

microbial biomass of soils reported during the snowmelt period (e.g., Brooks et al. 1998; Brooks and Williams 1999; Williams et al. 1996) may be caused by carbon limitation. In particular, long-duration, early developing snowpacks such as at our site may promote conditions where the otherwise common pulse of available carbon substrate during snowmelt is reduced (Brooks et al. 1997). The reason is that the long period of time with soils at or near 0°C results in an environment where heterotrophic activity may occur throughout the winter, utilizing available nutrients (Brooks and Williams 1999). Moreover, this carbon limitation under the snowpack has been shown to limit microbial respiration rates under snow and the resultant flux of CO_2 (Brooks et al. 2005).

Furthermore, after the early stages of snowmelt, inorganic nitrogen could become less available, contributing to the decrease in measured N_2O flux at this time. Nitrate and ammonium are released from the seasonal snowpack in the form of an ionic pulse (Williams and Melack 1991; Williams et al. 1996). Williams et al. (2009) computed the total NO_3^- -N loading in the snowpack at this site for both years at 1.4 – 1.7 kg N ha^{-1} , a range that exceeds the N lost as N_2O by a factor of 5–8. In addition, the over-winter nitrate production rate through nitrification is considerably high at this site (Brooks et al. 1996). Consequently, nitrate should not be a limiting factor for denitrification. However, Williams et al. (2009) have shown that DON was the dominant form of nitrogen in the soil solution at this site, with concentrations as high as $50 \mu\text{moles L}^{-1}$, and that concentrations of nitrate in nearby surface waters were always negligible. The cause of this discrepancy may be found in the inorganic N immobilization process: the recovery of isotopically labeled inorganic N in the soil (Williams et al. 1996) and in plants (Bilbrough et al. 2000) under the snow cover of nearby alpine areas, together with direct evidence of microbial N immobilization within the soil (Brooks and Williams 1999), suggests that immobilization of inorganic N in soil is an important control on the retention of N in the terrestrial environment. Thus, there may have been little nitrate available in underlying soils as a substrate for denitrification processes towards the later stages of snowmelt as a consequence of enhanced N immobilization and uptake.

Conclusions

N₂O and NO_x efflux from this subalpine meadow soil were detected at significant levels through the snowpack all winter long. Winter N₂O fluxes at NWT are among the highest ever measured in high elevation ecosystems, and are comparable to those observed in highly productive ecosystems, such as fertilized grasslands and high-N-cycling forests. The relatively well developed soils together with the ~210 days of prolonged deep snow cover developing at NWT result in subnivean N₂O production that might account for between 19 and 28% of the annual flux.

During the summer precipitation was identified as the main factor controlling N₂O flux. Our data suggest that during winter soil water content and substrate availability determines a N₂O pulse and subsequent decline during the snowmelt phase. Consequently, the snowmelt period appears to be critical for biological processes in seasonally snow-covered ecosystems.

This investigation on N gaseous emission through the seasonal snowpack represents an important extension to wintertime trace gas flux research that hitherto has mainly focused on CO₂ production. Increasing understanding on N₂O emission in seasonally snow-covered ecosystems may perhaps fill the gaps that still exist in evaluating the N budget at a global scale. Moreover, as N₂O is an important gas in determining the atmospheric radiative forcing, climate change-induced changes in snowpack conditions will likely feedback on wintertime N₂O flux. Given our current understanding of the controls on the biological processes resulting in N₂O flux in the wide array of seasonally snow-covered ecosystems, there is still a high uncertainty with both the direction and the magnitude of potential changes in these fluxes in a changing climate.

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