Lawrence Berkeley National Laboratory

Recent Work

Title CLOSE-COUPLING CALCULATION OF Li+-H2 DIFFUSION

Permalink https://escholarship.org/uc/item/7b50r10v

Author Lester, W.A.

Publication Date 1981-11-01

RESOURCE FOR COMPUTATION IN CHEMISTRY

Submitted to the Journal of Physics B: DOCUMENTS SECTION Atomic and Molecular Physics

CLOSE-COUPLING CALCULATION OF LI⁺-H₂ DIFFUSION

A.S. Dickinson, M.S. Lee, and W.A. Lester, Jr.

November 1981

TWO-WEEK LOAN COPY

LBL=13983 Přeprint

BERKELEY LABODATORY

This is a Library Circulating Copy which may be borrowed for two weeks. For a personal retention copy, call Tech. Info. Division, Ext. 6782

LAWRENCE BERKELEY LABORATORY UNIVERSITY OF CALIFORNIA

Prepared for the U.S. Department of Energy under Contract W-7405-ENG-48 and for the National Science Foundation under Interagency Agreement CHE-7721305

DISCLAIMER

This document was prepared as an account of work sponsored by the United States Government. While this document is believed to contain correct information, neither the United States Government nor any agency thereof, nor the Regents of the University of California, nor any of their employees, makes any warranty, express or implied, or assumes any legal responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by its trade name, trademark, manufacturer, or otherwise, does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof, or the Regents of the University of California. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof or the Regents of the University of California. Close-Coupling Calculation of Li⁺-H₂ Diffusion Cross Sections

A.S. Dickinson and M.S. Lee⁺

Department of Atomic Physics, School of Physics, The University, NEWCASTLE UPON TYNE, NE1 7RU, U.K.

and

W.A. Lester, Jr.*

National Resource for Computation in Chemistry Lawrence Berkeley Laboratory University of California Berkeley, CA 94720

+Permanent address:

Department of Material Science, Ulsan Institute of Technology, Ulsan, Korea.

*Present address: Materials and Molecular Research Division, Lawrence Berkeley Laboratory and Department of Chemistry, University of California, Berkeley, California 94720, U.S.A.

This work was supported by the Director, Office of Energy Research, Office of Basic Energy Sciences, Chemical Sciences Division of the U.S. Department of Energy under Contract No. W-7405-ENG-48 and by the National Science Foundation under Grant No. CHE-7721305.

ABSTRACT

The diffusion cross sections $Q_d(j \rightarrow j')$ have been calculated for the $0 \rightarrow 0$, $0 \rightarrow 2$ and $2 \rightarrow 2$ transitions in Li⁺-H₂ collisions at total energies between 46.4 and 51 meV. For the elastic cross sections, the results were within 15% of those obtained classically using the spherical component of the H₂ polarizability. The value of $Q_d(0 \rightarrow 2)$ is much larger than that obtained previously by Alexander (1976) but is too small to be significant for the room temperature mobility.

1. Introduction

Diffusion cross sections are required to evaluate ionic mobilities in gases. Precise experiments on these are available (McDaniel and Mason 1973, Ellis et al 1976a, 1978) for many ion-atom and ion-molecule combinations and provide a valuable source of information at energies not easily accessible by beam techniques. Recent advances in kinetic theory (Viehland and Mason 1978, Viehland et al 1981 and references therein) have opened up the possibility of utilizing mobility measurements at arbitrary field strengths to bridge the gap between the thermal energies sampled in previous zero-field mobility studies and the eV energies conveniently accessible in crossed-beam experiments. Mobility measurements on Li^+-H_2 have been performed by Ellis et al (1976b) and the effective collision energy goes up to about 19.3 eV.

While fully quantal calculations of the diffusion cross section, Q_d , for ion-atom collisions have been available for some time (McDaniel and Mason 1973), see also Dickinson and Lee (1978) and Wadehra et al (1978) for recent work, very much less has been done on quantal calculations of ion-molecule diffusion cross sections. Alexander (1976) has made a valuable introductory study of Li⁺-H₂ diffusion cross sections. However, the close-coupling bases he used were rather small. Here we re-examine this problem using a larger basis set for energies around 50 meV. This energy region is important for calculating the room-temperature mobility.

Atom-molecule diffusion coefficients have been calculated quantally by Monchick and Green (1977), see also Maitland et al (1981) for recent calculations of transport properties using the Infinite-Order-Sudden approximation (Parker and Pack 1978). However, these calculations have been dominated by short-ranged repulsive interactions while ion-molecule collisions are expected to be influenced most by the stronger long-range polarizability and quadrupole interactions. Also the large rotational spacings in H_2 may lead to differences from results obtained with heavier molecules.

In the following section we describe the scattering calculations and the evaluation of the diffusion cross section. Section 3 contains the results and discussion while our conclusions are presented in section 4.

2. Method

2.1 Scattering Calculations

We have followed the standard close-coupling procedure (Arthurs and Dalgarno 1960, Lester 1976). For total angular momentum quantum numbers $J \leq 20$, we have used the S-matrices obtained using the matrix elements and solution techniques of Schaefer and Lester (1973), employed previously by Dickinson et al (1976) to calculate the integral inelastic cross section Q(0+2) between rotational levels j = 0 and j = 2. Those S-matrices were calculated using a basis set $0 \le j \le 6$ in v = 0 and $0 \le j \le 4$ in v = 1, v denoting the vibrational quantum number. For the larger J values needed to ensure convergence of elastic diffusion cross sections, we have used the matrix elements and solution techniques of Kouri and Wells (1974) except that a new quadrature was employed in solving the integral equations. A basis set $0 \le j \le 4$ in v = 0 was employed. The fit to V_0 , the spherically symmetric component of the ion-molecule potential was altered slightly so that the polarization limit was employed for ion-molecule separations $R \ge R_0 = 7a_0$, R_0 being the point where the polarization potential crossed the fit used previously. For $R \ge 8a_0$ this polarization potential is within 2×10^{-5} a.u. of the ab initio calculation of Kutzelnigg et al (1973). Comparison between S-matrices calculated for J = 20 with the two potentials showed good agreement, the largest discrepancy being in the S^J(OJ,OJ) element where differences up to the equivalent of 0.1 radian in the phase shift occurred. (At this value of J, $S^{J}(0J, 2\ell)$ is very small.) However, for the diffusion cross section it is primarily the difference between adjacent phase shifts which is required and for these the two potentials gave results differing

by less than 0.03 rad.

To evaluate the 2+2 component of the diffusion cross section, it was necessary to calculate the additional S-matrix elements of odd parity, $S^{J}(2J\pm1, 2J\pm1)$. These were calculated, as described above for $J \ge 20$, with a basis set $2 \le j \le 6$ in v = 0 and $2 \le j \le 4$ in v = 1. Because of the substantial computing time needed, this calculation was performed at energies 46.9, 48.5, 50 and 51 meV only, spanning the range covered more closely in the other parity.

2.2 Diffusion Cross Section Calculations

We have used the procedures of Alexander (1976) for evaluating $Q_d(0+0)$ and $Q_d(0+2)$, where $O_d(j+j')$ is the diffusion cross section for transitions from level j to level j' (Arthurs and Dalgarno 1960). Two minor typographical errors in Alexander's paper are: (i) in his equation (6) $\ell_{max}/4$ should be $\ell_{max}/2$, and (ii) in his equation (11) $h_{\ell,\ell'}$ should be $h_{\ell,\ell'}^{J}/(J+1)$ (M.H. Alexander, private communication 1976).

We have formed an explicit expression for $Q_d(2\rightarrow 2)$ in terms of elements of the transition matrix \underline{T}^J . From Arthurs and Dalgarno (1960) we have, in their notation,

$$Q_{4}(j \rightarrow j') = Q(j \rightarrow j') - \pi k' A_{1}(j,j')/3k^{3}(2j+1).$$
 (1)

Here

$$A_{1}(2,2) = 3 \sum_{\substack{J \\ \ell_{1} \\ \ell_{1}'}} C(\ell_{1},\ell_{2})C(\ell_{1}',\ell_{2}') \left[T^{J^{*}}(2\ell_{1};2\ell_{1}') T^{J^{+1}}(2\ell_{2};2\ell_{2}') + \text{complex conjugate (c.c.)}\right]$$
(2)

Non-zero values of the C coefficient, which are obtained from certain Z coefficients of Blatt and Biedenharn (1952), are shown in Table 1. These coefficients have been checked against the more general expressions given by Hickman and Smith (1978).

3. Results and Discussion

The diffusion cross section $Q_d(0\rightarrow0)$ is shown in figure 1 as a function of energy, along with the value obtained by Alexander (1976) and the classical polarization cross section. The slight structure is due to resonances, both those discussed by Dickinson et al (1976) occurring for $J \leq 20$ and, more importantly, additional shape resonances in the j=0 channel for J=28. The behaviour of the partial diffusion cross sections, $Q_d^{\ J}(0\rightarrow0)$ as a function of J where

$$Q_{d}(j \neq j') = \sum_{J} Q_{d}^{J}(j \neq j'), \qquad (3)$$

¢

is similar to that obtained by Alexander (1976) at 50 meV (see his figure 1). This cross section is always within 15% of the classical Langevin result (McDaniel and Mason 1973) for a pure spherically symmetric polarization interaction for the energies considered and is generally much closer. The value obtained here at 50 meV is somewhat lower than that obtained by Alexander (1976), because of a resonance.

The even parity contribution to $Q_d(2+2)$ is also shown in figure 1. Because fewer J values contribute significantly at the same total energy, the resonance structure is more marked. The comparison with half the Langevin cross section shows that polarization forces are again dominant. While for j>0 the longest-ranged ion-homonuclear molecule interaction is the chargequadrupole, its relative magnitude for H₂ is so small that the polarization potential is more important out to about 40 a₀. Thus the distances at which the quadrupole interaction is dominant make negligible contributions to the cross section except extremely close to threshold. The behaviour of $Q_d^J(2+2)$ as a function of J is qualitatively similar to that for $Q_d^J(0+0)$, although for the 2+2 cross section the glory minimum occurs at J = 7 or 8.

In table 2 are shown the contributions to $Q_d(2\rightarrow 2)$ from T-matrices of both parities. While the even and odd contributions are comparable in magnitude some variation in their ratio occurs due to resonances. The full cross

section is within 5% of the classical polarization limit even though considerable mixing occurs between the different j = 2 channels for all J values contributing significantly.

We have found that over the whole energy range considered, $Q_d(0+2)$ lies between 86% and 100% of Q(0+2), which is shown in figure 1 of Dickinson et al (1976). At 50 meV the value of $Q_d(0+2)$ is considerably larger (0.36 Å² against 0.1 Å²) than that calculated by Alexander (1976) using the minimal basis set, v=0, j=0 and 2. Resonance effects are small at this energy.

The closeness of the integral and diffusion 0+2 cross sections, even though they vary in size by more than an order of magnitude, arises because for A₁(0+2), on which the difference depends, see eq.(1), we can write (Alexander 1976)

$$A_{1}(O+2) \simeq 3 \sum_{J} (J+1) \{ \sum_{\ell} [T^{J*}(OJ;2\ell) \ T^{J+1}(OJ+1;2\ell+1) + c.c.] + O(1/J^{2}) \}.$$
(4)

Since the phases of the products of T-matrix elements will in general differ substantially, considerable cancellation will occur in forming the sums. On the other hand Q(j + j') is formed by summing positive quantities and so is normally much larger. In addition, $A_1(O+2)$ is weighted by the factor k'/k = 1/3 at the energies of interest here. An alternative way of seeing this near equality is to note that the difference depends on the differential cross section integrated over all angles of scattering but weighted by $\cos \theta$, θ being the scattering angle. Thus considerable cancellation occurs between contributions from the forward and backward hemispheres, which will have a marked effect for cross sections not showing a strong forward or backward peak.

4. Summary and Conclusions

N. 3

v

We have shown that at around 50 meV both the 0-0 and 2-2 diffusion cross sections are dominated by the spherical component of the polarizability, consistent with the good agreement between the measured room-temperature

mobility, 12.4²cm /V-sec (Ellis et al 1976b) and the value obtained, 12.5 cm²/ V-sec, using the Langevin model. While the inelastic O+2 diffusion cross section is more than a factor of three larger than that calculated previously by Alexander (1976), its absolute magnitude is much too small for it to be significant for the room temperature mobility. Although $Q_d(O+2) \approx Q(O+2)$ at the energies considered here, this equality does not hold at higher energies 0.6-1.2 eV, comparing the results of Schaefer and Lester (1975) for Q(O+2) with those of Alexander (1976) for $Q_d(O+2)$. At these higher energies, the O+2 differential cross section is forward-peaked so that the cancellation which occurred at lower energies no longer occurs.

Since a converged fully quantal calculation of the diffusion cross section at the close grid of energies necessary to obtain the mobility up to the very high effective temperatures accessible in current experiments (Ellis et al 1976b) would be very expensive, it would be of interest to investigate a classical trajectory approach to these cross sections. As the Li^+-H_2 system should have one of the most accurate ab initio potential energy surfaces available for ion-molecule interactions, it could serve as a test-bed for transport properties of ions in molecular gases.

References

Alexander MH 1976 J. Chem. Phys. 64 4498-503

Arthurs AM and Dalgarno A 1960 Proc. Roy. Soc. A 256 540-51

7

Blatt JM and Biedenharn LC 1952 Rev. Mod. Phys. 24 258-72

Dickinson AS, Kouri DJ, Wells CA and Lester WA, Jr. 1976 J. Chem. Phys. <u>65</u> 1501-5

Dickinson AS and Lee MS 1978 J. Phys. B: At. Mol. Phys. 11 L377-9

Ellis HW, Pai RY, McDaniel EW, Mason EA and Viehland LA 1976a Atomic Data and Nuclear Data Tables <u>17</u> 177-210

Ellis HW, Pai RY and McDaniel EW 1976b J. Chem. Phys. 64 3492-3

Ellis HW, McDaniel EW, Albritton DL, Viehland LA, Lin SL and Mason EA 1978 Atomic Data and Nuclear Data Tables <u>22</u> 179-217

Hickman AP and Smith FT 1978 Phys. Rev. A 17 968-75

Kouri DJ and Wells CA 1974 J. Chem. Phys. 60 2296-2304

Kutzelnigg W, Staemmler V and Hoheisel C 1973 Chem. Phys. 1 27-44

Lester WA, Jr. 1976 in Dynamics of Molecular Collisions Part A, ed. WH Miller (New York: Plenum) pp.1-32

McDaniel EW and Mason EA 1973 The Mobility and Diffusion of Ions in Gases (New York: Wiley)

Maitland GC, Vesovic V and Wakeham WA 1981 Mol. Phys. 42 803-15

Monchick L and Green S 1977 J. Chem. Phys. 66 3085-93

Parker GA and Pack RT 1978 J. Chem. Phys. 68 1585-601

Schaefer J and Lester WA Jr. 1973 Chem. Phys. Lett. 20 575-80

Schaefer J and Lester WA 1975 J. Chem. Phys. 62 1913-24

Viehland LA, Lin SL and Mason EA 1981 Chem. Phys. 54 341-64

Viehland LA and Mason EA 1978 Annals of Physics 110 287-328

Wadehra JM, Cohen JS and Bardsley JN 1978 Phys. Rev. A 18 1009-11

Table 1

Non-zero values of the coefficients $C(l_1, l_2)$ introduced in eq.(2)

2₁ 2₂

С

J-1, J	$[3(J+2)(J-1)(2J+3)/(J+1)(2J+1)]^{\frac{1}{2}}$		
J+1, J	$3[3/(J+1)(2J+1)(2J+3)]^{\frac{1}{2}}$		
J+1, J+2	$[3J(J+3)(2J+1)/(J+1)(2J+3)]^{\frac{1}{2}}$		
J-2, J-1	$[3(J-1)(2J+3)/(2J-1)]^{\frac{1}{2}}$		
J, J-1	$3[2/(J+1)(2J-1)(2J+1)]^{\frac{1}{2}}$		
J, J+1	$[3J(J+2)(2J-1)(2J+5)/(J+1)(2J+3)]^{\frac{1}{2}}$		
J+2, J+1	$3[2/(J+1)(2J+3)(2J+5)]^{\frac{1}{2}}$		
J+2, J+3	$[3(J+3)(2J+1)/(2J+5)]^{\frac{1}{2}}$		

<u>Table 2</u>

Even and odd parity contributions to $Q_d(2\rightarrow 2)$ and comparison with the Langevin polarization cross section. All cross sections are in units of \mathbb{A}^2 .

E(eV)	odd	even	Q _d (2→2)	Pol. ^(a)
0.0469	473	590	1063	1068
0.0485	377	458	835	861
0.050	356	397	754	747
0.051	335	326	661	692

(a) The Langevin classical polarization cross section

Fig. 1

Plot of $Q_d(O \gg 0)$ and the even parity contribution to $Q_d(2 \rightarrow 2)$ as functions of energy. The energies at which the calculations were performed is shown thus, **e**. Also shown — — — are the results of the classical Langevin calculation using the spherical component of the polarizability (see text). The value of $Q_d(0 \rightarrow 0)$ obtained by Alexander (1976) is shown O.



Ş.,

This report was done with support from the Department of Energy. Any conclusions or opinions expressed in this report represent solely those of the author(s) and not necessarily those of The Regents of the University of California, the Lawrence Berkeley Laboratory or the Department of Energy. Reference to a company or product name does

Reference to a company or product name does not imply approval or recommendation of the product by the University of California or the U.S. Department of Energy to the exclusion of others that may be suitable.

TECHNICAL INFORMATION DEPARTMENT LAWRENCE BERKELEY LABORATORY BERKELEY, CALIFORNIA 94720 UNIVERSITY OF CALIFORNIA