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Application of Videomicroscopy to In-Situ Studies of Zinc Electrodepostion

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# **Authors**

Mc Vay, L Muller, R H Tobias, C W

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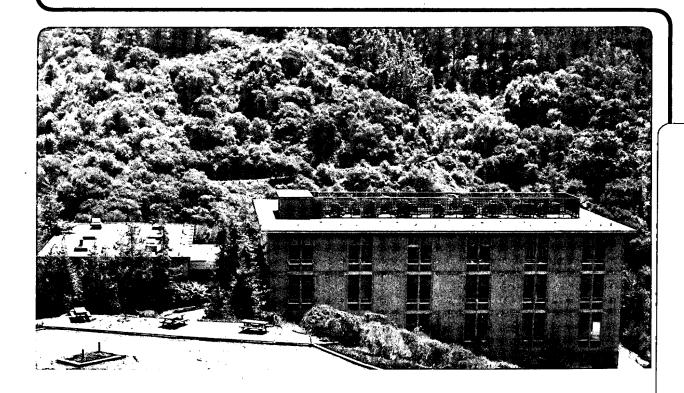
# Materials & Chemical Sciences Division

To be presented at the Electrochemical Society Fall Meeting, Hollywood, FL, October 15–20, 1989

**Application of Videomicroscopy to In-Situ Studies of Zinc Electrodepostion** 

L. Mc Vay, R.H. Muller, and C.W. Tobias

May 1989



Prepared for the U.S. Department of Energy under Contract Number DE-AC03-76SF00098.

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# APPLICATION OF VIDEOMICROSCOPY TO IN-SITU STUDIES OF ZINC ELECTRODEPOSITION

by

Laura Mc Vay, Rolf H. Muller, and Charles W. Tobias

Materials and Chemical Sciences Division Lawrence Berkeley Laboratory 1 Cyclotron Road Berkeley, Ca. 94720

May, 1989

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In electrodeposition processes, surface texture development is of considerable scientific and technological interest. Although many researchers have used ex-situ techniques to study deposit structures, these methods don't allow the uninterrupted observation of the developing morphology over the long periods of time typically required for the formation of thick deposits. In this investigation, the development of a videomicroscopy technique is reported that can be used for in-situ studies of microscopic-scale surface features.

The growth of zinc striations on a synthetic graphite substrate from flowing, well supported, acidic chloride electrolytes was observed to evaluate the capabilities of the video microscopy system. Micrographs were made at a magnification of 160X, which corresponds to a video image width of 800  $\mu$ m. The videotapes were recorded while the experiment was in progress and then edited to produce time lapse movies of the salient features of slowly occurring events. Still photographs were taken from the videotape with a Kodak SV6500 color video printer.

As illustrated in the videomicrographs of Figure 1, several steps are involved in striation formation. Nodules are first deposited randomly over the cathode surface (1a). With the passage of time, the protrusions coalesce (1b), and some agglomerates become large enough to affect the flow in the boundary layer. The eddies formed reduce the mass transport overpotential, causing the growth rate to increase in the direction parallel to the electrolyte flow (1c). As more charge is passed, the nodules coalesce to form ridges which then grow evenly in all directions.

Videomicroscopy can be used to record the progress of electrodeposition in real time. The technique shown here is currently capable of recording the deposition process at a maximum magnifications of 375X, allowing

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observation of an electrode area of approximately 300 x 200  $\mu$ m. A magnification of 600X can be achieved with a more intense light source and a higher power eyepiece. An objective lens with a long working distance has been used (50X, N. A. = 0.6). It provides a diffraction limited resolution of 0.5  $\mu$ m; the actual resolution is somewhat less because the lens is not designed for the 2 mm thick window used in the channel flow cell. Several areas of the surface can be quasi-simultaneously observed by using a traveling stage, allowing the experimenter to determine the consequences of surface events. With fiberoptic light sources, the images obtained can be of remarkably good quality.

### **ACKNOWLEDGEMENT**

This work was supported by the Assistant Secretary for Conservation and Renewable Energy, Office of Energy Storage and Distribution, Energy Storage Division of the U. S. Department of Energy, under Contract No. DE-AC03-76SF00098 with the Lawrence Berkeley Laboratory.

### REFERENCE

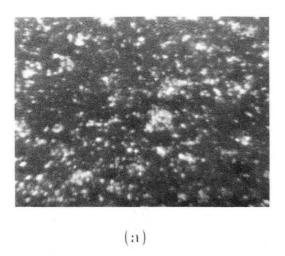
1. Mc Vay, L., R. H. Muller and C. W. Tobias, Application of Videomicroscopy to In-situ Studies of Electrodeposition, LBL report no. 26426, December (1988), Submitted to J. Electrochem. Soc.

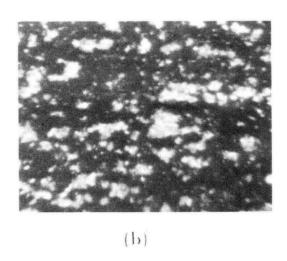
### FIGURE CAPTION

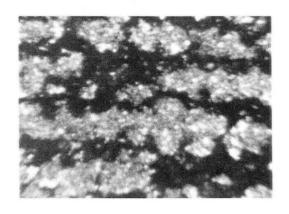
Figure 1- Genesis of a striation from a set of unaligned protrusions: (a) 1 min; (b) 3 min; (c) 5 min.

Deposition conditions: 3M ZnCl<sub>2</sub>, 3M KCl, pH = 1.5, 28.5 mA/cm<sup>2</sup>, Re = 4000, 800  $\mu$ m/image width.

The flow direction is from left to right.







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Figure 1

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