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Journal

Biogeosciences, 9(12)

ISSN

1726-4189

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Publication Date

2012-12-06

DOI

10.5194/bg-9-4979-2012

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Internal respiration of Amazon tree stems greatly exceeds external CO₂ efflux

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Received: 14 August 2012 – Published in Biogeosciences Discuss.: 27 August 2012

Revised: 4 November 2012 – Accepted: 16 November 2012 – Published: 6 December 2012

Abstract. Respiration in tree stems is an important component of forest carbon balance. The rate of CO₂ efflux from the stem has often been assumed to be a measure of stem respiration. However, recent work in temperate forests has demonstrated that stem CO₂ efflux can either overestimate or underestimate respiration rate because of emission or removal of CO₂ by transport in xylem water. Here, we studied gas exchange from stems of tropical forest trees using a new approach to better understand respiration in an ecosystem that plays a key role in the global carbon cycle. Our main questions were (1) is internal CO₂ transport important in tropical trees, and, if so, (2) does this transport result in net release of CO₂ respired in the roots at the stem, or does it cause the opposite effect of net removal of stem-respired CO₂? To answer these questions, we measured the ratio of stem CO₂ efflux to O₂ influx. This ratio, defined here as apparent respiratory quotient (ARQ), is expected to equal 1.0 if carbohydrates are the substrate for respiration, and the net transport of CO₂ in the xylem water is negligible. Using a stem chamber approach to quantifying ARQ, we found values of 0.66 ± 0.18 . These low ARQ values indicate that a large portion of respired CO₂ (~35%) is not emitted locally, and is probably transported upward in the stem. ARQ values of 0.21 ± 0.10 were found for the steady-state gas concentration within the stem, sampled by in-stem equilibration probes. These lower values may result from the proximity to the xylem water stream. In contrast, we found ARQ values of

1.00 ± 0.13 for soil respiration. Our results indicate the existence of a considerable internal flux of CO₂ in the stems of tropical trees. If the transported CO₂ is used in the canopy as a substrate for photosynthesis, it could account for up to 10% of the C fixed by the tree, and perhaps serve as a mechanism that buffers the response of the tree to changing CO₂ levels. Our results also indicate, in agreement with previous work, that the widely used CO₂ efflux approach for determining stem respiration is unreliable. We demonstrate here a field applicable approach for measuring the O₂ uptake rate, which we suggest to be a more appropriate method to estimate stem respiration rates.

1 Introduction

Respiration in tree stems is an important component of the terrestrial carbon cycle, and the emission of CO₂ from tree stems amounts to ~16% of the forest annual gross photosynthesis flux (Litton et al., 2007; Ryan et al., 1997; Waring et al., 1998). In a central Amazon forest, woody tissue respiration as estimated from stem efflux accounted for ~20–30% of total autotrophic respiration (Cavaleri et al., 2006; Chambers et al., 2004). Aerobic respiration results in the production of CO₂ and the consumption of O₂, thus measurements of both gases can be used to quantify the rate of respiration. While measurements of O₂ are technically

demanding, mainly because of the high natural background level that makes it difficult to detect changes, measuring CO₂ directly in the field (in situ) is both easy and affordable. Hence, it has become a common approach to measure the emission of CO₂ from a stem to the atmosphere as a proxy for the stem's respiration (Sprugel, 1990; Tranquillini, 1959). Lately, however, the assumption that stem CO₂ efflux provides a good measure of stem respiration has been questioned, and the question of how to correctly quantify stem respiration is still open (Teskey et al., 2008).

Any CO₂ respired by living cells inside a tree's stem has first to pass barriers to diffusion in the bark and/or the xylem before it is emitted into the atmosphere. Lenticels can probably facilitate this diffusion, but the diffusivity of stems for gases is somewhat limited (Hook et al., 1972). As a result, the concentration of CO₂ within stems builds up due to diffusive limitation. High internal CO₂ mixing ratios (up to 25 %) have been reported for a range of tree species (Teskey et al., 2008). These values are usually measured within the gas phase, which is presumed to be in equilibrium with the water that is lifted towards the crown in the transpiration stream. How much gas actually dissolves in or exsolves from the transpiration stream depends on several factors, such as pH and temperature, the concentration in the gas phase, and degree of saturation in the xylem water (Stumm and Morgan, 1995). It can be assumed that any gas that dissolves in the transpiration stream will be transported upward in the xylem, possibly all the way up to the crown. Alternatively, it will be lifted to any point within the stem where the CO₂ concentration within the gas phase is lower, and hence CO₂ from the liquid phase will escape back into the gas phase.

The degree to which CO₂ produced within the stem by respiration is emitted to the atmosphere versus transported in dissolved form likely varies and is a subject of current debate. Some findings suggest that xylem CO₂ transport is considerable (Teskey et al., 2008), and that the source of CO₂ diffusing out of the stem also includes CO₂ produced by respiration in other parts of the plant, such as roots (Aubrey and Teskey, 2009).

The main argument supporting the importance of CO₂ transport in the transpiration stream has been an observed correlation between sap flow velocity and stem CO₂ efflux in temperate trees (Levy et al., 1999; McGuire and Teskey, 2004). Recently, Kunert and Mercado Cardenas (2012) reported such a correlation for a tropical tree. Net export of CO₂ in the transpiration stream has been used to explain lower-than-expected CO₂ efflux from Eucalyptus growing in Brazil (Ryan et al., 2010) and very high rates of CO₂ efflux from canopy branches in tropical forest (Cavaleri et al., 2006). Additional support for this theory comes from experiments in which isotopically labeled carbonate was injected directly in the transpiration stream (Powers and Marshall, 2011). Apart from internal transport of tree-produced CO₂, some authors claim that CO₂ respired by soil microorgan-

isms might be taken up by roots and then transported upwards (Ford et al., 2007; Vapaavuori and Pelkonen, 1985).

In contrast, other authors have found no relationship between sap flow and stem CO₂ efflux (Maier and Clinton, 2006; Ubierna et al., 2009a). A field study in which trees were watered with isotopically labeled water and dissolved CO₂ found clear evidence for the uptake of the water, but no evidence for uptake of the dissolved CO₂ from the soil (Ubierna et al., 2009a). A possible explanation for these contrasting results might be differences in wood anatomy (e.g. diffuse porous, ring porous, and tracheid trees) or other traits among species. To summarize, we can say that the question of "To what extent does CO₂ efflux from a given portion of tree stem reflect transport versus in situ respiration?" remains unanswered.

To further add to the complexity of this situation, it has recently been reported (Berveiller and Damesin, 2008; Hibberd and Quick, 2002) that C₃ plants can use a mechanism typically associated with C₄ metabolism to take up and transport CO₂. By using the enzyme phosphoenolpyruvate carboxylase (PEPC), trees can bind CO₂ to phosphoenolpyruvate, resulting in the formation of oxalacetate which is then quickly transformed into malate that will also be dissolved and transported in the transpiration stream. By using decarboxylation enzymes, this process can be reversed, re-releasing the CO₂, presumably at a location where the CO₂ is needed (photosynthetically active tissue). It is not yet known to which extent this mechanism is used within trees, but in 9 temperate tree species tested by Berveiller and Damesin (2008), all species had significantly increased ratios of PEPC activity in stem wood compared to what is usually reported for C₃ metabolism.

It should be noted that none of the studies mentioned above were conducted in tropical forests, which are responsible for ~40 % of global terrestrial GPP (Beer et al., 2010), and many of methods to study internal carbon transport are not applicable in remote locations. Recently, Angert and Sherer (2011) demonstrated that the combined measurement of O₂ uptake in addition to CO₂ efflux can potentially separate transport from respiration fluxes, because the lower solubility of O₂ in water (28 times lower than that of CO₂ at 20 °C) should limit its contribution due to gas exchange with xylem water. In addition, in contrast to O₂, the dissolved CO₂ forms additional chemical species (bicarbonate, carbonate) by reacting with water. These reactions are pH-dependent and can increase the total inorganic carbon capacity. These differences between CO₂ and O₂ have been used at the global scale to separate the land and the ocean carbon sinks (Keeling et al., 1996). Here, we further develop the method proposed by Angert and Sherer (2011) to measure the ratio between CO₂ efflux and O₂ uptake in tropical forest tree stems, and use this information to determine if internal CO₂ transport is important in tropical trees. Moreover, this approach will allow us to determine whether this transport results in net release of CO₂ respired lower in the stem or in the roots, or

alternatively if it causes the opposite effect of net removal of stem-respired CO₂ upward towards the canopy.

This method is based on the ratio of CO₂ emission to O₂ uptake in respiration, which is known as the Respiratory Quotient (RQ). The RQ is 1.0 when carbohydrates are the substrate for respiration, but it is for example ~ 0.7 for fats and ~ 1.3 for malic acid (Stiles and Leach, 1933). Nitrate assimilation in roots can also cause RQ values above 1.0 (Bloom et al., 1989). We will define the ratio between the stem's CO₂ efflux to O₂ influx as ARQ (Apparent Respiratory Quotient). If the emissions from a stem are controlled only by respiration, then the value of ARQ will be equal to RQ and ARQ = 1.0 (assuming that carbohydrates are the main substrate for stem respiration). Alternatively, if net transport of CO₂ from outside the region where the measurement is made contributes significantly to CO₂ stem efflux (Aubrey and Teskey, 2009), we would expect to measure local ARQ > 1.0. Conversely, if there is net removal of locally produced CO₂ by dissolution in the xylem sap or fixation by PEPC, we would expect to measure ARQ < 1.0 in the part of the stem where this occurs. Here, we applied this approach to tropical forest trees. In addition to fluxes emitted from the stem surface, we also measured the concentrations of CO₂ and O₂ inside the stems and in the soil, which help to better constrain the CO₂ sources and sinks.

2 Methods

2.1 Site description

The study was carried out at the Center for Research and Forest Learning (CIEFOR) of the National University of the Peruvian Amazon (UNAP) in the community of Puerto Almenaras, which is located 16 km southwest of the city of Iquitos, Peru. CIEFOR is centered at 3°49'53.8" N, 73°22'28.2" W, encompassing a forested area of 1300 ha and managed by the Faculty of Forest Engineering-UNAP. Landforms in this area include plateaus, slopes and small valleys associated with perennial streams. The average canopy height in this area is 30 m and the most abundant taxonomic families include Lauraceae, Apocynaceae, Lecythidaceae, Fabaceae, Lauraceae, Burseraceae, Symaroubaceae, Myristicaceae, Simaroubaceae, and Annonaceae. More than 250 tree species are found in this area, and some common species include *Hymenolobium pulcherrimum* Ducke (Mari Mari), *Tachigali paniculata* Aublet (Tangarana), *Simarouba amara* Aublet (Marupa), *Euterpe precatória* C. Martius (Huasai), and *Guarea glabra* M. Vahl (Requia).

The meteorological station at CIEFOR, under the responsibility of SENAMHI (Meteorological and Hydrological National Service of Peru), reports a climatological annual rainfall of 2979 mm and maximum, average and minimum temperatures of 31.6 °C, 26.7 °C, and 21.6 °C, respectively. There is a dry season with reduced monthly averaged rain-

Table 1. Trees dimensions.

Tree	Diameter (m)	Estimated height (m)
Mari Mari 1	1.34	25
Mari Mari 2	0.39	18
Mari Mari 3	1.13	30
Marupa 1	0.33	15
Marupa 2	0.43	18
Marupa 3	0.38	16
Tangarana 1	0.71	23
Tangarana 2	0.42	20
Tangarana 3	0.88	23

fall, which usually extends from May to October. The 2010 dry season was characterized by a widespread drought in the Amazon Basin (Lewis et al., 2011; Marengo et al., 2011), and rainfall deficits were observed in this site from May 2010 to February 2011. The rainfall was especially low in August (43 mm) and September (102 mm). These values correspond to 20 % and 41 % of the climatology values for those months (www.senamhi.gob.pe).

We sampled stem CO₂/O₂ fluxes in both in the dry (27–30 September 2010) and wet (April 2011) seasons. We sampled in-stem gases twice, in October and December 2010. All the experiments in this study were conducted on a total of nine trees from the following species (three each): *Tachigali paniculata* (Tangarana) and *Hymenolobium* sp. (Mari Mari) from the Fabaceae family, with typical wood density values of 0.53 and 0.65 g mL⁻¹, respectively, and *Simarouba amara* (Marupa) from the Simaroubaceae family with wood density of 0.35 g mL⁻¹ (Chambers et al., 2004). All three species have diffuse-porous xylem anatomy. The stem chambers and in-stem probes were attached at heights of ~ 1.6 to ~ 2 m above the soil surface. Description of tree dimensions for the individuals sampled are summarized in Table 1.

2.2 Stem chambers, in-stem probes, and soil air sampling

Different stem chamber designs were used for each season. The stem chambers used for the dry season campaign are described in Angert and Sherer (2011). Each chamber was constructed from two rectangular clear Perspex parts: (1) a frame base equipped with closed-cell foam on the stem side, and (2) a lid equipped with plastic connectors for sampling and a 60 mL syringe, with its bottom part sawed-off to allow decreasing of the system's volume while taking an air sample. The total volume of the system was ~ 550 mL. The chamber was sealed to the stem by hot glue. For installation on trees with rough bark surfaces that complicated air-tight sealing, we first removed some bark, and then smoothed the surface with a file, while being careful not to damage the phloem and the cambium. In case of trees with smooth bark, we only removed loose bark and lichen before installing the

chambers. After closing the lid, we checked that the seal was air-tight by pulling the piston of the bottomless syringe attached to the lid. In all trees except “Tangarana 2” (see below), the piston returned in after releasing it, while in that tree there was only some resistance to the pull, even after adding more hot glue. Hence, it seems that air could still enter the chamber through small pores in the bark, but only at low rates. We conclude that for all trees, in the absence of strong winds (which were not present during our field campaign), the mixing between the chambers and the atmosphere was dominated by diffusion rather than by mass flow. Samples of the air in the chamber were collected in two pre-evacuated ~ 3.6 mL glass flasks with a Louwers–HapertTM O-ring valve (one for O₂ analysis and one for CO₂). Before sampling, the dead volume in the tubing and flask necks were purged with 30 mL of air from the chamber. Each chamber was sampled twice; the first sampling occurred 2–3 h after the start of the experiment (sealing of the chamber), while the second one occurred the following morning, at least 17 h later.

In the wet season sampling (April 2011), we used chambers based on the design reported in Ubierna et al. (2009b). The new chambers were custom built from polypropylene (PP) tubing material (11 cm OD). We used T-pieces (Ostendorf Kunststoffe GmbH, Vechta, Germany, HTRE DN 110) that are originally equipped with a threaded lid to close the third opening. The other two ends were welded shut with PP disks. These completely closed T-pieces were then cut longitudinally, thus removing a segment of the tube opposite to the threaded lid, resulting in an opening along the whole length of the tubing (27.2 cm) and 7.0 cm wide. The chambers were fit to the shape of the tree stem at the exact spot of installation, and initially attached by using two sets of lashing straps. To further stabilize the chambers and provide a gas-tight seal, the outline of the chamber was then glued to the stem with hot glue. As soon as the hot glue had hardened, the chambers were tested for leaks (and sealed again if necessary, until no leaks were found). Leak-testing was performed by measuring CO₂ inside the chamber and blowing respiratory air through a piece of tubing on all potentially leaky spots. Due to the high concentration of CO₂ in respiratory air, this method is both easy and highly sensitive.

The wet season chambers provided a permanently installed plastic chamber that could be closed for incubation or opened for ventilation between measurements dates. To avoid insect infestation, the chambers were covered with stainless steel mesh whenever opened for ventilation. These chambers were also used to measure the CO₂ efflux rate from the trees by attaching an Infra-Red Gas Analyzer (IRGA, LI-820 LI-COR, Lincoln NE, USA). Gas from the chamber was pumped through a water trap filled with Drierite into the IRGA at a constant flow rate of ~ 600 mL min⁻¹, and then pumped back into the chamber (closed dynamic chamber measurements; Pumpanen et al., 2004). Data was logged on a portable computer, using the Li-820 software. A linear

regression was performed on the [CO₂] data as a function of time to determine a flux rate, which was corrected for atmospheric pressure and temperature.

Using a modified lid, four flasks could be connected simultaneously to each chamber. Leaving the flasks' valves open allowed for CO₂ and O₂ to diffuse in and out from the chamber. Two 3.6 mL flasks (one for O₂ analysis and one for CO₂) were closed ~ 6 h after the beginning of the experiment, and then removed. The second pair of 3.6 mL flasks was closed 10 days after the experiment had begun.

The probes used to sample in-stem air were inspired by the design of Ubierna et al. (2009b). This approach is based on drilling a 6 cm deep hole into the stem after the removal of cracked bark, and hammering in a stainless steel tube (the probe), with an outer diameter that slightly exceeds that of the hole (we have used a 6 mm drill bit and a 1/4" OD tube). The probe was then connected to a flask filled with air and left for several days to weeks to equilibrate with the gases inside the stem. In our design, the equilibrium volume was two 3.6 mL sampling flasks which were connected to the probe by a plastic T connector and rubber tubing (10 ID mm 20 OD mm). This approach is simpler than the original design of Ubierna et al. (2009b), which required injecting acidified water into the equilibration volume, while collecting the sample. In the current design, sampling is simply done by closing the valves of the flasks and removing them from the probe.

Since the probe installation could potentially invoke a wound response, we installed the probes only after the chamber measurements ended, and left the equilibrating flasks connected to trees for an extra 2 weeks in the probe sampling time “A”. The flasks were closed and removed on 29 October. Probe sampling time “B” started on the same day, and ended on 6 December, allowing ample time for the air in the flasks to equilibrate.

Soil air was sampled, during the dry season, using a stainless steel tube (10 mm ID, 12.5 mm OD) that was hammered into the soil, as in Angert et al. (2001). The tube end was pointed to ensure easy insertion, and 2-mm-diameter holes were drilled above the pointed end for soil air collection. An 8-mm-diameter plastic rod inserted inside the tube reduced its dead volume. For the sampling of soil air, we used the same connectors and flasks used to sample the dry-season chambers.

2.3 Analytical methods

The CO₂ concentrations in the 3.6 mL flasks were measured in the lab by an IRGA (Li-840A) connected to a circulating system, as described in Angert and Sherer (2011). In addition to the measurements of CO₂ in flasks sent to the lab, we also conducted measurements in the field during the dry season campaign. In these chamber experiments, 10 mL of chamber air were sampled into a 60 mL syringe, containing 50 mL of CO₂-free air. The diluted sample was then immediately

introduced to the IRGA. To determine CO₂ concentrations in the soil air, we connected the IRGA by a three-way valve to the syringe used for flushing the soil tube and the flasks' necks. Good agreement was found between the measurements made in the lab and in the field. The relative error of the [CO₂] field measurements, based on the difference between duplicates, averaged 2.5 %, which is similar to that achieved in measuring the flasks in the lab. The oxygen concentrations were calculated from the O₂/Ar ratio (expressed as δO₂/Ar) determined by mass spectrometric analysis, under the assumption of constant Ar concentration. Sample preparation and mass spectrometry were according to Barkan and Luz (2003), which gives an accuracy of 0.02 % in O₂ concentrations (which translates to a relative accuracy of ~ 0.1 %).

3 The models for stem and chamber gases

3.1 Analytical 1-box model

Estimating the ratio between the stem's efflux of CO₂ and influx of O₂ from the concentration measurements requires some simple modeling. Our 1-box analytical model follows the one we have presented earlier (Angert and Sherer, 2011). In this model, a box represents the chamber and the top layer of the stem, which are assumed to have the same gas concentrations and to be in steady state on the timescale integrated by our sampling. For CO₂ this steady state results from a balance between CO₂ emitted from deeper layers of the stem to this box, and the CO₂ that diffuses out of the box to the atmosphere (Fig. 1). This balance can be described by the following equation:

$$E_C = g_C \Delta_C \quad (1)$$

where E_C is the CO₂ efflux to the box ($\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$), g_C is CO₂ conductance in the wood between the box and the atmosphere ($\mu\text{mol m}^{-2} \text{ s}^{-1}$), and Δ_C is the difference between the box and the ambient CO₂ molar fractions ($\mu\text{mol mol}^{-1}$).

We can write similar equation for O₂ by replacing the subscript "C" with "O" (here both E_O and Δ_O are negative):

$$E_O = g_O \Delta_O \quad (2)$$

Dividing Eq. (1) by Eq. (2) yields:

$$\frac{E_C}{E_O} = \frac{g_C}{g_O} \frac{\Delta_C}{\Delta_O} \quad (3)$$

The term $(-E_C/E_O)$ is, by definition (see introduction), equal to ARQ, and is controlled by both the stem's RQ, and by processes that remove or import CO₂ to the portion of the stem sampled by the chamber.

The conductance (g) for each gas depends on its diffusivity in air, on the length and area of the conducting wood section, and on the structure of the air-filled pore spaces (Millington and Shearer, 1971). In most trees, lenticels in the stem

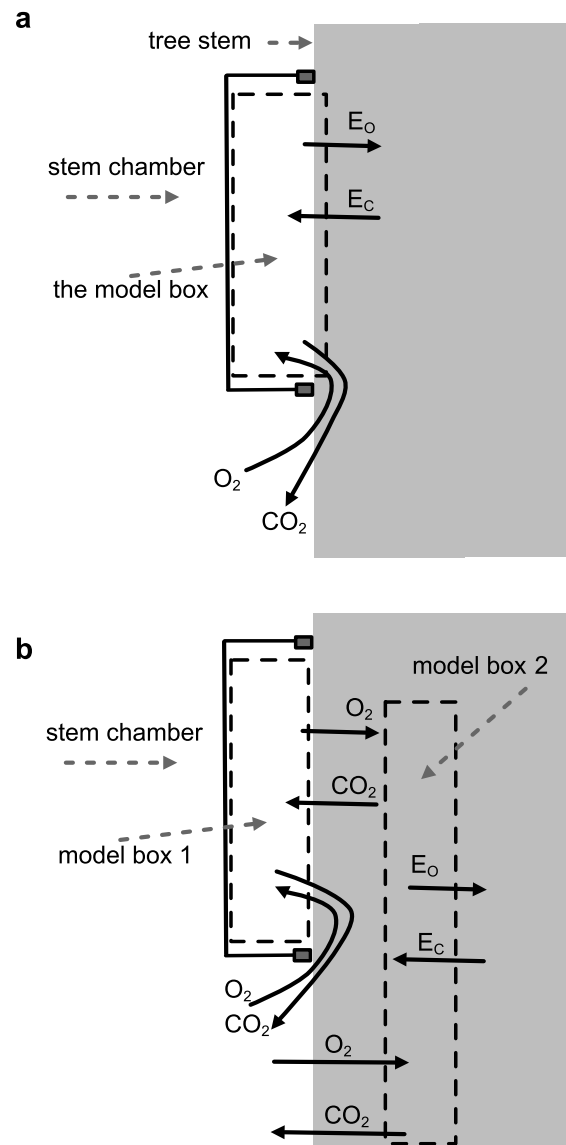


Fig. 1. Schematic drawing of the model boxes and fluxes: (a) 1-box model, (b) 2-box model. The O₂ and CO₂ diffusive fluxes, as well as the fluxes of O₂ consumption (E_O) and CO₂ release (E_C) are shown by solid lines.

provide the necessary aeration pathways (Hook et al., 1972). This structure determines the length of the conducting elements, which largely control the effective diffusivity in tree stems (Soriz and Hietz, 2006). Since the structure, the length, and the area, are identical for both gases, the ratio g_C/g_O in Eq. (1) is controlled only by the ratio of diffusivity of the two gases in air. For CO₂/O₂ this ratio of diffusivity in air is 0.76 ($0.138 \text{ cm}^2 \text{ s}^{-1}/0.182 \text{ cm}^2 \text{ s}^{-1}$ at STP) and is independent of temperature, since for different temperatures both diffusivity coefficients will change by the same factor (Massman, 1998). Thus, Eq. (1) becomes

$$\text{ARQ} = -0.76(\Delta_C/\Delta_O) \quad (4)$$

For calculating Δ_C and Δ_O we have assumed that ambient concentrations are identical to the atmospheric ones. This approximation will have negligible effects, e.g. an error of the order of $50 \mu\text{mol CO}_2 \text{ mol}^{-1}$ in an overall difference of several thousand to tens of thousands of $50 \mu\text{mol CO}_2 \text{ mol}^{-1}$.

3.2 Numerical 1-box model

To deal with the non-steady-state cases, we developed a simple numerical model. This 1-box numerical model is based on the same box and flux definitions as the analytical model above. The main difference is that no steady state is assumed between the box and the atmosphere. The model is initialized to start with atmospheric air in the box, and the changes in the O₂ and CO₂ are solved by a finite-differences approach. An example of a model run with a RQ = 1.0 (and arbitrary respiration and conductance values) is presented in Fig. 2.

After this time (~ 10 h in the model run pictured in Fig. 2) the gas concentrations approach their steady-state values. The time required to achieve steady state is a function of the value chosen for conductance. However, previous experiments with the same stem chambers on trees have shown that steady state is achieved in \sim half day (Angert and Sherer, 2011).

As predicted by the analytical model, the value of $-0.76(\Delta_C/\Delta_O)$ approaches 1.0 as the model approaches steady state. In contrast, at the beginning of the run before there was sufficient time to establish a diffusive steady state, this value is 0.76, so the value of $-\Delta_C/\Delta_O$ is 1.0 (i.e. it is equal to the ratio of E_C to E_O , which for the purpose of the model was set to be 1.0). Thus, in our chamber experiments the ARQ can be estimated from the value of $-0.76(\Delta_C/\Delta_O)$ for samples taken when the system is close to steady state, and from the value of $-\Delta_C/\Delta_O$ for samples taken shortly after sealing the chambers. It should be noted that ARQ is exactly equal to $-\Delta_C/\Delta_O$ only immediately after sealing of the chambers, when the [O₂] changes are not measurable. However, as illustrated by Fig. 2, sampling a few hours after sealing of the chamber will result in an error in estimating ARQ from $-\Delta_C/\Delta_O$ which is of the same order as the analytical uncertainty. One goal in this work was to compare results when sampling during non-steady-state and steady-state cases.

In order to investigate how the simplification of a 1-box model affects our estimate of the ARQ, and to acknowledge issues associated with slow diffusion of air deeper in the stem, we have also developed a 2-box numerical model (Fig. 1b). In this model, one box represents the air in the stem, while the other box represents the chamber air. The chamber air exchanges O₂ and CO₂ with the stem by diffusion, while both the stem and the chamber exchange gases with the atmosphere. The stem box also loses O₂ and gains CO₂ from respiration. In this model run, we allowed the stem to achieve steady state in terms of O₂ and CO₂ concentrations, while the chamber was assumed to be venti-

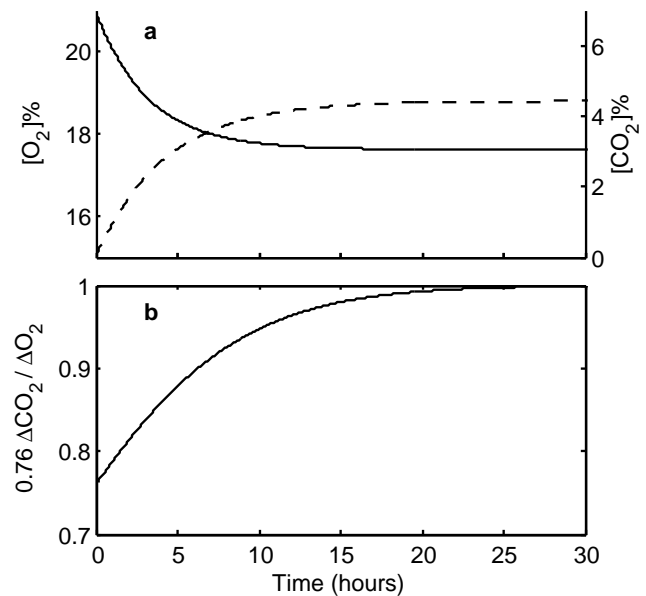


Fig. 2. (a) Modeled O₂ (solid line) and CO₂ (dashed line) concentrations, and (b) the relationship between the increase in CO₂ and the decrease in O₂ (multiplied by the ratio of these gases diffusivities in air, which is 0.76), as produced by a 1-box numerical model.

lated with atmospheric air (“open”). After the stem achieved steady state, the chamber was “sealed”, to simulate a chamber experiment in the field. The resulting increase in CO₂ and decrease in O₂ were similar to the results of the 1-box model, and again the value of $-\Delta_C/\Delta_O$ was 1.0 at the beginning of the simulated experiment, while later the value of $-0.76(\Delta_C/\Delta_O)$ approached 1.0.

Additional complexity can be introduced if the rate of oxygen uptake by the stem exceeds the rate of CO₂ emission. This would be accompanied by a reduction in pressure with a consequent mass flow of atmospheric air into the box. This will not only bring in O₂, but also N₂ and Ar. As a result, N₂ and Ar concentrations in the stem will increase, and the O₂/Ar ratio will not simply indicate O₂ concentrations. Further, the assumption of transport only by diffusion will not be valid in such a situation. However, including this effect of pressure induced mass flow in the model has shown that even in an extreme case with RQ = 0.5 and O₂ decreased to less than 15 %, the effect on the estimated value of ARQ is of the order of only few percent. This small effect results from the fact that the diffusion of Ar and N₂ tends to cancel the concentration gradients, and because the effect on the O₂/Ar ratio tends to cancel the effect of non-pure diffusion transport.

We have also considered the “water vapor flux fractionation effect”, which is driven by the diffusion of water vapor (Severinghaus et al., 1996). If a gradient in the concentration of water vapor is present in the outer bark layer, this effect can have some impact on the gas concentrations. However,

even if we assume that water vapor concentration changes from 4 % to 1 % within the bark, it will only change the CO₂/O₂ ratio by ~ 1 % and the O₂/Ar by 0.15 %. Thus, even under extreme water vapor gradients this effect will not have a measurable impact on our results.

3.3 Calculating O₂ respiration rates

The respiration rate can be calculated from the change in the [O₂] in the chamber headspace with time. The simplest approach is to use the first sampling time, and assume that [O₂] linearly decreased with time. This approach, which we call a “1-point approach”, neglects the increased diffusion of O₂ into the chamber as the concentration inside drops. Another approach is to solve the following differential equation, which describes the changes in [O₂]:

$$\frac{d[\text{O}_2]}{dt} = g_{\text{O}}\Delta_{\text{O}} - R \quad (5)$$

The solution of this equation is

$$R = -V(O_0 - O_{\text{ss}}) \ln\left(\frac{O_t - O_{\text{ss}}}{O_0 - O_{\text{ss}}}\right) / t \quad (6)$$

where R is the respiration rate (O₂ consumption rate in L O₂ s⁻¹), V is the chamber volume (L), O_0 is the atmospheric concentration (20.95 %), O_{ss} is the concentration in steady state (which is assumed to be the concentration of the second sampling), and O_t is the concentration at time t , which will be the time of the first sampling. The built-in assumption of this approach, which we call a “2-point approach”, is that respiration rate does not vary with time.

4 Results

4.1 Soil air measurements

The CO₂ concentrations in the soil air ranged from 0.57 % to 2.60 %, while the O₂ concentrations ranged from 19.22 % to 20.57 % (Table 2). The ARQ can be estimated from $-0.76(\Delta_{\text{C}}/\Delta_{\text{O}})$ and varied from 0.83 to 1.14, with an average of 1.00.

4.2 Stem chamber measurements

In the dry season, 2–3 h after the chambers were installed, the CO₂ concentrations were in the range of 0.77–2.51 %. After ~ 20 h, they had increased to 1.45–8.67 % (Table 3). The O₂ concentrations were in the range of 16.22–19.68 % (after 2–3 h) and 7.91–18.41 % after ~ 20 h. The ARQ ratio, which we estimated as $-\Delta_{\text{C}}/\Delta_{\text{O}}$ for the first sampling time, when the chamber headspace was far from steady state, ranged from 0.51–0.93. The ARQ estimated from $-0.76(\Delta_{\text{C}}/\Delta_{\text{O}})$ at the second sampling time (~ 20 h), when we assumed the chamber headspace air was at steady state, ranged from 0.23–0.89. The average estimates of ARQ agreed well between the

Table 2. Results of soil air sampling in the dry season. The analytical error in ARQ estimate is 0.05 on average.

Sampled next to tree	Soil [CO ₂] %	Soil [O ₂] %	ARQ
Mari Mari 1	n.a.	20.05	n.a.
Mari Mari 2	n.a.	n.a.	n.a.
Mari Mari 3	1.35	19.92	0.99
Marupa 1	1.09	20.03	0.90
Marupa 2	0.6	n.a.	n.a.
Marupa 3	0.73	20.28	0.83
Tangarana 1	n.a.	20.36	n.a.
Tangarana 2	0.57	20.57	1.14
Tangarana 3	2.6	19.22	1.14
Average			1.00
std dev			0.18

two calculation methods, 0.65 using the non-steady-state approach (first sampling) and 0.57 using the steady-state approach (second sampling) (Table 3).

In the wet season, the chamber headspace had CO₂ concentrations ranging from 0.30–1.49 % after 6–8 h (first sampling), which increased to 1.23–4.26 % by the time of the second sampling (10 days). The corresponding O₂ concentrations were 18.63–20.71 % and 16.15–19.57 %, respectively. The ARQ estimated from $-\Delta_{\text{C}}/\Delta_{\text{O}}$ for the first sampling ranged from 0.47–1.09, again in good agreement with those calculated assuming steady state ($-0.76\Delta_{\text{C}}/\Delta_{\text{O}}$) at the second sampling, 0.47–0.90. The average estimate of ARQ was 0.73 in the first sampling and 0.69 in the second (Table 3).

4.3 In-stem probe measurements

The CO₂ concentrations for air sampled by the in-stem probes ranged from 1.12–9.68 % for the month following the dry season chamber sampling (A in Table 4), and increased slightly in the following month 1.40–10.4 % (B in Table 4). The corresponding O₂ concentrations decreased from 3.53–18.06 % (A) to 0.0–17.98 % (B). The ARQ values estimated for sampling period A and assuming steady state (i.e. ARQ = $-0.76\Delta_{\text{C}}/\Delta_{\text{O}}$) ranged from 0.13–0.88, in good agreement with the subsequent sampling period B with ARQ of 0.14–0.77. Excluding tree “Marupa 3”, which gave constantly higher values, the average ARQ was 0.29 in experiment A and 0.17 in experiment B. Some of the equilibrating flasks filled with water which prevented measurements of air O₂ and CO₂, but allowed for measurements of water pH (using pH strips with ±0.5 units resolution). The pH of the xylem sap was 4.5 for trees “Tangarana 2” and “Mari Mari 3”, and 7.0 for trees “Mari Mari 2” and “Tangarana 3”.

Table 3. Results of the stem chambers experiments. First sampling: 2–3 h after the start of the experiment in the dry season, 6–8 h in the wet season. Second sampling: ~20 h in the dry season, 10 days in the wet season. The ARQ values are calculated in the two ways explained in the text. The analytical error in ARQ estimate is 0.02, on average, in the dry season and 0.03 in the wet season.

Tree	Season	First sampling [CO ₂] %	First sampling [O ₂] %	ARQ	Second sampling [CO ₂] %	Second sampling [O ₂] %	ARQ
Mari Mari 1	dry	0.77	19.51	0.51	5.75	16.05	0.89
Mari Mari 2	dry	1.13	n.a.	n.a.	1.50	n.a.	n.a.
Mari Mari 3	dry	1.34	19.55	0.93	1.45	18.41	0.42
Marupa 1	dry	n.a.	19.68	n.a.	2.56	12.62	0.23
Marupa 2	dry	0.81	n.a.	n.a.	2.80	15.61	0.40
Marupa 3	dry	1.36	19.06	0.70	4.83	16.07	0.75
Tangarana 1	dry	2.51	16.22	0.52	8.67	7.91	0.51
Tangarana 2	dry	1.14	19.16	0.61	5.87	12.59	0.53
Tangarana 3	dry	2.21	17.42	0.62	7.79	13.55	0.80
Dry Season average				0.65			0.57
std dev				0.15			0.23
Mari Mari 1	wet	0.54	20.16	0.63	1.40	19.55	0.74
Mari Mari 2	wet	0.55	19.87	0.47	1.23	19.57	0.66
Mari Mari 3	wet	0.63	20.08	0.68	2.06	18.83	0.73
Marupa 1	wet	1.16	19.86	1.03	4.26	17.37	0.90
Marupa 2	wet	0.62	20.12	0.70	3.86	16.76	0.69
Marupa 3	wet	0.30	20.71	1.09	2.46	18.36	0.71
Tangarana 1	wet	1.16	19.47	0.76	1.40	19.48	0.71
Tangarana 2	wet	0.69	19.86	0.60	3.01	16.15	0.47
Tangarana 3	wet	1.49	18.63	0.63	2.66	17.70	0.61
Wet season average				0.73			0.69
std dev				0.20			0.11

5 Discussion

5.1 Explaining the measured ARQ values

We found average ARQ values (\pm standard deviation) of 1.00 ± 0.13 for the soil pore space but considerably lower values in all measurements for tree stems: 0.21 ± 0.10 in the air sampled by in-stem probes (excluding tree “Mari Mari 3”), and 0.66 ± 0.18 in the chamber experiments (Fig. 3). These low ARQ values were found in all tree species studied, and in both the dry and wet season. The variability in ARQ is higher than expected from propagation of the analytical uncertainties for [CO₂] and [O₂] measurements, which result in uncertainties of 0.05 in the value of ARQ for the soil air pore space samples, 0.02–0.03 for the chamber headspace gases, and 0.01 for the gases sampled within stem probes. Thus, the observed variability could represent (1) differences in the RQ of the substrates used for respiration, (2) small deviations from steady state, or (3) processes other than respiration that affect [CO₂] and [O₂] differently. The low ARQ values, which are considerably below the expected value of 1.0, need to be explained.

First, the low ARQ values may be the result of some artifact associated with the assumptions we used in the calculations of ARQ. However, results for samples taken many

hours after the chamber headspace was isolated, and therefore including the 0.76 factor associated with the assumption of steady-state gas-phase diffusion, yielded similar ARQ values to that of samples taken a few hours after chamber closure, where no diffusion correction was necessary ($ARQ = -\Delta_C/\Delta_O$). We thus conclude that the finding of low ARQ values (0.66 averaged across all chamber measurements and all trees) is robust. Further confidence that we have no systematic sampling or analytical errors comes from the soil pore space samples, which give the expected ARQ of close to 1.0.

A second possibility is that the low ARQ values actually represent low RQ values. However, in order to explain RQ values of around 0.6 this way, fats would have to be the main substrate for respiration in both the dry and wet seasons, which is extremely unlikely. Moreover, ARQ values of around 0.5, as found for some of our trees in the stem chambers experiments, and 0.3, as found in the in-stem probe experiments, cannot be explained by the known range of substrate dependent RQ. Corticular photosynthesis will not cause deviations in the ARQ since the photosynthetic exchange of CO₂ and O₂ is with a ratio of about 1 : 1, and photorespiration will be inhibited by the high CO₂ concentration

Table 4. Results of in-stem probe sampling. During sampling period A the gas samplers attached to the probes were opened on 30 September and closed on 29 October 2011. In sampling period B, the gas flasks were attached and opened on 29 October and closed on 6 December 2011. The analytical error in ARQ estimate is 0.01.

Tree	Period	[CO ₂] %	[O ₂] %	ARQ
Mari Mari 1	A	n.a.	3.53	n.a.
Mari Mari 2	A	9.68	n.a.	n.a.
Mari Mari 3	A	n.a.	n.a.	n.a.
Marupa 1	A	2.89	3.66	0.13
Marupa 2	A	1.12	18.06	0.29
Marupa 3	A	3.37	18.05	0.88
Tangarana 1	A	8.29	6.42	0.43
Tangarana 2	A	5.03	n.a.	n.a.
Tangarana 3	A	8.12	n.a.	n.a.
Mari Mari 1	B	10.4	n.a.	n.a.
Mari Mari 2	B	5	0.47	0.19
Mari Mari 3	B	n.a.	0.13	n.a.
Marupa 1	B	2.8	7.45	0.16
Marupa 2	B	1.4	14.6	0.17
Marupa 3	B	3	17.98	0.77
Tangarana 1	B	2.6	10.33	0.18
Tangarana 2	B	3.8	0	0.14
Tangarana 3	B	5	0.01	0.18
Average (excluding “Marupa 3”)				0.21
std dev				0.10

in the stem. As a result, we conclude that the low ARQ values are not the result of low RQ.

A third possibility is that some of the O₂ uptake is not driven by respiration but by dissolution in the xylem water. However, even if we assume that the water arrives at the base of the stem with no O₂, this water could take up dissolved O₂ only up to a concentration in equilibrium with atmospheric air (~0.25 mmol L⁻¹). For a tree with a stem diameter of 0.5 m, a respiration rate of 200 mg C m⁻² h⁻¹, and a xylem water flux of 500 L per day, this removal of O₂ by dissolution will amount to only 10 % of the O₂ consumed by respiration up to the height of 1 m. However, above this height the water will be saturated and will not be able to take up more O₂. A more reasonable assumption is that the water O₂ was equilibrated in the roots, and hence was close to equilibration with the stem O₂ and can take up even smaller amounts of O₂. Some studies have suggested that the transpiration stream can provide O₂ to the sapwood (Eklund, 2000; Gansert, 2003). However, given the low solubility of O₂ in respect to respiration rates (as discussed above), this flux of O₂ is small, and thus can be important only in areas where respiration rates are extremely low and atmospheric O₂ diffusion is restricted. Moreover, a considerable supply of dissolved O₂ to the stem through this process will result in ARQ > 1, which is inconsistent with the observations we present here.

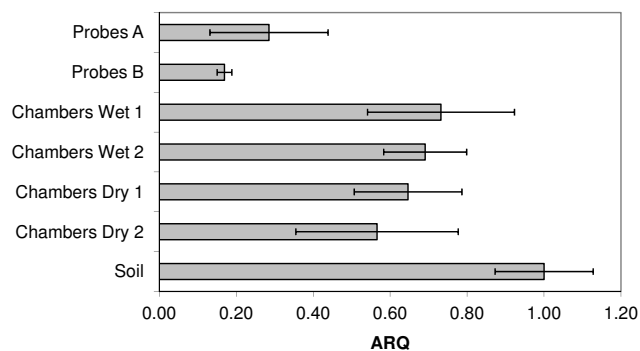


Fig. 3. The average ARQ for the gases sampled from in-stem probes, chamber headspace, and soil pore space, averaged over all trees. The error bars represent one standard deviation. The average for in-stem probes does not include tree “Marupa 3”, which gave a value of 0.88 in sampling period A and 0.77 in sampling period B (Table 4).

We conclude that O₂ dissolution and transport is not a large contributor to the fluxes we observed.

The fourth possibility, which is the only one left after rejecting the other three, is that a large portion of the respired CO₂ is removed from the stem sections studied, as was shown by previous studies in temperate forests (Teskey et al., 2008). It was also suggested (Teskey and McGuire, 2002, 2007) that storage and transport of dissolved CO₂ and bicarbonate in the xylem play an important role in controlling the CO₂ efflux from the stem. In our case, the storage option is ruled out, since storage effects would be averaged out in our longer experiments, like the 10 day chamber deployment, and the stem probe flasks that were left on for several weeks.

Given the high concentrations of CO₂ we measured in the chambers, it can be expected that a substantial amount of CO₂ will be dissolved in the xylem water. Hence, it seems plausible that xylem transport of DIC (Dissolved Inorganic Carbon) upward from the chambers, can at least partly explain our results. To estimate whether the xylem transport of DIC upward from the chambers is sufficient to explain the observed ARQ values, we performed a simple calculation based on the following assumptions: first, we assumed that water arriving at the base of the stem contains no dissolved CO₂. Second, we used the steady-state CO₂ concentrations in the dry-season chamber experiment to calculate the concentration of CO₂ dissolved in the xylem water, assuming the two are in equilibrium according to Henry’s law. In a first approximation, we ignored the possibility of DIC being present in other forms but dissolved CO₂ (a reasonable assumption for a tree with a xylem pH below 5.0).

Based on the average observed ARQ of ~0.6, we concluded that our measurements missed ~35 % of the CO₂ produced by in situ respiration, so respiration rates could be used to quantify the actual amount of “missing” CO₂. Based

on the above, explaining the missing amount of CO₂ with transport of dissolved CO₂ only (pH < 5.0) requires a flux of water in the range of 420 to 5590 L day⁻¹ for the 9 trees studied, which is higher than reported transpiration rates for tropical trees with comparable sizes (Jordan and Kline, 1977). For trees with a pH below 5.0 (trees #4 and #9), this issue remains unsolved, indicating that the transport of dissolved CO₂ only is not sufficient to explain the missing CO₂. For trees with a higher pH, additional C can be transported in the form of bicarbonate, e.g. at a sap pH of 7.0, the total DIC concentration can be ~ 6 times of the dissolved CO₂ concentration, which will bring the sap flow required to explain the ARQ values within the range of observed values. However, the above calculation generally overestimates the DIC transport capacity in the xylem, as it was made under the unrealistic assumption that no root-respired CO₂ was transported upward in the xylem water. Thus, while DIC transport in the xylem water is very likely to carry a considerable amount of respired CO₂ with it, it may fall short of fully explaining our observations. Conversion of CO₂ into malate by PEPC and a transport of this malate in the xylem water represents another possible pathway for transporting CO₂ away from the site of respiration (Berveiller and Damesin, 2008; Hibberd and Quick, 2002).

It has been suggested (Aubrey and Teskey, 2009) that in some tree species part of the CO₂ emitted by tree stems may originate from respiration taking place in the roots, and imported to the stem by transport in the xylem water. Aubrey and Teskey (2009) suggest that such transport could result in underestimation of root respiration from studies that use soil respiration measurements, since some of the root-respired CO₂ bypasses the soil and is instead emitted from tree stems. Such transport can also potentially bias CO₂-based stem respiration measurements, and cause the CO₂ efflux to exceed the local respiration rate. However, these should result in ARQ values of stems that are above 1.0 and, inconsistent with our observations, which point to net removal of stem CO₂ in the trees we studied.

The in-stem gas measurements present an additional challenge. The ARQ values found in these experiments were especially low (averaging 0.21 versus 0.66 for the chamber headspace samples). In general, we saw no big differences between tree species in our study. However, one individual tree (“Marupa 3”) had higher ARQ values in both probe experiments (0.77 and 0.88). The other two Marupa trees had low ARQ values in the stem gases sampled with probes, but did show high values of ARQ (0.90–1.09) in some of the chamber headspace samples. Thus, it is possible that this species, which belongs to a different family than the other two studied species, has different characteristics related to the “removal” of CO₂.

In any case, a more important question is why are most of the ARQ values obtained for in-stem gases so much lower than those sampled using chambers at the stem surface? One possible answer is that the O₂ and CO₂ pools in the deeper

parts of the stem have poor contact with the outer parts. As a result, the processes taking place in those deeper layers and sampled by the probes, have only a small impact on the stem’s CO₂ efflux and O₂ influx, which are sampled by the chambers. Alternatively, the lower ARQ deeper in the stem may be the result of the proximity to the xylem water, which removes the CO₂ and locally reduces the ARQ.

5.2 Implications of the low ARQ values

The respired CO₂ that is not emitted from the stem is most probably transported upward in the transpiration stream, as either dissolved CO₂ or bicarbonate, and maybe as malate as well. This CO₂ can be emitted to the atmosphere higher in the stem, or in the canopy. While a shift in the height at which CO₂ is emitted from the stem will not affect the entire tree carbon balance, this shift is important for the following reasons. First, measurements of stem CO₂ efflux are normally made at ~ 1.3 m height and then extrapolated to the surface area of the rest of the tree. The upward transport of CO₂ we report would create a systematic error in such extrapolated stem respiration rates. Indeed, a recent study (Cavaleri et al., 2006) in a wet tropical forest found that most of the CO₂ efflux came from the smallest branches (perhaps from out-gassing of transported CO₂). Such a systematic error implies that our models will fail to predict the response of trees to novel conditions, like climate change and increasing CO₂.

Moreover, at least part of the CO₂ transported up the stem could potentially be re-fixed by photosynthesis. Given the high concentration of CO₂ within stems, which reaches above 8.5 % in the current study in contrast to the 0.039 % currently in the atmosphere, and given the low affinity of RuBisCO to CO₂, there is a clear advantage for plants to use this internal CO₂ for photosynthesis. Indeed, there is evidence for such re-fixing or internal recycling of carbon. For example, Hibberd and Quick (2002) found that ¹⁴C labeled bicarbonate and malate added to the xylem water of celery and tobacco were fixed in bundle-sheath cells in the stem and leaves. Similarly, Stringer and Kimmerer (1993) found that excised leaves of *Populus deltoides* trees which transpired ¹⁴C labeled bicarbonate, fixed 99.6 % of the label. Similar ¹³C labeling of the bicarbonate in water transpired by *Platanus occidentalis* L. branches showed that 35 % of the label was fixed (McGuire et al., 2009). Recently, Powers and Marshall (2011) showed that introducing ¹³C labeled bicarbonate to tree xylem resulted with the label appearing in the phloem contents within a few days. This experiment provides a clever demonstration that such an internal carbon recycling mechanism is active in trees. Bloemen et al. (2012) used a similar labeling experiment to show that a significant proportion of the label was respired in the canopy, with some fixation in canopy branches, petioles and, to a lesser extent, leaves. However, the question of what is the relative contribution of this mechanism to overall forest primary productivity is still open. Based on our results, we can only estimate the maximal

possible contribution of this process to canopy photosynthesis.

When assuming that (1) the actual RQ value for respiration in the stem is 1.0, and (2) that [O₂] is affected only by respiration, we can use measurements of [O₂] to estimate how much CO₂ is produced locally by respiration. This approach suggests that, on average, about 35 % of the respired CO₂ has to be taken up by the transpiration stream to explain our observed ARQ values. This estimate is in accord with previous estimates based on other methods (McGuire and Teskey, 2004), and with previous implementation of the current method on different trees species (Angert and Sherer, 2011). If this estimate holds true for the entire stem and if all of the transported CO₂ is re-fixed in the canopy, and based on the estimate that the emission of CO₂ from tree stems amounts to ~ 16 % of the gross photosynthesis flux (Litton et al., 2007; Ryan et al., 1997; Waring et al., 1998), carbon originally respired within the tree stem could at the very maximum contribute ~ 10 % to a tree's gross productivity.

Internal CO₂ recycling would be even higher if root-respired CO₂ also enters the transpiration stream (Aubrey and Teskey, 2009). Assuming that 35 % (i.e. the same proportion as for stem-respired CO₂) of the root-respired CO₂ is also transported to the canopy to be re-fixed, then the internal transport and re-fixing mechanism can contribute to a maximum of ~ 20 % of the tree productivity. The re-fixing of internal carbon is insensitive to the atmospheric CO₂ concentration, and hence, will show no “carbon fertilization” effect. In addition, this re-fixation is expected to increase the plant resilience to drought, since partial closure of the stomata, which lowers the leaves internal CO₂ concentration and slows down photosynthesis, has only limited effect on the xylem CO₂ concentration (Teskey et al., 2008).

5.3 Estimating stem respiration from the consumption of O₂

Regardless of the nature of the processes that remove CO₂ from the stem, our results agree with previous studies showing that the CO₂ efflux does not correctly capture the stem respiration rate. This is not only because a fraction of the CO₂ respired is not emitted, but also because this fraction is highly variable and covers the range of 0–50 % of the respired CO₂. Understanding the processes that control stem respiration is thus impossible to build based solely on CO₂ efflux measurements. We suggest here that measurements of the O₂ influx are a better indicator of stem respiration in the field, as was previously suggested for respiration measurements in the lab (e.g. Davey et al., 2004). Such measurements may also be influenced by O₂ transport in the xylem, but to a much lesser extent compared to CO₂.

Calculated respiration rates based on the two approaches discussed in Sect. 3.3 are presented in Table 5. As expected, the respiration estimated using the 2-point approach is higher than the one calculated by the 1-point approach. The 2-point

Table 5. Stem respiration rate (*R*) based on O₂ (1-point and 2-point approaches) and CO₂ efflux (dynamic chamber). The units for all columns are μmol m⁻² s⁻¹.

Species	Season	<i>R</i> (1-point)	<i>R</i> (2-points)	CO ₂ efflux
Mari Mari 1	dry	5.9	7.0	6.3
Mari Mari 2	dry	n.a.	n.a.	3.5
Mari Mari 3	dry	3.6	5.0	2.7
Marupa 1	dry	3.5	3.8	3.1
Marupa 2	dry	n.a.	n.a.	2.8
Marupa 3	dry	n.a.	n.a.	2.5
Tangarana 1	dry	17.7	21.9	7.7
Tangarana 2	dry	n.a.	n.a.	2.2
Tangarana 3	dry	7.6	10.5	5.7
Mari Mari 1	wet	1.9	2.8	n.a.
Mari Mari 2	wet	2.7	5.3	n.a.
Mari Mari 3	wet	2.1	2.8	n.a.
Marupa 1	wet	2.6	3.1	n.a.
Marupa 2	wet	2.1	2.3	n.a.
Marupa 3	wet	0.6	0.6	n.a.
Tangarana 1	wet	n.a.	n.a.	n.a.
Tangarana 2	wet	2.6	3.0	n.a.
Tangarana 3	wet	5.9	10.3	n.a.

respiration is also higher than the CO₂ efflux estimated by monitoring the CO₂ concentration change in the headspace of the chamber. This result agrees with our finding of ARQ values below 1.0 (based on the flux ratios in Table 5, they range between 0.35 and 1.0). Since it is not possible, at present, to measure the small O₂ changes created only a few minutes after sealing the chamber, the diffusion into the chamber will always bias the 1-point approach estimates. As our calculations show, the correction for that bias is not negligible. Thus, we recommend using the 2-point approach to estimate O₂ uptake, and hence respiration. Simplifying the [O₂] measurements technique (e.g. by adapting fuel-cell-based O₂ analyzers for this task) is necessary to make this method widely applicable.

6 Conclusions

The average ratio between the CO₂ efflux and O₂ influx, which we defined here as ARQ, was found to be 0.66 ± 0.18 for three species of tropical forest trees, using the gases sampled with stem chambers. For the in-stem gases we found ARQ values of 0.21 ± 0.10 . The low ARQ values in both the in-stem and chambers measurements indicate that a large portion of the CO₂ respired (~ 35 %) in these tropical trees is transported upward in the stem by the xylem water. This CO₂ can be transported in both inorganic and organic forms. If the transported carbon is later fixed in the canopy, then a potentially important photosynthesis flux is also missing from current carbon balance estimates. Future work should reveal the fate of this transported carbon, and estimate the effect of this “recycled” photosynthesis flux on the sensitivity of forests to atmospheric CO₂ changes, and to drought stress.

The removal of CO₂ by the xylem water causes an underestimation of the stem respiration flux by the standard CO₂ efflux techniques in the tropical trees we studied. We thus conclude that measuring O₂ consumption provides a better way to quantify stem respiration rates. Although these kinds of measurements are technically more demanding, we show here a field applicable approach to perform it.

Acknowledgements. The authors thank Menachem Moshelion and Hezi Gildor for fruitful discussion and Eyal Wurgaft for his help with the O₂ analysis. AA was partly supported by ISF grant #870/08 and by Ring Foundation grant.

Edited by: J.-A. Subke

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