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Publication Date

2015

DOI

10.1016/j.proeng.2015.08.830

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Peer reviewed

EUROSENSORS 2015

Optical hydrogen sensing based on hybrid 2D MoO₃/Au nanoparticles

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Abstract

In this work for the first time, we report the use of molybdenum oxide (MoO₃) nanoflakes on gold as active material for plasmonic optical gas sensor. The MoO₃ flakes were deposited over a monolayer of gold nanoparticles, chemically attached to a functionalized fused silica substrate. The coupling between MoO₃ and gold nanoparticles led to reversible optical changes of the localized surface plasmon resonance of gold nanoparticles upon exposure to H₂.

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Peer-review under responsibility of the organizing committee of EUROSENSORS 2015

Keywords: molybdenum oxide; gold nanoparticles; surface plasmon resonance; optical sensor

1. Introduction

Among the two dimensional (2D) transition metal oxides, molybdenum trioxide (MoO₃) is one of the most studied gas sensitive material. 2D α -MoO₃, which is the layered crystal phase of this compound, can be useful especially for H₂ gas interaction since the small H₂ molecules can interact on its surface, diffuse across the layers, spill over and readily intercalate into the layers [1,2]. MoO₃ is an *n*-type semiconductor with a relatively wide

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bandgap energy of ~ 3.2 eV [3]. As a low cost and chemically stable compound, nanostructured MoO_3 also plays an important part in many industrial applications. This material has been integrated into a wide range of transducing platforms [4], such as electrochromic [5], photochromic [6], surface acoustic wave (SAW) [7] and conductometric [8].

In order to demonstrate the high efficiency of using 2D MoO_3 as an optical sensor, nanoflakes have been coupled with Au nanoparticles (NPs). Au NPs have been widely used as optical probe thanks to their Localized Surface Plasmon Resonance (LSPR). Recently, we demonstrated the possibility to fabricate optical gas sensor combining graphene oxides flakes with an Au monolayer [9]. Here we extend such a strategy to 2D MoO_3 .

2. Experimental

A detailed description of the MoO_3 nanoflakes synthesis is reported in [8]. The obtained 2D MoO_3 dispersed in an ethanol/water mixture and was drop casted (300 μl of solution for two times) on previously prepared Au monolayer deposited on pure silica glass slides from colloidal gold of 15 nm diameter as reported in [10]. A sample was also made of the pure 2D MoO_3 film obtained with the same drop casting procedure but on bare silicon substrate as a reference. The obtained films were heat treated at 400 $^\circ\text{C}$ in air for one hour.

Optical absorption spectra were obtained in the 200-900 nm wavelength range using a UV-Vis dual-beam spectrophotometer. Crystalline phases were assessed using glancing angle X-ray diffraction (GAXRD), using a 1° incidence angle and $\text{CuK}\alpha$ radiation at 30 kV and 40 mA. The films microstructures were examined using field emission scanning electron microscopy (FESEM) operated at 30 kV.

Optical gas sensing tests were performed on samples deposited on pure silica glass slides (with and without the gold layer) with optical absorption measurements in the 200-900 nm wavelength range using a Harrick gas flow cell (with 5.5 cm path length) coupled with a UV-Vis spectrophotometer. The substrate dimensions were reduced to 10 mm \times 20 mm by scoring the substrate and breaking it to this size. The standard operating temperature (OT) was set at 300 $^\circ\text{C}$ and gases at concentrations of 1 vol% CO or H_2 in dry air at a flow rate of 0.4 L/min were used. The incident spectrophotometer beam was normal to the film surface and illuminated an area of 9 mm \times 1.5 mm.

3. Results and discussion

Figure 1a shows the GAXRD pattern of the Au monolayer coated with the MoO_3 annealed at 400 $^\circ\text{C}$. It is possible to appreciate the presence of orthorhombic MoO_3 crystalline phase (JCPDS No. 050508) as well as the Au diffraction peaks (JCPDS No. 040784) of the underneath gold monolayer. Looking at the relative intensities of the MoO_3 diffraction peaks it is possible to note a preferred orientation along the (040) plane. The film microstructure is presented in figure 1b, with MoO_3 flakes of dimensions in the 200-300 nm range.

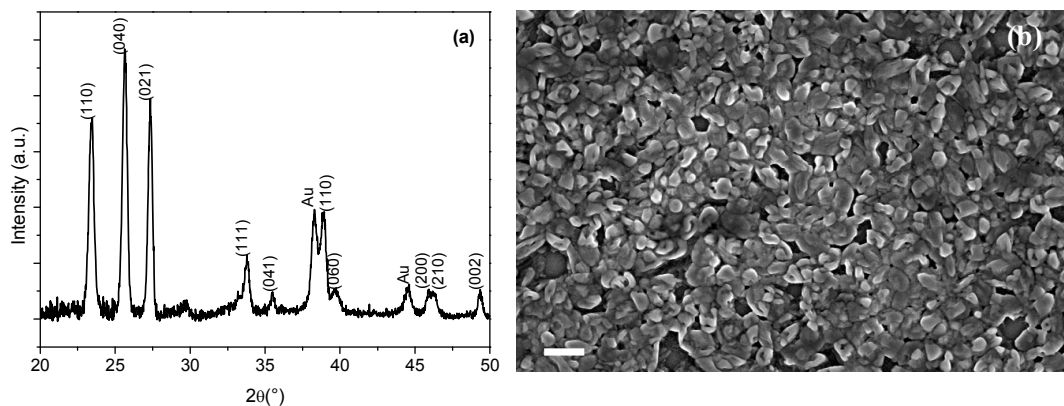


Figure 1 (a) GAXRD pattern of Au monolayer coated with the MoO_3 annealed at 400 $^\circ\text{C}$. (b) FESEM image of MoO_3 film drop casted on silicon substrate (with Au underneath layer) and annealed at 400 $^\circ\text{C}$. The bar correspond to 1 μm .

The absorption spectra of bare Au monolayer and Au monolayer covered with MoO₃ nanoflakes are presented in figure 2a. For the bare Au monolayer, it is possible to note the Au LSPR peak centred at 538 nm. The deposition of the MoO₃ nanoflakes induces a general increase in absorbance, probably due to an increase of the scattering, as well as a shift of the LSPR peak to a longer wavelength due to the higher dielectric constant of the MoO₃ nanoflakes with respect to air. It is also possible to note the absorption in the area of less than 450 nm due to the MoO₃ nanoflakes.

The effect of the target gas on the absorbance has been monitored both for the Au monolayer coated with MoO₃ and for the reference pure MoO₃. The sensing performances were monitored using the Optical Absorption Change parameter, defined as the difference in absorbance during target gas exposure and during air exposure ($OAC = A_{gas} - A_{air}$). Figure 2 shows the OAC after the H₂ exposure for both the Au monolayer coated with MoO₃ and pure MoO₃. It is evident that the presence of Au NPs is essential for having a gas response. The larger variation in the absorbance can be detected after the 570 nm region, corresponding to the position of the Au LSPR peak, and in the near infrared region. The prepared sensor did not respond to CO.

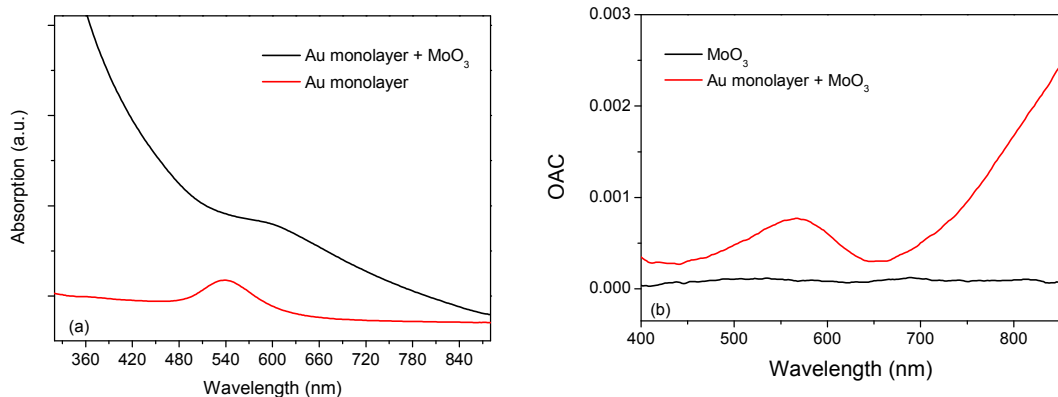


Figure 2 (a) Absorption spectra of Au monolayer (red line) and Au monolayer coated with MoO₃ nanoflakes (black line) annealed at 400 °C. (b) OAC curves ($OAC = A_{H_2} - A_{Air}$) for Au monolayer coated with MoO₃ (red line) and for the pure MoO₃ (black) under 1vol% H₂ exposure at 300 °C OT.

The samples show good dynamic behavior after repeated exposures to H₂ as reported in Figure 3: the response time is short, while the recovery time is slightly longer but still satisfactory. It is worth noting that at 570 nm the dynamic measurements are not as stable as those at 800 nm.

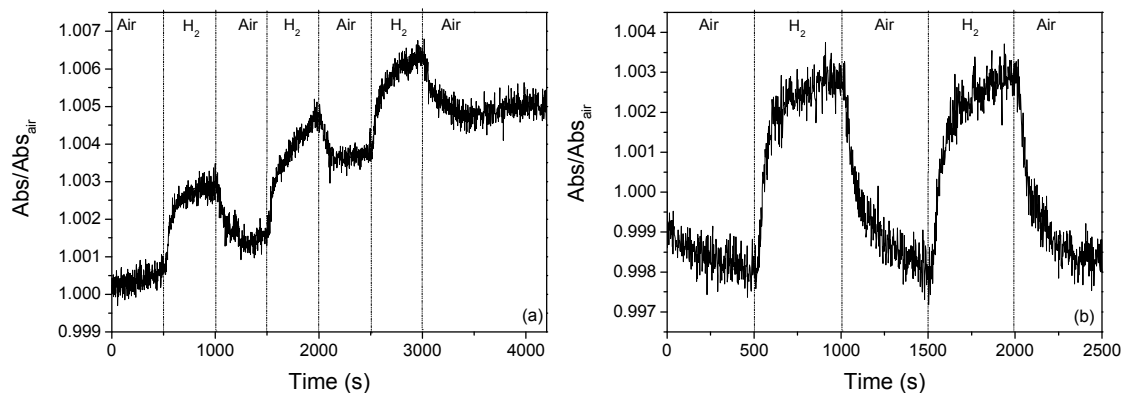


Figure 3 Dynamic response of the Au monolayer coated with MoO₃ nanoflakes at 300 °C OT, during exposure to different air-H₂-air cycles at (a) 570 nm and (b) 800 nm.

The sensing mechanism may be interpreted considering the interaction of H₂ with the oxygen adsorbed on the MoO₃ surface which leads to the formation of H₂O and the release of electrons, as already reported in [5]. This charge injection can be monitored looking at the LSPR peak of Au NPs which is in contact with the MoO₃ film, as already reported for other transition metal oxide – Au NPs composites [12]. It is worth noting that pure MoO₃ films did not show any appreciable variation of the absorbance when exposed to H₂, demonstrating the importance of the coupling with the Au NPs.

4. Conclusions

2D MoO₃ has been successfully used as a sensitive layer for LSPR based optical sensor. While bare MoO₃ was not sensitive to the target gas, the coupling between MoO₃ nanoflakes and Au NPs is effective for the realization of sensitive H₂ optical gas sensor. The plasmonic presented sensor has a good potential for future H₂ sensing applications.

Acknowledgements

M. Angiola gratefully acknowledges the financial support from the CARIPARO Foundation for the project “Dottorati di ricerca”.

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