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Enantiospecific Total Synthesis of *N*-Methylwelwitindolinone D Isonitrile

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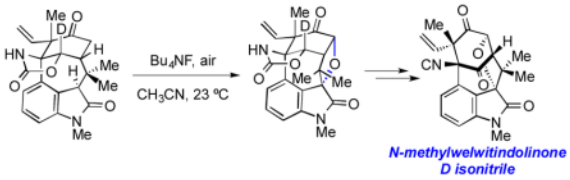
Abstract



We report the enantiospecific total synthesis of *N*-methylwelwitindolinone D isonitrile. Our route features a double C-H functionalization event involving a keto oxindole substrate to introduce the tetrahydrofuran ring of the natural product.

Keywords

C-H functionalization; natural products; nitrene insertion; total synthesis; welwitindolinone

The welwitindolinone family of natural products (e.g., **1**–**2**, Scheme 1) has attracted tremendous attention from the synthetic community over the past two decades.[1,2,3,4,5] Interest in these compounds stems from their promising biological profiles, in addition to their compact, yet daunting structures. Synthetic efforts toward the welwitindolinones have led to at least ten methods for building the bicyclo[4.3.1] core that is common to most of these natural products.[1,4] However, the sheer difficulty associated with late-stage manipulations has plagued most synthetic routes and only a few completed syntheses have been reported in recent years.[5]

One exceptionally challenging synthetic target is *N*-methylwelwitindolinone D isonitrile (**2**).[6, 7] The compound possesses five stereocenters, two quaternary carbons, and a heavily substituted cyclohexyl ring. Compared to other related family members, **2** also possesses an ether linkage between C3 and C14. Thus, a successful synthesis of **2** would not only have to assemble the congested oxindole-fused bicyclo[4.3.1] framework, but would also have to allow for introduction of the ethereal linkage on the sterically congested face of the bicycle. Highlights of synthetic efforts toward **2** include the Wood group's assembly of the spirocyclic oxindole[8] and Rawal's elegant total synthesis of (±)-**2** in 2011.[5a] Herein, we report our synthetic forays toward **2**, which culminate in an enantiospecific synthesis.

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Supporting information for this article is available on the WWW under http://www.angewandte.org or from the author.

Our retrosynthetic plan for the synthesis of **2** is presented in Scheme 2. The natural product would be accessed from **3** via late-stage manipulations. In a key disconnection, the tetrahydrofuran ring would be installed from keto-oxindole derivative **4**. Of note, the ability to elaborate **4** to **3** would hinge on our ability to perform chemoselective and diastereoselective manipulations adjacent to the two carbonyls. The cyclic carbamate was thought to be accessible using an intramolecular nitrene insertion reaction[9] involving oxindole substrate **5**. Substrate **5** would be derived from ketone **6**, which in turn can be readily prepared from known carvone derivative **7**[10] in just four steps using our previously established procedure involving an indolyne cyclization.[5b,11]

Our approach toward implementing the retrosynthetic plan is highlighted in Scheme 3. Indole **6** was converted to oxindole **8** using a one-pot oxidation/hydrolysis sequence. As the acidic conditions led to desilylation, reprotection of the alcohol was necessary to provide **9**. Deuteride reduction and carbamoylation proceeded without event to furnish **5** in quantitative yield. To our delight, exposure of **5** to Ag-promoted nitrene insertion conditions[12,5e] furnished **10** in 70% yield. It should be noted that attempts to use the proteo analog of **5** gave only 44% yield of the corresponding insertion product, along with 19% of recovered ketone **9**. Thus, consistent with our previous findings on an alternate substrate,[5e] the strategic use of deuterium minimizes an undesirable competitive reaction, thus giving synthetically useful yields of the desired insertion product **10**. From **10**, a standard deprotection/oxidation sequence delivered key intermediate **4**.

Many attempts to introduce the tetrahydrofuran ring from **4** were put forth. Unfortunately, efforts toward site-selective functionalization of one carbonyl over the other via enol ethers were unsuccessful. After considerable experimentation, it was found that the keto carbonyl could be α-functionalized first upon treatment of **4** with $CuBr_2$ in THF at ambient temperature to yield **11** as a single diastereomer (Scheme 4). It was hoped that C3-oxidation would provide an alcohol intermediate that would cyclize to give the necessary tetrahydrofuran ring. However, upon treatment of **11** with C3 oxidation conditions,[5b] the desired oxidation and cyclization did not occur. Instead, we unexpectedly obtained cyclobutane **13** in high yield, presumably via direct cyclization of the oxindole enolate (see transition structure **12**).[13] X-ray analysis of a single crystal of **13** validated our structural assignment.[14,7]

As a workaround, we opted to introduce a protected hydroxyl group directly onto C3 of substrate **11**. $Mn(OAc)_3$ was deemed a potential reagent for selective C3-acetoxylation, based on its use in benzylic acetoxylation reactions.[15] As shown in Table 1, treatment of oxindole **11** with $Mn(OAc)_3$ in AcOH at 80 °C provided acetoxylated product **14** (entry 1). Interestingly, when the corresponding reaction was conducted at 150 °C, we obtained a 53% yield of **3**, which possesses the desired tetrahydrofuran ring. Alternatively, **3** could also be prepared in one-pot by performing the acetoxylation at 80 °C, removing the volatiles, and exposing the crude intermediate to K_2CO_3 in MeOH and H_2O at 70 °C.

We also explored the feasibility of directly converting keto oxindole **4** to **3** (Scheme 5). Of note, the Wood group was able to elegantly install a tetrahydrofuran ring from a keto oxindole substrate using basic conditions and O_2.[8] Despite the modest yield, this key precedent laid the groundwork for additional experimentation. To our delight, we found that simple exposure of **4** to tetrabutylammonium fluoride in acetonitrile in the presence of air efficiently delivered **3**.[16] In previous studies, we[17] and others[18] have found that TBAF/air can facilitate C3 oxidation of oxindoles containing the welwitindolinone scaffold, but the use of TBAF/air to build an ethereal linkage via double C-H functionalization was unknown. It should be noted that the use of other bases in place of TBAF, such as K_2CO_3 and Cs_2CO_3, also promoted the formation of **3**, albeit in lower yields. It is likely that this efficient method

for introducing the tetrahydrofuran ring proceeds by initial diastereoselective C3 oxidation, followed by cyclization.[19] Related C3-peroxy compounds have been observed in our studies[20] and in Wood's.[8]

To complete the total synthesis, it remained to elaborate the cyclic carbamate to the ketone and isonitrile functional groups present in **2** (Scheme 6). Unexpectedly, attempted hydrolysis of **3** led to cyclohexyl ring fragmentation, a process that was attributed to the reactivity of the ketone. To circumvent this, ketone **3** was reduced to alcohol **15** with $LiAlH_4$. Fortunately, upon exposure of **15** to hydrolysis conditions, cyclohexyl ring fragmentation was not observed. Hydrolysis gave the desired diol intermediate, which was oxidized with IBX to provide diketone **16**. Finally, formylation provided **17**, which was directly exposed to standard dehydration conditions to deliver (+)-**2**.

In summary, we have completed the enantiospecific total synthesis of *N*-methylwelwitindolinone D isonitrile. Several unexpected hurdles, including the formation of the unusual cyclobutane-containing compound **13** were overcome en route to the natural product. Our total synthesis features a double C-H functionalization event of keto oxindole **4** to introduce the tetrahydrofuran ring of **2** and is achieved in 17 steps from readily available carvone derivative **7**.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

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13. Variations in reaction conditions (e.g., employing a variety of bases, saturating with O_2) did not overcome the formation of **13**.
14. CCDC 960226 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
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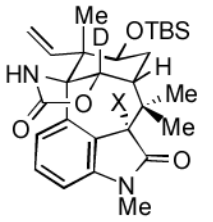
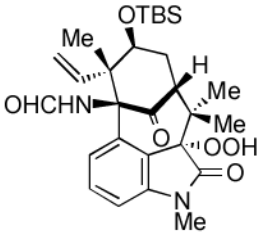
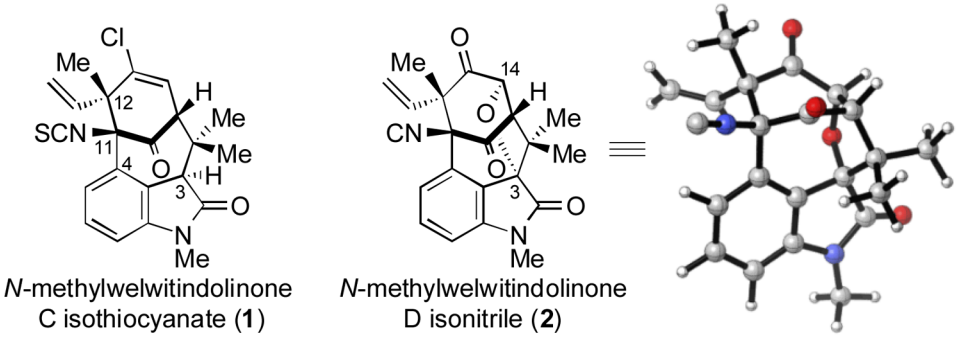
19. Treatment of **4** with 3.0 equiv TBAF and 50.0 equiv MeOD in CH_3CN under an atmosphere of N_2 at ambient temperature gave 40% deuterium incorporation at C3 after 10 min, whereas treatment under the same conditions for 1 h gave 50% deuterium incorporation at C3 and 25% deuterium incorporation at C14.
20. Efforts to isolate the putative peroxy species (Scheme 5) have been unsuccessful; however, we have isolated several related compounds, such as **i** and **ii**, by oxidation of the corresponding oxindoles.



i (X = OOH)



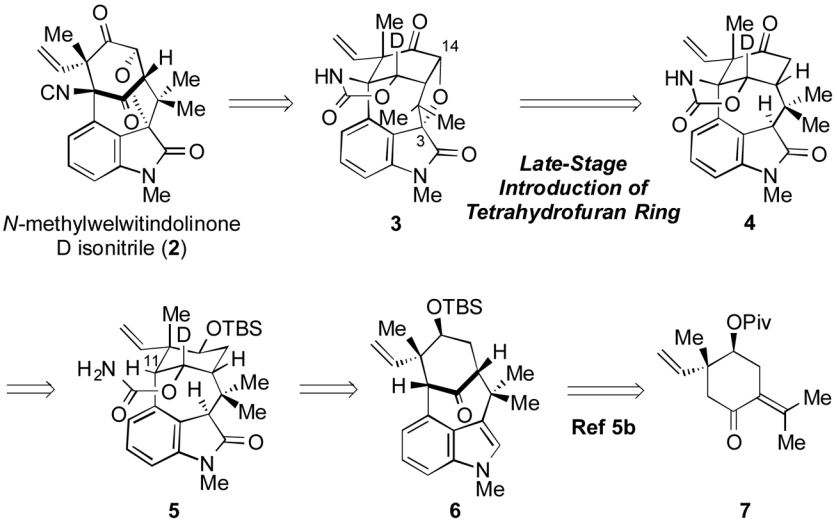
ii



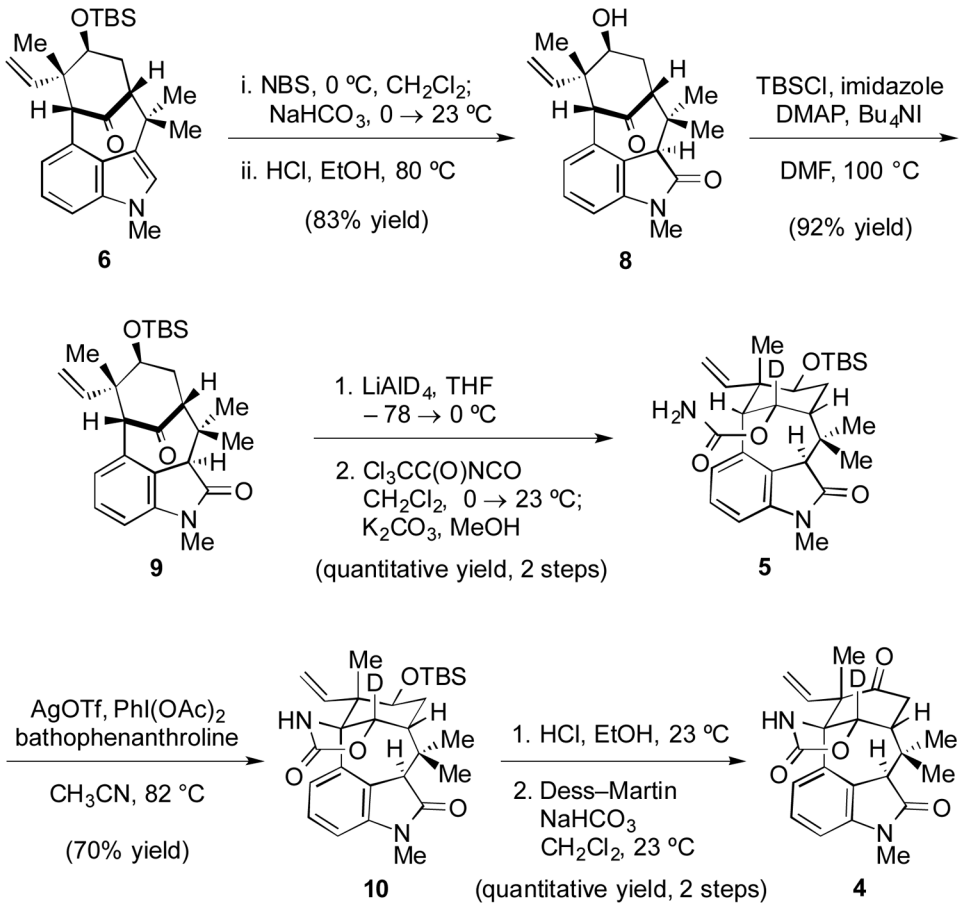
N-methylwelwitindolinone C isothiocyanate (**1**)

N-methylwelwitindolinone D isonitrile (**2**)

Scheme 1.
Welwitindolinones **1** and **2**.

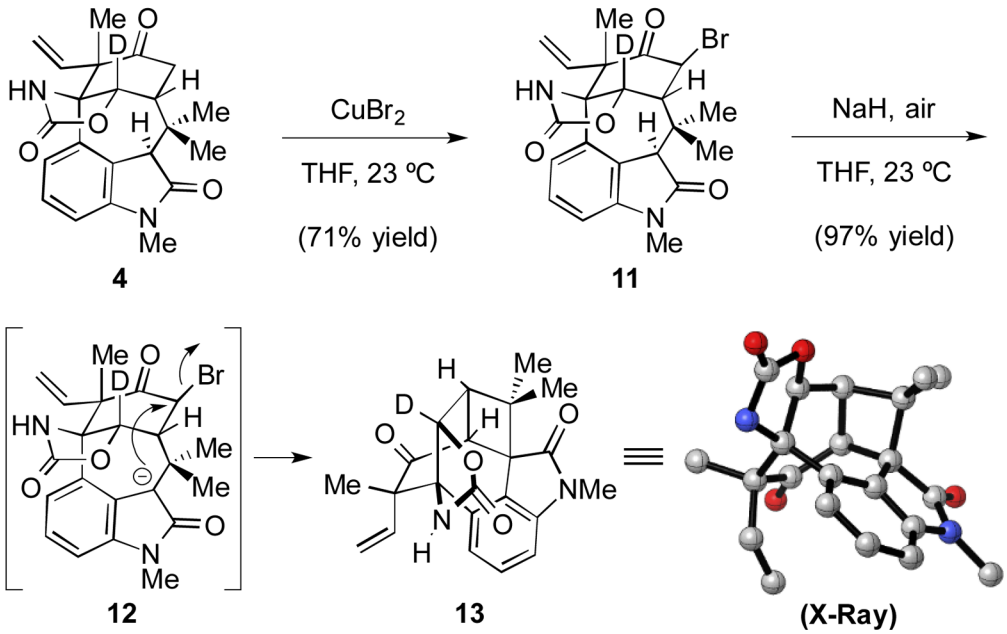


Scheme 2.
Retrosynthetic analysis of **2**.



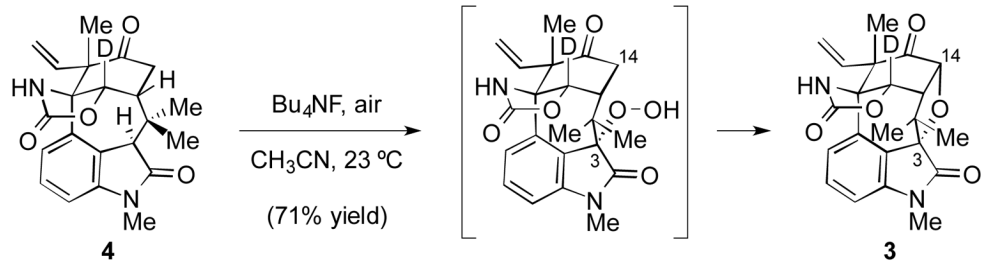
Scheme 3.

Elaboration of **6** to keto oxindole **4**; TBS=*tert*-butyldimethylsilyl, NBS=*N*-bromosuccinimide, DMAP=4-dimethylaminopyridine, DMF=dimethylformamide, THF=tetrahydrofuran, Tf=trifluoromethanesulfonyl, OAc=acetate, bathophenanthroline=4,7-diphenyl-1,10-phenanthroline, Dess–Martin=1,1,1-triacetoxy-1,1-dihydro-1,2-benziodoxol-3(1H)-one.

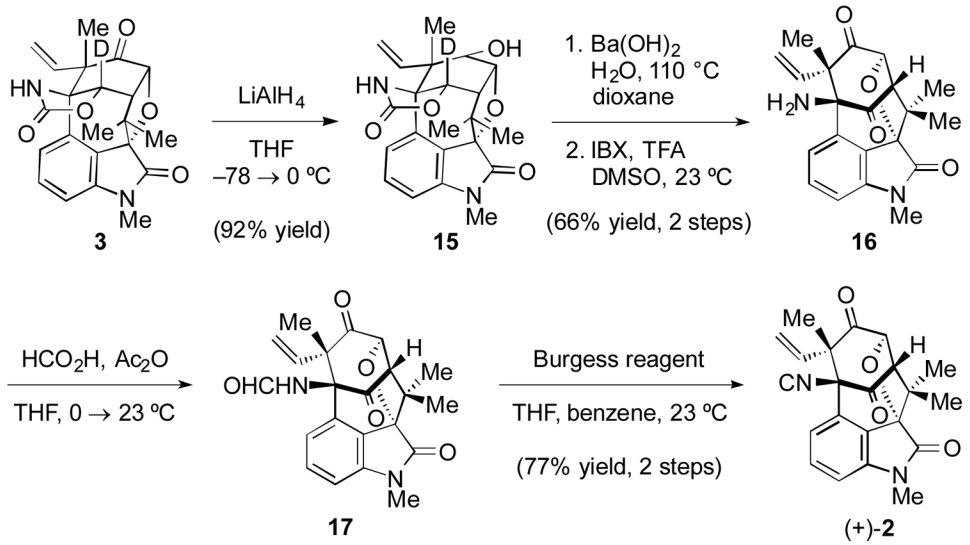


Scheme 4.

Unexpected formation of cyclobutane **13**; THF = tetrahydrofuran.



Scheme 5.
Double C-H functionalization of substrate **4** to install the tetrahydrofuran ring.

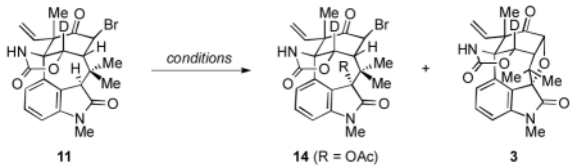


Scheme 6.

Completion of (+)-**2**; THF=tetrahydrofuran, dioxane=1,4-dioxane, IBX=2-iodoxybenzoic acid, TFA=trifluoroacetic acid, DMSO=dimethylsulfoxide, Burgess reagent=methyl *N*-(triethylammoniumsulfonyl)carbamate.

Table 1

Conversion of **11** to acetate **14** and cyclized product **3**.



entry	conditions	conversion to products 14 and 3	
1	$Mn(OAc)_3$ (4.0 equiv), AcOH, 80 °C	74	0
2	$Mn(OAc)_3$ (4.0 equiv), AcOH, 150 °C	2	53
3	$Mn(OAc)_3$ (4.0 equiv), AcOH, 80 °C; K_2CO_3, MeOH, H_2O, 70 °C	0	56