Title

Correction to “Synthesis, Structure, and Spectroscopy of the Biscarboranyl Stannylenes (bc)Sn·THF and K2[(bc)Sn]2 (bc = 1,1′(ortho-Biscarborane)) and Dibiscarboranyl Ethene (bc)CH ▫ CH(bc)”

Journal

Organometallics, 42(16)

ISSN

0276-7333

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Publication Date

2023-08-28

DOI

10.1021/acs.organomet.3c00319

Peer reviewed

ORGANOMETALLICS

pubs.acs.org/Organometallics

Addition/Correction

Correction to "Synthesis, Structure, and Spectroscopy of the Biscarboranyl Stannylenes (bc)Sn·THF and $K_2[(bc)Sn]_2$ (bc = 1,1′(*ortho*-Biscarborane)) and Dibiscarboranyl Ethene (bc)CH═CH(bc)"

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Organometallics **2023**, *42* (13), 1649−1657. DOI: 10.1021/acs.organomet.3c00190

Cite This: *Organometallics* 2023, 42, 2319−2320

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In the original title publication the compound **2** was reported to have the formula $K_2[(bc)Sn]_2$ (**bc**=1,1′-bis(*ortho*-biscarborane)). But a re-examination and rerefinement of the X-ray crystallographic data for **2** indicate that the correct formula is $[(bc)Sn]_2KCl$. As reported earlier, one of the K^+ ions acts as a countercation which is coordinated to the B−H vertices of the **bc** cage, but the second K^+ ion, which was said to bridge the two tin atoms in the original report, has been reformulated as a Cl^- ion, thereby forming a Sn−Cl−Sn bridging unit between two (**bc**)Sn moieties, as in the $\{(bc)Sn\text{-}\mu\text{-}Cl\text{-}Sn(bc)\}^-$ anion, as well as lowering the *R* value from 2.43% to 2.23%. Organotin complexes containing a Sn−Cl−Sn moiety typically have bridging Sn−Cl distances in the range 2.540(2)−2.967(1) Å,[1−10] and this fragment of compound **2** features Sn−Cl bond distances of 2.5868(8) and 2.5874(8) Å, which are at the lower end of the above range. The relatively short Sn−Cl distances in **2** indicate a strong interaction between the two ions, but whether this is a consequence of the rigid structure or electron-withdrawing influence of the **bc** ligand is not apparent. The drawings in Figure 1 below illustrate the corrected X-ray crystal formula for compound **2**, and Figure 1c in particular clearly illustrates the bridging atom as Cl^- seems more chemically sensible than when the species is formulated with a bridging K^+ cation.

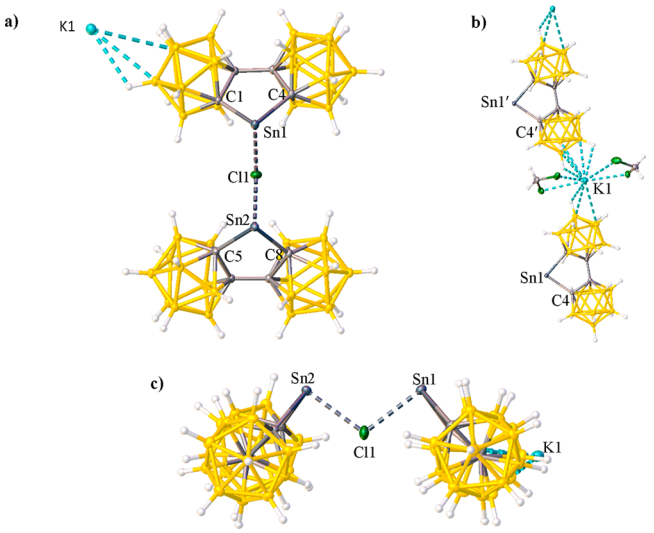

Figure 1. Thermal ellipsoid plot (50%) of **2**. (a) View of **2** to show coordination positions of K1 and Cl1. The CH_2Cl_2 solvent molecules are not shown for clarity. (b) Expanded view of **2** to show coordination of K1. (c) View of **2** to show coordination of Cl1. CH_2Cl_2 solvent molecules are not shown for clarity. Selected bond lengths (Å) and angles (deg): C1−Sn1 = 2.276(3), C4−Sn1 = 2.309(3), C5−Sn2 = 2.288(4), C8−Sn2 = 2.289(3), Cl1−Sn1 = 2.5868(8), Cl1−Sn2 = 2.5874(8), C1−Sn1−C4 = 81.69(11), C5−Sn2−C8 = 81.86(12), Sn1−Cl1−Sn2 = 106.98(3), C1−Sn1−Cl1 = 88.27(7), C4−Sn1−Cl1 = 94.11(7), C5−Sn2−Cl1 = 92.62(7), C8−Sn2−Cl1 = 90.91(7).

■ ACKNOWLEDGMENTS

We thank Prof. Igor Sivaev for alerting us to this error.

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Published: August 10, 2023

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https://doi.org/10.1021/acs.organomet.3c00319
Organometallics 2023, 42, 2319−2320