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Soil aggregates as biogeochemical reactors and implications for soil-atmosphere exchange of greenhouse gases—A concept

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#### 25 Abstract

Soil-atmosphere exchange significantly influences the global atmospheric abundances of 26 27 carbon dioxide ( $CO_2$ ), methane ( $CH_4$ ), and nitrous oxide ( $N_2O$ ). These greenhouse gases ( $GHG_3$ ) have been extensively studied at the soil profile level and extrapolated to coarser scales (regional 28 and global). However, finer scale studies of soil aggregation have not received much attention, 29 30 even though elucidating the GHG activities at the full spectrum of scales rather than just coarse levels is essential for reducing the large uncertainties in the current atmospheric budgets of these 31 gases. Through synthesizing relevant studies, we propose that aggregates, as relatively separate 32 33 micro-environments embedded in a complex soil matrix, can be viewed as biogeochemical reactors of GHGs. Aggregate reactivity is determined by both aggregate size (which determines the reactor 34 size) and the bulk soil environment including both biotic and abiotic factors (which further 35 influence the reaction conditions). With a systematic, dynamic view of the soil system, 36 implications of aggregate reactors for soil-atmosphere GHG exchange are determined by both an 37 38 individual reactor's reactivity and dynamics in aggregate size distributions. Emerging evidence supports the contention that aggregate reactors significantly influence soil-atmosphere GHG 39 exchange and may have global implications for carbon and nitrogen cycling. In the context of 40 41 increasingly frequent and severe disturbances, we advocate more analyses of GHG activities at the 42 aggregate scale. To complement data on aggregate reactors, we suggest developing bottom-up 43 aggregate-based models (ABMs) that apply a trait-based approach and incorporate soil system 44 heterogeneity.

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#### 48 1 Introduction

Biogeochemical reactions are manifested at spatial scales ranging from molecule to globe 49 50 (McClain et al., 2003). In the case of the biogeochemical processes specifically responsible for GHG production/consumption in soils, our understanding, however, is mostly derived from the 51 soil profile level because of methodological constraints (Mosier et al., 1991; Smith et al., 2003). 52 53 Larger scale understanding (from landscape through regional to global) is extrapolated from knowledge gained at this scale through scale-up exercises via a combination of land surface 54 modeling and remote sensing of land cover (e.g., McClain et al., 2003). From these scaling efforts, 55 it is clear that soil-atmosphere exchange significantly affects the atmospheric abundances of  $CO_2$ , 56 CH<sub>4</sub>, and N<sub>2</sub>O regionally and globally (Ciais et al., 2013). This conclusion is confirmed by top-57 down constraints derived from satellite measurements (Ciais et al., 2013). Nevertheless, we still 58 have very poor constraints on the global and regional balances of GHGs (e.g., Ciais et al., 2013; 59 Saunois et al., 2016), though the underlying biogeochemical reactions responsible for the 60 61 production and/or consumption of these gases in soils are relatively well understood [but see Wang et al. (2017a) for an emerging paradigm of GHG production via non-microbial pathways]. 62

63 These uncertainties necessitate a more complete understanding of GHG activities at the 64 full spectrum of spatial scales, especially the soil aggregate scale which is relatively less 65 understood. Such information is essential for further elucidating complex processes resulting from 66 soil heterogeneity and for guiding bottom-up modeling of soil-atmosphere exchange of GHGs 67 (e.g., Hinckley et al., 2014; Ebrahimi & Or, 2016). This endeavor will eventually contribute to a 68 better understanding of the mechanisms regulating atmospheric concentrations of GHGs and to 69 improved strategies for mitigating soil GHG emissions in the context of global environmental 70 changes, i.e., achieving 'climate-smart' soils (Paustian et al., 2016).

The soil system shows extremely high heterogeneity, and microbial activities are not 71 spatially homogeneous in the soil matrix (Young & Crawford, 2004). From the perspective of 72 73 structure, soil aggregates and pore spaces create fine-scale spatial heterogeneity (e.g., Elliott & Coleman, 1988; Horn et al., 1994; Rillig et al., 2017). Primary soil mineral particles (clay, silt, and 74 75 sand) chemically interact with organic matter (historically classified as 'primary organo-mineral 76 complexes'), forming the basic units of soil aggregates. These basic units can accrete into larger aggregates, depending on the availability of a diverse suite of binding agents (e.g., polyvalent 77 cations: Ca<sup>2+</sup> or Al<sup>3+</sup>) and various forms of organic matter (e.g., polysaccharides, organic acids, 78 79 plant debris, roots, and hyphae). From these aggregation processes a hierarchical system of soil aggregates emerges (Tisdall & Oades, 1982; Oades, 1991; Lehmann et al., 2007). Such aggregates 80 generate additional soil heterogeneity, along with other 'hotspots' associated with the rhizosphere, 81 detritusphere, and biopores that affect the distribution of substrates and microbial communities 82 (Kuzyakov & Blagodatskaya, 2015). 83

84 Since the soil aggregate concept was first proposed about a century ago, extensive studies have been conducted on physiochemical and biological properties at this scale, as well as their 85 86 responses to disturbances including soil management, land use change, and global change (e.g., 87 Elliott, 1986; Oades, 1991; Jastrow, 1996; Six et al., 2000a, 2000b, 2004; Blanco-Canqui & Lal, 88 2004; Lehmann et al. 2017). However, knowledge about aggregate-scale GHG dynamics is still 89 fairly scarce. To address GHG activities at this fine scale and hence their implications for coarse 90 scale GHG exchange, here we propose that aggregates—as relatively independent micro-habitats 91 in the soil matrix—can be viewed as biogeochemical reactors (hereafter referred to as aggregate 92 reactors) that produce GHGs including CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O. Correspondingly, aggregate reactivity 93 is defined as the potential for GHG production and the rate and duration of these reactions. Because

of varying sizes and turnover rates, we further argue that aggregate reactors in a soil system should
be viewed in a systematic and dynamic way. By proposing this aggregate reactor concept in a
dynamic framework, ecological theory can be applied to studies of GHG exchange by examining
both the reactivity of an aggregate reactor (physiology) and the compositional dynamics of
differing aggregate reactors (community ecology).

99 The overarching purpose of this paper is to demonstrate the rationale behind this concept, 100 discuss its implications for soil-atmosphere GHG exchange, and propose ways forward to improve our understanding of aggregate reactors. By synthesizing previous studies, we first examine basic 101 102 characteristics of aggregate reactors. We then discuss factors influencing aggregate reactivity, such 103 as aggregate size, soil bulk properties, and fractionation methodology. Implications of aggregate reactors for soil-atmosphere GHG exchange are also addressed. In light of increasingly frequent 104 105 and severe perturbations to soil systems, we emphasize impacts of soil management and global change on aggregate reactors. We conclude by identifying current knowledge gaps and research 106 107 opportunities, including the potential to develop aggregate-based models (ABMs) that can 108 explicitly incorporate the structural heterogeneity of soil systems. We recommend integration of 109 soil science, ecology, and climate science communities to advance the aggregate reactor concept 110 and to develop a predictive framework based on aggregate reactors in the context of global change. 111 These efforts should eventually help reduce the large uncertainties in GHG exchange associated 112 with soil heterogeneity.

113

### 114 2 The aggregate as a 'biogeochemical reactor' of GHG

Embedded in the soil matrix, soil aggregates exhibit physical, chemical, and biological
properties that differ from the bulk soil (Fig. 1). Soil aggregates contain a three-dimensional

structure with pores of varying sizes (e.g., Ebrahimi & Or, 2016). Organic matter becomes 117 occluded during the aggregation process. Identifiable components of the occluded fraction include 118 119 small particles of incompletely decomposed organic residues, pollen grains, and particles of plant tissue such as lignin coils and phytoliths. This physically bound organic matter, compared with 120 121 free organic matter, often has relatively higher carbon and nitrogen concentrations and contains 122 more alkyl carbon that is recalcitrant (Golchin et al. 1994a, b; Six et al., 2001b). Further, oxygen  $(O_2)$ , water, nutrients, and dissolved organic carbon diffuse into the aggregates from the inter-123 aggregate voids or macropores of bulk soil (Keiluweit et al., 2016). For example, O<sub>2</sub> diffusion 124 125 from the soil matrix (where the  $O_2$  concentration is further controlled by bulk soil moisture, plant 126 roots, and other factors) is limited by the pore networks of aggregates, whereas O<sub>2</sub> consumption in aggregates is controlled by microbial activities, organic carbon, and nutrient availability. The net 127 result is that aggregates can experience  $O_2$  limitations even within aerobic, well-drained soils, 128 making anoxia the most notable environmental characteristics of aggregates (e.g., Tiedje et al. 129 130 1984; Sexstone et al., 1985; Elliott & Coleman, 1988; Sexstone et al. 1988; Højberg et al., 1994; Diba et al., 2011; Keiluweit et al., 2016). 131

A variety of anaerobic metabolic pathways can occur in aggregates (Ebrahimi & Or, 2015; Keiluweit et al. 2016), including processes responsible for CH<sub>4</sub> and N<sub>2</sub>O production such as denitrification and methanogenesis (e.g., Sexstone et al. 1988; von Fischer & Hedin 2002; Keiluweit et al. 2016; summarized in **Tables 1-3**). Therefore, aggregates can be viewed as segregated biogeochemical reactors of GHG embedded in a complex soil matrix (**Fig. 1**). The connectivity and tortuosity of pores and other bulk soil properties (e.g., soil texture, moisture, and biological activities) determine the micro-environment in aggregates by regulating O<sub>2</sub> diffusion, distribution of water films, and substrate and nutrient accessibility for microbes, as well as thecomposition and structure of soil microbial community.

Microbial communities inhabit soil aggregates and exhibit dynamics in composition and activity (e.g., Ebrahimi & Or, 2016). Recently, based on aggregates' isolation feature Rillig et al. (2017) proposed that aggregates are 'incubators' of microbial evolution that allow processes including genetic drift, natural selection, and mutation to occur and that likely produce an overall effect of an increase of microbial diversity. This microbial evolution dimension further adds complexity to understanding aggregate reactors.

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#### 148 **3** Factors regulating aggregate reactivity

#### 149 **3.1 Aggregate reactor size**

Aggregate reactor size is typically measured in terms of diameter. Aggregate reactor size is an important factor for GHG reactivity because of size impacts on other physical, chemical, and biological activities. Just as GHG exchange exhibits substantial variation at coarse spatial scales, aggregate reactivity is also expected to vary across aggregate size classes. Here, we offer a nonexhaustive review of the major differences between macro- (>0.25 mm) and micro-aggregates (<0.25 mm), focusing on the aspects that could potentially result in reactivity differences.

First, aggregates of different sizes have differing geometry. One important aspect of the geometry is mean pore size which is smaller for micro-aggregates than for macro-aggregates (Dexter, 1988). This difference affects diffusion of O<sub>2</sub>, nutrients, and dissolved organic carbon (DOC). Compared with macro-aggregates, O<sub>2</sub> diffusion is slower into micro-aggregates (Sexstone *et al.*, 1985; Elliott & Coleman, 1988; Højberg *et al.*, 1994; Denef *et al.*, 2001; Diba *et al.*, 2011). 161 Second, the chemical composition of substrates for carbon and nitrogen mineralization is 162 different. Macro-aggregates often have higher carbon and nitrogen concentrations (*e.g.*, Elliott, 163 1986; Gupta & Germida, 1988; Cambardella & Elliott, 1993). Younger and more labile organic 164 matter (with a higher C/N ratio) constitutes more of the organic matter pool in macro-aggregates 165 than in micro-aggregates (Elliott, 1986; Elliott and Coleman 1988; Six et al., 2004).

Third, microbial community composition and structure are influenced by aggregate size
(e.g., Van Gestel et al., 1996; Mummey *et al.*, 2006; Kravchenko et al., 2014; Rabbi et al., 2016;
Ebrahimi and Or, 2016). For instance, Mummey *et al.*, 2006 found that microaggregates select for
specific microbial lineages across disparate soils. Bach et al. (2018) reported that microaggregates
hold more diverse microbial communities than macroaggregates.

Many studies have found differences in GHG process rates among aggregates of varying 171 sizes, though some studies have found no differences (Table 1-3). Sexstone et al. (1985) and later 172 studies found that the composition of microbial communities responsible for N<sub>2</sub>O production 173 174 varies with aggregate size. Sey et al. (2008) suggested that ammonium oxidizers are most abundant in macro-aggregates, while denitrifiers, which preferentially colonize anaerobic environments, are 175 more abundant in micro-aggregates. This study also found that denitrification pathways of N<sub>2</sub>O 176 177 production dominate in smaller aggregates, whereas nitrification dominates N<sub>2</sub>O production in 178 larger aggregates. In addition, greater denitrification rates occurred in the smallest aggregate size 179 fractions when acetylene was applied to prevent the complete reduction of N<sub>2</sub>O to N<sub>2</sub> during 180 denitrification. Higher denitrification in smaller aggregates was also reported by Seech and 181 Beauchamp (1988) and Uchida et al., (2008). Sey et al. (2008) attributed this pattern to a higher proportion of complete denitrification (NO<sub>3</sub><sup>-</sup>  $\rightarrow$  N<sub>2</sub>O  $\rightarrow$  N<sub>2</sub>) because of very low or effectively 182 183 absent O<sub>2</sub> in smaller aggregates. An alternative explanation, proposed by Ebrahimi and Or (2016),

is that N<sub>2</sub>O takes longer to diffuse out of larger aggregates and thus has more time to be completely
reduced from N<sub>2</sub>O to N<sub>2</sub>.

186 Overall, consistent relationships between aggregate reactor size and reactivity are difficult to establish, even qualitatively (**Table 1-3**). Although the majority of studies with different types 187 188 of soils (>60%) support an overall positive relationship between aggregate size and CO<sub>2</sub> 189 production, some studies observed a negative relationship. For N<sub>2</sub>O production, the majority of 190 studies (almost 70%) support an overall positive relationship with aggregate size, of which more than half found more N<sub>2</sub>O production from macro-aggregates than micro-aggregates (Table S3). 191 192 Regarding CH<sub>4</sub>, more studies observe that smaller aggregates act as consumers and larger aggregates as producers, but this is rarely true for the specific comparison of macro- and micro-193 194 aggregates (Table S4).

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#### **3.2 Bulk soil properties**

Bulk soil properties determine the environment surrounding the aggregate reactors, which largely shapes conditions in the reactors. The observed inconsistent relationships between aggregate reactor size and reactivity across different studies, as discussed above (**Table 1-3**), support this postulation: aggregate reactivity is not just determined by size but also by the bulk soil properties. These bulk properties can be abiotic or biotic factors that regulate the soil physicchemical and biological environment.

Although it remains challenging to establish direct causal connections between aggregate reactivity and bulk soil properties, a few studies have addressed these relationships. For instance, soil water content can significantly affect aggregate-level CH<sub>4</sub> activities (Sey *et al.* 2008). Aggregates with sizes < 0.25 mm and 0.25-2 mm consumed CH<sub>4</sub> at low water content but began

to produce CH<sub>4</sub> at higher water content (Sey et al. 2008). This same study also observed a 207 maximum CH<sub>4</sub> production rate at 40% water-filled pore space (WFPS) for 2-6 mm aggregates. 208 209 This pattern can be explained by greater methanogenesis when high water content prevents  $O_2$ diffusion (Yavitt et al., 1990). Regarding CO<sub>2</sub>, studies by Drury et al. (2004) and Mangalassery et 210 211 al. (2013) both observed the largest CO<sub>2</sub> effluxes from small sized aggregates in clay loam soils, 212 in contrast to other studies using different soils (Table 1). This observed difference in CO<sub>2</sub> production may have resulted from texture differences that influence soil porosity and water film 213 distribution. 214

Many other factors, especially biotic ones, that can also affect aggregate reactor conditions have not been studied at all. For instance, a soil O<sub>2</sub> concentration decline can be induced directly by root respiration and/or by root exudation that stimulates heterotrophic respiraton in the rhizosphere (Keiluweit *et al.*, 2016). Other soil organisms, such as soil fauna and fungal hyphae, can affect soil porosity and change the diffusivity of O<sub>2</sub> into aggregates or significantly affect the formation of soil aggregates and their associated C pools (Lehmann *et al.*, 2017). Based on this reasoning, biological activities should exert a variety of effects on aggregate-level GHG dynamics.

223 **3.3 Fractionation method** 

Theoretically, aggregate reactivity should be determined by both soil properties and aggregate reactor size. In practice, however, the observed variability of aggregate reactivity (**Table 1-3**) may also reflect differences in fractionation techniques. Separation methods are not uniform across studies; specifically, dry-sieving is used more often than wet-sieving (**Table 1-3**). Separation exerts significant influences on physical, chemical, and biological properties of different sized aggregates (Ashman *et al.*, 2003; Bach & Hofmockel, 2014; Kaiser *et al.*, 2015).

For instance, air-drying can increase the mechanical strength of aggregates, and wet-sieving can 230 231 increase potential enzyme activity (Bach & Holmockel, 2014). A recent review by Kaiser et al. 232 (2015) provides further details on air-drying and rewetting effects on soil aggregate stability. Currently, we know of only one study by Beauchamp & Seech (1990) that evaluated impacts of 233 dry- and wet-sieving methods on aggregate reactivity. They observed decreased denitrification 234 235 rates as the dry-sieved aggregate size increased, but the opposite relationship for wet-sieved aggregates. More experiments testing the effects of separation techniques on GHG activities across 236 aggregates of different sizes are needed to inform future studies of environmental effects on 237 238 aggregate reactivity.

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#### 240 **4** Aggregate reactors and soil-atmosphere GHG exchange

241 To link the fine-scale aggregate reactors and soil profile GHG exchange, here we argue that a systematic, dynamic view of the soil system is required. Specifically, we need to focus on 242 243 both the reactivity of an individual aggregate and also the composition of aggregate reactors of 244 different sizes in a soil system. This dual focus is necessary because aggregate reactors of different 245 sizes collectively make different contributions to soil profile GHG exchange. For example, 246 Bandyopadhyay & Lal (2014) report a much higher contribution to bulk soil CO<sub>2</sub> emissions from 247 macro-aggregates than micro-aggregates. The relative proportions of different size aggregate 248 reactors in a soil system are dynamically changing, and these changes are determined by the 249 turnover of aggregates (i.e. aggregate stability). Aggregate turnover is strongly influenced by 250 external disturbances to soil systems (Six et al. 2004). This by analogy is similar to the space-251 lifetime hypothesis for organisms proposed by Ginzburg & Damuth (2008). We should view an 252 aggregate reactor in four dimensions-in addition to its three dimensional spatial structure, one

more temporal dimension, aggregate lifetime, should be included. Therefore, as with ecological systems (e.g., Levin 1998; Grimm et al. 2005; Wang et al. 2017b), we need a dynamic view of aggregate reactors to account for their compositional dynamics (**Fig. 1**).

Current evidence has already unequivocally suggested that aggregate reactors have 256 significant implications for soil-atmosphere exchange of GHG. Formation of these reactors 257 258 suppresses carbon oxidation and CO<sub>2</sub> release because of oxygen limitation, promoting carbon sequestration (Six et al. 2002; Keiluweit et al. 2016; Keiluweit et al. 2017). While revisiting two 259 prior studies (Greenwood 1961; Sexstone et al. 1985), Keiluweit et al. (2016) offered an initial 260 261 estimate of aggregation effects on bulk soil carbon mineralization, showing a striking suppression in the range of 23-97.5% relative to fully aerobic soils. By contrast, when these reactors are 262 'destroyed' (e.g., by disturbance from tillage), carbon mineralization increases rapidly because of 263 increased O<sub>2</sub> availability (e.g. Elliott 1986, Beare et al. 1994, and Drury et al. 2004). In particular, 264 Keiluweit et al. (2017) recently reported that shifting from anaerobic to aerobic conditions leads 265 266 to a 10-fold increase in volume-specific mineralization rate, illustrating the sensitivity of anaerobically protected carbon to disturbance. These results, meanwhile, indirectly substantiate 267 the strong physical protection of organic matter against decomposition offered by the aggregates 268 269 (Six et al., 2000a, 2002).

In addition, the aggregate reactor concept illustrates that most field measurements of net soil–atmosphere exchange mask significant gross production and consumption of CH<sub>4</sub> and N<sub>2</sub>O. Fine-scale activities are disguised in the traditional metrics of soil-atmosphere exchange of CH<sub>4</sub> at larger scales (von Fischer & Hedin, 2002), where a soil is considered either a sink or a source. Multiple studies have shown that higher *in situ* gross CH<sub>4</sub> production could stimulate higher gross consumption, resulting in little difference in surface fluxes (Kammann *et al.*, 2009; Mangalassery

*et al.*, 2013; Yang & Silver, 2016). Similar to CH4, N<sub>2</sub>O is could also be both produced and
consumed within a soil (Chapuis-Lardy *et al.*, 2007), so aggregate scale production of N<sub>2</sub>O may
not always increase surface fluxes (Yang & Silver, 2016). In summary, aggregate reactors have
significant implications for soil-atmosphere exchange of GHG.

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### 281 5 Aggregate reactors in the context of global change

The soils beneath our feet are strongly affected by coarse-scale disturbances including soil 282 management practices (mostly agronomic practices), land use change, and global changes resulting 283 from growing human activities in the Anthropocene Epoch (Bronick & Lal 2005; Hinckley, 2014; 284 Paustian et al. 2016). With the aggregate reactor concept and a systematic view of soil systems as 285 discussed above, coarse scale perturbations of soil systems are postulated to affect soil-atmosphere 286 287 GHG exchange both directly by influencing conditions for aggregate reactivity and indirectly by altering the distribution of aggregate reactor sizes (Fig. 1). Soil management (e.g., soil cultivation, 288 289 fertilization, crop rotation, irrigation, biochar addition, and compaction) and land use change can significantly affect the size distribution of aggregates (Six et al., 1999, 2000b; Young et al., 2000; 290 Wang et al., 2013b). Climate warming increases the soil temperature, while precipitation shifts 291 292 alter soil moisture content, shaping the environment of aggregate reactors and thus aggregate-level microbial activities (e.g., Fang et al. 2015). Moreover, climate change can also indirectly affect 293 294 soil aggregate properties by influencing vegetation activity (Torn et al., 2015). Additionally, 295 atmospheric changes can indirectly affect soil structure by influencing above-ground vegetation 296 activities and carbon and nitrogen allocation. Rising CO<sub>2</sub> levels can alter soil structure and increase 297 soil aggregation and carbon sequestration (Cotrufo et al., 2013; Rillig et al., 1999; Six et al., 2001a; 298 Jastrow et al., 2005; Dorodnikov et al., 2009). Nitrogen deposition can also shape soil aggregate

properties by influencing rhizodeposition, microbial biomass and microbial activity (Janssens *et al.*, 2010). Increasing abundance of tropospheric ozone (O<sub>3</sub>), the most important secondary air
pollutant, can modify the soil structure in terms of aggregate properties and distribution and soilatmosphere GHG exchange (Kou *et al.*, 2014; Wang et al. 2017c).

Still, relatively few studies have addressed direct connections between these disturbances 303 304 and reactions responsible for GHG production from aggregate reactors. The available studies mainly focus on aggregate responses to tillage and fertilization, while the indirect effects mediated 305 by aggregate turnover and size distribution changes are still unknown. In general, macro-306 307 aggregates from no-till soils have higher CO<sub>2</sub> production than those from soils under conventional 308 tillage (Franzluebbers & Arshad, 1997; Fernández et al., 2010). Moreover, the tillage impact depends upon soil depth. Fernández et al. (2010) demonstrated that differences in CO<sub>2</sub> production 309 310 between tillage practices disappear for deeper soils. One possible explanation is that no-till soils show a pattern of decreasing of SOC with depth whereas conventional soils have uniformly 311 312 distributed SOC (Fernández et al., 2010; Plaza-Bonilla et al., 2014). This pattern might also explain why Plaza-Bonilla et al. (2014) did not observe differences between no-till and 313 conventional tillage. 314

Similar to CO<sub>2</sub>, CH<sub>4</sub> production and consumption is affected by tillage. A study by Plaza-Bonilla *et al.* (2014) reported that macro-aggregates act as CH<sub>4</sub> sources under conventional tillage and sinks under no tillage. This change could be attributed to inhibited methanotrophic activity induced by aggregate destruction under tillage, or, alternatively, to a smaller quantity of anoxic microsites within the no-tillage macro-aggregates maintained by intra-aggregate pore architecture and connectivity (*e.g.*, Brewer *et al.*,2018). In contrast, both Jiang *et al.* (2011) and Plaza-Bonilla *et al.* (2014) reported that soil tillage did not affect aggregate N<sub>2</sub>O production.

Fertilization has been reported to affect aggregate-level N<sub>2</sub>O production with the effects dependent on fertilizer type. For example, Plaza-Bonilla *et al.* (2014) concluded that mineral and organic nitrogen fertilizers can lead to differences in the relative importance of nitrification versus denitrification in macro-aggregates: nitrification dominates with mineral fertilizer whereas denitrification dominates with organic fertilizer (pig slurry). This difference was attributed to changes in the proportion of C and N substrates and in microbial activities (Plaza-Bonilla *et al.* 2014).

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#### **6** Prospects and aggregate-based modeling (ABM)

Generations of research have built a relatively solid knowledge of reactions broadly 331 responsible for GHG production and of aggregates' inherent biophysical and chemical properties. 332 Yet an understanding of the direct relationships between these properties and reactions specifically 333 at the aggregate level remains elusive. Emerging new mechanisms, such as anaerobic oxidation of 334 335 CH<sub>4</sub>, require additional investigation. Likewise, bulk soil conditions are still not yet fully linked to aggregate reactivity. Aside from the soil hydration-O<sub>2</sub> diffusion relationship (e.g., Ebrahimi & 336 Or, 2017), many abiotic and biotic factors that could regulate aggregate reactivity have not been 337 338 studied. As pointed out by Torn et al. (2015), two of the most widespread impacts of anthropogenic activities on soils in this century will be warmer temperatures and altered plant allocation 339 340 belowground because of rising CO<sub>2</sub> and nitrogen deposition. Therefore, more studies are needed 341 to understand how soil temperature and biotic factors (e.g., root activity, plant species, and soil 342 macro-fauna), as well as soil management practices, affect aggregate GHG fluxes. Additionally, how microbial community composition and dynamics control these activities in aggregates is 343 344 almost unknown (Allison et al. 2013; Ebrahimi & Or, 2016; Buchkowski et al. 2017). Finally,

future studies should move beyond the reactivity of an individual aggregate. A systematic, dynamic view of these reactors in soil systems is equally important for fully quantifying the implications of aggregate reactors for soil-atmosphere GHG exchange.

Driven by these grand questions, we should simultaneously refine and develop more 348 standardized methodologies for the study of aggregate reactors. Undoubtedly, the ideal route is to 349 350 conduct in situ measurements, which is a great challenge at the micron-scale with current techniques. Therefore, we recommend a combination of the following techniques. First, it is 351 essential to establish a uniform experimental protocol for aggregate fractionation. The approach 352 353 of optimal moisture fractionation by Bach and Hofmockel (2014) has a high potential to separate aggregates with minimal disturbance to chemical and biological properties and might be ideal for 354 studies of aggregre reactors. As a complement to aggregate fractionation, artificial aggregates are 355 a good means for exploring the relationships between aggregate reactivity and physical structure 356 (e.g., Ebrahimi and Or 2017; Schlüter et al. 2018). Another area of technique development could 357 358 exploit isotope pool dilution to measure gross fluxes of GHGs, followed by separation of soil aggregates to determine correlations between aggregate size distributions, physio-chemical 359 properties, and the gross gas fluxes [e.g., CH<sub>4</sub> by von Fischer & Hedin (2002) and N<sub>2</sub>O by Yang 360 361 et al., (2011)]. These methods could be combined with technique advancements in computer-aided tomography (CT) and electron microscopy (e.g., SEM and TEM) (Williams & Carter 1996) that 362 363 provide soil structural information in terms of aggregate reactor size and distribution (e.g., Young 364 & Crawford 2004; Rabbi et al. 2016).

Finally, a major research need is to develop computational models that can quantify and predict aggregate reactivity. These models could untangle the non-linearities between aggregate reactors and soil profile GHG exchange. Such research would address the grand challenge of

modelling soil biogeochemical processes at larger spatial and temporal scales in the context of 368 369 global environmental changes. With a systematic, dynamic view of aggregate reactors in soil 370 systems as discussed above, we propose a bottom-up strategy to develop aggregate-based models (ABM) that explicitly represent the 'behavior' of aggregate reactors of different sizes (Fig. 1). This 371 372 approach is inspired by the agent-based or individual-based modeling (IBM) strategy that largely 373 originated in ecological systems in the 1960s (Grimm et al. 2005; Shugart et al. 2018). For a soil system composed of aggregate reactors of different sizes, an ABM framework could be developed 374 to represent these different aggregates. For each aggregate reactor, a single IBM would be 375 376 developed to explicitly simulate microbial communities and their functions. Therefore, an ABM 377 is expected to be a hierarchy of individual-based models simulating each of an aggregate element and its dynamic properties. Such a hierarchically constructed ABM contrasts with traditional 378 models that represent soils as a set of discrete carbon fractions with an implicit treatment of 379 microbial diversity (e.g., reviewed in Bradford et al., 2016). We also advocate the incorporation 380 381 of a trait-based approach in the ABM based on trait data from aggregate reactors. These data could 382 include the physical (e.g., aggregate size, pore size, gas diffusion coefficient), chemical (e.g., substrates, enzyme, O<sub>2</sub>, and moisture), and biological properties (i.e., turnover rate and microbial 383 384 diversity) of soil aggregates and could be obtained by the techniques described above (i.e. 'top-385 down experiment'; Fig.1). We anticipate that a global soil aggregate trait database can be 386 established and that tradeoffs among aggregate traits (e.g., reactor size and O<sub>2</sub> abundance) could 387 be uncovered to facilitate the model construction, similar to previous successful applications with 388 plant traits (e.g., Kattge et al. 2011; Wright et al. 2014) and even litter decomposition (Allison 389 2012). Biophysical equations that may be helpful in building such an ABM are listed in the 390 Supporting Information.

Previous workers have built a solid foundation for developing ABMs based on mechanistic 391 modelling of soil processes that occur in aggregate reactors. Smith (1980) developed a model of 392 393 the variation in the extent of anaerobiosis in aggregated soils by extending previously published models of radial diffusion into individual aggregates (e.g., Currie 1962). This work assumed a log-394 normally distributed population of aggregate sizes to calculate denitrification rates. Recently, 395 396 Ebrahimi & Or (2015) embedded an individual-based microbial model [inherited from Kreft et al. (1998)] into an idealized artificial aggregate and developed an analytical model for 397 biogeochemical processes in aggregates. The model was later expanded to include aggregates of 398 399 different sizes to simulate CO<sub>2</sub> and N<sub>2</sub>O fluxes (Ebrahimi & Or 2016). Moreover, Ebrahimi & Or 400 (2018) applied their aggregate-based model to scale up microbial processes in aggregates of different sizes. They used spatial data on soil type and land cover to simulate GHG exchange at 401 402 the landscape scale. Future modeling efforts should dynamically represent aggregate reactivity and distribution and microbial community composition over space. The resulting models should be 403 404 validated under different soil conditions and management practices across spatial and temporal scales. These efforts will likely require cooperation among modelers, ecologists, microbiologists, 405 and climate scientists to advance a predictive science of land-atmosphere exchange of GHG in the 406 407 context of global environmental change (BERAC 2017).

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#### 409 **7** Conclusions

410 Soil GHG exchanges are manifested at a wide spectrum of scales from the aggregate 411 through the soil profile to the landscape, region, and globe. Understanding GHG exchange at these 412 different scales is essential for a more accurate quantification of trace gas fluxes and better 413 evaluation of land-atmosphere interactions in the context of accelerating global change. Based on

a synthesis of prior studies, we argue that aggregates can be viewed as biogeochemical reactors of 414 GHGs, with reactivities dependent on aggregate size and bulk soil abiotic and biotic factors that 415 shape the reaction environment. We also suggest a systematic, dynamic approach to link the 416 individual aggregate reactor with GHG exchange from the soil profile. Prior work has already 417 418 revealed implications of aggregate reactors for the soil-atmosphere exchange of GHG-for 419 example, soil carbon oxidation can be suppressed in aggregate reactors to facilitate carbon sequestration. However, our understanding of aggregate reactors is far from complete. We 420 advocate for more research on techniques, environmental drivers, and cross-scale linkages related 421 422 to the aggregate reactor concept. There is also great potential for developing mechanistic, aggregate-based models that use a trait-based approach to represent soil systems and reduce 423 uncertainties about soil-atmosphere GHG exchange in the face of human impacts. 424

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Land use*	Soil texture	Method	Size class (mm)	Moisture‡	Rate vs size∥	Reference
Grassland	Silty loam	Dry†	6.3-2, <2	FC	+	Bimüller et al. 2016
Pasture	Clay	Wet	2-0.25, 0.25-0.053, <0.053	70% FC	=	Rabbi et al. 2015
Crop	Clay	Wet	2-0.25, 0.25-0.053, <0.053	70% FC	=	Rabbi et al. 2015
Forest	Clay	Wet	2-0.25, 0.25-0.053, <0.053	70% FC	=	Rabbi et al. 2015
Crop	Silt loam	Wet	>2, 2-0.25, 0.25-0.053, <0.053	FC	+	Bandyopadhyay & Lal 2014
Crop (NT)	Silt loam	Wet	>2, 2-0.25, 0.25-0.053, <0.053	FC	+	Bandyopadhyay & Lal 2014
Forest	Silt loam	Wet	>2, 2-0.25, 0.25-0.053, <0.053	FC	+	Bandyopadhyay & Lal 2014
NA	Clay loam	Dry	4-2, 2-1, 1-0.5, <0.5	FC	-	Mangalassery et al. 2013
NA	Sandy loam	Dry	4-2, 2-1, 1-0.5, <0.5	FC	+	Mangalassery et al. 2013
Crop	Silt loam	Dry	>6, 6-4, 4-2, 2-1, 1-0.5, 0.5-0.25	60% FC	-	Muñoz et al. 2012
Tropical forest	Peat	Dry	20-8, 8-2, <2	30%&70% FC	+	Kimura et al. 2012
Oil palm	Peat	Dry	20-8, 8-2, <2	30%&70% FC	=	Kimura et al. 2012
Grassland	NA	Dry	4.5-2, <2	60% & 80% FC	+	Diba et al. 2011
Crop (NT/CT)	Sandy loam	Dry	>4, 4-1, <1	80% FC	+	Fernández et al. 2010
Crop	Sandy loam	Dry	6-2, 2-0.25, <0.25	20% - 80% WFPS	-	Sey et al. 2008
Crop	Clay loam	Dry	8-4, 4-2, 2-1, 1-0.5, 0.5-0.25, <0.25	30% GWC	-	Drury et al. 2004
Crop	Silt loam	Dry	5-2, 2-0.25,<0.25	67% FC	+	Schutter & Dick 2002
Crop	Loam	Wet	>1, 1-0.5, 0.5-0.25, <0.25	NA	+	Aoyama et al. 1999
Crop	Loam	Wet	5.6-1.0, 1.0-0.25, 0.25-0.05, <0.05	FC	+	Franzluebbers & Arshad 1997
Crop	Silt loam	Wet	5.6-1.0, 1.0-0.25, 0.25-0.05, <0.05	FC	+	Franzluebbers & Arshad 1997
Crop	Clay loam	Wet	5.6-1.0, 1.0-0.25, 0.25-0.05, <0.05	FC	+	Franzluebbers & Arshad 1997
Crop	Clay	Wet	5.6-1.0, 1.0-0.25, 0.25-0.05, <0.05	FC	+	Franzluebbers & Arshad 1997
Crop	Clay loam	Dry	20-10, 10-5, 5-2, 2-1, 1-0.5, 0.5-0.25, <0.25	FC	-	Seech & Beauchamp 1988
Prairie	Sandy loam	Dry†	8-1, 1 - 0.5, 0.5 - 0.25, 0.25 - 0.1, < 0.1	Field-moist	+	Gupta & Germida 1988
Prairie (CT)	Sandy loam	Dry†	8-1, 1 - 0.5, 0.5 - 0.25, 0.25 - 0.1, < 0.1	Field-moist	+	Gupta & Germida 1988

 Table 1 A compilation of research on aggregate reactors in terms of CO2.

\* The blank denotes information not available. NT and CT denote no tillage and conventional tillage, respectively.

† These studies directly used field-moist soils for dry sieving, which remain being labeled as dry sieving.

‡ Incubation moisture expressed in % field capacity (FC), in water-filled pore space (WFPS), or in gravimetric water content (GWC) based on literature.

I Sign '+' generally denotes a positive relationship of CO<sub>2</sub> production rate with aggregate size, '-' negative, while '=' no significant relationship.

For more detailed information of each study, see Table S1,S2 in supplementary information.

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Land use	Soil texture	Method	Size class (mm)	Moisture	Rate vs size	Reference
Crop	Clay loam	Dry	5.6-4, 4-2, 2-1	Aeration	+	Robinson et al. 2014
Crop	Silt loam	Wet	>2, 2-0.25, 0.25-0.053, <0.053	FC	+	Bandyopadhyay & Lal 2014
Crop(NT)	Silt loam	Wet	>2, 2-0.25, 0.25-0.053, <0.053	FC	+	Bandyopadhyay & Lal 2014
Forest	Silt loam Sandy	Wet	>2, 2-0.25, 0.25-0.053, <0.053	FC	+	Bandyopadhyay & Lal 2014
NA	loam	Dry	4-2, 2-1, 1-0.5, <0.5	FC	=	Mangalassery et al. 2013
NA	Clay loam	Dry	4-2, 2-1, 1-0.5, <0.5	FC	=	Mangalassery et al. 2013
Crop	Silt loam	Dry	>6, 6-4, 4-2, 2-1, 1-0.5, 0.5-0.25	60% FC	-	Muñoz et al. 2012
Oil palm	Peat	Dry	20-8, 8-2, <2	30%&70% FC	+	Kimura et al. 2012
Tropical						Kimura et al. 2012
forest	Peat	Dry	20-8, 8-2, <2	30%&70% FC	+	
Crop	NA	Wet	>2, 2-0.25, 0.25-0.053, <0.053	60% FC	+	Jiang et al. 2011
Grassland	NA Sandy	Dry	4.5-2, <2	60% & 80% FC	+	Diba et al. 2011
Crop	loam	Dry	6-2, 2-0.25, <0.25	20% - 80% WFPS	+	Sey et al. 2008
Pasture	Silt loam	Dry	5.6-4, 4-2, 2-1, <1	FC	-	Uchida et al. 2008
Crop	Clay loam	Dry	8-4, 4-2, 2-1, 1-0.5, 0.5-0.25, <0.25	30% GWC	+	Drury et al. 2004
NA	Loamy	Wet	5.0-3.0, 3.0-2.0, 2.0-1.0, 1.0-0.5, 0.5-0.25,<0.25	Dry	+	Manucharova et al. 2001
NA	NA	NA	10-4.0	NA	+	Stepanov et al. 1997
Crop	Silt loam	Wet	>4.7, 4.7-2, 2-1, 1-0.5, 0.5-0.25, 0.25-0.1, <0.1	FC	+	Beauchamp et al. 1990
Crop	Silt loam	Dry	>20, 20-10, 10-5, 5-2, 2-1, 1-0.5, 0.5-0.25, 0.25- 0.15,0.15-0.05,<0.05	FC	-	Beauchamp et al. 1990
Crop	Silt loam	Dry	20-10, 10-5, 5-2, 2-1, 1-0.5, 0.5-0.25, <0.25	Saturation	-	Seech & Beauchamp 1988

Table 2. A compilation of research on aggregate reactors in terms of N<sub>2</sub>O.

See Table 1 notes on information listed. For more detailed information of each study, see Table S1,S3 in supplementary information.

Land use	Soil texture	Method	Size class (mm)	Moisture	Rate vs size†	Reference
Crop	Silt loam	Wet	>2, 2-0.25, 0.25-0.053, <0.053	FC	=	Bandyopadhyay & Lal 2014
Crop (NT)	Silt loam	Wet	>2, 2-0.25, 0.25-0.053, <0.053	FC	=	Bandyopadhyay & Lal 2014
Forest	Silt loam	Wet	>2, 2-0.25, 0.25-0.053, <0.053	FC	=	Bandyopadhyay & Lal 2014
NA	Clay loam	Dry	4-2, 2-1, 1-0.5, <0.5	FC	-	Mangalassery et al. 2013
NA	Sandy loam	Dry	4-2, 2-1, 1-0.5, <0.5	FC	-	Mangalassery et al. 2013
Tropical forest	Peat	Dry	20-8, 8-2, <2	30% & 70% FC	+	Kimura et al. 2012
Oil palm	Peat	Dry	20-8, 8-2, <2	30% & 70% FC	+	Kimura et al. 2012
Crop	Sandy loam	Dry	6-2, 2-0.25, <0.25	20% - 80% WFPS	+/=/-	Sey et al. 2008
Crop	Loam	Dry	<2, 2-4, 4-6, 6-8, 8-10, >10	7-97% FC	+	Jäckel et al. 2001
Forest	Loam	Wet	>2, 2-0.25, 0.25-0.053, < 0.053	Dry	+	Wang et al. 2013a‡

Table 3. A compilation of research on aggregate reactors in terms of CH<sub>4</sub>.

\* See Table 1 notes on information listed;

† Pattern '+' denotes larger aggregates are CH<sub>4</sub> producers and smaller aggregates are consumers (or less production than larger aggregates); '-' denotes larger aggregates have less production than smaller aggregates; and '=' denotes no significant pattern is observed;

<sup>‡</sup> The only study on aggregate-scale non-microbial CH<sub>4</sub>.

For more detailed information of each study, see Table S1,S4 in supplementary information.

754	Fig. 1 Schematic of soils as a system of aggregate reactors of different sizes. At the profile level,
755	soils act as a source of $CO_2$ , either a source or a sink of $CH_4$ (denoted by the upward and downward
756	arrow, respectively), and a source of N2O. At fine scales, soil consists of aggregate reactors of
757	differing sizes. Each individual aggregate reactor can be described by physical (e.g., pore size,
758	diffusion coefficient, and aggregate size), chemical (e.g., concentration of O <sub>2</sub> , H <sub>2</sub> O, dissolved
759	organic carbon-DOC, substrates), and biological traits (turnover rate, microbial community
760	composition and dynamics). Top-down experiment refers to studying these properties by 'digging'
761	into soils. Aggregate reactivity depends on aggregate size (denoted by the irregular circles with
762	differing colors) and bulk soil properties including both abiotic and biotic factors, as well as coarser
763	scale anthropogenic disturbances. Different widths of the red arrows denote the reactor size-
764	induced variations in GHGs. Soil systems composed of aggregates of different sizes are
765	dynamically changing because of aggregate turnover (or aggregate stability), which is not
766	illustrated here. Bottom-up modelling refers to building models based on aggregate reactor that
767	can represent soil system composition and dynamics and simulate soil profile GHG exchange as
768	an emergent process.
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Bottom-up modelling: aggregate-based model (ABM)