

## **UC Irvine**

### **UC Irvine Electronic Theses and Dissertations**

**Title**

An Enantiospecific Formal Synthesis of (+)-7,20-Diisocyanoadociane

**Permalink**

<https://escholarship.org/uc/item/3w69n33g>

**Author**

Roosen, Philipp Christopher

**Publication Date**

2016

Peer reviewed|Thesis/dissertation

UNIVERSITY OF CALIFORNIA,  
IRVINE

An Enantiospecific Formal Synthesis of (+)-7,20-Diisocyanoadociane

DISSERTATION

submitted in partial satisfaction of the requirements for the degree of

DOCTOR OF PHILOSOPHY

In Chemistry

By

Philipp Christopher Roosen

Dissertation Committee:  
Professor Christopher D. Vanderwal, Chair  
Professor Larry E. Overman  
Professor Sergey V. Pronin

2016

Portions of Chapter 3 are adapted, according to the ACS AuthorChoice agreement, from:  
Roosen, P. C.; Vanderwal, C. D. *Org. Lett.* **2014**, *16*, 4368 © American Chemical Society

Portions of Chapter 5 are adapted from: Roosen, P. C.; Vanderwal, C. D. *Angew. Chem. Int. Ed.*  
**2016**, *55*, 7180–718355; *Angew. Chem.* **2016**, *128*, 7296–7299.

© 2016 Philipp Christopher Roosen

**DEDICATION**

*To My Loving Wife Michelle Coscia-Roosen*



## TABLE OF CONTENTS

	Page
<b>LIST OF FIGURES</b>	<b>vi</b>
<b>LIST OF TABLES</b>	<b>vii</b>
<b>LIST OF SCHEMES</b>	<b>viii</b>
<b>LIST OF ACROYNMS AND ABBREVIATIONS</b>	<b>xii</b>
<b>ACKNOWLEDGMENTS</b>	<b>xv</b>
<b>CURRICULUM VITAE</b>	<b>xvii</b>
<b>ABSTRACT OF THE DISSERTATION</b>	<b>xx</b>
<b>CHAPTER 1: BACKGROUND TO (+)-7,20-DIISOCYANOADOCIANE</b>	<b>1</b>
1.1 Introduction	1
1.2 Isolation and Structure Determination	1
1.3 Biosynthesis	4
1.4 Function and Ecology	7
1.5 Biological Activity	8
1.5.1 Introduction	8
1.5.2 Studies on Antiplasmodial Activity	10
1.6 Previous Syntheses of Cycloamphilectanes and Isocycloamphilectanes	12
1.6.1 Corey's Synthesis (+)-7,20-Diisocyanoadociane (1987)	12
1.6.2 Piers's Synthesis (+)-8-Isocyano-10-Cyloamphilectene (1998)	14
1.6.3 Mander's Formal Synthesis of (±)-7,20-Diisocyanoadociane (2006)	17
1.6.4 Miyaoka's Formal Synthesis of (+)-7,20-Diisocyanoadociane (2011)	20
1.7 Goals for the Synthesis	22
1.8 References and Notes	23
<b>CHAPTER 2: INITIAL EFFORTS TOWARDS 7,20-DIISOCYANOADOCIANE</b>	<b>27</b>
2.1 Introduction	27
2.2 A Polyene Imino–Pinacol Coupling Approach	27
2.3 A Diels–Alder/Aldol Condensation Pursuit	29
2.4 The Methylenation/Reductive Ring Closure Approach	32
2.4.1 Application of Danishefky's <i>trans</i> -Diels–Alder Paradigm	32
2.4.2 Efforts Towards a <i>trans</i> -Selective Conjugate Hydrosilylation Approach	34

2.4.3	A Reductive Stork Alkylation	38
2.5	Conclusions	41
2.6	Experimental Procedures	42
2.7	References and Notes	71
<b>CHAPTER 3: AN OXY-COPE/TRANSANNULAR MICHAEL APPROACH TOWARDS 7,20-DIISOCYANOADOCIANE</b>		<b>73</b>
3.1	Background on the Electronic Reorganization of 3-Oxy-1,5-Hexadiene Systems (the oxy-Cope Rearrangement)	73
3.2	Literature Precedent of an oxy-Cope Rearrangement with Relevance to 7,20-Diisocyanoadociane	77
3.3	Synthetic Efforts to Realize the oxy-Cope/Michael Cascade Towards 7,20-Diisocyanoadociane	79
3.4	A Literature Reevaluation of the Transannular Michael Stereochemistry	87
3.5	Evaluating the Reversibility of the Transannular Michael Reaction	89
3.6	Outlook for the oxy-Cope/Michael Reaction in Natural Product Synthesis	91
3.7	Conclusions	94
3.8	Experimental Procedures	96
3.9	References and Notes	118
<b>CHAPTER 4: EXPLORING A PHENANTHRENONE REDUCTION APPROACH TO 7,20-DIISOCYANOADOCIANE</b>		<b>122</b>
4.1	Introduction	122
4.2	Entry into a Phenanthrenone Reduction Route	122
4.3	Results from a Phenanthrenone Reduction Approach	125
4.3.1	Phenanthrenone Synthesis and Confirmation of Basic Birch Reactivity	125
4.3.2	A Reworked Phenanthrenone Approach Emphasizing Stereocontrol	128
4.3.3	Identifying a Conflicting Stereochemical Outcome	133
4.3.4	Settling the Phenanthrenone Birch Reduction Stereochemical Conundrum	135
4.4	Phenanthrenone Birch Reduction Optimization Attempts	138
4.5	Conclusions	141
4.6	Experimental Procedures	142
4.7	References and Notes	164
<b>CHAPTER 5: A FORMAL SYNTHESIS OF (+)-7,20-DIISOCYANOADOCIANE</b>		<b>167</b>
5.1	Introduction	167
5.2	A New Design to Control the All- <i>trans</i> Stereochemistry	167
5.3	The C8–C9 Ring Closing Strategy	168
5.3.1	Successes in Generating <i>trans</i> -Ring Fusions	168

5.3.2	Considering a Sc(OTf) <sub>3</sub> -Mediated Isonitrile Installation to 7,20-Diisocyanoadociane	172
5.4	A Global Approach to 7,20-Diisocyanoadociane and Other (Iso)Cycloamphilectanes	177
5.4.1	The Evolution to a C10–C11 Ring Closing Strategy	177
5.4.2	A Proof-of-Principle Preparation of a Perhydrophyrene Relevant to 7,20-Diisocyanoadociane	179
5.4.3	Improving the THF C–H Oxidation to a Butyrolactone	187
5.4.4	Finishing the Synthesis of Corey’s Dione	190
5.5	An Asymmetric Formal Synthesis of (+)-7,20-Diisocyanoadociane	192
5.5.1	The Asymmetric Robinson Annulation Reaction	192
5.5.2	A Perillaldehyde-Based Formal Synthesis	195
5.6	Outlook Towards the Synthesis of 7-Isocyano-(Iso)Cycloamphilectanes	197
5.7	Conclusions	200
5.8	Experimental Procedures	201
5.9	References and Notes	289
<b>APPENDIX A: NMR Data</b>		<b>293</b>
<b>APPENDIX B: X-ray Crystallographic Data</b>		<b>513</b>

## LIST OF FIGURES

		Page
Figure 1.1	The structure of (+)-DICA.	1
Figure 1.2	Isoprene units of the cycloamphilectane skeleton and C16 to C20 methyl shift to access the Isocycloamphilectane skeleton	5
Figure 3.1	Kempene and rippertane structures.	83
Figure 3.2	Stereochemical determination of oxy-Cope/Michael reaction product by Swaminathan. "Dione 6" refers to structure <b>3.20</b> .	87
Figure 3.3	NMR study in CD <sub>3</sub> OD to determine the reversibility of the Michael reaction.	90
Figure 3.4	Pseudopterosin aglycones	92
Figure 4.1	Rationalizing the phenanthrenone reduction stereochemistry.	140
Figure 5.1	Comparison of protonation events to explain the preference for <i>cis</i> -reduction.	184
Figure 5.2	A collection of (iso)cycloamphilectane natural products.	197

## LIST OF TABLES

		Page
Table 1.1	Initial evaluation of antiplasmodial activity and cytotoxicity of ICTs.	9
Table 1.2	Evaluation of Isocycloamphilectane ICTs against <i>P. falciparum</i> .	9
Table 2.1	Consolidation of conjugate hydrosilylation reactions.	37
Table 3.1	Evaluation of copper mediated conjugate additions into <b>3.31</b> .	80
Table 4.1	Small screen of reduction conditions in attempting to fully reduce <b>4.49</b> .	131
Table 4.2	Impact of alcohol additive on stereochemistry.	139
Table 4.3	Correcting the stereochemical outcome described in Table 4.1.	139
Table 5.1	Optimizing for a <i>trans</i> -reduction of enone <b>5.70</b> .	184
Table 5.2	Screening RuO <sub>4</sub> mediated THF C–H Oxidations.	188
Table 5.3	A Cr-based screen for C–H oxidation.	189
Table 5.4	Miscellaneous C–H oxidation attempts to install a lactone functionality.	190
Table 5.5	Screening the asymmetric conjugate addition reaction.	193
Table 5.6	Conditions to effect the aldol condensation of enantioenriched Michael adduct <b>5.93</b> .	195

## LIST OF SCHEMES

	Page
Scheme 1.1	Assignment of the absolute stereochemistry of (+)-DICA. 3
Scheme 1.2	Established biosynthetic relationship between cyanide, thiocyanate and their respective terpene derivatives. 6
Scheme 1.3	Corey's retrosynthetic analysis of (+)-DICA via double Diels–Alder reaction. 12
Scheme 1.4	Corey's synthesis of (+)-DICA. 14
Scheme 1.5	Piers's retrosynthetic analysis of (+)-8-isocyano-10-cycloamphilectene based on a Piers–Stille annulation and previous amphilectane syntheses. 15
Scheme 1.6	Piers's synthesis of (+)-8-isocyano-10-cycloamphilectene. 16
Scheme 1.7	Mander's impetus and inspiration for the synthesis of DICA. 18
Scheme 1.8	Mander's formal synthesis of (±)-DICA. 19
Scheme 1.9	Miyaoka's double Diels–Alder strategy to Corey's dione <b>1.37</b> . 20
Scheme 1.10	Miyaoka's formal synthesis of (+)-DICA. 21
Scheme 2.1	An initial polyene imino–pinacol coupling idea. 28
Scheme 2.2	An attempted bis-oxy-Cope rearrangement. 28
Scheme 2.3	A Diels–Alder/aldol condensation proposal to DICA. 29
Scheme 2.4	Initial success of the Diels–Alder/aldol condensation idea. 30
Scheme 2.5	Attempted Diels–Alder reaction via dienophile activation. 31
Scheme 2.6	Incomplete efforts towards a Diels–Alder/aldol condensation sequence. 32
Scheme 2.7	An ICT synthesis by Shenvi. 32
Scheme 2.8	General proposal for a final reductive ring closure 33
Scheme 2.9	Diels–Alder reactivity of an activated nitroalkene. 34

Scheme 2.10	Tracing tricycle bis-enoxysilane 2.41 back to a simple octalone starting material.	35
Scheme 2.11	Initial success of a trans-selective conjugate hydrosilylation.	35
Scheme 2.12	Exploring a Robinson annulation onto <b>2.79</b> .	38
Scheme 2.13	Allylation and subsequent ring closures based on a reductive Stork alkylation.	39
Scheme 2.14	A reductive Stork alkylation towards DICA.	40
Scheme 3.1	Rearrangement of the 3-oxy-1,5-hexadiene system.	74
Scheme 3.2	The two proposed mechanisms to explain the conversion of <b>3.4</b> to <b>3.5</b> .	74
Scheme 3.3	Initial mechanistic implications for a vinylogous retro-aldol mechanism	75
Scheme 3.4	Determining the mechanism for the rearrangement of 3.4 with the optically active tetrasubstituted alkene surrogate (+)- <b>3.12</b> .	77
Scheme 3.5	Heathcock's alternative reaction mechanisms depending of reaction conditions.	77
Scheme 3.6	Retrosynthetic analysis of DICA back to a spiro-fused oxy-Cope system.	79
Scheme 3.7	Generation of spirocyclic hydroxyenone <b>3.30</b> and <b>3.37</b> .	82
Scheme 3.8	Attempting an oxy-Cope/Michael cascade.	83
Scheme 3.9	An envisioned oxy-Cope/carbonyl-ene approach to kempane natural products.	84
Scheme 3.10	Silylation of spirocyclic enones in efforts towards an oxy-Cope rearrangement.	85
Scheme 3.11	Improved synthesis of [7.3.1]-bicycle <b>3.54</b> via an anionic oxy-Cope rearrangement and subsequent transannular Michael reaction.	86
Scheme 3.12	Reevaluation of the original oxy-Cope/Michael cascade.	88
Scheme 3.13	Proposed pathway for the equilibration of a perhydrophenalenedione system via a retro-Michael/transannular Michael sequence and calculations of perhydrophenalene stability.	89
Scheme 3.14	Proposed approach to all pseudopterosine aglycones.	93

Scheme 4.1	Origin of the phenanthrenone reduction idea.	123
Scheme 4.2	Proposed reaction course for the Birch reduction of a phenanthrenone system relevant to DICA.	123
Scheme 4.3	The stereochemical course of phenanthrenone Birch reductions.	125
Scheme 4.4	A tetralone synthesis and first phenanthrenone derivative.	126
Scheme 4.5	A successful phenanthrenone Birch reduction.	127
Scheme 4.6	A redesigned phenanthrenone Birch reduction route.	128
Scheme 4.7	First pass at the new phenanthrenone-based synthesis of DICA.	129
Scheme 4.8	A selective phenanthrenone reduction with Li/NH <sub>3</sub> .	130
Scheme 4.9	Successful ethylene diamine based reduction of phenanthrenone <b>4.49</b> .	133
Scheme 4.10	A des-methyl phenanthrenone undergoes successful Birch reduction.	134
Scheme 4.11	Alkene hydrogenation to determine the stereochemical course of the Birch reduction in conjugated phenanthrenones.	136
Scheme 4.12	Proposed difficulties isomerizing a skipped enone into conjugation.	137
Scheme 5.1	A previous attempt and a revision in using a Birch reduction approach to DICA.	168
Scheme 5.2	A Robinson annulation entry towards dihydronaphthalene studies of DICA.	169
Scheme 5.3	Efficient preparation of <b>5.2</b> by dihydrophthalene reduction.	170
Scheme 5.4	Proposed path forward to Corey's dione	170
Scheme 5.5	Early developments installing the two-carbon bridge.	171
Scheme 5.6	Use of an invertive isonitrile synthesis towards DICA.	173
Scheme 5.7	Exploring options for a selective diol synthesis.	174
Scheme 5.8	Efforts to install the C17 methyl group from the aromatic Grignard reagent.	175



Scheme 5.9	C17 Methyl installation by alkylation and subsequent ring closure strategies.	176
Scheme 5.10	Cycloamphilectane and isocycloamphilectanes with a conserved C7 and diverse C20 region.	178
Scheme 5.11	Exploring the reactivity after a vicinal difunctionalization.	180
Scheme 5.12	An initial procedure through a successful Birch reduction.	181
Scheme 5.13	An optimized vicinal difunctionalization and subsequent synthetic elaboration.	182
Scheme 5.14	Another unexpected stereochemical outcome from a dissolving metal reduction.	183
Scheme 5.15	Remaining requirements to complete Corey's dione.	185
Scheme 5.16	A successful fourth ring closure.	186
Scheme 5.17	End game to Corey's Dione.	192
Scheme 5.18	General approach to enantioenriched <b>5.7</b> .	193
Scheme 5.19	An enantiospecific formal synthesis of (+)-DICA from (–)-perillaldehyde.	196
Scheme 5.20	Proposed access to diol <b>5.26</b> utilizing reductive stereocontrol elements.	198
Scheme 5.21	General approach to 7-isocyano(iso)cycloamphilectanes.	199

## LIST OF ACROYNMS AND ABBREVIATIONS

Å	Ångstrom
°C	degrees Celsius
$[\alpha]_D^T$	specific rotation at wavelength of sodium D line at temperature T
Ac	acetate
acac	acetoacetate
AIBN	2,2'-azobisisobutyronitrile
aq.	aqueous
Ar	aryl
Bn	benzyl
bp	boiling point
bs	broad singlet
Bu	butyl
Bz	benzoyl
C <sub>6</sub> H <sub>6</sub>	benzene
cat.	catalytic
CCDC	Cambridge Crystallographic Data Centre
cm <sup>-1</sup>	wavenumber(s)
d	doublet
DABCO	1,4-diazabicyclo[2.2.2]octane
DBU	1,8-diazabicyclo[5.4.0]undec-7-ene
1,2-DCB	1,2-dichlorobenzene
DCE	1,2-dichloroethane
DIBAL	diisobutylaluminum hydride
DICA	7,20-diisocyanoadociane
DMAP	4-dimethylaminopyridine
DME	1,2-dimethoxyethane
DMF	dimethylformamide
3,5-DMP	3,5-dimethylpyrazole
DMP	Dess–Martin periodinane
DMPU	N,N'-dimethylpropylene urea
DMSO	dimethylsulfoxide
dr	diastereomeric ratio
EC <sub>50</sub>	half maximal effective concentration
EDA	1,2-ethylenediamine
EDTA	ethylenediaminetetraacetic acid
ee	enantiomeric excess
ESI	electrospray ionization
Et	ethyl
g	grams
HMDS	hexamethyldisilazide
HMPA	hexamethylphosphoramide
HPLC	high-performance liquid chromatography
HRMS	high-resolution mass spectrometry
HWE	Horner–Wadsworth–Emmons

Hz	hertz
i-Pr	iso-propyl
IBX	2-iodoxybenzoic acid
IC <sub>50</sub>	half maximal inhibitory concentration
ICT	isocyanoterpene
IR	infrared (spectroscopy)
<i>J</i>	coupling constant
Karstedt	platinum(0)-1,3-divinyl-1,1,3,3-tetramethyldisiloxane complex
LA	Lewis acid
LDA	lithium diisopropylamide
lit.	literature
mCPBA	meta-chloroperoxybenzoic acid
Me	methyl
mg	milligrams
MHz	megahertz
mL	milliliters
mmHg	millimeters of mercury
mmol	millimole
mp	melting point
Ms	methansulfonyl
MS	molecular sieves
MVK	methyl vinyl ketone
n-	normal
NBS	N-bromosuccinimide
NMO	N-methylmorpholine N-oxide
NMR	nuclear magnetic resonance
noe	nuclear Overhauser effect
NR	no reaction
PCC	pyridinium chlorochromate
PDC	pyridinium dichromate
Ph	phenyl
Phen	1,10-phenanthroline
Piv	pivaloyl
PPA	polyphosphoric acid
ppm	parts per million
PPTS	pyridinium para-toluenesulfonate
Pr	propyl
psi	pounds per square inch
q	quartet
Red-Al	sodium bis(2-methoxyethoxy)aluminum hydride
RSM	recovered starting material
RT	room temperature
s	singlet
<i>Si</i>	silicon-based protecting group
SI	selectivity index
SM	starting material

Stryker's	(triphenylphosphine)copper hydride hexamer
t	triplet
t-Bu	tert-butyl
TBAF	tetrabutylammonium fluoride
TBAI	tetrabutylammonium iodide
TBD	1,5,7-triazabicyclo[4.4.0]dec-5-ene
TBS	tert-butyldimethylsilyl
TEMPO	2,2,6,6-tetramethyl-1-piperidinyloxy
TES	triethylsilyl
Tf	trifluorosulfonyl
TFA	trifluoroacetic acid
TFAA	trifluoroacetic anhydride
TFDO	methyl(trifluoromethyl)dioxirane
THF	tetrahydrofuran
TIPS	triisopropylsilyl
TLC	thin layer chromatography
TMEDA	N,N,N',N'-tetramethylethane-1,2-diamine
TMS	trimethylsilyl
TPAP	tetrapropylammonium perruthenate
Trt	triphenylmethyl
Ts	para-toluenesulfonyl
UHP	urea hydrogen peroxide

## ACKNOWLEDGMENTS

I would like to thank my graduate advisor Prof. Chris Vanderwal for inviting me to join the lab and providing the right environment to explore chemistry for five years. Your dedication to total synthesis has been infectious. I appreciate the independence you offered in the whole of my experience, from the chemistry to investigate, the route selection and the decision to stop and pivot approaches. In addition to chemistry, I thank you for the flexibility during the growth of my family. The support and time off for the birth of my two daughters is not necessarily provided by every advisor, but it's something I will take with me.

Thank you Prof. Larry Overman for being both on my second year oral committee and my dissertation committee. The scholarship, curiosity, attention to detail and professionalism with which you approach chemical problems has been more than inspiring. I would like to also thank Prof. Sergey Pronin for being on my dissertation committee. Seeing first hand how dedicated you are to starting a lab has been eye opening. I look forward to the many successes to come out of your and your student's efforts. As a side note, knowing you were working on DICA was very intimidating but I was then very happy when you stopped and joined UCI.

The Vanderwal lab has been an amazing place to work. I thoroughly enjoyed our powerhouse bay (ex. circus bay) of Jon, Won-jin, Alex and Sharon. So many people came through the fifth floor bays that it's difficult to acknowledge everyone. Special thanks however should be given to Carl for being my first bay mate and Peter for the many late night discussions and chemistry fun. The ongoing projects are all exciting and I thank you group for tolerating my (typically unsolicited) advice. The group is an amazing place for friendships and I am thrilled to continue them after departing.

The whole UCI chemistry community is wonderfully integrated. I appreciate the friendship of people in many groups, whose interaction during colloquium organization, admissions work and just simple fun will stay with me.

My interest in chemistry did not arise out of thin air. My guidance and mentoring started with Profs. Vanessa McCaffrey, Cliff Harris and Andrew French at Albion College. This small college can boast of three fantastic organic professors, which all three opened me up to chemistry. My graduate advisor Prof. Mitch Smith cemented my concentration in chemistry. I appreciate his knowledge and ability to drive projects further than I had ever witnessed before and he was a fantastic mentor. Although our time together was short, without his mentorship I would not have the opportunities before me today and thank him for his continued relationship.

My family means the world to me. I couldn't thank you enough mom and dad for the foundation, support and persistence you instilled in me. This appreciation has only deepened as I now have children of my own. The hours of sacrifice to place education first have paid off in the most unexpected of ways. I look forward to many years of visits and enjoyment together.

After meeting each other at Albion, I could never have foreseen a decade like the one we experienced together, Coscia. An initial stint in the middle of Michigan and shuttling between Albion, Mason and East Lansing seemed too long ago. Thank you for being brave with me and

taking a leap out to California. Getting married and raising two beautiful girls has been most fulfilling. I admire you for a huge number of things, but most recently your dedication to our girls and the love you commit to our family. A special thank you also needs to be offered to the Coscia family for the continued support and love. We are going to have a fantastic life together.  
:o)

I would like to acknowledge *J. Am. Chem. Soc.* **1976**, *98*, 4010 for reprint of the crystal structure of 7,20-diisocyanoadociane and *J. Chem. Soc. Perkin Trans I* **1988**, 1003 for reprint of the crystal structure of a p-bromobenzoate derivative of 7,20-diisocyanoadociane.

## CURRICULUM VITAE

### *PHILIPP CHRISTOPHER ROOSEN*

1102 Natural Sciences II • University of California • Irvine, CA 92697-2025

Email: philipp.roosen@gmail.com

#### *Education*

---

**University of California, Irvine, CA** 2016

Doctor of Philosophy in Organic Chemistry

Thesis: “An Enantiospecific Formal Synthesis of (+)-7,20-Diisocyanoadociane”

**Michigan State University, East Lansing, MI** 2011

Master of Science in Organic Chemistry

Thesis: “A Hydrogen Bonding *ortho*-Directing Effect for the Iridium-Catalyzed Borylation of NH(*t*-Boc) Aromatics”

**Albion College, Albion, MI** 2008

Bachelor of Arts in Biology, minor in Chemistry, Cum Laude

#### *Research Experience*

---

**University of California, Department of Chemistry, Irvine, CA** 2011 – 2016

Advisor: Prof. Christopher Vanderwal

- › Developed an enantiospecific formal synthesis of the antiplasmodial, marine isocyanoterpene (+)-7,20-diisocyanoadociane
- › Initiated a synthesis of the antivirals wickerols A and B

**Michigan State University, Department of Chemistry, East Lansing, MI** 2008 – 2011

Advisor: Prof. Milton Smith, III in collaboration with Prof. Robert Maleczka, Jr.

- › Discovered and elucidated the mechanism of the hydrogen bonding *ortho*-directing effect of iridium catalyzed C–H borylation of N-(*tert*-butoxycarbonyl) anilines
- › Married the *ortho*-directed borylation of N-(*tert*-butoxycarbonyl) anilines with subsequent coupling reactions to access 3,4-dihydroquinolin-2-one and N-Boc-indole heterocycles
- › Initiated the systematic evaluation of iridium C–H borylation catalyst systems and reaction conditions to establish reaction parameters

*Research Experience (continued)*

---

**Albion College, Department of Chemistry, Albion, MI**

2007 – 2008

Advisor: Prof. Andrew French

- › Designed and synthesized novel chiral hypervalent iodine (III) reagents for oxidative transformations

**Henry Ford Hospital, Department of Neurology, Detroit, MI**

summer 2007

Advisor: Dr. Michael Chopp

- › Assisted in studying the enzyme ADAM17/TACE's influence on glioblastoma tumor angiogenesis

*Publications*

---

- Roosen, P. C.; Vanderwal, C. D. "A Formal Enantiospecific Synthesis of 7,20-Diisocyanoadociane" *Angew. Chem. Int. Ed.* **2016**, *55*, 7180–718355; *Angew. Chem.* **2016**, *128*, 7296–7299.
- Roosen, P. C.; Vanderwal, C. D. "Investigations into an Anionic Oxy-Cope/Transannular Conjugate Addition Approach to 7,20-Diisocyanoadociane" *Org. Lett.* **2014**, *16*, 4368–4371.
- Roosen, P. C.; Kallepalli, V. A.; Chattopadhyay, B.; Singleton, D. A.; Maleczka, R. E., Jr.; Smith, M. R., III "Outer-Sphere Direction in Iridium C–H Borylation" *J. Am. Chem. Soc.* **2012**, *134*, 11350–11353.
- Vanchura, B. V., II; Preshlock, S. M.; Roosen, P. C.; Kallepalli, V. A.; Staples, R. J.; Maleczka, R. E., Jr.; Singleton, D. A.; Smith, M. R., III "Electronic Effects in Iridium C–H Borylations: Insights from Unencumbered Substrates and Variation of Boryl Ligand Substituents" *Chem. Commun.* **2010**, *46*, 7724–7726.

*Research Presentations*

---

- Roosen, P. C.; Vanderwal, C. D. "Progress Towards the Synthesis of 7,20-Diisocyanoadociane" American Chemical Society Annual Meeting (Denver, CO) March 2015.
- Roosen, P. C.; Kallepalli, V. A.; Smith, M. R., III "*ortho*-Directed Iridium Mediate C–H Activation/Borylation of N-(*tert*-butoxycarbonyl)-Anilines" Boron in the Americas XII Conference (East Lansing, MI) June 2010.
- Roosen, P. C.; Logan, M. M.; French, A. N. "Modifications of Chiral Iodoarenes *Ortho*: Progress on the Synthesis and Examination of Select Target Molecules" Elkin R. Isaacs Student Research Symposium (Albion, MI) April 2008.



### *Honors and Awards*

---

- University of California Regents' Dissertation Fellowship (2016)
- UCI Graduate Award for Departmental Service (2015)
- UCI Department of Chemistry Safety Award (2015)
- UCI School of Physical Sciences travel award (2015)
- Associated Graduate Students travel grant (2015)
- Sigma Xi (2008)
- Beta Beta Beta (2007)
- Foundation for Undergraduate Research Scholarship and Creative Activity (FURSCA) grant (2007 – 2008)

### *Professional Organizations*

---

- American Chemical Society member (2007 – present)

## ABSTRACT OF THE DISSERTATION

An Enantiospecific Formal Synthesis of (+)-7,20-Diisocyanoadociane

By

Philipp Christopher Roosen

Doctor of Philosophy in Chemistry

University of California, Irvine, 2016

Professor Christopher D. Vanderwal, Chair

Described in this dissertation work are the paths, dead-ends and detours involved in eventually securing an enantiospecific formal synthesis of (+)-7,20-diisocyanoadociane (DICA) via Corey's dione, a project spanning early 2012 to early 2016. Highlighted is the isolation and biological background to DICA and isocyanoterpenes in general, in addition to the prior synthetic art for the preparation of cycloamphilectane and isocycloamphilectanes. Initial attempts at an entry into the DICA framework included a polyene cyclization, Diels–Alder/aldol condensation and reductive enone coupling strategy. Although these routes were unsuccessful they served as essential evolutionary precursors to the eventual completion of Corey's dione. Efforts to utilize Swaminathans oxy-Cope/transannular Michael cascade to construct the perhydropyrene scaffold of DICA led to a structural reassignment. Further mechanistic investigations into this reaction elucidated that the transannular Michael reaction was in fact under kinetic control. A phenanthrene reduction route led to an unexpected stereochemical outcome that could be useful for predictive control of either cis or trans outcomes. An enantiospecific formal synthesis of DICA was eventually secured through a dihydronaphthalene reduction. A general approach to multiple C7-isocyano(iso)cycloamphilectanes is also discussed.

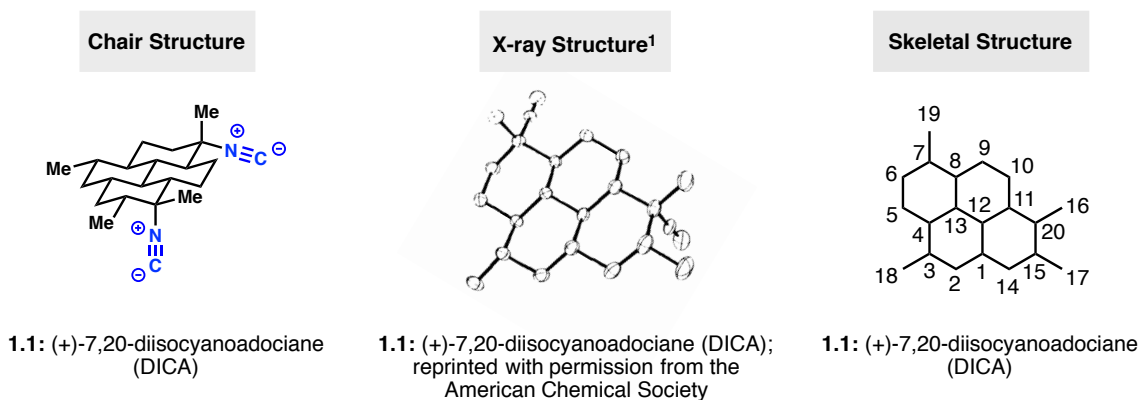
# CHAPTER 1:

## BACKGROUND TO (+)-7,20-DIISOCYANOADOCIANE

### 1.1 Introduction

(+)-7,20-Diisocyanoadociane<sup>1</sup> (**1.1**, DICA) is a polycyclic marine isocyanoterpene (ICT) with potent antiplasmodial activity that has attracted synthetic interest for over 30 years. Its *trans*-fused perhydropyrene ring system and salient isocyanides are marked challenges that have yet to be both addressed by the same synthetic design. In addition to a synthetic allure, the biology of DICA and ICTs more generally has further elevated interest in this family.

**Figure 1.1** The structure of (+)-DICA.



### 1.2 Isolation and Structure Determination

The marine sponges *Cymbastella hooperi* (ex. *Adocia* sp.) and *Amphimedon terpenensis* are native to the Great Barrier Reef, Australia and are identified by their cupped morphology and by their secondary metabolites.<sup>2</sup> Controversy surrounds whether these isolated sponges are identical species or two similar organisms that are both miscategorized.<sup>3</sup> DICA was isolated in the early 1970s from a collection of *Adocia* sp., now identified as *C. hooperi*,<sup>2</sup> in 2% yield by

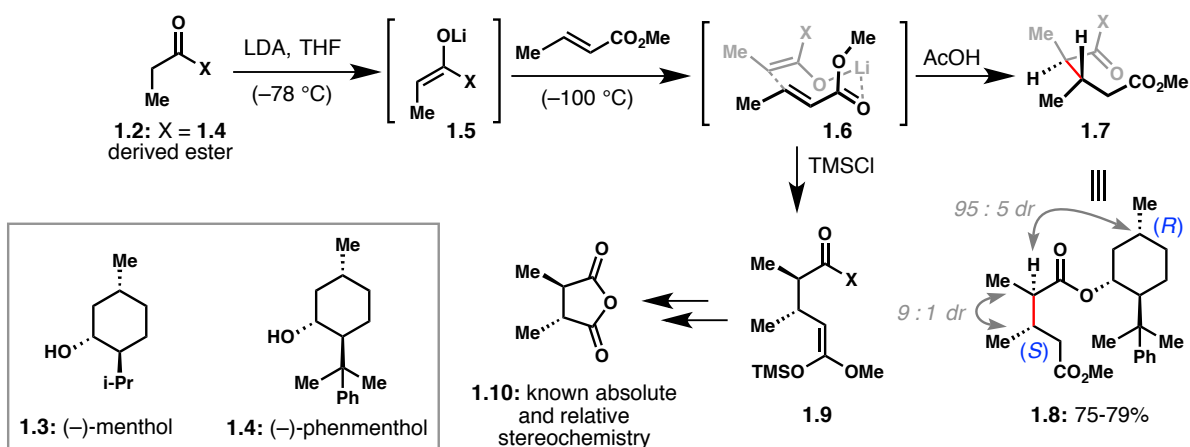
cold petroleum ether extraction of the milled freeze-dried sponge.<sup>1</sup> Its optical rotation was measured at  $[\alpha]_D^{22} +47.4^\circ$  ( $c$  0.7,  $\text{CH}_2\text{Cl}_2$ ). High-resolution mass spectrometry obtained  $e/z = 324.2565$ , suggesting the molecular formula of  $\text{C}_{22}\text{H}_{32}\text{N}_2$ . The IR spectrum showed peaks at 2130 and 2140  $\text{cm}^{-1}$  (s), indicating the presence of isonitrile functionalities. A 100-MHz  $^1\text{H}$  NMR spectrum revealed two methyl signals at  $\delta$  1.37 and 1.29 and two methyl doublets at  $\delta$  1.06 ( $J = 6$  Hz) and 0.88 ( $J = 6$  Hz), with no other resonances identifiable. Full high field  $^1\text{H}$ ,  $^{13}\text{C}$  and 2D NMR analysis was published two decades later.<sup>4</sup> The presence of two isonitriles was further confirmed by hydrolysis with mild acid to generate a structure with molecular formula  $\text{C}_{22}\text{H}_{36}\text{N}_2\text{O}_2$ , bearing appropriate  $^1\text{H}$  NMR data for a bis-secondary formamide molecule. Double recrystallization of the original isolate from hexane afforded crystals in the orthorhombic space group,  $P2_12_12_1$  with  $a = 7.086 \pm 0.004$  Å,  $b = 21.630 \pm 0.011$  Å,  $c = 13.104 \pm 0.007$  Å,  $Z = 4$ ,  $d_c = d_m = 1.079$   $\text{g}/\text{cm}^3$  and solved as structure **1.1** (Figure 1.1).

The absolute stereochemistry of DICA was assigned first by synthesis (see Scheme 1.1).<sup>5</sup> Corey's use of chiral-auxiliary-based Michael addition onto methyl crotonate was used to set the C3–C4 bond in a relative and absolute sense (Scheme 1.1B).<sup>6,7</sup> Although diastereocontrol in the simple propionate **1.2** case is excellent, it erodes for the substrate required for synthesis. Improved diastereocontrol on **1.11** is observed for the (–)-phenmenthol auxiliary at  $-100$  °C over the (–)-menthol based auxiliary at  $-78$  °C; however, for the purposes of synthesis, the menthol auxiliary was chosen for cost effectiveness. Even with a 60% ee, this preparation of DICA enabled the absolute stereochemical assignment as dextrorotatory.

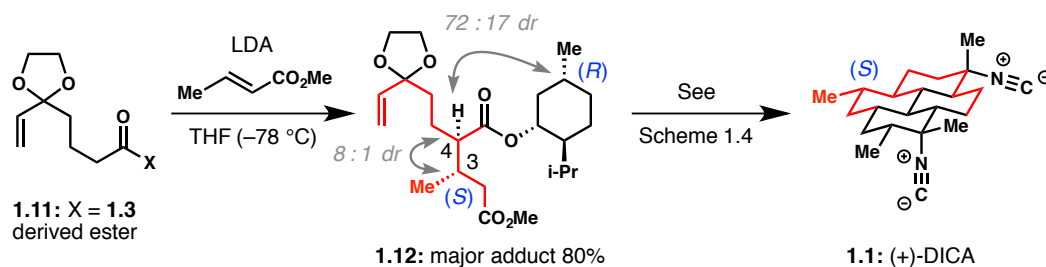
Shortly after Corey's assignment of (+)-DICA, a crystal structure of benzoate derivative **1.15** was reported and was in agreement with the absolute stereochemistry established by synthesis.<sup>8</sup>

**Scheme 1.1** Assignment of the absolute stereochemistry of (+)-DICA.

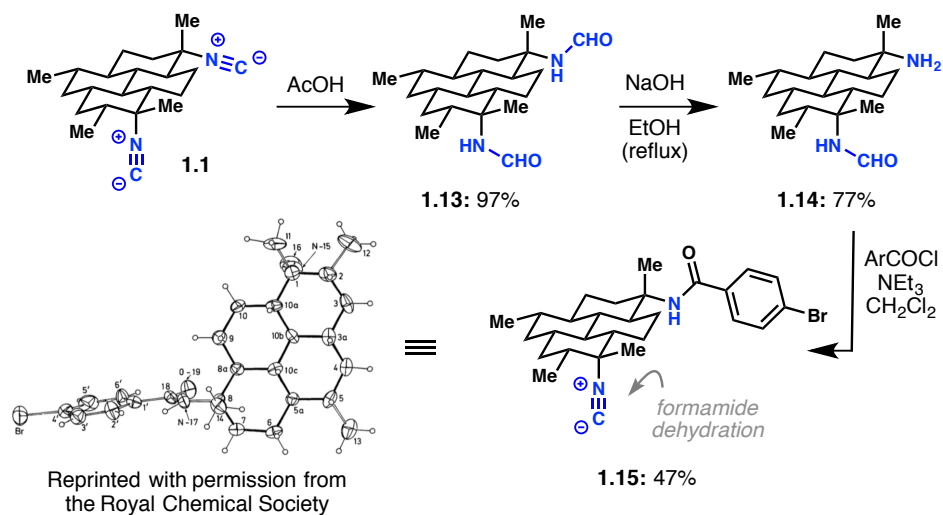
**A. Diastereoselective lithium enolate conjugate addition**



**B. Enantiocontrol in Corey's synthesis of (+)-DICA**



**C. X-ray crystal structure of a benzoate derivative of (+)-DICA**



### 1.3 Biosynthesis

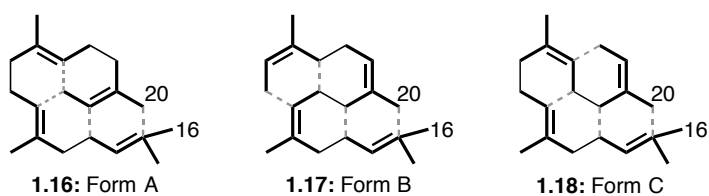
The complex relationship between sponges and the flora and fauna in their environment greatly challenges the delineation of the biosynthesis of their isolated secondary metabolites. Nonetheless, the origin and synthesis of DICA's carbon skeleton and isonitrile units has gathered significant attention.

DICA is comprised of a C<sub>20</sub> skeleton and hence classified as a diterpene. There are several options for housing four isoprene units in the methylated pyrene structure, recognizing a single C<sub>16</sub>-methyl shift (Figure 1.2A).<sup>1,9</sup> Garson has elegantly confirmed DICA is associated with the sponge and not a symbiotic or closely associated organism, for example cyanobacteria.<sup>10</sup> The difficulty comes in determining the source of the C<sub>20</sub> fragment, since it was shown that sodium [2-<sup>14</sup>C]acetate is not a carbon-source for DICA.<sup>11</sup> Typical terpene precursors like acetate, mevalonate, glucose and leucine are generally not incorporated into sponge terpenes as determined by feeding studies.<sup>8,11,12</sup> This result indicates that either the marine sponges *de novo* terpene synthesis requires other building blocks or more elaborated hydrocarbon fragments are obtained from external sources. Incorporation of [2-<sup>14</sup>C]acetate into carotenoids known to arise from blue-green algae symbiosis during feeding studies of other sponges supports the latter proposal.<sup>13</sup> Still, by examining the isolated natural products of *C. hooperi* a biosynthetic relationship between ICT diterpenes has been proposed (Figure 1.2B).<sup>14</sup> The cyclization of geranylgeranyl pyrophosphate **1.19** affords the cavernane scaffold **1.20**, a main branching point to ICT diterpenes. Further cyclization of **1.20** via amphilectane **1.21** and cycloamphilectane **1.22** sets up a methyl shift to the isocycloamphilectane **1.23** scaffold. Based on the proposed biosynthetic relationship between ICT sub-classes, the four isoprene units required to generate DICA are most likely arranged as depicted in **1.18**, Form C. The exact source of carbon, the

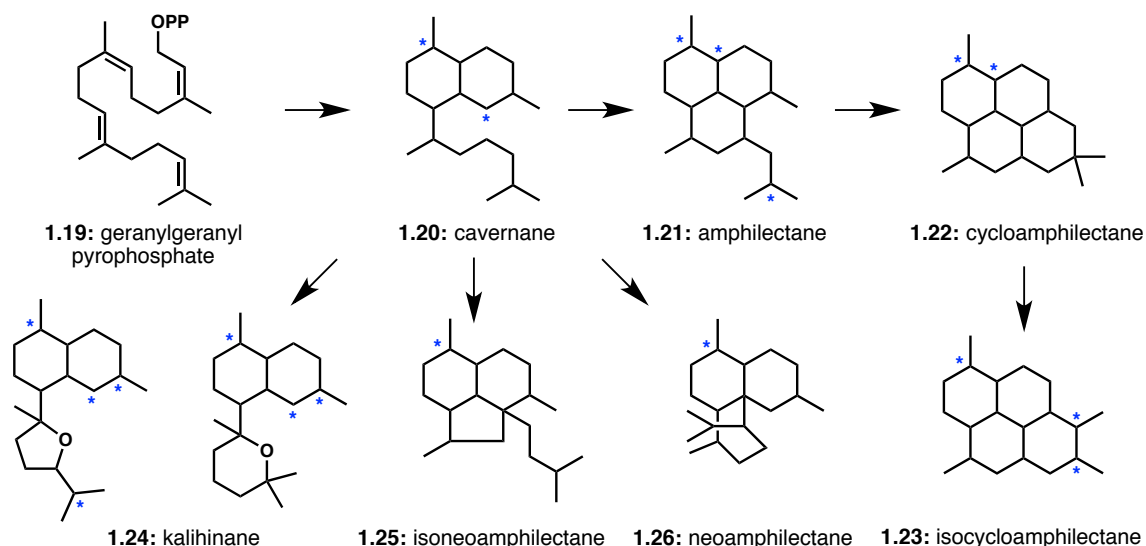
location in the sponge where cyclization is performed and the manner in which folding occurs remains unconfirmed for DICA, and ICTs more broadly.

**Figure 1.2** Isoprene units of the cycloamphilectane skeleton and C16 to C20 methyl shift to access the isocycloamphilectane skeleton.

**A. Incorporation of isoprene units into the cycloamphilectane scaffold**



**B. Proposed biosynthetic origin of ICT diterpenes via cyclization 1.18: Form C**

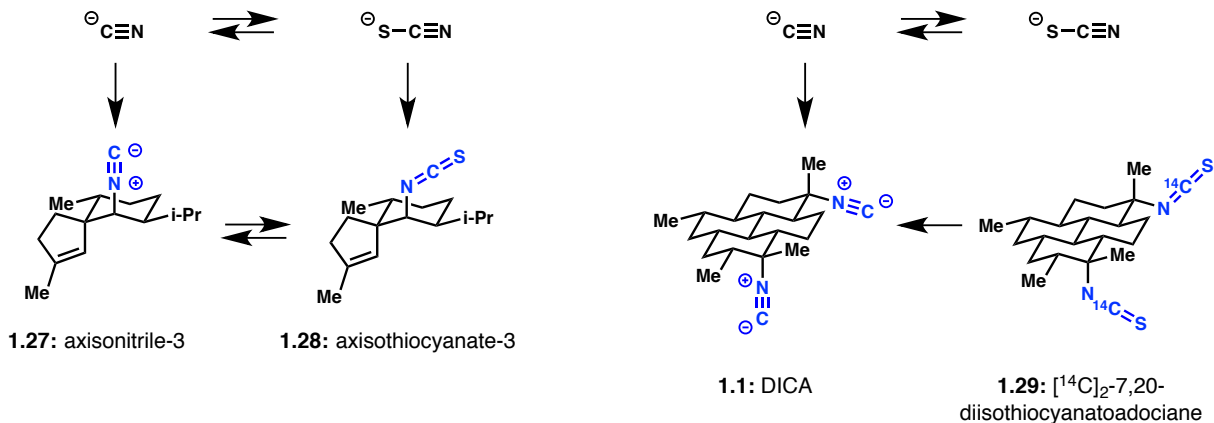


\* indicates the location of nitrogenous functional groups: isonitrile, isocyanate, isothiocyanate, formamide, amine

Arguably more interesting than terpene biosynthesis is the inclusion of cyanide into marine ICTs. Radiolabelling studies feeding  $\text{Na}^{[14]\text{C}}\text{N}$  to *A. terpenensis* followed by isolation of DICA showed incorporation of  $^{14}\text{C}$ .<sup>11</sup> Radiolabelling was retained upon hydrolysis of DICA to the bis-formamide. Upon cleavage of one formamide to the amine, 49% radioactivity remained and upon double deformylation only trace background activity was observed. This sequence

shows that the carbon of exogenous cyanide is incorporated into the isonitriles of DICA. A variety of other marine ICT natural products also show integration of radiolabeled cyanide during feeding studies.<sup>14</sup> Additionally, feeding of doubly labeled cyanide, Na[<sup>13</sup>C][<sup>15</sup>N], confirmed that both carbon and nitrogen of free cyanide are incorporated into the isonitrile group of the related ICT 9-isocyanoneopupukeanane.<sup>15</sup> Although the doubly labeled cyanide feeding study has not been performed for DICA to date, the example of 9-isocyanoneopupukeanane is likely representative for all ICTs.

**Scheme 1.2** Established biosynthetic relationship between cyanide, thiocyanate and their respective terpene derivatives.



Studies were undertaken to detect the origin of inorganic cyanide available to the organism and the manifold by which the isocyanate, isothiocyante, formamide and amine functionalities are born out. By observations and labeling studies of cyanide and axisonitrile-3 **1.27**, it was found that cyanide and thiocyanate are interconverted at an inorganic level (Scheme 1.2).<sup>16</sup> Additionally, isonitrile **1.27** and isothiocyante **1.28** are each interconverted, and importantly, without expulsion of either isonitrile or isothiocyante. These experiments were also performed in part on DICA and showed that radiolabelled isothiocyante is also biosynthetically



introduced into DICA via cyanide and that the organism is capable of desulfurizing the non-natural product [ $^{14}\text{C}$ ]<sub>2-7,20</sub>-diisothiocyanatoadociane **1.29**.<sup>17</sup> Inclusion of cyanide into ICTs and the interconversion of isocyano- and isothiocyanoterpenes have both been shown to be enzyme-mediated processes.<sup>14,15</sup> The occurrence of isocyanate, formamide and free amine derivatives has not been deeply studied but could be reasonably explained from hydrolysis of the isothiocyanate and isonitrile natural products.

Further investigation into the origin of the carbon and nitrogen in cyanide used by marine sponges is warranted. As determined by the significant production of ICTs, these sponges require significant quantities of the otherwise toxic cyanide. The seawater ecosystem home to ICT-producing sponges most likely contains minimal if any free cyanide, suggesting that cyanide is produced biosynthetically.<sup>14</sup> Labeling studies have indicated that the cyanide used to produce DICA is not of amino acid origin.<sup>8</sup> Although not experimentally determined, the biosynthetic origin of cyanide may involve microbial symbionts, as documented for other secondary metabolites.<sup>14</sup>

#### **1.4 Function and Ecology**

The abundance of ICTs in sponges of marine origin begs the question of their function and ecology. A wide variety of proposals exist for the various sesqui- and diterpenes, including structural, defense and symbiotic.<sup>14</sup> For the purposes of this dissertation, only information relevant to DICA will be presented.

Marine sponges generate DICA in significant quantities; it is obtained in 2% of the dry weight of collected sponge.<sup>1</sup> The production of DICA does not vary significantly with season or geography.<sup>13</sup> Upon dissection of *A. terpenensis*, DICA is found in both superficial ectosome and

deeper choanosome sponge tissue.<sup>10</sup> Higher concentrations of DICA relative to sterols were observed in larger nucleolated cells compared to small non-nucleated cells; the concentration of sterols in larger nucleolated cells was negligible. Further study revealed DICA to be associated with the sponge cell membrane; however, in vitro analysis using conventional phospholipids showed that DICA was not integral.<sup>18</sup> *Amphimedon* sp. phospholipids are unconventional by virtue of their significant quantities of unusual, brominated fatty acids.<sup>19,20</sup> Studies examining DICA in this lipid system have not been disclosed. The location and quantities of DICA available to the sponge suggest a structural role.<sup>18</sup> Further study on the precise roles of DICA and other ICTs are warranted.

## 1.5 Biological Activity

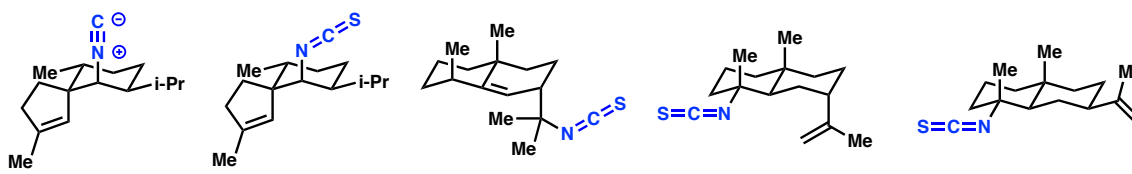
### 1.5.1 Introduction

A distinctive characteristic of ICTs is not only their structural diversity, but also their biological profile. This class of molecules, which encompasses both sesqui- and diterpenes, has a vast array of biological activity that for the purposes of this dissertation will focus on DICA.

DICA was first shown to exhibit in vitro antimicrobial activity against gram-positive bacteria although no precise detail was provided.<sup>21</sup> DICA was also shown to have vasodilative properties on isolated guinea pig hearts ( $EC_{50} = 22 \mu\text{g}/\text{heart}$ ), although no follow-up studies have been disclosed.<sup>22</sup> Significant interest in ICTs came out of an early evaluation of antiplasmodial activity. In collaborative work between the Angerhofer and König groups, ICTs **1.27**, **1.28** and **1.30–1.32** were screened against chloroquine-sensitive D6 and chloroquine-resistant W2 strains of *Plasmodium falciparum*, a malaria-causing parasite (Table 1.1).<sup>23</sup> The isonitrile **1.27** was markedly more potent than the isothiocyanates screened. These preliminary results led to a larger

screening program, in which DICA was also involved.<sup>24,25</sup>

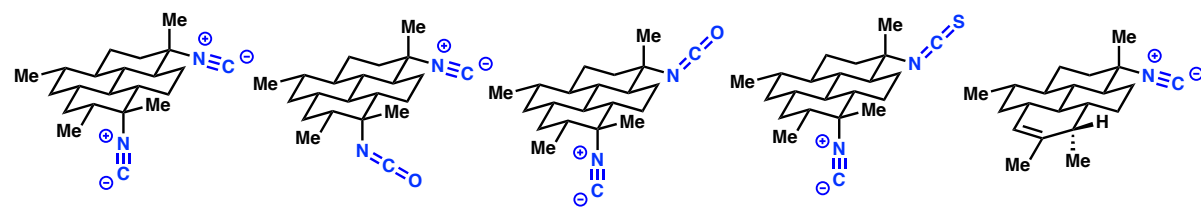
**Table 1.1** Initial evaluation of antiplasmodial activity and cytotoxicity of ICTs.



	1.27: axisonitrile-3	1.28: axisothiocyanate-3	1.30	1.31	1.32
D6 <sup>a</sup>	610	47,000	8,500	15,000	>38,000
W2 <sup>a</sup>	71	12,000	2,300	2,100	>38,000

<sup>a</sup> IC<sub>50</sub> values in nM; all ICT IC<sub>50</sub> values of human KB cell cytotoxicity were >75,000 nM

**Table 1.2** Evaluation of isocycloamphilectane ICTs against *P. falciparum*.



	1.1: (+)-DICA	1.33	1.34	1.35	1.36
D6 <sup>a</sup>	14	9	220	126	210
W2 <sup>a</sup>	13	7	265	80	66
KB Cells <sup>b</sup>	14,483	12,628	5,874	4,487	61,179

<sup>a</sup> IC<sub>50</sub> values in nM <sup>b</sup> IC<sub>50</sub> values of human KB cells cytotoxicity in nM

DICA exhibited significant in vitro activity against both D6 and W2 strains of *P. falciparum*, while also maintaining high selectivity for the parasite over human cells (Table 1.2). DICA derivatives **1.33–1.36**, bearing isothiocyanate or cyanate functionality and also isolated from *C. hooperi* show dramatic differences in activity. The position of the isocyanate is critically linked to biological activity as shown in **1.33** and **1.34**. C7 isothiocyanate **1.34** and isocyanate **1.35** show a 10–20 fold drop in activity, while a C20 isocyanate increases potency. Isocycloamphilectane **1.36**, containing only a single isonitrile, also showed a significant drop in

activity compared to DICA. The scope is too small to deduce structure-activity relationships, although the isonitriles are clearly critical for antiplasmodial activity.

The strong antiplasmodial activity inspired more biological evaluation of ICTs. Wright and coworkers approached with this opportunity, screened DICA and other ICTs against various targets.<sup>26</sup> The experiments were performed with 50  $\mu\text{g disk}^{-1}$  and the inhibition was measured in millimeters from the edge of the disk. DICA was examined for antibacterial activity against Gram-negative *Escherichia coli* (1 mm) and *Vibrio harveyi* ( $\text{IC}_{50}$  of 2,500 nM) and Gram-positive *Bacillus megaterium* (3 mm), with success. Antifungal activity was found against *Eurotium repens* (3 mm) and *Mycotypha microspore* (4 mm) but none against *Fusarium oxysporum*. Antialgal activity was assessed also by using 50  $\mu\text{g disk}^{-1}$  against *Chlorella fusca* (2 mm). Photosynthesis could be reduced by DICA by 36.9% at a test concentration of 0.2  $\text{mg mL}^{-1}$ . DICA's antitubercular activity against *Mycobacterium tuberculosis* had an  $\text{IC}_{50}$  value of 8,000 nM. Although DICA has a broad-ranging biological profile, the most potent and arguably most interesting phenotype is its antiplasmodial activity.

### 1.5.2 Studies on Antiplasmodial Activity

Malaria afflicts millions of people and causes approximately 1 million deaths annually, mostly in developing countries.<sup>27</sup> The current frontline treatment of artemisinin combination therapies has revolutionized the management of this illness; however, malaria-causing parasites are continuously evolving and have already begun showing signs of artemisinin resistance.<sup>28</sup> This is an early warning that new antimalarial medicines are needed, especially those containing novel pharmacophores or mechanisms of action. The current pipeline offers only more of the same medicinal derivatives, making a major conceptual leap in antimalarial therapeutics greatly

desired.<sup>29</sup> ICTs precisely fill this requirement as a potential lead since the isonitrile functional group constitutes a new warhead for exploration. DICA's potent activity against drug-resistant *P. falciparum* has elevated the molecule to further detailed study.

During erythrocyte infection, the parasite actively feeds on hemoglobin, thereby releasing toxic heme and hydrogen peroxide.<sup>30</sup> These molecules require neutralizing to avoid parasite self-destruction. This process is performed by the biocrystallization of heme into hemozoin and by various peroxidative pathways.<sup>31</sup> Interruption of these processes is proposed to be a major antiparasmodial mechanism of action for current antimalarial medicines and is a logical starting point for studying ICT-parasite interaction.<sup>32</sup> Wright and coworkers have performed initial studies, but the mechanism of action for DICA's potent biological activity is yet to be elucidated.<sup>24,33</sup>

A dual computational and experimental approach evaluated DICAs coordination to heme and subsequent inhibition of hemozoin formation as an explanation for antiparasmodial activity. The combination of 3D-QSAR and receptor modeling methodologies determined that the parasite target requires a hydrophobic pocket and is capable of forming electrostatic interactions.<sup>33</sup> There is a positive correlation between the structural homology of other ICTs and DICA to their antiparasmodial activity. This correlation was also observed experimentally as DICA showed strong coordination to heme by UV-vis, while less active ICTs showed subdued coordination. Experimental observations that DICA inhibited peroxidase-like activity of heme, oxidative heme decomposition and glutathione-dependent heme decomposition were also noted. Further studies revealed DICA inhibited the in vitro formation of  $\beta$ -hematin (the synthetic equivalent to hemozoin) from free heme.<sup>34</sup> These computational and experimental data were interpreted as key evidence that antiparasmodial activity is integrally related to interrupting

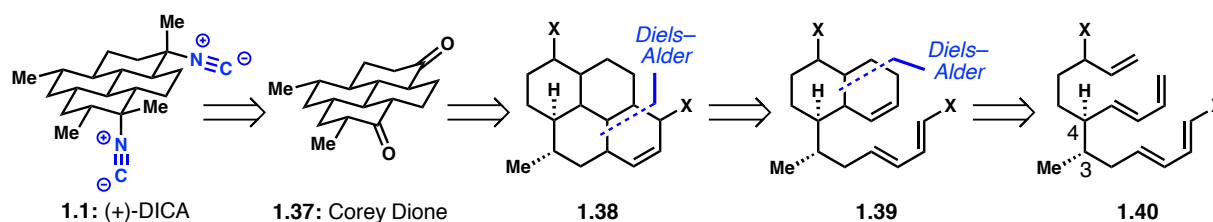
detoxification processes. These exciting results justify a continued pursuit in explaining DICAs strong potency.

## 1.6 Previous Syntheses of Cycloamphilectanes and Isocycloamphilectanes

### 1.6.1 Corey's Synthesis of (+)-7,20-Diisocyanoadociane (1987)

Corey and Magriotis reported the first synthesis of an isocycloamphilectane ICT by completing the synthesis of (+)-DICA.<sup>5</sup> Motivation for this synthetic effort was the unusual perhydropyrene structure, unprecedented biosynthesis and unknown absolute configuration. The Corey group disconnected DICA at the isonitrile carbons back to dione intermediate **1.37** (Corey's dione). Two elegant Diels–Alder disconnections take **1.37** back to the hypothetical molecule **1.40**. Of beauty is how one Diels–Alder sets up for the next, thereby generating five stereodefined centers of the perhydropyrene scaffold from the initially set C3–C4 connection.

**Scheme 1.3** Corey's retrosynthetic analysis of (+)-DICA via double Diels–Alder reaction.

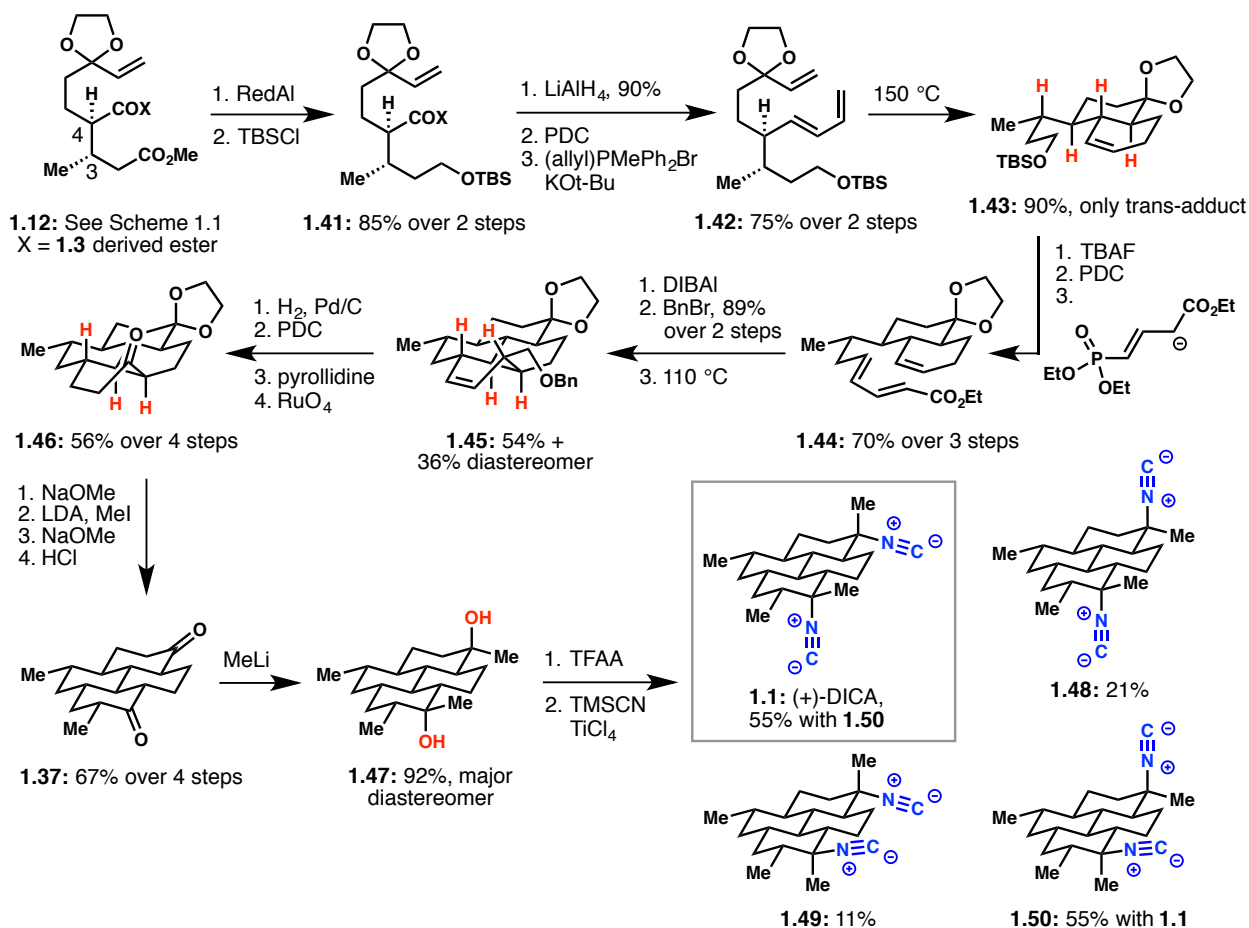


The synthesis began with establishing the C3–C4 stereochemical relationship via Corey's auxiliary-controlled asymmetric enolate conjugate addition (see Scheme 1.1).<sup>6</sup> Functional group interconversions led to Diels–Alder precursor **1.42**. Thermal cycloaddition afforded **1.43** in 90% yield as a single *trans*-decalin. Revealing the masked C1 aldehyde by TBAF deprotection and PDC oxidation enabled installation of the remaining skeletal carbons at the alkene oxidation states of **1.44**. Cycloaddition of ester **1.44** was unable to provide the correct stereochemistry;

reduction, benzyl protection and thermal cycloaddition generated **1.45**, which bore the correct C1–C12 *trans*-relationship. Oxidative removal of the superfluous carbon was accomplished by RuO<sub>4</sub> cleavage of the corresponding exocyclic enamine to afford **1.46**. After C17 methyl installation, the stereochemistry needed to be adjusted to the all-*trans* **1.37**. Double methyllithium addition afforded **1.47** as the diaxial diol. Trifluoroacetylation and ionization of the bistrifluoroacetates with TiCl<sub>4</sub> and capture with TMSCN generated (+)-DICA and three more diastereomers. Purification by silica gel chromatography afforded a mixture of (+)-DICA and **1.50** in 55% yield. (+)-DICA was isolated upon further purification by HPLC; however, the final yield remained unreported.

This first synthetic preparation of (+)-DICA confirmed the absolute stereochemistry of the natural product. The double Diels–Alder approach is an elegant solution to the construction of a perhydropyrene scaffold in which one Diels–Alder sets up for the next. Although the natural product synthesis was successful, certain elements could still be improved. First, while beautiful in design, the synthesis relies on several redox and protecting group manipulations to generate DICA in a total of 31 steps and a longest linear of 29 steps from commercial material. A concise, efficient and stereoselective synthesis of (+)-DICA or dione precursor **1.37** would be a worthwhile contribution. Second, stereocontrol of the second Diels–Alder to generate the perhydropyrene scaffold was sufficient for completion of the synthesis, but the 1.5:1 reaction selectivity could still be improved. When examining this synthesis, still the most pressing issue is the nonselective isonitrile installation. Ionization of the trifluoroacetates creates competitive axial and equatorial attack of TMSCN. One solution to this problem uses Sc(OTf)<sub>3</sub> as described by Shenvi, which promotes a mostly stereoinvertive-isonitrile displacement.<sup>35,36</sup> Additional efforts in the realm of direct isonitrile installations are however still warranted.

### Scheme 1.4 Corey's synthesis of (+)-DICA.



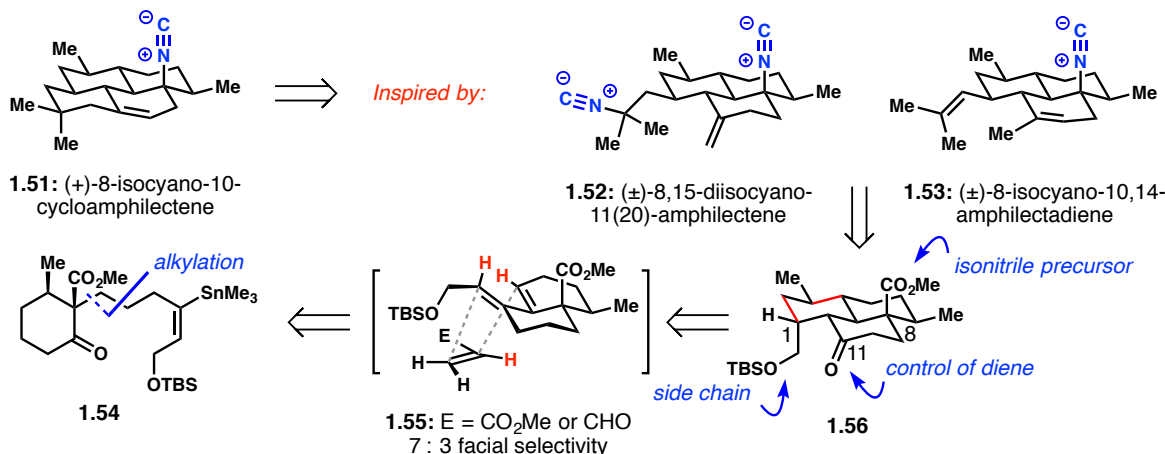
### 1.6.2 Piers's Synthesis of (+)-8-Isocyano-10-Cycloamphilectene (1998)

The only reported preparation of a cycloamphilectane ICT was completed by Piers and Schindeler and published only in a thesis.<sup>9</sup> The Piers lab's interest in methodology-inspired natural product synthesis drove the ICT syntheses of **1.52** and **1.53**.<sup>37–39</sup> The synthesis of **1.51** was initiated to further highlight the versatility of **1.56**, an intermediate scaffold obtained by an intramolecular Piers–Stille annulation and Diels–Alder reaction, and to determine the absolute configuration of **1.51** (Scheme 1.5). The synthetic branching point **1.56** contains three key functionalities: (1) a C8 ester as an isonitrile precursor, (2) a C11 ketone to control the alkene



functionality or ring closure to the cycloamphilectane target and (3) a C1 silyloxymethyl unit for diversification of the side chain.<sup>7</sup>

**Scheme 1.5** Piers's retrosynthetic analysis of (+)-8-isocyano-10-cycloamphilectene based on a Piers–Stille annulation and previous amphilectane syntheses.

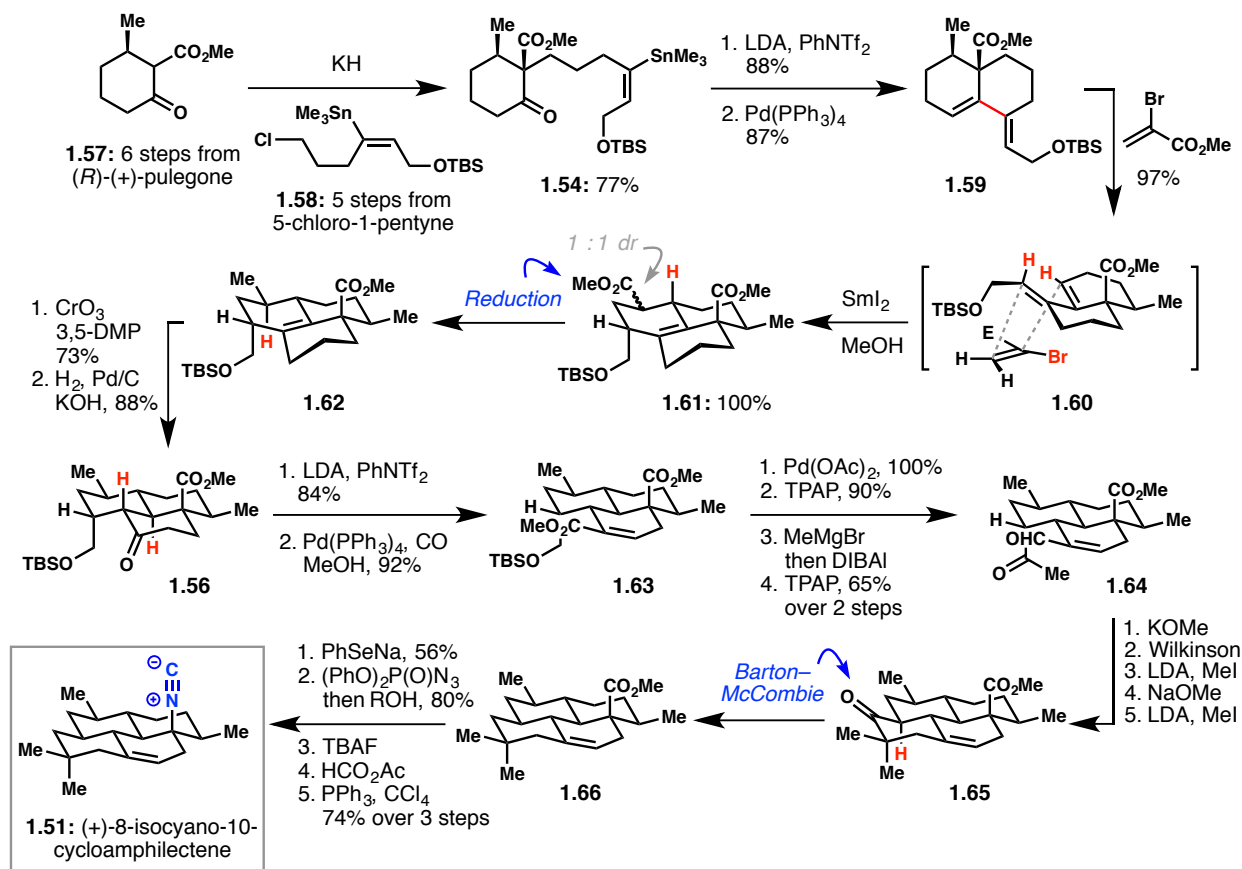


The first objective for a synthesis of **1.51** was improving the preparation of **1.56** for three reasons: (1) the earlier amphilectane syntheses were racemic, (2) the Diels–Alder reaction provided low facial selectivity and (3) an enone reduction to set the internal stereocenter of **1.56** was capricious. Piers and Schindeler addressed the first problem by starting from chiral pool material (*R*)-(+)-pulegone. The absolute stereochemistry of several amphilectanes and DICA was established with uniformity at the time of Piers and Schindeler's synthetic work, therefore it is unclear why the use of (*R*)-(+)-pulegone was chosen as it would generate the unnatural enantiomer. (*S*)-(–)-pulegone is significantly more expensive, but still readily prepared from (*S*)-(–)-citronellal. Upon alkylation of **1.57** with **1.58** and palladium-mediate cross coupling, the Diels–Alder reaction of **1.59** with methyl 1-bromoacrylate proceeded with complete facial selectivity. Sequential protodehalogenation and methyl ester reduction generated **1.62**, which was oxidized to the enone with CrO<sub>3</sub>•3,5-dimethylpyrazole and then reduced selectively via

hydrogenation in a strongly basic medium to afford the branching intermediate **1.56** in significantly optimized form as a single enantiomer.

Annulation of the fourth ring became the next challenge. The C11 ketone was derivatized to an unsaturated aldehyde and the C1 silyloxymethyl to a methyl ketone which upon exposure to base, cyclized to a dienone. Selective reduction of the disubstituted alkene followed by gem-dimethyl installation generated structure **1.65**. A Barton–McCombie radical deoxygenation was chosen to remove the C14 ketone, leaving only conversion of the ester into the C8 isonitrile to finish the synthesis of **1.51**.

**Scheme 1.6** Piers's synthesis of (+)-8-isocyano-10-cycloamphilectene.

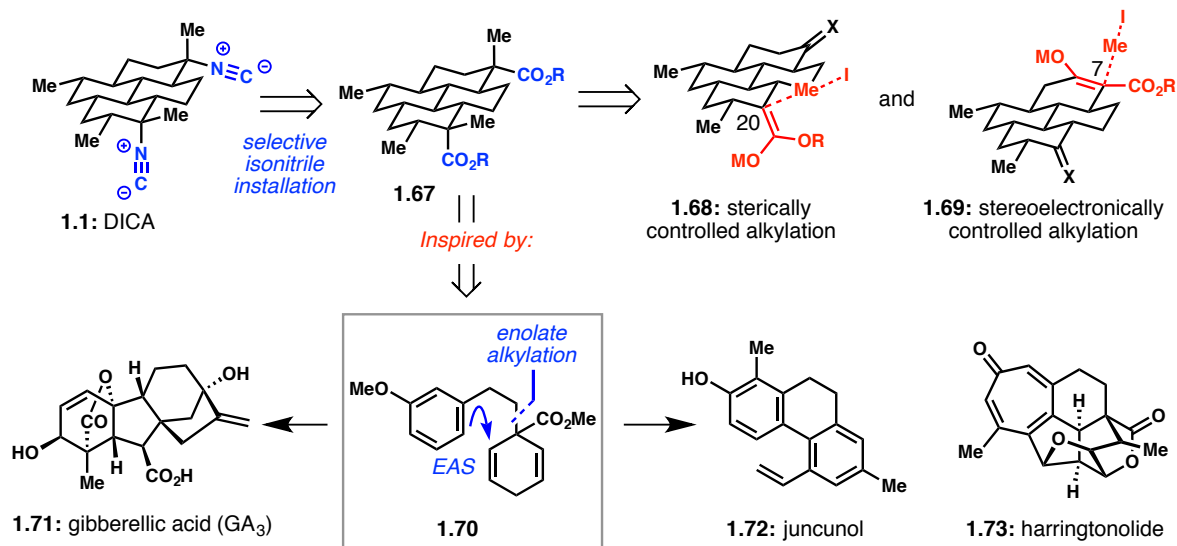


The Piers and Schindeler synthesis of **1.51** is an extension of earlier work from the Piers laboratory. A Piers–Stille annulation/Diels–Alder strategy enabled access to not only tricyclic amphilectane ICTs but also tetracyclic congeners. This synthesis addressed some of the shortcomings of earlier amphilectane syntheses, most notably the enantiospecific entry into ICTs by using chiral pool starting material, fashioning a highly facially selective Diels–Alder reaction, and optimizing for a robust enone reduction with excellent stereocontrol. To date, it is the only reported synthesis of a cycloamphilectane. The main drawback of this design is the step count, with a total of 46 steps and a longest linear of 36 steps from commercial material.

### *1.6.3 Mander's Formal Synthesis of (±)-7,20-Diisocyanoadociane (2006)*

The selective installation of tertiary isonitriles has been a formidable challenge in the context of ICT syntheses.<sup>40</sup> At the outset of Mander's interest in DICA, which started in the 1980's,<sup>41</sup> the most substantive solution was the Curtius rearrangement, formylation and dehydration first applied to ICTs by Piers.<sup>9,37–39</sup> With this method at hand, the most logical synthesis of the ester precursors is by stereoselective alkylation (Scheme 1.7). Alkylation under steric control was proposed to install the equatorial methyl at C20, while stereoelectronic control using a  $\beta$ -ketoester would provide axial methyl approach at C7.<sup>42</sup> The Curtius rearrangement's stereoretentive nature ensures that the configuration of the ester is relayed to the amine. Inspiration for construction of the perhydropyrene skeleton came from Mander's strong interest in using polyaromatics in natural product synthesis (Scheme 1.7).<sup>43,44</sup>

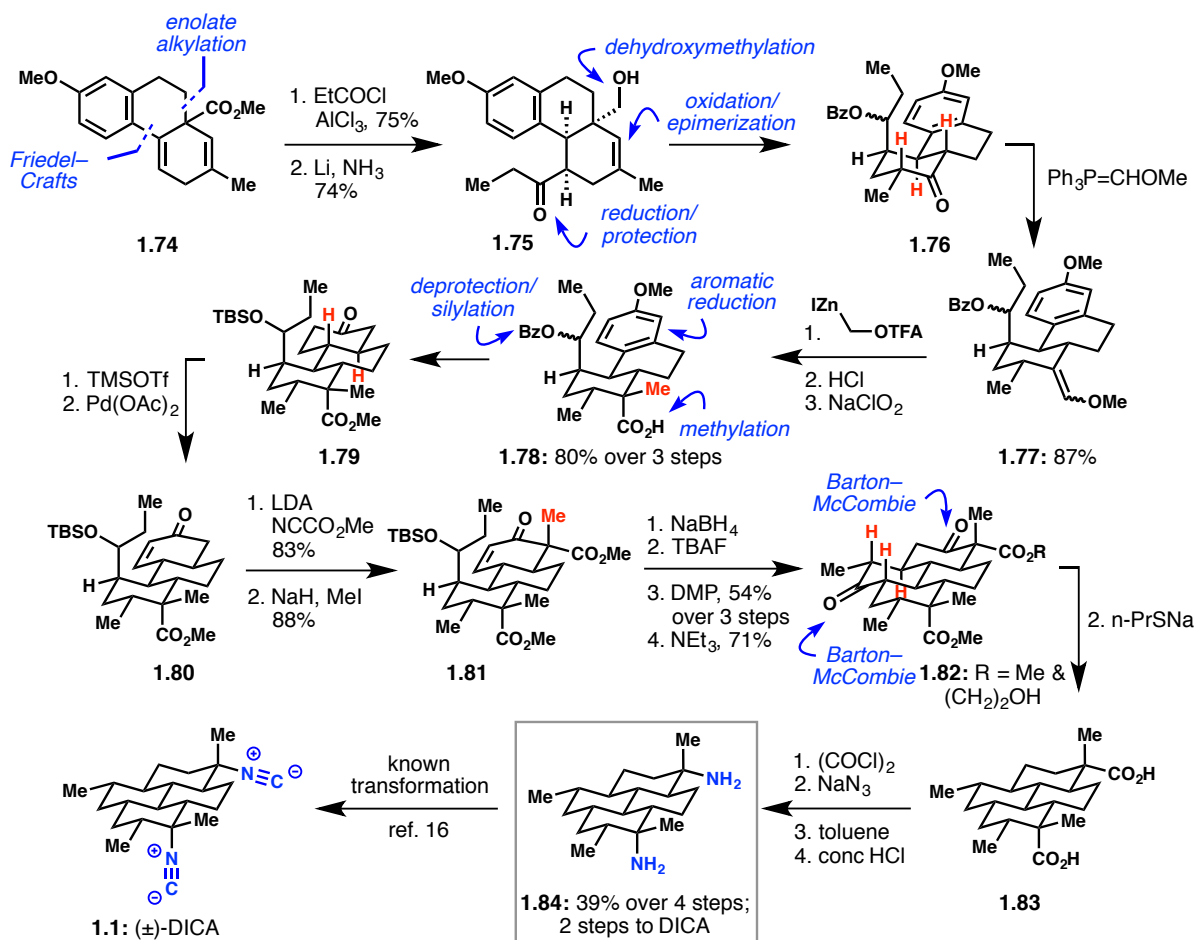
**Scheme 1.7** Mander's impetus and inspiration for the synthesis of DICA.



Phenanthrene **1.74**, generated by preceded Birch reduction/Friedel–Crafts condensation, became the unsaturated building block for Mander's stereoselective synthesis of DICA. An asymmetric manifold is only alluded to as potentially possible from diastereoselective Birch alkylation using a chiral benzamide.<sup>45</sup> From **1.74**, acylation and enone reduction with Li/NH<sub>3</sub> afforded **1.75**. Curiously the stereochemistry of reduction is *cis* (see Chapter 4 for discussion of similar reactivity). Further functional group manipulation and stereochemical correction sets the stage for the first methyl installation at C20. Although direct alkylation failed, cyclopropanation and acid-mediated ring opening afforded carboxylic acid **1.78**. Reduction of the aromatic and significant functional group manipulations generated **1.80**, ready for C7 alkylation. Deprotonation and alkylation of the corresponding  $\beta$ -ketoester produced axially methylated product **1.81**. The final ring was closed by intramolecular Michael addition of the revealed ethyl ketone and fused cyclohexenone. Barton–McCombie deoxygenation of **1.82** and functional group manipulation afforded diacid **1.83**. In a sequence analogous to the work by Piers, the diacid **1.83** was converted into free diamine **1.84**, a target readily transformed to DICA

via known means.<sup>16</sup>

**Scheme 1.8** Mander's formal synthesis of (±)-DICA.

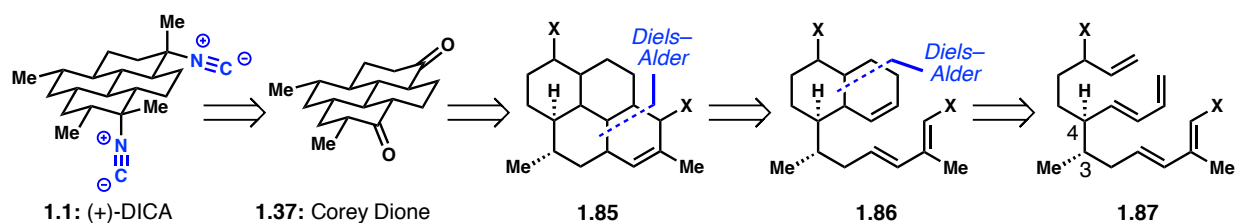


Mander's synthesis of DICA is the first and only synthesis with stereocontrol of the isonitrile moieties. It is designed around well-established enolate alkylation methods to control for equatorial and axial methylation. The obtained diester is transformed to the diamine via stereoretentive Curtius rearrangement, thereby completing a formal synthesis. The approach generates all stereocenters with excellent diastereocontrol and the diamine is completed in 48 total steps and 42 steps longest linear from commercial materials; an additional two steps are required to complete DICA.

### 1.6.4 Miyaoka's Formal Synthesis of (+)-7,20-Diisocyanoadociane (2011)

Miyaoka's formal synthesis of (+)-DICA follows almost the identical path as Corey's, 24 years prior (see Section 1.6.1).<sup>5,46</sup> The key C–C forming strategy also involved two Diels–Alder reactions to generate all stereocenters except C3 and C4 (Scheme 1.9). Distinct from Corey, Miyaoka sets the C3 and C4 stereocenters by alkylation of chiral lactone (*S*)-(-)-**1.88** (Scheme 1.10).

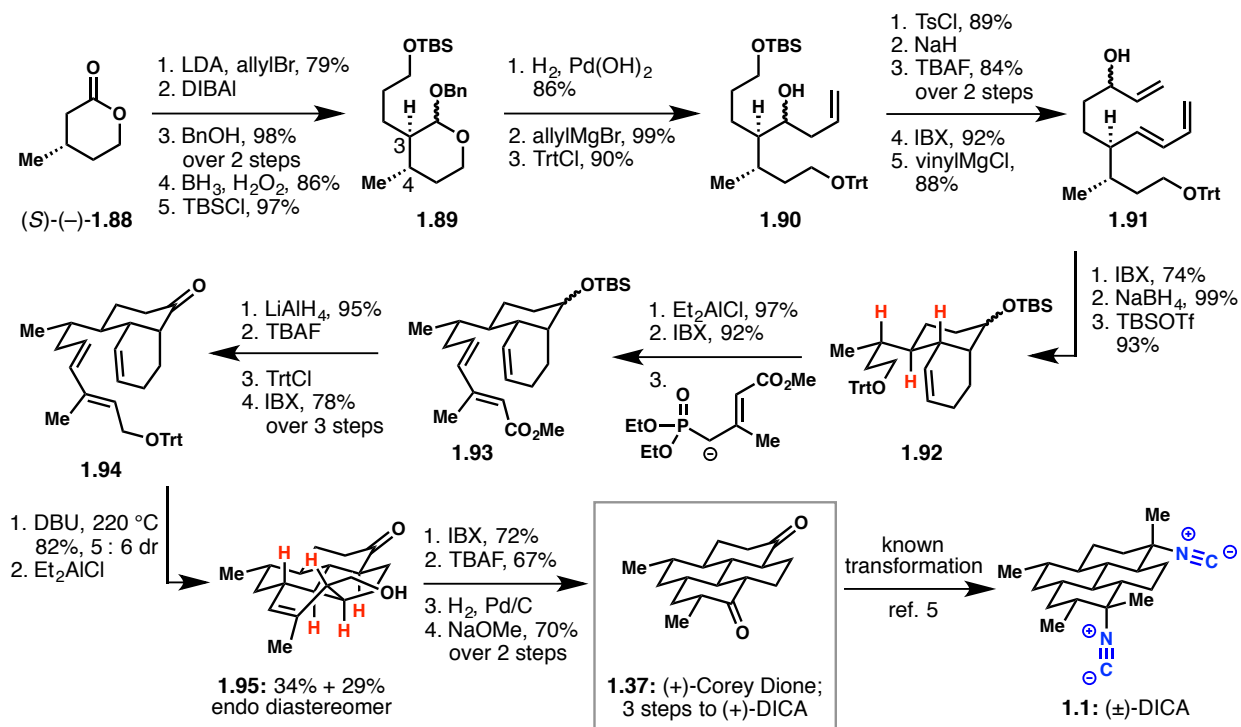
**Scheme 1.9** Miyaoka's double Diels–Alder strategy to Corey's dione.



Miyaoka's synthesis began with a known synthesis of (*S*)-(-)-**1.88**, via an esterase catalyzed desymmetrization.<sup>47</sup> Ester enolate allylation secured the *trans* relationship between C3 and C4, a vital control element for the subsequent Diels–Alder reactions (Scheme 1.10). Redox and protecting group manipulations afforded homoallylic alcohol **1.90**. Tosylation and elimination generated the diene, while silyl deprotection, oxidation and vinyl Grignard addition set up the alkenes for an oxidation triggered Diels–Alder (an observation also made by Corey in the context of DICA).<sup>5</sup> The remaining carbons were introduced via Horner–Wadsworth–Emmons (HWE) reaction of the corresponding **1.92** C1 aldehyde. Conversion of the unsaturated ester to the protected alcohol and oxidation of the C7 alcohol to the ketone afforded triene **1.94**. The next cycloaddition was performed with added DBU to ensure epimerization of the *cis*-decalin. Treatment of **1.94** at high temperatures afforded cycloaddition in 5 : 6 dr favoring an undesired endo diastereomer. Deprotection of the trityl group ensured clean isolation of the

desired diastereomer **1.95**. Oxidation to the skipped enal and unusual TBAF mediated oxidative deformylation<sup>48</sup> followed by hydrogenation and epimerization provided Corey's dione **1.37** in enantioenriched form to claim a formal synthesis of (+)-DICA.

**Scheme 1.10** Miyaoka's formal synthesis of (+)-DICA.



The purpose of Miyaoka's DICA synthesis is unclear as the main double Diels–Alder strategy is identical to Corey's (see Section 1.6.1).<sup>5</sup> Speculatively, Miyaoka's DICA synthesis was to fit with a broader approach to ICTs since an amphilectane<sup>49</sup> and several kalihinanes<sup>50–52</sup> were prepared by a similar Diels–Alder cycloaddition. There are only small differences compared to Corey's previously published work. First, the C3–C4 stereochemical relationship in Diels–Alder triene precursor **1.91** arises from substrate-controlled alkylation of lactone (*S*)-(-)-**1.88** instead of auxiliary controlled Michael addition. Second, the C17 methyl is incorporated into the HWE reagent instead of later installation via alkylation. And lastly, oxidative

demethylation of the extraneous carbon in **1.95** was accomplished via unique TBAF method<sup>53</sup> instead of oxidative cleavage of an enamine. Miyaoka's preparation of Corey's dione takes a total of 39 steps and 35 steps longest linear from commercial material, and requires another three steps to DICA.<sup>54</sup>

### **1.7 Goals for the Synthesis of (+)-Diisocyanoadociane**

ICTs, and specifically DICA, have attracted synthetic chemists with their unique structures and rich biological activity.<sup>55</sup> The widespread affliction of malaria and its negative impact on societies, especially for developing nations, requires continuous and persistent effort in building new methods of combat. In addition, the fundamental science revolving around antiplasmodial agents and the biology of this parasitic disease continues to be of interest. In this vein, the Vanderwal group initiated a research program to prepare and study ICTs. Synthetic studies towards (+)-DICA have been a cornerstone in this program since early 2012.

As DICA has already been prepared on multiple occasions, scaling the mountain of synthesis to this molecule again needs strong justification. First, DICA remains a structural terrain that has yet to succumb to concise synthesis and the construction of the all-*trans* perhydropyrene scaffold could still benefit from strategic chemical exercise. Second, the selective installation of the two isonitriles has been accomplished only by a single effort via a rather long sequence. Alternatives to the Curtius rearrangement have yet to be explored in the context of DICA. The need to install both an axial and equatorial isonitrile for DICA elevates this challenge. Overall, the structural, chemical and biological potential of DICA make it an attractive target.



## 1.8 References and Notes

- (1) Baker, J. T.; Wells, J. R.; Oberhänsli, W. E.; Hawes, G. B. *J. Am. Chem. Soc.* **1976**, *98*, 4010.
- (2) van Soest, W. M.; Desqueyroux-Faúdez, R.; Wright, A. D.; König, G. M. *Bull. Ins. R. Sci. Nat. Belg. Biol.* **1996**, *66*, 103.
- (3) König, G. M.; Wright, A. D. *Mem. Queensl. Museum* **1999**, *44*, 281.
- (4) König, G. M.; Wright, A. D. *Magn. Reson. Chem.* **1995**, *33*, 694.
- (5) Corey, E. J.; Magriotis, P. A. *J. Am. Chem. Soc.* **1987**, *109*, 287.
- (6) Corey, E. J.; Peterson, R. T. *Tetrahedron Lett.* **1985**, *26*, 5025.
- (7) Numbering matches the adociane scaffold, i.e. 7, 20-Diisocyanoadociane.
- (8) Fookes, C. J. R.; Garson, M. J.; MacLeod, J. K.; Skelton, B. W.; White, A. H. *J. Chem. Soc. Perkin Trans. 1* **1988**, 1003.
- (9) Schindeler, T. W. Ph.D. Dissertation, University of British Columbia, 1998.
- (10) Garson, M. J.; Thompson, J. E.; Larsen, R. M.; Battershill, C. N.; Murphy, P. T.; Bergquist, P. R. *Lipids* **1992**, *27*, 378.
- (11) Garson, M. J. *J. Chem. Soc. Chem. Commun.* **1986**, 35.
- (12) De Roa, M.; Minale, L.; Sodano, G. *Comp. Biochem. Physiol.* **1973**, *45B*, 883.
- (13) Garson, M. J. *Nat. Prod. Rep.* **1989**, *6*, 143.
- (14) Garson, M. J.; Simpson, J. S. *Nat. Prod. Rep.* **2004**, *21*, 164.
- (15) Karuso, P.; Scheuer, P. J. *J. Org. Chem.* **1989**, *54*, 2092.
- (16) Simpson, J. S.; Garson, M. J. *Org. Biomol. Chem.* **2004**, *2*, 939.
- (17) Simpson, J. S.; Garson, M. J. *Tetrahedron Lett.* **1999**, *40*, 3909.
- (18) Garson, M. J. *Chem. Rev.* **1993**, *93*, 1699.

- (19) Garson, M. J.; Zimmermann, M. P.; Battershill, C. N.; Holden, J. L.; Murphy, P. T. *Lipids* **1994**, *29*, 509.
- (20) Garson, M. J.; Zimmermann, M. P.; Hoberg, M.; Larsen, R. M.; Battershill, C. N.; Murphy, P. T. *Lipids* **1993**, *28*, 1011.
- (21) Kazlauskas, R.; Murphy, P. T.; Wells, R. J.; Blount, J. F. *Tetrahedron Lett.* **1980**, *21*, 315.
- (22) Morita, M.; Endo, M. *J. Pharmacobio-Dyn.* **1985**, *8*, s74.
- (23) Angerhofer, C. K.; Pezzuto, J. M.; König, G. M.; Wright, A. D.; Sticher, O. *J. Nat. Prod.* **1992**, *55*, 1787.
- (24) Wright, A. D.; König, G. M.; Angerhofer, C. K.; Greenidge, P.; Linden, A.; Desqueyroux-Faúndez, R. *J. Nat. Prod.* **1996**, *59*, 710.
- (25) König, G. M.; Wright, A. D.; Angerhofer, C. K. *J. Org. Chem.* **1996**, *61* (10), 3259.
- (26) Wright, A. D.; McCluskey, A.; Robertson, M. J.; MacGregor, K. A.; Gordon, C. P.; Guenther, J. *Org. Biomol. Chem.* **2011**, *9*, 400.
- (27) Gelb, M. H. *Curr. Opin. Chem. Biol.* **2007**, *11*, 440.
- (28) Dondorp, A. M.; Nosten, F.; Poravuth, Y.; Das, D.; Phyoo, A. P.; Tarning, J.; Lwin, K. M.; Arie, F.; Hanpithakpong, W.; Lee, S. J.; Ringwald, P.; Silamut, K.; Imwong, M.; Chotivanich, K.; Lim, P.; Herdman, T.; An, S. S.; Yeung, S.; Singhasivanon, P.; Day, N. *J. N. Eng. J. Med.* **2009**, *361*, 455.
- (29) Flannery, E. L.; Chatterjee, A. K.; Winzeler, E. A. *Nat. Rev. Microbiol.* **2013**, *11*, 849.
- (30) Miller, L. H.; Ackerman, H. C.; Su, X.-z.; Wellems, T. E. *Nat. Med.* **2013**, *19*, 156.
- (31) Weissbuch, I.; Leiserowitz, L. *Chem. Rev.* **2008**, *108*, 4899.
- (32) Wang, J.; Zhang, C.-J.; Chia, W. N.; Loh, C. C. Y.; Li, Z.; Lee, Y. M.; He, Y.; Yuan, L.-X.; Lim, T. K.; Liu, M.; Liew, C. X.; Lee, Y. Q.; Zhang, J.; Lu, N.; Lim, C. T.; Hua, Z.-

- C.; Liu, B.; Shen, H.-M.; Tan, K. S. W.; Lin, Q. *Nat. Commun.* **2015**, *6*, 10111.
- (33) Wright, A. D.; Wang, H.; Gurrath, M.; König, G. M.; Kocak, G.; Neumann, G.; Loria, P.; Foley, M.; Tilley, L. *J. Med. Chem.* **2001**, *44*, 873.
- (34) Young, R. M.; Adendorff, M. R.; Wright, A. D.; Davies-Coleman, M. T. *Eur. J. Med. Chem.* **2015**, *93*, 373.
- (35) Pronin, S. V.; Shenvi, R. A. *J. Am. Chem. Soc.* **2012**, *134*, 19604.
- (36) Pronin, S. V.; Reiher, C. A.; Shenvi, R. A. *Nature* **2013**, *501*, 195.
- (37) Piers, E.; Llinas-Brunet, M. *J. Org. Chem.* **1989**, *54*, 1483.
- (38) Piers, E.; Llinas-Brunet, M.; Oballa, R. M. *Can. J. Chem.* **1993**, *71*, 1484.
- (39) Piers, E.; Romero, M. A. *Tetrahedron* **1993**, *49*, 5791.
- (40) Schnermann, M. J.; Shenvi, R. A. *Nat. Prod. Rep.* **2015**, *32*, 543.
- (41) Crabtree, S. R. Ph.D. Dissertation, Australia National University, 1990.
- (42) Evans, D. A. "Stereoselective Alkylation Reactions of Chiral Metal Enolates" In *Asymmetric Synthesis*; Morrison, J. D., Ed.; Academic Press: New York, 1984; p 2.
- (43) Cossey, A. L.; Gunter, M. J.; Mander, L. N. *Tetrahedron Lett.* **1980**, *21*, 3309.
- (44) Mander, L. N. *Synlett* **1991**, 134.
- (45) Schultz, A. G.; Macielag, M.; Podhorez, D. E.; Suhadolnik, J. C.; Kullnig, R. K. *J. Org. Chem.* **1988**, *53*, 2456.
- (46) Miyaoka, H.; Okubo, Y.; Muroi, M.; Mitome, H.; Kawashima, E. *Chem. Lett.* **2011**, *40*, 246.
- (47) Francis, C. J.; Jones, J. B. *J. Chem. Soc. Chem. Commun.* **1984**, 579.
- (48) Kuroda, C.; Murase, A.; Suzuki, H.; Endo, S.; Anzai, S. *Bull. Chem. Soc. Jpn.* **1998**, *71*, 1639.

- (49) Miyaoka, H.; Okubo, Y. *Synlett* **2011**, 547.
- (50) Miyaoka, H.; Abe, Y.; Sekiya, N.; Mitome, H.; Kawashima, E. *Chem. Commun.* **2012**, 48, 901.
- (51) Miyaoka, H.; Shida, H.; Yamada, N.; Mitome, H.; Yamada, Y. *Tetrahedron Lett.* **2002**, 43, 2227.
- (52) Miyaoka, H.; Abe, Y.; Kawashima, E. *Chem. Pharm. Bull.* **2012**, 60, 1224.
- (53) Kuroda, C.; Murase, A.; Suzuki, H.; Endo, T.; Anzai, S. *Bull. Chem. Soc. Jpn.* **1998**, 71, 1639.
- (54) The preparation of **1.88** is counted as taking three steps from commercial 3-methylglutaric acid.
- (55) During the preparation of this dissertation a synthesis of DICA and preliminary antiplasmodial activity during the liver stage was published: Lu, H.-H.; Pronin, S. V.; Antonova-Koch, Y.; Meister, S.; Winzeler, E. A.; Shenvi, R. A. *J. Am. Chem. Soc.* **2016**, DOI: 10.1021/jacs.6b03899.

## CHAPTER 2: INITIAL EFFORTS TOWARDS 7,20-DIISOCYANOADOCIANE

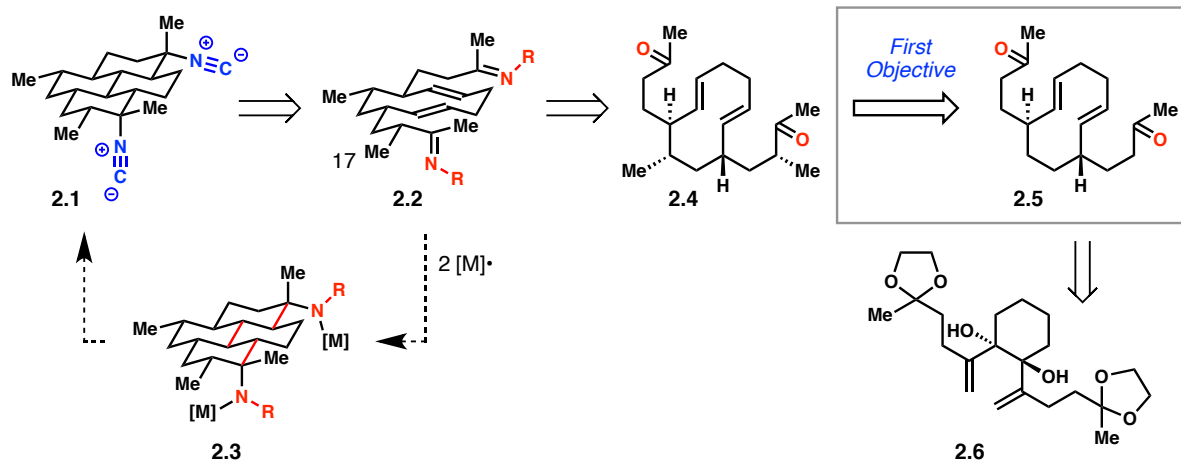
### 2.1 Introduction

7,20-Diisocyanoadociane (**2.1**, DICA) was chosen as a synthetic target for its complex tetracyclic structure, unusual incorporation of isonitriles and because so far it has not succumbed to a concise and efficient synthesis.<sup>1-3</sup> This synthetic program was not initiated because of a specific methodology or targeted post-synthetic study and thereby provided a freedom in ideas and execution. This autonomy offered both a great opportunity to explore reactivity and designs, but also at times became a challenge when prioritizing multiple avenues. Described here are the initial pursuits to DICA.

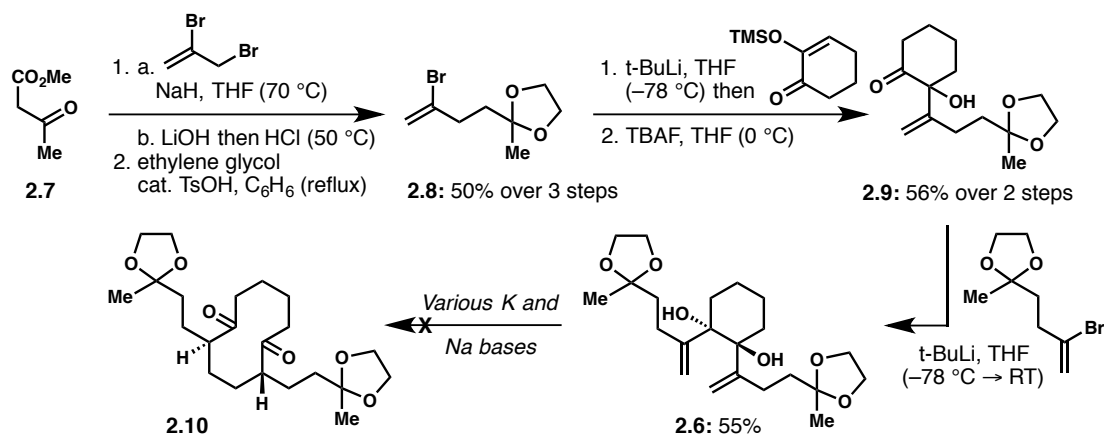
### 2.2 A Polyene Imino–Pinacol Coupling Approach

The first synthetic idea for tackling DICA was a proposed radical polyene cyclization.<sup>4</sup> The framework of DICA would be constructed by single electron reduction of an imine in **2.2** to the radical anion, two six-endo cyclizations and a six-exo termination onto the second imine followed by final reduction with a second equivalent of reductant (Scheme 2.1). The resulting bis-amine could be elaborated into DICA. Overall, the desired transformation can be classified as a polyene imino–pinacol coupling,<sup>5</sup> a currently unknown transformation. The stereochemical outcome of this reaction was not considered, but the C17 methyl group could potentially induce some stereocontrol. Synthetically, the first objective was to prepare **2.5**, a desmethyl analog to diketone **2.4**. Among the various ideas proposed, a bis-anionic oxy-Cope reaction was attractive for its use of symmetry.

**Scheme 2.1** An initial polyene imino–pinacol coupling idea.



**Scheme 2.2** An attempted bis-oxy-Cope rearrangement.



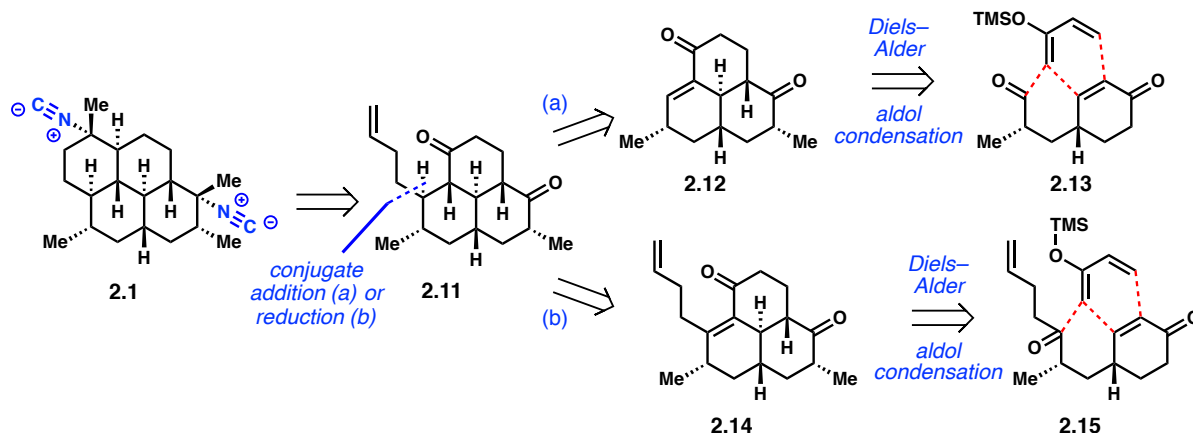
The approach was initiated by the preparation of vinyl bromide **2.8** via  $\beta$ -ketoester alkylation, decarboxylation and ketal protection. Lithium/halogen exchange with *t*-BuLi and addition to 2-(trimethylsiloxy)cyclohex-2-enone afforded **2.9** following TBAF desilylation. A second vinyl lithium addition afforded *trans*-diol **2.6**. Unfortunately using KHMDS, KO*t*-Bu, K<sub>2</sub>CO<sub>3</sub>, or NaH in THF up to reflux and in the microwave in DMF and DCB, no product suggestive of **2.10** was observed. Literature precedent indicates that after a bis-oxy-Cope reaction, the substrates are well established to undergo transannular aldol reactions.<sup>6–8</sup> This

reactivity could also be imagined complicating the outcome of the attempted bis-oxy-Cope rearrangements of **2.6**.

### 2.3 A Diels–Alder/Aldol Condensation Pursuit

The polyene cyclization idea was set aside in favor of an anionic oxy-Cope/Michael approach inspired by a literature reaction (see Chapter 3). Upon recognizing and explaining the unexpected stereochemical complexities,<sup>9</sup> the targeted intermediate **2.11** was still attractive for another attempt (Scheme 2.3). Two ideas were conceived: (1) a conjugate addition to **2.12**, and (2) a conjugate reduction of **2.14**. The synthesis of both enones **2.12** and **2.14** was disconnected via a Diels–Alder/aldol condensation cascade. The construction of aldehyde **2.13** was considered easier and therefore pursued first.

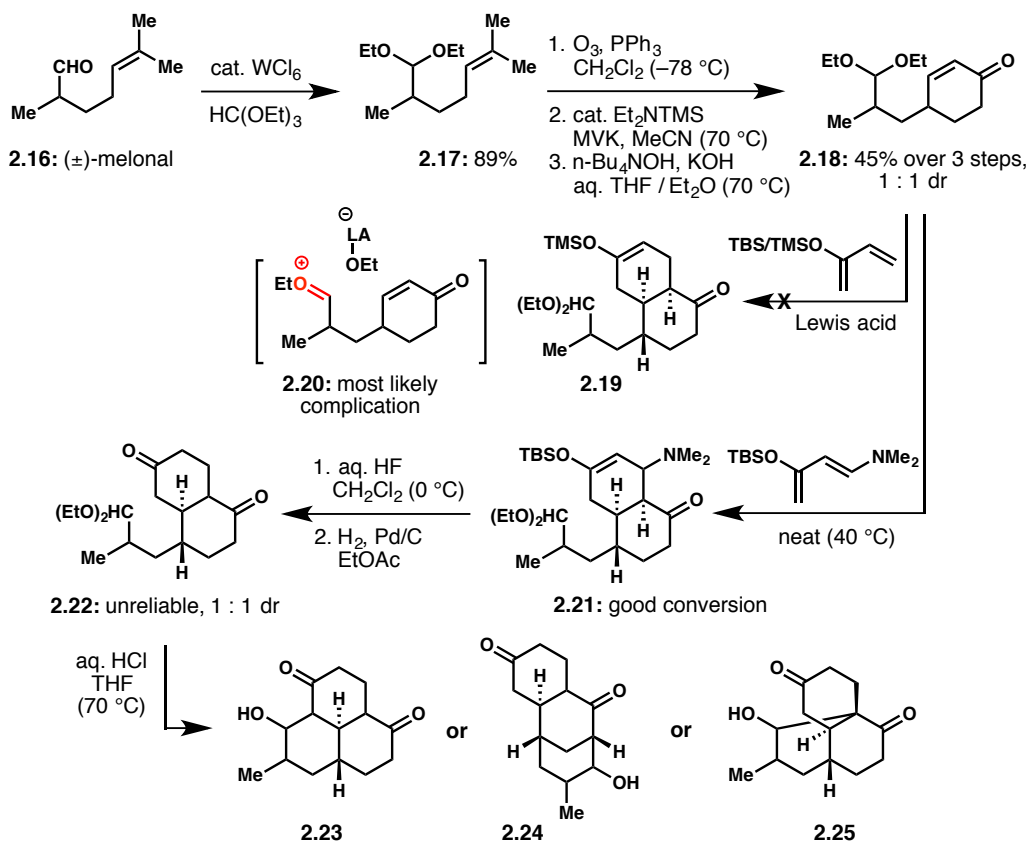
**Scheme 2.3** A Diels–Alder/aldol condensation proposal to DICA.



The route began with a ketal protection of melonal **2.16**,<sup>10</sup> followed by an ozonolysis and Robinson annulation to generate **2.18** (Scheme 2.4). A ketal-protecting group was chosen because a one step deprotection and aldol condensation was envisioned to forge the third ring. Unfortunately, the ketal caused significant problems during the Diels–Alder reaction. Lewis acid

mediated cyclization did not provide cycloadduct **2.19**, most likely because of competitive ionization of the ketal (**2.20**). Heating a neat mixture of 2-(trimethylsiloxy)butadiene<sup>11</sup> and **2.18** was also unsuccessful. Good reactivity was found when neat Rawal's diene<sup>12</sup> and **2.18** were heated at slightly elevated temperature. Disappointingly, the deprotection and installation of the enone was not successful. In retrospect, running the reaction in an HF soluble solvent such as MeCN would have been a wiser decision than CH<sub>2</sub>Cl<sub>2</sub> and could have led to success. Still, small quantities of **2.22** were obtained, which when heated in the presence of aqueous HCl generated a single diastereomer of an alcohol-bearing aldol product. Unfortunately, the structure was not determined, but is likely to be **2.23**, **2.24** or **2.25**. To better understand this reactivity, a more reliable route to **2.22** was sought.

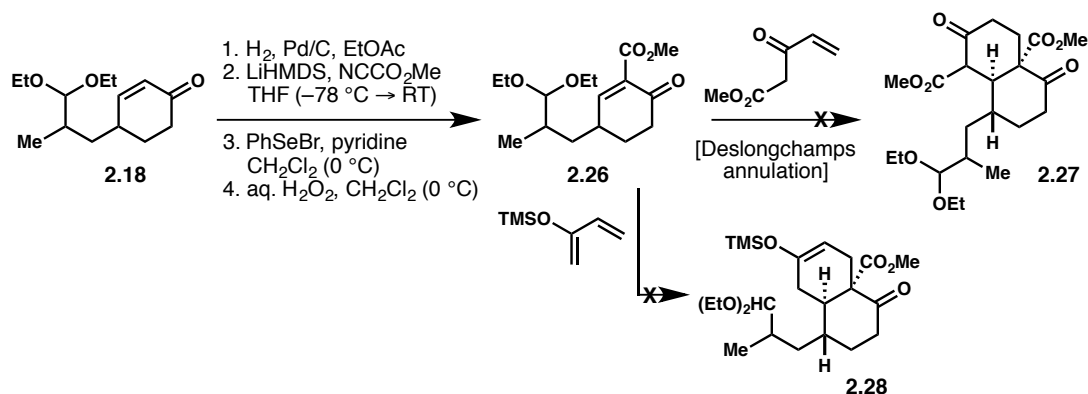
**Scheme 2.4** Initial success of the Diels–Alder/aldol condensation idea.





To improve the material throughput to **2.22**, a dienophile-activating group was introduced (Scheme 2.5). The ester unit on **2.26** was installed using Mander's reagent followed by selenylation and selenoxide elimination. Deslongchamps annulation<sup>13,14</sup> using Nazarov reagent<sup>15</sup> **2.26** under acidic, basic or neutral conditions in numerous solvents did not afford **2.27**. The Deslongchamps annulation has been used extensively in total synthesis, but a unifying feature is that all examples feature a  $\beta$ -substituted Nazarov reagent. A control reaction using such a  $\beta$ -substituted Nazarov reagent was not attempted, but would give good indication on whether **2.26** is a competent coupling partner. Thermal conditions using 2-(trimethylsilyloxy)butadiene only returned starting material.<sup>16</sup> In the wake of semi-success at the Diels–Alder/aldol condensation idea, elimination of the problematic ketal in **2.18** was pursued.

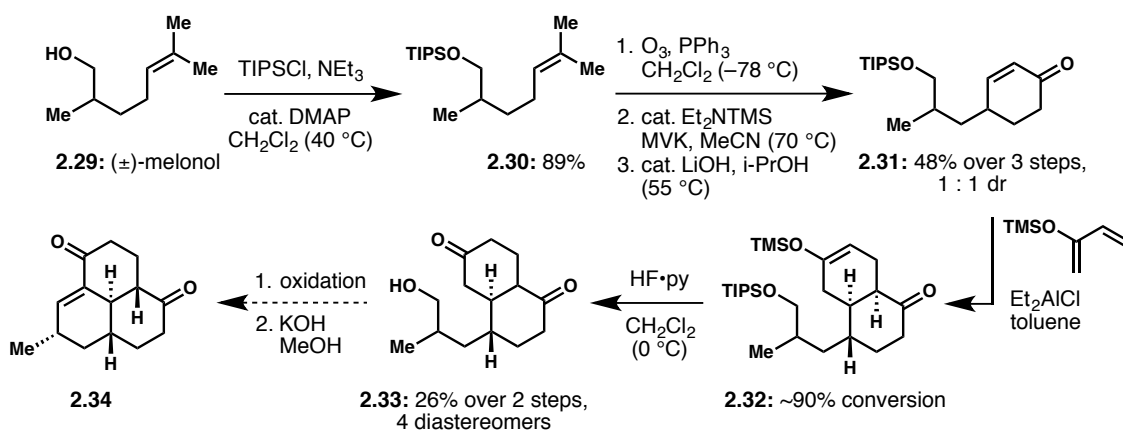
**Scheme 2.5** Attempted Diels–Alder reaction via dienophile activation.



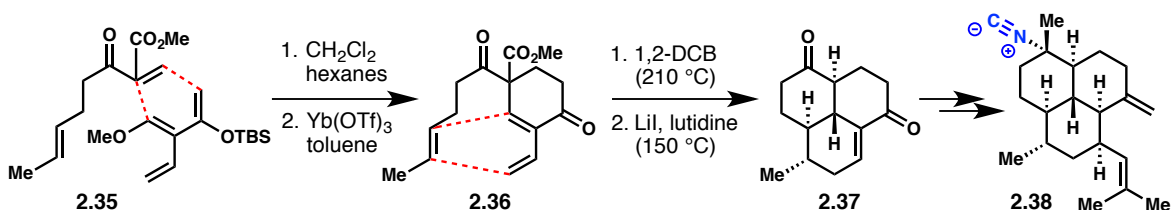
Instead of using a ketal as the masked aldehyde, which complicated Lewis acid activation of the dienophile for Diels–Alder reactivity, **2.29** could be converted into TIPS protected alcohol **2.30** (Scheme 2.6). Oxidative cleavage of the alkene then set up for another Robinson annulation. Standard diethylaminotrimethylsilane-catalyzed conjugate addition conditions were retained, but homogenous LiOH-mediated aldol condensation conditions<sup>17</sup> were implemented to furnish **2.31** instead, as biphasic conditions were unsuccessful in generating **2.18**. Diels–Alder with 2-

(trimethylsiloxy)butadiene in the presence of two equivalents of  $\text{Et}_2\text{AlCl}$  led to good conversion to **2.32**. The last aldol condensation was not attempted on this system because at this point the general approach was considered too close to an ICT synthesis reported concurrently by Shenvi (Scheme 2.7).<sup>18</sup>

**Scheme 2.6** Incomplete efforts towards a Diels–Alder/aldol condensation sequence.



**Scheme 2.7** An ICT synthesis by Shenvi.



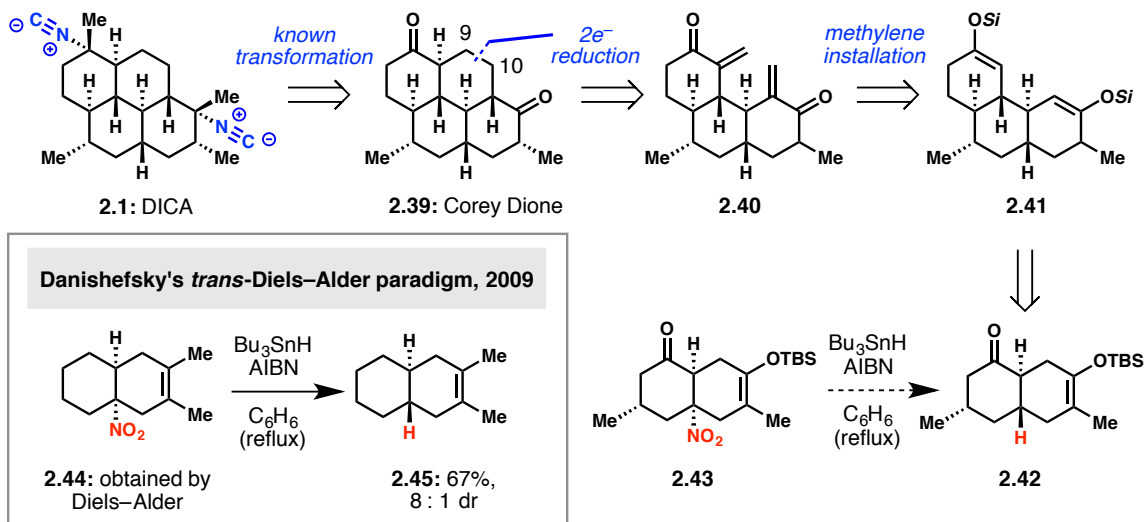
## 2.4 The Methylenation/Reductive Ring Closure Approach

### 2.4.1 Application of Danishefsky's *trans*-Diels–Alder Paradigm

The next set of ideas revolved around a complete redesign of a synthesis of DICA (Scheme 2.8). Corey's dione **2.39** was disconnected between C9–C10 back to bis-enone **2.40**. The last ring closure would occur by a two-electron reductive coupling of bis-enone **2.40**.<sup>19</sup> The

pseudo-symmetric **2.40** was further disconnected to bis-enoxysilane **2.41**. Danishefsky's *trans*-Diels–Alder reaction was considered advantageous for gaining entry into the required *trans*-stereochemistry of decalin **2.42**.<sup>20</sup>

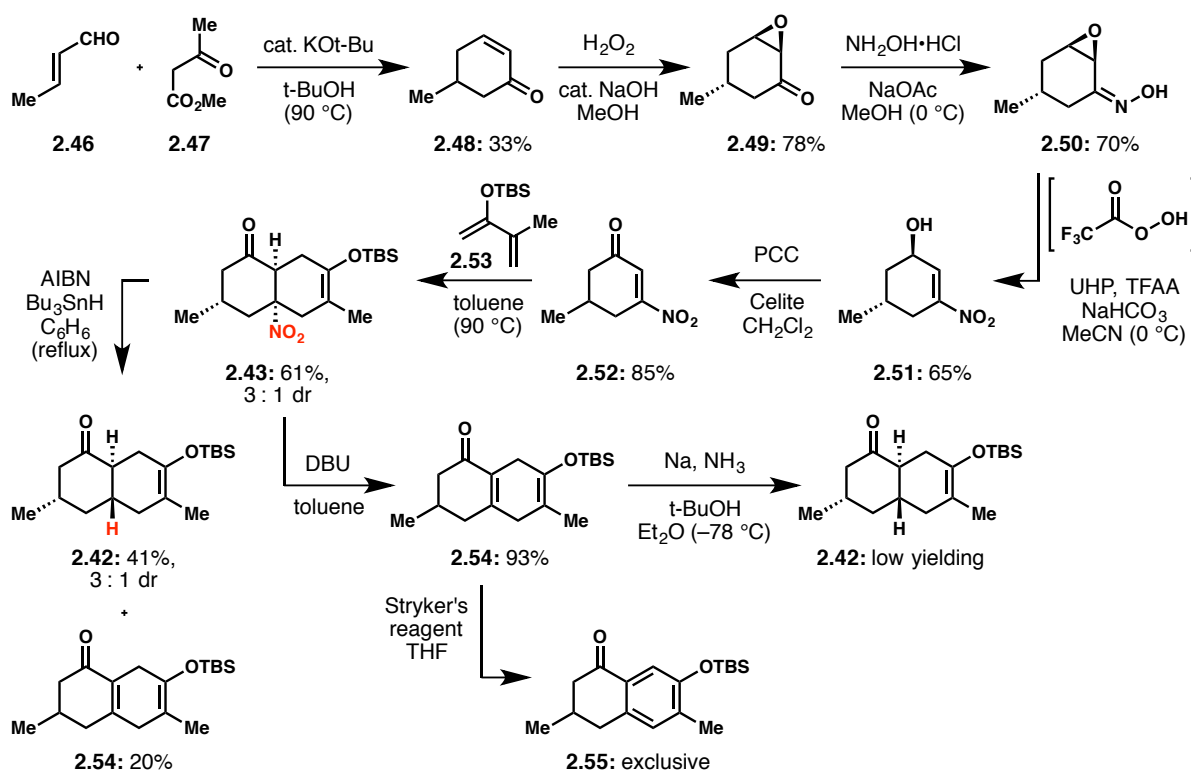
**Scheme 2.8** General proposal for a final reductive ring closure.



The synthesis of **2.43** was performed in an analogous manner to a Corey procedure (Scheme 2.9).<sup>21</sup> Cyclohexenone **2.48** was smoothly epoxidized under basic conditions and then treated with hydroxylamine to generate oxime **2.50**. Trifluoroperoxyacetic acid oxidized the oxime to a nitro functional group, which triggered opening of the epoxide. After oxidation of the alcohol, the doubly activated alkene **2.52** was subjected to Diels–Alder reaction with dienoxysilane **2.53** to generate **2.43** in 3 : 1 dr. The stereochemistry depicted was assigned based on an analogous literature reaction.<sup>22</sup> Tributyltin hydride radical denitration successfully afforded **2.42**, although significant elimination product **2.54** was also observed. Alternatively, the nitro could be eliminated with DBU to afford **2.54**. Conjugate reduction with Stryker's reagent afforded only tetralone **2.55**. Birch reduction of **2.54** generated **2.42**, however in low yields.

After a more critical examination of this route, the undesired alkene-orientation of **2.42** and the fairly involved sequence of steps made a different approach more appealing.

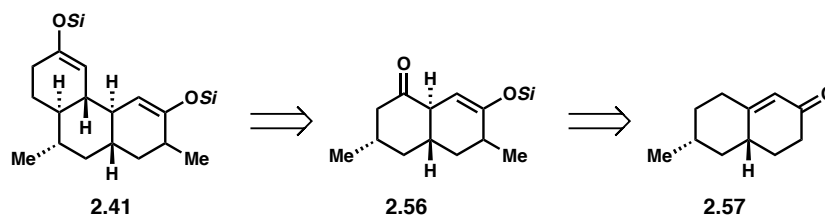
**Scheme 2.9.** Diels–Alder reactivity of an activated nitroalkene.



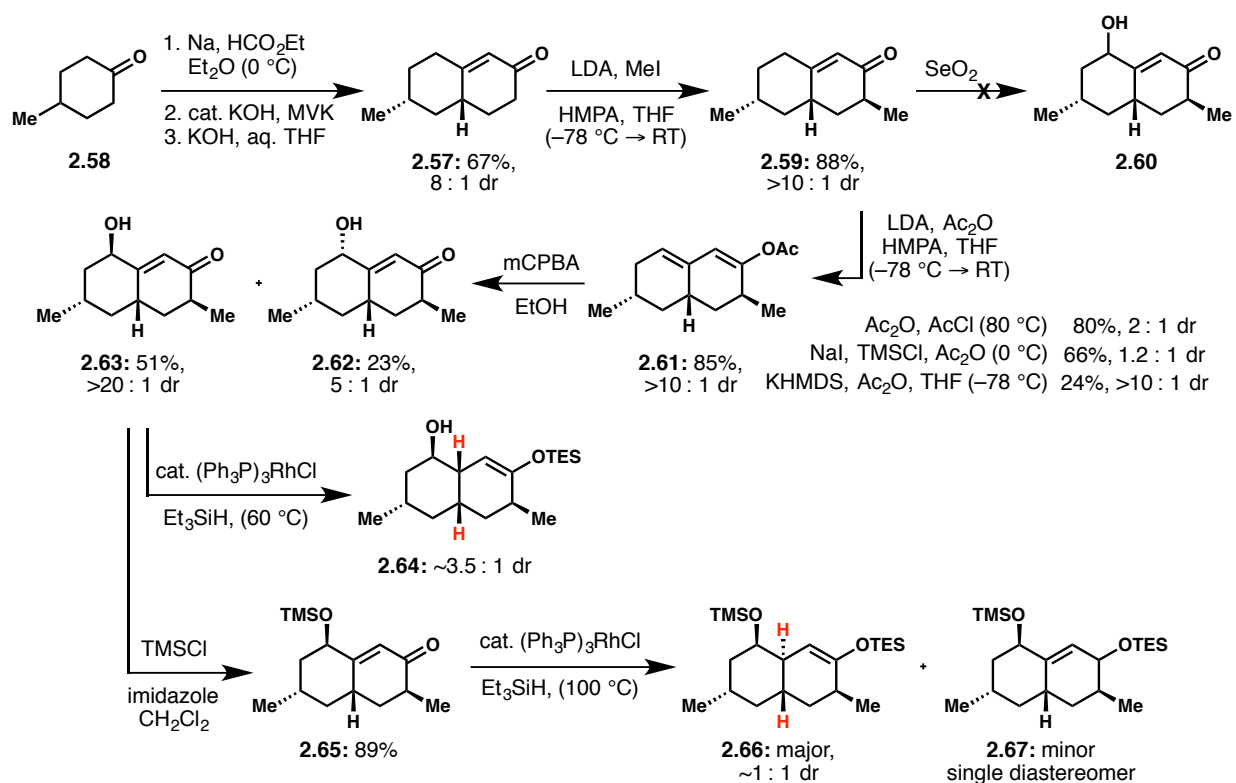
#### 2.4.2 Efforts Towards a *trans*-Selective Conjugate Hydrosilylation Approach

The previous approach was changed because of perceived complications in having the enoxysilane enolized in the incorrect direction (Scheme 2.10). The basic idea of a reductive enone coupling idea was still exciting, but correct orientation of the C20 enoxysilane in **2.41** was considered important. In a second-generation approach, **2.41** was disconnected back to enone **2.57**.

**Scheme 2.10** Tracing tricyclic bis-enoxysilane **2.41** back to a simple octalone starting material.



**Scheme 2.11** Initial success of a *trans*-selective conjugate hydrosilylation.

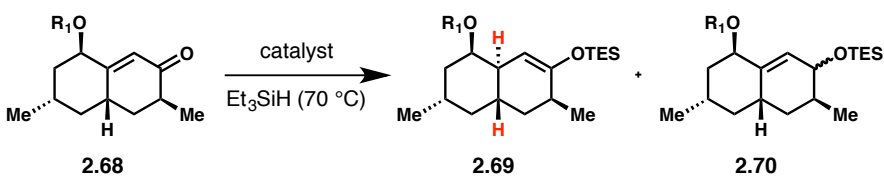


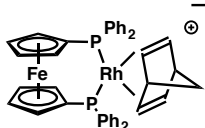
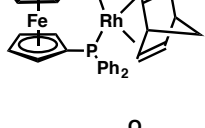
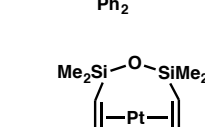
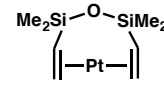
The octalone **2.57** was constructed by Robinson annulation onto 4-methylcyclohexenone **2.58** (Scheme 2.11). Alkylation of the cross-conjugated dienolate with MeI installed the C17 methyl group in excellent diastereoselectivity. A direct allylic oxidation with selenium dioxide did not afford any **2.60** and therefore a two-step protocol was implemented instead. Generation of the conjugated dienoxy acetate was accomplished under a variety of conditions; however, epimerization of the C17 methyl group was consistently observed. The use of LDA with HMPA

and Ac<sub>2</sub>O was eventually successful in affording **2.61** as a single diastereomer. Oxidation of **2.61** with in situ generated DMDO was successful in generating **2.62** and **2.63**; however, low yields and a messy reaction profile were reason to switch to mCPBA. Conjugate reduction of **2.63** with Li/NH<sub>3</sub> only caused elimination of the gamma-hydroxide. Conjugate hydrosilylation was accomplished with Wilkinson's catalyst in triethylsilane at elevated temperatures to afford **2.64** bearing a *cis*-ring fusion. Installing a bulky trimethylsilyl group on the alcohol was designed to block the *cis*-face of the enone and direct to a more *trans*-selective hydrosilylation. Delightfully, this design was successful! The reduction of **2.65** with Wilkinson's catalyst afforded a 1 : 1 mixture of *cis*- and *trans*-ring fusions. Unfortunately, 1,2-reduction of the enone was also observed. The next step was to improve the 1 : 1 dr by using larger alcohol protecting groups and evaluating different catalysts.

Several combinations of alcohol protecting group and catalyst were examined (Table 2.1). A general reactivity screen narrowed the catalysts to those providing clean conjugate reduction. Pivaloyl protecting the alcohol of **2.63** enhanced the selectivity from *cis*-conjugate silylation to an approximately 1 : 1 mixture. Karstedt's catalyst was determined to be the most reliable in affording 1,4 over 1,2-reductions (entry 6). Other platinum sources such as Pt(PPh<sub>3</sub>)<sub>4</sub> and PtO<sub>2</sub> provided similar diastereoselectivity, but were much less active than Karstedt's catalyst or competitively conducted 1,2-reduction (entries 8, 9).

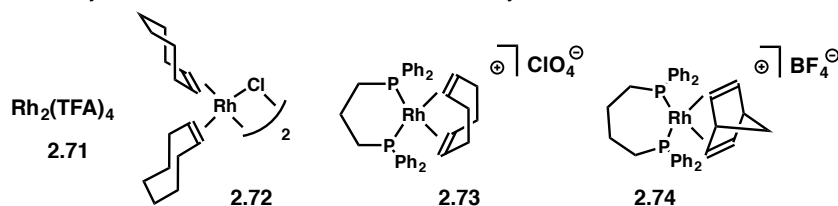
**Table 2.1** Consolidation of conjugate hydrosilylation reactions.



Entry	Catalyst <sup>b</sup>	R <sub>1</sub>	Yield	2.69 trans / cis <sup>a</sup>	2.70
1	Rh <sub>2</sub> (OAc) <sub>4</sub>	TMS		1 : 4	
2		TMS		1 : 4	
3		Bz		1 : 9	
4		Piv		1 : 1.4	
5		Bz		1 : 3.5	
6	Karstedt's Catalyst	Piv	80%	1.1 : 1	
7	Grubbs I, II Stewart–Grubbs	Piv			exclusive
8	Pt(PPh <sub>3</sub> ) <sub>4</sub>	Piv		1.3 : 1	
9	PtO <sub>2</sub>	Piv		1.2 : 1	major
10	Pt/C	Piv			major

<sup>a</sup> Ratio by <sup>1</sup>H NMR of crude reaction mixture without an internal standard

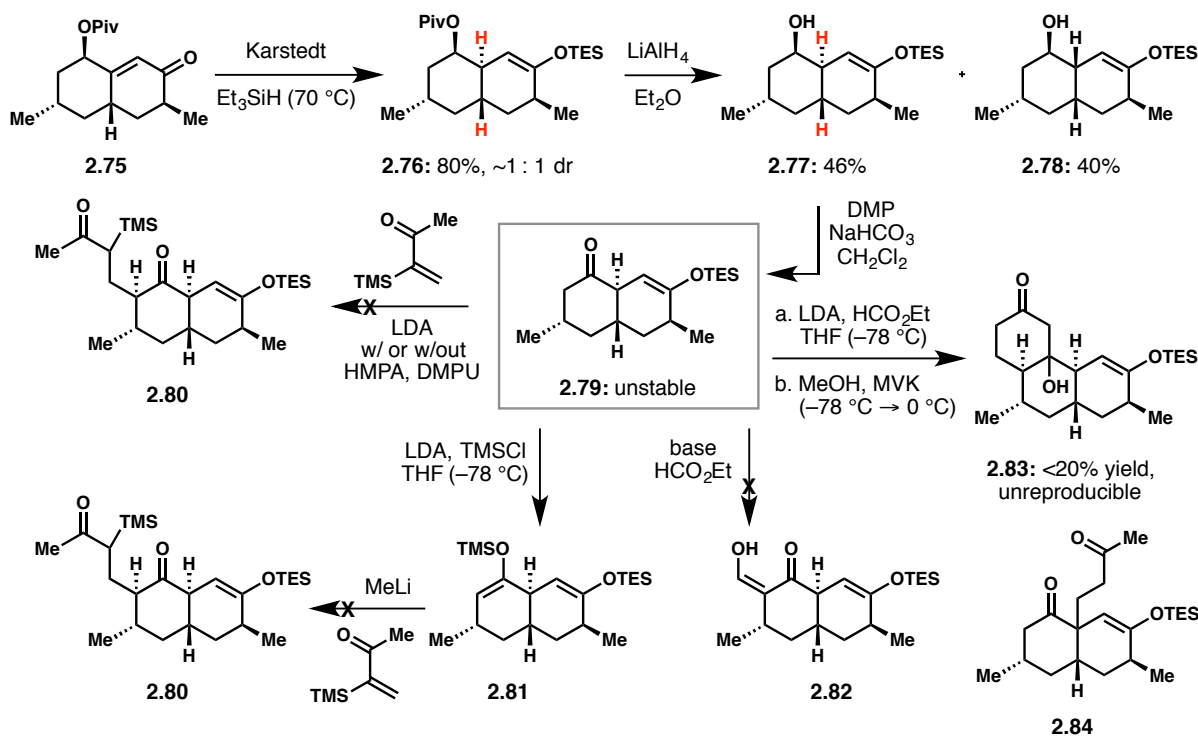
<sup>b</sup> Catalysts with unsuccessful conversions or messy reaction outcomes:



With a relatively acceptable 1.1 : 1 *trans*-selective hydrosilylation established, the route was continued (Scheme 2.12). Deprotection of the pivaloyl group with LiAlH<sub>4</sub> generated separable alcohols **2.77** and **2.78**, both of which were readily handled, stable to silica gel and storable at room temperature. Upon oxidation of **2.77** to ketone **2.79**, the material became much more unstable and needed to be used without purification within an hour. The first attempted Robinson annulation used the Stork–Ganem reagent.<sup>23</sup> However, enolizing **2.79** with LDA and attempting conjugate addition did not furnish any product reminiscent of **2.80**. The

enoxytrimethylsilane **2.81** could be generated cleanly by action of LDA, indicating that enolization occurs as expected. Attempts to reveal the enolate with MeLi were unsuccessful, leading to only decomposition, even upon simple aqueous quench. Standard Claisen condensation conditions of treating **2.79** with base, such as KOt-Bu, NaH or LDA and ethyl formate did not provide any **2.82**. The only conditions successful at generating the desired C–C bond involved treating **2.79** with LDA and ethyl formate, followed by quenching with methanol and excess methyl vinyl ketone. An additional compound, proposed to be **2.84** was also isolated. The low yield and unreliable preparation of **2.83** led to a reevaluation of strategy.

**Scheme 2.12** Exploring a Robinson annulation onto **2.79**.



#### 2.4.3 A Reductive Stork Alkylation

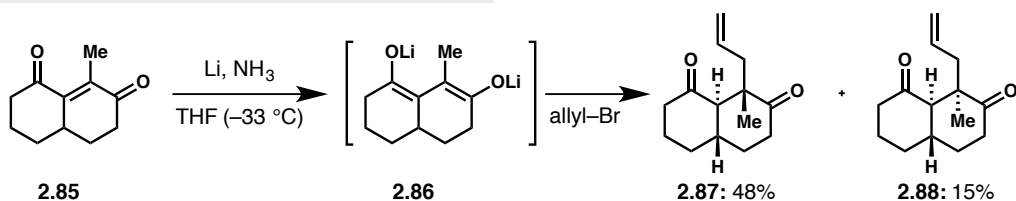
A more robust approach was attempted owing to the difficulty in executing a Robinson



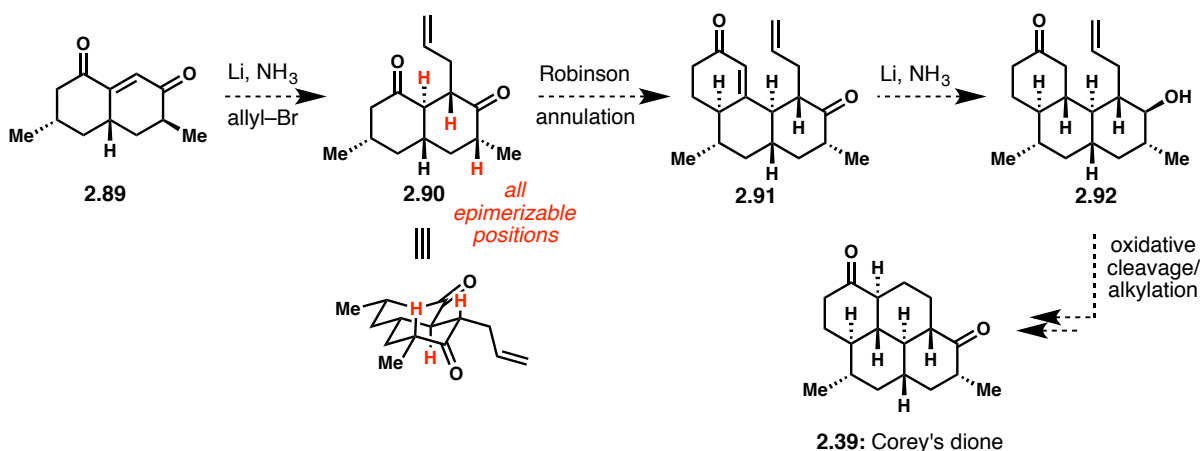
annulation onto **2.79**. An inspiration for the next attempt was Stork's reduction of enedione **2.85** to the dienolate **2.86** and subsequent alkylation (Scheme 2.13A).<sup>24</sup> This procedure has found extensive utility in steroid chemistry<sup>25–28</sup> and other total syntheses,<sup>29,30</sup> making it reliable enough to commit towards DICA. Reductive alkylation of **2.89** would generate **2.90** after epimerization of the three enolizable positions (Scheme 2.13B). A Robinson annulation onto **2.90** via activation through the keto-aldehyde would ensure selective enolization. A final ring closure would then prepare Corey's dione **2.39**.

**Scheme 2.13** Allylation and subsequent ring closures based on a reductive Stork alkylation.

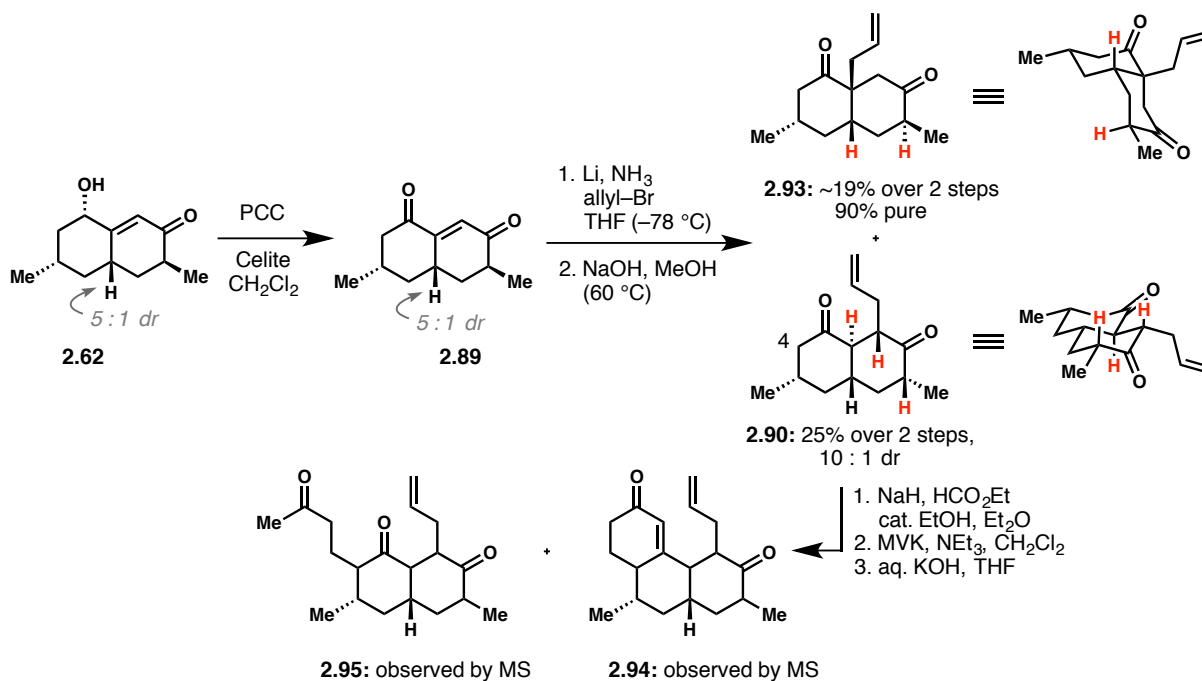
**A. Stork reduction and alkylation of an enedione**



**B. Proposal to incorporate the Stork alkylation methodology towards DICA**



**Scheme 2.14** A reductive Stork alkylation towards DICA.



The undesired minor diastereomer from the hydrosilylation approach, **2.62** was oxidized to enedione **2.89** (Scheme 2.14). Addition of **2.89** to a solution of Li/NH<sub>3</sub> and quenching with allyl bromide generated a mixture of compounds that could be converged to two mono-allylated products **2.93** and **2.90** upon equilibration with base. Di-allylated products were also observed, but not quantified or characterized. The undesired mono-allyl product **2.93** arose from alkylation at the ring fusion. The *cis*-alkylation was assigned because equilibrating conditions after alkylation did not change the <sup>1</sup>H NMR spectrum and this could only be possible if the β-methyl group is already in an equatorial orientation as in **2.93**. The desired allylation product was equilibrated to the all-equatorially oriented **2.90** and then subjected to Robinson annulation. Activation of the C4 position as the keto-aldehyde set up for a selective Michael addition and ring closure. This procedure produced a mixture of compounds, but by mass spectrometry both **2.95** and the fully condensed **2.94** were observed, indicating feasibility of the original plan. The route however, fell out of favor owing to the irreproducibility and low yield of the enolate

alkylation and conception of a more robust entry to molecules similar to **2.94** (Chapter 4). In retrospect, the complications experienced during the reductive alkylation are most likely due to the inexperience in handling dissolving metal reductions.

## **2.5 Conclusions**

The project of synthesizing 7,20-diisocyanoadociane was initiated with the goal of contributing a short and selective synthesis. The synthetic journey started with a bold radical cascade idea. Although it was never attempted, a ketyl or amine radical anion polyene cyclization could still be an interesting methodology to investigate. This early work was essential in probing possible avenues for a synthetic program. Retrospectively, these early explorations had fatal flaws, were discontinued due to inexperience or were simply supplanted by “better” ideas. A continuous evolution in ideas and change in strategy continuously sharpened the trajectory towards a synthesis of DICA and were directly continued in Chapter 4.

## 2.6 Experimental Procedures

### Purifications –

*Solvents:* Dry tetrahydrofuran (THF), diethyl ether (Et<sub>2</sub>O), dichloromethane (CH<sub>2</sub>Cl<sub>2</sub>), toluene, benzene (C<sub>6</sub>H<sub>6</sub>), dimethylformamide (DMF), acetonitrile (MeCN), methanol (MeOH) were obtained by passing commercially available formulations through activated alumina columns. tert-Butyl alcohol (t-BuOH) was purified by distillation from CaH<sub>2</sub>.

*Amines:* Triethylamine (NEt<sub>3</sub>), N,N'-dimethylpropylene urea (DMPU) hexamethylphosphoramide (HMPA), pyridine (py) was purified by distillation from CaH<sub>2</sub>.

*Halides:* Trimethylsilyl chloride was purified by distillation from CaH<sub>2</sub>.

*Miscellaneous:* Tributyltin hydride was purified by distillation. Melonal was obtained from Sigma-Aldrich and purified by column chromatography (100:1 hexanes/EtOAc). Methyl vinyl ketone (MVK) was purified by distillation. Acetic anhydride was purified by fractional distillation. Allyl bromide was purified by distillation and stored over copper beads at –20 °C. Trifluoroacetic anhydride was purified by distillation from CaH<sub>2</sub>.

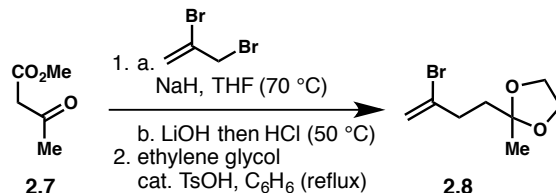
**Titration** – Alkyl lithium reagents were titrated using 2,6-di-(tert-butyl)-4-methylphenol (BHT) as the sacrificial proton source and fluorene as an indicator in THF or using diphenylacetic acid in THF. Grignard reagents were titrated using salicylaldehyde phenylhydrazone in THF.<sup>31</sup>

**Reaction Setup** – All reactions were carried out under an argon atmosphere with dry solvents under anhydrous conditions, unless otherwise noted. Argon balloons were the sole inert

atmosphere used. Reactions run at an ambient temperature of 20–25 °C are designated as room temperature. Microwave reactions were performed in an Anton Paar Microwave. Yields refer to chromatographically and spectroscopically homogeneous materials, unless otherwise stated.

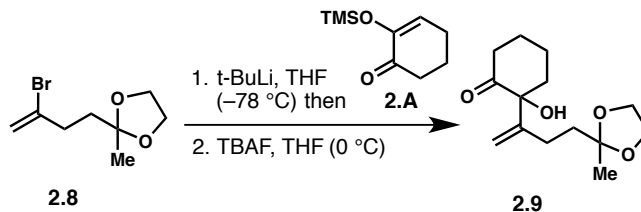
**Analysis** – Thin layer chromatography was performed on 0.25 mm EMD glass-backed TLC plates impregnated with a fluorescent dye and visualized with UV light and KMnO<sub>4</sub> in K<sub>2</sub>CO<sub>3</sub>/NaOH/water or *p*-anisaldehyde in ethanol/aqueous H<sub>2</sub>SO<sub>4</sub>/AcOH and heat as a developing agent. Forced flow (flash) chromatography was performed on EMD Silica 60, mesh 0.04-0.063 silica gel. NMR spectra were recorded on Bruker 500 MHz instrument, obtained at 298 K unless otherwise noted and calibrated to residual undeuterated solvent as an internal reference. Chemical shifts are reported in ppm with the following abbreviations to explain multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, quin = quintuplet, sext = setet, sep = septet, bs = broad signal, m = multiplet. All coupling constants are apparent *J* values measured at the indicated field strengths. FT-IR spectra were recorded on a Perkin-Elmer spectrum RX1 spectrometer. High-resolution mass spectra (HRMS) were recorded on a H<sub>2</sub>O<sub>s</sub> LCT Premier spectrometer using ESI-TOF (electrospray ionization-time of flight). Melting points were measured on a MEL-TEMP II capillary apparatus and stand uncorrected.

## Vinyl Bromide 2.8



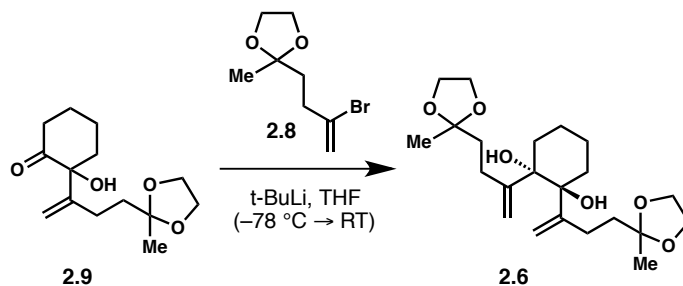
To a stirring suspension of 1.1 g (27 mmol) NaH (60% in mineral oil) in 30 mL THF cooled in an ice bath, was added 3.2 mL (30 mmol) **2.7** before the addition of 2.1 mL (22 mmol) 2,3-dibromopropene.<sup>32</sup> The reaction was warmed to room temperature, fitted with a reflux condenser and heated at 70 °C for 1 hour. After cooling the reaction to room temperature, 60 mL water, an additional 30 mL THF and 2.52 g (60 mmol) LiOH•H<sub>2</sub>O were added and the reaction heated at 50 °C. After 23 hours the reaction was cooled to room temperature and 6 M aq. HCl was added until bubbling subsided. The solution was extracted thrice with EtOAc. All organic layers were collected, washed thrice with sat. aq. NaHCO<sub>3</sub> and brine before being dried over MgSO<sub>4</sub> and stripped of all volatiles. The oil was dissolved in 60 mL C<sub>6</sub>H<sub>6</sub> and refluxed under a Dean-Stark apparatus with 50 mg (0.26 mmol) *p*-TsOH•H<sub>2</sub>O and 1.7 mL (30 mmol) ethylene glycol. After cooling to room temperature, the organic layer was washed thrice with sat. aq. NaHCO<sub>3</sub>, then water and brine before being dried over MgSO<sub>4</sub> and all volatiles removed in vacuo. The material was purified by column chromatography (30:1 hexanes/EtOAc) to afford 2.41 g (50% over 3 steps) **2.8** as a yellow oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.59 (d, *J* = 1.4, 1H), 5.39 (d, *J* = 1.7, 1H), 3.99-3.92 (m, 4H), 2.56-2.53 (m, 2H), 1.95-1.92 (m, 2H), 1.35 (s, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 134.3, 116.3, 109.2, 64.8, 37.7, 36.2, 24.0; IR (thin film) 2982, 2881, 1630, 1055 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>8</sub>H<sub>13</sub>O<sub>2</sub>Br [M+Na]<sup>+</sup> 221.0177 found 221.0181.

## Hydroxyketone 2.9



A solution of 0.82 g (7.3 mmol) 1,2-cyclohexanedione, 1.9 mL (15.0 mmol) TMSCl and 20 mL CH<sub>2</sub>Cl<sub>2</sub> was treated with 3 mL (21.5 mmol) NEt<sub>3</sub> at room temperature. After 5 minutes, the solution was filtered and all volatiles removed in vacuo. The remaining solid was filtered through a fine frit with hexanes and all volatiles removed in vacuo to afford **2.A**. A solution of 1.95 g (8.8 mmol) **2.8** in 20 mL THF was cooled to -78 °C then treated with 13 mL t-BuLi (19.5 mmol, 1.5 M in pentane). After 10 minutes **2.A** in 6 mL THF was slowly added with a 2 mL THF wash. After 2 hours at -78 °C, the reaction was quenched with sat. aq. NH<sub>4</sub>Cl, warmed to room temperature and extracted thrice with EtOAc. The organic layer was dried over MgSO<sub>4</sub> and all volatiles removed in vacuo. The crude oil was dissolved in 20 mL THF, cooled in an ice bath and diluted with 15 mL TBAF (15 mmol, 1 M in THF). The reaction was warmed to room temperature over 20 minutes, diluted with water and extracted thrice with EtOAc. The organic layer was dried over MgSO<sub>4</sub>, all volatiles removed in vacuo and the crude oil purified by column chromatography (5:1→4:1 hexanes/EtOAc) to afford 1.04 g (56%) **2.9** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.14 (t, *J* = 1.6, 1H), 5.12 (s, 1H), 4.19 (s, 1H), 3.97-3.91 (m, 4H), 2.64 (dq, *J* = 13.9, 3.0, 1H), 2.59-2.52 (m, 2H), 2.18-2.06 (m, 2H), 2.01-1.94 (m, 1H), 1.87-1.77 (m, 3H), 1.73 (dt, *J* = 12.9, 3.2, 1H), 1.70-1.56 (m, 2H), 1.33 (s, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 213.4, 147.7, 113.0, 109.6, 81.2, 64.6, 39.0, 38.5, 37.6, 28.2, 25.2, 23.9, 23.0; IR (thin film) 3457, 2938, 2870, 1711, 1641, 1055 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>14</sub>H<sub>22</sub>O<sub>4</sub> [M+Na]<sup>+</sup> 277.1416 found 277.1410.

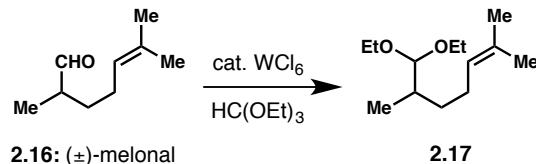
## Diol 2.6



A solution of 495 mg (2.24 mmol) **2.8** in 20 mL THF was cooled to  $-78\text{ }^{\circ}\text{C}$  then treated with 3.3 mL t-BuLi (4.95 mmol, 1.5 M in pentane). The reaction was stirred for 10 minutes then 150 mg (0.59 mmol) **2.9** in 1 mL THF with 0.4 mL THF wash was slowly added. The reaction was warmed to room temperature over 17 hours. Sat. aq.  $\text{NH}_4\text{Cl}$  was added and the reaction extracted thrice with EtOAc. The organic layers were collected, dried over  $\text{MgSO}_4$ , all volatiles removed in vacuo and the crude oil purified by column chromatography (10:1→4:1 hexanes/EtOAc) to afford 130 mg (55%) **2.6** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.01 (s, 2H), 4.96 (s, 2H), 3.97-3.88 (m, 8H), 2.36 (s, 2H), 2.30-2.20 (m, 4H), 2.08-2.01 (m, 2H), 1.85 (td,  $J = 12.5, 4.5$ , 2H), 1.73-1.70 (m, 4H), 1.52-1.50 (m, 2H), 1.45 (d,  $J = 12.7$ , 2H), 1.34 (s, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  154.6, 110.6, 110.0, 77.2, 64.6, 38.6, 33.5, 27.8, 23.7, 20.6; IR (thin film) 3475, 3092, 2980, 2930, 2881, 1630, 1058  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{22}\text{H}_{36}\text{O}_6$   $[\text{M}+\text{Na}]^+$  419.2410 found 419.2403.

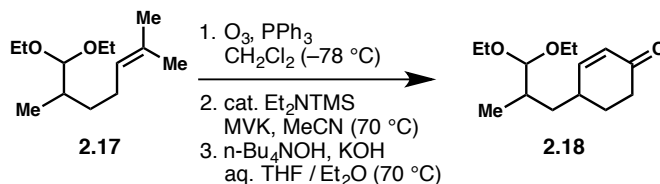


**Ketal 2.17** [Adapted from the literature.]<sup>33</sup>



In a glove box, 101 mg (0.25 mmol)  $\text{WCl}_6$  was weighed into a round bottom flask, the flask sealed with a rubber septum and moved into a fume hood. To this flask was added 3.60 g (25.7 mmol) **2.16** and 10.7 mL (64.2 mmol) ethyl orthoformate. The reaction was stirred for 3 hours after which 15 mL 1M NaOH was added and the contents extracted twice with  $\text{CH}_2\text{Cl}_2$ . The organic layers were combined, washed with water, then brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. Distillation (90 °C/3.5 mmHg) afforded 3.4 g (61%) **2.17** as a colorless liquid.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.17 (tdt,  $J = 7.1, 2.7, 1.4$ , 1H), 4.23 (d,  $J = 6.4$ , 1H), 3.72 (ddq,  $J = 11.2, 9.4, 7.0$ , 2H), 3.55 (dq,  $J = 9.3, 7.1, 2.3$ , 2H), 2.15-2.08 (m, 1H), 2.04-1.96 (m, 1H), 1.82-1.77 (m, 1H), 1.74 (d,  $J = 1.1$ , 3H), 1.67 (s, 3H), 1.65-1.58 (m, 1H), 1.27 (td,  $J = 7.1, 1.7$ , 6H), 1.24-1.16 (m, 1H), 0.977 (d,  $J = 6.8$ , 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  131.3, 124.7, 106.9, 62.0, 36.0, 32.0, 25.7, 25.4, 17.6, 15.3, 14.4.

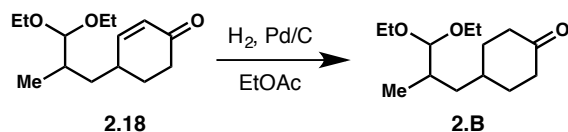
**Cyclohexenone 2.18**



A solution of 2.30 g (10.7 mmol) **2.17** in 90 mL  $\text{CH}_2\text{Cl}_2$  was ozonoloyzed at  $-78$  °C. After excess ozone was purged from the solution, 5.60 g (21.4 mmol) triphenylphosphine was added and the reaction warmed to room temperature. After stirring for 2 hours all volatiles were

removed in vacuo. The solid was stirred with 30 mL pentane at 0 °C, filtered and all volatiles removed in vacuo. The crude material was heated at 70 °C with 0.5 mL (2.7 mmol) Et<sub>2</sub>NTMS and 1.3 mL (16.2 mmol) MVK in 60 mL MeCN. After 38 hours all volatiles were removed in vacuo. The crude material was heated at 70 °C with 300 mg (5.35 mmol) KOH, 0.8 mL 40% aq. n-Bu<sub>4</sub>NOH in 63 mL THF/Et<sub>2</sub>O/H<sub>2</sub>O (1:4:4) mixture. After 5 hours the layers were separated. The aqueous layer was extracted twice with Et<sub>2</sub>O. All organic layers were combined, washed twice with brine, dried over MgSO<sub>4</sub>, filtered and purified by column chromatograph (10:1→8:1 hexanes/EtOAc) to afford 1.40 g (54%, 1:1 dr) **2.18** as colorless liquid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 6.90-6.87 (m, 0.5H), 6.84-6.82 (m, 0.5H), 5.98 (dt, *J* = 10.1, 1.8 Hz, 1H), 4.20 (d, *J* = 5.7 Hz, 0.5H), 4.19 (d, *J* = 5.7 Hz, 0.5H), 3.72-3.65 (m, 2H), 3.54-3.47 (m, 2H), 2.55-2.48 (m, 2H), 2.41-2.32 (m, 1H), 2.15-2.07 (m, 1H), 1.93-1.85 (m, 1H), 1.72 (ddd, *J* = 13.5, 9.2, 4.4 Hz, 1H), 1.67-1.59 (m, 1H), 1.34 (ddd, *J* = 14.0, 9.6, 4.7 Hz, 0.5H), 1.25-1.20 (m, 6.5H), 0.99 (d, *J* = 6.9 Hz, 1.5H), 0.97 (d, *J* = 6.9 Hz, 1.5H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 200.0, 199.9, 156.0, 154.5, 128.9, 128.8, 107.0, 63.0, 62.7, 62.4, 37.0, 36.7, 36.6, 36.3, 34.1, 33.9, 33.6, 33.4, 29.8, 28.0, 15.34, 15.32, 15.2, 14.6; IR (thin film) 2973, 2874, 1680, 1450, 1114, 1059 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>14</sub>H<sub>24</sub>O<sub>3</sub> [M+NH<sub>4</sub>]<sup>+</sup> 258.2069 found 258.2076.

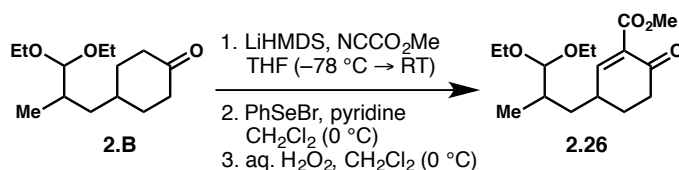
### Cyclohexanone **2.B**



A solution of 305 mg (1.27 mmol) **2.18** in 5 mL EtOAc was stirred over 13 mg (0.4 mol% Pd) 10% Pd/C (55% wetted) under a hydrogen balloon at room temperature. After 7 hours celite was added and the reaction filtered through a pad of celite to afford 302 mg (98%) ketone as a

colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  4.17 (d,  $J = 5.9$  Hz, 1H), 3.67 (dq,  $J = 9.4, 7.0, 4.5$  Hz, 2H), 3.53-3.46 (m, 2H), 2.40-2.28 (m, 4H), 2.09-1.99 (m, 2H), 1.86-1.80 (m, 2H), 1.48 (dtd,  $J = 14.9, 10.5, 4.5$  Hz, 2H), 1.32 (qd,  $J = 12.0, 4.7$  Hz, 1H), 1.21 (td,  $J = 7.0, 2.7$  Hz, 6H), 1.15 (ddd,  $J = 13.9, 9.6, 4.5$  Hz, 1H), 0.95 (d,  $J = 6.8$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  212.4, 107.2, 62.5, 62.4, 40.9, 40.7, 37.4, 34.1, 34.0, 33.2, 31.7, 15.3, 14.9; IR (thin film) 2923, 1716, 1113, 1060  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{14}\text{H}_{26}\text{O}_3$   $[\text{M}+\text{NH}_4]^+$  260.2226 found 260.2237.

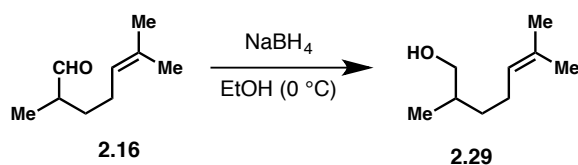
### Cyclohexenone 2.26



To 0.5 mL (0.5 mmol) LiHMDS (1.0 M in THF) cooled to  $-78$  °C was added 50 mg (0.21 mmol) **2.B** in 0.3 mL THF and washed with 0.2 + 0.2 mL THF. After 30 minutes, 23 mg (0.27 mmol) methyl cyanoformate in 0.3 mL THF was added. The reaction was stirred for 10 minutes, the cold bath removed and stirring continued. After 2 hours, sat. aq. NaCl was added and the aqueous layer extracted twice with EtOAc. The organic layers were collected dried over  $\text{MgSO}_4$ , filtered, all volatiles removed in vacuo and the crude material taken on to the next step. To an ice cold solution of 55 mg (0.23 mmol) PhSeBr in 0.3 mL  $\text{CH}_2\text{Cl}_2$  was added 0.1 mL (0.25 mmol, 2.5 M in  $\text{CH}_2\text{Cl}_2$ ) pyridine. After 5 minutes, crude  $\beta$ -ketoester was added with 0.6 mL  $\text{CH}_2\text{Cl}_2$ . The reaction was stirred at 0 °C for 30 minutes then room temperature for 1.5 hours. The organic layer was washed with sat. aq.  $\text{NH}_4\text{Cl}$ , sat. aq.  $\text{NaHCO}_3$  and brine, then dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. This material was dissolved in 1 mL  $\text{CH}_2\text{Cl}_2$ , cooled

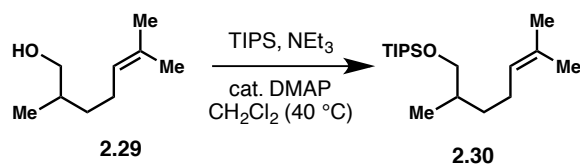
to 0 °C and treated with aq. H<sub>2</sub>O<sub>2</sub> until complete conversion by TLC. The organic layer was washed with sat. NaHCO<sub>3</sub>, sat. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and brine then dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford 57 mg (90% over 3 steps) of crude **2.26** with good purity (>90%). The obtained material was found to be incompatible with silica gel, even when treated with 2% NEt<sub>3</sub> and Florisil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 194.7, 165.33, 165.25, 161.6, 160.3, 131.8, 131.7, 106.93, 106.87, 63.4, 63.0, 62.5, 52.2, 37.8, 37.5, 36.3, 36.0, 34.4, 34.2, 34.1, 34.0, 29.0, 27.4, 15.4, 15.35, 15.33, 14.7; IR (thin film) 2973, 1745, 1691, 1721, 1113, 1060 cm<sup>-1</sup>.

**Melonol 2.29** [Adapted from the literature.]<sup>34</sup>



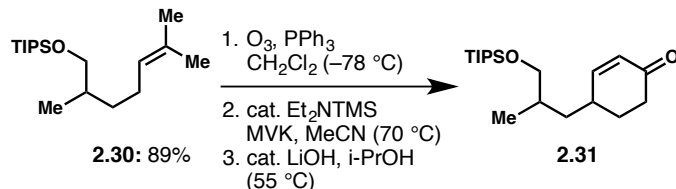
To 30 mL EtOH were added 580 mg (15.3 mmol) NaBH<sub>4</sub> at 0 °C. After 15 minutes, 2.14 g (15.3 mmol) **2.16** was added dropwise. The reaction was stirred for 10 minutes at 0 °C before being diluted with EtOAc and quenched slowly with sat. aq. NH<sub>4</sub>Cl. After warming up to room temperature and ensuring all solids were dissolved by the addition of water, the layers were separated and the aqueous layer extracted thrice with EtOAc. All organic layers were collected, solid NaCl was added, the layers separated. The organic layer was dried over MgSO<sub>4</sub>, filtered, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→7:1 hexanes/EtOAc) to provide 2.01 g (93%) **2.29** as a colorless liquid. The spectral data matched the literature.<sup>34</sup>

## Silylether 2.30



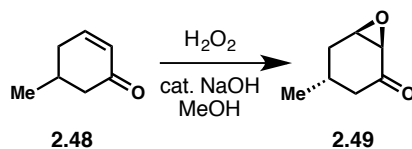
A solution of 2.00 g (14.1 mmol) **2.18**, 4.4 mL (21.1 mmol) TIPSCl, 3.3 mL (23.9 mmol) and 171 mg (1.4 mmol) DMAP in 30 mL CH<sub>2</sub>Cl<sub>2</sub> was heated at 40 °C for 20 hours. After the allotted time, 10 mL sat. aq. NH<sub>4</sub>Cl and 40 mL EtOAc were added and transferred to a separatory funnel. The flask was rinsed with 20 mL (1:1 sat. aq. NH<sub>4</sub>Cl/EtOAc), 10 mL EtOAc and 10 mL water. The layers were separated and the organic layer washed once with sat. aq. NH<sub>4</sub>Cl, water and brine, then dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The material was purified by column chromatography (hexanes) to afford 3.72 g (89%) **2.30** as a colorless liquid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.12 (t, *J* = 7.0 Hz, 1H), 3.55 (dd, *J* = 9.5, 5.7 Hz, 1H), 3.47 (dd, *J* = 9.5, 6.4 Hz, 1H), 2.07-1.93 (m, 2H), 1.69 (s, 3H), 1.65-1.59 (m, 1H), 1.61 (s, 3H), 1.50-1.43 (m, 1H), 1.16-1.01 (m, 22H), 0.91 (d, *J* = 6.7 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 131.1, 125.0, 68.5, 35.6, 33.3, 25.7, 25.6, 18.1, 17.6, 16.8, 12.0; IR (thin film) 2942, 2865, 1653 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>18</sub>H<sub>38</sub>OSi [M+H]<sup>+</sup> 299.2770 found 299.2775.

## Enone 2.31



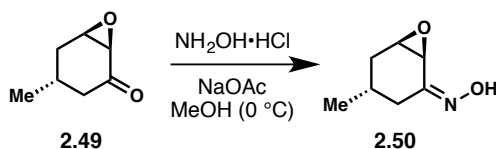
A solution of 3.70 g (12.4 mmol) **2.30** in 60 mL  $\text{CH}_2\text{Cl}_2$  was ozonolyzed at  $-78\text{ }^\circ\text{C}$ . After the excess ozone was purged from the solution, 7.25 g (27.6 mmol) triphenylphosphine were added and the reaction warmed to room temperature. After stirring for 2 hours all volatiles were removed in vacuo. The solid was stirred with 30 mL hexanes at  $0\text{ }^\circ\text{C}$ , filtered, washed in 3 portions with 40 mL ice cold hexanes and all volatiles removed in vacuo. The crude material was heated at  $70\text{ }^\circ\text{C}$  with 0.6 mL (3.1 mmol)  $\text{Et}_2\text{NTMS}$  and 1.5 mL (18.6 mmol) MVK in 30 mL MeCN. After 15 hours all volatiles were removed in vacuo. The crude material was heated at  $55\text{ }^\circ\text{C}$  with 50 mg (1.2 mmol) LiOH in 30 mL *i*-PrOH. After 3 hours all volatiles were removed in vacuo and the crude oil separated between EtOAc and sat. aq.  $\text{NH}_4\text{Cl}$ . The layers were separated and the organic layer then washed once with  $\text{NH}_4\text{Cl}$ , twice with sat.  $\text{NaHCO}_3$ , brine, dried over  $\text{MgSO}_4$ , filtered and purified by column chromatography (20:1 hexanes/EtOAc) to afford 1.9 g (48%, 1:1 dr) **2.31** as a colorless liquid.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  6.90-6.85 (m, 1H), 5.99 (d,  $J = 2.1\text{ Hz}$ , 0.5H), 5.97 (d,  $J = 2.2\text{ Hz}$ , 0.5H), 3.59-3.48 (m, 2H), 2.56-2.48 (m, 2H), 2.41-2.32 (m, 1H), 2.17-2.11 (m, 1H), 1.83-1.77 (m, 1H), 1.74-1.56 (m, 2H), 1.33-1.26 (m, 1H), 1.23-1.18 (m, 1H), 1.14-1.01 (m, 20H), 0.97 (d,  $J = 6.7\text{ Hz}$ , 1.5H), 0.95 (d,  $J = 6.7\text{ Hz}$ , 1.5H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  200.04, 199.97, 155.9, 155.1, 128.82, 128.78, 68.7, 68.4, 38.6, 38.3, 36.9, 36.7, 33.7, 33.6, 33.3, 29.5, 28.5, 18.0, 17.7, 17.3, 16.6, 12.0; IR (thin film) 2942, 2865, 1685, 1462, 1101  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{36}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  347.2382 found 347.2381.

## Epoxide 2.49



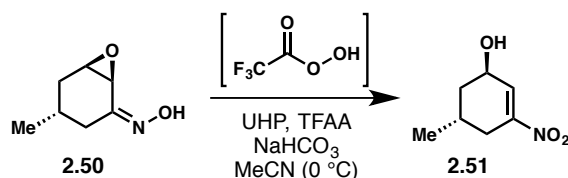
To a solution of 2.86 g (26.0 mmol) **2.48** in 25 mL MeOH cooled in an ice bath, was added 7.8 mL (77.9 mmol) 30% aq. H<sub>2</sub>O<sub>2</sub>. The reaction was initiated with 0.1 mL (0.5 mmol) 5 M aq. NaOH. The internal temperature rose to 10 °C and kept at 10-15 °C with the dropwise addition of an additional 0.1 mL (0.5 mmol) 5 M aq. NaOH. The reaction was stirred for a total time of 30 minutes then poured into 60 mL brine and 40 g ice. The aqueous solution was extracted with 60 mL CH<sub>2</sub>Cl<sub>2</sub> then 4 times with 30 mL CH<sub>2</sub>Cl<sub>2</sub>. The organic layers were combined and washed with 40 mL 1:1 sat. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>/H<sub>2</sub>O solution, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by distillation (98 °C/25 mmHg) to afford 2.54 g (78%) **2.49** as a colorless liquid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 3.56 (dd, *J* = 4.6, 1.9 Hz, 1H), 3.22 (d, *J* = 3.8 Hz, 1H), 2.54 (dd, *J* = 18.0, 4.8 Hz, 1H), 2.36-2.31 (m, 1H), 2.26-2.16 (m, 1H), 1.77 (dd, *J* = 18.0, 11.2 Hz, 1H), 1.60 (ddd, *J* = 14.8, 10.7, 1.0 Hz, 1H), 0.96 (d, *J* = 6.8 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 205.5, 54.9, 54.3, 44.8, 31.4, 22.6, 21.1; IR (thin film) 2958, 1711 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>7</sub>H<sub>10</sub>O<sub>2</sub> [M+NH<sub>4</sub>]<sup>+</sup> 144.1024 found 144.1017.

## Oxime 2.50



To a solution of 2.53 g (20.0 mmol) **2.49** in 15 mL MeOH was added 3.59 g (43.8 mmol) NaOAc and 1.52 g (21.9 mmol) NH<sub>2</sub>OH·HCl sequentially at 0°C. After 1.5 hours at 0 °C the reaction was poured into 60 mL brine and 40 g ice. The aqueous solution was extracted with 60 mL CH<sub>2</sub>Cl<sub>2</sub> then 4 times 30 mL CH<sub>2</sub>Cl<sub>2</sub>. The organic layers were combined and washed with 50 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (2:1 hexanes/EtOAc) and recrystallized from Et<sub>2</sub>O/hexanes to afford 1.78 g (62%) **2.50** as a white solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) mixture of oxime diastereomers δ 8.32-8.29 (m), 8.15-8.06 (m), 4.11 (d, *J* = 3.8 Hz), 3.52 (dd, *J* = 8.6, 3.5 Hz), 3.44 (t, *J* = 1.8 Hz), 2.83-2.78 (m), 2.36 (dd, *J* = 14.9, 2.3 Hz), 2.27 (dt, *J* = 15.0, 2.1 Hz), 1.84-1.80 (m), 1.72 (dd, *J* = 14.9, 11.6 Hz), 1.65 (d, *J* = 12.3 Hz), 1.47 (ddd, *J* = 14.9, 10.2, 1.6 Hz), 0.97 (d, *J* = 6.7 Hz); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) major δ 154.6, 53.0, 44.1, 35.0, 32.8, 23.6, 21.2, minor 155.3, 53.2, 51.7, 32.2, 30.0, 21.3, 20.9; IR (thin film) 3246, 2954, 1654, 1457, 971 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>7</sub>H<sub>10</sub>NO<sub>2</sub> [M-H]<sup>-</sup> 140.0712 found 140.0715.

## Nitroalkene 2.51

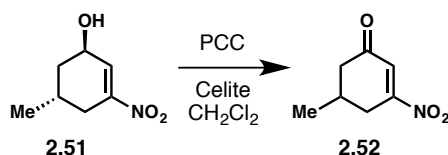


Trifluoroperoxyacetic acid was prepared by the dropwise addition of 5.7 mL (40.3 mmol) trifluoroacetic anhydride in 10 mL MeCN to 4.57 g (48.5 mmol) urea hydrogen peroxide in 50



mL MeCN at 0 °C. This solution was added dropwise via teflon canula to a stirring mixture of 1.96 g (13.9 mmol) **2.50** and 7.09 g (84.4 mmol) NaHCO<sub>3</sub> in 30 mL MeCN at 0 °C. After a total time of 2 hours at 0 °C, the reaction was carefully quenched with 25 mL sat. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and 25 mL water then 80 mL of volatiles removed in vacuo. The aqueous phase was diluted with 20 mL water and extracted 7 times with 50 mL EtOAc (until no oxime was observed by TLC, 2:1 hexanes/EtOAc, R<sub>f</sub> = 0.36). The organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (2:1 hexanes/EtOAc) to afford 1.42 g (65%) **2.51** as yellow sheets. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.21-7.20 (m, 1H), 4.57 (d, *J* = 2.1 Hz, 1H), 2.83 (q, *J* = 11.1 Hz, 1H), 2.09 (t, *J* = 10.9 Hz, 2H), 1.84 (t, *J* = 16.3 Hz, 2H), 1.51-1.45 (m, 1H), 1.11 (d, *J* = 5.9 Hz, 4H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 151.8, 131.0, 63.8, 38.0, 32.1, 24.1, 20.7; IR (thin film) 3368, 2956, 1521, 1337 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>7</sub>H<sub>10</sub>NO<sub>3</sub> [M-H]<sup>-</sup> 156.0661 found 156.0667.

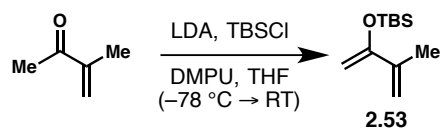
### Nitroenone **2.52**



A flask was charged with 1.41 g (8.97 mmol) **2.51**, 52 mL CH<sub>2</sub>Cl<sub>2</sub> and 9.7 g Celite. To the reaction was added 4.83 g (22.4 mmol) PCC portionwise and the contents stirred for 4 hours. The reaction was concentrated to 25% volume, filtered through silica gel with 500 mL Et<sub>2</sub>O and all volatiles removed in vacuo. The crude material was purified by column chromatography (4:1 hexanes/EtOAc) to afford 1.19 g (85%) **2.52** as a yellow oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 6.94 (s, 1H), 3.07 (dd, *J* = 18.8, 4.6 Hz, 1H), 2.62 (dd, *J* = 16.4, 3.4 Hz, 1H), 2.57-2.51 (m, 1H), 2.41-2.35 (m, 1H), 2.21 (dd, *J* = 16.3, 12.2 Hz, 1H), 1.21 (d, *J* = 6.6 Hz, 3H); IR (thin

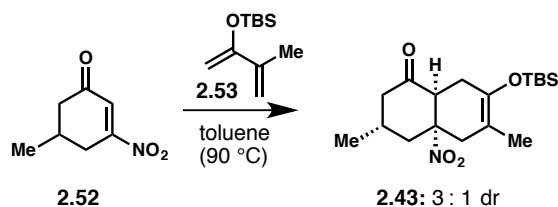
film) 2960, 1694, 1531, 1343, 1324, 754, 714  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_7\text{H}_8\text{NO}_3$  [ $\text{M}-\text{H}$ ] $^-$  154.0504 found 154.0502.

**Diene 2.53** [Adapted from the literature.]<sup>35</sup>



To a solution of LDA, prepared from 5.5 mL (14.3 mmol, 2.60 M / hexanes) n-BuLi and 2.1 mL (15.0 mmol) diisopropylamine in 15 mL THF was added 1.10 g (13.0 mmol) methyl isopropenyl ketone in 1 mL THF at  $-78\text{ }^\circ\text{C}$ . After 10 minutes, 3.5 mL (28.9 mmol) DMPU was added neat. The reaction was stirred at  $-78\text{ }^\circ\text{C}$  for 10 minutes then treated with 1.95 g (13.0 mmol) TBSCl in 1 mL THF and stirred for 10 minutes before the cold bath was removed. After 50 minutes the reaction was quenched with the addition of 60 mL half sat.  $\text{NaHCO}_3$  at  $0\text{ }^\circ\text{C}$ . The solution was extracted with 75 mL hexanes and the organic phase washed with 40 mL water, 20 mL brine, then dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The residue was purified by column chromatography (200:1 hexanes/ $\text{NEt}_3$ ) to afford 1.50 g (58%) **2.53** as a colorless liquid. The spectral data was identical to the literature.<sup>35</sup>

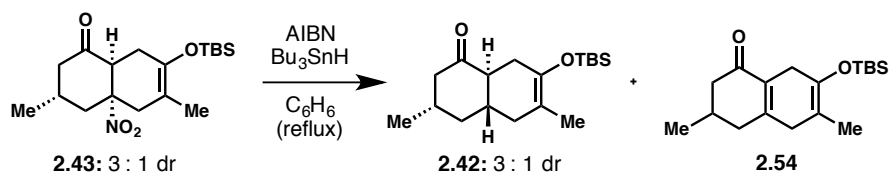
**Cycloadduct 2.43**



A microwave vial was charged with 107 mg (0.69 mmol) **2.52** and 255 mg (1.28 mmol) **2.53**, placed under argon and set into an oil bath at  $90\text{ }^\circ\text{C}$ . After 7 hours the reaction was cooled and

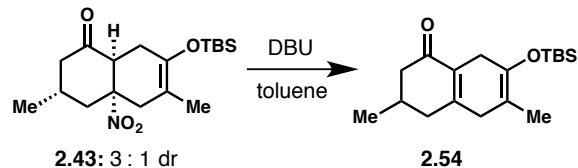
purified by column chromatography (10:1 hexanes/EtOAc) to afford 145 mg (59%, 3:1 dr) **2.43** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$   $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm) major  $\delta$  204.8, 140.4, 105.9, 91.2, 50.7, 48.0, 44.2, 34.0, 29.8, 25.8, 25.7, 25.3, 21.9, 16.1, 3.9, 4.2; IR (thin film) 2923, 1722, 1542, 1257, 1194  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{31}\text{NO}_4\text{Si}$   $[\text{M}+\text{Na}]^+$  376.1920 found 376.1918.

### Decalin 2.42



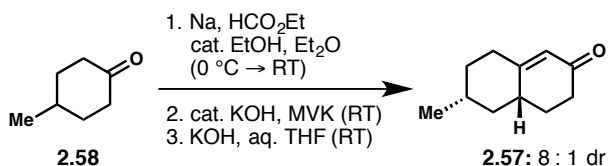
A 10 mL round bottom flask was charged with 50 mg (0.14 mmol) **2.43**, 0.13 mL (0.47 mmol)  $\text{Bu}_3\text{SnH}$  and 0.5 mL  $\text{C}_6\text{H}_6$  then the reaction was heated to 80  $^\circ\text{C}$ . AIBN was added portionwise until complete conversion by TLC. The mixture was concentrated in vacuo and purified by column chromatography (20:1 hexanes/EtOAc) to afford 18 mg (41%) **2.42** and 9 mg (20%) **2.54**. **2.42**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  2.41-1.87 (m, 10H), 1.66-1.53 (m, 5H), 1.21-0.93 (m, 14H), 0.24-0.07 (m, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  211.3, 141.9, 110.0, 50.5, 49.8, 40.9, 39.2, 38.4, 33.5, 29.4, 26.0, 25.8, 22.4, 16.0, -3.7, -3.9; IR (thin film) 2927, 2853, 1712, 1686, 1176  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{32}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  331.2069 found 331.2060. **2.54**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  2.86 (s, 4H), 2.51-2.47 (m, 1H), 2.28-2.00 (m, 5H), 1.61 (m, 1H), 1.06 (d,  $J = 6.4$  Hz, 3H), 0.97 (s, 10H), 0.15 (s, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  198.6, 153.0, 141.5, 129.3, 106.3, 45.7, 39.5, 38.4, 29.7, 28.6, 25.8, 21.3, 15.1, -3.7; IR (thin film) 2954, 1670  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{30}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  329.1913 found 329.1912.

## Cyclohexadiene 2.54



To a solution of 53 mg (0.15 mmol) **2.43** in 0.4 mL toluene was added 27 mg (0.18 mmol) DBU in 0.3 mL toluene at room temperature. After 30 minutes the solution was loaded onto a column and eluted (10:1 hexanes/EtOAc) to provide 42 mg (93%) **2.54** as a white solid.

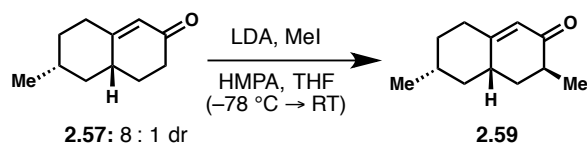
## Enone 2.57



To 100 mL ice cold Et<sub>2</sub>O, under a purging argon atmosphere was added 1.2 g (52.2 mmol) Na metal cut and flattened then concomitantly 5.00 g (44.6 mmol) **2.58** and 5.3 mL (66.2 mmol) ethyl formate via syringe. The reaction was initiated with 0.3 mL EtOH and stirred for 1 hour at 0 °C during which the reaction precipitated a thick orange solid. The ice bath was removed and stirring continued for 9 hours. A single portion of 50 mL water was added slowly, the layers separated and the aqueous phase washed twice with 50 mL Et<sub>2</sub>O. The aqueous layer was acidified with 20 mL 6M HCl and extracted twice with 50 mL Et<sub>2</sub>O. The organic layers were combined, washed with 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. To crude β-ketoaldehyde and 8.5 mL (102 mmol) methyl vinyl ketone was added 200 mg (3.56 mmol) powdered KOH in 50 mg portion at room temperature until the reaction produced an exotherm. After 45 min, the contents were diluted with 50 mL EtOAc and 50 mL sat. NH<sub>4</sub>Cl, the contents transferred to a separatory funnel and the flask rinsed twice with 25 mL EtOAc and

10 mL water. The layers were separated, then the organic layer washed with 20 mL water, 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. To crude tricarbonyl dissolved in 150 mL THF and cooled in an ice bath was added dropwise an ice cold solution of 11.2 g (200 mmol) KOH in 150 mL water. After the addition over 0.5 hour, the ice bath was removed and the reaction stirred for an additional 16.5 hours before 120 mL volatiles were removed in vacuo. The remaining solution was extracted with 100 mL and two times 50 mL EtOAc, then the organic layers were combined, washed with 20 mL water, 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The residue was purified by column chromatography (8:1 hexanes/EtOAc) to afford 4.72 g (67% over 3 steps, 8:1 dr) **2.57** as a colorless liquid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) major δ 5.82 (s, 1H), 2.46-2.23 (m, 5H), 2.08 (dq, J = 13.7, 4.7 Hz, 1H), 1.92-1.83 (m, 2H), 1.72-1.57 (m, 3H), 1.14-1.05 (m, 1H), 0.95 (d, J = 6.6 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 200.1, 166.9, 124.3, 42.9, 37.5, 36.6, 35.3, 35.0, 31.9, 29.2, 21.9; IR (thin film) 2923, 2865, 1668, 1620, 1455 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>11</sub>H<sub>16</sub>O [M+NH<sub>4</sub>]<sup>+</sup> 182.1545 found 182.1552.

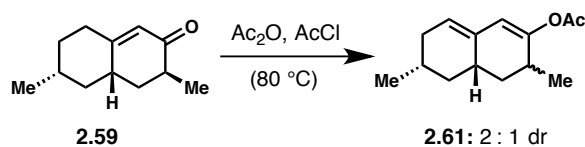
### Enone **2.59**



To a solution of LDA, prepared from 12.7 mL (35.8 mmol) 2.61 M n-BuLi/hexanes and 5.5 mL (39.2 mmol) diisopropylamine in 30 mL THF at 0 °C, was added at -78 °C, 4.90 g (29.8 mmol) **2.57** with 30 mL THF. After 30 minutes, 6.3 mL (36.2 mmol) HMPA was added neat. The reaction was kept at -78 °C for 10 minutes, then treated with 5.6 mL (90.0 mmol) methyl iodide neat and stirred for 10 minutes before the cold bath was removed. After 50 minutes the reaction

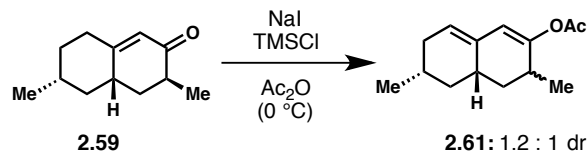
was quenched with the addition of 150 mL half sat.  $\text{NH}_4\text{Cl}$  at 0 °C and 40 mL of volatiles were removed in vacuo. The mixture was poured into a separatory funnel and the flask rinsed with 150 mL EtOAc. The layers were separated and the organic phase washed with 50 mL water and 30 mL brine. All aqueous layers were collected and extracted with 50 mL EtOAc. This organic layer was washed with 10 mL brine, combined with the previous organic phase, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The residue was purified by column chromatography (12:1 hexanes/EtOAc) to afford 4.71 g (88%) **2.59** as a light yellow liquid.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.74 (s, 1H), 2.43-2.37 (m, 3H), 2.23 (td,  $J = 13.6, 4.5$  Hz, 1H), 1.91-1.85 (m, 3H), 1.78-1.67 (m, 2H), 1.11 (t,  $J = 6.2$  Hz, 3H), 1.14-1.02 (m, 2H), 0.94 (d,  $J = 6.6$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  203.1, 166.2, 122.4, 42.6, 38.6, 35.9, 35.9, 35.5, 35.4, 32.3, 21.8, 15.8; HRMS (ESI) calculated for  $\text{C}_{12}\text{H}_{18}\text{O}$   $[\text{M}+\text{H}]^+$  179.1442 found 179.1436.

### Dienol Acetate 2.61



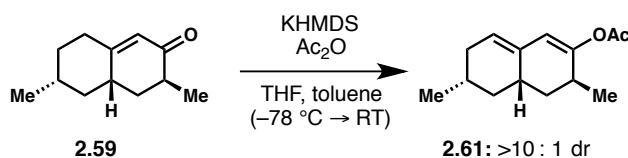
A solution of 50 mg (0.28 mmol) **2.59** in 3.7 mL  $\text{Ac}_2\text{O}/\text{AcCl}$  (2.6:1) was heated at 80 °C for 1 hour. The reaction was cooled and all volatiles removed in vacuo then taken up in 20 mL EtOAc and washed thrice with  $\text{NaHCO}_3$ , brine and dried over  $\text{MgSO}_4$ , filtered. All volatiles were removed and the crude material purified by column chromatography (20:1 hexanes/EtOAc) to afford 49 mg (80%, 2:1 dr) **2.61** acetate as a colorless oil.

## Dienol acetate 2.61



To a solution of 50 mg (0.28 mmol) **2.59** in 1.4 mL Ac<sub>2</sub>O was added sequentially 128 mg (0.86 mmol) NaI then 0.11 mL (0.86 mmol) TMSCl at 0 °C. After 2 hours in the ice bath, 0.5 mL NEt<sub>3</sub> and 10 mL sat. aq. NaHCO<sub>3</sub> were added. The reaction was further diluted with 20 mL EtOAc and the layers separated. The organic layer was washed with 10 mL 1 M NaOH, 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. Purification of the crude material by column chromatography (30:1 hexanes/EtOAc) provided 41 mg (66%, 1.2:1 dr) **2.61** as a colorless oil.

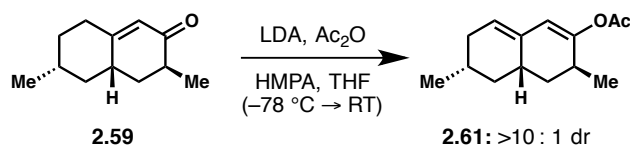
## Dienol Acetate 2.61



A solution of 14.5 mL (7.25 mmol) 0.5 M KHMDS/toluene in 10 mL THF was treated with 1.00 g (5.60 mmol) enone dissolved in 20 mL THF at -78 °C. After 20 minutes the solution was canula transferred to a stirring mixture of 1.6 mL (7.0 mmol) Ac<sub>2</sub>O in 20 mL THF cooled to -78 °C. The cold bath was removed and stirring continued for 1 h as the reaction warmed to room temperature. The reaction was quenched with 100 mL water, volatiles removed in vacuo and the remaining liquid extracted with 60 mL EtOAc. The organic layer was washed with 50 mL water, 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. Purification of the

residue by column chromatography (100:1 hexanes/EtOAc) provided 298 mg (24%) dienol acetate as a colorless oil.

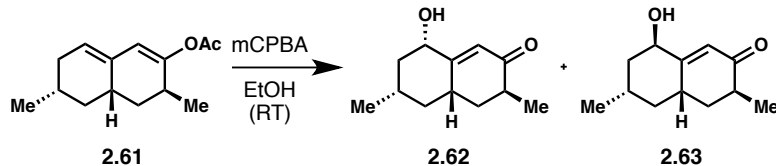
### Dienol acetate **2.61**



LDA was prepared from 4.8 mL (34.2 mmol) *i*-Pr<sub>2</sub>NH, 12.0 mL (31.3 mmol) 2.61 M *n*-BuLi/hexanes in 40 mL THF at 0 °C was cooled to -78 °C and treated with 12.0 mL (69.0 mmol) HMPA. After 20 minutes, 4.65 g (26.1 mmol) **2.59** was added with 30 mL THF. After 20 minutes the solution was canula transferred into a solution of 7.4 mL (78.4 mmol) Ac<sub>2</sub>O in 50 mL THF stirring at -78 °C. Cooling was continued for 10 minutes before the bath was removed. After an additional 50 minutes, 100 mL sat. NaHCO<sub>3</sub> was added, approximately 80 mL volatiles removed in vacuo and the remaining liquid partitioned between an additional 100 mL water and 150 mL EtOAc. The layers were separated and the organic layer washed with 100 mL water, 50 mL water and 30 mL brine, then dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (30:1 hexanes/EtOAc) to provide 4.92 g (85%) **2.61** as a light yellow oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.77 (s, 1H), 5.47 (t, *J* = 2.4 Hz, 1H), 2.45 (t, *J* = 6.7 Hz, 1H), 2.37-2.34 (m, 1H), 2.20-2.17 (m, 1H), 1.78-1.57 (m, 6H), 1.12 (d, *J* = 7.2 Hz, 3H), 0.98 (d, *J* = 6.4 Hz, 3H), 1.02-0.92 (m, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 169.4, 152.3, 134.8, 124.4, 117.1, 38.7, 36.7, 34.7, 31.8, 30.1, 29.2, 22.4, 21.2, 18.4; IR (thin film) 2926, 2871, 1742 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>14</sub>H<sub>20</sub>O<sub>2</sub> [M+H]<sup>+</sup> 221.1542 found 221.1534.



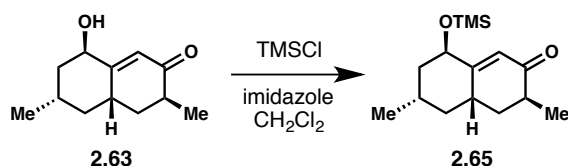
## Alcohols 2.62 and 2.63



To 8.55 g (38.1 mmol, 77% purity) mCPBA in 80 mL EtOH cooled to 0 °C was added 4.92 g (22.3 mmol) **2.61** in 40 mL EtOH. The ice bath was removed and stirred for 1 hour before slowly quenching with 150 mL equal parts sat. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>/sat. NaHCO<sub>3</sub>/water at 0 °C. Approximately 100 mL volatiles were removed in vacuo and the mixture extracted with 150 mL EtOAc. The organic phase was washed with 60 mL equal parts sat. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>/sat. NaHCO<sub>3</sub>/water, 60 mL water and 30 mL brine. All aqueous washings were combined and extracted thrice with 50 mL EtOAc. These organic layers were combined, washed with 30 mL brine and combined with the previous organic solution. After drying over MgSO<sub>4</sub>, the solution was filtered and all volatiles removed in vacuo. The residue was purified by column chromatograph (6:1 hexanes/EtOAc) to provide 2.25 g (51%, >20:1 dr) **2.63** as a white solid, recrystallization from EtOAc/hexanes afforded colorless prisms (mp = 51–53 °C). The column was flushed with EtOAc and the remainder purified by column chromatography (2:1 hexanes/EtOAc) to afford 1.04 g (23%, 5:1 dr) **2.62** as a light yellow oil. **2.62**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 6.10 (s, 1H), 4.24 (dd, *J* = 5.9, 5.6 Hz, 1H), 2.50–2.44 (m, 1H), 2.35 (dq, *J* = 12.6, 4.6 Hz, 1H), 2.25–2.20 (m, 1H), 1.98–1.77 (m, 5H), 1.18–1.10 (m, 2H), 1.12 (d, *J* = 6.9 Hz, 3H), 1.00 (d, *J* = 6.4 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 202.7, 167.1, 118.5, 71.3, 45.4, 41.8, 38.1, 35.6, 30.9, 30.3, 21.3, 15.4; IR (thin film) 3412, 2953, 2925, 2869, 1664, 1457, 1065 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>12</sub>H<sub>18</sub>O<sub>2</sub> [M+NH<sub>4</sub>]<sup>+</sup> 212.1651 found 212.1650. **2.63**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.82 (d, *J* = 1.1 Hz, 1H), 4.40 (s, 1H), 2.90 (dq, *J* = 11.5, 5.6 Hz, 1H), 2.49–2.42 (m, 1H), 2.21–2.16 (m,

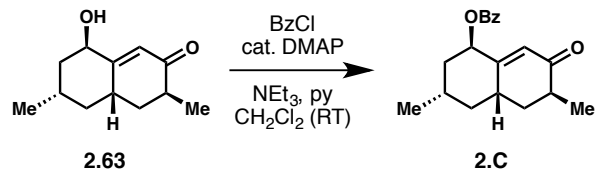
1H), 2.04 (dq,  $J = 14.4, 2.8$  Hz, 1H), 1.95-1.85 (m, 3H), 1.79 (dt,  $J = 13.8, 5.3$  Hz, 1H), 1.28 (td,  $J = 13.2, 2.3$  Hz, 1H), 1.12 (d,  $J = 7.0$  Hz, 3H), 1.16-1.08 (m, 1H), 0.95 (d,  $J = 6.6$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  203.5, 165.1, 122.7, 72.2, 42.6, 42.1, 38.3, 35.5, 30.9, 26.0, 21.5, 25.4; IR (thin film) 3412, 2951, 2923, 2870, 1667  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{12}\text{H}_{18}\text{O}_2$   $[\text{M}+\text{Na}]^+$  217.1205 found 217.1204.

### Silyl Ether **2.65**



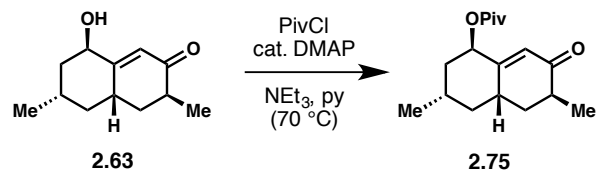
A solution of 37 mg (0.19 mmol) **2.63** and 45 mg (0.66 mmol) imidazole in 2 mL  $\text{CH}_2\text{Cl}_2$  was added 0.040 mL (0.32 mmol) TMSCl at 0 °C. The ice bath was removed and the reaction stirred for 7 hours. After being recooled to 0 °C, the reaction was quenched with 10 mL half sat.  $\text{NaHCO}_3$  and partitioned with 10 mL EtOAc. The layers were separated and the organic layer washed with 3 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The residue was purified by column chromatography (40:2:1 hexanes/EtOAc/ $\text{NEt}_3$ ) to afford 45 mg (89%) **2.65** as a colorless liquid.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.77 (s, 1H), 4.31 (d,  $J = 2.4$  Hz, 1H), 2.85 (dd,  $J = 12.5, 4.7$  Hz, 1H), 2.46-2.41 (m, 1H), 2.20 (td,  $J = 6.4, 2.9$  Hz, 1H), 1.92-1.75 (m, 4H), 1.26-1.15 (m, 2H), 1.11 (d,  $J = 6.9$  Hz, 3H), 0.91 (d,  $J = 6.7$  Hz, 3H), 0.09 (s, 9H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  203.2, 166.4, 121.6, 72.7, 44.3, 42.0, 37.8, 35.9, 31.4, 26.0, 21.5, 15.2, 0.1; IR (thin film) 2954, 2924, 2872, 1680  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{15}\text{H}_{26}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  289.1600 found 289.1597.

## Benzoate 2.C



A vial containing 20 mg (0.104 mmol) **2.63**, 1.2 mg (0.010 mmol) DMAP, 0.05 mL (0.36 mmol) NEt<sub>3</sub>, 0.2 mL (2.48 mmol) pyridine and 0.02 mL (0.17 mmol) benzoyl chloride in 1 mL CH<sub>2</sub>Cl<sub>2</sub> was stirred at room temperature for 16 hours. The reaction was partitioned between sat. NaHCO<sub>3</sub> and EtOAc. The layers were separated and the organic phase washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (8:1 hexanes/EtOAc) to afford 27 mg (86%) **2.68Bz** as a white solid (mp = 93–95 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 8.04 (dd, *J* = 8.3, 1.2 Hz, 2H), 7.59 (t, *J* = 7.4 Hz, 1H), 7.46 (t, *J* = 7.7 Hz, 2H), 6.04 (s, 1H), 5.65 (s, 1H), 2.85 (dd, *J* = 12.0, 5.3 Hz, 1H), 2.51–2.44 (m, 1H), 2.27–2.20 (m, 2H), 1.97 (ddd, *J* = 13.8, 10.2, 5.6 Hz, 2H), 1.83 (dt, *J* = 13.9, 4.9 Hz, 1H), 1.44–1.38 (m, 1H), 1.32–1.24 (m, 1H), 1.11 (d, *J* = 6.9 Hz, 3H), 1.00 (d, *J* = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 202.5, 165.3, 159.7, 133.2, 130.2, 129.6, 128.5, 124.9, 74.6, 41.7, 40.6, 38.1, 35.4, 32.3, 27.2, 21.5, 15.2; IR (thin film) 2954, 2924, 1719, 1679, 1268 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>19</sub>H<sub>22</sub>O<sub>3</sub> [M+Na]<sup>+</sup> 321.1467 found 321.1462.

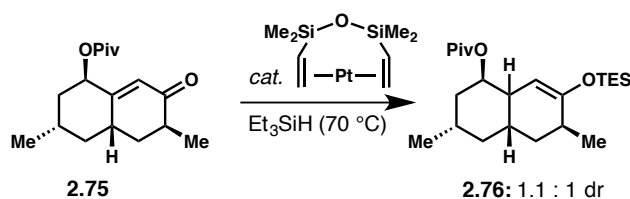
## Pivalate 2.75



A flask was charged with 2.10 g (10.8 mmol) **2.63**, 40 mL pyridine, 3.0 mL (21.5 mmol) NEt<sub>3</sub>, 130 mg (1.06 mmol) DMAP and 5.3 mL (43.1 mmol) trimethylacetyl chloride then stirred at 70

°C. After 18 hours the reaction was cooled, volatiles removed in vacuo and the contents partitioned between 150 mL EtOAc and 150 mL water. The layers were separated and the organic phase washed twice with 100 mL 3M HCl and 50 mL water. The aqueous phases were combined and extracted with 50 mL EtOAc. This organic layer was washed with 20 mL water and combined with the previous organic layer, then washed with 50 mL sat. NaHCO<sub>3</sub>, 50 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The residue was purified by column chromatography (8:1 hexanes/EtOAc) to afford 2.33 g (77%) **2.75** as a light yellow oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.94 (s, 1H), 5.37 (s, 1H), 2.70 (dq, *J* = 11.7, 5.7 Hz, 1H), 2.49-2.42 (m, 1H), 2.06 (dd, *J* = 11.7, 2.6 Hz, 2H), 1.97-1.88 (m, 2H), 1.80 (dt, *J* = 13.8, 5.2 Hz, 1H), 1.33-1.25 (m, 1H), 1.20 (s, 9H), 1.20-1.15 (m, 1H), 1.12 (t, *J* = 12.9 Hz, 3H), 0.97 (d, *J* = 6.4 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 202.7, 177.1, 159.6, 124.7, 73.6, 41.8, 40.3, 38.9, 38.3, 35.3, 31.8, 27.1, 26.9, 21.5, 15.3; IR (thin film) 2957, 2927, 2871, 1730, 1681, 1150 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>17</sub>H<sub>26</sub>O<sub>3</sub> [M+Na]<sup>+</sup> 301.1780 found 301.1783.

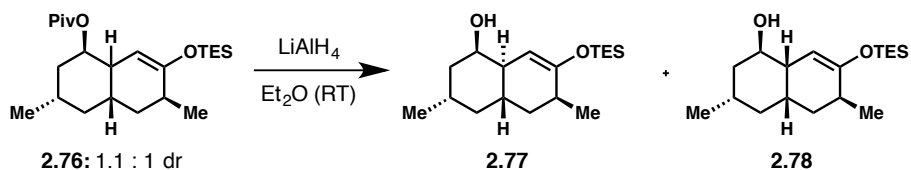
### Enoxysilane 2.76



A flask was charged with 206 mg (0.74 mmol) **2.75**, 36 mg (~2% Pt in xylenes, 0.0037 mmol) Karstedt's Complex and 1.0 mL (6.3 mmol) Et<sub>3</sub>SiH, then heated at 70 °C for 4 hours. The reaction was cooled, all volatiles removed in vacuo and the remaining crude material purified by column chromatography (50:1:1 hexanes/EtOAc/NEt<sub>3</sub>) to afford 251 mg (86%, 1.2:1 dr) enoxysilane as a colorless oil. Mixture of diastereomers: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27

ppm) mixture of diastereomers  $\delta$  5.04 (s), 4.83 (s), 4.48 (s), 4.40 (s), 2.49 (s), 2.21-2.18 (m), 2.13 (t,  $J = 6.7$  Hz), 2.01-1.98 (m), 1.91 (d,  $J = 10.5$  Hz), 1.86-1.73 (m), 1.62 (d,  $J = 12.7$  Hz), 1.53 (td,  $J = 12.7, 6.2$  Hz), 1.42 (d,  $J = 12.8$  Hz), 1.39-1.32 (m), 1.21 (s), 1.19 (s), 1.12-1.03 (m), 1.02-0.92 (m), 0.88 (dd,  $J = 19.6, 6.5$  Hz), 0.74-0.60 (m);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  178.1, 177.8, 155.7, 154.5, 103.0, 102.1, 74.2, 71.9, 44.8, 41.3, 39.0, 38.9, 38.8, 37.8, 37.3, 34.2, 34.0, 33.8, 30.5, 30.1, 28.6, 27.7, 27.3, 27.2, 26.1, 22.5, 22.3, 19.7, 18.6, 6.8, 6.7, 5.1, 4.9; IR (thin film) 2955, 2912, 2876, 1727, 1656, 1164  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{23}\text{H}_{42}\text{O}_3\text{Si}$   $[\text{M}+\text{H}]^+$  395.2982 found 395.2984.

### ***Trans*-Decalin 2.77 and *Cis*-Decalin 2.78**

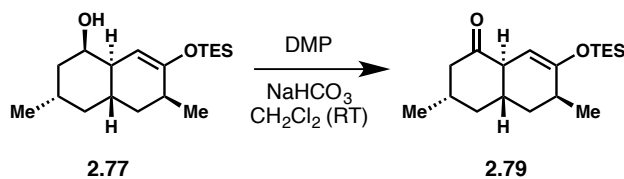


To a slurry of 49 mg (1.29 mmol)  $\text{LiAlH}_4$  in 3.5 mL  $\text{Et}_2\text{O}$  was added 123 mg (0.312 mmol) **2.76** in 1.5 mL  $\text{Et}_2\text{O}$  at 0 °C. The ice bath was removed and the reaction stirred for 30 minutes. Quenching was performed at 0 °C by sequential addition of 0.1 mL  $\text{EtOAc}$ , 0.05 mL water, 0.05 mL 5M  $\text{NaOH}$  and 0.15 mL water, then stirring rapidly for 30 minutes at room temperature. After the addition of  $\text{Na}_2\text{SO}_4$ , the mixture was filtered over Celite, washed through with  $\text{Et}_2\text{O}$  and all volatiles removed in vacuo. The crude material was purified by column chromatography (50:5:1 hexanes/ $\text{EtOAc}$ / $\text{NEt}_3$ ) to afford 41 mg (42%) **2.77** as a colorless oil and 35 mg (36%) **2.78** as a white solid, recrystallization from  $\text{EtOAc}$  afford a white solid (mp = 68–70 °C). **2.77**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  4.57 (s, 1H), 3.86 (s, 1H), 2.16 (t,  $J = 6.8$  Hz, 1H), 1.89 (dt,  $J = 23.4, 9.3$  Hz, 3H), 1.69 (d,  $J = 13.9$  Hz, 1H), 1.59-1.50 (m, 2H), 1.41 (d,  $J = 11.8$

Hz, 2H), 1.13-1.03 (m, 2H), 1.12 (d,  $J = 7.1$  Hz, 3H), 1.03 (s, 9H), 0.90 (d,  $J = 6.5$  Hz, 3H), 0.68 (q,  $J = 8.0$  Hz, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  158.2, 103.7, 70.1, 46.3, 41.3, 40.6, 37.1, 34.0, 27.6, 27.0, 22.3, 19.6, 6.7, 5.1; IR (thin film) 3449, 2953, 2911, 2875, 1656, 1200, 1172  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{17}\text{H}_{34}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  333.2226 found 333.2224.

**2.78:**  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  4.38 (s, 1H), 3.83 (s, 1H), 2.47 (s, 1H), 2.21-2.18 (m, 1H), 2.06 (td,  $J = 8.1, 4.1$  Hz, 1H), 1.86-1.80 (m, 1H), 1.75 (ddd,  $J = 13.4, 6.0, 3.3$  Hz, 1H), 1.60-1.58 (m, 2H), 1.41 (s, 1H), 1.38-1.26 (m, 2H), 1.17-1.07 (m, 2H), 1.05 (d,  $J = 6.9$  Hz, 3H), 0.99 (t,  $J = 8.0$  Hz, 9H), 0.87 (d,  $J = 6.6$  Hz, 3H), 0.68 (q,  $J = 7.9$  Hz, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  154.1, 103.3, 72.3, 42.0, 37.9, 37.0, 34.4, 30.7, 29.4, 25.2, 22.5, 18.6, 6.8, 5.1; IR (thin film) 3344, 2952, 2912, 2875, 1655, 1196  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{34}\text{O}_2\text{Si}$   $[\text{M}+\text{H}]^+$  311.2406 found 311.2403.

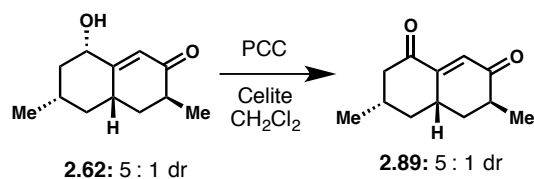
### Ketone 2.79



To a solution of 39 mg (0.126 mmol) **2.77** in 0.6 mL  $\text{CH}_2\text{Cl}_2$  was added 32 mg (0.381 mmol)  $\text{NaHCO}_3$  and 75 mg (0.177) Dess-Martin Periodinane at 0 °C. The ice bath was immediately removed and the contents stirred for 30 minutes. The reaction was poured into 3 mL stirring hexanes, filtered through a cotton filter with hexanes, then washed with 10 mL 1:1 sat.  $\text{NaHCO}_3$ /sat.  $\text{Na}_2\text{S}_2\text{O}_3$ , 5 mL half sat.  $\text{NaHCO}_3$ , 5 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo to afford **2.79** as a colorless oil that decomposed upon purification by chromatography.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.09 (s, 1H), 2.75 (d,  $J = 11.2$  Hz,

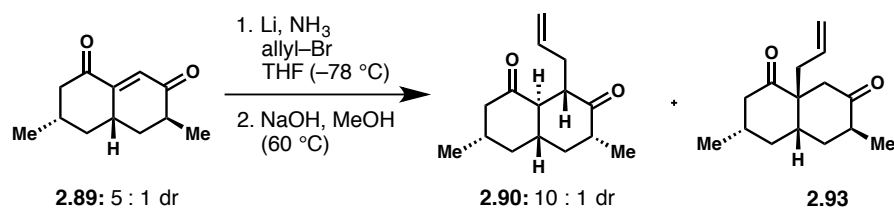
1H), 2.36 (d,  $J = 11.4$  Hz, 1H), 2.19 (t,  $J = 6.7$  Hz, 1H), 2.06-1.98 (m, 2H), 1.80-1.71 (m, 1H), 1.64-1.56 (m, 2H), 1.09 (t,  $J = 7.5$  Hz, 3H), 1.04 (d,  $J = 6.1$  Hz, 3H), 1.10-0.91 (m, 2H), 1.00-0.93 (m, 9H), 0.70 (q,  $J = 7.8$  Hz, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  209.8, 155.4, 98.6, 52.9, 49.2, 40.7, 37.8, 35.7, 34.3, 34.0, 22.4, 19.5, 6.8, 5.0; IR (thin film) 2954, 2913, 2874, 1714, 1657, 1195  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{32}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  331.2069 found 331.2064.

### Enedione **2.89**



A 50 mL round bottom flask was charged with 840 mg (4.32 mmol) **2.62**, 20 mL  $\text{CH}_2\text{Cl}_2$  and 4 g Celite. To the reaction was added 1.55 g (7.2 mmol) PCC and the contents stirred for 2 hours. After the addition of another 0.50 g (2.3 mmol) PCC and 3 hours of stirring, the reaction was diluted with 20 mL  $\text{Et}_2\text{O}$ , filtered through silica gel with 350 mL  $\text{Et}_2\text{O}$  and all volatiles removed in vacuo. The crude material was purified by column chromatography (3:1 hexanes/ $\text{EtOAc}$ ) to afford 740 mg (89%) **2.89** as a yellow wax.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  6.40 (d,  $J = 2.6$  Hz, 1H), 2.86-2.79 (m, 1H), 2.73 (dd,  $J = 12.8, 2.4$  Hz, 1H), 2.63-2.57 (m, 1H), 2.09 (m, 1H), 2.00-1.96 (m, 2H), 1.24 (t,  $J = 12.3$  Hz, 1H), 1.17 (d,  $J = 7.5$  Hz, 3H), 1.16 (m, 2H), 1.09 (d,  $J = 6.0$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  203.3, 200.7, 153.4, 126.6, 49.1, 40.3, 39.5, 36.7, 33.1, 30.0, 22.0, 15.7.

## Allyl Diketone **2.90** and **2.93**



A 25 mL round bottom flask was charged with 9 mg (1.2 mmol) lithium metal and ammonia at –78 °C. To the stirring blue solution was slowly added 60 mg (0.31 mmol) **2.89**. After complete addition isoprene and 0.05 mL (0.58 mmol) allyl bromide were added. The reaction was stirred 10 minutes before solid NH<sub>4</sub>Cl was added and the mixture warmed to room temperature. The mixture was partitioned between water and EtOAc. The organic layers were combined, washed with sat. aq. brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was dissolved in 3 mL methanol and 0.3 mL 1 M NaOH/methanol. The reaction was heated to 60 °C for 3 hours then cooled, partitioned between aq. NH<sub>4</sub>Cl and EtOAc. The organic layers were combined, washed with sat. aq. brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (10:1 hexanes/EtOAc) to afford 18 mg (25%) **2.90** and 14 mg (19%) **2.93**. **2.90**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.81-5.73 (m, 1H), 4.96 (dd, J = 21.5, 12.6 Hz, 2H), 2.74 (t, J = 8.9 Hz, 1H), 2.44-2.26 (m, 5H), 2.09-1.88 (m, 5H), 1.34-1.16 (m, 2H), 1.06 (d, J = 6.1 Hz, 3H), 1.02 (d, J = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 212.4, 209.5, 136.9, 116.4, 58.4, 49.9, 47.6, 44.1, 43.1, 42.5, 41.1, 35.3, 30.9, 22.3, 14.4; **2.93**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.59 (dtd, J = 15.9, 10.5, 5.3 Hz, 1H), 5.00 (dd, J = 16.5, 8.2 Hz, 2H), 2.75 (dd, J = 13.9, 4.5 Hz, 2H), 2.64 (t, J = 6.5 Hz, 1H), 2.38-2.35 (m, 1H), 2.29-2.23 (m, 2H), 2.11 (d, J = 4.8 Hz, 1H), 2.01-1.77 (m, 5H), 1.10-1.01 (m, 7H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 210.0, 209.8, 133.6, 117.9, 55.4, 46.1, 45.7, 40.4, 37.2, 37.0, 36.5, 36.1, 32.0, 22.3, 14.3.



## 2.7 References and Notes

- (1) Corey, E. J.; Magriotis, P. A. *J. Am. Chem. Soc.* **1987**, *109*, 287.
- (2) Miyaoka, H.; Okubo, Y.; Muroi, M.; Mitome, H.; Kawashima, E. *Chem. Lett.* **2011**, *40*, 246.
- (3) Fairweather, K. A.; Mander, L. N. *Org. Lett.* **2006**, *8*, 3395.
- (4) Yoder, R. A.; Johnston, J. N. *Chem. Rev.* **2005**, *105*, 4730.
- (5) Faugeron, V.; Génisson, Y. *Curr. Org. Chem.* **2008**, *12*, 751.
- (6) Fallis, A. . *Tetrahedron Lett.* **1969**, *10*, 1337.
- (7) Clausen, C.; Wartchow, R.; Butenschön, H. *European J. Org. Chem.* **2001**, 93.
- (8) Cossy, J.; BouzBouz, S.; Laghgar, M.; Tabyaoui, B. *Tetrahedron Lett.* **2002**, *43*, 823.
- (9) Roosen, P. C.; Vanderwal, C. D. *Org. Lett.* **2014**, *16*, 4368.
- (10) Firouzabadi, H.; Iranpoor, N.; Karimi, B. *Synth. Commun.* **1999**, *29*, 2255.
- (11) Jung, M. E.; McCombs, C. A. *Org. Synth.* **1978**, *58*, 163.
- (12) Kozmin, S. A.; Rawal, V. H. *J. Org. Chem.* **1997**, *62*, 5252.
- (13) Tricotet, T.; Brückner, R. *European J. Org. Chem.* **2007**, 1069.
- (14) Lavallée, J.-F.; Spino, C.; Ruel, R.; Hogan, K. T.; Deslongchamps, P. *Can. J. Chem.* **1992**, *70*, 1406.
- (15) Audran, G.; Brémond, P.; Feuerstein, M.; Marque, S. R. A.; Santelli, M. *Tetrahedron* **2013**, *69*, 8325.
- (16) Liu, H.-J.; Ngooi, T. K.; Browne, E. N. C. *Can. J. Chem.* **1988**, *66*, 3143.
- (17) Chen, K.; Baran, P. S. *Nature* **2009**, *459*, 824.
- (18) Pronin, S. V.; Shenvi, R. A. *J. Am. Chem. Soc.* **2012**, *134*, 19604.
- (19) Streuff, J. *Synthesis* **2013**, 281.

- (20) Kim, W. H.; Lee, J. H.; Danishefsky, S. J. *J. Am. Chem. Soc.* **2009**, *131*, 12576.
- (21) Corey, E. J.; Estreicher, H. *Tetrahedron Lett.* **1981**, *22*, 603.
- (22) Cheng, X.; Waters, S. P. *Org. Lett.* **2010**, *12*, 205.
- (23) Stork, G.; Ganem, B. *J. Am. Chem. Soc.* **1973**, *95*, 6152.
- (24) Stork, G.; Logusch, E. W. *J. Am. Chem. Soc.* **1980**, *102*, 1218.
- (25) Stork, G.; Logusch, E. W. *J. Am. Chem. Soc.* **1980**, *102*, 1219.
- (26) Stork, G.; Winkler, J. D.; Shiner, C. S. *J. Am. Chem. Soc.* **1982**, *104*, 3767.
- (27) Nemoto, H.; Matsuhashi, N.; Imaizumi, M.; Nagai, M.; Fukumoto, K. *J. Org. Chem.* **1990**, *55*, 5625.
- (28) Jung, M. E.; Yoo, D. *Org. Lett.* **2011**, *13*, 2698.
- (29) Ushakov, D. B.; Maier, M. E. *Synlett* **2013**, 705.
- (30) Orru, R. V. A.; Wijnberg, J. B. P. A.; de Groot, A. *J. Org. Chem.* **1995**, *60*, 4233.
- (31) Love, B. E.; Jones, E. G. *J. Org. Chem.* **1999**, *64*, 3755.
- (32) Lespieau, R.; Bourguel, M. *Org. Synth.* **1925**, *5*, 49.
- (33) Firouzabadi, H.; Iranpoor, N.; Karimi, B. *Synth. Commun.* **1999**, *29*, 2255.
- (34) Lee, J.; Ryu, T.; Park, S.; Lee, P. H. *J. Org. Chem.* **2012**, *77*, 4821.
- (35) Ireland, R. E.; Thompson, W. J. *J. Org. Chem.* **1979**, *44*, 3041.

**CHAPTER 3:**  
**AN OXY-COPE/TRANSANNULAR MICHAEL APPROACH**  
**TOWARDS 7,20-DIISOCYANOADOCIANE**

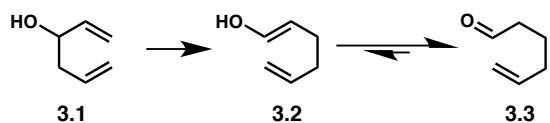
**3.1 Background on the Electronic Reorganization of 3-Oxy-1,5-Hexadiene Systems (the oxy-Cope Rearrangement)**

Pericyclic reactions have enabled the construction of a myriad of complex structures because of their ability to perform difficult C–C bond forming transformations and the facility of their coupling in tandem with a host of other reactions.<sup>1</sup> Among these, the oxy-Cope rearrangement (Scheme 3.1A) has been widely adopted for its mild conditions, predictable outcome and ease of starting material synthesis.<sup>2-4</sup>

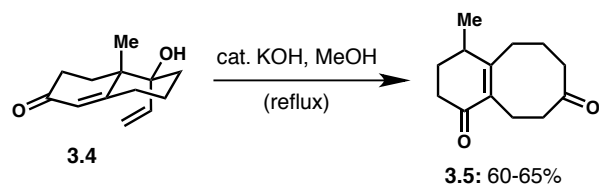
One of the earliest investigators of oxy-Cope systems was Swaminathan with the groundwork laid by rearrangement of 1,5-hexadiene **3.4** to **3.5** in the presence of catalytic amounts of KOH (Scheme 3.1B).<sup>5</sup> This first report of a base-catalyzed reorganization of an oxy-Cope system was reported<sup>6</sup> before the term “oxy-Cope” was coined<sup>7</sup> and over a decade prior to the classic disclosure of significant rate increase using potassium alkoxide by Evans and Golob.<sup>8</sup> Several mechanistic pathways have been proposed for the transformation (Scheme 3.2). The original report suggests the product arises via a vinylogous retro-aldol/conjugate addition/isomerization cascade (Scheme 3.2A); however, after description of the oxy-Cope reaction,<sup>7</sup> an alternative oxy-Cope reaction/isomerization was also suggested (Scheme 3.2B).<sup>9</sup>

**Scheme 3.1** Rearrangement of the 3-oxy-1,5-hexadiene system.

**A. General mechanism of the [3,3] oxy-Cope rearrangement**

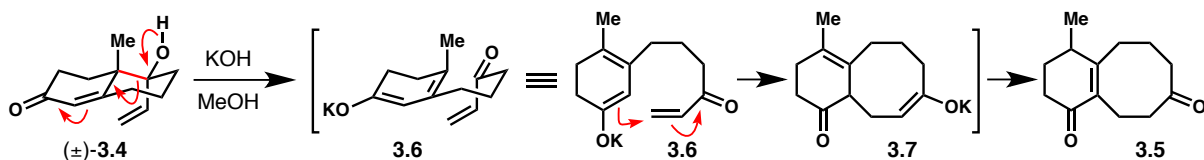


**B. First reported base-induced rearrangement of an oxy-Cope system**

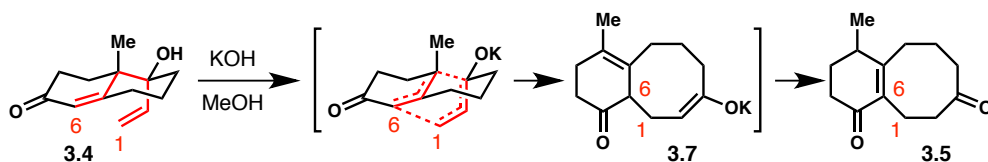


**Scheme 3.2** The two proposed mechanisms to explain the conversion of 3.4 to 3.5.

**A. The vinylogous retro-aldol / conjugate addition / isomerization pathway**



**B. The anionic oxy-Cope / isomerization pathway**

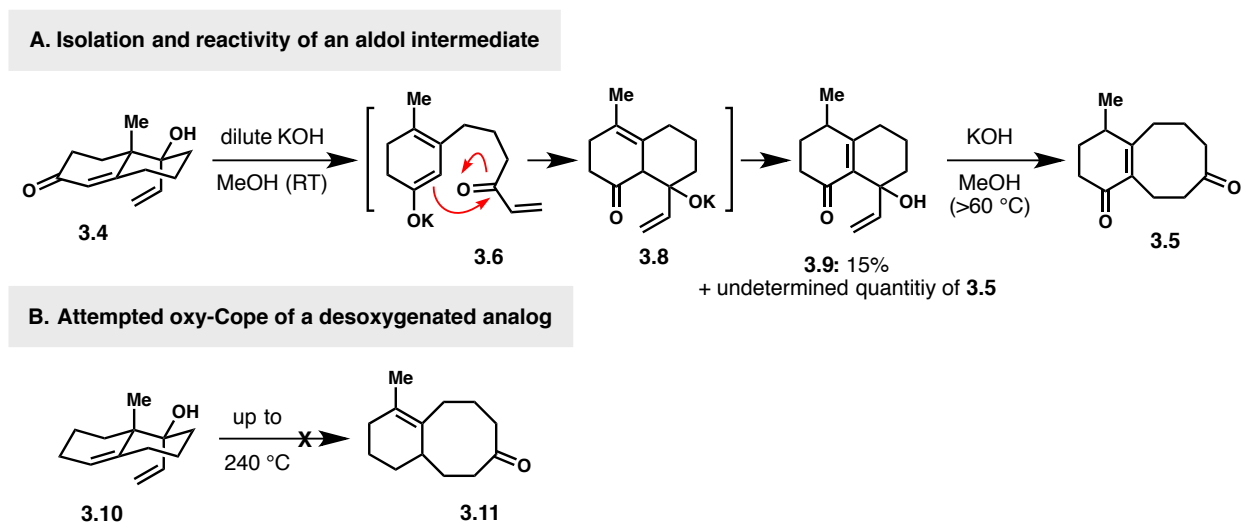


Significant effort was put into probing whether the mechanism of this transformation proceeded via Scheme 3.2A or 3.2B.<sup>5</sup> Three initial results attest to favoring the retro-aldol mechanism (Scheme 3.3). First, the alcohol epimer of 3.4 in basic methanol still affords 3.5. Second, running the reaction more dilute at a lower temperature with less base enabled the isolation of 3.9, which arises from the vinylogous retro-aldol/aldol addition/isomerization reaction of 3.4. Upon heating to >60 °C 3.9 cleanly converted to 3.5. Finally, 3.10, which has the

enone replaced with an electron neutral alkene is thermally stable up to 240 °C in the absence of base. Contrarily, enone **3.4** provides **3.5** when refluxed in ethylene glycol.<sup>9,10</sup>

Although these experiments may imply the relevance of the vinylogous retro-aldol mechanism, they are not completely conclusive. First, the equatorial vinyl-substituted **3.4** may operate under a different mechanism than the axial vinyl-substituted **3.4**. Second, under the more mild reaction conditions in which **3.9** was isolated, there is no mention of the quantities of **3.5** obtained. This lack of information severely undermines the formation of **3.9** as mechanistic evidence since intermediate **3.9** does not afford **3.5** until heated to 60 °C. Thus, it is not clear how **3.5** is originally formed. Third, no experiments on exposing **3.10** to the standard conditions of base at elevated temperatures were reported. The absence of this experiment makes it difficult to gauge the influence of anionic rate acceleration.

**Scheme 3.3** Initial mechanistic implications for a vinylogous retro-aldol mechanism.



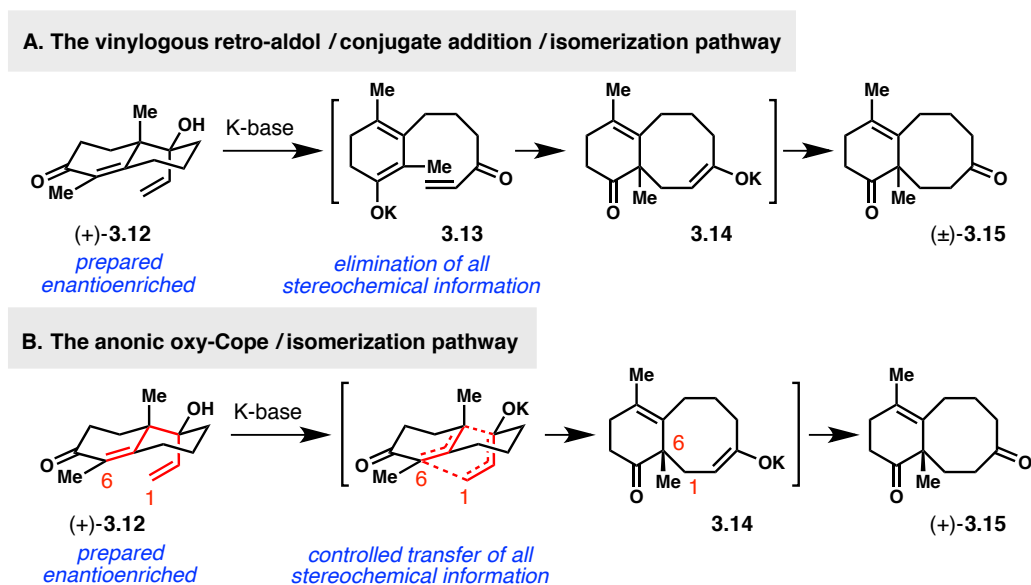
Not until 22 years after the initial disclosure was more light shed on the mechanism in an elegantly designed set of experiments.<sup>11</sup> To distinguish between the two mechanisms, the retention (or erosion) of enantioenrichment in **3.12** was evaluated (Scheme 3.4). If the retro-aldol

mechanism were operable, any enantioenrichment of **3.12** would be erased as the intermediate dienolate **3.13** is achiral. If, however, the oxy-Cope mechanism were operable, then the enantiopurity of the starting material would be transferred to the ring-expanded product. The last enone isomerization step needs to be eliminated to unequivocally use optical purity as a mechanistic probe. This was accomplished by using the tetra-substituted enone **3.12**.

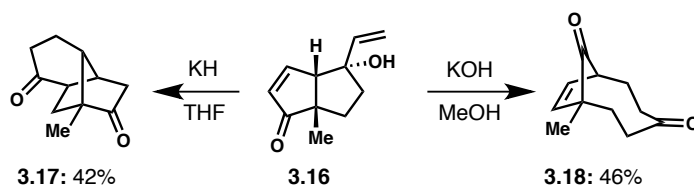
When (+)-**3.12** was exposed to basic methanol, the isolated product **3.15** had no optical rotation. This result implies the reaction operates via a vinylogous retro-aldol fragmentation. Interestingly, when (+)-**3.12** was reacted with KH in THF at room temperature, product **3.15** did have significant optical rotation, implying a concerted oxy-Cope mechanism!<sup>12</sup> These results suggest that the mechanism is dependent upon the reaction conditions. Disappointingly, a control experiment ensuring that (+)-**3.15** does not racemize under standard methoxide conditions was not mentioned. To the best of the presented evidence the rearrangement of **3.4** with basic methanol follows a fragmentation/recombination pathway.

Swaminathan and coworkers are not the only group to have observed a difference in reaction progression between aprotic and protic rearrangement conditions of oxy-Cope systems. Heathcock<sup>13</sup> observed an interesting divergence of reactivity when bicycle **3.26** was treated with either KH or KOH in methanol (Scheme 3.5).

**Scheme 3.4** Determining the mechanism for the rearrangement of **3.4** with the optically active tetrasubstituted alkene surrogate (+)-**3.12**.



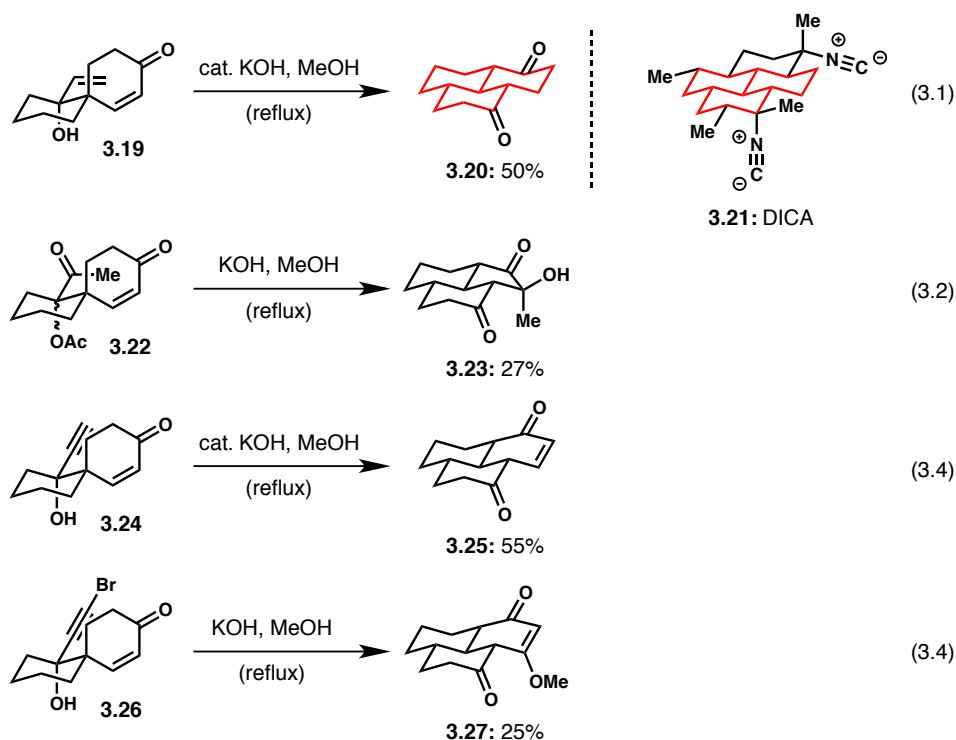
**Scheme 3.5** Heathcock's alternative reaction mechanisms depending on reaction conditions.



### 3.2 Literature Precedent of an oxy-Cope Rearrangement with Relevance To 7,20-Diisocyanoadociane

Among the explored base-induced rearrangements related to **3.4** to **3.5**, one directly addresses our interest in synthetic work towards the marine isocyanoterpene 7,20-diisocyanoadociane (DICA, **3.21**). In the presence of base, spirocycle **3.19** was reported to yield the perhydropyrene scaffold **3.20** (Equation 3.1).<sup>14</sup> The proposed all-*trans* stereochemistry and location of functional groups map directly onto the core structure of DICA.

This transformation was proposed to operate by a retro-vinylogous aldol/double Michael addition, as this fragmentation pathway appears to be related to mechanistic study on the system **3.4**. No experimental evidence for either mechanism was reported. The rearrangement of several other similar spirocyclic structures have also been disclosed (Equations 3.2-3.4).<sup>14-16</sup> The reaction of acetate **3.22** with base most reasonably caused deprotection followed by a retro-vinylogous aldol and recombination to **3.23**. Consistent with the conversion of **3.19** to **3.20**, alkyne **3.24** rearranges to enone **3.25**. Bromoalkyne **3.26** also proceeds similarly to **3.24**, but undergoes halogen displacement with methoxide to provide vinylogous ester **3.27**.

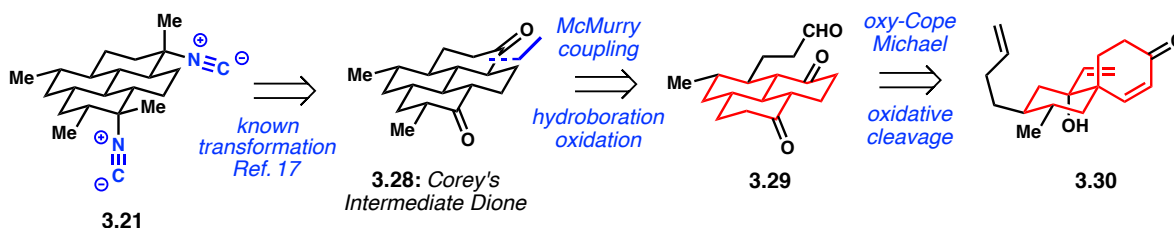


With this inspiration, a synthetic approach to DICA was devised (Scheme 3.6); a summary of the work presented in this chapter was published.<sup>17</sup> Corey's intermediate dione **3.28** was initially targeted.<sup>18</sup> The last ring would be annulated via a McMurry coupling and a hydroboration/oxidation sequence, leading back to the oxy-Cope rearrangement product, perhydrophenalene **3.29**. The spirocyclic substrate **3.30** bears an additional methyl and butenyl



chain required for further elaboration. Spirocycle **3.30** could be constructed in a similar fashion to the literature precedent from Swaminathan.<sup>14,19,20</sup>

**Scheme 3.6** Retrosynthetic analysis of DICA back to a spiro-fused oxy-Cope system.



### 3.3 Synthetic Efforts to Realize the oxy-Cope/Michael Cascade Towards 7,20-Diisocyanoadociane

Application of the oxy-Cope/Michael reaction towards DICA was initiated because its application would contribute a vastly lower step count than previous syntheses and would provide an interesting and strategically different disconnection. Syntheses of polycyclic molecules, and the previous approaches to DICA specifically,<sup>18,21,22</sup> typically progress via the sequential annulation of rings. The oxy-Cope/Michael cascade generates three rings in a single reaction from a spirocycle without conservation of any original rings: a unique and unusual feature in synthetic planning.

**Table 3.1** Evaluation of copper mediated conjugate additions into **3.31**.

Entry	Conditions	Observations Yield	<i>dr</i> <sup>a</sup>	X = Ligand Addition
1	CuI, CuCN or CuBr•DMS w/ & w/out TMSCl or HMPA in THF or Et <sub>2</sub> O	Various temperatures (-78 °C, -60 °C, 0 °C, RT)	competitive 1,2 addition, low conversions	
2	cat. CuBr•DMS, BF <sub>3</sub> •OEt <sub>2</sub>	(78 °C → RT)	67%	6 : 1
3	(butenyl) <sub>2</sub> CuMgBr, BF <sub>3</sub> •OEt <sub>2</sub>	(-78 °C)	–	7 : 1
4	(butenyl)Cu, BF <sub>3</sub> •OEt <sub>2</sub>	(-78 °C)	–	5 : 1
5	(COD)CuBr•(butenylMgBr)	(-78 °C)	–	12 : 1
	CuBr•DMS, X–Li <sup>b</sup> then butenylMgBr <sup>c</sup>	(-78 °C)		
6	PhO–Cu•(butenylMgBr)		65%	>20 : 1
7	t-BuO–Cu•(butenylMgBr)		–	19 : 1
8	t-Bu–Cu•(butenylMgBr)		–	19 : 1
9	TMSM–Cu•(butenylMgBr)		–	>20 : 1
10	Mes–Cu•(butenylMgBr)		–	>20 : 1
11	(butenyl) <sub>2</sub> CuMgBr	(-78 °C)	–	19 : 1

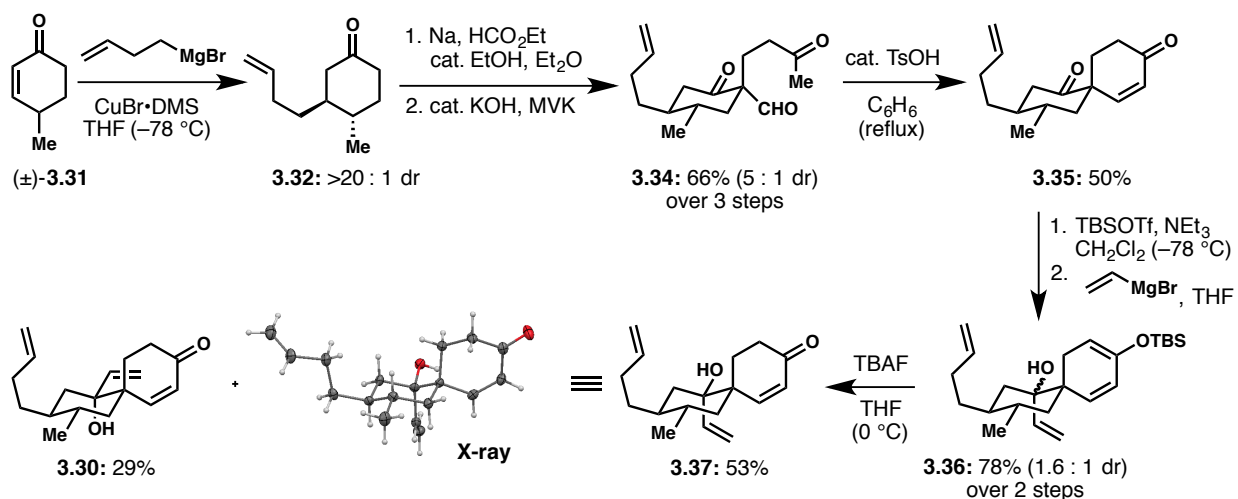
<sup>a</sup> *dr* was determined by NMR or GC-FID <sup>b</sup> at 0 °C <sup>c</sup> at -78 °C

The synthesis of spirocycle **3.30** commenced with a conjugate addition to enone **3.31**. Although the reaction should have been a straightforward transformation to **3.32**, optimization to avoid 1,2-addition and to improve *trans* diastereoselectivity was required (Table 3.1). Of the copper salts screened, CuBr•DMS enabled the cleanest formation of alkylcopper reagents, as determined by homogeneity of the reaction mixture. Lewis acids, such as TMSCl and BF<sub>3</sub>•OEt<sub>2</sub> did not have a beneficial effect and led to relatively low diastereoselectivity (entries 1-4). The copper source (COD)CuBr<sup>23,24</sup> was evaluated for being a stable copper source that requires only one equivalent of nucleophile. While it afforded an improved 12:1 *dr*, the stoichiometric COD made this copper source unappealing. Mixed cuprates,<sup>25</sup> such as Posner's heteroatom cuprates,<sup>26,27</sup> Bertz's TMSM-cuprate<sup>28</sup> and Saegusa's Mes-cuprate<sup>29</sup> afforded excellent *trans* diastereoselectivity. In situ-generated copper phenoxide afforded >20:1 *dr*; however, isolation

was complicated by the difficult separation of stoichiometric phenol (entry 6). All mixed cuprates required the pregeneration of an active copper species, an inconvenience in reaction setup. Additionally, *t*-Bu and Mes transferred competitively with the desired butenyl group (entries 8, 10). Although the TMSM-cuprate was found to provide superior diastereoselectivity and reactivity, its expense mitigated further implementation (entry 9). Ultimately the most user-friendly and reliable conditions were using the basic Gilman cuprate, generated from two equivalents of butenylmagnesium bromide, one equivalent CuBr•DMS and without Lewis acid. These simple conditions afforded excellent diastereoselectivity, as well as selectivity for conjugate addition over 1,2-carbonyl addition, and provided material of sufficient quality to continue without purification.

With practical conjugate addition conditions available, the spirocyclic Robinson annulation was performed to afford **3.35** as a single diastereomer (Scheme 3.7). Silylation of enone **3.35** ensured vinylmagnesium bromide addition occurred solely into the ketone, affording **3.36** as a 1.6 : 1 mixture of equatorial and axial alcohols. Desilylation of **3.36** with TBAF generated the separable enones **3.37** and **3.30**. Structure determination was accomplished by X-ray analysis of **3.37**.

**Scheme 3.7** Generation of spirocyclic hydroxyenones **3.30** and **3.37**.

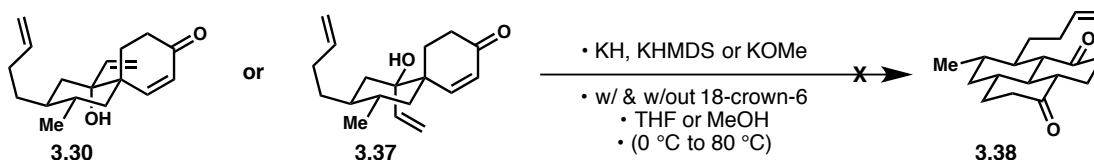


The desired anionic oxy-Cope/Michael reactivity was explored using both diastereomeric enones **3.30** and **3.37** (Scheme 3.8). To our surprise and disappointment, exposing either alcohol diastereomer to the standard conditions of KOH in methanol afforded an intractable mixture of compounds. Further evaluation via kationation did not lead to any success (Scheme 3.8A). Metal-mediated reactions using Pd(II)<sup>30</sup> or Hg(II) also proved unsuccessful (Scheme 3.8B).<sup>31,32</sup>

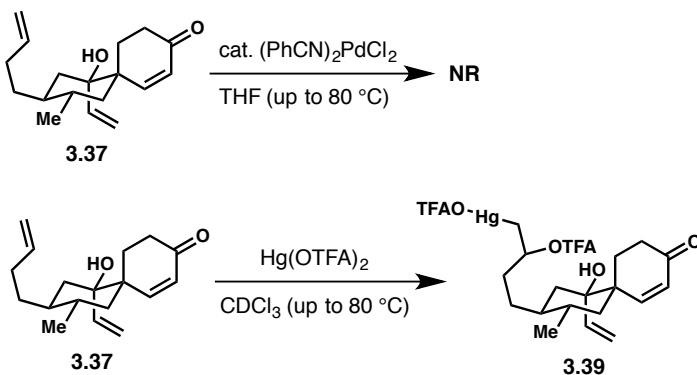
Although the anionic version of the oxy-Cope is the standard rearrangement of 3-oxy-1,5-hexadiene systems, a neutral variant is also available. In contrast to the experiments under basic conditions, simply heating diene **3.30** in an inert solvent led to a clean reaction; however, instead of isolating the expected oxy-Cope product [7.3.1]-bicycle **3.40**, tricycle **3.41** was the only observed product (Scheme 3.8C). Although **3.41** is not the desired compound, this experiment validated that the oxy-Cope rearrangement of **3.30** can occur! Not unsurprisingly, upon oxy-Cope and enol tautomerization, the cyclodecenone **3.40** is well positioned to undergo a transannular carbonyl-ene reaction to afford **3.41** as a single diastereomer.

**Scheme 3.8** Attempting an oxy-Cope/Michael cascade.

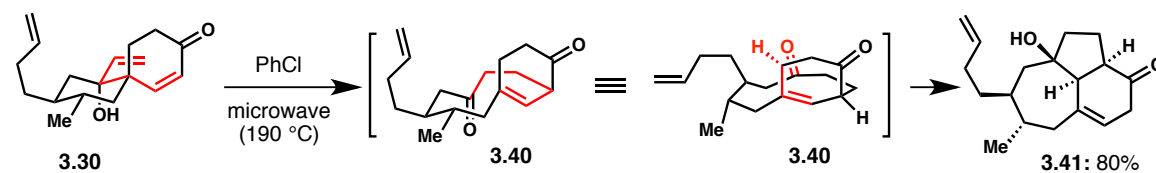
**A. Desired oxy-Cope / Michael cascade**



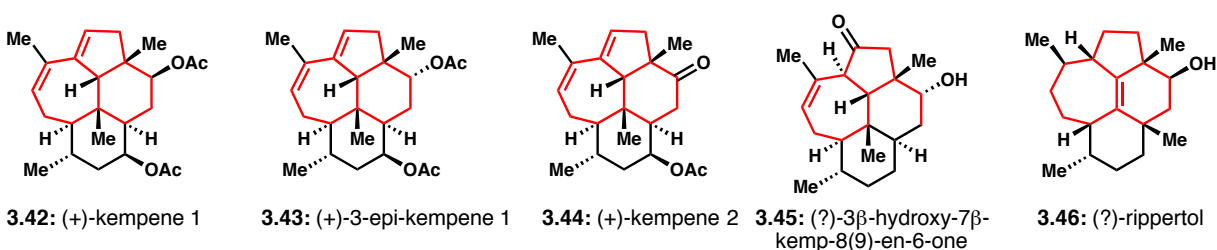
**B. Metal-catalyzed attempts at a rearrangement**



**C. Thermal rearrangement of oxy-Cope system 3.30**



**Figure 3.1** Kempene and rippertane structures.

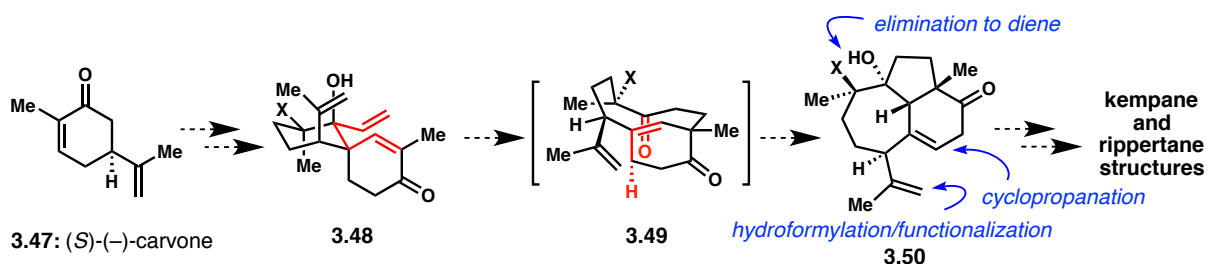


The tricyclic ring system of **3.41** is reminiscent of the kempene and rippertane core (Figure 3.1) and an approach using this tandem thermal reaction could be imagined.<sup>33,34</sup>

Currently no efforts using an oxy-Cope/carbonyl-ene reaction towards these molecules have been reported, though other natural products have been prepared using this tandem methodology.<sup>35,36</sup> The synthesis of these natural products still holds exciting challenges.

Starting from (*S*)-(-)-carvone, **3.47**, Li/NH<sub>3</sub> reduction of the enone and capture of the resultant enolate with a group capable of reforming the alkene (SPh, SePh, etc.) followed by spirocyclic Robinson annulation—MVK would arrive *trans* to the isopropenyl—could generate the ketone precursor to **3.48**. The sequence of enone silylation, Grignard reagent addition and deprotection would generate decorated spirocyclic enone **3.48**. Exposing this substrate to thermal conditions could trigger the desired oxy-Cope/carbonyl-ene rearrangement to the core kempane structure (Scheme 3.9). The 1,5-diene of **3.48** highlighted in red would selectively undergo oxy-Cope rearrangement and upon carbonyl-ene, generate a tertiary alcohol for elimination. The isopropenyl substituent could be elaborated via hydroformylation to address annulation of the southern ring and a cyclopropanation/fragmentation to introduce the remaining methyl group.

**Scheme 3.9** An envisioned oxy-Cope/carbonyl-ene approach to kempane natural products.

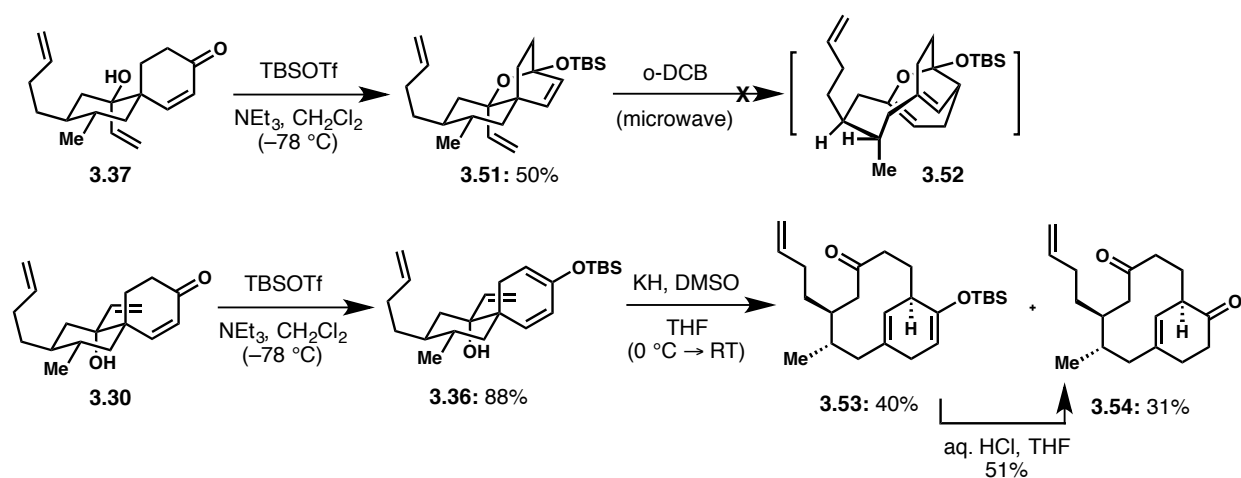


Although the anionic oxy-Cope transformation of enones **3.30** and **3.37** had failed, the success of a thermal oxy-Cope encouraged a sustained effort. In this vein, masking the enone functionality was evaluated next (Scheme 3.10). Attempted enoxysilane formation of **3.37** with TBSOTf led to [2.2.2]-oxabicyclic **3.49**. Oxy-Cope rearrangement of this molecule was

unsuccessful, most likely because of the strain and frailty of **3.50** generated upon [3,3]-rearrangement, so silylation of the other allylic alcohol diastereomer was pursued.

Treating enone **3.30** with TBSOTf smoothly afforded siloxydiene **3.36**. Deprotonating the alcohol with KH and DMSO at 0 °C and warming the reaction to room temperature led to formation of a single product by TLC analysis. Upon quenching with ethanol the two products **3.53** and **3.54** were isolated. Both compounds were generated via [3,3]-rearrangement, and could be funneled to the skipped enone **3.54** upon deprotecting enoxysilane **3.53** with acid.

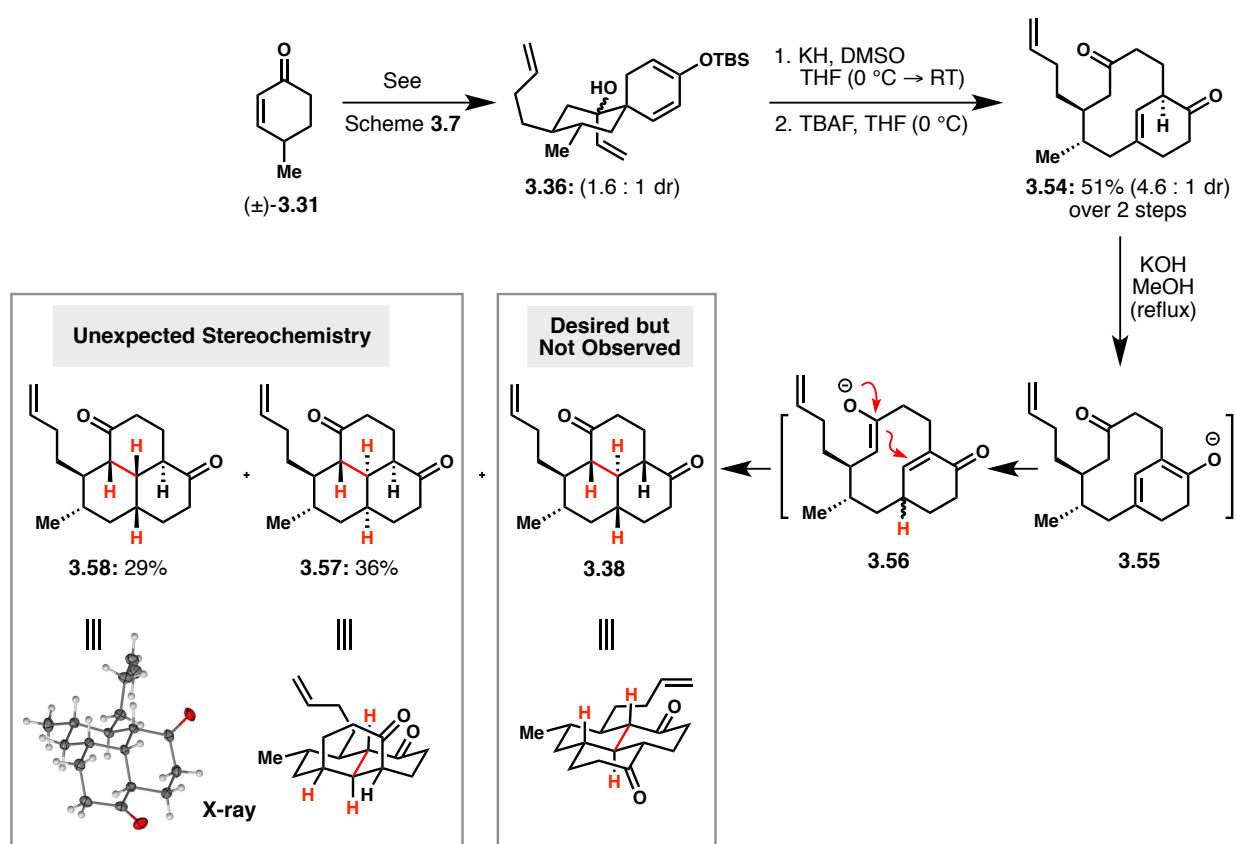
**Scheme 3.10** Silylation of spirocyclic enones in efforts towards an oxy-Cope rearrangement.



With a basic synthetic outline to obtain oxy-Cope product **3.54** established, a more streamlined preparation was sought. As the synthesis outlined in Scheme 3.7 and 3.10 proceeds through enoxysilane **3.36** as a mixture of alcohol diastereomers, the anionic oxy-Cope could be attempted at this stage. Two outcomes were envisioned: 1. only the axial-alcohol diastereomer undergoes rearrangement or 2. both diastereomers could rearrange leading to better material throughput. Gratifyingly, when **3.36** was deprotonated with KH and warmed to room temperature, a single compound was obtained; both diastereomers of **3.36** cleanly underwent ring

expansion (Scheme 3.11). Treating the crude cyclohexadiene with cold TBAF afforded **3.54**. Although the sigmatropic rearrangement is high yielding, the relatively low isolated yield of **3.54** is due to decomposition during desilylation. Still, this consolidated approach allows for a rapid build up of material when compared to the previous sequence.

**Scheme 3.11** Improved synthesis of [7.3.1]-bicycle **3.54** via an anionic oxy-Cope rearrangement and subsequent transannular Michael reaction.



With the oxy-Cope ring expansion reaction completed, transannular cyclization of bicycle **3.54** was examined. Treating **3.54** with hot, basic methanol afforded two products. The desired pathway would isomerize the skipped enone into conjugation to **3.55**, and be followed by *trans*-selective cyclization to **3.38** as preceded by Swaminathan (Equation 3.1).<sup>14</sup> Instead, the two products formed arose from indiscriminant protonation of the extended enolate **3.55** at the



gamma position, followed by a *cis*-selective cyclization to generate **3.57** and **3.58**. No desired *trans* ring-closure was observed. This disappointing result brought into question the original literature report.

### 3.4 A Literature Reevaluation of the Transannular Michael Stereochemistry

The original Swaminathan report of the oxy-Cope/Michael cascade of **3.19** reported the product perhydrophenalene to be in the all-*trans* configuration (Equation 3.1, Figure 3.2).<sup>14</sup> When this reaction sequence was adapted towards DICA, the transannular Michael reaction provided two products via a *cis*-selective ring closure (Scheme 3.11). These two contradicting results and the lack of rigorous stereochemical determination originally performed by Swaminathan (Figure 3.2), made a reinvestigation prudent.

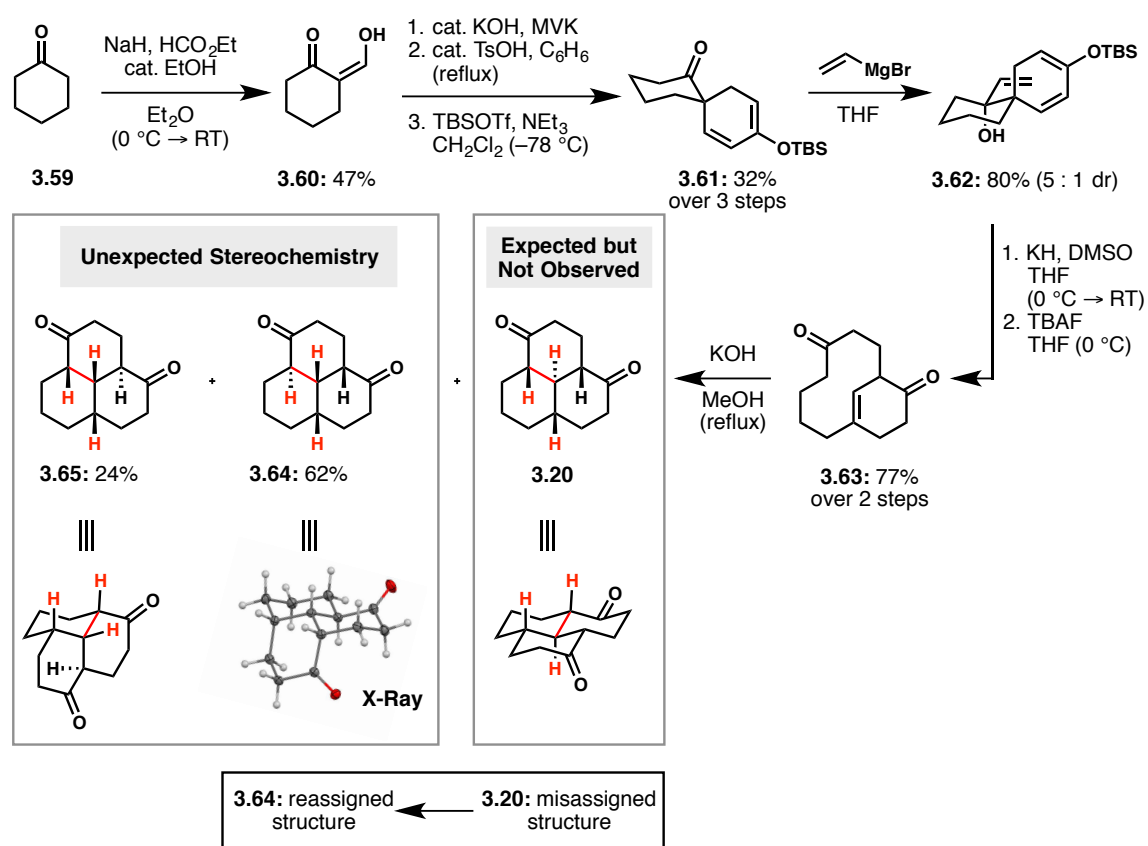
**Figure 3.2** Stereochemical determination of oxy-Cope/Michael reaction product by Swaminathan. “Dione 6” refers to structure **3.20**.<sup>14</sup>

Since the rearrangement is being carried out under equilibrating conditions, the dione **6** may be expected to take up the most stable conformation *viz.* the all *trans* conformation **6a**.

A synthetic sequence similar to the approach for DICA was initiated for the simple unsubstituted literature system (Scheme 3.12). Spirocyclic Robinson annulation on cyclohexanone **3.59** and silylation afforded silyloxydiene **3.61**. Vinylmagnesium bromide addition followed by deprotonation with KH and deprotection with TBAF generated the ring expanded **3.63**. Treatment of **3.63** with potassium hydroxide in methanol triggered the transannular Michael addition and afforded a major compound with identical <sup>1</sup>H and <sup>13</sup>C NMR

data as reported by Swaminathan<sup>14</sup> who assigned structure **3.20**. X-Ray crystallography was chosen as the most rigorous means of structural determination. Instead of the expected all-*trans* **3.20**, analysis revealed the stereochemistry was **3.64**. The structure reported by Swaminathan was thus misassigned and was thereby corrected to **3.64**, a compound obtained from a *cis*-selective Michael addition (the bond highlight in red). A second compound (**3.65**) was isolated as a minor component and clearly arises via epimerization of the  $\alpha$ -carbons. The structural reassignment of **3.20** to **3.64** requires that caution be exercised with the other reported structures obtained via such cascades (Equations 3.2-3.4).<sup>14-16</sup>

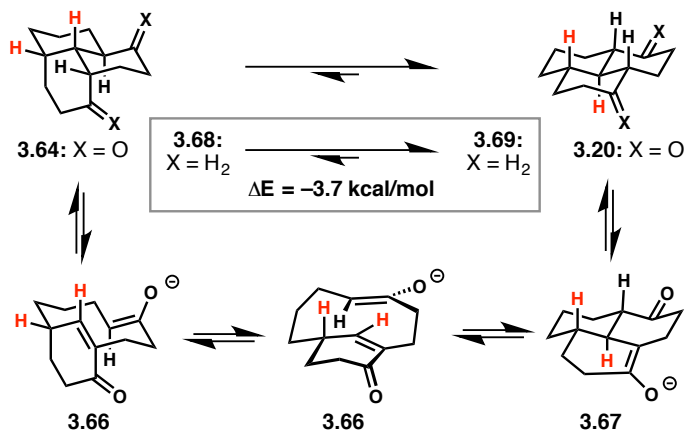
**Scheme 3.12** Reevaluation of the original oxy-Cope/Michael cascade.



### 3.5 Evaluating the Reversibility of the Transannular Michael Reaction

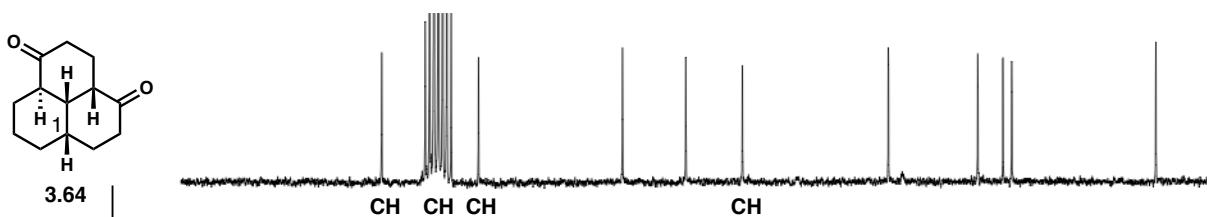
Swaminathan's assertion that the oxy-Cope/Michael reaction of **3.19** should provide a perhydrophenalendione in an all-*trans* relationship because of the equilibrating conditions, even though incorrect, is still interesting (Figure 3.2). The reaction of **3.63** with basic methanol to afford **3.64** and **3.65** could simply be the outcome of a kinetic conjugate addition. Under thermodynamic control, the ring closure would be reversible and therefore could still equilibrate to the more stable all-*trans* **3.20** (Scheme 3.13). Computations have already established the all-*trans*, all-chair conformation of unsubstituted perhydrophenalene **3.69** to be most thermodynamically stable (Scheme 3.13).<sup>37</sup> The known reversibility of Michael reactions lends credence to the possible equilibration of **3.64** to **3.20**.<sup>38,39</sup> Disappointingly, exposing **3.64** to higher temperatures in methanol or basic ethylene glycol did not lead to any change in product distributions. Also, if the retro-Michael were to have occurred, at elevated temperatures, a carbonyl-ene product similar to **3.41** could also be imagined; however, such a product was never observed.

**Scheme 3.13** Proposed pathway for the equilibration of a perhydrophenalenedione system via a retro-Michael/transannular Michael sequence and calculations of perhydrophenalene stability.<sup>37</sup>

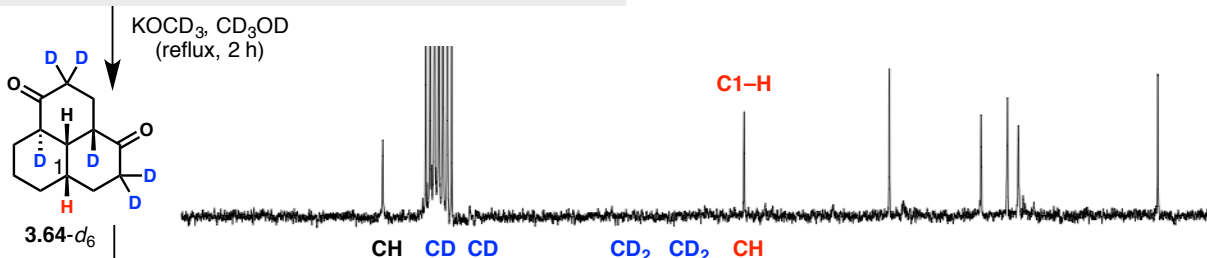


**Figure 3.3** NMR study in CD<sub>3</sub>OD to determine the reversibility of the Michael reaction.

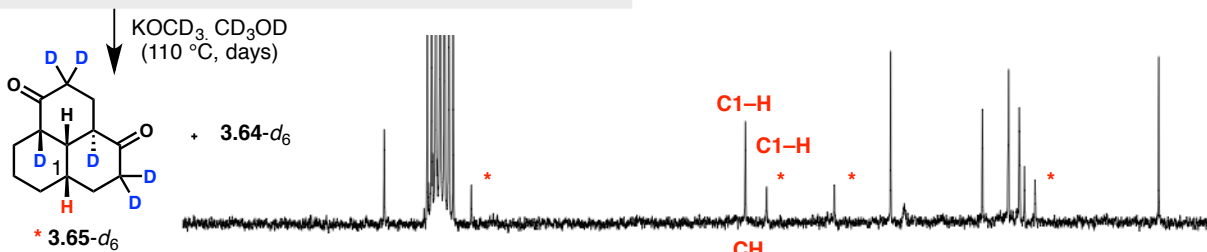
**A. Tricycle 3.64; <sup>13</sup>C NMR, CD<sub>3</sub>OD δ 49.15**



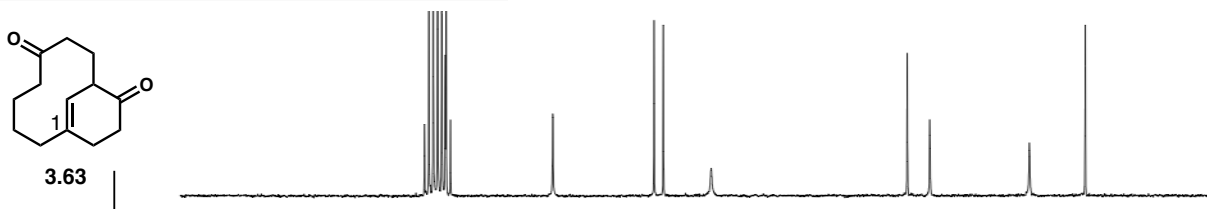
**B. Standard reaction conditions; <sup>13</sup>C NMR, CD<sub>3</sub>OD δ 49.15**



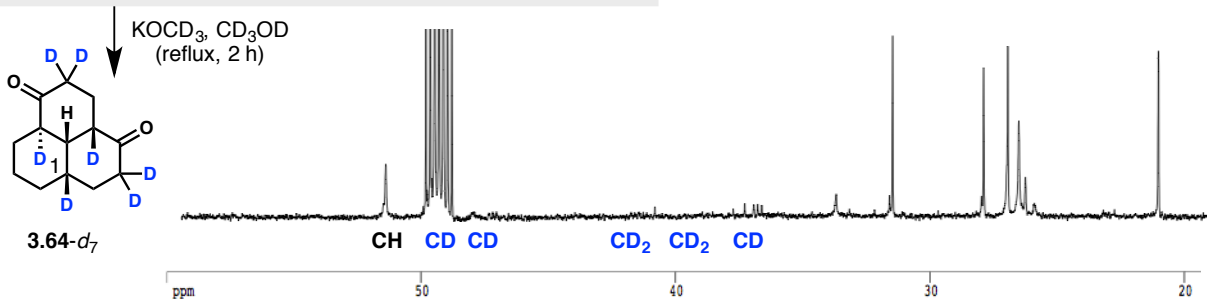
**C. Extended reaction conditions; <sup>13</sup>C NMR, CD<sub>3</sub>OD δ 49.15**



**D. [7.3.1] Bicycle 3.63; <sup>13</sup>C NMR, CD<sub>3</sub>OD δ 49.15**



**E. Assignment of the C1 resonance; <sup>13</sup>C NMR, CD<sub>3</sub>OD δ 49.15**



Because the equilibration of **3.64** to **3.20** relies on a retro-Michael reaction, a detailed investigation into the feasibility of the retro-transannular Michael reaction was initiated to better understand the failed equilibration. Two possibilities exist for the observed lack of equilibration: 1. the retro-Michael reaction may not be occurring under the experimental conditions or 2. the retro-Michael may be facile, but the transannular ring closure is only *cis*-selective. Deuterium labeling studies were devised to distinguish between the two possibilities.

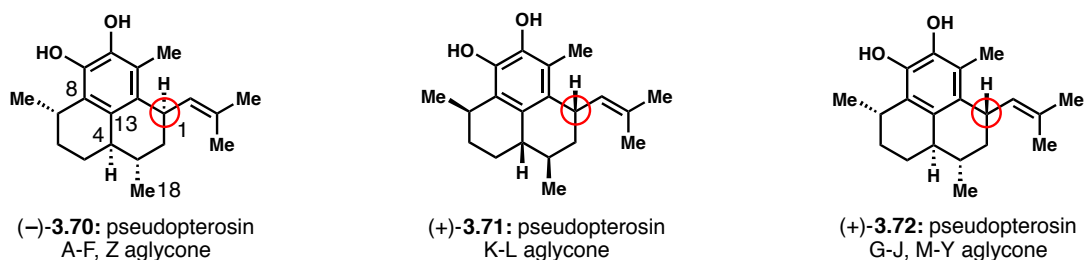
Tracking the incorporation of deuterium provided insight into the reversibility of the transannular Michael reaction (Figure 3.3). Treating **3.64** with KOCD<sub>3</sub>/CD<sub>3</sub>OD under typical cyclization conditions incorporated deuterium at all enolizable positions to afford **3.64-d<sub>6</sub>** (Figure 3.3A-B). No deuterium incorporation was observed at the gamma-position, C1.<sup>40</sup> Extended heating of **3.64-d<sub>6</sub>** afforded only a mixture of **3.64-d<sub>6</sub>** and **3.65-d<sub>6</sub>**; still, no incorporation of deuterium at C1 was observed (Figure 3.3B-C). To assign the residual methine carbons  $\delta$  51.4 and 37.1, the cyclodecenone **3.63** was closed in CD<sub>3</sub>OD (Figure 3.3D-E). The C1 position was deuterated upon enone isomerization to enable the cyclization to **3.64-d<sub>7</sub>**. The peak at  $\delta$  37.1 was assigned to C1 as  $\delta$  51.4 remained protiated. Based on these reactivity studies, the transannular Michael reaction, which is necessary for conversion of **3.64** to **3.20** is irreversible in basic methanol (Scheme 3.13).

### 3.6 Outlook for the oxy-Cope/Michael Reaction in Natural Product Synthesis

Swaminathan's oxy-Cope/Michael cascade proceeded smoothly to construct the perhydrophenalene C–C bonds; however, in the context of DICA, not with the desired stereochemical outcome. The ease of starting material synthesis and rapid construction of molecular complexity makes the oxy-Cope/Michael reaction a powerful method of polycyclic

assembly. The attempted implementation of this strategy towards DICA supported this point, but because of a stereochemical misassignment, DICA ended up as the wrong target for this reaction.

**Figure 3.4** Pseudopterodin aglycones.



Although disappointing that the designed strategy towards DICA did not bear fruit, Swaminathan's transformation to the perhydrophenalene scaffold could still be useful for natural product synthesis. Of the phenalene containing natural products, the pseudopterodins may be most readily accessed (Figure 3.4).<sup>41-43</sup> This class of molecules consists of numerous members with variation in glycosylation pattern of a substituted phenalene core. The cores can be divided into one set of enantiomers, (-)-**3.70** and (+)-**3.71**, and into the side chain epimer, (+)-**3.72**. The pseudopterodin aglycones have been prepared synthetically on 15 occasions, with the latest by Sherburn in 2015.<sup>44,45</sup>



ring, directly into the seven-carbon tether. The stereocenters generated at C8, C12 and C13<sup>46</sup> are inconsequential, as the cyclohexenone ring would be immediately oxidized to phenol **3.79**, an intermediate prepared previously.<sup>43,47-49</sup> The remaining functionalizations include an aromatic hydroxylation, followed by the installation of the side chain. Implementation of Fu's enantioselective cross-coupling technology<sup>50,51</sup> could enable the selective synthesis of all pseudopterosin aglycones (-)-**3.70**, (+)-**3.71** and (+)-**3.72**, *a feat not achieved by any of the 15 currently published total synthesis strategies.*<sup>45</sup>

Use of the oxy-Cope/Michael rearrangement towards the pseudopterosins would contribute an entirely new synthetic design for a popular target. All approaches, except for the recent Sherburn synthesis, rely either on a terpene or aromatic starting material as a scaffold to construct subsequent rings: termed "structure-goal strategy."<sup>44</sup> An oxy-Cope/Michael strategy originating from a spirocycle starting material would be the only synthesis to construct all three rings simultaneously and also deviates from the previously relied on "structure-goal strategy". It would also reprise a strategy that was unsuccessful in the context of DICA to another natural product system.

### **3.7 Conclusions**

The oxy-Cope/transannular Michael cascade of spirocyclic enones was uncovered in the literature and evaluated in the interest of a synthesis of DICA. The course of the reaction generated the desired carbocyclic framework required for the proposed synthesis, but upon examining the products in detail, an undesired *cis* ring-fusion was observed. This turned our attention to reevaluating the literature report's proposed claims. A structural misassignment of the literature compound was uncovered by X-ray analysis. As previously noticed in an



application towards DICA, the transannular Michael undergoes only a *cis*-selective addition. Further deuterium studies confirmed a suspicion that the conjugate addition is irreversible in refluxing basic methanol, counter to Swaminathan's assertions. Although the oxy-Cope/Michael reaction strategy towards DICA was unsuccessful, the reaction could still have a bright future in natural product synthesis; accessing the pseudopterosins is one example.

### 3.8 Experimental Procedures

#### Purifications –

Solvents: Dry tetrahydrofuran (THF), diethyl ether (Et<sub>2</sub>O), dichloromethane (CH<sub>2</sub>Cl<sub>2</sub>), toluene, benzene (C<sub>6</sub>H<sub>6</sub>), dimethylformamide (DMF), methanol (MeOH) were obtained by passing commercially available formulations through activated alumina columns

Amines: Triethylamine (NEt<sub>3</sub>) was purified by distillation from CaH<sub>2</sub>.

Triflates: *tert*-Butyldimethylsilyl trifluoromethanesulfonate (TBSOTf) was purified by distillation over CaH<sub>2</sub>.

Metals: Copper(I) iodide (CuI) was purified by Soxhlet extraction with THF followed by drying of the solid under high vacuum. (COD)CuBr was prepared according to the literature.<sup>24</sup>

Miscellaneous: Methyl vinyl ketone (MVK) was purified by distillation. BF<sub>3</sub>•OEt<sub>2</sub> was purified by distillation from CaH<sub>2</sub>. KOCD<sub>3</sub> was prepared by addition of CD<sub>3</sub>OD to a suspension of KH in toluene, removing all volatiles in vacuo and washing the remaining solid with pentane.

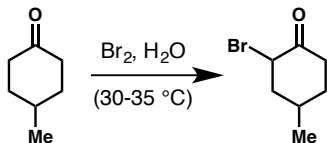
**Titration** – Alkyl lithium reagents were titrated using 2,6-di-(*tert*-butyl)-4-methylphenol (BHT) as the sacrificial proton source and fluorene as an indicator in THF or using diphenylacetic acid in THF. Grignard reagents were titrated using salicylaldehyde phenylhydrazone in THF.<sup>52</sup>

**Reaction Setup** – All reactions were carried out under an argon atmosphere with dry solvents under anhydrous conditions, unless otherwise noted. Argon balloons were the sole inert atmosphere used. Reactions run at an ambient temperature of 20–25 °C are designated as room

temperature. Microwave reactions were performed in an Anton Paar Microwave. Yields refer to chromatographically and spectroscopically homogeneous materials, unless otherwise stated.

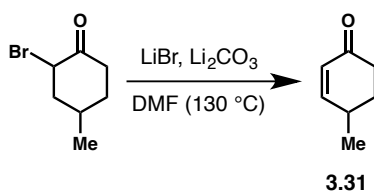
**Analysis** – Thin layer chromatography was performed on 0.25 mm EMD glass-backed TLC plates impregnated with a fluorescent dye and visualized with UV light and  $\text{KMnO}_4$  in  $\text{K}_2\text{CO}_3/\text{NaOH}/\text{water}$  or *p*-anisaldehyde in ethanol/aqueous  $\text{H}_2\text{SO}_4/\text{AcOH}$  and heat as a developing agent. Forced flow (flash) chromatography was performed on EMD Silica 60, mesh 0.04-0.063 silica gel. NMR spectra were recorded on Bruker 500 MHz instrument, obtained at 298 K unless otherwise noted and calibrated to residual undeuterated solvent as an internal reference. Chemical shifts are reported in ppm with the following abbreviations to explain multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, quin = quintuplet, sext = setet, sep = septet, bs = broad signal, m = multiplet. All coupling constants are apparent *J* values measured at the indicated field strengths. FT-IR spectra were recorded on a Perkin-Elmer spectrum RX1 spectrometer. High-resolution mass spectra (HRMS) were recorded on a H2Os LCT Premier spectrometer using ESI-TOF (electrospray ionization-time of flight). Melting points were measured on a MEL-TEMP II capillary apparatus and stand uncorrected.

## 2-Bromo-4-methyl cyclohexanone [Adapted from the literature]<sup>53</sup>



A 250 mL round bottom flask was charged with 20.0 g (0.178 mol) 4-methylcyclohexanone and 60 mL water and placed in an empty crystallizing dish. The rapidly stirring solution was treated dropwise with 8.4 mL (0.164 mol) bromine over the course of 30 minutes, keeping an internal reaction temperature between 30-35 °C by adding ice to the crystallizing dish. After stirring for an additional 1.5 hours the reaction turned colorless, then diluted with 20 mL sat. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and 40 mL water, partitioned with 200 mL EtOAc and the layers separated. The organic phase was washed twice with 50 mL sat. NaHCO<sub>3</sub> and 50 mL brine. All aqueous layers were collected and back extracted with 50 mL EtOAc. This organic layer was washed twice with 20 mL sat. NaHCO<sub>3</sub>, 20 mL brine and combined with the previously obtained organic phase, dried over MgSO<sub>4</sub>, filtered, all volatiles removed in vacuo and the crude material distilled (60-73 °C/0.6 mmHg) to afford 24.2 g (77%, 2:1 dr) 2-bromo-4-methyl cyclohexanone as a colorless liquid. The spectral data was identical to the literature.<sup>54</sup>

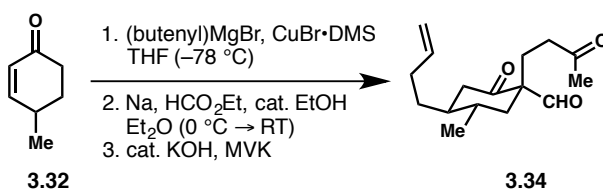
## 4-Methylcyclohex-2-en-1-one 3.31 [Adapted from the literature]<sup>55</sup>



A three neck 1 L flask was fitted with a mechanical stirrer, a rubber septum and a reflux condenser, and flushed with argon. The flask was charged with 24.1 g (0.127 mol) 2-bromo-4-methyl cyclohexanone, 250 mL DMF, 23.5 g (0.318 mol) Li<sub>2</sub>CO<sub>3</sub> and 27.8 g (0.320 mol) LiBr,

the septum replaced with a glass stopper and the reaction placed in an oil bath set to 130 °C. After 3 hours the reaction was cooled to room temperature, diluted with 250 mL EtOAc and filtered through Celite. The filter cake was washed four times with 250 mL EtOAc portions. Five separatory funnels were set up in tandem, the first loaded with the filtrate solution and the other four with 250 mL fresh EtOAc each. To the first funnel was added 1 L water and after vigorous shaking, the aqueous layer drained and poured into the second funnel. After vigorous shaking, the aqueous layer was drained and poured into the third funnel, shaken and continued with the fourth and fifth separatory funnel. This process was repeated three additional times with 750 and twice with 500 mL water portions. The first separatory funnel was washed with 150 mL brine, the other four with 50 mL brine, then all organic layers collected, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was distilled (48-52 °C/4.5 mmHg) to afford 8.62 g (62%, ~3% DMF contamination) **3.31** as a colorless liquid. The spectral data was identical to the literature.<sup>54</sup>

### Michael Adduct 3.34

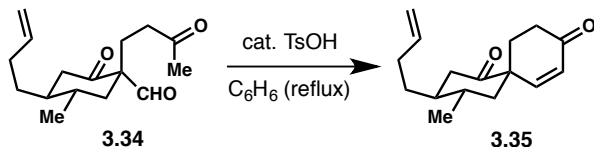


A 250 mL round bottom flask was charged with 70 mL (25.6 mmol) freshly prepared 0.37 M butenylmagnesium bromide/THF. The reaction was cooled to -78 °C then treated with 2.67 g (13.0 mmol) CuBr·DMS. The cuprate was aged 45 minutes before 1.19 g (10.8 mmol) **3.32** in 15 mL THF was added over the course of 15 minutes. After stirring at -78 °C for a total of 3 hours, the reaction was quenched with 10 mL sat. NH<sub>4</sub>Cl and filtered through Celite with the assistance

of 150 mL EtOAc. All volatiles were removed, the residue dissolved in 150 mL EtOAc, washed thrice with 50 mL portions 2:1 1 M NaOH/sat.  $\text{NH}_4\text{Cl}$  then 30 mL brine. The organic layer was dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. Crude ketone (>20:1 dr) was of sufficient purity to move forward. To an ice cooled 50 mL round bottom flask containing 20 mL  $\text{Et}_2\text{O}$  was added 300 mg (13 mol) Na metal cut and flattened, then crude ketone from above in 5 mL  $\text{Et}_2\text{O}$  and 1.3 mL (16.1 mmol) ethyl formate were added. The reaction was initiated with 0.1 mL EtOH and the ice bath allowed to melt over the course of 5 hours. The contents were stirred for 10 minutes after the addition of 6 mL 0.5 M NaOH, diluted with 24 mL 0.5 M NaOH and the layers separated. The aqueous layer was washed twice with 20 mL  $\text{Et}_2\text{O}$ . All organic layers were combined and back extracted twice with 20 mL 0.5 M NaOH. The aqueous phases were collected, acidified with 20 mL 6 M HCl then extracted with 40 mL and thrice with 20 mL  $\text{Et}_2\text{O}$ . These organic layers were combined, washed with 10 mL water, 10 mL half sat.  $\text{NaHCO}_3$ , 10 mL brine, then dried over  $\text{MgSO}_4$  and all volatiles removed in vacuo to afford a ketoaldehyde of sufficient purity to proceed. A 50 mL round bottom flask was charged with crude ketoaldehyde, 1.3 mL (16.1 mmol) methyl vinyl ketone and 33 mg (0.59 mmol) powdered KOH at room temperature (exotherm). After stirring for 30 minutes, the contents were diluted with 60 mL  $\text{Et}_2\text{O}$  and washed twice with 10 mL water then 10 mL brine. The organic layer was dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The residue was purified by column chromatography (5:1 hexanes/EtOAc) to afford 1.92 g (66% over 3 steps, 5.2:1 dr) **3.34** as a light yellow oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm) major  $\delta$  9.61 (s, 1H), 9.41 (s, 1H), 5.76 (dddd,  $J = 16.1, 11.9, 8.6, 4.1, 1\text{H}$ ), 5.03 (d,  $J = 17.1, 1\text{H}$ ), 4.99 (d,  $J = 12.2, 1\text{H}$ ), 2.62-2.38 (m, 2H), 2.28-2.12 (m, 3H), 2.10 (s, 3H), 2.06-1.85 (m, 3H), 1.76-1.69 (m, 2H), 1.62 (dd,  $J = 13.8, 3.5, 1\text{H}$ ), 1.53-1.45 (m, 1H), 1.43-1.24 (m, 1H), 1.03 (d,  $J = 6.4, 3\text{H}$ );  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)

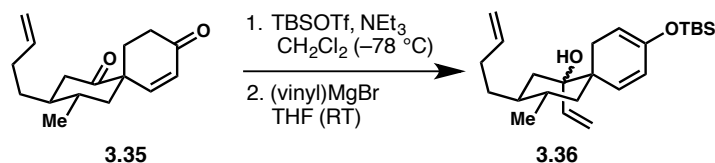
major  $\delta$  212.3, 207.3, 200.9, 137.9, 115.1, 62.3, 43.7, 43.5, 39.5, 38.1, 32.5, 30.9, 30.0, 29.9, 25.1, 18.8; IR (thin film) 2926, 2854, 2740, 1718, 1639  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{16}\text{H}_{24}\text{O}_3$   $[\text{M}+\text{Na}]^+$  287.1623 found 287.1621.

### Spiro Enone 3.35



An azeotropically dried solution of 70 mg (0.37 mmol) p-TsOH $\cdot$ H<sub>2</sub>O in 40 mL benzene was added to 1.90 g (7.19 mmol) **3.34** in a 100 mL round bottom flask and refluxed over a Hickman still. After 2 hours the reaction was cooled to room temperature, diluted with 50 mL Et<sub>2</sub>O, washed twice with 10 mL sat. NaHCO<sub>3</sub>, 10 mL water, 10 mL brine then dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (9:1 hexanes/EtOAc) then recrystallized from Et<sub>2</sub>O/hexanes to afford 902 mg (50%) **3.35** as fluffy white needles (mp = 42-44 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm)  $\delta$  6.77 (d, J = 10.3, 1H), 6.06 (d, J = 10.3, 1H), 5.79 (ddt, J = 16.9, 10.2, 6.7, 1H), 5.05 (d, J = 17.1, 1H), 4.99 (d, J = 10.2, 1H), 2.52-2.45 (m, 2H), 2.37-2.12 (m, 5H), 2.01-1.94 (m, 2H), 1.88-1.81 (m, 1H), 1.76 (ddt, J = 13.3, 6.7, 3.3, 1H), 1.55 (t, J = 13.1, 1H), 1.52-1.46 (m, 1H), 1.34 (dtd, J = 13.7, 9.1, 4.8, 1H), 1.04 (d, J = 6.4, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm)  $\delta$  211.5, 198.3, 152.5, 138.0, 129.5, 115.1, 50.9, 44.3, 44.2, 42.7, 33.3, 32.7, 31.5, 30.4, 29.9, 18.8; IR (thin film) 2919, 1702, 1684, 1450, 1227  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{16}\text{H}_{22}\text{O}_2$   $[\text{M}+\text{Na}]^+$  269.1518 found 269.1518.

### Siloxydienes 3.36

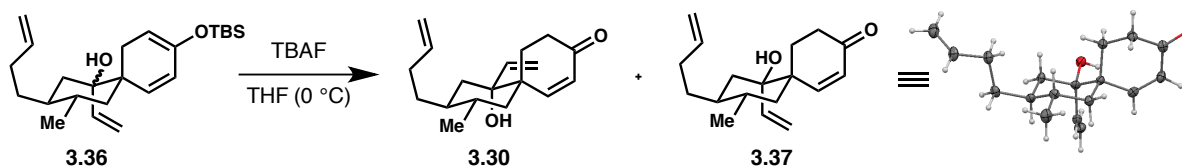


To a solution of 303 mg (1.23 mmol) **3.35** in 13 mL CH<sub>2</sub>Cl<sub>2</sub> was added 0.60 mL (4.3 mmol) NEt<sub>3</sub> and 0.32 mL (1.4 mmol) TBSOTf at -78 °C. After 3 hours of stirring at -78 °C the reaction was diluted with 10 mL hexanes and poured into a separatory funnel containing 10 mL sat. NaHCO<sub>3</sub>. The flask was rinsed twice with 10 mL pentane, the organic layer washed with 10 mL sat. NaHCO<sub>3</sub>, 10 mL water, 5 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford a compound of sufficient purity to proceed. The crude material from the previous step in 7 mL THF was treated with 7.0 mL (5.0 mmol) 0.71 M vinylmagnesium bromide/THF at 0 °C. The ice bath was removed and stirring continued for 8 hours. The reaction was quenched at 0 °C with 20 mL half sat. NH<sub>4</sub>Cl, diluted with 30 mL EtOAc and the phases separated. The organic layer was washed with 10 mL sat. NaHCO<sub>3</sub>, 5 mL water and 10 mL brine, then dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (160:4:1 hexanes/EtOAc/NEt<sub>3</sub>) to afford 373 mg (78% over two steps, 1:1.6 dr) **3.36** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) major δ 6.39 (dd, J = 17.0, 10.9, 1H), 5.86-5.75 (m, 2H), 5.71 (dd, J = 10.2, 2.2, 1H), 5.44 (d, J = 10.2, 1H), 5.42 (dd, J = 17.0, 1.8, 1H), 4.83 (dt, J = 4.7, 2.2, 1H), 2.86 (dd, J = 17.6, 4.5, 1H), 1.78 (s, 1H), 1.74 (dd, J = 14.0, 4.0, 1H), 1.49 (t, J = 12.7, 1H), minor δ 6.06 (dd, J = 17.4, 10.9, 1H), 5.86-5.75 (m, 1H), 5.65 (dd, J = 10.2, 2.1, 1H), 4.67-4.65 (m, 1H) (mixed) 5.20-5.08 (m, 3H), 5.04-4.92 (m, 4H), 2.36-2.22 (m, 3H), 2.18-2.08 (m, 2H), 2.00-1.89 (m, 2H), 1.69-1.56 (m, 6H), 1.43-1.33 (m, 4H), 1.22-1.13 (m, 4H), 0.95-0.90 (m, 24H), 0.13 (s, 6H), 0.13 (s, 3H), 0.12 (s, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) combined δ 147.2, 146.7, 143.9, 141.3, 139.2, 138.9, 134.7, 134.1, 127.0,



125.6, 114.3, 114.2, 113.9, 112.7, 101.9, 100.4, 77.9, 76.1, 42.2, 42.1, 41.7, 39.9, 39.8, 39.0, 37.7, 37.6, 32.8, 32.3, 30.9, 30.52, 30.47, 29.1, 27.6, 25.7, 25.7, 19.6, 19.5, 18.0, 18.0, -4.4, -4.45, -4.48; IR (thin film) 2928, 1654, 1252, 1201  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{24}\text{H}_{40}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  411.2695 found 411.2693.

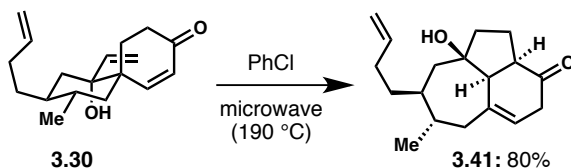
### Spirocyclic Enones **3.30** and **3.37**



To a solution of 0.99 g (2.54 mmol) **3.36** in 10 mL THF was added 4 mL TBAF (1 M in THF) at 0 °C. After 2 minutes at 0 °C, the reaction was diluted with sat. aq.  $\text{NaHCO}_3$  and poured into EtOAc. The layers were separated and the organic layer washed twice with sat. aq.  $\text{NaHCO}_3$ , brine, dried over  $\text{MgSO}_4$  and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1 hexanes/EtOAc) to afford 201 mg (28%) **3.30** as a colorless oil and 394 mg (56%) **3.37** as a white solid. **3.30**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.21 (dd,  $J = 10.4, 1.8, 1\text{H}$ ), 5.95 (d,  $J = 17.4, 1\text{H}$ ), 5.93 (dd,  $J = 17.4, 10.7, 1\text{H}$ ), 5.82 (ddt,  $J = 17.0, 10.3, 6.7, 1\text{H}$ ), 5.25 (dd,  $J = 17.4, 0.9, 1\text{H}$ ), 5.16 (d,  $J = 10.9, 1\text{H}$ ), 5.03 (dd,  $J = 17.1, 1.6, 1\text{H}$ ), 4.98-4.95 (m, 1H), 2.49-2.41 (m, 1H), 2.35 (dt,  $J = 17.4, 4.5, 1\text{H}$ ), 2.20-2.12 (m, 1H), 2.08-1.88 (m, 3H), 1.75-1.64 (m, 4H), 1.58-1.43 (m, 3H), 1.38 (s, 1H), 1.28-1.21 (m, 1H), 0.97 (d,  $J = 6.1, 3\text{H}$ );  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  199.4, 156.1, 142.6, 138.9, 128.7, 114.7, 114.4, 75.1, 43.3, 37.47, 37.46, 37.2, 33.6, 32.2, 31.7, 30.5, 27.0, 19.6; IR (thin film) 3435, 2924, 1664  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{26}\text{O}_2$   $[\text{M}+\text{Na}]^+$  297.1830 found 297.1830. **3.37**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  6.68 (d,  $J = 10.4, 1\text{H}$ ), 6.50 (dd,  $J = 17.1, 10.9, 1\text{H}$ ), 5.97 (d,  $J$

= 10.4, 1H), 5.78 (ddt,  $J = 17.0, 10.3, 6.7$ , 1H), 5.40 (dd,  $J = 17.1, 1.2$ , 1H), 5.28 (dd,  $J = 10.9, 1.1$ , 1H), 5.00 (dq,  $J = 17.1, 1.6$ , 1H), 4.95 (dt,  $J = 10.2, 0.8$ , 1H), 2.68 (ddd,  $J = 17.2, 7.1, 5.3$ , 1H), 2.54 (ddd,  $J = 14.1, 9.4, 4.9$ , 1H), 2.41 (ddd,  $J = 17.2, 9.7, 5.4$ , 1H), 2.17-2.10 (m, 1H), 2.01-1.92 (m, 2H), 1.72-1.63 (m, 3H), 1.59 (dd,  $J = 13.6, 3.6$ , 1H), 1.54 (s, 1H), 1.53-1.45 (m, 1H), 1.41 (t,  $J = 12.9$ , 1H), 1.31-1.26 (m, 1H), 1.19 (dtd,  $J = 13.7, 9.1, 4.7$ , 1H), 0.97 (d,  $J = 6.2, 3H$ );  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  200.0, 155.3, 141.0, 138.6, 130.0, 114.63, 114.58, 77.3, 43.4, 41.7, 41.1, 40.0, 34.4, 32.7, 31.6, 30.4, 26.5, 19.5; IR (thin film) 3433, 2920, 1663, 1450  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{26}\text{O}_2$   $[\text{M}+\text{Na}]^+$  297.1830 found 297.1829. X-Ray quality crystals were grown from pentane /  $\text{CH}_2\text{Cl}_2$  (mp = 123-125  $^\circ\text{C}$ ).

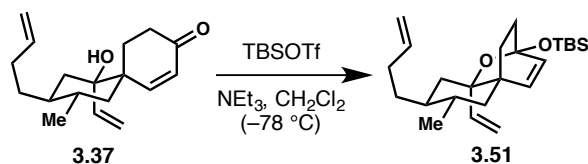
### Tricycle 3.41



A microwave vial was charged with 18.8 mg (0.0685 mmol) **3.30** and 1 mL chlorobenzene, sealed and heated at 190  $^\circ\text{C}$  in a microwave for 4 hours + 4 hours. All volatiles were removed in vacuo and the crude material purified by column chromatography (6:1 hexanes/EtOAc) to afford 15.0 mg (80%) **3.41** as a white solid, which partially isomerized upon attempted recrystallization.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.81-5.73 (m, 1H), 5.64 (s, 1H), 5.00 (d,  $J = 17.2$ , 1H), 4.94 (d,  $J = 10.0$ , 1H), 2.97 (d,  $J = 21.0$ , 1H), 2.82-2.90 (m, 2H), 2.56-2.50 (m, 1H), 2.37 (d,  $J = 14.0$ , 1H), 2.15-2.05 (m, 2H), 2.02-1.44 (m, 12H), 1.04 (d,  $J = 6.4, 3H$ );  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  209.4, 138.8, 136.7, 121.5, 114.5, 82.2, 57.3,

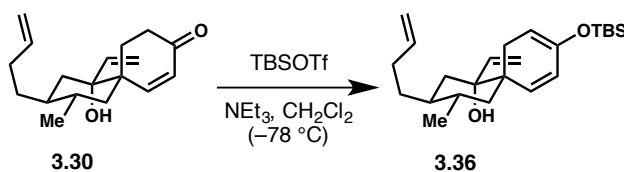
48.6, 43.7, 41.8, 40.4, 39.7, 39.2, 39.0, 34.6, 31.8, 22.9, 21.6; IR (thin film) 3467, 2917, 2850, 1708, 1459  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{26}\text{O}_2$   $[\text{M}+\text{Na}]^+$  297.1830 found 297.1834.

### [2.2.2] Bicycle 3.51



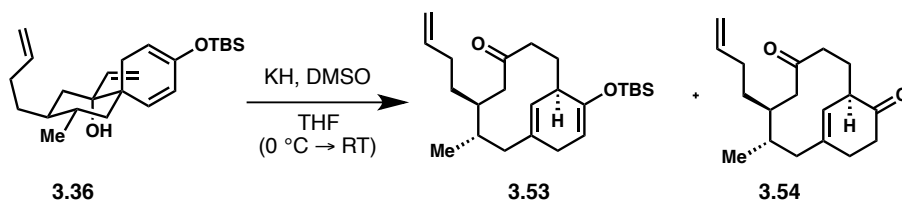
To a solution of 10 mg (0.036 mmol) **42** in 0.5 mL  $\text{CH}_2\text{Cl}_2$  was added 0.2 mL (1.4 mmol)  $\text{NEt}_3$  and 50  $\mu\text{L}$  (0.22 mmol) TBSOTf in  $-78\text{ }^\circ\text{C}$ . After 1 hours of stirring at  $-78\text{ }^\circ\text{C}$  the reaction was diluted with  $\text{CH}_2\text{Cl}_2$  and poured into a separatory funnel containing sat. aq.  $\text{NaHCO}_3$ . The flask was rinsed twice with  $\text{CH}_2\text{Cl}_2$  and the organic layer washed with sat. aq.  $\text{NaHCO}_3$ , brine, dried over  $\text{MgSO}_4$  and all volatiles removed in vacuo. The crude material was purified by column chromatography to afford 7 mg (50%) **44** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  6.12-6.06 (m, 2H), 5.86 (d,  $J = 8.3$ , 1H), 5.80 (ddt,  $J = 17.0, 10.3, 6.7$ , 1H), 5.16 (dd,  $J = 17.0, 2.3$ , 1H), 5.04-4.99 (m, 2H), 4.94 (d,  $J = 10.1$ , 1H), 2.29 (ddd,  $J = 12.6, 9.8, 2.8$ , 1H), 2.19-2.12 (m, 1H), 1.95 (dq,  $J = 14.7, 7.4$ , 1H), 1.90-1.85 (m, 2H), 1.70-1.63 (m, 2H), 1.61-1.46 (m, 4H), 1.23-1.15 (m, 2H), 1.04 (td,  $J = 12.5, 4.8$ , 1H), 0.94 (d,  $J = 5.6$ , 3H), 0.91 (s, 9H), 0.19 (s, 3H), 0.15 (s, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  141.7, 139.1, 137.0, 136.1, 115.1, 114.3, 97.6, 79.8, 40.11, 39.9, 39.62, 39.56, 34.6, 33.1, 31.6, 30.6, 25.9, 25.7, 20.0, 17.9,  $-2.4$ ,  $-2.7$ ; IR (thin film) 2928, 1684, 1640, 1212  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{24}\text{H}_{40}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  411.2695 found 411.2667

### Dienoxysilane 3.36



To a solution of 52 mg (0.19 mmol) **3.30** in 4 mL CH<sub>2</sub>Cl<sub>2</sub> was added 0.4 mL (2.9 mmol) NEt<sub>3</sub> and 0.1 mL (0.44 mmol) TBSOTf at -78 °C. After 1 hours of stirring at -78 °C the reaction was diluted with CH<sub>2</sub>Cl<sub>2</sub> and poured into a separatory funnel containing sat. aq. NaHCO<sub>3</sub>. The flask was rinsed twice with CH<sub>2</sub>Cl<sub>2</sub> and the organic layer washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (hexanes→20:1 hexanes/EtOAc) to afford 62 mg (85%) **3.36** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 6.06 (dd, J = 17.4, 10.9, 1H), 5.83-5.75 (m, 1H), 5.81 (d, J = 10.1, 1H), 5.65 (dd, J = 10.2, 2.1, 1H), 5.17 (dd, J = 17.4, 1.3, 1H), 5.10 (dd, J = 10.9, 1.3, 1H), 5.02 (dq, J = 17.1, 1.8, 1H), 4.96 (ddt, J = 10.2, 2.3, 1.1, 1H), 4.67-4.65 (m, 1H), 2.35 (dd, J = 17.2, 3.9, 1H), 2.25 (dd, J = 17.3, 5.5, 1H), 2.17-2.11 (m, 1H), 2.00-1.93 (m, 1H), 1.72-1.64 (m, 2H), 1.58 (dd, J = 13.6, 3.4, 1H), 1.49 (t, J = 12.7, 2H), 1.44-1.37 (m, 3H), 1.24-1.18 (m, 1H), 0.94-0.90 (m, 12H), 0.13 (s, 3H), 0.12 (s, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 147.2, 143.9, 139.2, 134.7, 125.6, 114.2, 112.7, 100.4, 76.1, 41.7, 39.0, 37.7, 37.6, 32.3, 30.54, 30.52, 29.1, 25.7, 19.7, 18.0, -4.4, -4.5; IR (thin film) 3435, 2928, 2856, 1652 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>24</sub>H<sub>40</sub>O<sub>2</sub>Si [M+Na]<sup>+</sup> 411.2695 found 411.2693.

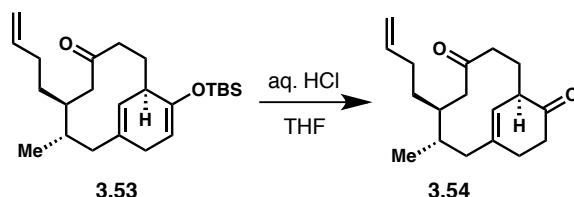
## Enoxysilane **3.53** and Skipped Enone **3.54**



To a stirring solution of oil free KH in 0.5 mL THF was added 20 mg (0.26 mmol) DMSO in 0.5 mL THF at room temperature, followed by 50 mg (0.13 mmol) **3.36** in 0.6 mL THF with 0.4 mL and 0.5 mL THF rinses at 0 °C. The ice bath was removed after 10 minutes and the reaction stirred for 35 minutes. The reaction was quenched with 10 drops EtOH, then sat. aq. NaHCO<sub>3</sub>. The aqueous phase was extracted twice with EtOAc. The organic phases were combined, washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (hexanes→40:1 hexanes/EtOAc) to afford 20 mg (40%) **3.53** as a waxy solid and 11 mg (31%) **3.54** as a waxy solid. **3.53**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.83 (ddt, J = 17.0, 10.3, 6.6, 1H), 5.03-4.92 (m, 4H), 3.05-3.00 (m, 1H), 2.79 (s, 1H), 2.69-2.60 (m, 2H), 2.43 (td, J = 13.2, 1.6, 1H), 2.27-2.20 (m, 1H), 2.12-1.99 (m, 4H), 1.97-1.90 (m, 4H), 1.68-1.62 (m, 1H), 1.54-1.48 (m, 1H), 1.40-1.33 (m, 1H), 0.93 (s, 12H), 0.17 (s, 6H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 212.4, 149.1, 139.1, 134.4, 129.0, 114.1, 101.5, 50.4, 48.0, 40.3, 37.9, 37.8, 34.8, 34.3, 31.6, 29.7, 25.7, 21.4, 18.1, -4.4, -4.5; IR (thin film) 2854, 2927, 2856, 1705, 1658, 1203 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>24</sub>H<sub>40</sub>O<sub>2</sub>Si [M+Na]<sup>+</sup> 411.2695 found 411.2687. **3.54**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ (major) 5.80 (ddt, J = 16.9, 10.2, 6.7, 1H), 5.21 (s, 1H), 5.00 (d, J = 17.2, 1H), 4.93 (d, J = 10.1, 1H), 2.93 (s, 1H), 2.69-2.47 (m, 5H), 2.38-2.33 (m, 1H), 2.25-1.87 (m, 9H), 1.63-1.55 (m, 1H), 1.53-1.47 (m, 1H), 1.42-1.34 (m, 1H), 0.95 (d, J = 6.7 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ (major) 212.0, 211.1, 138.8, 138.5, 130.1, 114.3, 51.0, 48.9, 47.6, 39.4, 39.3, 37.8, 35.0, 34.2, 30.22,

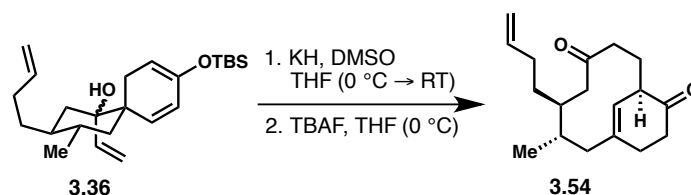
29.4, 28.7, 21.6; IR (thin film) 2955, 2924, 2914, 1706, 1436  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{26}\text{O}_2$   $[\text{M}+\text{Na}]^+$  297.1830 found 297.1831.

### Skipped Enone 3.54



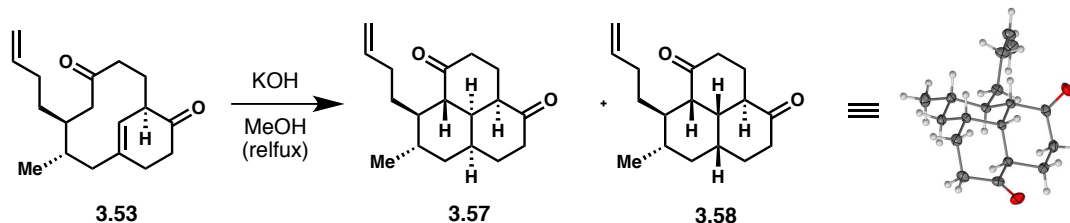
A solution of 18 mg (0.046 mmol) **3.53** in 0.8 mL THF was treated with 1.6 mL 1M aq. HCl. After stirring overnight, 6M aq. HCl was added until most starting material had disappeared and new polar spots appeared. After 14 h the reaction was quenched with sat. aq.  $\text{NaHCO}_3$  and extracted with EtOAc. The organic layer was washed with brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (hexanes $\rightarrow$ 30:1 $\rightarrow$ 8:1 $\rightarrow$ 1:1 hexanes/EtOAc) to afford 3 mg (17%) recovered **3.53**, 6.5 mg (51%) **3.54** as a waxy solid and 2 mg of two unknown compounds.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm) major  $\delta$  5.80 (ddt,  $J = 16.9, 10.2, 6.7, 1\text{H}$ ), 5.21 (s, 1H), 5.00 (d,  $J = 17.2, 1\text{H}$ ), 4.93 (d,  $J = 10.1, 1\text{H}$ ), 2.93 (s, 1H), 2.69-2.47 (m, 5H), 2.38-2.33 (m, 1H), 2.25-1.87 (m, 9H), 1.63-1.55 (m, 1H), 1.53-1.47 (m, 1H), 1.42-1.34 (m, 1H), 0.95 (d,  $J = 6.7\text{ Hz}, 3\text{H}$ );  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm) major  $\delta$  212.0, 211.1, 138.8, 138.5, 130.1, 114.3, 51.0, 48.9, 47.6, 39.4, 39.3, 37.8, 35.0, 34.2, 30.22, 29.4, 28.7, 21.6; IR (thin film) 2955, 2924, 2914, 1706, 1436  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{26}\text{O}_2$   $[\text{M}+\text{Na}]^+$  297.1830 found 297.1831.

### Skipped Enone 3.54



To a slurry of 152 mg (3.79 mmol) oil free potassium hydride in 7 mL THF was added 0.13 mL (1.8 mmol) DMSO at 0 °C. After stirring for 10 minutes, 363 mg (0.933 mmol) **3.36** in 3 mL THF was added at 0 °C. The ice bath was removed and stirring continued for 1 hour before being cooled back to 0 °C and quenched with 0.25 mL AcOH. The reaction was partitioned between 45 mL third sat. NaHCO<sub>3</sub> and 50 mL EtOAc. The organic layer was washed twice with 10 mL water, then 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed to afford a crude residue. The crude material was dissolved in 8.5 mL THF, cooled to 0 °C and treated with 2.3 mL (2.3 mmol) 1 M TBAF/THF. After 10 minutes, 30 mL half sat. NH<sub>4</sub>Cl and 40 mL EtOAc was added and the layers separated. The organic layer was washed with 10 mL water, 5 mL brine, dried over MgSO<sub>4</sub>, filtered and volatiles removed in vacuo. The crude material was purified by column chromatography (4:1 hexanes/EtOAc) to afford 130 mg (51% over 2 steps, 4.6:1 dr) **3.54** as a white wax. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) major δ 5.80 (ddt, J = 16.9, 10.2, 6.7, 1H), 5.21 (s, 1H), 5.00 (d, J = 17.2, 1H), 4.93 (d, J = 10.1, 1H), 2.93 (s, 1H), 2.69-2.47 (m, 5H), 2.38-2.33 (m, 1H), 2.25-1.87 (m, 9H), 1.63-1.55 (m, 1H), 1.53-1.47 (m, 1H), 1.42-1.34 (m, 1H), 0.95 (d, J = 6.7 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) major δ 212.0, 211.1, 138.8, 138.5, 130.1, 114.3, 51.0, 48.9, 47.6, 39.4, 39.3, 37.8, 35.0, 34.2, 30.22, 29.4, 28.7, 21.6; IR (thin film) 2955, 2924, 2914, 1706, 1436 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>18</sub>H<sub>26</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 297.1830 found 297.1831.

## Perhydrophenalenediones **3.57** and **3.58**

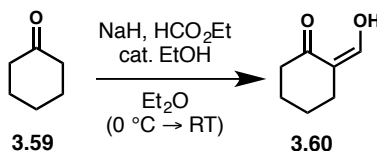


A 10 mL round bottom flask was charged with 42.0 mg (0.153 mmol) **3.53**, 2.0 mL (0.11 mmol) 0.054 M KOH/MeOH, fitted with a reflux condenser and heated at 75 °C under rigorous exclusion of oxygen. The reaction was cooled to room temperature after 3 hours, all volatiles removed in vacuo and the residue dissolved in 5 mL CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was washed with 5 mL water, 4 mL half sat. NH<sub>4</sub>Cl, 3 mL brine, then dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford a mixture of **3.57** and **3.58** in >80% crude yield. The crude oil purified by column chromatography (100:1→40:1 C<sub>6</sub>H<sub>6</sub>/EtOAc) to afford 12.1 mg (29%) **3.58** as a white solid which was recrystallized from CH<sub>2</sub>Cl<sub>2</sub>/pentane to afford colorless needles (mp = 111-113°C) and 15.1 mg (36%) **3.57** as a colorless oil. **3.57**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.74 (ddt, J = 16.9, 10.2, 6.6, 1H), 4.96 (d, J = 17.1, 1H), 4.89 (d, J = 10.2, 1H), 2.71-2.65 (m, 2H), 2.57 (dd, J = 13.3, 6.1, 1H), 2.51-2.46 (m, 2H), 2.40-2.30 (m, 3H), 2.21 (dd, J = 11.9, 4.7, 1H), 2.16-2.05 (m, 2H), 1.91-1.83 (m, 2H), 1.62-1.52 (m, 4H), 1.47-1.38 (m, 2H), 1.32 (td, J = 12.7, 3.6, 1H), 0.91 (d, J = 6.2, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 213.6, 211.2, 139.3, 113.9, 51.6, 50.4, 48.7, 41.01, 41.00, 40.1, 39.0, 36.1, 30.1, 30.0, 28.5, 27.3, 27.0, 20.2; IR (thin film) 2927, 1710, 1443 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>18</sub>H<sub>26</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 297.1830 found 297.1840. **3.58**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.75 (ddt, J = 16.9, 10.2, 6.7, 1H), 5.01 (d, J = 17.1, 1H), 4.93 (d, J = 10.1, 1H), 2.95 (t, J = 12.3, 1H), 2.56-2.40 (m, 3H), 2.34-2.27 (m, 3H), 2.17-2.08 (m, 2H), 1.94-1.85 (m, 3H), 1.77 (q, J = 12.4, 1H), 1.70-1.59 (m, 3H), 1.55-1.50 (m, 1H), 1.26-1.18 (m, 3H), 1.06 (d, J = 6.3, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77



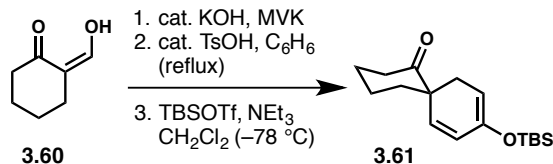
ppm)  $\delta$  212.9, 211.4, 138.7, 114.6, 56.7, 47.9, 43.8, 39.3, 37.7, 37.4, 34.9, 34.8, 34.7, 32.0, 28.1, 27.9, 26.2, 19.9; IR (thin film) 2921, 1709  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{26}\text{O}_2$   $[\text{M}+\text{Na}]^+$  297.1830 found 297.1838.

**2-Hydroxymethylenecyclohexanone 3.60** [Adapted from the literature]<sup>56</sup>



To an ice cooled solution of 1.11 g (48.3 mmol) Na metal cut and flattened in 100 mL Et<sub>2</sub>O was added 4.6 mL (44.4 mmol) cyclohexanone, 5.3 mL (65.6 mmol) ethyl formate, and 0.2 mL EtOH. The ice bath was allowed to melt over the course of 2 hours, during which time the reaction turned into a thick orange mass. The contents were stirred for an additional 2 hours, quenched with 50 mL water, the phases separated and the aqueous phase washed twice with 50 mL Et<sub>2</sub>O. The aqueous layer was acidified with 10 mL 6 M HCl and extracted twice with 50 mL Et<sub>2</sub>O. These organic layers were combined, washed with 10 mL water, 10 mL half sat. NaHCO<sub>3</sub>, 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The yellow oil was distilled (63-65 °C/3.5 mmHg) to afford 2.64 g (47%) **3.60** as a colorless liquid. The spectral data was identical to the literature.<sup>57</sup>

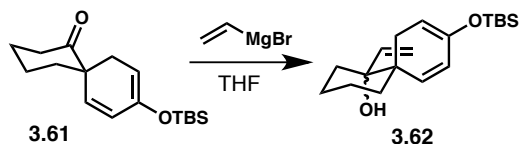
### Silyloxydiene 3.61



A flask was charged with 2.63 g (20.9 mmol) **3.60**, 4.3 mL (52.1 mmol) methyl vinyl ketone and 75 mg (1.3 mmol) powdered KOH at room temperature (exotherm). After stirring for 9 hours the contents were diluted with EtOAc, washed twice with sat. NaHCO<sub>3</sub>, once with sat. NH<sub>4</sub>Cl then brine. The organic layer was dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford a yellow oil. The crude oil was dissolved in 100 mL benzene and refluxed over Dean-Stark with 200 mg (1.0 mmol) p-TsOH•H<sub>2</sub>O. After 3 hours the reaction was cooled to room temperature and washed thrice with sat. NaHCO<sub>3</sub>, then brine. The organic layer was dried over MgSO<sub>4</sub> and all volatiles removed in vacuo. The dark oil was purified by column chromatography (6:1→4:1 hexanes/EtOAc) to afford 1.73 g of an impure product. To 1.65 g impure enone in 50 mL CH<sub>2</sub>Cl<sub>2</sub> cooled to -78 °C was added 5.2 mL (37 mmol) NEt<sub>3</sub> and 2.1 mL (9.26 mmol) TBSOTf sequentially. After 3 hours the flask was removed from the cold bath, diluted with 25 mL pentane and poured into stirring sat. NaHCO<sub>3</sub>. The flask was rinsed with an additional 50 mL pentane. The layers were separated and the organic layer washed once with sat. NaHCO<sub>3</sub>, water and brine, then dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The yellow oil was purified by column (50:1→20:1 hexanes/EtOAc) to afford 1.96 g (32% over 3 steps) **3.61** as a colorless wax. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.90 (d, *J* = 10.1, 1H), 5.73 (dd, *J* = 10.0, 2.1, 1H), 4.78-4.76 (m, 1H), 2.67 (dd, *J* = 17.0, 4.0, 1H), 2.51 (ddd, *J* = 14.0, 8.7, 5.4, 1H), 2.33 (ddd, *J* = 13.5, 7.8, 5.4, 1H), 2.19 (dd, *J* = 17.0, 5.4, 1H), 1.89-1.65 (m, 6H), 0.90 (s, 9H), 0.11 (s, 3H), 0.10 (s, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 212.3, 144.9, 130.8, 126.6,

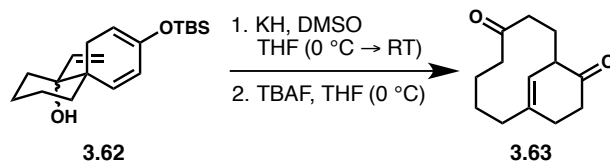
100.5, 49.8, 38.4, 36.8, 29.9, 27.8, 25.6, 20.8, 18.0, -4.58, -4.63; IR (thin film) 3046, 2931, 2894, 2857, 1708, 1653, 1254, 1225, 1212, 1182, 1127  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{17}\text{H}_{28}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  315.1756 found 315.1751.

### Allylic Alcohols 3.62



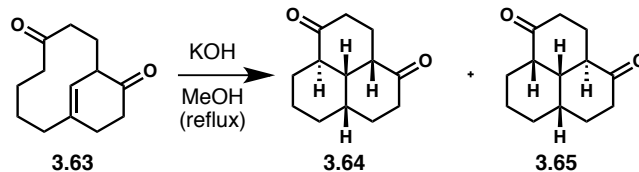
To a solution of 1.92 g (6.56 mmol) **3.61** in 30 mL THF was added 27 mL (27 mmol) 1.0 M vinylmagnesium bromide/THF at 0 °C. After 10 minutes the cold bath was removed and the contents stirred for 6 hours. The reaction was quenched at 0 °C with 1 mL sat.  $\text{NH}_4\text{Cl}$  and 5 mL sat.  $\text{NaHCO}_3$ , then passed through a pad of Celite with 300 mL  $\text{Et}_2\text{O}$ . The organic layer was washed once with sat.  $\text{NaHCO}_3$ , water, brine then dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The oil was purified by column chromatography (20:1→10:1 hexanes/ $\text{EtOAc}$ ) to afford 1.69 g (80%, 5.4:1 dr) **3.62** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm) major  $\delta$  6.07 (dd,  $J = 17.2, 10.9$ , 1H), 5.81 (d,  $J = 10.3$ , 1H), 5.67 (dd,  $J = 10.3, 2.2$ , 1H), 5.26 (dd,  $J = 17.2, 1.5$ , 1H), 5.11 (dd,  $J = 10.9, 1.4$ , 1H), 4.75 (dt,  $J = 5.9, 2.9$ , 1H), 2.60 (dd,  $J = 17.2, 3.3$ , 1H), 1.96 (dd,  $J = 17.2, 6.1$ , 1H), 1.92-1.81 (m, 2H), 1.70-1.61 (m, 2H), 1.58-1.44 (m, 4H), 1.41 (s, 1H), 0.92 (s, 9H), 0.13 (m, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm) major  $\delta$  146.3, 144.3, 133.8, 126.0, 112.3, 101.4, 76.4, 41.1, 34.5, 32.6, 29.9, 25.7, 21.5, 21.1, 18.0, -4.49, -4.51; IR (thin film) 3482, 2931, 2858, 1654, 1252, 1223, 1190  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{19}\text{H}_{32}\text{O}_2\text{Si}$   $[\text{M}+\text{H}]^+$  321.2250 found 321.2239.

### Skipped Enone 3.63



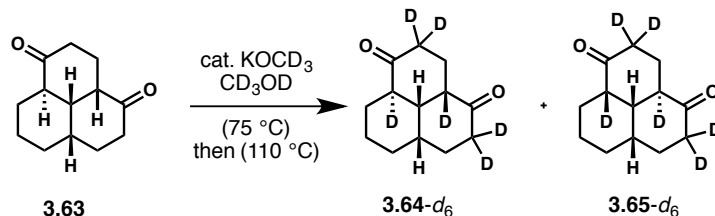
To slurry of 87 mg (2.2 mmol) oil free potassium hydride in 3 mL THF was added 0.09 mL (1.27 mmol) DMSO at 0 °C. After stirring for 10 minutes, 205 mg (0.639 mmol) **3.62** in 1.5 mL THF was added at 0 °C. The ice bath was removed and stirring continued for 1 hour before being cooled back to 0 °C and quenched with 0.15 mL AcOH. The reaction was partitioned between 20 mL half sat. NaHCO<sub>3</sub> and 20 mL EtOAc. The layers were separated and the organic phase washed with 5 mL sat. NaHCO<sub>3</sub>, 5 mL water, 5 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed. The crude material was dissolved in 6 mL THF, cooled to 0 °C and treated with 1.9 mL (1.9 mmol) 1 M TBAF/THF. After 10 minutes, 30 mL half sat. NH<sub>4</sub>Cl and 20 mL EtOAc was added, the layers separated and the aqueous phase washed with 10 mL EtOAc. The organic layers were combined, washed with 10 mL water, 5 mL brine, dried over MgSO<sub>4</sub>, filtered and volatiles removed in vacuo. The crude material was purified by column chromatography (5:1 hexanes/EtOAc) to afford 102 mg (77% over 2 steps) **3.63** as a white solid, which was recrystallized from Et<sub>2</sub>O/hexanes to afford white needles (mp = 50-51 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.25 (s, 1H), 2.94 (bs, 1H), 2.63-2.46 (m, 4H), 2.42-2.20 (m, 5H), 2.10-1.98 (m, 4H), 1.78-1.72 (m, 1H), 1.62-1.56 (m, 1H), 1.52-1.45 (m, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 212.2, 212.1, 138.5, 127.8, 47.5, 43.8, 39.8, 39.3, 37.1, 29.5, 28.8, 24.7, 22.4; IR (thin film) 2926, 1704, 1437 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>13</sub>H<sub>18</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 229.1205 found 229.1206.

## Tricycles 3.64 & 3.65



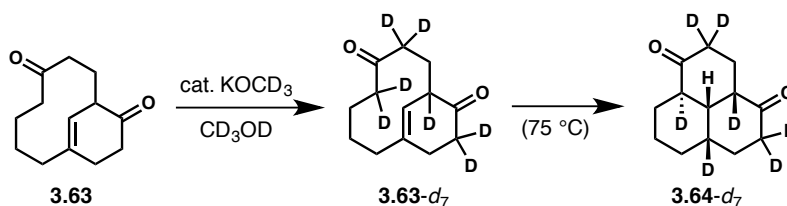
A 10 mL round bottom flask was charged with 41.6 mg (0.202 mmol) **3.63** and 2 mL (0.14 mmol) 0.07 M KOH/MeOH before being fitted with a reflux condenser and heated at 75 °C under rigorous exclusion of oxygen. The reaction was cooled to room temperature after 2 hours, partitioned between 5 mL EtOAc and 5 mL half sat. NH<sub>4</sub>Cl. The aqueous layer was extracted twice with 3 mL EtOAc. The organic layers were combined, washed with 2 mL brine, dried over MgSO<sub>4</sub>, filtered, the volatiles removed in vacuo and the crude oil purified by column chromatography (8:1→3:2 hexanes/EtOAc) to afford 25.6 mg (62%) **3.64** as a white solid, which was recrystallized from CH<sub>2</sub>Cl<sub>2</sub>/pentane to afford a single colorless prism (mp = 91-93 °C) and 10.2 mg (24%) **3.65** as a colorless oil. **3.64**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 2.70-2.63 (m, 2H), 2.545-2.47 (m, 3H), 2.42-2.39 (m, 1H), 2.34-2.31 (m, 1H), 2.26-2.15 (m, 3H), 1.93 (d, J = 13.8, 1H), 1.84-1.80 (m, 1H), 1.65-1.52 (m, 4H), 1.46-1.36 (m, 1H), 1.28-1.20 (m, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 212.6, 211.2, 49.7, 48.1, 46.3, 41.0, 38.4, 35.8, 30.1, 26.7, 25.4, 25.3, 19.7; IR (thin film) 2917, 2848, 1709 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>13</sub>H<sub>18</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 229.1205 found 229.1206. **3.65**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 2.93 (t, J = 12.3, 1H), 2.57-2.42 (m, 3H), 2.35-2.27 (m, 3H), 2.17 (dt, J = 12.7, 4.0, 1H), 2.01 (dt, J = 13.2, 2.9, 1H), 1.96-1.86 (m, 3H), 1.79 (dt, J = 12.8, 3.2, 1H), 1.70-1.57 (m, 4H), 1.51-1.42 (m, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 213.0, 211.6, 52.8, 46.2, 43.1, 37.4, 36.9, 34.6, 32.4, 25.4, 25.2, 25.0, 24.6; IR (thin film) 2927, 2860, 1706 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>13</sub>H<sub>18</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 229.1205 found 229.1206.

### Deuteration Experiments on 3.64



A J-Young tube was charged with a solution of 6.0 mg (0.029 mmol) **3.64** in 0.6 mL CD<sub>3</sub>OD and analyzed by <sup>1</sup>H, <sup>13</sup>C and DEPT-Q NMR. Then 0.05 mL (0.02 mmol) 0.4 M KOCD<sub>3</sub>/DOCD<sub>3</sub> (prepared from potassium hydride and CD<sub>3</sub>OD in toluene, followed by removal of all volatiles and washing with pentane) was added and placed in an oil bath preheated to 75 °C and reacted for 2 hours. Analysis was performed by <sup>1</sup>H and <sup>13</sup>C NMR; **3.64-d<sub>6</sub>** was observed. After 4 days at 110 °C a ~2:1 mixture **3.64-d<sub>6</sub>** and **3.65-d<sub>6</sub>** was observed by <sup>13</sup>C NMR.

### Deuteration Experiments on 3.63



A J-Young tube was charged with a solution of 17.0 mg (0.0824 mmol) **3.63** in 0.6 mL CD<sub>3</sub>OD and analyzed by <sup>1</sup>H, <sup>13</sup>C and DEPT-Q NMR. Then 0.05 mL (0.055 mmol) 1.1 M KOCD<sub>3</sub>/DOCD<sub>3</sub> (prepared from potassium hydride and CD<sub>3</sub>OD in toluene, followed by removal of all volatiles and washing with pentane) was added and sonicated for a half hour and analyzed by <sup>1</sup>H and <sup>13</sup>C NMR; **3.63-d<sub>7</sub>** was observed. The reaction was quenched with 0.1 mL CD<sub>3</sub>CO<sub>2</sub>D, diluted with 3 mL water and extracted thrice with 3 mL Et<sub>2</sub>O. The organic layers were combined, washed with 1 mL water, 1 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The residue was taken up in 0.6 mL CD<sub>3</sub>OD and treated with with 0.05 mL (0.055 mmol)

KOCD<sub>3</sub>/DOCD<sub>3</sub> and the NMR tube placed in an oil bath preheated to 75 °C and reacted for 2 hours. Analysis was performed by <sup>1</sup>H and <sup>13</sup>C NMR; **3.64-*d*<sub>7</sub>** was observed.

### 3.9 References and Notes

- (1) Fleming, I. *Pericyclic Reactions*, 1st ed.; Oxford University Press, 1998.
- (2) Paquette, L. A. *Synlett* **1990**, 67.
- (3) Wilson, S. R. *Org. React.* **1993**, 43, 93.
- (4) Paquette, L. A. *Tetrahedron* **1997**, 53, 13971.
- (5) Swaminathan, S. *J. Indian Chem. Soc.* **1984**, 61, 99.
- (6) Swaminathan, S.; John, J. P.; Ramachandran, S. *Tetrahedron Lett.* **1962**, 3, 729.
- (7) Berson, J. A.; Jones, M. J. *Am. Chem. Soc.* **1964**, 86, 5019.
- (8) Evans, D. A.; Golob, A. M. *J. Am. Chem. Soc.* **1975**, 97, 4765.
- (9) Swaminathan, S.; Srinivasan, K. G.; Vankataramani, P. S. *Tetrahedron* **1970**, 26, 1453.
- (10) Chandrasekaran, S.; Vankataramani, P. S.; Srinivasan, K. G. . S. S. *Tetrahedron Lett.* **1973**, 14, 991.
- (11) Uma, R.; Swaminathan, S.; Rajagopalan, K. *Tetrahedron Lett.* **1984**, 25, 5825.
- (12) An additional compound was isolated. Its structure was not conclusively assigned, but the proposed structure would have arisen via a [1, 3] rearrangement. .
- (13) Tice, C. M.; Heathcock, C. H. *J. Org. Chem.* **1981**, 46, 9.
- (14) Rao, C. S. S.; Kumar, G.; Rajagopalan, K.; Swaminathan, S. *Tetrahedron* **1982**, 38, 2195.
- (15) Shanmugam, P.; Devan, B.; Srinivasan, R.; Rajagopalan, K. *Tetrahedron* **1997**, 53, 12637.
- (16) Ravikumar, V. T.; Rajagopalan, K.; Swaminathan, S. *Tetrahedron Lett.* **1985**, 26, 6137.
- (17) Roosen, P. C.; Vanderwal, C. D. *Org. Lett.* **2014**, 16, 4368.
- (18) Corey, E. J.; Magriotis, P. A. *J. Am. Chem. Soc.* **1987**, 109, 287.
- (19) Dave, V.; Whitehurst, J. S. *J. Chem. Soc. Perkin I* **1973**, 393.



- (20) Eaton, P. E.; Jobe, P. G. *Synthesis* **1983**, 796.
- (21) Fairweather, K. A.; Mander, L. N. *Org. Lett.* **2006**, 8, 3395.
- (22) Miyaoka, H.; Okubo, Y.; Muroi, M.; Mitome, H.; Kawashima, E. *Chem. Lett.* **2011**, 40, 246.
- (23) Leyendecker, F.; Jesser, F. *Tetrahedron Lett.* **1980**, 21, 1311.
- (24) Cook, B. W.; Miller, R. G. J.; Todd, P. F. *J. Organomet. Chem.* **1969**, 19, 421.
- (25) Woodward, S. *Chem. Soc. Rev.* **2000**, 29, 393.
- (26) Posner, G. H.; Whitten, C. E. *Tetrahedron Lett.* **1973**, 14, 1815.
- (27) Posner, G. H.; Whitten, C. E.; Sterling, J. J. *J. Am. Chem. Soc.* **1973**, 95, 7788.
- (28) Bertz, S. H.; Eriksson, M.; Miao, G.; Snyder, J. P. *J. Am. Chem. Soc.* **1996**, 118, 10906.
- (29) Tsuda, T.; Yazawa, T.; Watanabe, K.; Fujii, T.; Saegusa, T. *J. Org. Chem.* **1981**, 46, 192.
- (30) Bluthe, N.; Malacria, M.; Gore, J. *Tetrahedron Lett.* **1983**, 24, 1157.
- (31) Bluthe, N.; Malacria, M.; Gore, J. *Tetrahedron* **1984**, 40, 3277.
- (32) Bluthe, N.; Malacria, M.; Gore, J. *Tetrahedron Lett.* **1982**, 23, 4263.
- (33) Prestwich, G. D.; Solheim, B. A.; Clardy, J.; Pilkiewicz, F. G.; Miura, L.; Tanis, S. P.; Nakanishi, K. *J. Am. Chem. Soc.* **1977**, 99, 8082.
- (34) Schubert, M.; Metz, P. *Angew. Chemie Int. Ed.* **2011**, 50, 2954.
- (35) Anagnostaki, E. E.; Zografos, A. L. *Org. Lett.* **2013**, 15, 152.
- (36) Barriault, L.; Deon, D. H. *Org. Lett.* **2001**, 3, 1925.
- (37) Dillen, J. L. M. *J. Org. Chem.* **1984**, 49, 3800.
- (38) Julia, S. A.; Eschenmoser, A.; Heusser, H.; Tarköy, N. *Helv. Chim. Acta* **1953**, 36, 1885.
- (39) Kuo, F.; Fuchs, P. L. *Synth. Commun.* **1986**, 16, 1745.
- (40) Numbering matches the isocycloamphilectane scaffold, i.e. 7, 20-diisocyanoadociane.

- (41) Berrue, F.; Kerr, R. G. *Nat. Prod. Rep.* **2009**, *26*, 681.
- (42) Heckrodt, T. J.; Mulzer, J. *Top. Curr. Chem.* **2005**, *244*, 1.
- (43) Newton, C. G.; Sherburn, M. S. *Nat. Prod. Rep.* **2015**, *32*, 865.
- (44) Newton, C. G.; Drew, S. L.; Lawrence, A. L.; Willis, A. C.; Paddon-Row, M. N.; Sherburn, M. S. *Nat. Chem.* **2015**, *7*, 82.
- (45) During the preparation of this dissertation a report on the synthesis of various amphilectane and serrulatane diterpenoids described the synthetic divergence to pseudopterosins aglycons of type **3.70**, **3.71** and **3.72** based on an unselective cyclization to both C1 epimers: Yu, X.; Su, F.; Liu, C.; Yuan, H.; Zhao, S.; Zhou, Z.; Quan, T.; Luo, T. *J. Am. Chem. Soc.* **2016**, *138*, 6261.
- (46) Numbering matches pseudopterosin structures.
- (47) Yadav, J. S.; Bhasker, E. V.; Srihari, P. *Tetrahedron* **2010**, *66*, 1997.
- (48) Corey, E. J.; Carpino, P. *J. Am. Chem. Soc.* **1989**, *111*, 5472.
- (49) Kocienski, P. J.; Pontiroli, A.; Qun, L. *J. Chem. Soc. Perkin Trans. 1* **2001**, 2356.
- (50) Arp, F. O.; Fu, G. C. *J. Am. Chem. Soc.* **2005**, *127*, 10482.
- (51) Do, H.-Q.; Chandrashekar, E. R. R.; Fu, G. C. *J. Am. Chem. Soc.* **2013**, *135*, 16288.
- (52) Love, B. E.; Jones, E. G. *J. Org. Chem.* **1999**, *64*, 3755.
- (53) Draper, A. L.; Heilman, W. J.; Schaefer, W. E.; Shine, H. J.; Shoolery, J. N. *J. Org. Chem.* **1962**, *27*, 2727.
- (54) Hoke, M. E.; Brescia, M.-R.; Bogaczyk, S.; DeShong, P.; King, B. W.; Crimmins, M. T. *J. Org. Chem.* **2002**, *67*, 327.
- (55) Beckwith, A. L. J.; Phillipou, G. *Aust. J. Chem.* **1976**, *29*, 1277.
- (56) Ainsworth, C. *Org. Synth.* **1959**, *39*, 27.

(57) Wu, Z.; Li, Y.; Cai, Y.; Yuan, J.; Yuan, C. *Bioorg. Med. Chem. Lett.* **2013**, *23*, 4903.

**CHAPTER 4:**  
**EXPLORING A PHENANTHRENONE REDUCTION APPROACH**  
**TO 7,20-DIISOCYANOADOCIANE**

#### **4.1 Introduction**

The goal of efficient access to 7,20-diisocyanoadociane (**4.5**, DICA) continued to drive the project. An allylic oxidation/hydrosilylation/Robinson annulation design, which intended to establish all the functionality for succinct ring closure, was not successfully executed (Scheme 4.1, see Chapter 2 for more detail), but became the foundation for a new idea. Several rounds of evolution were realized during the following phenanthrenone-based Birch reduction approach towards DICA. A study on the Birch reduction of phenanthrenone substrates, with particular attention paid to stereochemistry is described.

#### **4.2 Entry into a Phenanthrenone Reduction Route**

Robinson annulation onto decalone **4.1** into tricyclic enone **4.2** was intended to lead to bis-enone **4.3**. This bis-enone, upon two electron reduction,<sup>1</sup> would create Corey's dione **4.4**, a direct precursor to DICA.<sup>2</sup> Problems executing the Robinson annulation led to a reevaluation of strategy. Tricyclic enone of the type **4.3** was still considered an attractive target, but the fragile triethylenoxysilane moiety needed replacing. Thus, the synthetically equivalent molecule **4.6**, bearing instead a protected alcohol, was considered. The same reductive coupling of bis-enone **4.3** to **4.4** was still intended as a final ring closure. Although the proposed elaboration of protected alcohol **4.6** to enoxysilane **4.7** would most likely require more steps than directly having the enoxysilane, as in **4.2**, this substrate opened up a highly attractive Birch reduction



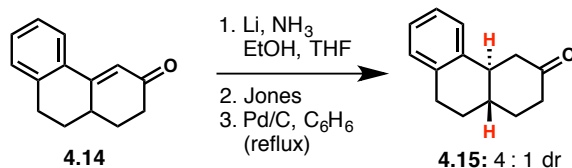
within it, an enone and aromatic functionality, both of which can be reduced under dissolving metal conditions. With excess metal in ammonia and alcohol, the enone is predicted to reduce to enolate **4.10**.<sup>3,4</sup> Subsequent protonation and ketone reduction affords equatorial alcohol **4.11**.<sup>5</sup> The anisole moiety is then cleaved and isomerized to enone **4.13** in classic Birch reduction fashion.<sup>6,7</sup>

With the general course of the reaction rationalized, the issue of stereochemistry needed to be addressed. The literature revealed four conflicting reports on the stereochemical outcomes of conjugated phenanthrenone reductions with dissolving metals. Two reports of a phenanthrenone enone-reduction claim *trans*-selectivity,<sup>8,9</sup> while the other two claim a *cis*-selective reduction (Scheme 4.3A).<sup>10,11</sup> Either small differences in Li/NH<sub>3</sub> reduction conditions change the selectivity between *cis* and *trans*, or two groups are correct in their stereochemical assignment, while the other two are not. Interestingly, both Jacquesy<sup>10</sup> and Eisenbraun<sup>8</sup> reduce the identical, unsubstituted phenanthrenone **4.14** and reported different results under essentially identical conditions; one group claims to have performed a *trans*-reduction, the other group claims a *cis*-reduction. Support for the groups claiming a *trans*-reduction can be found in the reduction of octalone **4.21** to **4.22** or well preceded steroid A/B ring reductions both of which give excellent *trans*-selectivity (Scheme 4.3B).<sup>3,12</sup> Faced with these literature discrepancies, the best course forward was to evaluate the desired system on route to DICA and determine the stereochemistry independently.

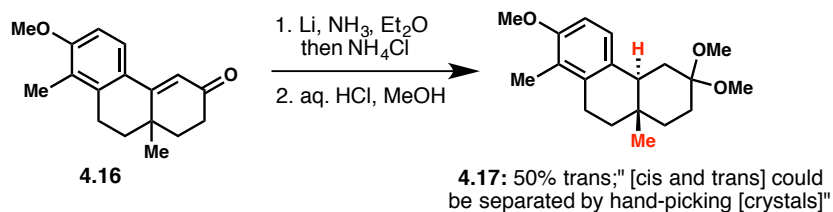
## Scheme 4.3 The stereochemical course of phenanthrenone Birch reductions.

### A. Conflicting literature results of phenanthrenone dissolving metal reductions

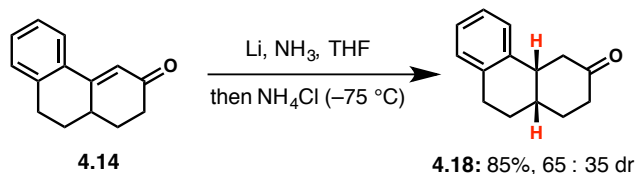
*Jacquesy, 1977*



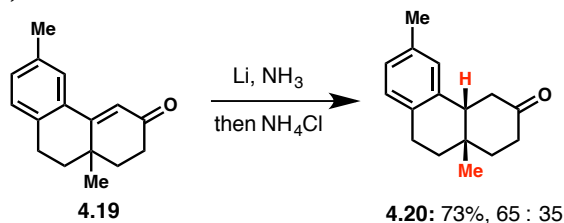
*Robinson, 1957*



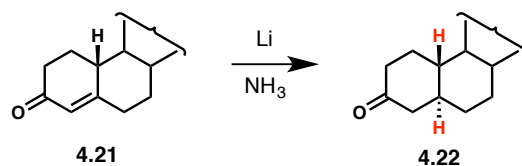
*Eisenbraun, 1988*



*Chatterjee, 1982*



### B. Well-established stereocontrol of a dissolving metal reduction



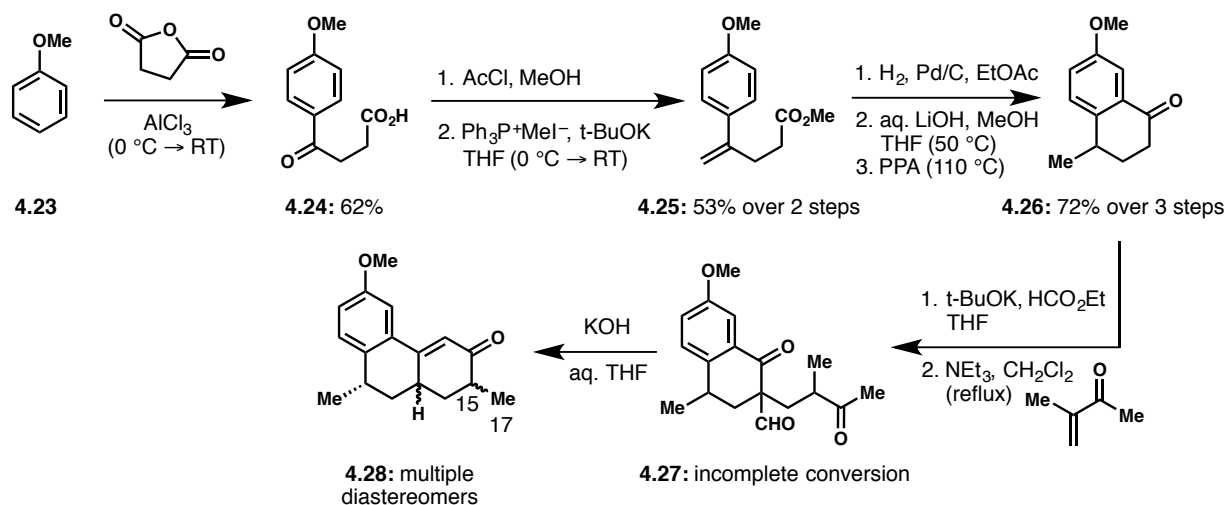
## 4.3 Results from a Phenanthrenone Reduction Approach

### 4.3.1 Phenanthrenone Synthesis and Confirmation of Basic Birch Reactivity

Phenanthrenones are typically accessed from a tetralone precursor, and the preparation of **4.28** was no different. Although shorter and more elegant syntheses of tetralone **4.26** could be

imagined, a literature report sufficed for exploratory chemistry.<sup>13</sup> Starting from anisole, a six-step sequence was completed to afford multi-gram quantities of tetralone **4.26** on a first pass (Scheme 4.4). A Robinson annulation of **4.26** with methyl isopropenyl ketone was implemented to generate the desired phenanthrenone bearing the required C17 methyl group.<sup>14</sup> Experimentally, the corresponding  $\beta$ -ketoaldehyde of **4.26** performed poorly in a conjugate addition with methyl isopropenyl ketone. Although incomplete conversion was observed, the crude product was treated with aqueous base to trigger the subsequent cyclization. Disappointingly, the obtained phenanthrenone **4.28** was isolated as a mixture of diastereomers, both at the ring fusion and at the C15 methyl-bearing stereogenic center. To better understand this system the C17 des-methyl analog of **4.28**, **4.29** was targeted (Scheme 4.5).

**Scheme 4.4** A tetralone synthesis and first phenanthrenone derivative.

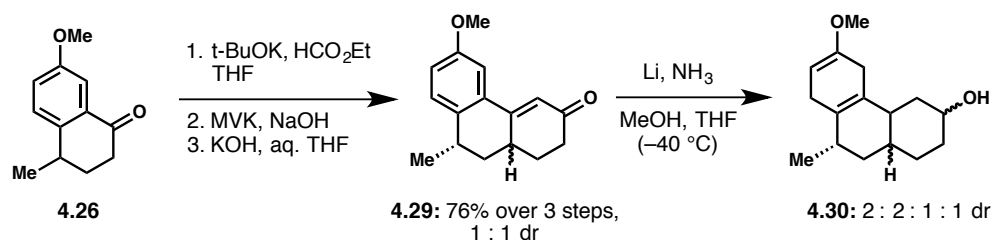


The Robinson annulation of **4.26** with methyl vinyl ketone performed better than with methyl isopropenyl ketone and generated phenanthrenone **4.29** in good yield; however, it too lacked stereocontrol (Scheme 4.5). Regardless, it was anticipated that the correct diastereomer could be separated at a later stage. Treating **4.29** with excess Li/NH<sub>3</sub> accomplished the desired reduction



to **4.30**. Therefore the proposed one-pot enone and aromatic reduction was successful! Although the desired reduction was accomplished, stereocontrol was problematic. The original 1 : 1 dr in **4.29** was further aggravated to a 2 : 2 : 1 : 1 mixture. This route could have been pursued further, but an approach more strongly emphasizing stereocontrol was prioritized.

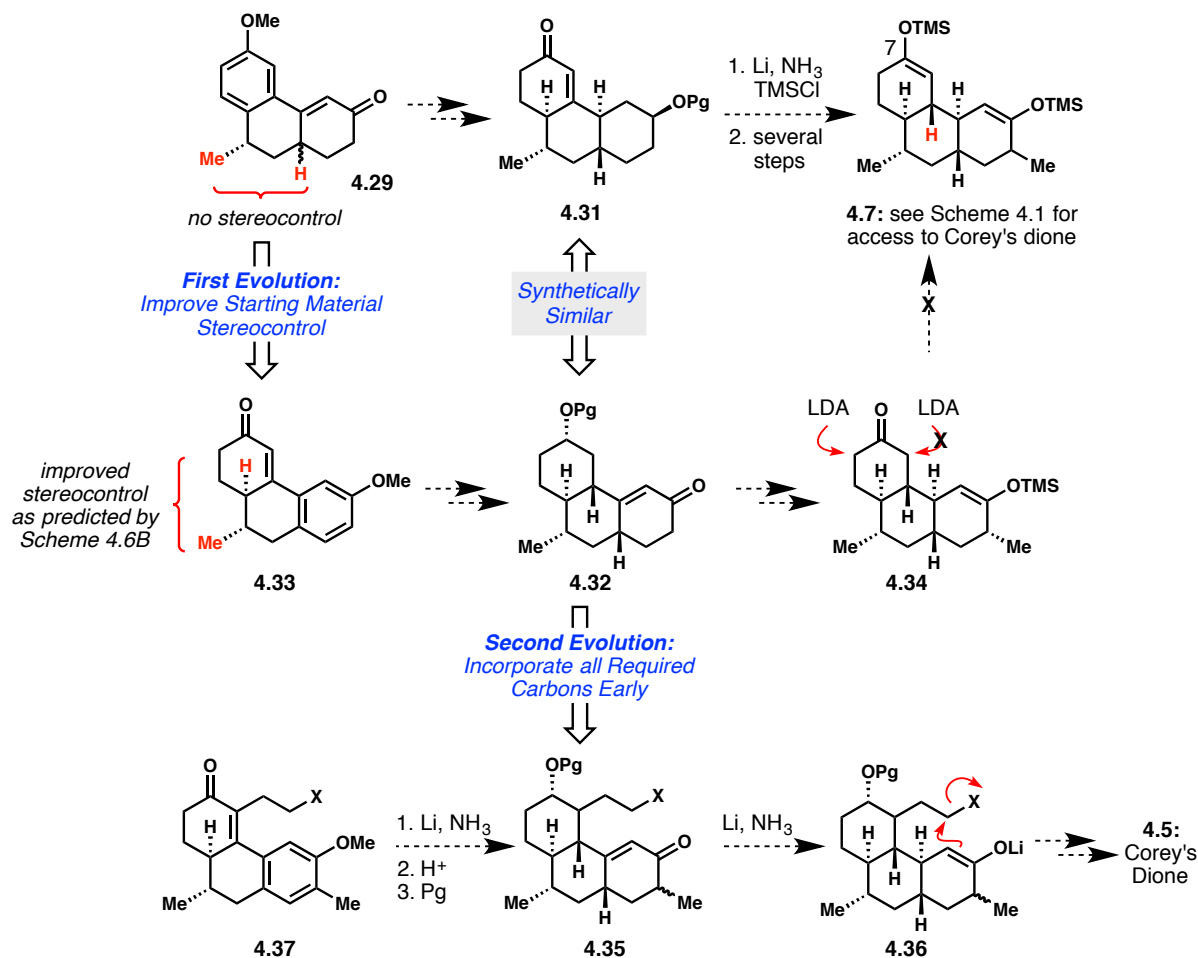
**Scheme 4.5** A successful phenanthrenone Birch reduction.



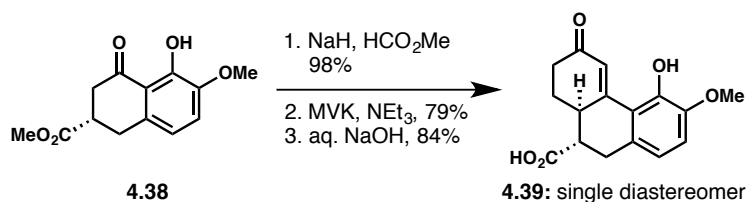
While stereocontrol in the synthesis of **4.30** was lacking, its reactivity to produce a structure of type **4.31** was not in doubt (Scheme 4.6A), but the poor diastereocontrol obtained by Robinson annulation and from the Birch reduction triggered reevaluation. Better control in the Robinson annulation was imagined if the two stereocenters were vicinal, as in **4.33**. Literature precedent supported this proposal as **4.39** could be prepared as a single diastereomer (Scheme 4.6B).<sup>15</sup> Subsequent Birch reduction of **4.33** would lead to intermediate **4.32**, a synthetically similar molecule to the originally targeted **4.31**. In a forward sense, however, cyclohexenone **4.34** was expected to be problematic. At a stage in which the C7 enoxysilane<sup>14</sup> would be installed, such as on **4.34**, correct regiochemical enolization would be unfavored, barring chiral base deprotonation.<sup>16</sup> This problem of selectivity could be addressed by changing the fourth ring closure to an alkylation proceeding through enolate **4.36**. This second evolution meant an elaborated vinyl ketone would be needed to generate **4.37** by Robinson annulation.

**Scheme 4.6.** A redesigned phenanthrenone Birch reduction route.

**A. The evolution of a phenanthrenone based Birch reduction**



**B. Stereochemical precedence for a Robinson annulation**

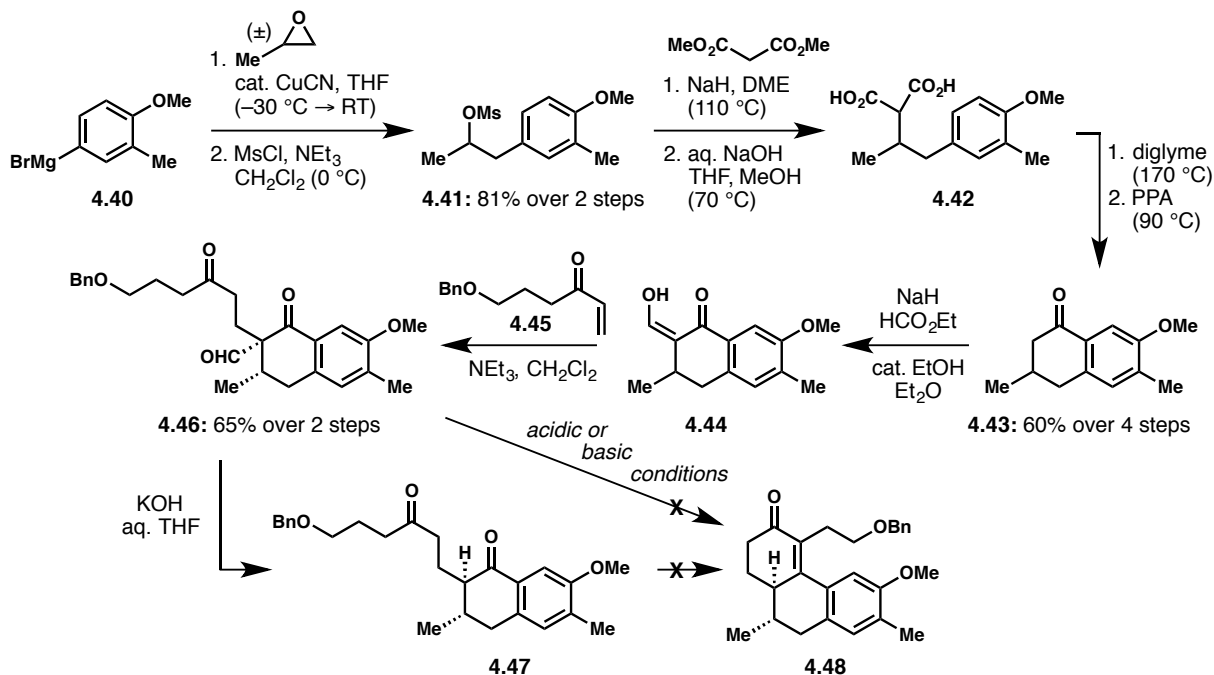


**4.3.2 A Reworked Phenanthrenone Approach Emphasizing Stereocontrol**

The synthesis of a phenanthrenone of the type **4.37** and evaluation of the subsequent Birch reduction became the next objective. A synthesis of its tetralone precursor was conceived

of de novo, and started by opening of propene oxide with aromatic Grignard reagent **4.40**, followed by mesylation (Scheme 4.7). A malonic ester synthesis and carboxylic acid cyclization of **4.42** to **4.30** required optimization, but eventually led to significant quantities of tetralone **4.43**. An asymmetric synthesis of tetralone **4.43** would simply require enantioenriched propylene oxide.<sup>17</sup> Formylation of **4.43** was complicated by aromatization if not monitored. Michael addition of **4.44** with **4.45** proceeded smoothly. Disappointingly the ring closing aldol condensation was problematic and cyclization was not observed. The only products were those of deformylation under basic conditions and recovered starting material.

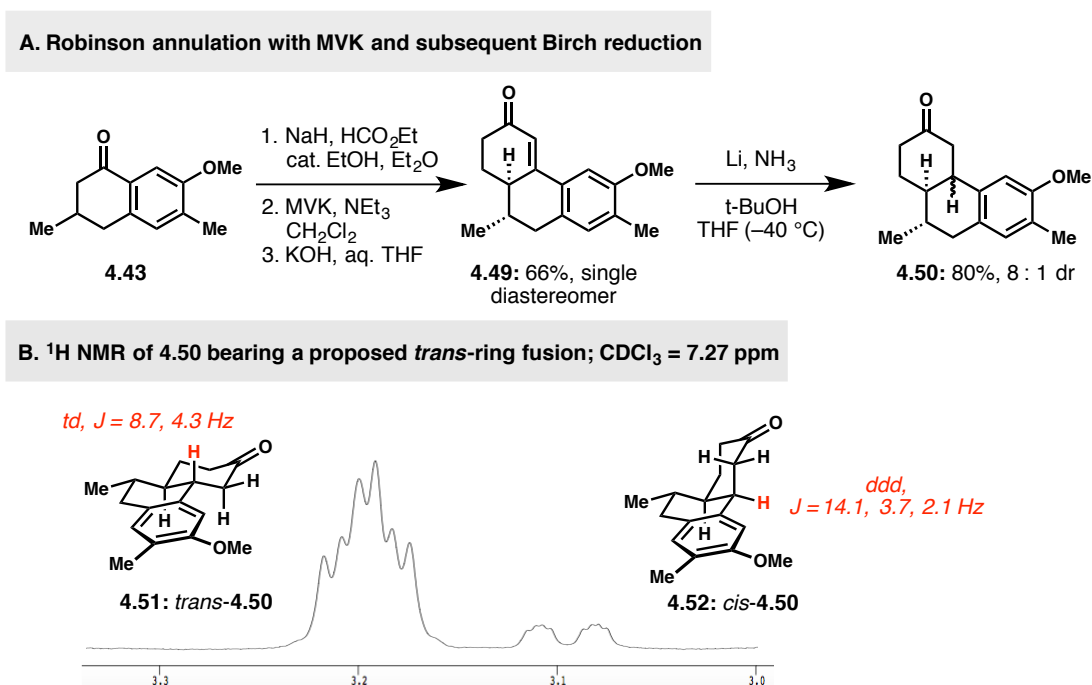
**Scheme 4.7** First pass at the new phenanthrenone-based synthesis of DICA.



Although it was attractive to bring all carbons in early with vinyl ketone **4.45**, the problem of phenanthrenone reduction stereochemistry and reactivity still needed addressing. At this point, the elaborated **4.45** was replaced with methyl vinyl ketone, the product of which underwent the Robinson annulation efficiently to afford **4.49** (Scheme 4.8A). Birch reduction of

**4.49** with Li/NH<sub>3</sub> in the presence of excess t-BuOH provided only reduction of the enone to ketone **4.50**; a curiosity when compared to the earlier reduction successes of **4.29** (Scheme 4.5). The advantage at this point was that the relative stereochemistry of enone reduction could be obtained. Analysis of the <sup>1</sup>H NMR determined that the major species in an 8 : 1 mixture of **4.50** bore the *trans*-ring junction (Scheme 4.8B).

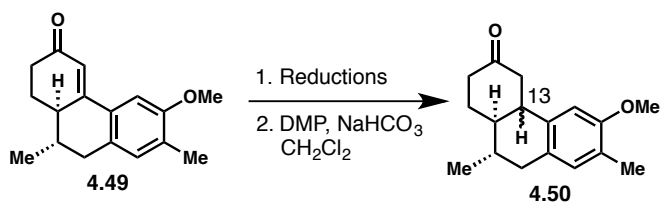
**Scheme 4.8** A selective phenanthrenone reduction with Li/NH<sub>3</sub>.



In stark contrast to the previously run Birch reduction (Scheme 4.5), several attempts at reducing the aromatic **4.49** were not successful. Even under the influence of a large excess of Li and t-BuOH, neither the ketone nor the aromatic was reduced. In addition to t-BuOH, isopropanol and ethanol were also screened as proton sources for attempted aromatic cleavage. Both ethanol and isopropanol reduced the ketone, but still did not reduce the aromatic. Phenanthrenone **4.49**'s limited solubility in THF, insolubility in diethyl ether and insolubility in ammonia made evaluating this reaction difficult. The lack of solubility may have also

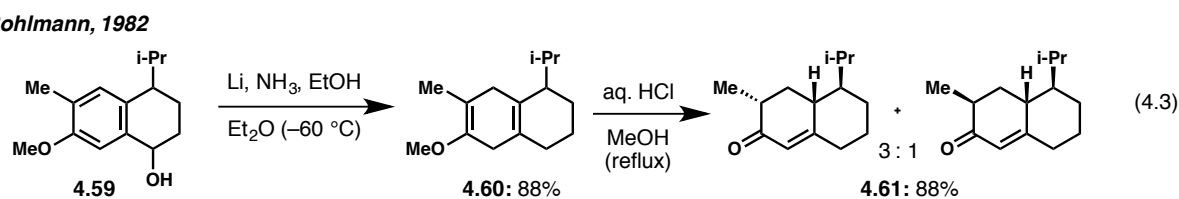
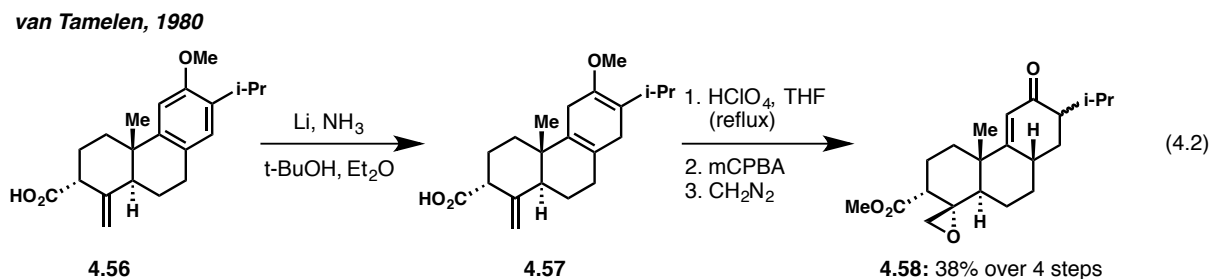
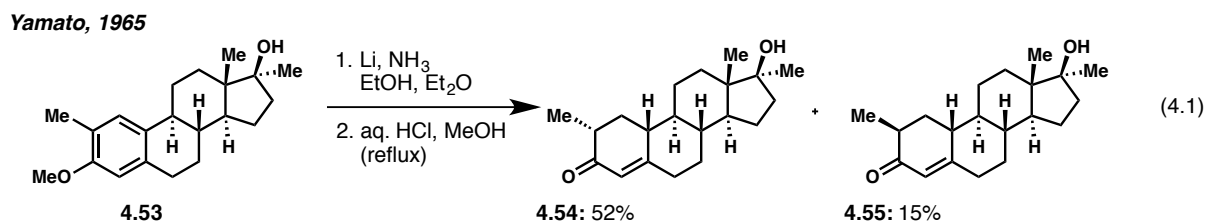
contributed to the absence of reactivity. A further screen of Mg in methanol and hydrogenation were screened to possibly improve setting the C13 stereocenter, and unsurprisingly did not reduce the aromatic ring. The lack of aromatic reduction in **4.49** is difficult to explain because the literature contains many examples of Birch reductions of 1,2,4,5-tetrasubstituted anisoles; select examples are provided in equations 4.1-4.3.<sup>18-20</sup> The issue of difficult aromatic reductions has been addressed before, most notably by the procedures of Benkeser using alkylamine solvent.<sup>21</sup> These conditions have the added advantage of being performed above the boiling point of ammonia at  $-33\text{ }^{\circ}\text{C}$ .<sup>22</sup>

**Table 4.1** Small screen of reduction conditions in attempting to fully reduce **4.49**.



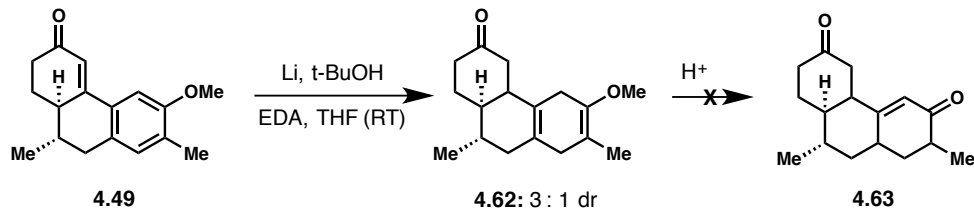
Entry	Reductions	4:51 <i>trans</i> <sup>a</sup>	4:52: <i>cis</i> <sup>a</sup>
1	Li, NH <sub>3</sub> with EtOH, i-PrOH or t-BuOH THF ( $-40\text{ }^{\circ}\text{C}$ )	variable	
2	Mg, cat. HgCl <sub>2</sub> , MeOH	1.1	1
3	H <sub>2</sub> , Pd/C, EtOAc	1	3
4	H <sub>2</sub> , PtO <sub>2</sub> , EtOAc	1	1.7

<sup>a</sup> as proposed initially by NMR data (see Table 4.3 for more information)



Stronger and higher temperature reducing agents were evaluated for the recalcitrant reduction of **4.49**. Classic Benkeser reductions in methylamine and ethylamine were not pursued because these controlled substances are inconvenient to obtain. Alternative Benkeser reductions were tried instead. Using Ca in ethylene diamine, propylamine, and THF provided only starting material.<sup>23</sup> Li or Na in neat HMPA caused decomposition of material.<sup>24</sup> Modified ethylenediamine based conditions eventually were successful in reducing the aromatic ring (Scheme 4.9). Several literature reports include propylamine as a solvent; however, it was found not necessary for effective reduction.<sup>25,26</sup> Although the aromatic moiety was reduced, the ketone was not! Presumably, ethylenediamine condensed either onto the enone or the transiently generated ketone, thereby masking the carbonyl reduction.

**Scheme 4.9** Successful ethylene diamine based reduction of phenanthrenone **4.49**.



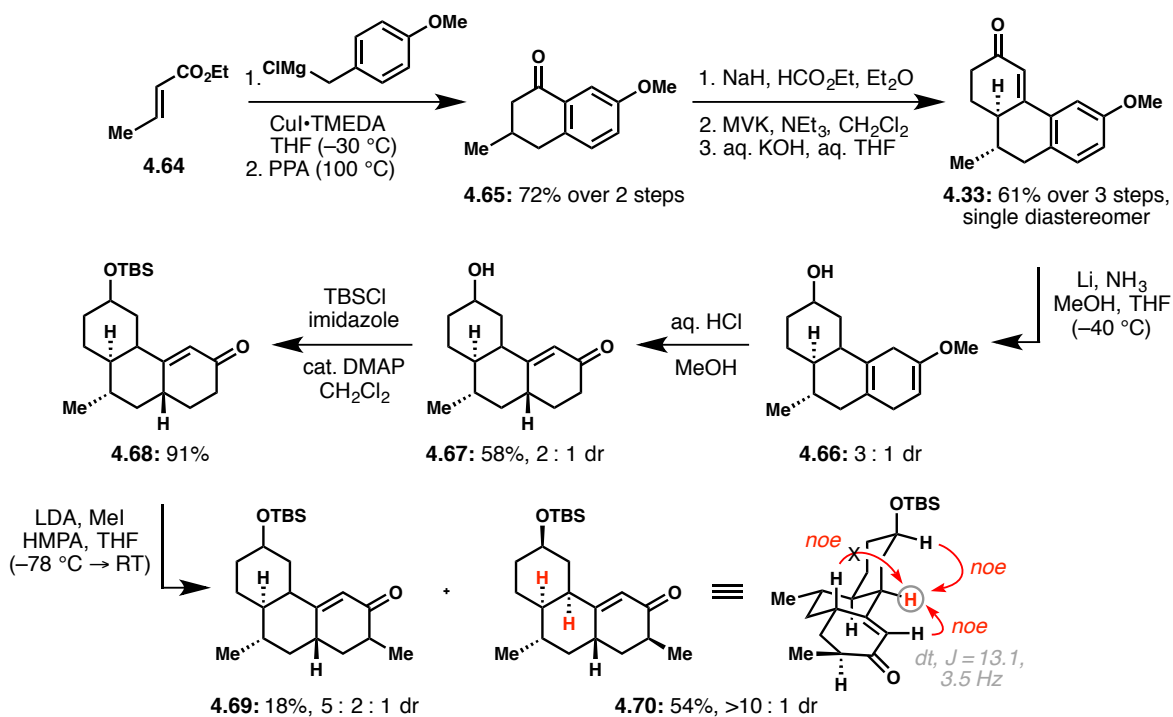
Having successfully reduced the aromatic, the next step was hydrolysis and isomerization to the conjugated enone **4.63** (Scheme 4.9). Standard conditions of aqueous HCl in methanol failed. Hydrolysis of the vinyl ether to the corresponding ketone was achieved with mild acid, but attempts at isomerizing the skipped enone into conjugation under acid or basic conditions were once again unsuccessful. Complexity in the lack of reduction of the aromatic under standard Birch conditions, problems with low mass recovery in the ethylene diamine-based reductions, and difficulties in enone isomerization were all attributed to the C17 methyl group.<sup>14</sup> An approach that installs this methyl later was therefore pursued.

#### 4.3.3 Identifying a Conflicting Stereochemical Outcome

To examine whether the singular C17 methyl group may inhibit aromatic reduction, the des-methyl analog of **4.49** was prepared (Scheme 4.10). The TMEDA-ligated p-methoxybenzylcopper reagent was added to ethyl crotonate and the corresponding ester treated with polyphosphoric acid to generate tetralone **4.65** via a slightly modified literature procedure.<sup>27</sup> Robinson annulation with methyl vinyl ketone generated **4.33**, and in analogy to **4.49**, also as a single diastereomer. Treatment of this phenanthrenone with excess Li/NH<sub>3</sub>, in the presence of excess methanol in THF, cleanly reduced the system down to alcohol **4.66**. Although the hydrolysis and enone isomerization to **4.67** was difficult, it could be accomplished with aqueous acid while carefully monitoring by TLC. The obtained 2 : 1 mixture of diastereomers was

silylated and then methylated via the cross-conjugated dienolate. Two sets of fractions were separated, one consisted of a mixture of diastereomers and assigned as structure **4.69** and one was highly enriched in the major diastereomer. As judged by NMR analysis, the major diastereomer was consistent with structure **4.70**, bearing a *cis*-ring fusion. This ring fusion came from reduction of phenanthrenone **4.33** using Li/NH<sub>3</sub> and is in direct contradiction to the previously proposed stereochemistry of reduction (Scheme 4.8)! It is impossible for the assignment of both **4.51** and **4.70** to be correct. This discrepancy required further clarification.

**Scheme 4.10** A des-methyl phenanthrenone undergoes successful Birch reduction.



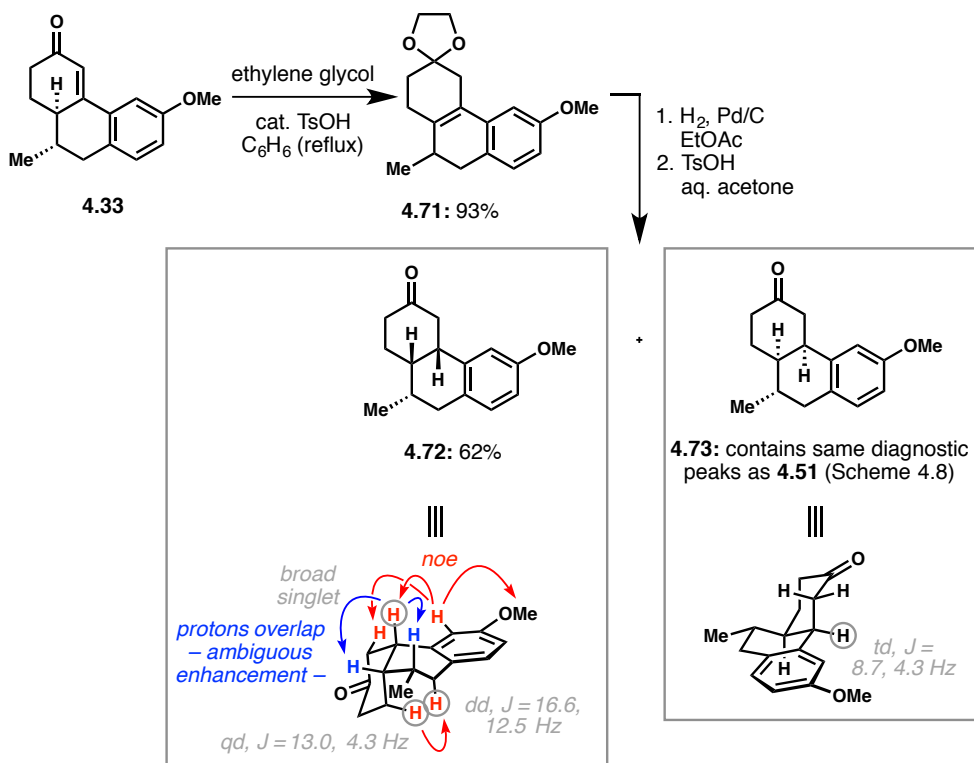


#### 4.3.4 Settling the Phenanthrenone Birch Reduction Stereochemical Conundrum

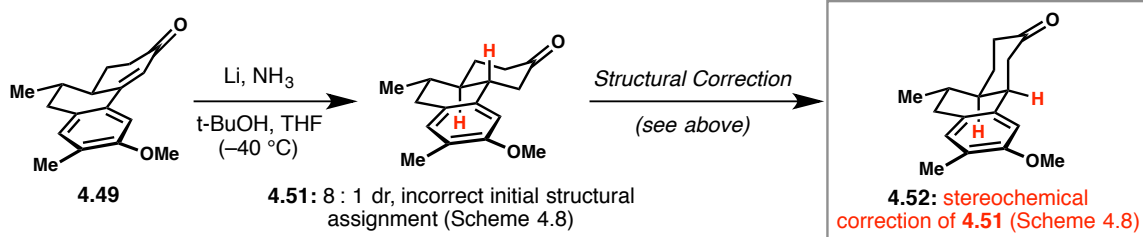
An empirically guided approach to determining the stereochemical course of phenanthrenone Birch reductions was sought. Phenanthrenone **4.33** was chosen as the substrate to evaluate since large quantities were available at the time and **4.49** was not. The C17 methyl<sup>14</sup> group was considered too distal to be considered relevant to the herein described experiments. Ketalization of **4.33** migrated the double bond out of its original position to afford **4.71** (Scheme 4.11A). Hydrogenation of alkene **4.71** over Pd/C provided a 4 : 1 mixture of diastereomers. Since surface catalysts strongly enforce *cis*-reduction, the two diastereomers obtained arose from hydrogen delivery to the face opposite of the methyl and to the same face as the methyl. Acidic ketal deprotection yielded two separable ketones. The major constituent was identified as structure **4.72**. The minor isomer was therefore structure **4.73**. This material had an almost identical <sup>1</sup>H NMR spectrum to the major product obtained from Birch reduction of **4.49** (Scheme 4.5). The major isomer upon reduction of **4.49** was thus corrected to **4.52** (Scheme 4.11B).

**Scheme 4.11** Alkene hydrogenation to determine the stereochemical course of the Birch reduction in conjugated phenanthrenones.

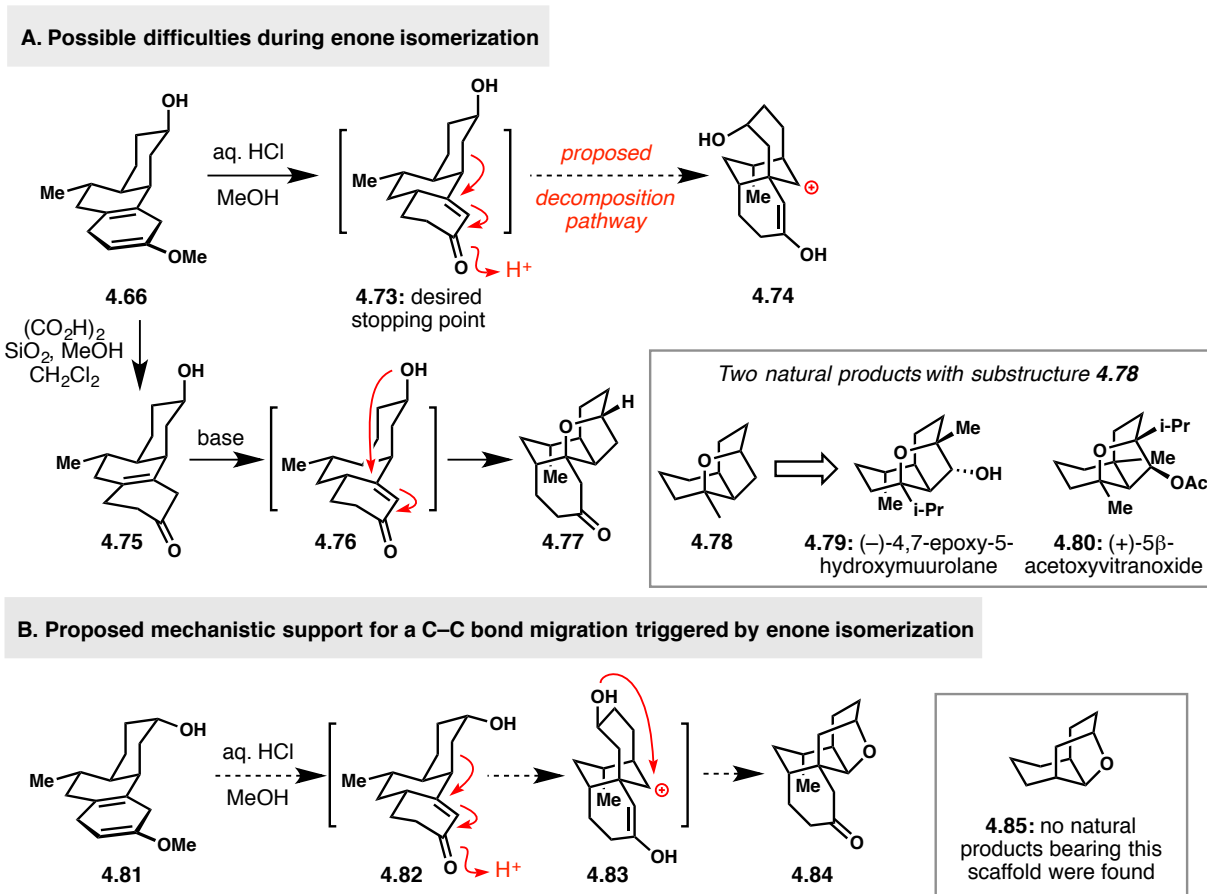
**A. Using *cis*-selective hydrogenation to tackle a stereochemical problem**



**B. Stereochemical correction of a previous Birch reaction (Scheme 4.8)**



**Scheme 4.12** Proposed difficulties isomerizing a skipped enone into conjugation.



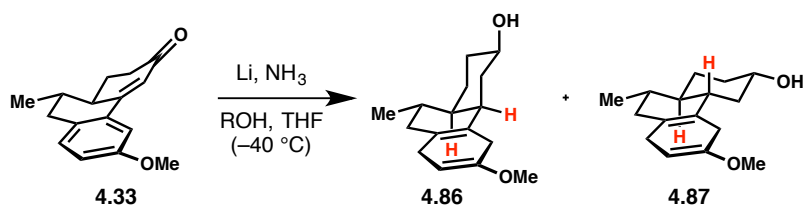
The stereochemical correction of **4.51** to **4.52** allows for a reflection on the conflicting literature reports (Scheme 4.3A). Based on the results described, the stereochemistry of all four reports is highly likely *cis*. Caution should be applied when using the Jacquesy<sup>8</sup> and Robinson<sup>9</sup> examples as precedent since their stereochemical assignment is questionable. In addition, the observed *cis* stereochemistry explains the difficulty experienced in isomerizing enones **4.62** and **4.66** into conjugation (Scheme 4.9 and 4.10). For representative example **4.66**, under acidic conditions, carbonyl protonation may cause C–C bond migration to cation **4.74**, an intermediate with the ability to decompose further (Scheme 4.12A). The migrating bond has good overlap with the enone, and could explain why careful monitoring of the isomerization was essential. A probe for this decomposition pathway could shed light onto its existence. If

migration of the C–C bond is occurring, then the alcohol epimer **4.81** may possibly trap the formed cation and generate tetrahydrofuran **4.84** (Scheme 4.12B). Basic isomerization attempts of **4.75**, which was generated with mild acid from **4.66**, afforded among other compounds, a ketone containing product. Isolation and full characterization was not performed, but NMR of the crude reaction mixture and TLC analysis suggests **4.77** as a possibility. If this tetrahydrofuran formation is occurring, then an oxy-Michael reaction could be envisioned for preparing natural products **4.79**<sup>28</sup> and **4.80**.<sup>29</sup>

#### 4.4 Phenanthrenone Birch Reduction Optimization Attempts

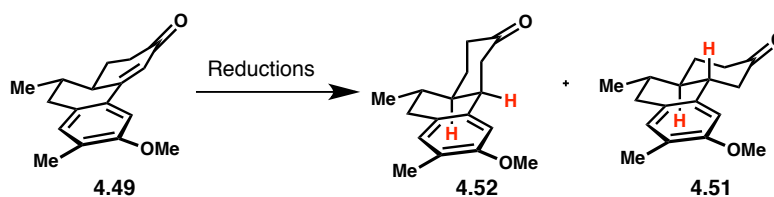
Since the relative stereochemistry of phenanthrenone Birch reductions has now been established as *cis*, altering the selectivity was attempted. An inspiration for this possibility was that the reduction of **4.49** with Li/NH<sub>3</sub> and t-BuOH provided **4.50** in 8 : 1 dr, while the literature examples using EtOH provided 2-3 : 1 *cis/trans* selectivity.

Using phenanthrenone **4.33**, a small screen of alcoholic additives displayed a clear trend between alcohol sterics and *cis/trans* selectivity upon reduction. While methanol provided a 3 : 1 mixture of *cis/trans*, the selectivity for *cis*-reduction increased in correlation with alcohol bulk up to 8 : 1 *cis/trans* with t-BuOH (Table 4.2 and 4.3). This trend can be rationalized by comparing both the *cis*- and *trans*-protonation states (Figure 4.1). Although *trans*-reduction of enones is electronically favored for dissolving metal reductions,<sup>3,4</sup> it is likely that the allylic strain present in *trans*-protonation state **4.88** actually switches inherent favorability to the *cis*-protonation state **4.89** (Figure 4.1A). An additional contribution is that the *cis*-face of the enone is sterically more available (Figure 4.1B). Bulky alcohols push an already *cis*-selective reduction further to **4.91** because of a steric preference to approach from the more open enone face.

**Table 4.2** Impact of alcohol additive on stereochemistry.

Entry	ROH	4.86: cis <sup>a</sup>	:	4.87: trans <sup>a</sup>
1	MeOH	3	:	1
2	EtOH	4	:	1
3	i-PrOH <sup>b</sup>	5	:	1

<sup>a</sup> ratios obtained by crude <sup>1</sup>H NMR <sup>b</sup> MeOH quench

**Table 4.3** Correcting the stereochemical outcome described in Table 4.1.

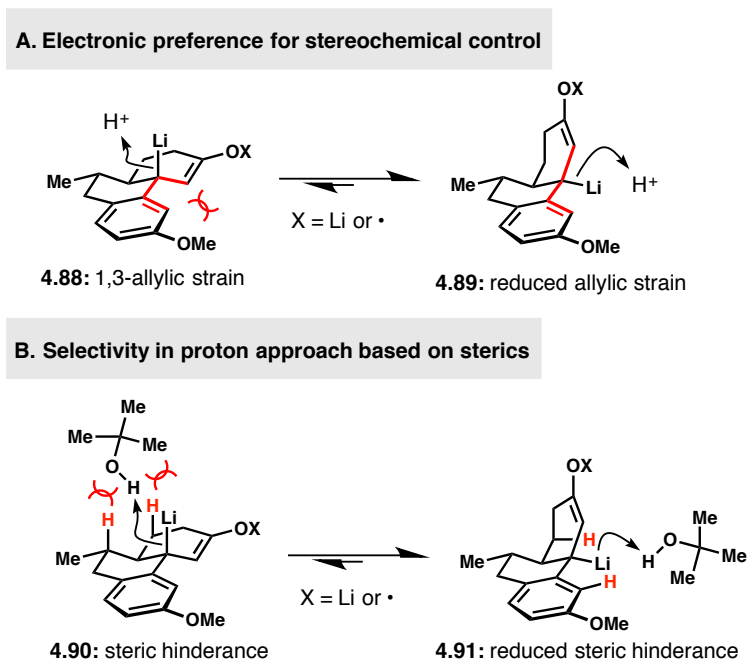
Entry	Reductions	4.52: cis <sup>a</sup>	:	4.51: trans <sup>a</sup>
1	Li, NH <sub>3</sub> , t-BuOH, THF (-40 °C) (see Scheme 4.8)	8	:	1
2	Mg, cat. HgCl <sub>2</sub> , MeOH <sup>b</sup>	1.1	:	1
3	H <sub>2</sub> , Pd/C, EtOAc <sup>b</sup>	1	:	3
4	H <sub>2</sub> , PtO <sub>2</sub> , EtOAc <sup>b</sup>	1	:	1.7

<sup>a</sup> ratios obtained by crude <sup>1</sup>H NMR <sup>b</sup> 2. DMP, NaHCO<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>

Because of the correction in stereochemical assignment (Scheme 4.3.4), the reduction attempts on **4.49**, initially presented in Table 4.1, also require correcting (Table 4.3). Curiously, Mg in methanol provided the best *trans*-selectivity via a single electron reduction pathway providing an almost 1 : 1 *cis/trans* ratio. Hydrogenation catalysts afford the *trans*-isomer as the major product. These hydrogenation conditions were not pursued further towards DICA because of the relatively low selectivity in both enone reduction and subsequent ketone reduction and requirement for an additional aromatic reduction. The ideal of reducing an aromatic substrate in

a single step was still a main objective.

**Figure 4.1** Rationalizing the phenanthrone reduction stereochemistry.



Although it was discouraging that changing alcohol additives in these Birch reductions did not afford the *trans* isomer as the major product, this screen holds interesting insight in a broader sense. The literature on dissolving metal reductions of enones suggest each substrate has an innate selectivity that is solely dependent on its structure and not reaction conditions.<sup>3</sup> *The experiments just presented fully contradict these claims.*<sup>30</sup> Further collaborative investigation using experimental and computational methods could elucidate this curious trend in stereoselectivity, and perhaps lead to greater selectivity in one direction or another. Obvious evaluation of reaction conditions such as metal, temperature, cosolvents and additives were not performed, but are expected, as already shown with Mg in methanol, to have significant impact on diastereoselectivity. Overall, this inquiry would be interesting from a fundamental reactivity standpoint and expand on 50 years of dissolving metal *trans*-reduction dogma.

## 4.5 Conclusions

A rationally derived phenanthrenone Birch reduction was conceived to address problems of the all-*trans* stereochemistry required for a synthetic preparation of DICA. With the literature split on the outcome of such a reduction, an independent evaluation was important not only as a direct entry into perhydrophenanthrene systems, but also as a contribution to clear up ambiguity in the literature. An initial *trans* stereochemical assignment inspired continued efforts in this area. After a successful evolution in approaches addressing the lack of aromatic reduction, the major observed product was found to have a *cis*-ring fusion! Stereochemical assignment obtained by hydrogenation was found to support a *cis*-ring fusion, thereby correcting the previously assigned *trans* stereochemistry and indicating which literature precedent is correct in their assignment. In attempts to switch the stereochemical reduction to *trans*, a positive correlation between alcohol size and improved *cis*-selectivity were observed. The phenanthrenone reduction approach could still be continued using a mostly *trans*-selective hydrogenation, but was replaced by a better approach to DICA (see Chapter 5).

## 4.6 Experimental Procedures

### Purifications –

Solvents: Dry tetrahydrofuran (THF), diethyl ether (Et<sub>2</sub>O), dichloromethane (CH<sub>2</sub>Cl<sub>2</sub>), toluene, benzene (C<sub>6</sub>H<sub>6</sub>) and methanol (MeOH) were obtained by passing commercially available formulations through activated alumina columns. tert-Butyl alcohol (t-BuOH) was purified by distillation over CaH<sub>2</sub>.

Amines: Diisopropylamine (i-Pr<sub>2</sub>NH), triethylamine (NEt<sub>3</sub>), N,N'-dimethylpropylene urea (DMPU), propylamine (n-PrNH<sub>2</sub>), N,N,N',N'-tetramethylethylene-1,2-diamine (TMEDA) and hexamethylphosphoramide (HMPA) were purified by distillation from CaH<sub>2</sub>. Ethylenediamine (EDA) was purified by distillation from sodium metal.

Chlorides: Acetyl chloride (AcCl) was purified by distillation from phosphorus pentachloride.

Metals: Copper(I) cyanide (CuCN) was purified by refluxing in degassed water, filtering, washing with ethanol, diethyl ether and drying under high vacuum. Copper(I) iodide (CuI) was purified by Soxhlet extractor with THF then drying the solid under high vacuum.

Miscellaneous: Methyl vinyl ketone (MVK) and (±)-propylene oxide were purified by distillation.

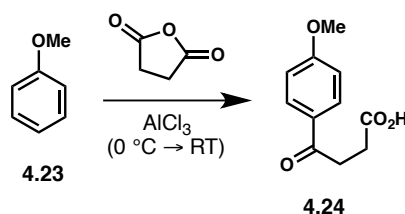
**Titration** – Alkyl lithium reagents were titrated using 2,6-di-(tert-butyl)-4-methylphenol (BHT) as the sacrificial proton source and fluorene as an indicator in THF or using diphenylacetic acid in THF. Grignard reagents were titrated using salicylaldehyde phenylhydrazone in THF.<sup>31</sup>



**Reaction Setup** – All reactions were carried out under an argon atmosphere with dry solvents under anhydrous conditions, unless otherwise noted. Argon balloons were the sole inert atmosphere used. Reactions run at an ambient temperature of 20–25 °C are designated as room temperature. Yields refer to chromatographically and spectroscopically homogeneous materials, unless otherwise stated.

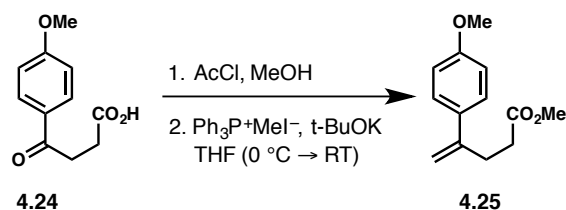
**Analysis** – Thin layer chromatography was performed on 0.25 mm EMD glass-backed TLC plates impregnated with a fluorescent dye and visualized with UV light and KMnO<sub>4</sub> in K<sub>2</sub>CO<sub>3</sub>/NaOH/water or *p*-anisaldehyde in ethanol/aqueous H<sub>2</sub>SO<sub>4</sub>/AcOH and heat as a developing agent. Forced flow (flash) chromatography was performed on EMD Silica 60, mesh 0.04-0.063 silica gel. NMR spectra were recorded on Bruker 500 MHz instrument, obtained at 298 K unless otherwise noted and calibrated to residual undeuterated solvent as an internal reference. Chemical shifts are reported in ppm with the following abbreviations to explain multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, quin = quintuplet, sext = setet, sep = septet, bs = broad signal, m = multiplet. All coupling constants are apparent *J* values measured at the indicated field strengths. FT-IR spectra were recorded on a Perkin-Elmer spectrum RX1 spectrometer. High-resolution mass spectra (HRMS) were recorded on a H2Os LCT Premier spectrometer using ESI-TOF (electrospray ionization-time of flight). Melting points were measured on a MEL-TEMP II capillary apparatus and stand uncorrected.

**Carboxylic Acid 4.24** [Adapted from the literature]<sup>32</sup>



A 500 mL 3-neck round bottom flask, fitted with an overhead stirrer, a septum and a pressure release tube submerged in an aqueous NaOH trap, was charged with 75 mL (0.690 mmol) **4.23**, 10.2 g (0.102 mmol) succinic anhydride and cooled in an ice bath. In one portion 27.0 g (0.202 mmol)  $\text{AlCl}_3$  was added. The reaction was stirred at  $0\text{ }^\circ\text{C}$  for 2 hours then the bath was removed. After 2 hours the contents were poured into 250 mL conc. HCl cooled to  $0\text{ }^\circ\text{C}$  under overhead stirring. The pink solution was stirred at room temperature for 1 hour, filtered over a fritted funnel and the contents washed with 200 mL water and thrice with 100 mL water. The solid was azeotropically dried with toluene then crystallized from 300 mL toluene and recrystallized from 100 mL toluene to afford 13.4 g (62%) **4.24**. The spectral data was identical to the literature.<sup>33</sup>

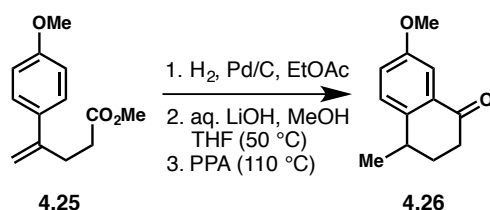
**Ester 4.25** [Adapted from the literature]<sup>13</sup>



A 250 mL round bottom flask was charged with 13.13 g (63.1 mmol) **4.24**, 150 mL MeOH and 5.0 mL (70.3 mmol) acetyl chloride. After stirring for 18 hours at room temperature, 80 mL sat.  $\text{NaHCO}_3$  was added and 60 mL volatiles removed in vacuo. The solution was diluted with 80 mL water and extracted with 200 mL  $\text{Et}_2\text{O}$ . The organic phase was washed with 50 mL water. All aqueous phases were combined and extracted with 50 mL  $\text{Et}_2\text{O}$ . The organic phase was washed

with 20 mL water and combined with the previous organic phase. This process was repeated a total of four times. The combined organic layer was washed with 25 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. A 1 L round bottom flask containing crude keto-ester, 33 g (81 mmol) methyltriphenylphosphonium iodide and 150 mL THF was cooled in an ice bath and fitted with an addition funnel. A solution of 8.5 g (76 mmol) t-BuOK in 60 mL THF was added dropwise over the course of 15 minutes, the addition funnel washed with 50 mL THF and the reaction stirred for 30 minutes at 0 °C, then 30 minutes after removing the cold bath. After the addition of 100 mL water, 200 mL volatiles were removed and the solution extracted with 200 mL  $\text{Et}_2\text{O}$ , then thrice with 50 mL  $\text{Et}_2\text{O}$ . All organic layers were combined, washed with 30 mL third sat.  $\text{Na}_2\text{S}_2\text{O}_3$ , 30 mL water, 25 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude solid was triturated with 50 mL portions of hexanes and filtered over celite for a total of ~800 mL hexanes. All washings were combined and the volatiles removed in vacuo. The crude oil was purified by column chromatograph (8:1 hexanes/EtOAc) to afford 7.39 g (53% over 2 steps) **4.25** as a yellow wax.

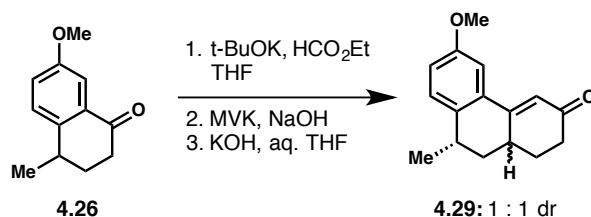
**Tetralone 4.26** [Adapted from the literature]<sup>13</sup>



A 500 mL round bottom flask was charged with 7.39 g (33.5 mmol) **4.25**, 100 mL EtOAc, 1.04 g 10% Pd/C and stirred under a hydrogen balloon. After 15 hours, 5 g Celite was added and the contents filtered over a celite plug with the assistance of fresh EtOAc. All volatiles were removed in vacuo. A 500 mL round bottom flask containing crude ester was stirred with 100 mL

THF/MeOH/water (2:1:1) and 2.58 g (61.5 mmol) LiOH•H<sub>2</sub>O at 50 °C. After 1 hour the reaction was quenched with 40 mL 3M HCl, and extracted with 100 mL, then thrice with 50 mL Et<sub>2</sub>O. The organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. Polyphosphoric acid (70 g P<sub>2</sub>O<sub>5</sub> and 60 g H<sub>3</sub>PO<sub>4</sub>, 100 °C, 9 h) was added to crude acid and heated at 110 °C for 2 hours in a 500 mL round bottom flask. The reaction was partially cooled and 300 g ice added slowly. The solution was extracted 8 times with 50 mL portions CH<sub>2</sub>Cl<sub>2</sub>. The organic layers were combined and washed with 100 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The black residue was purified by column chromatography (10:1→9:1 hexanes/EtOAc) to afford 4.62 g (70% over 3 steps) **4.26** as a yellow oil. The spectral data was identical to the literature.<sup>13</sup>

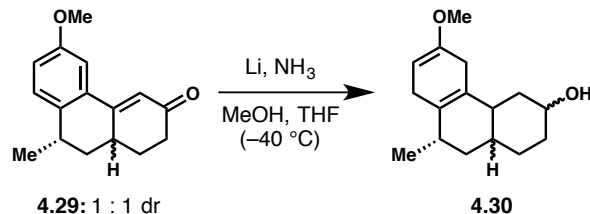
### Phenanthrenone **4.29**



A 25 mL round bottom flask containing 201 mg (1.06 mmol) **4.26** and 0.15 mL (1.86 mmol) ethyl formate in 5 mL THF was cooled in an ice bath and treated with 1.3 mL (1.3 mmol) 1 M t-BuOK/THF. The ice bath was removed upon complete addition and stirring continued for 1 hour. The reaction was treated with 2 mL 6 M HCl and 8 mL water, then extracted 3 times with 5 mL CH<sub>2</sub>Cl<sub>2</sub>. The organic layers were combined, washed with 10 mL half sat. brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude mixture in a 10 mL round bottom flask was stirred with 0.30 mL (3.60 mmol) MVK and 7 mg (0.18 mmol) powdered NaOH (exotherm). After 20 minutes the reaction was diluted with 5 mL sat. NH<sub>4</sub>Cl and extracted twice

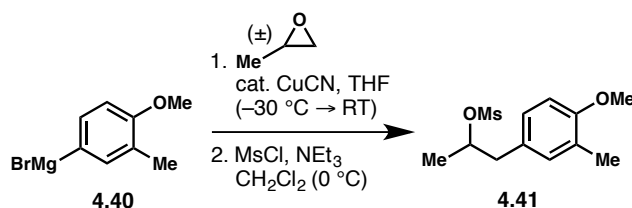
with 5 mL CH<sub>2</sub>Cl<sub>2</sub>. The organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. A 25 mL round bottom flask containing the crude material in 4 mL THF was treated with 210 mg (5.25 mmol) NaOH in 4 mL water at 0 °C. The ice bath was immediately removed and stirring continued for 36 hours. The reaction was quenched with 10 mL half sat. NH<sub>4</sub>Cl and extracted twice with 10 mL CH<sub>2</sub>Cl<sub>2</sub>. The organic layers were combined, washed with 5 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles were removed. The residue was purified by column chromatography (5:1 hexanes/EtOAc) to afford 195 mg (74% over 3 steps, 1:1 dr) **4.29** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.33 (d, *J* = 8.7 Hz, 1H), 7.25 (d, *J* = 2.6 Hz, 1H), 7.21 (d, *J* = 2.6 Hz, 1H), 7.13 (d, *J* = 8.5 Hz, 1H), 6.98 (dd, *J* = 8.7, 2.7 Hz, 1H), 6.95 (dd, *J* = 8.5, 2.6 Hz, 1H), 6.61 (s, 2H), 3.82 (s, 3H), 3.81 (s, 3H), 3.09 (qt, *J* = 7.2, 3.6 Hz, 1H), 2.97 (dq, *J* = 12.0, 6.0 Hz, 1H), 2.92-2.85 (m, 1H), 2.73-2.66 (m, 1H), 2.62-2.54 (m, 2H), 2.53-2.42 (m, 2H), 2.23-2.12 (m, 3H), 2.04 (dt, *J* = 12.9, 4.2 Hz, 1H), 1.84-1.75 (m, 4H), 1.40-1.38 (m, 2H), 1.37 (d, *J* = 6.8 Hz, 3H), 1.32 (d, *J* = 7.2 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 200.4, 200.3, 159.1, 158.3, 158.0, 157.9, 137.4, 137.0, 132.1, 131.3, 130.5, 128.1, 120.6, 120.3, 118.5, 118.2, 108.7, 108.3, 55.3 (2 C), 40.2, 37.5, 37.2, 36.9, 36.8, 32.4, 32.1, 31.0, 30.4, 30.1, 22.7, 21.1; IR (thin film) 2927, 2856, 1660, 1611, 1589, 1493 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>16</sub>H<sub>18</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 265.1205 found 265.1210.

## Birch Reduction to 4.30



A 25 mL round bottom flask was charged with 8 mL ammonia, 34 mg (0.14 mmol) **4.29** in 1.4 mL THF and 0.25 mL (6.2 mmol) MeOH. Slow, portionwise addition of 43 mg (6.2 mmol) lithium metal was conducted at  $-40\text{ }^{\circ}\text{C}$ . After complete addition of metal, 450 mg (8.5 mmol)  $\text{NH}_4\text{Cl}$  was added, the ammonia evaporated, water and EtOAc added. The organic layer was washed with brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo to afford **4.30** as a mixture of diastereomers (2:2:1:1 dr).

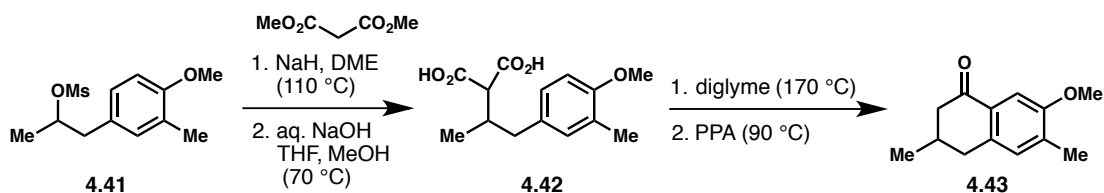
## Mesyate 4.41



A 100 mL round bottom flask containing 720 mg (29.6 mmol) magnesium metal was flame-dried, layered with 5 mL THF (rigorously degassed solvent was used during this procedure), treated with a couple drops of dibromoethane and heated to  $70\text{ }^{\circ}\text{C}$ . A solution of 5.00 g (24.9 mmol) 4-bromo-2-methylanisole in 25 mL THF was added dropwise as to maintain a gentle reflux. After complete addition the reaction was stirred at  $70\text{ }^{\circ}\text{C}$  another hour then cooled to  $-30\text{ }^{\circ}\text{C}$ . After the addition of 70 mg (0.80 mmol) CuCN and 2.0 mL (28.6 mmol) (rac)-propylene oxide the reaction was stirred for 1 hour with gradual warming to  $0\text{ }^{\circ}\text{C}$ , then stirred an additional half hour at room temperature. The reaction was quenched with 100 mL half sat.  $\text{NH}_4\text{Cl}$  and

extracted with 100 mL EtOAc. The aqueous layer was separated and back extracted with 20 mL EtOAc. Both organic layers were combined, washed with 50 mL sat.  $\text{NH}_4\text{Cl}$ , 30 mL third sat.  $\text{NH}_4\text{Cl}$ , 20 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The residue was used for the next without purification. To a 250 mL round bottom flask containing crude alcohol, was added 80 mL  $\text{CH}_2\text{Cl}_2$  and 15 mL (108 mmol)  $\text{NEt}_3$ . After the dropwise addition of 3.0 mL (38.8 mmol) methanesulfonyl chloride at 0 °C, the reaction was stirred at 0 °C for 2 hours. To the stirring mixture was added 150 mL sat.  $\text{NaHCO}_3$  and vigorous stirring continued without external cooling for 20 minutes. Layers were separated and the aqueous layer washed with 20 mL  $\text{CH}_2\text{Cl}_2$ . The organic layers were combined, washed with 40 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The residue was purified by column chromatography (5:1 hexanes/EtOAc) then recrystallized from  $\text{Et}_2\text{O}$ /pentane to afford 5.24 g (81% over 2 steps) **4.41** as white crystals (mp = 60–62 °C).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.03-6.98 (m, 2H), 6.77 (d,  $J$  = 8.1 Hz, 1H), 4.85 (ttd,  $J$  = 6.6, 6.4, 6.2 Hz, 1H), 3.82 (s, 3H), 2.91 (dd,  $J$  = 14.1, 8.0 Hz, 1H), 2.81 (dd,  $J$  = 14.0, 5.4 Hz, 1H), 2.54-2.52 (m, 3H), 2.18-2.16 (m, 3H), 1.46 (d,  $J$  = 6.2 Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  156.8, 131.8, 128.3, 127.7, 126.8, 109.9, 81.8, 55.3, 42.1, 37.8, 21.4, 16.2; IR (thin film) 2935, 2836, 1612, 1505, 1346, 1253, 1172  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{12}\text{H}_{18}\text{O}_4\text{S}$   $[\text{M}+\text{Na}]^+$  281.0833 found 281.0823.

## Tetralone 4.43

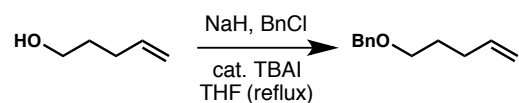


A 100 mL round bottom flask containing 24 mL DME and 0.970 g (24.3 mmol) NaH 60% in mineral oil was cooled in an ice bath and 2.7 mL (23.6 mmol) dimethylmalonate added dropwise. After the reaction was warmed to room temperature 3.00 g (11.6 mmol) **4.41** was added in one portion, the flask fitted with a good reflux condenser and the reaction placed in a 110 °C oil bath. After 20 hours, the reaction was cooled, the volatiles stripped in vacuo and the solid mass dissolved in 60 mL two-thirds sat. NH<sub>4</sub>Cl and 60 mL EtOAc. The aqueous layer was washed with an additional 20 mL EtOAc. The combined organic phases were washed with 20 mL brine, dried over MgSO<sub>4</sub>, filtered all volatiles removed in vacuo. The crude material was taken forward. The crude diester was placed into a 250 mL round bottom flask, diluted with 60 mL 1:1:1 THF/MeOH/water and treated with 4.64 g (116 mmol) powdered NaOH. The flask was placed into a 70 °C oil bath. After 1.5 hours, the reaction was cooled, the volatiles stripped in vacuo and crude material taken up in 120 mL water. The aqueous solution was washed with two portions of 20 mL Et<sub>2</sub>O, then 40 mL conc. HCl added and the aqueous phase extracted twice with 60 mL Et<sub>2</sub>O. The organic layers were combined, washed with 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The solid was subjected to the next reaction crude. Crude diacid in a 100 mL round bottom flask was dissolved in 35 mL diglyme and placed in a 170 °C oil bath for 1 hour. The reaction was cooled and all volatiles were removed in vacuo. The obtained oil was treated to the next reaction crude. Crude acid in a 100 mL round bottom flask was treated with 38 g polyphosphoric acid (prepared from 25 g H<sub>3</sub>PO<sub>4</sub> and 15 g P<sub>2</sub>O<sub>5</sub>) and



heated at 90 °C for 3 hours. External heating was removed and the reaction treated slowly with 150 g ice, then 20 mL 5 M NaOH and the resulting aqueous solution extracted with two portions of 60 mL CH<sub>2</sub>Cl<sub>2</sub>. The organic layers were collected, washed with 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude solid was dissolved in 20 mL hexanes, filtered over Celite, concentrated to approximately half volume and crystallized in a freezer overnight. The solvent was removed via cannula, the crystals washed twice with 2 mL portions of hexanes then the remaining solvent removed in vacuo to afford 1.44 g (60% over 4 steps) **4.43** as a white solid (mp = 68–69 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.44 (s, 1H), 7.01 (s, 1H), 3.87 (s, 3H), 2.88 (dd, *J* = 15.9, 3.6 Hz, 1H), 2.70 (dd, *J* = 13.1, 1.5 Hz, 1H), 2.59 (dd, *J* = 15.9, 10.3 Hz, 1H), 2.33–2.26 (m, 2H), 2.26–2.24 (m, 3H), 1.13 (d, *J* = 6.1 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 198.5, 156.6, 136.4, 133.9, 130.9, 130.9, 106.6, 55.5, 46.9, 37.2, 30.9, 21.4, 16.6; IR (thin film) 2951, 1678, 1510, 1496, 1302, 1210 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>13</sub>H<sub>16</sub>O<sub>2</sub> [M+H]<sup>+</sup> 205.1228 found 205.1221.

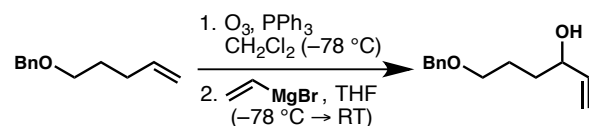
### 1-Benzoyloxy-4-pentene [Adapted from the literature]<sup>34</sup>



A 250 mL round bottom flask was charged with 70 mL THF and 5.2 mL (50.4 mmol) 4-penten-1-ol and treated portionwise with 2.1 g (52.5 mmol) 60% NaH in mineral oil at 0 °C. The reaction was warmed to room temperature and 6.4 mL (55.6 mmol) benzyl chloride and 185 mg (0.5 mmol) tetrabutylammonium iodide were added. The reaction was heated to 80 °C in an oil bath for 18 hours. After cooling to room temperature, 50 mL 1 M HCl were added, most volatiles removed in vacuo and the contents partitioned with 100 mL Et<sub>2</sub>O. The aqueous phase was

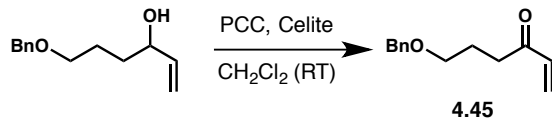
removed and back extracted with 25 mL Et<sub>2</sub>O. Both organic layers were combined, washed twice with 25 mL water, 25 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford a liquid that was distilled (75-78 °C/0.5 mmHg) to afford 7.10 g (80%) 1-benzyloxy-4-pentene as a colorless liquid. Spectral data was identical to the literature.<sup>34</sup>

### 6-Benzyloxyhex-1-en-3-ol



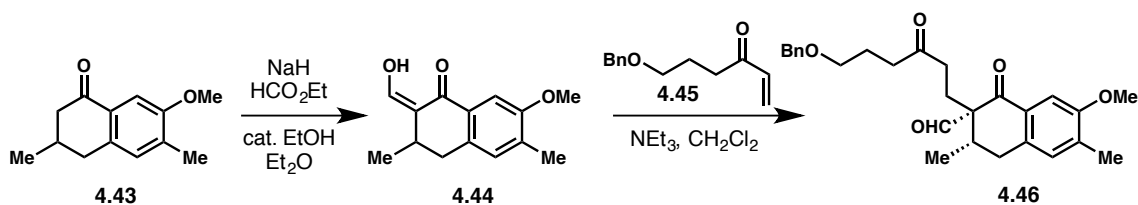
A 50 mL round bottom flask containing 563 mg (3.19 mmol) 1-benzyloxy-4-pentene in 20 mL CH<sub>2</sub>Cl<sub>2</sub> cooled to -78 °C was bubbled a stream of ozone until a blue color persisted. The contents were treated with 1.10 g (4.19 mmol) PPh<sub>3</sub> and the reaction warmed to room temperature. After all ozonide was quenched the reaction was dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude aldehyde in a 100 mL round bottom flask was diluted with 15 mL THF, cooled to -78 °C and 4.5 mL (4.5 mmol) 1.0 M vinylmagnesium bromide/THF was added dropwise. After 15 minutes the cold bath was removed. After 8 hours of stirring 10 mL 3 M HCl and 25 mL EtOAc was added. The layers were separated and the organic layer washed twice with 10 mL 3 M HCl. The aqueous washings were collected, back extracted twice with 10 mL EtOAc. The organic layers were combined, washed with 5 mL water, 10 mL sat. NaHCO<sub>3</sub>, 5 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The residue was purified by column chromatography (4:1 hexanes/EtOAc) to afford 285 mg (43% over 2 steps) 6-benzyloxyhex-1-en-3-ol as a colorless oil and 131 mg (23% over 2 steps) 4-benzyloxy-1-butanol.<sup>35</sup> The spectral data was identical to the literature.<sup>36</sup>

### 6-(Benzyloxy)hex-1-en-3-one 4.45



A 20 mL scintillation vial was charged with 270 mg (1.31 mmol) 6-benzyloxyhex-1-en-3-ol, 5 mL CH<sub>2</sub>Cl<sub>2</sub>, 1.2 g celite and 588 mg (2.73 mmol) PCC. After 2 hours the reaction was diluted with Et<sub>2</sub>O and filtered over Et<sub>2</sub>O to afford 128 mg (48%) **4.45** as a colorless liquid. The spectral data was identical to the literature.<sup>37</sup>

### Michael Adduct 4.46



A 1 dram vial containing 14 mg (0.35 mmol) NaH (60% in mineral oil) in 0.7 mL Et<sub>2</sub>O was cooled in an ice bath and treated sequentially with 0.05 mL (0.62 mmol) ethyl formate, 30 mg (0.145 mmol) **4.43** and 1 drop ethanol. Stirring was continued at 0 °C for 20 minutes before the bath was removed. After an additional hour and 40 minutes 1 M NaOH was added and the solution extracted thrice with Et<sub>2</sub>O. The aqueous layer was back extracted with 1 M NaOH. The aqueous layer was acidified with 6 M HCl and extracted thrice with Et<sub>2</sub>O. The organic layers were combined, washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude vinylogous acid in a 1 dram vial was dissolved in 0.3 mL CH<sub>2</sub>Cl<sub>2</sub> and 0.2 mL (1.43 mmol) NEt<sub>3</sub> at room temperature and treated with 45 mg (0.22 mmol) enone. After 30 hours the reaction wash treated with 1 M HCl, extracted with EtOAc, dried over MgSO<sub>4</sub> and all volatiles were removed in vacuo. The crude material was purified by column chromatography (5:1

hexanes/EtOAc) to afford 41 mg (65% over 2 steps, single diastereomer) **4.46** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 9.98 (s, 1H), 7.38 (s, 1H), 7.30-7.22 (m, 5H), 6.95 (s, 1H), 4.41 (s, 2H), 3.82 (s, 3H), 3.40 (t, *J* = 6.1 Hz, 2H), 3.14 (dd, *J* = 17.1, 4.7 Hz, 1H), 2.68 (dd, *J* = 17.1, 5.6 Hz, 1H), 2.50-2.34 (m, 4H), 2.32-2.26 (m, 2H), 2.22 (s, 3H), 2.12-2.05 (m, 1H), 1.81 (quintet, *J* = 6.6 Hz, 2H), 1.07 (d, *J* = 7.0 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 209.1, 203.6, 197.1, 156.8, 138.3, 135.2, 133.6, 131.3, 129.6, 128.3, 127.6, 127.5, 106.8, 72.8, 69.2, 62.4, 55.4, 39.4, 37.0, 35.3, 32.7, 24.7, 23.8, 16.7, 16.1.

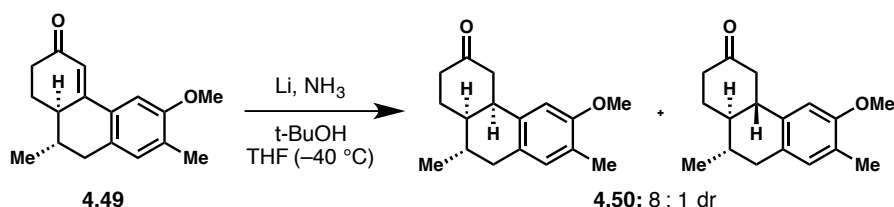
### Phenanthrenone 4.49



A 50 mL round bottom flask containing 304 mg (7.60 mmol) NaH (60% in mineral oil) in 15 mL Et<sub>2</sub>O was cooled in an ice bath and treated sequentially with 0.95 mL (11.8 mmol) ethyl formate, 602 mg (2.95 mmol) **4.43** and 0.05 mL (0.86 mmol) ethanol. Stirring was continued at 0 °C for 20 minutes before the bath was removed. After an additional hour and 40 minutes 40 mL 0.25 M NaOH was added, the layers separated and the aqueous phase extracted twice with 10 mL Et<sub>2</sub>O. The aqueous layer was acidified with 10 mL 6 M HCl and extracted with 40 mL and 10 mL Et<sub>2</sub>O. The organic layers were combined, washed with 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude vinylogous acid in a 100 mL round bottom flask was dissolved in 3 mL CH<sub>2</sub>Cl<sub>2</sub> and 3 mL (21.5 mmol) NEt<sub>3</sub> at room temperature and treated with 0.75 mL (9.00 mmol) MVK. All volatiles were removed in vacuo after 13 hours at room temperature. The crude solid was dissolved in 8 mL THF, cooled to 0°C and treated by the

dropwise addition of 980 mg (17.5 mmol) KOH in 8 mL water. After complete addition the cold bath was removed and the reaction stirred for 5 days. Volatiles were removed in vacuo and the remaining solution partitioned between 10 mL sat. NH<sub>4</sub>Cl and 30 mL EtOAc and the layers separated. The aqueous layer was extracted with 10 mL EtOAc. The combined organic layers were washed with 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was resubjected to the same conditions as described above for an additional 4 days. Work up consisted of the above procedure. Crystallization from CH<sub>2</sub>Cl<sub>2</sub>/hexanes afford 501 mg (66%, single diastereomer) **4.49** (mp = 188–190 °C) as an off-white crystalline solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.15 (s, 1H), 6.94 (s, 1H), 6.64 (s, 1H), 3.83 (s, 3H), 2.79 (dd, *J* = 16.3, 4.1 Hz, 1H), 2.63-2.54 (m, 2H), 2.45-2.37 (m, 2H), 2.28-2.25 (m, 1H), 2.22 (s, 3H), 1.74-1.61 (m, 2H), 1.16 (d, *J* = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 200.3, 158.8, 156.4, 132.4, 131.2, 130.9, 129.5, 119.8, 105.1, 55.3, 43.3, 38.3, 37.2, 35.3, 27.0, 19.5, 16.2; IR (thin film) 2949, 2913, 1654, 1205 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>17</sub>H<sub>20</sub>O<sub>2</sub> [M+H]<sup>+</sup> 257.1541 found 257.1529.

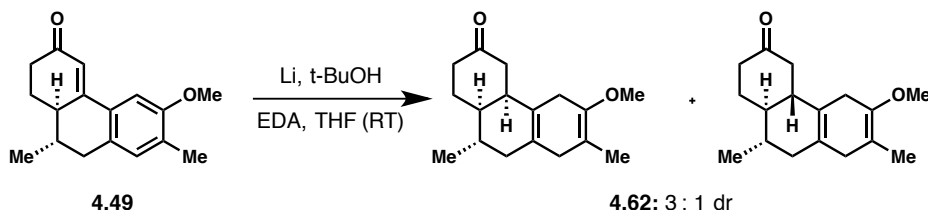
#### Anisole 4.50



A 10 mL round bottom flask was charged with excess lithium metal and ammonia at –40 °C, then 10 mg (0.039 mmol) **4.49** and 58 mg (0.78 mmol) t-BuOH in 0.5 mL THF was added dropwise. After 30 min another portion of 58 mg (0.78 mmol) t-BuOH in 0.2 mL THF was added. The reaction was quenched with solid NH<sub>4</sub>Cl, warmed to room temperature, diluted with water

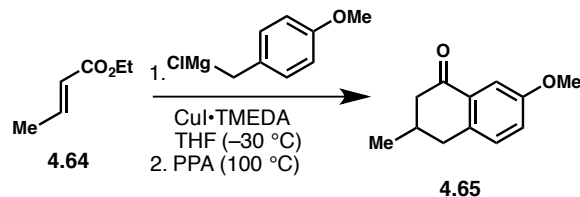
and extracted with EtOAc. The organic layer was dried over MgSO<sub>4</sub>, filtered and all volatiles removed. The crude solid was purified by column chromatography (8:1 hexanes/EtOAc) to afford 8.0 mg (80%, 8:1 dr) **4.50** as a white wax. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) major δ 6.87 (s, 1H), 6.50 (s, 1H), 3.79 (s, 3H), 3.20 (td, *J* = 8.6, 4.3 Hz, 1H), 3.09 (ddd, *J* = 14.1, 3.7, 2.1 Hz, 3H), 2.86 (dd, *J* = 16.3, 5.1 Hz, 1H), 2.57-2.55 (m, 2H), 2.47-2.19 (m, 5H), 2.19 (s, 3H), 1.86 (ddq, *J* = 14.1, 9.2, 4.6 Hz, 2H), 1.14 (d, *J* = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) major δ 211.5, 156.0, 137.5, 130.9, 127.0, 125.0, 109.0, 55.3, 47.6, 42.4, 39.0, 37.3, 29.7, 27.3, 26.3, 19.2, 15.8; IR (thin film) 2919, 2853, 1710 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>17</sub>H<sub>22</sub>O<sub>2</sub> [M+NH<sub>4</sub>]<sup>+</sup> 276.1964 found 276.1967.

### Cyclohexadiene 4.62



To a 25 mL round bottom flask was added 20 mg (0.080 mmol) **4.49**, 0.38 mL (4.0 mmol) *t*-BuOH, 0.8 mL ethylenediamine, 0.8 mL THF and a vigorously stirring glass stir bar. At room temperature 25 mg (3.6 mmol) lithium metal was added over the course of 2 hours. After stirring overnight the mixture was diluted with 5 mL THF and cannula transferred into a stirring mixture of 10 mL water and 15 mL Et<sub>2</sub>O. The layers were separated and the aqueous solution extracted with Et<sub>2</sub>O. The organic layers were combined, washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford **4.62** (3:1 dr).

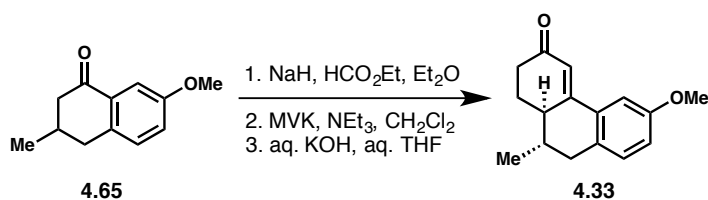
PCR 3.225-227 [Adapted from the literature]<sup>27</sup>



A 50 mL 2 neck round bottom flask filled with 2.29 g (94.4 mmol) magnesium metal was flame-dried under vacuum, allowed to cool, flushed with argon, fitted with a thermometer and an addition funnel, then layered with 9 mL THF. To the vigorously stirring mixture was added 0.05 mL 1,2-dibromoethane, then 3.2 mL (23.6 mmol) p-methoxybenzyl chloride in 6 mL THF was added from the addition funnel to maintain an internal temperature of 60-65 °C. After complete addition the solution was stirred for an additional 2 hours. A 250 mL round bottom flask was charged with 3.62 g (19.0 mmol) CuI, 80 mL THF and 3.1 mL (21.4 mmol) TMEDA at room temperature and stirred until a homogeneous solution was obtained. The flask was fitted with an addition funnel and cooled to -78 °C. The freshly prepared Grignard reagent was loaded to the addition funnel and added dropwise. After stirring for 30 minutes at -78 °C a new addition funnel was fitted and loaded with 1.9 mL (15.3 mmol) ethyl trans-2-butenoate, 4.9 mL (38.6 mmol) TMSCl and 20 mL THF. After dropwise addition of this mixture, the reaction was warmed to -30 °C. After 18 hours stirring at -30 °C, the reaction was poured into 125 mL sat. NH<sub>4</sub>Cl/75 mL sat. NH<sub>4</sub>OH and partitioned with 100 mL Et<sub>2</sub>O. The layers were separated, the aqueous layer extracted with 25 mL Et<sub>2</sub>O and all organic layers combined. The ethereal phase was washed twice with 50 mL water, 20 mL brine, dried over MgSO<sub>4</sub>, filtered all volatiles removed in vacuo. The crude mixture was swirled with 40 mL hexanes, filtered and washed with thrice with 10 mL hexanes. The hexanes washes were concentrated and the crude material carried on to the next step. The crude ester was treated with 30 g PPA and heated at 100 °C for

1.5 hour. The heating source was removed and 100 g ice water was added. The aqueous solution was extracted twice with 40 mL EtOAc. All organic layers were combined, washed with 30 mL sat. NaHCO<sub>3</sub>, 20 mL brine, dried over MgSO<sub>4</sub>, stirred vigorously with 6 g charcoal overnight, filtered and all volatiles removed in vacuo. The crude material was purified by crystallization from hexanes to afford 1.47 g (51%) **4.65** as small, off-white needles (mp = 43–44 °C). The spectral data was identical to the literature.<sup>27</sup>

### Phenanthrenone 4.33

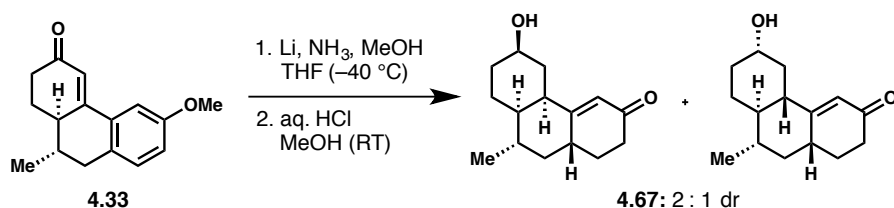


A 100 mL round bottom flask containing 750 mg (18.8 mmol) NaH (60% in mineral oil) in 50 mL Et<sub>2</sub>O was cooled in an ice bath. Sequentially, 2.4 mL (30 mmol) ethyl formate, 1.42 g (7.46 mmol) **4.65** and 0.1 mL (1.7 mmol) ethanol was added. Stirring was continued at 0 °C for 20 minutes before the bath was removed. After 6 hours stirring, 30 mL water was added, the layers separated and the aqueous phase extracted twice with 10 mL Et<sub>2</sub>O. The ethereal layers were washed twice with 10 mL 1 M NaOH. The aqueous layers were combined, acidified with 10 mL 6 M HCl and extracted with 30 mL and 10 mL Et<sub>2</sub>O. The organic layers were combined, washed with 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude vinylogous acid in a 200 mL round bottom flask was dissolved in 10 mL CH<sub>2</sub>Cl<sub>2</sub> and 5.2 mL (37 mmol) NEt<sub>3</sub> at room temperature and treated with 1.9 mL (23 mmol) MVK. All volatiles were removed in vacuo after 25 hours at room temperature. The crude solid was dissolved in 35 mL THF, cooled at 0°C and treated by the dropwise addition of 2.1 g (37 mmol) KOH in 35 mL



water. After complete addition the cold bath was removed and the reaction stirred for 2 days. All volatiles were removed in vacuo and the remaining solution partitioned between 20 mL sat.  $\text{NH}_4\text{Cl}$  and 40 mL EtOAc and the layers separated. The aqueous layer was extracted with 10 mL EtOAc. The combined organic layers were washed with 10 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. Crystallization from EtOAc afforded 1.12 g (61%) **4.33** (mp = 143–145 °C) as an off-white crystalline solid.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.25 (d,  $J = 2.4$  Hz, 1H), 7.09 (d,  $J = 8.4$  Hz, 1H), 6.93 (dd,  $J = 8.4, 2.5$  Hz, 1H), 6.65 (d,  $J = 1.8$  Hz, 1H), 3.82 (s, 3H), 2.84 (dd,  $J = 16.3, 4.1$  Hz, 1H), 2.64–2.57 (m, 2H), 2.46–2.37 (m, 2H), 2.30–2.24 (m, 1H), 1.75–1.63 (m, 2H), 1.17 (d,  $J = 6.5$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  200.3, 158.3, 158.1, 132.5, 132.1, 130.2, 120.8, 118.2, 108.3, 55.4, 43.3, 38.4, 37.2, 35.2, 27.0, 19.5; IR (thin film) 2946, 2907, 1655, 1498  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{16}\text{H}_{18}\text{O}_2$   $[\text{M}+\text{H}]^+$  243.1385 found 243.1376.

### Enone 4.67

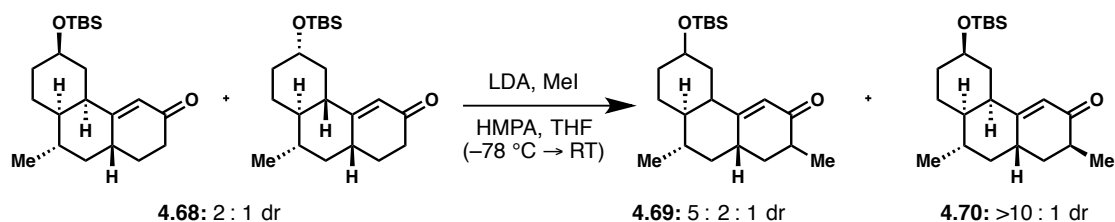


A 250 mL round bottom flask was charged with 50 mL ammonia, 25 mL THF, 5 mL MeOH and 500 mg (2.06 mmol) **4.33**. Slow, portionwise addition of 470 mg (68 mmol) lithium metal was conducted at  $-40$  °C. After complete addition of metal, 1 g  $\text{NH}_4\text{Cl}$  was added, the ammonia evaporated, 30 mL water and 50 mL EtOAc added. The organic layer was washed with 10 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude oil in a 25 mL round bottom flask was dissolved in 15 mL MeOH, treated with 1.5 mL 6 M HCl. After 24 hours



2.16-1.82 (m, 4H), 1.72-1.53 (m, 3H), 1.41-1.09 (m, 3H), 1.05-0.74 (m, 14H), 0.07 (s, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm) major  $\delta$  200.5, 170.6, 124.0, 71.5, 46.1, 43.5, 42.1, 38.7, 36.8, 34.5, 30.0, 29.4, 27.1, 25.9, 25.6, 25.5, 18.9, 4.65, minor  $\delta$  200.3, 168.8, 121.1, 71.3, 48.9, 45.3, 43.5, 38.1, 37.7, 36.8, 35.6, 35.4, 29.4, 28.5, 25.9, 19.2, 18.2, 4.6; IR (thin film) 2930, 2856, 1676, 1618, 1253, 1089  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{21}\text{H}_{36}\text{O}_2\text{Si}$   $[\text{M}+\text{H}]^+$  349.2563 found major diastereomer 349.2558, minor diastereomer 349.2574.

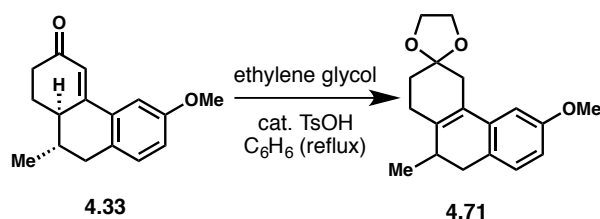
### Alkylated 4.69 and 4.70



LDA was prepared in a 25 mL round bottom flask by addition of 0.45 mL (1.2 mmol) 2.7 M n-BuLi/hexanes to 0.20 mL (1.4 mmol) diisopropylamine in 4 mL THF at 0 °C. To the stirring LDA solution at -78 °C was added 346 mg (0.99 mmol) **4.68** with the assistance of 2 mL THF. After 10 minutes, 0.22 mL (1.3 mmol) HMPA was added neat followed by 0.18 mL (2.9 mmol) methyl iodide. The cold bath was removed after an additional 10 minutes and the reaction stirred for 1 hour before 10 mL half sat.  $\text{NH}_4\text{Cl}$  and 20 mL EtOAc was added. The organic layer was separated, washed with 5 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude mixture was purified by column chromatography (40:1  $\rightarrow$  30:1 hexanes/EtOAc) to afford 28 mg (26%, 5:2:1 dr) **4.69** and 194 mg (54%, >10:1 dr) **4.70** as a colorless oil. **4.70**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.75 (s, 1H), 3.63 (tt,  $J = 10.0, 4.9$  Hz, 1H), 2.67-2.63 (m, 1H), 2.43 (dt,  $J = 12.4, 6.1$  Hz, 2H), 2.07-2.01 (m, 1H), 1.95 (d,  $J = 12.1$  Hz, 1H), 1.89-1.82 (m, 2H), 1.75-1.64 (m, 3H), 1.57 (d,  $J = 12.5$  Hz, 1H), 1.39-1.26 (m, 3H), 1.11 (d,  $J = 7.2$  Hz,

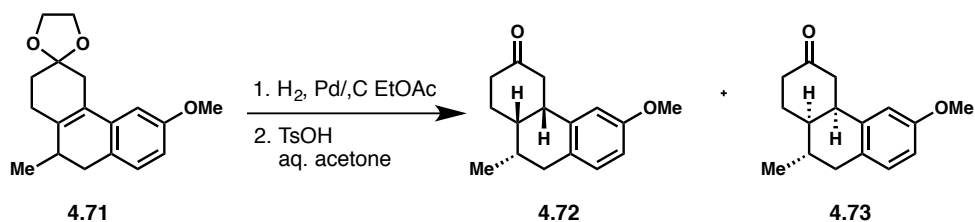
3H), 1.01 (s, 1H), 0.92-0.88 (m, 3H), 0.88 (s, 9H), 0.06 (d,  $J = 2.5$  Hz, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  203.7, 169.3, 122.1, 71.5, 46.2, 43.2, 42.7, 39.1, 37.9, 35.7, 31.2, 30.1, 27.6, 25.9, 25.6, 19.0, 18.2, 16.1, 4.6; IR (thin film) 2929, 2856, 1675, 1622, 1460, 1255, 1090  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{22}\text{H}_{38}\text{O}_2\text{Si}$   $[\text{M}+\text{H}]^+$  363.2719 found 363.2728.

### Ketal 4.71



A 10 mL round bottom flask containing 0.05 mL (0.9 mmol) ethylene glycol and 8 mg (0.04 mmol)  $\text{TsOH}\cdot\text{H}_2\text{O}$  in 5 mL benzene was refluxed under a Hickman still for 20 minutes. The solution was cooled, 100 mg (0.413 mmol) **4.33** added and the reaction refluxed under a Hickman still. After 2 hours the reaction was cooled, diluted with 10 mL EtOAc, washed with 3 mL sat.  $\text{NaHCO}_3$ , 2 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1 hexanes/EtOAc) to afford 110 mg (93%) **4.71** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.02 (d,  $J = 8.1$  Hz, 1H), 6.70-6.66 (m, 2H), 4.08-4.02 (m, 4H), 3.80 (s, 3H), 2.96 (dd,  $J = 15.1, 6.5$  Hz, 1H), 2.73 (d,  $J = 16.6$  Hz, 1H), 2.52 (d,  $J = 17.9$  Hz, 1H), 2.50-2.44 (m, 3H), 2.30-2.23 (m, 1H), 1.91-1.80 (m, 2H), 0.94 (d,  $J = 7.2$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  158.2, 138.3, 136.3, 128.5, 125.8, 123.4, 110.5, 108.5, 107.7, 64.4, 55.2, 35.7, 34.7, 32.7, 31.2, 28.0, 16.9; IR (thin film) 2953, 2883, 2830, 1605, 1574, 1493, 1210, 1039  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{22}\text{O}_3$   $[\text{M}+\text{H}]^+$  287.1647 found 287.1640.

### Ketone 4.72 and 4.73



To a 10 mL round bottom flask was added 31 mg (0.11 mmol) **4.71**, 1.2 mL EtOAc, 5 mg 10% Pd/C. The reaction was stirred under a hydrogen balloon for 20 hours at room temperature. Celite was added and the reaction was filtered over Celite, washing with EtOAc to afford ~95% conversion to a 4:1 mixture of diastereomers. The crude mixture was dissolved in 1 mL 5:1 acetone/water then catalytic TsOH•H<sub>2</sub>O added. After 4 days at room temperature 3 mL water and 2 mL sat. NaHCO<sub>3</sub> was added and the solution extracted with 5 mL EtOAc. The organic layer was washed with 2 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude was purified by column chromatography (8:1 hexanes/EtOAc) to afford 6 mg **4.73** as an impure oil and 16 mg (62%) **4.72** as a white solid that was recrystallized from Et<sub>2</sub>O to provide white fluffy needles (mp = 140–142 °C). The minor diastereomer <sup>1</sup>H NMR spectrum significantly matched that of **4.50**. **4.72**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 6.98 (d, *J* = 8.4 Hz, 1H), 6.86 (s, 1H), 6.70 (d, *J* = 8.3 Hz, 1H), 3.78 (s, 3H), 3.48 (s, 1H), 3.10 (d, *J* = 14.7 Hz, 1H), 2.77-2.73 (m, 2H), 2.44 (t, *J* = 14.6 Hz, 1H), 2.34 (dd, *J* = 13.6, 6.2 Hz, 1H), 2.30-2.27 (m, 1H), 2.20 (t, *J* = 12.1 Hz, 2H), 1.93-1.91 (m, 1H), 1.51 (dd, *J* = 13.0, 4.9 Hz, 1H), 1.12 (d, *J* = 6.7 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 210.9, 158.0, 136.9, 130.0, 128.1, 112.3, 112.1, 55.3, 44.6, 42.4, 40.5, 40.2, 33.4, 32.3, 20.2, 19.7; IR (thin film) 2952, 1701, 1495, 1239, 1040 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>16</sub>H<sub>20</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 267.1361 found 267.1357.

#### 4.7 References and Notes

- (1) Streuff, J. *Synthesis* **2013**, 281.
- (2) Corey, E. J.; Magriotis, P. A. *J. Am. Chem. Soc.* **1987**, *109*, 287.
- (3) Caine, D. *Org. React.* **1976**, *23*, 1.
- (4) Keinan, E.; Greenspoon, N. "Reduction of  $\alpha,\beta$ -unsaturated carbonyl compounds" In *The Chemistry of Enones*; Patai, S.; Rappoport, Z., Ed.; John Wiley & Sons, Ltd, 1989; pp 923–1022.
- (5) Huffman, J. W. "Reduction of C=X to CHXH by Dissolving Metals and Related Methods" In *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I., Ed.; Elsevier Science Ltd., 1991; pp 107–127.
- (6) Hook, J. M.; Mander, L. N. *Nat. Prod. Rep.* **1986**, *3*, 35.
- (7) Rabideau, P. W.; Marcinow, Z. *Org. React.* **1992**, *42*, 1.
- (8) Gesson, J.-P.; Jacquesy, J.-C.; Jacquesy, R. *Nouv. J. Chim.* **1977**, *1*, 511.
- (9) Robinson, M. J. T. *Tetrahedron* **1957**, *1*, 49.
- (10) Bansal, R. C.; Browne, C. E.; Eisenbraun, E. J.; Thomson, J. S. *J. Org. Chem.* **1988**, *53*, 452.
- (11) Chatterjee, A.; Chaudhuri, S. R. R.; Chatterjee, S. K. *J. Chem. Soc., Chem. Commun.* **1982**, 24.
- (12) Barton, D. H. R.; Robinson, C. H. *J. Chem. Soc.* **1954**, 3045.
- (13) Nadeau, E.; Ventura, D. L.; Brekan, J. A.; Davies, H. M. *J. Org. Chem.* **2010**, *75*, 1927.
- (14) Numbering matches the isocycloamphilectane scaffold, i.e. 7,20-diisocyanoadociane.
- (15) White, J. D.; Hrcniar, P.; Stappenbeck, F. *J. Org. Chem.* **1999**, *64*, 7871.
- (16) Sobukawa, M.; Nakajima, M.; Koga, K. *Tetrahedron: Asymmetry* **1990**, *1*, 295.

- (17) Taguri, T.; Yamamoto, M.; Fujii, T.; Muraki, Y.; Ando, T. *European J. Org. Chem.* **2013**, 6924.
- (18) van Tamelen, E. E.; Taylor, E. G. *J. Am. Chem. Soc.* **1980**, *102*, 1202.
- (19) Yamato, Y.; Kaneko, H. *Tetrahedron* **1965**, *21*, 2501.
- (20) Bohlmann, F.; Giencke, W. *Tetrahedron* **1983**, *39*, 443.
- (21) Benkeser, R. A.; Sgihotri, R. K.; Burbous, M. L.; Kaiser, E. M.; Mallan, J. M.; Ryan, P. *W. J. Org. Chem.* **1964**, *29*, 1313.
- (22) Kaiser, E. M. *Synthesis* **1972**, 391.
- (23) Benkeser, R. A.; Laugal, J.; Rappa, A. *Tetrahedron Lett.* **1984**, *25*, 2089.
- (24) Kotlarek, W. *Pol. J. Chem.* **1983**, *57*, 809.
- (25) Kwart, H.; Conley, R. A. *J. Org. Chem.* **1973**, *38*, 2011.
- (26) Garst, M. E.; Dolby, L. J.; Esfandiari, S.; Fedoruk, N. A.; Chamberlain, N. C.; Avey, A. *A. J. Org. Chem.* **2000**, *65*, 7098.
- (27) Scopton, A.; Kelly, T. R. *J. Org. Chem.* **2005**, *70*, 10004.
- (28) Brown, G. D. *Phytochemistry* **1994**, *35*, 425.
- (29) Shu, Y.-F.; Wei, H.-C.; Wu, C.-L. *Phytochemistry* **1994**, *37*, 773.
- (30) Granger, R.; Chapat, J. P.; Crassous, J.; Simon, F. *Bull. Chim. Soc. Fr.* **1968**, 4265.
- (31) Love, B. E.; Jones, E. G. *J. Org. Chem.* **1999**, *64*, 3755.
- (32) Seed, A. J.; Sonpatki, V.; Herbert, M. R. *Org. Synth.* **2002**, *79*, 204.
- (33) Rassadin, V. A.; Six, Y. *Tetrahedron* **2014**, *70*, 787.
- (34) Lowik, D. W. P. M.; Liskamp, R. M. *European J. Org. Chem.* **2000**, 1219.
- (35) Kiddle, J. J.; Green, D. L. C.; Thompson, C. M. *Tetrahedron* **1995**, *51*, 2851.
- (36) Kyne, R. E.; Ryan, M. C.; Kliman, L. T.; Morken, J. P. *Org. Lett.* **2010**, *12*, 3796.

(37) Iyengar, R.; Schildknecht, K.; Morton, M.; Aubé, J. *J. Org. Chem.* **2005**, *70*, 10645.



## CHAPTER 5: A FORMAL SYNTHESIS OF (+)-7,20-DIISOCYANOADOCIANE

### 5.1 Introduction

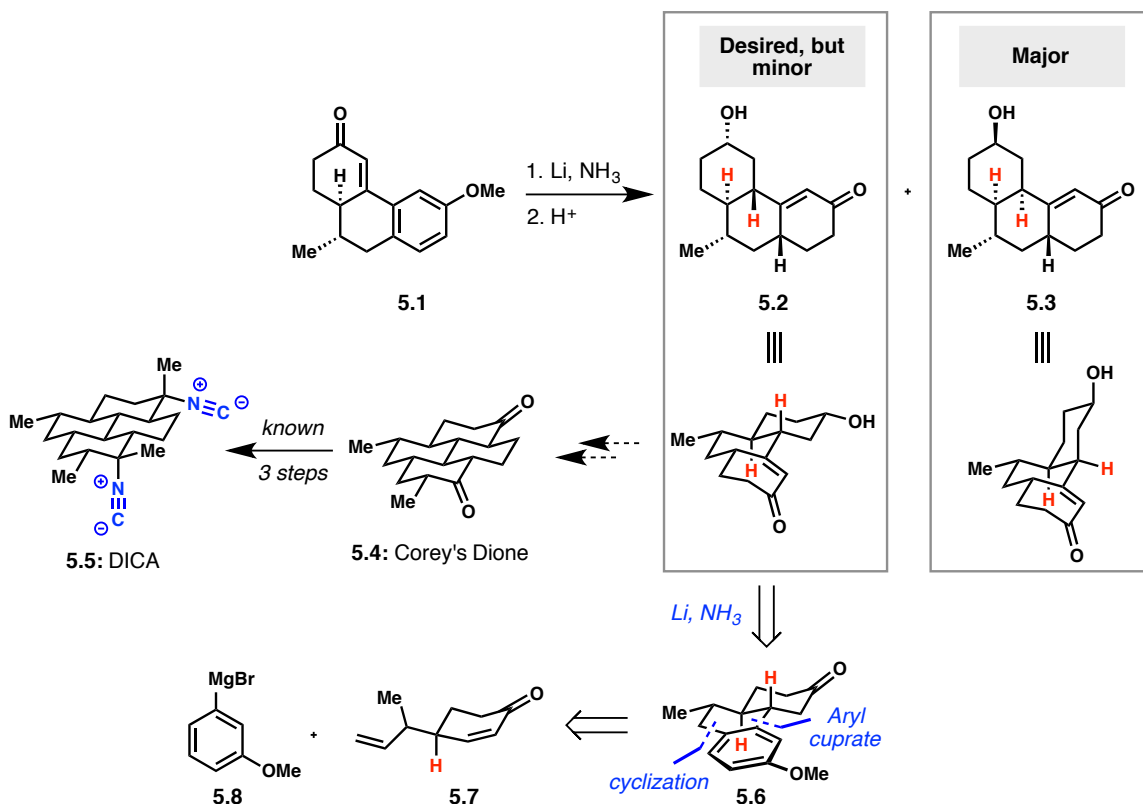
A synthesis of 7,20-diisocyanoadociane (**5.5**, DICA) via reduction of an unsaturated polycyclic intermediate was considered highly attractive both because unsaturated molecules can be rapidly constructed and can provide multiple options in controlling stereochemistry. Previously, Birch reduction of a phenanthrenone substrate successfully generated a tricyclic scaffold relevant to DICA in short order, but failed due to an undesired stereochemical outcome (see Chapter 4). Described here is a revised approach that successfully secured an enantiospecific formal synthesis of (+)-DICA and could serve as a platform to access more 7-isocyano(iso)cycloamphilectanes.

### 5.2 A New Design to Control the All-*trans* Stereochemistry

Initial studies concerning the reduction of phenanthrenones towards DICA showed an undesirable tendency for *cis*-reduction (Scheme 5.1). Attempts at switching selectivity to the *trans*-ring fusion using hydrogenation were modestly promising, but a synthetic redesign to substantially improve stereocontrol was preferred. The minor diastereomer obtained by phenanthrenone reduction, **5.2**, was still a desirable target and considered a key intermediate. If a synthesis of **5.2** could be secured in an elegant manner, then a single ring closure to Corey's dione **5.4** would be the only operation remaining.<sup>1</sup> Rapidly generating phenanthrenone **5.1** highlights the advantage aromatic starting materials may have in synthetic design;<sup>2,3</sup> therefore the enone of **5.2** was still retrosynthetically traced back to an anisole, such as **5.6**. Based on previous

experience, the challenge of setting the *trans*-ring fusion was to be tackled directly by a robust and stereochemically assured method. A cuprate conjugate addition would be suited to the task.

**Scheme 5.1** A previous attempt and a revision in using a Birch reduction approach to DICA.



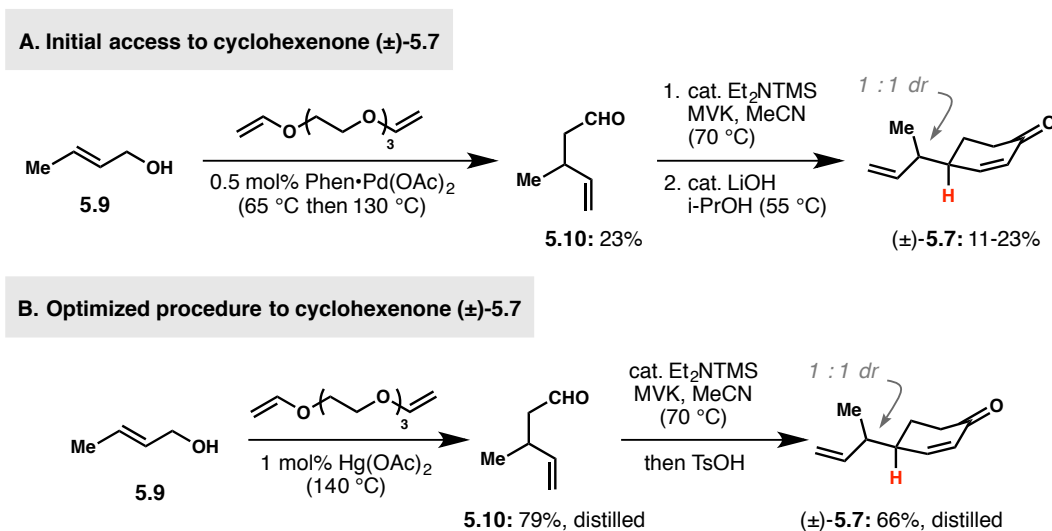
## 5.3 The C8–C9 Ring Closing Strategy

### 5.3.1 Successes in Generating *trans*-Ring Fusions

A Robinson annulation to generate **5.7** was devised, surmising that application of a chiral organocatalyst to the Michael addition step would render the route highly enantioselective.<sup>4–6</sup> The synthesis began with a Claisen rearrangement of crotyl alcohol (**5.9**) by intermediacy of the vinyl enol ether (Scheme 5.2). Palladium-mediated conditions were originally used, but suffered from incomplete conversion and difficulty in purification.<sup>7</sup> Diethyltrimethylsilylamine facilitated

a Michael addition to methyl vinyl ketone to provide the adduct in excellent yield; however, aldol ring closure under literature<sup>8</sup> conditions, previously successful in a slightly different context (see Chapter 2), afforded **5.7** in low yield. Although this three-step sequence of generating **5.7** was sufficient for the exploratory stage, optimization quickly became a priority. Returning to classic Hg(OAc)<sub>2</sub> catalyzed Claisen rearrangement afforded **5.10** in a reliable and scalable manner.<sup>9</sup> No change was made to the Michael addition; however, a screen of various reactions was performed to improve the aldol condensation. The most effective conditions were TsOH in acetonitrile, meaning that TsOH could simply be added upon complete Michael addition to afford **5.7**. This three-step/two-pot process prepared enough **5.7** to fuel studies.

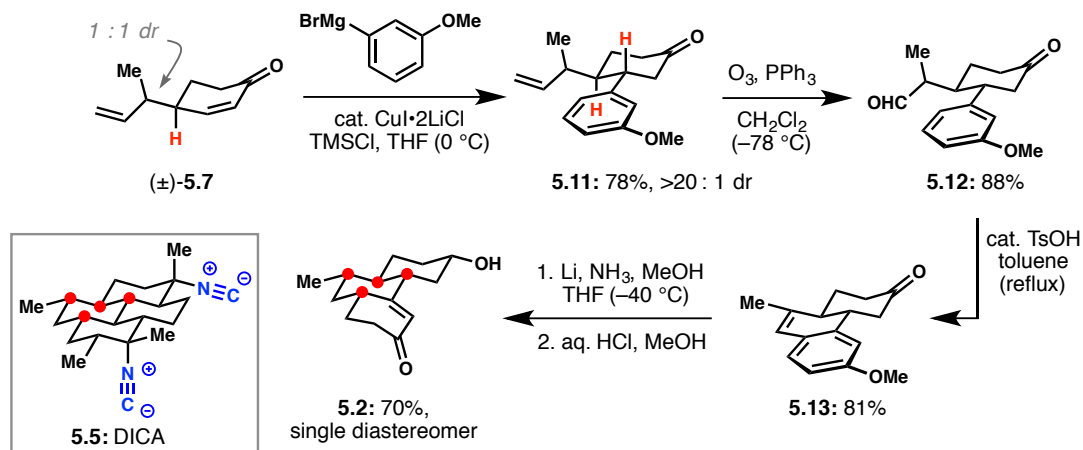
**Scheme 5.2** A Robinson annulation entry towards dihydronaphthalene studies of DICA.



The dihydronaphthalene Birch reduction idea was continued via cuprate addition to **5.7** (Scheme 5.3). Initial stoichiometric cuprate conditions with CuI afforded **5.11** in ~50% yield with tedious workup. Employing catalytic quantities of the soluble Retz CuI•2LiCl not only improved the yield, but also the ease of reaction setup and workup.<sup>10</sup> Ozonolysis and Friedel–Crafts cyclodehydration afforded dihydronaphthalene **5.13**. Treating **5.13** with Li/NH<sub>3</sub> and

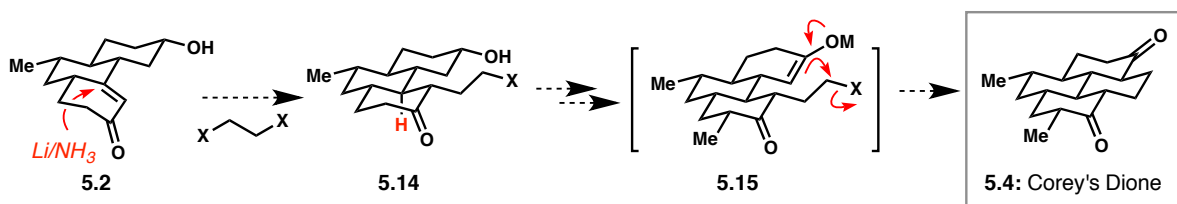
excess methanol followed by aqueous HCl furnished **5.2** without issue as a single diastereomer on the first attempt. The synthesis of a scaffold bearing four desired stereocenters reminiscent of DICA in an all *trans*-arrangement was finally successful.

**Scheme 5.3** Efficient preparation of **5.2** by dihydrophthalene reduction.



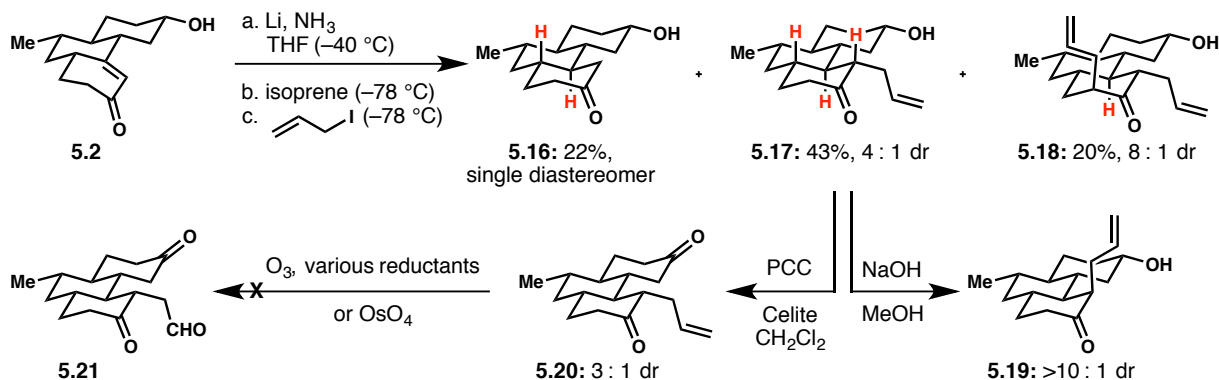
Tricycle **5.2** needs to be stitched together with an appropriate two-carbon unit to intercept Corey's dione (Scheme 5.4). The first step requires introducing a proton at the beta carbon of enone **5.2** in an axial orientation by well preceded  $\text{Li}/\text{NH}_3$  reduction.<sup>11,12</sup> The hereby-generated enolate could then be alkylated with a 1,2-difunctionalized ethane to provide **5.14**. The last C–C bond could then be formed by enolate alkylation to generate Corey's dione **5.4**.

**Scheme 5.4** Proposed path forward to Corey's dione.

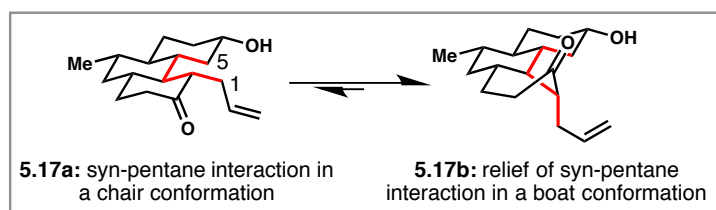


## Scheme 5.5 Early developments installing the two-carbon bridge.

### A. Installing a two carbon unit towards Corey's dione



### B. Proposed conformational flexibility in 5.17



As a first pass, the free C7 alcohol<sup>13</sup> of **5.2** was not protected and submitted directly to Li/NH<sub>3</sub> (Scheme 5.5A). Upon reduction, the enolate was allylated to afford the three products **5.16**, **5.17** and **5.18**. The des-allyl **5.16** was isolated as a single diastereomer, indicating that the  $\beta$ -position had exclusively protonated axially. The desired compound **5.17** was purified as a 3 : 1 mixture of diastereomers at the allyl position. The diastereoselectivity could not be improved in the desired direction by epimerization, because the thermodynamic preference of the allyl is not the expected equatorial orientation, but is axial instead. Syn-pentane interactions in the chair conformation are believed to cause this discrepancy (Scheme 5.5B). The bis-allylated **5.18** was also separated and arose from allylation of the enolate of **5.17**, which was generated by deprotonation with lithium amide. Next, the alkene of **5.17** needed to be dehomologated and converted into a leaving group. Disappointingly, cleavage by ozonolysis or OsO<sub>4</sub> led to heavy

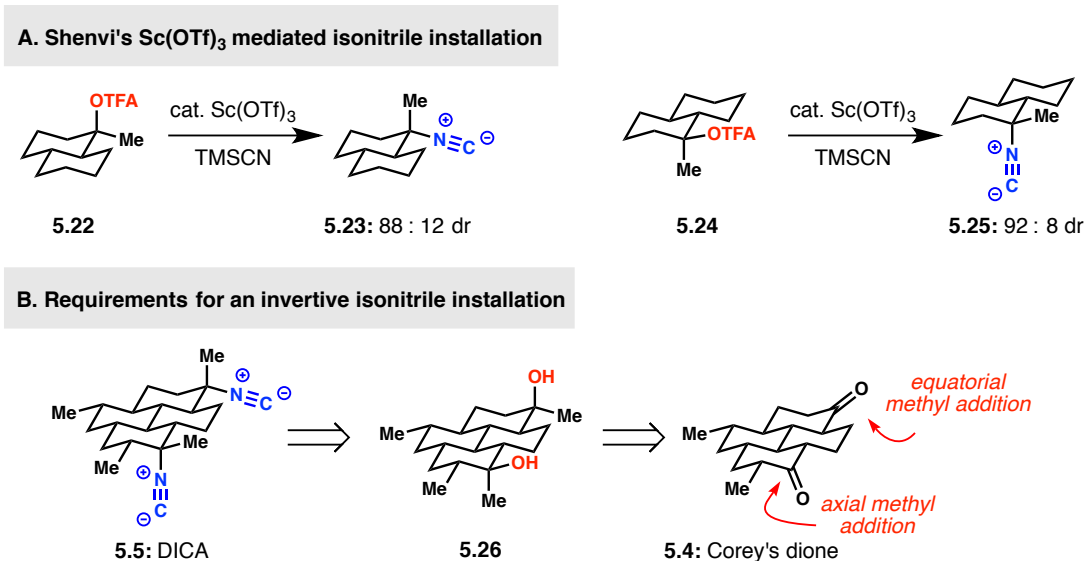
mixtures of compounds that were unable to be carried forward, even in crude form through the aldol condensation.

### 5.3.2 Considering a $Sc(OTf)_3$ -Mediated Isonitrile Installation to 7,20-Diisocyanoadociane

Concurrent with the success in generating all-*trans* ring fusions (Scheme 5.5), the endgame isonitrile installation began to be considered. At the outset of this project, a path through Corey's dione was sought; however, during the course of these studies, Shenvi<sup>14,15</sup> disclosed an invertive displacement of tertiary trifluoroacetates with TMS-CN using  $Sc(OTf)_3$  as an elegant and selective advancement to the originally used  $TiCl_4$ <sup>1,16</sup> (Scheme 5.6A). This methodology was applied to the synthesis of 7-isocyano-11(20),14-epiamphilectadiene by Shenvi,<sup>15</sup> and its implementation to DICA quickly became a goal. The selective synthesis of **5.26**, bearing both an axial and an equatorial alcohol is required for this approach (Scheme 5.6B). This challenge could be addressed either by derivatizing Corey's dione **5.4** to **5.26**, or by controlling the axial or equatorial configuration of the tertiary alcohols independently.

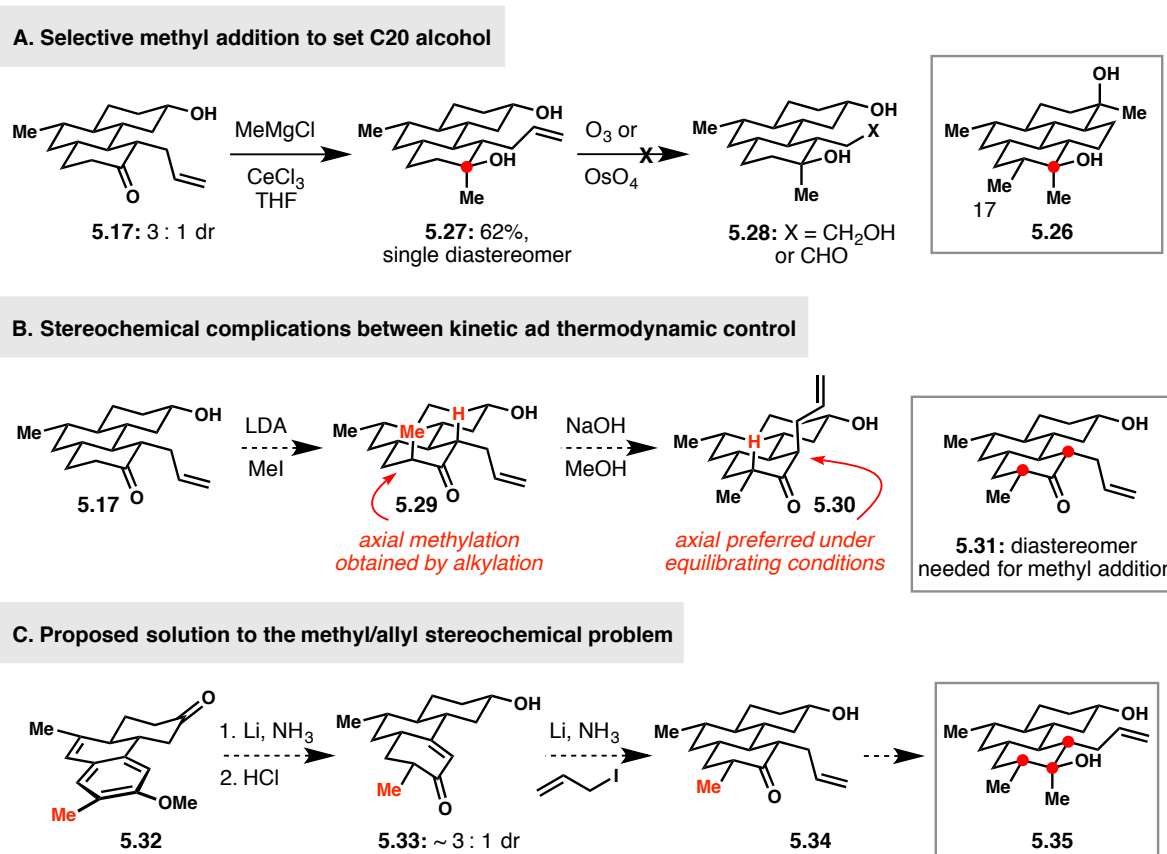
For the current approach, intermediate **5.17** would be appropriate to begin derivatization to diol **5.26** as the C7 and C20 positions are already differentiated (Scheme 5.7). After a screen of MAD,<sup>17</sup>  $AlMe_3$ ,<sup>18</sup> organomanganese reagents,<sup>19</sup> methylmagnesium halide and methyllithium, a clean methyl addition to **5.17** to afford **5.27** was found with methylmagnesium chloride/ $CeCl_3$  complex. The unusual axial selectivity presumably arose from the equilibrium between **5.17a** and **5.17b** favoring the boat conformation and thereby presenting only a "convex" face for nucleophile addition (Scheme 5.5B).

**Scheme 5.6** Use of an invertive isonitrile synthesis towards DICA.



Although this nucleophilic methyl addition conveniently generates the desired C20 tertiary alcohol, the C17 methyl group in **5.27** is absent and still needs to be installed. Overall, correctly placing this equatorial methyl group is complicated (Scheme 5.7). Alkylation of **5.17** could install the methyl group; however, it would be introduced axial. While equilibrating conditions would reasonably epimerize the methyl group to the equatorial orientation, these same conditions would also epimerize the allyl group axial to form **5.30** (Scheme 5.7B). The stereochemical arrangement of **5.30** precludes methyl addition to the ketone, as there is no manner of correcting the allyl moiety. A proposed solution to this problem could be the incorporation of the C17 methyl group earlier. Although at equilibrium the methyl enone is predicted to arrive at ~2 : 1 dr, the mixture could be advanced through the reductive allylation and methyl addition to generate the stereotriad of **5.35** (Scheme 5.7C). Even though the reduction of 1,2,4,5-tetrasubstituted anisoles was previously problematic (Chapter 4), this idea was still pursued.

**Scheme 5.7** Exploring options for a selective diol synthesis.

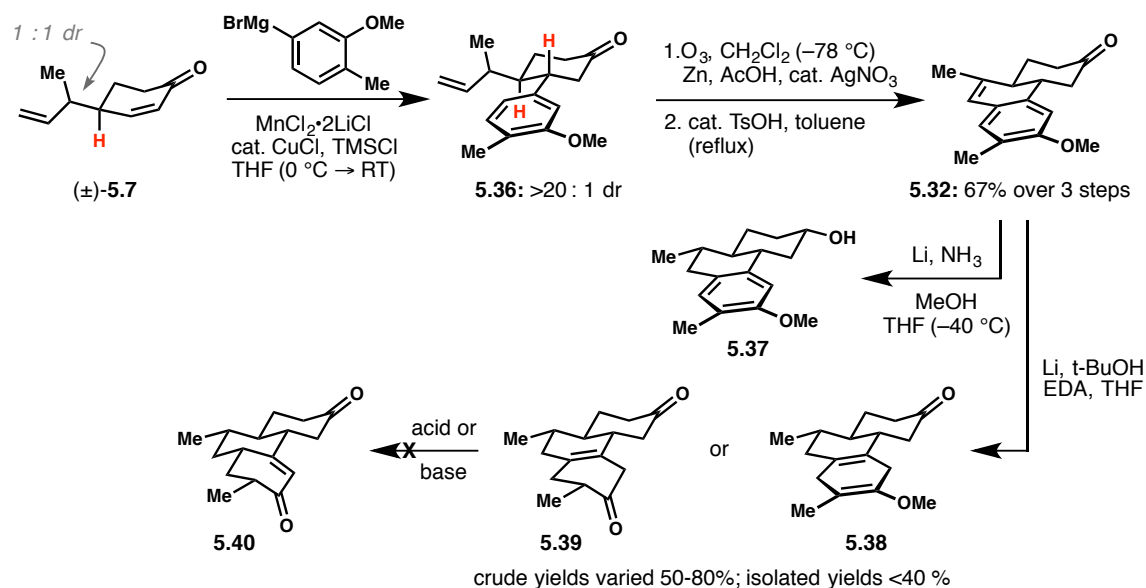


Cuprate addition into enone **5.7** using Cahiez organomanganate<sup>19</sup> conditions afforded cleaner conversion to **5.36** than previously under Retz<sup>10</sup> conditions and was advantageous in carrying **5.36** forward without purification (Scheme 5.8). Ozonolysis and quench by zinc reduction also facilitated in moving the aldehyde through the Friedel–Crafts cyclodehydration in crude form. Dihydronaphthalene **5.32** was thus generated in 67% yield over three steps with a single purification, an advantage over the three purifications previously required. At this point, dissolving metal reduction using Li/NH<sub>3</sub> and methanol once again failed to reduce the aromatic moiety as had been previously observed with the *cis*-diastereomer of **5.32** (see Chapter 4). The ethylene diamine conditions<sup>20</sup> adapted in Chapter 4 performed just as expected and reduced the dihydronaphthalene successfully, but not the ketone of **5.32**. Depending on the workup



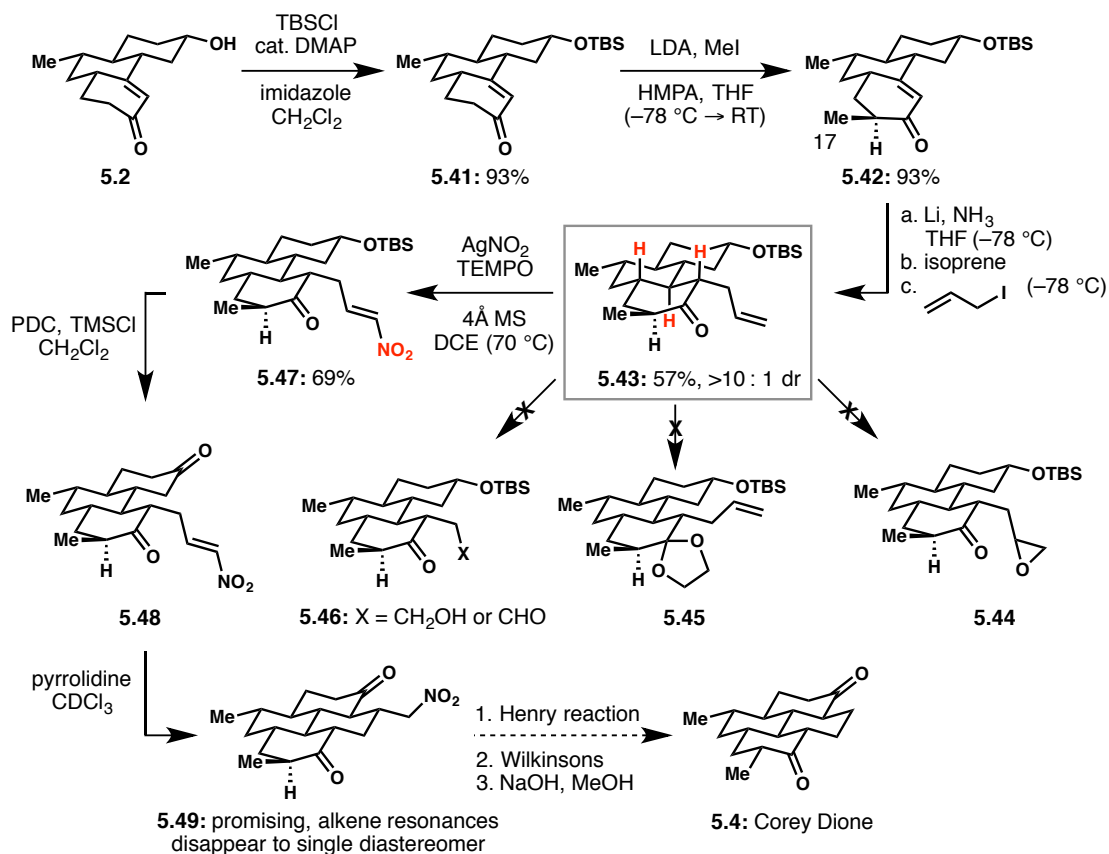
conditions, either the cyclohexadiene **5.38** (using aqueous  $\text{NH}_4\text{Cl}$ ) or the skipped enone **5.39** (using aqueous  $\text{HCl}$ ) could be isolated, although both in relatively low yields. A short screen of acidic and basic conditions did not effectively provide the desired conjugated enone and therefore attention was turned to methyl group installation by enone alkylation.

**Scheme 5.8** Efforts to install the C17 methyl group from the aromatic Grignard reagent.



The alcohol of **5.2** was protected and the methyl group was installed by action of LDA and methyl iodide (Scheme 5.9). At this point, no effort was made in addressing the strategy of targeting diol **5.26** for the  $\text{Sc}(\text{OTf})_3$  mediated TMS-CN displacement methodology (Scheme 5.7C); instead, the C20 ketone was maintained for another charge to Corey's dione **5.4**. Enone reduction with carefully measured equivalents of Li followed by allylation afforded **5.43** with excellent diastereocontrol.

**Scheme 5.9** C17 Methyl installation by alkylation and subsequent ring closure strategies.



Tricycle **5.43** became the platform for numerous ring closure attempts. Epoxidation with mCPBA, DMDO or trifluoroperoxyacetic acid led only to multiple unidentifiable products. Under standard ketal protection conditions (TsOH in refluxing benzene), **5.43** was recovered unchanged, while toluene at reflux afforded several unidentified compounds. Similarly to the previous attempts on **5.20**, ozonolysis or dihydroxylation and diol cleavage of **5.43** also formed multiple unidentified products.

A reaction that functioned surprisingly well in this context was radical alkene nitration. Several conditions<sup>21–24</sup> were tried but the most successful procedure<sup>25</sup> involved heating a sealed vial without rigorous exclusion of air or moisture, containing substrate, excess AgNO<sub>2</sub>, excess TEMPO, molecular sieves and DCE at 70 °C for 30 hours. At this point deprotection of the tert-

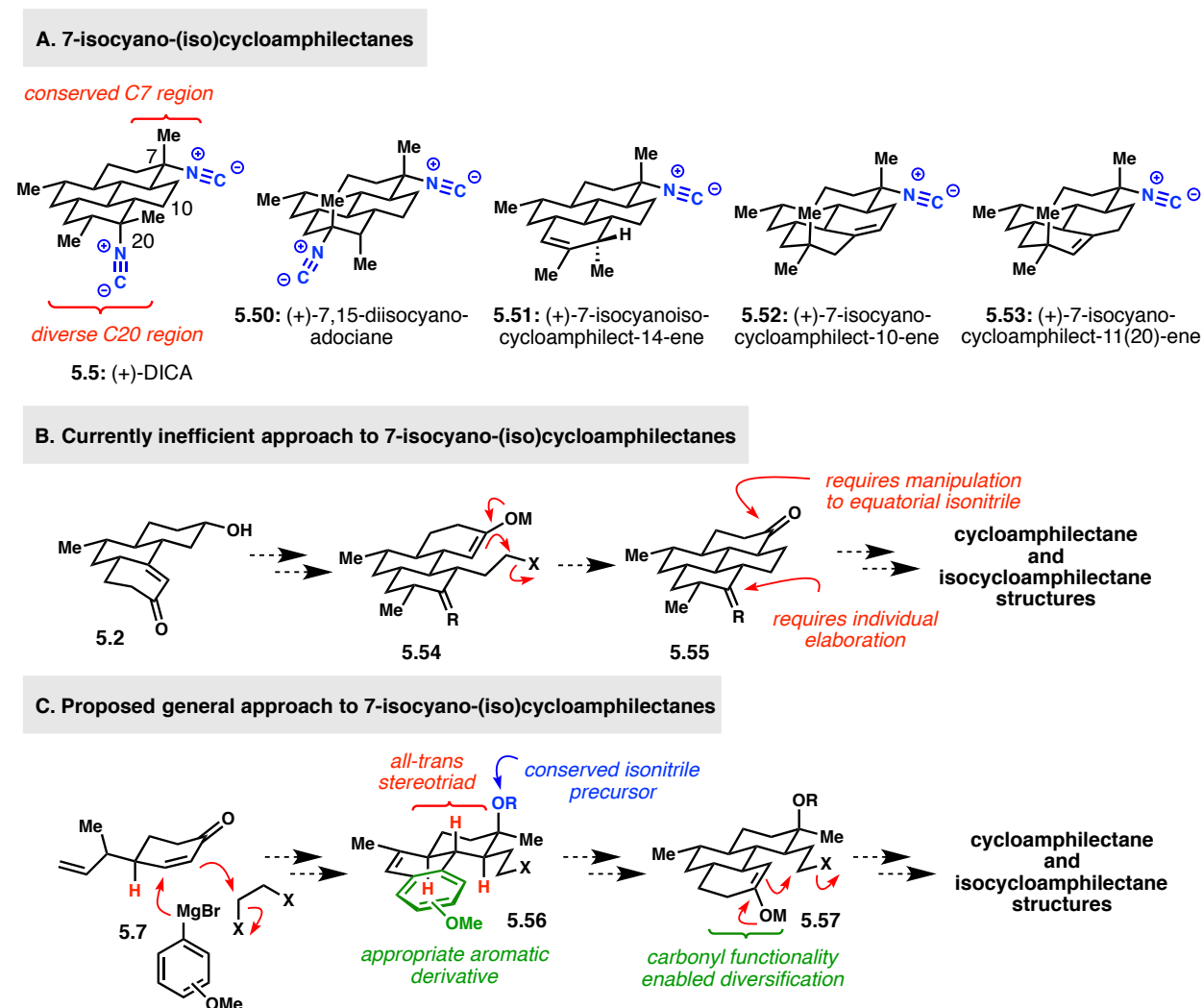
butyldimethylsilyl group failed using TBAF, aqueous HCl, TsOH or HF. The sensitive nitroalkene most likely complicated this operation. Luckily, in situ generated trimethylsilyl chlorochromate<sup>26</sup> was successful in direct deprotection and oxidation to **5.47**. Only one conjugate addition was attempted with the small quantities of material obtained. Addition of pyrrolidine to **5.48** in CDCl<sub>3</sub> led to almost immediate disappearance of the alkene resonances by <sup>1</sup>H NMR to afford a single diastereomer of an enamine-containing compound of unknown structure. Although this result was likely a first success in forging the tetracyclic scaffold of DICA, at this point a broader analysis of the ICT family<sup>27</sup> was performed and a slightly modified, more global approach was given attention.

## **5.4 A Global Approach to 7,20-Diisocyanoadociane and Other (Iso)Cycloamphilectanes**

### *5.4.1 The Evolution to a C10–C11 Ring Closing Strategy*

The close structural similarity in ICT natural products provides an opportunity not just to synthesize a single member, but instead to develop a more general approach. A number of cycloamphilectanes and isocycloamphilectanes exhibit a conserved C7 region and a diversified C20 region (Scheme 5.10A). A general approach would rest on a common advanced intermediate, from which **5.5**, **5.50–5.53** are accessed. The current C8–C9 ring closing approach to DICA requires a late stage manipulation of both the C7 and C20 regions to access (iso)cycloamphilectanes since a ketone at C7 is essential for ring closure (Scheme 5.10B). Ideally, an advanced common intermediate should contain a conserved C7 isonitrile precursor and appropriate functionality to diversify the C20 region (Scheme 5.10C). A C7 axial tertiary alcohol would be ideal as an isonitrile precursor since Sc(OTf)<sub>3</sub> and TMSCN could readily install the equatorial isonitrile.<sup>14,15</sup> A possible structure that meets these requirements is **5.56**.

**Scheme 5.10** Cycloamphilectane and isocycloamphilectanes with a conserved C7 and diverse C20 region.

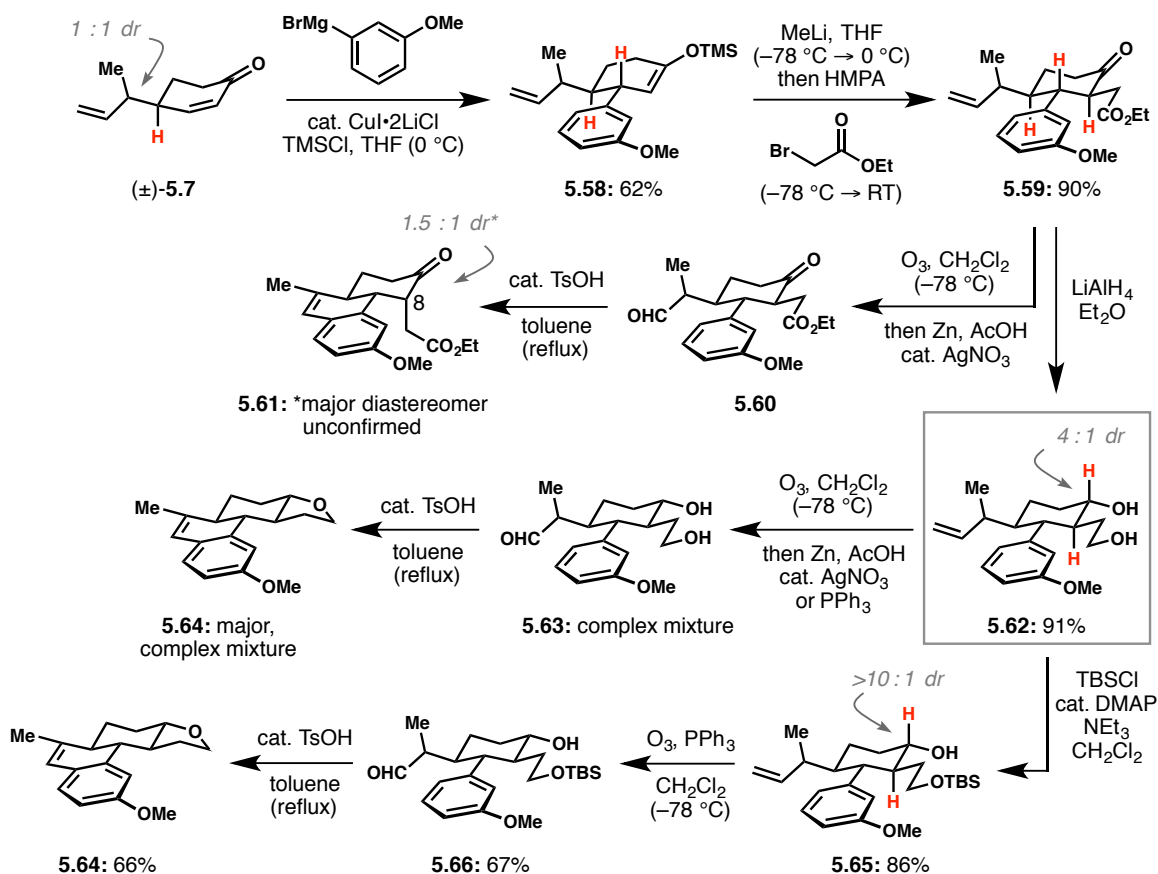


Using a tertiary alcohol at C7 has several advantages (Scheme 5.10C). First, this idea requires revising the ring closing location to the C10–C11 bond, suggesting the key dihydronaphthalene intermediate **5.56** could be obtained via a vicinal difunctionalization of **5.7**. Also, the C7 tertiary alcohol locks the *all-trans* stereotriad generated by vicinal difunctionalization and enables selective manipulation of the C20 region to **5.5**, **5.50–5.53**. A first proof-of-principle synthesis to determine the feasibility of this approach would be access to Corey's dione **5.4**.

#### 5.4.2 A Proof-of-Principle Preparation of a Perhydropyrene Relevant to 7,20-Diisocyanoadociane

A unified approach to (iso)cycloamphilectane was initiated by examining the vicinal difunctionalization of **5.7** (Scheme 5.11). The enoxysilane generated upon cuprate addition could be retained with careful workup and purification using Florisil® silica gel column chromatography. Revealing the free enolate of **5.58** was accomplished with MeLi and alkylation with ethyl bromoacetate accessed **5.59** in >20:1 selectivity for the all-*trans* stereotriad. Ozonolysis and acid-mediated cyclodehydration generated the required carbon skeleton of **5.61**, but in 1.5 : 1 dr. The acidic conditions caused epimerization of the C8 position<sup>13</sup> to avoid negative steric interaction between the aromatic and ester side chain. This epimerization issue was resolved by reducing **5.59** to diol **5.62**. Ozonolysis of **5.62** led to a complex mixture of products that could be taken on in crude form to the Friedel–Crafts cyclodehydration. However, instead of obtaining the expected dihydronaphthalene bearing a diol, **5.64** was isolated; the diol had condensed to a THF ring. Trying to avoid this acid mediated cyclization, the primary alcohol of **5.62** was TBS protected. Mono-TBS ether **5.65** was taken through the sequence, but was also found to cyclize to THF **5.64**. Although other protecting group schemes could be devised, the diol's propensity to cyclize indicated a path forward.

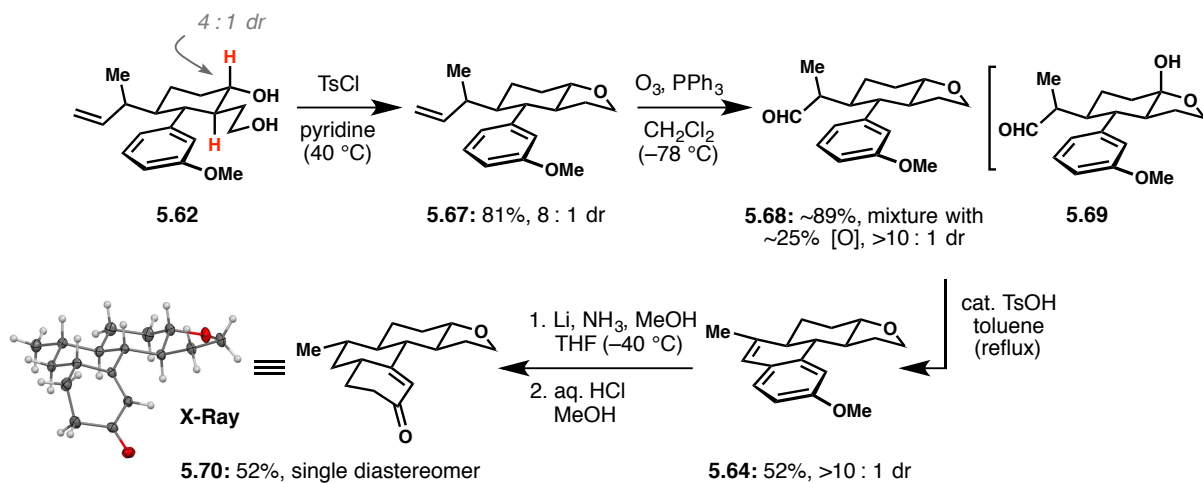
**Scheme 5.11** Exploring the reactivity after a vicinal difunctionalization.



Capitalizing on the inherent reactivity of the molecule, the diol was intentionally condensed into a THF ring. This diol masking strategy enabled a smooth transformation through the Birch reduction phase (Scheme 5.12). Tosylation of the primary alcohol and intramolecular displacement generated the *trans*-perhydrobenzofuran **5.67**. Ozonolysis of this compound provided the desired aldehyde **5.68**, but contaminated with an unknown oxidation product. Retrospectively, this could be the C–H oxidized **5.69** (see Section 5.4.3 for more details). Cyclodehydration of **5.68** in the usual way generated dihydronaphthalene **5.64** that was subjected to Birch reduction and isomerization to readily afford **5.70** in excellent diastereocontrol. X-ray crystallographic analysis confirmed the structure of **5.70**. With the general reactivity established, but seeking to avoid the tedious intermediacy of enoxysilane **5.58**, the procedure was optimized

(Scheme 5.13).

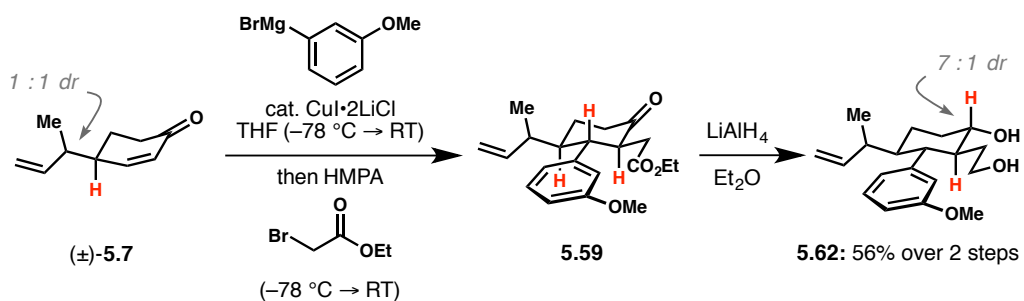
**Scheme 5.12** An initial procedure through a successful Birch reduction.



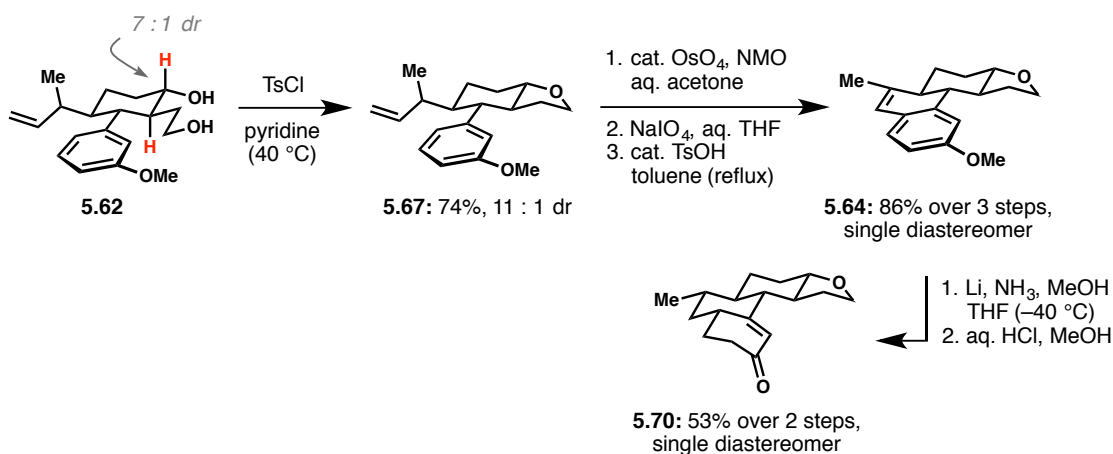
Conjugate addition of 3-methoxymagnesium bromide operated effectively without added TMSCl (Scheme 5.13A). The obtained enolate was alkylated with ethyl bromoacetate, but required 30 hours at room temperature for satisfactory conversion. Reduction of the crude reaction product with LiAlH<sub>4</sub> provided diol **5.62** in an improved 7 : 1 dr, most likely because of the residual HMPA. At this point two options exist for generating **5.70**. The first involved enriching the major diastereomer over the course of multiple steps to eventually arrive at **5.70** as a single diastereomer (Scheme 5.13B). Since for this first foray of targeting Corey's dione **5.4** the C7 alcohol<sup>13</sup> epimer is inconsequential, both diastereomers can also be carried through to **5.70** (Scheme 5.13C).

**Scheme 5.13** An optimized vicinal difunctionalization and subsequent synthetic elaboration.

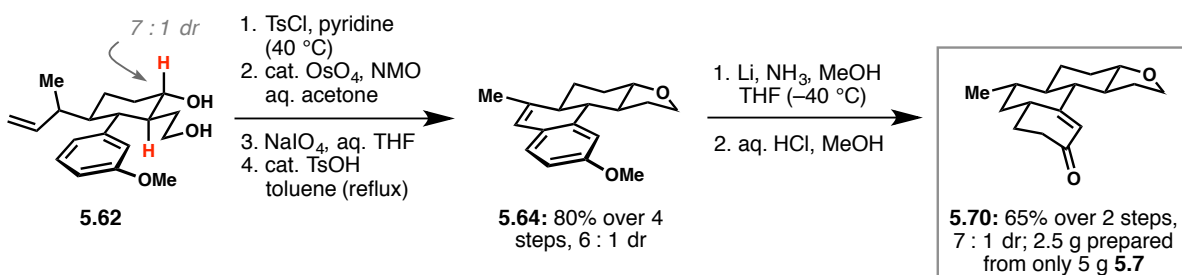
**A. A reliable vicinal difunctionalization and reduction**



**B. Procedure to obtain 5.70 as a single diastereomer**



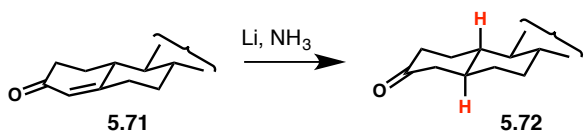
**C. Procedure carrying a mixture of inconsequential diastereomers through**



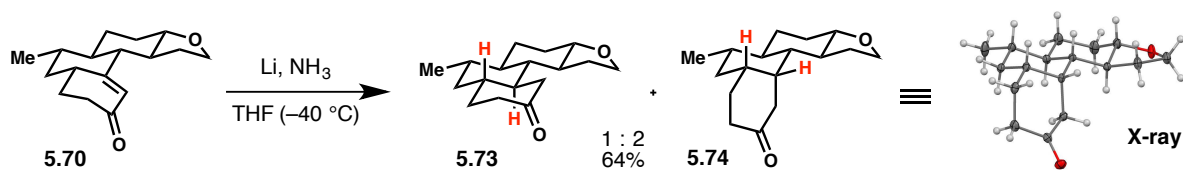


**Scheme 5.14** Another unexpected stereochemical outcome from a dissolving metal reduction.

**A. Classic dissolving metal reduction to set a *trans*-ring fusion from an enone**



**B. A surprising stereochemical outcome upon dissolving metal reduction**

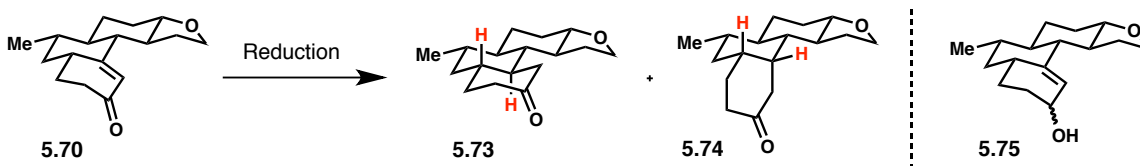


The next step of reducing enone **5.70** was expected to be accomplished with Li/NH<sub>3</sub>, as it is widely used to generate *trans*-decalin ring junctions (Scheme 5.14A).<sup>11,12,28</sup> Surprisingly, treating **5.70** with Li/NH<sub>3</sub> afforded a 2 : 1 mixture of ketones, favoring the *cis*-ring junction **5.74** (Scheme 5.14B). Further exploring the reducing metal did not change the selectivity dramatically (Table 5.1, entries 1–2), while adding a bulky proton source simply improved *cis*-selectivity (entry 3). Selecting for the *trans*-ring fusion proved challenging in this case, most likely due to the *syn*-pentane interaction present in the key reduction intermediate (Figure 5.1). Homogeneous hydride reagents such as Karstedt's catalyst<sup>29</sup> and t-BuCu/DIBAL enhanced the *cis*-selectivity, even though the latter is exceptional in providing *trans*-selectivity in other decalin systems<sup>30</sup> and the Hajos–Parrish ketone<sup>31</sup> (entries 5–6). Fortunately, heterogeneous reducing agents provided mixtures significantly enriched in the *trans* product **5.73** (entries 7–9). A small screen of hydrogenation catalysts led to the discovery that reductions with Rh/C were highly *trans*-selective (15 : 1 dr). High-pressure reductions with Pt/C and Ru/C fared poorly, giving trace reaction or the allylic alcohol **5.75**.

Since the *trans*-reduction of **5.70** using Rh/C provided high selectivity, only two more

transformations are required: (1) THF ring opening and alkylation and (2) installation of the C17 methyl group by alkylation (Scheme 5.15).

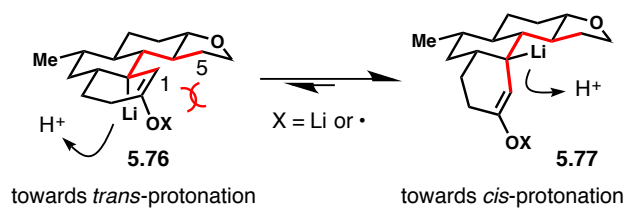
**Table 5.1** Optimizing for a *trans*-reduction of enone **5.70**.



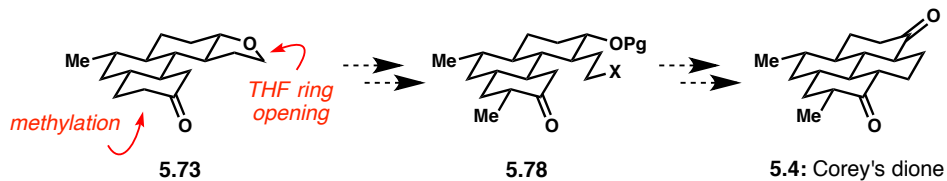
Entry	Reduction	Yield <sup>a</sup>	5.73: <i>trans</i>	5.74: <i>cis</i>
1	Na, NH <sub>3</sub> , THF (-78 °C)	85%	1	: 1
2	K, NH <sub>3</sub> , THF (-78 °C)	82%	1	: 1
3	K, t-BuOH, NH <sub>3</sub> , THF (-78 °C)	80%	1	: 3
4	Mg, MeOH	<b>decomposition</b>		
5	1. cat. Karstedt, Et <sub>3</sub> SiH (70 °C) 2. TBAF, THF	92%	1	: 5
6	t-BuCu, DIBALH, HMPA, THF (-50 °C)	86%	1	: >20
7	H <sub>2</sub> balloon, Pd/C, EtOAc	94%	6	: 1
8	400 psi H <sub>2</sub> , Rh/alumina, EtOAc <sup>b</sup>	93%	8	: 1
9	500 psi H <sub>2</sub> , Rh/C, EtOAc <sup>b</sup>	93%	15	: 1
10	1300 psi H <sub>2</sub> , Pt/C, EtOAc	<b>trace</b>		
11	1300 psi H <sub>2</sub> , Ru/C, EtOAc	<b>5.75 as major</b>	~10	: 1

<sup>a</sup> Yield of purified material after column chromatography <sup>b</sup> 2. PCC, Celite, CH<sub>2</sub>Cl<sub>2</sub>

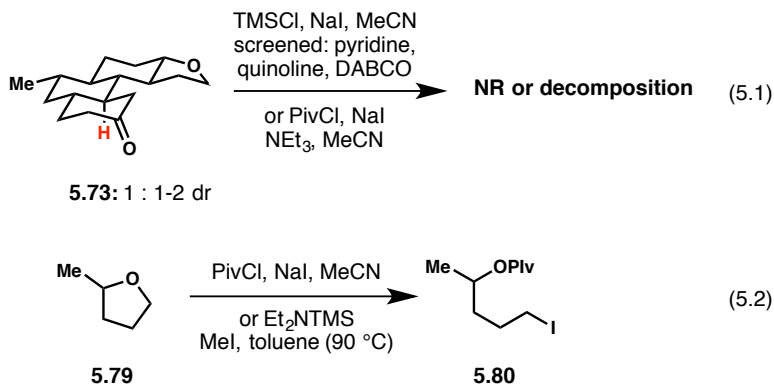
**Figure 5.1** Comparison of protonation events to explain the preference for *cis*-reduction.



**Scheme 5.15** Remaining requirements to complete Corey's dione.



The planned THF ring opening was to be facilitated by a Lewis acid/nucleophile combination.<sup>32-41</sup> Only a couple conditions were screened, and were unsuccessful in opening the THF ring of **5.73** (Equation 5.1). Evaluating select conditions on 2-methyltetrahydrofuran (**5.79**) showed success only at concentrations of 1 Molar and on greater than 0.5 mmol scale (Equation 5.2). At the time, large quantities of **5.73** were not available and the material on hand was considered too precious to commit. A nucleophilic ring opening, a C–H oxidation and reductive ring opening advancement was pursued instead.

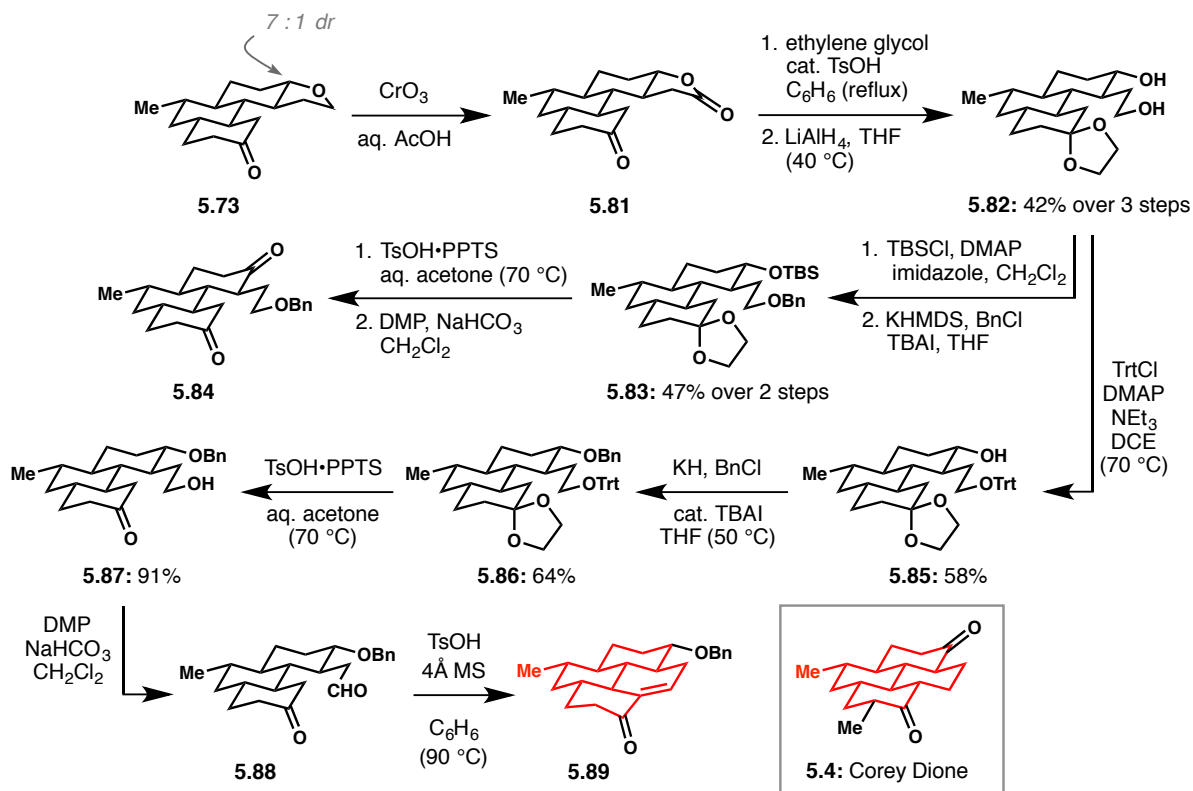


One of the first C–H oxidation reactions transforming **5.73** to lactone **5.81** was successful (Scheme 5.16). Stirring **5.73** in the presence of a slight excess of CrO<sub>3</sub> in aqueous acetic acid (Fieser's reagent) afforded **5.81**.<sup>42</sup> The C20 ketone<sup>13</sup> was protected and the lactone opened with LiAlH<sub>4</sub> to generate diol **5.82**. At this point the secondary diol needed to be protected. A scheme involving double protection followed by deprotection of the primary alcohol was envisioned. Silylation of the primary alcohol followed by benzyl protection smoothly generated **5.83**. It was discovered after continuing with the sequence that the TBS group migrated to the secondary

alcohol during benzyl protection. Deprotection of **5.83** with buffered acid and oxidation afforded a ketone and not an aldehyde carbonyl.

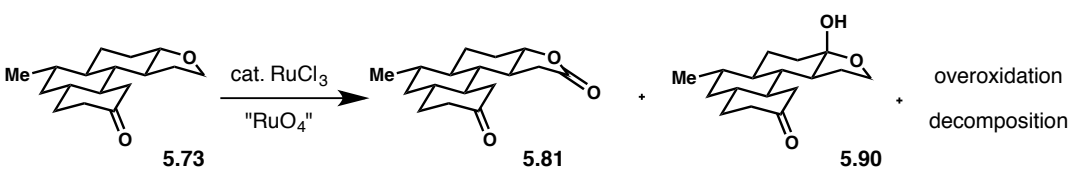
The successful sequence started with a trityl protection of **5.82**, followed by benzyl protection. Then, acidic conditions deprotected both the trityl and ketal groups to afford **5.87**. The fourth ring was cyclized by alcohol oxidation and then aldol condensation between the C20 ketone and C10 aldehyde to afford enone **5.89** in sub-milligram quantities.<sup>13</sup> This marks the preparation of all four rings of DICA in the all-*trans* relationship by means of a dihydronaphthalene reduction route.

**Scheme 5.16** A successful fourth ring closure.



### 5.4.3 Improving the THF C–H Oxidation to a Butyrolactone

The C–H oxidation to lactone **5.81** was a low yielding reaction with CrO<sub>3</sub> and required improvement.<sup>43</sup> A standard method of THF to lactone oxidation is the use of RuO<sub>4</sub>.<sup>44,45</sup> A number of different conditions were tried, but unfortunately none was more effective than Fieser's reagent (Table 5.2). Although all reactions supposedly generate RuO<sub>4</sub>, each set of conditions provided a different product profile. In addition to lactone **5.81**, competitive oxidation provided lactol **5.90** and three other oxidation products in variable quantities. A general trend showed that more polar organic reaction media favored formation of **5.90**, while the less polar CCl<sub>4</sub> generated parity between **5.81** and **5.90** without preference. Inexplicably, co-oxidant H<sub>5</sub>IO<sub>6</sub> led to complete formation of lactone **5.81**; however, heavy side oxidation was problematic (entry 6). Overall, the choice of co-oxidant impacted not only product distribution but also side product formation. These results are a curious anomaly since all reagent combinations should form the same "RuO<sub>4</sub>" and would be predicted to react identically.

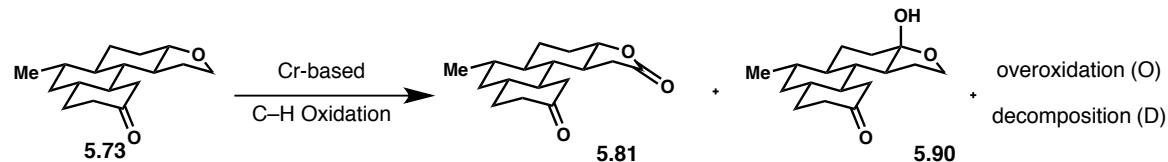
**Table 5.2** Screening RuO<sub>4</sub> mediated THF C–H Oxidations.


Entry	Conditions	SM <sup>a</sup>	5.81	5.90	5.81 : 5.90	Unidentified Products A, B & C
1	EtOAc, MeCN aq. Na <sub>2</sub> EDTA NaHCO <sub>3</sub> , oxone	20%	15%	50%	1 : 3.3	15%
2	NaHCO <sub>3</sub> , NaIO <sub>4</sub>	8%	21%	71%	1 : 3.4	trace
3	CCl <sub>4</sub> , aq. MeCN NaHCO <sub>3</sub> , oxone	45%	13%	32%	1 : 2.5	10%
4	NaHCO <sub>3</sub> , NaIO <sub>4</sub>	10%	33%	33%	1 : 1	24%
5	NaHCO <sub>3</sub> , NaOCl		33%	67%	1 : 2	trace
6	H <sub>5</sub> IO <sub>6</sub>		16%		>20 : 1	26% 57%
7	NaHCO <sub>3</sub> , NaIO <sub>4</sub> aq. acetone	59%	7%	34%	1 : 4.9	
8	Pb(OAc) <sub>4</sub> , aq. AcOH	46%				18% 36%

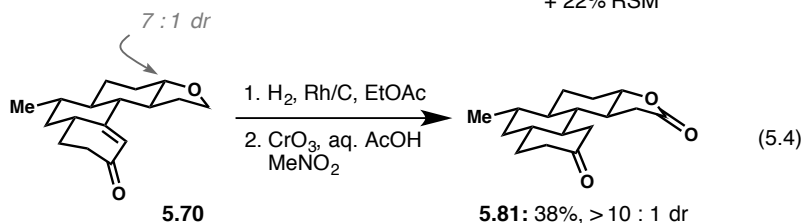
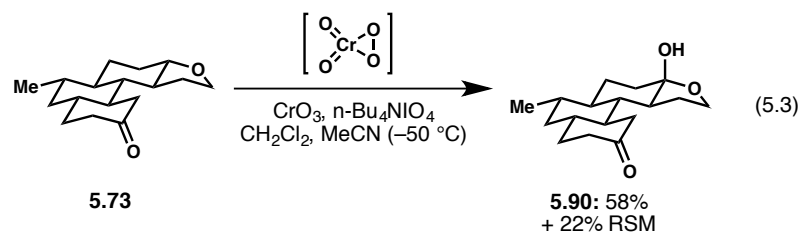
<sup>a</sup> Percentages were obtained by <sup>1</sup>H NMR integration without internal standard

Early successes with Fieser's reagent guided the exploration of more Cr-based C–H oxidation conditions.<sup>46</sup> Lactone **5.81** was frequently observed during these studies (Table 5.3), but disappointingly, these reactions were also contaminated with unidentified overoxidation products and general decomposition. From this compilation of oxidations three interesting results are of further note. Jones reagent (CrO<sub>3</sub> in aqueous sulfuric acid) was highly selective in providing only **5.81** without side oxidations. Unfortunately, the reaction was plagued by low conversions and low mass recovery. Only trace reactivity was observed in the absence of sulfuric acid. The Fuchs reagent (chromyl peroxide "CrO<sub>4</sub>") selectively and cleanly provided only lactol **5.91** with recovered starting material (Equation 5.3).<sup>47</sup> No other oxidation products or decomposition was observed. Alternatively, concentrated Fieser's reagent dissolved in MeNO<sub>2</sub> improved the yield of **5.81** to 38% (Equation 5.4). The increased yield is most likely to due to the ease with which MeNO<sub>2</sub> is worked up rather than improvement in the course of the oxidation.

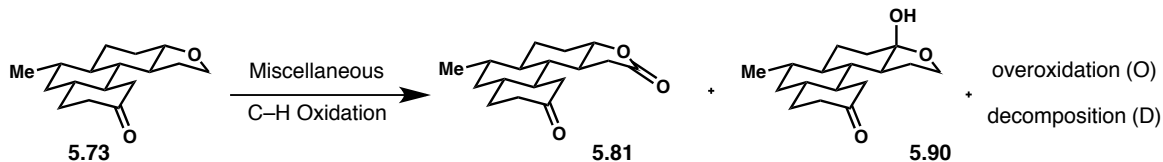
**Table 5.3** A Cr-based screen for C–H oxidation.



Entry	Conditions	Results	Corresponding Literature
1	CrO <sub>3</sub> , aq. AcOH in MeCN, acetone, DMF, C <sub>6</sub> H <sub>6</sub> , CH <sub>2</sub> Cl <sub>2</sub> or MeNO <sub>2</sub>	<b>5.81</b> , O, D	among others: Wettstein, A.; Mischer, K. <i>Helv. Chim. Acta</i> <b>1942</b> , <i>25</i> , 718
2	CrO <sub>3</sub> , aq. H <sub>2</sub> SO <sub>4</sub> in AcOH, MeCN, acetone, Et <sub>2</sub> O, CH <sub>2</sub> Cl <sub>2</sub> , C <sub>6</sub> H <sub>6</sub> or MeNO <sub>2</sub>	<b>5.81</b> , D	among others: (1) Brown, H. C.; Garg, C. P.; Liu, K.-T. <i>J. Org. Chem.</i> <b>1971</b> , <i>36</i> , 387 (2) Kropp, P. J.; Worsham, P. R.; Davidson, R. I.; Jones, T. H. <i>J. Am. Chem. Soc.</i> <b>1982</b> , <i>104</i> , 3972
3	CrO <sub>3</sub> , Ac <sub>2</sub> O, C <sub>6</sub> H <sub>6</sub> (up to reflux)	D	Frauenrath, H.; Philipps, T. <i>Liebigs Ann. Chem.</i> <b>1985</b> , 1951
4	CrO <sub>3</sub> , TMSO <sub>2</sub> , MeCN, CH <sub>2</sub> Cl <sub>2</sub>	<b>5.81</b> , O, D	Shahi, S. P.; Gupta, A.; Pitre, S. V.; Reddy, M. V. R.; Kumareswaran, S.; Vankar, Y. D. <i>J. Org. Chem.</i> <b>1999</b> , <i>64</i> , 4509
5	CrO <sub>3</sub> , pyridine, DCE (up to 50 °C)	NR	Okabe, M.; Abe, M.; Tada, M. <i>J. Org. Chem.</i> <b>1982</b> , <i>47</i> , 1775
6	CrO <sub>3</sub> , 3,5-DMP, CH <sub>2</sub> Cl <sub>2</sub> (–20 °C up to RT)	<b>5.81</b> , O, D	Salmond, W. G.; Barta, M. A.; Havens, J. L. <i>J. Org. Chem.</i> <b>1978</b> , <i>43</i> , 2057
7	PCC, H <sub>5</sub> IO <sub>6</sub> , MeCN	<b>5.81</b> , O, D	Piccialli, V.; Zaccaria, S.; Oliviero, G.; D'Errico, S.; D'Atri, V.; Borbone, N. <i>Eur. J. Org. Chem.</i> <b>2012</b> , 4293
8	PCC, Celite, C <sub>6</sub> H <sub>6</sub> (reflux)	NR	
9	PDC, t-BuO <sub>2</sub> H, C <sub>6</sub> H <sub>6</sub>	<b>5.81</b> , O, D	Chidambaram, N.; Chandrasekaran, S. <i>J. Org. Chem.</i> <b>1987</b> , <i>52</i> , 5048
10	CrO <sub>3</sub> , n-Bu <sub>4</sub> NIO <sub>4</sub> , CH <sub>2</sub> Cl <sub>2</sub> , MeCN (–50 °C)	<b>5.90</b>	Lee, S.; Fuchs, P. L. <i>Org. Lett.</i> <b>2004</b> , <i>6</i> , 1437



**Table 5.4** Miscellaneous C–H oxidation attempts to install a lactone functionality.



Entry	Conditions	Results	Corresponding Literature
1	Pb(OAc) <sub>4</sub> , C <sub>6</sub> H <sub>6</sub> (reflux)	NR	
2	KMnO <sub>4</sub> /CuSO <sub>4</sub> ·5H <sub>2</sub> O w/ and w/out alumina CH <sub>2</sub> Cl <sub>2</sub> , DCE, neat (up to reflux)	1 : 2 to 1 : 1 <b>5.81</b> / <b>5.90</b> O, D	among others: Zhao, D.; Lee, D. G. <i>Synthesis</i> <b>1994</b> , 910
3	O <sub>3</sub> , SiO <sub>2</sub>	<b>5.81</b> , O, D	Cohen, Z.; Keinan, E.; Mazur, Y.; Varkony, T. H. <i>J. Org. Chem.</i> <b>1975</b> , <i>40</i> , 2141
4	CoCl <sub>2</sub> or Co(acac) <sub>3</sub> , O <sub>2</sub> DME (reflux)	NR	Reetz, M. T.; Töllner, K. <i>Tetrahedron Lett.</i> <b>1995</b> , <i>36</i> , 9461
5	PhI(OAc) <sub>2</sub> , t-BuO <sub>2</sub> H, MeNO <sub>2</sub>	D	Zhao, Y.; Ang, J. Q. L.; Ng, A. W. T.; Yeung, Y.-Y. <i>RSC Adv.</i> <b>2013</b> , <i>3</i> , 19765
6	Ca(OCl) <sub>2</sub> AcOH, MeCN	D	Nwaukwa, S. O.; Keehn, P. M. <i>Tetrahedron Lett.</i> <b>1982</b> , <i>23</i> , 35
7	KBrO <sub>3</sub> , NaHSO <sub>4</sub> aq. CH <sub>2</sub> Cl <sub>2</sub>	NR	Metsger, L.; Bittner, S. <i>Tetrahedron</i> <b>2000</b> , <i>56</i> , 1905
8	TFDO, aq. MeCN (0 °C)	<b>5.90</b> , D	Curci, R.; D'Accolti, L.; Fiorentino, M.; Fusco, C.; Adam, W.; González-Nuñez, M. E.; Mello, R. <i>Tetrahedron Lett.</i> <b>1992</b> , <i>33</i> , 4225

A variety of conditions that are not Ru or Cr-based are also known to provide THF oxidation (Table 5.4). A handful of these were screened in the context of **5.73**. Strong oxidants like KMnO<sub>4</sub> once again competitively underwent 3° C–H oxidation to lactol **5.91** (Entry 2). In situ generated TFDO oxidized **5.73** to only **5.91**, although not as cleanly as with CrO<sub>4</sub> (Equation 5.3). After having spent effort evaluating a wide variety of C–H oxidation reactions, CrO<sub>3</sub> in acetic acid and MeNO<sub>2</sub> could be considered as the currently optimal conditions.

#### 5.4.4 Finishing the Synthesis of Corey's Dione

An improved preparation of **5.89** and completion of Corey's intermediate was pursued next (Scheme 5.17). The first improvement was the conversion of **5.81** to **5.87** without intermediate purification. The alcohol oxidation and acid-mediated aldol condensation afforded

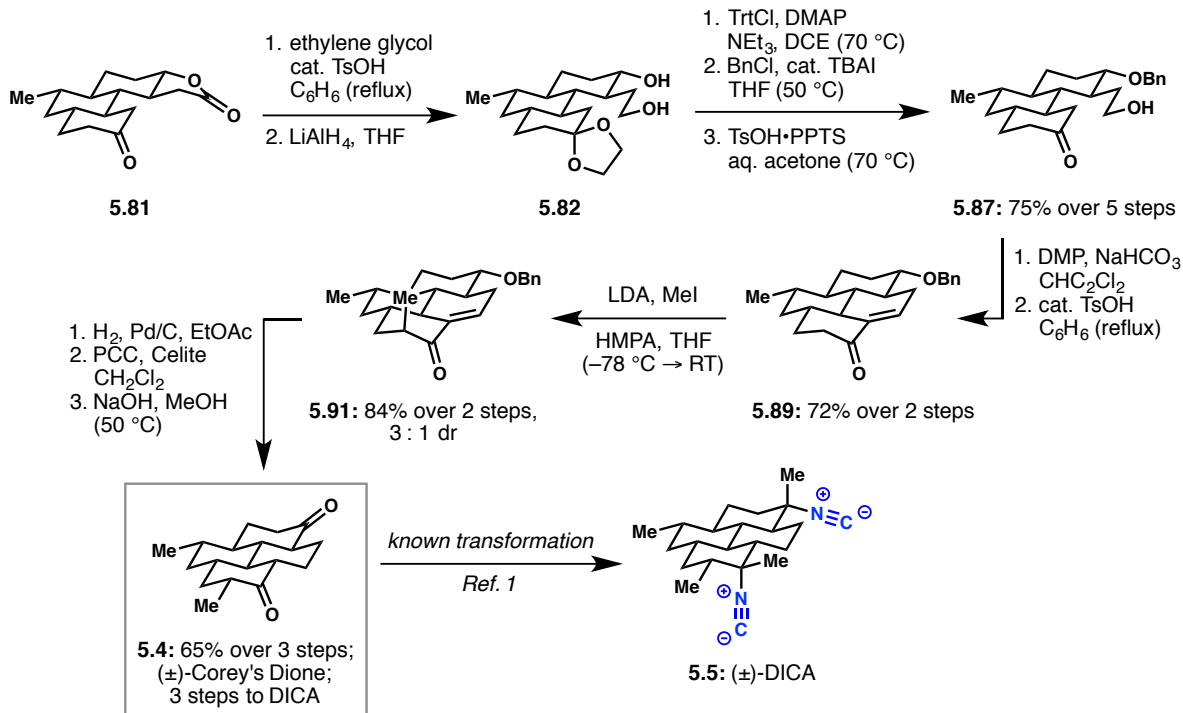


enone **5.89** efficiently. The cross-conjugated dienolate derived from enone **5.89** was methylated with a moderate axial preference; after enone hydrogenation/benzyl ether hydrogenolysis and oxidation of the C7 alcohol, base-mediated equilibration afforded Corey's dione **5.4**. Access to this target, completed a formal synthesis of racemic DICA.

Corey's intermediate was prepared in a total of 24 steps and a longest linear sequence of 24 steps from commercially available crotyl alcohol and tri(ethyleneglycol) divinyl ether. Highlights of this synthesis include the telescoping of several steps and the high degree of stereocontrol. Corey's dione **5.4** was prepared using only two distillations and eight chromatographic purifications. Excellent relative stereochemical control at all eight centers of the perhydrophyrene scaffold was obtained; all stereocenters were introduced with >20 : 1 dr, except C12 which was installed in 15 : 1 dr. The 1.7% overall yield from commercial material is unoptimized. To date this report is the shortest and most selective synthesis of DICA. Introduction of the isonitriles has not been attempted, but was described by Corey.<sup>1</sup> This synthesis affirms that the general design in Scheme 5.10C is useful in preparing the isocycloamphilectane scaffold.

This proof-of-principle synthesis opens several directions for future effort. A top priority was rendering the synthesis enantioselective. Additionally, accessing Corey's dione currently does not selectively generate DICA since the isonitrile installation is not stereorcontrolled. A second-generation synthesis needs to address selective installation of these salient isonitriles. During the next charge, the redox and protecting group steps of the current iteration will need to be reworked. And lastly, the dihydronaphthalene Birch reduction design needs to be applied to other isocycloamphilectanes and cycloamphilectanes.

### Scheme 5.17 End game to Corey's Dione.

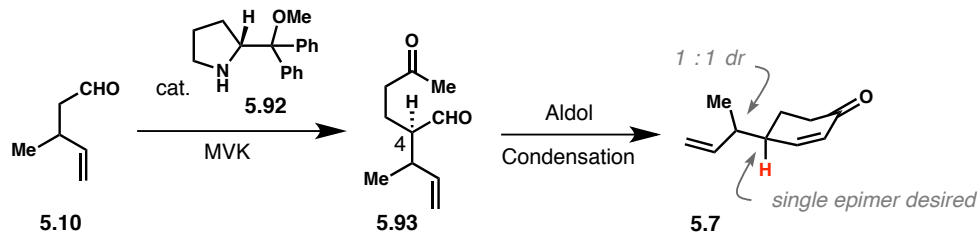


## 5.5 An Asymmetric Formal Synthesis of (+)-7,20-Diisocyanoadociane

### 5.5.1 The Asymmetric Robinson Annulation Reaction

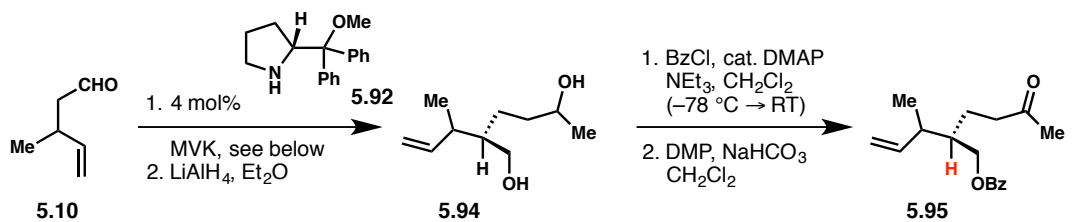
From the outset, the synthesis of DICA was intended to be asymmetric. The attractiveness of generating cyclohexenone **5.7** via Robinson annulation was that its preparation could be rapidly adapted asymmetric via chiral catalysis.<sup>4,8,48-51</sup> Starting from racemic **5.10**, an asymmetric conjugate addition using Gellman's catalyst<sup>4</sup> would set the C4 stereocenter (Scheme 5.18). The complicated part of the annulation is retaining the aldehyde stereocenter while enolizing the methyl ketone for aldol condensation. Several reports indicate that this is a surmountable challenge.<sup>6,8,48,49</sup>

**Scheme 5.18** General approach to enantioenriched **5.7**.



A goal for beginning this synthesis was the scalability and operational ease for at least the early steps. This was accomplished for the racemic synthesis of cyclohexenone **5.7** and consisted of two straightforward distillations. The prolinol catalyzed conjugate addition of aldehydes onto vinyl ketones requires significant quantities of catechol additives under the literature conditions. This additional reagent requires removal by column chromatography and was therefore avoided. With this intended divergence from literature conditions, the enantioselectivity of the conjugate addition needed evaluating (Table 5.5).

**Table 5.5** Screening the asymmetric conjugate addition reaction.



Entry	Solvent <sup>a</sup>	Conditions <sup>a</sup>	Yield of 5.94	dr (d1 : d2) <sup>b</sup>	Diastereomer 1 (%ee) <sup>b</sup>	Diastereomer 2 (%ee) <sup>b</sup>
1	neat	RT	7%	c	c	c
2	Et <sub>2</sub> O	RT	7%	52 : 48	81	84
3	MeOH	RT	24%	c	c	c
4	neat	40 °C	40%	51 : 49	88	89
5	MeOH	40 °C	77%	49 : 51	81	80

<sup>a</sup> MeCN, toluene, EtOAc (RT) gave < 3% yield

<sup>b</sup> Analyzed by chiral HPLC

<sup>c</sup> Chiral HPLC analysis was inconclusive

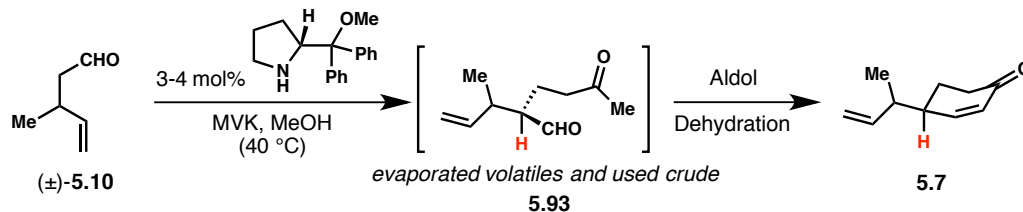
Since removing the catechol significantly suppressed reactivity (Table 5.5, entry 1), a solvent and temperature screen was performed. A spectrum of solvents ranging from non-polar

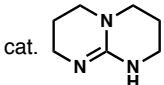
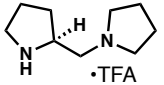
(toluene), medium polarity (Et<sub>2</sub>O, EtOAc) to polar (MeCN) and polar protic (methanol) was evaluated. Methanol provided the best reactivity, affording diol **5.94** in 77% yield over 2 steps after immediate LiAlH<sub>4</sub> reduction (entry 5). Derivatization to the primary benzoylated product **5.95** and chiral HPLC analysis showed both diastereomers to have ~80% ee. Although the enantiomeric excess was not ideal, a fluorinated version of prolinol catalyst **5.92** has been shown to catalyze conjugate additions to methyl vinyl ketone in 95% ee without catechol additive and could be used instead.<sup>51</sup> At the time, this prolinol catalyst was simply not considered worth the effort to prepare.

With an enantioselective conjugate addition available, the next consideration became retention of the newly forged stereocenter during the intramolecular aldol addition. A number of conditions were tried (Table 5.6). Literature precedent of using catalytic LiOH in isopropanol<sup>8</sup> or phase-transfer conditions of Bu<sub>4</sub>NOH in aqueous THF/Et<sub>2</sub>O<sup>49</sup> afforded <15% yields and mostly decomposition. Literature conditions of using DBU provided almost complete racemization (entries 1–2).<sup>48</sup> The more basic and nucleophilic TBD slightly improved retention of stereochemistry, but still only provided maximum 46% ee (entries 4–8). The McQuade catalyst was most successful since a good yield of **5.7** was obtained (entry 9), but a 66% ee and an enantiospecificity of 83% was still not satisfactory. The best conservation of stereochemical information was obtained with LDA, but the low yield precluded its further implementation (entry 10).

Although the asymmetric Robinson annulation did not render the synthesis highly enantioselective, this reaction still provided material that was moderately enantioenriched. A more effective asymmetric entry to DICA and thereby the general family of 7-isocyano(iso)cycloamphilectane was still desired.

**Table 5.6** Conditions to effect the aldol condensation of enantioenriched Michael adduct **5.93**.



Entry	Aldol	Dehydration	Yield <sup>a</sup>	Diastereomer 1 (% ee / % es) <sup>b</sup>	Diastereomer 2 (% ee / % es) <sup>b</sup>
1	DBU, CH <sub>2</sub> Cl <sub>2</sub>	then MsCl	59%	4 / 5	6 / 8
2	"wet" DBU, CH <sub>2</sub> Cl <sub>2</sub>	MsCl, cat. DMAP, NEt <sub>3</sub> , CH <sub>2</sub> Cl <sub>2</sub>	54%	4 / 5	6 / 8
3	TsOH, MeCN		62%	34 / 43	32 / 40
	cat. 				
4	THF	then MsCl, cat. DMAP, NEt <sub>3</sub>	59%	32 / 40	34 / 43
5	THF	then Ac <sub>2</sub> O, cat. DMAP, NEt <sub>3</sub>	66%	32 / 40	32 / 40
6	MeCN	then TsOH	68%	34 / 43	36 / 45
7	CH <sub>2</sub> Cl <sub>2</sub>	then MsCl, cat. DMAP, NEt <sub>3</sub>	56%	34 / 43	36 / 45
8	MeOH	MsCl, cat. DMAP, NEt <sub>3</sub> , CH <sub>2</sub> Cl <sub>2</sub>	55%	46 / 58	46 / 58
9		cat.  ·TFA	62%	66 / 83	66 / 83
		hexanes (RT, 24 h)			
10	LDA, THF (-78 °C)	MsCl, cat. DMAP NEt <sub>3</sub> , CH <sub>2</sub> Cl <sub>2</sub>	23% <sup>c</sup>	76 / 95	78 / 98
11	10 mol% t-BuOK THF (0 °C)	MsCl, cat. DMAP NEt <sub>3</sub> , CH <sub>2</sub> Cl <sub>2</sub>	43%	62 / 78	56 / 70

<sup>a</sup> Yields reflect material after silica-gel column chromatography

<sup>b</sup> Analyzed by chiral HPLC; es = [(%ee<sub>5,7</sub>) / (%ee<sub>5,95</sub>)] x 100; dr ~ 1 : 1 for all reactions

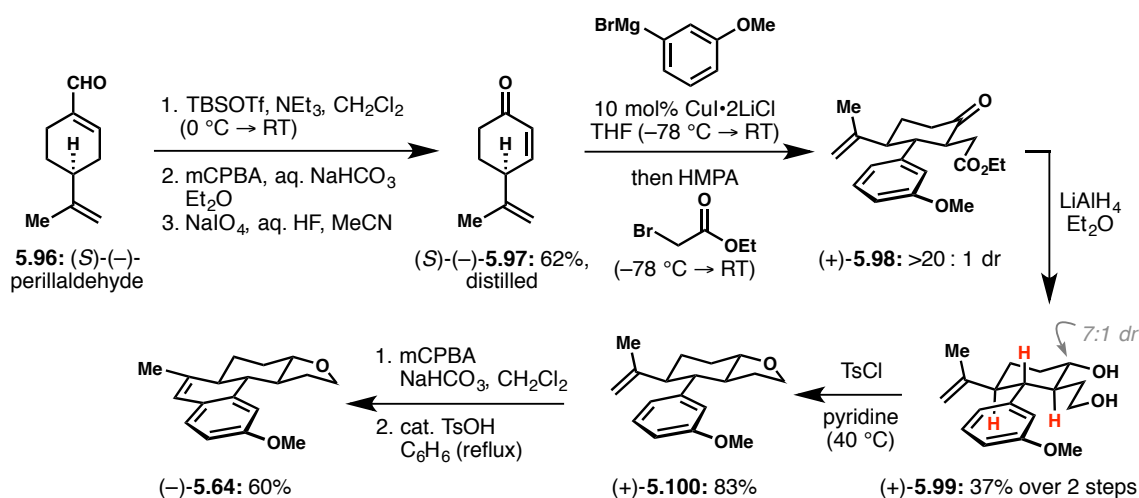
<sup>c</sup> 27% Michael adduct isolated

### 5.5.2 A Perillaldehyde-Based Formal Synthesis

After attempts at improving a less than ideal asymmetric Robinson annulation, a new enantiocontrolled entry was required. Inspiration for a new starting point came from analysis of the synthesis in hand. While the preparation of dihydronaphthalene **5.64** requires an oxidative

cleavage of the alkene, this intermediate aldehyde could instead arise from a dehomologated alkene via oxidation. This insight identified the known cyclohexenone (–)-**5.97**, available in enantiopure form,<sup>52,53</sup> as an attractive starting material. The inexpensive terpene (–)-perillaldehyde’s conversion to cyclohexenone (–)-**5.97** on gram scale has been previously reported;<sup>53</sup> however, purification by distillation was preferable to improve throughput (Scheme 5.19).

**Scheme 5.19** An enantiospecific formal synthesis of (+)-DICA from (–)-perillaldehyde.



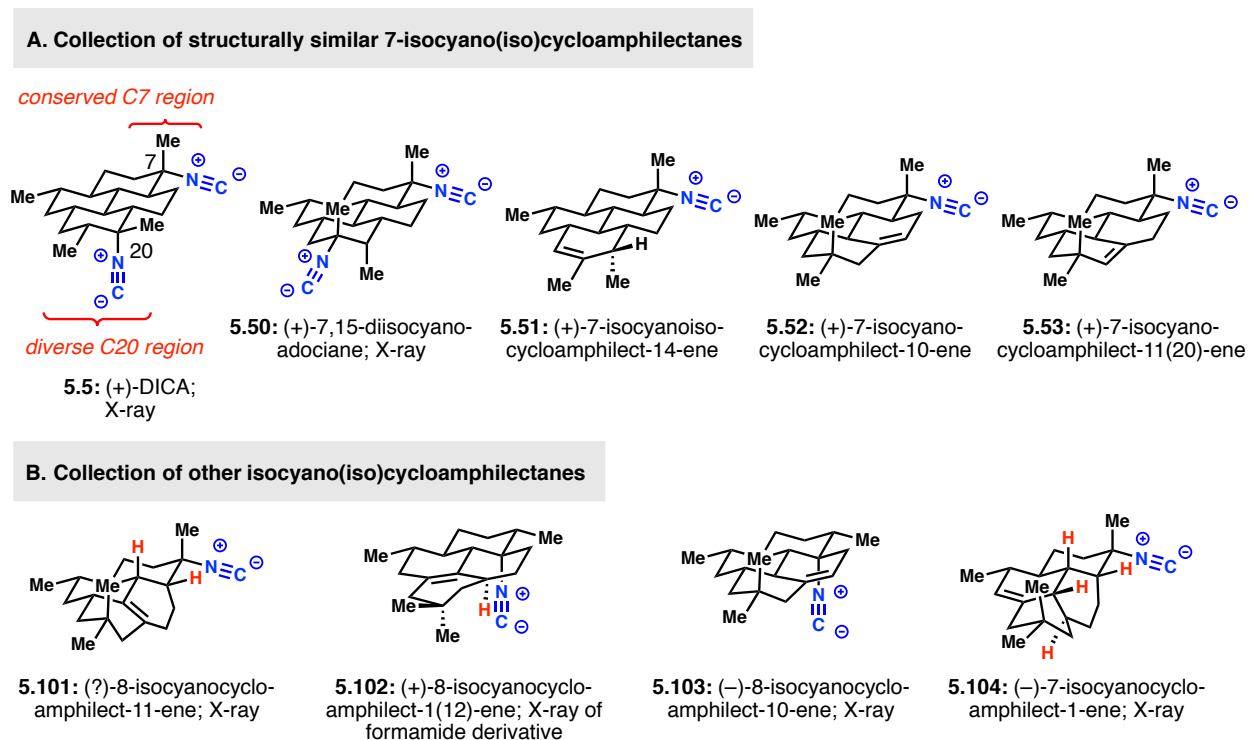
In an unoptimized sequence, conjugate addition and alkylation selectively installed the three stereogenic centers with desired diastereoselectivity as seen in (+)-**5.99**, in a similar manner previously shown to make **5.62**. Condensation of diol (+)-**5.99** via the tosylate furnished *trans*-fused THF (+)-**5.100**. Alkene epoxidation was followed by heating at reflux with acid, which presumably triggered epoxide rearrangement to the aldehyde followed by in situ cyclodehydration to dihydronaphthalene (–)-**5.64**. This sequence intersects the racemic synthesis of DICA described earlier (Section 5.4), permitting access to all later intermediates in enantiopure form. Because of the desire to improve several aspects of the overall synthesis, the

formal synthesis of DICA using this optically active material was not revisited, though it is clear that this chiral pool approach is suitable for doing so with respect both to control of absolute configuration and material throughput concerns. In its current form, this route is still 24 steps in longest linear sequence and total steps from perillaldehyde with only 10 purifications in an unoptimized form.

## 5.6 Outlook Towards the Synthesis of 7-Isocyano-(Iso)Cycloamphilectanes

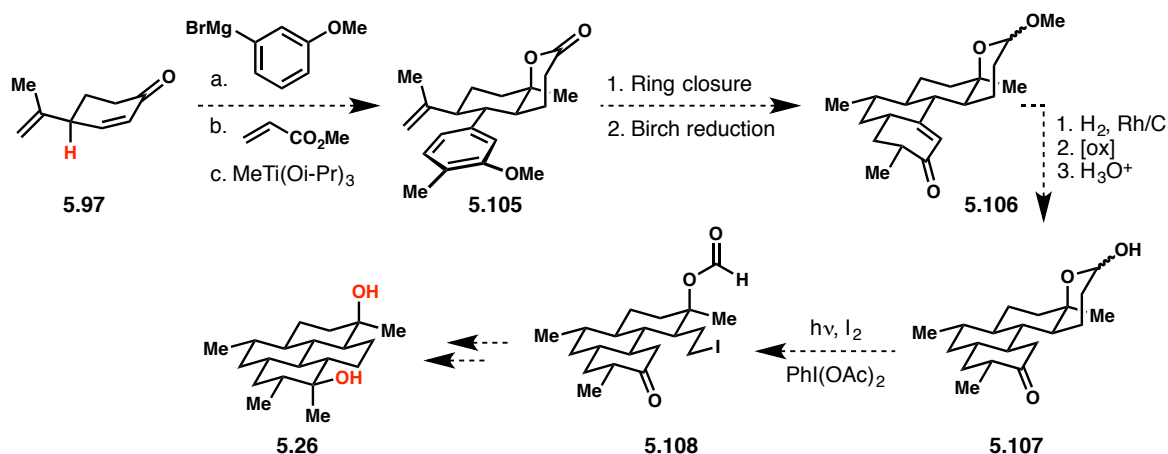
Having accomplished the formal synthesis of DICA in asymmetric form, two directions can be envisioned for this project: (1) adapt the knowledge gained about stereocontrol to a concise synthesis of DICA that includes introduction of the salient isonitriles, and (2) showcase the generality of this approach to other (iso)cycloamphilectane ICTs (Figure 5.2).<sup>27</sup>

**Figure 5.2** A collection of (iso)cycloamphilectane natural products.



To improve the synthesis of DICA, several aspects of the current formal synthesis need to be addressed, mainly the drawback of forming the THF ring. Ideally, the functionality at C10 should be directly amenable to ring closure and not require extensive redox and protecting group manipulations. Additionally, to address selective introduction of the isonitriles, the current state-of-the-art  $\text{Sc}(\text{OTf})_3$ -mediated invertive isonitrile installation would be most attractive. In a forward sense, the stereocontrol elements used in the formal synthesis would be applied to a second-generation synthesis. The synthesis could start from the same perillaldehyde derived enone **5.97** (Scheme 5.20). Several routes could be imagined, but the following idea would address several of the required improvements. A conjugate addition/Michael reaction would set the required stereocenters *trans*. Selective methyl addition into the ketone from the equatorial face would close the lactone to **5.105**.<sup>19,54–57</sup> Enone **5.106** would be formed by the previously used acid-induced cyclization and Birch reduction. Hydrogenation would generate the last internal stereocenter and sets up **5.107** for a photoinduced  $\beta$ -fragmentation of an alkoxy radical to generate alkyl iodide **5.108**.<sup>58,59</sup> Alkylation and methylation would generate the appropriate diol for an invertive isonitrile displacement.

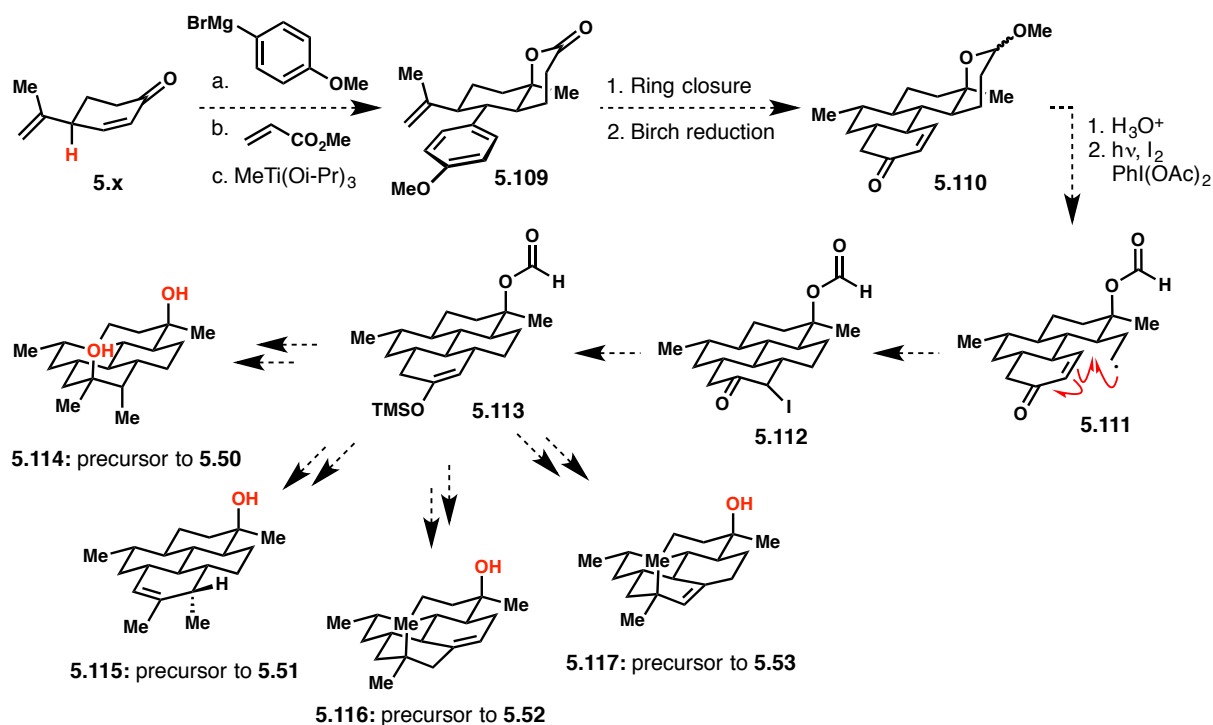
**Scheme 5.20** Proposed access to diol **5.26** utilizing reductive stereocontrol elements.





The formation of the perhydropyrene scaffold via alkylation and reduction previously used to access DICA could also be implemented to prepare other 7-isocyano(iso)cycloamphilectanes (Scheme 5.21). A conjugate addition of 4-methoxyphenylmagnesium bromide puts the oxidation state in a more favorable position in comparison to 3-methoxyphenylmagnesium bromide. The oxygen radical fragmentation may directly add into the C11–C20 electron deficient alkene of **5.111** or be trapped with  $I_2$  and set up for an anionic conjugate addition. Alternatively,  $SmI_2$  combined with the Suárez photofragmentation may allow for nucleophilic conjugate addition in the presence of the radical fragmentation. The enoxysilane **5.113** bears all the handles necessary to prepare **5.114–5.117**, the direct precursors to their respective ICTs (Figure 5.2).

**Scheme 5.21** General approach to 7-isocyano(iso)cycloamphilectanes.



## 5.7 Conclusions

The formal synthesis of DICA was accomplished by a continuous evolution of ideas. The failures of previous routes led to both subtle and dramatic changes in strategies that culminated in a highly stereoselective perhydropyrene synthesis using a tandem vicinal difunctionalization, dissolving metal reduction and hydrogenation reactions. When examining the synthesis as a whole, it becomes clear how overwhelmingly important the carbonyl group was in this effort. A carbonyl was used in every productive C–C bond constructing step, using either its natural electrophilic reactivity, nucleophilic tendency in the forms of the enolate, or activating ability as an electron withdrawing group. The main takeaway from this adventure has been the development of excellent stereocontrol strategies with the power to address a variety of perhydropyrene ICT (iso)cycloamphilectanes.

## 5.8 Experimental Procedures

### Purifications –

Solvents: Dry tetrahydrofuran (THF), diethyl ether (Et<sub>2</sub>O), dichloromethane (CH<sub>2</sub>Cl<sub>2</sub>), toluene, benzene (C<sub>6</sub>H<sub>6</sub>), acetonitrile (MeCN), dimethylformamide (DMF) and methanol (MeOH) were obtained by passing commercially available formulations through activated alumina columns. tert-Butyl alcohol (t-BuOH) was purified by distillation over CaH<sub>2</sub>.

Amines: Diisopropylamine (i-Pr<sub>2</sub>NH), triethylamine (NEt<sub>3</sub>), pyridine (py), and hexamethylphosphoramide (HMPA) were purified by distillation from CaH<sub>2</sub>. Ethylene diamine (EDA) was purified by distillation from sodium metal.

Chlorides and Triflates: tert-Butyldimethylsilyl trifluoromethanesulfonate (TBSOTf), trimethylsilyl chloride (TMSCl), and benzyl chloride (BnCl) were purified by distillation over CaH<sub>2</sub>.

Metals: Copper(I) iodide (CuI) was purified by Soxhlet extractor with THF then drying the solid under high vacuum. Copper(I) chloride (CuCl) was purified by dissolving in conc. HCl then adding water until all material precipitated. The material was collected by filtration, washed generously with water, ethanol then ether and dried in vacuo. Manganese(II) chloride tetrahydrate was azeotropically dried with toluene under a Dean–Stark apparatus, then concentrated in vacuo.

Miscellaneous: Methyl vinyl ketone (MVK) was purified by distillation. Allyl iodide was purified by distillation and stored over copper beads at –20 °C. Ethyl bromoacetate was purified by washing thrice with 2 M Na<sub>2</sub>CO<sub>3</sub>, twice with brine, drying over MgSO<sub>4</sub>, filtering and distillation.

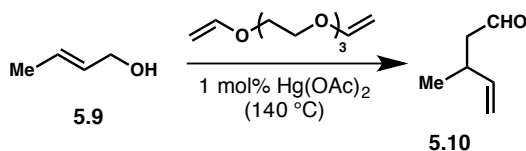
**Titration** – Alkyl lithium reagents were titrated using 2,6-di-(tert-butyl)-4-methylphenol (BHT) as the sacrificial proton source and fluorene as an indicator in THF or using diphenylacetic acid in THF. Grignard reagents were titrated using salicylaldehyde phenylhydrazone in THF.<sup>60</sup>

**Reaction Setup** – All reactions were carried out under an argon atmosphere with dry solvents under anhydrous conditions, unless otherwise noted. Argon balloons were the sole inert atmosphere used. Reactions run at an ambient temperature of 20–25 °C are designated as room temperature. Yields refer to chromatographically and spectroscopically homogeneous materials, unless otherwise stated.

**Analysis** – Thin layer chromatography was performed on 0.25 mm EMD glass-backed TLC plates impregnated with a fluorescent dye and visualized with UV light and KMnO<sub>4</sub> in K<sub>2</sub>CO<sub>3</sub>/NaOH/water or *p*-anisaldehyde in ethanol/aqueous H<sub>2</sub>SO<sub>4</sub>/AcOH and heat as a developing agent. Forced flow (flash) chromatography was performed on EMD Silica 60, mesh 0.04-0.063 silica gel. NMR spectra were recorded on Bruker 500 MHz instrument, obtained at 298 K unless otherwise noted and calibrated to residual undeuterated solvent as an internal reference. Chemical shifts are reported in ppm with the following abbreviations to explain multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, quin = quintuplet, sext = setet, sep = septet, bs = broad signal, m = multiplet. All coupling constants are apparent *J* values measured at the indicated field strengths. FT-IR spectra were recorded on a Perkin-Elmer spectrum RX1 spectrometer. High-resolution mass spectra (HRMS) were recorded on a H2Os LCT Premier spectrometer using ESI-TOF (electrospray ionization-time of flight). Melting points were measured on a MEL-TEMP II capillary apparatus and stand uncorrected. Optical rotations were

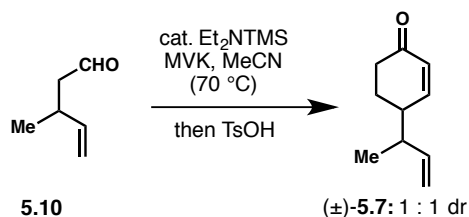
measured with a Jasco P-1010 polarimeter operating on the sodium D-line (589 nm) using a 50 mm path-length cell and are reported as:  $[\alpha]_D^T$  (concentration in g/100 mL, solvent). Analytical chiral HPLC was performed with an Agilent 110 Series HPLC using a Chiralpak AS-H column (0.46 cm x 25 cm) obtained from Daicel Chemical Industries Ltd. and Chiralpak AD-H (0.46 cm x 25 cm) obtained from Daicel Chemical Industries Ltd. with visualization at the mentioned wavelength.

### 3-Methyl-4-pentenal **5.10** [Adapted from the literature]<sup>7,9</sup>



A 500 mL round bottom flask was charged with a stir bar, 90 g (1.25 mol) **5.9**, 250 mL (1.25 mol) tri(ethylene glycol) divinyl ether and 2.01 g (6.35 mmol)  $\text{Hg}(\text{OAc})_2$ . An excellent reflux condenser was greased<sup>61</sup> and fitted, then the flask immersed in a  $140\text{ }^\circ\text{C}$  oil bath for 8 hours. The reaction was cooled to room temperature and an additional 1.98 g (6.21 mmol)  $\text{Hg}(\text{OAc})_2$  was added before the reaction was reheated to  $140\text{ }^\circ\text{C}$ . After 20 hours the reaction was cooled, the reflux condenser replaced for a simple distillation head and distilled ( $80\text{ }^\circ\text{C}/200\text{ mmHg}$ ) directly into an iced receiving flask to afford 95 g (77%) **5.10** as a colorless liquid. Spectral data was identical to the literature.<sup>7</sup> Note: The general procedure for Wei et al.<sup>7</sup> palladium catalyzed *trans*-vinylation/Claisen rearrangement was inferior in our hands to the procedure described above using mercury. Although mercury is highly toxic, this procedure was more reliable in advancing material. All glassware was washed with concentrated nitric acid after use, then diluted with water.

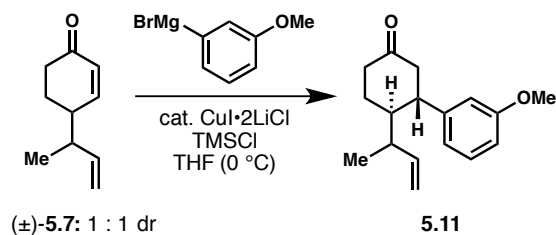
### Cyclohexenone ( $\pm$ )-**5.7**



A 1 L round bottom flask was charged with 9.80 g (99.8 mmol) **5.10**, 280 mL MeCN, 10 mL (123 mmol) MVK and 3.5 mL (18.5 mmol)  $\text{Et}_2\text{NTMS}$ . The flask was fitted with a reflux

condenser and immersed into a 90 °C oil bath for 21 hours. The reaction was cooled to room temperature and 38 g (200 mmol) TsOH•H<sub>2</sub>O added. After 8 hours the reaction was cooled in an ice bath and 120 mL 1 M NaOH added, followed by 120 mL sat. aq. NaHCO<sub>3</sub>, ensuring the temperature remained below 10 °C. Approximately 280 mL of volatiles were removed in vacuo and the remaining liquid diluted with 90 mL water and 200 mL Et<sub>2</sub>O. The phases were separated and the aqueous phase extracted twice with 100 mL Et<sub>2</sub>O. All organic layers were combined, washed twice with 75 mL sat. aq. NaHCO<sub>3</sub> and back extracted with 30 mL Et<sub>2</sub>O. The organic layers were combined, washed with 50 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by distillation (77-80 °C/1.0 mmHg) to afford 10.0 g (66%, 1:1 dr) **5.7** as a colorless liquid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 6.94 (dt, *J* = 10.3, 2.0 Hz, 0.5H), 6.91 (ddd, *J* = 10.3, 2.3, 1.6 Hz, 0.5H), 6.02 (dt, *J* = 10.3, 2.6, 0.5H), 6.02 (dt, *J* = 10.3, 2.6, 0.5H), 5.75 (m, 1H), 5.09-5.05 (m, 2H), 2.54 (t, *J* = 4.3 Hz, 0.5H), 2.50 (t, *J* = 4.3 Hz, 0.5H), 2.47-2.29 (m, 3H), 2.08-2.01 (m, 1H), 1.78 (m, 1H), 1.09 (d, *J* = 7.2 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 199.89, 199.87, 153.5, 153.4, 141.2, 141.0, 129.8, 129.7, 115.4, 114.9, 41.7, 41.2, 41.1, 40.9, 37.4, 37.2, 26.1, 26.0, 17.2, 16.4; IR (thin film) 3077, 2963, 2871, 1682 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>10</sub>H<sub>14</sub>O [M+Na]<sup>+</sup> 173.0942 found 173.0950.

## Cyclohexanone 5.11

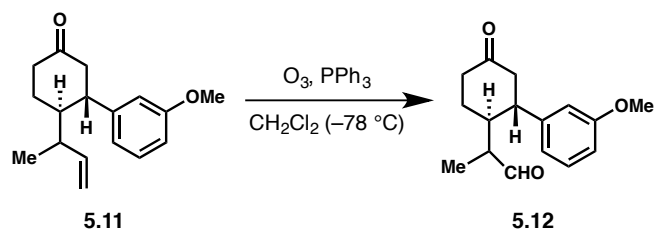


3-Methoxyphenylmagnesium bromide was prepared by addition of 3.0 mL (23.7 mmol) 3-bromoanisole in 3.0 mL THF to 0.865 g (35.6 mmol) magnesium metal in 12 mL THF activated by dibromoethane, maintaining reflux. To a 250 mL round bottom flask charged with 2.00 g (13.3 mmol) **5.7**, 0.255 g (0.133 mmol) CuI, 0.125 g (2.95 mmol) LiCl and 1.9 mL (15.0 mmol) TMSCl in 80 mL THF, cooled to 0 °C was added 13.5 mL (16.2 mmol) 1.2 M 3-methoxyphenylmagnesium bromide dropwise over 30 minutes. The reaction was stirred for an additional hour then quenched at 0 °C by addition of 50 mL 1 M HCl and warmed to room temperature. The mixture was extracted with 50 mL and 25 mL Et<sub>2</sub>O. The organic layer was washed with 40 mL 3 M HCl, 20 mL water, 40 mL 3:1 sat. aq. NH<sub>4</sub>Cl/5 M NaOH, 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (10:1 hexanes/EtOAc) to afford 2.67 g (78%, 1:1 dr) **5.11** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.28-7.25 (m, 2H), 6.79 (td, *J* = 15.7, 7.6 Hz, 6H), 5.81 (ddd, *J* = 17.0, 10.7, 6.1 Hz, 1H), 5.67 (dt, *J* = 17.8, 8.9 Hz, 1H), 5.00 (d, *J* = 10.5 Hz, 2H), 4.93 (d, *J* = 17.3 Hz, 1H), 4.82 (d, *J* = 17.1 Hz, 1H), 3.82 (s, 6H), 2.88 (td, *J* = 20.4, 11.3 Hz, 2H), 2.59-2.41 (m, 8H), 2.22-2.09 (m, 4H), 2.05-1.99 (m, 2H), 1.55 (quintetd, *J* = 12.9, 4.9 Hz, 2H), 1.01 (d, *J* = 7.0 Hz, 3H), 0.84 (d, *J* = 6.9 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 210.69, 210.66, 159.9, 159.8, 145.0, 144.8, 143.3, 138.8, 129.9, 129.8, 119.8, 119.5, 115.7, 113.6, 113.5, 113.4, 111.5, 111.5, 55.2, 49.7, 49.5, 48.2, 47.7, 46.4, 45.7, 41.1, 41.0, 37.2,



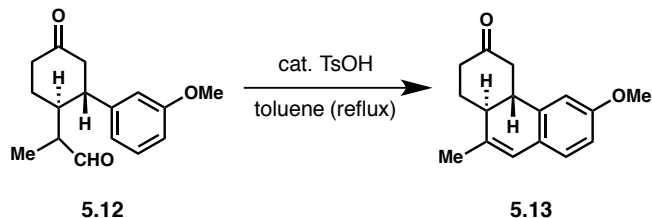
36.2, 25.3, 25.1, 18.9, 11.5; IR (thin film) 2959, 1714, 1600, 1487, 1263, 1046  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{17}\text{H}_{22}\text{O}_2$   $[\text{M}+\text{Na}]^+$  281.1518 found 281.1511.

### Aldehyde 5.12



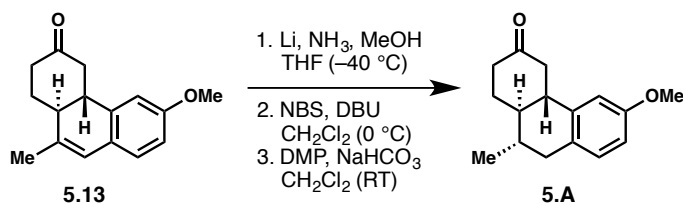
A 250 mL round bottom flask containing 2.67 g (10.3 mmol) **5.11** in 100 mL  $\text{CH}_2\text{Cl}_2$  was treated with ozone at  $-78^\circ\text{C}$ . After a persistent blue color, the color was discharged by bubbling oxygen and 3.22 g (12.2 mmol)  $\text{PPh}_3$  added. The reaction was allowed to warm to room temperature over several hours. All volatiles were removed and the residue purified by column chromatography (3:1 hexanes/EtOAc) to afford 2.40 g (88%, 1:1 dr) **5.12** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  9.59 (s, 1H), 9.47 (s, 1H), 7.28 (q,  $J = 8.0$  Hz, 2H), 6.79 (td,  $J = 17.7, 7.7$  Hz, 6H), 3.79 (d,  $J = 8.0$  Hz, 6H), 3.12 (q,  $J = 9.8$  Hz, 1H), 2.88 (td,  $J = 11.5, 5.4$  Hz, 1H), 2.73-2.67 (m, 1H), 2.64-2.46 (m, 8H), 2.41-2.36 (m, 2H), 2.25 (qd,  $J = 7.2, 2.5$  Hz, 1H), 2.20-2.14 (m, 1H), 1.87-1.67 (m, 2H), 1.60 (dtd,  $J = 16.7, 11.3, 5.0$  Hz, 1H), 1.11 (d,  $J = 7.0$  Hz, 3H), 1.01 (d,  $J = 7.3$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  209.50, 209.47, 205.5, 203.2, 160.1, 160.0, 143.72, 143.69, 130.3, 130.1, 119.6, 119.3, 113.6, 113.3, 112.2, 112.0, 55.2, 49.4, 49.2, 47.4, 47.2, 47.0, 46.7, 44.8, 41.0, 41.1, 40.2, 27.9, 26.2, 11.2, 7.1; IR (thin film) 2943, 2719, 1716, 1600, 1488, 1263, 1046  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{16}\text{H}_{20}\text{O}_3$   $[\text{M}+\text{Na}]^+$  283.1310 found 283.1307.

## Dihydronaphthalene 5.13



A 100 mL round bottom flask containing 2.36 g (9.07 mmol) **5.12**, 86 mg (0.45 mmol) TsOH·H<sub>2</sub>O and 70 mL toluene was refluxed under a Dean-Stark trap. After 2 hours the reaction was cooled, diluted with 100 mL Et<sub>2</sub>O, washed with 20 mL sat. aq. NaHCO<sub>3</sub>, 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (6:1 hexanes/EtOAc) to afford 1.78 g (81%) **5.13** as a white solid that was crystallized from Et<sub>2</sub>O to afford a single cubic crystal (mp = 132–134 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 6.97 (d, *J* = 7.9 Hz, 1H), 6.73 (d, *J* = 7.5 Hz, 1H), 6.72–6.67 (m, 1H), 6.29 (s, 1H), 3.80 (s, 3H), 3.04 (d, *J* = 13.6 Hz, 1H), 2.87 (t, *J* = 12.7 Hz, 1H), 2.58–2.39 (m, 5H), 1.92 (s, 3H), 1.63 (t, *J* = 11.4 Hz, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 210.5, 158.5, 138.1, 136.0, 127.7, 126.3, 124.1, 110.8, 110.1, 55.2, 43.8, 42.2, 40.9, 40.7, 28.7, 20.6; IR (thin film) 2918, 2849, 1713, 1606, 1571, 1242, 1159 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>16</sub>H<sub>18</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 265.1205 found 265.1210.

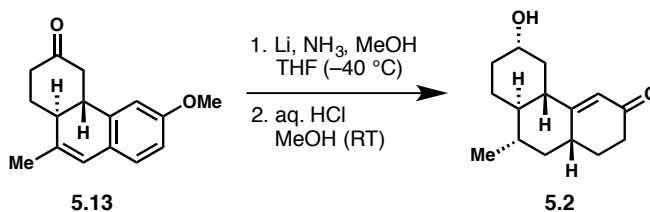
## Independent Analysis of Birch Reduction Stereochemistry



To a 10 mL round bottom flask charged with a glass stir bar was condensed 2 mL ammonia followed by the addition of 12 mg (0.050 mmol) **5.13** in 0.5 mL THF and 0.2 mL MeOH. Slowly, 12 mg (1.73 mmol) lithium metal was added in small pieces at -40 °C. After complete addition, ammonia was evaporated and the residue then taken up in EtOAc and water. The organic layer was separated and washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. To a 10 mL round bottom flask containing crude cyclohexadiene was added 0.5 mL CH<sub>2</sub>Cl<sub>2</sub> and 0.02 mL (0.13 mmol) DBU, cooled to 0 °C and treated with 14 mg (0.079 mmol) NBS. After 20 minutes at 0 °C sat. aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, water and EtOAc was added. The layers were separated and the organic layer was washed with water and brine. All aqueous washings were combined, back extracted with EtOAc and all organic layers combined. The EtOAc layer was dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. To a 10 mL round bottom flask containing crude alcohol was added 0.4 mL CH<sub>2</sub>Cl<sub>2</sub>, 40 mg (0.48 mmol) NaHCO<sub>3</sub> and at 0 °C 32 mg (0.075 mmol) DMP. The ice bath was removed. After overnight stirring EtOAc was added and the reaction filtered over Celite. Sat. aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and water were added, the layers separated and the organic phase washed with sat. aq. NaHCO<sub>3</sub>. The EtOAc layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1 hexanes/EtOAc) to afford 5.0 mg (41%, single diastereomer) **5.A** as a white solid that was recrystallized from Et<sub>2</sub>O to afford white needles (mp = 111–113 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.03 (d, *J* = 8.3 Hz, 1H), 6.73 (dd, *J*

= 8.3, 2.4 Hz, 1H), 6.71 (d,  $J = 2.1$  Hz, 1H), 3.78 (s, 3H), 3.08 (ddd,  $J = 14.2, 4.0, 2.2$  Hz, 1H), 2.85-2.81 (m, 2H), 2.57-2.51 (m, 2H), 2.43-2.32 (m, 3H), 1.74-1.64 (m, 1H), 1.51-1.40 (m, 2H), 1.11 (t,  $J = 10.8$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  211.2, 157.9, 139.4, 129.7, 128.8, 112.2, 110.4, 55.3, 46.5, 45.1, 43.8, 41.1, 38.4, 33.5, 29.4, 19.6; IR (thin film) 2950, 2920, 1709, 1606, 1501, 1239, 1038  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{16}\text{H}_{20}\text{O}_2$   $[\text{M}+\text{Na}]^+$  267.1361 found 267.1371.

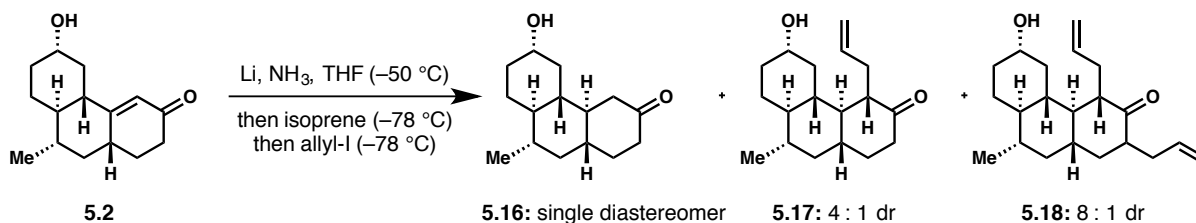
### Enone 5.2



To a 1 L 3-neck round bottom flask fitted with a low-temperature thermometer, was charged with a glass stir bar, 1.77 g (7.30 mmol) **5.13** in 75 mL THF and 230 mL ammonia and 15 mL (370 mmol) MeOH at  $-60$  °C. Slowly, 2.50 g (260 mmol) lithium metal was added in small pieces at  $-40$  °C. After the blue color discharged, 20 g (374 mmol) solid  $\text{NH}_4\text{Cl}$  was added and the ammonia evaporated overnight. The white residue taken up in 50 mL water and extracted with 100 mL, 50 mL and 25 mL EtOAc. The organic layers were combined, washed with 20 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. To a 300 mL flask containing crude material was added 75 mL MeOH followed by 7.5 mL 6 M HCl. After 8 hours 100 mL sat. aq.  $\text{NaHCO}_3$  was added and after 1 hour of stirring  $\sim 60$  mL volatiles were removed in vacuo. The remaining solution was extracted with 75 mL, 50 mL and twice with 25 mL EtOAc. All organic layers were combined, washed with 30 mL brine and back extracted with 25 mL EtOAc, then dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude

material was purified by column chromatography (1:1→1:3 hexanes/EtOAc) to afford 1.20 g (70%) **5.2** as a white solid that was recrystallized from Et<sub>2</sub>O to afford colorless prisms (mp = 139–140 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.80 (s, 1H), 3.64 (tt, *J* = 10.3, 4.8 Hz, 1H), 2.40–2.24 (m, 4H), 2.15–2.02 (m, 5H), 1.89–1.85 (m, 2H), 1.66–1.58 (m, 1H), 1.38 (dddd, *J* = 12.5, 9.4, 6.3, 3.1 Hz, 1H), 1.26–1.09 (m, 4H), 1.07–0.98 (m, 1H), 0.92 (t, *J* = 9.6 Hz, 3H), 0.90–0.85 (m, 1H), 0.78 (qd, *J* = 10.9, 3.1 Hz, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 200.6, 169.0, 121.0, 70.5, 48.9, 45.2, 43.4, 38.1, 37.4, 36.8, 35.5, 34.9, 29.3, 28.5, 19.3; IR (thin film) 3392, 2926, 2857, 1658, 1057 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>15</sub>H<sub>22</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 257.1518 found 257.1521.

### Allylation to **5.16**, **5.17** and **5.18**

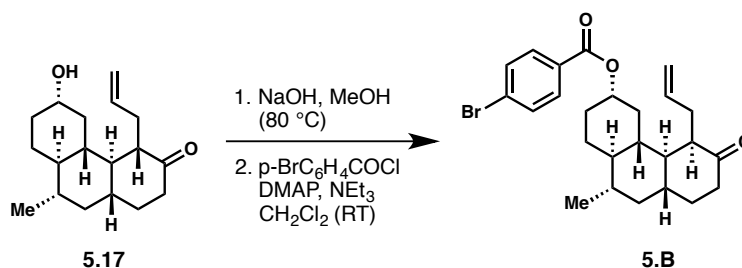


To a 25 mL round bottom flask containing a glass stir bar and 7.7 mg (1.1 mmol) lithium metal was condensed 7 mL ammonia and 1.5 mL THF added. At -50 °C 76 mg (0.32 mmol) **5.2** in 1.5 mL THF was added. The reaction was cooled to -78 °C then treated with 0.1 mL isoprene and stirred until all lithium was discharged. To the white slurry was added 69 mg (0.41 mmol) allyl iodide at -78 °C. The reaction was stirred for 2.5 hours before 3 mL sat. aq. NH<sub>4</sub>Cl was added slowly and warmed to room temperature. The mixture was partitioned between 5 mL water and 15 mL EtOAc. The organic layer was washed with 5 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (4:1→1:2 hexanes/EtOAc) to afford 17 mg (22%, >20:1 dr) **5.16** as a white solid which was

recrystallized from Et<sub>2</sub>O to afford white crystals (mp = 139–140 °C), 38 mg (43%, 4:1 dr) **5.17** as a white wax and 20 mg (20%, 8:1 dr) **5.18** as a white solid which was recrystallized from Et<sub>2</sub>O to afford a single diastereomer as white needles (mp = 118–120 °C). **5.16**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 3.57-3.53 (m, 1H), 2.62 (dd, *J* = 14.0, 1.8 Hz, 1H), 2.40-2.29 (m, 2H), 2.08-2.00 (m, 3H), 1.96 (ddt, *J* = 13.1, 6.2, 3.1 Hz, 1H), 1.88 (t, *J* = 13.3 Hz, 1H), 1.72 (dt, *J* = 13.1, 3.4 Hz, 1H), 1.48 (qt, *J* = 11.6, 3.1 Hz, 1H), 1.35 (qd, *J* = 12.6, 4.9 Hz, 1H), 1.25-1.15 (m, 4H), 0.92 (d, *J* = 6.5 Hz, 3H), 1.00-0.77 (m, 4H), 0.59 (qd, *J* = 10.8, 2.9 Hz, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 211.9, 70.7, 47.4, 47.3, 46.1, 45.1, 41.8, 41.3, 40.9, 38.8, 36.5, 35.3, 33.2, 28.3, 19.9; IR (thin film) 3400, 2921, 2855, 1711, 1455, 1041 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>15</sub>H<sub>24</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 259.1674 found 259.1674. **5.17**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.71-5.60 (m, 1H), 5.12-4.97 (m, 2H), 3.54 (t, *J* = 10.4 Hz, 1H), 2.48 (td, *J* = 15.7, 8.2 Hz, 2H), 2.32-2.22 (m, 3H), 2.11 (d, *J* = 9.7 Hz, 1H), 2.02-1.96 (m, 2H), 1.75-1.59 (m, 4H), 1.32-1.11 (m, 5H), 0.99-0.75 (m, 6H), 0.61 (q, *J* = 10.4 Hz, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) major δ 216.0, 134.9, 117.2, 70.7, 53.2, 50.9, 48.0, 46.6, 42.6, 39.9, 39.8, 39.4, 37.2, 36.5, 35.5, 28.8, 28.3, 20.0, minor δ 214.4, 134.9, 116.6, 70.6, 51.3, 50.2, 47.3, 42.4, 40.8, 38.1, 38.0, 36.3, 35.4, 34.4, 34.2, 30.4, 28.4, 19.9; IR (thin film) 3388, 2923, 2858, 1702, 1444, 1051, 1023 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>18</sub>H<sub>28</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 299.1987 found 299.1994. **5.18**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.78-5.63 (m, 2H), 5.05-4.98 (m, 4H), 3.57-3.52 (m, 1H), 2.61-2.55 (m, 1H), 2.52-2.45 (m, 2H), 2.36-2.21 (m, 2H), 2.13-1.96 (m, 4H), 1.86-1.75 (m, 1H), 1.73-1.50 (m, 2H), 1.33-1.01 (m, 5H), 1.00-0.76 (m, 7H), 0.59 (dt, *J* = 18.4, 9.2 Hz, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 216.2, 136.4, 135.0, 116.8, 116.4, 70.7, 54.1, 51.8, 47.6, 46.8, 42.50, 42.47, 40.0, 39.0, 39.4, 37.6, 36.4, 35.4, 35.3, 34.6, 28.2, 19.9; IR (thin film)

3262, 2920, 2853, 1698, 1441, 1059  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{21}\text{H}_{32}\text{O}_2$   $[\text{M}+\text{Na}]^+$  339.2300 found 339.2307.

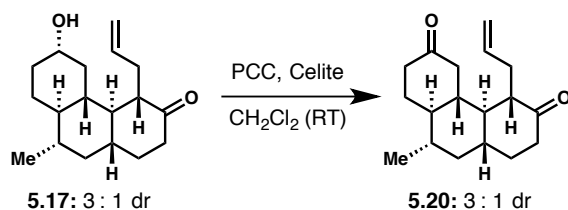
### Axial Allyl **5.19** and Benzoate **5.B**



To 1 dram vial containing 6 mg (0.025 mmol, 3:1 dr) **5.17** in 0.3 mL MeOH was added 0.05 mL (0.05 mmol) 1 M NaOH/MeOH at room temperature, the flask sealed and the reaction heated to 80 °C. After 2 hours, sat. aq.  $\text{NH}_4\text{Cl}$  was added and the mixture extracted thrice with  $\text{Et}_2\text{O}$ . The organic layers were combined, washed with brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo to afford a single diastereomer. The crude material in a 1 dram vial and 4 mg (0.033 mmol) DMAP was diluted with 0.4 mL 1:1  $\text{CH}_2\text{Cl}_2/\text{NEt}_3$  and treated with 11 mg (0.050 mmol) p-bromobenzoyl chloride at room temperature. The reaction was stirred for 1 hour then quenched with sat. aq.  $\text{NaHCO}_3$  and extracted four times with EtOAc. The organic layers were combined, washed with 1:1 brine/ $\text{NaHCO}_3$ , dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1 hexanes/EtOAc) to provide 7 mg (70%, >10:1 dr) **5.B** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.90-7.88 (m, 2H), 7.59-7.57 (m, 2H), 5.62 (dddd,  $J = 16.7, 10.3, 8.3, 6.1$  Hz, 1H), 5.05 (dd,  $J = 17.0, 1.0$  Hz, 1H), 4.99 (d,  $J = 10.1$  Hz, 1H), 4.91 (tt,  $J = 11.4, 4.5$  Hz, 1H), 2.60 (dt,  $J = 8.0, 3.8$  Hz, 1H), 2.49 (td,  $J = 14.0, 6.4$  Hz, 1H), 2.39-2.34 (m, 1H), 2.29 (dd,  $J = 12.0, 8.4$  Hz, 1H), 2.24-2.09 (m, 4H), 1.98 (ddt,  $J = 13.0, 6.6, 3.1$  Hz, 1H), 1.82-1.75 (m, 2H),

1.45-1.37 (m, 1H), 1.36-1.25 (m, 3H), 1.23-1.17 (m, 1H), 1.10-0.99 (m, 2H), 0.94 (d,  $J = 6.5$  Hz, 3H), 0.80 (q,  $J = 12.7$  Hz, 1H), 0.70-0.64 (m, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  214.2, 165.3, 134.8, 131.6, 131.1, 129.6, 127.9, 116.7, 73.9, 51.4, 50.4, 47.4, 42.4, 40.8, 38.1, 36.2, 34.3, 34.2, 34.2, 31.5, 30.4, 28.2, 19.8; IR (thin film) 2922, 2865, 1713, 1590, 1272, 1103  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{25}\text{H}_{31}\text{BrO}_3$   $[\text{M}+\text{Na}]^+$  481.1354 found 481.1342.

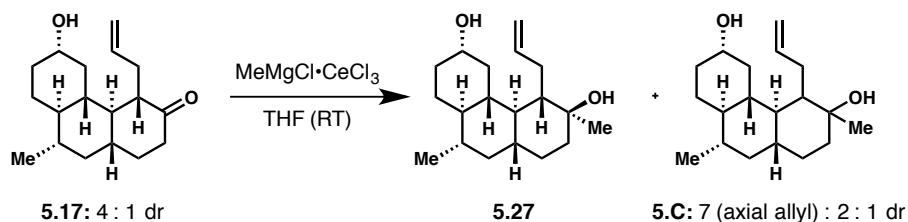
### Ketone 5.20



To 1 dram vial of 14 mg (0.051 mmol, 3:1 dr) **5.17** and 60 mg celite in 0.8 mL  $\text{CH}_2\text{Cl}_2$  was added 30 mg (0.14 mmol) PCC at room temperature. After 16 hours the mixture was diluted with  $\text{Et}_2\text{O}$ , passed through a silica gel column and concentrated. The crude material was purified by column chromatography (4:1 hexanes/ $\text{EtOAc}$ ) to provide 12 mg (85%, 3:1 dr) **5.20** as a white solid.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.70-5.57 (m, 1H), 5.07-4.98 (m, 2H), 2.56-2.21 (m, 10H), 2.05-1.89 (m, 1H), 1.81-1.72 (m, 2H), 1.68-1.58 (m, 1H), 1.41-1.32 (m, 1H), 1.30-1.17 (m, 3H), 1.10 (q,  $J = 10.6$  Hz, 1H), 1.03-0.82 (m, 4H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm) major  $\delta$  214.7, 210.7, 134.6, 117.6, 52.7, 51.6, 48.7, 47.6, 45.5, 42.3, 41.0, 39.8, 39.7, 37.2, 36.3, 29.9, 28.6, 20.0, minor  $\delta$  213.7, 211.0, 134.3, 116.9, 51.2, 51.1, 46.9, 44.0, 43.1, 42.1, 41.0, 37.9, 36.1, 34.3, 33.9, 30.2, 30.1, 19.9; IR (thin film) 2960, 2916, 2859, 1713  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{26}\text{O}_2$   $[\text{M}+\text{Na}]^+$  297.1830 found 297.1836.



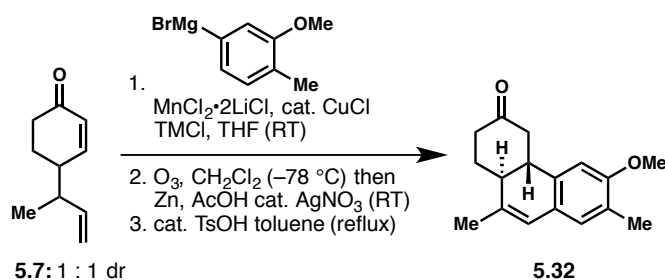
## Diol 5.27 and 5.C



To 1 dram vial was added 45 mg (0.18 mmol)  $\text{CeCl}_3$  and dried under vacuum at 160 °C for 5 hours. The solid was cooled under argon and stirred with 0.6 mL THF for 30 minutes before being cooled in an ice bath and treated with 0.07 mL (0.19 mmol) 2.7 M  $\text{MeMgCl/THF}$ . After 30 minutes at 0 °C a solution of 10 mg (0.036 mmol) **5.17** in 0.3 mL THF was added dropwise at 0°C. The ice bath was removed after 10 minutes. After 5 hours the reaction was recooled to 0°C and quenched with 1 mL 3 M HCl, then extracted thrice with  $\text{Et}_2\text{O}$ . The organic layers were combined, washed with sat. aq.  $\text{NaHCO}_3$ , dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (3:1→1:1 hexanes/ $\text{EtOAc}$ ) to provide 6.5 mg (62%, single diastereomer) **5.27** as a white solid and 2.6 mg (25%, 7 (axial allyl):2:1 dr) **5.C** as a colorless oil. **5.27**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  6.03-5.95 (m, 1H), 5.09-5.02 (m, 2H), 3.50-3.46 (m, 1H), 2.71-2.66 (m, 1H), 2.34-2.32 (m, 1H), 2.23-2.18 (m, 1H), 2.06-1.95 (m, 2H), 1.65 (dt,  $J = 13.2, 3.6$  Hz, 1H), 1.55 (dt,  $J = 12.9, 3.3$  Hz, 1H), 1.50-1.44 (m, 1H), 1.41-1.30 (m, 4H), 1.27 (t,  $J = 7.1$  Hz, 1H), 1.22 (s, 3H), 1.21-1.02 (m, 6H), 0.90 (t,  $J = 5.7$  Hz, 3H), 0.95-0.83 (m, 2H), 0.70 (t,  $J = 9.4$  Hz, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  140.5, 115.1, 73.6, 71.6, 49.1, 48.0, 47.4, 44.2, 42.6, 42.1, 39.4, 37.4, 35.1, 34.8, 30.8, 29.3, 28.4, 20.4; IR (thin film) 3389, 3322, 2922, 1638, 1403, 1045  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{19}\text{H}_{32}\text{O}_2$   $[\text{M}+\text{NH}_4]^+$  310.2746 found 310.2743. **5.C**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm) mixture  $\delta$  6.12-6.04 (m, 0.09H), 6.03-5.94 (m, 0.6H), 5.90-5.81 (m, 0.2H), 5.15-5.00 (m, 1.2H), 4.97-4.93 (m, 0.7H), 3.58-3.43 (m, 1H), 2.35-2.27 (m, 1H), 2.15-2.00 (m, 3H),

1.94-1.86 (m, 1H), 1.79-1.51 (m, 4H), 1.49-1.37 (m, 2H), 1.33-1.13 (m, 5H), 1.12-0.98 (m, 4H), 0.95-0.81 (m, 5H), 0.80-0.68 (m, 2H), 0.61-0.56 (m, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm) major  $\delta$  141.7, 114.2, 70.7, 48.1, 47.8, 45.8, 43.3, 41.2, 38.8, 36.3, 35.4, 34.6, 34.4, 31.7, 29.4, 28.6, 27.7, 20.0; IR (thin film) 3369, 2924, 2856, 1637, 1459, 1050  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{19}\text{H}_{32}\text{O}_2$   $[\text{M}+\text{Na}]^+$  315.2300 found 315.2308.

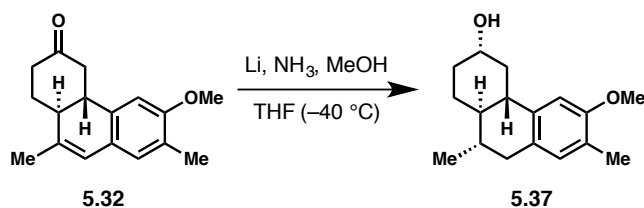
### Dihydronaphthalene 5.32



3-Methoxy-4-methylphenylmagnesium bromide was prepared by addition of 1.60 g (7.96 mmol) 5-bromo-2-methylanisole in 1.0 mL THF to 0.220 g (9.0 mmol) magnesium metal in 5.0 mL THF activated by dibromoethane maintaining 40–50 °C. A 50 mL round bottom flask containing 154 mg (1.22 mmol)  $\text{MnCl}_2$  and 109 mg (2.57 mmol)  $\text{LiCl}$  was dried under high vacuum in a 150 °C oil bath for 10 hours. The flask was cooled, placed under an argon balloon and 10 mL THF added to dissolve the salts. At 0 °C, 540 mg (3.60 mmol) **5.7** in 4 mL THF and 10 mg (0.010 mmol)  $\text{CuCl}$  were added. After 10 minutes 0.64 mL (5.9 mmol)  $\text{TMSCl}$  was added, followed by the dropwise addition of 3.9 mL (4.1 mmol) 1.05 M 3-methoxy-4-methylphenylmagnesium bromide over the course of 5 minutes. Stirring at 0 °C was continued for 10 minutes before the cold bath was removed. After 5 hours at room temperature, the flask was again cooled in an ice bath and 20 mL 3 M  $\text{HCl}$  added slowly. The ice bath was removed and stirred for 10 minutes. The mixture was partitioned between an additional 10 mL 3 M  $\text{HCl}$  and 30 mL  $\text{Et}_2\text{O}$ . The layers

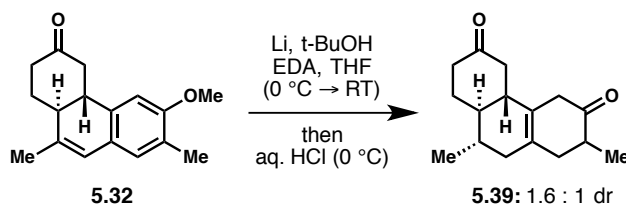
were separated and the aqueous layer extracted with 10 mL Et<sub>2</sub>O. The organic layers were combined, washed with 10 mL sat. aq. NaHCO<sub>3</sub>, 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was carried on crude. Crude alkene in a 100 mL round bottom flask was dissolved in 30 mL CH<sub>2</sub>Cl<sub>2</sub>, cooled to -78 °C and ozonized. After the solution became blue, the solution was sparged with oxygen. A single portion of 3 mL AcOH, 1.2 g (18 mmol) Zn powder and 15 mg (0.088 mmol) AgNO<sub>3</sub> was added. After stirring at -78 °C for 2 hours, the bath was removed and stirring continued for 2 hours. The reduction was determined complete by TLC and the solution was filtered over Celite, washing with CH<sub>2</sub>Cl<sub>2</sub> (~100 mL as determined by TLC). The filtrate was washed with 50 mL water, 50 mL sat. aq. NaHCO<sub>3</sub>, 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. A 100 mL round bottom flask containing crude aldehyde in 50 mL toluene was treated with 34 mg (0.18 mmol) TsOH•H<sub>2</sub>O and heated to 140 °C under a Dean–Stark trap. After 2 hours the reaction was cooled, diluted with 20 mL Et<sub>2</sub>O, washed with 20 mL half sat. aq. NaHCO<sub>3</sub> and 10 mL brine. The aqueous layers were combined and back extracted with 10 mL Et<sub>2</sub>O. All organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by trituration with hexanes to afford 628 mg (67%) **5.32** as a tan solid which was recrystallized from EtOAc to afford wispy white needles (mp = 193–194 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 6.86 (s, 1H), 6.62 (s, 1H), 6.27 (s, 1H), 3.84 (s, 4H), 3.11 (ddd, *J* = 14.1, 4.0, 2.2 Hz, 1H), 2.91 (td, *J* = 13.9, 3.8 Hz, 1H), 2.61–2.41 (m, 6H), 2.20 (s, 4H), 1.93 (s, 3H), 1.69 (qd, *J* = 12.6, 4.7 Hz, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 210.9, 156.5, 135.8, 135.1, 128.0, 127.2, 124.7, 124.2, 105.6, 55.5, 44.2, 42.2, 41.0, 40.9, 28.8, 20.6, 15.8; IR (thin film) 2923, 2852, 1714, 1611 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>17</sub>H<sub>20</sub>O<sub>2</sub> [M+NH<sub>4</sub>]<sup>+</sup> 274.1807 found 274.1800.

### Anisole 5.37



To a 10 mL round bottom flask charged with a glass stir bar was condensed 3 mL ammonia followed by the addition of 24 mg (0.094 mmol) **5.32** in 1 mL THF and 0.3 mL MeOH. Slowly, 24 mg (3.46 mmol) lithium metal was added in small pieces at -40 °C. After the blue color discharged, 0.140 g (2.62 mmol) solid NH<sub>4</sub>Cl was added and the ammonia evaporated. The white residue taken up in water and EtOAc. The organic layer was separated and washed with brine. The aqueous layers were combined and back extracted with EtOAc. The organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford a crude aromatic material assigned as **5.37**.

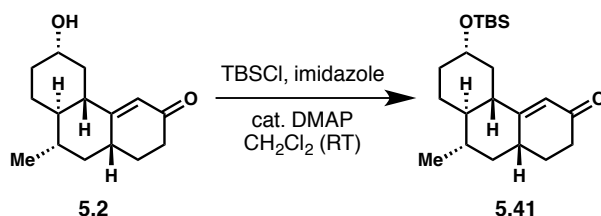
### Skipped Enone 5.39



To a 25 mL round bottom flask charged with a glass stir bar was added 125 mg (0.488 mmol) **5.32**, 5.0 mL THF and 2.5 mL (37 mmol) ethylene diamine before being cooled in an ice bath. After the addition of 1.2 mL (12.5 mmol) t-BuOH, 80 mg (11.5 mmol) lithium metal was added in small pieces over the course of 3 hours at 0 °C. After complete consumption of lithium metal, the cold bath was removed and stirring continued for 1 hour. The flask was recooled to 0 °C and canula transferred into a vigorously stirring flask of 10 mL conc. HCl and 50 g ice. Best results

were obtained by submerging the canula into the aqueous solution. After complete transfer the reaction flask was washed twice with 1.5 mL Et<sub>2</sub>O. The aqueous mixture was warmed to room temperature and extracted with 25 mL EtOAc, then four times with 10 mL EtOAc. The organic layers were combined, washed with 10 mL 6 M HCl then back extracted with 5 mL EtOAc and combined with the remaining organic layers. The organic phase was washed with 5 mL water, 5 mL sat. aq. NaHCO<sub>3</sub>, 10 mL brine, dried over MgSO<sub>4</sub> and all volatiles removed in vacuo to afford 100 mg (83%, 1.6:1 dr) crude material of good purity. Column chromatography (4:1 hexanes/EtOAc) afforded 40 mg (33%, 1.8:1 dr) **5.39** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 2.91 (d, *J* = 19.8 Hz, 1H), 2.81 (d, *J* = 20.5 Hz, 0.5H), 2.69 (td, *J* = 12.9, 6.7 Hz, 1H), 2.60-2.56 (m, 2H), 2.51-2.41 (m, 3H), 2.37-2.27 (m, 3H), 2.21-1.80 (m, 8H), 1.58-1.54 (m, 1H), 1.52-1.45 (m, 1H), 1.39-1.20 (m, 4H), 1.11 (d, *J* = 6.7 Hz, 1.5H), 1.07 (d, *J* = 6.5 Hz, 3H), 1.02 (d, *J* = 6.3 Hz, 1.5H), 1.01 (d, *J* = 6.5 Hz, 3H), 0.94-0.86 (m, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) major δ 211.6, 211.0, 130.0, 126.7, 44.9, 44.8, 44.7, 41.9, 41.2, 41.1, 40.1, 39.4, 32.7, 28.7, 18.9, 13.5, minor δ 212.1, 211.1, 129.6, 126.3, 45.1, 45.0, 44.5, 42.4, 41.1, 40.4, 39.8, 39.1, 32.9, 28.9, 19.0, 14.6; IR (thin film) 2962, 2926, 2874, 1716, 1453 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>16</sub>H<sub>22</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 269.1518 found 269.1516.

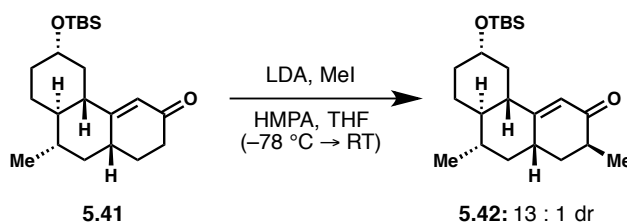
### Enone 5.41



To 100 mL round bottom flask containing 1.20 g (5.12 mmol) **5.2**, 0.525 g (7.71) imidazole and 60 mg (0.49 mmol) DMAP in 25 mL CH<sub>2</sub>Cl<sub>2</sub> was added 0.950 g (6.30 mmol) TBSCl at room

temperature. After 15 hours, 100 mL half sat. aq. NaHCO<sub>3</sub> was added and the mixture extracted with 100 mL and 50 mL EtOAc. The organic layers were combined, washed with 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (6:1 hexanes/EtOAc) to provide 1.67 g (93%) **5.41** as a white solid which was recrystallized from pentane to afford colorless prisms (mp = 110–112 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.84 (s, 1H), 3.61 (tt, *J* = 10.8, 4.3 Hz, 1H), 2.42–2.26 (m, 3H), 2.16–2.10 (m, 1H), 2.08–2.00 (m, 2H), 1.94–1.82 (m, 3H), 1.68–1.62 (m, 1H), 1.42–1.34 (m, 1H), 1.32–1.21 (m, 2H), 1.14 (q, *J* = 12.5 Hz, 1H), 1.01 (qd, *J* = 12.5, 3.2 Hz, 1H), 0.95 (d, *J* = 6.5 Hz, 3H), 0.89 (s, 9H), 0.78 (qd, *J* = 10.9, 3.0 Hz, 1H), 0.06 (d, *J* = 1.8 Hz, 6H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 200.4, 168.9, 121.1, 71.3, 48.9, 45.3, 43.5, 38.1, 37.7, 36.9, 35.6, 35.4, 29.5, 28.5, 25.9, 19.2, 18.2, –4.6; IR (thin film) 2929, 2857, 1677, 1092 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>21</sub>H<sub>36</sub>O<sub>2</sub>Si [M+Na]<sup>+</sup> 371.2382 found 371.2388.

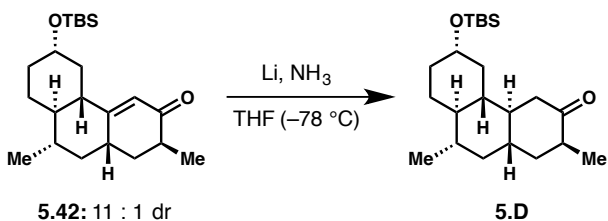
### Enone 5.42



LDA was prepared in a 50 mL round bottom flask by addition of 1.90 mL (5.13 mmol) 2.7 M n-BuLi/hexanes to 0.81 mL (5.78 mmol) diisopropylamine in 17 mL THF at 0 °C. To the stirring LDA solution at –78 °C was added 1.55 g (4.45 mmol) **5.41** with the assistance of 9 mL THF. After 10 minutes, 1.0 mL (5.75 mmol) HMPA was added neat followed by 0.83 mL (13.3 mmol) methyl iodide. The cold bath was removed after an additional 10 minutes and the reaction stirred for 50 minutes before 20 mL half sat. aq. NH<sub>4</sub>Cl was added. The solution was extracted with 50

mL and 25 mL EtOAc. The organic layers were combined, washed with 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude mixture was purified by column chromatography (10:1 hexanes/EtOAc) to afford 1.51 g (93%, 13:1 dr) **5.42** as a white wax. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.75 (s, 1H), 3.63 (tt, *J* = 10.8, 4.4 Hz, 1H), 2.47-2.40 (m, 1H), 2.35 (dq, *J* = 11.6, 5.5 Hz, 1H), 2.05 (dq, *J* = 13.3, 3.6 Hz, 1H), 2.00-1.83 (m, 5H), 1.77 (dt, *J* = 13.7, 4.7 Hz, 1H), 1.44-1.23 (m, 4H), 1.11 (d, *J* = 6.9 Hz, 3H), 1.06-0.98 (m, 1H), 0.95 (d, *J* = 6.3 Hz, 3H), 0.89 (s, 9H), 0.74 (qd, *J* = 10.6, 3.1 Hz, 1H), 0.06 (d, *J* = 1.8 Hz, 6H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 202.9, 168.1, 119.5, 71.2, 49.9, 45.7, 42.8, 37.8, 37.6, 37.4, 37.2, 36.0, 35.5, 29.6, 25.9, 19.2, 18.2, 15.4, -4.6; IR (thin film) 2928, 2856, 1675, 1627, 1249, 1092, 863, 835, 775 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>22</sub>H<sub>38</sub>O<sub>2</sub>Si [M+H]<sup>+</sup> 363.2719 found 363.2715.

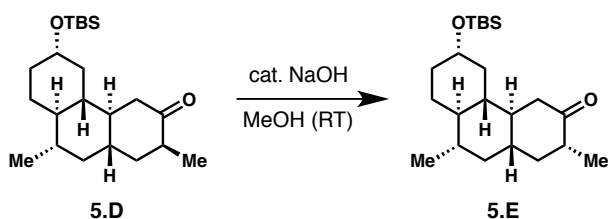
### Ketone 5.D



To a 10 mL round bottom flask containing a glass stir bar and 5 mg (0.72 mmol) lithium metal then 3 mL ammonia was condensed and 0.4 mL THF added. At -78 °C 10 mg (0.028 mmol) **5.42** in 0.4 mL THF was added. After 2 minutes, 0.1 mL isoprene and 91 mg (1.7 mmol) solid NH<sub>4</sub>Cl was added and the flask warmed to room temperature. The mixture was partitioned between 2 mL water and 5 mL EtOAc. The organic layer was washed with 2 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (20:1 hexanes/EtOAc) to afford 8.9 mg (89%, >20:1 dr) **5.D** as a

colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  3.54-3.48 (tt,  $J = 10.8, 4.4$  Hz, 1H), 2.56-2.49 (m, 2H), 2.05 (dd,  $J = 14.7, 12.8$  Hz, 1H), 2.01-1.93 (m, 2H), 1.91-1.86 (m, 1H), 1.68-1.54 (m, 4H), 1.27-1.12 (m, 3H), 1.20 (d,  $J = 7.3$  Hz, 3H), 0.96-0.80 (m, 4H), 0.92 (d,  $J = 6.5$  Hz, 3H), 0.88 (d,  $J = 9.4$  Hz, 9H), 0.56 (qd,  $J = 10.8, 3.3$  Hz, 1H), 0.04 (s, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  216.0, 71.5, 47.3, 47.2, 46.2, 43.8, 42.1, 41.6, 39.5, 39.1, 36.7, 35.7, 35.3, 28.5, 20.0, 18.2, 17.7, -4.55, -4.60; IR (thin film) 2927, 2856, 1711, 1460, 1087, 860, 835, 774  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{22}\text{H}_{40}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  387.2695 found 387.2703.

### Ketone 5.E

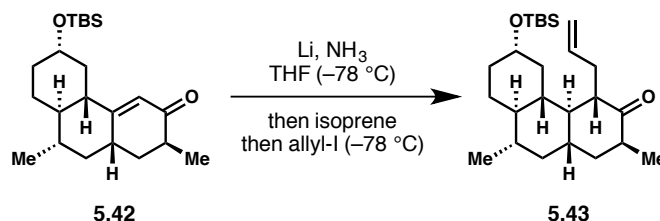


To a 1 dram vial was added 8.1 mg (0.022 mmol) **5.D** in 0.2 mL MeOH and 0.1 mL (0.013 mmol) 0.13 M NaOH/MeOH. After 2 hours at room temperature, 2 mL sat. aq.  $\text{NH}_4\text{Cl}$  and 5 mL EtOAc were added. The organic layer was separated, washed with 2 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo to afford 7.9 mg (97%, >10:1 dr) **5.E** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  3.53-3.47 (tt,  $J = 10.8, 4.4$  Hz, 1H), 2.64 (dd,  $J = 13.3, 3.9$  Hz, 1H), 2.42 (dq,  $J = 12.9, 6.4$  Hz, 1H), 2.01-1.87 (m, 5H), 1.69 (dt,  $J = 13.0, 3.5$  Hz, 1H), 1.54 (qt,  $J = 10.7, 3.2$ , 1H), 1.26-1.07 (m, 4H), 1.02 (d,  $J = 6.5$  Hz, 3H), 0.95-0.75 (m, 4H), 0.91 (s, 3H), 0.88 (s, 9H), 0.55 (qd,  $J = 10.8, 3.2$  Hz, 1H), 0.06 (s, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  213.2, 71.5, 48.4, 47.3, 46.2, 45.2, 44.7, 42.6, 41.7, 41.5, 39.3, 36.6, 35.7, 28.5, 25.9, 20.0, 18.2, 14.3, -4.54, -4.59; IR (thin film) 2927, 2856, 1713,



1460, 1250, 1090, 860, 835, 774  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{22}\text{H}_{40}\text{O}_2\text{Si}$   $[\text{M}+\text{Na}]^+$  387.2695 found 387.2687.

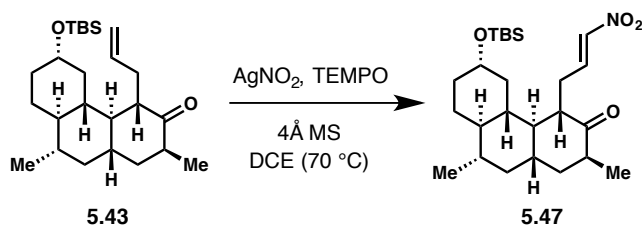
### Allyl 5.43



To a 25 mL round bottom flask containing a glass stir bar and 4.0 mg (0.58 mmol) lithium metal was condensed 7 mL ammonia and 4 mL THF added. At  $-78\text{ }^{\circ}\text{C}$  85 mg (0.23 mmol) **5.42**, 17 mg (0.23 mmol) t-BuOH in 3 mL THF was added. After stirring for 10 minutes, 0.1 mL isoprene was added and stirred until all lithium was discharged. To the white slurry was added 0.045 mL (0.50 mmol) allyl iodide at  $-78\text{ }^{\circ}\text{C}$ . The reaction was stirred for 2.5 hours before 3 mL sat. aq.  $\text{NH}_4\text{Cl}$  was added slowly and warmed to room temperature. The mixture was partitioned between 5 mL water and 15 mL EtOAc. The organic layer was washed with 5 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (30:1 hexanes/EtOAc) to afford 54 mg (56%, >10:1 dr) **5.43** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.68 (ddt,  $J = 16.9, 10.0, 7.0$  Hz, 1H), 5.01 (d,  $J = 7.0$  Hz, 1H), 4.99 (s, 1H), 3.49 (tt,  $J = 10.1, 4.8$  Hz, 1H), 2.64 (dq,  $J = 10.3, 6.9, 3.4$  Hz, 1H), 2.44 (dt,  $J = 13.3, 6.3$  Hz, 1H), 2.31-2.25 (m, 1H), 2.19 (dt,  $J = 10.1, 4.9$  Hz, 1H), 2.02-1.88 (m, 4H), 1.64-1.57 (m, 2H), 1.39 (dt,  $J = 13.5, 3.3$  Hz, 1H), 1.28-1.21 (m, 1H), 1.16-1.06 (m, 3H), 1.04 (d,  $J = 6.9$  Hz, 3H), 0.89 (s, 9H), 0.97-0.77 (m, 6H), 0.56 (qd,  $J = 10.5, 2.5$  Hz, 1H), 0.06 (s, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  217.9, 135.1, 116.7, 71.6, 54.4, 51.9, 47.7, 47.1, 42.7, 40.1, 40.0, 38.5, 37.7, 37.5, 36.5, 35.8, 28.4, 25.9, 19.9, 18.3, 16.3,  $-4.48$ ,

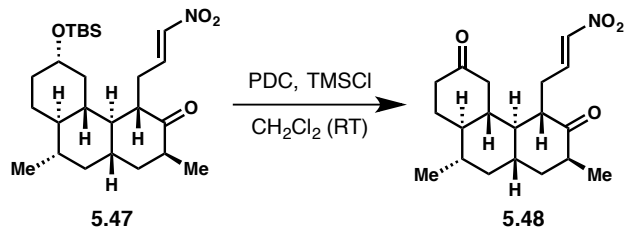
−4.55; IR (thin film) 2927, 2857, 1709, 1461, 1249, 1089, 835, 775  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{25}\text{H}_{44}\text{O}_2$   $[\text{M}+\text{Na}]^+$  427.3008 found 427.3014.

### Nitroalkene 5.47



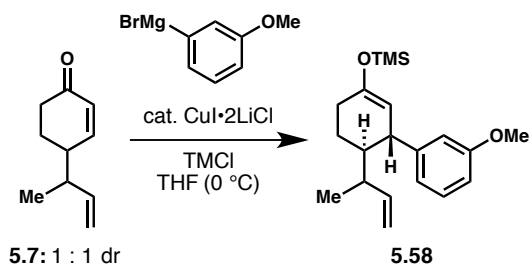
A 1 dram vial containing 30 mg (0.0741 mmol) **5.43** and 32 mg 4 Å MS in 0.5 mL DCE was stirred for 15 minutes before 116 mg (0.754 mmol)  $\text{AgNO}_2$  and 20 mg (0.128 mmol) TEMPO was added. The flask was sealed and heated at 70 °C for 30 hours. The flask was cooled, filtered over Celite with EtOAc, then concentrated in vacuo. The crude material was purified by column chromatography (10:1 hexanes/EtOAc) to afford 23 mg (69%, >20:1 E/Z) **5.47** as a light red oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.15 (dt,  $J = 13.5, 7.0$  Hz, 1H), 6.98 (d,  $J = 13.5$  Hz, 1H), 3.51 (d,  $J = 0.4$  Hz, 1H), 2.65-2.60 (m, 2H), 2.48 (dt,  $J = 15.9, 8.4$  Hz, 1H), 2.30-2.28 (m, 1H), 1.66 (t,  $J = 13.9$  Hz, 2H), 1.52-1.43 (m, 3H), 1.09 (d,  $J = 6.5$  Hz, 3H), 1.31-0.78 (m, 12H), 0.91 (s, 9H), 0.62-0.58 (m, 1H), 0.07 (s, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  215.8, 140.1, 138.3, 71.0, 52.3, 51.5, 47.2, 46.9, 42.3, 39.9, 38.0, 37.7, 36.9, 36.2, 35.4, 33.1, 28.0, 25.6, 19.6, 17.9, 16.0, −4.7, −4.9; IR (thin film) 2928, 2857, 1706, 1527, 1461, 1349, 1089, 860, 835, 775  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{25}\text{H}_{43}\text{NO}_4\text{Si}$   $[\text{M}+\text{Na}]^+$  472.2859 found 472.2862.

## Ketone 5.48



A 1 dram vial containing 5 mg (0.013 mmol) PDC, 0.1 mL CH<sub>2</sub>Cl<sub>2</sub> at 0 °C was treated with 0.1 mL (0.031 mmol) 0.31 M TMSCl/CH<sub>2</sub>Cl<sub>2</sub>. After 10 minutes ~2 mg (0.0044 mmol) **5.47** in 0.2 mL CH<sub>2</sub>Cl<sub>2</sub> was added. After 5 minutes the cold bath was removed and after 12 hours silica gel was added and the contents filtered over silica gel with EtOAc. The filtrate was concentrated and the crude material was purified by column chromatography (3:1 hexanes/EtOAc) to afford ~0.5 mg (33%) **5.48**.

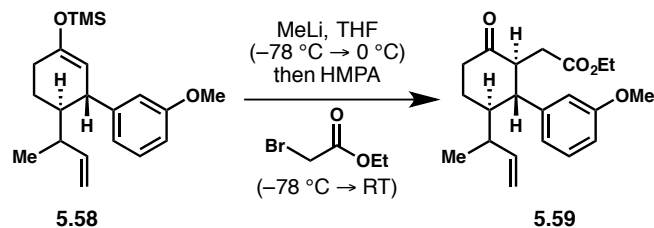
## Enoxysilane 5.58



3-Methoxyphenylmagnesium bromide was prepared by addition of 1.9 mL (15.0 mmol) 3-bromoanisole in 1.9 mL THF to 440 mg (18.1 mmol) magnesium metal in 7.5 mL THF activated by dibromoethane maintaining 40–50 °C. A 50 mL round bottom flask was charged with 17 mL THF (rigorously degassed by purging with argon under sonication) then 0.80 mL (0.40 mmol) 0.50 M CuI·2LiCl/THF. At 0 °C, 602 mg (4.01 mmol) **5.7** in 6 mL THF and 0.55 mL (4.3 mmol) TMSCl was added, followed by the dropwise addition of 3.4 mL (4.8 mmol) 1.41 M 3-methoxyphenylmagnesium bromide over the course of 5 minutes. Stirring at 0 °C was continued

for 1.5 hours before 2 mL  $\text{NEt}_3$  was added and the reaction poured into a rapidly stirring solution of 25 mL sat. aq.  $\text{NH}_4\text{Cl}$ , 25 g ice, 25 mL  $\text{Et}_2\text{O}$  and 2 mL  $\text{NEt}_3$  cooled in an ice bath. After all salts dissolved an additional 25 mL  $\text{Et}_2\text{O}$  was added, the layers were separated and the organic layer washed with 20 mL half sat. aq.  $\text{NaHCO}_3$ , 15 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude oil was flushed through a pad of Florisil®, pretreated with hexanes 1%  $\text{NEt}_3$ , with hexanes 1%  $\text{NEt}_3$  to afford 830 mg (62%, >20:1 conjugate addition dr) **5.58** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{C}_6\text{D}_6$  at 7.15 ppm)  $\delta$  7.16-7.12 (m, 1H), 7.01 (dt,  $J$  = 6.7, 1.9 Hz, 1H), 6.93 (d,  $J$  = 7.6 Hz, 0.5H), 6.89 (d,  $J$  = 7.6 Hz, 0.5H), 6.71-6.68 (m, 1H), 5.70-5.62 (m, 1H), 5.03-4.88 (m, 3H), 3.41 (dq,  $J$  = 8.6, 2.6 Hz, 1H), 3.37 (s, 1.5H), 3.36 (s, 1.5 H), 2.35-2.28 (m, 1H), 2.19-2.13 (m, 2H), 1.69-1.58 (m, 2H), 1.51 (ddt,  $J$  = 11.4, 8.3, 3.1 Hz, 1H), 1.41-1.32 (m, 1H), 0.88 (d,  $J$  = 6.9 Hz, 1.5H), 0.86 (d,  $J$  = 6.9 Hz, 1.5H), 0.17 (s, 4.5H), 0.16 (s, 4.5H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{C}_6\text{D}_6$  at 128 ppm) mixture  $\delta$  160.4, 151.6, 151.3, 148.50, 148.46, 141.4, 141.1, 129.6, 129.5, 121.2, 121.1, 115.0, 114.8, 114.6, 113.5, 111.8, 111.7, 108.2, 107.5, 54.7, 47.2, 46.1, 45.4, 44.7, 38.1, 37.3, 30.2, 29.6, 22.5, 21.8, 19.5, 14.2, 0.4; IR (thin film) 2958, 2873, 2835, 1665, 1251, 1184, 845  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for desilylation only,  $\text{C}_{17}\text{H}_{22}\text{O}_2$   $[\text{M}+\text{Na}]^+$  281.1518 found 281.1523.

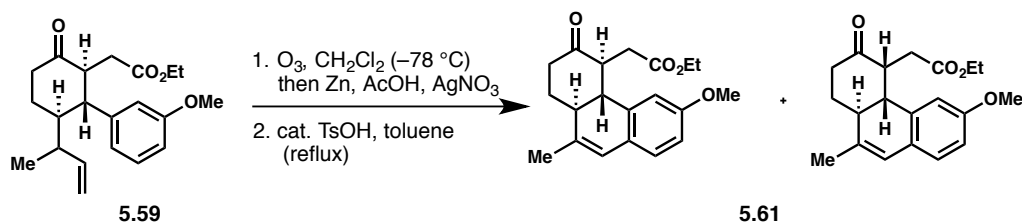
## Ketoester 5.59



To 25 mL round bottom flask containing 819 mg (2.48 mmol) **5.58** in 10 mL THF was added 1.9 mL (2.6 mmol) 1.40 M MeLi/Et<sub>2</sub>O at -78 °C. After 30 minutes the flask was placed into an ice bath and stirring continued for 30 minutes. The reaction was recooled to -78 °C then treated with 0.90 mL (5.2 mmol) HMPA and stirred until homogeneous. After the addition of 0.70 mL (6.3 mmol) ethyl bromoacetate the reaction was kept at -78 °C for 30 minutes. The cold bath was removed. After 1 hour the reaction was poured into 30 mL half sat. aq. NH<sub>4</sub>Cl and extracted with 30 mL EtOAc and 10 mL EtOAc. The organic layers were combined, washed with 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (5:1 hexanes/EtOAc) to afford 768 mg (90%, >20:1 alkylation) **5.59** as a white wax. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.28-7.24 (m, 1H), 6.80-6.72 (m, 3H), 5.78 (ddd, *J* = 17.2, 10.6, 5.8 Hz, 0.5H), 5.65 (ddd, *J* = 17.2, 10.3, 8.0 Hz, 0.5H), 5.01-4.96 (m, 1H), 4.89 (dt, *J* = 17.3, 1.6 Hz, 0.5H), 4.84-4.80 (m, 0.5H), 4.06-3.94 (m, 2H), 3.81 (s, 3H), 3.09 (dddd, *J* = 16.7, 12.5, 8.9, 3.8 Hz, 1H), 2.63 (t, *J* = 11.7 Hz, 0.5H), 2.60-2.50 (m, 2.5H), 2.43 (ddd, *J* = 16.5, 9.4, 6.9 Hz, 1H), 2.22-2.00 (m, 3H), 1.92 (ddd, *J* = 20.2, 16.7, 3.5 Hz, 1H), 1.54 (ddd, *J* = 12.6, 7.8, 5.0 Hz, 1H), 1.20-1.16 (m, 3H), 0.98 (d, *J* = 7.0 Hz, 1.5H), 0.81 (d, *J* = 6.9 Hz, 1.5H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) mixture δ 210.02, 209.97, 172.7, 172.7, 159.97, 159.85, 143.3, 143.0, 142.8, 138.5, 130.0, 129.8, 115.9, 113.5, 111.90, 111.88, 60.31, 60.29, 55.19, 55.18, 53.6, 53.0, 52.3, 52.5, 47.4, 46.7, 41.2, 41.0, 37.4, 36.4, 32.45, 32.44, 26.0, 25.7, 19.0, 14.1, 11.0; IR (thin film) 3076, 2962, 2938, 2836, 1731, 1715, 1599, 1584, 1262,

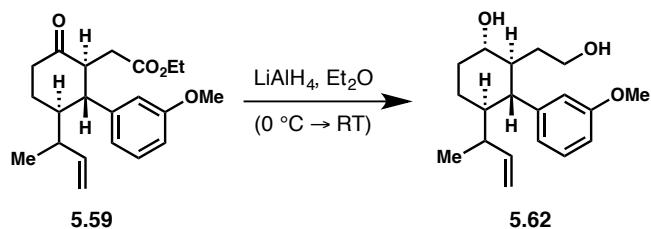
1195, 1156, 1041  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{21}\text{H}_{28}\text{O}_4$   $[\text{M}+\text{Na}]^+$  367.1885 found 367.1866.

### Ketoester 5.61



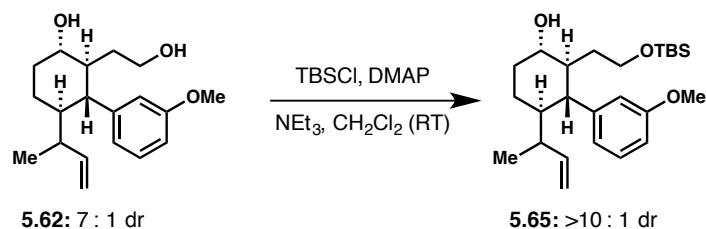
A 100 mL round bottom flask containing 30 mg (0.087 mmol) **5.59** dissolved in 2 mL  $\text{CH}_2\text{Cl}_2$  was cooled to  $-78\text{ }^\circ\text{C}$  and treated with a stream of ozone. After the solution became blue, the solution was sparged with oxygen. A single portion of 0.2 mL  $\text{AcOH}$ , 30 mg (0.46 mmol)  $\text{Zn}$  powder and 1 mg (0.0059 mmol)  $\text{AgNO}_3$  was added. After stirring at  $-78\text{ }^\circ\text{C}$  for 2 hours, the bath was removed and stirring continued for 2 hours. The reduction was determined complete by TLC and the solution was filtered over Celite, washing with  $\text{CH}_2\text{Cl}_2$ . The filtrate was washed with water, sat. aq.  $\text{NaHCO}_3$ , brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. A round bottom flask containing crude aldehyde in toluene was treated with  $\text{TsOH}\cdot\text{H}_2\text{O}$  and heated to  $140\text{ }^\circ\text{C}$  under a Dean–Stark trap. After 2 hours the reaction was cooled, diluted with  $\text{Et}_2\text{O}$ , washed with half sat. aq.  $\text{NaHCO}_3$  and brine. The aqueous layers were combined and back extracted with  $\text{Et}_2\text{O}$ . All organic layers were combined, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo to afford a crude material assigned as **5.61** as a 1.6:1 mixture of diastereomers.

### Diol 5.62



A 100 mL 3-neck round bottom flask fitted with a thermometer was charged with 15 mL Et<sub>2</sub>O, cooled in an ice bath and 260 mg (6.85 mmol) LiAlH<sub>4</sub> added. After 10 minutes 768 mg (2.23 mmol) dicarbonyl was added with the assistance of 7 mL Et<sub>2</sub>O ensuring the internal temperature remained below 4 °C over the course of 30 minutes. The ice bath was removed. After 3 hours 0.25 mL EtOAc was added at 0 °C then 0.25 mL water, 0.25 5 M NaOH, 0.75 mL water and 0.50 g Na<sub>2</sub>SO<sub>4</sub>. After 30 minutes of vigorous stirring the contents were filtered over Celite®; the filter cake was washed with 150 mL Et<sub>2</sub>O. The filtrate was concentrated and the residue purified by column chromatography (1:2 hexanes/EtOAc) to afford 624 mg (91%, 4:1 dr) **5.62** as a viscous colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) mixture of diastereomers and slow rotation δ 7.22 (bs, 1H), 6.76-6.69 (m, 3H), 5.83-5.67 (m, 1H), 4.99 (dd, *J* = 10.3, 1.4 Hz, 0.5), 4.91-4.89 (m, 0.5), 4.85-4.79 (m, 1H), 3.81 (s, 3H), 3.59-3.54 (m, 1H), 3.45-3.37 (m, 1H), 3.31-3.23 (m, 1H), 2.27-2.20 (m, 1H), 2.12 (ddq, *J* = 12.6, 8.5, 4.1 Hz, 1H), 1.93-1.82 (m, 2H), 1.71-1.40 (m, 5H), 1.19 (quintett, *J* = 12.7, 2.7 Hz, 1H), 0.89 (d, *J* = 6.8 Hz, 0.2H), 0.88 (d, *J* = 7.0 Hz, 1.3H), 0.86 (d, *J* = 6.9 Hz, 0.2H), 0.80 (d, *J* = 6.9 Hz, 1.3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) mixture of diastereomers and slow rotation δ 145.0, 144.8, 144.1, 139.6, 115.1, 112.7, 74.8, 74.6, 62.2, 62.1, 55.1, 55.1, 49.91, 49.87, 48.1, 47.4, 37.6, 36.6, 35.05, 35.00, 34.7, 34.5, 23.3, 23.2, 19.0, 11.1; IR (thin film) 3304, 3077, 2997, 2932, 2882, 1599, 1584, 1487, 1260, 1046 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>19</sub>H<sub>28</sub>O<sub>3</sub> [M+Na]<sup>+</sup> 327.1936 found 327.1933.

### TBS ether **5.65**

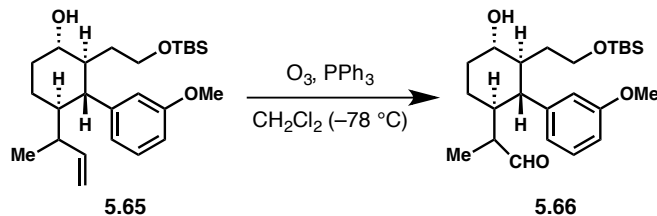


To a 1 dram vial containing 15 mg (0.050 mmol) **5.62** in 1 mL CH<sub>2</sub>Cl<sub>2</sub> was added 7 mg (0.057 mmol) DMAP and 0.05 mL (0.36 mmol) NEt<sub>3</sub>. After the addition of 19 mg (0.13 mmol) TBSCl the reaction stirred for 3 days. The reaction was quenched with sat. aq. NaHCO<sub>3</sub>, extracted thrice with EtOAc. The organic layers were combined and washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (10:1 hexanes/EtOAc) to afford 18 mg (86%, >10:1 dr) **5.65** as a colorless oil.

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.21 (bs, 1H), 6.81-6.61 (m, 3H), 5.75-5.68 (m, 1H), 4.98 (dd, *J* = 10.3, 2.0 Hz, 0.5H), 4.89 (dt, *J* = 10.5, 1.6 Hz, 0.5H), 4.84-4.78 (m, 1H), 4.66 (bs, 1H), 3.80 (s, 3H), 3.60-3.54 (m, 1H), 3.34 (dtd, *J* = 14.7, 10.1, 4.4 Hz, 1H), 3.28-3.20 (m, 1H), 2.25-2.12 (m, 2H), 1.93-1.80 (m, 1H), 1.74-1.50 (m, 4H), 1.45-1.34 (m, 2H), 1.29-1.12 (m, 1H), 0.88 (d, *J* = 1.6 Hz, 10.5H), 0.80 (d, *J* = 6.9 Hz, 1.5H), 0.043 (s, 1.5H), 0.038 (s, 1.5H), 0.035 (s, 1.5H), 0.031 (s, 1.5H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 145.4, 145.1, 144.2, 139.9, 115.0, 112.6, 74.3, 74.2, 63.1, 55.12, 55.11, 50.60, 50.54, 48.3, 47.6, 37.8, 36.7, 36.6, 35.0, 34.9, 34.4, 34.3, 25.9, 23.4, 23.2, 19.0, 18.2, 11.2, -5.5, -5.6; IR (thin film) 3416, 3077, 2953, 2929, 2858, 2882, 1599, 1584, 1257, 1080, 1048, 836, 777 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>25</sub>H<sub>42</sub>O<sub>3</sub>Si [M+Na]<sup>+</sup> 441.2801 found 441.2784.

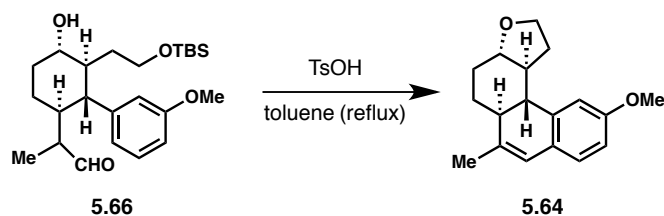


## Aldehyde 5.66



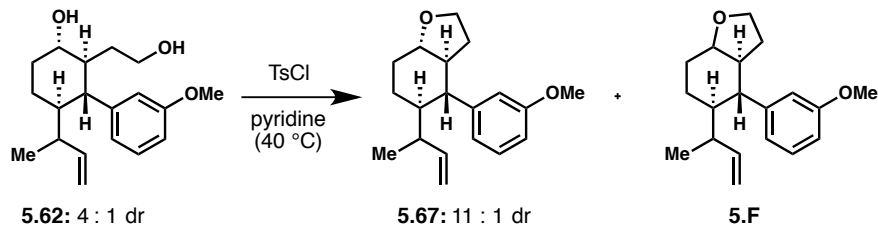
To a 1 dram vial containing 12 mg (0.029 mmol) **5.65** in 0.5 mL  $\text{CH}_2\text{Cl}_2$  was bubbled  $\text{O}_3/\text{O}_2$  at  $-78\text{ }^\circ\text{C}$  until the solution turned blue. The flask was purged with  $\text{O}_2$  until the blue color faded, then 9 mg (0.034 mmol)  $\text{PPh}_3$  in 0.2 mL  $\text{CH}_2\text{Cl}_2$  was added. Stirring was continued at  $-78\text{ }^\circ\text{C}$  for 1 hour then the bath was removed. After 2 hours the contents were diluted with EtOAc, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (5:1 hexanes/EtOAc) to afford 8 mg (67%) **5.66** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  9.47 (s, 0.5H), 9.40 (s, 0.5H), 7.21 (bs, 1H), 6.88-6.57 (m, 3H), 4.80 (bs, 1H), 3.81 (s, 1.5H), 3.80 (s, 1.5H), 3.61-3.58 (m, 1H), 3.39 (qd,  $J = 10.1, 3.7$  Hz, 1H), 3.31-3.24 (m, 1H), 2.25-2.13 (m, 2H), 1.97-1.86 (m, 2H), 1.74-1.22 (m, 5H), 1.00 (d,  $J = 7.1$  Hz, 1.5H), 1.01-0.89 (m, 1H), 0.95 (d,  $J = 7.1$  Hz, 1.5H), 0.89 (s, 4.5H), 0.89 (s, 4.5H), 0.05 (s, 6H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  205.1, 203.9, 144.2, 144.1, 73.8, 73.8, 63.0, 55.2, 48.2, 47.8, 46.5, 35.1, 34.8, 34.4, 34.0, 26.8, 25.9, 24.6, 18.2, 10.9, 6.8,  $-5.5, -5.6$ ; IR (thin film) 3410, 2929, 2857, 1721, 1599, 1464, 1256, 1081, 1047, 836, 778  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{24}\text{H}_{40}\text{O}_4\text{Si}$   $[\text{M}+\text{Na}]^+$  443.2594 found 443.2595.

## Dihydronaphthalene 5.64



To a 5 mL round bottom flask containing 7 mg (0.017 mmol) **5.66** was added a solution of 2 mg (0.011 mmol) TsOH•H<sub>2</sub>O predried with 2 mL toluene over a Hickmann still. The reaction was refluxed over a Hickmann still for 2 hours. The reaction was cooled, diluted with sat. aq. NaHCO<sub>3</sub>, EtOAc and the layers separated. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (5:1 hexanes/EtOAc) to afford 3 mg (66%, single diastereomer) **5.64** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.05 (s, 1H), 6.94 (d, *J* = 8.2 Hz, 1H), 6.71 (dd, *J* = 8.3, 2.4 Hz, 1H), 6.25 (s, 1H), 4.09 (q, *J* = 8.1 Hz, 1H), 3.97 (td, *J* = 8.6, 3.5 Hz, 1H), 3.81 (s, 3H), 3.20 (ddd, *J* = 11.0, 9.7, 3.5 Hz, 1H), 2.75 (dtd, *J* = 11.1, 7.5, 3.6 Hz, 1H), 2.51 (dd, *J* = 14.0, 10.5 Hz, 1H), 2.24 (ddq, *J* = 14.0, 10.5, 3.5 Hz, 2H), 2.12 (t, *J* = 14.0 Hz, 1H), 1.91 (s, 3H), 1.88-1.74 (m, 2H), 1.55-1.39 (m, 2H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 157.9, 139.5, 138.4, 128.7, 125.7, 123.6, 112.4, 109.9, 83.7, 67.1, 55.2, 46.8, 46.4, 41.9, 31.8, 29.9, 27.0, 20.8; IR (thin film) 2935, 2869, 1607, 1489, 1157, 1064, 1041 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>18</sub>H<sub>22</sub>O<sub>2</sub> [M+NH<sub>4</sub>]<sup>+</sup> 288.1964 found 288.1955.

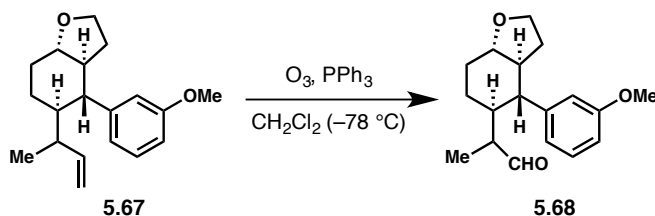
## Perhydrobenzofuran 5.67 and 5.F



To a 25 mL round bottom flask containing 510 mg (1.68 mmol) **5.62** in 8 mL pyridine was added 820 mg (4.30 mmol) *p*-toluenesulfonyl chloride at room temperature. The flask was placed in a 40 °C oil bath. After 15 hours the reaction was cooled to room temperature and quenched with 10 mL sat. aq. NaHCO<sub>3</sub>, 5 mL water and extracted with 30 mL EtOAc. The organic layer was washed thrice with 10 mL 6 M HCl then 5 mL water. The acidic aqueous washings were combined, back extracted with 10 mL EtOAc and all organic phases combined. The organic medium was washed with 10 mL sat. aq. NaHCO<sub>3</sub>, 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (10:1 hexanes/EtOAc) to afford 358 mg (74%, 11:1 dr) **5.67** as a colorless oil and 74 mg (15%, 2.7:1.8:1.7:1 dr) **5.F** as a colorless oil. **5.67**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) spectrum for an 8:1 mixture of diastereomers δ 7.24 (t, *J* = 7.8 Hz, 0.5H), 7.23 (t, *J* = 7.9 Hz, 0.5H), 6.79-6.72 (m, 3H), 5.80-5.70 (m, 1H), 4.99 (d, *J* = 10.5 Hz, ), 4.92 (d, *J* = 10.5 Hz, 1H), 4.84 (dd, *J* = 17.2, 11.2 Hz, 1H), 3.88 (qd, *J* = 9.1, 2.6 Hz, 1H), 3.84-3.79 (m, 1H), 3.82 (s, 3H), 3.16 (ddq, *J* = 16.4, 10.6, 5.5 Hz, 1H), 2.31 (dt, *J* = 15.5, 10.9 Hz, 1H), 2.21 (ddq, *J* = 12.0, 8.2, 3.9 Hz, 1H), 2.12-2.04 (m, 1H), 1.97 (dq, *J* = 13.7, 3.5 Hz, 1H), 1.83-1.40 (m, 7H), 1.32-1.20 (m, 1H), 0.95 (d, *J* = 7.0 Hz, 1.4H), 0.91 (d, *J* = 7.0 Hz, 0.2H), 0.84 (d, *J* = 7.0 Hz, 1.4H), 0.79 (d, *J* = 6.9 Hz, 0.2H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) major δ 144.8, 144.6, 144.1, 139.8, 129.5, 129.4, 115.0, 112.7, 110.9, 82.8, 82.6, 67.1, 67.1, 55.1, 52.1, 52.0, 50.3, 49.9, 48.0, 47.4, 37.3, 36.2, 30.6, 30.5, 30.0, 30.0, 23.51, 23.45, 19.1, 11.6; IR (thin film) 3073, 2934, 2871, 1600, 1583,

1261, 1047, 911, 778, 701  $\text{cm}^{-1}$ ; HRMS could not be obtained. **5.F**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.25-7.21 (m, 1H), 6.80-6.72 (m, 3H), 5.80-5.66 (m, 1H), 5.00-4.73 (m, 2H), 4.02 (dq,  $J = 14.7, 7.6$  Hz, 0.6H), 3.90-3.80 (m, 5H), 3.20-3.12 (m, 0.4H), 2.34-2.05 (m, 3H), 1.98-1.89 (m, 1H), 1.85-1.31 (m, 7H), 1.29-1.21 (m, 1H), 0.95 (d,  $J = 7.0$  Hz, 0.7H), 0.91 (d,  $J = 7.0$  Hz, 1.3H), 0.84 (d,  $J = 7.0$  Hz, 0.4H), 0.79 (d,  $J = 6.9$  Hz, 0.8H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm) mixture  $\delta$  159.6, 159.5, 146.0, 145.7, 144.8, 144.6, 144.2, 144.1, 139.80, 139.76, 129.4, 129.4, 129.3, 129.1, 115.0, 114.8, 112.7, 112.6, 110.94, 110.86, 110.8, 82.8, 82.6, 77.6, 77.4, 67.1, 67.1, 65.8, 65.7, 55.13, 55.11, 52.1, 52.0, 50.3, 49.9, 48.1, 48.0, 47.7, 47.4, 46.6, 46.1, 46.0, 45.8, 37.9, 37.3, 36.8, 36.2, 30.9, 30.9, 30.6, 30.5, 30.0, 28.09, 28.06, 23.51, 23.45, 19.7, 19.6, 19.1, 18.9, 11.6, 11.2; IR (thin film) 2934, 2869, 1599, 1486, 1458, 1261, 1047, 667, 702  $\text{cm}^{-1}$ ; HRMS could not be obtained.

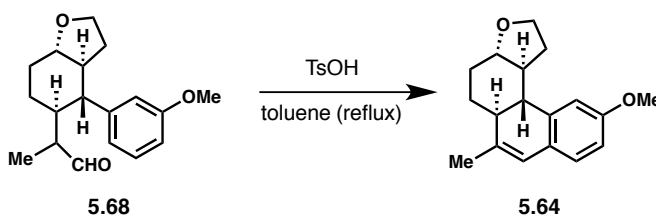
### Aldehyde **5.68**



To a 1 dram vial containing 73 mg (0.255 mmol) **5.67** in 2.5 mL  $\text{CH}_2\text{Cl}_2$  was bubbled  $\text{O}_3/\text{O}_2$  at  $-78^\circ\text{C}$  until the solution turned blue. The flask was purged with  $\text{O}_2$  until the blue color faded, then 80 mg (0.305 mmol)  $\text{PPh}_3$  was added. Stirring was continued at  $-78^\circ\text{C}$  for 2 hours then the bath was removed. After 6 hours the reaction was dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (3:1 $\rightarrow$ 1:2 hexanes/EtOAc) to afford 65 mg (89%, with  $\sim 25\%$  over oxidation) **5.68** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm) major  $\delta$  9.55 (s, 0.5H), 9.49 (s, 0.5H), 7.29-7.23 (m, 1H),

6.81-6.74 (m, 3H), 4.09 (s, 1H), 3.94-3.85 (m, 2H), 3.83 (s, 1.5H), 3.82 (s, 1.5H), 3.21 (dtd,  $J = 17.8, 10.5, 3.5$  Hz, 1H), 2.56 (t,  $J = 10.9$  Hz, 1H), 2.39-2.16 (m, 3H), 2.10-1.99 (m, 1H), 1.83-1.35 (m, 4H), 1.07 (d,  $J = 7.1$  Hz, 1.5H), 1.00 (d,  $J = 7.1$  Hz, 1.5H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  205.1, 204.1, 143.62, 143.59, 129.9, 129.84, 129.80, 111.8, 67.3, 67.2, 55.3, 55.2, 51.9, 51.7, 47.6, 47.0, 46.4, 42.2, 30.8, 30.4, 30.1, 29.9, 27.0, 24.8, 14.3, 11.3, 7.1; IR (thin film) 2936, 2879, 1720, 1600, 1485, 1262, 1158, 1045  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{24}\text{O}_3$  [ $\text{M} + \text{Na}$ ] $^+$  311.1623 found 311.1635.

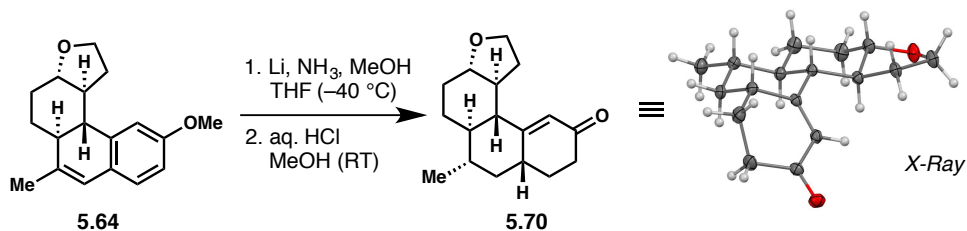
### Dihydronaphthalene 5.64



To a 10 mL round bottom flask containing 63 mg (0.218 mmol) **5.68** was added a solution of 2 mg (0.011 mmol)  $\text{TsOH} \cdot \text{H}_2\text{O}$  predried with 3 mL toluene under a Hickmann still. The reaction was heated to reflux under a Hickmann still for 2 hours. The reaction was cooled, diluted with sat. aq.  $\text{NaHCO}_3$ ,  $\text{EtOAc}$  and the layers separated. The organic layer was washed with brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (5:1 hexanes/ $\text{EtOAc}$ ) to afford 31 mg (52%, single diastereomer) **5.64** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.05 (s, 1H), 6.94 (d,  $J = 8.2$  Hz, 1H), 6.71 (dd,  $J = 8.3, 2.4$  Hz, 1H), 6.25 (s, 1H), 4.09 (q,  $J = 8.1$  Hz, 1H), 3.97 (td,  $J = 8.6, 3.5$  Hz, 1H), 3.81 (s, 3H), 3.20 (ddd,  $J = 11.0, 9.7, 3.5$  Hz, 1H), 2.75 (dtd,  $J = 11.1, 7.5, 3.6$  Hz, 1H), 2.51 (dd,  $J = 14.0, 10.5$  Hz, 1H), 2.24 (ddq,  $J = 14.0, 10.5, 3.5$  Hz, 2H), 2.12 (t,  $J = 14.0$  Hz, 1H), 1.91 (s, 3H), 1.88-1.74 (m, 2H), 1.55-1.39 (m, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$

157.9, 139.5, 138.4, 128.7, 125.7, 123.6, 112.4, 109.9, 83.7, 67.1, 55.2, 46.8, 46.4, 41.9, 31.8, 29.9, 27.0, 20.8; IR (thin film) 2935, 2869, 1607, 1489, 1157, 1064, 1041  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{22}\text{O}_2$   $[\text{M}+\text{NH}_4]^+$  288.1964 found 288.1955.

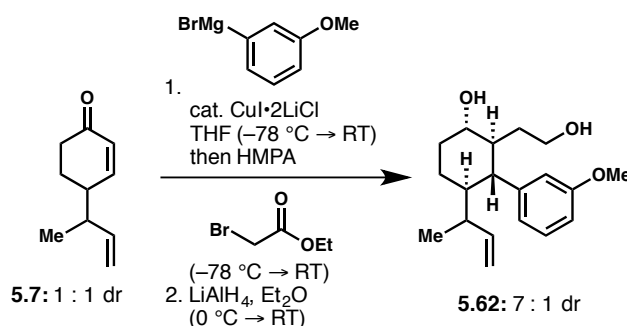
### Enone 5.70



To a 10 mL round bottom flask charged with a glass stir bar was condensed 4 mL ammonia followed by the addition of 30 mg (0.11 mmol) **5.64** in 1.1 mL THF and 0.30 mL (4.9 mmol) MeOH. Slowly, 34 mg (4.9 mmol) lithium metal was added in small pieces at  $-40$  °C. After complete addition and the blue color discharged, 275mg solid  $\text{NH}_4\text{Cl}$  was added and the ammonia evaporated. The white residue taken up in 3 mL water and 4 mL EtOAc. The organic layer was separated and washed with 1 mL brine. The aqueous layers were combined and back extracted with 2 mL EtOAc. The organic layers were combined, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material in a 10 mL round bottom flask was dissolved in 1.1 mL MeOH and treated with 0.1 mL 6 M HCl. After 16 hours at room temperature half sat. aq.  $\text{NaHCO}_3$  and EtOAc were added. The layers were separated and the aqueous extracted with EtOAc. All organic layers were collected washed with brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (3:1 hexanes/EtOAc) to afford 16 mg (52% over 2 steps) **5.70** as a white solid that recrystallized from benzene to afford colorless prisms (mp =  $85\text{--}86$  °C).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.85 (s, 1H), 3.97-3.88 (m, 2H), 3.15 (ddd,  $J = 11.4, 9.6, 3.4$ , 1H), 2.42-2.29 (m, 4H), 2.23-2.12

(m, 3H), 1.90 (dt,  $J = 12.7, 4.2$  Hz, 1H), 1.84 (t,  $J = 10.5$  Hz, 1H), 1.75-1.66 (m, 1H), 1.56-1.26 (m, 5H), 1.15-1.08 (m, 1H), 0.98-0.89 (m, 1H), 0.96 (d,  $J = 6.5$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  199.9, 168.8, 121.1, 82.9, 67.1, 51.6, 51.4, 45.1, 43.2, 38.4, 37.4, 34.1, 29.9, 29.7, 29.2, 27.4, 19.6; IR (thin film) 2927, 2868, 1667, 1622  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{17}\text{H}_{24}\text{O}_2$   $[\text{M}+\text{Na}]^+$  283.1674 found 283.1682.

### Diol 5.62

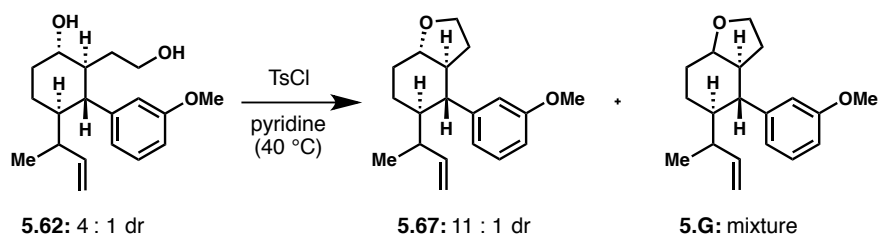


3-Methoxyphenylmagnesium bromide was prepared by addition of 7 mL (56 mmol) 3-bromoanisole in 7 mL THF to 1.77 g (73 mmol) magnesium metal in 28 mL THF activated by dibromoethane maintaining 40-50  $^\circ\text{C}$ . A 250 mL 3-neck round bottom flask fitted with a low temperature thermometer was charged with 110 mL THF and 31 mL (39 mmol) 1.26 M 3-methoxyphenylmagnesium bromide. At  $-78\text{ }^\circ\text{C}$ , 9 mL (3.3 mmol) 0.37 M  $\text{CuI}\cdot 2\text{LiCl}/\text{THF}$  was added followed by 5.01 g (33.3 mmol, 1:1 dr) **5.7** in 15 mL THF while maintaining at least  $-72\text{ }^\circ\text{C}$ . Stirring at  $-78\text{ }^\circ\text{C}$  was continued for 2 hours before the cold bath was removed. After 1 hour the reaction was recooled to  $-78\text{ }^\circ\text{C}$ . To the solution was added 12 mL (82.8 mmol) HMPA. After 1 hour 15 mL (136 mmol) ethyl bromoacetate was added. The reaction was stirred at  $-78\text{ }^\circ\text{C}$  for 10 minutes then the cold bath removed. After stirring for 28 hours at room temperature, the reaction was cooled in an ice bath and treated with 100 mL sat. aq.  $\text{NH}_4\text{Cl}$ , 100 mL water, extracted with

100 mL and twice with 50 mL Et<sub>2</sub>O. The organic layers were combined, washed with 50 mL half sat. aq. NH<sub>4</sub>Cl, 30 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford a crude oil. To an ice cooled 1 L 3-neck round bottom flask fitted with an overhead stirrer and thermometer, containing 300 mL Et<sub>2</sub>O was added 5.05 g (133 mmol) LiAlH<sub>4</sub>. The crude oil in 50 mL Et<sub>2</sub>O was added via canula and washed with an additional 50 mL Et<sub>2</sub>O, maintaining an internal temperature below 6 °C over the course of 1.5 hours. The ice bath was removed after 30 minutes. After 5 hours the flask was recooled in an ice bath and 5 mL EtOAc added over the course of 30 minutes. An oil bubbler was connected and 5 mL water, 5 mL 5 M NaOH, 15 mL water and 40 g Na<sub>2</sub>SO<sub>4</sub> were added with careful temperature monitoring. After stirring the contents overnight, the solution was filtered over Celite and the filter cake washed with Et<sub>2</sub>O (~1.2 L as determined by TLC). The filtrate was concentrated and the residue purified by column chromatography (1:2→1:1 hexanes/EtOAc) to afford 5.68 g (56% over 2 steps, 7:1 dr) **5.62** as a viscous colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) mixture of diastereomers and slow rotation δ 7.22 (bs, 1H), 6.76-6.69 (m, 3H), 5.83-5.67 (m, 1H), 4.99 (dd, *J* = 10.3, 1.4 Hz, 0.5), 4.91-4.89 (m, 0.5), 4.85-4.79 (m, 1H), 3.81 (s, 3H), 3.59-3.54 (m, 1H), 3.45-3.37 (m, 1H), 3.31-3.23 (m, 1H), 2.27-2.20 (m, 1H), 2.12 (ddq, *J* = 12.6, 8.5, 4.1 Hz, 1H), 1.93-1.82 (m, 2H), 1.71-1.40 (m, 5H), 1.19 (quintett, *J* = 12.7, 2.7 Hz, 1H), 0.89 (d, *J* = 6.8 Hz, 0.2H), 0.88 (d, *J* = 7.0 Hz, 1.3H), 0.86 (d, *J* = 6.9 Hz, 0.2H), 0.80 (d, *J* = 6.9 Hz, 1.3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) mixture of diastereomers and slow rotation δ 145.0, 144.8, 144.1, 139.6, 115.1, 112.7, 74.8, 74.6, 62.2, 62.1, 55.1, 55.1, 49.91, 49.87, 48.1, 47.4, 37.6, 36.6, 35.05, 35.00, 34.7, 34.5, 23.3, 23.2, 19.0, 11.1; IR (thin film) 3304, 3077, 2997, 2932, 2882, 1599, 1584, 1487, 1260, 1046 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>19</sub>H<sub>28</sub>O<sub>3</sub> [M+Na]<sup>+</sup> 327.1936 found 327.1933.



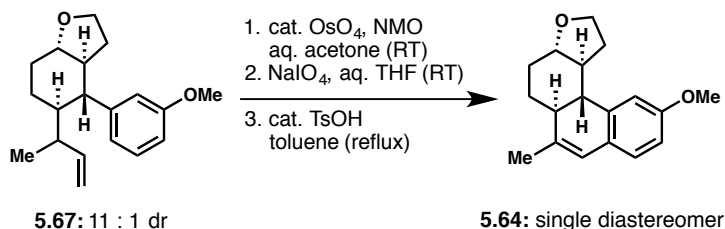
### **Trans-Perhydrobenzofuran 5.67 and 5.G**



To a 25 mL round bottom flask containing 510 mg (1.68 mmol, 4:1 dr) **5.62** in 8 mL pyridine was added 820 mg (4.30 mmol) p-toluenesulfonyl chloride at room temperature. The flask was placed in a 40 °C oil bath. After 15 hours the reaction was cooled to room temperature and quenched with 10 mL sat. aq. NaHCO<sub>3</sub>, 5 mL water and extracted with 30 mL EtOAc. The organic layer was washed thrice with 10 mL 6 M HCl then 5 mL water. The acidic aqueous washings were combined, back extracted with 10 mL EtOAc and all organic phases combined. The organic medium was washed with 10 mL sat. aq. NaHCO<sub>3</sub>, 10 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (10:1 hexanes/EtOAc) to afford 358 mg (74%, 11:1 dr) **5.67** as a colorless oil and 74 mg (15%, 2.7:1.8:1.7:1 dr) **5.G** as a colorless oil. **5.67**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) spectrum for an 8:1 mixture of diastereomers δ 7.24 (t, *J* = 7.8 Hz, 0.5H), 7.23 (t, *J* = 7.9 Hz, 0.5H), 6.79-6.72 (m, 3H), 5.80-5.70 (m, 1H), 4.99 (d, *J* = 10.5 Hz, ), 4.92 (d, *J* = 10.5 Hz, 1H), 4.84 (dd, *J* = 17.2, 11.2 Hz, 1H), 3.88 (qd, *J* = 9.1, 2.6 Hz, 1H), 3.84-3.79 (m, 1H), 3.82 (s, 3H), 3.16 (ddq, *J* = 16.4, 10.6, 5.5 Hz, 1H), 2.31 (dt, *J* = 15.5, 10.9 Hz, 1H), 2.21 (ddq, *J* = 12.0, 8.2, 3.9 Hz, 1H), 2.12-2.04 (m, 1H), 1.97 (dq, *J* = 13.7, 3.5 Hz, 1H), 1.83-1.40 (m, 7H), 1.32-1.20 (m, 1H), 0.95 (d, *J* = 7.0 Hz, 1.4H), 0.91 (d, *J* = 7.0 Hz, 0.2H), 0.84 (d, *J* = 7.0 Hz, 1.4H), 0.79 (d, *J* = 6.9 Hz, 0.2H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) major δ 144.8, 144.6, 144.1, 139.8, 129.5, 129.4, 115.0, 112.7, 110.9, 82.8, 82.6, 67.1, 67.1, 55.1, 52.1, 52.0, 50.3, 49.9, 48.0, 47.4, 37.3, 36.2, 30.6, 30.5, 30.0, 30.0, 23.51, 23.45, 19.1, 11.6; IR (thin film) 3073,

2934, 2871, 1600, 1583, 1261, 1047, 911, 778, 701  $\text{cm}^{-1}$ ; HRMS could not be obtained. **5.G**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.25-7.21 (m, 1H), 6.80-6.72 (m, 3H), 5.80-5.66 (m, 1H), 5.00-4.73 (m, 2H), 4.02 (dq,  $J = 14.7, 7.6$  Hz, 0.6H), 3.90-3.80 (m, 5H), 3.20-3.12 (m, 0.4H), 2.34-2.05 (m, 3H), 1.98-1.89 (m, 1H), 1.85-1.31 (m, 7H), 1.29-1.21 (m, 1H), 0.95 (d,  $J = 7.0$  Hz, 0.7H), 0.91 (d,  $J = 7.0$  Hz, 1.3H), 0.84 (d,  $J = 7.0$  Hz, 0.4H), 0.79 (d,  $J = 6.9$  Hz, 0.8H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm) mixture  $\delta$  159.6, 159.5, 146.0, 145.7, 144.8, 144.6, 144.2, 144.1, 139.80, 139.76, 129.4, 129.4, 129.3, 129.1, 115.0, 114.8, 112.7, 112.6, 110.94, 110.86, 110.8, 82.8, 82.6, 77.6, 77.4, 67.1, 67.1, 65.8, 65.7, 55.13, 55.11, 52.1, 52.0, 50.3, 49.9, 48.1, 48.0, 47.7, 47.4, 46.6, 46.1, 46.0, 45.8, 37.9, 37.3, 36.8, 36.2, 30.9, 30.9, 30.6, 30.5, 30.0, 28.09, 28.06, 23.51, 23.45, 19.7, 19.6, 19.1, 18.9, 11.6, 11.2; IR (thin film) 2934, 2869, 1599, 1486, 1458, 1261, 1047, 667, 702  $\text{cm}^{-1}$ ; HRMS could not be obtained.

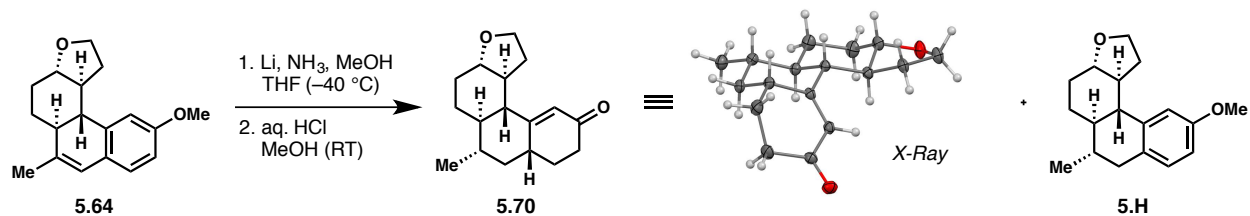
### Dihydronaphthalene 5.64



To a 25 mL round bottom flask containing 320 mg (1.12 mmol, 11:1 dr) **5.67** in 8 mL acetone and 2.6 mL nano-pure water was added 0.14 mL (0.022 mmol) 4 wt%  $\text{OsO}_4$  in water then 270 mg (2.30 mmol) NMO at 0 °C open to air. The ice bath was immediately removed and after stirring 12 hours at room temperature 10 mL water and 1.20 g  $\text{NaHSO}_3$  was added. After 10 minutes the solution was extracted thrice with 15 mL EtOAc. All organic layers were combined, washed with 10 mL sat. aq.  $\text{NaHCO}_3$  and 10 mL brine. The aqueous layers were combined and back extracted with 5 mL EtOAc. The organic layers were combined dried over  $\text{MgSO}_4$ , filtered

and all volatiles removed in vacuo. The crude material was subjected to the next reaction without further purification. To a 100 mL round bottom flask containing the crude oil was added 10 mL 1:1 THF/water then 485 mg (2.27 mmol) NaIO<sub>4</sub> open to air. After 1 hour 10 mL water was added followed by 1.20 g NaHSO<sub>3</sub>. The mixture was extracted thrice with 15 mL EtOAc. All organic layers were combined, washed with 10 mL sat. aq. NaHCO<sub>3</sub> and 10 mL brine. The aqueous layers were combined and back extracted with 5 mL EtOAc. The organic layers were combined dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was subjected to the next reaction without further purification. To a 25 mL round bottom flask containing crude aldehyde was added a solution of 12 mg (0.063 mmol) TsOH•H<sub>2</sub>O predried with 11 mL toluene over a Hickmann still. The reaction was refluxed over a Hickmann still for 2 hours. The reaction was cooled, diluted with 10 mL sat. aq. NaHCO<sub>3</sub>, 20 mL EtOAc and the layers separated. The organic layer was washed with 5 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (7:1→5:1 hexanes/EtOAc) to afford 262 mg (86% over 3 steps, single diastereomer) **5.64** as a white solid that was recrystallized from Et<sub>2</sub>O to afford colorless prisms (mp =134–136 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.05 (s, 1H), 6.94 (d, *J* = 8.2 Hz, 1H), 6.71 (dd, *J* = 8.3, 2.4 Hz, 1H), 6.25 (s, 1H), 4.09 (q, *J* = 8.1 Hz, 1H), 3.97 (td, *J* = 8.6, 3.5 Hz, 1H), 3.81 (s, 3H), 3.20 (ddd, *J* = 11.0, 9.7, 3.5 Hz, 1H), 2.75 (dtd, *J* = 11.1, 7.5, 3.6 Hz, 1H), 2.51 (dd, *J* = 14.0, 10.5 Hz, 1H), 2.24 (ddq, *J* = 14.0, 10.5, 3.5 Hz, 2H), 2.12 (t, *J* = 14.0 Hz, 1H), 1.91 (s, 3H), 1.88-1.74 (m, 2H), 1.55-1.39 (m, 2H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 157.9, 139.5, 138.4, 128.7, 125.7, 123.6, 112.4, 109.9, 83.7, 67.1, 55.2, 46.8, 46.4, 41.9, 31.8, 29.9, 27.0, 20.8; IR (thin film) 2935, 2869, 1607, 1489, 1157, 1064, 1041 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>18</sub>H<sub>22</sub>O<sub>2</sub> [M+NH<sub>4</sub>]<sup>+</sup> 288.1964 found 288.1955.

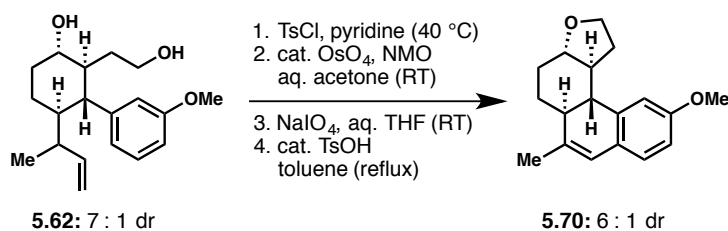
## Tetracyclic Enone 5.70 and Anisole 5.H



To a 50 mL round bottom flask charged with a glass stir bar was condensed 20 mL ammonia followed by the addition of 248 mg (0.917 mmol) **12** in 4.5 mL THF and 1.9 mL (47 mmol) MeOH. Slowly, 250 mg (36 mmol) lithium metal was added in small pieces at -40 °C. The side of the flasks were washed with 0.3 mL (7.4 mmol) MeOH after 150 mg lithium metal was added. After complete addition and the blue color discharged, 2.1 g (39 mmol) solid NH<sub>4</sub>Cl was added and the ammonia evaporated. The white residue was taken up in 5 mL water and 20 mL EtOAc. The organic layer was separated and washed with 5 mL brine. The aqueous layers were combined and back extracted with 5 mL EtOAc. The organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material in a 50 mL round bottom flask was dissolved in 10 mL MeOH and treated with 0.5 mL 6 M HCl. After 16 hours at room temperature 20 mL half sat. aq. NaHCO<sub>3</sub> and 20 mL EtOAc were added. The layers were separated and the aqueous extracted with 10 mL and 5 mL EtOAc. All organic layers were collected washed with 5 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (4:1→2:1 hexanes/EtOAc) to afford 44 mg (18% over 2 steps) **5.G** as a white solid that was recrystallized from Et<sub>2</sub>O to afford colorless needles (mp = 80–82 °C) and 128 mg (53% over 2 steps) **5.70** as a white solid that recrystallized from benzene to afford colorless prisms (mp = 85–86 °C). **5.H**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.04 (d, *J* = 8.3 Hz, 1H), 6.96 (d, *J* = 1.9 Hz, 1H), 6.72 (dd, *J* = 8.2, 2.3 Hz, 1H), 4.03-4.00 (m, 2H), 3.83 (s, 3H), 3.31 (td, *J* = 10.6, 3.1 Hz, 1H), 2.79 (dd, *J* = 15.3,

6.3 Hz, 1H), 2.61 (dq,  $J = 11.5, 5.8$  Hz, 1H), 2.41 (dd,  $J = 15.3, 6.7$  Hz, 1H), 2.33 (t,  $J = 10.7$  Hz, 1H), 2.22 (dq,  $J = 11.5, 3.6$  Hz, 1H), 2.13 (dq,  $J = 13.4, 3.5$  Hz, 1H), 1.97 (ddd,  $J = 20.8, 11.5, 9.3$  Hz, 1H), 1.75-1.64 (m, 2H), 1.45 (qd,  $J = 12.1, 3.5$  Hz, 1H), 1.29-1.21 (m, 1H), 0.96-0.89 (m, 1H), 0.94 (d,  $J = 6.8$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  157.6, 141.87, 129.9, 129.2, 112.1, 110.4, 83.7, 67.1, 55.2, 48.5, 47.7, 46.1, 37.3, 33.8, 31.8, 30.5, 29.1, 21.1; IR (thin film) 2929, 2870, 1610, 1495, 1237, 1062, 1040, 805  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{24}\text{O}_2$   $[\text{M}+\text{NH}_4]^+$  290.2120 found 290.2113. **5.70**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  5.85 (s, 1H), 3.97-3.88 (m, 2H), 3.15 (ddd,  $J = 11.4, 9.6, 3.4$ , 1H), 2.42-2.29 (m, 4H), 2.23-2.12 (m, 3H), 1.90 (dt,  $J = 12.7, 4.2$  Hz, 1H), 1.84 (t,  $J = 10.5$  Hz, 1H), 1.75-1.66 (m, 1H), 1.56-1.26 (m, 5H), 1.15-1.08 (m, 1H), 0.98-0.89 (m, 1H), 0.96 (d,  $J = 6.5$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  199.9, 168.8, 121.1, 82.9, 67.1, 51.6, 51.4, 45.1, 43.2, 38.4, 37.4, 34.1, 29.9, 29.7, 29.2, 27.4, 19.6; IR (thin film) 2927, 2868, 1667, 1622  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{17}\text{H}_{24}\text{O}_2$   $[\text{M}+\text{Na}]^+$  283.1674 found 283.1682.

### Dihydronaphthalene 5.64

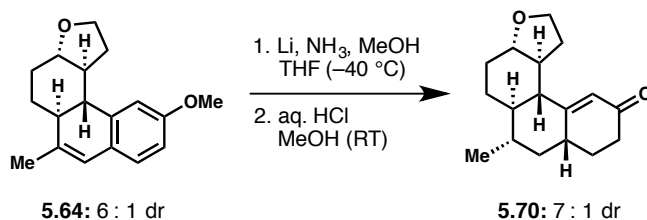


To a 500 mL round bottom flask containing 5.68 g (18.7 mmol, 7:1 dr) **5.62** in 95 mL pyridine was added 8.91 g (47.1 mmol) p-toluenesulfonyl chloride at room temperature. The flask was placed in a 40 °C oil bath. After 15 hours the reaction was cooled to room temperature and quenched with 100 mL sat. aq.  $\text{NaHCO}_3$ , 50 mL water and extracted with 300 mL and 50 mL EtOAc. The organic layers were washed thrice with ice cold 100 mL 6 M HCl then 50 mL water.

The acidic aqueous washings were combined, back extracted with 100 mL and 50 mL EtOAc and all organic phases combined. The organic medium was washed with 50 mL sat. aq. NaHCO<sub>3</sub>, 50 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was taken on to the next step. To a 500 mL round bottom flask containing crude alkene in 150 mL acetone and 50 mL nano-pure water was added 1.2 mL (0.189 mmol) 4 wt% OsO<sub>4</sub> in water then 4.40 g (37.6 mmol) NMO at 0 °C open to air. The ice bath was immediately removed and after stirring 20 hours at room temperature 9.5 g (75 mmol) Na<sub>2</sub>SO<sub>3</sub> in 50 mL water was added. After 20 minutes, 5 g NaCl was added and the solution was extracted with 150 mL, 100 mL and 50 mL EtOAc. All organic layers were combined, washed with 50 mL sat. aq. NaHCO<sub>3</sub> and 50 mL brine. The aqueous layers were combined and back extracted with 50 mL EtOAc. The organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was subjected to the next reaction without further purification. To a 1 L round bottom flask containing the crude oil was added 190 mL 1:1 THF/water then immersed in a room temperature water bath and 8.14 g (38.1 mmol) NaIO<sub>4</sub> added open to air. After 1 hour 18 g (150 mmol) Na<sub>2</sub>SO<sub>3</sub> suspended in 150 mL water was added. The mixture was extracted with 150 mL and 100 mL EtOAc. The organic layers were combined and washed with 50 mL sat. aq. NaHCO<sub>3</sub> and 50 mL brine. The aqueous layers were combined and back extracted with 50 mL EtOAc. The organic layers were combined dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was subjected to the next reaction without further purification. To a 250 mL round bottom flask containing crude aldehyde was added 180 mL toluene and 180 mg (0.95 mmol) TsOH•H<sub>2</sub>O. The reaction was refluxed over a Dean-Stark trap for 2 hours. The reaction was cooled, diluted with 50 mL sat. aq. NaHCO<sub>3</sub>, 100 mL EtOAc and the layers separated. The organic layer was washed with 50 mL brine, dried over MgSO<sub>4</sub>, filtered and all

volatiles removed in vacuo. The crude oil was purified by column chromatography (5:1 hexanes/EtOAc) to afford 4.04 g (80% over 4 steps, 6:1 dr) **5.70** as a white wax.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm) major  $\delta$  7.05 (s, 1H), 6.94 (d,  $J = 8.2$  Hz, 1H), 6.71 (dd,  $J = 8.3, 2.4$  Hz, 1H), 6.25 (s, 1H), 4.09 (q,  $J = 8.1$  Hz, 1H), 3.97 (td,  $J = 8.6, 3.5$  Hz, 1H), 3.81 (s, 3H), 3.20 (ddd,  $J = 11.0, 9.7, 3.5$  Hz, 1H), 2.75 (dtd,  $J = 11.1, 7.5, 3.6$  Hz, 1H), 2.51 (dd,  $J = 14.0, 10.5$  Hz, 1H), 2.24 (ddq,  $J = 14.0, 10.5, 3.5$  Hz, 2H), 2.12 (t,  $J = 14.0$  Hz, 1H), 1.91 (s, 3H), 1.88-1.74 (m, 2H), 1.55-1.39 (m, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm) major  $\delta$  157.9, 139.5, 138.4, 128.7, 125.7, 123.6, 112.4, 109.9, 83.7, 67.1, 55.2, 46.8, 46.4, 41.9, 31.8, 29.9, 27.0, 20.8; IR (thin film) 2935, 2869, 1607, 1489, 1157, 1064, 1041  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{18}\text{H}_{22}\text{O}_2$   $[\text{M}+\text{NH}_4]^+$  288.1964 found 288.1955.

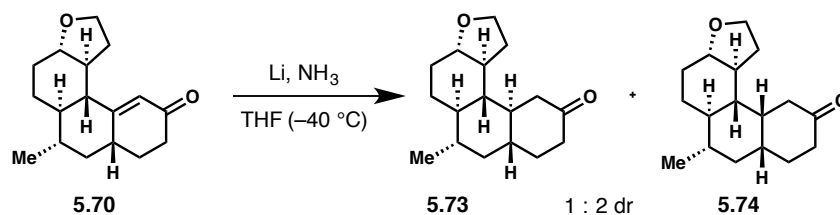
### Tetracyclic Enone **5.70**



To a 1 L 3-neck round bottom flask fitted with an overhead stirrer was added 4.04 g (14.9 mmol, 6:1 dr) **5.64** in 100 mL THF followed by 300 mL ammonia and 6 mL MeOH. At  $-40$  °C, a total of 4.17 g (600 mmol) lithium metal was added while additional 6 mL MeOH was added after 0.93 g, 1.80 g and 2.86 g lithium. After complete addition of lithium and disappearance of a blue color, an additional 1 mL MeOH and 35 g (654 mmol) solid  $\text{NH}_4\text{Cl}$  was added and the ammonia evaporated overnight. The white residue taken up in 100 mL water and extracted with 200 mL and 100 mL EtOAc. The organic layers were combined, washed twice with 50 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material in a 1 L round

bottom flask was dissolved in 150 mL MeOH and treated with 15 mL 6 M HCl. After 10 hours at room temperature 200 mL sat. aq. NaHCO<sub>3</sub> and 100 mL water were added while controlling the temperature. The mixture was extracted with 300 mL and thrice with 100 mL EtOAc. All organic layers were collected washed with 50 mL sat. aq. NaHCO<sub>3</sub>, 50 mL water and 50 mL brine. These aqueous washings were combined and back extracted with 50 mL EtOAc. The organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (4:1→2:1 hexanes/EtOAc) to afford 2.55 g (65% over 2 steps, 7:1 dr) **5.70** as a white wax. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) major δ 5.85 (s, 1H), 3.97-3.88 (m, 2H), 3.15 (ddd, *J* = 11.4, 9.6, 3.4, 1H), 2.42-2.29 (m, 4H), 2.23-2.12 (m, 3H), 1.90 (dt, *J* = 12.7, 4.2 Hz, 1H), 1.84 (t, *J* = 10.5 Hz, 1H), 1.75-1.66 (m, 1H), 1.56-1.26 (m, 5H), 1.15-1.08 (m, 1H), 0.98-0.89 (m, 1H), 0.96 (d, *J* = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) major δ 199.9, 168.8, 121.1, 82.9, 67.1, 51.6, 51.4, 45.1, 43.2, 38.4, 37.4, 34.1, 29.9, 29.7, 29.2, 27.4, 19.6; IR (thin film) 2927, 2868, 1667, 1622 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>17</sub>H<sub>24</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 283.1674 found 283.1682.

### Li/NH<sub>3</sub> Reduction to **5.73** and **5.74**

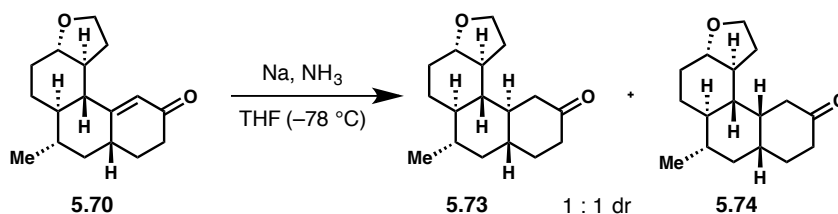


A 10 mL round bottom flask was charged with a glass stir bar and filled with 3 mL liquid ammonia. To the stirring solution was added 3 mg (0.43 mmol) lithium metal and 0.2 mL THF. At -40 °C 14 mg (0.054 mmol) **5.70** was added with the assistance of 0.5 mL THF. After 2 minutes 0.1 mL isoprene was added, followed by 75 mg solid NH<sub>4</sub>Cl. The flask was warmed to



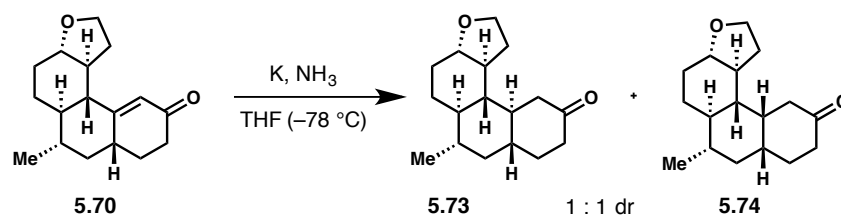
room temperature. After evaporation of the ammonia, 1 mL water and 2 mL EtOAc were added. The layers separated and the organic phase washed with 1 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude solid was purified by column chromatography (2:1 hexanes/EtOAc) to afford 9.0 mg (64%, 1:2 dr) **5.73/5.74** as a colorless oil.

**Table 5.1, Entry 1: Na/NH<sub>3</sub> Reduction to 5.73 and 5.74**



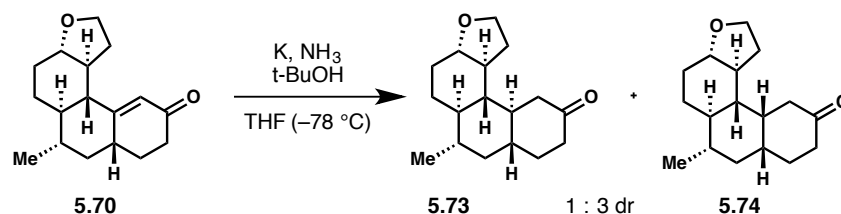
A 10 mL round bottom flask was charged with a glass stir bar and filled with 3 mL liquid ammonia. To the stirring solution was added 10 mg (0.038 mmol) **5.70** with 0.8 mL THF followed by 16 mg (0.70 mmol) sodium metal at -78 °C. After 3 minutes 0.15 mL isoprene was added, followed by 62 mg solid NH<sub>4</sub>Cl. The flask was warmed to room temperature. After evaporation of the ammonia, 1 mL water and 2 mL EtOAc were added. The layers separated and the organic phase washed with 1 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude solid was purified by column chromatography (2:1 hexanes/EtOAc) to afford 8.5 mg (85%, 1:1 dr) **5.73/5.74** as a colorless oil.

**Table 5.1, Entry 2: K/NH<sub>3</sub> Reduction to 5.73 and 5.74**



A 10 mL round bottom flask was charged with a glass stir bar and filled with 3 mL liquid ammonia. To the stirring solution was added 10.4 mg (0.040 mmol) **5.70** with 0.8 mL THF followed by 30 mg (0.70 mmol) potassium metal at  $-78\text{ }^\circ\text{C}$ . After 30 minutes 0.10 mL isoprene was added, followed by 71 mg solid  $\text{NH}_4\text{Cl}$ . The flask was warmed to room temperature. After evaporation of the ammonia, 2 mL water and 5 mL EtOAc were added. The layers separated and the organic phase washed with 2 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude solid was purified by column chromatography (3:1 hexanes/EtOAc) to afford 8.6 mg (82%, 1:1 dr) **5.73/5.74** as a colorless oil.

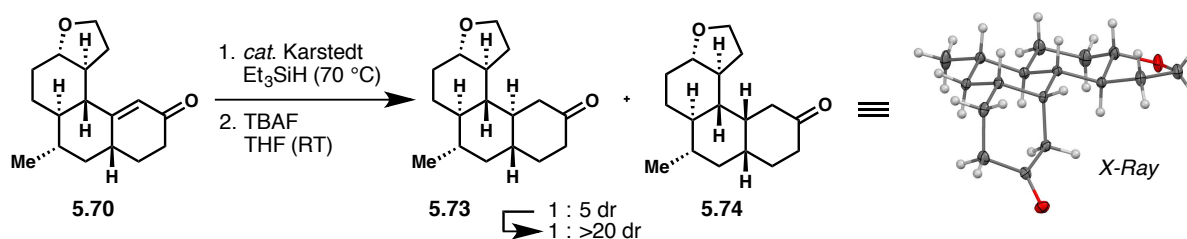
**Table 5.1, Entry 3: K/NH<sub>3</sub>/t-BuOH Reduction 5.73 and 5.74**



A 10 mL round bottom flask was charged with a glass stir bar and filled with 3 mL liquid ammonia. To the stirring solution was added 10 mg (0.038 mmol) **5.70** with 0.8 mL THF and 0.18 mL (1.9 mmol)  $t\text{-BuOH}$  followed by 30 mg (0.77 mmol) potassium metal at  $-78\text{ }^\circ\text{C}$ . After 30 minutes, 0.10 mL isoprene was added, followed by 71 mg solid  $\text{NH}_4\text{Cl}$ . The flask was warmed to room temperature. After evaporation of the ammonia, 2 mL water and 3 mL EtOAc were added. The layers separated and the organic phase washed with 2 mL brine, dried over

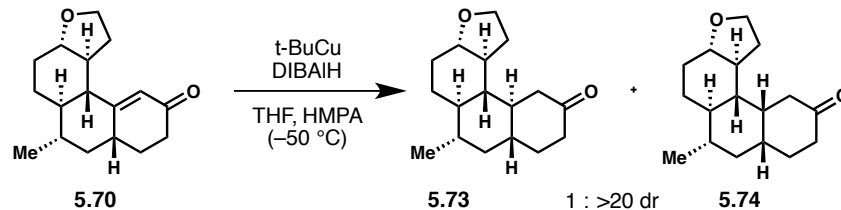
MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude solid was purified by column chromatography (3:1 hexanes/EtOAc) to afford 8.0 mg (80%, 1:3 dr) **5.73/5.74** as a white wax.

**Table 5.1, Entry 5: Karstedt/Et<sub>3</sub>SiH Reduction to 5.73 and 5.74**



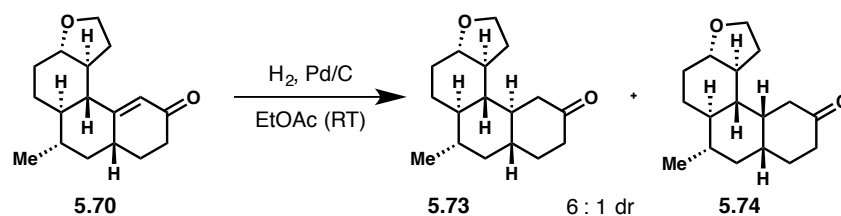
A 1 dram vial was charged with 15 mg (0.058 mmol) **5.70**, 3 mg (0.00016 mmol) Karstedt's catalyst 2 wt% in xylene and 0.4 mL triethylsilane, capped and heated to 70 °C. After 12 hours the reaction was cooled and all volatiles removed in vacuo. The crude material was dissolved in 0.4 mL THF and treated with 0.15 mL (0.15 mmol) TBAF 1 M/THF open to air. After 30 minutes 1 mL sat. aq. NH<sub>4</sub>Cl was added and the mixture extracted twice with 2 mL EtOAc. The organic layers were combined, washed with 1 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude solid was purified by column chromatography (2:1 hexanes/EtOAc) to afford 14 mg (92% over 2 steps, 1:5 dr) **5.73/5.74** as a white solid. Crystallization from pentane/EtOAc afforded **5.74** as a single diastereomer in colorless prisms (mp = 137–139 °C).

**Table 5.1, Entry 6: t-BuCu/DIBAL Reduction to 5.73 and 5.74**



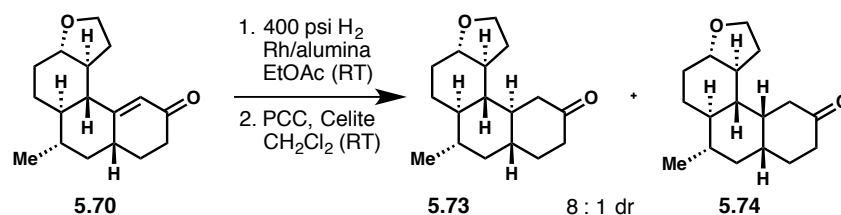
A 1 dram vial containing 7.9 mg (0.041 mmol)  $\text{CuI}$  in 0.4 mL THF was cooled to  $-78\text{ }^\circ\text{C}$  and treated with 0.03 mL (0.042 mmol) 1.4 M  $t\text{-BuLi}$ /pentane. After stirring at  $-50\text{ }^\circ\text{C}$  for 15 minutes the flask was recooled to  $-78\text{ }^\circ\text{C}$  and treated with 0.10 mL (0.57 mmol) HMPA then 0.10 mL (0.1 mmol) 1.0 M DIBALH/hexanes. After 20 minutes at  $-50\text{ }^\circ\text{C}$  the flask was recooled to  $-78\text{ }^\circ\text{C}$  and 11.3 mg (0.043 mmol) **5.70** in 0.4 mL THF was added slowly. The reaction was stirred at  $-50\text{ }^\circ\text{C}$  for 8 hours, after which it was removed from the cold bath and quenched with 2 mL 2 M HCl. The flask was warmed to room temperature and partitioned with 6 mL EtOAc. The organic layer was washed with 2 mL 3:1 sat. aq.  $\text{NH}_4\text{Cl}$ /5 M NaOH twice then 2 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (3:1 hexanes/EtOAc) to afford 9.8 mg (86%, 1:>20 dr) **5.73/5.74** as a white solid. Crystallization from pentane/EtOAc afforded colorless prisms (mp =  $137\text{--}139\text{ }^\circ\text{C}$ ). **5.74**:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  3.95-3.88 (m, 2H), 3.11-3.06 (m, 1H), 2.47-2.38 (m, 2H), 2.25-2.10 (m, 5H), 2.00 (dtd,  $J = 11.0, 7.4, 3.3\text{ Hz}$ , 1H), 1.92-1.88 (m, 3H), 1.54 (dd,  $J = 24.5, 11.9\text{ Hz}$ , 1H), 1.49-1.43 (m, 2H), 1.35-1.13 (m, 4H), 1.00-0.93 (m, 2H) 1.00 (d,  $J = 6.5\text{ Hz}$ , 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  212.7, 83.2, 67.3, 47.3, 46.5, 41.1, 40.4, 38.3, 37.5, 37.2, 35.2, 34.3, 31.3, 30.5, 28.5, 28.3, 20.4; IR (thin film) 2924, 2866, 1711, 1453, 1058  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{17}\text{H}_{26}\text{O}_2$   $[\text{M}+\text{Na}]^+$  285.1830 found 285.1843.

**Table 5.1, Entry 7: Pd/C Reduction to 5.73 and 5.74**



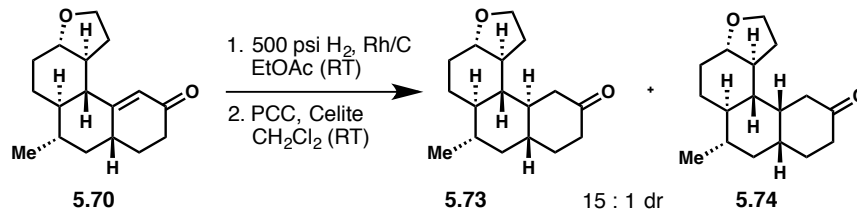
A 1 dram vial containing 5.3 mg (0.020 mmol) **5.70** and catalytic Pd/C in 0.4 mL EtOAc was stirred under a hydrogen balloon for 20 hours. The reaction was filtered over Celite, eluting with EtOAc and all volatiles removed in vacuo. The crude solid was purified by column chromatography (3:1 hexanes/EtOAc) to afford 5.0 mg (94%, 6:1 dr) **5.73/5.74** as a white semi-solid.

**Table 5.1, Entry 8: Rh/alumina Reduction to 5.73 and 5.74**



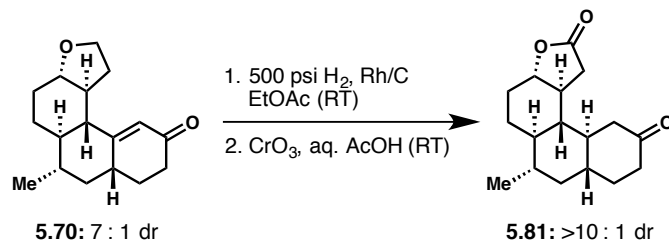
A 1 dram vial containing 5.9 mg (0.023 mmol) **5.70** and catalytic Rh/alumina in 0.4 mL EtOAc was stirred under 400 psi H<sub>2</sub> in a bomb reactor. After 20 hours, the reaction was filtered over Celite, eluting with EtOAc and all volatiles removed in vacuo to afford an alcohol as a mixture of diastereomers. The crude material was dissolved in 0.5 mL CH<sub>2</sub>Cl<sub>2</sub> and 22 mg Celite added followed by 21 mg (0.097 mmol) PCC open to air. After 5 hours the reaction was diluted with Et<sub>2</sub>O and filtered over silica gel, eluting with Et<sub>2</sub>O. All volatiles were removed in vacuo and the crude material was purified by column chromatography (3:1 hexanes/EtOAc) to afford 5.5 mg (93% over 2 steps, 8:1 dr) **5.73/5.74** as a white semi-solid.

**Table 5.1, Entry 9: Rh/C Reduction to 5.73 and 5.74**



A 25 mL round bottom containing 563 mg (1.26 mmol, 7:1 dr) **5.70** and 46 mg (0.022 mmol) 5% Rh/C in 10 mL EtOAc was stirred under 500 psi H<sub>2</sub> in a bomb reactor. After 24 hours, the reaction was filtered over Celite, eluting with 30 mL EtOAc and all volatiles removed in vacuo to afford a mixture of ketone and alcohol. The crude material in a 50 mL round bottom flask was dissolved in 10 mL CH<sub>2</sub>Cl<sub>2</sub> and 550 mg Celite added followed by 470 mg (2.18 mmol) PCC open to air. After 5 hours the reaction was diluted with 20 mL Et<sub>2</sub>O and filtered over silica gel, eluting with 120 mL Et<sub>2</sub>O. All volatiles were removed in vacuo and the crude material was purified by column chromatography (3:1 hexanes/EtOAc) to afford 525 mg (93% over 2 steps, 15:1 dr) **5.73/5.74** as a white semi-solid. **5.73**: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 3.88-3.85 (m, 2H), 3.07 (ddd, *J* = 11.2, 10.0, 3.7 Hz, 1H), 2.82 (ddd, *J* = 14.1, 3.9, 1.8 Hz, 1H), 2.41-2.31 (m, 2H), 2.25 (dq, *J* = 11.5, 5.8 Hz, 1H), 2.17-2.09 (m, 2H), 2.02 (t, *J* = 13.3 Hz, 1H), 1.96 (ddt, *J* = 13.2, 6.0, 3.0 Hz, 1H), 1.73 (dt, *J* = 13.1, 3.5 Hz, 1H), 1.70-1.62 (m, 1H), 1.54-1.46 (m, 1H), 1.43-1.19 (m, 5H), 1.03-0.72 (m, 4H), 0.95 (d, *J* = 6.5, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 212.0 83.5, 66.9, 50.7, 49.3, 48.9, 47.9, 47.4, 42.4, 41.2, 41.1, 36.5, 33.3, 32.6, 30.5, 28.2, 20.8; IR (thin film) 2925, 2865, 1711, 1455, 1062 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>17</sub>H<sub>26</sub>O<sub>2</sub> [M+NH<sub>4</sub>]<sup>+</sup> 280.2277 found 280.2279.

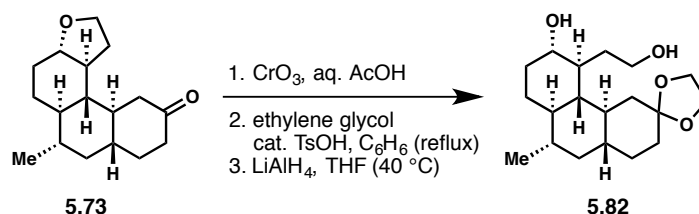
## Lactone 5.81 using Fieser's Reagent



A 25 mL round bottom flask containing dram vial containing 292 mg (1.12 mmol, 7:1 dr) **5.70** and 35 mg (0.017) 5% Rh/C in 5.5 mL EtOAc was stirred under 500 psi H<sub>2</sub> in a bomb reactor. After 18 hours, the reaction was filtered over Celite, eluting with EtOAc and all volatiles removed in vacuo to afford a mixture of ketone and alcohol. Only 247 mg (~81%) crude material in a 20 mL scintillation vial was dissolved in 4 mL AcOH and treated with 410 mg (4.10 mmol) CrO<sub>3</sub> in 6 mL 5:1 AcOH/water open to air. After 20 hours, 0.20 mL i-PrOH was added and stirring continued for 20 minutes. After being poured into 70 mL ice cold stirring water, the solution was treated with 30 mL 5 M NaOH at 0 °C and extracted with 30 mL, 20 mL and 10 mL Et<sub>2</sub>O. The organic layers were combined, washed with 10 mL water, 10 mL sat. aq. NaHCO<sub>3</sub> and 10 mL brine. The aqueous washings were combined and back extracted with 20 mL Et<sub>2</sub>O. All organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (1:2 hexanes/Et<sub>2</sub>O) to afford 69 mg (27% over 2 steps, >10:1 dr) **5.81** as a white semi-solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 3.84 (td, *J* = 11.1, 3.5 Hz, 1H), 2.76 (dd, *J* = 16.1, 6.4 Hz, 1H), 2.57 (ddd, *J* = 13.9, 3.9, 1.8 Hz, 1H), 2.40-2.20 (m, 5H), 2.03-1.95 (m, 2H), 1.79-1.74 (m, 2H), 1.58-1.49 (m, 3H), 1.42-1.06 (m, 2H), 1.19-1.04 (m, 2H), 0.96-0.83 (m, 2H), 0.96 (d, *J* = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 210.6, 175.9, 84.9, 50.2, 48.4, 48.2, 47.1, 46.8, 42.0, 41.0, 40.9, 37.4,

36.5, 32.9, 29.6, 27.7, 20.3; IR (thin film) 2920, 2867, 1781, 1710, 1210, 1041, 966  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{17}\text{H}_{24}\text{O}_3$   $[\text{M}+\text{Na}]^+$  299.1623 found 299.1624.

### Diol 5.82

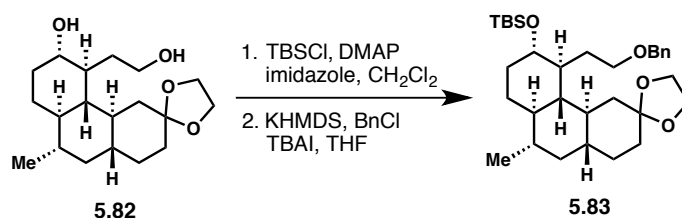


To a 1 dram vial containing 30 mg (0.111 mmol) **5.73** in 0.3 mL AcOH was added 51 mg (0.51 mmol)  $\text{CrO}_3$  in 0.8 mL 9:1 AcOH/water open to air. After 23 hours, 0.10 mL i-PrOH was added and stirring continued for 20 minutes. After being poured into 7 mL ice cold stirring water and 5 mL  $\text{Et}_2\text{O}$ , the solution was treated with 3 mL 5 M NaOH at  $0^\circ\text{C}$  and extracted with 5 mL  $\text{Et}_2\text{O}$ . The organic layers were combined, washed twice with 2 mL water and 2 mL sat. aq.  $\text{NaHCO}_3$ . The aqueous washings were combined and back extracted with 3 mL  $\text{Et}_2\text{O}$ . All organic layers were combined, washed with 2 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. A 10 mL round bottom flask containing crude C–H oxidation product in 2.5 mL  $\text{C}_6\text{H}_6$  was refluxed with 0.03 mL (0.54 mmol) ethylene glycol and 2 mg (0.011 mmol)  $\text{TsOH}\cdot\text{H}_2\text{O}$  under a Hickman Still. After 2 hours the reaction was cooled, diluted with 5 mL EtOAc and washed with 2 mL sat. aq.  $\text{NaHCO}_3$  and 2 mL brine. The organic layer was dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude lactone was added with the assistance of 0.1 mL THF to 16 mg (0.42 mmol)  $\text{LiAlH}_4$  in 1 mL THF at  $0^\circ\text{C}$ . The flask was immediately removed from the cold bath and stirred for 10 minutes before being heated to  $40^\circ\text{C}$  for 1 hour. After cooling the flask to  $0^\circ\text{C}$ , 0.05 mL EtOAc, 0.02 mL water, 0.02 mL 5 M NaOH, 0.06 mL water and  $\text{Na}_2\text{SO}_4$  were carefully added sequentially. The solution was filtered over Celite and



washed with Et<sub>2</sub>O (~40 mL as determined by TLC) and all volatiles removed in vacuo. The crude material was purified by column chromatography (EtOAc) to afford 15 mg (42% over 3 steps, >10:1 dr) **5.82** as a white semi-solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 3.94 (s, 4H), 3.81-3.62 (m, 2H), 3.47-3.31 (m, 2H), 2.05-1.46 (m, 10H), 1.35-1.10 (m, 7H), 0.92-0.81 (m, 7H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 109.4, 76.2, 64.3, 64.1, 62.3, 50.1, 47.9, 47.7, 45.6, 43.3, 41.6, 40.7, 39.4, 37.6, 34.2, 32.1, 31.4, 26.0, 20.2.

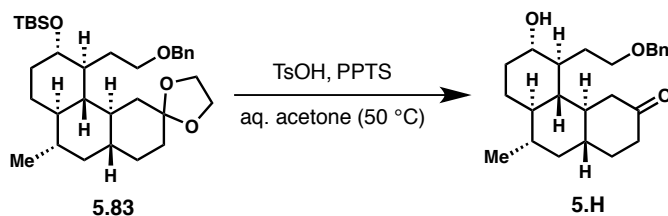
### Benzyl Ether **5.83**



To a 1 dram vial containing 11 mg (0.034 mmol) **5.82** was added 5.5 (0.081 mmol) imidazole, 2.5 mg (0.020 mmol) DMAP and 0.5 mL CH<sub>2</sub>Cl<sub>2</sub>. At 0 °C 7.0 mg (0.046 mmol) TBSCl was added and the cooling bath removed. After 10 hours sat. aq. NaHCO<sub>3</sub> was added and the mixture extracted with EtOAc. The organic layers were combined, washed with sat. aq. brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material and 2 mg (0.0054 mmol) TBAI in a 1 dram vial were dissolved in 0.4 mL THF and at 0 °C treated with 0.2 mL (0.10 mmol) KHMDS 0.5M/toluene and 0.02 mL (0.17 mmol) benzyl chloride. The cold bath was immediately removed. After 1 hour the reaction was recooled to 0 °C and quenched with half sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layers were combined, washed with sat. aq. brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed. The crude material was purified by column chromatography (20:1 hexanes/EtOAc) to afford 8.5 mg (47%) **5.83** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.34-7.26 (m, 5H), 4.53-4.41 (m, 2H), 3.93-3.87 (m,

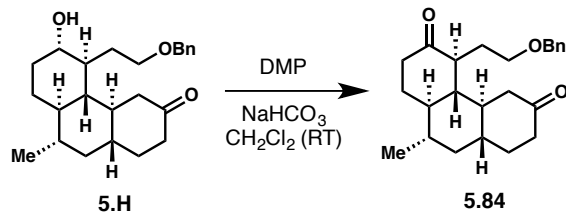
2H), 3.81-3.76 (m, 1H), 3.75-3.64 (m, 2H), 3.55-3.48 (m, 2H), 2.11 (d,  $J = 12.0$  Hz, 1H), 2.04-2.02 (m, 1H), 1.92-1.86 (m, 1H), 1.74-1.35 (m, 10H), 1.32-0.95 (m, 5H), 0.95-0.77 (m, 13H), 0.62-0.50 (m, 1H), 0.04 (s, 3H), 0.03 (s, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  138.6, 128.3, 127.81, 127.77, 127.44, 124.41, 109.5, 73.2, 71.8, 69.3, 64.0, 63.9, 48.6, 47.3, 42.9, 42.0, 41.8, 41.7, 38.6, 38.4, 35.0, 34.8, 31.3, 26.4, 25.9, 24.2, 20.0, 17.9, -4.6, -4.7.

### Benzyl Ether **5.H**



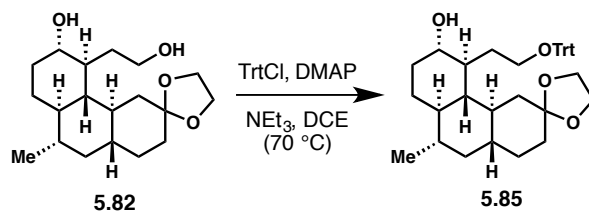
A 1 dram vial containing 8.5 mg (0.016 mmol) **5.82**, 4 mg (0.016 mmol) PPTS and 3 mg (0.016 mmol)  $\text{TsOH}\cdot\text{H}_2\text{O}$  in 0.2 mL 3:1 acetone/water was sealed and heated to 70 °C for 6 hours. The reaction was cooled, diluted with sat. aq.  $\text{NaHCO}_3$  and extracted with EtOAc. The organic layers were combined, washed with sat. aq. brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was purified by column chromatograph (3:1 hexanes/EtOAc) to afford 3.7 mg (62%) **5.H** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.37-7.29 (m, 4H), 4.57 (d,  $J = 11.8$  Hz, 1H), 4.51 (d,  $J = 11.8$  Hz, 1H), 3.63-3.55 (m, 2H), 3.49-3.44 (m, 2H), 2.54 (dd,  $J = 12.8, 3.1$  Hz, 1H), 2.35 (m, 2H), 2.13 (t,  $J = 13.0$  Hz, 1H), 2.00-1.89 (m, 3H), 1.81-1.51 (m, 6H), 1.42-1.15 (m, 6H), 0.91 (d,  $J = 6.5$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  211.7, 137.5, 128.5, 127.8, 74.6, 73.4, 69.7, 50.43, 50.39, 47.7, 46.7, 45.7, 42.6, 41.7, 41.0, 37.4, 35.9, 33.7, 30.5, 25.5, 20.1.

### Ketone 5.84



A 1 dram vial containing **5.H**, DMP and NaHCO<sub>3</sub> in 0.2 mL CH<sub>2</sub>Cl<sub>2</sub> was stirred for 30 minutes. The reaction was cooled, diluted with sat. aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layers were combined, washed with sat. aq. NaHCO<sub>3</sub>, sat. aq. brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford a crude material assigned as **5.84**.

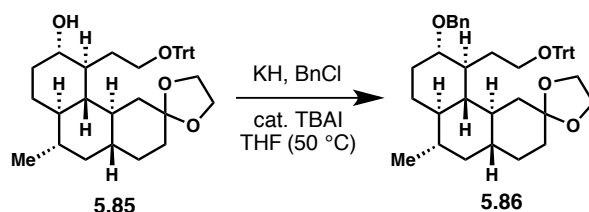
### Trityl Alcohol 5.85



A 5 mL round bottom flask was charged with 15 mg (0.046 mmol) **5.82**, 6.8 mg (0.056 mmol) DMAP, 0.03 mL (0.22 mmol) NEt<sub>3</sub> and 18 mg (0.064 mmol) triphenylmethyl chloride in 1 mL DCE. After stirring the reaction at 85 °C for 3 hours the solution was cooled, diluted with sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layers were combined, washed with sat. aq. brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (5:1 hexanes/EtOAc) to afford 15 mg (58%) **5.85** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.44 (d, *J* = 7.5 Hz, 6H), 7.31 (t, *J* = 7.6 Hz, 6H), 7.23 (t, *J* = 7.3 Hz, 3H), 3.85-3.79 (m, 2H), 3.72-3.67 (m, 1H), 3.63-3.58 (m, 1H), 3.47-3.43 (m, 1H), 3.30-3.24 (m, 1H), 3.13 (dt, *J* = 9.1, 6.1 Hz, 1H), 1.92-1.80 (m, 4H), 1.69-1.46 (m,

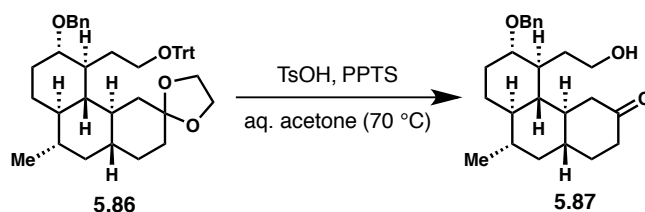
7H), 1.39-0.98 (m, 9H), 0.87 (d,  $J = 6.4$  Hz, 3H), 0.91-0.78 (m, 3H), 0.70 (t,  $J = 8.9$  Hz, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  171.1, 143.9, 128.7, 128.6, 127.8, 127.0, 109.4, 87.4, 77.3, 77.0, 76.8, 74.0, 64.2, 64.0, 63.3, 60.4, 49.5, 47.5, 45.7, 44.4, 43.0, 41.6, 40.0, 37.9, 36.0, 34.5, 31.4, 29.3, 25.1, 21.1, 20.1, 14.2.

### Benzyl Ether **5.86**



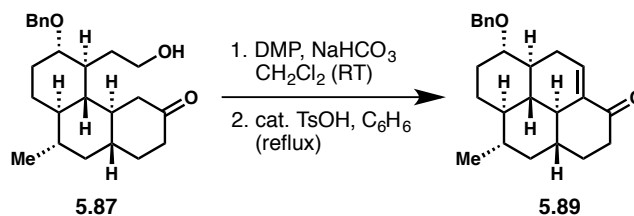
A 1 dram vial was charged with 5 mg (~0.12 mmol) oil free KH and layered with 0.2 mL THF. At 0 °C a solution of 1 mg (0.0026 mmol) TBAI and 15 mg (0.026 mmol) **5.82** was added with a total of 0.5 mL THF. After the addition of 1 drop benzyl chloride the reaction was heated to 50 °C for 2 hours. The reaction was cooled, then treated with water and extracted with EtOAc. The organic layers were combined, washed with sat. aq. brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (15:1 hexanes/EtOAc) to afford 11 mg (64%) **5.86** as a colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.47 (d,  $J = 7.4$  Hz, 5H), 7.34-7.22 (m, 15H), 4.44 (d,  $J = 12.3$  Hz, 1H), 4.40 (d,  $J = 12.3$  Hz, 1H), 3.87-3.81 (m, 2H), 3.66-3.64 (m, 1H), 3.29 (bs, 1H), 3.10 (t,  $J = 6.0$  Hz, 2H), 2.19 (d,  $J = 10.8$  Hz, 1H), 1.98-1.84 (m, 3H), 1.78-1.55 (m, 7H), 1.46-1.37 (m, 2H), 1.36-1.23 (m, 6H), 1.13-1.03 (m, 2H), 0.99-0.86 (m, 3H), 0.91 (d,  $J = 6.3$  Hz, 3H), 0.70-0.65 (m, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  144.4, 139.4, 128.7, 128.2, 127.71, 127.68, 127.65, 127.2, 127.0, 126.8, 109.5, 78.0, 69.6, 64.3, 64.0, 62.2, 60.4, 48.4, 47.3, 42.7, 42.0, 41.5, 39.5, 38.7, 38.1, 35.1, 34.8, 31.2, 24.1, 22.7, 21.1, 19.9, 14.2.

## Ketoalcohol 5.87



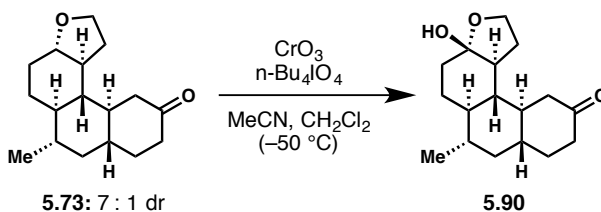
A 1 dram vial containing 11 mg (0.016 mmol) **5.86**, 4.5 mg (0.018 mmol) PPTS and 6.8 mg (0.036 mmol) TsOH•H<sub>2</sub>O dissolved in 0.6 mL 5:1 acetone/water was sealed and heated to 70 °C. After 1 hour the reaction was cooled, diluted with sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layers were combined, washed with brine, dried over MgSO<sub>4</sub> and filtered. The brine layer was back extracted with EtOAc, dried over MgSO<sub>4</sub>, filtered, combined with the previously dried organic layer and all volatiles removed in vacuo. The crude material was purified by column chromatography (2:1→1:1 hexanes/EtOAc) to afford 5.7 mg (91%) **5.87** as a white semi-solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.36-7.26 (m, 5H), 4.57 (d, *J* = 12.1 Hz, 1H), 4.46 (d, *J* = 12.1 Hz, 1H), 3.66-3.57 (m, 2H), 3.33 (q, *J* = 4.4 Hz, 1H), 2.66 (dd, *J* = 13.3, 3.1 Hz, 1H), 2.35 (dd, *J* = 10.4, 4.9 Hz, 2H), 2.08 (t, *J* = 13.6 Hz, 1H), 1.96 (ddq, *J* = 13.3, 8.6, 4.5 Hz, 2H), 1.86-1.79 (m, 1H), 1.73-1.56 (m, 4H), 1.53-1.40 (m, 2H), 1.39-1.23 (m, 4H), 1.19-1.11 (m, 1H), 1.00-0.83 (m, 3H), 0.91 (d, *J* = 6.4 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 211.9, 138.7, 128.3, 127.6, 127.5, 79.2, 69.9, 61.4, 50.4, 50.0, 46.5, 43.1, 42.2, 41.4, 41.1, 39.2, 38.3, 38.0, 33.5, 24.2, 23.8, 19.9; IR (thin film) 3418, 2921, 1711, 1065, 734 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>24</sub>H<sub>34</sub>O<sub>3</sub> [M+Na]<sup>+</sup> 393.2406 found 393.2389.

## Enone 5.89



A 1 dram vial containing **5.87**, NaHCO<sub>3</sub> and DMP in 0.2 mL CH<sub>2</sub>Cl<sub>2</sub> was stirred at room temperature. After 10 minutes the reaction was treated with Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, water and extracted with EtOAc. The organic layer was separated, washed with sat. aq. NaHCO<sub>3</sub>, sat. aq. brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. A 1 dram vial containing crude **5.88** in 0.5 mL C<sub>6</sub>H<sub>6</sub> was refluxed with TsOH•H<sub>2</sub>O and 4Å MS. After 20 minutes the reaction was cooled, treated with sat. aq. NaHCO<sub>3</sub>, diluted with EtOAc and the layers separated. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford crude **5.89**.

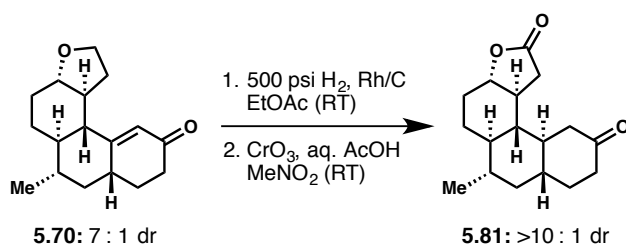
## Lactol 5.90



A 1 dram vial containing 6.9 mg (0.069 mmol) CrO<sub>3</sub> dissolved in 0.4 mL MeCN was cooled to -40 °C and 4.4 mg (0.017 mmol) **5.73** added with 0.2 mL CH<sub>2</sub>Cl<sub>2</sub>. The reaction was cooled to -50 to -55 °C under vigorous stirring and treated with 30 mg (0.069 mmol) n-Bu<sub>4</sub>NIO<sub>4</sub> in 0.2 mL MeCN. After 15 minutes 1 mL 1:1:2 sat. aq. NaHCO<sub>3</sub>/sat. aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>/water was added, the mixture warmed to room temperature and extracted twice with 2 mL EtOAc. The organic layers were combined, washed with 1:1:2 sat. aq. NaHCO<sub>3</sub>/sat. aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>/water, 1 mL brine, dried

over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (1:1→1:4 hexanes/EtOAc) to afford 2.7 mg (58%) **5.90** as a colorless oil and 1.0 mg (22%) recovered starting material as a thin film. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 4.02 (td, *J* = 8.8, 3.4 Hz, 1H), 3.90 (q, *J* = 8.1 Hz, 1H), 2.63-2.59 (m, 1H), 2.55-2.47 (m, 1H), 2.38-2.36 (m, 1H), 2.15-2.10 (m, 1H), 2.08-1.49 (m, 8H), 1.41-1.15 (m, 4H), 1.11-0.98 (m, 2H), 0.94 (d, *J* = 6.5 Hz, 3H), 0.96-0.73 (m, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 211.6, 105.6, 65.8, 50.6, 49.1, 47.6, 47.4, 46.8, 42.5, 41.0, 40.9, 36.7, 35.3, 33.6, 32.7, 25.1, 20.1; IR (thin film) 3405, 2923, 2863, 1713, 1455, 1116, 1052, 1029 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>17</sub>H<sub>26</sub>O<sub>3</sub> [M+Na]<sup>+</sup> 301.1780 found 301.1773.

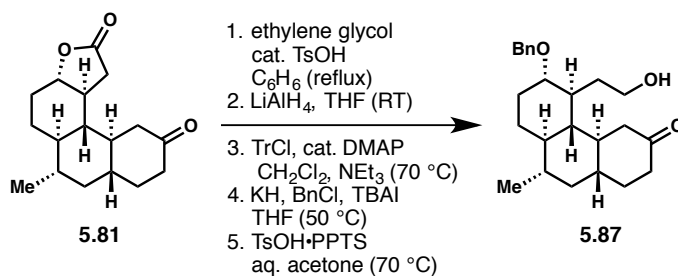
### Lactone **5.81** using Fieser's Reagent in MeNO<sub>2</sub>



A 10 mL round bottom flask containing 108 mg (0.414 mmol, 7:1 dr) **5.70** and 9 mg (0.0044 mmol) 5% Rh/C in 1.8 mL EtOAc was stirred under 500 psi H<sub>2</sub> in a bomb reactor. After 23 hours, the reaction was filtered over Celite, eluting with EtOAc and all volatiles removed in vacuo to afford a mixture of ketone and alcohol. In a 25 mL round bottom flask, crude THF tetracycle was dissolved in 4 mL MeNO<sub>2</sub>, cooled to 0 °C and treated with 166 mg (1.66 mmol) CrO<sub>3</sub> dissolved in 0.8 mL 10:1 AcOH/water open to air. After 20 hours at room temperature, the reaction was recooled to 0°C then 5 mL half sat. aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and 5 mL water added. The solution was extracted with 15 mL and 5 mL EtOAc. The organic layers were combined, washed

with 3 mL sat. aq. NaHCO<sub>3</sub> and 3 mL brine. The aqueous washings were combined and backextracted with 3 mL EtOAc. All organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (1:2→1:4 hexanes/Et<sub>2</sub>O) to afford 43 mg (38% over 2 steps, >10:1 dr) **5.81** as a white semi-solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 3.84 (td, *J* = 11.1, 3.5 Hz, 1H), 2.76 (dd, *J* = 16.1, 6.4 Hz, 1H), 2.57 (ddd, *J* = 13.9, 3.9, 1.8 Hz, 1H), 2.40-2.20 (m, 5H), 2.03-1.95 (m, 2H), 1.79-1.74 (m, 2H), 1.58-1.49 (m, 3H), 1.42-1.06 (m, 2H), 1.19-1.04 (m, 2H), 0.96-0.83 (m, 2H), 0.96 (d, *J* = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 210.6, 175.9, 84.9, 50.2, 48.4, 48.2, 47.1, 46.8, 42.0, 41.0, 40.9, 37.4, 36.5, 32.9, 29.6, 27.7, 20.3; IR (thin film) 2920, 2867, 1781, 1710, 1210, 1041, 966 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>17</sub>H<sub>24</sub>O<sub>3</sub> [M+Na]<sup>+</sup> 299.1623 found 299.1624.

### Hydroxyketone 17



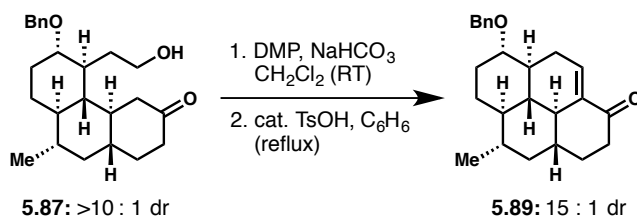
A 10 mL round bottom flask containing 67 mg (0.242 mmol >10:1 dr) **5.81** in 3.5 mL benzene was refluxed with 0.07 mL (1.25 mmol) ethylene glycol and 3 mg (0.016 mmol) TsOH·H<sub>2</sub>O over a Hickman Still. After 2 hours the reaction was cooled, diluted with 10 mL EtOAc and washed with 2 mL sat. aq. NaHCO<sub>3</sub> and 2 mL brine. The organic layer was dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude lactone was added with the assistance of 0.9 mL THF to 19 mg (0.501 mmol) LiAlH<sub>4</sub> in 1 mL THF at 0 °C. The flask was immediately



removed from the cold bath and stirred for 14 hours. After cooling the flask to 0 °C, 0.02 mL water, 0.02 mL 5 M NaOH, 0.06 mL water and Na<sub>2</sub>SO<sub>4</sub> were carefully added sequentially. The solution was filtered over Celite and washed with Et<sub>2</sub>O (~40 mL as determined by TLC) and all volatiles removed in vacuo. To a 10 mL round bottom flask containing crude diol in 2.2 mL DCE and 0.2 mL (1.43 mmol) NEt<sub>3</sub> was added 3 mg (0.024 mmol) DMAP and 68 mg (0.244 mmol) triphenylmethyl chloride. After heating the mixture at 70 °C for 9 hours, the reaction was cooled, treated with 2 mL sat. aq. NaHCO<sub>3</sub> and extracted with 10 mL EtOAc. The organic layer was separated, washed with 2 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. A 10 mL round bottom flask containing 36 mg (0.90 mmol) KH freed from oil with three pentane washings was added 0.5 mL THF and 10 mg (0.027 mmol) TBAI at 0 °C. Crude alcohol was added with the assistance of 2 mL THF followed by 0.06 mL (0.52 mmol) benzyl chloride. The cold bath was removed and the reaction stirred for 30 minutes, after which the reaction was heated to 50 °C. After 2 hours the flask was cooled in an ice bath, treated carefully with 2 mL water and extracted with 10 mL EtOAc. The organic layer was separated, washed with 2 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material in a 10 mL round bottom flask was dissolved in 4 mL 5:1 acetone/water then treated with 21 mg (0.084 mmol) PPTS and 35 mg (0.18 mmol) TsOH•H<sub>2</sub>O. A reflux condenser was fitted and the reaction immersed in a 70 °C oil bath open to air. After 2 hours the reaction was cooled, diluted with 2 mL sat. aq. NaHCO<sub>3</sub> and extracted with 10 mL and 5 mL EtOAc. The organic layers were combined, washed with 2 mL brine, dried over MgSO<sub>4</sub> and filtered. The brine layer was back extracted with 3 mL EtOAc, dried over MgSO<sub>4</sub>, filtered, combined with the previously dried organic layer and all volatiles removed in vacuo. The crude material was purified by column chromatography (2:1 → 1:1 hexanes/EtOAc) to afford 68 mg (75% over 5 steps, >10:1 dr) **5.87** as

a white semi-solid.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.36-7.26 (m, 5H), 4.57 (d,  $J = 12.1$  Hz, 1H), 4.46 (d,  $J = 12.1$  Hz, 1H), 3.66-3.57 (m, 2H), 3.33 (q,  $J = 4.4$  Hz, 1H), 2.66 (dd,  $J = 13.3, 3.1$  Hz, 1H), 2.35 (dd,  $J = 10.4, 4.9$  Hz, 2H), 2.08 (t,  $J = 13.6$  Hz, 1H), 1.96 (ddq,  $J = 13.3, 8.6, 4.5$  Hz, 2H), 1.86-1.79 (m, 1H), 1.73-1.56 (m, 4H), 1.53-1.40 (m, 2H), 1.39-1.23 (m, 4H), 1.19-1.11 (m, 1H), 1.00-0.83 (m, 3H), 0.91 (d,  $J = 6.4$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  211.9, 138.7, 128.3, 127.6, 127.5, 79.2, 69.9, 61.4, 50.4, 50.0, 46.5, 43.1, 42.2, 41.4, 41.1, 39.2, 38.3, 38.0, 33.5, 24.2, 23.8, 19.9; IR (thin film) 3418, 2921, 1711, 1065, 734  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{24}\text{H}_{34}\text{O}_3$   $[\text{M}+\text{Na}]^+$  393.2406 found 393.2389.

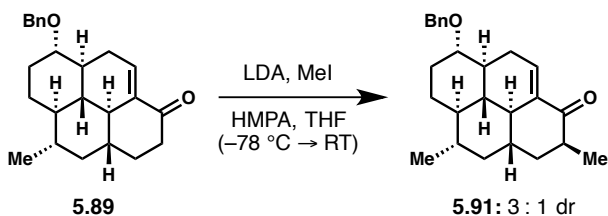
### Tetracyclic Enone 5.89



A 10 mL round bottom flask containing 65 mg (0.175 mmol, >10:1 dr) **17** and 264 mg (3.14 mmol)  $\text{NaHCO}_3$  in 2 mL  $\text{CH}_2\text{Cl}_2$  was added 149 mg (0.351 mmol) DMP at room temperature open to air. After 20 minutes the reaction was treated carefully with 1 mL  $\text{Na}_2\text{S}_2\text{O}_3$  and stirred for 20 minutes before being diluted with an additional 1 mL  $\text{Na}_2\text{S}_2\text{O}_3$  and 2 mL water and extracted with 10 mL EtOAc. The organic layer was separated, washed twice with 2 mL sat. aq.  $\text{NaHCO}_3$ , 2 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. A 10 mL round bottom flask containing crude keto-aldehyde in 3 mL benzene was refluxed with 4 mg (0.021 mmol)  $\text{TsOH}\cdot\text{H}_2\text{O}$  over a Hickman Still. After 20 minutes the reaction was cooled, treated with 2 mL sat. aq.  $\text{NaHCO}_3$ , diluted with 10 mL EtOAc and the layers separated. The organic layer was washed with 2 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed

in vacuo. The crude material was purified by column chromatography (6:1 hexanes/EtOAc) to afford 44 mg (72%, 15:1 dr) **5.89** as a light yellow semi-solid.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$  at 7.27 ppm)  $\delta$  7.36-7.28 (m, 5H), 6.79 (t,  $J = 2.3$  Hz, 1H), 4.69 (d,  $J = 11.6$  Hz, 1H), 4.43 (d,  $J = 11.6$  Hz, 1H), 3.01 (td,  $J = 10.4, 3.9$  Hz, 1H), 2.80 (dtd,  $J = 20.4, 5.0, 2.6$  Hz, 1H), 2.59-2.55 (m, 1H), 2.28 (tdd,  $J = 21.5, 11.5, 5.3$  Hz, 2H), 2.07 (dq,  $J = 9.8, 3.4$  Hz, 1H), 1.87-1.76 (m, 4H), 1.54-1.38 (m, 3H), 1.33-1.22 (m, 2H), 0.98-0.73 (m, 4H), 0.95 (d,  $J = 6.5$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$  at 77 ppm)  $\delta$  200.2, 138.8, 136.8, 135.6, 128.3, 127.7, 127.5, 82.9, 70.5, 46.9, 46.8, 45.7, 42.4, 41.4, 39.5, 39.3, 36.8, 30.7, 30.1, 29.9, 27.5, 19.7; IR (thin film) 2917, 2858, 1686, 1619, 1454, 1256  $\text{cm}^{-1}$ ; HRMS (ESI) calculated for  $\text{C}_{24}\text{H}_{30}\text{O}_2$   $[\text{M}+\text{Na}]^+$  373.2144 found 373.2159.

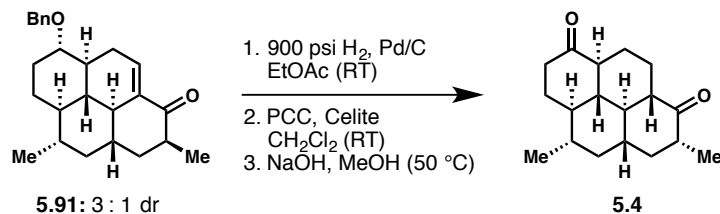
### Methyl Enone 5.91



LDA was prepared in a 1 dram vial by addition of 0.3 mL (0.17 mmol) 0.57 M diisopropylamine/THF followed by 0.06 mL (0.15 mmol) 2.5 M/hexanes at 0 °C. To the stirring LDA solution at  $-78$  °C was added 42 mg (0.12 mmol,  $>10:1$  dr) **5.89** with the assistance of 0.5 mL THF. After 10 minutes, 0.03 mL (0.17 mmol) HMPA was added neat followed by 0.03 mL (0.48 mmol) methyl iodide. The cold bath was removed after an additional 10 minutes and the reaction stirred for 50 minutes before 2 mL half sat. aq.  $\text{NH}_4\text{Cl}$  was added. The solution was extracted with 10 mL EtOAc. The organic layer was separated, washed with 2 mL brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude mixture was purified by

column chromatography (10:1→5:1 hexanes/EtOAc) to afford 37 mg (84%, 3:1 dr, ~15% dimethylation?) **5.91** as a yellow solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 7.36-7.29 (m, 5H), 6.79 (t, *J* = 2.3 Hz, 0.7H), 6.69 (t, *J* = 2.3 Hz, 0.2), 4.69 (d, *J* = 11.6 Hz, 1H), 4.43 (d, *J* = 11.6 Hz, 1H), 3.01 (td, *J* = 10.4, 3.9 Hz, 1H), 2.83-2.77 (m, 1H), 2.65-2.59 (m, 1H), 2.31-2.24 (m, 1H), 2.09-2.06 (m, 1H), 1.87-1.45 (m, 6H), 1.33-1.21 (m, 3H), 1.14 (d, *J* = 7.4 Hz, 3H), 0.97-0.74 (m, 4H), 0.96 (d, *J* = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) major δ (major) 204.0, (mixture of diastereomers) 138.82, 136.0, 135.9, 134.5, 128.3, 127.7, 127.7, 127.5, (major) 82.9, 70.5, 46.9, 46.8, 46.7, 45.9, 42.7, 41.5, 41.3, 37.1, 36.9, 34.1, 30.9, 30.2, 27.6, 19.7, 18.6; IR (thin film) 2917, 2864, 1684, 1769, 1453, 1093, 735 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>25</sub>H<sub>32</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 387.2300 found 387.2293.

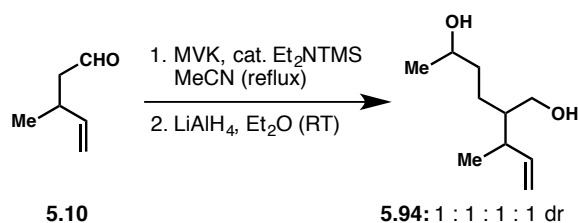
#### Corey's Dione **5.4**



A 1 dram vial containing 27 mg (0.074 mmol, 3:1 dr) **5.91** and 9 mg (0.0042 mmol) 10% Pd/C (50% wetted) in 0.8 mL EtOAc was stirred under 900 psi H<sub>2</sub> in a bomb reactor. After 23 hours, the reaction was filtered over Celite, eluting with EtOAc (~10 mL as determined by TLC) and all volatiles removed in vacuo to afford a keto-alcohol as a mixture of diastereomers. The crude material in a 1 dram vial was dissolved in 0.4 mL CH<sub>2</sub>Cl<sub>2</sub> and 64 mg Celite added followed by 60 mg (0.278 mmol) PCC open to air. After 2 hours the reaction was diluted with 1 mL Et<sub>2</sub>O and filtered over silica gel, eluting with Et<sub>2</sub>O (~30 mL as determined by TLC) and all volatiles were removed in vacuo. The crude material in a 1 dram vial was dissolved in 0.6 mL MeOH and 0.6

mL (0.078 mmol) 0.13 M NaOH/MeOH. After 30 minutes at 50 °C the reaction was cooled, diluted with 2 mL half sat. aq. NH<sub>4</sub>Cl and extracted with 10 mL EtOAc. The organic layer was washed with 2 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (5:1 hexanes/EtOAc) to afford 13 mg (65% over 3 steps, >20:1 dr) **5.4** as a white solid that was recrystallized from Et<sub>2</sub>O to afford wispy white needles (mp = 160–161 °C, Corey's mp (no solvent specified, 60% ee) = 137–140 °C<sup>1</sup>, Miyaoka's mp (no solvent specified, no enantiopurity mentioned) = 123–125 °C<sup>62</sup>). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 2.50-2.32 (m, 4H), 2.05-1.97 (m, 5H), 1.75 (dt, *J* = 13.1, 3.4 Hz, 1H), 1.67 (qt, *J* = 11.5, 3.4 Hz, 1H), 1.34-1.13 (m, 6H), 1.10-1.03 (m, 2H), 1.01 (d, *J* = 6.5 Hz, 3H), 0.99 (d, *J* = 6.5 Hz, 3H), 0.87 (q, *J* = 12.3 Hz, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 213.0, 212.0, 53.6, 52.7, 52.1, 52.1, 46.3, 44.3, 43.0, 41.2, 41.0, 40.4, 36.4, 31.0, 23.7, 23.6, 19.9, 14.4; IR (thin film) 2923, 2859, 1710, 1455, 1259, 1088, 800 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>18</sub>H<sub>26</sub>O<sub>2</sub> [M]<sup>+</sup> 274.1933 found 274.1944.

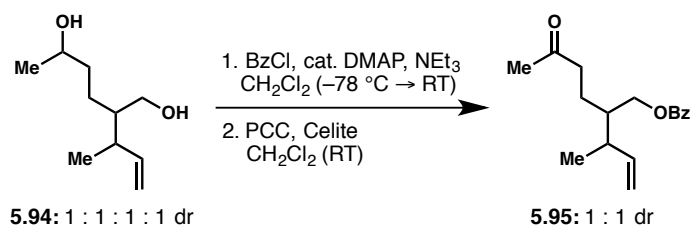
### Diol (±)-**5.94**



A 1 dram vial was charged with 52 mg (0.530 mmol) **5.10**, 1.5 mL MeCN, 0.05 mL (0.83 mmol) MVK and 0.02 mL (0.11 mmol) Et<sub>2</sub>NTMS. The vial sealed and immersed into a 90 °C oil bath for 21 hours. The reaction was cooled to room temperature and all volatiles removed in vacuo. The crude Michael adduct was added to 21 mg (0.52 mmol) LiAlH<sub>4</sub> in 3 mL Et<sub>2</sub>O at 0 °C with the assistance of 2 mL Et<sub>2</sub>O. The ice bath was removed and stirring continued for 30 minutes

before being recooled to 0 °C and treated with 0.09 mL water, 0.02 mL 5 M NaOH and Na<sub>2</sub>SO<sub>4</sub>. After 30 minutes of vigorous stirring the solution was filtered over Celite, eluting with Et<sub>2</sub>O (~15 mL as determined by TLC) and all volatiles removed in vacuo. The crude material was purified by column chromatography (1:2→1:3 Hexanes/EtOAc) to afford 77 mg (85%, 1:1:1:1 dr) (±)-**5.94** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 5.83-5.74 (m, 1H), 5.03-4.98 (m, 2H), 3.83-3.75 (m, 1H), 3.65-3.53 (m, 2H), 2.37-2.23 (m, 1H), 2.13-2.04 (m, 2H), 1.54-1.32 (m, 5H), 1.19 (d, *J* = 6.2 Hz, 3H), 1.02 (d, *J* = 6.9 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 142.97, 142.87, 142.85, 142.77, 114.0, 113.80, 113.76, 68.4, 68.0, 63.6, 63.5, 63.35, 63.25, 45.5, 45.4, 45.09, 45.06, 38.8, 38.74, 38.71, 38.7, 37.1, 36.8, 36.7, 36.4, 24.1, 24.0, 23.8, 23.6, 23.6, 23.5, 17.0, 16.9, 16.8, 16.8; IR (thin film) 3335, 2964, 2928, 2875 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>10</sub>H<sub>20</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 195.1361 found 195.1353.

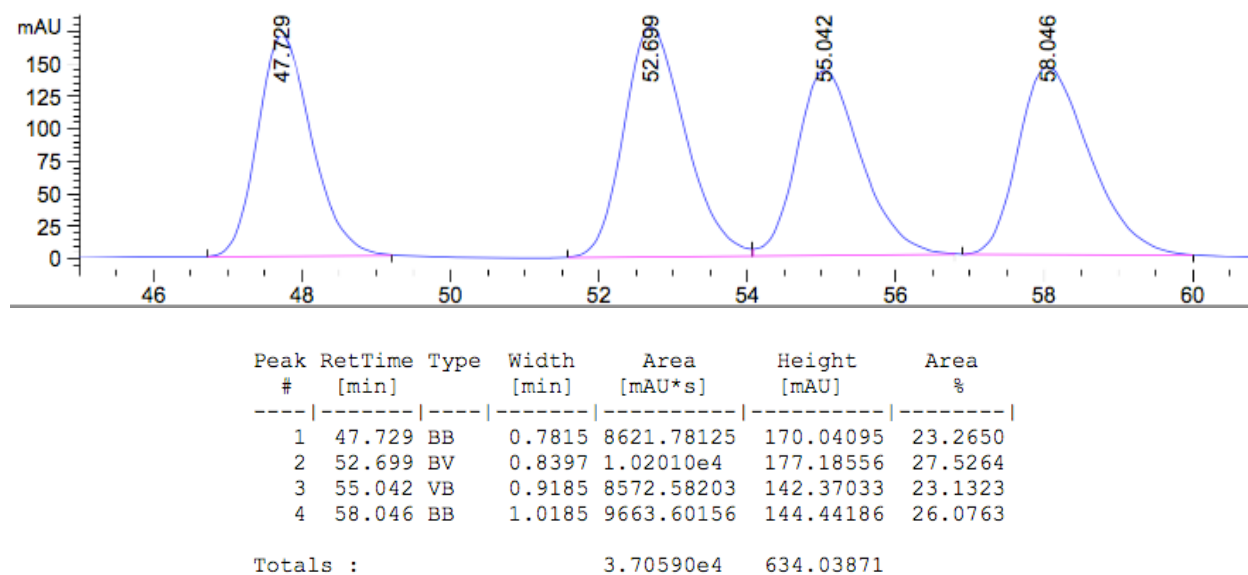
### Benzoate (±)-**5.95**



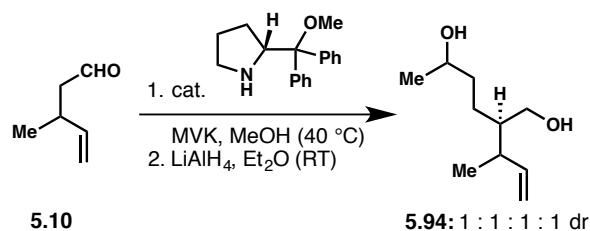
A 1 dram vial containing 12 mg (0.070 mmol, 1:1:1:1 dr) (±)-**S94**, 1.7 mg (0.014 mmol) DMAP, 0.5 mL CH<sub>2</sub>Cl<sub>2</sub> and 0.1 mL (0.72 mmol) NEt<sub>3</sub> was cooled to -78 °C. The reaction was treated with 0.1 mL (0.070 mmol) 0.7 M BzCl/CH<sub>2</sub>Cl<sub>2</sub> and stirred for 1 hour before the cold bath was removed. After 30 minutes at room temperature the reaction was quenched with 1 mL sat. aq. NaHCO<sub>3</sub> and extracted with 5 mL EtOAc. The organic layer was separated, washed with 1 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude alcohol was dissolved in 0.4 mL CH<sub>2</sub>Cl<sub>2</sub> and stirred for 2 hours with 48 mg Celite and 32 mg (0.15 mmol)

PCC open to air. The reaction was diluted with 2 mL Et<sub>2</sub>O and filtered over silica with Et<sub>2</sub>O (~15 mL as determined by TLC). All volatiles were removed in vacuo and purified by column chromatography to afford 8.2 mg (42%, 1:1 dr) (±)-**5.95** as colorless oil. Chiral HPLC analysis was obtained on an AD-H column, 99:1 hexanes/IPA at 0.4 mL/min, visualization at 230 nm, 54:46 dr, 50.1:49.9 (0%ee) diastereomer 1, 51.4:48.6 (3%ee) diastereomer 2. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) δ 8.04 (d, *J* = 8.2 Hz, 2H), 7.58 (t, *J* = 7.4 Hz, 1H), 7.46 (t, *J* = 7.4 Hz, 2H), 5.82-5.75 (m, 1H), 5.08-5.03 (m, 2H), 4.35-4.23 (m, 2H), 2.63-2.48 (m, 2H), 2.42 (dq, *J* = 12.8, 6.4 Hz, 1H), 2.15 (s, 3H), 1.90-1.72 (m, 2H), 1.68-1.55 (m, 1H), 1.10 (d, *J* = 6.9 Hz, 1.5H), 1.09 (d, *J* = 6.9 Hz, 1.5H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ 208.5, 166.6, 141.5, 133.0, 130.2, 129.5, 128.4, 114.8, 114.7, 65.3, 65.2, 42.0, 41.8, 41.6, 41.4, 39.0, 38.9, 30.0, 22.5, 22.5, 17.0, 16.8; IR (thin film) 2963, 2928 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>17</sub>H<sub>22</sub>O<sub>3</sub> [M+Na]<sup>+</sup> 297.1467 found 297.1464.

**Figure 5.3.** Chiral HPLC chromatogram of (±)-**5.95**.

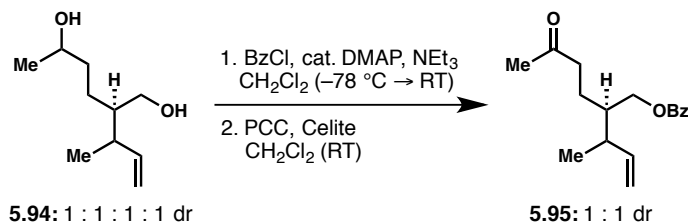


## Diol 5.94



A 1 dram vial was charged with 5.7 mg (0.021 mmol) prolinol catalyst,<sup>63</sup> 52 mg (0.53 mmol) **5.10**, 1 mL MeOH and 0.05 mL (0.83 mmol) MVK. The vial sealed and heated at 40 °C. After 36 hours, the reaction was cooled to room temperature and all volatiles removed in vacuo. The crude Michael adduct was added to 25 mg (0.66 mmol) LiAlH<sub>4</sub> in 3 mL Et<sub>2</sub>O at 0 °C with the assistance of 2 mL Et<sub>2</sub>O. The ice bath was removed and stirring continued for 30 minutes before being recooled to 0 °C and treated with 0.09 mL water, 0.02 mL 5 M NaOH and Na<sub>2</sub>SO<sub>4</sub>. After a minimum of 30 minutes of vigorous stirring, the solution was filtered over Celite, eluting with Et<sub>2</sub>O (~15 mL as determined by TLC) and all volatiles removed in vacuo. The crude material was purified by column chromatography (1:2→1:3 Hexanes/EtOAc) to afford 70 mg (77%) **5.94** as a colorless oil. The spectral data was identical to (±)-**5.94**.

## Benzoate 5.95

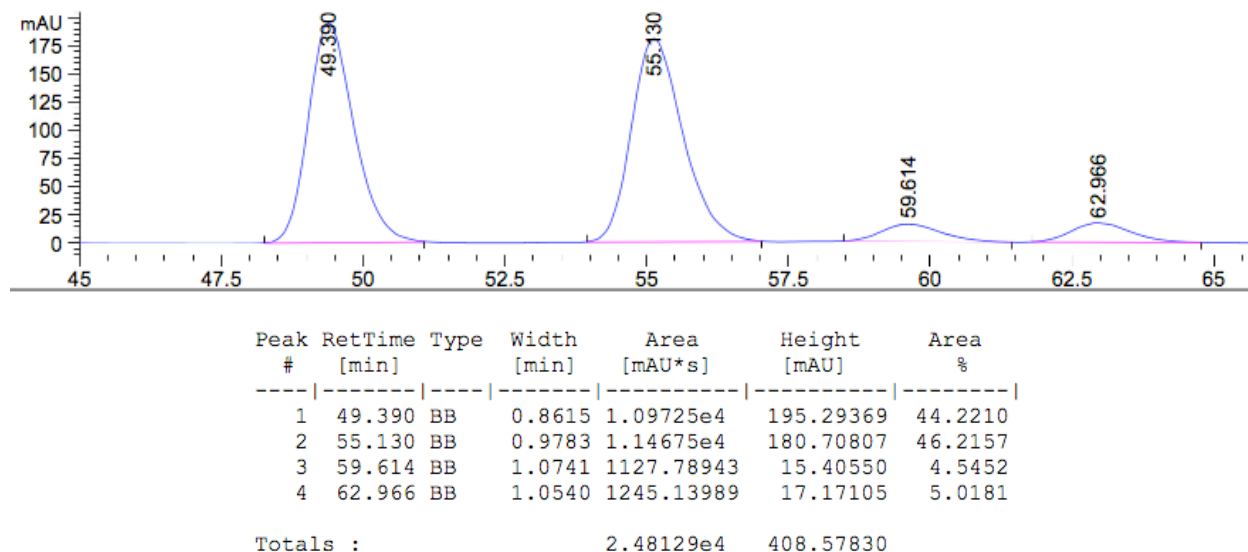


A 1 dram vial containing 70 mg (0.41 mmol) **5.94**, 10 mg (0.82 mmol) DMAP, 3 mL CH<sub>2</sub>Cl<sub>2</sub> and 0.4 mL (2.8 mmol) NEt<sub>3</sub> was cooled to -78 °C. The reaction was treated with 0.2 mL (0.50 mmol) 0.25 M BzCl/CH<sub>2</sub>Cl<sub>2</sub> and stirred for 0.5 hour before the cold bath was removed. After 1

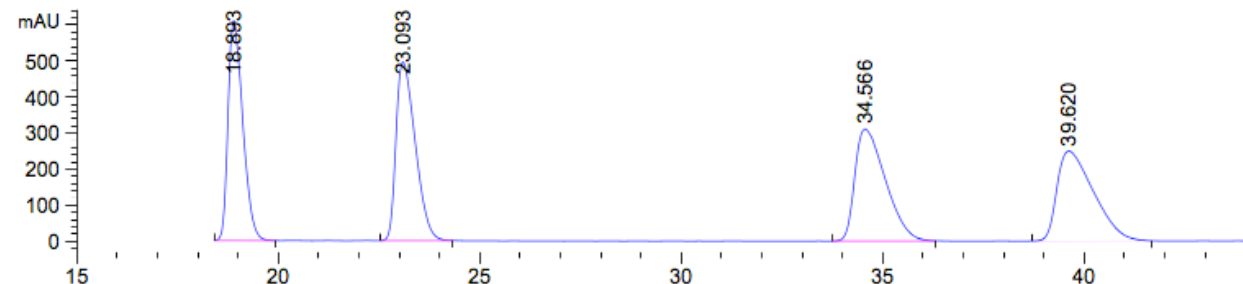


hour at room temperature the reaction was quenched with NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was separated, washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude alcohol was dissolved in 4 mL CH<sub>2</sub>Cl<sub>2</sub> and treated with 350 mg (4.2 mmol) NaHCO<sub>3</sub> and 257 mg (0.61 mmol) DMP, then stirred until complete open to air. The reaction was quenched with sat. aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and sat. aq. NaHCO<sub>3</sub>, before being extracted with EtOAc. The organic layer was separated and washed with sat. aq. NaHCO<sub>3</sub>, brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (8:1 hexanes/EtOAc) to afford 82 mg (74%, 49:51 dr) **5.95** as a colorless oil, 90.7:9.3 (81%ee) diastereomer 1, 90.2:9.8 (80%ee) diastereomer 2. Chiral HPLC analysis was obtained on an AD-H column, using 99:1 hexanes/IPA at 0.4 mL/min and visualized at 230 nm. The spectral data was identical to (±)-**5.95**.

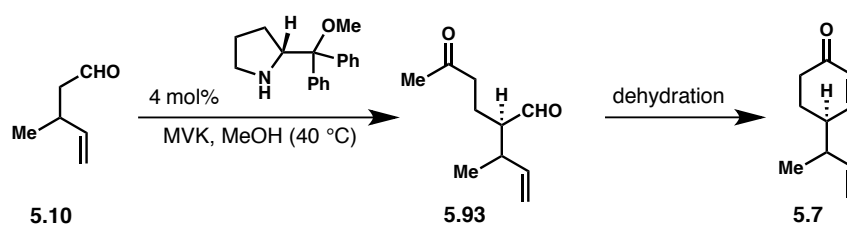
**Figure 5.4.** Chiral HPLC chromatogram for Table 5.5, Entry5 for **5.95**



**Figure 5.5.** Chiral HPLC chromatogram for cyclohexenone ( $\pm$ )-**5.7**.



Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	18.893	BB	0.3973	1.55348e4	609.68811	24.2550
2	23.093	BB	0.5215	1.64443e4	495.12198	25.6750
3	34.566	BB	0.8165	1.63599e4	310.61008	25.5433
4	39.620	BB	0.9747	1.57088e4	249.44130	24.5267
Totals :				6.40477e4	1664.86147	

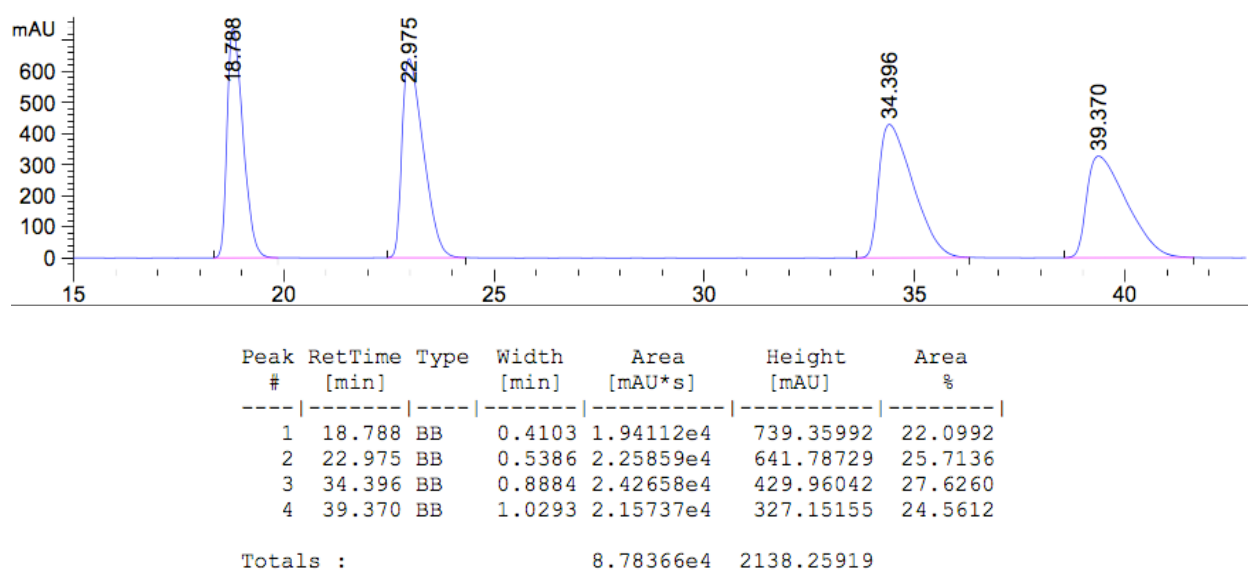


A 1 dram vial was charged with 11 mg (0.042 mmol) prolinol,<sup>63</sup> 108 mg (1.10 mmol) **5.10**, 1.1 mL MeOH and 0.13 mL (1.6 mmol) MVK. The vial was sealed and immersed in a 40 °C oil bath. After 39 hours the reaction was cooled and all volatiles removed in vacuo. Chiral HPLC analysis was obtained on an AS-H column, using 90:10 hexanes/IPA at 0.75 mL/min and visualized at 230 nm. The crude Michael adduct **5.93** was divided evenly into three 1 dram vials.

**Table 5.6, Entry 1: 1. DBU 2. MsCl:**<sup>64</sup> The crude **5.93** was dissolved in 1.2 mL dry CH<sub>2</sub>Cl<sub>2</sub>, cooled to 0 °C and treated with 0.17 mL (1.1 mmol) DBU. The cold bath was immediately removed and the reaction stirred for 10 hours. The reaction was then treated with 0.05 mL (0.65 mmol) MsCl at 0 °C. The cold bath was removed. After stirring for 12 hours the reaction was

cooled to 0 °C, quenched with sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 33 mg (59%) enantioenriched **5.7** as a colorless oil, 53:47 dr, 52:48 (4%ee) diastereomer 1, 53:47 (6%ee) diastereomer 2.

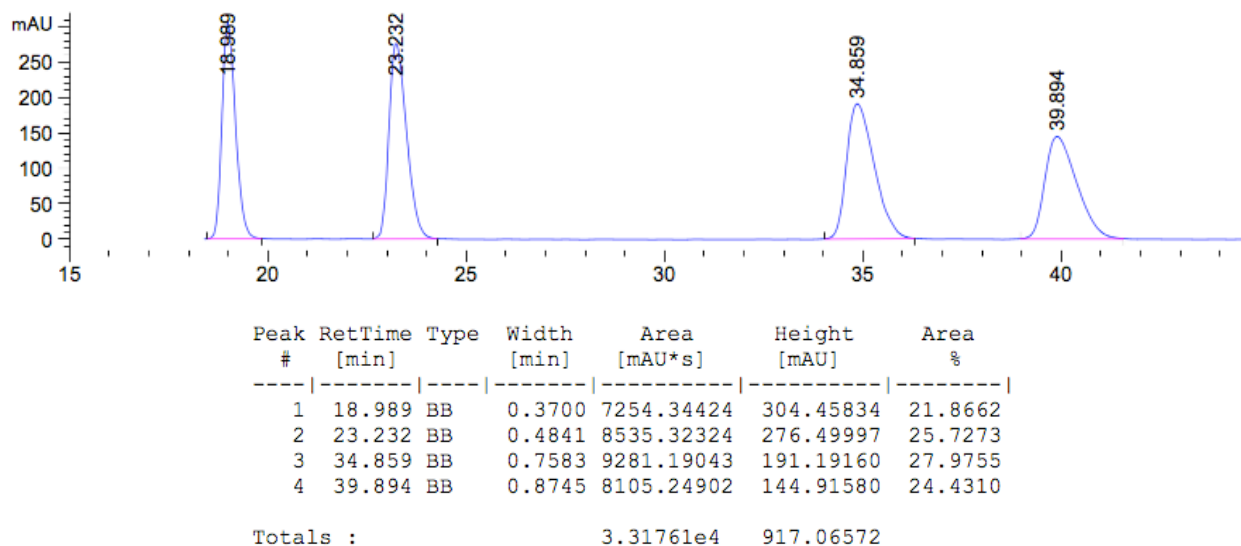
**Figure 5.6.** Chiral HPLC chromatogram for Table 5.6, entry 1.



**Table 5.6, Entry 2: 1. DBU 2. MsCl:**<sup>64</sup> The crude **5.93** was dissolved in 1.2 mL “wet” CH<sub>2</sub>Cl<sub>2</sub>, cooled to 0 °C and treated with 0.17 mL (1.1 mmol) DBU open to air. The cold was immediately removed and stirred for 8 hours. The reaction was quenched with brine and extracted with EtOAc. The organic phase was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude hydroxyketone was dissolved in 1 mL CH<sub>2</sub>Cl<sub>2</sub> and treated with 0.27 mL (1.9 mmol) NEt<sub>3</sub> and 0.05 mL (0.65 mmol) MsCl at 0 °C. The cold bath was removed. After stirring for 30 hours the reaction was cooled to 0 °C, quenched with sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and

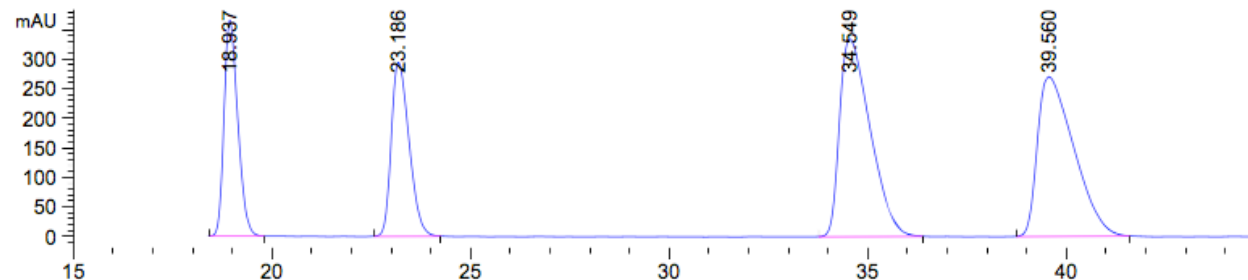
all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 30 mg (54%) enantioenriched **5.7** as a colorless oil, 54:46 dr, 52:48 (4%ee) diastereomer 1, 53:47 (6%ee) diastereomer 2.

**Figure 5.7.** Chiral HPLC chromatogram for Table 5.6, entry 2.

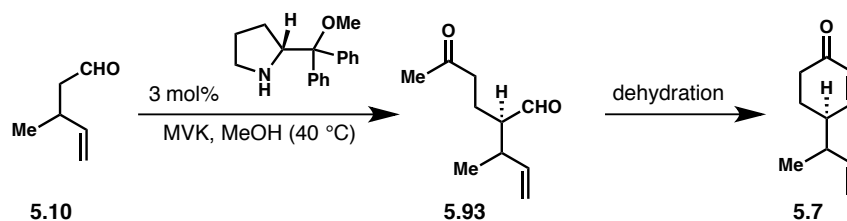


**Table 5.6, Entry 3: TsOH, MeCN:** The crude **5.93** was dissolved in 1 mL MeCN then 140 mg (0.74 mmol) TsOH•H<sub>2</sub>O added open to air. After 8 hours, the reaction was cooled to 0 °C and quenched with 0.7 mL 1 M NaOH and sat. aq. NaHCO<sub>3</sub>. All volatiles were removed in vacuo and the aqueous layer extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 36 mg (62%) enantioenriched **5.7** as a colorless oil, 51:49 dr, 67:33 (34%ee) diastereomer 1, 66:34 (32%ee) diastereomer 2.

**Figure 5.8.** Chiral HPLC chromatogram for Table 5.6, entry 3.



Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	18.937	BB	0.3673	8803.39941	365.06793	16.5764
2	23.186	BB	0.4860	9190.24023	294.55118	17.3048
3	34.549	BB	0.8253	1.78740e4	334.55463	33.6559
4	39.560	BB	0.9282	1.72404e4	270.27594	32.4629
Totals :				5.31080e4	1264.44968	

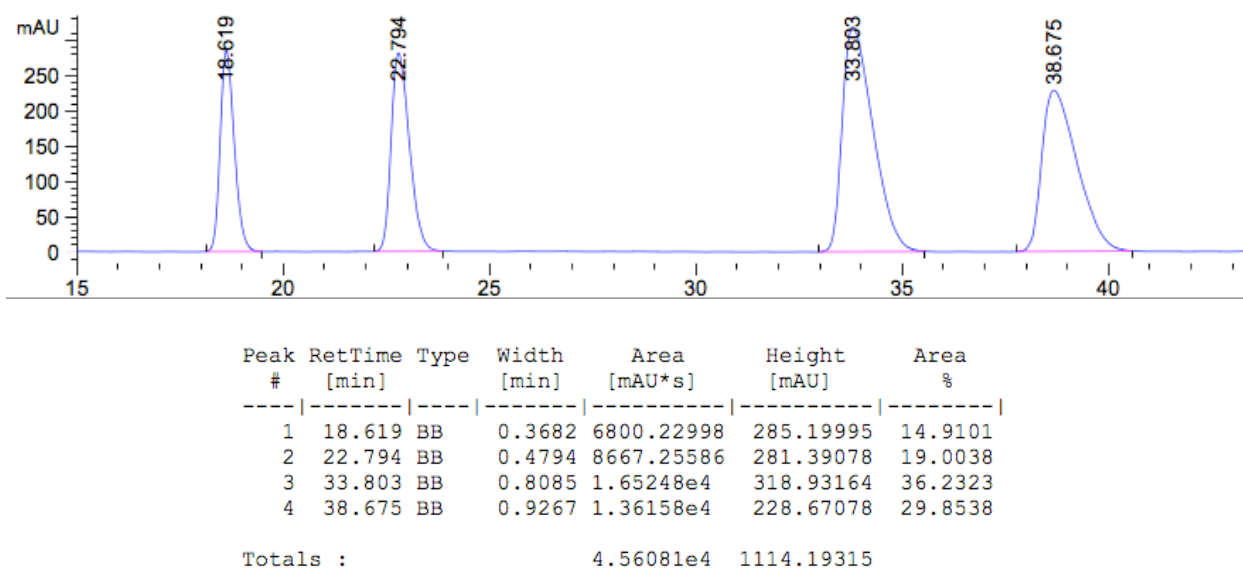


A 1 dram vial was charged with 11 mg (0.043 mmol) prolinol,<sup>63</sup> 113 mg (1.15 mmol) **5.10**, 1.1 mL MeOH and 0.13 mL (1.6 mmol) MVK. The vial was sealed and immersed in a 40 °C oil bath. After 36 hours the reaction was cooled and all volatiles removed in vacuo. Chiral HPLC analysis was obtained on an AS-H column, using 90:10 hexanes/IPA at 0.75 mL/min and visualized at 230 nm. The crude Michael adduct **5.93** was divided evenly into four 1 dram vials.

**Table 5.6, Entry 4: TBD, THF then MsCl:** The crude **5.93** was dissolved in 1 mL THF and treated with 2.6 mg (0.02 mmol) TBD. Stirring was continued for 20 minutes after which time the reaction was cooled to 0 °C and 0.16 mL (1.1 mL) NEt<sub>3</sub>, ~2 mg DMAP and 0.03 mL (0.39 mmol) MsCl added. The cold bath was removed. After stirring for 22 hours the reaction was

cooled to 0 °C, quenched with sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 26 mg (59%) enantioenriched **5.7** as a colorless liquid, 55:45 dr, 66:34 (32%ee) diastereomer 1, 67:33 (34%ee) diastereomer 2.

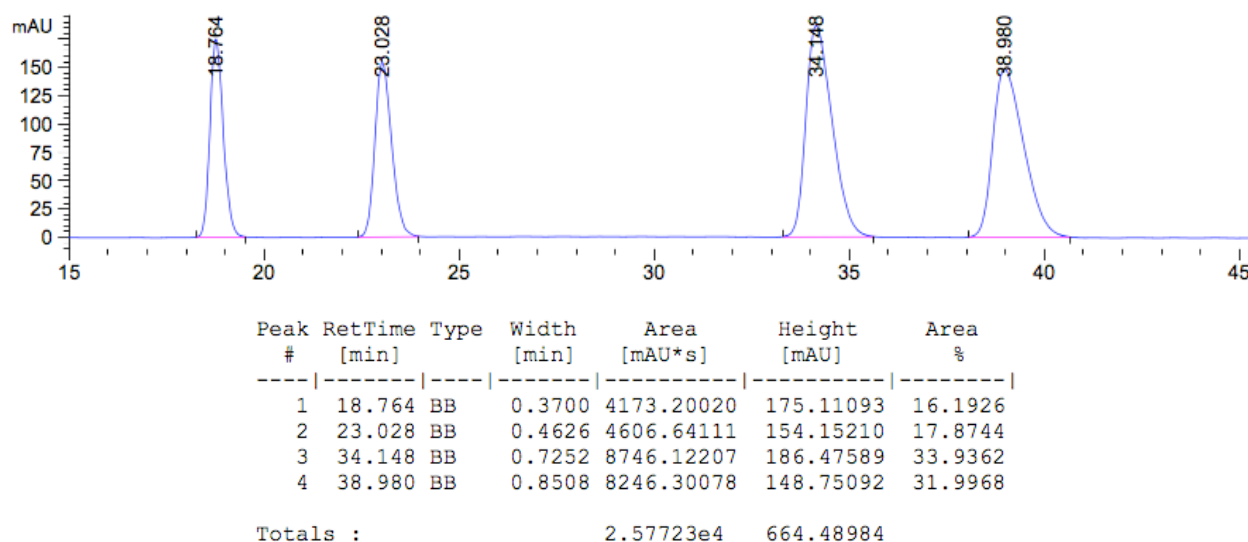
**Figure 5.9.** Chiral HPLC chromatogram for Table 5.6, entry 4.



**Table 5.6, Entry 5: TBD, THF then Ac<sub>2</sub>O:** The crude **5.93** was dissolved in 1 mL THF and treated with 2.6 mg (0.02 mmol) TBD. Stirring was continued for 20 minutes after which time the reaction was cooled to 0 °C and 0.16 mL (1.1 mL) NEt<sub>3</sub>, ~2 mg DMAP and 0.04 mL (0.42 mmol) Ac<sub>2</sub>O added. The cold bath was removed. After stirring for 22 hours the reaction was cooled to 0 °C, quenched with sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 29 mg

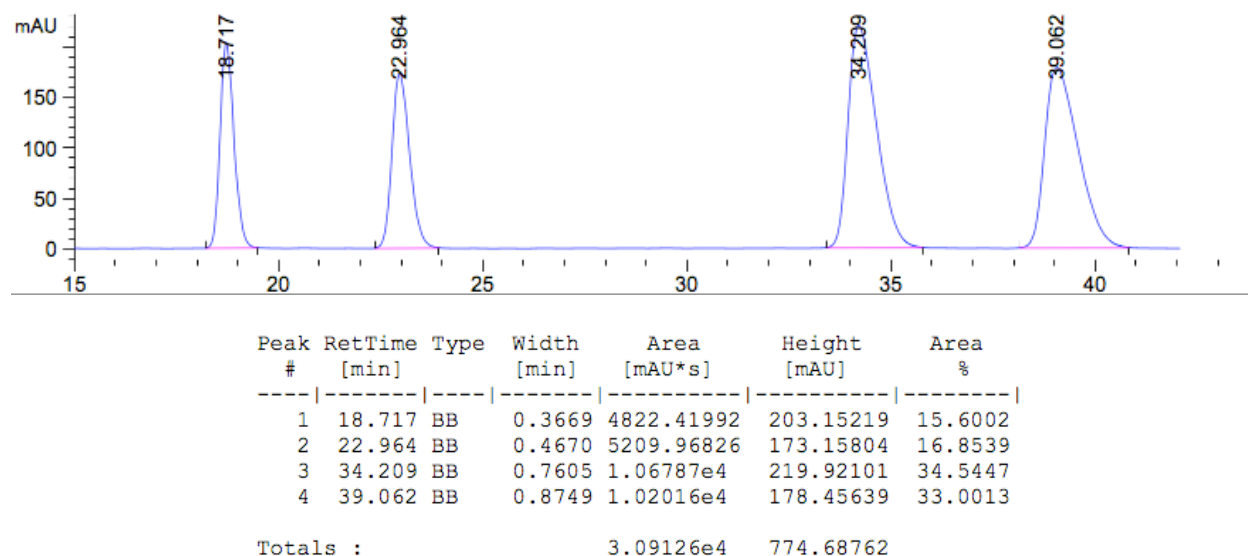
(66%) enantioenriched **5.7** as a colorless liquid, 52:48 dr, 66:34 (32%ee) diastereomer 1, 66:34 (32%ee) diastereomer 2.

**Figure 5.10.** Chiral HPLC chromatogram for Table 5.6, entry 5.



**Table 5.6, Entry 6: TBD, MeCN then TsOH:** The crude **5.93** was dissolved in 1 mL MeCN and treated with 2.7 mg (0.02 mmol) TBD. Stirring was continued for 20 minutes after which time 54 mg (0.28 mmol) TsOH•H<sub>2</sub>O was added. After stirring for 4 hours the reaction was cooled to 0 °C, quenched with 0.2 mL 1 M NaOH, sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 30 mg (68%) enantioenriched **5.7** as a colorless liquid, 51:49 dr, 67:33 (34%ee) diastereomer 1, 68:32 (36%ee) diastereomer 2.

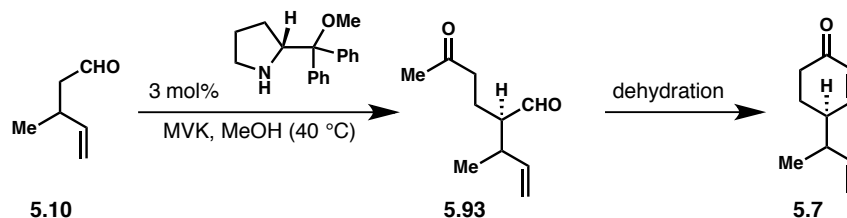
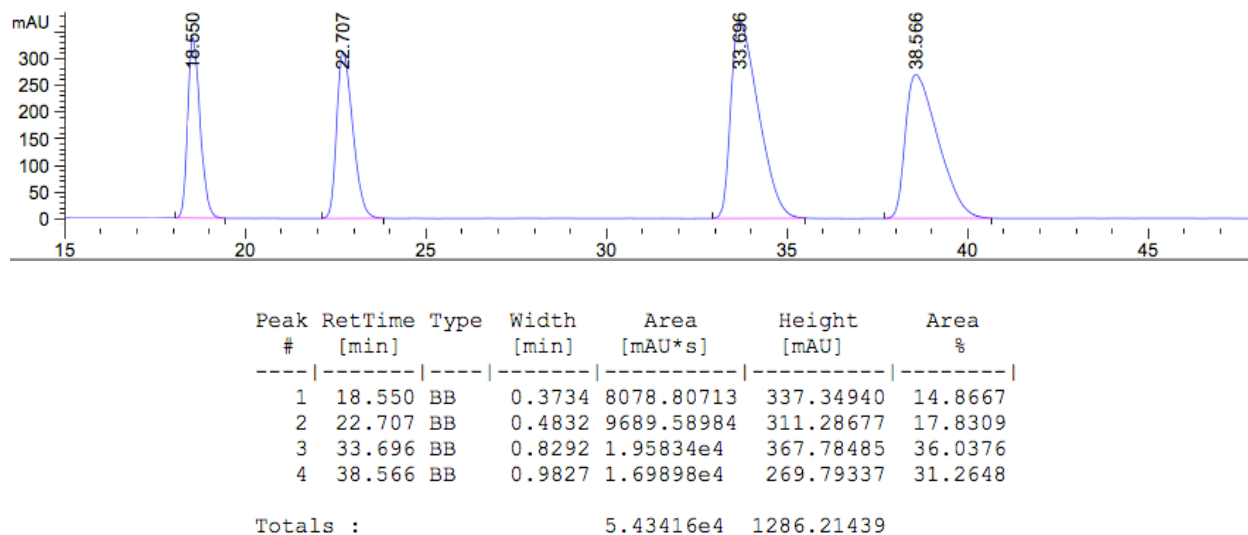
**Figure 5.11.** Chiral HPLC chromatogram for Table 5.6, entry 6.



**Table 5.6, Entry 7: TBD, CH<sub>2</sub>Cl<sub>2</sub> then MsCl:** The crude **5.93** was dissolved in 1 mL CH<sub>2</sub>Cl<sub>2</sub> and treated with 2.8 mg (0.02 mmol) TBD. Stirring was continued for 3.5 hours after which time the reaction was cooled to 0 °C and 0.16 mL (1.1 mmol) NEt<sub>3</sub>, ~2 mg DMAP and 0.03 mL (0.39 mmol) MsCl added. The cold bath was removed. After stirring for 18 hours the reaction was cooled to 0 °C, quenched with sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 25 mg (56%) enantioenriched **5.7** as a colorless liquid, 54:46 dr, 67:33 (34%ee) diastereomer 1, 68:32 (36%ee) diastereomer 2.



**Figure 5.12.** Chiral HPLC chromatogram for Table 5.6, entry 7.

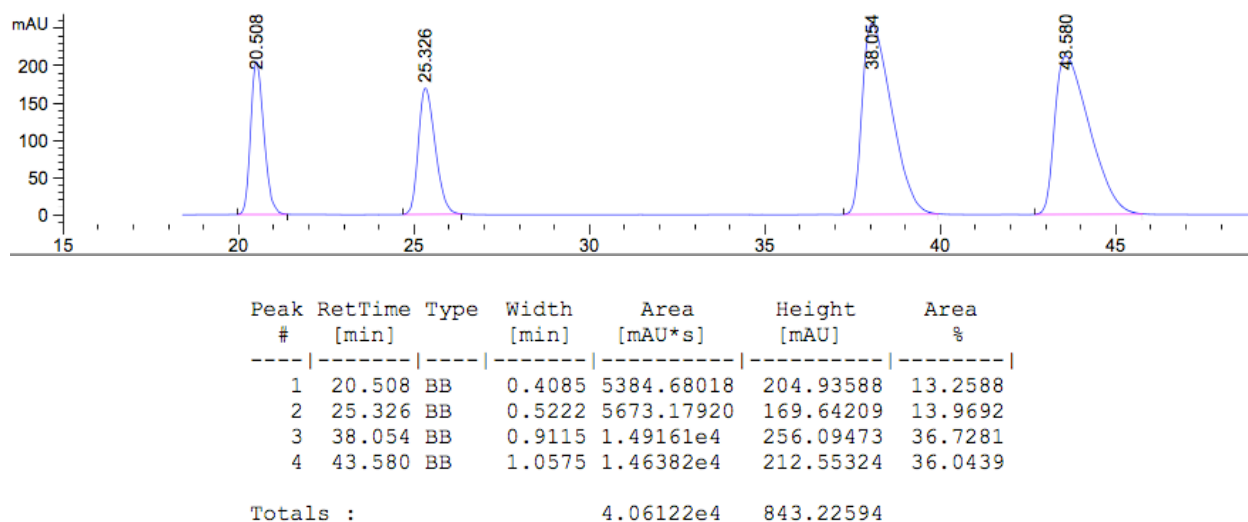


A 1 dram vial was charged with 14 mg (0.050 mmol) prolinol,<sup>63</sup> 166 mg (1.69 mmol) **5.10**, 1.5 mL MeOH and 0.2 mL (2.5 mmol) MVK. The vial was sealed and immersed in a 40 °C oil bath. After 36 hours the reaction was cooled and all volatiles removed in vacuo. Chiral HPLC analysis was obtained on an AS-H column, using 90:10 hexanes/IPA at 0.75 mL/min and visualized at 230 nm. The crude Michael adduct was divided evenly into six 1 dram vials.

**Table 5.6, Entry 8: 1. TBD, MeOH 2. MsCl:** The crude **5.93** was dissolved in 1 mL MeOH and treated with 1.7 mg (0.012 mmol) TBD. Stirring was continued for 4 hours after which time the reaction was quenched with sat. aq.  $\text{NH}_4\text{Cl}$  and extracted with EtOAc. The organic layer was separated, washed with brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The

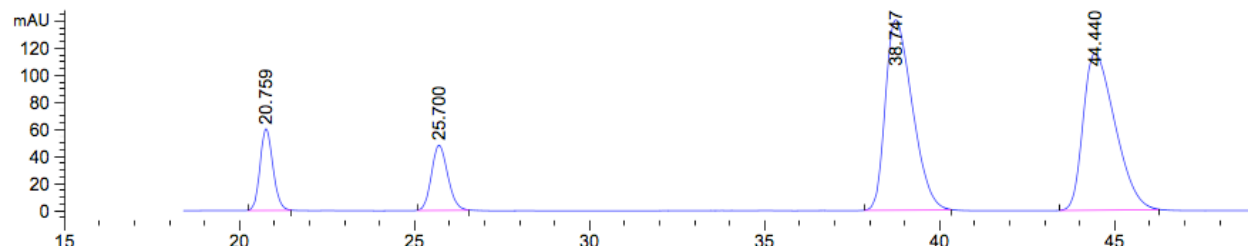
crude material was dissolved in 1 mL CH<sub>2</sub>Cl<sub>2</sub>, cooled to 0 °C and treated with 0.25 mL (1.8 mmol) NEt<sub>3</sub>, ~3 mg DMAP and 0.03 mL (0.39 mmol) MsCl added. The cold bath was removed. After stirring for 18 hours the reaction was cooled to 0 °C, quenched with sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 23 mg (55%) enantioenriched **5.7** as a colorless liquid, 51:49 dr, 73:27 (46%ee) diastereomer 1, 73:27 (46%ee) diastereomer 2.

**Figure 5.13.** Chiral HPLC chromatogram for Table 5.6, entry 8.



**Table 5.6, Entry 9: McQuade Diamine Catalyst:**<sup>6</sup> The crude **5.93** was dissolved in 1 mL hexanes and treated with 22 mg (0.087 mmol) McQuade diamine catalyst. Stirring was continued for 24 hours after which time the reaction was diluted with water and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 26 mg (62%) enantioenriched **5.7** as a colorless liquid, 51:49 dr, 83:17 (66%ee) diastereomer 1, 83:17 (66%ee) diastereomer 2.

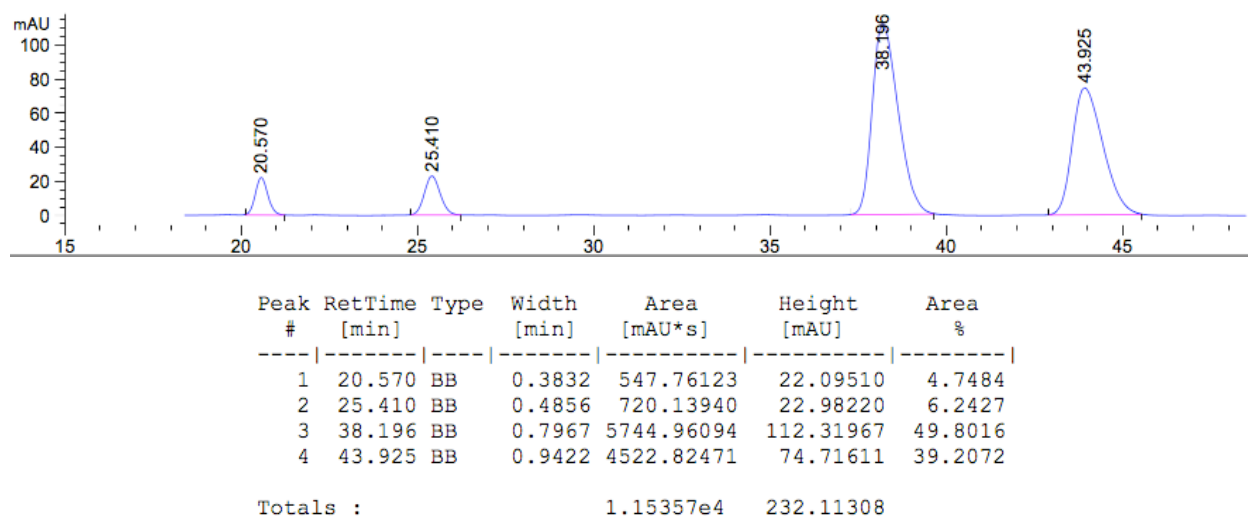
**Figure 5.14.** Chiral HPLC chromatogram for Table 5.6, entry 9.



Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	20.759	BB	0.3977	1548.58093	60.28382	8.5989
2	25.700	BB	0.5025	1563.21143	48.17989	8.6801
3	38.747	BB	0.8317	7550.24658	140.78036	41.9247
4	44.440	BB	0.9838	7347.04687	115.85931	40.7963
Totals :				1.80091e4	365.10339	

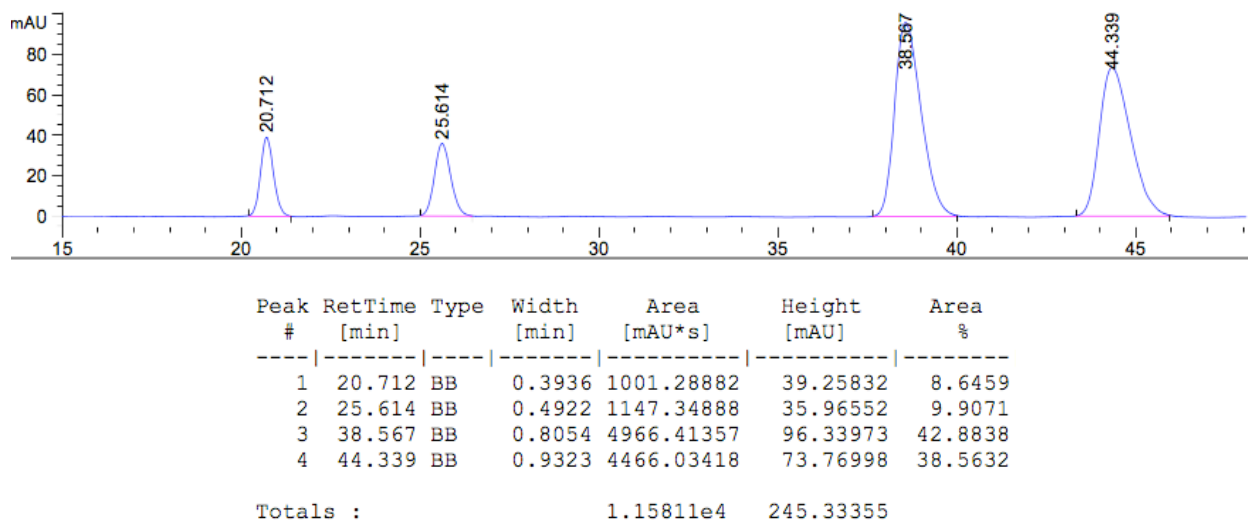
**Table 5.6, Entry 10: 1. LDA 2. MsCl:** The crude **5.93** was dissolved in 1 mL THF and treated with 0.3 mL (0.32 mmol) 0.93 M LDA/THF slowly at  $-78$  °C. Stirring was continued for 1 minute after which time the reaction was quenched with half sat. aq.  $\text{NH}_4\text{Cl}$  and extracted with EtOAc. The organic layer was separated, washed with brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was dissolved in 1 mL  $\text{CH}_2\text{Cl}_2$ , cooled to  $0$  °C and treated with 0.25 mL (1.8 mmol)  $\text{NEt}_3$ , ~3 mg DMAP and 0.03 mL (0.39 mmol) MsCl added. The cold bath was removed. After stirring for 18 hours the reaction was cooled to  $0$  °C, quenched with sat. aq.  $\text{NaHCO}_3$  and extracted with EtOAc. The organic layer was washed with brine, dried over  $\text{MgSO}_4$ , filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 10 mg (23%) enantioenriched **5.7** as a colorless liquid, 56:44 dr, 88:12 (76%ee) diastereomer 1, 89:11 (78%ee) diastereomer 2 and 13 mg (27%) Michael adduct.

**Figure 5.15.** Chiral HPLC chromatogram for Table 5.6, entry 10.

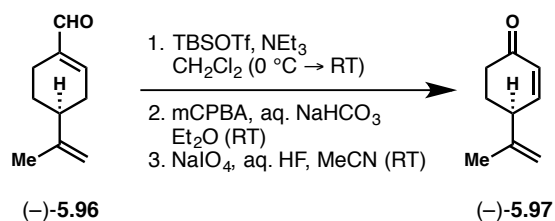


**Table 5.6, Entry 11: 1. KOt-Bu 2. MsCl:** A 1 dram vial was charged with 0.5 mL (0.08 mmol) 0.16 M KOt-Bu/THF and cooled to 0 °C. The crude **5.93** was added dropwise with the assistance of 1 mL THF. Stirring was continued for 1 minute after which time the reaction was quenched with half sat. aq. NH<sub>4</sub>Cl and extracted with EtOAc. The organic layer was separated, washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was dissolved in 1 mL CH<sub>2</sub>Cl<sub>2</sub>, cooled to 0 °C and treated with 0.25 mL (1.8 mmol) NEt<sub>3</sub>, ~3 mg DMAP and 0.03 mL (0.39 mmol) MsCl added. The cold bath was removed. After stirring for 18 hours the reaction was cooled to 0 °C, quenched with sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by column chromatography (10:1→8:1 hexanes/EtOAc) to afford 18 mg (43%) enantioenriched **5.7** as a colorless liquid, 53:47 dr, 81:19 (62%ee) diastereomer 1, 78:22 (56%ee) diastereomer 2.

**Figure 5.16.** Chiral HPLC chromatogram for Table 5.6, entry 11.



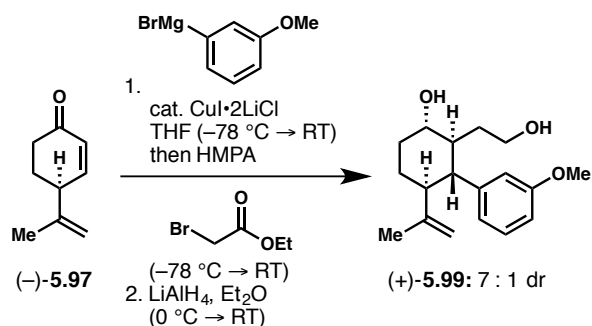
(S)-(-)-**97** [Adapted from the literature]<sup>65</sup>



A 250 mL round bottom flask was charged with 5.20 g (34.6 mmol) 90% pure (S)-(-)-perillaldehyde [(S)-(-)-**5.96**] and 130 mL CH<sub>2</sub>Cl<sub>2</sub>, cooled to 0 °C and sequentially treated with 10 mL (71.7 mmol) NEt<sub>3</sub> and 8.7 mL (37.9 mmol) TBSOTf. After 10 minutes at 0 °C, the flask was removed and stirring continued for 20 minutes. The contents were poured into 60 mL stirring sat. aq. NaHCO<sub>3</sub> and the flask was rinsed with 20 mL CH<sub>2</sub>Cl<sub>2</sub>. The layers were separated and the aqueous layer extracted with 20 mL CH<sub>2</sub>Cl<sub>2</sub>. The organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude dienoxysilane in a 500 mL round bottom flask was dissolved in 130 mL Et<sub>2</sub>O and 150 mL sat. aq. NaHCO<sub>3</sub> open to air. To the vigorously stirring mixture was added portionwise 14.7 g (59.7-63.9 mmol) 70-75% pure

mCPBA. Upon complete consumption of dienoxysilane by TLC the layers were separated and extracted with 50 mL Et<sub>2</sub>O. The organic layers were combined, washed with 20 mL sat. aq. NaHCO<sub>3</sub>, 20 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. To a 500 mL round bottom flask containing crude epoxide was added 100 mL MeCN then at 0 °C, 10.3 (48.2 mmol) NaIO<sub>4</sub> and slowly 2.5 mL HF in 25 mL water open to air. The reaction was further stirred for 5 minutes then the cold bath removed. After 2 hours 250 mL water was added and the solution extracted thrice with 125 mL Et<sub>2</sub>O. The organic layers were combined, washed with 25 mL sat. aq. NaHCO<sub>3</sub> and 25 mL brine. The last two aqueous washings were combined and back extracted with 30 mL Et<sub>2</sub>O. All organic layers were combined, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude material was purified by distillation (68–75 °C/1.5 mmHg) to afford 2.93 g (62% over 3 steps) (S)-(-)-**5.97** as a colorless solution. Spectral data was identical to the literature.<sup>65</sup> Optical rotation  $[\alpha]_{\text{D}}^{23} -159.1^{\circ}$  (*c* 1.03, MeOH) and  $[\alpha]_{\text{D}}^{24} -189.4^{\circ}$  (*c* 0.0162, MeOH); lit. for (S)-4-(2-propenyl)-2-cyclohexen-1-one is  $[\alpha]_{\text{D}}^{26} -153.8^{\circ}$  (*c* 1.03, MeOH),<sup>66</sup> for (R)-4-(2-propenyl)-2-cyclohexen-1-one is  $[\alpha]_{\text{D}}^{26} +192.2^{\circ}$  (*c* 0.0162, MeOH)<sup>65</sup> and  $[\alpha]_{\text{D}}^{22} +157.6^{\circ}$  (*c* 1.37, CHCl<sub>3</sub>).<sup>67</sup>

## Diol (+)-5.100



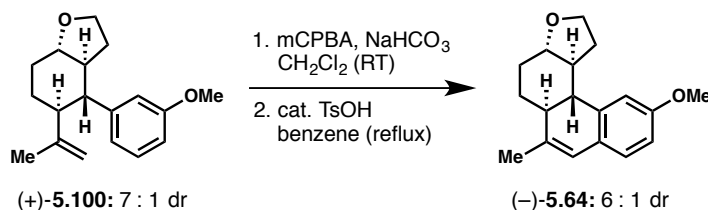
3-Methoxyphenylmagnesium bromide was prepared by addition of 0.7 mL (5.6 mmol) 3-bromoanisole in 0.7 mL THF to 0.177 g (7.3 mmol) magnesium metal in 2.8 mL THF activated by dibromoethane maintaining 40–50 °C. A 10 mL round bottom flask was charged with 2 mL THF and 0.7 mL (0.90 mmol) 1.29 M 3-methoxyphenylmagnesium bromide. At –78 °C, 0.3 mL (0.075 mmol) 0.25 M CuI·2LiCl/THF was added followed by 100 mg (0.73 mmol) (S)-(-)-5.97 in 0.6 mL THF. Stirring at –78 °C was continued for 2 hours before the cold bath was removed. After 1 hour the reaction was recooled to –78 °C. To the solution was added 0.26 mL (1.50 mmol) HMPA. After 1 hour 0.30 mL (3.62 mmol) ethyl bromoacetate was added. The reaction was stirred at –78 °C for 10 minutes then the cold bath removed. After stirring for 29 hours at room temperature, the reaction was cooled in an ice bath and treated with 1.5 mL sat. aq. NH<sub>4</sub>Cl, 1.5 mL water, extracted with 10 mL and twice with 5 mL Et<sub>2</sub>O. The organic layers were combined, washed with 1 mL half sat. aq. NH<sub>4</sub>Cl, 1 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo to afford a crude oil. To an ice cooled 25 mL round bottom flask containing 5 mL Et<sub>2</sub>O was added 158 mg (4.2 mmol) LiAlH<sub>4</sub>. The crude oil was transferred using 5 mL Et<sub>2</sub>O dropwise. The ice bath was removed after 5 minutes. After 5 hours the reaction was placed into a room temperature water bath and 0.3 mL EtOAc added, then the flask was recooled in an ice bath and 0.16 mL water, 0.16 mL 5 M NaOH, 0.45 mL water and MgSO<sub>4</sub> was added. After stirring the contents for several hours, the solution was filtered over Celite and the





mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (5:1 hexanes/EtOAc) to afford 57 mg (83%, 7:1 dr) (+)-**5.100** as a colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub> at 7.27 ppm) major δ 7.19 (t, *J* = 7.9 Hz, 1H), 6.74-6.68 (m, 3H), 4.57 (s, 2H), 3.91 (td, *J* = 9.1, 2.7 Hz, 1H), 3.85 (q, *J* = 8.0 Hz, 1H), 3.80 (s, 3H), 3.26 (td, *J* = 10.1, 4.0 Hz, 1H), 2.45-2.35 (m, 2H), 2.24-2.21 (m, 1H), 1.92 (td, *J* = 6.6, 3.6 Hz, 1H), 1.88-1.77 (m, 1H), 1.70-1.49 (m, 4H), 1.54 (s, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub> at 77 ppm) δ (mixture of diastereomers) 159.4, 146.8, 144.5, 129.1, 128.9, 121.2, 120.0, 114.9, 113.7, 112.0, 111.3, 110.9, 110.7 (major) 82.6, 67.1, 55.1, 51.7, 51.6, 50.4, 30.8, 30.6, 30.0, 19.9; IR (thin film) 3070, 2934, 2876, 1645, 1601, 1489, 1261, 1048, 776, 700 cm<sup>-1</sup>; HRMS (ESI) calculated for C<sub>18</sub>H<sub>24</sub>O<sub>2</sub> [M+Na]<sup>+</sup> 295.1674 found 295.1664; [α]<sub>D</sub><sup>22</sup> +17.5° (*c* 0.50, CHCl<sub>3</sub>).

### Dihydronaphthalene (-)-**5.64**



To a 1 dram vial containing 5 mg (0.018 mmol, 7:1 dr) (+)-**5.100** and 19 mg (0.23 mmol) NaHCO<sub>3</sub> in 0.3 mL CH<sub>2</sub>Cl<sub>2</sub> was added 9 mg (0.037 mmol) 70% mCPBA at room temperature open to air. After 13 hours 1 mL half sat. aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> was added and the mixture extracted with 2 mL and 1 mL EtOAc. The organic layers were combined, washed with 1 mL sat. aq. NaHCO<sub>3</sub>, 1 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. To a 5 mL round bottom flask containing crude epoxide was added 2 mL benzene and 2 mg (0.011 mmol) TsOH•H<sub>2</sub>O. The reaction was refluxed over a Hickman still for 2 hours. The reaction was

cooled, diluted with 1 mL sat. aq. NaHCO<sub>3</sub>, 3 mL EtOAc and the layers separated. The organic layer was washed with 1 mL brine, dried over MgSO<sub>4</sub>, filtered and all volatiles removed in vacuo. The crude oil was purified by column chromatography (6:1 hexanes/EtOAc) to afford 3 mg (60% over 2 steps, 6:1 dr) (-)-**5.64** as a white wax. Spectral data was identical to (±)-**5.64**;  $[\alpha]_D^{22} -165.0^\circ$  (*c* 0.50, CHCl<sub>3</sub>).

## 5.9 References and Notes

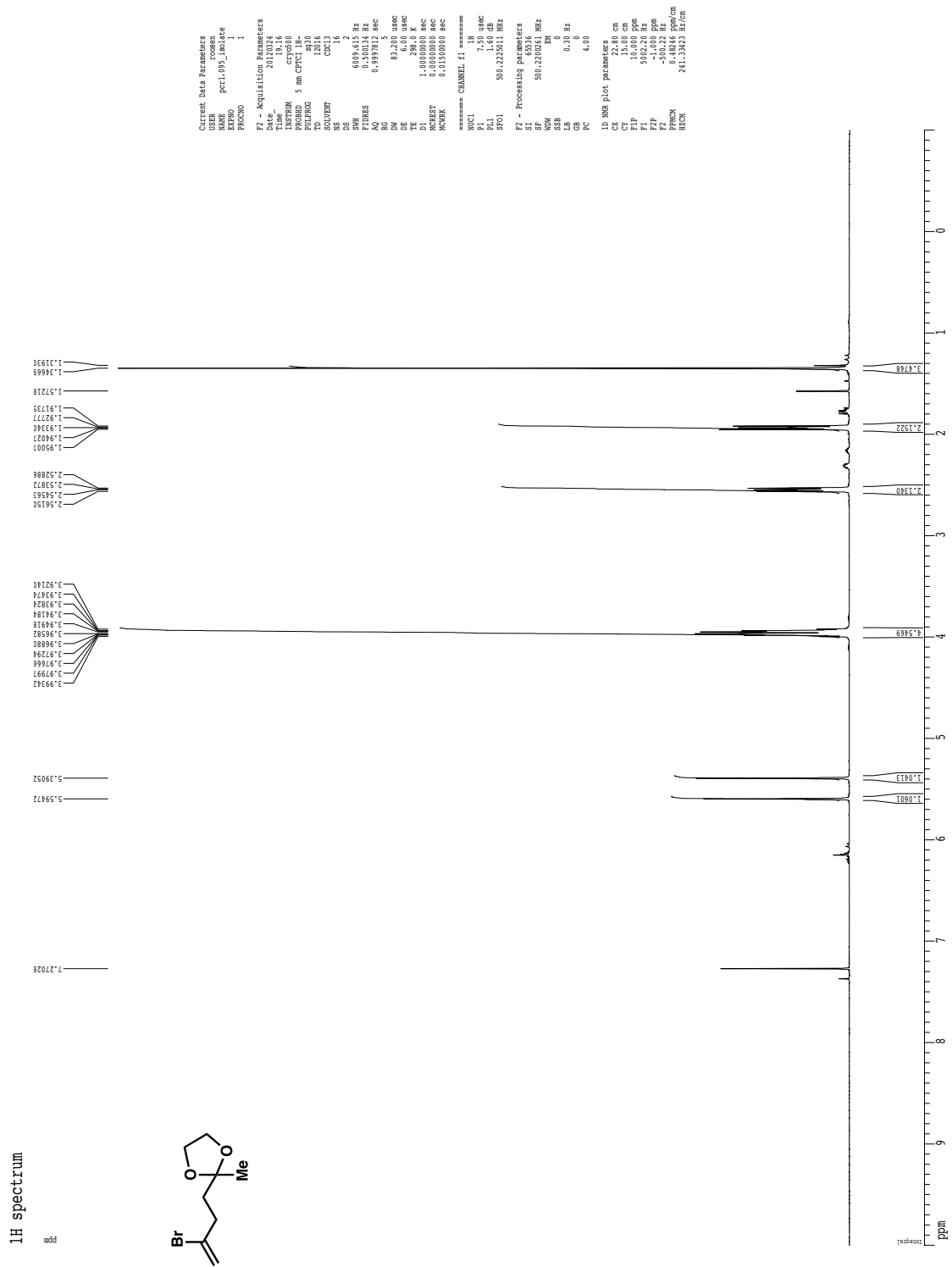
- (1) Corey, E. J.; Magriotis, P. A. *J. Am. Chem. Soc.* **1987**, *109*, 287.
- (2) Mander, L. N. *Synlett* **1991**, 134.
- (3) Parr, B. T.; Economou, C.; Herzon, S. B. *Nature* **2015**, *525*, 507.
- (4) Chi, Y.; Gellman, S. H. *Org. Lett.* **2005**, *7*, 4253.
- (5) Franzén, J.; Marigo, M.; Fielenbach, D.; Wabnitz, T. C.; Kjærsgaard, A.; Jørgensen, K. A. *J. Am. Chem. Soc.* **2005**, *127*, 18296.
- (6) Houjeiry, T. I.; Poe, S. L.; Mcquade, D. T. *Org. Lett.* **2012**, *14*, 4394.
- (7) Wei, X.; Lorenz, J. C.; Kapadia, S.; Saha, A.; Haddad, N.; Busacca, C. A.; Senanayake, C. *H. J. Org. Chem.* **2007**, *72*, 4250.
- (8) Chen, K.; Baran, P. S. *Nature* **2009**, *459*, 824.
- (9) Watanabe, W. H.; Conlon, L. E. *J. Am. Chem. Soc.* **1957**, *79*, 2828.
- (10) Reetz, M. T.; Kindler, A. *J. Organomet. Chem.* **1995**, *502*, C5.
- (11) Keinan, E.; Greenspoon, N. "Reduction of alpha,beta-Unsaturated Carbonyl Compounds" In *The Chemistry of Enones*; Patai, S.; Rappoport, Z., Ed.; John Wiley & Sons, Ltd, 1989; pp 923–1022.
- (12) Caine, D. *Org. React.* **1976**, *23*, 1.
- (13) Numbering matches the isocycloamphilectane scaffold, i.e. 7, 20-diisocyanoadociane.
- (14) Pronin, S. V.; Reiher, C. A.; Shenvi, R. A. *Nature* **2013**, *501*, 195.
- (15) Pronin, S. V.; Shenvi, R. A. *J. Am. Chem. Soc.* **2012**, *134*, 19604.
- (16) Sasaki, T.; Nakanishi, A.; Ohno, M. *J. Org. Chem.* **1981**, *46*, 5445.
- (17) Maruoka, K.; Itoh, T.; Yamamoto, H. *J. Am. Chem. Soc.* **1985**, *107*, 4573.
- (18) Suzuki, T.; Kobayashi, T.; Takegami, Y.; Kawasaki, Y. *Bull. Chem. Soc. Jpn.* **1974**, *47*,

- 1971.
- (19) Cahiez, G.; Duplais, C.; Buendia, J. *Chem. Rev.* **2009**, *109*, 1434.
- (20) Garst, M. E.; Dolby, L. J.; Esfandiari, S.; Fedoruk, N. A.; Chamberlain, N. C.; Avey, A. *J. Org. Chem.* **2000**, *65*, 7098.
- (21) Hassner, A.; Kropp, J. E.; Kent, G. J. *J. Org. Chem.* **1969**, *34*, 2628.
- (22) Sy, W.-W.; By, A. W. *Tetrahedron Lett.* **1985**, *26*, 1193.
- (23) Taniguchi, T.; Fujii, T.; Ishibashi, H. *J. Org. Chem.* **2010**, *75*, 8126.
- (24) Maity, S.; Naveen, T.; Sharma, U.; Maiti, D. *Org. Lett.* **2013**, *15*, 3384.
- (25) Maity, S.; Manna, S.; Rana, S.; Naveen, T.; Mallick, A.; Maita, D. *J. Am. Chem. Soc.* **2013**, *135*, 3355.
- (26) Cossio, F. P.; Aizpurua, J. M.; Palomo, C. *Can. J. Chem.* **1986**, *64*, 225.
- (27) Emsermann, J.; Kauh, U.; Opatz, T. *Mar. Drugs* **2016**, *14*, 16.
- (28) Huffman, J. W. "Reduction of C=X to CHXH by Dissolving Metals and Related Methods" In *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I., Ed.; Elsevier Science Ltd., 1991; pp 107–127.
- (29) Johnson, C. R.; Raheja, R. K. *J. Org. Chem.* **1994**, *59*, 2287.
- (30) Tsuda, T.; Hayashi, T.; Satomi, H.; Kawamoto, T.; Saegusa, T. *J. Org. Chem.* **1986**, *51*, 537.
- (31) Daniewski, A. R.; Kiegiel, J. *Synth. Commun.* **1988**, *18*, 115.
- (32) Krüerke, U. *Chem. Ber.* **1962**, *95*, 174.
- (33) Karger, M. H.; Mazur, Y. *J. Am. Chem. Soc.* **1969**, *91*, 5663.
- (34) Karger, M. H.; Mazur, Y. *J. Org. Chem.* **1971**, *36*, 532.
- (35) Beard, C. D.; Baum, K.; Grakauskas, V. *J. Org. Chem.* **1973**, *38*, 3673.

- (36) Guindon, Y.; Therien, M.; Girard, Y.; Yoakim, C. *J. Org. Chem.* **1987**, *52*, 1680.
- (37) Forbus, Jr, T. R.; Taylor, S. L.; Martin, J. C. *J. Org. Chem.* **1987**, *52*, 4156.
- (38) Ohshita, J.; Iwata, A.; Kanetani, F.; Kunai, A.; Yamamoto, Y.; Matui, C. *J. Org. Chem.* **1999**, *64*, 8024.
- (39) Iqbal, J.; Srivastava, R. R. *Tetrahedron* **1991**, *47*, 3155.
- (40) Node, M.; Kajimoto, T.; Nishide, K.; Fujita, E.; Fuji, K. *Tetrahedron Lett.* **1984**, *25*, 219.
- (41) Funk, R. L.; Mossman, C. J.; Zeller, W. E. *Tetrahedron Lett.* **1984**, *25*, 1655.
- (42) Wettstein, A.; Mischer, K. *Helv. Chim. Acta* **1942**, *25*, 718.
- (43) Godfrey, C. R. A. "Oxidation Adjacent to Oxygen of Ethers" In *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I., Ed.; Elsevier Science Ltd., 1991; Vol. 7, pp 235–250.
- (44) Courtney, J. L. "Ruthenium Tetroxide Oxidations" In *Organic Syntheses by Oxidation with Metal Compounds*; Mijs, W. J.; de Jonge, C. R. H. I., Ed.; Plenum Press: New York, 1986; pp 445–468.
- (45) Lee, J. S.; Cao, H.; Fuchs, P. L. *J. Org. Chem.* **2007**, *72*, 5820.
- (46) Freeman, F. "Oxidation by Chromium(VI) Compounds" In *Organic Syntheses by Oxidation with Metal Compounds*; Mijs, W. J.; de Jonge, C. R. H. I., Ed.; Plenum Press: New York, 1986; pp 41–118.
- (47) Lee, S.; Fuchs, P. L. *Org. Lett.* **2004**, *6*, 1437.
- (48) Wang, J.; Ma, A.; Ma, D. *Org. Lett.* **2008**, *10*, 5425.
- (49) Nicolaou, K. C.; Sarlah, D.; Shaw, D. M. *Angew. Chemie Int. Ed.* **2007**, *46*, 4708.
- (50) Daub, M. E.; Prudhomme, J.; Le Roch, K.; Vanderwal, C. D. *J. Am. Chem. Soc.* **2015**, *137*, 4912.
- (51) Franzén, J.; Marigo, M.; Fielenbach, D.; Wabnitz, T. C.; Kjærsgaard, A.; Jørgensen, K. A.

- J. Am. Chem. Soc.* **2005**, *127*, 18296.
- (52) Stevens, R. V.; Albizati, K. F. *J. Org. Chem.* **1985**, *50*, 632.
- (53) Siegel, C.; Gordon, P. M.; Uliss, D. B.; Handrick, G. R.; Dalzell, H. C.; Razdan, R. K. *J. Org. Chem.* **1991**, *56*, 6865.
- (54) Weidmann, B.; Seebach, D. *Angew. Chemie Int. Ed.* **1983**, *22*, 31.
- (55) Reetz, M. T.; Haning, H.; Stanchev, S. *Tetrahedron Lett.* **1992**, *33*, 6963.
- (56) Reetz, M. T. *Top. Curr. Chem.* **1982**, *106*, 1.
- (57) Ashby, E. C.; Laemmle, J. T. *Chem. Rev.* **1975**, *75* (4), 521.
- (58) Francisco, C. G.; Freire, R.; Rodríguez, M. S.; Suárez, E. *Tetrahedron Lett.* **1987**, *28*, 3397.
- (59) Suginome, H.; Yamada, S. *J. Org. Chem.* **1985**, *50*, 2489.
- (60) Love, B. E.; Jones, E. G. *J. Org. Chem.* **1999**, *64*, 3755.
- (61) An ungreased joint led to a decreased yield of more than 20%.
- (62) Miyaoka, H.; Okubo, Y.; Muroi, M.; Mitome, H.; Kawashima, E. *Chem. Lett.* **2011**, *40*, 246.
- (63) Enders, D.; Kipphardt, H.; Gerdes, P.; Breña-Valle, L. J.; Bhushan, V. *Bull. Soc. Chim. Belg.* **1988**, *97*, 693.
- (64) Wang, J.; Ma, A.; Ma, D. *Org. Lett.* **2008**, *10*, 5425.
- (65) Siegel, C.; Gordon, P. M.; Uliss, D. B.; Handrick, G. R.; Dalzell, H. C.; Razdan, R. K. *J. Org. Chem.* **1991**, *56*, 6865.
- (66) Stevens, R. V.; Albizati, K. F. *J. Org. Chem.* **1985**, *50*, 632.
- (67) Sato, T.; Gotoh, Y.; Watanabe, M.; Fujisawa, T. *Chem. Lett.* **1983**, 1533.

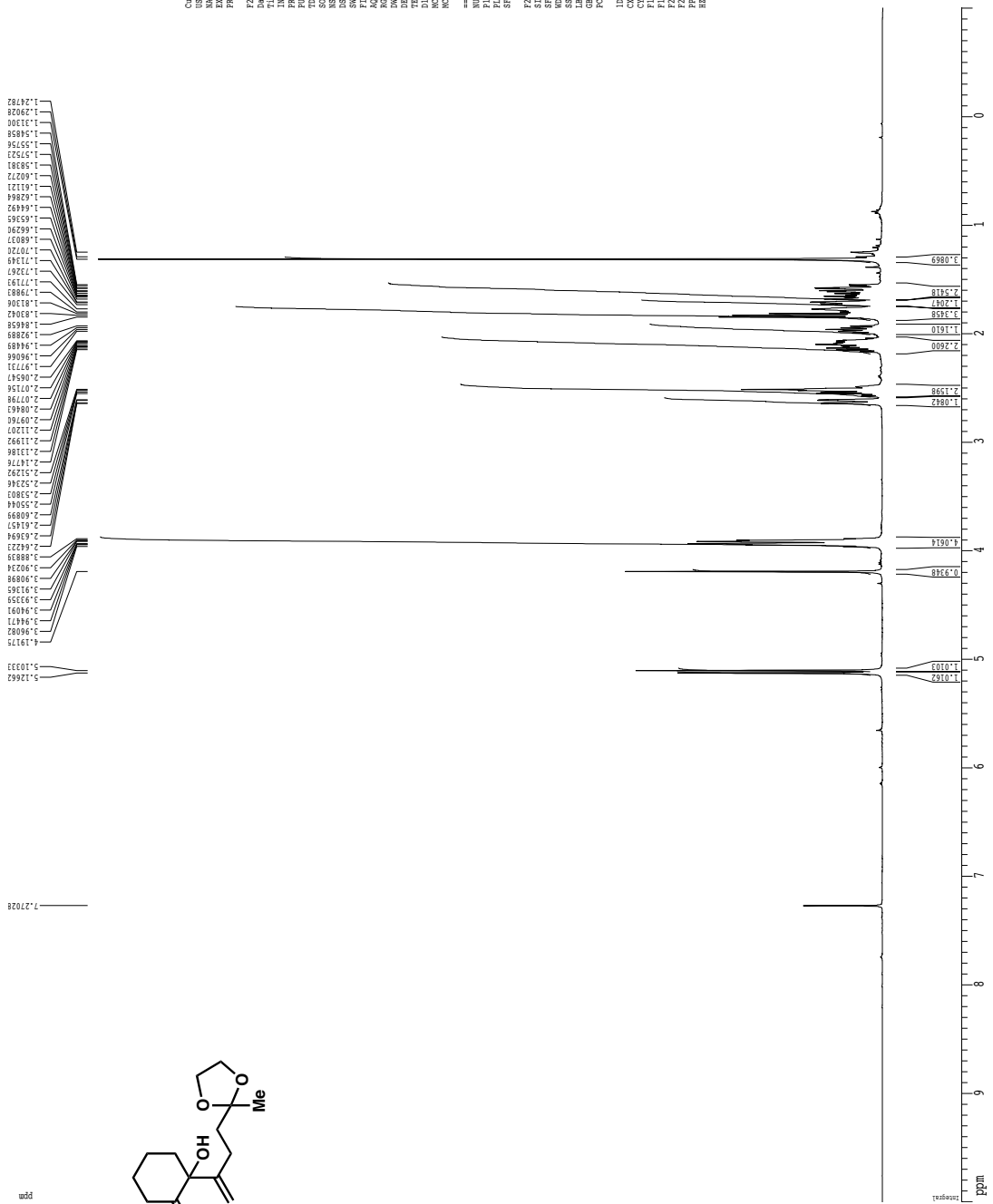
# APPENDIX A: NMR Data







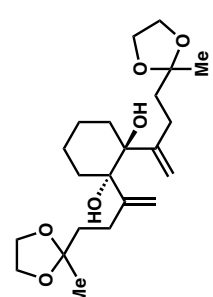
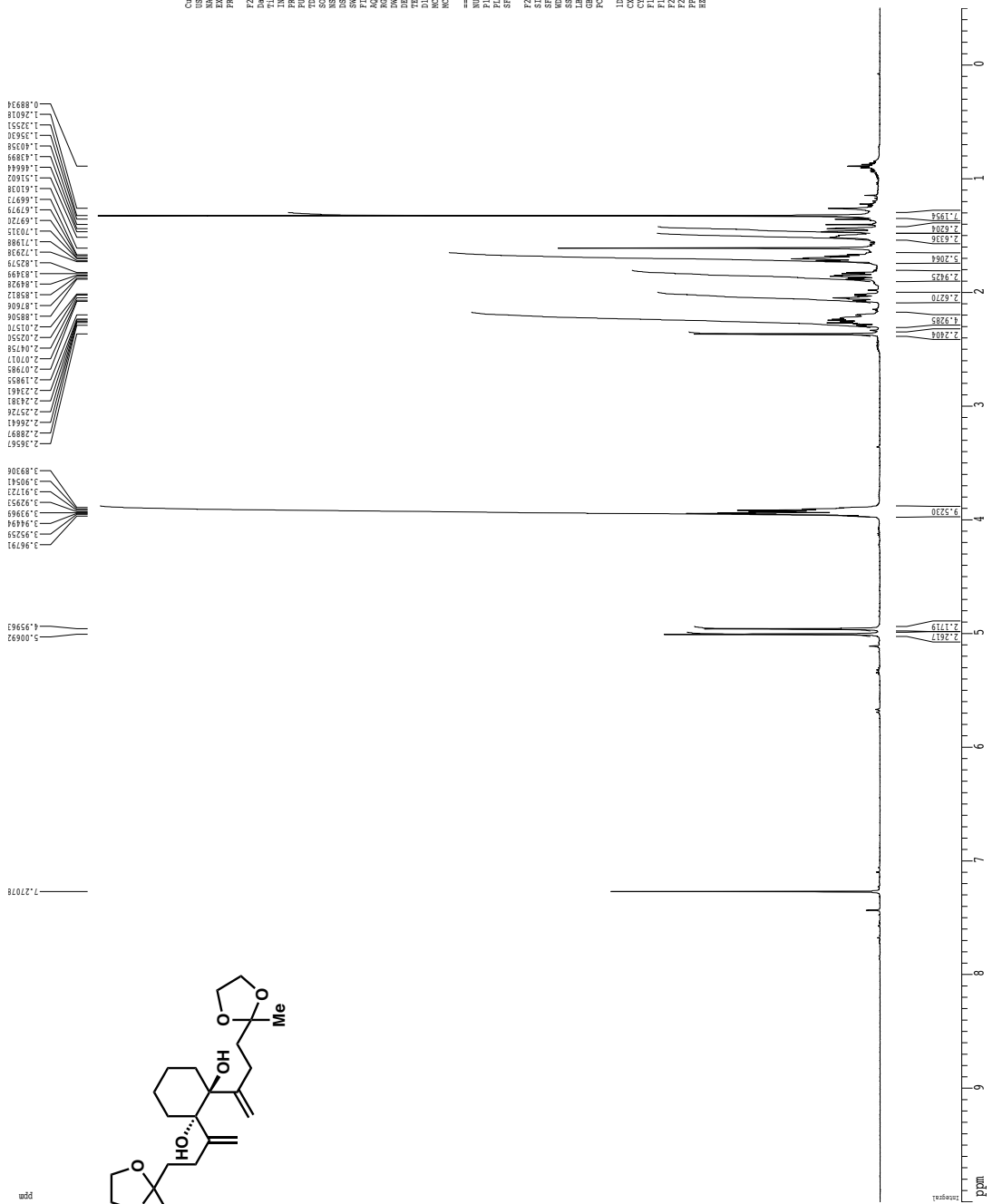
1H spectrum



Current Data Parameters  
 USER roosen  
 NAME PCL1102\_F16-21  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20120402  
 Time 13:12  
 INSTRUM crys000  
 PROBHD 5 mm CP1  
 PULPROG zg30  
 TD 16022  
 CH1INVT CH1 2  
 SS 2  
 DS 6100.2 Hz  
 FIDRES 0.375495 Hz  
 AQ 1.331804 sec  
 RG 7.1 usec  
 DE 6.00 usec  
 TE 300.2 K  
 INJECT 0.0100000 sec  
 ACQPRG 0.0100000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.225013 MHz  
 F2 - Processing parameters  
 SF 500.220166 MHz  
 EN  
 GB 0.30 Hz  
 CB 0  
 PC 4.00  
 ID NMR pipe parameters  
 CI 22.80 cm  
 FI 2.00 mm  
 FIDP 10.000 ppm  
 F1 500.220 MHz  
 F2 -500.220 MHz  
 FREQN 0.44246 ppm/cm  
 BICOE 241.3323 Hz/cm



<sup>1</sup>H spectrum

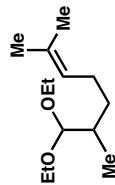
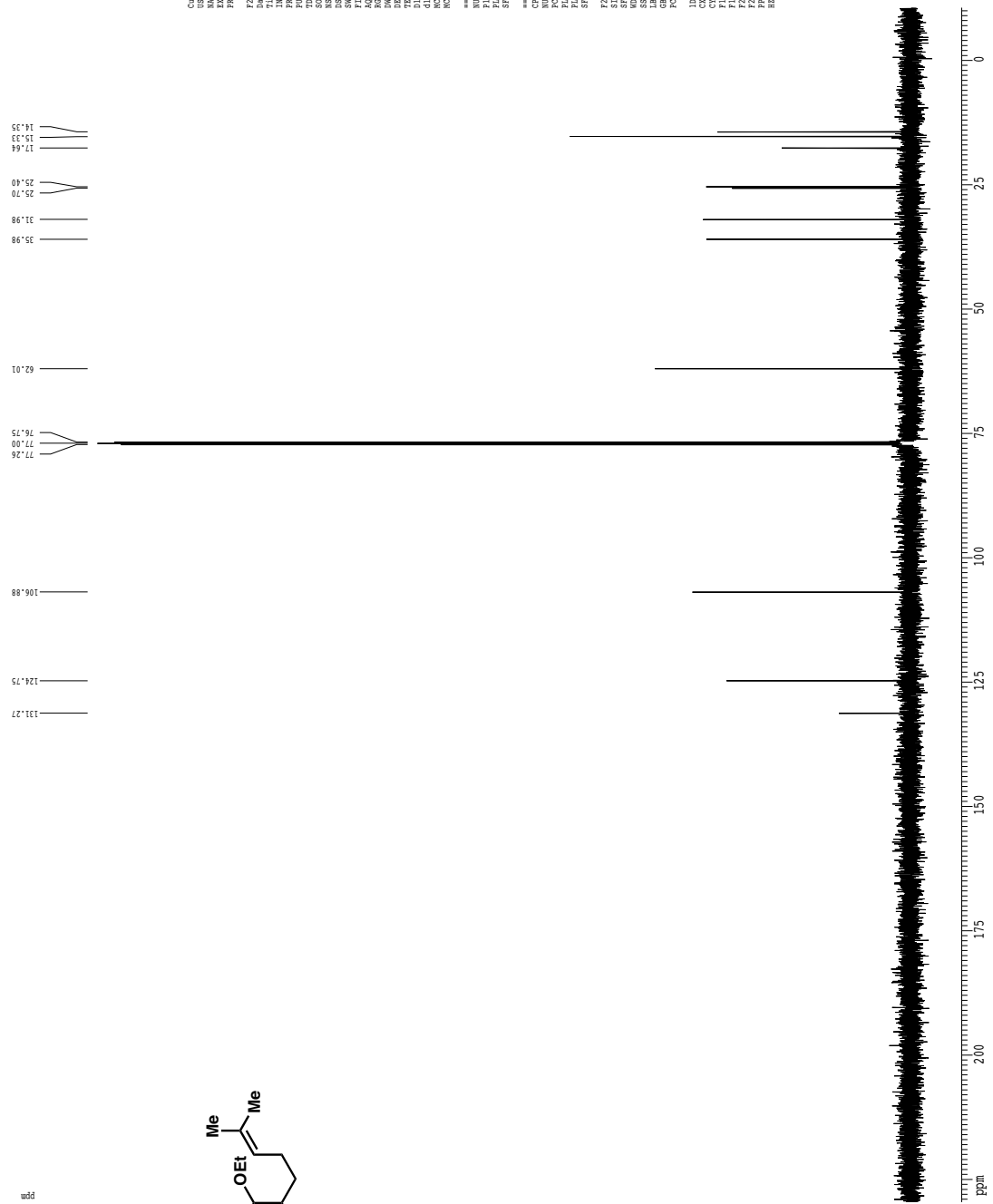


Current Data Parameters  
 USER roosen  
 NAME PCL1102\_F01-36  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2/12/04  
 Time 13:55  
 INSTRUM MPR5000  
 PULPROG zgpg30  
 TD 16322  
 SFO 500.131313  
 AQ 0.100000  
 RG 63.718  
 DE 6.00 uSBC  
 TE 300.2 K  
 ACQRES 0.100000 sec  
 NUC1 1H  
 P1 7.50 uSBC  
 PL 0.000000 sec  
 SFO1 500.225411 MHz  
 F2 - Processing parameters  
 SF 500.225411 MHz  
 EQ 0.000000 sec  
 ZS 0.30 Hz  
 GB 0  
 PC 4.00  
 ID MR parameters  
 CT 22.80 cm  
 FI 10.000 ppm  
 F1 500.220 Hz  
 F2 -250.11 Hz  
 FREQ 0.44053 ppm/cm  
 BICO 230.36450 Hz/cm



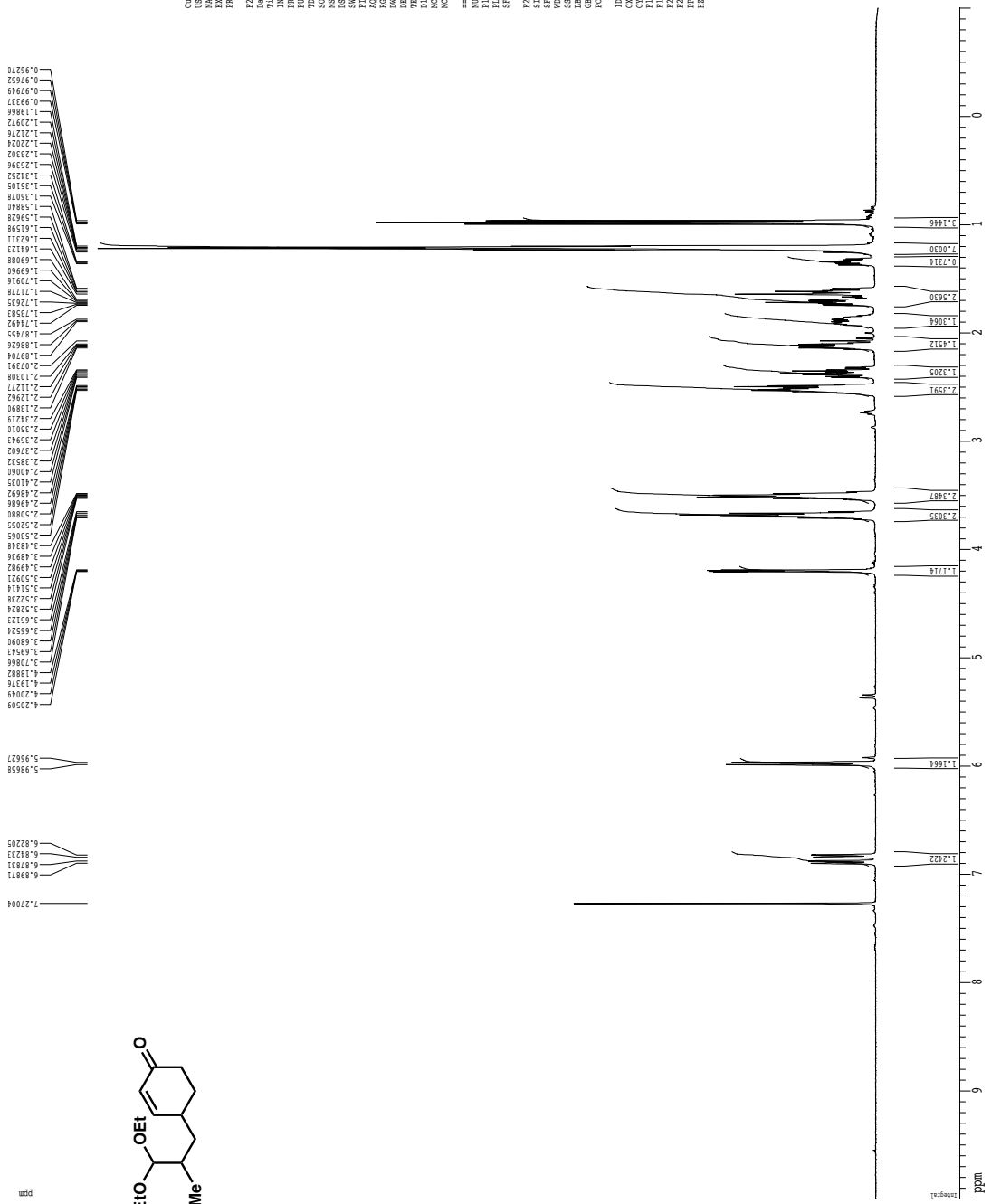


13C spectrum with 1H decoupling



Current Data Parameters  
 USES = PC2,102, distilled  
 NAME =  
 PROCNO = 1  
 F2 - Acquisition Parameters  
 Date\_ = 20101022  
 Time = 18:29  
 INSTRUM = spect  
 PROCNO = 5  
 PULPROG = zgpg30  
 AQ = 5.00  
 SOLVENT = CDCl3  
 NS = 200  
 DS = 4  
 SWH = 30303.431 Hz  
 FIDRES = 0.442388 Hz  
 AQ = 1.0000000 sec  
 RG = 327.68  
 SF = 125.760369 MHz  
 DQ = 16.500 usec  
 DE = 2.50 usec  
 TE = 300.2 K  
 D1 = 0.25000000 sec  
 d11 = 0.03000000 sec  
 DELT = 0.03000000 sec  
 MONK = 0.01500000 sec  
 ===== CHANNEL f1 =====  
 NU1 = 13C  
 P1 = 7.70 usec  
 PL1 = 0.00 dB  
 SFO1 = 125.760369 MHz  
 ===== CHANNEL f2 =====  
 CPDPRG2 = waltz1616  
 NU2 = 1H  
 P2 = 68.00 usec  
 PL2 = -2.00 dB  
 PL12 = 13.20 dB  
 SFO2 = 499.402570 MHz  
 F2 - Processing parameters  
 SI = 327.68  
 SF = 125.762370 MHz  
 WDM = 0  
 GB = 0  
 CB = 1.00 Hz  
 PC = 2.00  
 ID NMR plot parameters  
 CT = 22.00 cm  
 FIDP = 226.520 ppm  
 F1 = 28021.76 Hz  
 F2 = -1319.36 Hz  
 FREQC = 10.2747 GHz/cm  
 SFOC = 1321.91918 MHz/cm

1H spectrum

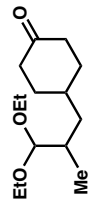
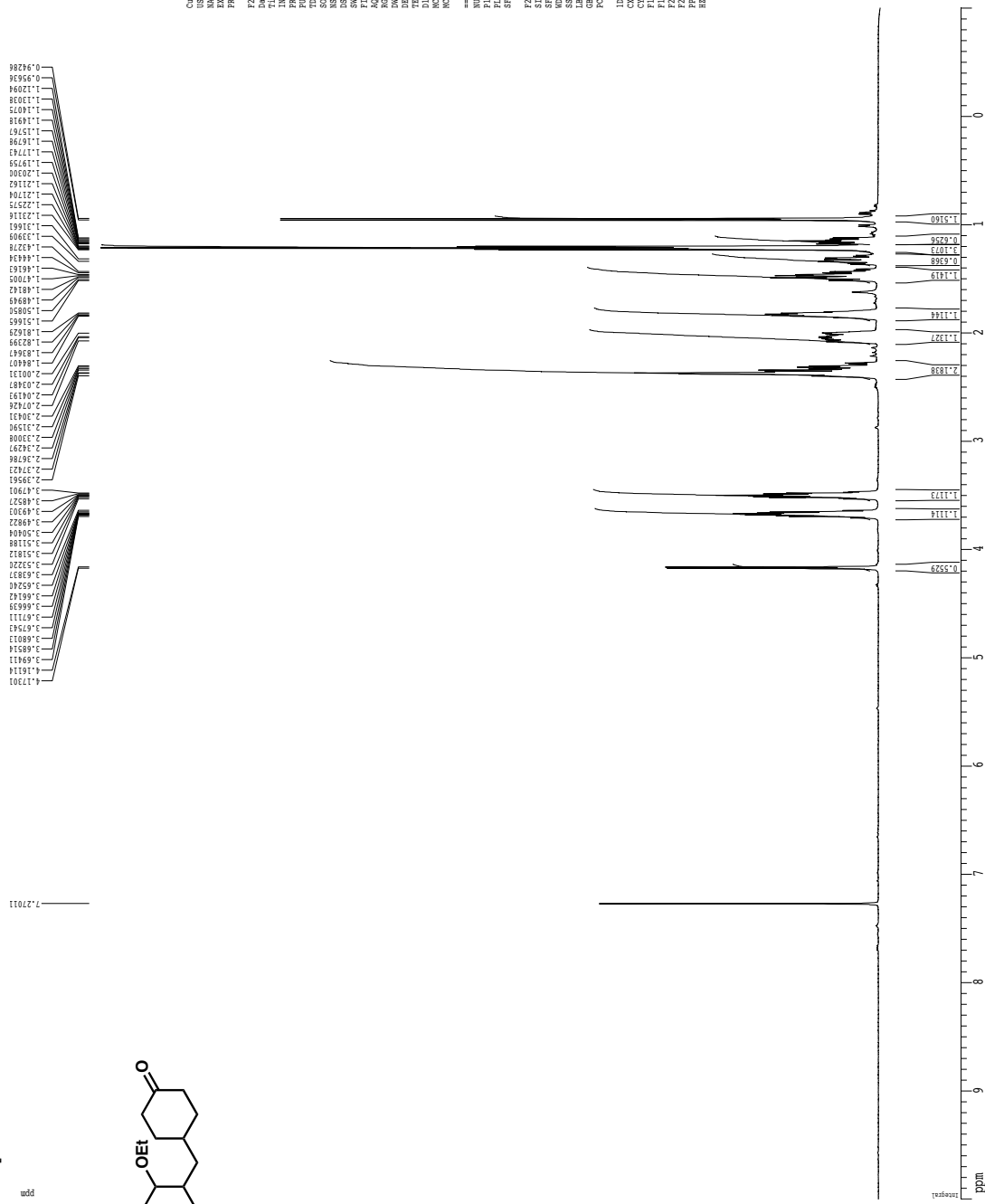


Current Data Parameters  
 USER roosen  
 NAME pct2.001\_isolate  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20100224  
 Time 10:04  
 INSTRUM crys001  
 PULPROG zgpg30  
 TD 81728  
 SFO 500.136261 MHz  
 AQ 0.16152 sec  
 RG 6.3 usec  
 DE 6.00 usec  
 TE 300.2 K  
 ACQRES 0.100000 sec  
 MCWPR 0.0100000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.1362613 MHz  
 F2 - Processing parameters  
 SF 500.1362613 MHz  
 EQ 2  
 ZS 0.30 Hz  
 GB 0  
 PC 4.00  
 ID NMR pilot parameters  
 CT 22.80 cm  
 PI 10.00 ppm  
 F1P 500.210 Hz  
 F2P 500.210 Hz  
 F3P -500.210 Hz  
 FREQCN 0.48246 ppm/cm  
 BECN 241.3423 Hz/cm



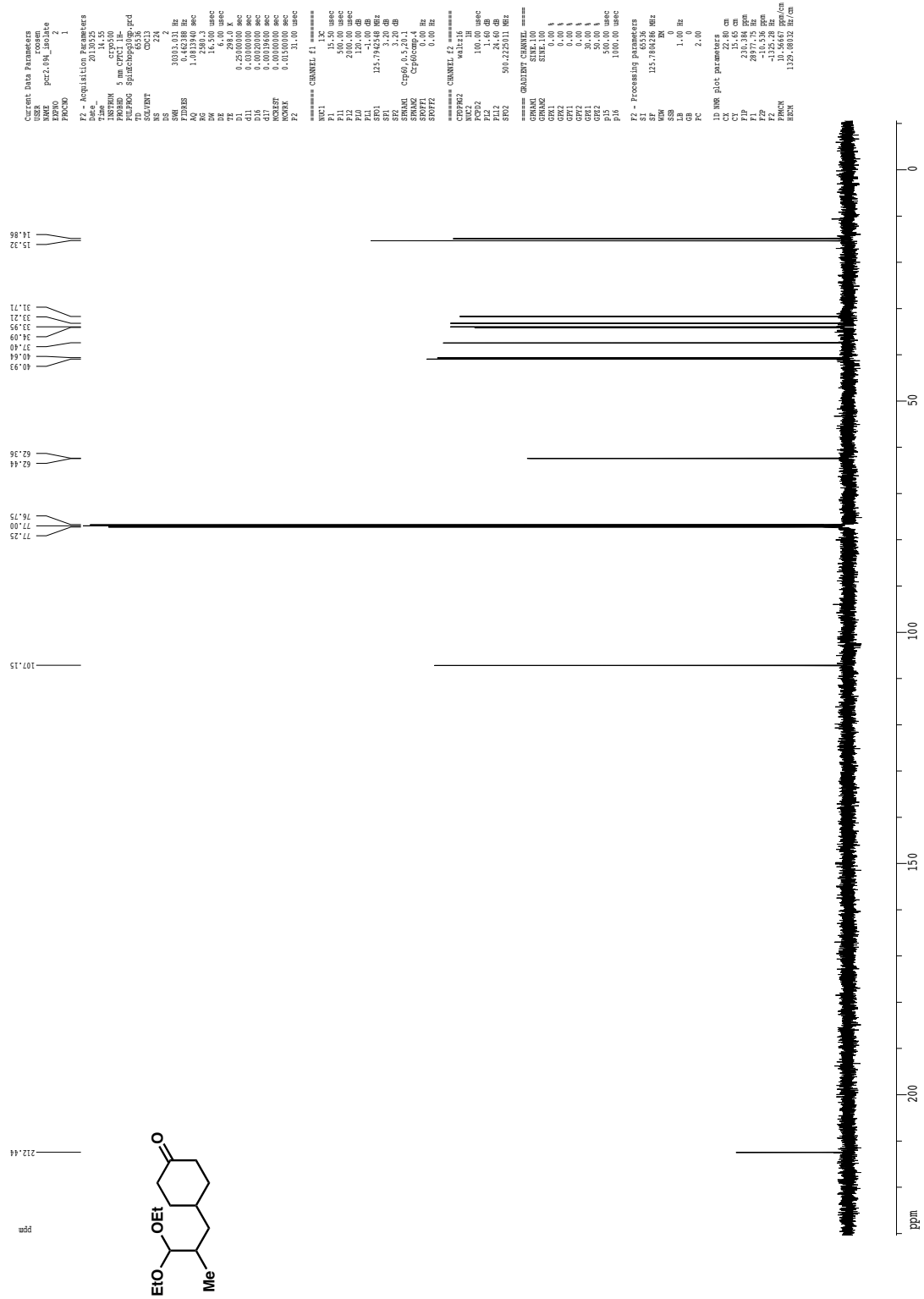


1H spectrum

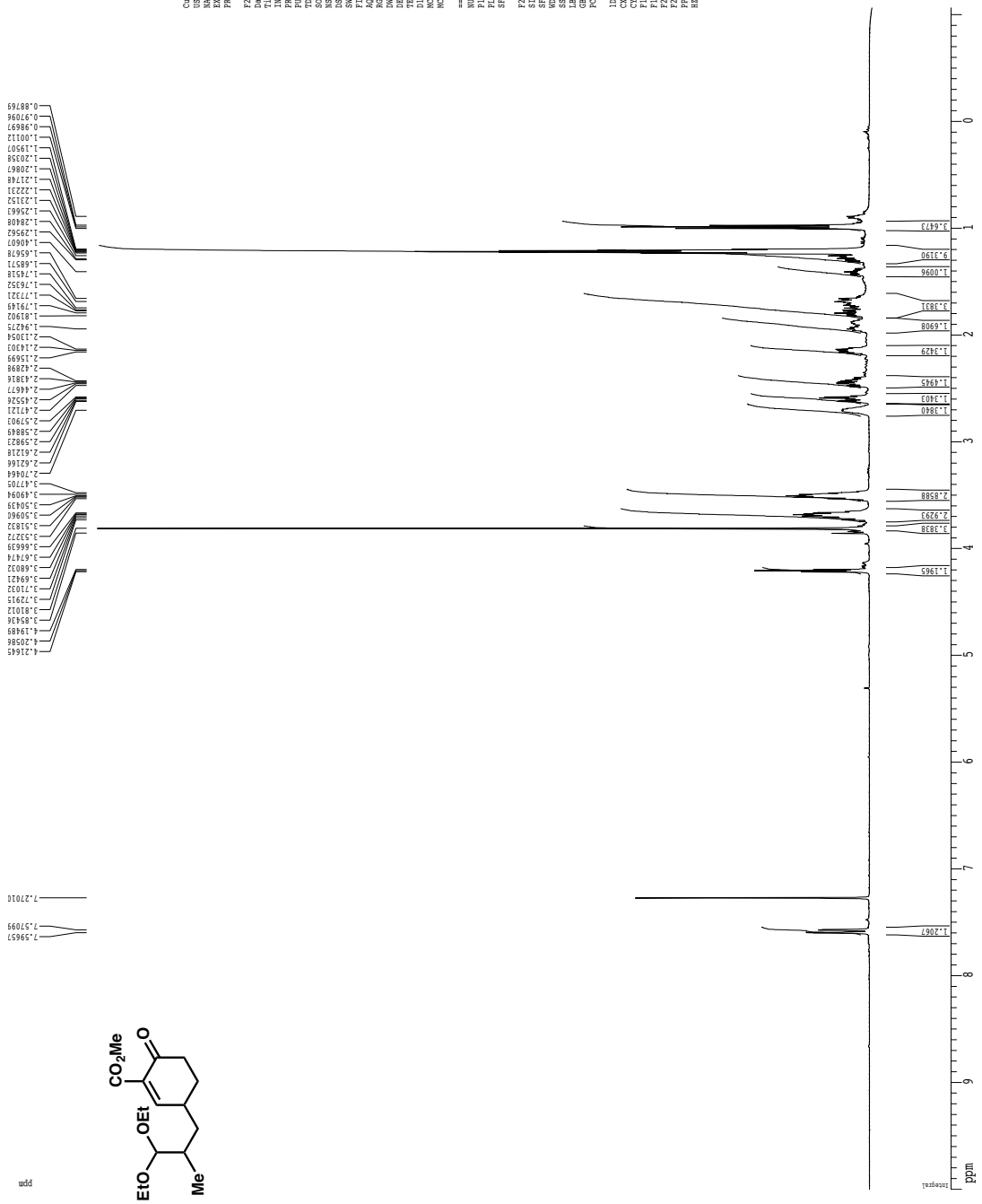


Current Data Parameters  
 USER roosen  
 NAME pct2.094\_isolate  
 PRODO 1  
 F2 - Acquisition Parameters  
 Date 2010/02/28  
 Time 14:53  
 INSTRUM avy500  
 PULPROG zgpg30  
 TD 81728  
 CH1VENT CH1  
 SI 0  
 DS 2  
 AS 0.05  
 BS 0.000000 Hz  
 AQ 5.0999774 sec  
 RG 63.71 usec  
 DE 6.00 usec  
 TE 298.0 K  
 INJECT 0.000000 sec  
 ACQPRG 0.000000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL 0.00 dB  
 FFO1 500.2254103 MHz  
 F2 - Processing parameters  
 SF 500.220562 MHz  
 EQ 0.00 usec  
 ZS 0.30 Hz  
 GB 0  
 PC 4.00  
 ID MR pulse parameters  
 CI 22.80 cm  
 CR 0.00 usec  
 F1P 10.000 ppm  
 F1 500.220 MHz  
 F2P 0.000 ppm  
 F2 -500.222 MHz  
 FREQCN 0.44246 ppm/cm  
 BPCN 241.3323 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



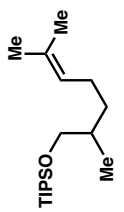
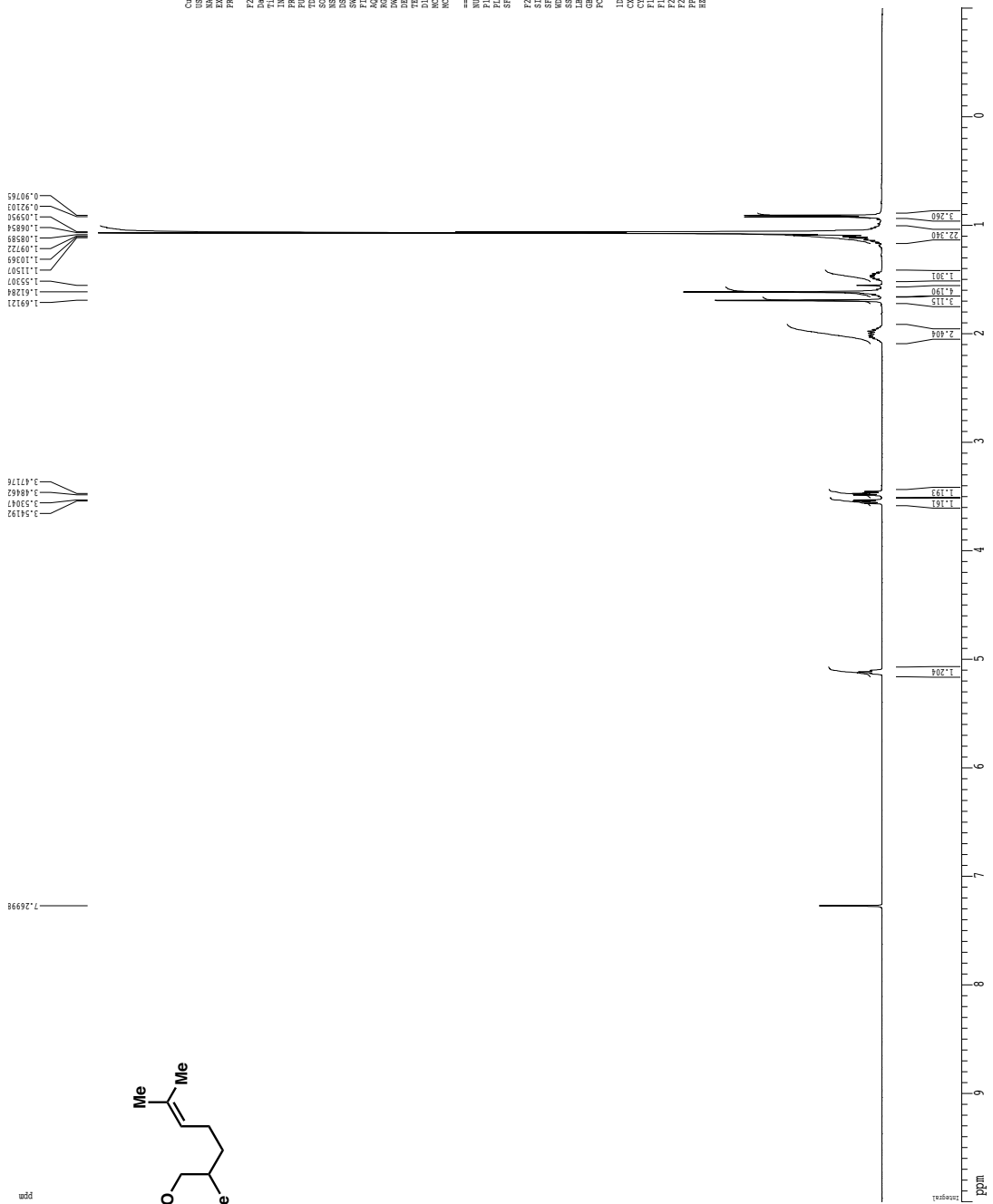
<sup>1</sup>H spectrum



Current Data Parameters  
 USER roosen  
 NAME P022.100\_0100  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20130604  
 Time 9:05  
 INSTRUM spect  
 PULPROG zgpg30  
 TD 81728  
 CH1NAME CH1  
 NS 8  
 DS 2  
 SWH 8165.2 Hz  
 FWHM 0.048643 Hz  
 AQ 5.1891774 sec  
 RG 8  
 GB 63.0 usec  
 DE 6.00 usec  
 TE 298.0 K  
 WDWEM 0.000000 sec  
 MCHWRT 0.0100000 sec  
 ===== CHANNEL f1 =====  
 NUC1 <sup>1</sup>H  
 P1 7.50 usec  
 PL1 0.00 dB  
 FFO1 500.225413 MHz  
 F2 - Processing parameters  
 SF 500.220566 MHz  
 EQ 0.00 usec  
 ZS 0.30 Hz  
 GB 0  
 PC 4.00  
 ID NMR pulse parameters  
 CE 22.80 cm  
 CR 0.00 usec  
 F1P 10.000 ppm  
 F1 500.220 MHz  
 F2 10.000 ppm  
 F2 451.45 Hz  
 FREQCN 0.44619 ppm/cm  
 BUCN 240.70378 Hz/cm



1H spectrum



Current Data Parameters  
 USER roosen  
 NAME PC22.130\_isolate  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20100828  
 Time 12:02  
 INSTRUM spect  
 PULPROG zgpg30  
 TD 65536  
 SFO 500.136260  
 AQ 0.100000  
 RG 65.711  
 FE 6.00 usec  
 TE 300.2 K  
 IC 1  
 ACQ 0.100000 sec  
 MCWRT 0.0100000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL 0.00 dB  
 SFO1 500.225413 MHz  
 F2 - Processing parameters  
 SF 500.220567 MHz  
 DS 4  
 GB 0  
 CB 0  
 PC 4.00  
 ID MRG plot parameters  
 CT 22.80 cm  
 CD 10.000 ppm  
 F1P 500.220 MHz  
 F2 500.220 MHz  
 F3 500.220 MHz  
 F4 500.220 MHz  
 FREQCN 0.44846 ppm/cm  
 BECN 241.32423 Hz/cm

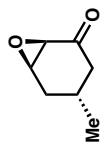
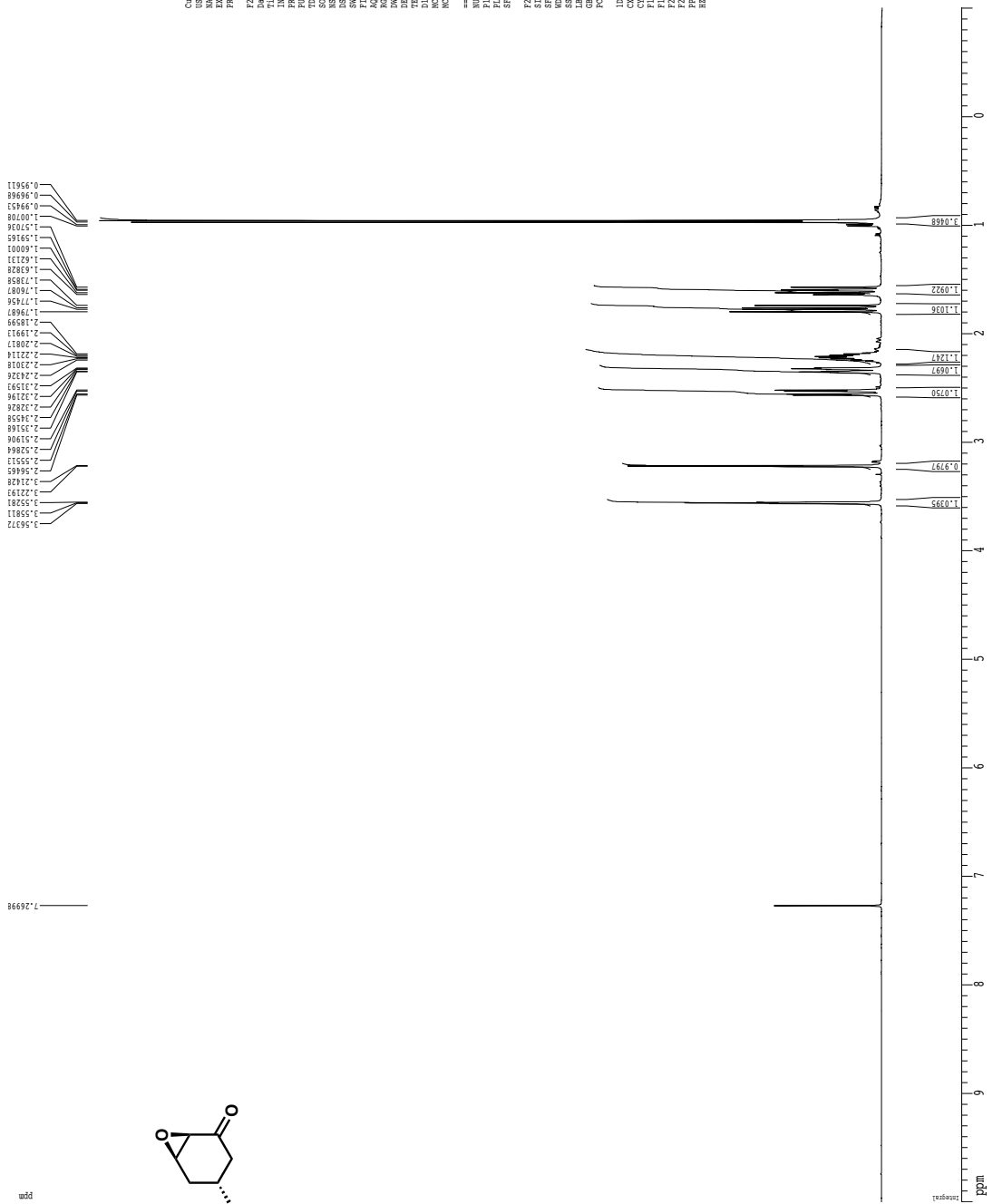






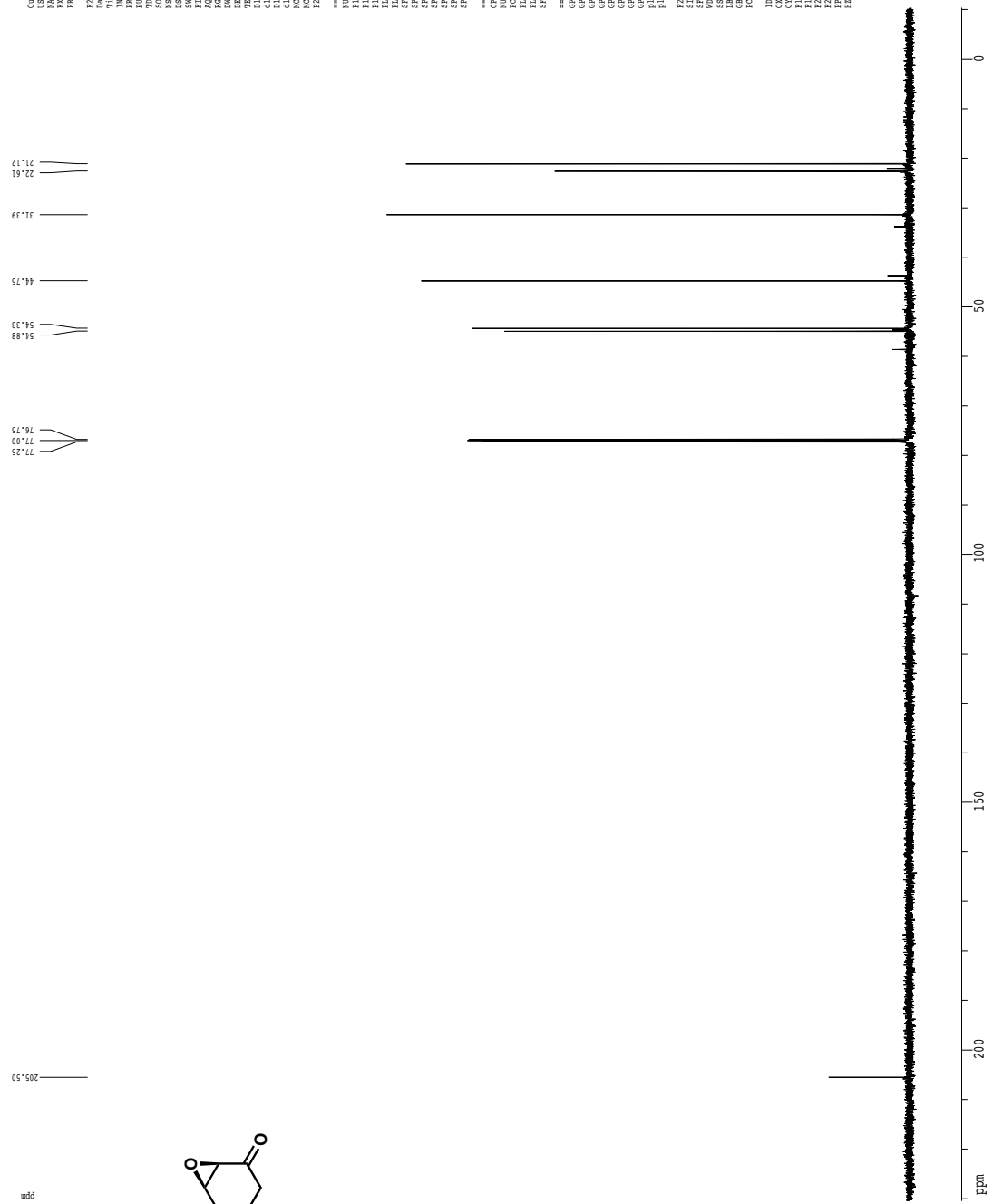


1H spectrum



Current Data Parameters  
 USER roosen  
 NAME PC22.199\_isolate  
 PROBO 1  
 F2 - Acquisition Parameters  
 Date 2/13/2026  
 Time 18:21  
 INSTRUM cryso00  
 PULPROG zgpg30  
 TD 32768  
 SFO 500.130761 MHz  
 AQ 0.250000 sec  
 RG 7.1 usec  
 DE 6.00 usec  
 TE 300.2 K  
 INJECT 0.0100000 sec  
 ACQPRG 0.0100000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.225413 MHz  
 F2 - Processing parameters  
 SF 500.225413 MHz  
 EQ 0.00 usec  
 ZS 0.30 Hz  
 GB 0  
 PC 4.00  
 ID NG parameters  
 CI 22.80 cm  
 FI 10.00 ppm  
 F1 500.225413 MHz  
 F2 -500.225413 MHz  
 FREQCN 0.44246 ppm/cm  
 BECN 241.3323 Hz/cm

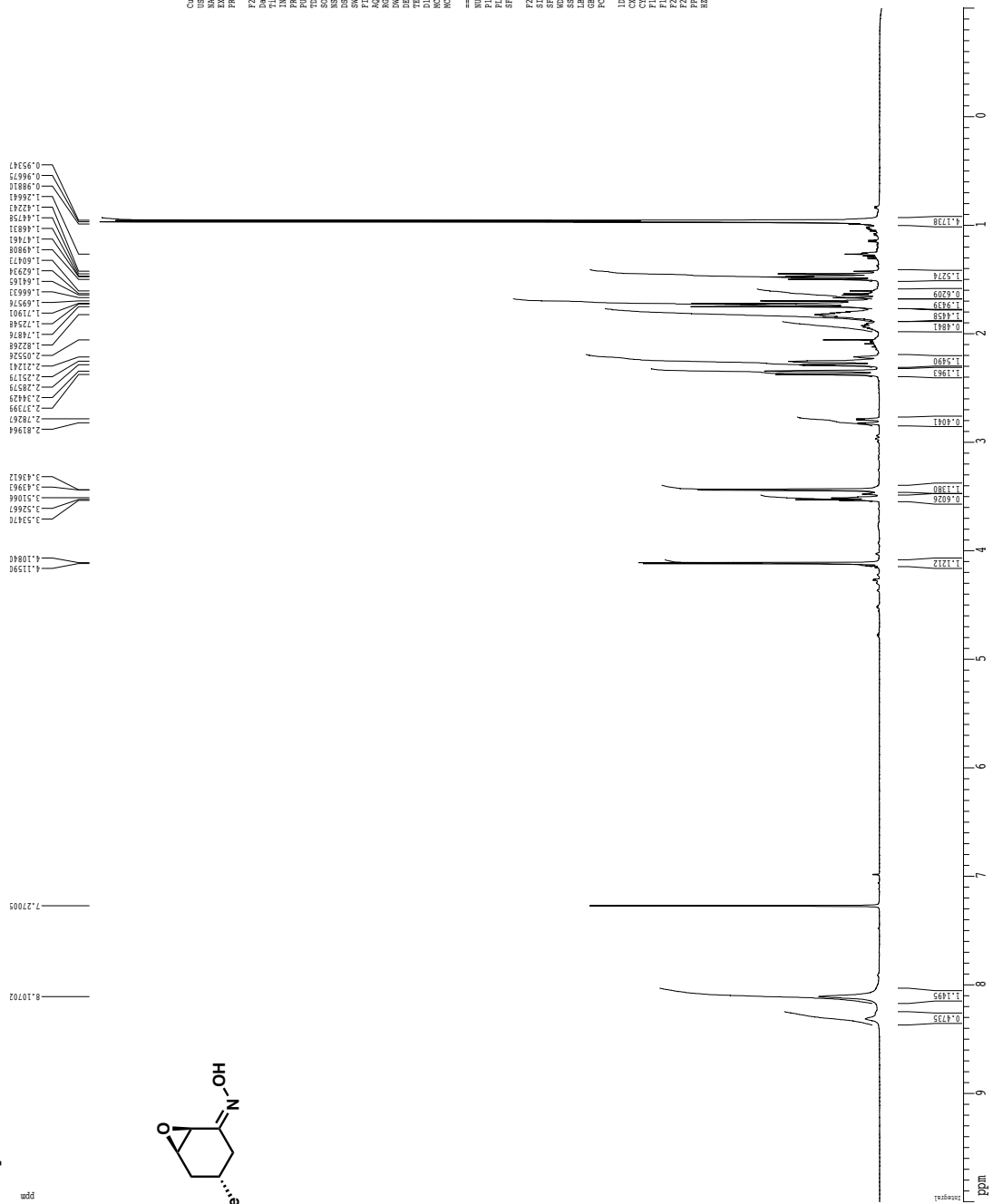
Z-restored spin-echo 13C spectrum with 1H decoupling



```

Current Data Parameters
NAME      pec2.190_1solate
PROCNO    1
F2 - Acquisition Parameters
Date_     20.10.05
Time      6.18.26
INSTRUM   spect
PROBHD    5 mm QNP1H
PULPROG   zgpg30
SOLVENT   CDCl3
SS        256
SFO1      125.762549 MHz
SF1        3.20 dB
SFRAC1    Cmp6,0.5,20.1 dB
SFRAC2    Cmp6comp,4
SFRAC3    1.0
SFRAC4    0.0
SFRAC5    0.0
===== CHANNEL f1 =====
NUC1       13C
P1         15.50 usec
PL1        0.00 dB
PC         200.00 usec
P2         120.00 dB
P3         120.00 dB
SFO2      125.762549 MHz
SF2        3.20 dB
SFRAC2    Cmp6,0.5,20.1 dB
SFRAC3    1.0
SFRAC4    0.0
SFRAC5    0.0
===== CHANNEL f2 =====
CPDPRG2   zgpg30
NUC2       1H
P2         100.00 usec
PL2        0.00 dB
PC         24.60 dB
SFO3      500.225011 MHz
===== GRABUNT CHANNEL =====
GRAB1     SINE,100
GRAB2     SINE,100
GPA1      0.00 V
GPA2      0.00 V
GPA3      0.00 V
GPA4      0.00 V
GPA5      0.00 V
GPA6      0.00 V
GPA7      0.00 V
GPA8      0.00 V
GPA9      0.00 V
GPA10     0.00 V
P16       500.00 usec
P16       1000.00 usec
F2 - Processing parameters
SI         65536
SF         125.762549 MHz
WDW        EM
SSB        0
GB         1.00 Hz
PC         2.00
=====
1D NMR plot parameters
CK         22.80 cm
SI         32768
SF         200.630 MHz
F1         29109.68 Hz
F2         110.80 Hz
F3         110.80 Hz
FREQN     10.56488 GHz/cm
BUCN      1329.10719 Hz/cm
    
```

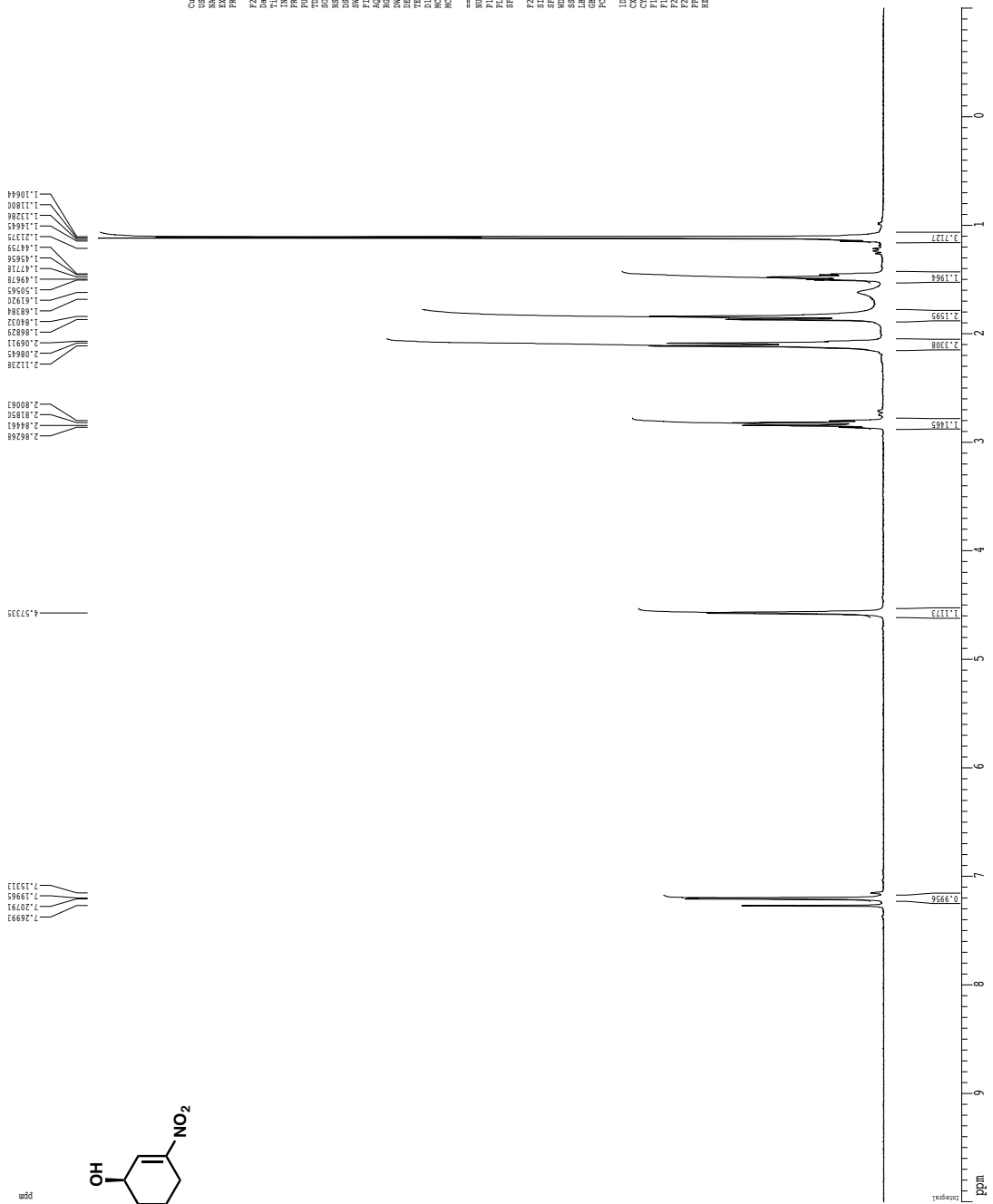
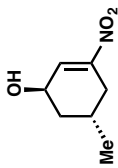
<sup>1</sup>H spectrum





<sup>1</sup>H spectrum

ppm



Current Data Parameters  
USER: pcoc08  
NAME: pcz2.177\_isolate  
EXPNO: 1  
PROCNO: 1

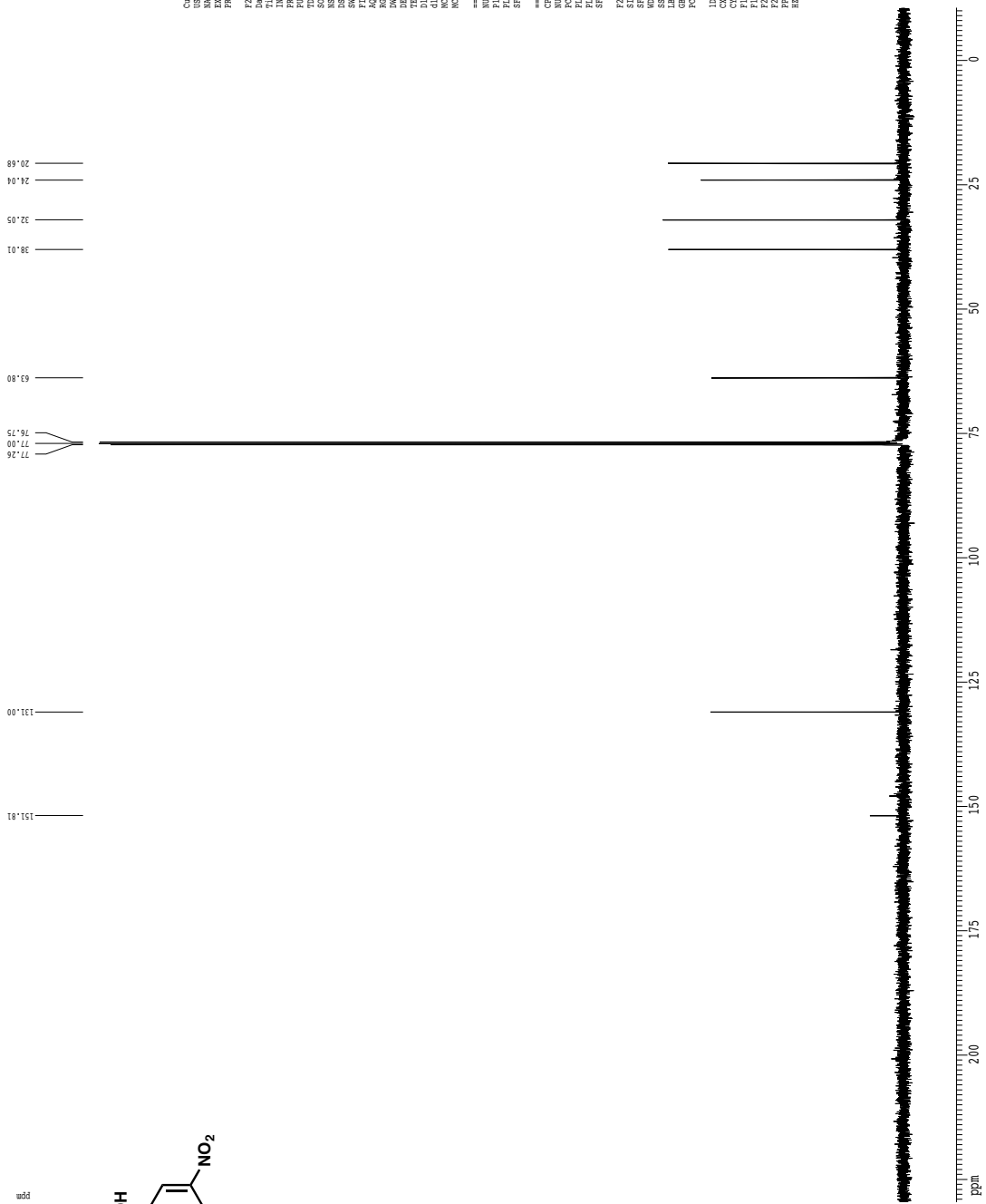
F2 - Acquisition Parameters  
Date\_ Time: 20100112 12:34  
INSTRUM: spect  
PROBHD: 5mm broadbnd  
PULPROG: zgpg30  
TD: 327488  
SOLVENT: CDCl3  
DS: 2  
SWH: 8012.820 Hz  
FIDRES: 0.15690 Hz  
AQ: 1.9998451 sec  
RG: 812.7  
AQ: 62.400 usec  
TE: 298.2 K  
D1: 0.1000000 sec  
DELTA: 0.0500000 sec  
ACQSK: 0.0500000 sec

===== CHANNEL f1 =====  
NUC1: 1H  
P1: 12.20 usec  
PL1: -2.00 dB  
SFO1: 499.9509500 MHz

F2 - Processing parameters  
SI: 32768  
SF: 499.9509500 MHz  
WDW: EM  
SSB: 0  
CB: 0  
PC: 1.00

ID NMR file parameters  
CX: 22.80 cm  
CT: 5.00 cm  
CF: 10.00 cm  
FI: 499.950 Hz  
FZ: -1.000 ppm  
PRGCM: 0.46246 ppm/cm  
BRDCM: 244.88556 Hz/cm

<sup>13</sup>C spectrum with <sup>1</sup>H decoupling



```

Current Data Parameters
=====
NAME      PC22.177_isolate
EXPNO     2
PROCNO    1

F2 - Acquisition Parameters
=====
Date_     20080817
Time      12.38
INSTRUM   gp500
PROBHD    5 mm Broadband
PULPROG   zgpg30
TD         65536
SOLVENT   CDCl3
DS         2
SH1       3003.001 Hz
NUC1       13C
NUC2       13C
RG         1.011848 e6
AQ         6.912 e6
RG         16.500 usec
TE         298.2 K
D1         0.2200000 sec
DELTA     0.0100000 sec
INTEGR    0.0100000 sec
SFO1      125.7603000 MHz
===== CHANNEL f1 =====
NUC1       13C
P1         7.70 usec
PL1       0.00 dB
SFO1      125.7603000 MHz
===== CHANNEL f2 =====
NUC2       1H
P2         161.10 usec
PL2       19.00 dB
SFO2      500.1360000 MHz
===== CHANNEL f3 =====
P3         80.00 usec
PL3       19.00 dB
SFO3      125.7603000 MHz
===== CHANNEL f4 =====
P4         499.4924864 usec
PL4       19.00 dB
SFO4      499.4924864 MHz
F2 - Processing parameters
=====
SI         65536
SF         125.7603000 MHz
WDW        EM
SSB        0
LB         1.00 Hz
GB         0
PC         2.00

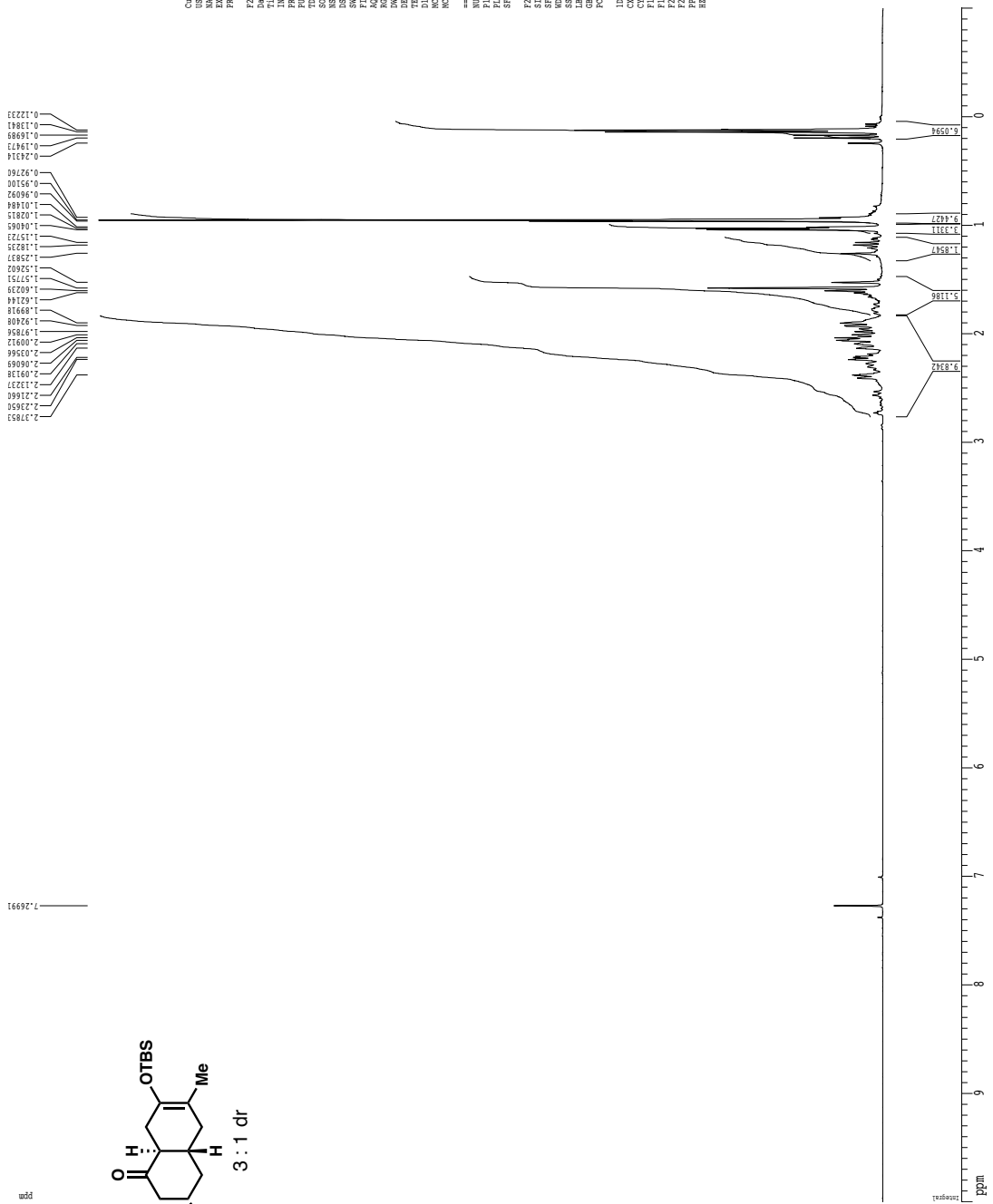
ID parameters
=====
CT         154.65 cm
PIF        229.520 ppm
PIB        15.150 ppm
PI         131.000 ppm
F2         -131.000 Hz
F3         131.000 Hz
F4         131.000 Hz
F5         131.000 Hz
F6         131.000 Hz
F7         131.000 Hz
F8         131.000 Hz
F9         131.000 Hz
F10        131.000 Hz
F11        131.000 Hz
F12        131.000 Hz
F13        131.000 Hz
F14        131.000 Hz
F15        131.000 Hz
F16        131.000 Hz
F17        131.000 Hz
F18        131.000 Hz
F19        131.000 Hz
F20        131.000 Hz
F21        131.000 Hz
F22        131.000 Hz
F23        131.000 Hz
F24        131.000 Hz
F25        131.000 Hz
F26        131.000 Hz
F27        131.000 Hz
F28        131.000 Hz
F29        131.000 Hz
F30        131.000 Hz
F31        131.000 Hz
F32        131.000 Hz
F33        131.000 Hz
F34        131.000 Hz
F35        131.000 Hz
F36        131.000 Hz
F37        131.000 Hz
F38        131.000 Hz
F39        131.000 Hz
F40        131.000 Hz
F41        131.000 Hz
F42        131.000 Hz
F43        131.000 Hz
F44        131.000 Hz
F45        131.000 Hz
F46        131.000 Hz
F47        131.000 Hz
F48        131.000 Hz
F49        131.000 Hz
F50        131.000 Hz
F51        131.000 Hz
F52        131.000 Hz
F53        131.000 Hz
F54        131.000 Hz
F55        131.000 Hz
F56        131.000 Hz
F57        131.000 Hz
F58        131.000 Hz
F59        131.000 Hz
F60        131.000 Hz
F61        131.000 Hz
F62        131.000 Hz
F63        131.000 Hz
F64        131.000 Hz
F65        131.000 Hz
F66        131.000 Hz
F67        131.000 Hz
F68        131.000 Hz
F69        131.000 Hz
F70        131.000 Hz
F71        131.000 Hz
F72        131.000 Hz
F73        131.000 Hz
F74        131.000 Hz
F75        131.000 Hz
F76        131.000 Hz
F77        131.000 Hz
F78        131.000 Hz
F79        131.000 Hz
F80        131.000 Hz
F81        131.000 Hz
F82        131.000 Hz
F83        131.000 Hz
F84        131.000 Hz
F85        131.000 Hz
F86        131.000 Hz
F87        131.000 Hz
F88        131.000 Hz
F89        131.000 Hz
F90        131.000 Hz
F91        131.000 Hz
F92        131.000 Hz
F93        131.000 Hz
F94        131.000 Hz
F95        131.000 Hz
F96        131.000 Hz
F97        131.000 Hz
F98        131.000 Hz
F99        131.000 Hz
F100       131.000 Hz
=====
  
```





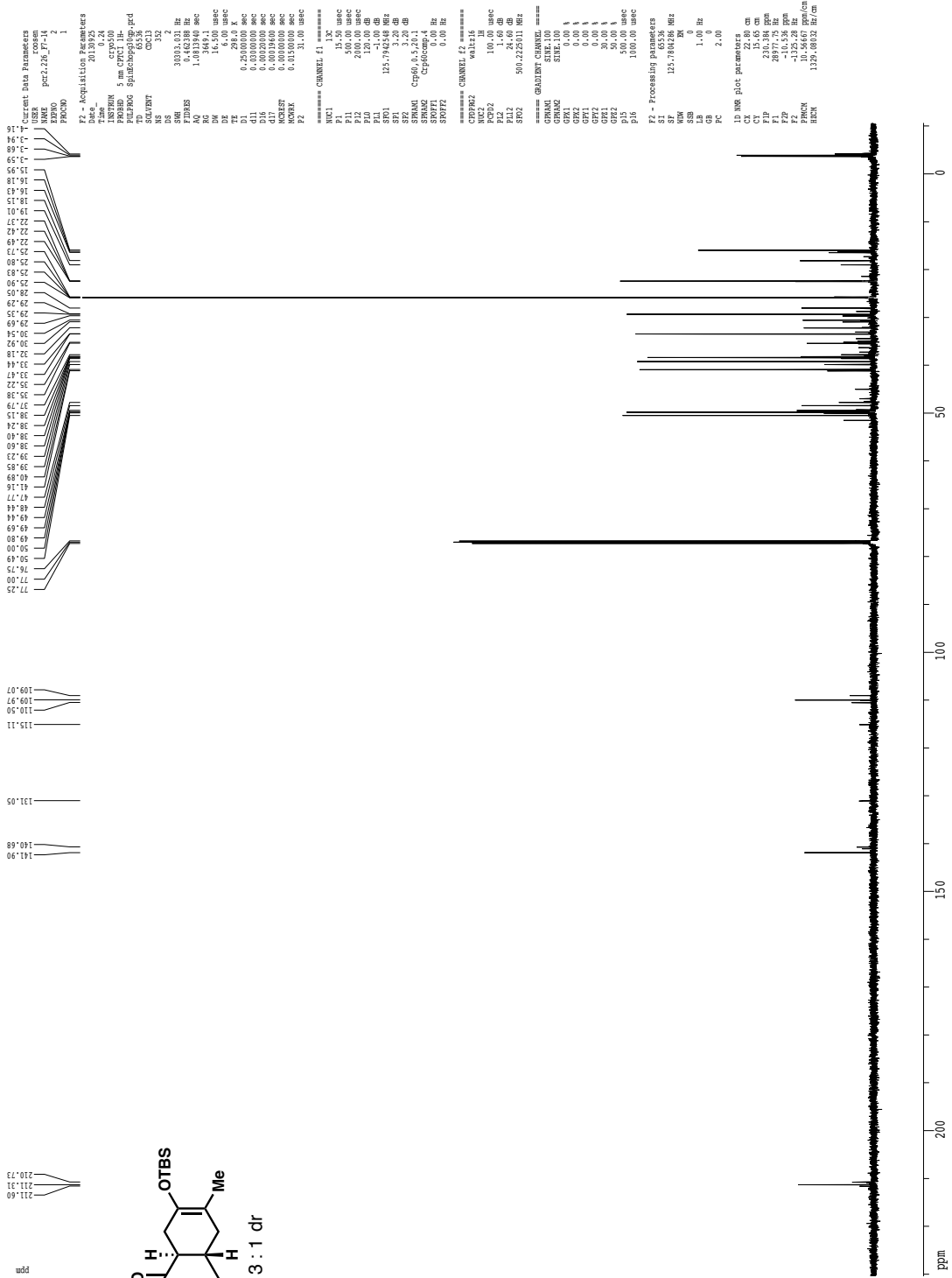
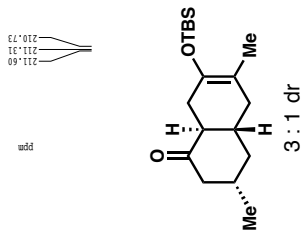


1H spectrum

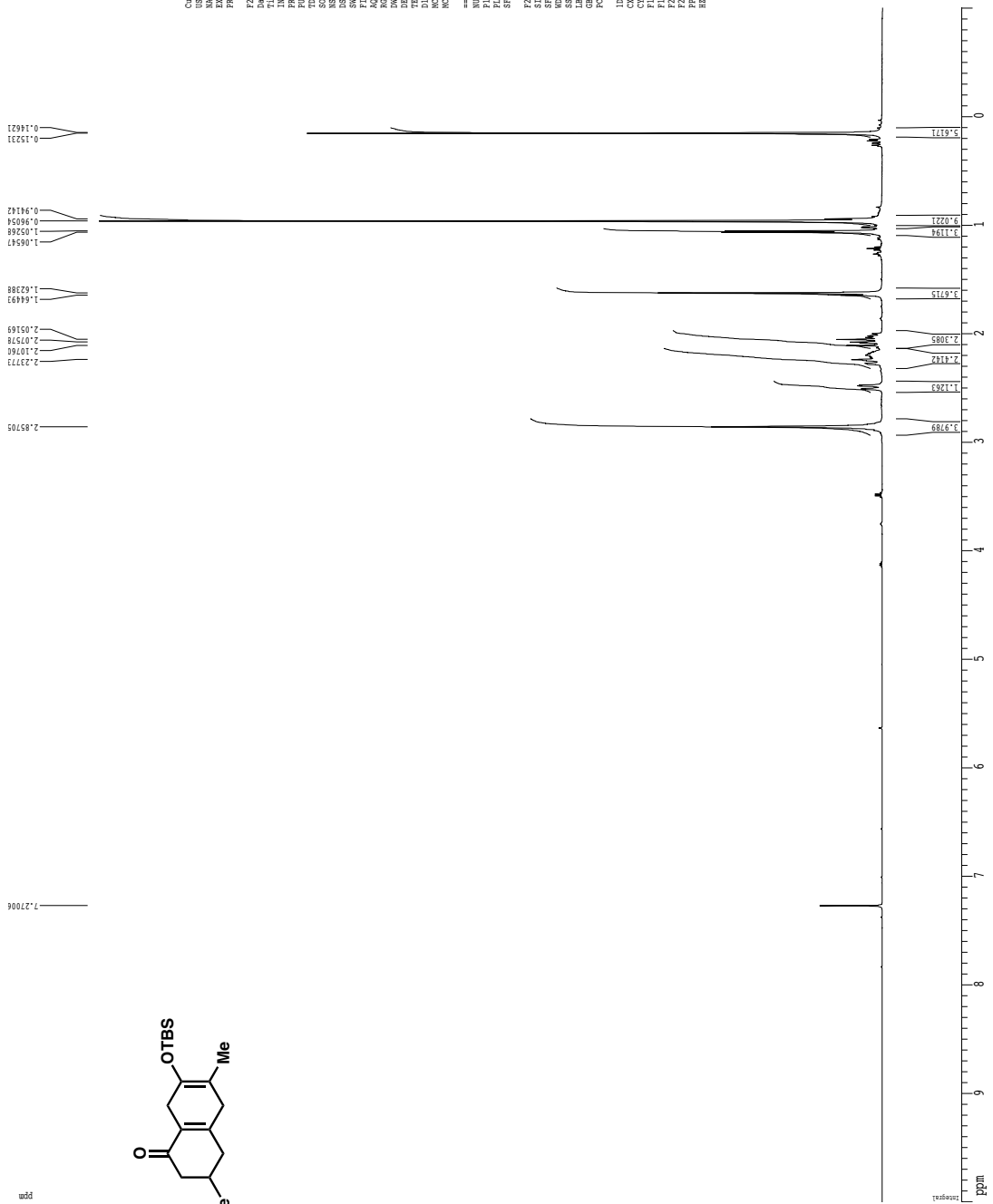


Current Data Parameters  
 USER roosen  
 NAME P02.122.F-14  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20100925  
 Time 0.22  
 INSTRUM EYEN500  
 PULPROG zgpg30  
 TD 32768  
 CHAN1ENT CHAN1  
 NS 16  
 DS 2  
 SFO 500.136261 MHz  
 AQ 0.254026 sec  
 RG 1.9998451 sec  
 BG 3.4  
 DE 6.00 usec  
 TE 298.0 K  
 WALTZ16 0.0100000 sec  
 WALTZ17 0.0100000 sec  
 WALTZ18 0.0100000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.225413 MHz  
 F2 - Processing parameters  
 SF 500.2201657 MHz  
 EQN EN  
 ZS 0.30 Hz  
 GB 0  
 PC 4.00  
 ID NMR pipe parameters  
 CT 22.80 cm  
 PI 1.60 mm  
 FIDP 10.000 ppm  
 F1 500.220166 MHz  
 F2 -500.220166 MHz  
 FREQCN 0.44246 ppm/cm  
 BECN 241.32423 Hz/cm

z-restored spin-echo 13C spectrum with 1H decoupling



<sup>1</sup>H spectrum



Current Data Parameters  
 USER roosen  
 NAME PC2.108\_isolate  
 NO 1  
 PRODO 1

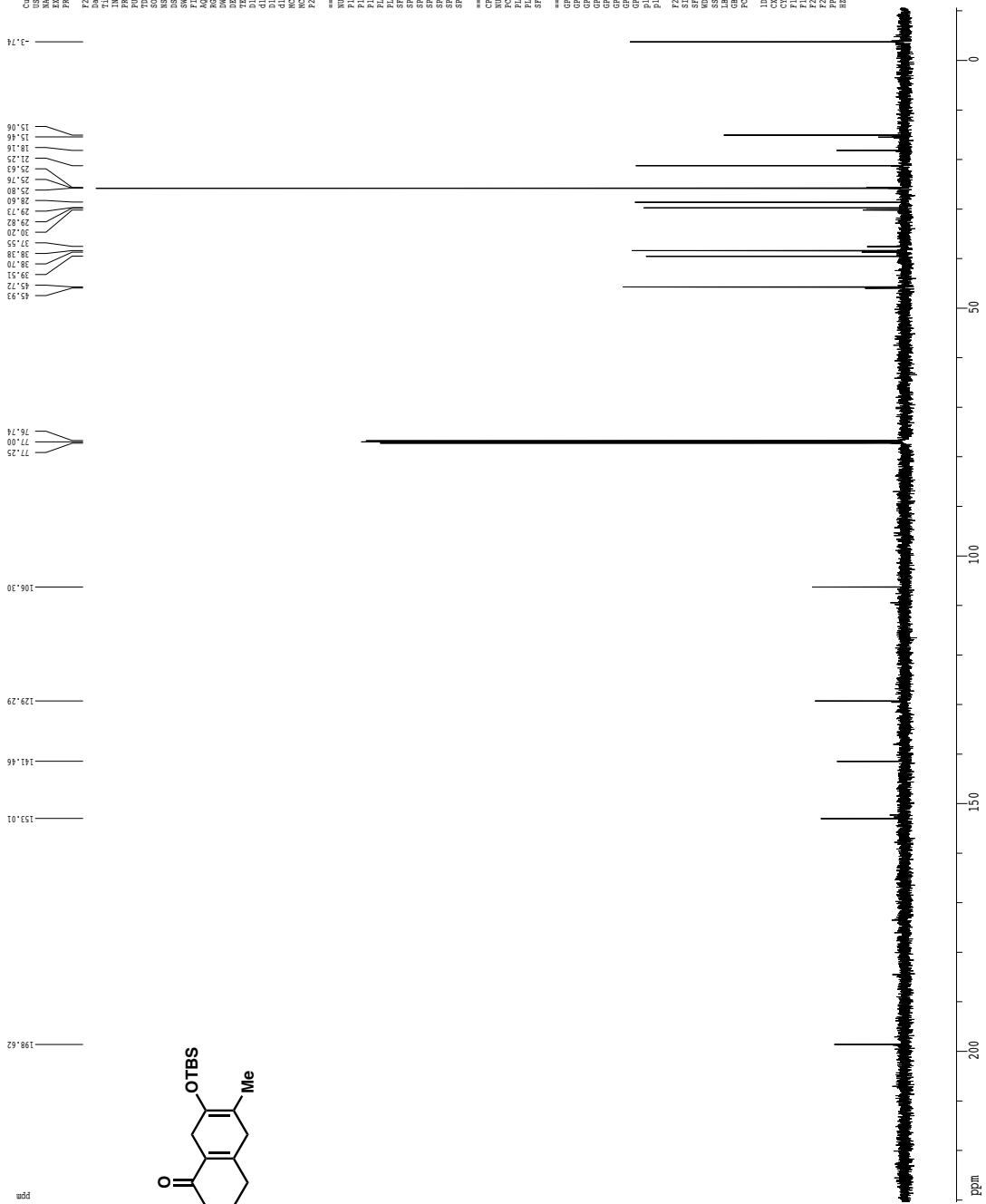
F2 - Acquisition Parameters  
 Date\_ 20100825  
 Time 2:23  
 INSTRUM cryo500  
 PULPROG zgpg30  
 TD 32768  
 CH2SOLVENT CDCl3  
 NS 16  
 DS 2  
 SWH 8163.2 Hz  
 FWHM 0.254025 Hz  
 AQ 1.999451 sec  
 RG 63.7  
 EQ 6.00 usec  
 TE 298.0 K  
 KW 0.100000 sec  
 MCKEY 0.000000 sec  
 MCHWK 0.0100000 sec

===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.225413 MHz

F2 - Processing parameters  
 SF 500.220371 MHz  
 EQ2 0.00 usec  
 ZS 0.30 Hz  
 GB 0  
 PC 4.00

1D NMR pilot parameters  
 CT 22.80 cm  
 CP 10.00 usec  
 F1P 10.000 ppm  
 F1 500.220 MHz  
 F2 500.220 MHz  
 F2P -500.220 ppm  
 FREQC 0.48246 ppm/cm  
 BECN 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



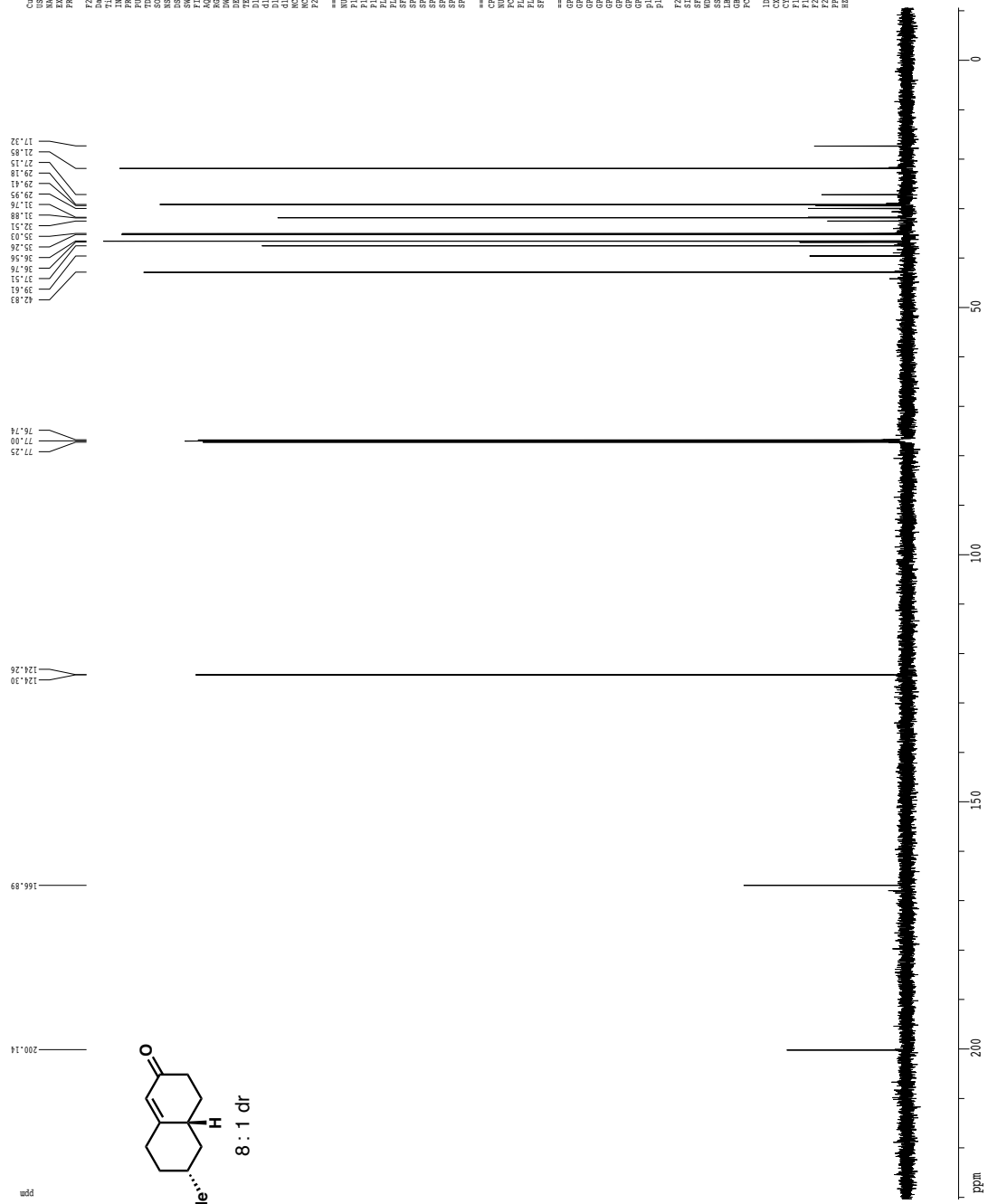
```

Current Data Parameters
NAME      pec2.108_1solate
PROCNO    1
=====
F2 - Acquisition Parameters
Date_     20.10.05
Time      6.26
INSTRUM   spect
PROBHD    5 mm QNP1H-
PULPROG   zgpg30
=====
NUC1       13C
NUC2       1H
=====
===== CHANNEL f1 =====
NUC1       13C
P1         15.50 usec
PL1        0.00 dB
PC         200.00 usec
=====
===== CHANNEL f2 =====
CPDPRG2   zgpg30
NUC2       1H
P2         10.00 usec
PL2        0.00 dB
PC         24.60 dB
=====
===== GRABENT CHANNEL =====
GRAB1      SINE
GRAB2      SINE
GRAB3      0.00 A
GRAB4      0.00 A
GRAB5      0.00 A
GRAB6      0.00 A
GRAB7      0.00 A
GRAB8      30.00 A
GRAB9      30.00 A
GRAB10     30.00 A
GRAB11     30.00 A
GRAB12     500.00 usec
GRAB13     1000.00 usec
=====
F2 - Processing parameters
SI         65536
SF         125.760420 MHz
WDW        EM
SSB        0
GB         1.00 Hz
PC         2.00
=====
10 NMR plot parameters
CX         22.80 cm
CY         13.80 cm
CZ         23.80 cm
F1         28977.29 Hz
F2         115.50 Hz
FREQN1     10.56467 GHz/cm
FREQN2     1329.0032 Hz/cm

```



Z-restored spin-echo <sup>13</sup>C spectrum with <sup>1</sup>H decoupling



```

Current Data Parameters
NAME      pec2230_F15-27
PROCNO    1
=====
F2 - Acquisition Parameters
Date_     2013/09/25
Time      19:13
INSTRUM   spect
PROBHD    5 mm QNP1H
PULPROG   zgpg30
=====
SOLVENT   CDCl3
NS        200
DS        4
SWH        30303.431 Hz
FIDRES     0.442388 Hz
AQ         1.003304 sec
RG         655
AQ1        16.500 usec
DE         6.00 usec
TE         300.2 K
D1         0.50000000 sec
d11        0.03000000 sec
d12        0.03000000 sec
d13        0.00019600 sec
d17        0.00000000 sec
NOEPRG1   0.00000000 sec
NUC1PROG  13C
NUC1       13C
=====
===== CHANNEL f1 =====
P1         15.50 usec
PL1        0.00 dB
PC1        200.00 usec
P2         120.00 usec
PL2        0.00 dB
PC2        120.00 usec
=====
SFO1       125.762549 MHz
SFO2       500.136490 MHz
SFO3       3.20 MHz
SFO4       500.136490 MHz
SFO5       500.136490 MHz
SFO6       500.136490 MHz
SFO7       500.136490 MHz
SFO8       500.136490 MHz
SFO9       500.136490 MHz
SFO10      500.136490 MHz
===== CHANNEL f2 =====
CPDPRG2   zgpg30
NUC2PROG  1H
NUC2       1H
P2         10.00 usec
PL2        0.00 dB
PC2        24.60 usec
SFO2       500.136490 MHz
===== GRADIENT CHANNEL =====
GRAD1     SIB
SIB1       0.00 usec
SIB2       0.00 usec
SIB3       0.00 usec
SIB4       0.00 usec
SIB5       0.00 usec
SIB6       0.00 usec
SIB7       0.00 usec
SIB8       0.00 usec
SIB9       0.00 usec
SIB10      0.00 usec
=====
F2 - Processing parameters
SI         65536
SF         125.762549 MHz
WDW        EM
SSB        0
GB         0
PC         2.00
=====
1D NMR plot parameters
CK         22.80 cm
SI         32768
SF         125.762549 MHz
F1         28875.80 Hz
F2         110.531 Hz
F3         110.531 Hz
F4         110.531 Hz
F5         110.531 Hz
F6         110.531 Hz
F7         110.531 Hz
F8         110.531 Hz
F9         110.531 Hz
F10        110.531 Hz
=====
  
```

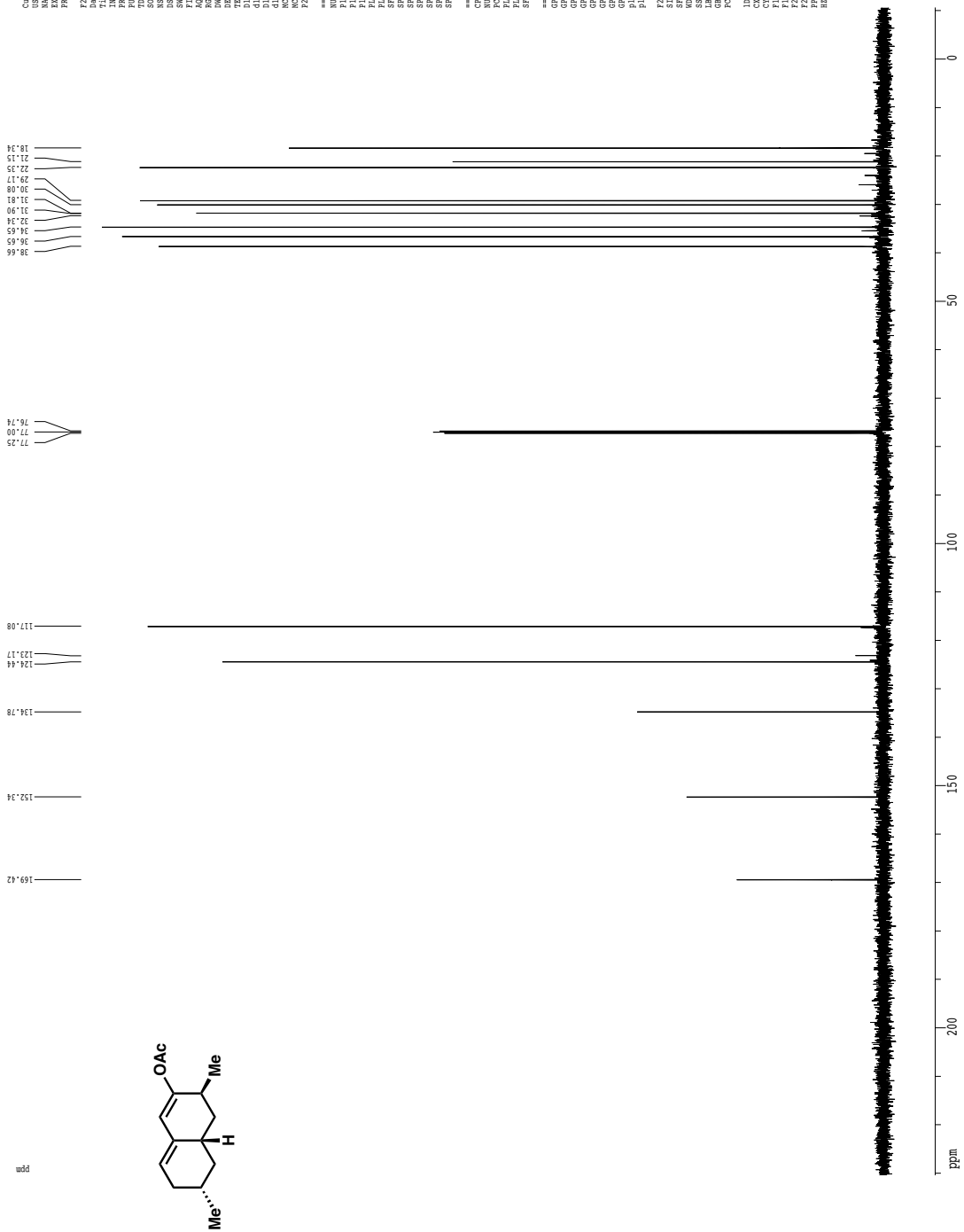








Z-restored spin-echo 13C spectrum with 1H decoupling

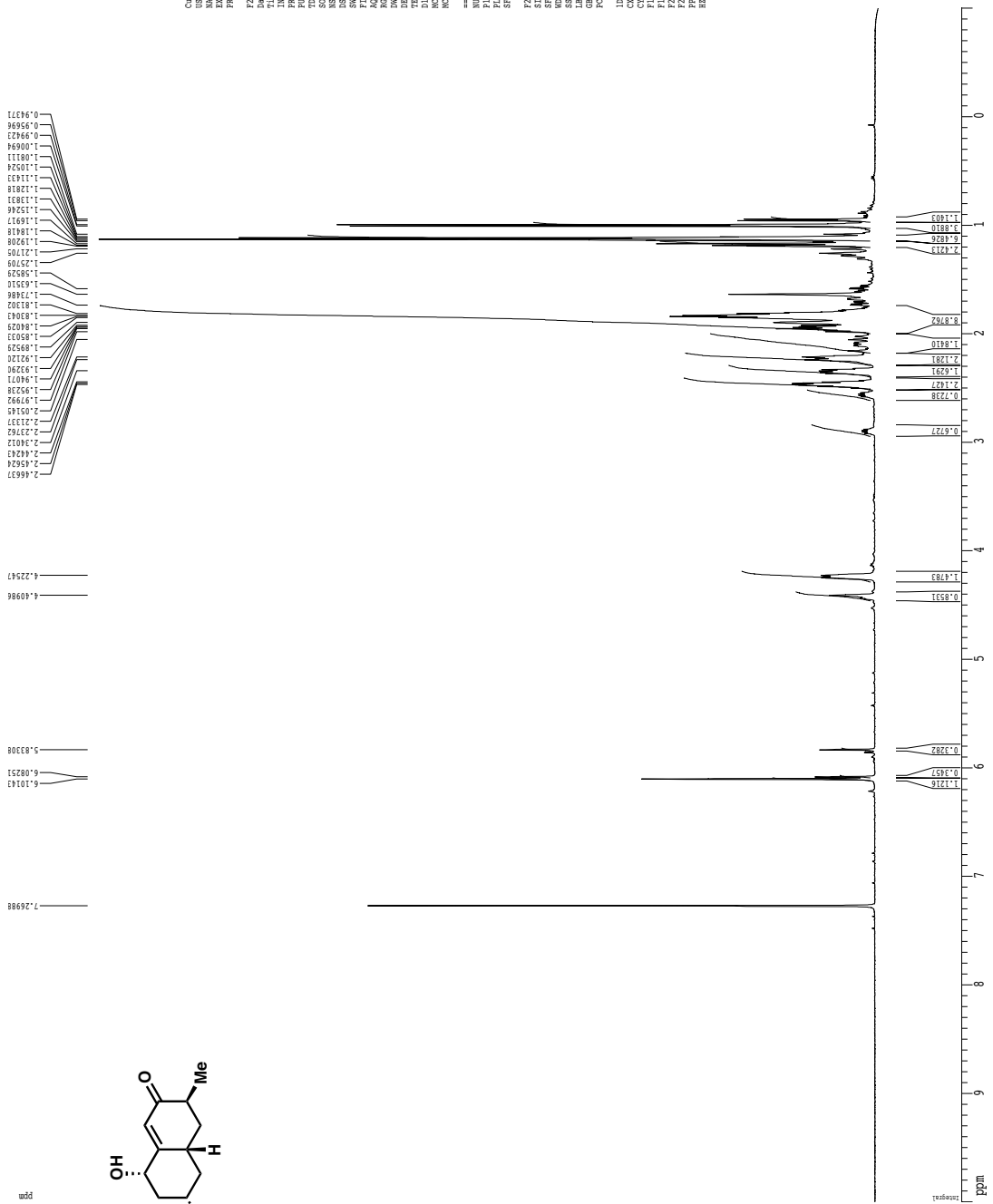


```

Current Data Parameters
NAME      pzt.262_inja
PROCNO    1
F2 - Acquisition Parameters
Date_     20131024
Time      18:24
INSTRUM   spect
PROBHD    5 mm QNP1H-
PULPROG   zgpg30
SOLVENT   CDCl3
NS        104
DS        4
SWH        33003.431 Hz
FIDRES    0.442388 Hz
AQ        1.093863 sec
RG        16.500 usec
DE        56.00 usec
TE        300.2 K
D1        0.25000000 sec
d11       0.03000000 sec
d12       0.03000000 sec
d13       0.00019600 sec
d14       0.00019600 sec
d15       0.00019600 sec
d16       0.00019600 sec
d17       0.00019600 sec
d18       0.00019600 sec
d19       0.00019600 sec
d20       0.00019600 sec
d21       0.00019600 sec
d22       0.00019600 sec
d23       0.00019600 sec
d24       0.00019600 sec
d25       0.00019600 sec
d26       0.00019600 sec
d27       0.00019600 sec
d28       0.00019600 sec
d29       0.00019600 sec
d30       0.00019600 sec
===== CHANNEL f1 =====
NUC1      13C
P1        15.50 usec
PL1       0.00 dB
PC1       2000.00 usec
P2        130.00 usec
PL2       19.00 dB
PC2       128.7942549 MHz
SFO1      125.762549 MHz
SFO2      500.136099 MHz
===== CHANNEL f2 =====
CPDPRG2   zgpg30
NUC2      13C
P2        15.50 usec
PL2       0.00 dB
PC2       2000.00 usec
P3        130.00 usec
PL3       19.00 dB
PC3       128.7942549 MHz
SFO3      125.762549 MHz
SFO4      500.136099 MHz
===== GRABUNT CHANNEL =====
GRABUN1   SINE
SINE1     100.00 usec
PL1       0.00 dB
PC1       1000.00 usec
GRABUN2   SINE
SINE2     100.00 usec
PL2       0.00 dB
PC2       1000.00 usec
GRABUN3   SINE
SINE3     100.00 usec
PL3       0.00 dB
PC3       1000.00 usec
GRABUN4   SINE
SINE4     100.00 usec
PL4       0.00 dB
PC4       1000.00 usec
===== F2 - Processing parameters =====
SI        65536
SF        125.762549 MHz
WDW        RM
SSB        0
GB        0
PC        2.00
===== 1D NMR plot parameters =====
CX        22.80 cm
CY        23.82 cm
F1        28976.25 Hz
F2        10.5668 Hz
F3        10.5668 Hz
F4        10.5668 Hz
F5        10.5668 Hz
F6        10.5668 Hz
F7        10.5668 Hz
F8        10.5668 Hz
F9        10.5668 Hz
F10       10.5668 Hz
F11       10.5668 Hz
F12       10.5668 Hz
F13       10.5668 Hz
F14       10.5668 Hz
F15       10.5668 Hz
F16       10.5668 Hz
F17       10.5668 Hz
F18       10.5668 Hz
F19       10.5668 Hz
F20       10.5668 Hz
F21       10.5668 Hz
F22       10.5668 Hz
F23       10.5668 Hz
F24       10.5668 Hz
F25       10.5668 Hz
F26       10.5668 Hz
F27       10.5668 Hz
F28       10.5668 Hz
F29       10.5668 Hz
F30       10.5668 Hz
=====

```

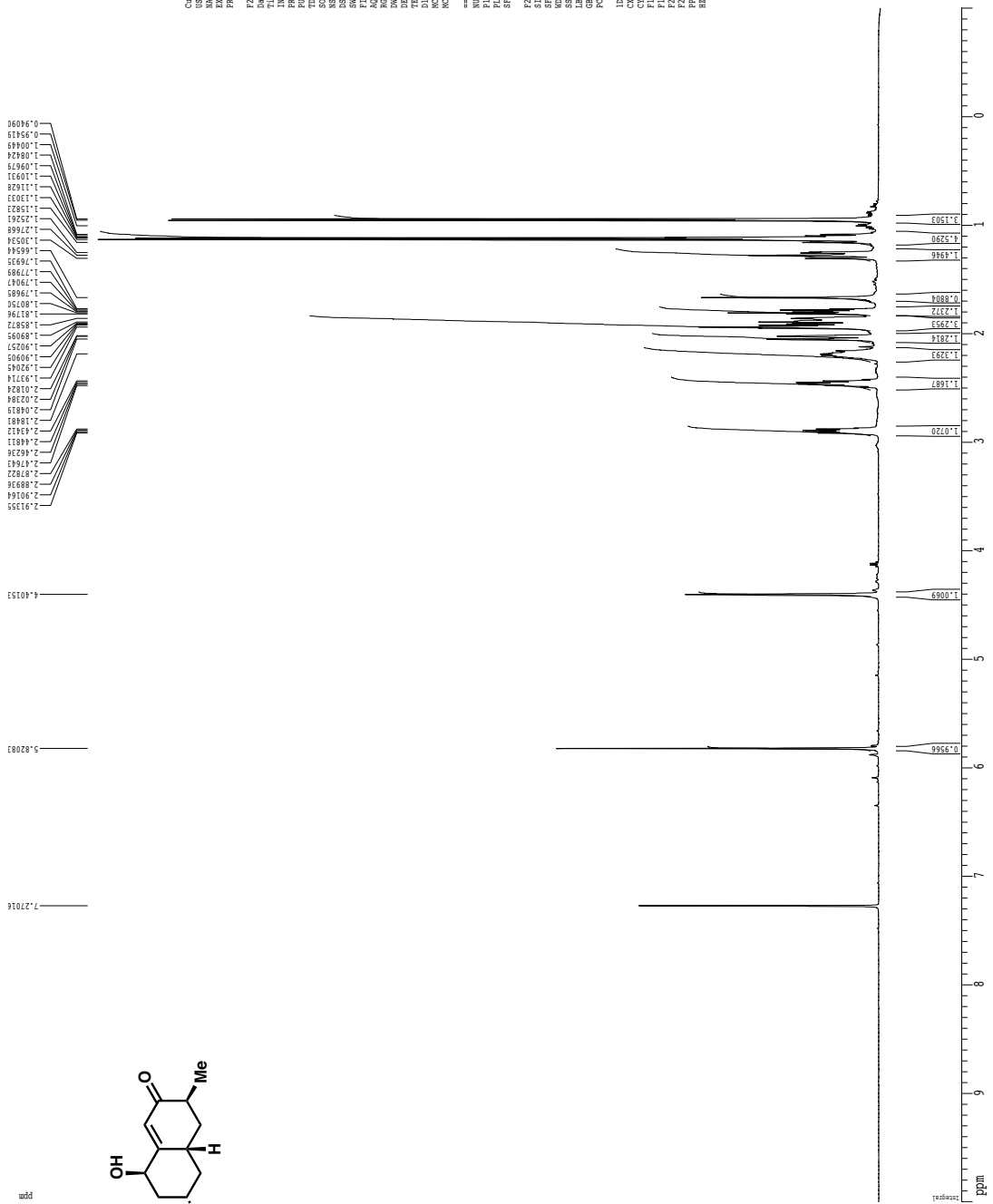
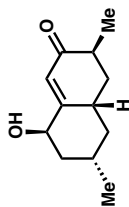
<sup>1</sup>H spectrum



Current Data Parameters  
 USER roosen  
 NAME P033.011\_milior  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20131231  
 Time 15:04  
 INSTRUM crys000  
 PULPROG zgpg30  
 TD 32768  
 SFO 500.136266  
 AQ 0.100000 sec  
 RG 64.3  
 DE 6.00 uS/pc  
 TE 300.2 K  
 INJECT 0.0100000 sec  
 ACQPRG 0.0100000 sec  
 ===== CHANNEL f1 =====  
 NUC1 <sup>1</sup>H  
 P1 7.50 uS/pc  
 PL1 0.00  
 PR1 510.223413 MHz  
 F2 - Processing parameters  
 SF 500.2201666 MHz  
 EQ 0.00  
 ZS 0.30 Hz  
 GB 0  
 PC 4.00  
 ID NMR pilot parameters  
 CT 22.80 cm  
 PI 10.000 ppm  
 F1P 10.000 ppm  
 F1 500.2201666 MHz  
 F2 500.2201666 MHz  
 FREQCN 0.44246 ppm/cm  
 BUCN 241.33423 Hz/cm

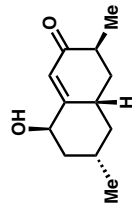
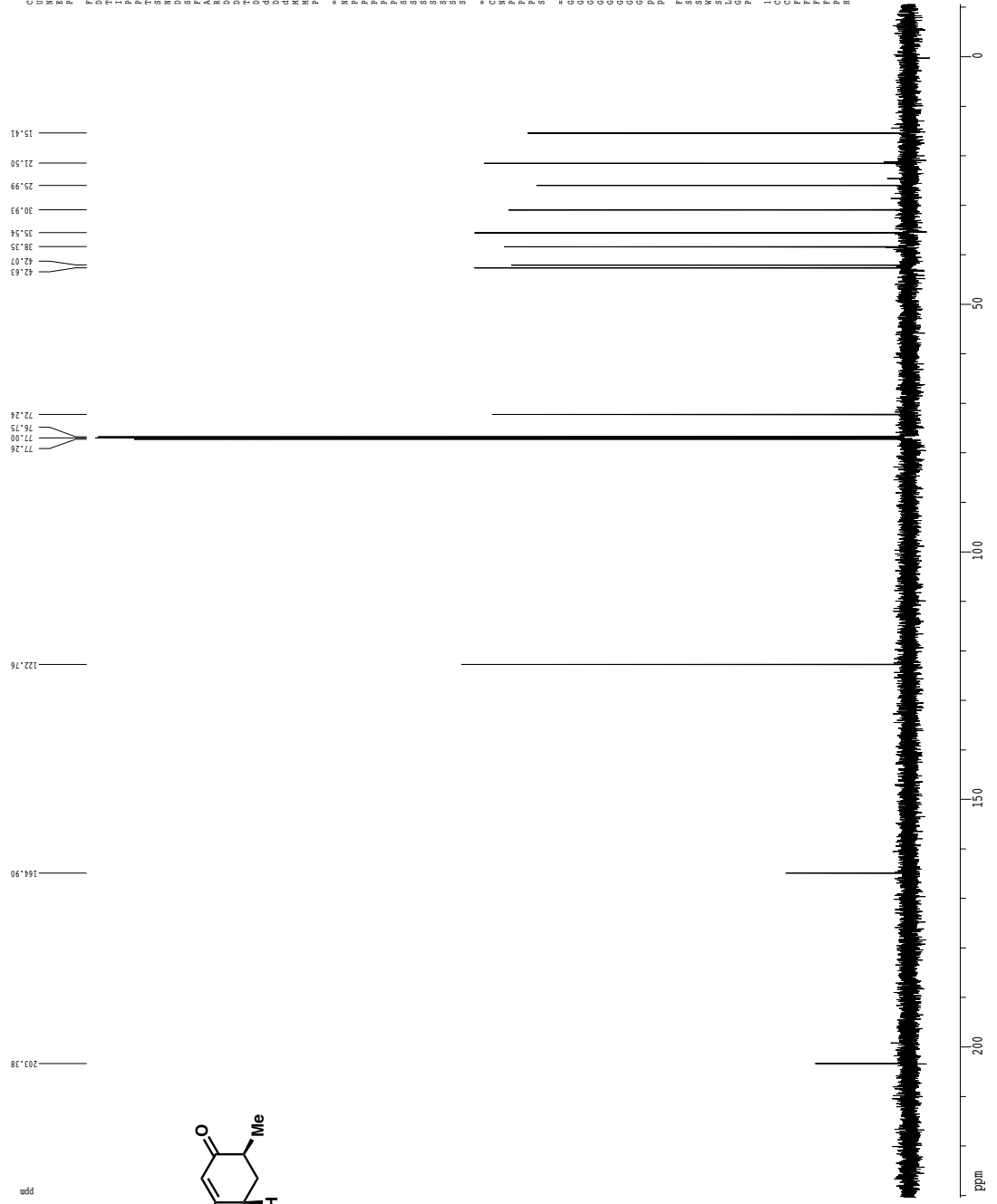


<sup>1</sup>H spectrum



Current Data Parameters  
 USER roosen  
 NAME P022.257\_F1-1-15  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20110208  
 Time 22:31  
 INSTRUM spect  
 PULPROG zgpg30  
 TD 32768  
 SFO 500.135  
 AQ 0.16  
 RG 64  
 DI 4  
 DE 6.00 uS  
 TE 298.15 K  
 ACQPROG mzgpg30  
 MCHUNK 0.0100000 sec  
 ===== CHANNEL f1 =====  
 NUC1 <sup>1</sup>H  
 P1 7.50 uS  
 PL1 0.00 dB  
 SFO1 500.2254113 MHz  
 F2 - Processing parameters  
 SF 500.2254113 MHz  
 EQ2 EN  
 ZS 0.30 Hz  
 GB 0  
 PC 4.00  
 ID NR0 pilot parameters  
 CT 22.80 cm  
 PI 1.00 dB  
 F1P 10.000 ppm  
 F1 500.2254113 MHz  
 F2 500.2254113 MHz  
 F2P -50.022 ppm  
 FREQH 0.44846 ppm/cm  
 BECN 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling

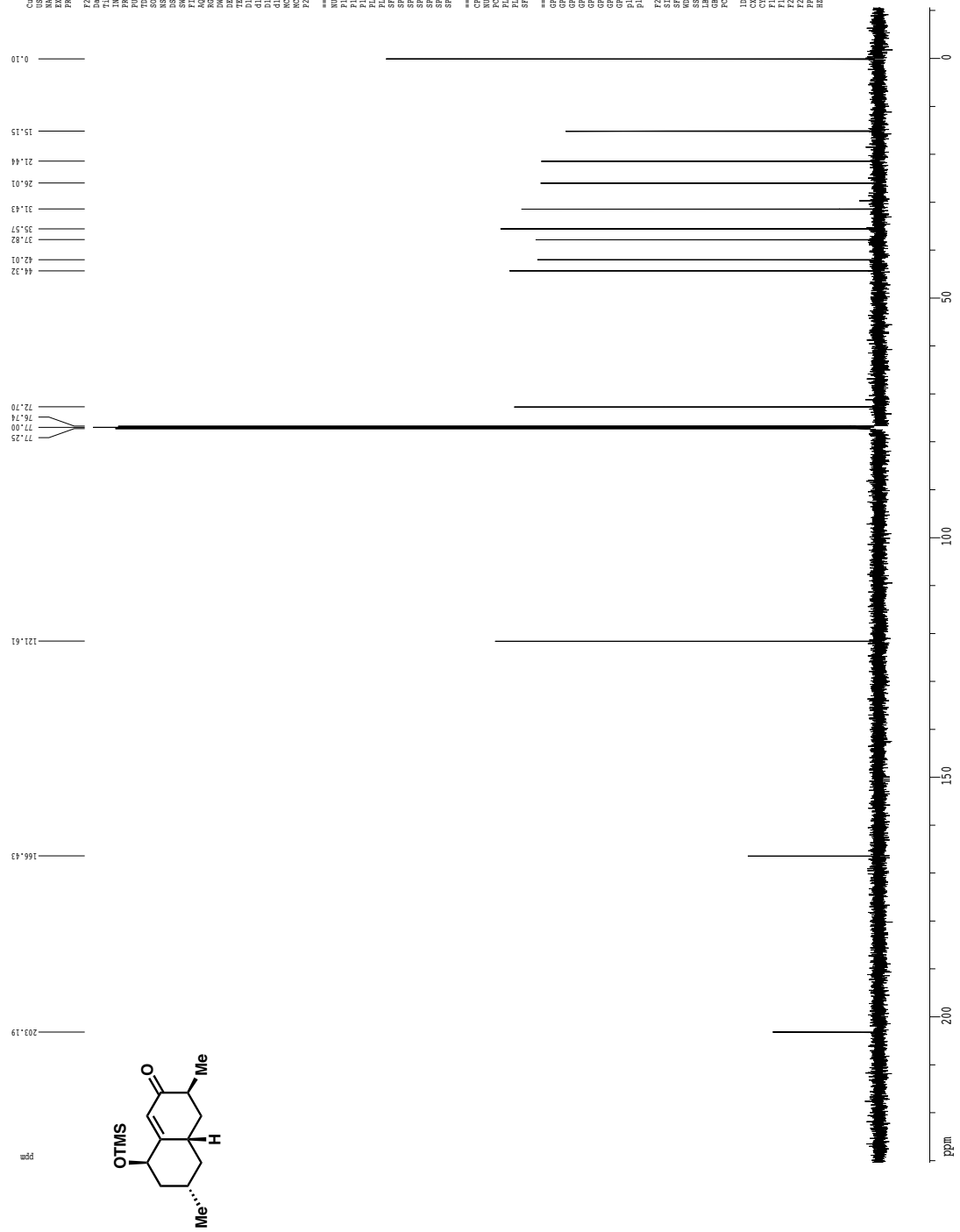


```

Current Data Parameters
NAME   pec2257_P1-15
PROCNO 1
===== Acquisition Parameters
Date_   20131020
Time    22.35
INSTRUM spect
PROBHD  5 mm QNP1H-
PULPROG zgpg30
AQ       0.0013600 sec
RG       327.5
SI       64
SF       125.7642400 MHz
SFO1     125.7642400 MHz
SFO2     500.225011 MHz
===== CHANNEL f1 =====
NUC1     13C
P1       15.50 usec
PL1      0.00 dB
PC1      200.00 usec
RF1      125.7642400 MHz
===== CHANNEL f2 =====
CPDPRG2 zgpg30
NUC2     131H
P2       100.00 usec
PL2      0.00 dB
PC2      24.60 dB
RF2      500.225011 MHz
===== GRABBER CHANNEL =====
GRABPR2 SINE
GRABNUC 131H
GRABPC  0.00 usec
GRABPL  0.00 dB
GRABRF  500.00 MHz
===== Processing parameters
SI       65536
SF       125.7642400 MHz
AQ       0.00000000 sec
RG       327.5
SI       64
SF       125.7642400 MHz
PC       2.00
===== 1D NMR list parameters
CK       22.80 cm
SI       23.850 cm
RG       28877.29 Hz
F1       115.50 ppm
F2       115.50 ppm
F3       115.50 ppm
F4       115.50 ppm
F5       115.50 ppm
F6       115.50 ppm
=====
  
```

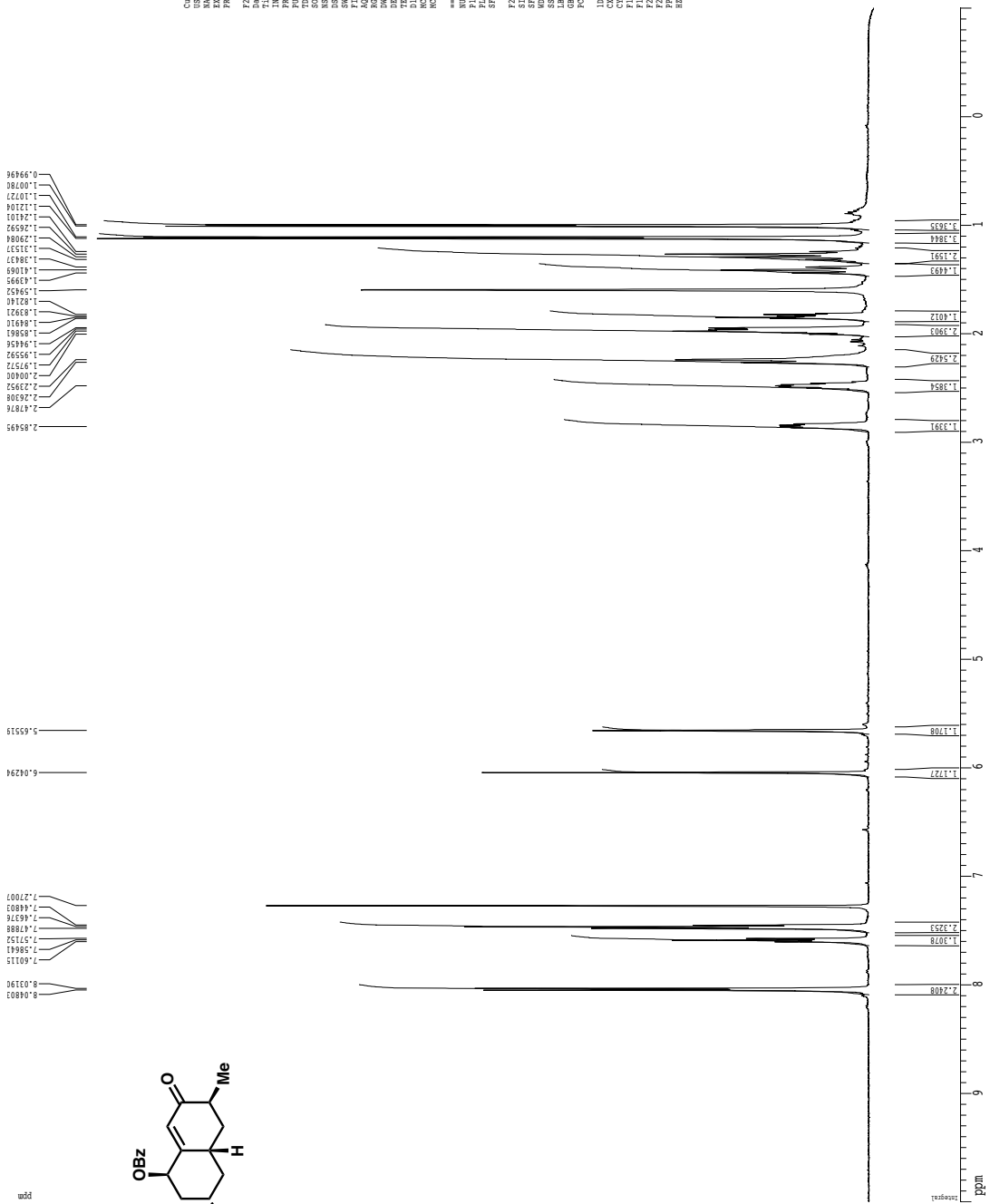


Z-restored spin-echo 13C spectrum with 1H decoupling

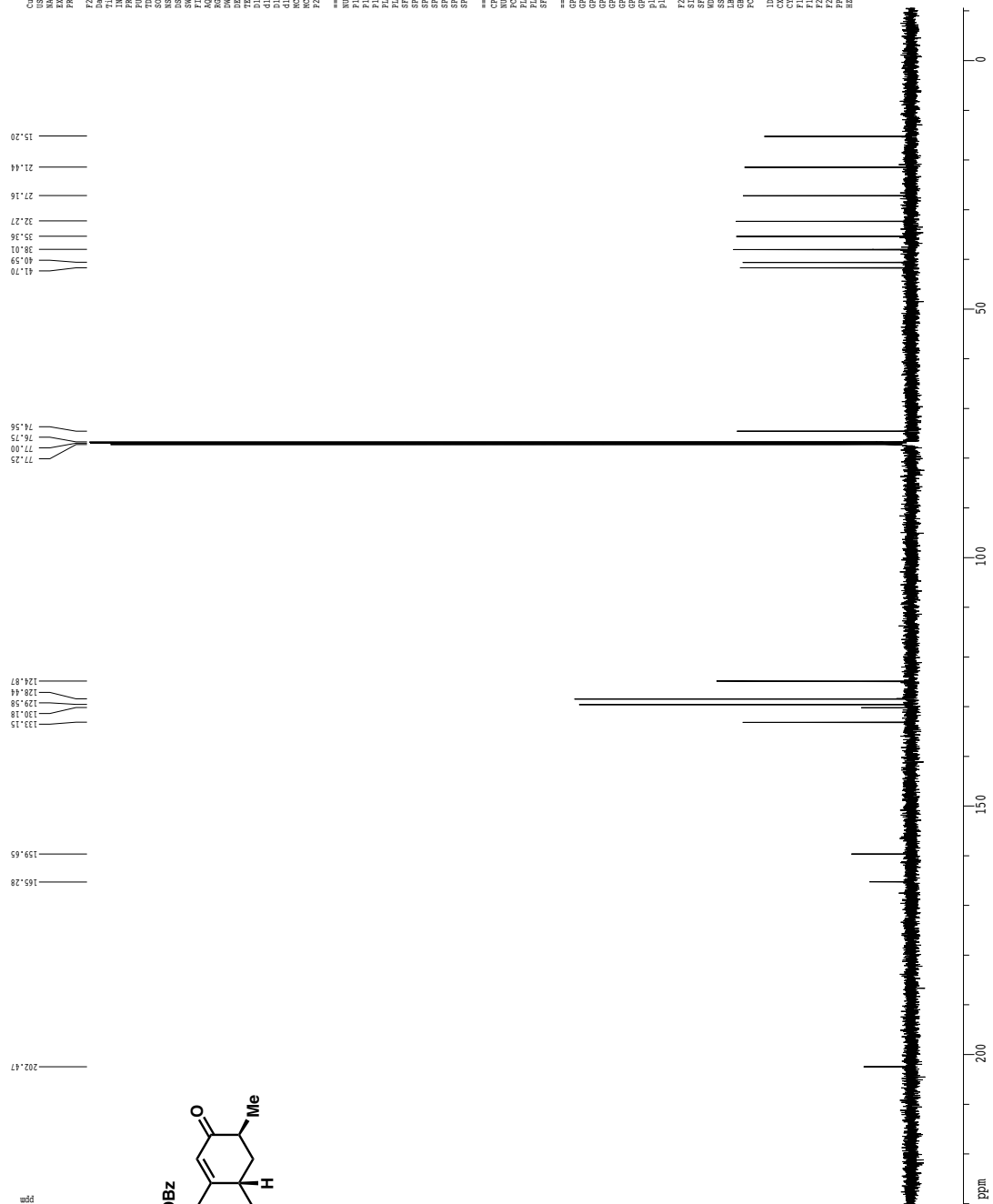
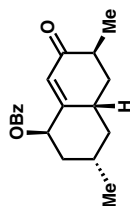




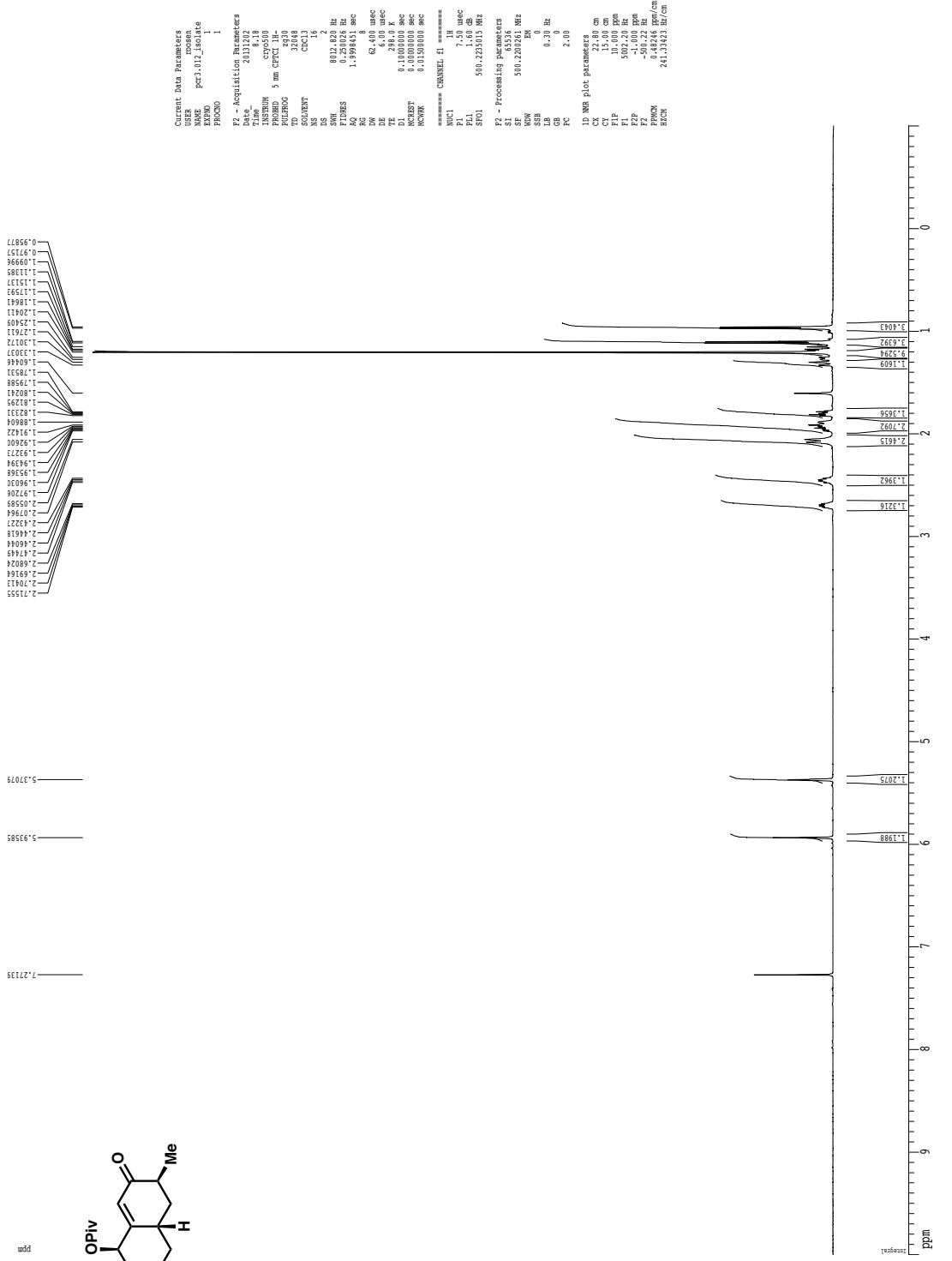
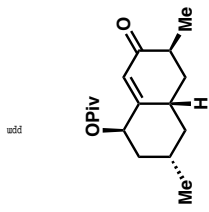
1H spectrum



# Z-restored spin-echo 13C spectrum with 1H decoupling



# 1H spectrum



Current Data Parameters  
 USER rcoosen  
 NAME P03.012\_isolate  
 PROBO 1  
 PRODO 1

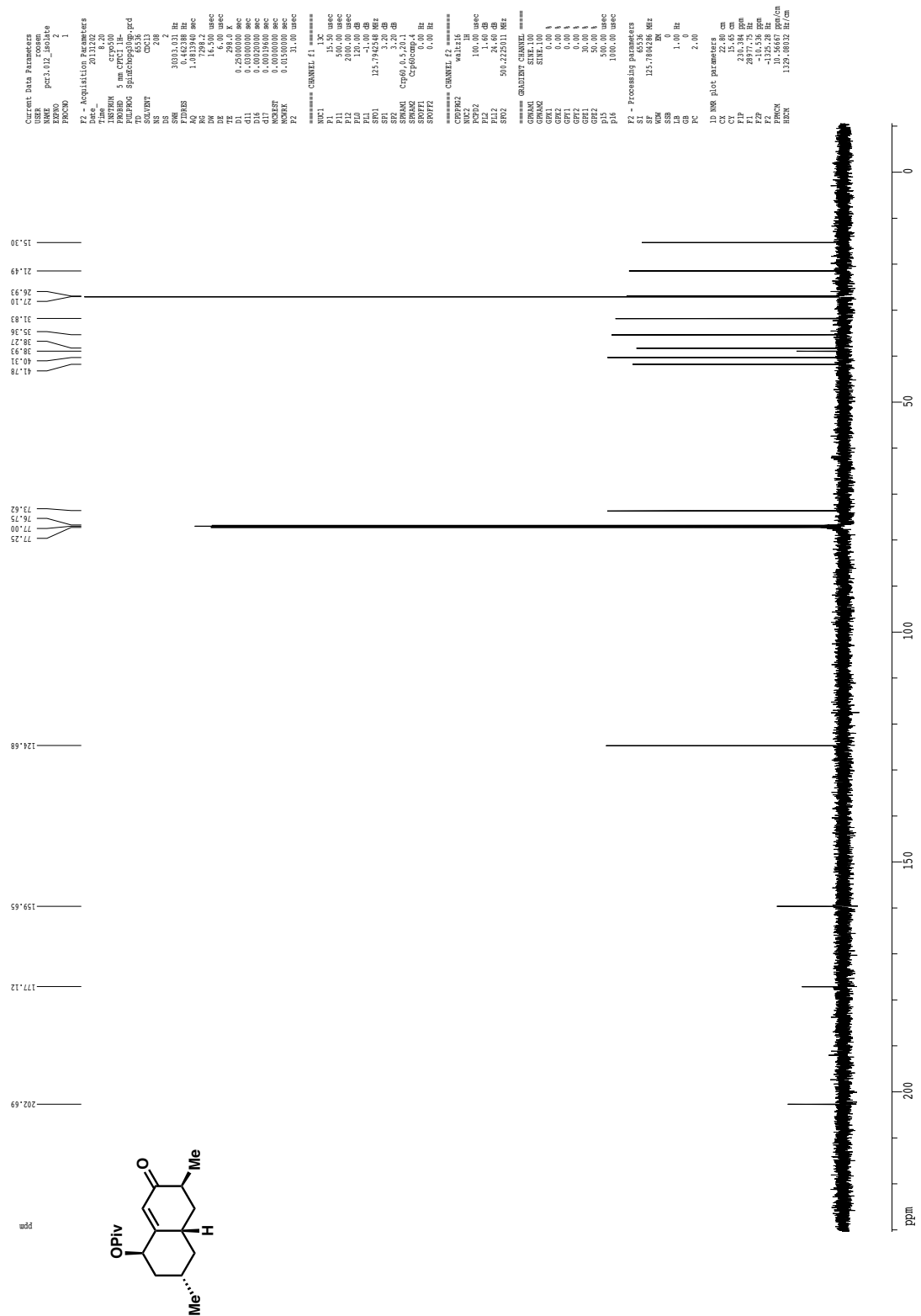
F2 - Acquisition Parameters  
 Date\_ 2013202  
 Time 8.18  
 INSTRUM crysol0  
 SPECTUM 1  
 PULPROG zgpg30  
 TD 32048  
 CH2SOLVENT CDCl3  
 NS 16  
 DS 2  
 SFO 8145.82 Hz  
 EQ 0.254028 Hz  
 AQ 1.7999451 sec  
 RG 8  
 BQ 63.6 usec  
 DE 6.00 usec  
 TE 0.298.0 K  
 INJECT 0.0100000 sec  
 INCR 0.0100000 sec  
 ACQUIS 0.0100000 sec

===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.2254143 MHz

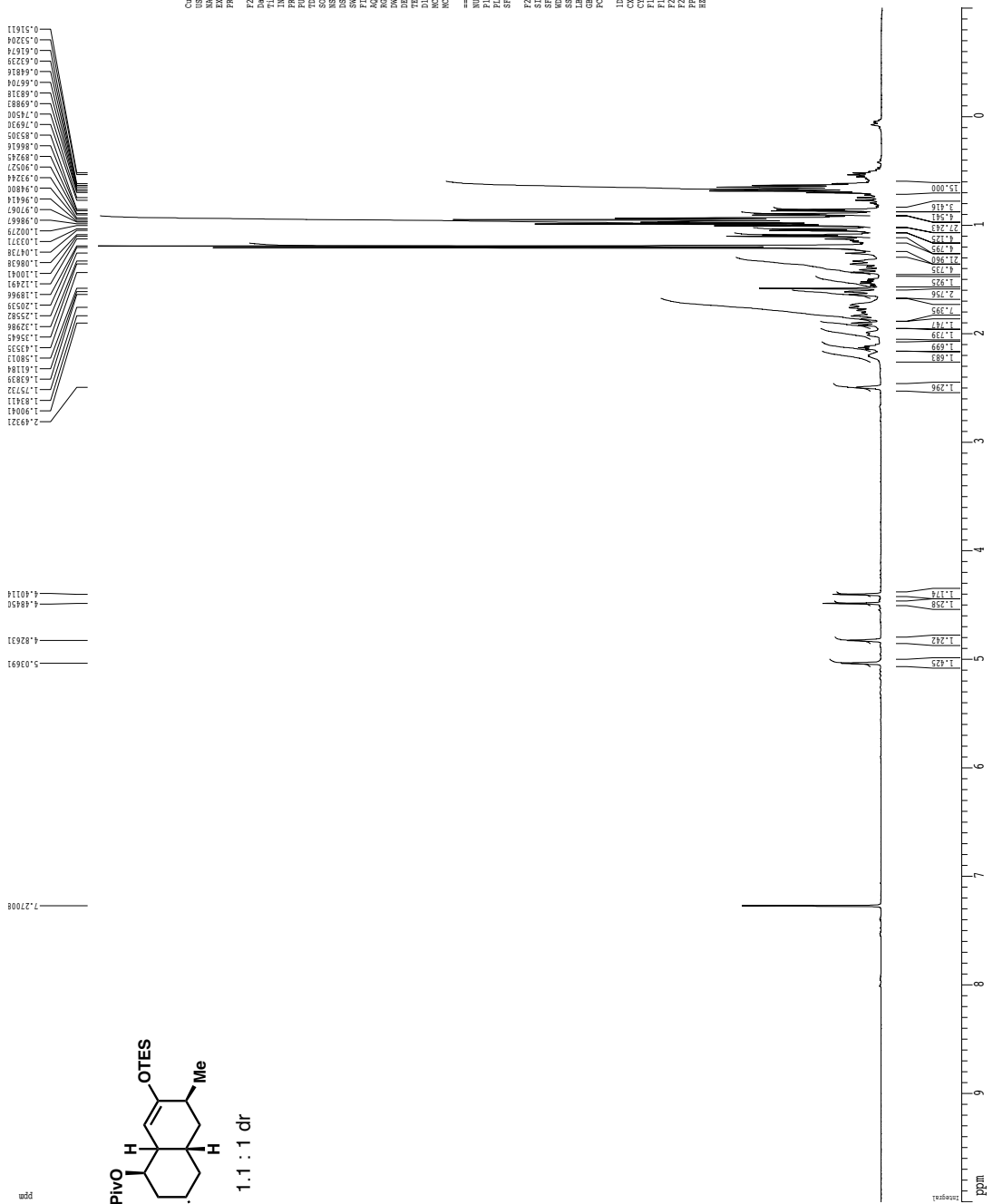
F2 - Processing parameters  
 SFO 500.2201651 MHz  
 EN  
 GB 0  
 CB 0.30 Hz  
 FC 2.00

1D MR plot parameters  
 CT 22.80 cm  
 CR 0.25 cm  
 FIDP 10.000 ppm  
 FI 500.220 Hz  
 F2 500.220 MHz  
 PRG 1D  
 FPCOR 0.44846 ppm/cm  
 BPCOR 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling

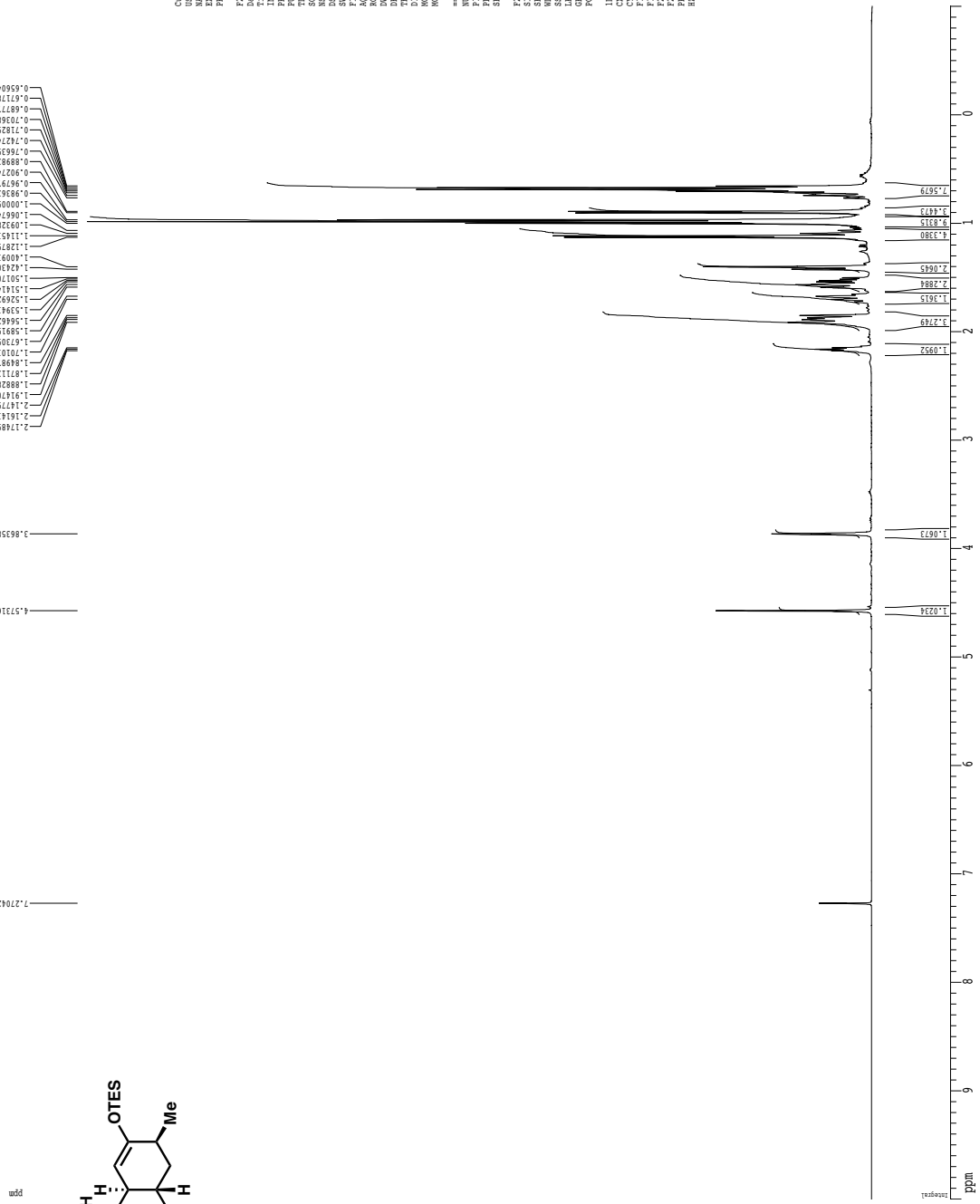


<sup>1</sup>H spectrum

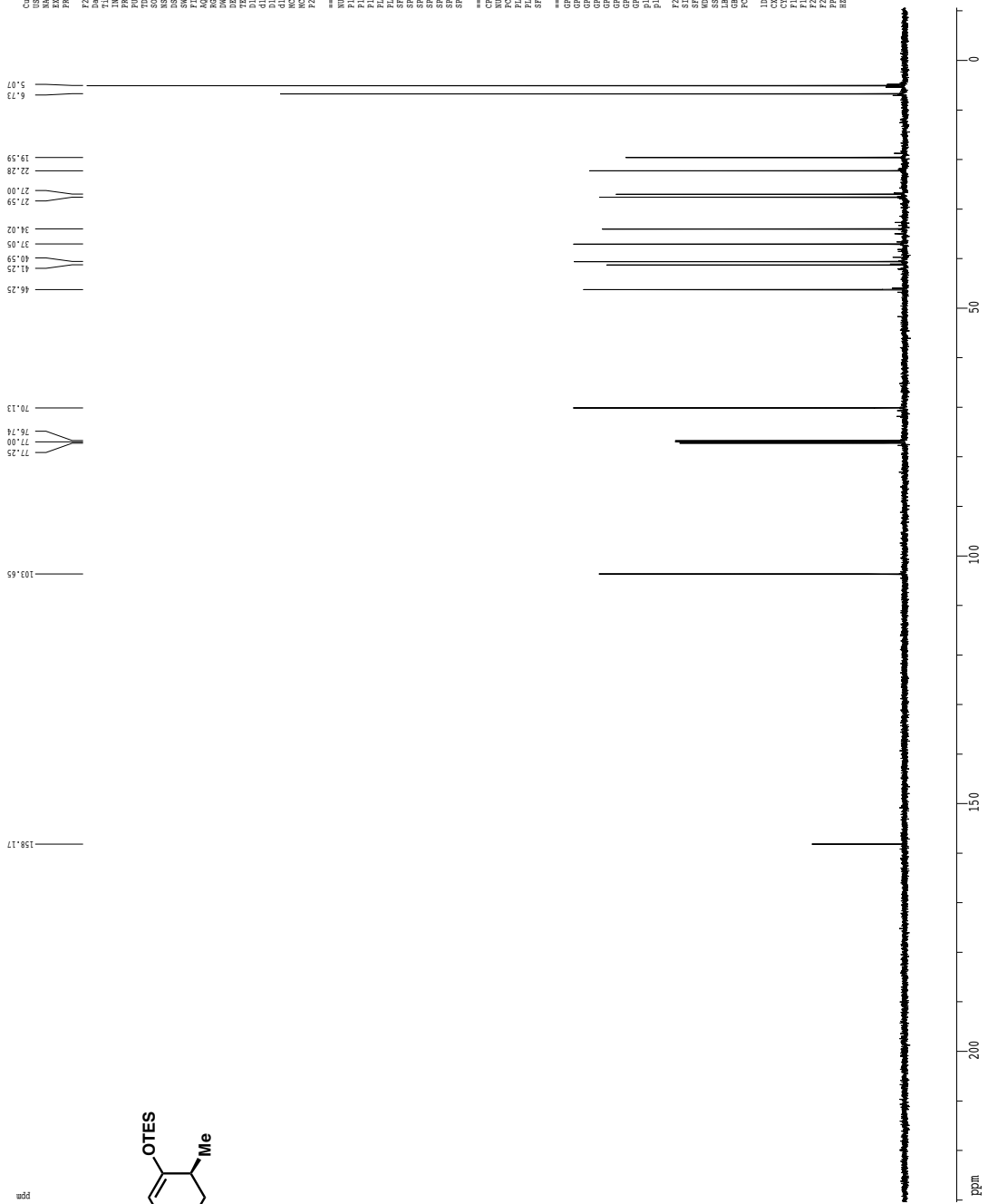




<sup>1</sup>H spectrum



Z-restored spin-echo 13C spectrum with 1H decoupling

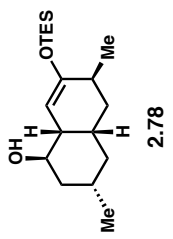


```

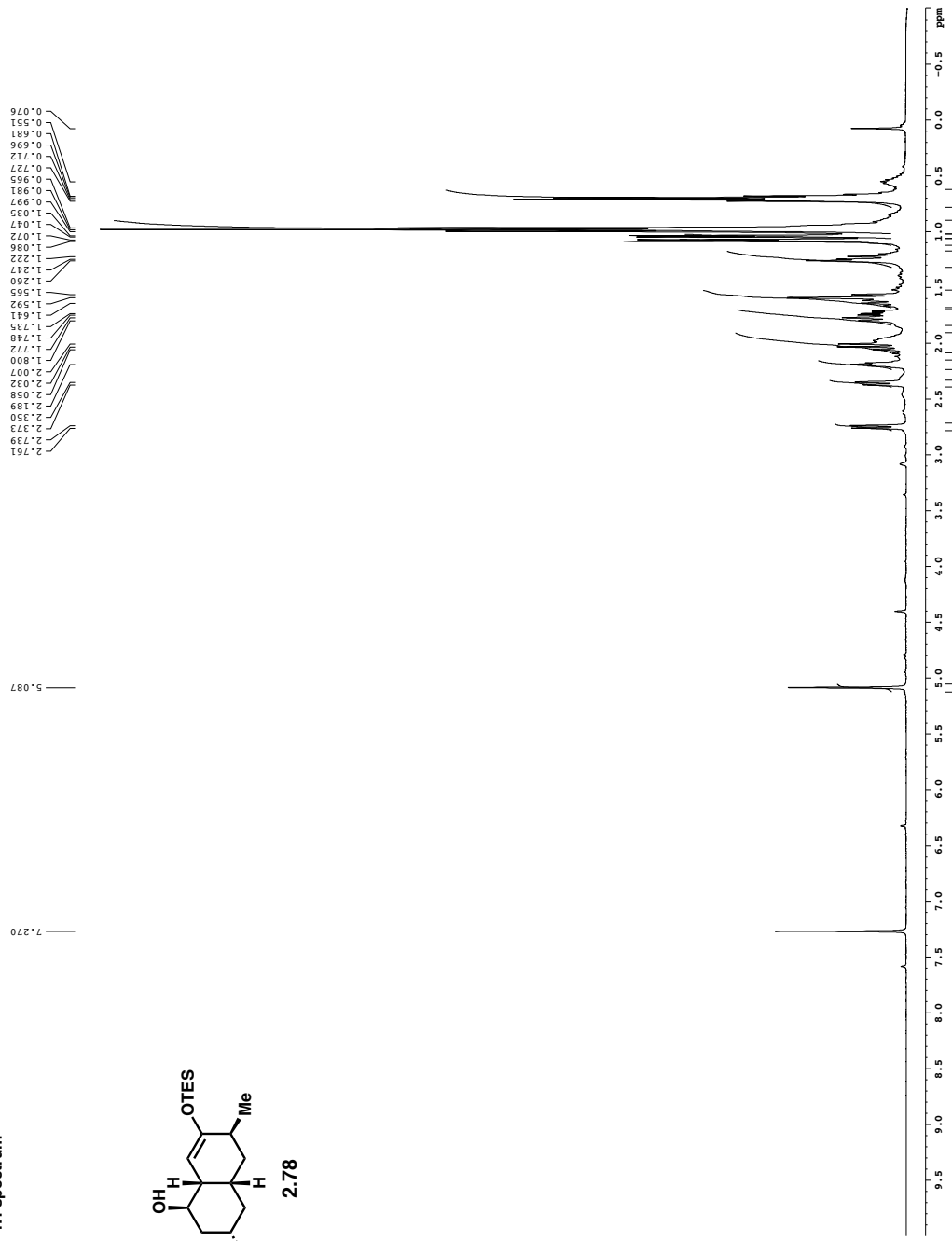
Current Data Parameters
NAME          gpc3.015_top
PROCNO       1
F2 - Acquisition Parameters
Date_         20131204
Time          6:17:22
INSTRUM      spect
PROBHD       5 mm QNP1H-
PULPROG      zgpg30
SOLVENT      CDCl3
SS           161
===== CHANNEL f1 =====
NUC1          13C
P1           15.50 usec
PL1          -1.00 dB
PC1          200.00 usec
RF1           120.00 dB
SFO1         125.7642449 MHz
SF1           3.20 dB
SFR1         Cmp66.6.5.201.0B
SFRAC2       Cmp66comp.4
SFRAC1       1.00
SFO2         0.00 Hz
SFR2         0.00 Hz
===== CHANNEL f2 =====
CPDPRG2      zgpg30
NUC2          1H
P2           10.00 usec
PL2          -1.00 dB
PC2          24.60 dB
SFO2         500.225011 MHz
===== GRABUNT CHANNEL =====
GRAMM        SINE
SFR1         0.00 kHz
SFR2         0.00 kHz
SFR3         0.00 kHz
SFR4         0.00 kHz
SFR5         0.00 kHz
SFR6         500.00 usec
SFR7         1000.00 usec
F2 - Processing parameters
SI           65536
AQ           125.7642449 MHz
RG           0
SFO          125.7642449 MHz
WDW          EM
SSB           0
LB           1.00 Hz
GB           0
PC           2.00
=====
1D NMR plot parameters
CK           22.80 cm
SI           23.80 cm
F1           28977.29 Hz
F2           110.50 Hz
F3           110.50 Hz
FREQN       103.56467 ppm/cm
BUCN       1329.0032 Hz/cm
  
```



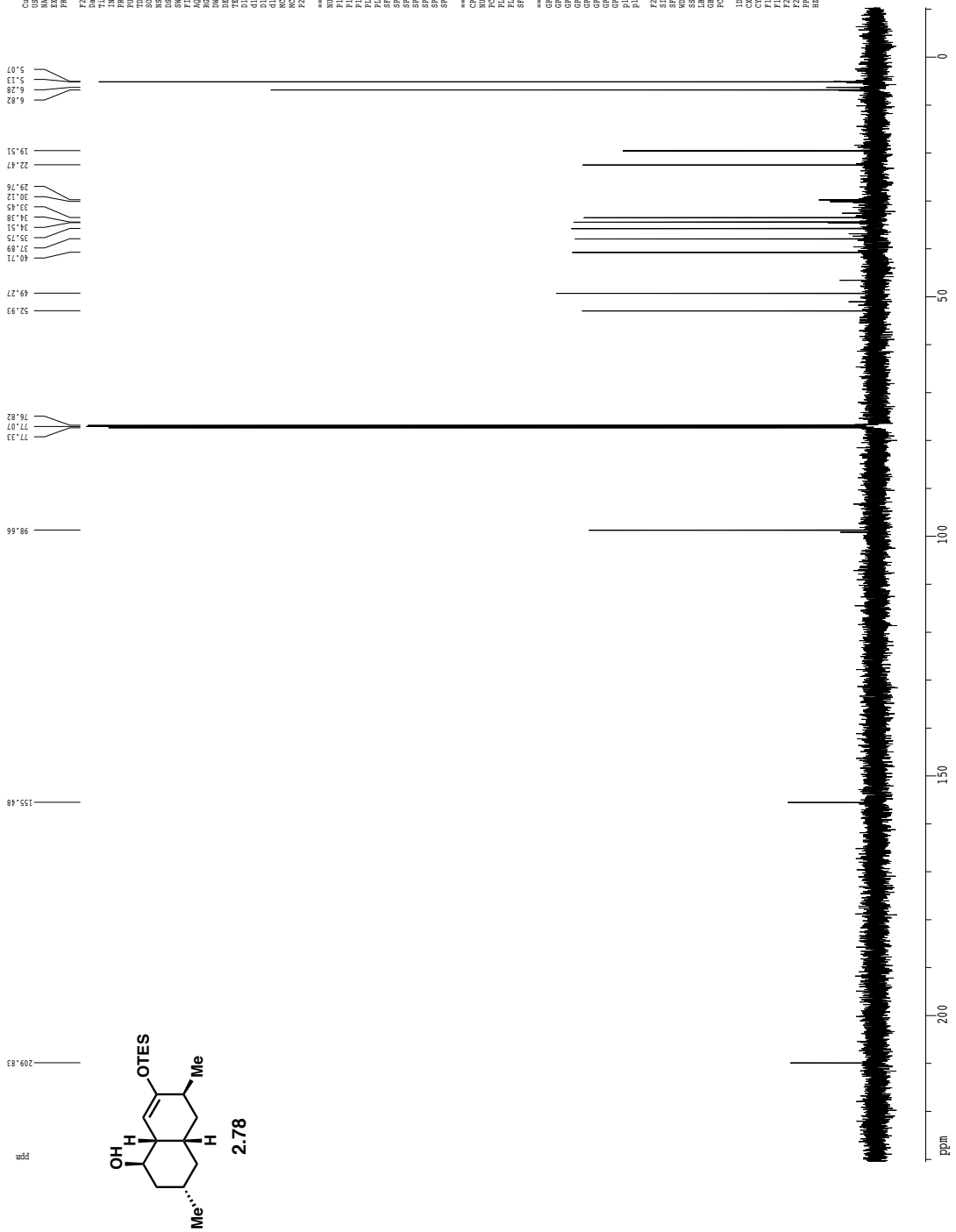
1H spectrum



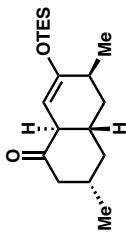
Current Data Parameters  
 NAME: Pur3\_017\_16oct16  
 PROCNO: 1  
 F2 - Acquisition Parameters  
 Date\_: 20111005  
 TIME: 13:33:00  
 INSTRUM: cryoind  
 PULPROG: zgpg30  
 TD: 32768  
 SFO1: 500.223515 MHz  
 NS: 16  
 DS: 4  
 SWH: 8012.820 Hz  
 FIDRES: 0.250026 Hz  
 AQ: 1.00000000 sec  
 RG: 327.68  
 GB: 4.15 usec  
 DB: 6.00 usec  
 DI: 0.10000000 sec  
 DECI: 0.01000000 sec  
 REFR: 0.01000000 sec  
 CHANNEL: 1H  
 NUC1: 1H  
 P1: 12.00 usec  
 PL1: 0.00 dB  
 SFO1: 500.223515 MHz  
 F2 - Processing parameters  
 SI: 32768  
 SF: 500.220266 MHz  
 DS: 4  
 SWH: 8012.820 Hz  
 LB: 0.30 Hz  
 GB: 0.00 Hz  
 PC: 4.00



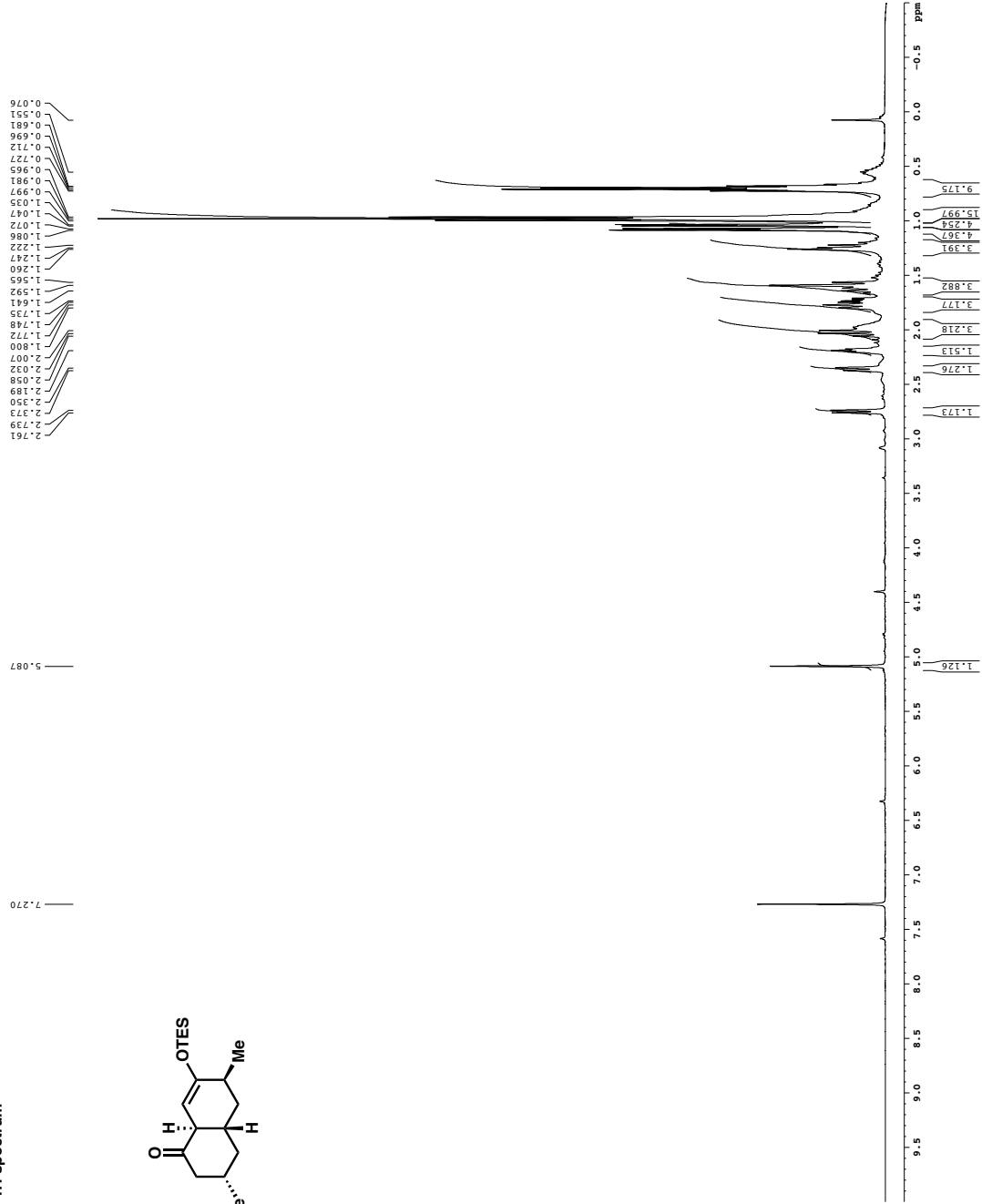
Z-restored spin-echo 13C spectrum with 1H decoupling



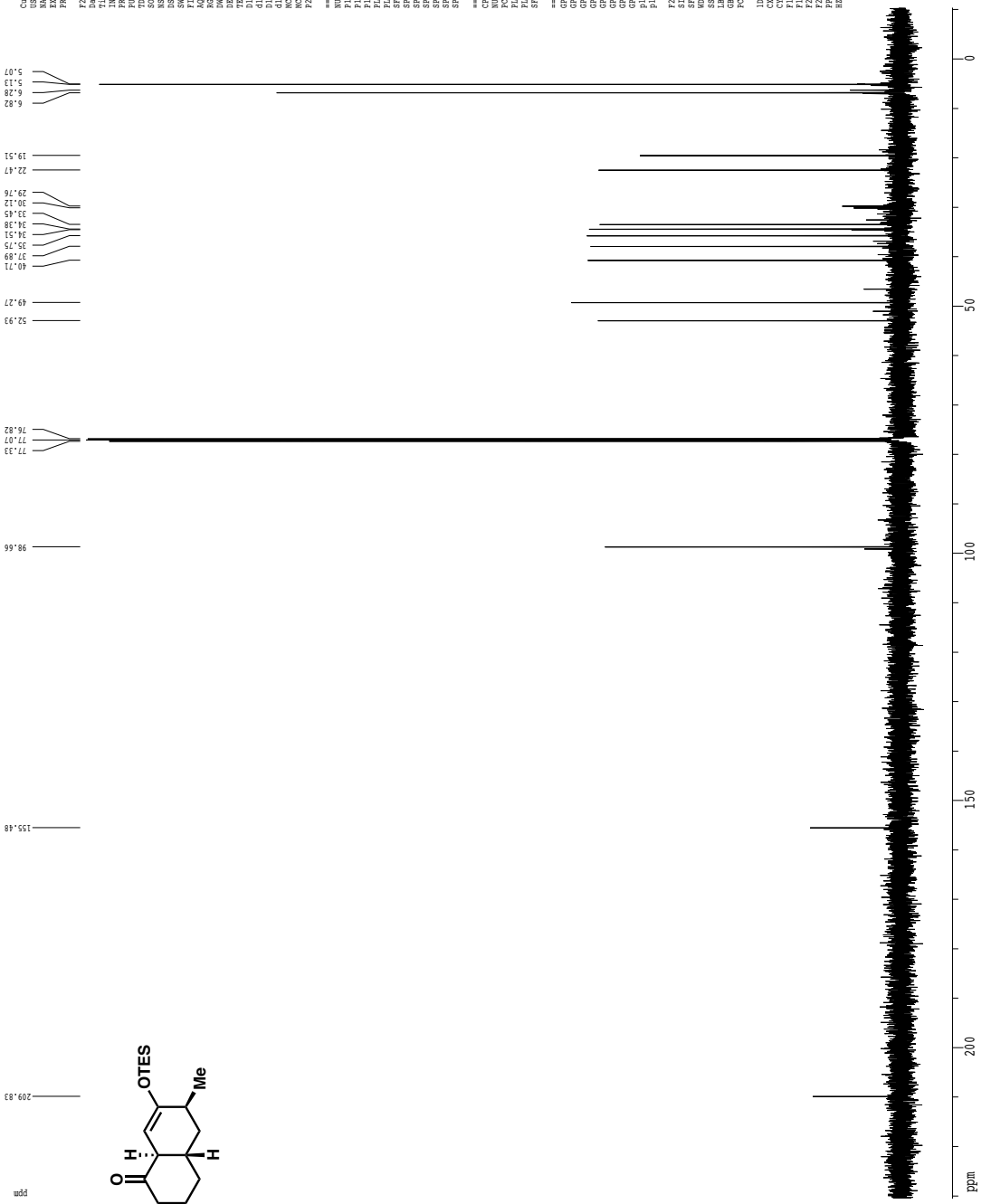
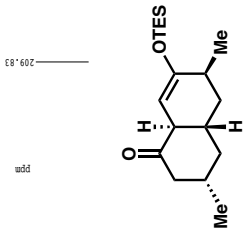
1H spectrum



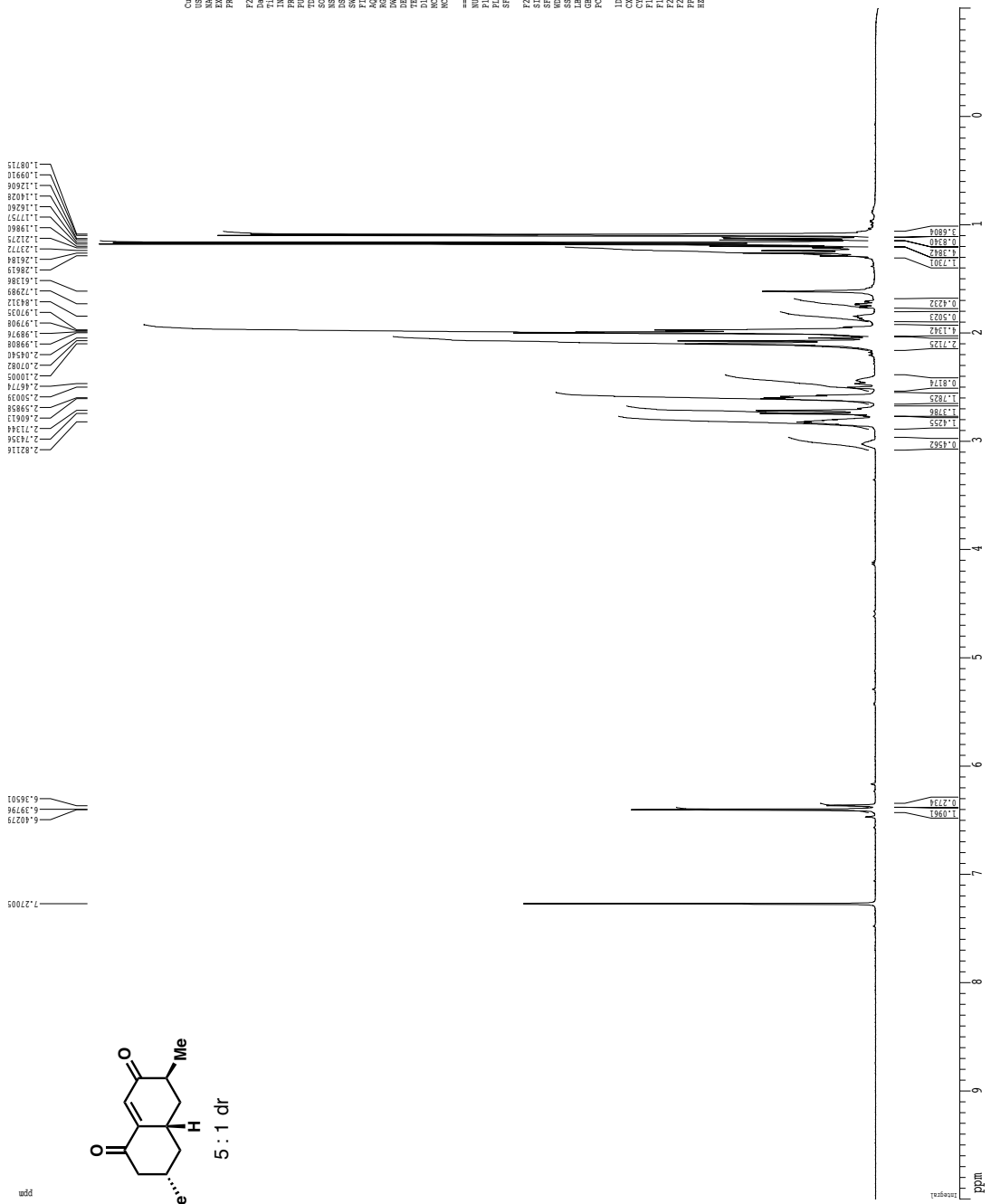
Current Data Parameters  
 NAME: Pur3\_017\_16oct16  
 PROCNO: 1  
 F2 - Acquisition Parameters  
 Date\_: 20111005  
 TIME: 13:33:11  
 INSTRUM: cryoind  
 PULPROG: zgpg30  
 TD: 32768  
 SFO: 500.1361994 MHz  
 NS: 16  
 DS: 4  
 SWH: 8012.820 Hz  
 FIDRES: 0.250026 Hz  
 AQ: 1.0000000 sec  
 RG: 4.15  
 GB: 0.0000000 sec  
 DB: 6.00 usec  
 DI: 0.1000000 sec  
 DECI: 0.0000000 sec  
 ACQRES: 0.1000000 sec  
 CHANNEL: C13NMR1  
 NUC1: 13C  
 P1: 1.00 usec  
 PL1: 0.00 dB  
 F2 - Processing parameters  
 SF: 500.1361994 MHz  
 SF2: 500.220266 MHz  
 LB: 0.30 Hz  
 GB: 0.0000000 sec  
 PC: 4.00



Z-restored spin-echo 13C spectrum with 1H decoupling

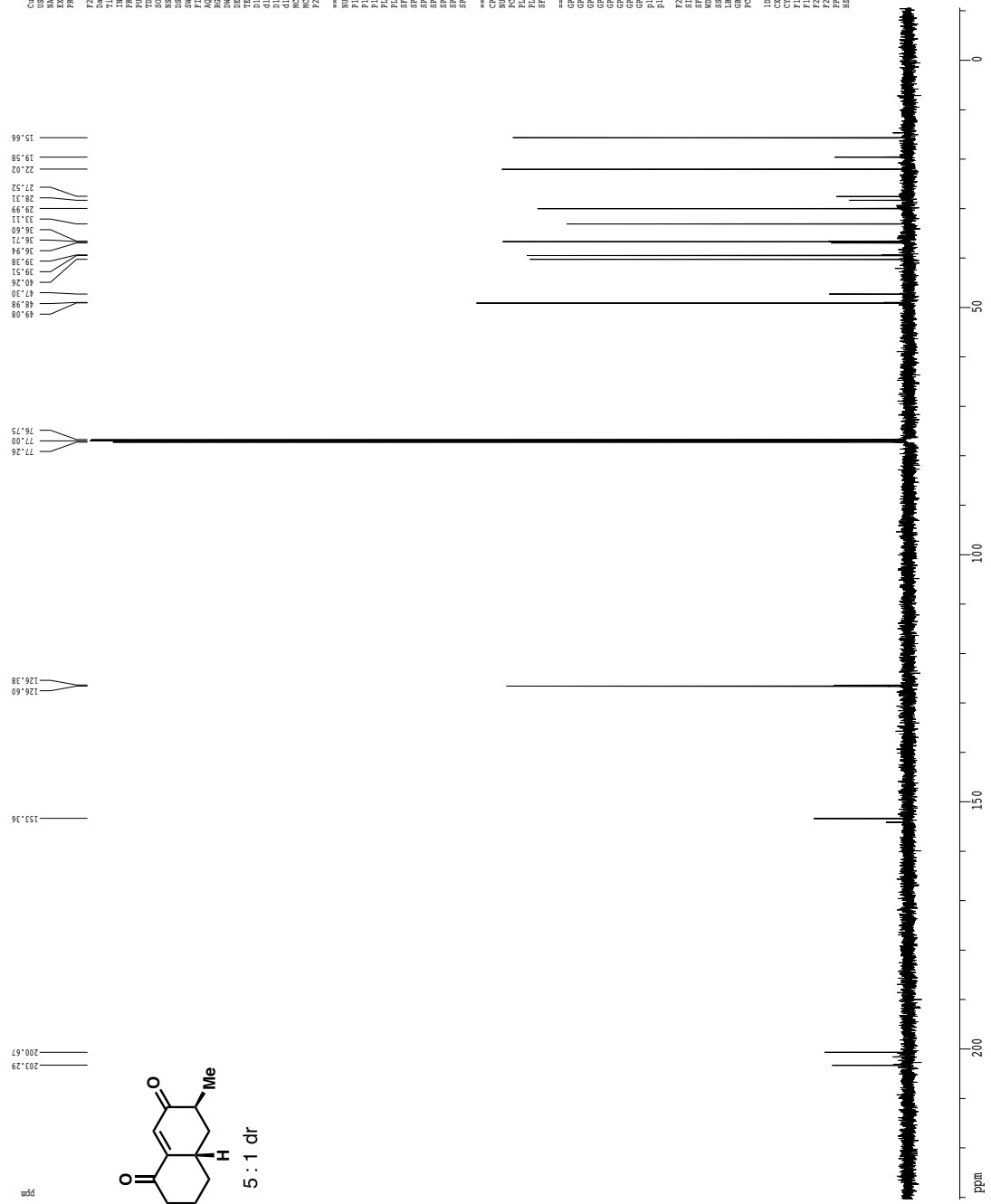


<sup>1</sup>H spectrum



Current Data Parameters  
 USER roosen  
 NAME PC3.022\_isolate  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20131211  
 Time 15:07  
 INSTRUM crys001  
 PULPROG zgpg30  
 TD 32768  
 SFO 500.136266  
 AQ 0.100000  
 RG 63.0  
 JE 6.00 uSBC  
 TE 300.2 K  
 FREQ 500.136266 MHz  
 NUC1 1H  
 P1 7.50 uSBC  
 PL 0.00  
 PRG1 500\_223413  
 F2 - Processing parameters  
 SI 500.2201666 MHz  
 SN EN  
 DS 0  
 GB 0  
 GC 4.00  
 ID NR0 pilot parameters  
 CI 22.80 cm  
 CR 10.000 ppm  
 FI 500.220166 MHz  
 F2 -500.220166 MHz  
 F3 -500.220166 MHz  
 F4 0.48246 ppm/cm  
 BICO 241.3323 Hz/cm

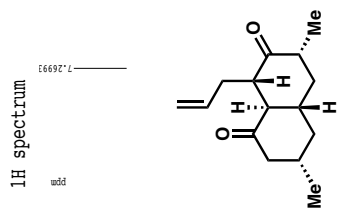
Z-restored spin-echo <sup>13</sup>C spectrum with <sup>1</sup>H decoupling



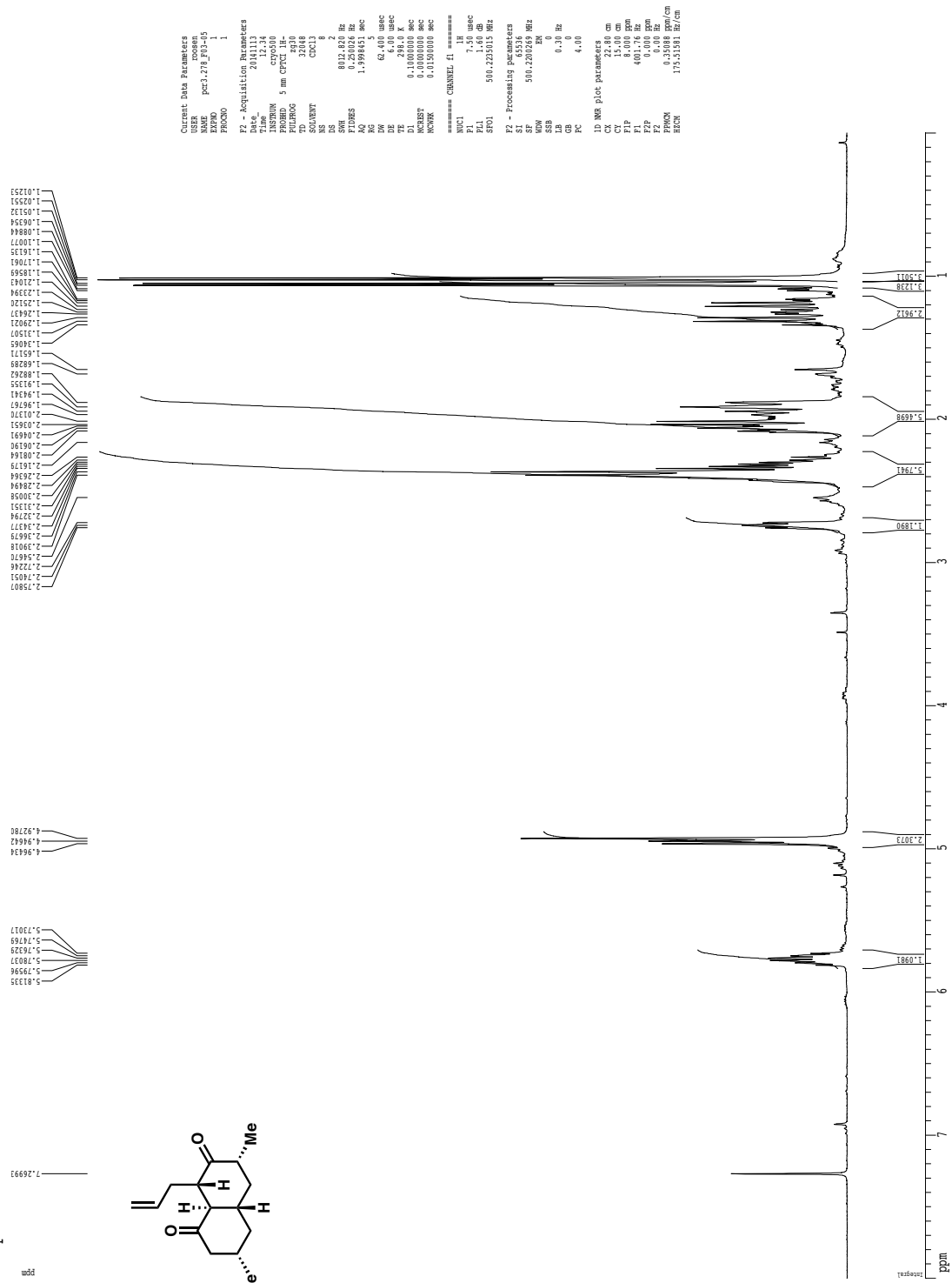
```

Current Data Parameters
NAME          pec3.02 isolate
PROCNO       1
=====
F2 - Acquisition Parameters
Date_         20131211
Time         15.11
INSTRUM      spect
PROBHD       5 mm QNP1H
PULPROG      zgpg30
PCPDPRG2     cpdprg2
SFO1         125.762548 MHz
SF2          500.136436 MHz
AQ           3.20
RG           320
WDW          EM
SS           2
RG2          320
=====
===== CHANNEL f1 =====
NUC1          13C
P1           15.50 usec
PL1          0.00 dB
PC1          200.00 usec
P2           120.00 usec
PL2          0.00 dB
PC2          200.00 usec
SFO1         125.762548 MHz
SF2          500.136436 MHz
SR1          3.20
SR2          3.20
SFO1         125.762548 MHz
SFO2         500.136436 MHz
===== CHANNEL f2 =====
CPDPRG2      zgpg30
NUC2          1H
P2           10.00 usec
PL2          0.00 dB
PC2          20.00 usec
SFO1         500.136436 MHz
SF2          400.146351 MHz
===== GRADIENT CHANNEL =====
GRAD1        SIB1.100
GRAD2        SIB1.100
GPR1         0.00 V
GPR2         0.00 V
GPR3         0.00 V
GPR4         0.00 V
GPR5         0.00 V
GPR6         0.00 V
GPR7         0.00 V
GPR8         0.00 V
=====
===== F2 - Processing parameters =====
SI           65536
SF           500.136436 MHz
WDW          EM
SS           0
RG           320
PC           1.80
=====
===== 1D NMR plot parameters =====
CK           22.80 cm
SI           32768
SF           125.762548 MHz
AQ           3.20 usec
RG           320
WDW          EM
SS           0
PC           1.80
=====
===== CHANNEL f1 =====
===== CHANNEL f2 =====
===== GRADIENT CHANNEL =====
=====
===== F2 - Processing parameters =====
SI           65536
SF           500.136436 MHz
WDW          EM
SS           0
RG           320
PC           1.80
=====
===== 1D NMR plot parameters =====
CK           22.80 cm
SI           32768
SF           125.762548 MHz
AQ           3.20 usec
RG           320
WDW          EM
SS           0
PC           1.80
=====
===== CHANNEL f1 =====
===== CHANNEL f2 =====
===== GRADIENT CHANNEL =====
=====
===== F2 - Processing parameters =====
SI           65536
SF           500.136436 MHz
WDW          EM
SS           0
RG           320
PC           1.80
=====
===== 1D NMR plot parameters =====
CK           22.80 cm
SI           32768
SF           125.762548 MHz
AQ           3.20 usec
RG           320
WDW          EM
SS           0
PC           1.80
=====

```



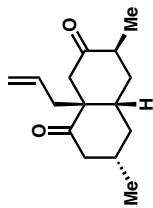
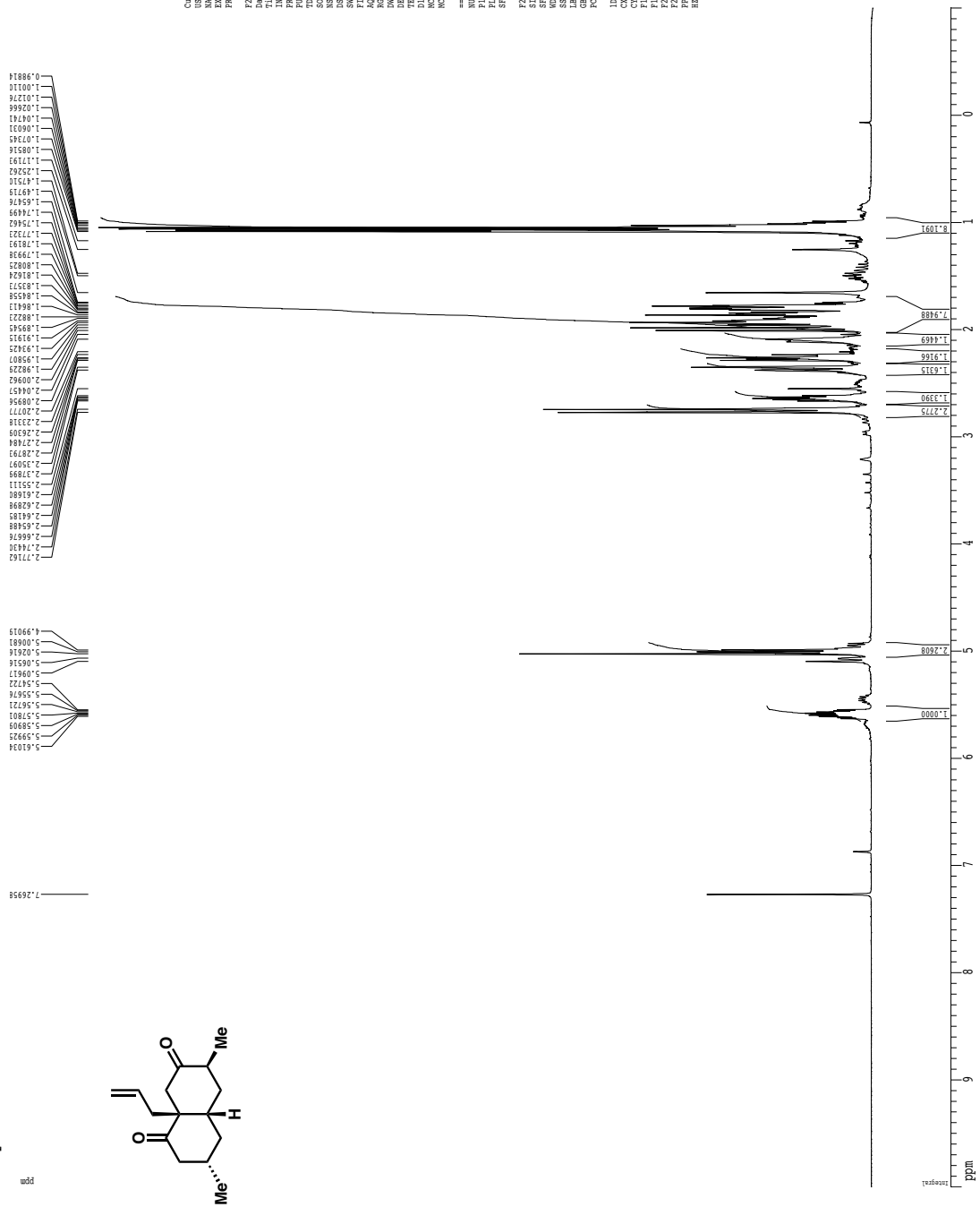
<sup>1</sup>H spectrum





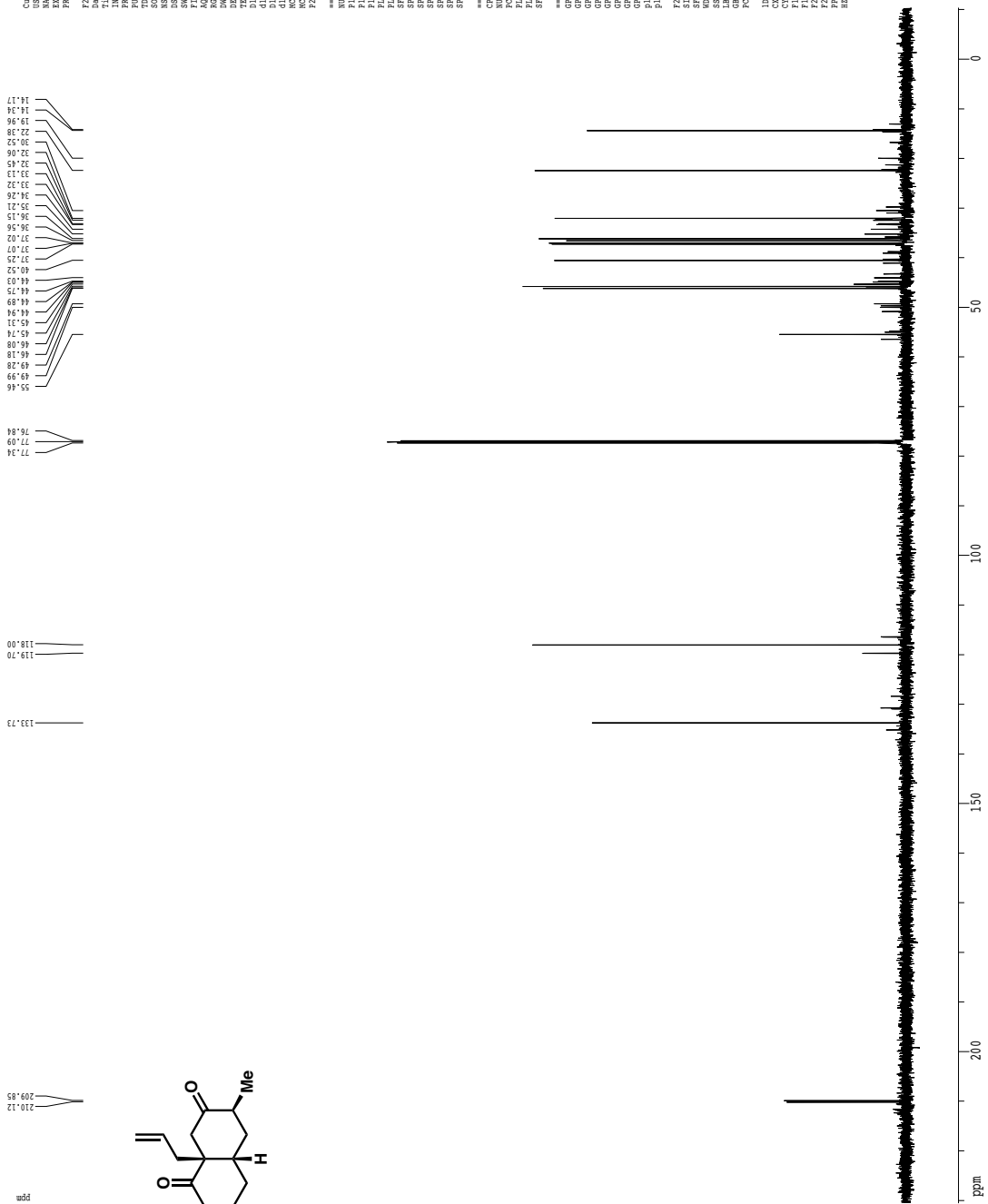


<sup>1</sup>H spectrum

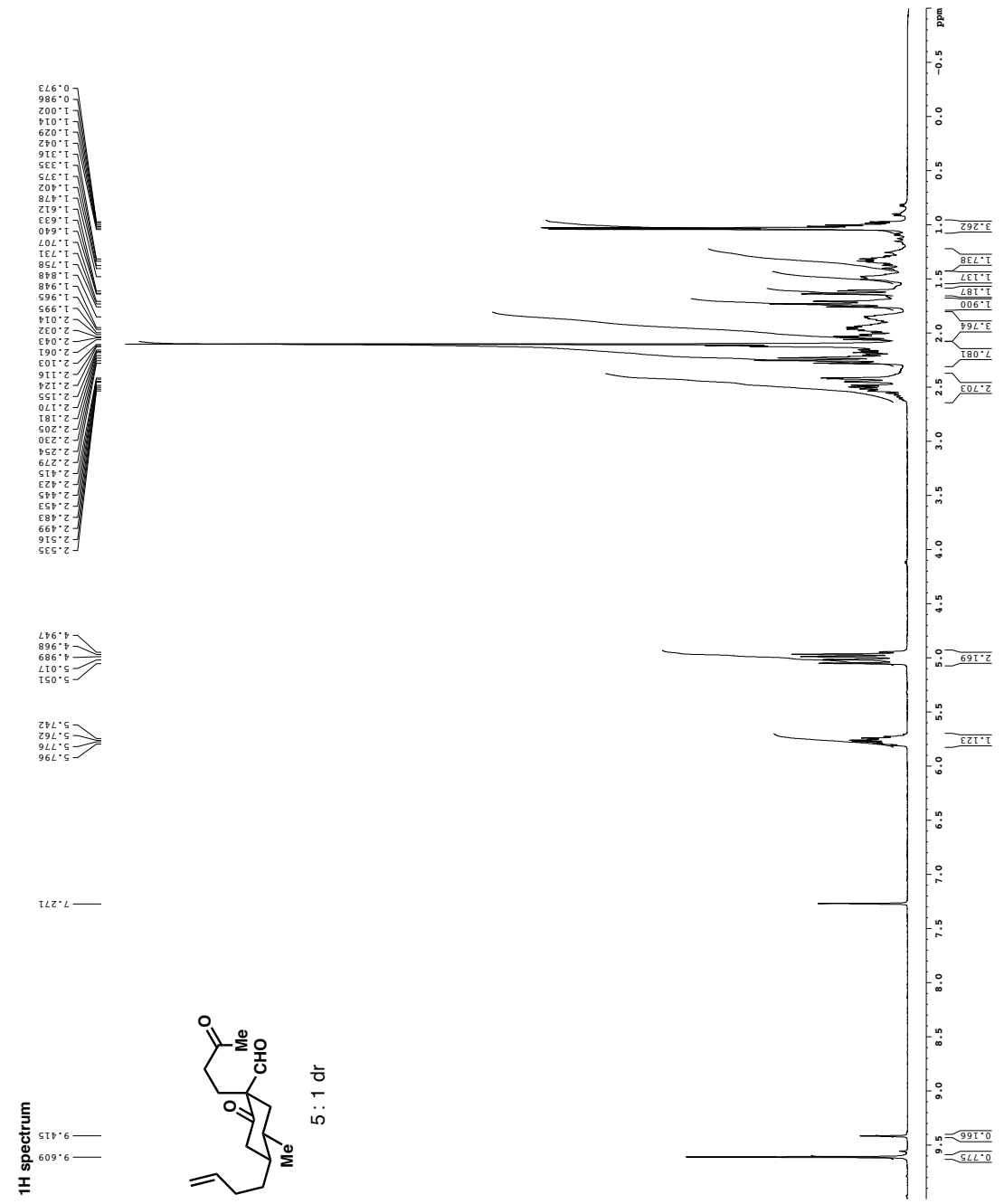


Current Data Parameters  
 USER roosen  
 NAME PC3.27E-F7-09  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2014113  
 Time 12:59  
 INSTRUM spect  
 PULPROG zgpg30  
 TD 32768  
 CHANVENT CHAN  
 SI 0  
 DS 2  
 AS 0  
 HYPERC 0  
 AQ 1.999451 sec  
 RG 6.3 usec  
 DE 6.00 usec  
 TE 0.1 298.0 K  
 KWAVE 0.000000 sec  
 KWAVE 0.000000 sec  
 ===== CHANNEL f1 =====  
 NUCL1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.225413 MHz  
 F2 - Processing parameters  
 SI 0  
 SF 500.225413 MHz  
 EQ 0  
 ZS 0.30 Hz  
 GB 0  
 PC 4.00  
 ID NMR pipe parameters  
 CT 22.80 cm  
 PI 10.00 mm  
 FIDP 10.00 ppm  
 FI 500.220 Hz  
 F2 500.225413 MHz  
 FREQN 0.44246 ppm/cm  
 BICOE 241.3323 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling

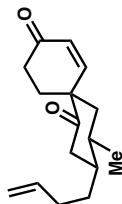


Current Data Parameters  
 USER tooren  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2015.11.11  
 Time 15:35  
 PROBHD 5 mm CPYCH-1H  
 PULPROG zgpg30  
 SOLVENT CDCl3  
 DS 2  
 B1 2  
 B2 2  
 FIDRES 0.226026 Hz  
 AQ 1.9998451 sec  
 RM 62.400 usec  
 DM 0.122880 K  
 TE 298.2 K  
 T1 0.10000000 sec  
 T2 0.10000000 sec  
 T3 0.10000000 sec  
 T4 0.10000000 sec  
 T5 0.10000000 sec  
 T6 0.10000000 sec  
 T7 0.10000000 sec  
 T8 0.10000000 sec  
 T9 0.10000000 sec  
 T10 0.10000000 sec  
 T11 0.10000000 sec  
 T12 0.10000000 sec  
 T13 0.10000000 sec  
 T14 0.10000000 sec  
 T15 0.10000000 sec  
 T16 0.10000000 sec  
 T17 0.10000000 sec  
 T18 0.10000000 sec  
 T19 0.10000000 sec  
 T20 0.10000000 sec  
 T21 0.10000000 sec  
 T22 0.10000000 sec  
 T23 0.10000000 sec  
 T24 0.10000000 sec  
 T25 0.10000000 sec  
 T26 0.10000000 sec  
 T27 0.10000000 sec  
 T28 0.10000000 sec  
 T29 0.10000000 sec  
 T30 0.10000000 sec  
 T31 0.10000000 sec  
 T32 0.10000000 sec  
 T33 0.10000000 sec  
 T34 0.10000000 sec  
 T35 0.10000000 sec  
 T36 0.10000000 sec  
 T37 0.10000000 sec  
 T38 0.10000000 sec  
 T39 0.10000000 sec  
 T40 0.10000000 sec  
 T41 0.10000000 sec  
 T42 0.10000000 sec  
 T43 0.10000000 sec  
 T44 0.10000000 sec  
 T45 0.10000000 sec  
 T46 0.10000000 sec  
 T47 0.10000000 sec  
 T48 0.10000000 sec  
 T49 0.10000000 sec  
 T50 0.10000000 sec  
 T51 0.10000000 sec  
 T52 0.10000000 sec  
 T53 0.10000000 sec  
 T54 0.10000000 sec  
 T55 0.10000000 sec  
 T56 0.10000000 sec  
 T57 0.10000000 sec  
 T58 0.10000000 sec  
 T59 0.10000000 sec  
 T60 0.10000000 sec  
 T61 0.10000000 sec  
 T62 0.10000000 sec  
 T63 0.10000000 sec  
 T64 0.10000000 sec  
 T65 0.10000000 sec  
 T66 0.10000000 sec  
 T67 0.10000000 sec  
 T68 0.10000000 sec  
 T69 0.10000000 sec  
 T70 0.10000000 sec  
 T71 0.10000000 sec  
 T72 0.10000000 sec  
 T73 0.10000000 sec  
 T74 0.10000000 sec  
 T75 0.10000000 sec  
 T76 0.10000000 sec  
 T77 0.10000000 sec  
 T78 0.10000000 sec  
 T79 0.10000000 sec  
 T80 0.10000000 sec  
 T81 0.10000000 sec  
 T82 0.10000000 sec  
 T83 0.10000000 sec  
 T84 0.10000000 sec  
 T85 0.10000000 sec  
 T86 0.10000000 sec  
 T87 0.10000000 sec  
 T88 0.10000000 sec  
 T89 0.10000000 sec  
 T90 0.10000000 sec  
 T91 0.10000000 sec  
 T92 0.10000000 sec  
 T93 0.10000000 sec  
 T94 0.10000000 sec  
 T95 0.10000000 sec  
 T96 0.10000000 sec  
 T97 0.10000000 sec  
 T98 0.10000000 sec  
 T99 0.10000000 sec  
 T100 0.10000000 sec  
 F2 - Processing parameters  
 SI 65536 Hz  
 SF 500.260781 MHz  
 WDW EM  
 SSB 0  
 GB 0  
 CB 0  
 PC 4.00





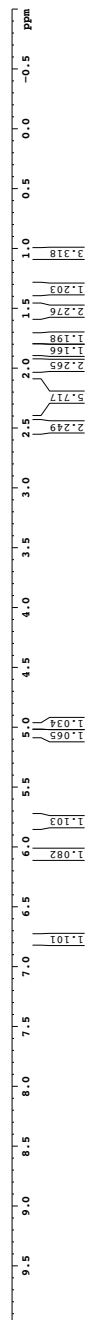
**<sup>1</sup>H spectrum**



Current Data Parameters  
 USER toosen  
 EXPNO 1  
 PROCNO 1  
 P2 - Acquisition Parameters  
 Date\_ 201012  
 Time 10:12  
 PROBHD 5 mm CPYCH-1H  
 PULPROG zgpg30  
 TOUPROG 2D  
 SOLVENT CDCl<sub>3</sub>  
 DS 2  
 B1 2  
 B2 2  
 F1 601.62 MHz  
 F2 601.62 MHz  
 F3 601.62 MHz  
 P1 7.50 usec  
 P2 7.50 usec  
 P3 7.50 usec  
 SFO1 500.22735015 MHz  
 SFO2 500.22735015 MHz  
 SFO3 500.22735015 MHz  
 P2 - Processing parameters  
 SI 65536  
 SF 500.22735015 MHz  
 WDW EM  
 SSB 0  
 LB 0.30 Hz  
 GB 0  
 PC 4.00

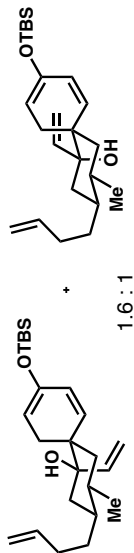
1.031  
1.044  
1.024  
1.337  
1.347  
1.500  
1.520  
1.547  
1.573  
1.619  
1.768  
1.838  
1.963  
1.984  
1.991  
2.146  
2.215  
2.245  
2.262  
2.272  
2.306  
2.320  
2.340  
2.356  
2.466  
2.474  
2.496  
2.504  
2.525

4.980  
5.000  
5.064  
5.746  
5.760  
5.780  
5.798  
5.814  
5.827  
5.850  
5.871  
6.783  
6.784  
7.269



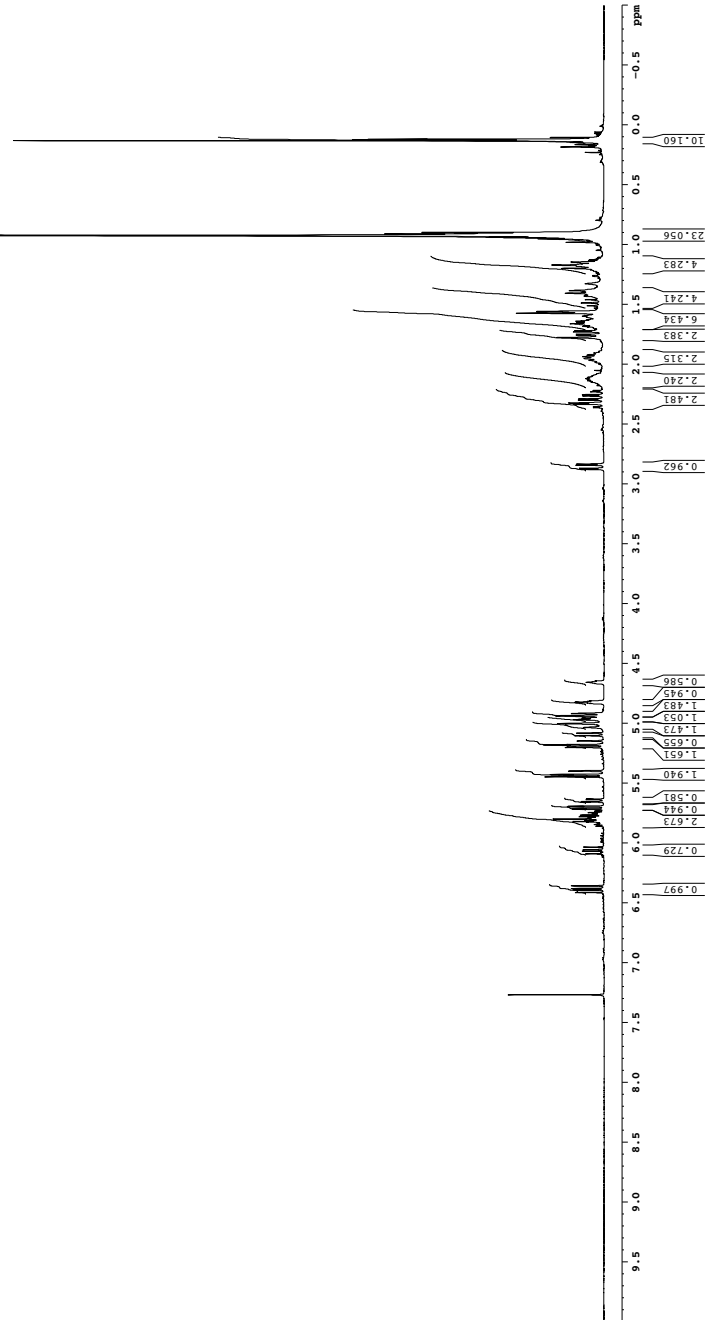


1H spectrum



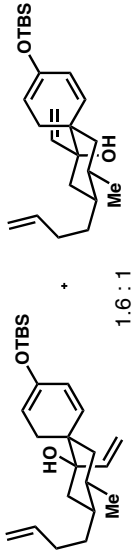
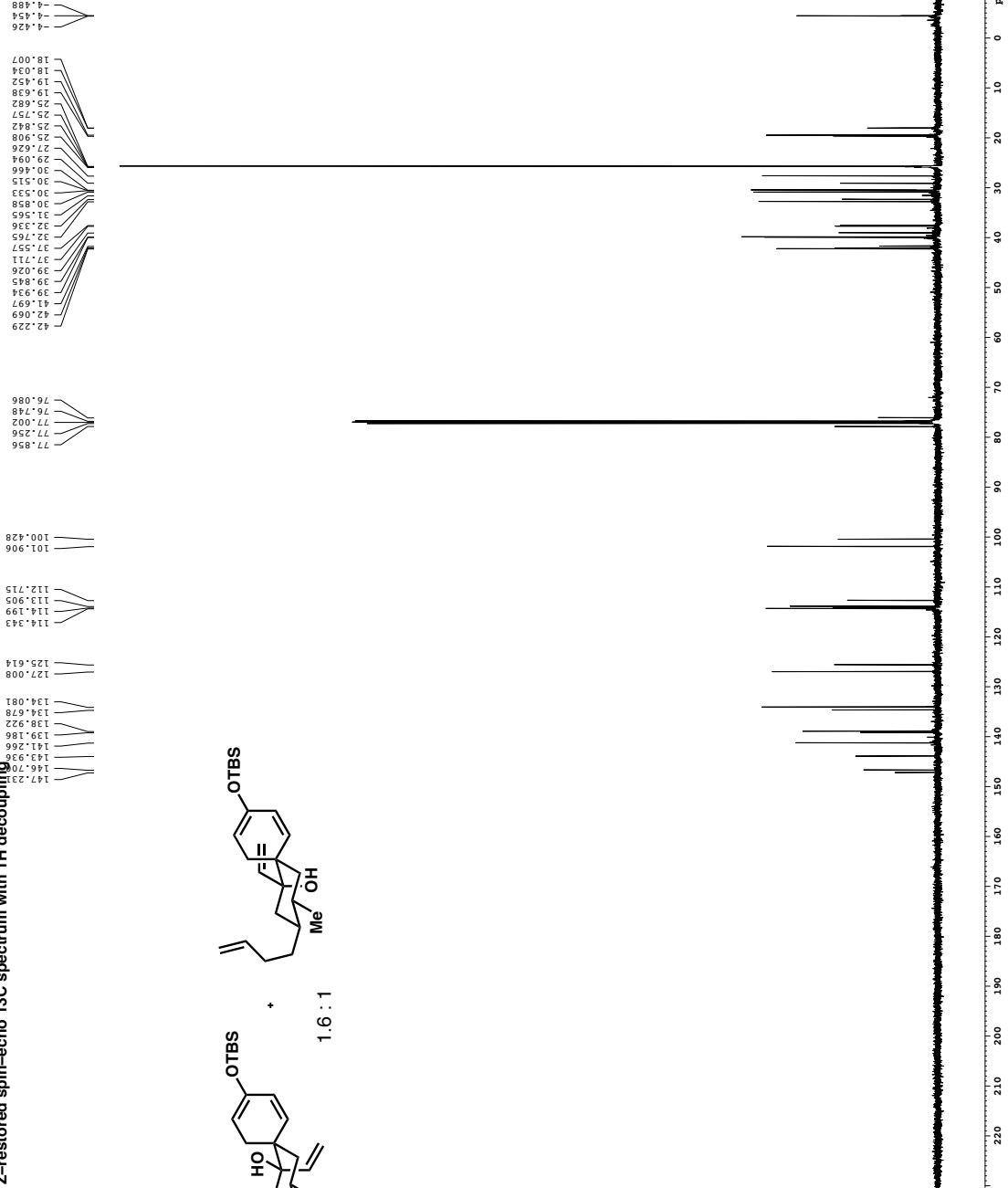
Current Data Parameters  
USER toosen  
EXPNO p01.237\_70914  
PROCNO 1  
F2 - Acquisition Parameters  
Date\_ 2010.03.16  
Time 10.36  
INSTRUM spect  
PROBHD 5 mm CPYCH 1H-  
PULPROG zgpg30  
TD 65536  
SOLVENT CDCl3  
DS 2  
AQ 0.12000000 sec  
RG 655.20000000 Hz  
FIDRES 0.11000000 Hz  
AQ 4.9999123 sec  
RG 62.40000000 Hz  
DM 0.10000000 sec  
TE 300.2 K  
FREQ 500.136260000 MHz  
NUC1 1H  
NUC2 1H  
PC 4.00

F2 - Processing parameters  
SI 65536  
SF 500.136260000 MHz  
WDW EM  
SSB 0  
GB 0  
PC 4.00



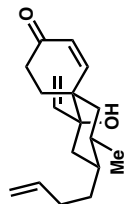
Z-restored spin-echo 13C spectrum with 1H decoupling

```
===== Acquisition Parameters =====
USRR          PULPROG  zgpg30
PROBHD        5 mm CPYCH 1H-1
P2            0.15000000
PT            2.00000000
FIDRES       0.01300000
AQ           1.00000000
RG           327.50
DM           16.5000 uSsec
TE           300.2 K
TR           0.50000000
D1           0.05000000
DELTA        0.00000000
NUC1          13C
NUC2          13C
NUC3          13C
P1           1.5000 uSsec
P2           1.5000 uSsec
PC           10.00 dB
PD           12.00 dB
PL12         0.00
PL12         0.00
PL12         0.00
PL12         0.00
===== Gradient Channels =====
GR1NAME      G1
GR1UNIT      V
GR1SCALE     1.0000
GR1OFF       0.0000
GR2NAME      G2
GR2UNIT      V
GR2SCALE     1.0000
GR2OFF       0.0000
===== Processing Parameters =====
SI           65536
SF           125.762200 MHz
WDW          EM
SSB          0
GB           0
PC           2.00
```

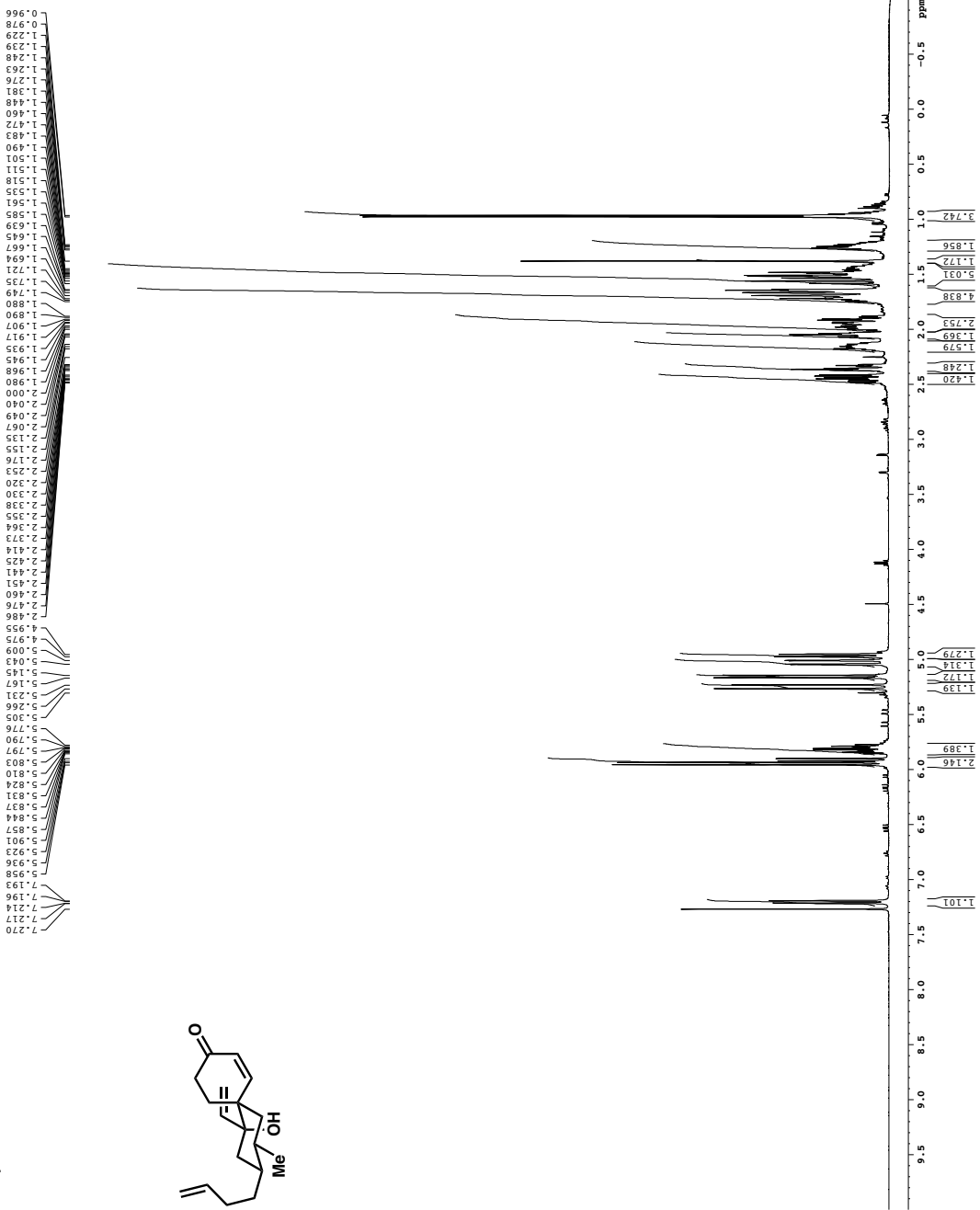




**<sup>1</sup>H spectrum**



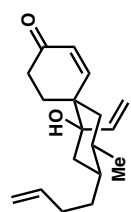
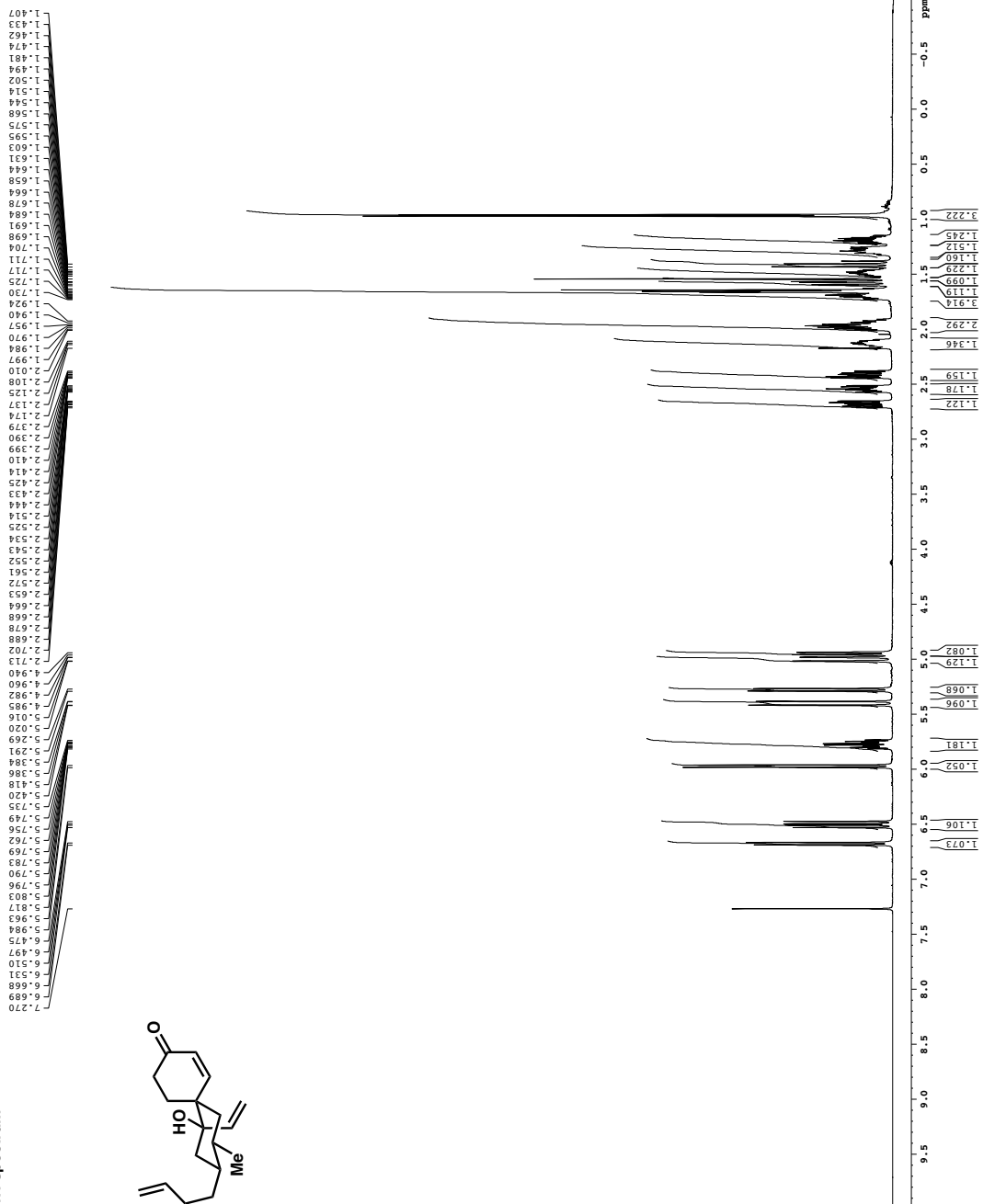
Current Data Parameters  
 USER toonen  
 EXPNO 1  
 PROCNO 1  
 P2 - Acquisition Parameters  
 Date\_ 2015.07.27  
 Time 15.27  
 INSTRUM spect  
 PROBM 5 mm CPYX  
 PULPROG zgpg30  
 SOLVENT CDCl3  
 DS 2  
 EQ 2  
 FIDRES 0.110003 Hz  
 AQ 4.9999123 sec  
 RM 62.400 usec  
 DM 0.100000 sec  
 TE 298.2 K  
 TD 65536  
 SFO1 500.136260 MHz  
 NUC1 13C  
 NUC2 13C  
 ACQRES 0.0130000 sec  
 CHANDEL E1  
 P1 7.50 usec  
 SFO2 500.1273505 MHz  
 P2 - Processing parameters  
 SI 65536  
 SF 500.136260 MHz  
 DS 4  
 GB 0  
 CB 4.00



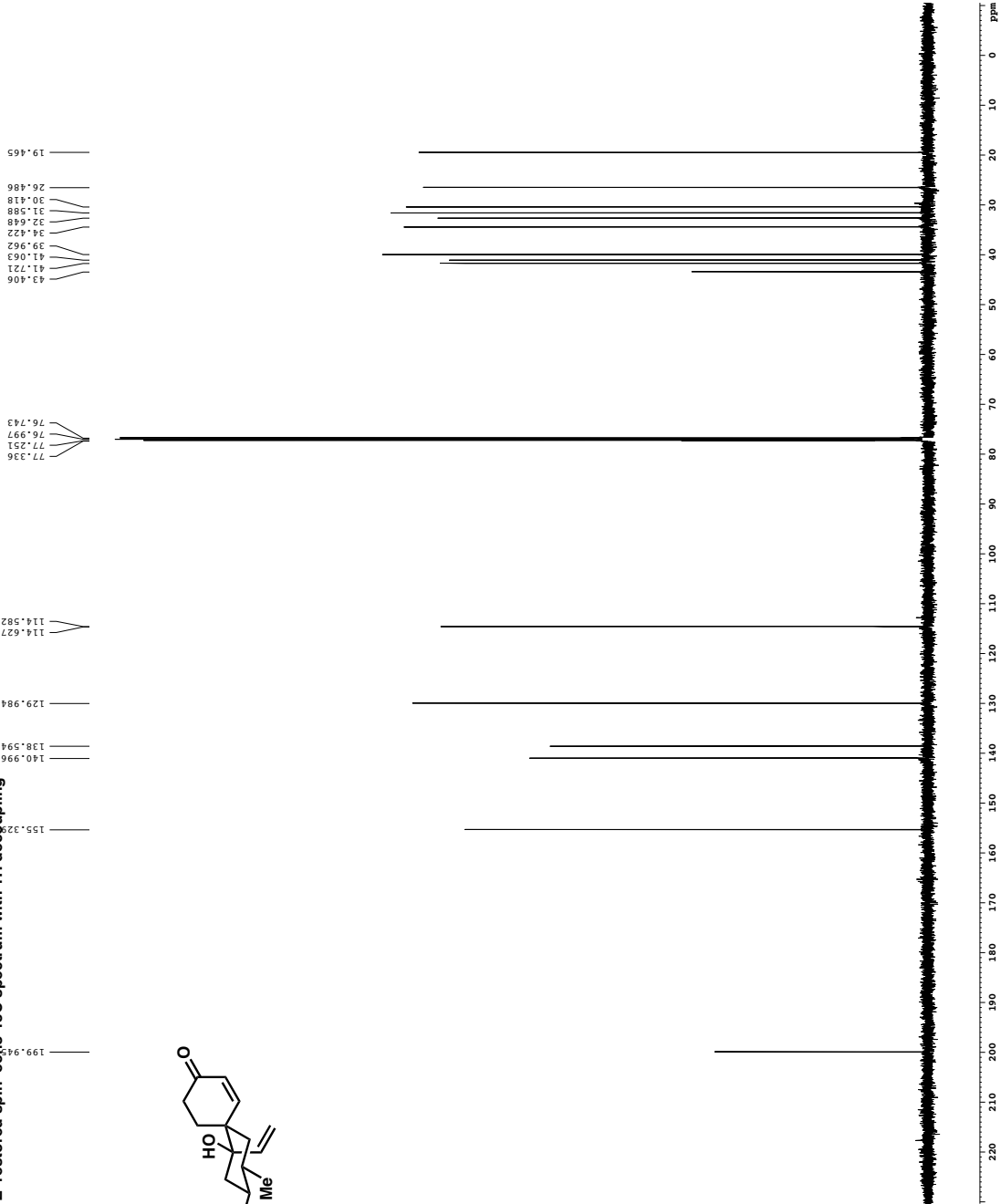
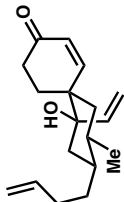


**<sup>1</sup>H spectrum**

Current Data Parameters  
 USER toonen  
 EXPNO 1  
 PROCNO 1  
 P1 1.00  
 P2 - Acquisition Parameters  
 T1 17.33  
 T2 17.33  
 F1 500.136199  
 F2 500.136199  
 AQ 0.01300000 sec  
 SOLVENT CDCl3  
 DS 2  
 BS 0.11  
 EQ 0.100000 Hz  
 FIDRES 4.9999123 sec  
 AQ 62.400 usec  
 DM 0.29820 K  
 TE 0.10000000 sec  
 ACQRES 0.10000000 sec  
 MCRES 0.01300000 sec  
 ===== CHANNEL f1 =====  
 P1 7.50 usec  
 P2 7.50 usec  
 SFO1 500.136199 MHz  
 ===== CHANNEL f2 =====  
 P1 7.50 usec  
 P2 7.50 usec  
 SFO2 500.136199 MHz  
 ===== CHANNEL f3 =====  
 P1 7.50 usec  
 P2 7.50 usec  
 SFO3 500.136199 MHz



Z-restored spin-echo 13C spectrum with 1H decoupling

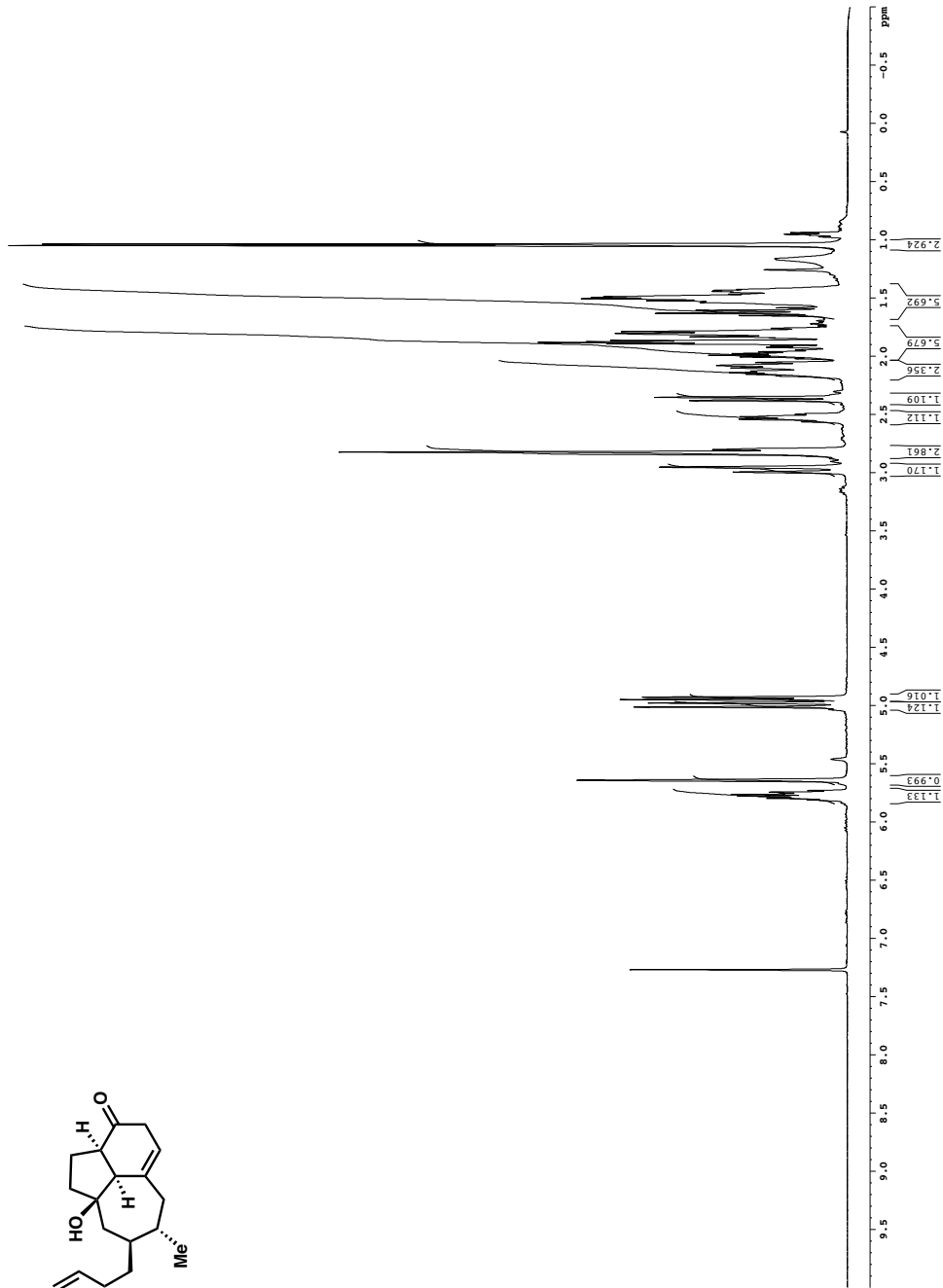
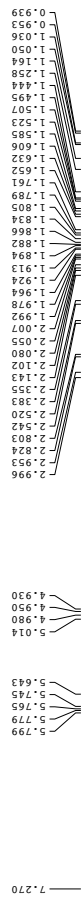
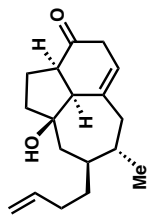


```

Current Data Parameters
NAME            C2020001
EXPNO           2
PROCNO          1
PR - Acquisition Parameters
Time           21.737
Date_          20170905
PROBHD         5 mm CPDCL 1H
PULPROG        zgpg30
SOLVENT        CDCl3
DS             3
F2 - Processing Parameters
SI              65536
SF             125.7604366
AQ             0.390625
SFO2           50.000
WDW            EM
SSB            0
GB             0
PC             2.00
===== CHANNEL F1 =====
NUC1            13
PCPD1          15.30
PCPD2          500.00
PFL1           19.20
PFL2           12.00
SFO1           125.7604366
SR1            3.20
SRMH1          C13P60.0.5.20.1
SFO2           50.000
SRMH2          C13P60.0.5.20.1
SFO3           125.7604366
===== CHANNEL F2 =====
NUC2            13
PCPD2          15.30
PCPD4          500.00
PFL2           19.20
PFL4           12.00
SFO2           125.7604366
SR2            3.20
SRMH1          C13P60.0.5.20.1
SFO3           50.000
SRMH2          C13P60.0.5.20.1
SFO4           125.7604366
===== CHANNEL F3 =====
NAME            CUMANEOL
EXPNO           1
PROCNO          1
PR - Acquisition Parameters
Time           21.737
Date_          20170905
PROBHD         5 mm CPDCL 1H
PULPROG        zgpg30
SOLVENT        CDCl3
DS             3
F2 - Processing Parameters
SI              65536
SF             125.7604366
AQ             0.390625
SFO2           50.000
WDW            EM
SSB            0
GB             0
PC             2.00
===== CHANNEL F1 =====
NUC1            13
PCPD1          15.30
PCPD2          500.00
PFL1           19.20
PFL2           12.00
SFO1           125.7604366
SR1            3.20
SRMH1          C13P60.0.5.20.1
SFO2           50.000
SRMH2          C13P60.0.5.20.1
SFO3           125.7604366
===== CHANNEL F2 =====
NAME            CUMANEOL
EXPNO           1
PROCNO          1
PR - Acquisition Parameters
Time           21.737
Date_          20170905
PROBHD         5 mm CPDCL 1H
PULPROG        zgpg30
SOLVENT        CDCl3
DS             3
F2 - Processing Parameters
SI              65536
SF             125.7604366
AQ             0.390625
SFO2           50.000
WDW            EM
SSB            0
GB             0
PC             2.00

```

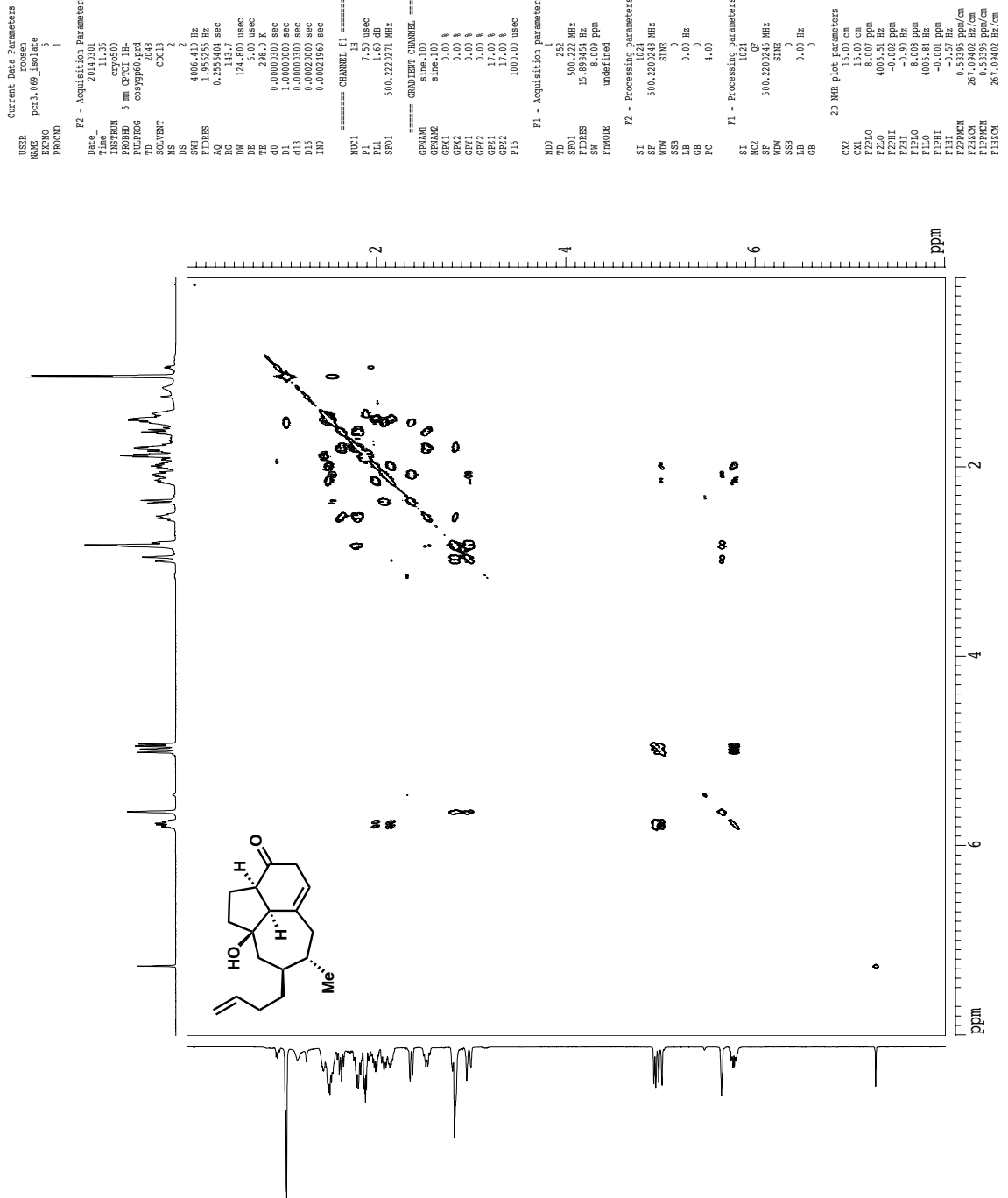
1H spectrum



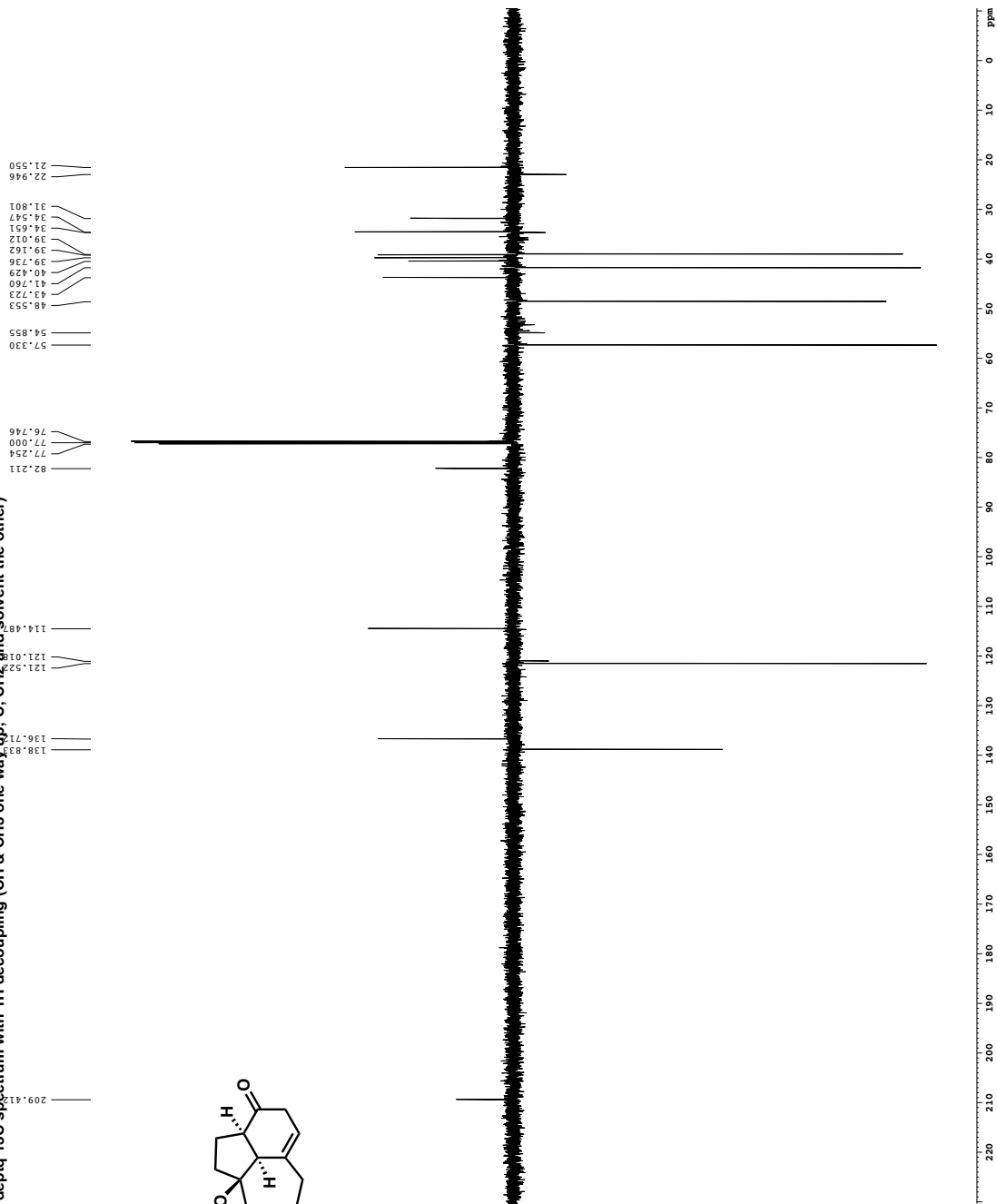
Current Data Parameters  
 USER toonen  
 EXPNO 1  
 PROCNO 1  
 P2 - Acquisition Parameters  
 Title 11110  
 Date\_ 2011.10  
 Time 20:00  
 PROBM 9 mm CPTCL 1H  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 DS 2  
 BS 1  
 F1 613.812 MHz  
 F2 220026.000 MHz  
 FIDRES 1.9998451 sec  
 AQ 62.400 usec  
 RM 1.00000000  
 TE 300.2 K  
 TR 0.12880000 sec  
 WDELTA 0.10000000 sec  
 WDELT2 0.01500000 sec  
 CHANNEL1 CHANDEL.EI  
 NU1 1H 500.136052 MHz  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.136051 MHz  
 P2 - Processing parameters  
 SI 655536 usec  
 SF 500.136052 MHz  
 WDW EM  
 GB 0  
 CB 4.00



gc0sy60



deptq 13C spectrum with 1H decoupling (CH & CH3 one way up; C, CH2 and solvent the other)



```

Current Data Parameters
=====
USRR          pcr3_00p_1decou
EXPNO         1
PROCNO        1
F2 - Acquisition Parameters
=====
Time          21.119
Date_         06/11/99
Time2         06/11/99
PROBHD        5 mm CPCLC 1H-
PULPROG       zgpg30
DECOUPLING    deptq30
SOLVENT       CDCl3
NS            3452
DS            2
FIDRES        0.442368 Hz
AQ            1.0683251 sec
RG            16.500
AQ            16.500 usec
RM            0.0000000
TE            300.2
TR            1.0
SFO           101.253 MHz
C1P1         15.5000000
C1P2         1.0000000
C1P3         1.0000000
C1P4         0.0000000
C1P5         0.0000000
C1P6         0.0000000
C1P7         0.0000000
C1P8         0.0000000
C1P9         0.0000000
C1P10        0.0000000
C1P11        0.0000000
C1P12        0.0000000
C1P13        0.0000000
C1P14        0.0000000
C1P15        0.0000000
C1P16        0.0000000
C1P17        0.0000000
C1P18        0.0000000
C1P19        0.0000000
C1P20        0.0000000
=====
Channel f1
=====
NUC1          13C
P1            15.50 usec
PL1           0.00 dB
PC1           1.00 usec
=====
Channel f2
=====
NUC2          13C
P2            15.50 usec
PL2           0.00 dB
PC2           1.00 usec
=====
Channel f3
=====
NUC3          13C
P3            15.50 usec
PL3           0.00 dB
PC3           1.00 usec
=====
Channel f4
=====
NUC4          13C
P4            15.50 usec
PL4           0.00 dB
PC4           1.00 usec
=====
Channel f5
=====
NUC5          13C
P5            15.50 usec
PL5           0.00 dB
PC5           1.00 usec
=====
Channel f6
=====
NUC6          13C
P6            15.50 usec
PL6           0.00 dB
PC6           1.00 usec
=====
Channel f7
=====
NUC7          13C
P7            15.50 usec
PL7           0.00 dB
PC7           1.00 usec
=====
Channel f8
=====
NUC8          13C
P8            15.50 usec
PL8           0.00 dB
PC8           1.00 usec
=====
Channel f9
=====
NUC9          13C
P9            15.50 usec
PL9           0.00 dB
PC9           1.00 usec
=====
Channel f10
=====
NUC10         13C
P10           15.50 usec
PL10          0.00 dB
PC10          1.00 usec
=====
Channel f11
=====
NUC11         13C
P11           15.50 usec
PL11          0.00 dB
PC11          1.00 usec
=====
Channel f12
=====
NUC12         13C
P12           15.50 usec
PL12          0.00 dB
PC12          1.00 usec
=====
Channel f13
=====
NUC13         13C
P13           15.50 usec
PL13          0.00 dB
PC13          1.00 usec
=====
Channel f14
=====
NUC14         13C
P14           15.50 usec
PL14          0.00 dB
PC14          1.00 usec
=====
Channel f15
=====
NUC15         13C
P15           15.50 usec
PL15          0.00 dB
PC15          1.00 usec
=====
Channel f16
=====
NUC16         13C
P16           15.50 usec
PL16          0.00 dB
PC16          1.00 usec
=====
Channel f17
=====
NUC17         13C
P17           15.50 usec
PL17          0.00 dB
PC17          1.00 usec
=====
Channel f18
=====
NUC18         13C
P18           15.50 usec
PL18          0.00 dB
PC18          1.00 usec
=====
Channel f19
=====
NUC19         13C
P19           15.50 usec
PL19          0.00 dB
PC19          1.00 usec
=====
Channel f20
=====
NUC20         13C
P20           15.50 usec
PL20          0.00 dB
PC20          1.00 usec
=====
Channel f21
=====
NUC21         13C
P21           15.50 usec
PL21          0.00 dB
PC21          1.00 usec
=====
Channel f22
=====
NUC22         13C
P22           15.50 usec
PL22          0.00 dB
PC22          1.00 usec
=====
Channel f23
=====
NUC23         13C
P23           15.50 usec
PL23          0.00 dB
PC23          1.00 usec
=====
Channel f24
=====
NUC24         13C
P24           15.50 usec
PL24          0.00 dB
PC24          1.00 usec
=====
Channel f25
=====
NUC25         13C
P25           15.50 usec
PL25          0.00 dB
PC25          1.00 usec
=====
Channel f26
=====
NUC26         13C
P26           15.50 usec
PL26          0.00 dB
PC26          1.00 usec
=====
Channel f27
=====
NUC27         13C
P27           15.50 usec
PL27          0.00 dB
PC27          1.00 usec
=====
Channel f28
=====
NUC28         13C
P28           15.50 usec
PL28          0.00 dB
PC28          1.00 usec
=====
Channel f29
=====
NUC29         13C
P29           15.50 usec
PL29          0.00 dB
PC29          1.00 usec
=====
Channel f30
=====
NUC30         13C
P30           15.50 usec
PL30          0.00 dB
PC30          1.00 usec
=====
Channel f31
=====
NUC31         13C
P31           15.50 usec
PL31          0.00 dB
PC31          1.00 usec
=====
Channel f32
=====
NUC32         13C
P32           15.50 usec
PL32          0.00 dB
PC32          1.00 usec
=====
Channel f33
=====
NUC33         13C
P33           15.50 usec
PL33          0.00 dB
PC33          1.00 usec
=====
Channel f34
=====
NUC34         13C
P34           15.50 usec
PL34          0.00 dB
PC34          1.00 usec
=====
Channel f35
=====
NUC35         13C
P35           15.50 usec
PL35          0.00 dB
PC35          1.00 usec
=====
Channel f36
=====
NUC36         13C
P36           15.50 usec
PL36          0.00 dB
PC36          1.00 usec
=====
Channel f37
=====
NUC37         13C
P37           15.50 usec
PL37          0.00 dB
PC37          1.00 usec
=====
Channel f38
=====
NUC38         13C
P38           15.50 usec
PL38          0.00 dB
PC38          1.00 usec
=====
Channel f39
=====
NUC39         13C
P39           15.50 usec
PL39          0.00 dB
PC39          1.00 usec
=====
Channel f40
=====
NUC40         13C
P40           15.50 usec
PL40          0.00 dB
PC40          1.00 usec
=====
Channel f41
=====
NUC41         13C
P41           15.50 usec
PL41          0.00 dB
PC41          1.00 usec
=====
Channel f42
=====
NUC42         13C
P42           15.50 usec
PL42          0.00 dB
PC42          1.00 usec
=====
Channel f43
=====
NUC43         13C
P43           15.50 usec
PL43          0.00 dB
PC43          1.00 usec
=====
Channel f44
=====
NUC44         13C
P44           15.50 usec
PL44          0.00 dB
PC44          1.00 usec
=====
Channel f45
=====
NUC45         13C
P45           15.50 usec
PL45          0.00 dB
PC45          1.00 usec
=====
Channel f46
=====
NUC46         13C
P46           15.50 usec
PL46          0.00 dB
PC46          1.00 usec
=====
Channel f47
=====
NUC47         13C
P47           15.50 usec
PL47          0.00 dB
PC47          1.00 usec
=====
Channel f48
=====
NUC48         13C
P48           15.50 usec
PL48          0.00 dB
PC48          1.00 usec
=====
Channel f49
=====
NUC49         13C
P49           15.50 usec
PL49          0.00 dB
PC49          1.00 usec
=====
Channel f50
=====
NUC50         13C
P50           15.50 usec
PL50          0.00 dB
PC50          1.00 usec
=====
Channel f51
=====
NUC51         13C
P51           15.50 usec
PL51          0.00 dB
PC51          1.00 usec
=====
Channel f52
=====
NUC52         13C
P52           15.50 usec
PL52          0.00 dB
PC52          1.00 usec
=====
Channel f53
=====
NUC53         13C
P53           15.50 usec
PL53          0.00 dB
PC53          1.00 usec
=====
Channel f54
=====
NUC54         13C
P54           15.50 usec
PL54          0.00 dB
PC54          1.00 usec
=====
Channel f55
=====
NUC55         13C
P55           15.50 usec
PL55          0.00 dB
PC55          1.00 usec
=====
Channel f56
=====
NUC56         13C
P56           15.50 usec
PL56          0.00 dB
PC56          1.00 usec
=====
Channel f57
=====
NUC57         13C
P57           15.50 usec
PL57          0.00 dB
PC57          1.00 usec
=====
Channel f58
=====
NUC58         13C
P58           15.50 usec
PL58          0.00 dB
PC58          1.00 usec
=====
Channel f59
=====
NUC59         13C
P59           15.50 usec
PL59          0.00 dB
PC59          1.00 usec
=====
Channel f60
=====
NUC60         13C
P60           15.50 usec
PL60          0.00 dB
PC60          1.00 usec
=====
Channel f61
=====
NUC61         13C
P61           15.50 usec
PL61          0.00 dB
PC61          1.00 usec
=====
Channel f62
=====
NUC62         13C
P62           15.50 usec
PL62          0.00 dB
PC62          1.00 usec
=====
Channel f63
=====
NUC63         13C
P63           15.50 usec
PL63          0.00 dB
PC63          1.00 usec
=====
Channel f64
=====
NUC64         13C
P64           15.50 usec
PL64          0.00 dB
PC64          1.00 usec
=====
Channel f65
=====
NUC65         13C
P65           15.50 usec
PL65          0.00 dB
PC65          1.00 usec
=====
Channel f66
=====
NUC66         13C
P66           15.50 usec
PL66          0.00 dB
PC66          1.00 usec
=====
Channel f67
=====
NUC67         13C
P67           15.50 usec
PL67          0.00 dB
PC67          1.00 usec
=====
Channel f68
=====
NUC68         13C
P68           15.50 usec
PL68          0.00 dB
PC68          1.00 usec
=====
Channel f69
=====
NUC69         13C
P69           15.50 usec
PL69          0.00 dB
PC69          1.00 usec
=====
Channel f70
=====
NUC70         13C
P70           15.50 usec
PL70          0.00 dB
PC70          1.00 usec
=====
Channel f71
=====
NUC71         13C
P71           15.50 usec
PL71          0.00 dB
PC71          1.00 usec
=====
Channel f72
=====
NUC72         13C
P72           15.50 usec
PL72          0.00 dB
PC72          1.00 usec
=====
Channel f73
=====
NUC73         13C
P73           15.50 usec
PL73          0.00 dB
PC73          1.00 usec
=====
Channel f74
=====
NUC74         13C
P74           15.50 usec
PL74          0.00 dB
PC74          1.00 usec
=====
Channel f75
=====
NUC75         13C
P75           15.50 usec
PL75          0.00 dB
PC75          1.00 usec
=====
Channel f76
=====
NUC76         13C
P76           15.50 usec
PL76          0.00 dB
PC76          1.00 usec
=====
Channel f77
=====
NUC77         13C
P77           15.50 usec
PL77          0.00 dB
PC77          1.00 usec
=====
Channel f78
=====
NUC78         13C
P78           15.50 usec
PL78          0.00 dB
PC78          1.00 usec
=====
Channel f79
=====
NUC79         13C
P79           15.50 usec
PL79          0.00 dB
PC79          1.00 usec
=====
Channel f80
=====
NUC80         13C
P80           15.50 usec
PL80          0.00 dB
PC80          1.00 usec
=====
Channel f81
=====
NUC81         13C
P81           15.50 usec
PL81          0.00 dB
PC81          1.00 usec
=====
Channel f82
=====
NUC82         13C
P82           15.50 usec
PL82          0.00 dB
PC82          1.00 usec
=====
Channel f83
=====
NUC83         13C
P83           15.50 usec
PL83          0.00 dB
PC83          1.00 usec
=====
Channel f84
=====
NUC84         13C
P84           15.50 usec
PL84          0.00 dB
PC84          1.00 usec
=====
Channel f85
=====
NUC85         13C
P85           15.50 usec
PL85          0.00 dB
PC85          1.00 usec
=====
Channel f86
=====
NUC86         13C
P86           15.50 usec
PL86          0.00 dB
PC86          1.00 usec
=====
Channel f87
=====
NUC87         13C
P87           15.50 usec
PL87          0.00 dB
PC87          1.00 usec
=====
Channel f88
=====
NUC88         13C
P88           15.50 usec
PL88          0.00 dB
PC88          1.00 usec
=====
Channel f89
=====
NUC89         13C
P89           15.50 usec
PL89          0.00 dB
PC89          1.00 usec
=====
Channel f90
=====
NUC90         13C
P90           15.50 usec
PL90          0.00 dB
PC90          1.00 usec
=====
Channel f91
=====
NUC91         13C
P91           15.50 usec
PL91          0.00 dB
PC91          1.00 usec
=====
Channel f92
=====
NUC92         13C
P92           15.50 usec
PL92          0.00 dB
PC92          1.00 usec
=====
Channel f93
=====
NUC93         13C
P93           15.50 usec
PL93          0.00 dB
PC93          1.00 usec
=====
Channel f94
=====
NUC94         13C
P94           15.50 usec
PL94          0.00 dB
PC94          1.00 usec
=====
Channel f95
=====
NUC95         13C
P95           15.50 usec
PL95          0.00 dB
PC95          1.00 usec
=====
Channel f96
=====
NUC96         13C
P96           15.50 usec
PL96          0.00 dB
PC96          1.00 usec
=====
Channel f97
=====
NUC97         13C
P97           15.50 usec
PL97          0.00 dB
PC97          1.00 usec
=====
Channel f98
=====
NUC98         13C
P98           15.50 usec
PL98          0.00 dB
PC98          1.00 usec
=====
Channel f99
=====
NUC99         13C
P99           15.50 usec
PL99          0.00 dB
PC99          1.00 usec
=====
Channel f100
=====
NUC100        13C
P100          15.50 usec
PL100         0.00 dB
PC100         1.00 usec
=====
Processing Parameters
=====
SI            65536
SF            125.760253 MHz
WDW           EM
SSB           0
GB            0
PC            2.00
  
```



9hmcqc

Current Data Parameters  
 USER NAME pr1.06 isolate  
 F2PROG F2PROG  
 F2NAME 1

Date F2 - Acquisition Parameters  
 Time 11.22  
 INSTRN cpy500  
 PULPROG zgpg30  
 TO Z048  
 NS 2048  
 NS 2  
 CQ1 2  
 CQ2 2

DE 6.50 usec  
 AQ 0.2556404 sec  
 FIDRES 1.94625 Hz  
 SFO 300.200 MHz  
 DE 6.50 usec  
 AQ 0.2556404 sec  
 FIDRES 1.94625 Hz  
 SFO 300.200 MHz

CH22 145.000000 Hz  
 d0 0.0000000 sec  
 d1 0.0000000 sec  
 d2 0.0000000 sec  
 d3 0.0000000 sec  
 d4 0.0000000 sec  
 d5 0.0000000 sec  
 d6 0.0000000 sec  
 d7 0.0000000 sec  
 d8 0.0000000 sec  
 d9 0.0000000 sec  
 d10 0.0000000 sec  
 d11 0.0000000 sec  
 d12 0.0000000 sec  
 d13 0.0000000 sec  
 d14 0.0000000 sec  
 d15 0.0000000 sec  
 d16 0.0000000 sec  
 d17 0.0000000 sec  
 d18 0.0000000 sec  
 d19 0.0000000 sec  
 d20 0.0000000 sec

==== CHANNEL F1 =====  
 NUC1 1H  
 P1 15.00 usec  
 PL1 15.00 dB  
 SFO1 300.200000 MHz  
 CQ1PROG zgpg30  
 NUC2 13C  
 P2 15.50 usec  
 PL2 19.00 dB  
 PL12 11.38 dB  
 SFO2 125.762500 MHz

==== CHANNEL F2 =====  
 NUC1 13C  
 P1 15.00 usec  
 PL1 15.00 dB  
 SFO1 300.200000 MHz  
 CQ1PROG zgpg30  
 NUC2 13C  
 P2 15.50 usec  
 PL2 19.00 dB  
 PL12 11.38 dB  
 SFO2 125.762500 MHz

==== GRUEN CHANNEL =====  
 GRUEN1 13C  
 P1 15.00 usec  
 PL1 15.00 dB  
 SFO1 300.200000 MHz  
 GRUEN2 13C  
 P1 15.00 usec  
 PL1 15.00 dB  
 SFO1 300.200000 MHz  
 GRUEN3 13C  
 P1 15.00 usec  
 PL1 15.00 dB  
 SFO1 300.200000 MHz

F1 - Acquisition parameters  
 NS0 32  
 SFO 300.200000 MHz  
 FIDRES 94.339622 Hz  
 SFO2 125.762500 MHz  
 F2PROG zgpg30  
 F2NAME isolate.f2

F2 - Processing parameters  
 SI 300.200000 MHz  
 SF 300.200000 MHz  
 IN 0  
 LB 5.00 Hz  
 GB 0  
 PC 4.00

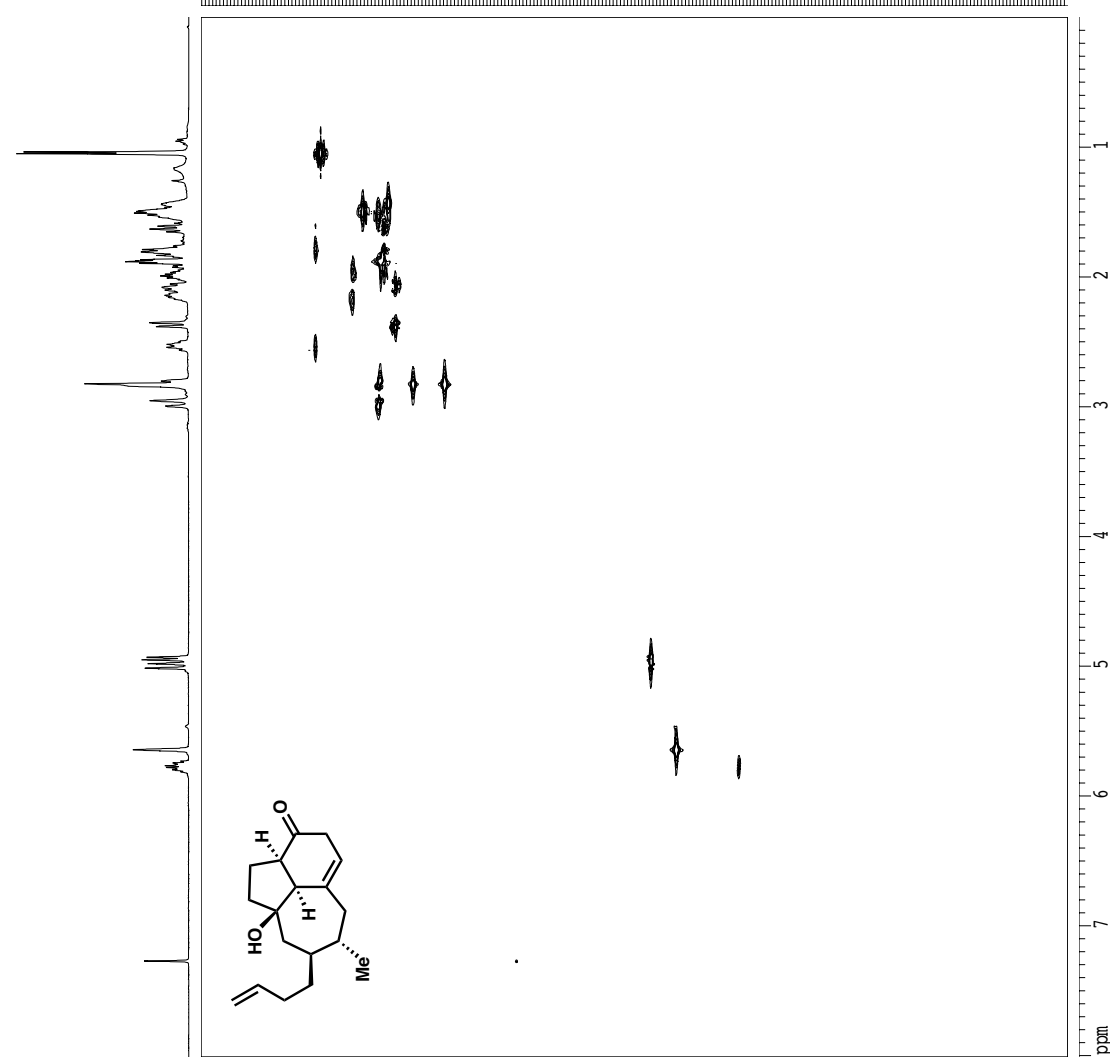
F1 - Acquisition parameters  
 NS0 32  
 SFO 300.200000 MHz  
 FIDRES 94.339622 Hz  
 SFO2 125.762500 MHz  
 F2PROG zgpg30  
 F2NAME isolate.f2

F2 - Processing parameters  
 SI 300.200000 MHz  
 SF 300.200000 MHz  
 IN 0  
 LB 5.00 Hz  
 GB 0  
 PC 4.00

F1 - Processing parameters  
 SI 300.200000 MHz  
 SF 300.200000 MHz  
 IN 0  
 LB 5.00 Hz  
 GB 0  
 PC 4.00

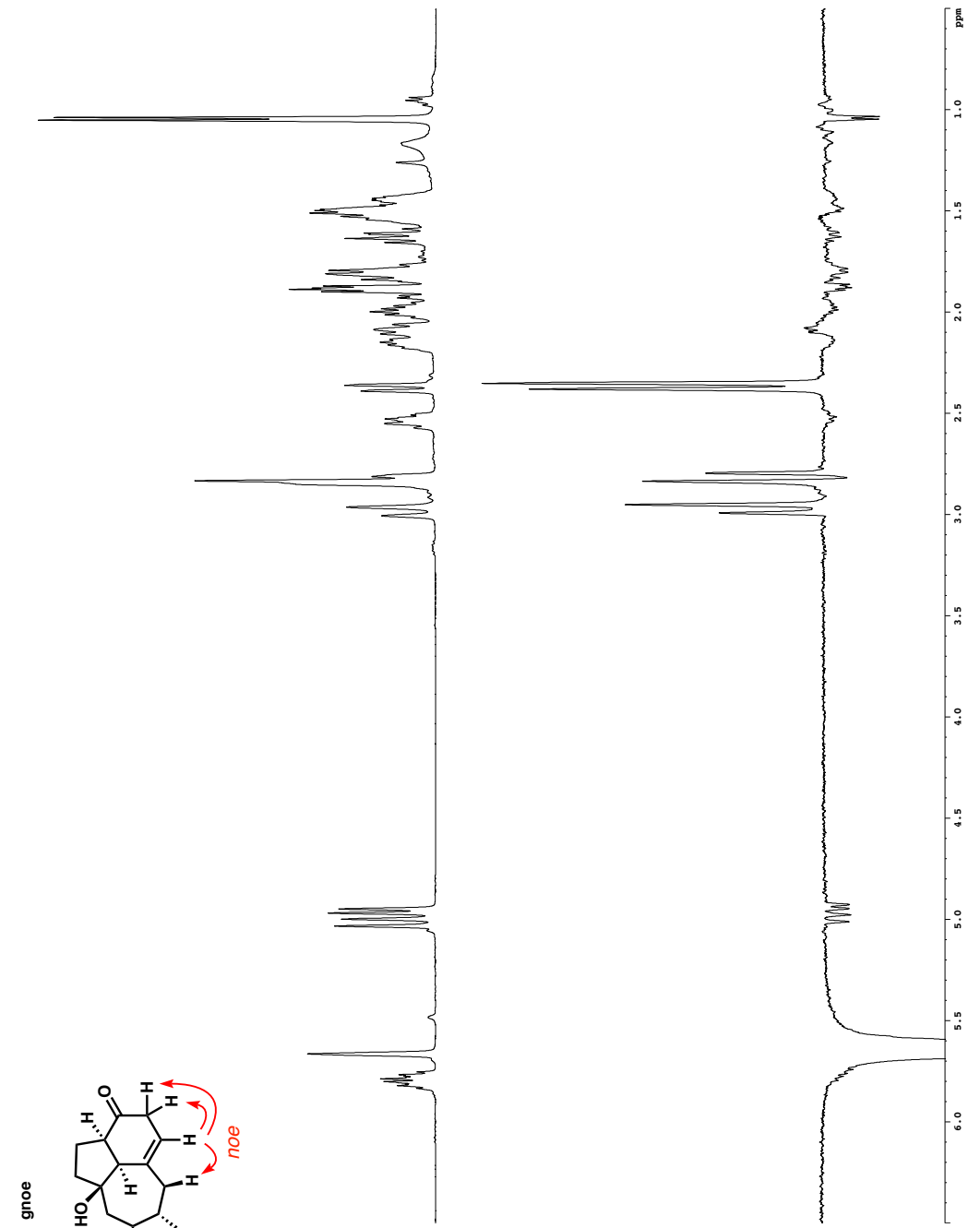
F2 - Processing parameters  
 SI 300.200000 MHz  
 SF 300.200000 MHz  
 IN 0  
 LB 5.00 Hz  
 GB 0  
 PC 4.00

2D NMR plot parameters  
 CQ1 2  
 CQ2 2  
 CH1 15.00 cm  
 CH2 15.00 cm  
 F2F0 8.016 ppm  
 F2F1 0.000000 Hz  
 F2F2 0.000000 Hz  
 F2F3 0.446 Hz  
 F2F4 2000.000000 Hz  
 F2F5 2000.000000 Hz  
 F2F6 -10.133 ppm  
 F2F7 -1274.332 Hz/cm  
 F2F8 222.51858 Hz/cm  
 F2F9 15.0073 ppm/cm  
 F2F10 2011.7049 Hz/cm

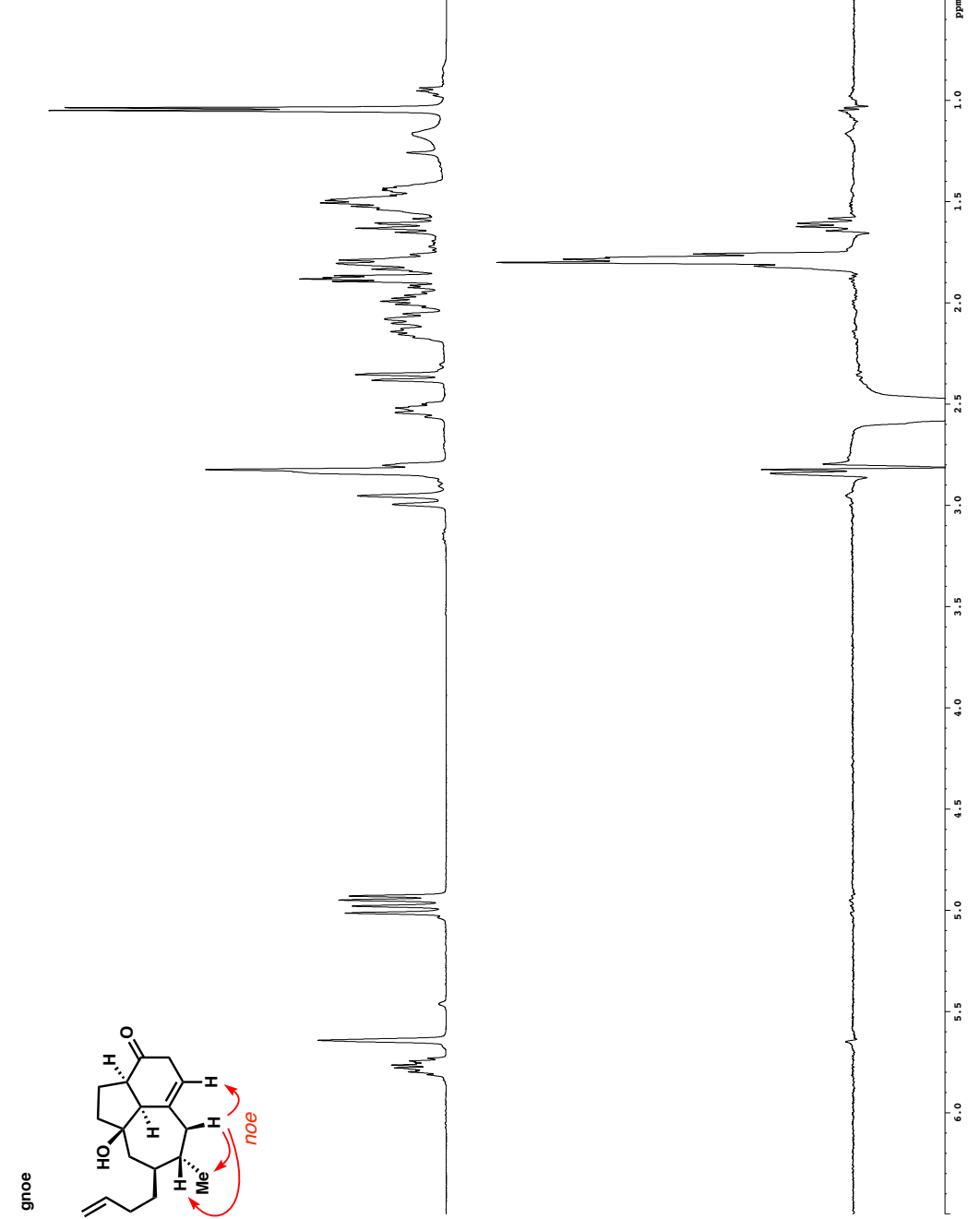


Current Data Parameters  
 USER: pcr3.009\_12oct06  
 EXPNO: 4  
 PROCNO: 1

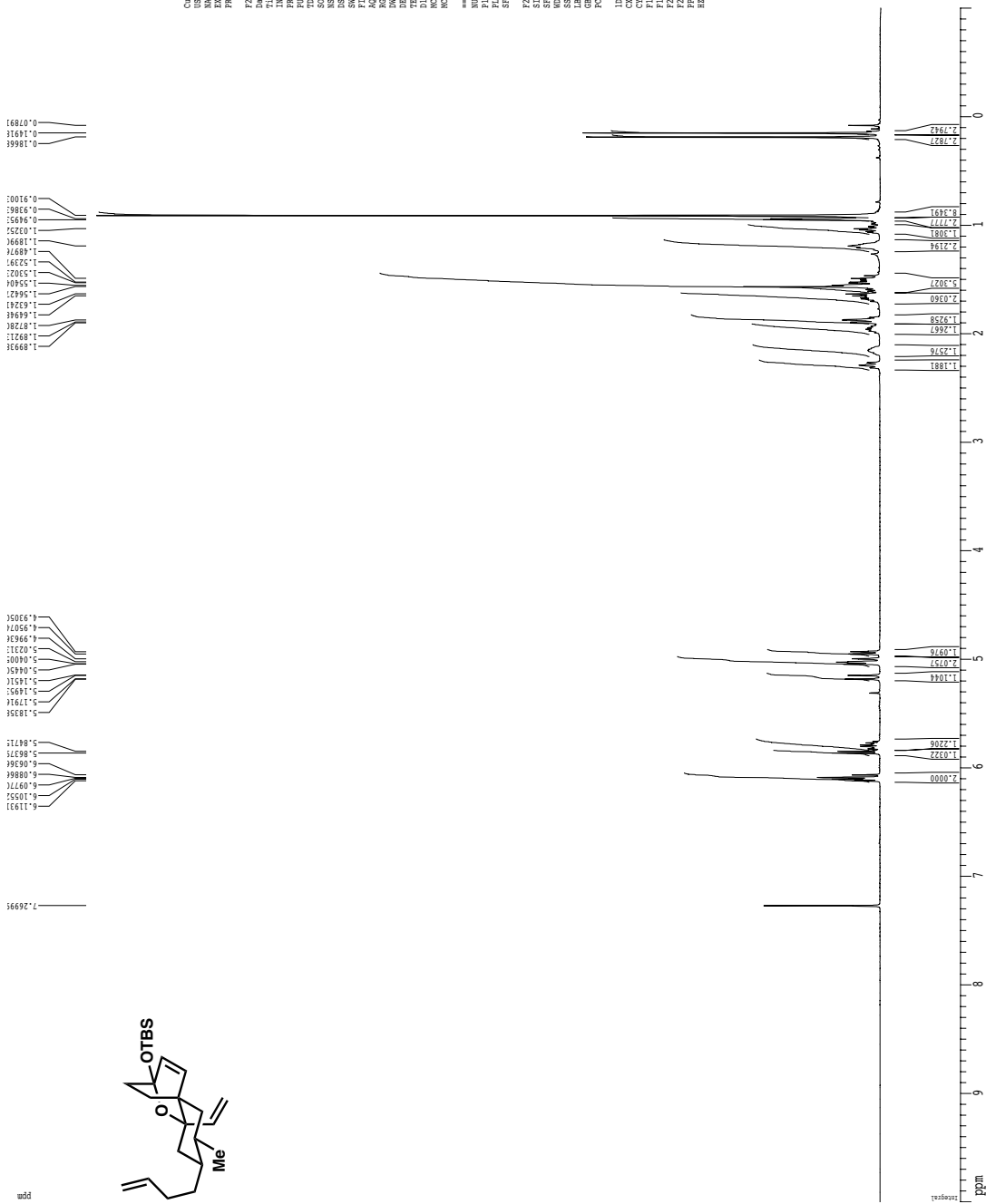
F2 - Acquisition Parameters  
 Date\_ Time: 2013.12.22 11:13:22  
 Time: 1.00000000 sec  
 PROBHD: 5 mm CPYCC1 1H-1  
 PULPROG: gnoe  
 TD: 65536  
 SOLVENT: CDCl3  
 DS: 4  
 B1: 0112.572 Hz  
 FIDRES: 0.1122266 Hz  
 AQ: 4.0889465 sec  
 RM: 62.400 usec  
 DE: 1.00000000 sec  
 TE: 300.2 K  
 TR: 1.0029820 sec  
 D1: 0.05000000 sec  
 D16: 0.05000000 sec  
 D17: 0.05000000 sec  
 D2: 0.16339699 sec  
 D22: 0.16339699 sec  
 D23: 0.16339699 sec  
 D24: 0.16339699 sec  
 D25: 0.16339699 sec  
 D26: 0.16339699 sec  
 D27: 0.16339699 sec  
 D28: 0.16339699 sec  
 D29: 0.16339699 sec  
 D30: 0.16339699 sec  
 D31: 0.16339699 sec  
 D32: 0.16339699 sec  
 D33: 0.16339699 sec  
 D34: 0.16339699 sec  
 D35: 0.16339699 sec  
 D36: 0.16339699 sec  
 D37: 0.16339699 sec  
 D38: 0.16339699 sec  
 D39: 0.16339699 sec  
 D40: 0.16339699 sec  
 D41: 0.16339699 sec  
 D42: 0.16339699 sec  
 D43: 0.16339699 sec  
 D44: 0.16339699 sec  
 D45: 0.16339699 sec  
 D46: 0.16339699 sec  
 D47: 0.16339699 sec  
 D48: 0.16339699 sec  
 D49: 0.16339699 sec  
 D50: 0.16339699 sec  
 D51: 0.16339699 sec  
 D52: 0.16339699 sec  
 D53: 0.16339699 sec  
 D54: 0.16339699 sec  
 D55: 0.16339699 sec  
 D56: 0.16339699 sec  
 D57: 0.16339699 sec  
 D58: 0.16339699 sec  
 D59: 0.16339699 sec  
 D60: 0.16339699 sec  
 D61: 0.16339699 sec  
 D62: 0.16339699 sec  
 D63: 0.16339699 sec  
 D64: 0.16339699 sec  
 D65: 0.16339699 sec  
 D66: 0.16339699 sec  
 D67: 0.16339699 sec  
 D68: 0.16339699 sec  
 D69: 0.16339699 sec  
 D70: 0.16339699 sec  
 D71: 0.16339699 sec  
 D72: 0.16339699 sec  
 D73: 0.16339699 sec  
 D74: 0.16339699 sec  
 D75: 0.16339699 sec  
 D76: 0.16339699 sec  
 D77: 0.16339699 sec  
 D78: 0.16339699 sec  
 D79: 0.16339699 sec  
 D80: 0.16339699 sec  
 D81: 0.16339699 sec  
 D82: 0.16339699 sec  
 D83: 0.16339699 sec  
 D84: 0.16339699 sec  
 D85: 0.16339699 sec  
 D86: 0.16339699 sec  
 D87: 0.16339699 sec  
 D88: 0.16339699 sec  
 D89: 0.16339699 sec  
 D90: 0.16339699 sec  
 D91: 0.16339699 sec  
 D92: 0.16339699 sec  
 D93: 0.16339699 sec  
 D94: 0.16339699 sec  
 D95: 0.16339699 sec  
 D96: 0.16339699 sec  
 D97: 0.16339699 sec  
 D98: 0.16339699 sec  
 D99: 0.16339699 sec  
 D100: 0.16339699 sec



Current Data Parameters  
 USER: pcr3.009\_120108  
 INSTR: spect  
 PROCNO: 1  
 P2 - Acquisition Parameters  
 Date\_ Time: 2013.11.30 11:37  
 TIME: 11.37  
 PROBM: 13C  
 PULPROG: zgpg30  
 TD: 65536  
 SOLVENT: CDCl3  
 DS: 4  
 SWH: 8112.572 Hz  
 FIDRES: 0.1122266 Hz  
 AQ: 4.0889465 sec  
 INJ: 62.400 uSec  
 DE: 1.0000000 K  
 TE: 1.1612982 K  
 DR: 0.5000000 sec  
 DL: 0.0000000 sec  
 DQ: 0.0000000 sec  
 CD2: 0.1633969 sec  
 PZ: 15.00 uSec  
 ===== CHANNEL f1 =====  
 NUC1: 13C  
 P1: 7.50 uSec  
 P4: 30.00 uSec  
 P2: 40000.00 uSec  
 SFO1: 500.2213799 MHz  
 SFO2:  
 SFO3:  
 SFO4:  
 SFO5:  
 SFO6:  
 SFO7:  
 ===== CHANNEL f2 =====  
 NUC2: 1H  
 P1: 7.50 uSec  
 P4: 30.00 uSec  
 P2: 40000.00 uSec  
 SFO1: 500.2213799 MHz  
 SFO2:  
 SFO3:  
 SFO4:  
 SFO5:  
 SFO6:  
 SFO7:  
 ===== GRADIENT CHANNEL =====  
 GRAM3: g100  
 GRAM4: s100  
 GRAM5: s100  
 GRAM6: s100  
 GRAM7: s100  
 GRAM8: s100  
 GRAM9: s100  
 GRAM10: s100  
 GRAM11: s100  
 GRAM12: s100  
 GRAM13: s100  
 GRAM14: s100  
 GRAM15: s100  
 GRAM16: s100  
 GRAM17: s100  
 GRAM18: s100  
 GRAM19: s100  
 GRAM20: s100  
 GRAM21: s100  
 GRAM22: s100  
 GRAM23: s100  
 GRAM24: s100  
 GRAM25: s100  
 GRAM26: s100  
 GRAM27: s100  
 GRAM28: s100  
 GRAM29: s100  
 GRAM30: s100  
 GRAM31: s100  
 GRAM32: s100  
 GRAM33: s100  
 GRAM34: s100  
 GRAM35: s100  
 GRAM36: s100  
 GRAM37: s100  
 GRAM38: s100  
 GRAM39: s100  
 GRAM40: s100  
 GRAM41: s100  
 GRAM42: s100  
 GRAM43: s100  
 GRAM44: s100  
 GRAM45: s100  
 GRAM46: s100  
 GRAM47: s100  
 GRAM48: s100  
 GRAM49: s100  
 GRAM50: s100  
 GRAM51: s100  
 GRAM52: s100  
 GRAM53: s100  
 GRAM54: s100  
 GRAM55: s100  
 GRAM56: s100  
 GRAM57: s100  
 GRAM58: s100  
 GRAM59: s100  
 GRAM60: s100  
 GRAM61: s100  
 GRAM62: s100  
 GRAM63: s100  
 GRAM64: s100  
 GRAM65: s100  
 GRAM66: s100  
 GRAM67: s100  
 GRAM68: s100  
 GRAM69: s100  
 GRAM70: s100  
 GRAM71: s100  
 GRAM72: s100  
 GRAM73: s100  
 GRAM74: s100  
 GRAM75: s100  
 GRAM76: s100  
 GRAM77: s100  
 GRAM78: s100  
 GRAM79: s100  
 GRAM80: s100  
 GRAM81: s100  
 GRAM82: s100  
 GRAM83: s100  
 GRAM84: s100  
 GRAM85: s100  
 GRAM86: s100  
 GRAM87: s100  
 GRAM88: s100  
 GRAM89: s100  
 GRAM90: s100  
 GRAM91: s100  
 GRAM92: s100  
 GRAM93: s100  
 GRAM94: s100  
 GRAM95: s100  
 GRAM96: s100  
 GRAM97: s100  
 GRAM98: s100  
 GRAM99: s100  
 GRAM100: s100  
 ===== Processing parameters =====  
 SI: 32768  
 SF: 500.2213799 MHz  
 DS: 4  
 SWH: 8112.572 MHz  
 LB: 1.00 Hz  
 GB: 0  
 PC: 1.00

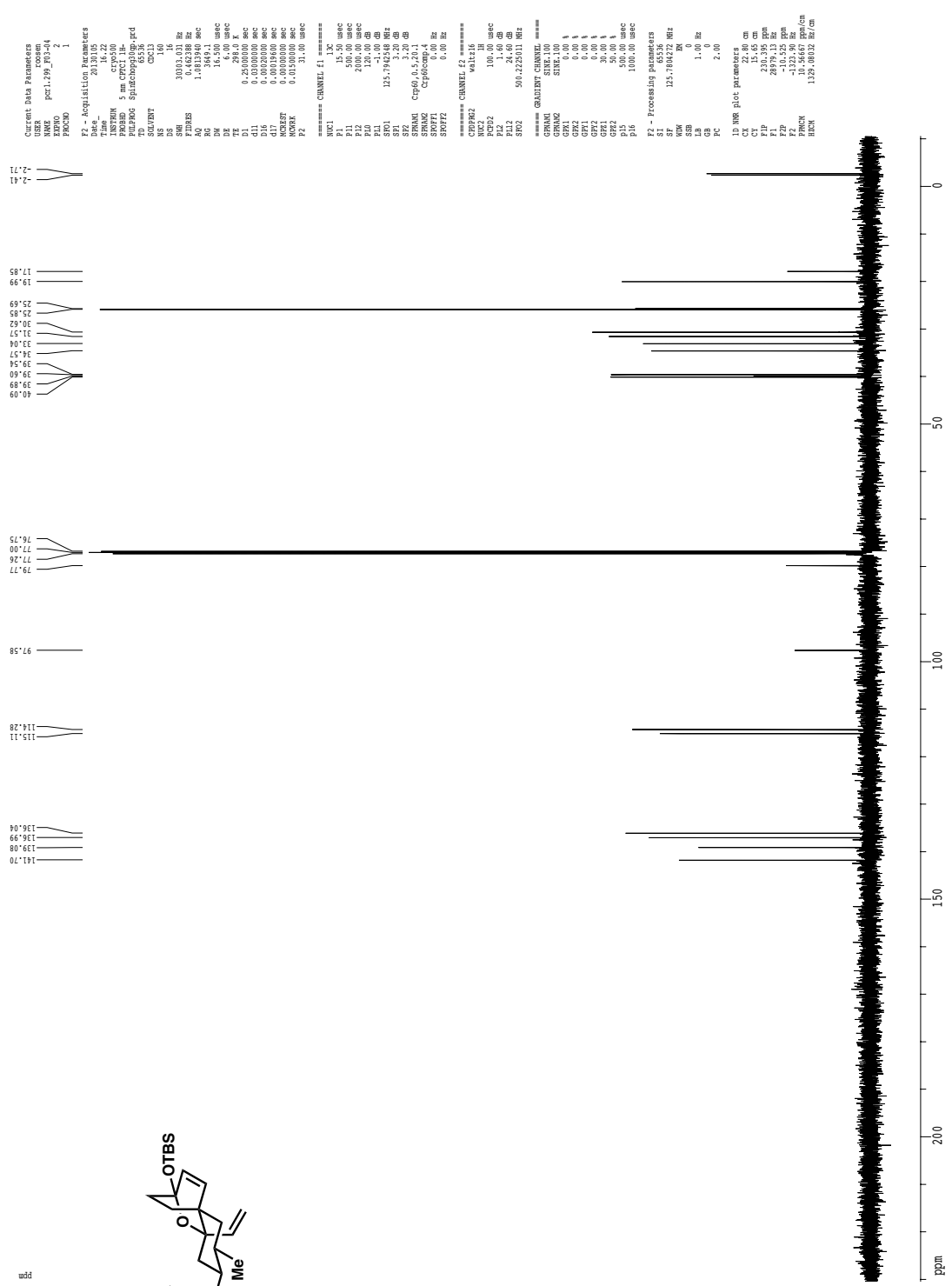


1H spectrum

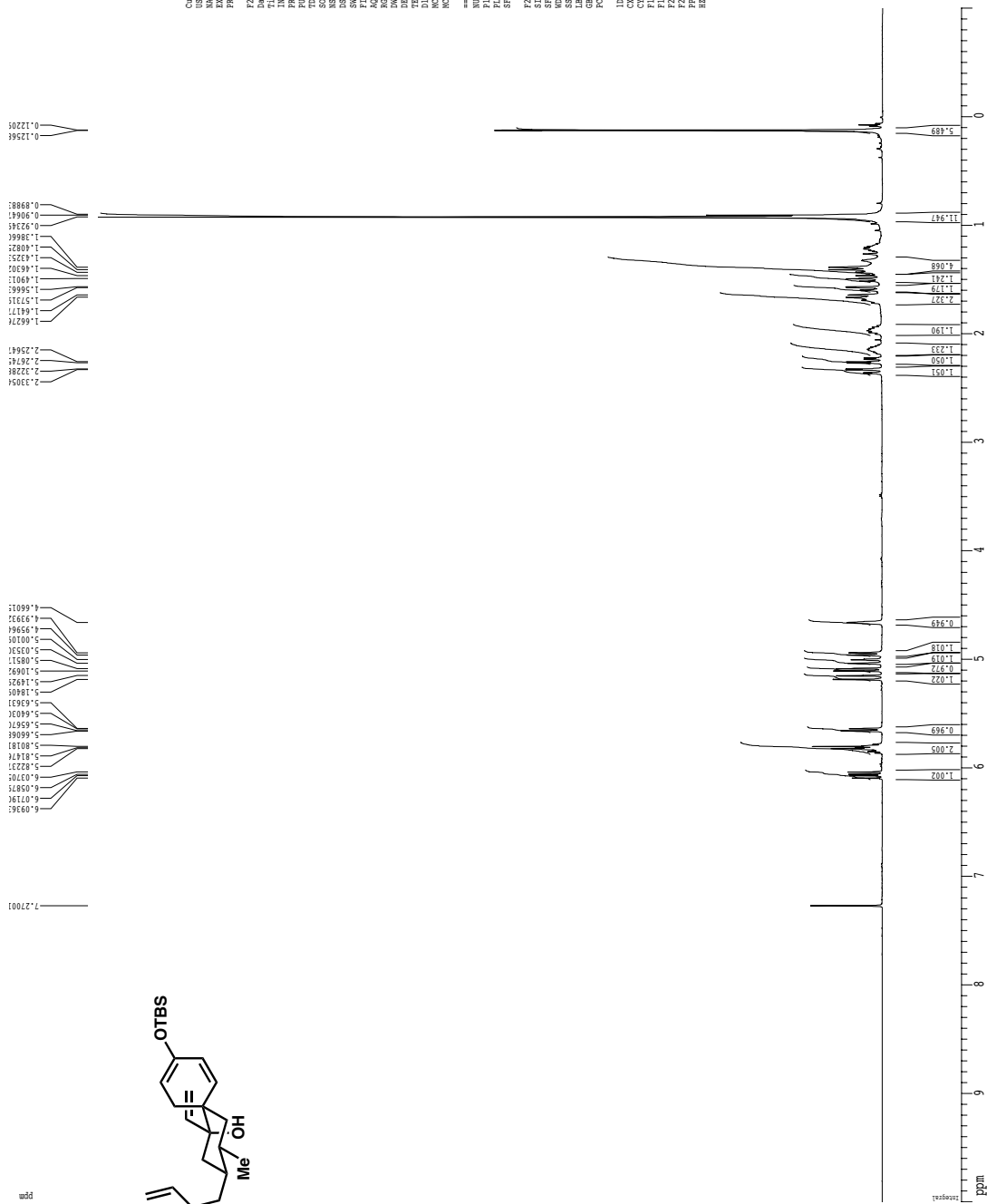


Current Data Parameters  
 NAME Pcl1\_298\_F03-04  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2010105  
 INSTRUM cty6500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 8192  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 802.820 Hz  
 FIDRES 0.100003 Hz  
 AQ 4.1997772 sec  
 RG 4  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 DI 0.1000000 sec  
 ACQSF 0.0000000 sec  
 SCANS 6  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.2200000 MHz  
 WWP 8K  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 F1 500.2200000 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 HWCN 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling

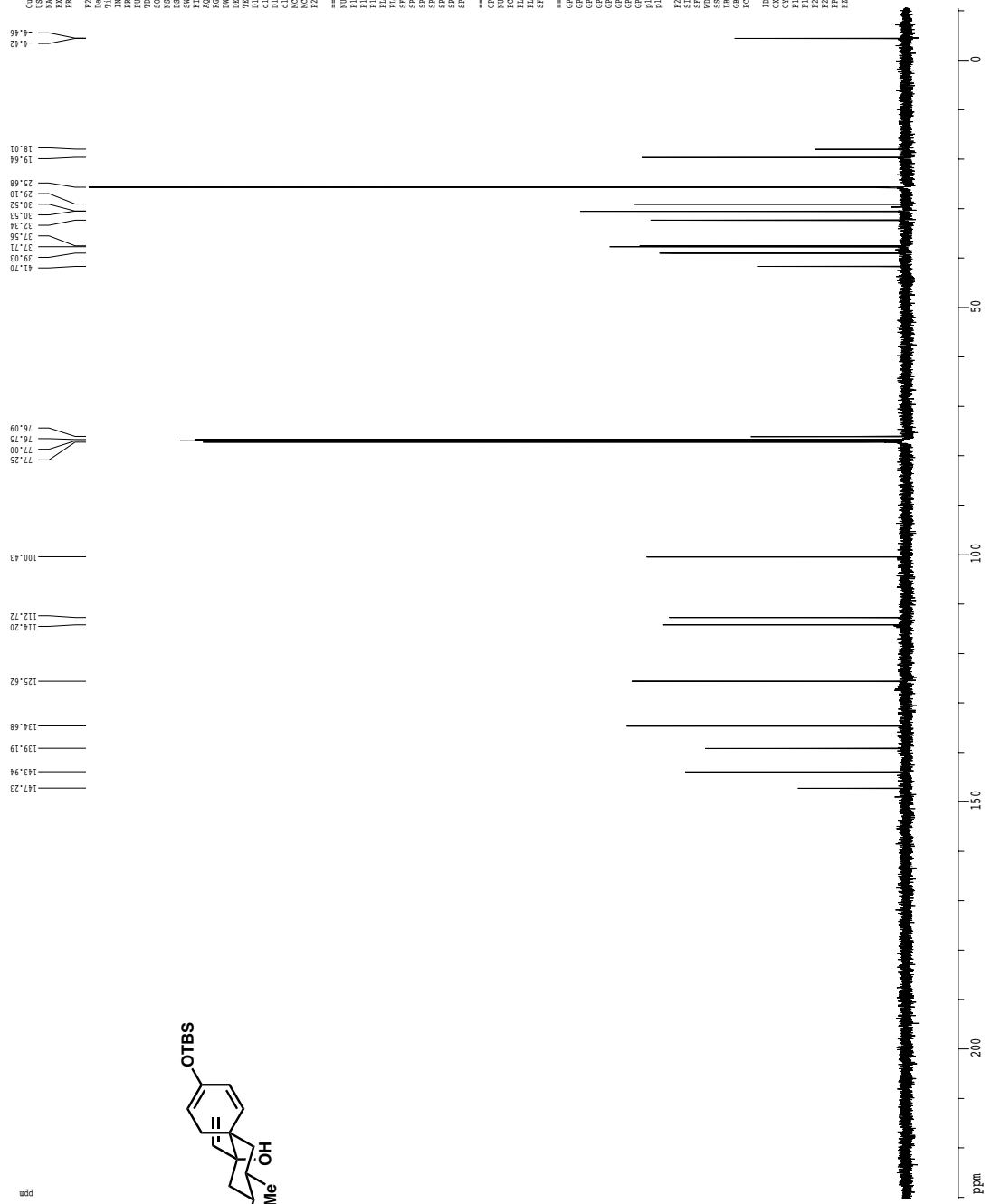


1H spectrum

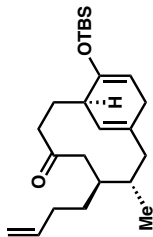
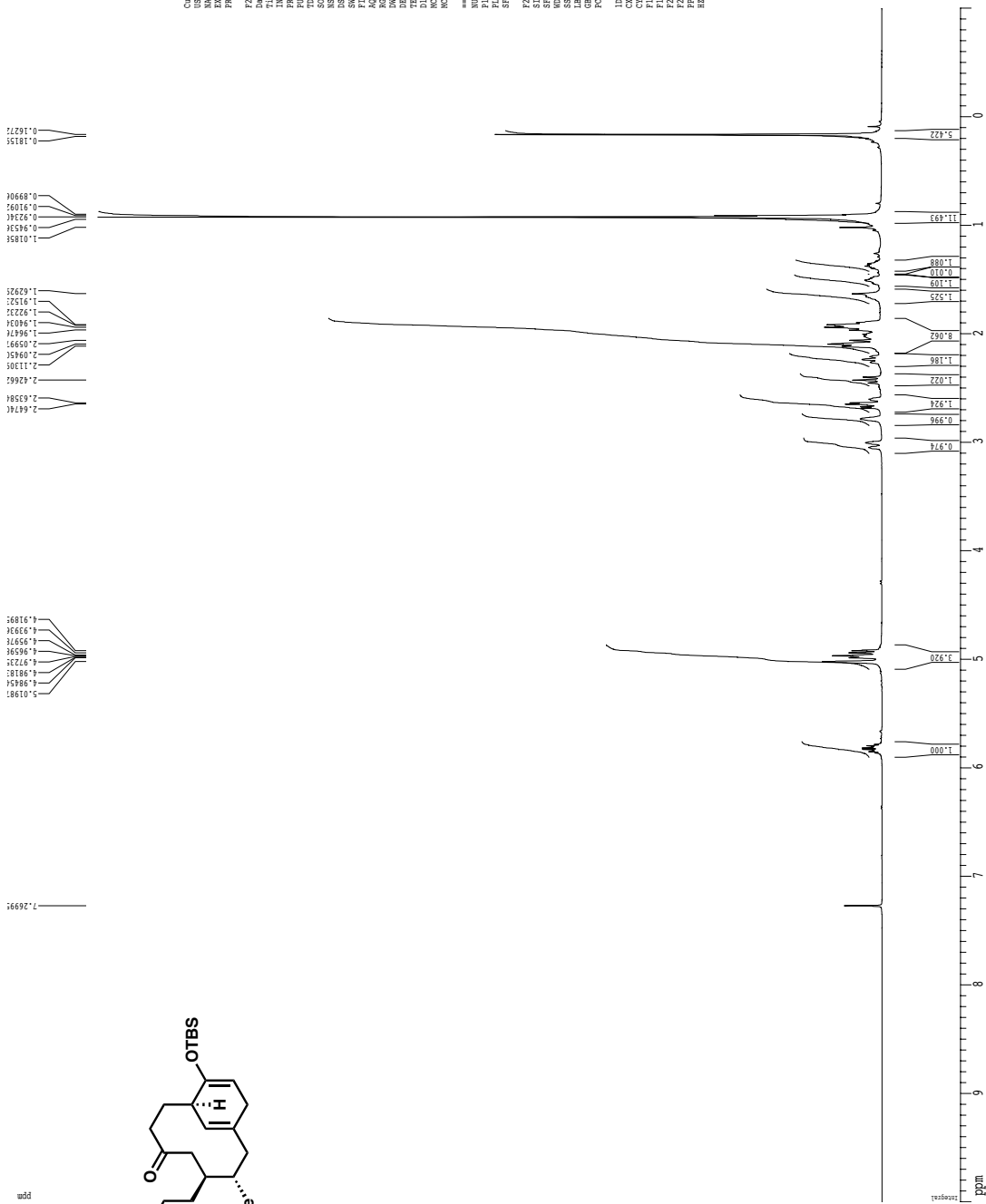


Current Data Parameters  
NAME Pcl1\_292\_F04-07  
EXPNO 1  
PROCNO 1  
F2 - Acquisition Parameters  
Date\_ 20121221  
Time 13.11.11  
INSTRUM cty6500  
PROBHD 5 mm CPXI 1H-  
PULPROG zgpg30  
TD 65536  
SOLVENT CDCl3  
NS 2  
DS 2  
SWH 8012.820 Hz  
FIDRES 0.100003 Hz  
AQ 4.199999999 sec  
RG 312  
DM 62.400 usec  
DE 6.00 usec  
TE 298.4 K  
D1 0.11000000 sec  
DCREST 0.00000000 sec  
SFO1 501.220000 MHz  
===== CHANNEL f1 =====  
NUC1 13C  
P1 7.50 usec  
PL1 1.60 dB  
SFO1 500.220000 MHz  
F2 - Processing parameters  
SI 65536  
SF 500.220000 MHz  
WDW EM  
SSB 0  
GB 0  
PC 4.00  
LO MRB pulse parameters  
CX 22.80 cm  
CY 15.00 cm  
CZ 15.00 cm  
F0 500.220000 MHz  
F1 500.220000 MHz  
F2 -1.000 ppm  
F3 -500.22 Hz  
GAMMA 13C  
NUC2 13C  
HPCON 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



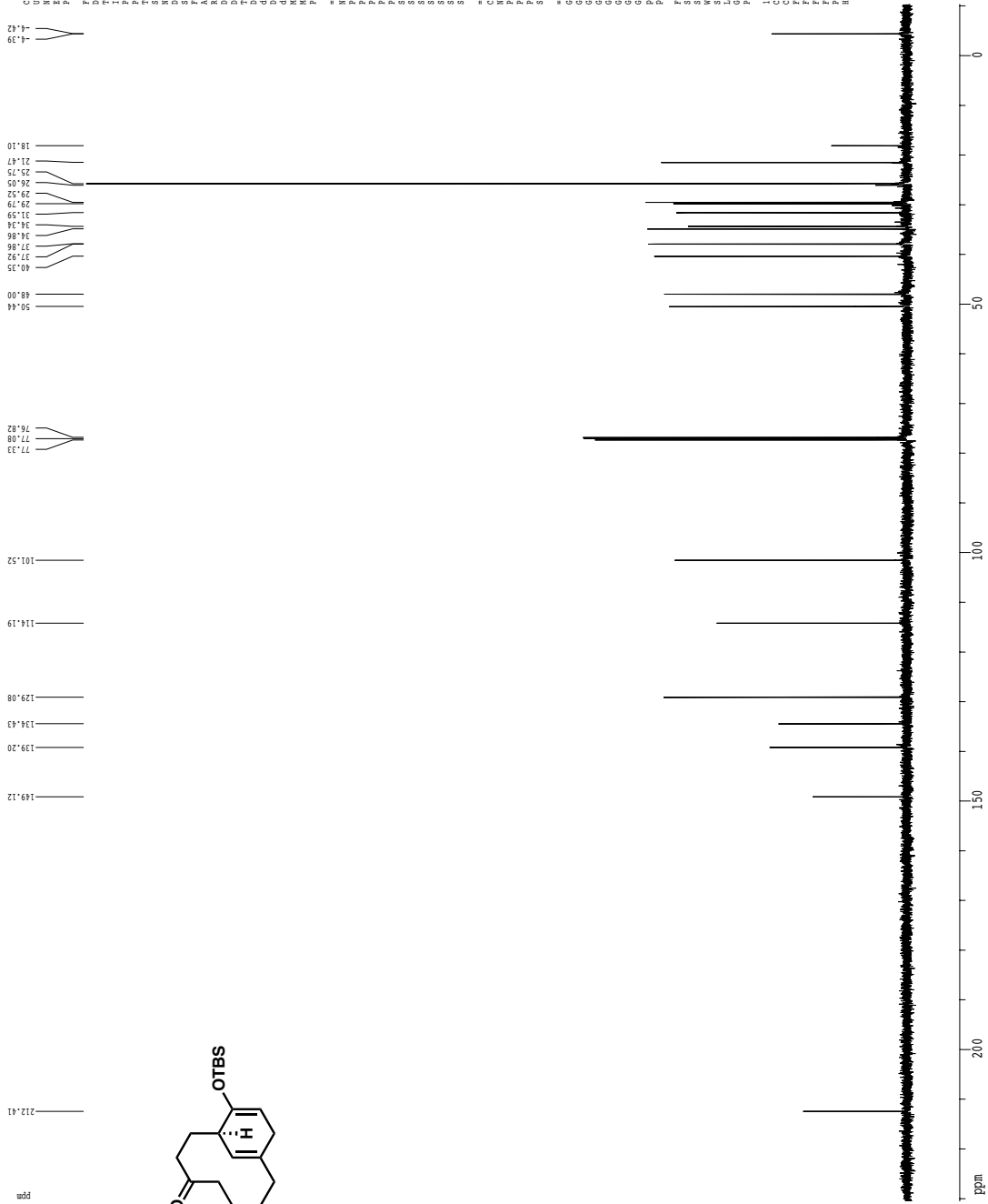
1H spectrum



Current Data Parameters  
 NAME Pct2.048 F05-08  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20130306  
 Time 08:12:28  
 INSTRUM cty6500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 65536  
 SOLVENT CDCl3  
 NS 2  
 DS 4  
 SWH 802.820 Hz  
 FIDRES 0.100002 Hz  
 AQ 4.199662 sec  
 RG 4.5  
 DQ 62.400 usec  
 WDE 0.390 usec  
 FWHM 294.4 Hz  
 D1 0.1000000 sec  
 D11 0.0000000 sec  
 ACQRES 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 F41 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 65536  
 SF 500.225015 MHz  
 WDW RM  
 SSB 0  
 GB 0  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CZ 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 WCON 241.33423 Hz/cm



Z-restored spin-echo 13C spectrum with 1H decoupling

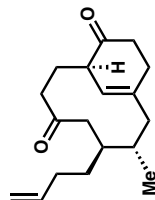


```

Current Data Parameters
NAME           pcz21a8_19a-98
EXPNO         2
PROCNO        1
F2 - Acq. Parameters
Date_         20161029
Time         22:53
INSTRUM      cryo500
PROBHD       zgpg30
PULPROG      zgpg30
TD           65536
SOLVENT      CDCl3
NS           132
DS           2
SWH           30000.000 Hz
FIDRES       1.0813940 Hz
AQ           7986.2
RG           650
SI           6500
SF           125.7603622 MHz
CT           0.25
PC          298.0 K
DT           0.433000000 sec
DE           0.000200000 sec
TE           3.20 dB
SFOFF        0.000000000 Hz
SFOFF2       3.20 dB
SFOFF3       0.000000000 Hz
===== CHANNEL f1 =====
NUC1          13C
P1           15.50
NUC2          13C
P2           15.50
SFO1         125.7603622 MHz
SFO2         -1.00 dB
SFO3         125.7603622 MHz
SFOFF        0.00 Hz
SFOFF2       3.20 dB
SFOFF3       0.00 Hz
===== CHANNEL f2 =====
CPDPRG2      waltz16
NUC1          13C
P1           100.00
NUC2          13C
P2           100.00
SFO1         125.7603622 MHz
SFO2         50.0225911 MHz
===== CHANNEL CHANNEL =====
GRANU        STXK.100
SINX         STXK.100
===== Processing parameters =====
SI           65536
SF           125.7603622 MHz
WDW          EM
SSB           0
LB           3.00 Hz
GB           0
PC           2.00
===== ID MSQ plot parameters =====
SI           65536
SF           125.7603622 MHz
WDW          EM
SSB           0
LB           3.00 Hz
GB           0
PC           2.00

```

**<sup>1</sup>H spectrum**



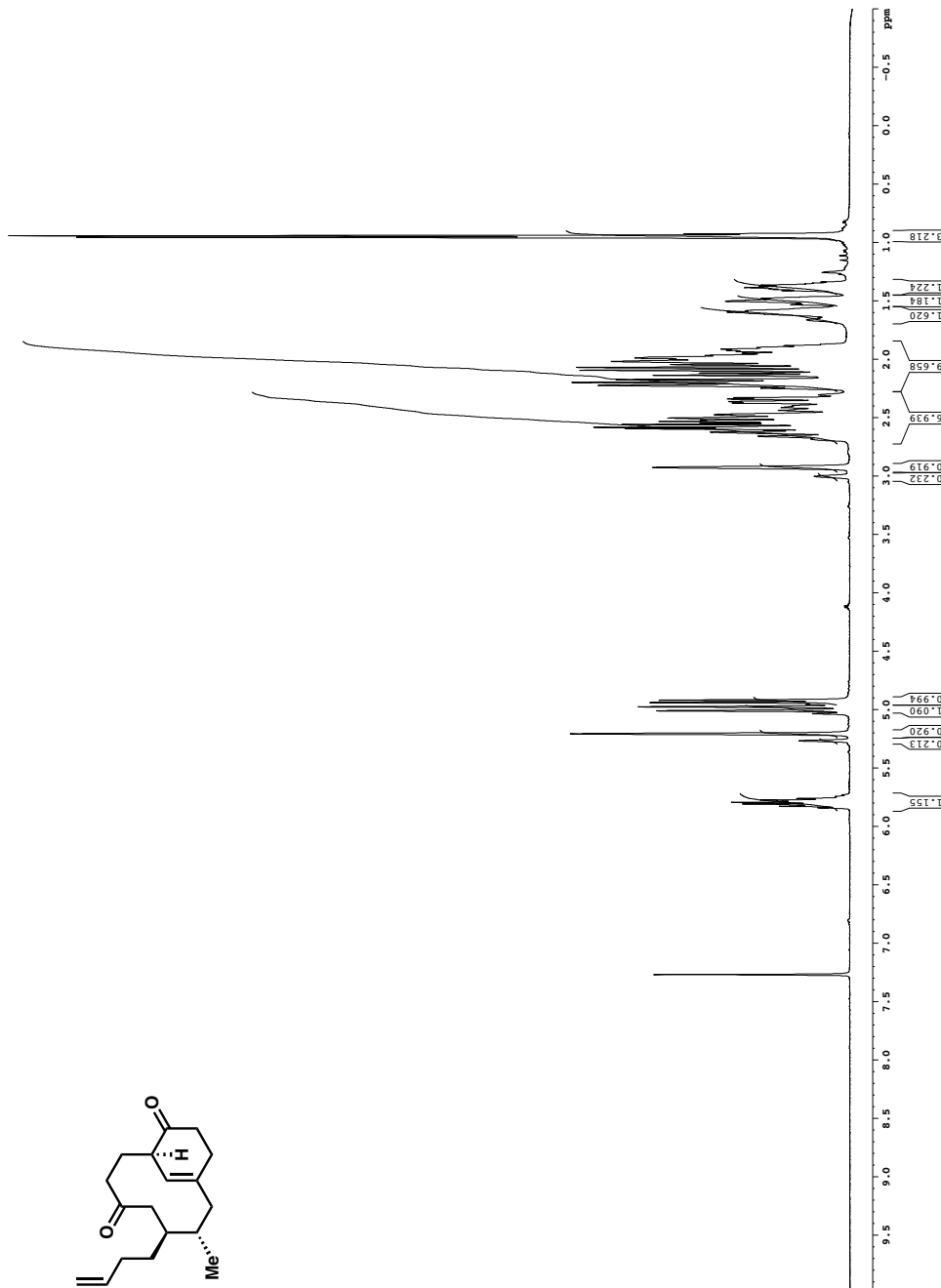
Current Data Parameters  
 USER toonen  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2011.05.11  
 Time 11:45  
 INSTRUM spect  
 PROBM 5 mm CPCLP-1H  
 PULPROG zgpg30  
 SOLVENT CDCl3  
 DS 2  
 BS 1012.62 Hz  
 FIDRES 0.226026 Hz  
 AQ 1.9998451 sec  
 RM 62.400 usec  
 DM 0.10000000 sec  
 TE 298.2 K  
 ACQ 0.10000000 sec  
 SCANS 1  
 SFO1 500.136099 MHz  
 CHANNEL f1  
 P1 7.50 usec  
 PL12 0.00 dB  
 SFO2 500.136099 MHz  
 F2 - Processing parameters  
 SI 65536  
 SF 500.136099 MHz  
 WDW EM  
 SSB 0  
 CB 0  
 GB 0  
 PC 4.00

3.004  
2.927  
2.660  
2.623  
2.611  
2.583  
2.559  
2.546  
2.534  
2.504  
2.476  
2.439  
2.363  
2.340  
2.330  
2.224  
2.199  
2.175  
2.139  
2.119  
2.095  
2.070  
2.045  
2.020  
1.990  
1.915  
1.915  
1.597  
1.555  
1.387  
1.258  
0.957  
0.926

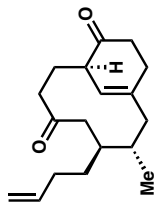
4.921  
4.941  
4.978  
5.012  
5.036  
5.208  
5.268

5.829  
5.809  
5.795  
5.775

7.270

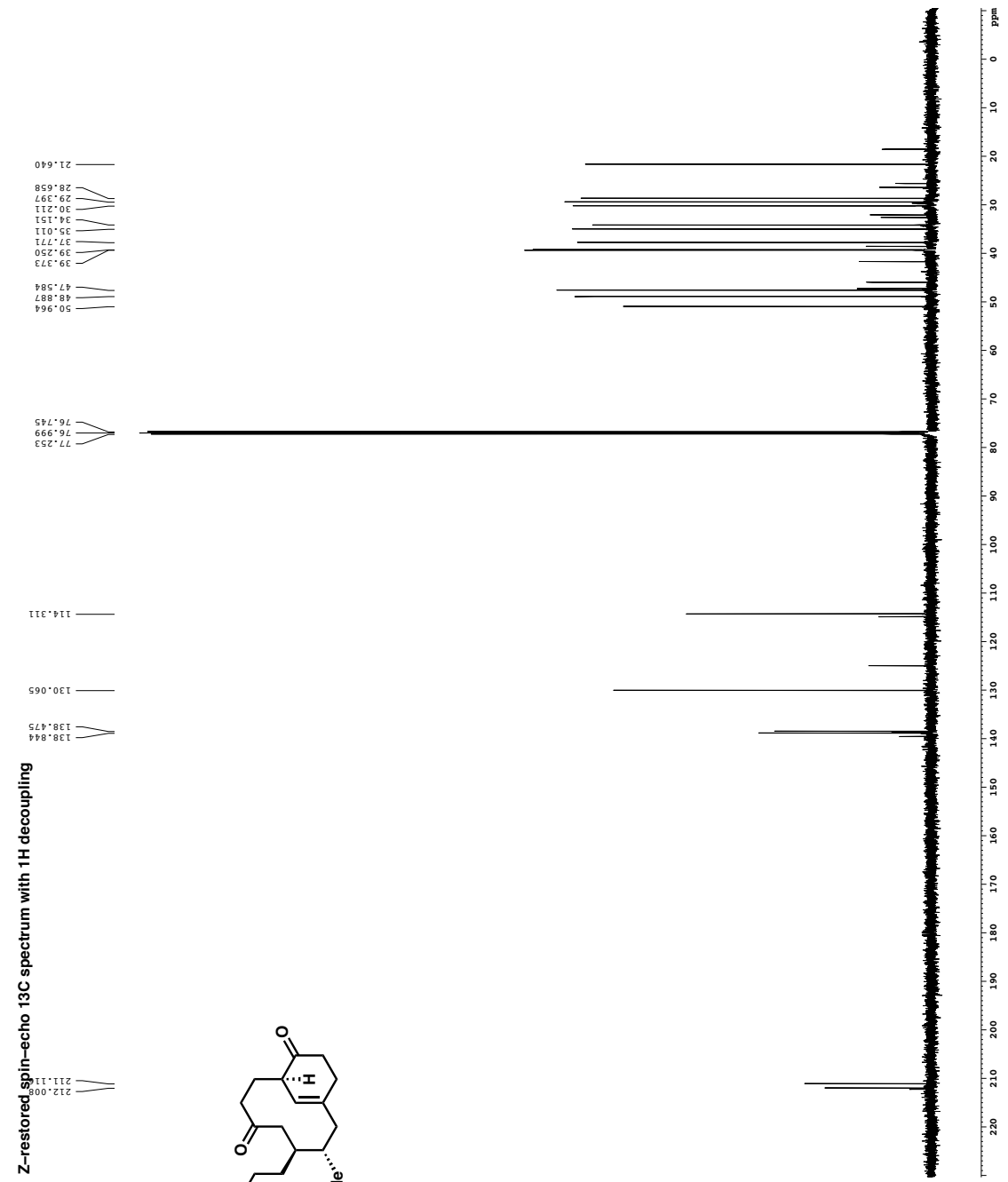


Z-restored spin-echo 13C spectrum with 1H decoupling

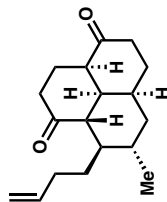


Current Data Parameters  
USER: pcr3\_06d\_1000en  
EXPNO: 2  
PROCNO: 1  
F2 - Acquisition Parameters  
Date\_ 2010.04  
Time 11.04  
PROBHD 5 mm CPYCC 1H  
NUC1 13C  
PULPROG zgpg30  
SOLVENT CDCl3  
DS 2  
AQ 3.61000000 sec  
FIDRES 0.442388 Hz  
RG 1.00000000 sec  
AQ 16.50000000 sec  
TE 300.2 K  
TD 65536  
SFO 125.7614500 MHz  
NUC2 13C  
P1 15.00000000 sec  
P2 0.0100000000 sec  
PC 2.0000000000 sec  
PR 0.0000000000 sec  
PT 0.0000000000 sec  
PS 0.0000000000 sec  
SR 0.0000000000 sec  
WDW 1.0000000000 sec  
SSB 0.0000000000 sec  
GB 0.0000000000 sec  
PC 2.0000000000 sec  
F2 - Processing parameters  
SI 65536  
SF 125.7614500 MHz  
WDW EM  
GB 0  
PC 2.00

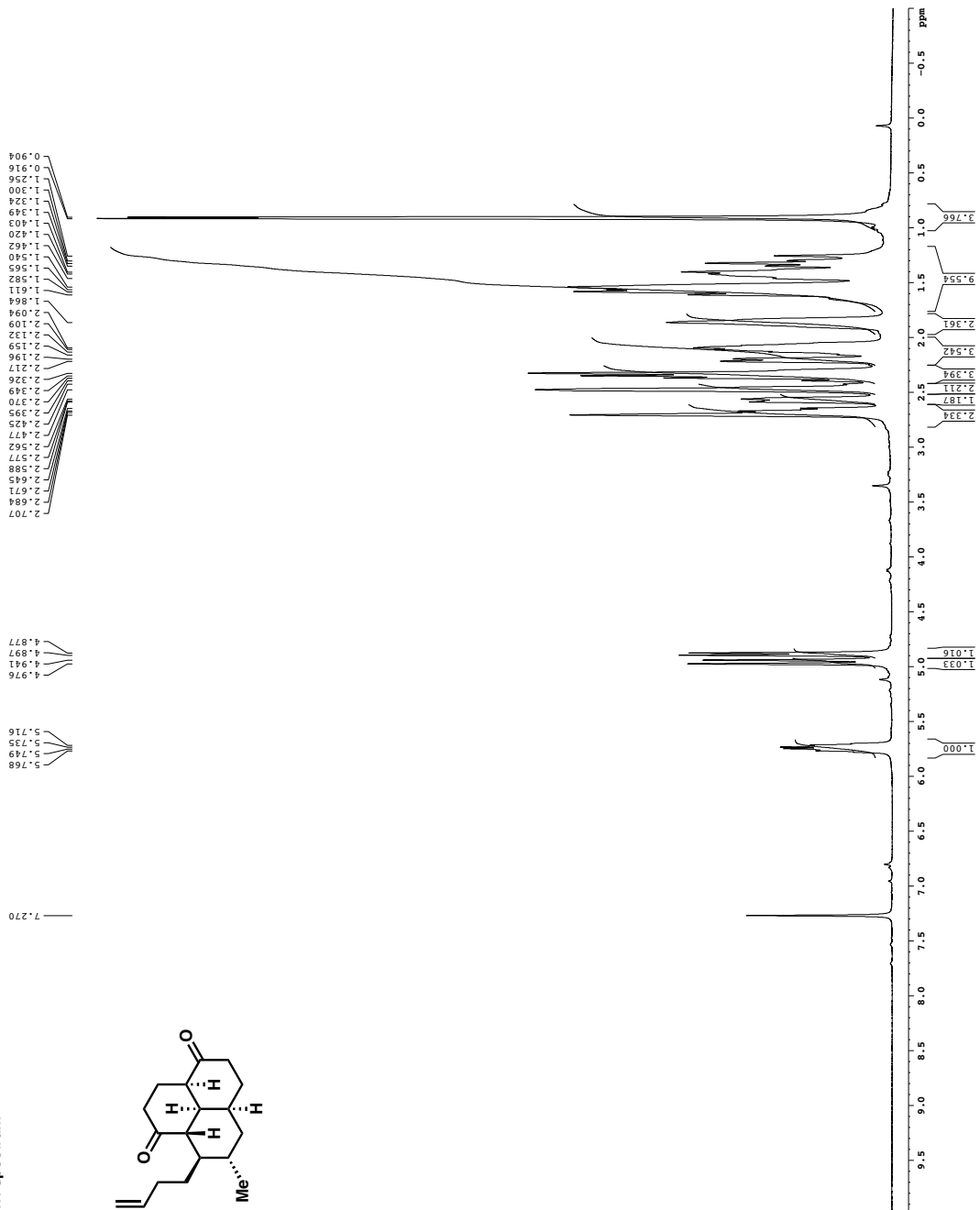
==== CHANNEL F1 =====  
NUC1 13C  
P1 15.00000000 sec  
P2 0.0100000000 sec  
PC 2.0000000000 sec  
PR 0.0000000000 sec  
PT 0.0000000000 sec  
PS 0.0000000000 sec  
SR 0.0000000000 sec  
WDW 1.0000000000 sec  
SSB 0.0000000000 sec  
GB 0.0000000000 sec  
PC 2.0000000000 sec  
===== CHANNEL F2 =====  
NUC2 13C  
P1 15.00000000 sec  
P2 0.0100000000 sec  
PC 2.0000000000 sec  
PR 0.0000000000 sec  
PT 0.0000000000 sec  
PS 0.0000000000 sec  
SR 0.0000000000 sec  
WDW 1.0000000000 sec  
SSB 0.0000000000 sec  
GB 0.0000000000 sec  
PC 2.0000000000 sec  
===== CHANNEL F3 =====  
NUC3 13C  
P3 15.00000000 sec  
P4 0.0100000000 sec  
PC 2.0000000000 sec  
PR 0.0000000000 sec  
PT 0.0000000000 sec  
PS 0.0000000000 sec  
SR 0.0000000000 sec  
WDW 1.0000000000 sec  
SSB 0.0000000000 sec  
GB 0.0000000000 sec  
PC 2.0000000000 sec



**<sup>1</sup>H spectrum**



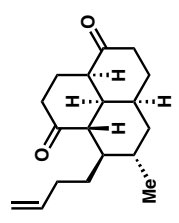
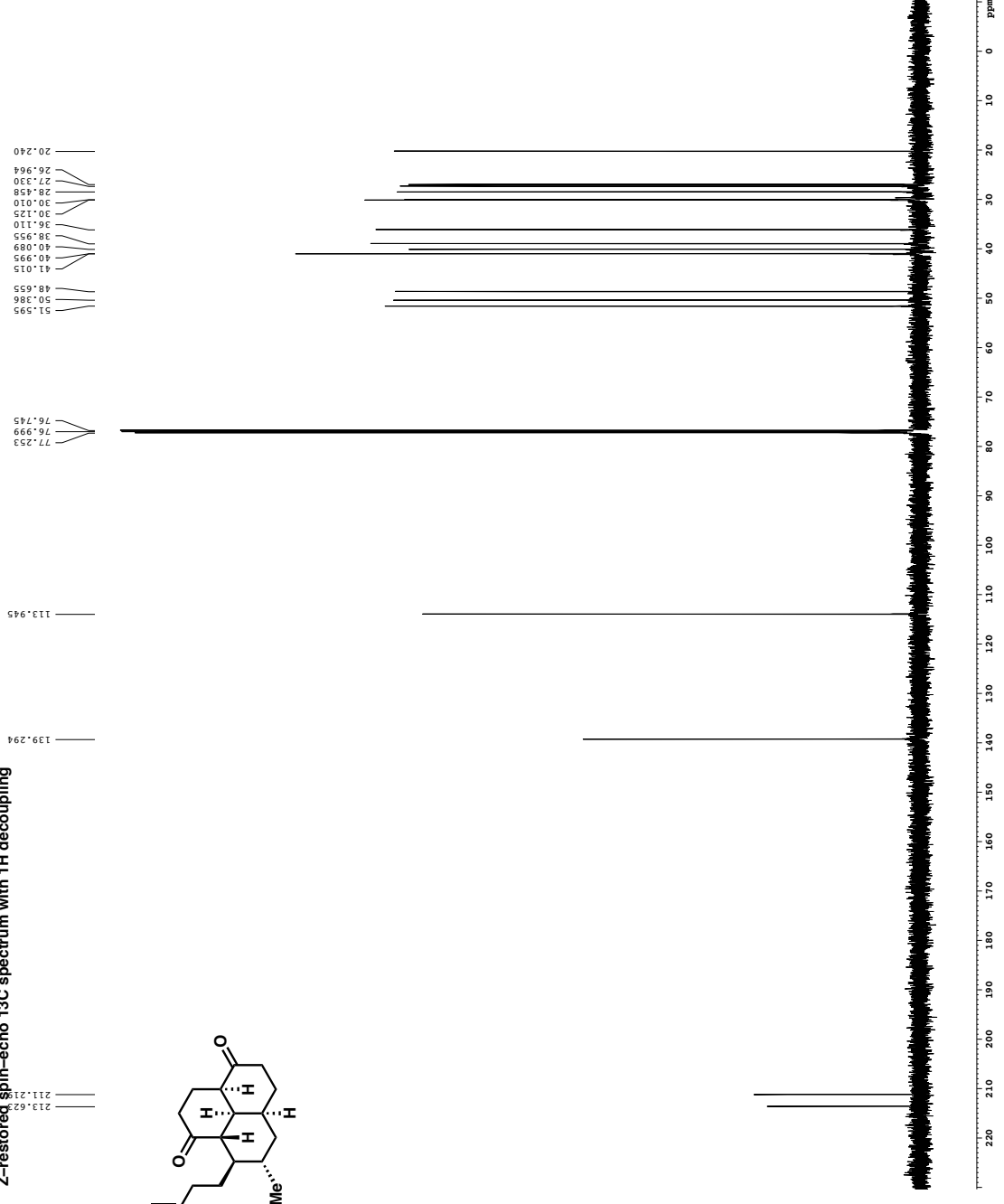
Current Data Parameters  
 USER: jtoonen  
 EXPNO: 1  
 PROCNO: 1  
 F2 - Acquisition Parameters  
 Date\_ : 2015-10-11  
 Time: 15:30  
 PROBHD: 5 mm CPYCH-1H  
 PULPROG: zgpg30  
 TOU: 2.00  
 SOLVENT: CDCl3  
 DS: 2  
 BS: 0.2  
 FIDRES: 0.12 Hz  
 AQ: 0.220026 sec  
 RG: 1.9998451  
 INJ: 62.400 usec  
 DE: 1.0000000  
 TE: 300.2 K  
 ACQ: 0.10000000 sec  
 SCANS: 64  
 MEASST: 0.01500000 sec  
 MEASST2: 0.01500000 sec  
 ===== CHANNEL f1 =====  
 NU1: 1  
 PR1: 7.50 usec  
 PL1: 0.00 dB  
 SF01: 500.2213015 MHz  
 F2 - Processing parameters  
 SI: 65536  
 SF: 500.2207289 MHz  
 WDW: EM  
 GB: 0  
 CB: 0  
 PC: 4.00



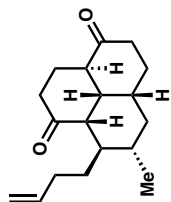
Z-restored spin-echo 13C spectrum with 1H decoupling

```

Current Data Parameters
=====
USER          PCT3.0v_1173
EXPNO         2
PROCNO        1
F2 - Acquisition Parameters
=====
Date_         20100115
Time         15:32
INSTRUM      spect
PROBHD       5 mm CPCC1 1H-
PULPROG      zgpg30
SOLVENT      DMSO-d6
NAME         CXC13
DS           4
SI           2
FIDRES       0.482368 Hz
AQ           1.009380 sec
RG           655
DM           16.500 usec
DE           0.29820 K
TE           300.2
D1           0.03000000 sec
d11          0.03000000 sec
d15          0.00000000 sec
DELTA        0.00000000 sec
NUC1         13
NUC2         13
NUS1         0.01000000 Hz
NUS2         0.01100000 Hz
===== CHANNEL F1 =====
NUC1         13C
P1           15.30 usec
PL1          0.00 dB
PC1          500.00 usec
PR1          19.00 usec
RG1          655
WDW1         EM
SSB1         0
GB1          0.00 usec
PC1          120.00 dB
===== CHANNEL F2 =====
NAME         GRADIENT CHANNEL
NUC1         13C
P1           15.30 usec
PL1          0.00 dB
PC1          500.00 usec
PR1          19.00 usec
RG1          655
WDW1         EM
SSB1         0
GB1          0.00 usec
PC1          120.00 dB
===== GRADIENT CHANNEL =====
SI           2
SF           125.761258 MHz
SFO1         125.761258 MHz
SFO2         125.761258 MHz
SFO3         125.761258 MHz
SFO4         125.761258 MHz
SFO5         125.761258 MHz
SFO6         125.761258 MHz
SFO7         125.761258 MHz
SFO8         125.761258 MHz
SFO9         125.761258 MHz
SFO10        125.761258 MHz
SFO11        125.761258 MHz
SFO12        125.761258 MHz
===== PROCESSING PARAMETERS =====
SI           2
SF           125.761258 MHz
WDW1         EM
SSB1         0
GB1          0.00 usec
PC1          120.00 dB
=====
  
```

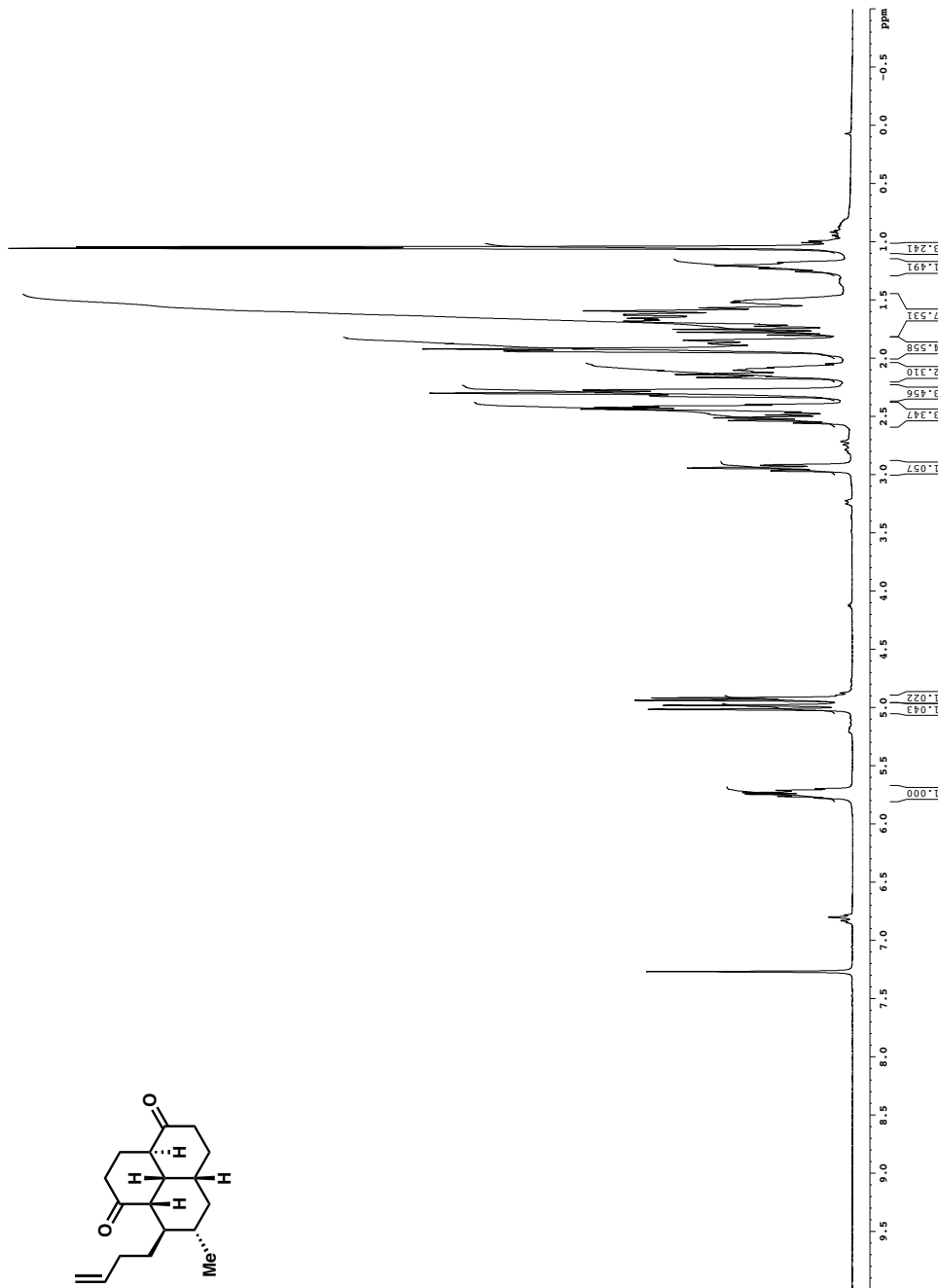


**<sup>1</sup>H spectrum**



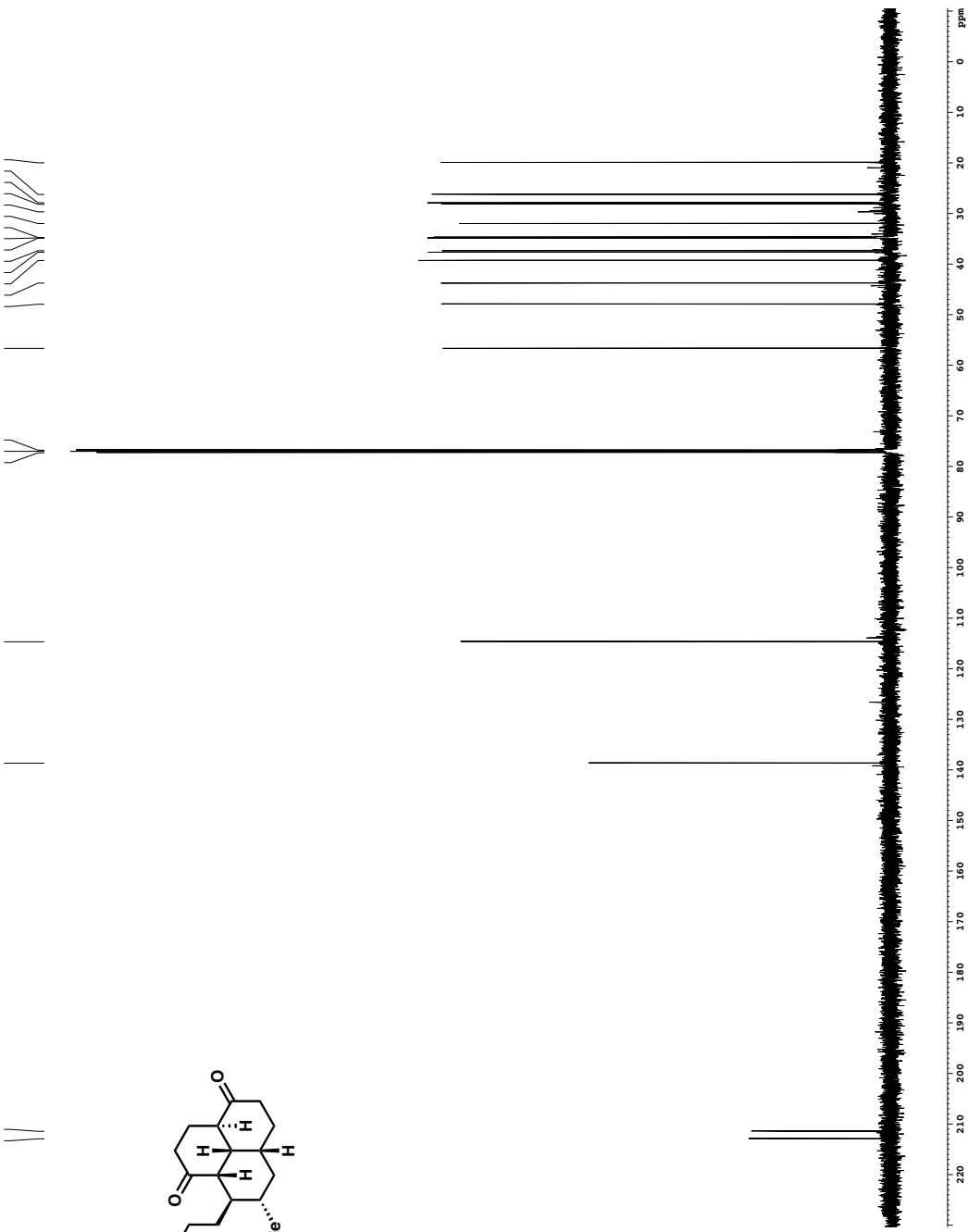
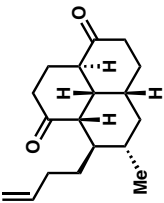
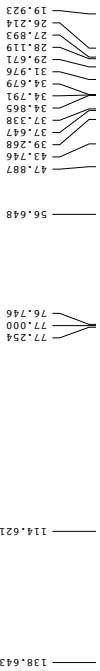
Current Data Parameters  
 USER toosen  
 EXPNO 3  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2010.06  
 Time 10.06  
 INSTRUM spect  
 PROBMG 9 mm CPDQZ 1H  
 PULPROG zgpg30  
 SOLVENT CDCl3  
 DS 2  
 B1 2  
 B2 2  
 F1 617.62 Hz  
 F2 617.62 Hz  
 FIDRES 0.2240226 Hz  
 AQ 1.9998451 sec  
 RM 62.400 usec  
 DM 62.400 usec  
 DE 1.29820 K  
 TE 300.2 K  
 ACQRES 0.00000000 sec  
 SCANS 1024  
 SFO1 500.1360000 MHz  
 SFO2 500.1360000 MHz  
 SFO3 500.1360000 MHz  
 P1 7.50 usec  
 PL1 0.00 dB  
 PL2 0.00 dB  
 PL3 0.00 dB  
 PZ 0.00 mm  
 PR 0.00 mm  
 F2 - Processing parameters  
 SI 65536  
 SF 500.1360000 MHz  
 WDW EM  
 GB 0  
 CB 4.00

0.999  
1.008  
1.017  
1.057  
1.180  
1.208  
1.231  
1.256  
1.512  
1.522  
1.565  
1.595  
1.625  
1.631  
1.644  
1.658  
1.670  
1.684  
1.691  
1.727  
1.753  
1.778  
1.802  
1.847  
1.873  
1.922  
1.939  
1.945  
2.080  
2.093  
2.104  
2.141  
2.165  
2.275  
2.301  
2.322  
2.359  
2.414  
2.426  
2.439  
2.467  
2.489  
2.513  
2.537  
2.558  
2.921  
2.945  
2.969  
4.920  
4.940  
4.983  
5.017  
5.700  
5.713  
5.733  
5.747  
5.759  
5.766  
5.780  
7.270



Z-restored spin-echo 13C spectrum with 1H decoupling

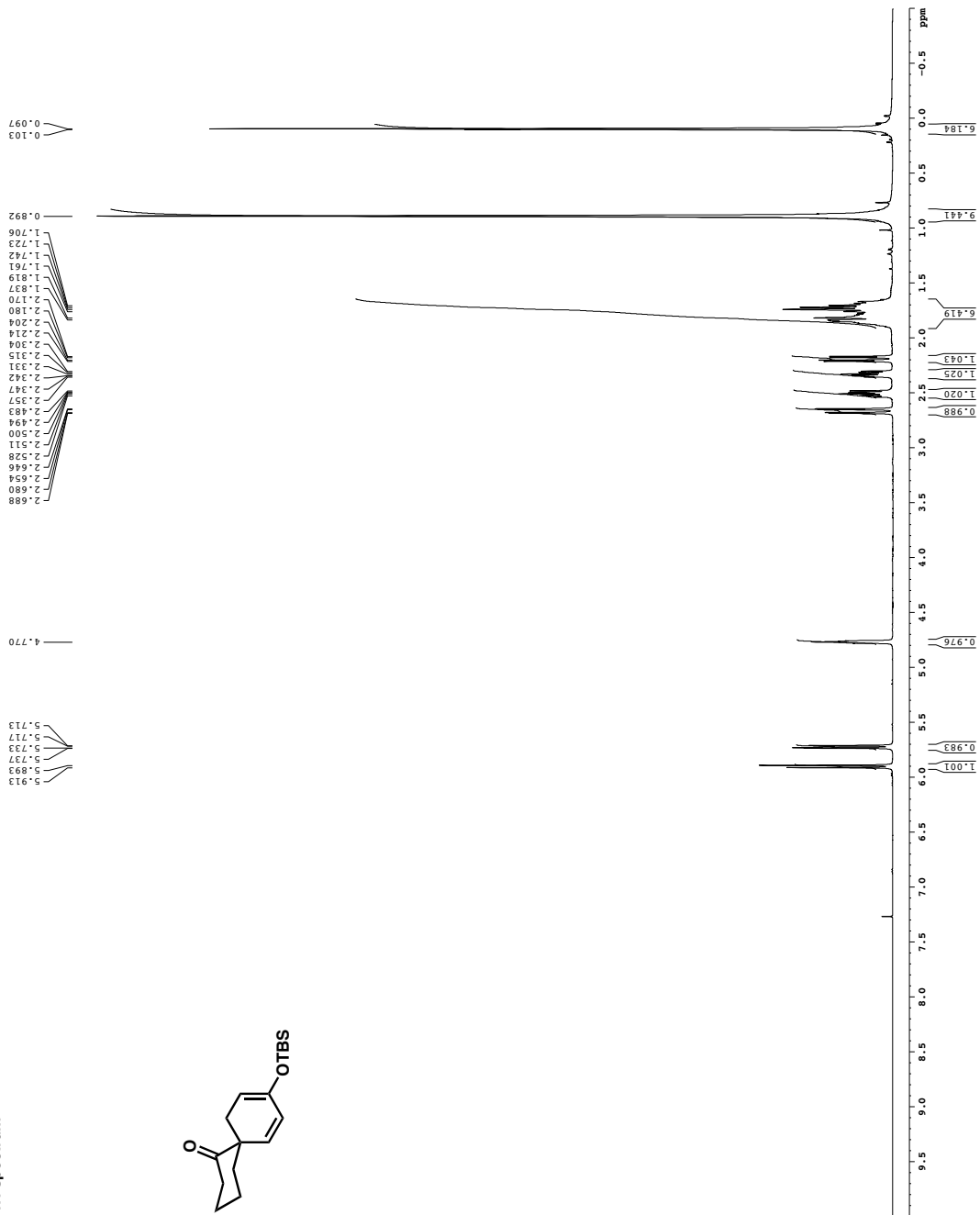
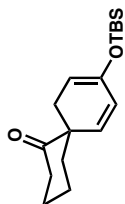
```
Current Data Parameters
=====
USER      pc13_06_1289
EXPNO     2
PROCNO    1
P2 - Acquisition Parameters
=====
Time         21.45
Date_        15.45
Time2         15.45
PROBHD      5 mm CPDCCI 1H-
PULPROG     zgpg30
SOLVENT     Spiroheptadiene
NS          655
DS          4
DSB         2
F2 - Acquisition Parameters
=====
FIDRES      3.017 Hz
AQ          0.482368 Hz
RG          1.068390 sec
DM          16.500 usec
DE          0.000000 sec
TE          300.2 K
TR          0.42860 sec
D11         0.0300000 sec
D12         0.0000000 sec
D13         0.0000000 sec
D14         0.0000000 sec
D15         0.0000000 sec
SFO1        125.762518 MHz
SFO2        500.136099 MHz
SFO3        125.762518 MHz
SFO4        3.20 MHz
SFO5        3.20 MHz
SFO6        3.20 MHz
SFO7        Cyp60.0.5.20.1 GHz
SFO8        Cyp60.0.5.20.1 GHz
SFO9        Cyp60.0.5.20.1 GHz
SFO10       Cyp60.0.5.20.1 GHz
SFO11       Cyp60.0.5.20.1 GHz
SFO12       Cyp60.0.5.20.1 GHz
SFO13       Cyp60.0.5.20.1 GHz
SFO14       Cyp60.0.5.20.1 GHz
SFO15       Cyp60.0.5.20.1 GHz
SFO16       Cyp60.0.5.20.1 GHz
SFO17       Cyp60.0.5.20.1 GHz
SFO18       Cyp60.0.5.20.1 GHz
SFO19       Cyp60.0.5.20.1 GHz
SFO20       Cyp60.0.5.20.1 GHz
SFO21       Cyp60.0.5.20.1 GHz
SFO22       Cyp60.0.5.20.1 GHz
SFO23       Cyp60.0.5.20.1 GHz
SFO24       Cyp60.0.5.20.1 GHz
SFO25       Cyp60.0.5.20.1 GHz
SFO26       Cyp60.0.5.20.1 GHz
SFO27       Cyp60.0.5.20.1 GHz
SFO28       Cyp60.0.5.20.1 GHz
SFO29       Cyp60.0.5.20.1 GHz
SFO30       Cyp60.0.5.20.1 GHz
=====
===== CHANNEL F1 =====
NUC1       13C
P1         15.00 usec
PL1        0.00 dB
PL2        19.00 dB
PL3        19.00 dB
PL4        19.00 dB
PL5        19.00 dB
PL6        19.00 dB
PL7        19.00 dB
PL8        19.00 dB
PL9        19.00 dB
PL10       19.00 dB
PL11       19.00 dB
PL12       19.00 dB
PL13       19.00 dB
PL14       19.00 dB
PL15       19.00 dB
PL16       19.00 dB
PL17       19.00 dB
PL18       19.00 dB
PL19       19.00 dB
PL20       19.00 dB
PL21       19.00 dB
PL22       19.00 dB
PL23       19.00 dB
PL24       19.00 dB
PL25       19.00 dB
PL26       19.00 dB
PL27       19.00 dB
PL28       19.00 dB
PL29       19.00 dB
PL30       19.00 dB
=====
===== CHANNEL F2 =====
NUC2       13C
P2         15.00 usec
PL2        0.00 dB
PL3        19.00 dB
PL4        19.00 dB
PL5        19.00 dB
PL6        19.00 dB
PL7        19.00 dB
PL8        19.00 dB
PL9        19.00 dB
PL10       19.00 dB
PL11       19.00 dB
PL12       19.00 dB
PL13       19.00 dB
PL14       19.00 dB
PL15       19.00 dB
PL16       19.00 dB
PL17       19.00 dB
PL18       19.00 dB
PL19       19.00 dB
PL20       19.00 dB
PL21       19.00 dB
PL22       19.00 dB
PL23       19.00 dB
PL24       19.00 dB
PL25       19.00 dB
PL26       19.00 dB
PL27       19.00 dB
PL28       19.00 dB
PL29       19.00 dB
PL30       19.00 dB
=====
===== CHANNEL F3 =====
=====
===== CHANNEL F4 =====
=====
===== CHANNEL F5 =====
=====
===== CHANNEL F6 =====
=====
===== CHANNEL F7 =====
=====
===== CHANNEL F8 =====
=====
===== CHANNEL F9 =====
=====
===== CHANNEL F10 =====
=====
===== CHANNEL F11 =====
=====
===== CHANNEL F12 =====
=====
===== CHANNEL F13 =====
=====
===== CHANNEL F14 =====
=====
===== CHANNEL F15 =====
=====
===== CHANNEL F16 =====
=====
===== CHANNEL F17 =====
=====
===== CHANNEL F18 =====
=====
===== CHANNEL F19 =====
=====
===== CHANNEL F20 =====
=====
===== CHANNEL F21 =====
=====
===== CHANNEL F22 =====
=====
===== CHANNEL F23 =====
=====
===== CHANNEL F24 =====
=====
===== CHANNEL F25 =====
=====
===== CHANNEL F26 =====
=====
===== CHANNEL F27 =====
=====
===== CHANNEL F28 =====
=====
===== CHANNEL F29 =====
=====
===== CHANNEL F30 =====
=====
===== CHANNEL F31 =====
=====
===== CHANNEL F32 =====
=====
===== CHANNEL F33 =====
=====
===== CHANNEL F34 =====
=====
===== CHANNEL F35 =====
=====
===== CHANNEL F36 =====
=====
===== CHANNEL F37 =====
=====
===== CHANNEL F38 =====
=====
===== CHANNEL F39 =====
=====
===== CHANNEL F40 =====
=====
===== CHANNEL F41 =====
=====
===== CHANNEL F42 =====
=====
===== CHANNEL F43 =====
=====
===== CHANNEL F44 =====
=====
===== CHANNEL F45 =====
=====
===== CHANNEL F46 =====
=====
===== CHANNEL F47 =====
=====
===== CHANNEL F48 =====
=====
===== CHANNEL F49 =====
=====
===== CHANNEL F50 =====
=====
===== CHANNEL F51 =====
=====
===== CHANNEL F52 =====
=====
===== CHANNEL F53 =====
=====
===== CHANNEL F54 =====
=====
===== CHANNEL F55 =====
=====
===== CHANNEL F56 =====
=====
===== CHANNEL F57 =====
=====
===== CHANNEL F58 =====
=====
===== CHANNEL F59 =====
=====
===== CHANNEL F60 =====
=====
===== CHANNEL F61 =====
=====
===== CHANNEL F62 =====
=====
===== CHANNEL F63 =====
=====
===== CHANNEL F64 =====
=====
===== CHANNEL F65 =====
=====
===== CHANNEL F66 =====
=====
===== CHANNEL F67 =====
=====
===== CHANNEL F68 =====
=====
===== CHANNEL F69 =====
=====
===== CHANNEL F70 =====
=====
===== CHANNEL F71 =====
=====
===== CHANNEL F72 =====
=====
===== CHANNEL F73 =====
=====
===== CHANNEL F74 =====
=====
===== CHANNEL F75 =====
=====
===== CHANNEL F76 =====
=====
===== CHANNEL F77 =====
=====
===== CHANNEL F78 =====
=====
===== CHANNEL F79 =====
=====
===== CHANNEL F80 =====
=====
===== CHANNEL F81 =====
=====
===== CHANNEL F82 =====
=====
===== CHANNEL F83 =====
=====
===== CHANNEL F84 =====
=====
===== CHANNEL F85 =====
=====
===== CHANNEL F86 =====
=====
===== CHANNEL F87 =====
=====
===== CHANNEL F88 =====
=====
===== CHANNEL F89 =====
=====
===== CHANNEL F90 =====
=====
===== CHANNEL F91 =====
=====
===== CHANNEL F92 =====
=====
===== CHANNEL F93 =====
=====
===== CHANNEL F94 =====
=====
===== CHANNEL F95 =====
=====
===== CHANNEL F96 =====
=====
===== CHANNEL F97 =====
=====
===== CHANNEL F98 =====
=====
===== CHANNEL F99 =====
=====
===== CHANNEL F100 =====
=====
```



**<sup>1</sup>H spectrum**

```

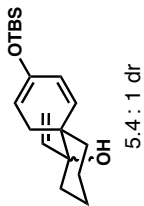
Current Data Parameters
=====
USER          : toonen
EXPNO         : 1
PROCNO        : 1
PROCPS        : 1
F2 - Acquisition Parameters
=====
Date_         : 2015_09
Time          : 15.09
PROBHD1       : 5 mm CPYCH 1H-
PULPROG       : zgpg30
SOLVENT       : CDCl3
DS            : 2
AQ            : 0.17000000 sec
RG            : 0.10000000 Hz
FIDRES        : 4.9999173 sec
AQ            : 62.400 usec
RG            : 0.10000000 Hz
TE            : 298.2 K
AQC           : 0.00000000 sec
MCBPC1        : 0.00000000 sec
MCBPC2        : 0.01500000 sec
=====
NAME          : CHANDEL.E1
P1            : 7.50 usec
PC            : 4.00
PR            : 200.22735015 MHz
SFO3          : 200.22735015 MHz
=====
F2 - Processing parameters
=====
SI            : 65536
SF            : 500.260768 MHz
WDW           : EM
SSB           : 0
GB            : 0
PC            : 4.00
  
```





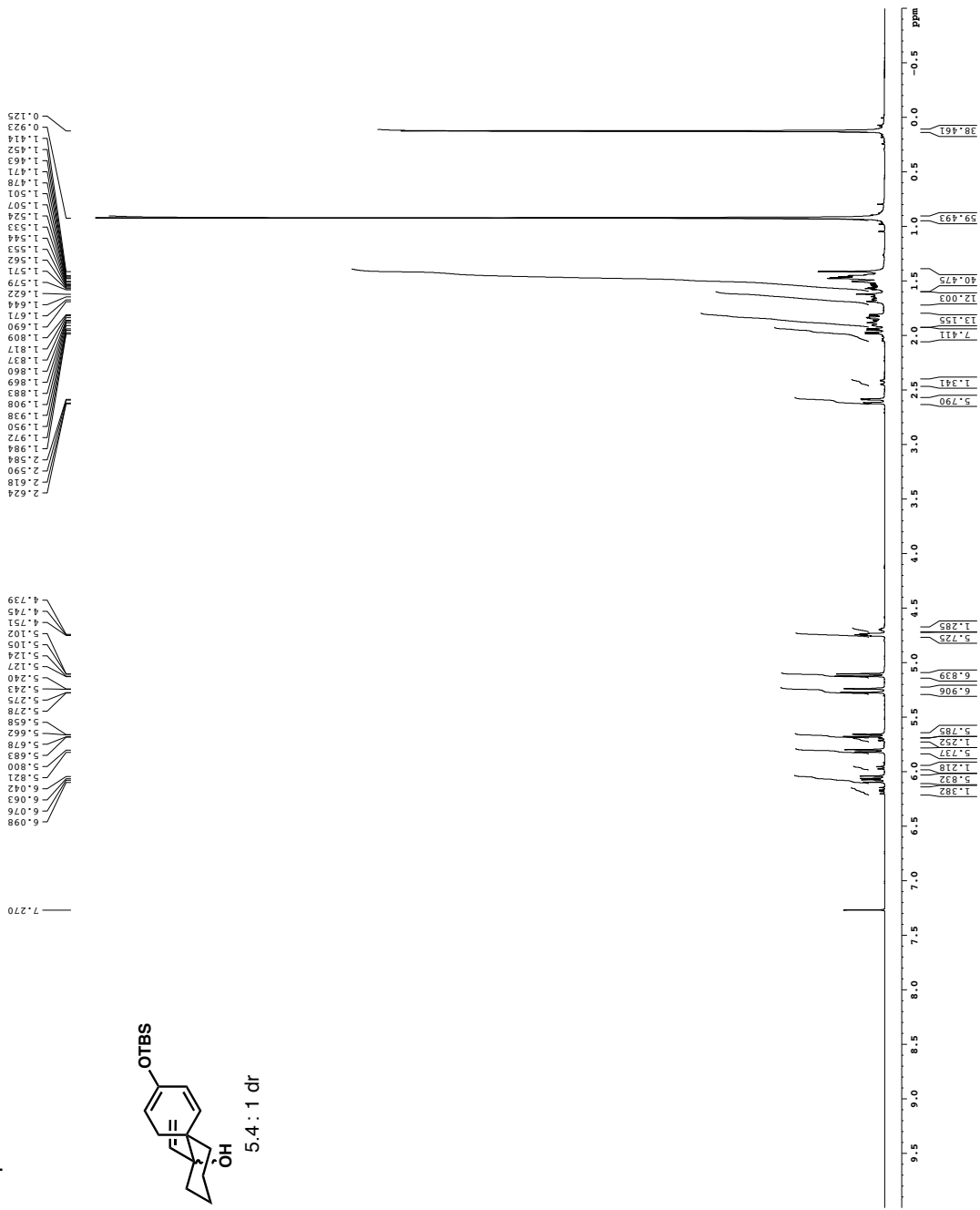


# **<sup>1</sup>H spectrum**



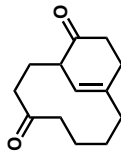
```

Current Data Parameters
=====
USER          : toonen
PROBHD        : PZ1.1MM.P1.19
PROCNO        : 1
F2 - Acquisition Parameters
=====
Date_         : 2010.09
Time          : 01.00
PROBHD        : 5 mm CPCCZ 1H-
PULPROG       : zgpg30
SOLVENT       : CDCl3
DS            : 2
AQ            : 6.675022 sec
FIDRES        : 0.0750020 Hz
RG            : 6.6665330 sec
DM            : 83.200 usec
DE            : 1.488.0 K
TE            : 300.2
AQC           : 0.00000000 sec
MCBRES1       : 0.01300000 sec
=====
NAME          : CHANREL.E1
P1            : 7.50 usec
PC            : 4.00
=====
F2 - Processing parameters
=====
SI            : 65536
SF            : 500.1370028 MHz
WDW           : EM
SSB           : 0
GB            : 0
PC            : 4.00
  
```

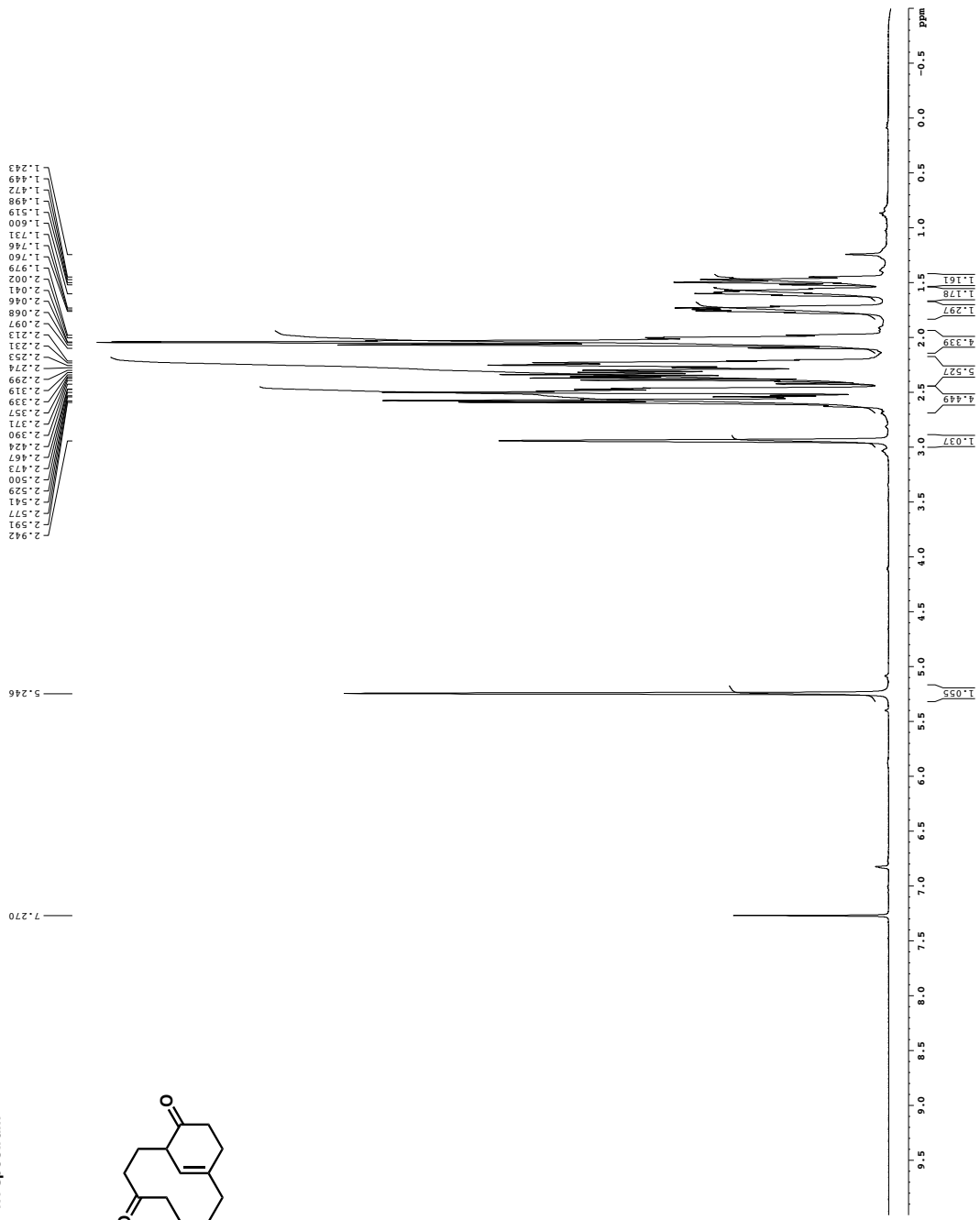




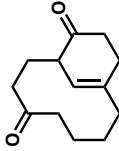
**<sup>1</sup>H spectrum**



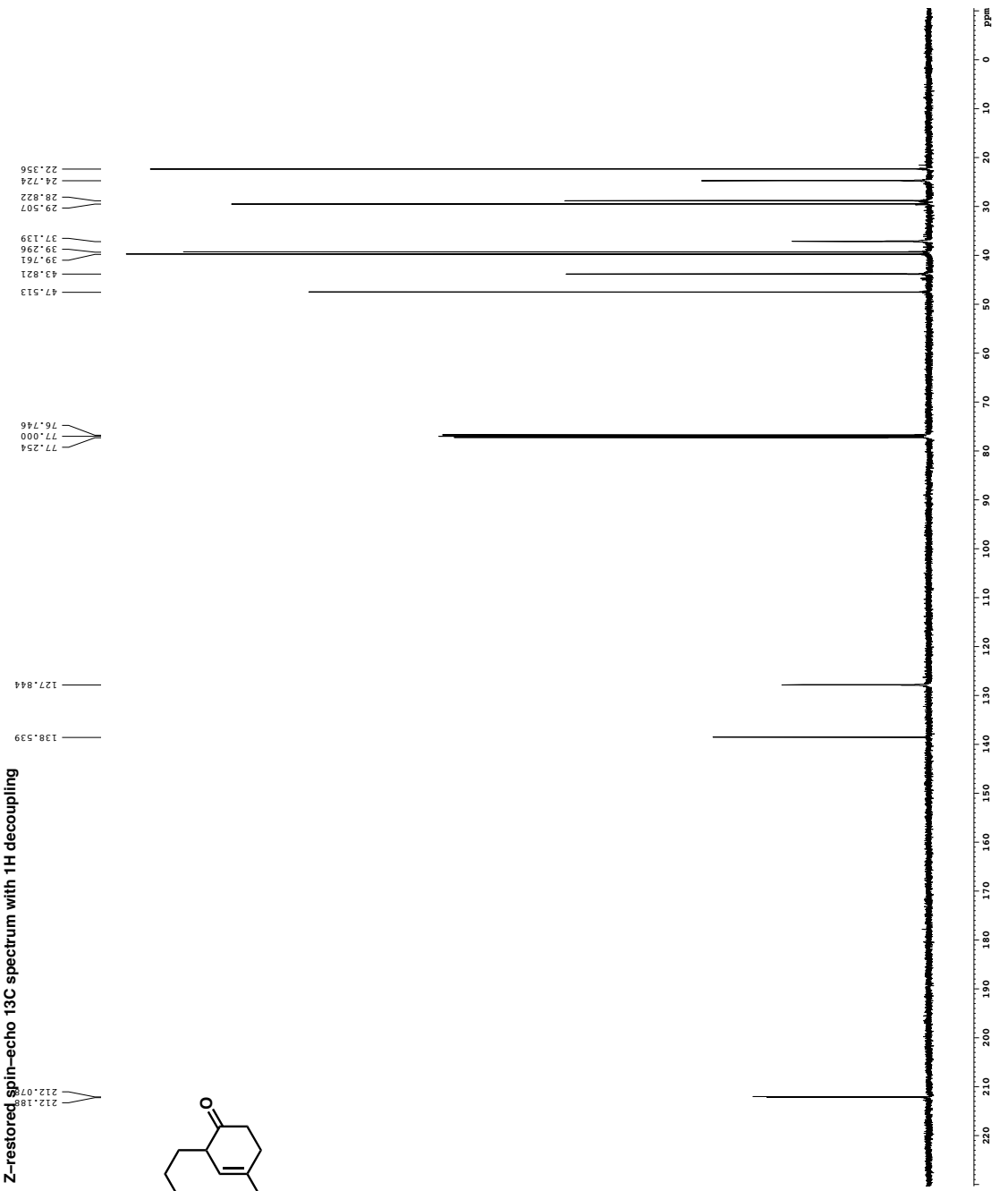
Current Data Parameters  
 USER pcr3\_041\_1000en  
 INPRO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Title\_ 2014\_08\_29  
 Date\_ 2014\_08\_29  
 Time\_ 8:29  
 INSTRUM spect  
 PROBMOD 2 mm CPYCH 1H-  
 PULPROG zgpg30  
 TD 32768  
 SFO 500.135436  
 SOLVENT CDCl3  
 DS 4  
 F1 7.50 uSAC  
 F2 7.50 uSAC  
 F3 7.50 uSAC  
 FWHM 0.220026 Hz  
 AQ 1.9986451 sec  
 RM 62.400 uSAC  
 DM 0.10000000 sec  
 TE 298.2 K  
 ACQ 0.10000000 sec  
 SCANS 1  
 ACRES 0.01300000 sec  
 MCRES 0.01300000 sec  
 ===== CHANNEL f1 =====  
 P1 7.50 uSAC  
 PL1 0.00000000 dB  
 SFO1 500.135436 MHz  
 F2 - Processing parameters  
 SI 655536 Hz  
 SF 500.135436 MHz  
 WDW EM  
 LB 0.30 Hz  
 GB 0  
 PC 4.00



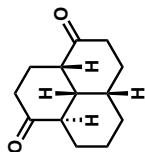
### Z-restored spin-echo 13C spectrum with 1H decoupling



```
Current Data Parameters
NAME: PC3_044_20200415_1
PROCNO: 1
=====
F2 - Acquisition Parameters
Date_ 2018. 08. 31
Time 08. 44. 11
PROBHD: 5 mm CPYXCH-1H
NUC1: 13C
PULPROG: zgpg30
SOLVENT: CDCl3
DS: 4
AQ: 3.61500000 sec
FIDRES: 0.482368 Hz
RG: 1.063260 sec
AQ: 16.500 usec
TE: 300.2 K
NUC1: 13C
DE: 2.982000 usec
AQ: 0.03000000 sec
NUC2: 1H
DE: 0.00000000 sec
AQ: 0.00000000 sec
NUC3: 1H
DE: 0.00000000 sec
AQ: 0.00000000 sec
P2: 0.01000000 sec
=====
F1 - Acquisition Parameters
NAME: PC3_044_20200415_2
PROCNO: 2
=====
F1 - Acquisition Parameters
NAME: PC3_044_20200415_1
PROCNO: 1
=====
F1 - Processing Parameters
SI: 65536
SF: 125.760420 MHz
WDW: EM
SSB: 0
GB: 0
PC: 2.00
=====
===== CHANNEL F1 =====
NUC1: 13C
P1: 15.30 usec
PL1: 0.00 dB
PL2: 0.00 dB
PL3: 0.00 dB
PL4: 0.00 dB
PL5: 0.00 dB
PL6: 0.00 dB
PL7: 0.00 dB
PL8: 0.00 dB
PL9: 0.00 dB
PL10: 0.00 dB
PL11: 0.00 dB
PL12: 0.00 dB
PL13: 0.00 dB
PL14: 0.00 dB
PL15: 0.00 dB
=====
===== CHANNEL F2 =====
NAME: PC3_044_20200415_2
PROCNO: 2
=====
F2 - Acquisition Parameters
Date_ 2018. 08. 31
Time 08. 44. 11
PROBHD: 5 mm CPYXCH-1H
NUC1: 13C
PULPROG: zgpg30
SOLVENT: CDCl3
DS: 4
AQ: 3.61500000 sec
FIDRES: 0.482368 Hz
RG: 1.063260 sec
AQ: 16.500 usec
TE: 300.2 K
NUC1: 13C
DE: 2.982000 usec
AQ: 0.03000000 sec
NUC2: 1H
DE: 0.00000000 sec
AQ: 0.00000000 sec
NUC3: 1H
DE: 0.00000000 sec
AQ: 0.00000000 sec
P2: 0.01000000 sec
=====
F2 - Processing Parameters
SI: 65536
SF: 125.760420 MHz
WDW: EM
SSB: 0
GB: 0
PC: 2.00
=====
===== CHANNEL F1 =====
NUC1: 13C
P1: 15.30 usec
PL1: 0.00 dB
PL2: 0.00 dB
PL3: 0.00 dB
PL4: 0.00 dB
PL5: 0.00 dB
PL6: 0.00 dB
PL7: 0.00 dB
PL8: 0.00 dB
PL9: 0.00 dB
PL10: 0.00 dB
PL11: 0.00 dB
PL12: 0.00 dB
PL13: 0.00 dB
PL14: 0.00 dB
PL15: 0.00 dB
=====
===== CHANNEL F2 =====
NAME: PC3_044_20200415_2
PROCNO: 2
=====
F2 - Acquisition Parameters
Date_ 2018. 08. 31
Time 08. 44. 11
PROBHD: 5 mm CPYXCH-1H
NUC1: 13C
PULPROG: zgpg30
SOLVENT: CDCl3
DS: 4
AQ: 3.61500000 sec
FIDRES: 0.482368 Hz
RG: 1.063260 sec
AQ: 16.500 usec
TE: 300.2 K
NUC1: 13C
DE: 2.982000 usec
AQ: 0.03000000 sec
NUC2: 1H
DE: 0.00000000 sec
AQ: 0.00000000 sec
NUC3: 1H
DE: 0.00000000 sec
AQ: 0.00000000 sec
P2: 0.01000000 sec
=====
F2 - Processing Parameters
SI: 65536
SF: 125.760420 MHz
WDW: EM
SSB: 0
GB: 0
PC: 2.00
=====
```



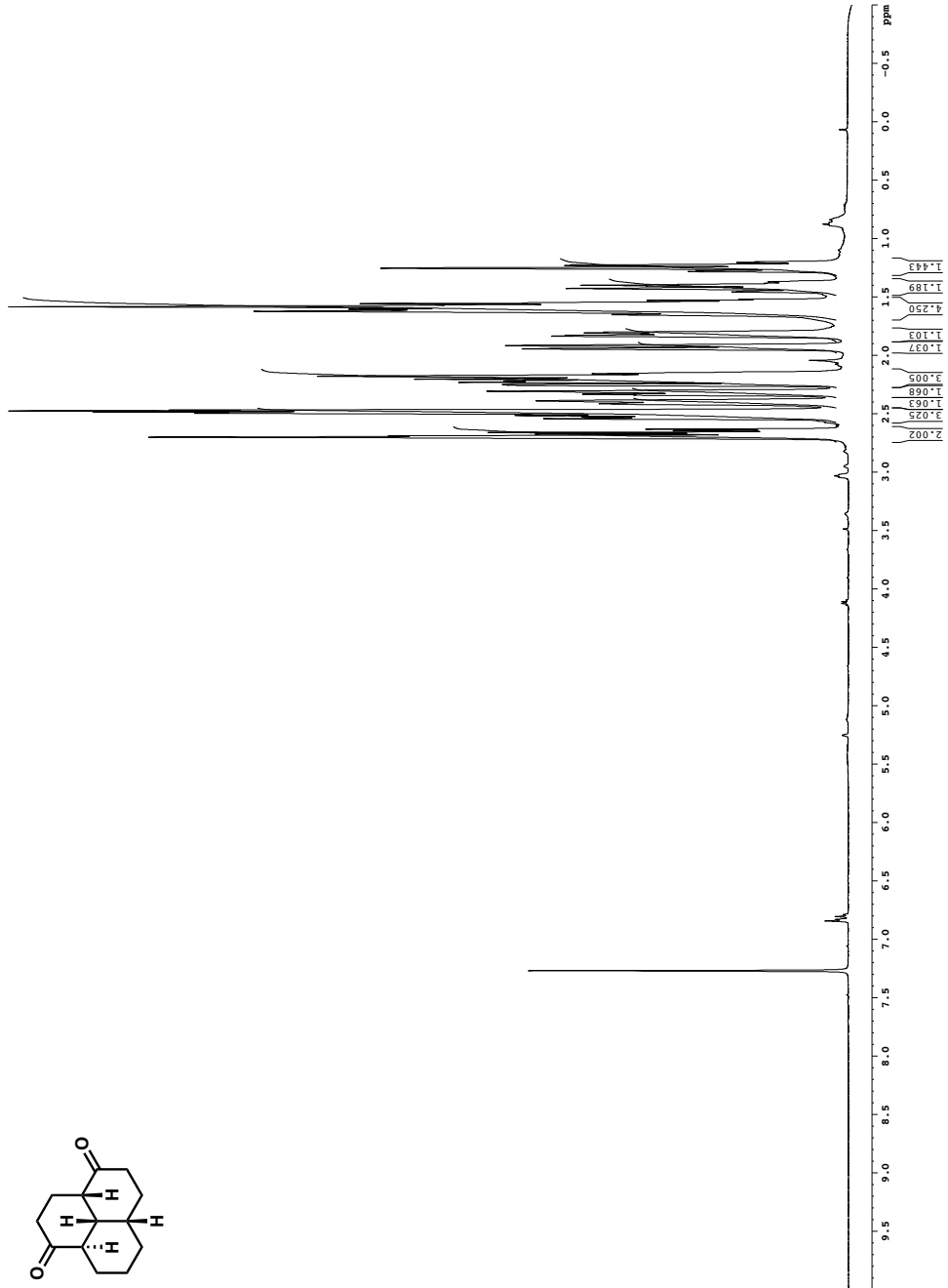
**<sup>1</sup>H spectrum**



7.270

2.701  
2.674  
2.661  
2.646  
2.633  
2.543  
2.530  
2.516  
2.497  
2.486  
2.477  
2.469  
2.416  
2.390  
2.383  
2.308  
2.251  
2.232  
2.205  
2.180  
2.158  
2.044  
1.945  
1.917  
1.936  
1.809  
1.652  
1.623  
1.598  
1.558  
1.530  
1.457  
1.428  
1.400  
1.283  
1.255  
1.232  
1.206

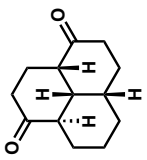
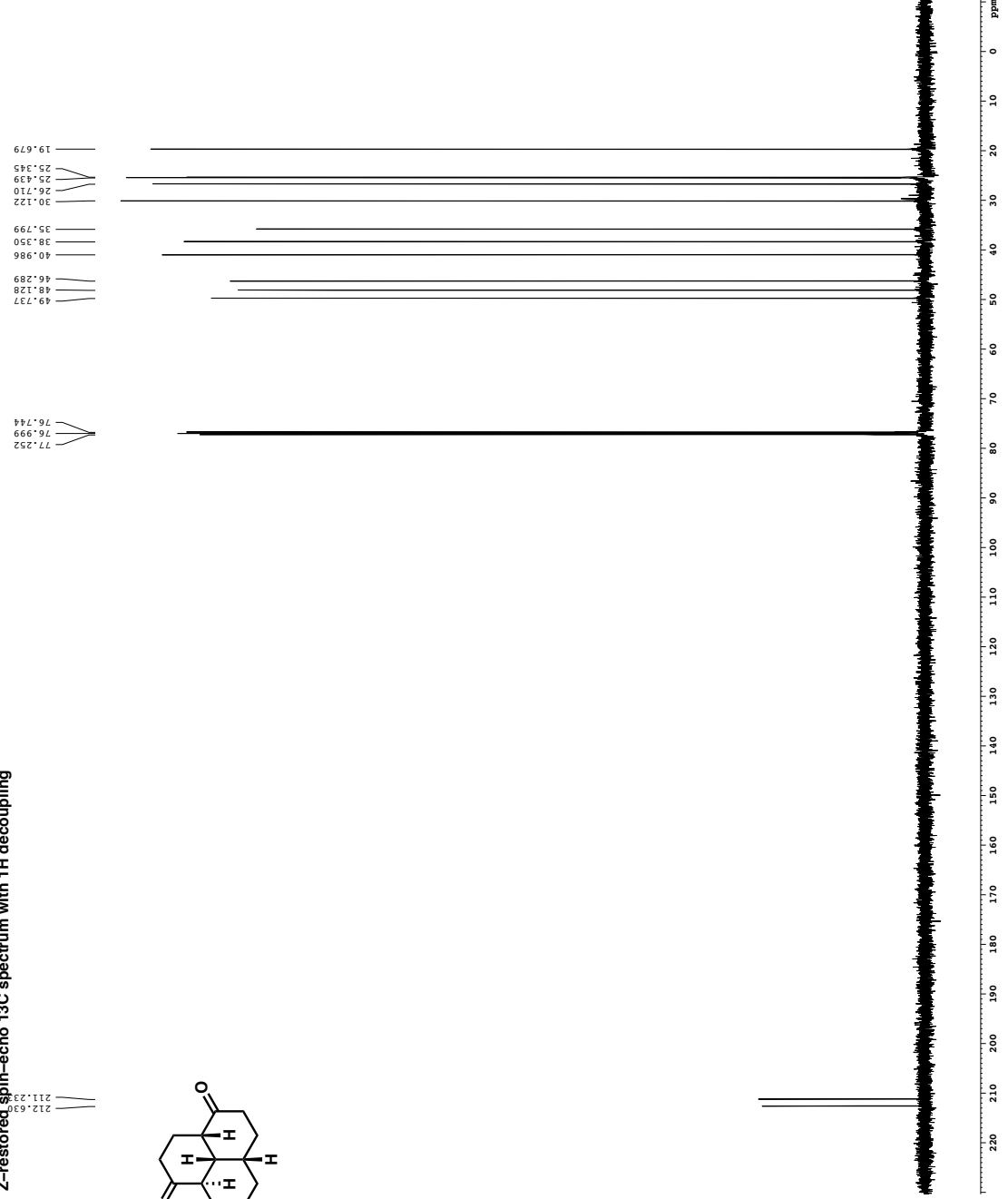
Current Data Parameters  
 USER boonen  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2018-08-30  
 Time 14:50  
 PROBM 5 mm CPYCH-1H  
 PULPROG zgpg30  
 TOU 2.00  
 SOLVENT CDCl3  
 DS 2  
 B1 0.12  
 B2 0.12  
 FIDRES 0.220026 Hz  
 AQ 1.9998451 sec  
 RM 62.400 usec  
 DM 0.129820 K  
 TE 300.2 K  
 ACF 0.00000000 sec  
 MCORE 0.01300000 sec  
 ===== CHANNEL f1 =====  
 P1 7.50 usec  
 PL 0.00 dB  
 SFO1 500.2213015 MHz  
 F2 - Processing parameters  
 SI 65536  
 SF 500.220726 MHz  
 WDW EM  
 SSB 0  
 LB 0.30 Hz  
 GB 0  
 PC 4.00



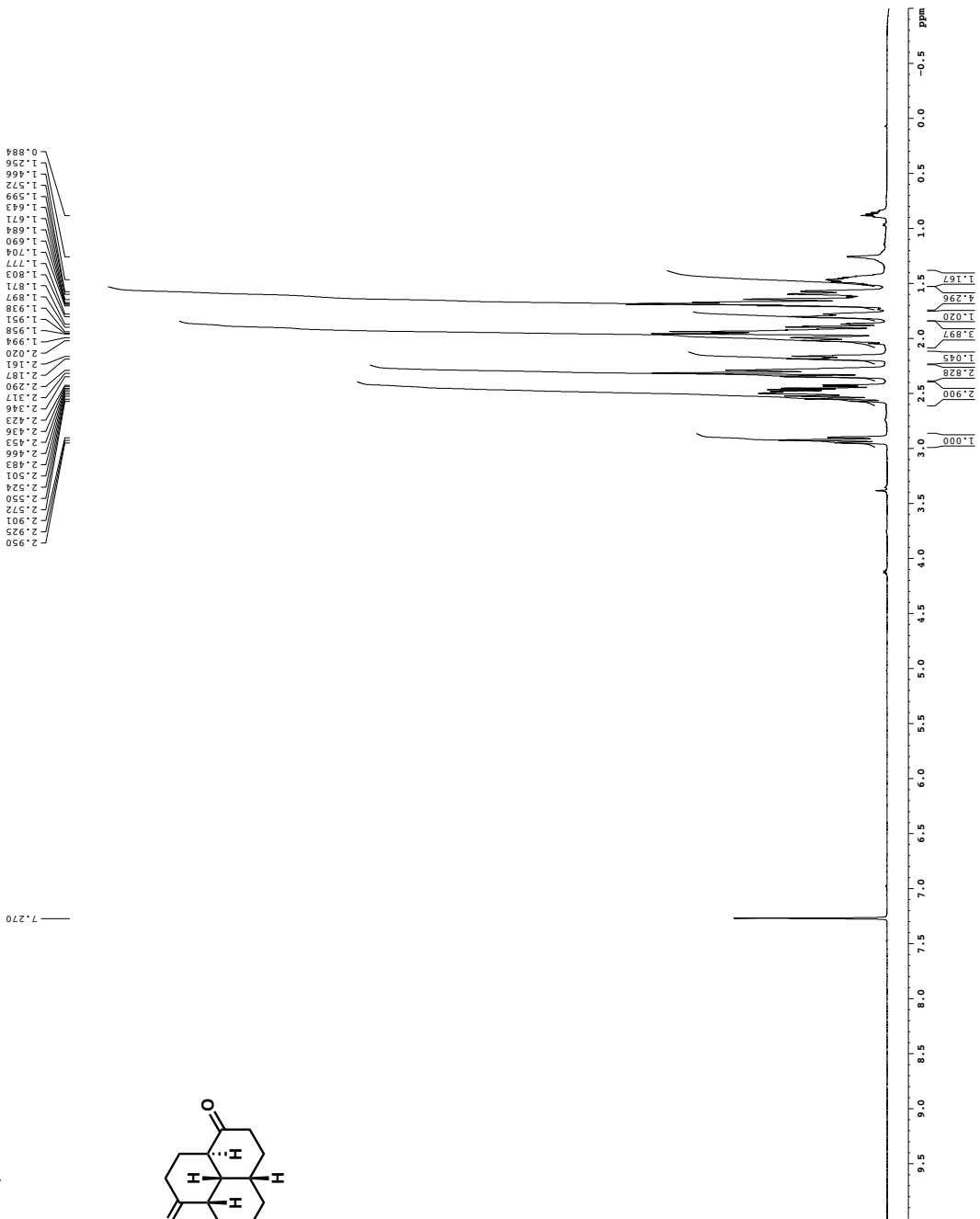
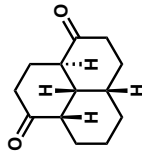
Z-restored spin-echo 13C spectrum with 1H decoupling

```

Current Data Parameters
=====
USER      Pct3_063_jm12
PROBHD    5 mm CPYCC 1H
P2 - Acquisition Parameters
=====
Date_     20160823
Time      8:53
PROBHD    5 mm CPYCC 1H
NUC1      13C
PULPROG   zgpg30
SOLVENT   CDCl3
DS         2
AQ         3.61700000 Hz
RG         1.00000000 sec
FIDRES     0.482388 Hz
AQ         1.00000000 sec
RG         16.50000000 Hz
TE         300.2 K
TR         0.29820000 sec
D1         0.0300000000 sec
D11        0.0000000000 sec
D15        0.0000000000 sec
DELTA     0.0000000000 sec
NUC2      13C
NUC2FREQ  125.7614588 MHz
P2         0.0100000000 sec
=====
===== CHANNEL F1 =====
NUC1      13C
P1         1.5000000000 sec
P11        500.0000000000 MHz
P12        12.0000000000 dB
P13        12.0000000000 dB
P14        12.0000000000 dB
P15        12.0000000000 dB
P16        12.0000000000 dB
P17        12.0000000000 dB
P18        12.0000000000 dB
P19        12.0000000000 dB
P20        12.0000000000 dB
P21        12.0000000000 dB
P22        12.0000000000 dB
P23        12.0000000000 dB
P24        12.0000000000 dB
P25        12.0000000000 dB
P26        12.0000000000 dB
P27        12.0000000000 dB
P28        12.0000000000 dB
P29        12.0000000000 dB
P30        12.0000000000 dB
===== CHANNEL F2 =====
NUC2      13C
P1         1.5000000000 sec
P11        500.0000000000 MHz
P12        12.0000000000 dB
P13        12.0000000000 dB
P14        12.0000000000 dB
P15        12.0000000000 dB
P16        12.0000000000 dB
P17        12.0000000000 dB
P18        12.0000000000 dB
P19        12.0000000000 dB
P20        12.0000000000 dB
P21        12.0000000000 dB
P22        12.0000000000 dB
P23        12.0000000000 dB
P24        12.0000000000 dB
P25        12.0000000000 dB
P26        12.0000000000 dB
P27        12.0000000000 dB
P28        12.0000000000 dB
P29        12.0000000000 dB
P30        12.0000000000 dB
===== GRABBER CHANNEL =====
GRABPRG   zgpg30
=====
===== Processing parameters =====
SI         65536
SF         125.7614588 MHz
WDW        EM
SSB        0
GB         0
PC         2.00
  
```



**<sup>1</sup>H spectrum**



Current Data Parameters  
 USER: hosen  
 EXPNO: 1  
 PROCNO: 1  
 F2 - Acquisition Parameters  
 Date\_ : 15.23  
 Time: 15.23  
 PROBHD: 5 mm CPCLP 1H-  
 PULPROG: zgpg30  
 SOLVENT: CDCl3  
 DS: 2  
 BS: 2  
 FIDRES: 0.1262 Hz  
 AQ: 0.226026 sec  
 RG: 1.9998451  
 INJ: 62.400 usec  
 DE: 0.19820 K  
 TE: 300.2 K  
 ACQRES: 0.10000000 sec  
 SCANS: 0.01300000 sec  
 CHANNEL: CHANDEL F1  
 NU1: 7.50 usec  
 PL1: 0.00000000  
 PR1: 500.22735015 MHz  
 FREQ: 500.22735015 MHz  
 F2 - Processing parameters  
 SI: 65536  
 SF: 500.22735015 MHz  
 WDW: EM  
 SSB: 0.00 Hz  
 CB: 4.00



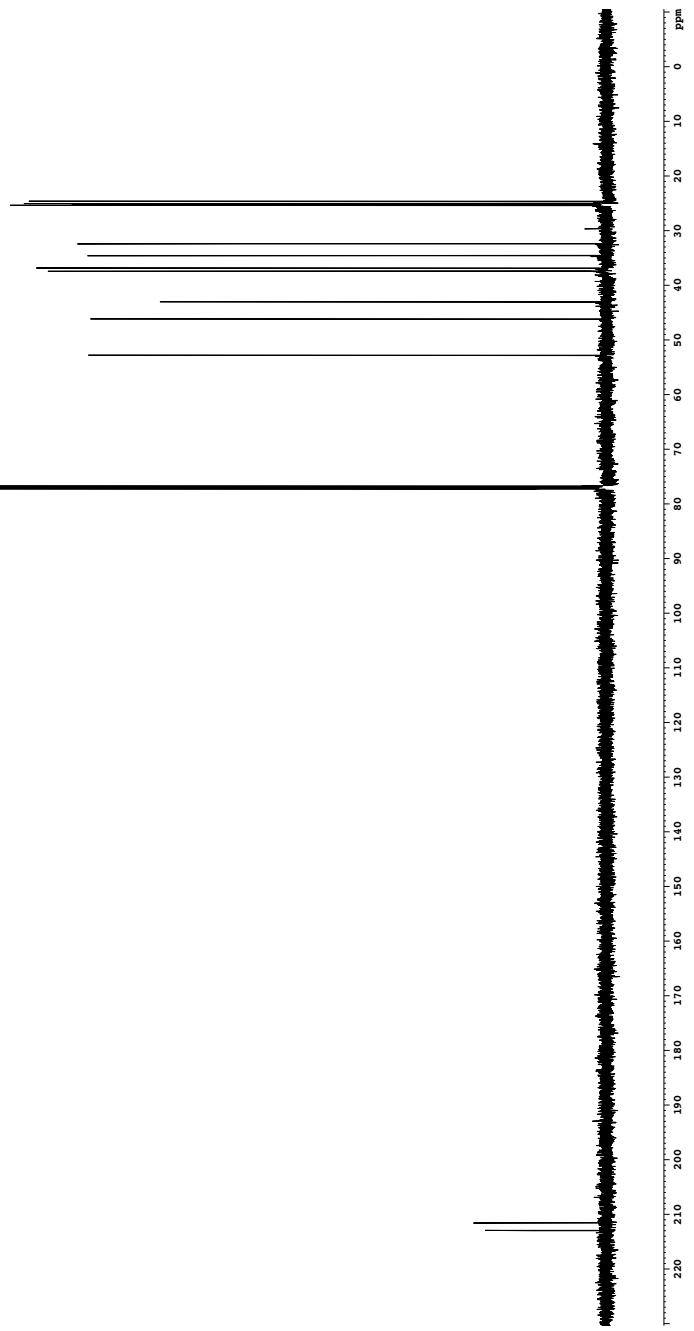
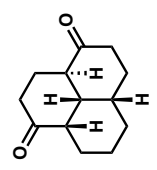
Z-restored spin-echo 13C spectrum with 1H decoupling

```

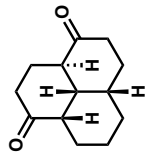
Current Data Parameters
=====
USER      pcr3_06_2008m_2
EXPNO     2
PROCNO    1

F2 - Acquisition Parameters
=====
Time      24.1525
Date_     041115
PROBHD    5 mm CPYCC 1H-
PULPROG   zgpg30
SOLVENT   DMSO-d6
SOLVENT2  CDCl3
DS         4
AQ         3.61700000 Hz
RG         0.482388 Hz
RG2        1.00000000 sec
RG3        16.50000000 sec
TE         300.2 K
TR         0.00000000 sec
TI1        0.00000000 sec
TI2        0.00000000 sec
TI3        0.00000000 sec
TI4        0.00000000 sec
TI5        0.00000000 sec
TI6        0.00000000 sec
TI7        0.00000000 sec
TI8        0.00000000 sec
TI9        0.00000000 sec
TI10       0.00000000 sec
TI11       0.00000000 sec
TI12       0.00000000 sec
TI13       0.00000000 sec
TI14       0.00000000 sec
TI15       0.00000000 sec
TI16       0.00000000 sec
TI17       0.00000000 sec
TI18       0.00000000 sec
TI19       0.00000000 sec
TI20       0.00000000 sec
=====
F1 - Processing Parameters
=====
NUC1       13C
P1         15.30000000 sec
P11        500.00000000 MHz
P2         12.00000000 sec
P21        120.00000000 MHz
P3         125.76825600 MHz
P31        3.20000000 dB
P32        3.20000000 dB
P33        3.20000000 dB
P34        3.20000000 dB
P35        3.20000000 dB
P36        3.20000000 dB
P37        3.20000000 dB
P38        3.20000000 dB
P39        3.20000000 dB
P40        3.20000000 dB
=====
F2 - Processing Parameters
=====
SI         65536
SF         125.76825600 MHz
WDW        EM
SSB        0
GB         0
PC         2.00
  
```

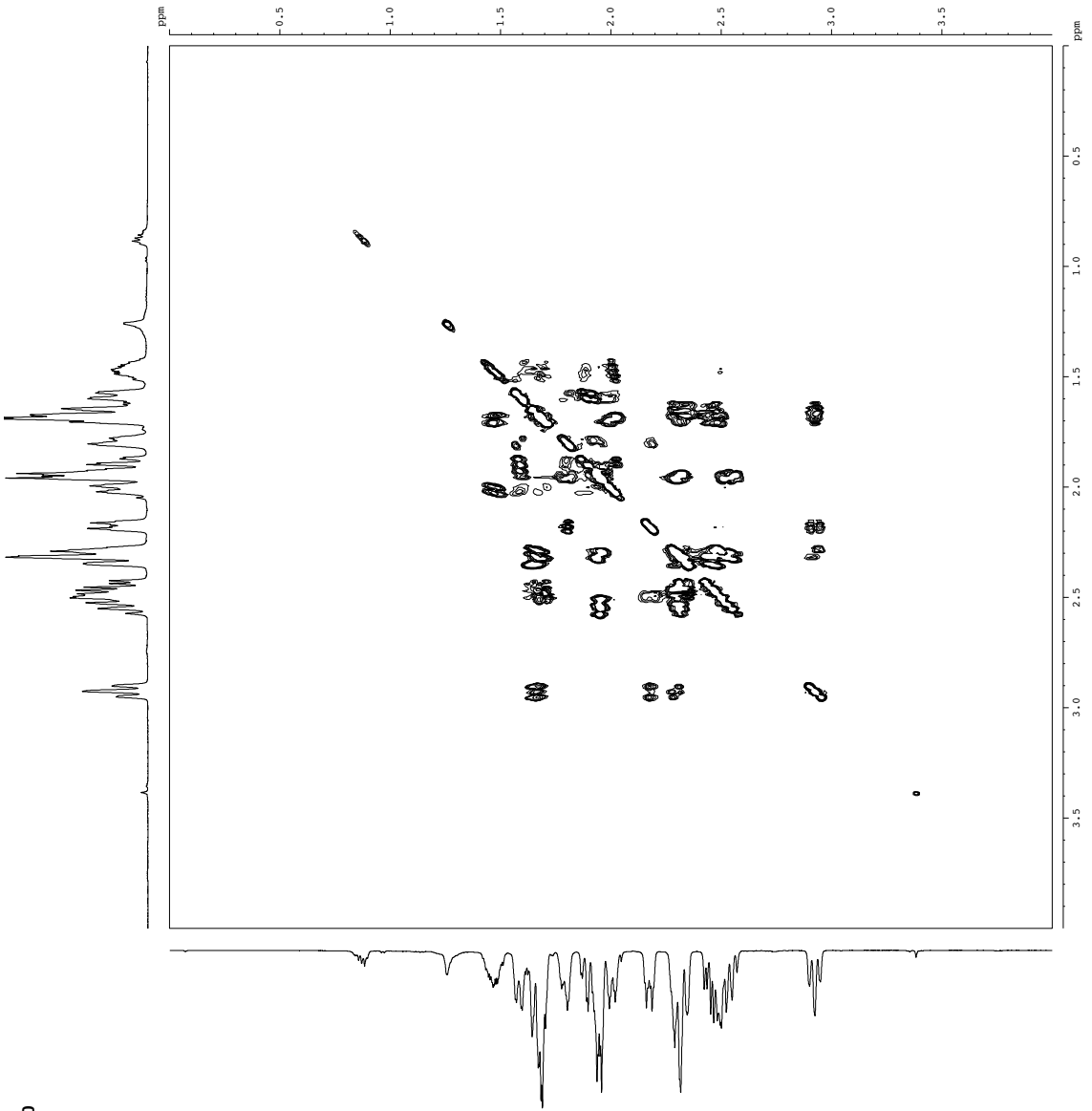
212.999  
 211.616  
 77.255  
 77.001  
 76.747  
 52.807  
 46.179  
 43.045  
 37.423  
 36.961  
 34.591  
 32.406  
 25.375  
 25.147  
 24.624



gcosy60



Current Data Parameters  
USER pcc1\_06\_mdir\_5  
EXPNO 1  
PROCNO 1  
Date\_ Acq 20110223  
Time 16:03  
PROBHD 5 mm CPXI 1H-  
TD 65536  
SFO1 500.2210270 MHz  
NUC1 1H  
P1 1.00 usec  
RG 327.680 Hz  
WDW EM  
SSB 0  
GB 0  
PC 4.00  
F1 - Acquisition parameters  
SFO1 500.2210270 MHz  
SFO2 150.222 MHz  
SFO3 40.009 MHz  
PROCNO unshifted  
F2 - Processing parameters  
SI 327.680 Hz  
SF 500.2210270 MHz  
WDW EM  
SSB 0  
GB 0  
PC 4.00  
F1 - Processing parameters  
SFO1 500.2210270 MHz  
SFO2 150.222 MHz  
SFO3 40.009 MHz

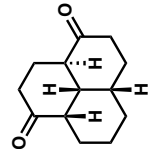
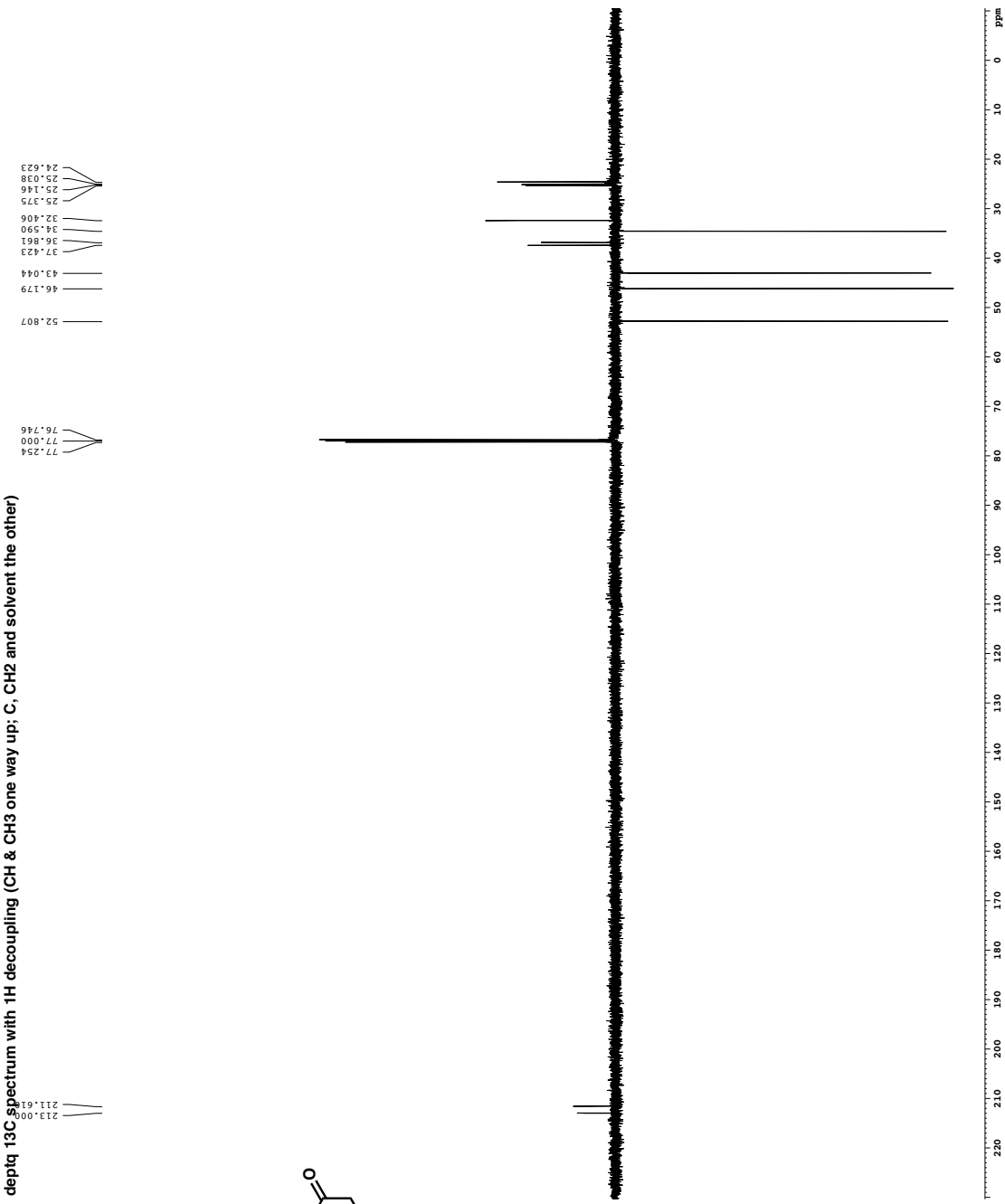


deptq 13C spectrum with 1H decoupling (CH & CH3 one way up; C, CH2 and solvent the other)

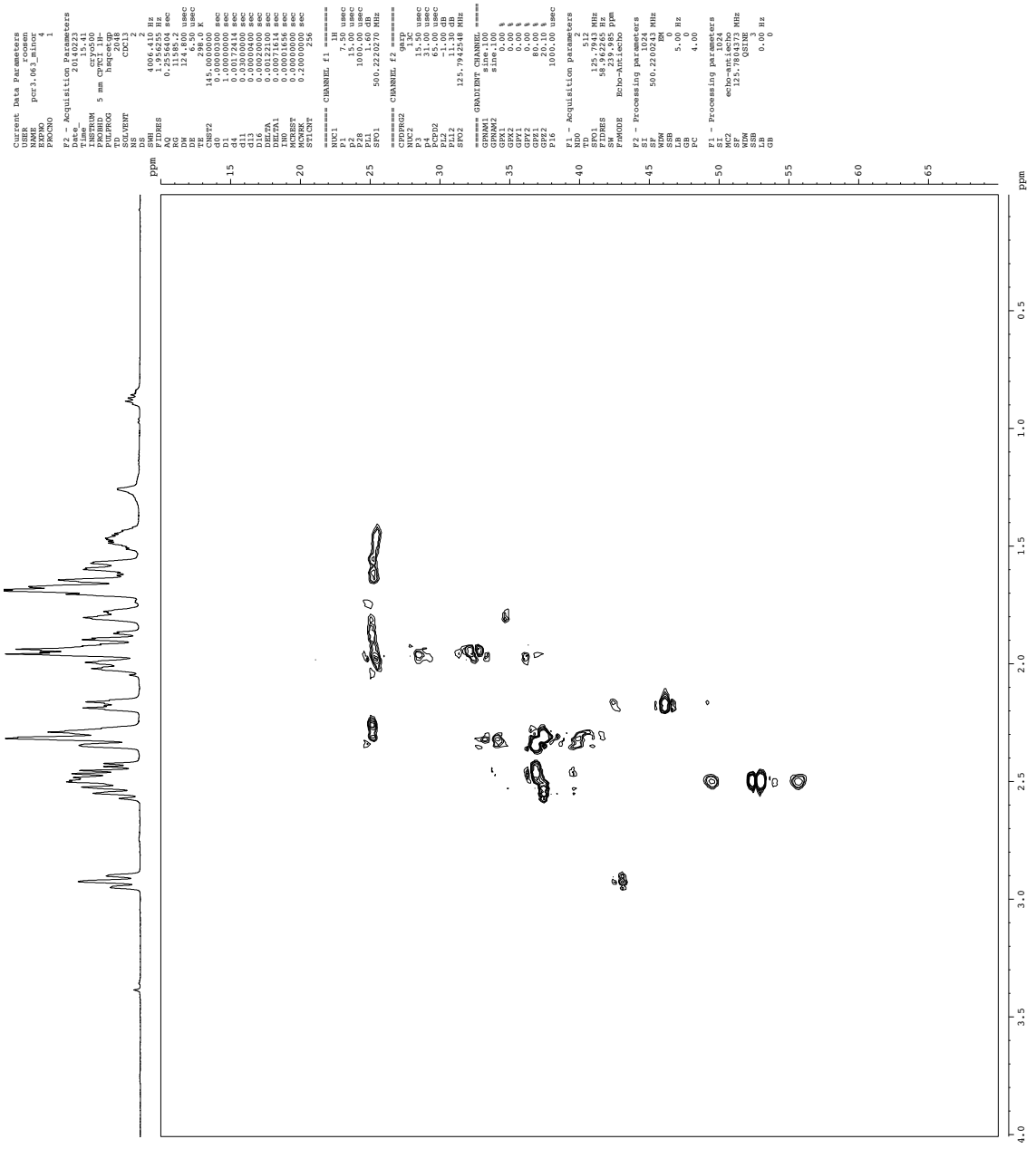
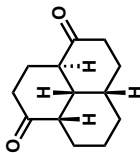
```

Current Data Parameters
=====
USRR          pc3_060_31mm_3
EXPNO         3
PROCNO        1

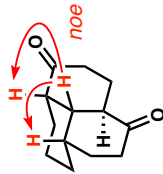
F2 - Acquisition Parameters
=====
Time          24.15.37
Date_         20060916
PROBHD        5 mm CPCLP 1H-
PULPROG       zgpg30
TD             65536
SOLVENT       CDCl3
AQ            0.0224828 sec
RG            312.5
DS            4
SWH            16.300 MHz
FIDRES        0.482368 Hz
AQ            1.0683048 sec
RG            312.5
DS            4
TE            300.2 K
=====
Channel f1
=====
NUC1           13C
P1            15.50 uSAC
P2            2000.00 uSAC
PC            1.00 dB
PL1           -1.00 dB
PL2           -1.00 dB
PL12          125.790000 MHz
SFO1           101.626126 MHz
SFO2           500.136300 MHz
=====
Channel f2
=====
CPDPRG2       waltz16
NUC2           13C
P1            11.25 uSAC
P2            17.50 uSAC
PC            1.00 dB
PL1           -1.00 dB
PL2           -1.00 dB
PL12          125.790000 MHz
SFO1           101.626126 MHz
SFO2           500.136300 MHz
=====
Gradient Channel
=====
GRADRG        GR1
GRADNUC        13C
GRADP1         0.00 %
GRADP2         0.00 %
GRADP3         0.00 %
GRADP4         0.00 %
GRADP5         0.00 %
GRADP6         0.00 %
GRADP7         0.00 %
GRADP8         0.00 %
GRADP9         0.00 %
GRADP10        0.00 %
GRADP11        0.00 %
GRADP12        0.00 %
GRADP13        0.00 %
GRADP14        0.00 %
GRADP15        0.00 %
GRADP16        0.00 %
GRADP17        0.00 %
GRADP18        0.00 %
GRADP19        0.00 %
GRADP20        0.00 %
=====
F3 - Processing Parameters
=====
SI            32768
SF            125.760426 MHz
WDW           EM
SSB           0
GB            0
PC            2.00
  
```



9HSQC



gnoe



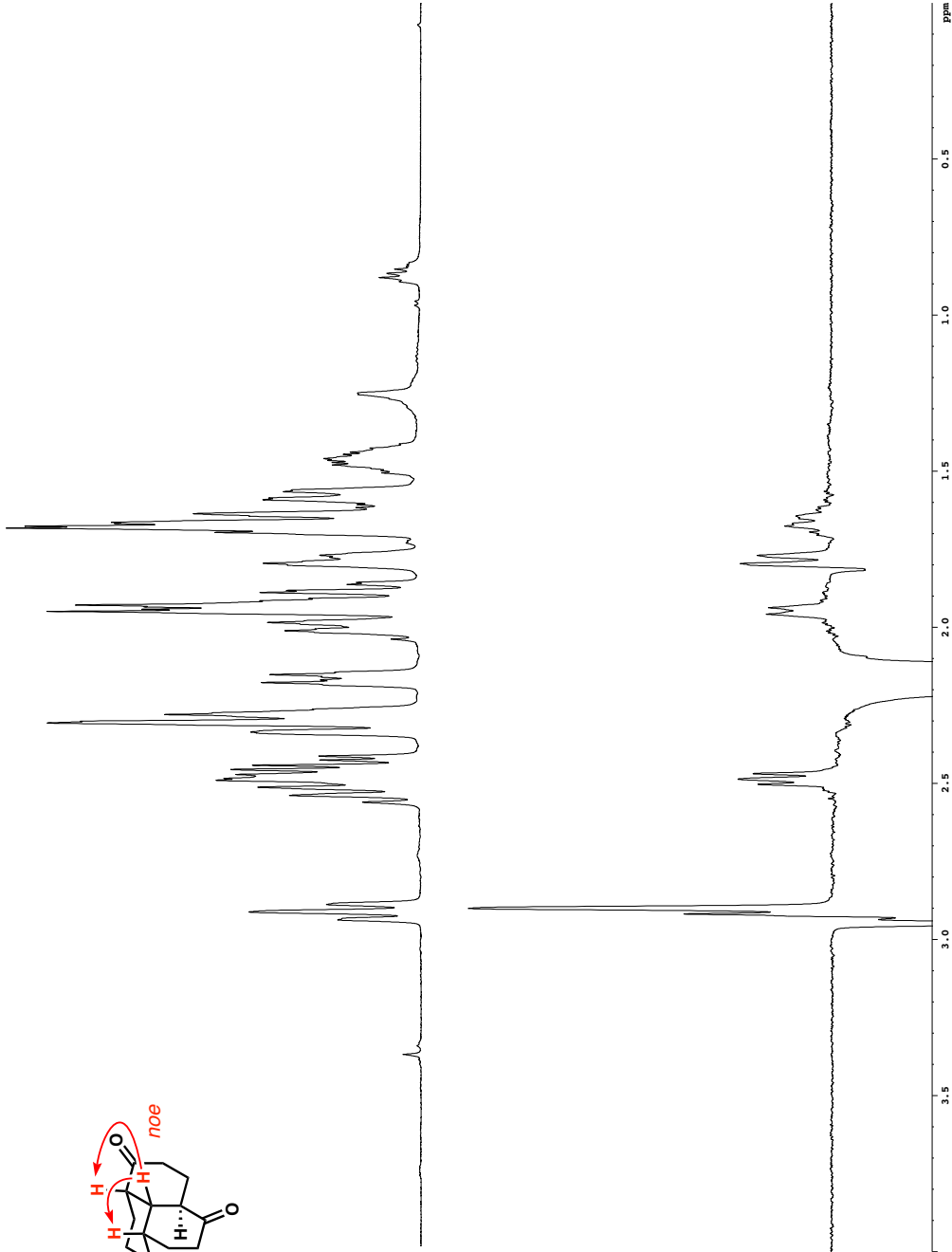
```
Current Data Parameters
=====
USER      PCT3_063_201107_7
EXPNO     1
PROCNO    1

F2 - Acquisition Parameters
=====
Time          21.1637
Date_         20110707
PROBHD       5 mm CPYCC 1H-
PULPROG      gnoe13
TD            65536
SOLVENT      CDCl3
DS           4
AQ           0.142
FIDRES       0.112266 Hz
AQ           4.0889956 sec
RG           62.400
DM           62.400 usec
TE           300.2
TR           1.1629820 sec
DE           0.0000000 sec
DB           0.0000000 sec
DQ           0.0000000 sec
CD2         0.1639699 sec
PC           15.00 usec

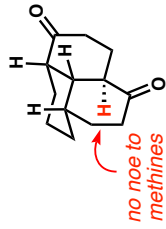
===== CHANNEL f1 =====
NUC1        CHANDEL F1
P1          7.50 usec
PA          30.00 usec
PR          30.00 usec
PL          40000.00 usec
PZ          0.0000000 usec
SFO1        500.2210961 MHz
SI1         gnoc1
SF1         500.2210961 MHz
SFOF1       0.100 Hz

===== GRADIENT CHANNEL =====
GDMG3       15.00 usec
GDMG4       15.00 usec
GDMG5       15.00 usec
GDMG6       15.00 usec
GDMG7       15.00 usec
GDMG8       15.00 usec
GDMG9       15.00 usec
GDMG10      15.00 usec
GDMG11      15.00 usec
GDMG12      15.00 usec
GDMG13      15.00 usec
GDMG14      15.00 usec
GDMG15      15.00 usec
GDMG16      15.00 usec
GDMG17      15.00 usec
GDMG18      15.00 usec
GDMG19      15.00 usec
GDMG20      15.00 usec
GDMG21      15.00 usec
GDMG22      15.00 usec
GDMG23      15.00 usec
GDMG24      15.00 usec
GDMG25      15.00 usec
GDMG26      15.00 usec
GDMG27      15.00 usec
GDMG28      15.00 usec
GDMG29      15.00 usec
GDMG30      15.00 usec
GDMG31      15.00 usec
GDMG32      15.00 usec
GDMG33      15.00 usec
GDMG34      15.00 usec
GDMG35      15.00 usec
GDMG36      15.00 usec
GDMG37      15.00 usec
GDMG38      15.00 usec
GDMG39      15.00 usec
GDMG40      15.00 usec
GDMG41      15.00 usec
GDMG42      15.00 usec
GDMG43      15.00 usec
GDMG44      15.00 usec
GDMG45      15.00 usec
GDMG46      15.00 usec
GDMG47      15.00 usec
GDMG48      15.00 usec
GDMG49      15.00 usec
GDMG50      15.00 usec
GDMG51      15.00 usec
GDMG52      15.00 usec
GDMG53      15.00 usec
GDMG54      15.00 usec
GDMG55      15.00 usec
GDMG56      15.00 usec
GDMG57      15.00 usec
GDMG58      15.00 usec
GDMG59      15.00 usec
GDMG60      15.00 usec
GDMG61      15.00 usec
GDMG62      15.00 usec
GDMG63      15.00 usec
GDMG64      15.00 usec
GDMG65      15.00 usec
GDMG66      15.00 usec
GDMG67      15.00 usec
GDMG68      15.00 usec
GDMG69      15.00 usec
GDMG70      15.00 usec
GDMG71      15.00 usec
GDMG72      15.00 usec
GDMG73      15.00 usec
GDMG74      15.00 usec
GDMG75      15.00 usec
GDMG76      15.00 usec
GDMG77      15.00 usec
GDMG78      15.00 usec
GDMG79      15.00 usec
GDMG80      15.00 usec
GDMG81      15.00 usec
GDMG82      15.00 usec
GDMG83      15.00 usec
GDMG84      15.00 usec
GDMG85      15.00 usec
GDMG86      15.00 usec
GDMG87      15.00 usec
GDMG88      15.00 usec
GDMG89      15.00 usec
GDMG90      15.00 usec
GDMG91      15.00 usec
GDMG92      15.00 usec
GDMG93      15.00 usec
GDMG94      15.00 usec
GDMG95      15.00 usec
GDMG96      15.00 usec
GDMG97      15.00 usec
GDMG98      15.00 usec
GDMG99      15.00 usec
GDMG100     15.00 usec

F2 - Processing parameters
=====
SI          65536
SF          500.2210961 MHz
WDW         EM
SSB         0
LB          1.00 Hz
GB          0
PC          1.00
```



gnoe



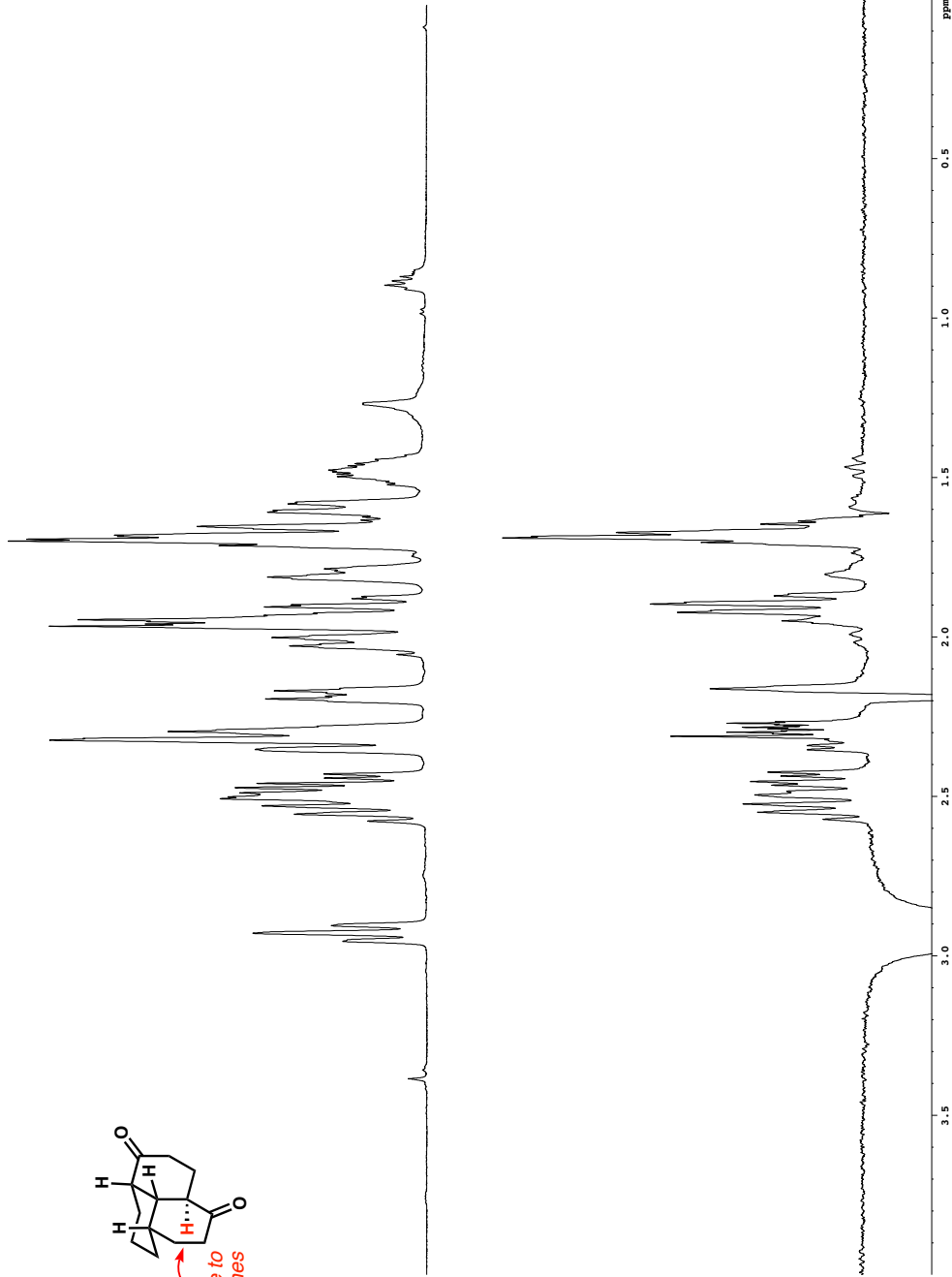
```

Current Data Parameters
=====
NAME      P03_060_010001_6
EXPNO     1
PROCNO    1

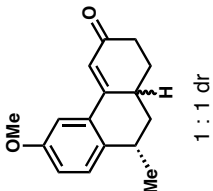
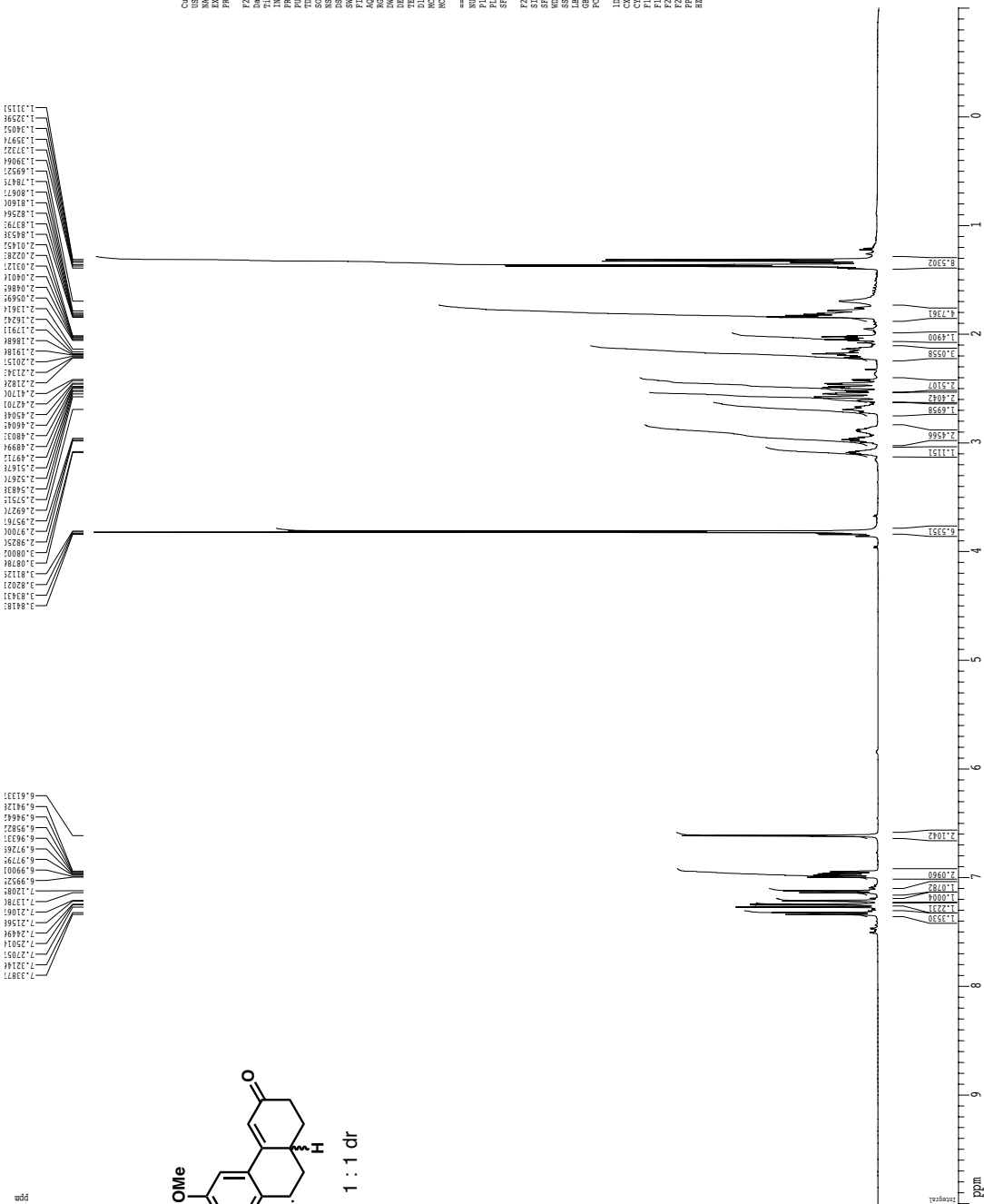
F2 - Acquisition Parameters
=====
Date_     2016-10-20
Time      11:46:20
INSTRUM   spect
PROBHD     5 mm CPYCC 1H-
PULPROG   zgpg30
TD         65536
SOLVENT    CDCl3
NS         2048
DS         4
SWH        611.242 MHz
FIDRES     0.122266 Hz
AQ         4.0889465 sec
RG         62.400 us/c
DM         1.6629820 K
TE         300.2
D8         0.50000000 sec
D9         0.00000000 sec
D16        0.00000000 sec
d22        0.16339699 sec
d2         0.16339699 sec
d222       15.000 us/c

===== CHANNEL f1 =====
NUC1       13C
P1         7.50 us/c
PC         30.00 us/c
PD         30.00 us/c
PF2        40000.00 us/c
SFO1       100.6261905 MHz
SFO2
SFO3
SFO4
SFO5
SFO6
SFO7
SFO8
SFO9
SFO10
SFO11
SFO12
SFO13
SFO14
SFO15
SFO16
SFO17
SFO18
SFO19
SFO20
SFO21
SFO22
SFO23
SFO24
SFO25
SFO26
SFO27
SFO28
SFO29
SFO30
SFO31
SFO32
SFO33
SFO34
SFO35
SFO36
SFO37
SFO38
SFO39
SFO40
SFO41
SFO42
SFO43
SFO44
SFO45
SFO46
SFO47
SFO48
SFO49
SFO50
SFO51
SFO52
SFO53
SFO54
SFO55
SFO56
SFO57
SFO58
SFO59
SFO60
SFO61
SFO62
SFO63
SFO64
SFO65
SFO66
SFO67
SFO68
SFO69
SFO70
SFO71
SFO72
SFO73
SFO74
SFO75
SFO76
SFO77
SFO78
SFO79
SFO80
SFO81
SFO82
SFO83
SFO84
SFO85
SFO86
SFO87
SFO88
SFO89
SFO90
SFO91
SFO92
SFO93
SFO94
SFO95
SFO96
SFO97
SFO98
SFO99
SFO100
===== GRADIENT CHANNEL =====
G1         0.00 us/c
G2         0.00 us/c
G3         0.00 us/c
G4         0.00 us/c
G5         0.00 us/c
G6         0.00 us/c
G7         0.00 us/c
G8         0.00 us/c
G9         0.00 us/c
G10        0.00 us/c
G11        0.00 us/c
G12        0.00 us/c
G13        0.00 us/c
G14        0.00 us/c
G15        0.00 us/c
G16        0.00 us/c
G17        0.00 us/c
G18        0.00 us/c
G19        0.00 us/c
G20        0.00 us/c
G21        0.00 us/c
G22        0.00 us/c
G23        0.00 us/c
G24        0.00 us/c
G25        0.00 us/c
G26        0.00 us/c
G27        0.00 us/c
G28        0.00 us/c
G29        0.00 us/c
G30        0.00 us/c
G31        0.00 us/c
G32        0.00 us/c
G33        0.00 us/c
G34        0.00 us/c
G35        0.00 us/c
G36        0.00 us/c
G37        0.00 us/c
G38        0.00 us/c
G39        0.00 us/c
G40        0.00 us/c
G41        0.00 us/c
G42        0.00 us/c
G43        0.00 us/c
G44        0.00 us/c
G45        0.00 us/c
G46        0.00 us/c
G47        0.00 us/c
G48        0.00 us/c
G49        0.00 us/c
G50        0.00 us/c
G51        0.00 us/c
G52        0.00 us/c
G53        0.00 us/c
G54        0.00 us/c
G55        0.00 us/c
G56        0.00 us/c
G57        0.00 us/c
G58        0.00 us/c
G59        0.00 us/c
G60        0.00 us/c
G61        0.00 us/c
G62        0.00 us/c
G63        0.00 us/c
G64        0.00 us/c
G65        0.00 us/c
G66        0.00 us/c
G67        0.00 us/c
G68        0.00 us/c
G69        0.00 us/c
G70        0.00 us/c
G71        0.00 us/c
G72        0.00 us/c
G73        0.00 us/c
G74        0.00 us/c
G75        0.00 us/c
G76        0.00 us/c
G77        0.00 us/c
G78        0.00 us/c
G79        0.00 us/c
G80        0.00 us/c
G81        0.00 us/c
G82        0.00 us/c
G83        0.00 us/c
G84        0.00 us/c
G85        0.00 us/c
G86        0.00 us/c
G87        0.00 us/c
G88        0.00 us/c
G89        0.00 us/c
G90        0.00 us/c
G91        0.00 us/c
G92        0.00 us/c
G93        0.00 us/c
G94        0.00 us/c
G95        0.00 us/c
G96        0.00 us/c
G97        0.00 us/c
G98        0.00 us/c
G99        0.00 us/c
G100       0.00 us/c

===== Processing parameters =====
SI         32768
SF         100.6261905 MHz
WDW        EM
SSB        0
LB         0.30 Hz
GB         0.00 Hz
PC         1.00 Hz
=====
  
```



1H spectrum



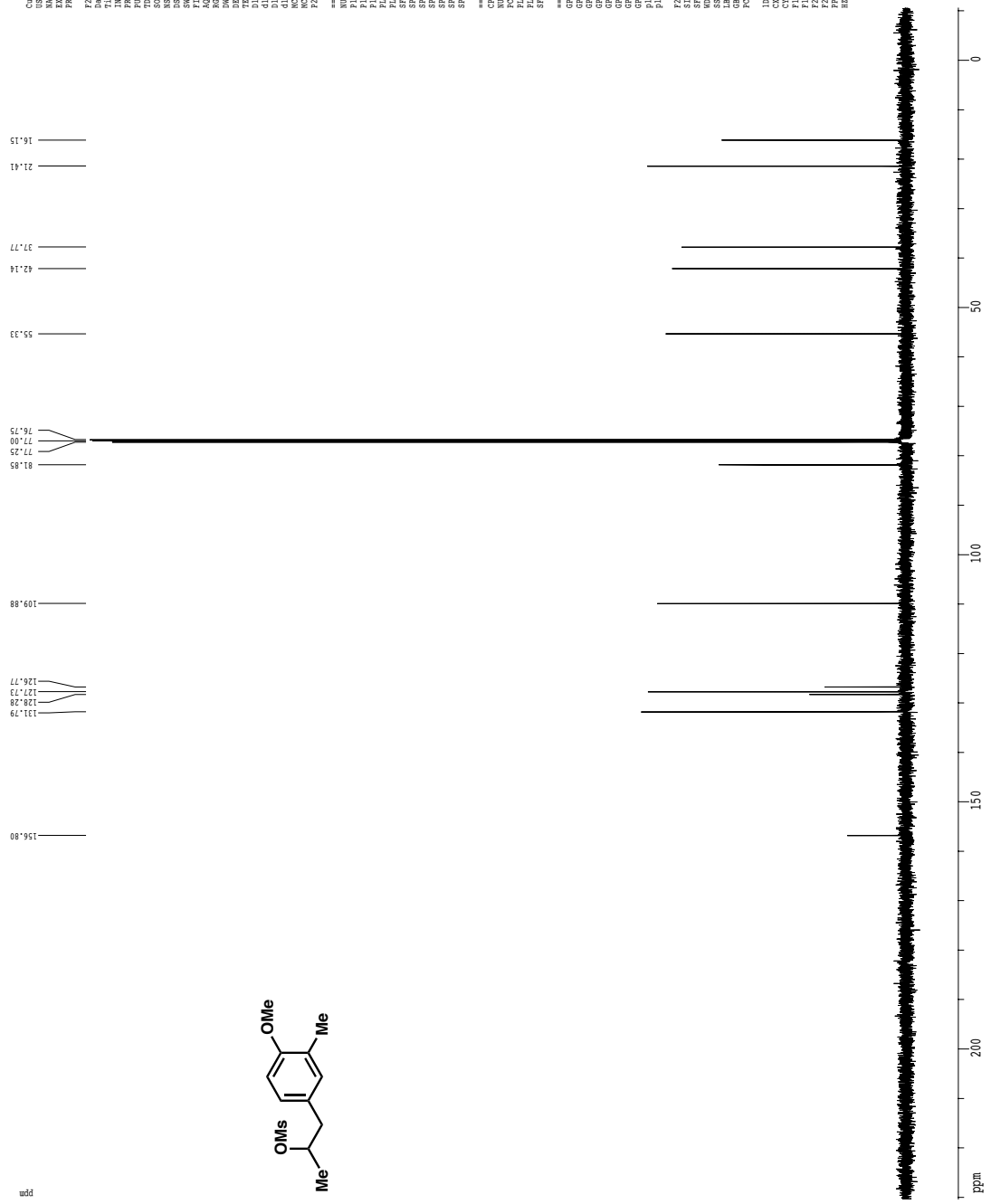
Current Data Parameters  
 NAME Pc3.113\_isolate  
 EXPNO 3  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2010/11  
 INSTRUM cty6500  
 PROBRD 5 mm CPXI H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 4  
 SWH 802.820 Hz  
 FIDRES 0.250024 Hz  
 AQ 1.1996614 sec  
 RG 317  
 EQ 1.7  
 DM 62.400 usec  
 DE 6.00 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 DCREST 0.0000000 sec  
 ACQNR 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.45 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 30.000000 MHz  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 ID MR plot parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 F1 5002.201 Hz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 GAMMA 13.703 Hz/cm  
 HZCN 241.33423 Hz/cm







Z-restored spin-echo 13C spectrum with 1H decoupling

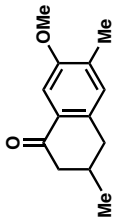
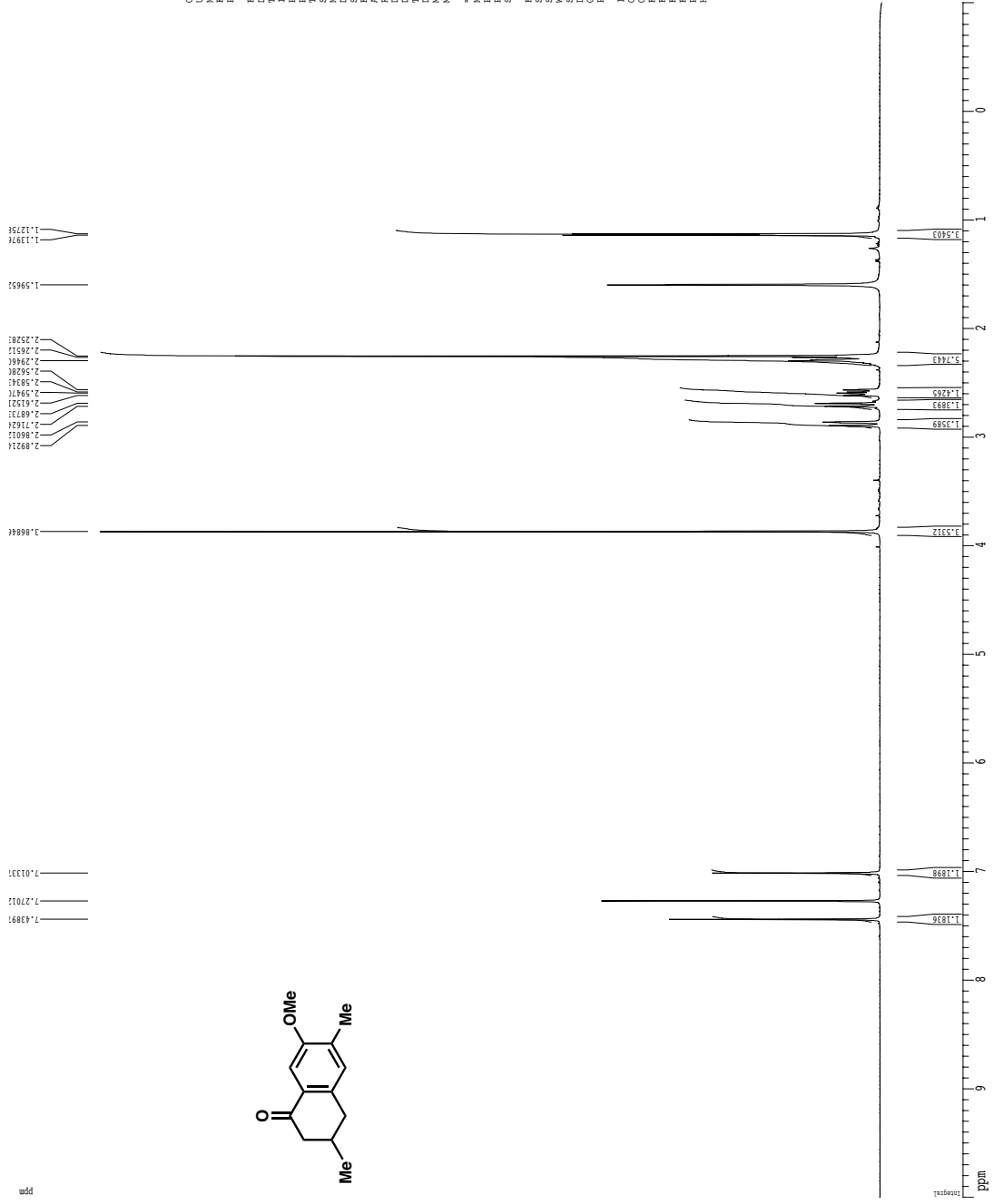


```

Current Data Parameters
NAME          pc3140_560Date
EXPNO         2
PROCNO        1
F2 - Acq. Parameters
Date_         20100704
Time          21.04
INSTRUM       cryo400
PROBHD        5mmCPY130
PULPROG       zgpg30
TD             65536
SOLVENT       CDCl3
NS            2
DS            4
SWH           30000.000 Hz
FIDRES        1.0813340 Hz
AQ            1.251
RG            3251
AQ            1.251
RG            3251
TE           300.2 K
NUC1          13C
NUC2          13C
PCPD2         100.00 uSAC
PCPD1         1.00 dB
SFO2          500.225011 MHz
SFO1          125.762200 MHz
SF2           3.20 dB
SF1           3.20 dB
SFO3          CPMAS, 6.5, 20.1
SFO4          CPMAS, 6.5, 20.1
SFO5          0.00 Hz
SFO6          0.00 Hz
SFO7          0.00 Hz
SFO8          0.00 Hz
===== CHANNEL F1 =====
NUC1          13C
P1            15.50 uSAC
PCPD2         100.00 uSAC
PCPD1         1.00 dB
SFO2          500.225011 MHz
===== CHANNEL F2 =====
NAME          vcp130
CPDPRG2       waltz16
PCPD2         100.00 uSAC
PCPD1         1.00 dB
SFO2          500.225011 MHz
===== CHANNEL CHANNEL =====
GRANU         32768
SINX          1.00
SINZ          1.00
GPRG1         0.00 A
GPRG2         0.00 A
GPRG3         0.00 A
GPRG4         0.00 A
GPRG5         0.00 A
GPRG6         0.00 A
GPRG7         0.00 A
GPRG8         0.00 A
GPRG9         0.00 A
GPRG10        0.00 A
GPRG11        0.00 A
GPRG12        0.00 A
GPRG13        0.00 A
GPRG14        0.00 A
GPRG15        0.00 A
GPRG16        0.00 A
GPRG17        0.00 A
GPRG18        0.00 A
GPRG19        0.00 A
GPRG20        0.00 A
GPRG21        0.00 A
GPRG22        0.00 A
GPRG23        0.00 A
GPRG24        0.00 A
GPRG25        0.00 A
GPRG26        0.00 A
GPRG27        0.00 A
GPRG28        0.00 A
GPRG29        0.00 A
GPRG30        0.00 A
GPRG31        0.00 A
GPRG32        0.00 A
GPRG33        0.00 A
GPRG34        0.00 A
GPRG35        0.00 A
GPRG36        0.00 A
GPRG37        0.00 A
GPRG38        0.00 A
GPRG39        0.00 A
GPRG40        0.00 A
GPRG41        0.00 A
GPRG42        0.00 A
GPRG43        0.00 A
GPRG44        0.00 A
GPRG45        0.00 A
GPRG46        0.00 A
GPRG47        0.00 A
GPRG48        0.00 A
GPRG49        0.00 A
GPRG50        0.00 A
GPRG51        0.00 A
GPRG52        0.00 A
GPRG53        0.00 A
GPRG54        0.00 A
GPRG55        0.00 A
GPRG56        0.00 A
GPRG57        0.00 A
GPRG58        0.00 A
GPRG59        0.00 A
GPRG60        0.00 A
GPRG61        0.00 A
GPRG62        0.00 A
GPRG63        0.00 A
GPRG64        0.00 A
GPRG65        0.00 A
GPRG66        0.00 A
GPRG67        0.00 A
GPRG68        0.00 A
GPRG69        0.00 A
GPRG70        0.00 A
GPRG71        0.00 A
GPRG72        0.00 A
GPRG73        0.00 A
GPRG74        0.00 A
GPRG75        0.00 A
GPRG76        0.00 A
GPRG77        0.00 A
GPRG78        0.00 A
GPRG79        0.00 A
GPRG80        0.00 A
GPRG81        0.00 A
GPRG82        0.00 A
GPRG83        0.00 A
GPRG84        0.00 A
GPRG85        0.00 A
GPRG86        0.00 A
GPRG87        0.00 A
GPRG88        0.00 A
GPRG89        0.00 A
GPRG90        0.00 A
GPRG91        0.00 A
GPRG92        0.00 A
GPRG93        0.00 A
GPRG94        0.00 A
GPRG95        0.00 A
GPRG96        0.00 A
GPRG97        0.00 A
GPRG98        0.00 A
GPRG99        0.00 A
GPRG100       0.00 A
===== Processing parameters =====
SI            32768
SF            125.7604866 MHz
WDW           EM
SSB           0
GB            0
PC            2.00
CT            15.45 cm
CT            15.45 cm
F1F2          230.84 ppm
F2F1          -10.536 ppm
F2           -125.29 Hz
F1           125.7604866 Hz
RG            1329.8832 Hz/cm
    
```

1H spectrum

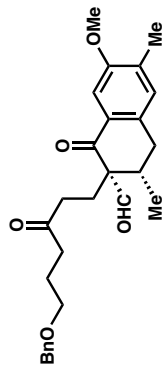
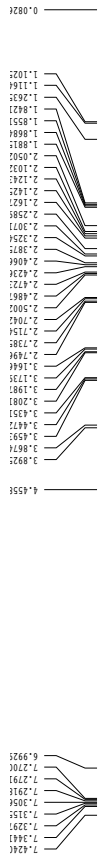
ppm



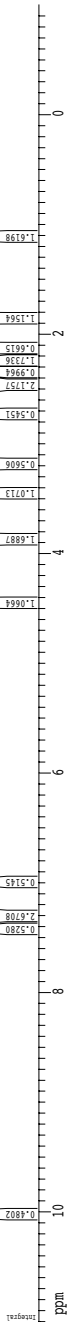
Current Data Parameters  
NAME Pcc3.158\_isolate  
EXPNO 1  
PROCNO 1  
F2 - Acquisition Parameters  
Date\_ 2010729  
Time 11:17:01  
INSTRUM cgc500  
PROBHD 5 mm CPXI 1H-  
PULPROG zgpg30  
TD 32768  
SOLVENT CDCl3  
NS 2  
DS 2  
SWH 8012.820 Hz  
FIDRES 0.250026 Hz  
AQ 1.199161 sec  
RG 5  
DM 62.400 usec  
DE 6.000 usec  
TE 298.2 K  
D1 0.1000000 sec  
ACQSF 0.0000000 sec  
SFO1 501.225015 MHz  
===== CHANNEL f1 =====  
NUC1 1H  
P1 7.45 usec  
PL1 0.00 dB  
PL2 1.60 dB  
SFO1 500.225015 MHz  
F2 - Processing parameters  
SI 6536  
SF 500.225015 MHz  
WDW EM  
SSB 0  
GB 0.00 Hz  
PC 4.00  
LO MR p1e1 parameters  
CX 22.80 cm  
CY 15.00 cm  
FZ 10.00 cm  
F1 500.225015 MHz  
F2 -1.000 ppm  
F3 -500.22 Hz  
HPCON 241.33423 Hz/cm



1H spectrum

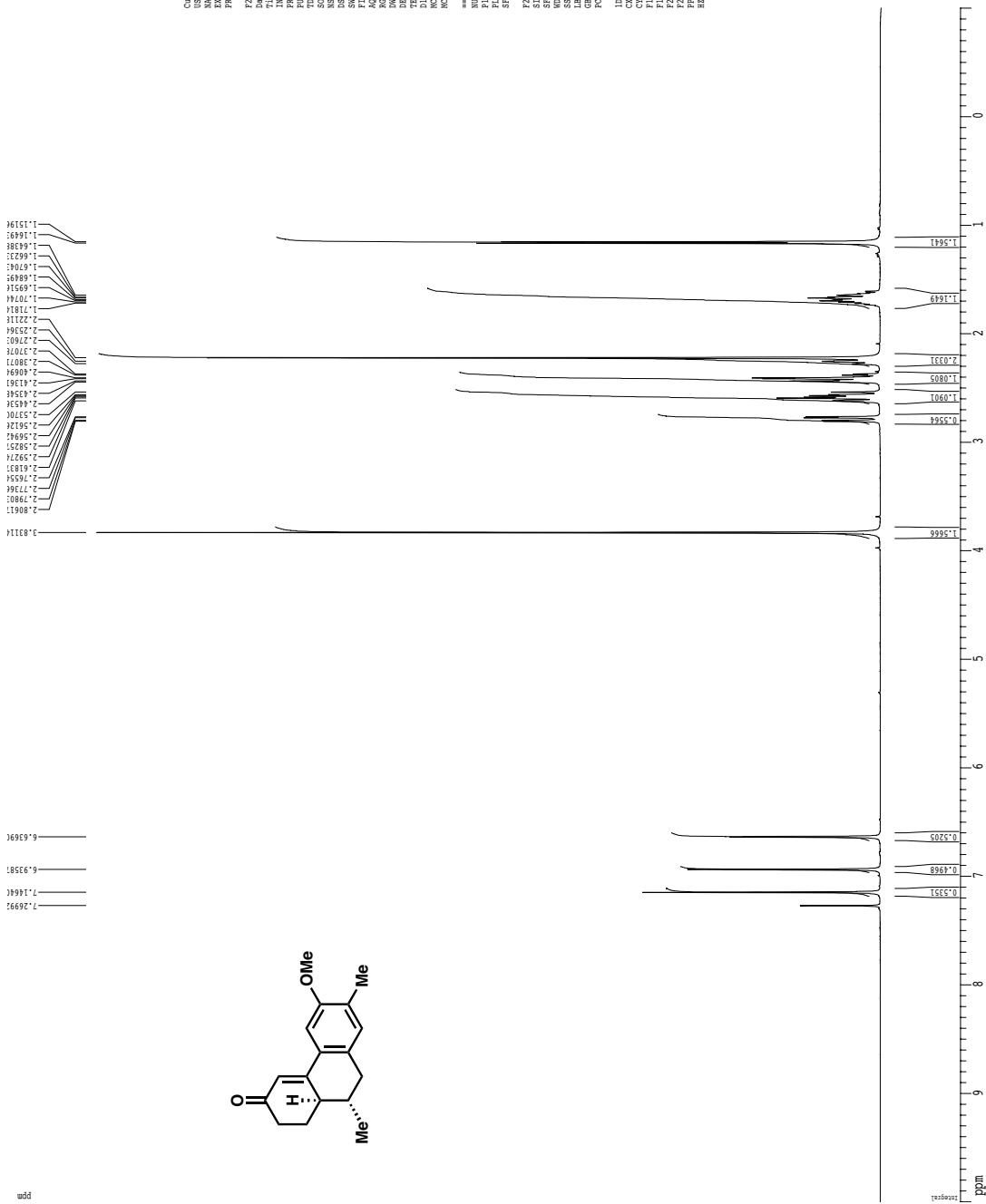


Current Data Parameters  
 NAME Pcc3.167\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2010093  
 Time 11:52:10  
 INSTRUM ctyes500  
 PROBHD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 4  
 SWH 8012.820 Hz  
 FIDRES 0.250026 Hz  
 AQ 1.199844 sec  
 RG 327.68  
 DQ 2.000000 sec  
 DE 2.000000 sec  
 TE 300.2 K  
 D1 0.1000000 sec  
 DCOffset 0.0000000 sec  
 ACQNR 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WWP 2.000000 sec  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 L0 MR parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 FX 500.225015 MHz  
 FY 500.225015 MHz  
 FZ -1.000 ppm  
 GY 200.122 Hz/cm  
 GZ 263.23371 Hz/cm





1H spectrum

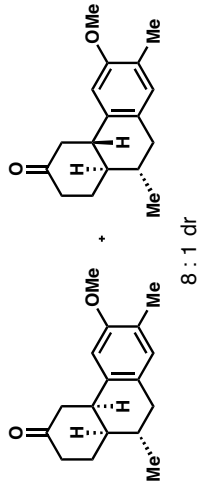
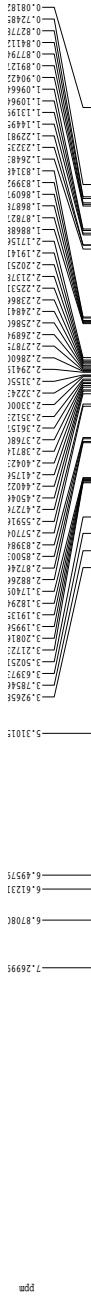


Current Data Parameters  
 NAME Pcc3.188\_crop1  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2010023  
 TIME 11:00:00  
 INSTRUM ctyes500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 0  
 DS 2  
 SFO1 500.225015 MHz  
 FIDRES 0.250026 Hz  
 RG 1.999944  
 ACQ 4.5  
 DM 62.400 usec  
 DE 6.00 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 ACQST 0.0000000 sec  
 ACQR 0.1000000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 0  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CT 15.00 cm  
 CR 1.00 cm  
 F1 500.201 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 HPCN 241.33423 Hz/cm



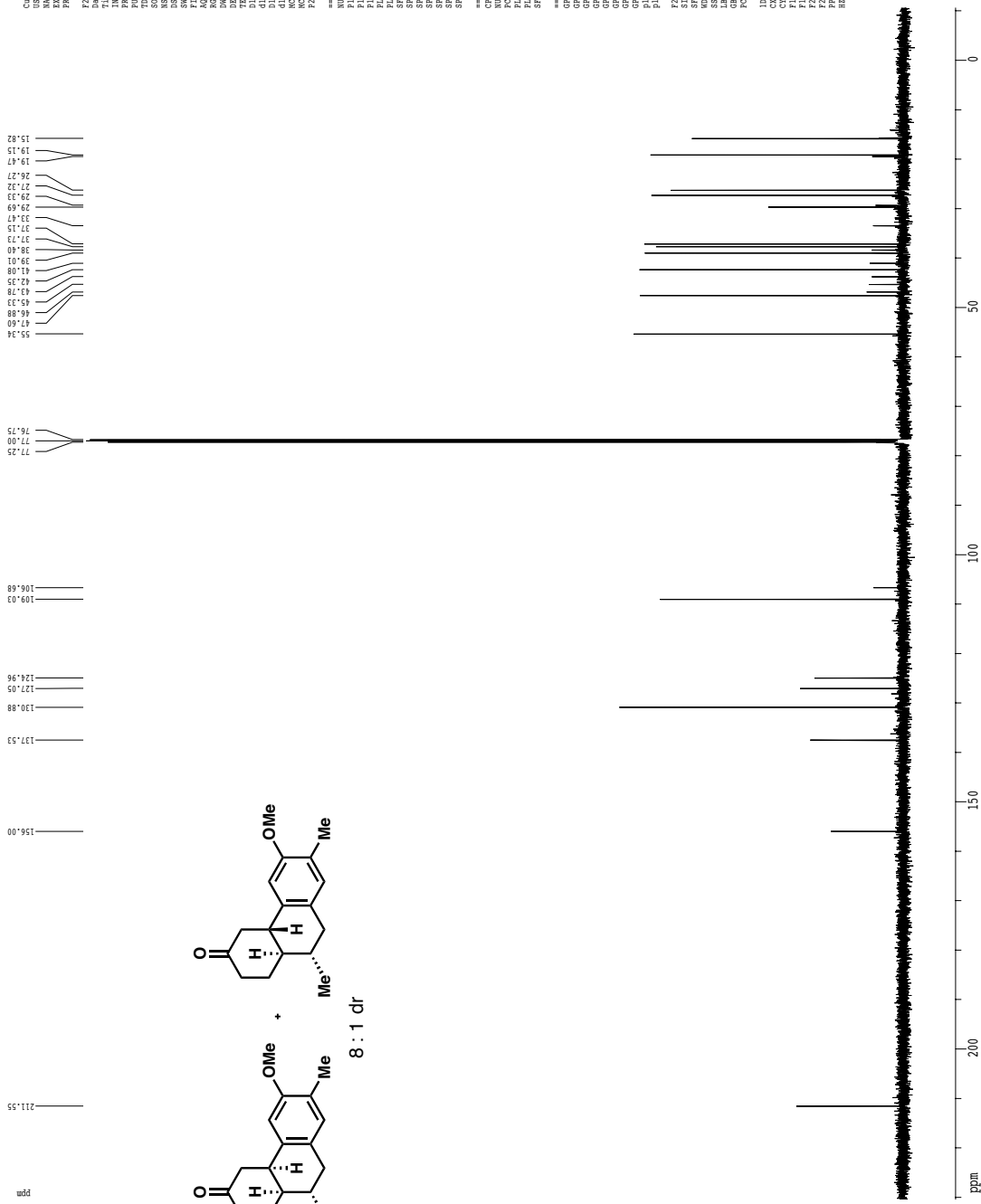


**<sup>1</sup>H spectrum**



Current Data Parameters  
 NAME Pcc3.103\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20100116  
 Time 10.58  
 INSTRUM cryo500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 4  
 SWH 8012.820 Hz  
 FIDRES 0.250026 Hz  
 AQ 1.1991616 sec  
 RG 5  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 DCOffset 0.0000000 sec  
 ACQBRD 0.1330000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.45 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WWP 8K  
 SSB 0  
 GB 0.0 Hz  
 PC 0.10  
 LO MR parameters  
 CX 22.80 cm  
 CZ 10.00 cm  
 CR 2.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 HPCON 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



Current Data Parameters  
 NAME per3185\_150Date  
 EXPNO 2  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2016.09.12 12.51  
 Time 20.18  
 INSTRUM cryo500  
 PULPROG zgpg30  
 PROCESOR sptproc3  
 F2 - Processing Parameters  
 DP 653.6  
 AS 2  
 CONVST COSY2  
 SFC 2  
 SFS 300.01538 Hz  
 SFR 1.500000 usec  
 AQ 1.0813440 sec  
 RG 798.2  
 EQ 5.000000 usec  
 DE 6.00 usec  
 TE 298.2 K  
 D1 0.500000 sec  
 d11 0.4300000 sec  
 d16 0.0020000 sec  
 d17 0.0000000 sec  
 d18 0.0000000 sec  
 d19 0.0000000 sec  
 d20 0.0000000 sec  
 ACQRES 0.0150000 sec  
 F2 31.00 usec  
 ===== CHANNEL F1 =====  
 F1 15.50 usec  
 F1 11  
 F2 300.00 usec  
 F2 12  
 F3 200.00 usec  
 F3 13  
 F4 200.00 dB  
 F4 14  
 F5 125.795220 Hz  
 F5 15  
 F6 3.20 dB  
 F6 16  
 SFO1 C600.632811 MHz  
 SFO2 C600.632811 MHz  
 SFOFF1 0.00 Hz  
 SFOFF2 0.00 Hz  
 ===== CHANNEL F2 =====  
 CDPF2 waltz16  
 PCPD2 100.00 usec  
 P2 3.00 dB  
 P2 13  
 SFO2 500.225411 MHz  
 ===== CHANNEL F3 =====  
 GRAB1 STW1.00  
 GRAB2 STW1.00  
 GRAB3 0.00 V  
 GRAB4 0.00 V  
 GRAB5 0.00 V  
 GRAB6 0.00 V  
 GRAB7 0.00 V  
 GRAB8 0.00 V  
 GRAB9 0.00 V  
 GRAB10 0.00 V  
 GRAB11 50.00 V  
 GRAB12 50.00 usec  
 GRAB13 1000.00 usec  
 ===== Processing Parameters =====  
 S1 653.6  
 SF 125.780486 MHz  
 AQ 1.0813440 sec  
 LB 1.00 Hz  
 GB 2.00  
 CT 15.465 cm  
 F1F 230.84 ppm  
 F2F -10.536 ppm  
 F2 1325.28 Hz  
 F2 1325.28 Hz/cm  
 F2CN 1325.08132 Hz/cm

Current Data Parameters  
 USER pcc1.185 isolate  
 NAME pcc1.185 isolate  
 PRNO 1

F2 - Acquisition Parameters  
 Date\_ 21.04.00  
 Time 13.40  
 INSTRM spect  
 PULPROG zgpg30  
 TD 65536  
 SOLVENT CDCl3  
 DS 2  
 SFO 300.130131 MHz  
 FIDRES 0.000180 Hz  
 AQ 1.013848 sec  
 RG 5160.4  
 DW 16.000 usec  
 DE 0.000000 usec  
 TE 298.2 K  
 CNUST2 145.000000  
 CNUST1 1.0000000 sec  
 d1 0.0034428 sec  
 d12 0.0002000 sec  
 DELTA 0.0003574 sec  
 DELTA1 0.0027878 sec  
 DELTA2 0.0000000 sec  
 DELTA3 0.0024428 sec  
 INVERT 0.0000000 sec  
 ACQPRG 0.0000000 sec

\*\*\*\*\* CHANNEL F1 \*\*\*\*\*  
 NUCL1 13C  
 P1 11.25 usec  
 PL1 0.00 dB  
 F1 120.00 MHz  
 SFO1 125.762548 MHz  
 SFO2 125.762548 MHz  
 SFO3 125.762548 MHz  
 SFO4 125.762548 MHz  
 SFO5 125.762548 MHz  
 SFO6 125.762548 MHz  
 SFO7 125.762548 MHz  
 SFO8 125.762548 MHz  
 SFO9 125.762548 MHz  
 SFO10 125.762548 MHz  
 SFO11 125.762548 MHz  
 SFO12 125.762548 MHz  
 SFO13 125.762548 MHz  
 SFO14 125.762548 MHz  
 SFO15 125.762548 MHz  
 SFO16 125.762548 MHz  
 SFO17 125.762548 MHz  
 SFO18 125.762548 MHz  
 SFO19 125.762548 MHz  
 SFO20 125.762548 MHz  
 SFO21 125.762548 MHz  
 SFO22 125.762548 MHz  
 SFO23 125.762548 MHz  
 SFO24 125.762548 MHz  
 SFO25 125.762548 MHz  
 SFO26 125.762548 MHz  
 SFO27 125.762548 MHz  
 SFO28 125.762548 MHz  
 SFO29 125.762548 MHz  
 SFO30 125.762548 MHz  
 SFO31 125.762548 MHz  
 SFO32 125.762548 MHz  
 SFO33 125.762548 MHz  
 SFO34 125.762548 MHz  
 SFO35 125.762548 MHz  
 SFO36 125.762548 MHz  
 SFO37 125.762548 MHz  
 SFO38 125.762548 MHz  
 SFO39 125.762548 MHz  
 SFO40 125.762548 MHz  
 SFO41 125.762548 MHz  
 SFO42 125.762548 MHz  
 SFO43 125.762548 MHz  
 SFO44 125.762548 MHz  
 SFO45 125.762548 MHz  
 SFO46 125.762548 MHz  
 SFO47 125.762548 MHz  
 SFO48 125.762548 MHz  
 SFO49 125.762548 MHz  
 SFO50 125.762548 MHz  
 SFO51 125.762548 MHz  
 SFO52 125.762548 MHz  
 SFO53 125.762548 MHz  
 SFO54 125.762548 MHz  
 SFO55 125.762548 MHz  
 SFO56 125.762548 MHz  
 SFO57 125.762548 MHz  
 SFO58 125.762548 MHz  
 SFO59 125.762548 MHz  
 SFO60 125.762548 MHz  
 SFO61 125.762548 MHz  
 SFO62 125.762548 MHz  
 SFO63 125.762548 MHz  
 SFO64 125.762548 MHz  
 SFO65 125.762548 MHz  
 SFO66 125.762548 MHz  
 SFO67 125.762548 MHz  
 SFO68 125.762548 MHz  
 SFO69 125.762548 MHz  
 SFO70 125.762548 MHz  
 SFO71 125.762548 MHz  
 SFO72 125.762548 MHz  
 SFO73 125.762548 MHz  
 SFO74 125.762548 MHz  
 SFO75 125.762548 MHz  
 SFO76 125.762548 MHz  
 SFO77 125.762548 MHz  
 SFO78 125.762548 MHz  
 SFO79 125.762548 MHz  
 SFO80 125.762548 MHz  
 SFO81 125.762548 MHz  
 SFO82 125.762548 MHz  
 SFO83 125.762548 MHz  
 SFO84 125.762548 MHz  
 SFO85 125.762548 MHz  
 SFO86 125.762548 MHz  
 SFO87 125.762548 MHz  
 SFO88 125.762548 MHz  
 SFO89 125.762548 MHz  
 SFO90 125.762548 MHz  
 SFO91 125.762548 MHz  
 SFO92 125.762548 MHz  
 SFO93 125.762548 MHz  
 SFO94 125.762548 MHz  
 SFO95 125.762548 MHz  
 SFO96 125.762548 MHz  
 SFO97 125.762548 MHz  
 SFO98 125.762548 MHz  
 SFO99 125.762548 MHz  
 SFO100 125.762548 MHz

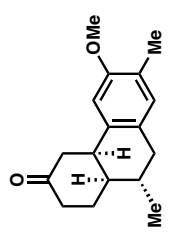
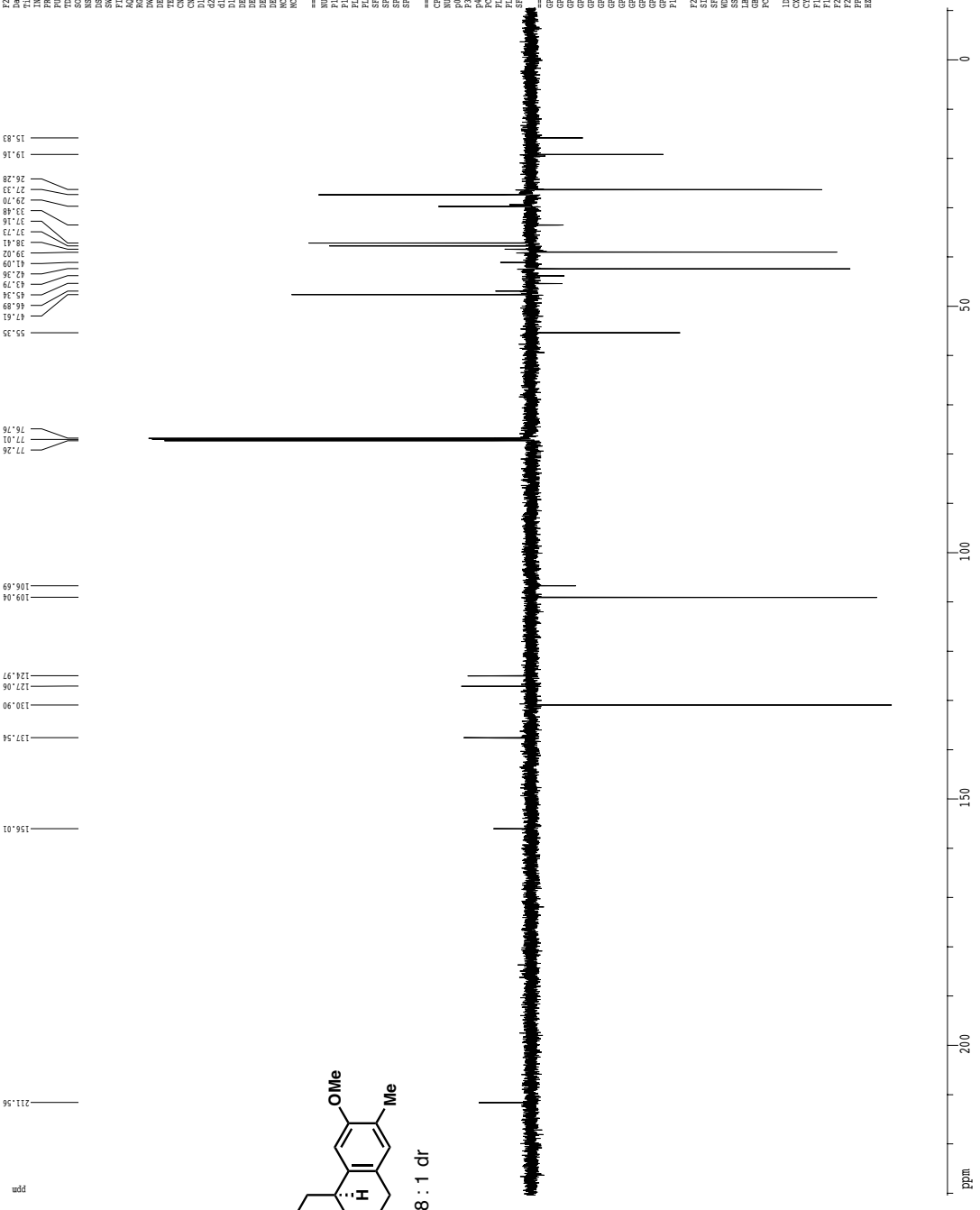
\*\*\*\*\* CHANNEL F2 \*\*\*\*\*  
 NUCL2 1H  
 P2 11.25 usec  
 PL2 0.00 dB  
 F2 400.146300 MHz  
 SFO1 400.146300 MHz  
 SFO2 400.146300 MHz  
 SFO3 400.146300 MHz  
 SFO4 400.146300 MHz  
 SFO5 400.146300 MHz  
 SFO6 400.146300 MHz  
 SFO7 400.146300 MHz  
 SFO8 400.146300 MHz  
 SFO9 400.146300 MHz  
 SFO10 400.146300 MHz  
 SFO11 400.146300 MHz  
 SFO12 400.146300 MHz  
 SFO13 400.146300 MHz  
 SFO14 400.146300 MHz  
 SFO15 400.146300 MHz  
 SFO16 400.146300 MHz  
 SFO17 400.146300 MHz  
 SFO18 400.146300 MHz  
 SFO19 400.146300 MHz  
 SFO20 400.146300 MHz  
 SFO21 400.146300 MHz  
 SFO22 400.146300 MHz  
 SFO23 400.146300 MHz  
 SFO24 400.146300 MHz  
 SFO25 400.146300 MHz  
 SFO26 400.146300 MHz  
 SFO27 400.146300 MHz  
 SFO28 400.146300 MHz  
 SFO29 400.146300 MHz  
 SFO30 400.146300 MHz  
 SFO31 400.146300 MHz  
 SFO32 400.146300 MHz  
 SFO33 400.146300 MHz  
 SFO34 400.146300 MHz  
 SFO35 400.146300 MHz  
 SFO36 400.146300 MHz  
 SFO37 400.146300 MHz  
 SFO38 400.146300 MHz  
 SFO39 400.146300 MHz  
 SFO40 400.146300 MHz  
 SFO41 400.146300 MHz  
 SFO42 400.146300 MHz  
 SFO43 400.146300 MHz  
 SFO44 400.146300 MHz  
 SFO45 400.146300 MHz  
 SFO46 400.146300 MHz  
 SFO47 400.146300 MHz  
 SFO48 400.146300 MHz  
 SFO49 400.146300 MHz  
 SFO50 400.146300 MHz  
 SFO51 400.146300 MHz  
 SFO52 400.146300 MHz  
 SFO53 400.146300 MHz  
 SFO54 400.146300 MHz  
 SFO55 400.146300 MHz  
 SFO56 400.146300 MHz  
 SFO57 400.146300 MHz  
 SFO58 400.146300 MHz  
 SFO59 400.146300 MHz  
 SFO60 400.146300 MHz  
 SFO61 400.146300 MHz  
 SFO62 400.146300 MHz  
 SFO63 400.146300 MHz  
 SFO64 400.146300 MHz  
 SFO65 400.146300 MHz  
 SFO66 400.146300 MHz  
 SFO67 400.146300 MHz  
 SFO68 400.146300 MHz  
 SFO69 400.146300 MHz  
 SFO70 400.146300 MHz  
 SFO71 400.146300 MHz  
 SFO72 400.146300 MHz  
 SFO73 400.146300 MHz  
 SFO74 400.146300 MHz  
 SFO75 400.146300 MHz  
 SFO76 400.146300 MHz  
 SFO77 400.146300 MHz  
 SFO78 400.146300 MHz  
 SFO79 400.146300 MHz  
 SFO80 400.146300 MHz  
 SFO81 400.146300 MHz  
 SFO82 400.146300 MHz  
 SFO83 400.146300 MHz  
 SFO84 400.146300 MHz  
 SFO85 400.146300 MHz  
 SFO86 400.146300 MHz  
 SFO87 400.146300 MHz  
 SFO88 400.146300 MHz  
 SFO89 400.146300 MHz  
 SFO90 400.146300 MHz  
 SFO91 400.146300 MHz  
 SFO92 400.146300 MHz  
 SFO93 400.146300 MHz  
 SFO94 400.146300 MHz  
 SFO95 400.146300 MHz  
 SFO96 400.146300 MHz  
 SFO97 400.146300 MHz  
 SFO98 400.146300 MHz  
 SFO99 400.146300 MHz  
 SFO100 400.146300 MHz

\*\*\*\*\* GRADIENT CHANNEL \*\*\*\*\*  
 GPCP01 0.00 usec  
 GPCP02 0.00 usec  
 GPCP03 0.00 usec  
 GPCP04 0.00 usec  
 GPCP05 0.00 usec  
 GPCP06 0.00 usec  
 GPCP07 0.00 usec  
 GPCP08 0.00 usec  
 GPCP09 0.00 usec  
 GPCP10 0.00 usec  
 GPCP11 0.00 usec  
 GPCP12 0.00 usec  
 GPCP13 0.00 usec  
 GPCP14 0.00 usec  
 GPCP15 0.00 usec  
 GPCP16 0.00 usec  
 GPCP17 0.00 usec  
 GPCP18 0.00 usec  
 GPCP19 0.00 usec  
 GPCP20 0.00 usec  
 GPCP21 0.00 usec  
 GPCP22 0.00 usec  
 GPCP23 0.00 usec  
 GPCP24 0.00 usec  
 GPCP25 0.00 usec  
 GPCP26 0.00 usec  
 GPCP27 0.00 usec  
 GPCP28 0.00 usec  
 GPCP29 0.00 usec  
 GPCP30 0.00 usec  
 GPCP31 0.00 usec  
 GPCP32 0.00 usec  
 GPCP33 0.00 usec  
 GPCP34 0.00 usec  
 GPCP35 0.00 usec  
 GPCP36 0.00 usec  
 GPCP37 0.00 usec  
 GPCP38 0.00 usec  
 GPCP39 0.00 usec  
 GPCP40 0.00 usec  
 GPCP41 0.00 usec  
 GPCP42 0.00 usec  
 GPCP43 0.00 usec  
 GPCP44 0.00 usec  
 GPCP45 0.00 usec  
 GPCP46 0.00 usec  
 GPCP47 0.00 usec  
 GPCP48 0.00 usec  
 GPCP49 0.00 usec  
 GPCP50 0.00 usec  
 GPCP51 0.00 usec  
 GPCP52 0.00 usec  
 GPCP53 0.00 usec  
 GPCP54 0.00 usec  
 GPCP55 0.00 usec  
 GPCP56 0.00 usec  
 GPCP57 0.00 usec  
 GPCP58 0.00 usec  
 GPCP59 0.00 usec  
 GPCP60 0.00 usec  
 GPCP61 0.00 usec  
 GPCP62 0.00 usec  
 GPCP63 0.00 usec  
 GPCP64 0.00 usec  
 GPCP65 0.00 usec  
 GPCP66 0.00 usec  
 GPCP67 0.00 usec  
 GPCP68 0.00 usec  
 GPCP69 0.00 usec  
 GPCP70 0.00 usec  
 GPCP71 0.00 usec  
 GPCP72 0.00 usec  
 GPCP73 0.00 usec  
 GPCP74 0.00 usec  
 GPCP75 0.00 usec  
 GPCP76 0.00 usec  
 GPCP77 0.00 usec  
 GPCP78 0.00 usec  
 GPCP79 0.00 usec  
 GPCP80 0.00 usec  
 GPCP81 0.00 usec  
 GPCP82 0.00 usec  
 GPCP83 0.00 usec  
 GPCP84 0.00 usec  
 GPCP85 0.00 usec  
 GPCP86 0.00 usec  
 GPCP87 0.00 usec  
 GPCP88 0.00 usec  
 GPCP89 0.00 usec  
 GPCP90 0.00 usec  
 GPCP91 0.00 usec  
 GPCP92 0.00 usec  
 GPCP93 0.00 usec  
 GPCP94 0.00 usec  
 GPCP95 0.00 usec  
 GPCP96 0.00 usec  
 GPCP97 0.00 usec  
 GPCP98 0.00 usec  
 GPCP99 0.00 usec  
 GPCP100 0.00 usec

F2 - Processing Parameters  
 SI 65536  
 SF 125.7604272 MHz  
 DS 2  
 SS 0  
 LB 1.00 Hz  
 GB 0  
 PC 2.00

1D NMR plot parameters  
 SI 65536  
 SF 125.7604272 MHz  
 DS 2  
 SS 0  
 LB 1.00 Hz  
 GB 0  
 PC 2.00

deftq 13C spectrum with 1H decoupling (CH & CH3 one way up; C, CH2 and solvent the other)



ghmcqc

Current Data Parameters  
USER ghmcqc  
NAME pr11b\_isolate  
PROCNO 1  
PRGNO 1

Date F1 - Acquisition Parameters  
Time 13.43  
INSTRUM cpcy500  
PROBHD 5 mm QNP5/1H  
PULPROG zgpg30  
TD 32768  
SFO 500.136250  
AQ 0.2556404 sec  
RG 327.680  
DE 6.50 usec  
TE 300.2 K

===== CHANNEL F1 =====  
NUC1 1H  
P1 15.50 usec  
PL1 0.00 dB  
SFO1 500.136250 MHz

===== CHANNEL F2 =====  
CPDPRG2 zgpg30  
NUC2 13C  
P2 15.50 usec  
PL2 0.00 dB  
SFO2 125.762500 MHz

===== GRABBER CHANNEL =====  
GPRG1 zgpg30  
GPRG2 zgpg30  
GPRG3 zgpg30  
GPRG4 zgpg30  
GPRG5 zgpg30  
GPRG6 zgpg30  
GPRG7 zgpg30  
GPRG8 zgpg30  
GPRG9 zgpg30  
GPRG10 zgpg30  
GPRG11 zgpg30  
GPRG12 zgpg30  
GPRG13 zgpg30  
GPRG14 zgpg30  
GPRG15 zgpg30  
GPRG16 zgpg30  
GPRG17 zgpg30  
GPRG18 zgpg30  
GPRG19 zgpg30  
GPRG20 zgpg30  
GPRG21 zgpg30  
GPRG22 zgpg30  
GPRG23 zgpg30  
GPRG24 zgpg30  
GPRG25 zgpg30  
GPRG26 zgpg30  
GPRG27 zgpg30  
GPRG28 zgpg30  
GPRG29 zgpg30  
GPRG30 zgpg30  
GPRG31 zgpg30  
GPRG32 zgpg30  
GPRG33 zgpg30  
GPRG34 zgpg30  
GPRG35 zgpg30  
GPRG36 zgpg30  
GPRG37 zgpg30  
GPRG38 zgpg30  
GPRG39 zgpg30  
GPRG40 zgpg30  
GPRG41 zgpg30  
GPRG42 zgpg30  
GPRG43 zgpg30  
GPRG44 zgpg30  
GPRG45 zgpg30  
GPRG46 zgpg30  
GPRG47 zgpg30  
GPRG48 zgpg30  
GPRG49 zgpg30  
GPRG50 zgpg30  
GPRG51 zgpg30  
GPRG52 zgpg30  
GPRG53 zgpg30  
GPRG54 zgpg30  
GPRG55 zgpg30  
GPRG56 zgpg30  
GPRG57 zgpg30  
GPRG58 zgpg30  
GPRG59 zgpg30  
GPRG60 zgpg30  
GPRG61 zgpg30  
GPRG62 zgpg30  
GPRG63 zgpg30  
GPRG64 zgpg30  
GPRG65 zgpg30  
GPRG66 zgpg30  
GPRG67 zgpg30  
GPRG68 zgpg30  
GPRG69 zgpg30  
GPRG70 zgpg30  
GPRG71 zgpg30  
GPRG72 zgpg30  
GPRG73 zgpg30  
GPRG74 zgpg30  
GPRG75 zgpg30  
GPRG76 zgpg30  
GPRG77 zgpg30  
GPRG78 zgpg30  
GPRG79 zgpg30  
GPRG80 zgpg30  
GPRG81 zgpg30  
GPRG82 zgpg30  
GPRG83 zgpg30  
GPRG84 zgpg30  
GPRG85 zgpg30  
GPRG86 zgpg30  
GPRG87 zgpg30  
GPRG88 zgpg30  
GPRG89 zgpg30  
GPRG90 zgpg30  
GPRG91 zgpg30  
GPRG92 zgpg30  
GPRG93 zgpg30  
GPRG94 zgpg30  
GPRG95 zgpg30  
GPRG96 zgpg30  
GPRG97 zgpg30  
GPRG98 zgpg30  
GPRG99 zgpg30  
GPRG100 zgpg30

F1 - Acquisition parameters  
NUC1 1H  
P1 15.50 usec  
PL1 0.00 dB  
SFO1 500.136250 MHz

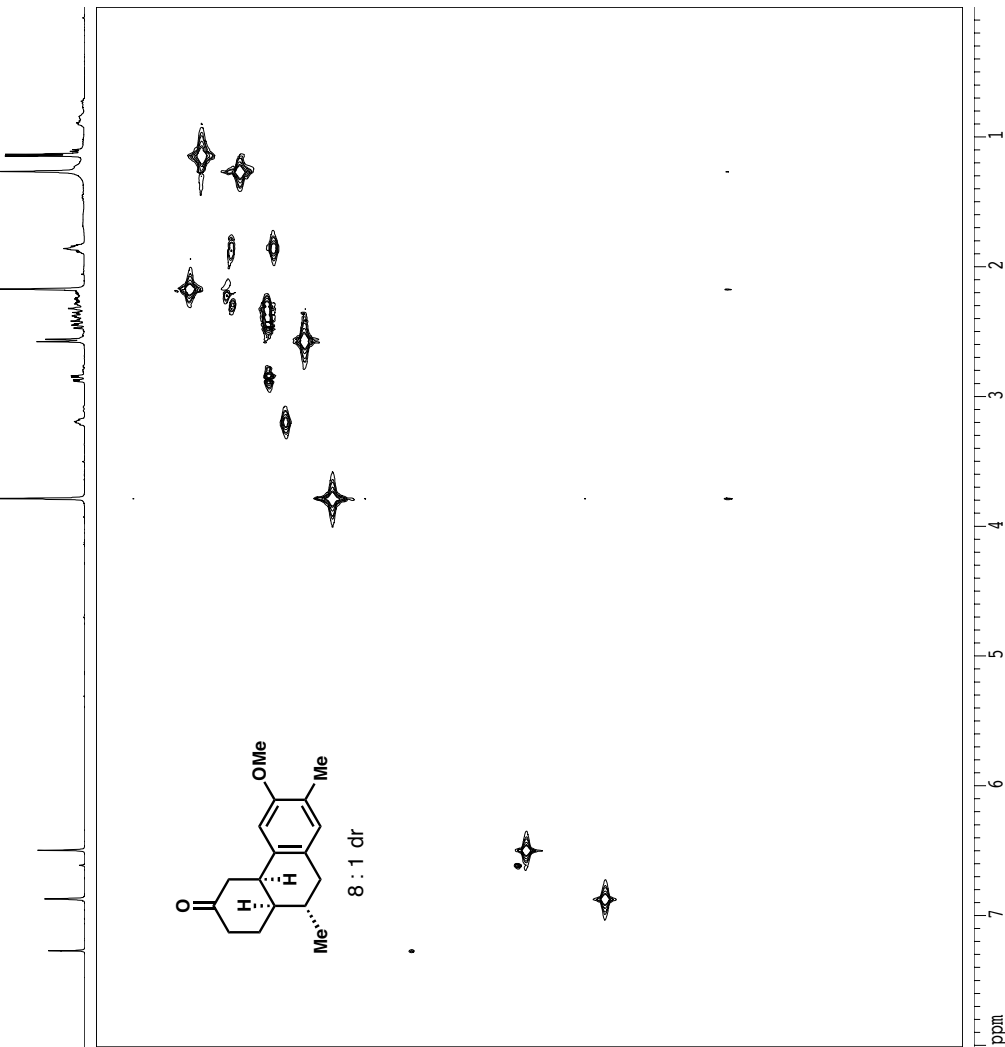
F2 - Acquisition parameters  
NUC2 13C  
P2 15.50 usec  
PL2 0.00 dB  
SFO2 125.762500 MHz

F1 - Processing parameters  
SI 32768  
SF 500.136250 MHz  
WDW EM  
SSB 0  
GB 0  
PC 4.00

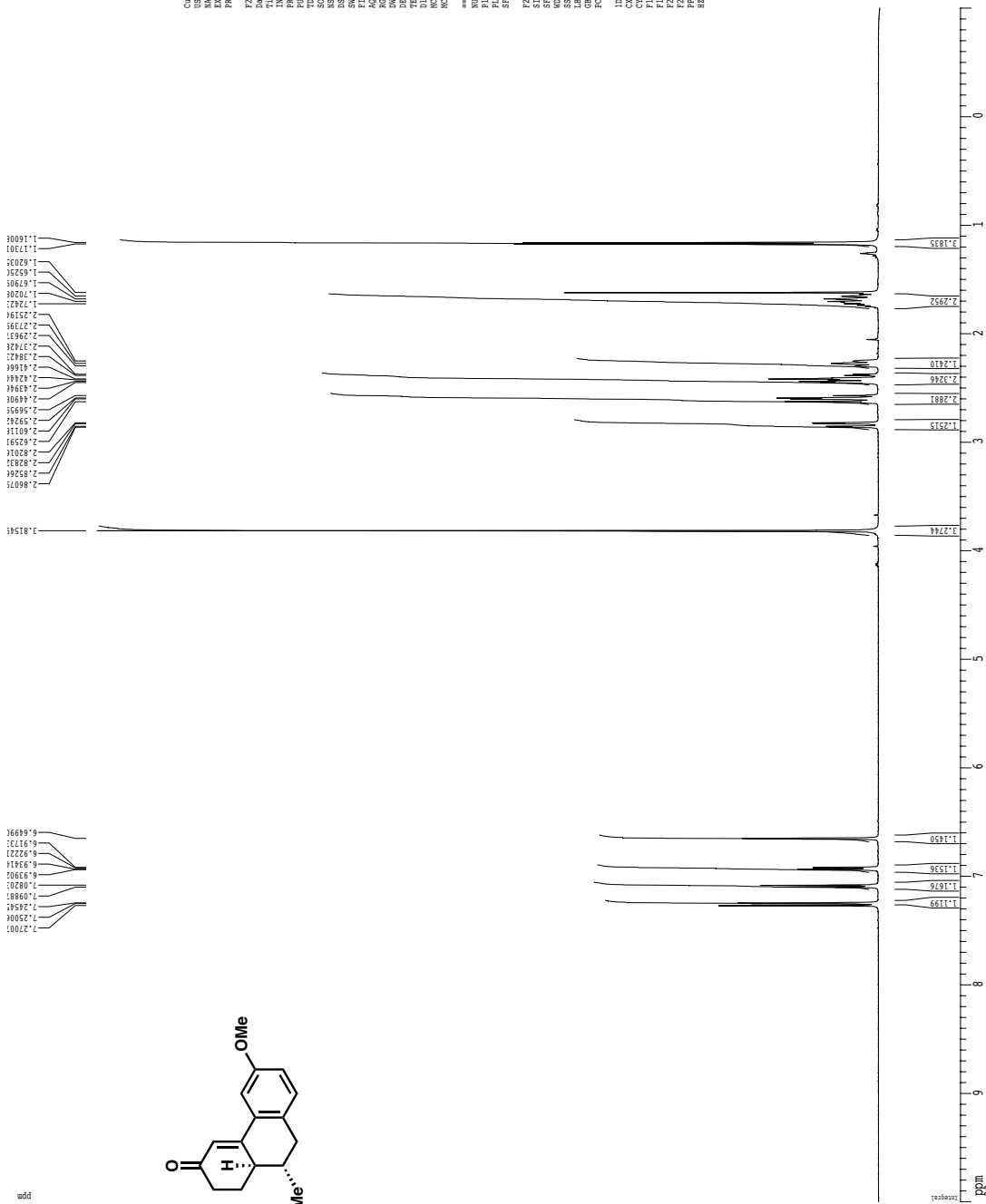
F2 - Processing parameters  
SI 32768  
SF 125.762500 MHz  
WDW EM  
SSB 0  
GB 0  
PC 4.00

2D NMR plot parameters  
SI 32768  
SF 500.136250 MHz  
WDW EM  
SSB 0  
GB 0  
PC 4.00

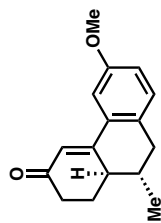
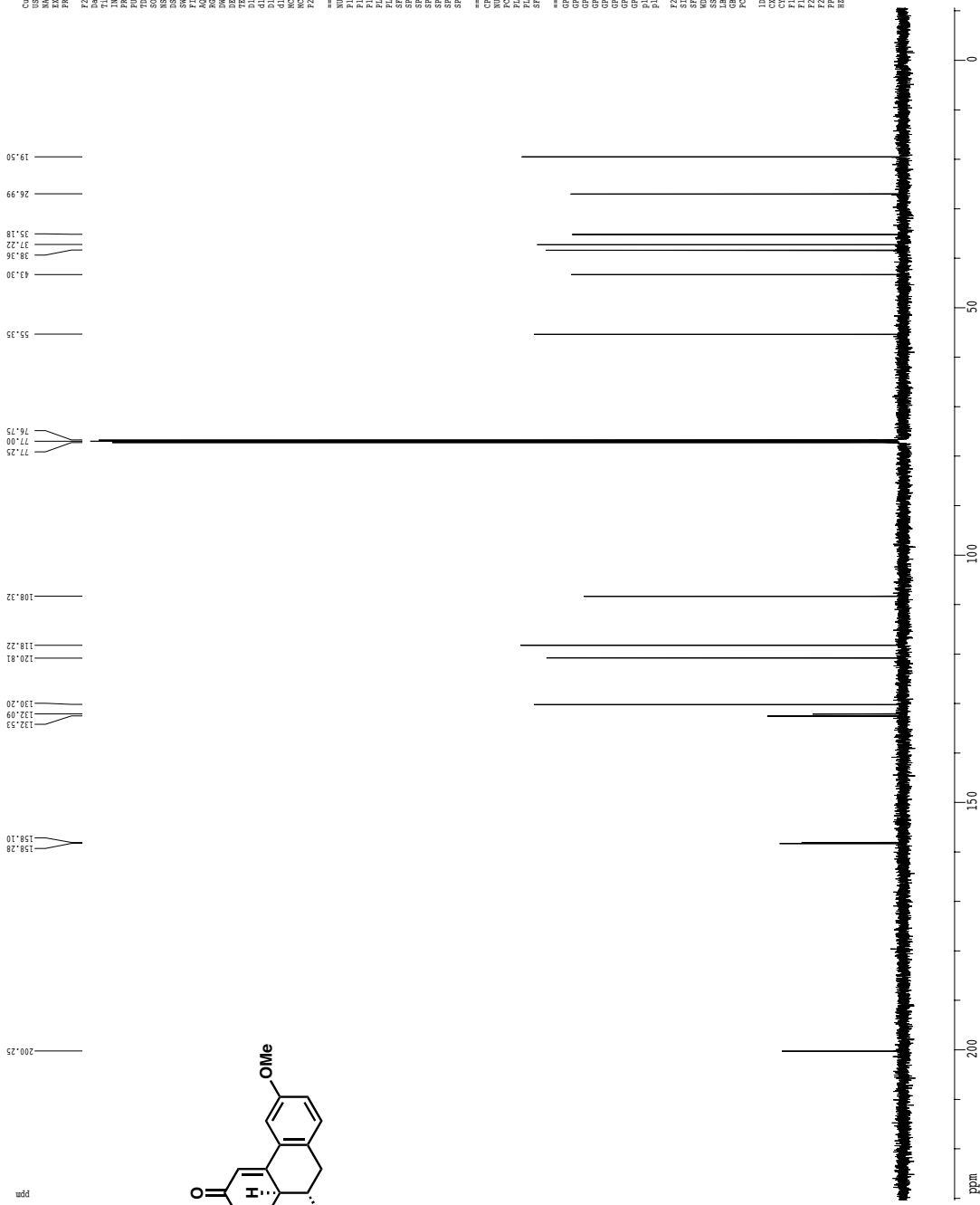
2D NMR plot parameters  
SI 32768  
SF 125.762500 MHz  
WDW EM  
SSB 0  
GB 0  
PC 4.00



1H spectrum



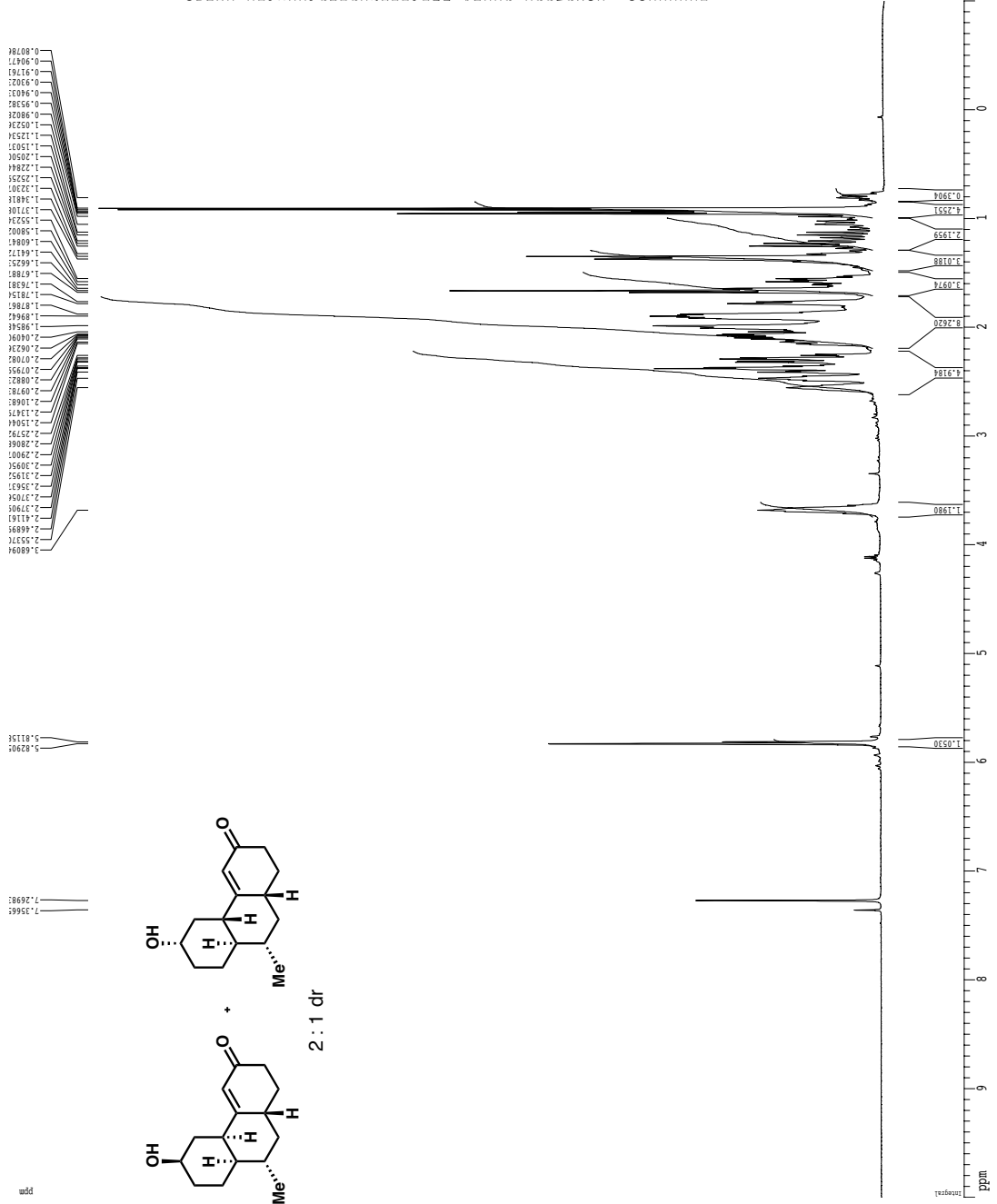
Z-restored spin-echo 13C spectrum with 1H decoupling



```

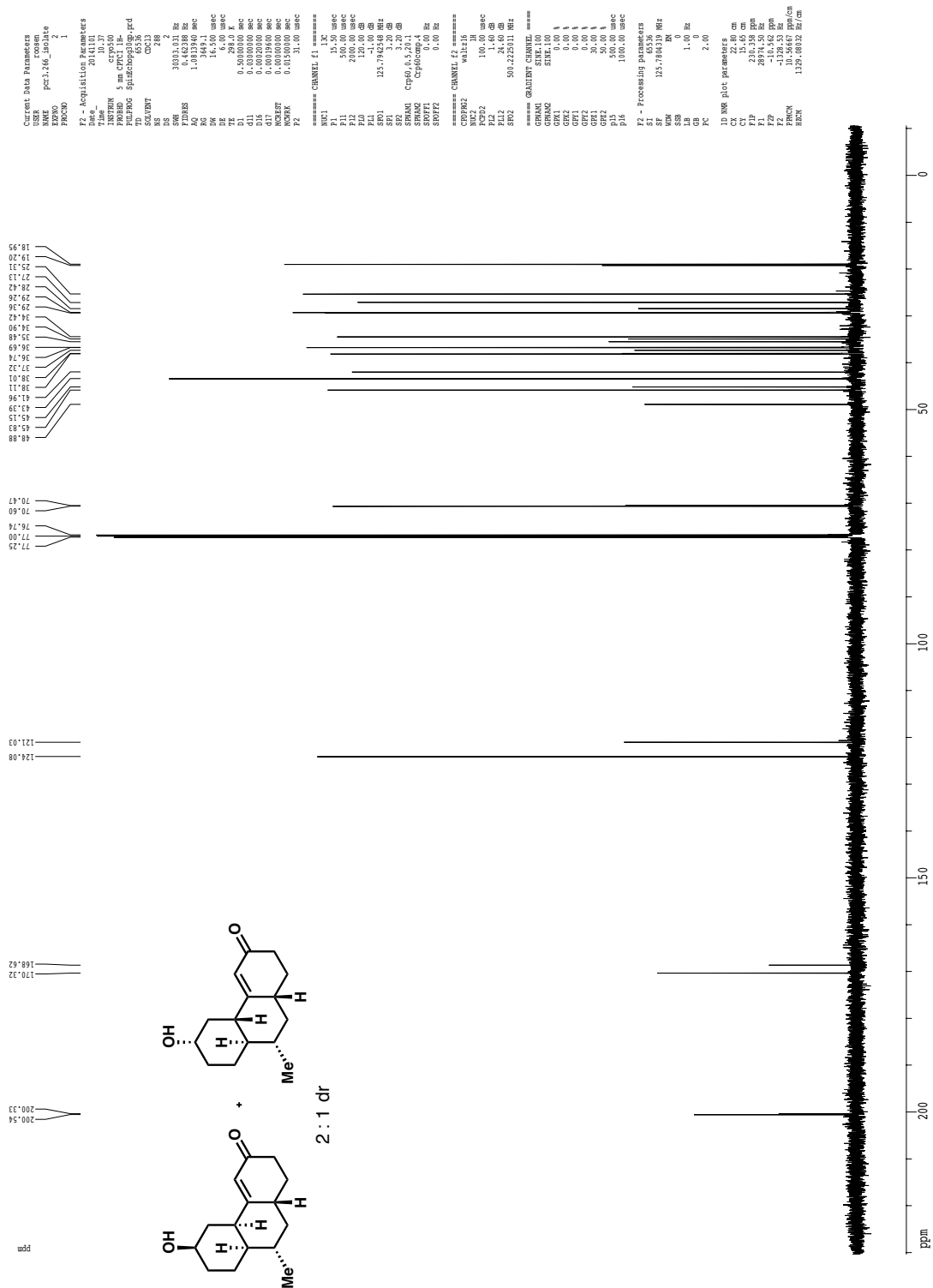
Current Data Parameters
=====
NAME      pcq3749_150Date
EXPNO     2
PROCNO    1
F2 - Acq. Parameters
=====
Date_     20111202
Time      12.10
INSTRUM   crys400
PROBHD    zgpg30
PULPROG   zgpg30
TD        65536
SOLVENT   CDCl3
NS        4096
DS        2
SWH       30313.938 Hz
FIDRES    1.4813340 Hz
AQ         2296.8
RG         656.00 usec
DE         6.500 usec
TE         0.2983 K
NUC1       13C
NUC2       13C
NUC3
NUC4
NUC5
NUC6
NUC7
NUC8
NUC9
NUC10
NUC11
NUC12
NUC13
NUC14
NUC15
NUC16
NUC17
NUC18
NUC19
NUC20
===== CHANNEL F1 =====
PC11      125.762200 MHz
PC12      15.50 usec
PC13      300.00 usec
PC14      200.00 dB
PC15      200.00 dB
PC16      1.00 dB
PC17      125.762200 MHz
PC18      15.50 usec
PC19      3.20 dB
PC20      CPU0:0.5281
PC21      CPU1:0.3536
PC22      CPU2:0.3536
PC23      CPU3:0.3536
PC24      CPU4:0.3536
PC25      CPU5:0.3536
PC26      CPU6:0.3536
PC27      CPU7:0.3536
PC28      CPU8:0.3536
PC29      CPU9:0.3536
PC30      CPU10:0.3536
===== CHANNEL F2 =====
PCF11      125.762200 MHz
PCF12      15.50 usec
PCF13      300.00 usec
PCF14      200.00 dB
PCF15      200.00 dB
PCF16      1.00 dB
PCF17      125.762200 MHz
PCF18      15.50 usec
PCF19      3.20 dB
PCF20      CPU0:0.5281
PCF21      CPU1:0.3536
PCF22      CPU2:0.3536
PCF23      CPU3:0.3536
PCF24      CPU4:0.3536
PCF25      CPU5:0.3536
PCF26      CPU6:0.3536
PCF27      CPU7:0.3536
PCF28      CPU8:0.3536
PCF29      CPU9:0.3536
PCF30      CPU10:0.3536
===== CHANNEL M1 =====
GRANU1    SINE,100
SINW1     0.00 A
SINW2     0.00 A
SINW3     0.00 A
SINW4     0.00 A
SINW5     0.00 A
SINW6     0.00 A
SINW7     0.00 A
SINW8     0.00 A
SINW9     0.00 A
SINW10    0.00 A
===== CHANNEL M2 =====
GRANU2    SINE,100
SINW21    0.00 A
SINW22    0.00 A
SINW23    0.00 A
SINW24    0.00 A
SINW25    0.00 A
SINW26    0.00 A
SINW27    0.00 A
SINW28    0.00 A
SINW29    0.00 A
SINW30    0.00 A
===== Processing parameters =====
SI        65536
SF        125.7604391 MHz
WDW       EM
SSB       0
LB        0
GB        0
PC        1.00 Hz
DE        2.00
===== ID MS parameters =====
CT        15.65 cm
PIF       230.380 ppm
FZP       -10.540 ppm
F2        -1225.74 Hz
F2CEN     1329.88120 Hz/cm
  
```

1H spectrum



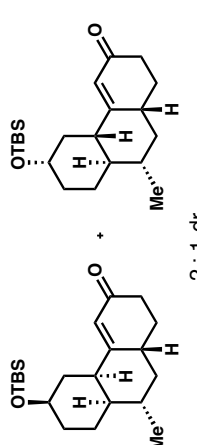
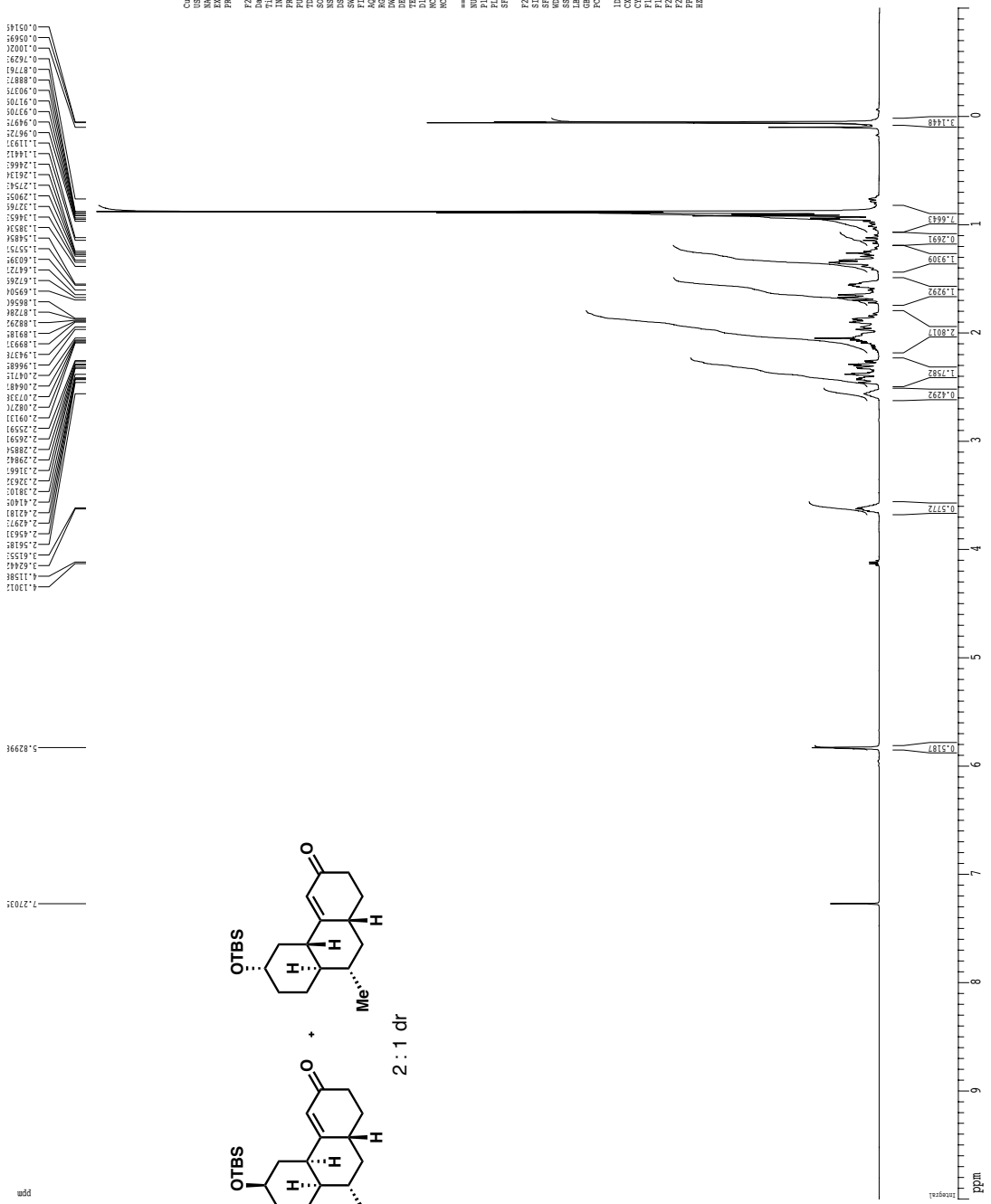
Current Data Parameters  
 NAME Pc3.266\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 201110  
 INSTRUM crys500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SFO1 500.225015 MHz  
 FIDRES 0.250024 Hz  
 AQ 1.399935 sec  
 RG 317  
 DQ 62.400 usec  
 DE 6.00 usec  
 TE 298.2 K  
 DI 0.1000000 sec  
 ACQBST 0.0000000 sec  
 ACQBUR 0.0100000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 30.00 usec  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 L0 MR parameters  
 CK 22.80 cm  
 CT 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 HZCN 241.33423 Hz/cm

### Z-restored spin-echo 13C spectrum with 1H decoupling



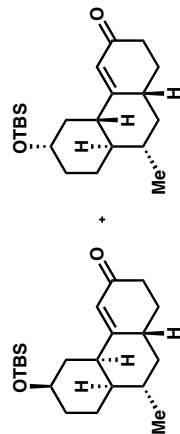
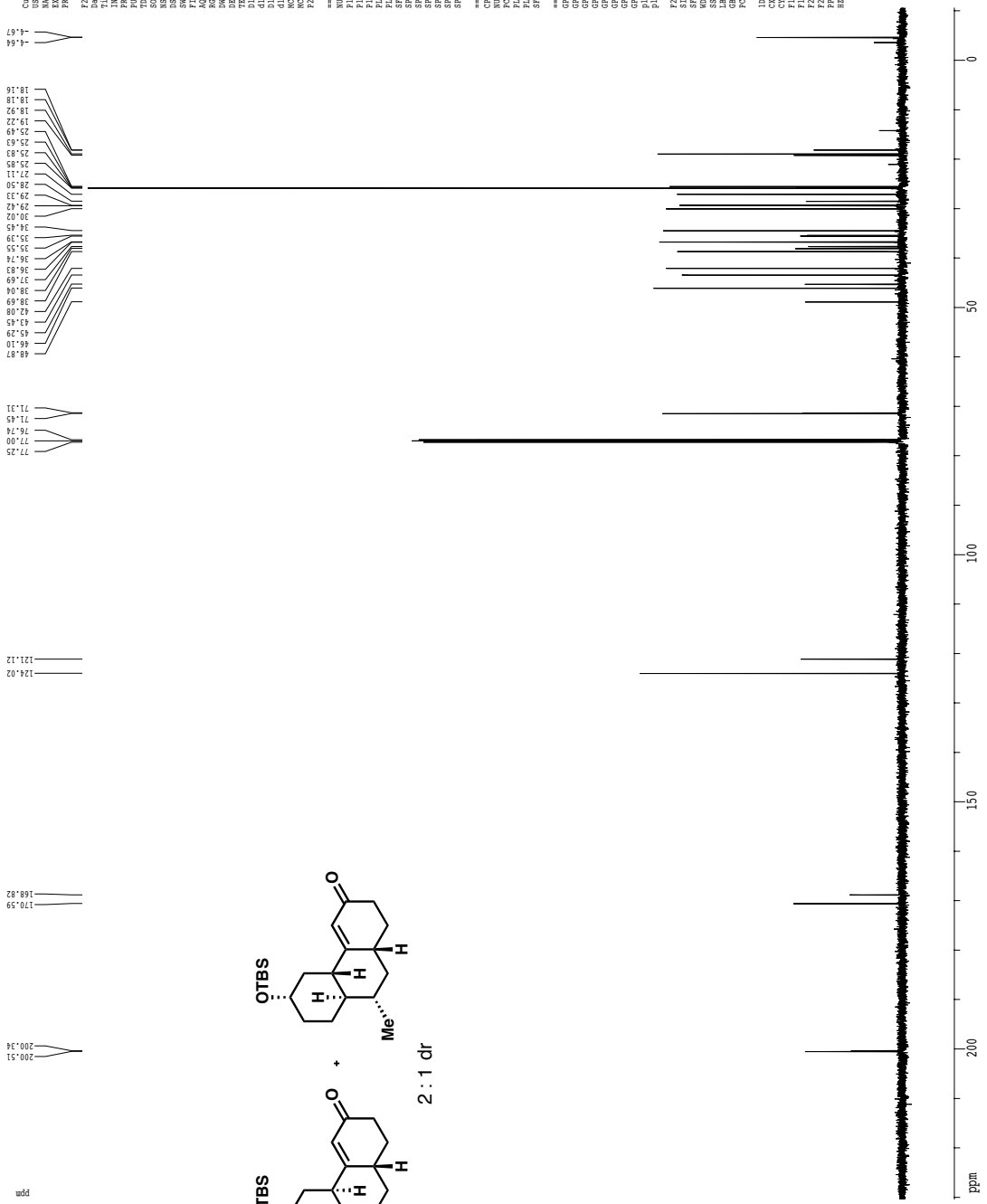


1H spectrum



Current Data Parameters  
 NAME Pcc3.271\_Isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 201010  
 INSTRUM cty6500  
 PROBRD 5 mm CPCL 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 802.820 Hz  
 FIDRES 0.250204 Hz  
 RG 1.1991517  
 GC 0.0000000 sec  
 DC 62.400 usec  
 DE 294.4 Hz  
 DI 0.1000000 sec  
 ACQBST 0.0000000 sec  
 ACQBR 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.45 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WDW RM  
 SSB 0  
 GB 0.00 Hz  
 PC 2.00  
 LO MR parameters  
 CX 22.80 cm  
 CT 15.00 cm  
 CP 1.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz/cm  
 HPCW 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



2 : 1 dr

Current Data Parameters  
 NAME per3270\_150Date  
 EXPNO 2  
 F2 - Acquisition Parameters  
 Date\_ 2011.11.15  
 Time 9.36  
 INSTRUM cryo500  
 PULPROG zgpg30  
 PROCNO 8  
 F2 - Processing parameters  
 DATE\_ 2011.11.15  
 TIME 13.29:08.02  
 INSTRUM spect  
 PULPROG zgpg30  
 PROCNO 8  
 F2 - Processing parameters  
 SI 65536  
 SF 125.7604395 MHz  
 DS 4  
 AS 0  
 L8 1.00 Hz  
 GB 2.00  
 PC 2.00  
 ID NOE plot parameters  
 CT 15.65 cm  
 F1F 230.78 ppm  
 F2F -10.544 ppm  
 FZ -1.23670 Hz  
 HCN 13.29:08.02 Hz/cm

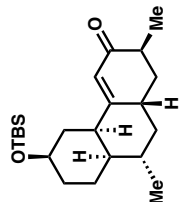
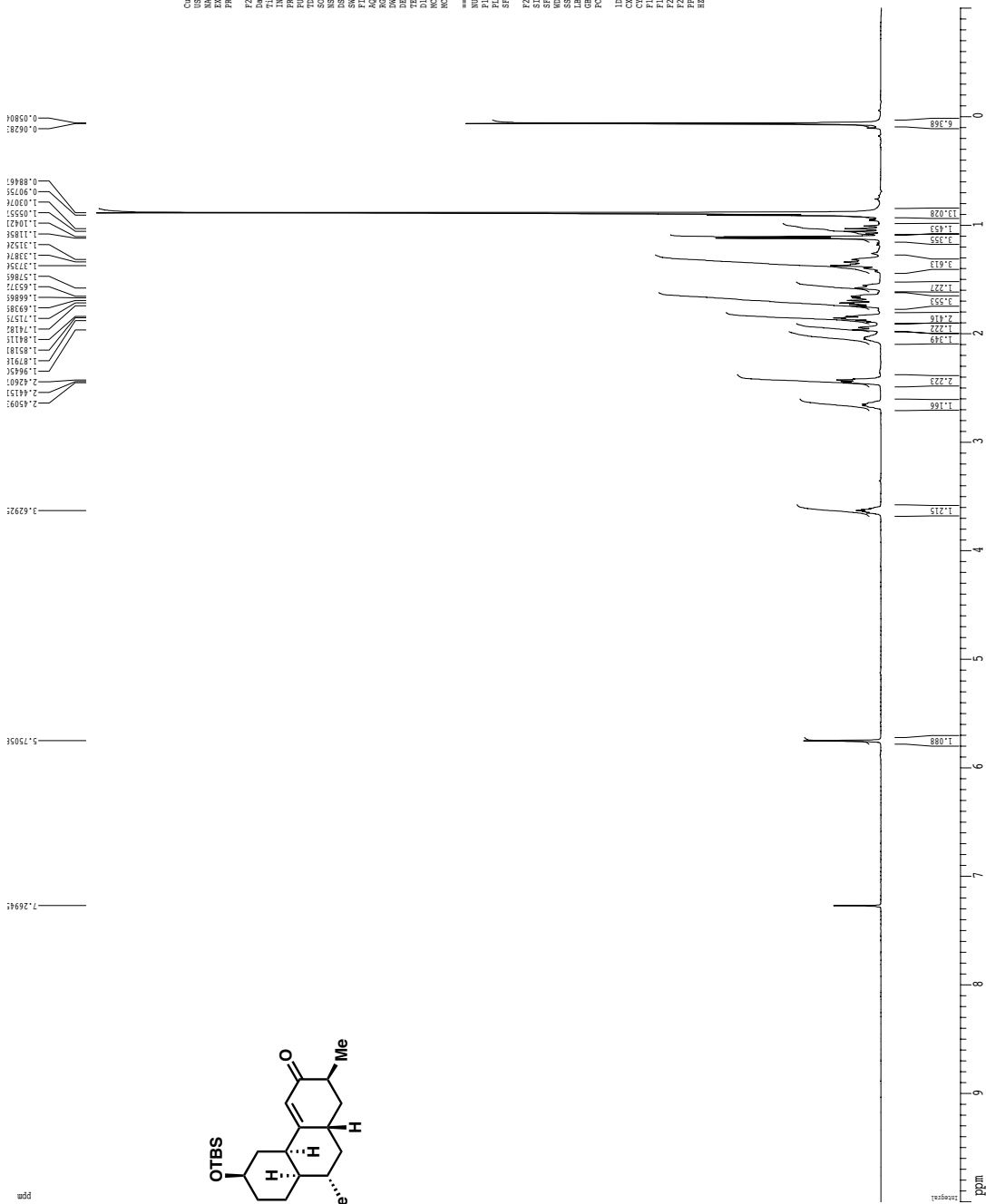
===== CHANNEL F1 =====  
 NU1 15.50 uSAC  
 F1 500.00 uSAC  
 F2 200.00 uSAC  
 F3 200.00 dB  
 F4 1.00 dB  
 F5 125.7604395 MHz  
 SFO1 3.20 dB  
 SFO2 CFC63.63281  
 SHIM0 C299.9678  
 SHIM1 0.00 Hz  
 SFOFF1 0.00 Hz  
 SFOFF2 0.00 Hz

===== CHANNEL F2 =====  
 CDPR2 wal25.6  
 PCP2 100.00 uSAC  
 F4 1.00 dB  
 F5 500.2254311 MHz

===== CHANNEL M1 =====  
 GRAN1 SINE.100  
 GRAN2 SINE.100  
 GPC1 0.00 A  
 GPC2 0.00 A  
 GPC3 0.00 A  
 GPC4 0.00 A  
 GPC5 0.00 A  
 GPC6 0.00 A  
 GPC7 50.00 A  
 GPC8 1000.00 uSAC  
 GPC9 0.00 A

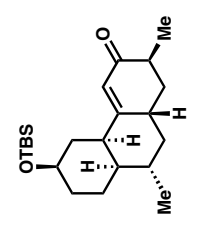
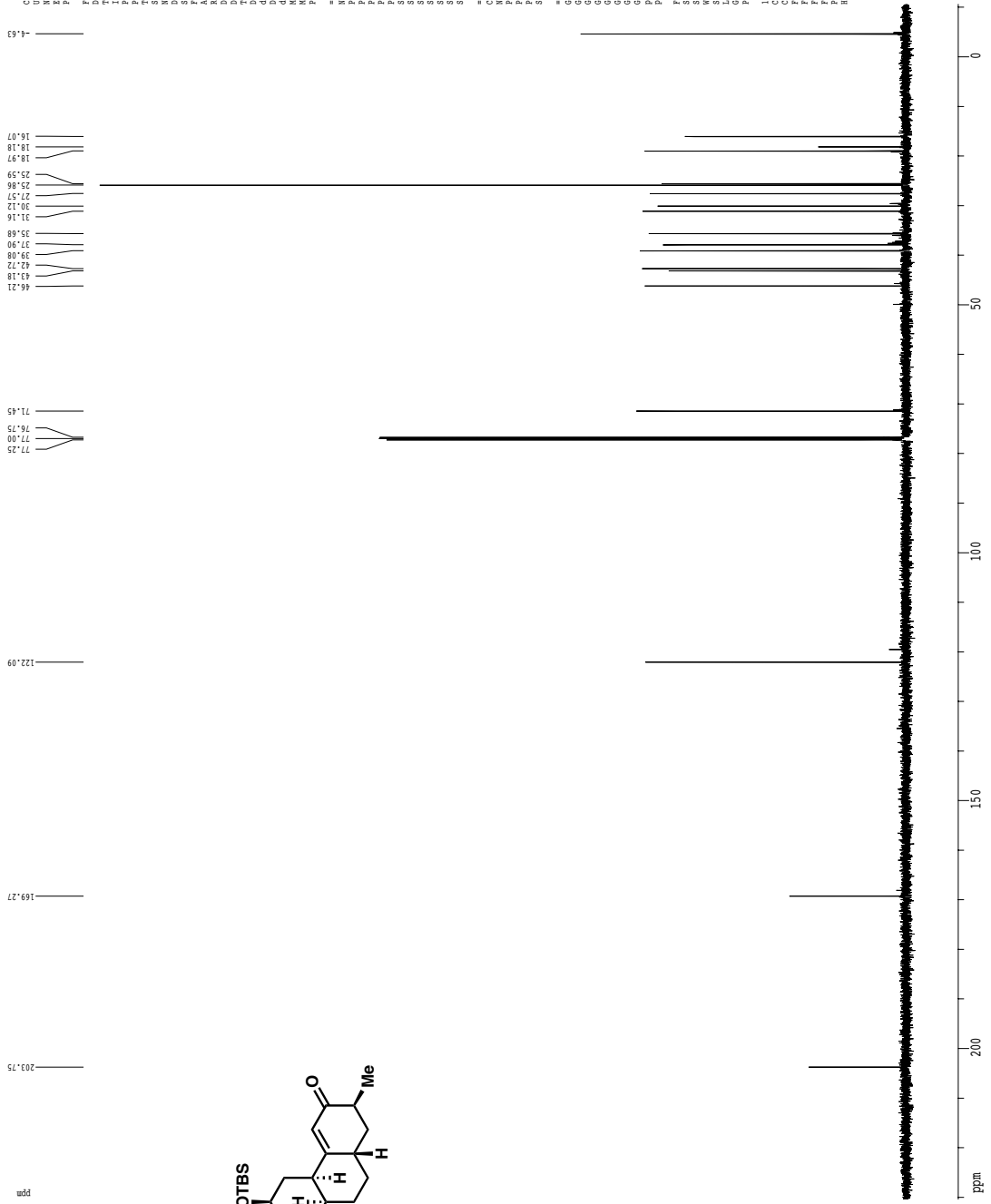
===== CHANNEL M2 =====  
 GRAN1 SINE.100  
 GRAN2 SINE.100  
 GPC1 0.00 A  
 GPC2 0.00 A  
 GPC3 0.00 A  
 GPC4 0.00 A  
 GPC5 0.00 A  
 GPC6 0.00 A  
 GPC7 50.00 A  
 GPC8 1000.00 uSAC  
 GPC9 0.00 A

1H spectrum



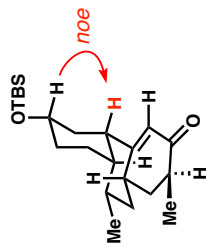
Current Data Parameters  
NAME Pcc3-271\_130-46  
EXPNO 1  
PROCNO 1  
F2 - Acquisition Parameters  
Date\_ 201110  
Time 11:05:00  
INSTRUM cty6500  
PROBHD 5 mm CPXI 1H-  
PULPROG zgpg30  
TD 32748  
SOLVENT CDCl3  
NS 2  
DS 2  
SWH 802.820 Hz  
FIDRES 0.25026 Hz  
AQ 1.19964 sec  
RG 4.5  
DM 62.400 usec  
DE 6.00 usec  
TE 298.2 K  
D1 0.1000000 sec  
ACQSF 0.0000000 sec  
SFO1 500.1300000 sec  
===== CHANNEL f1 =====  
NUC1 13C  
P1 7.50 usec  
PL1 0.00 dB  
PL2 1.60 dB  
SFO1 500.1300000 MHz  
F2 - Processing parameters  
SI 6536  
SF 500.1300000 MHz  
WDW EM  
SSB 0  
GB 0  
PC 4.00  
LO MRB pulse parameters  
CX 22.80 cm  
CY 15.00 cm  
CZ 15.00 cm  
F1 500.1300000 MHz  
F2 -1.000 ppm  
F3 -500.1300000 MHz  
HPCON 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



Current Data Parameters  
 NAME: pct3271\_1518-16  
 EXPNO: 2  
 F2 - Acquisition Parameters  
 PROCNO: 1  
 Date\_ Time: 2010.12.13  
 File: 12.13.10  
 INSTRUM: cryo500  
 PROBR: 5mmCPY130  
 PULPROG: zgpg30  
 ACQPROG: zgpg30  
 DP: 65536  
 ASSEMBLY: mkl011  
 SOLVENT: CDCl3  
 NS: 4  
 DS: 2  
 SWH: 30000.000 Hz  
 FWHM: 1.0013840 Hz  
 AQ: 1.251  
 RG: 1  
 SFO: 125.761200 MHz  
 EC: 6.00 usec  
 DE: 6.00 usec  
 TE: 0.29830 K  
 TD: 0.6300000 Hz  
 FIDRES: 0.0002000 Hz  
 DWDW: 0.0001000 Hz  
 NUC1: 13C  
 NUC2: 13C  
 NUC3: 13C  
 KWEST: 0.0000000 Hz  
 MONK: 0.0150000 Hz  
 F2: 31.00 usec  
 ===== CHANNEL f1 =====  
 CH1: 13C  
 NU1: 13  
 FL1: 15.50 usec  
 PL1: 0.00 dB  
 F2: 100.00 usec  
 PL2: 0.00 dB  
 F3: 100.00 usec  
 PL3: 0.00 dB  
 F4: 100.00 usec  
 PL4: 0.00 dB  
 F5: 125.761200 MHz  
 SFO1: 125.761200 MHz  
 SF1: 125.761200 MHz  
 SF2: 3.200 MHz  
 SF3: 3.200 MHz  
 SF4: 3.200 MHz  
 SF5: 3.200 MHz  
 SHWO: Cp(1)0.520.1  
 SHWB: 0.0000000 Hz  
 SHWF1: 0.00 Hz  
 SHWF2: 0.00 Hz  
 ===== CHANNEL f2 =====  
 CH2: 13C  
 NU2: 13  
 FL2: 15.50 usec  
 PL2: 0.00 dB  
 F2: 100.00 usec  
 PL2: 0.00 dB  
 F3: 100.00 usec  
 PL3: 0.00 dB  
 F4: 100.00 usec  
 PL4: 0.00 dB  
 F5: 500.225011 MHz  
 SFO2: 500.225011 MHz  
 ===== CHANNEL f3 =====  
 CH3: 13C  
 NU3: 13  
 FL3: 15.50 usec  
 PL3: 0.00 dB  
 F2: 100.00 usec  
 PL2: 0.00 dB  
 F3: 100.00 usec  
 PL3: 0.00 dB  
 F4: 100.00 usec  
 PL4: 0.00 dB  
 F5: 50.00 usec  
 PL5: 0.00 dB  
 F6: 50.00 usec  
 PL6: 0.00 dB  
 F7: 100.00 usec  
 PL7: 0.00 dB  
 ===== Processing parameters =====  
 SI: 65536  
 SF: 125.760491 MHz  
 DS: 2  
 AS: 0  
 L8: 1.00 Hz  
 L9: 1.00 Hz  
 GB: 0  
 PC: 2.00  
 ID: 000 plot parameters  
 CT: 15.45 cm  
 E1: 1.00 usec  
 F1P: 230.380 ppm  
 F2P: 0.000000 ppm  
 F3P: -10.540 ppm  
 F4P: 0.000000 ppm  
 F5P: -125.74 Hz / cm  
 F6P: 0.000000 Hz / cm  
 F7P: 1329.68020 Hz / cm

gnoe



Current Data Parameters  
USER pcf3.271\_1306\_6  
INPRO 6  
PROCNO 1

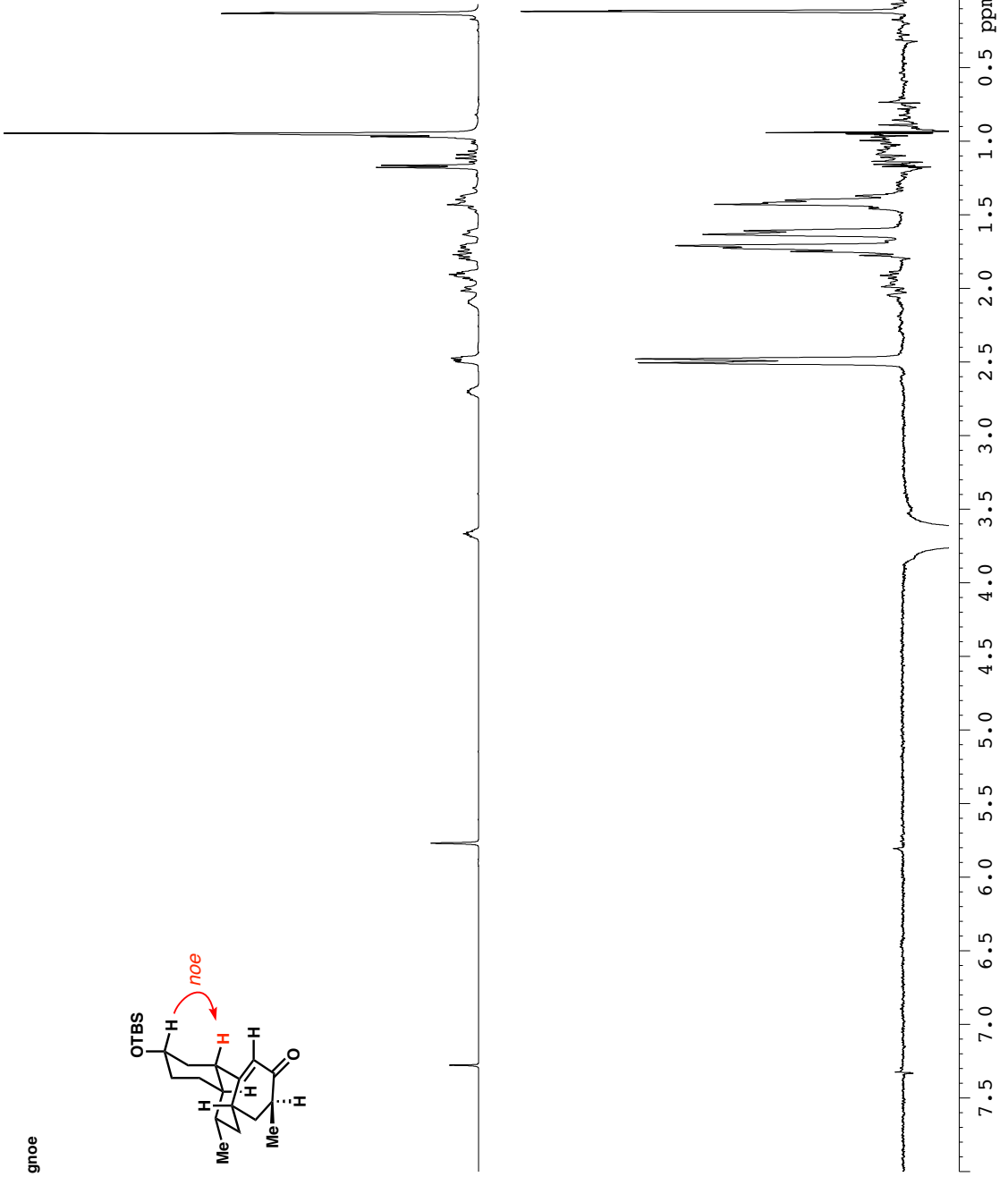
F2 - Acquisition Parameters  
Time 24.13.12  
Date 08-16-2011  
PROBHD 5 mm CPCC 1H-  
PULPROG zgpg30  
TD 65536  
SOLVENT CDCl3  
DS 2  
AQ 6.12.52 Hz  
FIDRES 0.122266 Hz  
RG 4.0889456 sec  
DM 62.400 usec  
DE 1.10000000 K  
TE 300.2 K  
D8 0.50000000 sec  
D16 0.00500000 sec  
d12 0.16339699 sec  
d2 0.16339699 sec  
F2 15.00 usec

==== CHANNEL F1 =====  
NUC1 1H  
P1 7.50 usec  
PL1 0.00 dB  
P4 30.00 usec  
PL4 0.00 dB  
P2 40000.00 usec  
PL2 0.00 dB  
SFO1 500.2218451 MHz  
SI 1 61.60  
SFOF1 gnuarf0 0.400 Hz

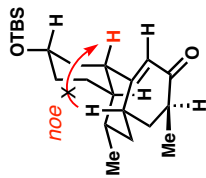
==== CHANNEL F2 =====  
GRAB1 1000.00 usec  
GRAB2 10.00 usec  
GRAB3 30.00 usec  
GRAB4 30.00 usec  
SFO2 500.2218451 MHz  
SI 2 61.60  
SFOF2 gnuarf0 0.400 Hz

==== GRADIENT CHANNEL =====  
GDMX3 10.00 usec  
GDMX4 10.00 usec  
GDMX5 10.00 usec  
GDMX6 10.00 usec  
GDMX7 10.00 usec  
GDMX8 10.00 usec  
GDMX9 10.00 usec  
GDMX10 10.00 usec  
GDMX11 10.00 usec  
GDMX12 10.00 usec  
GDMX13 10.00 usec  
GDMX14 10.00 usec  
GDMX15 10.00 usec  
GDMX16 10.00 usec  
GDMX17 10.00 usec  
GDMX18 10.00 usec  
GDMX19 10.00 usec  
GDMX20 10.00 usec  
GDMX21 10.00 usec  
GDMX22 10.00 usec  
GDMX23 10.00 usec  
GDMX24 10.00 usec  
GDMX25 10.00 usec  
GDMX26 10.00 usec  
GDMX27 10.00 usec  
GDMX28 10.00 usec  
GDMX29 10.00 usec  
GDMX30 10.00 usec  
GDMX31 10.00 usec  
GDMX32 10.00 usec  
GDMX33 10.00 usec  
GDMX34 10.00 usec  
GDMX35 10.00 usec  
GDMX36 10.00 usec  
GDMX37 10.00 usec  
GDMX38 10.00 usec  
GDMX39 10.00 usec  
GDMX40 10.00 usec  
GDMX41 10.00 usec  
GDMX42 10.00 usec  
GDMX43 10.00 usec  
GDMX44 10.00 usec  
GDMX45 10.00 usec  
GDMX46 10.00 usec  
GDMX47 10.00 usec  
GDMX48 10.00 usec  
GDMX49 10.00 usec  
GDMX50 10.00 usec

F2 - Processing parameters  
SI 2 61.60  
SFO 500.2218451 MHz  
WDW EM  
SSB 0  
LB 1.00 Hz  
GB 0  
PC 1.00

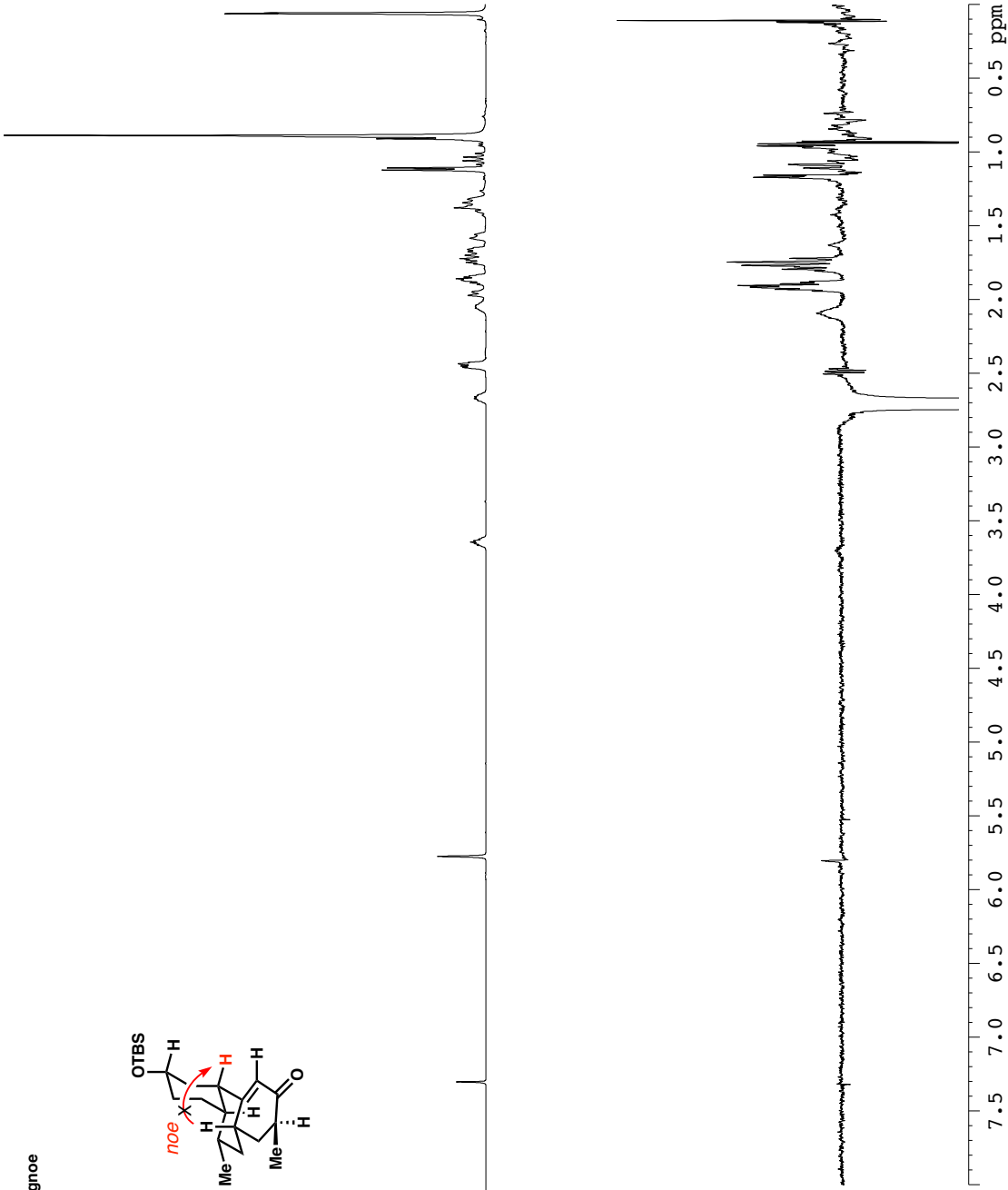


gnoe

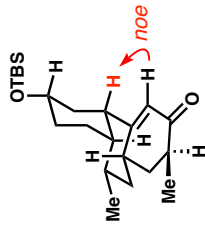


Current Data Parameters  
USER: pcr3.271\_1306\_6  
EXPNO: 6  
PROCNO: 1

F2 - Acquisition Parameters  
Date\_ Time: 2013.12.11 11:12  
Time: 1.13  
PROBHD: 5 mm CPCC 1H-  
PULPROG: gnoe  
SOLVENT: CDCl3  
DS: 2  
SS: 2  
AQ: 0.01250000 sec  
FIDRES: 0.1122266 Hz  
AQ: 62.400 usec  
RG: 327.500  
TE: 300.2 K  
TD: 65536  
DE: 1.00000000 sec  
DQ: 0.00000000 sec  
DQ2: 0.00000000 sec  
DQ3: 0.00000000 sec  
DQ4: 0.00000000 sec  
DQ5: 0.00000000 sec  
DQ6: 0.00000000 sec  
DQ7: 0.00000000 sec  
DQ8: 0.00000000 sec  
DQ9: 0.00000000 sec  
DQ10: 0.00000000 sec  
DQ11: 0.00000000 sec  
DQ12: 0.00000000 sec  
DQ13: 0.00000000 sec  
DQ14: 0.00000000 sec  
DQ15: 0.00000000 sec  
DQ16: 0.00000000 sec  
DQ17: 0.00000000 sec  
DQ18: 0.00000000 sec  
DQ19: 0.00000000 sec  
DQ20: 0.00000000 sec  
DQ21: 0.00000000 sec  
DQ22: 0.00000000 sec  
DQ23: 0.00000000 sec  
DQ24: 0.00000000 sec  
DQ25: 0.00000000 sec  
DQ26: 0.00000000 sec  
DQ27: 0.00000000 sec  
DQ28: 0.00000000 sec  
DQ29: 0.00000000 sec  
DQ30: 0.00000000 sec  
DQ31: 0.00000000 sec  
DQ32: 0.00000000 sec  
DQ33: 0.00000000 sec  
DQ34: 0.00000000 sec  
DQ35: 0.00000000 sec  
DQ36: 0.00000000 sec  
DQ37: 0.00000000 sec  
DQ38: 0.00000000 sec  
DQ39: 0.00000000 sec  
DQ40: 0.00000000 sec  
DQ41: 0.00000000 sec  
DQ42: 0.00000000 sec  
DQ43: 0.00000000 sec  
DQ44: 0.00000000 sec  
DQ45: 0.00000000 sec  
DQ46: 0.00000000 sec  
DQ47: 0.00000000 sec  
DQ48: 0.00000000 sec  
DQ49: 0.00000000 sec  
DQ50: 0.00000000 sec  
DQ51: 0.00000000 sec  
DQ52: 0.00000000 sec  
DQ53: 0.00000000 sec  
DQ54: 0.00000000 sec  
DQ55: 0.00000000 sec  
DQ56: 0.00000000 sec  
DQ57: 0.00000000 sec  
DQ58: 0.00000000 sec  
DQ59: 0.00000000 sec  
DQ60: 0.00000000 sec  
DQ61: 0.00000000 sec  
DQ62: 0.00000000 sec  
DQ63: 0.00000000 sec  
DQ64: 0.00000000 sec  
DQ65: 0.00000000 sec  
DQ66: 0.00000000 sec  
DQ67: 0.00000000 sec  
DQ68: 0.00000000 sec  
DQ69: 0.00000000 sec  
DQ70: 0.00000000 sec  
DQ71: 0.00000000 sec  
DQ72: 0.00000000 sec  
DQ73: 0.00000000 sec  
DQ74: 0.00000000 sec  
DQ75: 0.00000000 sec  
DQ76: 0.00000000 sec  
DQ77: 0.00000000 sec  
DQ78: 0.00000000 sec  
DQ79: 0.00000000 sec  
DQ80: 0.00000000 sec  
DQ81: 0.00000000 sec  
DQ82: 0.00000000 sec  
DQ83: 0.00000000 sec  
DQ84: 0.00000000 sec  
DQ85: 0.00000000 sec  
DQ86: 0.00000000 sec  
DQ87: 0.00000000 sec  
DQ88: 0.00000000 sec  
DQ89: 0.00000000 sec  
DQ90: 0.00000000 sec  
DQ91: 0.00000000 sec  
DQ92: 0.00000000 sec  
DQ93: 0.00000000 sec  
DQ94: 0.00000000 sec  
DQ95: 0.00000000 sec  
DQ96: 0.00000000 sec  
DQ97: 0.00000000 sec  
DQ98: 0.00000000 sec  
DQ99: 0.00000000 sec  
DQ100: 0.00000000 sec



gnoe



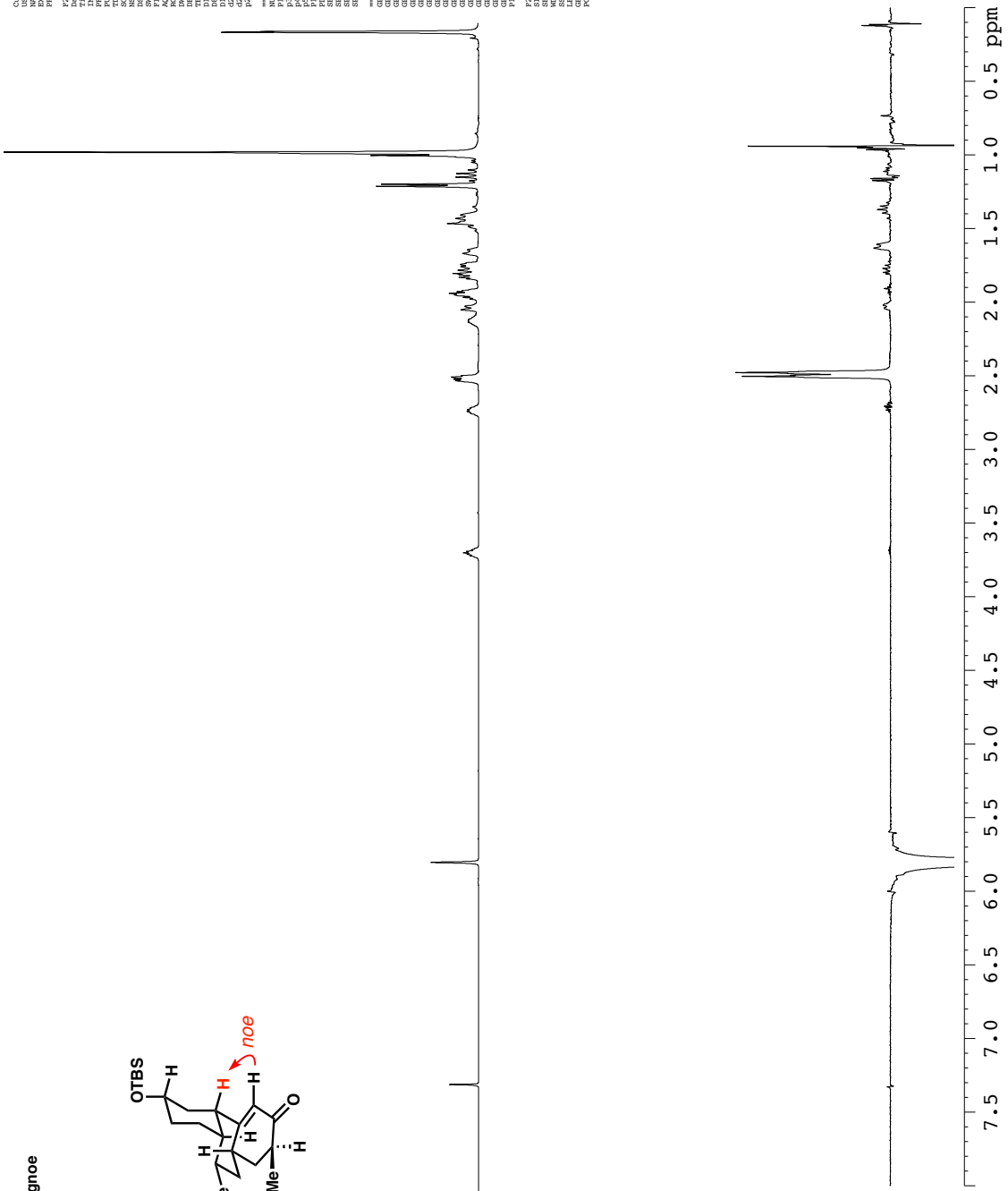
Current Data Parameters  
USER: pcr3.271\_1306\_6  
EXPNO: 6  
PROCNO: 1

F2 - Acquisition Parameters  
Date\_ Time: 2013.12.11 13:12:11  
PROBHD: 5 mm CPYCC 1H-  
PULPROG: gnoe  
SOLVENT: CDCl3  
DS: 2  
AQ: 0.11250000 sec  
FIDRES: 0.1122266 Hz  
AQ: 62.400 usec  
RG: 327.500  
TE: 300.2 K  
DE: 1.16529820 K  
D8: 0.50000000 sec  
D9: 0.00500000 sec  
D16: 0.00500000 sec  
D22: 0.16339699 sec  
D2: 15.000 usec

===== CHANNEL f1 =====  
NUC1: 1H  
P1: 7.50 usec  
PL1: 0.00 dB  
PC1: 30.00 usec  
PR1: 0.00 usec  
P2: 40.000.00 usec  
PC2: 0.00 dB  
PC3: 0.00 dB  
PC4: 0.00 dB  
PC5: 0.00 dB  
PC6: 0.00 dB  
PC7: 0.00 dB  
PC8: 0.00 dB  
PC9: 0.00 dB  
PC10: 0.00 dB  
PC11: 0.00 dB  
PC12: 0.00 dB  
PC13: 0.00 dB  
PC14: 0.00 dB  
PC15: 0.00 dB  
PC16: 0.00 dB  
PC17: 0.00 dB  
PC18: 0.00 dB  
PC19: 0.00 dB  
PC20: 0.00 dB  
PC21: 0.00 dB  
PC22: 0.00 dB  
PC23: 0.00 dB  
PC24: 0.00 dB  
PC25: 0.00 dB  
PC26: 0.00 dB  
PC27: 0.00 dB  
PC28: 0.00 dB  
PC29: 0.00 dB  
PC30: 0.00 dB  
PC31: 0.00 dB  
PC32: 0.00 dB  
PC33: 0.00 dB  
PC34: 0.00 dB  
PC35: 0.00 dB  
PC36: 0.00 dB  
PC37: 0.00 dB  
PC38: 0.00 dB  
PC39: 0.00 dB  
PC40: 0.00 dB  
PC41: 0.00 dB  
PC42: 0.00 dB  
PC43: 0.00 dB  
PC44: 0.00 dB  
PC45: 0.00 dB  
PC46: 0.00 dB  
PC47: 0.00 dB  
PC48: 0.00 dB  
PC49: 0.00 dB  
PC50: 0.00 dB  
PC51: 0.00 dB  
PC52: 0.00 dB  
PC53: 0.00 dB  
PC54: 0.00 dB  
PC55: 0.00 dB  
PC56: 0.00 dB  
PC57: 0.00 dB  
PC58: 0.00 dB  
PC59: 0.00 dB  
PC60: 0.00 dB  
PC61: 0.00 dB  
PC62: 0.00 dB  
PC63: 0.00 dB  
PC64: 0.00 dB  
PC65: 0.00 dB  
PC66: 0.00 dB  
PC67: 0.00 dB  
PC68: 0.00 dB  
PC69: 0.00 dB  
PC70: 0.00 dB  
PC71: 0.00 dB  
PC72: 0.00 dB  
PC73: 0.00 dB  
PC74: 0.00 dB  
PC75: 0.00 dB  
PC76: 0.00 dB  
PC77: 0.00 dB  
PC78: 0.00 dB  
PC79: 0.00 dB  
PC80: 0.00 dB  
PC81: 0.00 dB  
PC82: 0.00 dB  
PC83: 0.00 dB  
PC84: 0.00 dB  
PC85: 0.00 dB  
PC86: 0.00 dB  
PC87: 0.00 dB  
PC88: 0.00 dB  
PC89: 0.00 dB  
PC90: 0.00 dB  
PC91: 0.00 dB  
PC92: 0.00 dB  
PC93: 0.00 dB  
PC94: 0.00 dB  
PC95: 0.00 dB  
PC96: 0.00 dB  
PC97: 0.00 dB  
PC98: 0.00 dB  
PC99: 0.00 dB  
PC100: 0.00 dB

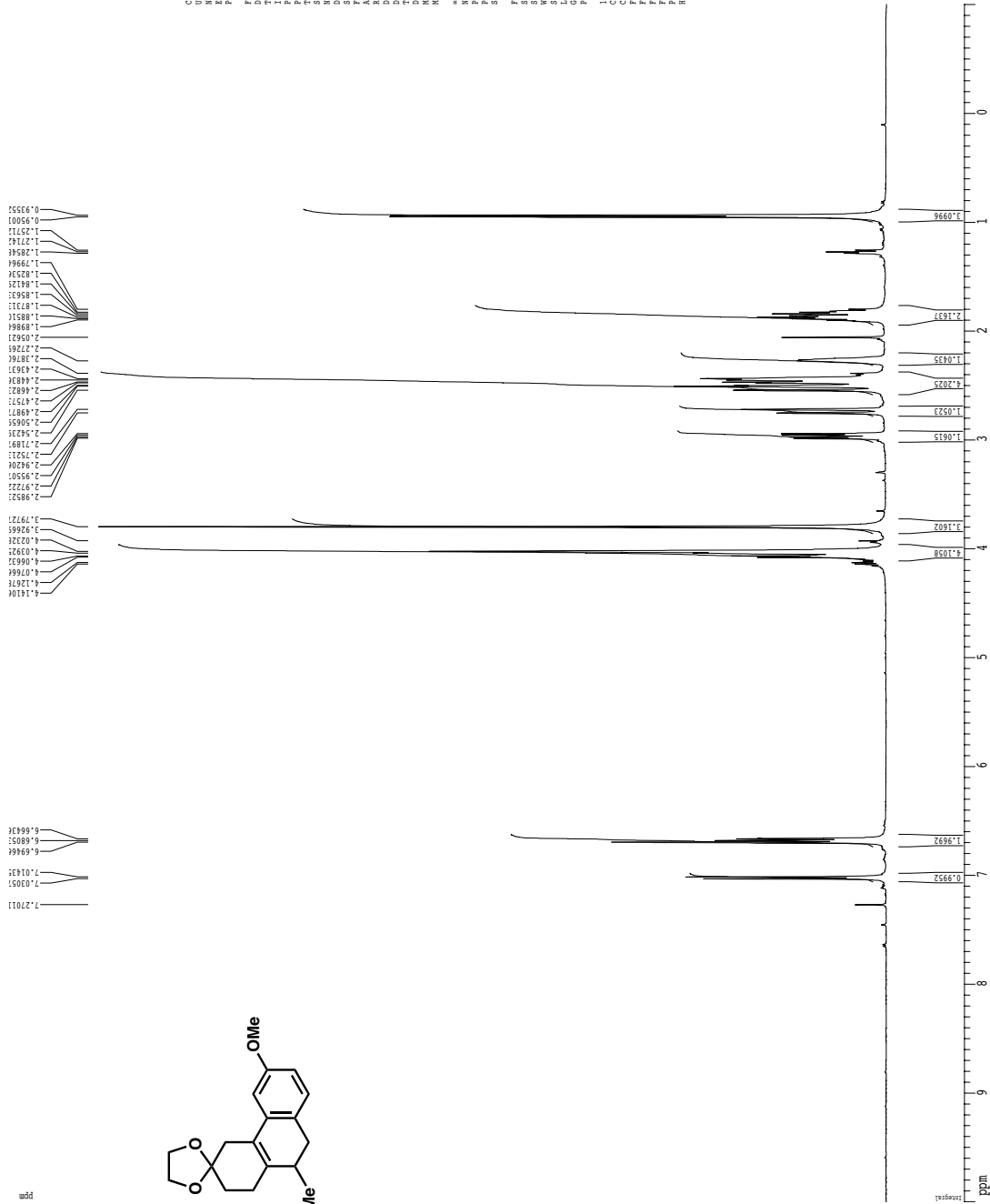
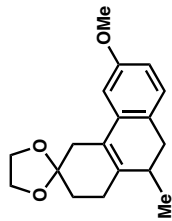
===== GRADIENT CHANNEL =====  
G1: 0.00 usec  
G2: 0.00 usec  
G3: 0.00 usec  
G4: 0.00 usec  
G5: 0.00 usec  
G6: 0.00 usec  
G7: 0.00 usec  
G8: 0.00 usec  
G9: 0.00 usec  
G10: 0.00 usec  
G11: 0.00 usec  
G12: 0.00 usec  
G13: 0.00 usec  
G14: 0.00 usec  
G15: 0.00 usec  
G16: 0.00 usec  
G17: 0.00 usec  
G18: 0.00 usec  
G19: 0.00 usec  
G20: 0.00 usec  
G21: 0.00 usec  
G22: 0.00 usec  
G23: 0.00 usec  
G24: 0.00 usec  
G25: 0.00 usec  
G26: 0.00 usec  
G27: 0.00 usec  
G28: 0.00 usec  
G29: 0.00 usec  
G30: 0.00 usec  
G31: 0.00 usec  
G32: 0.00 usec  
G33: 0.00 usec  
G34: 0.00 usec  
G35: 0.00 usec  
G36: 0.00 usec  
G37: 0.00 usec  
G38: 0.00 usec  
G39: 0.00 usec  
G40: 0.00 usec  
G41: 0.00 usec  
G42: 0.00 usec  
G43: 0.00 usec  
G44: 0.00 usec  
G45: 0.00 usec  
G46: 0.00 usec  
G47: 0.00 usec  
G48: 0.00 usec  
G49: 0.00 usec  
G50: 0.00 usec  
G51: 0.00 usec  
G52: 0.00 usec  
G53: 0.00 usec  
G54: 0.00 usec  
G55: 0.00 usec  
G56: 0.00 usec  
G57: 0.00 usec  
G58: 0.00 usec  
G59: 0.00 usec  
G60: 0.00 usec  
G61: 0.00 usec  
G62: 0.00 usec  
G63: 0.00 usec  
G64: 0.00 usec  
G65: 0.00 usec  
G66: 0.00 usec  
G67: 0.00 usec  
G68: 0.00 usec  
G69: 0.00 usec  
G70: 0.00 usec  
G71: 0.00 usec  
G72: 0.00 usec  
G73: 0.00 usec  
G74: 0.00 usec  
G75: 0.00 usec  
G76: 0.00 usec  
G77: 0.00 usec  
G78: 0.00 usec  
G79: 0.00 usec  
G80: 0.00 usec  
G81: 0.00 usec  
G82: 0.00 usec  
G83: 0.00 usec  
G84: 0.00 usec  
G85: 0.00 usec  
G86: 0.00 usec  
G87: 0.00 usec  
G88: 0.00 usec  
G89: 0.00 usec  
G90: 0.00 usec  
G91: 0.00 usec  
G92: 0.00 usec  
G93: 0.00 usec  
G94: 0.00 usec  
G95: 0.00 usec  
G96: 0.00 usec  
G97: 0.00 usec  
G98: 0.00 usec  
G99: 0.00 usec  
G100: 0.00 usec

F2 - Processing parameters  
SI: 327.500 MHz  
SF: 500.2200000 MHz  
WDW: EM  
SSB: 0  
LB: 1.00 Hz  
GB: 0  
PC: 1.00



1H spectrum

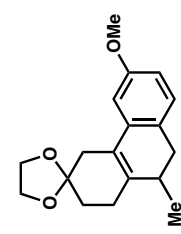
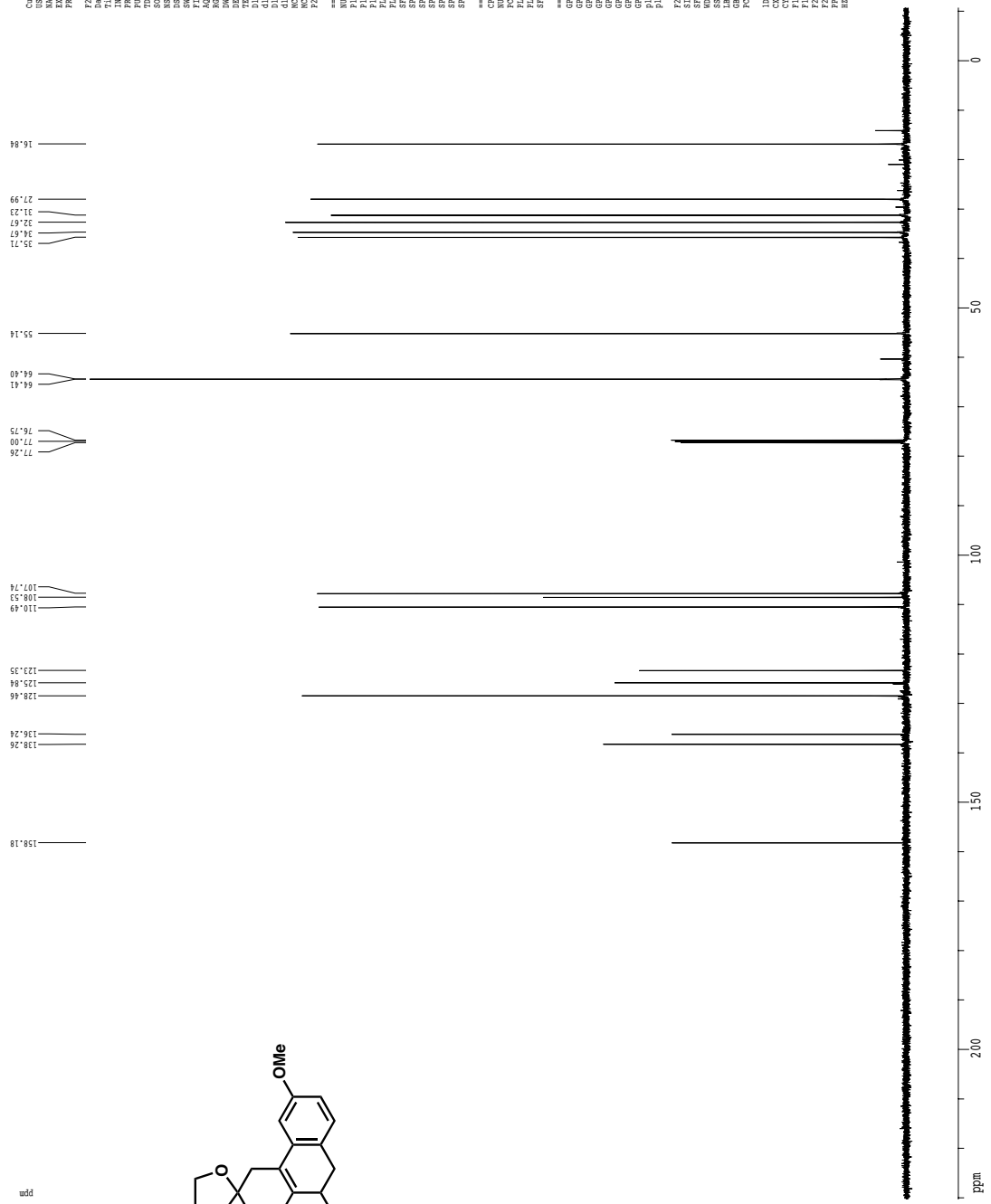
ppm



Current Data Parameters  
 NAME pac3.277\_1sp04a\_2  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20080806  
 Time 0.06  
 INSTRUM crys00  
 PULPROG zgpg30  
 TD 32768  
 CONVENTION IUPAC  
 NS 8  
 DS 8  
 SFO1 500.136261 MHz  
 AQ 1.9999451 sec  
 ZW 62.400 usec  
 DE 6.00 usec  
 DI 0.1000000 sec  
 NUCHE1 0.0000000 sec  
 NUCHE2 0.0150000 sec  
 ===== CHANNEL f1 =====  
 NUC1 7.25 usec  
 P1 1.60 dB  
 SFO1 500.223015 MHz  
 F2 - Processing parameters  
 S1 653.6  
 VM 500.223015 MHz  
 WID 0  
 GB 0  
 CB 0  
 PC 4.00  
 ID\_NMR plot parameters  
 CT 22.80 cm  
 F1P 10.000 GHz  
 F1 500.220 MHz  
 F2 500.223015 MHz  
 FREQN 0.48246 ppm/cm  
 HZCN 241.33822 Hz/cm



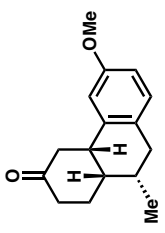
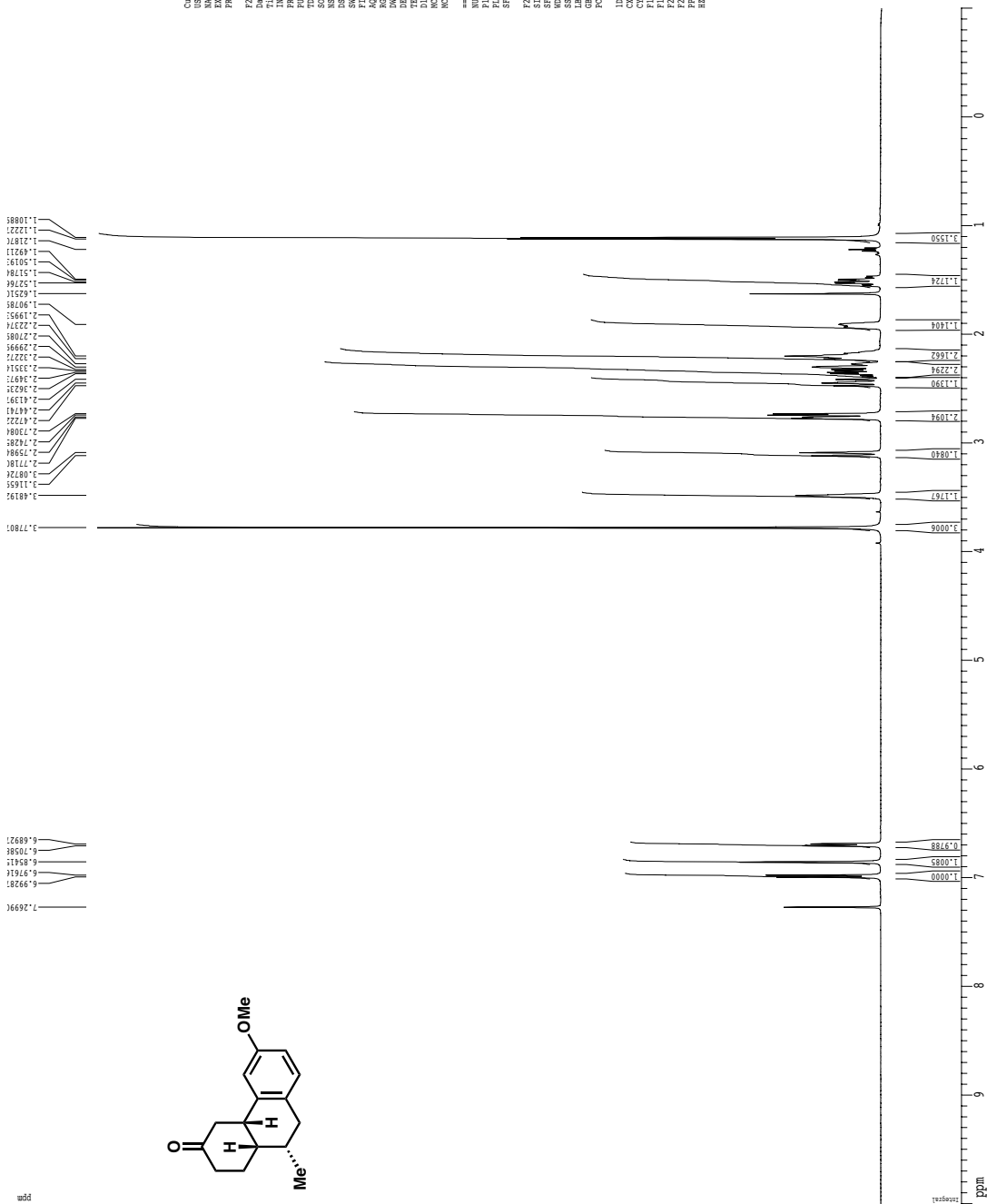
Z-restored spin-echo 13C spectrum with 1H decoupling



```

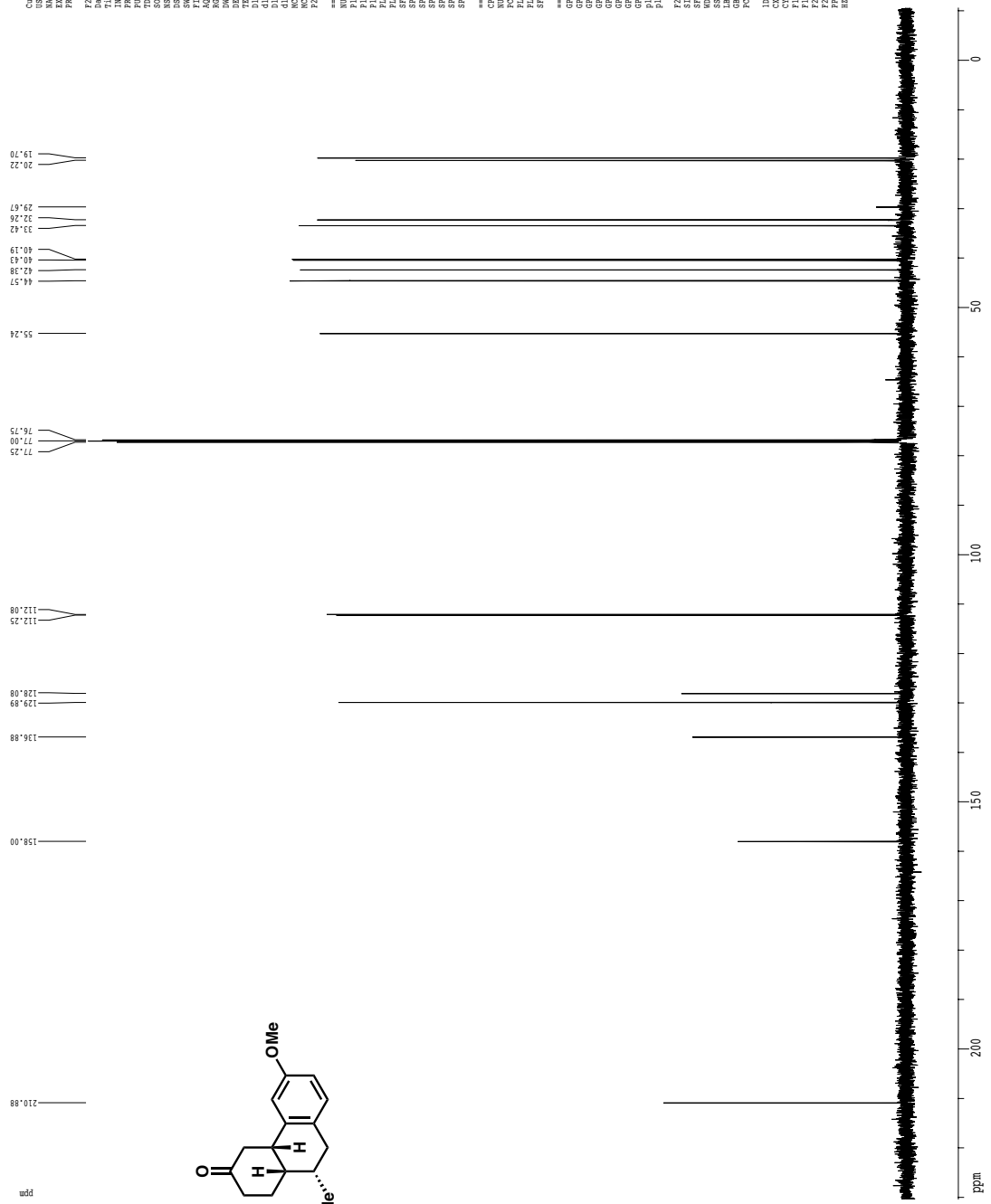
Current Data Parameters
NAME          pc3277_150146_2
EXPNO         2
PROCNO        1
F2 - Acquisition Parameters
Date_         20100808
Time          01:08
INSTRUM       cryo00
PROBHD        5mm CPY13
PULPROG       zgpg30
HARDWARE      spect
SOLVENT       CDCl3
NS            2
DS            4
SFO1          300.13538 Hz
SFO2          100.628150 Hz
AQ            1.0013340 sec
RG            632
RG2           632
DE            6.00 usec
TE            300.2 K
TD            65536
FIDRES        0.49300000 sec
AQRES         0.00200000 sec
SFO3          125.761272 Hz
SFO4          2.70 dB
SFO5          2.70 dB
SFO6          0.00000000 sec
SFO7          0.00000000 sec
SFO8          0.00000000 sec
SFO9          0.00000000 sec
SFO10         0.00000000 sec
SFO11         0.00000000 sec
SFO12         0.00 Hz
SFO13         0.00 Hz
===== CHANNEL F1 =====
NUC1          13
P1            16.55 usec
PL1           0.00 dB
PC1           100.00 usec
RG1           200.00
RG2           200.00 dB
F1            125.761272 MHz
SFO1          -1.00 dB
SFO2          2.70 dB
SFO3          2.70 dB
SFO4          0.00000000 sec
SFO5          0.00000000 sec
SFO6          0.00 Hz
SFO7          0.00 Hz
===== CHANNEL F2 =====
CPDPRG2       waltz16
NUC2          13
P2            16.55 usec
PL2           0.00 dB
PC2           100.00 usec
RG2           200.00
RG3           200.00 dB
F2            125.761272 MHz
SFO1          -1.00 dB
SFO2          2.70 dB
SFO3          2.70 dB
SFO4          0.00000000 sec
SFO5          0.00000000 sec
SFO6          0.00 Hz
SFO7          0.00 Hz
===== CHANNEL F3 =====
GRANU         STW1.100
GRAMS         STW1.100
SFO1          50.000000 MHz
SFO2          50.000000 MHz
SFO3          50.000000 MHz
SFO4          50.000000 MHz
SFO5          100.000000 MHz
===== Processing parameters =====
SI            65536
SF            125.760406 MHz
WDW           EM
SSB           0
LB            0
GB            0
PC            2.00
===== ID MS parameters =====
CT            15.45 cm
PIF           230.288 ppm
F2F           -10.432 ppm
F2            -1337.30 Hz
RG            1329.8032 Hz/cm
    
```

1H spectrum



Current Data Parameters  
 NAME Pcc3-298 f10-17  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2011209  
 Time 15:00  
 INSTRUM ctyc500  
 PROBRD 5 mm CPCL H-  
 PULPROG zgpg30  
 TD 32748  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250026 Hz  
 AQ 1.1994614 sec  
 RG 64.3  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 DI 0.11000000 sec  
 ACQSF 0.00000000 sec  
 ACQR 0.11000000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.50 usec  
 PL1 -1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 500.225015 MHz  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CK 22.80 cm  
 CT 15.00 cm  
 CF 150.144 MHz  
 FI 5002.20 Hz  
 FFP -1.000 ppm  
 FZ -500.22 Hz  
 HPCW 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



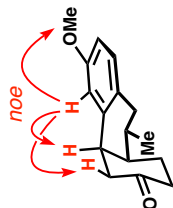
Current Data Parameters  
 NAME pcr3.298\_118-17  
 EXPNO 2  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20160916  
 Time 9:45  
 INSTRUM cryo500  
 PULPROG zgpg30  
 PROCPRG zgpg30  
 F2 - Processing parameters  
 SI 65536  
 SF 125.7604309 MHz  
 DS 4  
 AS 0  
 L8 1.00 Hz  
 SFO 2.00  
 ID NSG plot parameters  
 CT 15.45 cm  
 F1F 230.465 ppm  
 F2F -10.555 ppm  
 F2 -1227.59 Hz  
 F2C 13.29788332 Hz/cm

===== CHANNEL F1 =====  
 NU1 16.55 uSAC  
 F1 500.00 uSAC  
 F2 500.00 uSAC  
 F3 200.00 dB  
 F4 200.00 dB  
 F5 125.7604309 MHz  
 SFO 2.70 dB  
 SF2 2.70 dB  
 SFO2 CPG0.65281  
 SFOFF1 CPG0.65281  
 SFOFF2 0.00 Hz

===== CHANNEL F2 =====  
 CPGPR2 val12.6  
 PCPD2 100.00 uSAC  
 F1 1.00 dB  
 F2 1.00 dB  
 SF2 500.2225411 MHz

===== CHANNEL CHANNEL =====  
 GRANU 100  
 STW 100  
 GPR2 0.00 A  
 GPR3 0.00 A  
 GPR4 0.00 A  
 GPR5 0.00 A  
 GPR6 0.00 A  
 GPR7 0.00 A  
 GPR8 0.00 A  
 GPR9 0.00 A  
 GPR10 0.00 A  
 GPR11 0.00 A  
 GPR12 0.00 A  
 GPR13 0.00 A  
 GPR14 0.00 A  
 GPR15 0.00 A  
 GPR16 0.00 A  
 GPR17 0.00 A  
 GPR18 0.00 A  
 GPR19 0.00 A  
 GPR20 0.00 A  
 GPR21 0.00 A  
 GPR22 0.00 A  
 GPR23 0.00 A  
 GPR24 0.00 A  
 GPR25 0.00 A  
 GPR26 0.00 A  
 GPR27 0.00 A  
 GPR28 0.00 A  
 GPR29 0.00 A  
 GPR30 0.00 A  
 GPR31 0.00 A  
 GPR32 0.00 A  
 GPR33 0.00 A  
 GPR34 0.00 A  
 GPR35 0.00 A  
 GPR36 0.00 A  
 GPR37 0.00 A  
 GPR38 0.00 A  
 GPR39 0.00 A  
 GPR40 0.00 A  
 GPR41 0.00 A  
 GPR42 0.00 A  
 GPR43 0.00 A  
 GPR44 0.00 A  
 GPR45 0.00 A  
 GPR46 0.00 A  
 GPR47 0.00 A  
 GPR48 0.00 A  
 GPR49 0.00 A  
 GPR50 0.00 A  
 GPR51 0.00 A  
 GPR52 0.00 A  
 GPR53 0.00 A  
 GPR54 0.00 A  
 GPR55 0.00 A  
 GPR56 0.00 A  
 GPR57 0.00 A  
 GPR58 0.00 A  
 GPR59 0.00 A  
 GPR60 0.00 A  
 GPR61 0.00 A  
 GPR62 0.00 A  
 GPR63 0.00 A  
 GPR64 0.00 A  
 GPR65 0.00 A  
 GPR66 0.00 A  
 GPR67 0.00 A  
 GPR68 0.00 A  
 GPR69 0.00 A  
 GPR70 0.00 A  
 GPR71 0.00 A  
 GPR72 0.00 A  
 GPR73 0.00 A  
 GPR74 0.00 A  
 GPR75 0.00 A  
 GPR76 0.00 A  
 GPR77 0.00 A  
 GPR78 0.00 A  
 GPR79 0.00 A  
 GPR80 0.00 A  
 GPR81 0.00 A  
 GPR82 0.00 A  
 GPR83 0.00 A  
 GPR84 0.00 A  
 GPR85 0.00 A  
 GPR86 0.00 A  
 GPR87 0.00 A  
 GPR88 0.00 A  
 GPR89 0.00 A  
 GPR90 0.00 A  
 GPR91 0.00 A  
 GPR92 0.00 A  
 GPR93 0.00 A  
 GPR94 0.00 A  
 GPR95 0.00 A  
 GPR96 0.00 A  
 GPR97 0.00 A  
 GPR98 0.00 A  
 GPR99 0.00 A  
 GPR100 0.00 A

**<sup>1</sup>H spectrum**



```

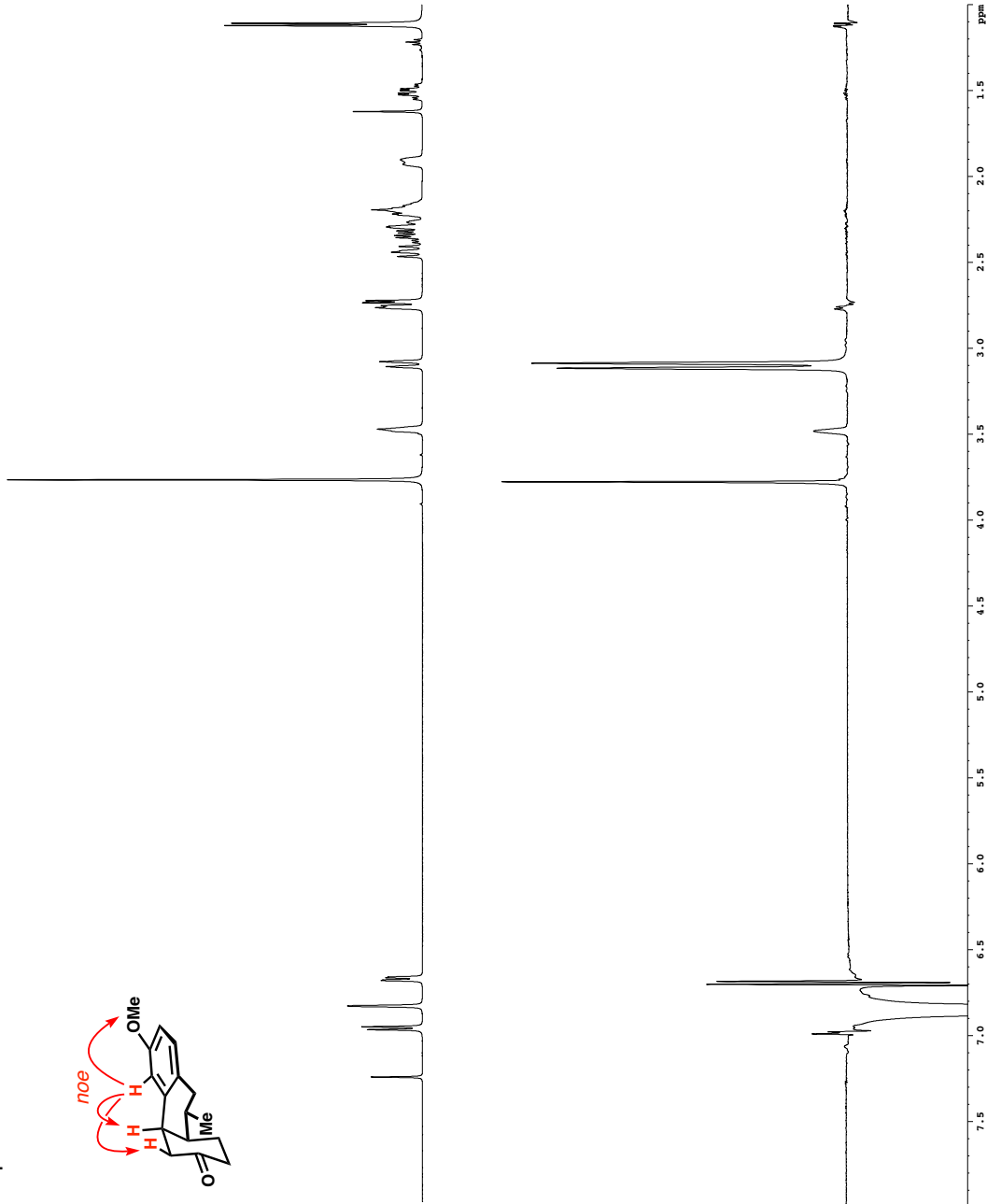
Current Data Parameters
=====
USER      PCT3_298_110-7
EXPNO     1
PROCNO    1

F2 - Acquisition Parameters
=====
Time      24.15.56
Date_     20080811
PROBHD    5 mm CPYCH 1H-
PULPROG   zgpg30
SOLVENT   CDCl3
DS         2
AQ         0.17200000 sec
RG         327.50000000
FIDRES    0.2260026 Hz
AQ        1.99984251 sec
RG         327.50000000
DE         62.40000000 usec
TE        300.2 K
TE        298.0 K
NUC1       13C
NUC2       1H
ACQRES    0.00000000 sec
SFO1       500.136199 MHz
SFO2       500.136199 MHz
SFO3       500.22735015 MHz

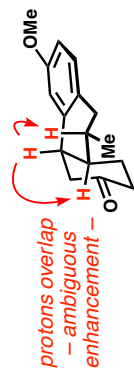
=====
F2 - Processing parameters
=====
SI         65536
SF         500.22735015 MHz
WDW        EM
SSB        0
GB         0
PC         4.00

=====
CX NMR plot parameters
=====
CT         15.000000 cm
CI         15.000000 cm
P1         500.220000 MHz
P2         -500.220000 MHz
PC         4.00000000 cm
=====

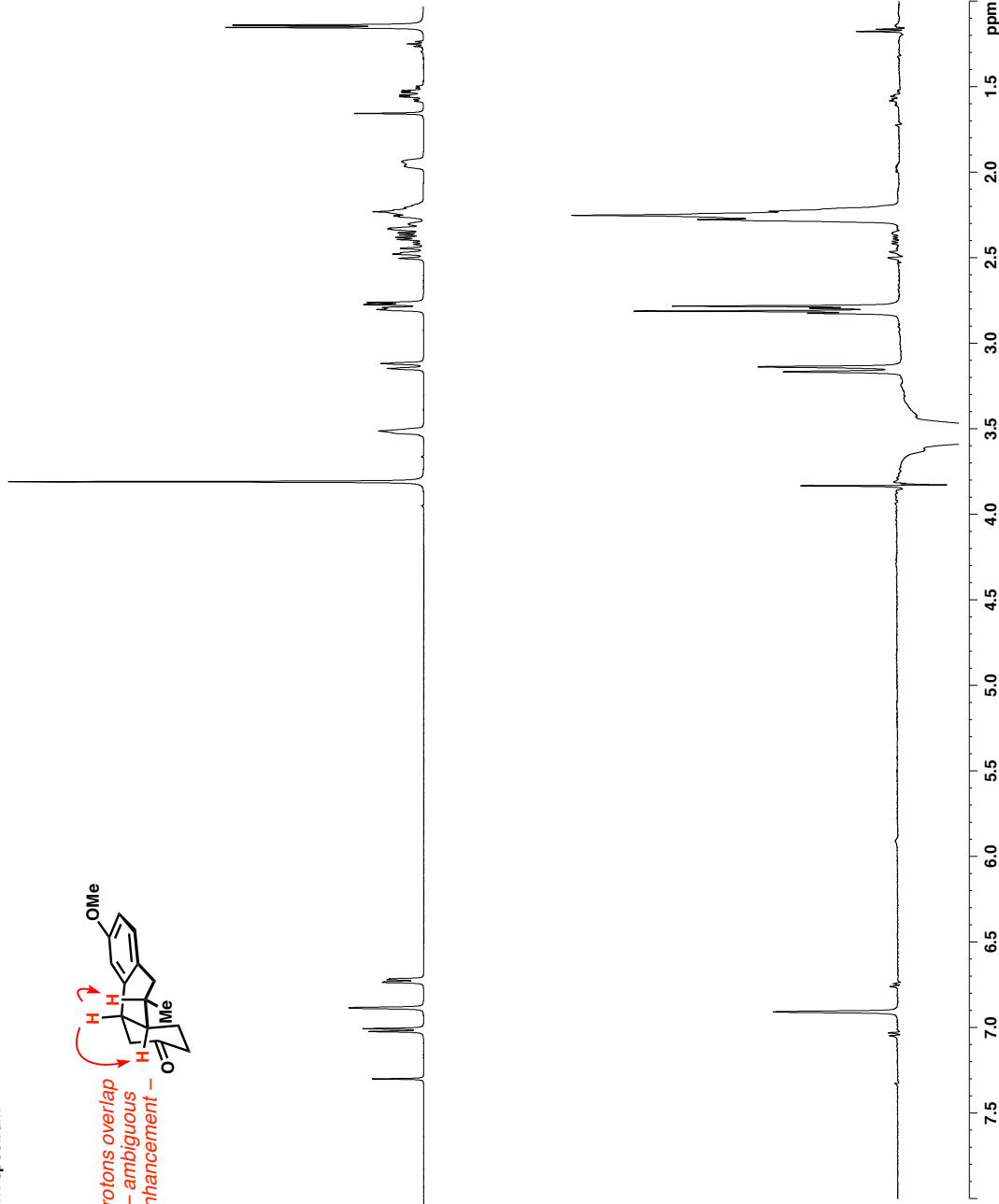
```



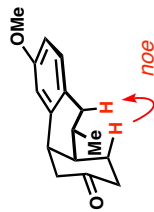
**<sup>1</sup>H spectrum**



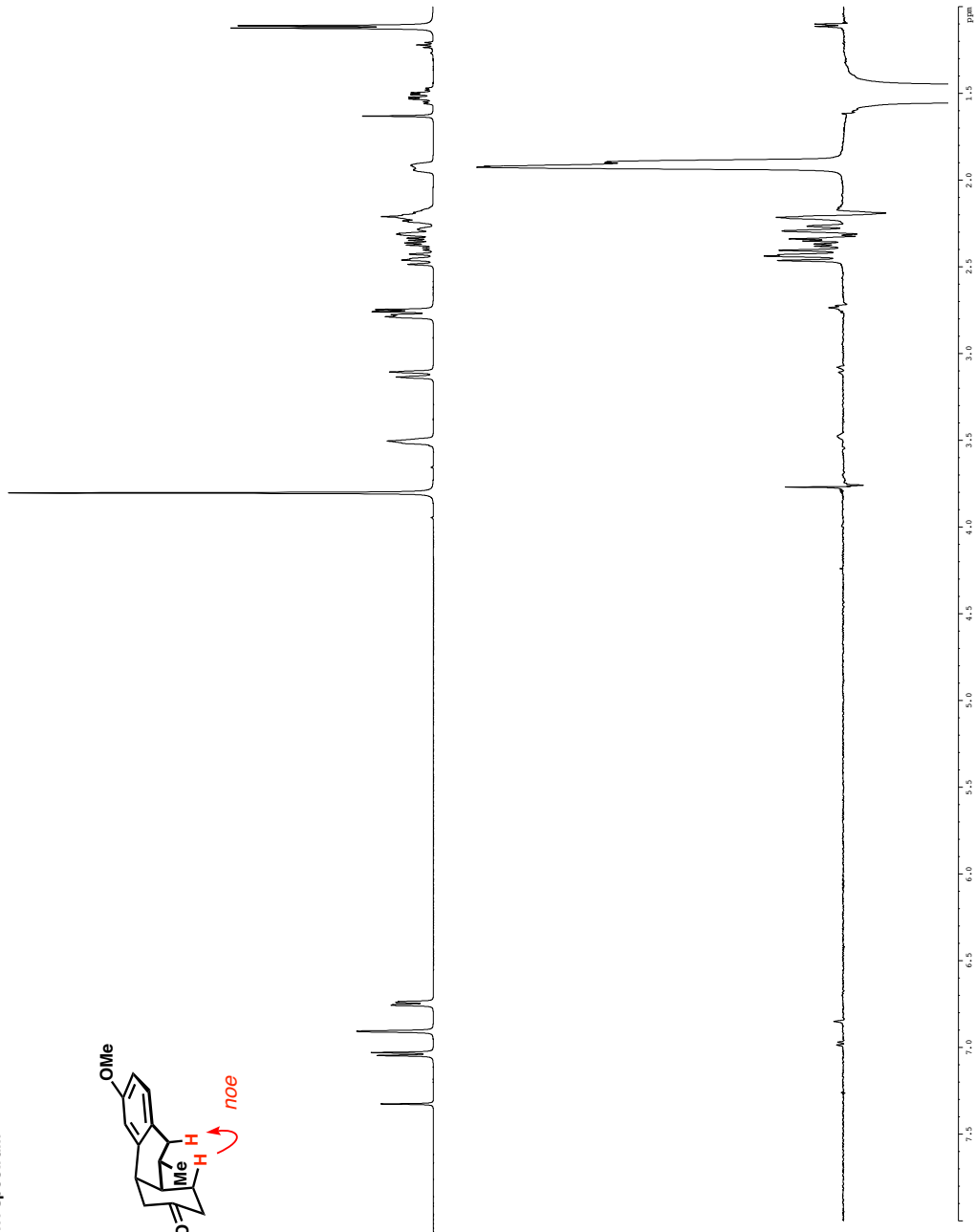
Current Data Parameters  
USER PCT3\_298\_1107  
INPRO 1  
PROCNO 1  
  
F2 - Acquisition Parameters  
Time 21.15.36  
Date\_ 20100726  
PROBHD 5 mm CPCLP1H  
PULPROG zgpg30  
SOLVENT CDCl3  
DS 2  
AQ 617.82 Hz  
FIDRES 0.220026 Hz  
AQ 1.9998451 sec  
RG 327.50 Hz  
INSTRUM spect  
TE 300.2 K  
TD 65536  
SFO1 500.136299 MHz  
SCANS 1024  
PROCNO 1  
P1 7.50 usec  
PC 4.00 usec  
CY 1000  
C1 15.00 usec  
C2 15.00 usec  
C3 15.00 usec  
P1 5002.20 Hz  
P2 -500.22 Hz  
P3 5002.20 Hz  
PC 241.33423 Hz  
PC 241.33423 Hz



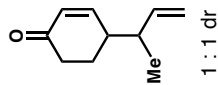
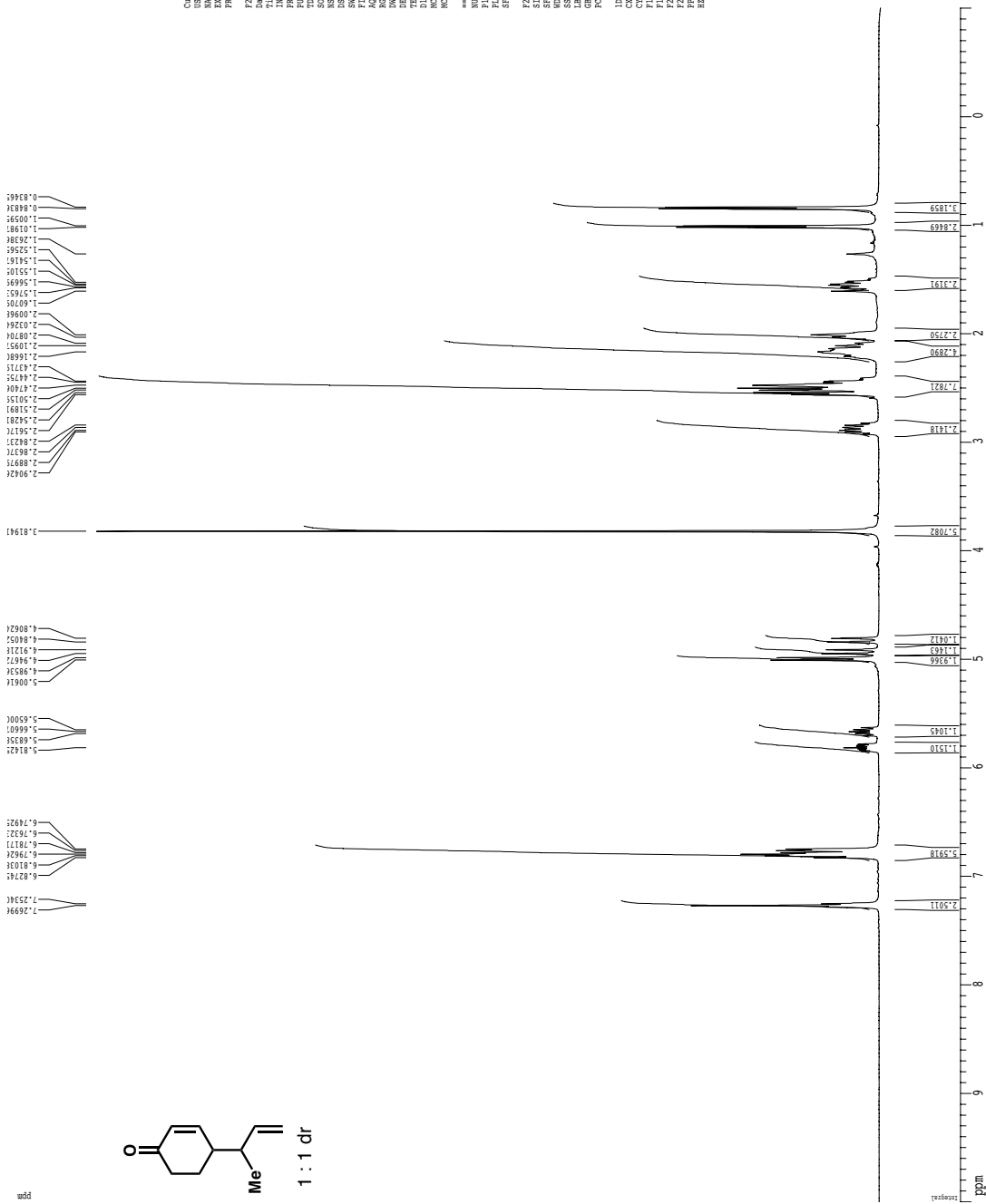
**<sup>1</sup>H spectrum**



Current Data Parameters  
USER toonen  
EXPNO p03\_298\_110\_7  
PROCNO 1  
F2 - Acquisition Parameters  
Date\_ 2015\_16  
Time 15:26  
PROBHD 5 mm CPYX 1H-  
PULPROG zgpg30  
SOLVENT CDCl3  
DS 2  
AQ 0.12000000 sec  
FIDRES 0.220026 Hz  
AQ 1.9998451 sec  
RG 62.400 usec  
DM 0.12000000 sec  
TE 300.2 K  
NOREST 0.00000000 sec  
MCORR 0.01000000 sec  
===== CHANNEL f1 =====  
NUC1 1H 7.50 usec  
P1 2.00 usec  
PL1 0.00 dB  
SFO1 500.2213015 MHz  
F2 - Processing parameters  
SI 65536 usec  
WDW EM  
SSB 0.30 Hz  
GB 0  
PC 4.00  
===== CHANNEL f2 =====  
CX MRB plot param  
CT 22.80 cm  
C1 15.00 cm  
P1 500.20 MHz  
F2 -500.22 MHz  
SFO2 241.31823 MHz

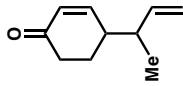
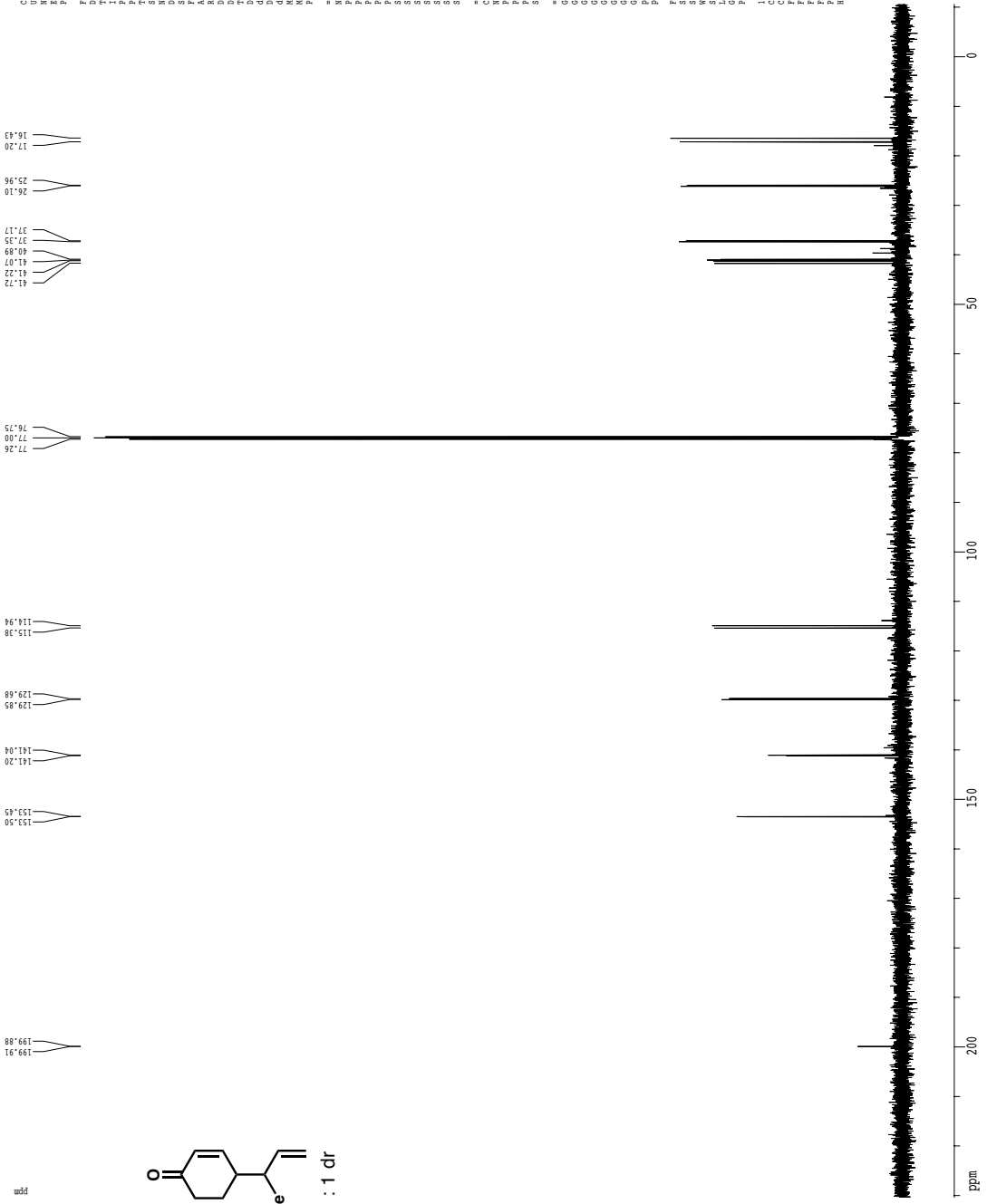


1H spectrum



Current Data Parameters  
 NAME Pct4.018 isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2010109  
 Time 12.00  
 INSTRUM cty5000  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 0  
 DS 2  
 SFO1 8012.820 Hz  
 FIDRES 0.250024 Hz  
 RG 1.591911 sec  
 ACQ 5  
 DM 62.400 usec  
 DE 0.000000 sec  
 TE 298.2 K  
 D1 0.1000000 sec  
 DCREST 0.0000000 sec  
 ACQRK 0.1330000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.2201500 MHz  
 WOP 0  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MRB p1cr parameters  
 CX 22.80 cm  
 CT 15.00 cm  
 F1 500.2201500 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 F4 0.0000000 MHz  
 HSCN 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling

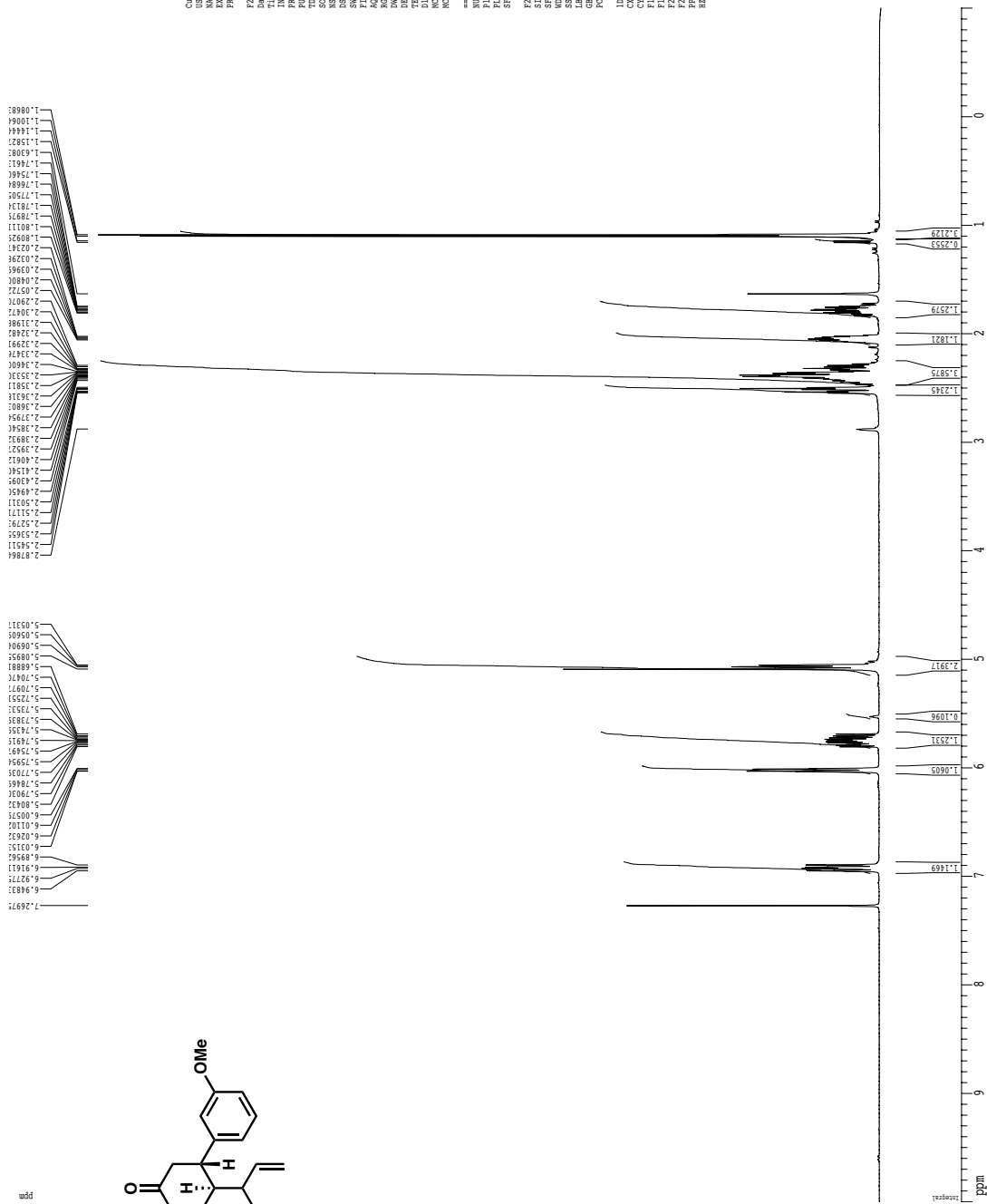
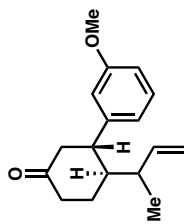


1 : 1 dr

```
Current Data Parameters
NAME          pect-115_77-80
EXPNO         2
PROCNO        1
F2 - Acquisition Parameters
Date_         20100814
Time          13:45
INSTRUM       cryo500
PROBHD        sbbbcpr13bmz1
PULPROG       zgpg30
TD             65536
SOLVENT       CDCl3
NS            14
DS            2
SWH           30000.000 Hz
FIDRES        1.4013340 Hz
AQ            7986.2
RG            656.00 usec
DE            6.500 usec
TE            0.29830 K
NUC1           13C
NUC2           13C
NUC3           13C
NUC4           13C
NUC5           13C
NUC6           13C
NUC7           13C
NUC8           13C
NUC9           13C
NUC10          13C
===== CHANNEL f1 =====
NUC1           13C
P1            16.55 usec
PL1           -1.00 dB
PL2           -1.00 dB
PL3           -1.00 dB
PL4           -1.00 dB
PL5           -1.00 dB
PL6           -1.00 dB
PL7           -1.00 dB
PL8           -1.00 dB
PL9           -1.00 dB
PL10           -1.00 dB
PL11           -1.00 dB
PL12           -1.00 dB
PL13           -1.00 dB
PL14           -1.00 dB
PL15           -1.00 dB
PL16           -1.00 dB
PL17           -1.00 dB
PL18           -1.00 dB
PL19           -1.00 dB
PL20           -1.00 dB
===== CHANNEL f2 =====
CPDPRG2       waltz16
NUC1           13C
NUC2           13C
PL1           100.00 usec
PL2           100.00 usec
PL3           100.00 usec
PL4           100.00 usec
PL5           100.00 usec
PL6           100.00 usec
PL7           100.00 usec
PL8           100.00 usec
PL9           100.00 usec
PL10           100.00 usec
PL11           100.00 usec
PL12           100.00 usec
PL13           100.00 usec
PL14           100.00 usec
PL15           100.00 usec
PL16           100.00 usec
PL17           100.00 usec
PL18           100.00 usec
PL19           100.00 usec
PL20           100.00 usec
===== CHANNEL CHANNEL =====
GRANU         65536
SINFT         1.00
SF            125.7604866 MHz
WDW           EM
SSB            0
GB            0.00 Hz
PC            0.00 Hz
SC              0
===== Processing parameters =====
SI            65536
SF            125.7604866 MHz
WDW           EM
SSB            0
GB            0.00 Hz
PC            0.00 Hz
SC              0
===== IDMG plot parameters =====
SI            65536
SF            125.7604866 MHz
WDW           EM
SSB            0
GB            0.00 Hz
PC            0.00 Hz
SC              0
===== =====
SI            65536
SF            125.7604866 MHz
WDW           EM
SSB            0
GB            0.00 Hz
PC            0.00 Hz
SC              0
```



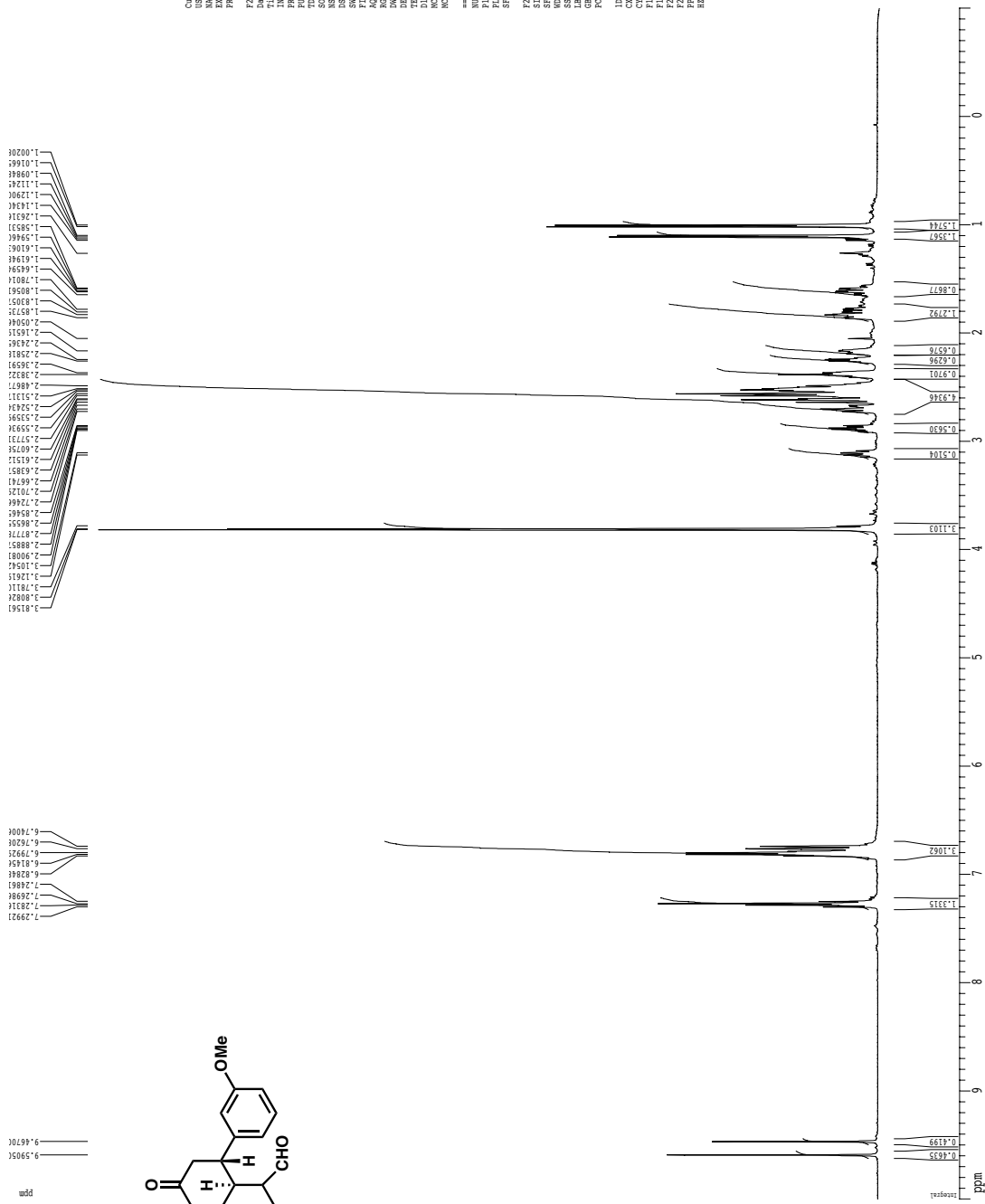
1H spectrum



Current Data Parameters  
NAME Pct4.113\_72-80  
EXPNO 1  
PROCNO 1  
Date\_ 20150117  
Time 17:56:05  
INSTRUM cty6500  
PROBHD 5 mm CPXI 1H-  
PULPROG zgpg30  
TD 32768  
SOLVENT CDCl3  
NS 2  
DS 2  
SWH 8012.820 Hz  
FIDRES 0.250024 Hz  
AQ 1.1991641 sec  
RG 3.7  
DM 62.400 usec  
DE 0.000000 sec  
TE 296.2 K  
D1 0.1000000 sec  
ACQST 0.0000000 sec  
SOLVR 0.1130000 sec  
===== CHANNEL f1 =====  
NUC1 1H  
P1 7.45 usec  
PL1 0 dB  
PL2 1.60 dB  
PL3 0 dB  
SFO1 500.225015 MHz  
F2 - Processing parameters  
SI 6534  
SF 500.225015 MHz  
WDW EM  
SSB 0  
GB 0  
PC 4.00  
LO MR parameters  
CX 22.80 cm  
CY 15.00 cm  
CZ 15.00 cm  
FX 500.225015 MHz  
FY 500.225015 MHz  
FZ -1.000 ppm  
TX 500.225015 MHz  
TY 500.225015 MHz  
FIDRES 0.250024 Hz  
AQ 1.1991641 sec  
RG 3.7  
DM 62.400 usec  
DE 0.000000 sec  
TE 296.2 K  
D1 0.1000000 sec  
ACQST 0.0000000 sec  
SOLVR 0.1130000 sec

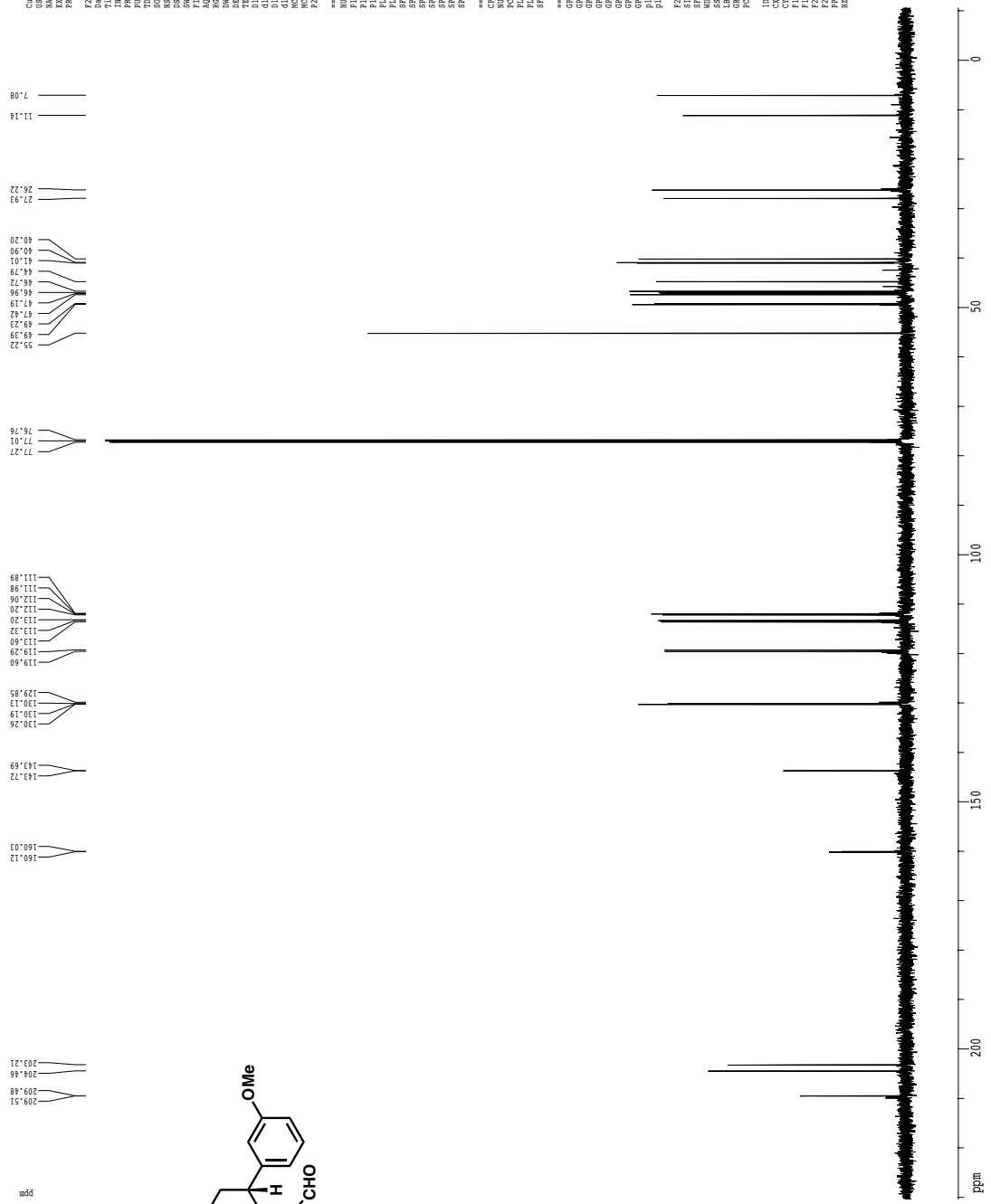


1H spectrum



Current Data Parameters  
 NAME Pct4.014\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20101218  
 Time 11:29:07  
 INSTRUM cty5000  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 802.820 Hz  
 FIDRES 0.250204 Hz  
 AQ 1.199191 sec  
 RG 4.5  
 DQ 62.400 usec  
 DE 694.0 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 DCREST 0.0000000 sec  
 ACQBR 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 F41 500.225015 MHz  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 S1 6536  
 SI 500.220000 MHz  
 KW 64  
 SSB 0  
 CB 0.00 Hz  
 PC 4.00  
 LD MR parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 FX 500.220000 MHz  
 FY 500.220000 MHz  
 FZ -1.000 ppm  
 GY 2500.22 Hz/cm  
 HZ 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling

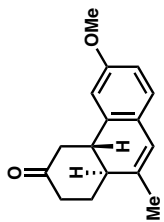
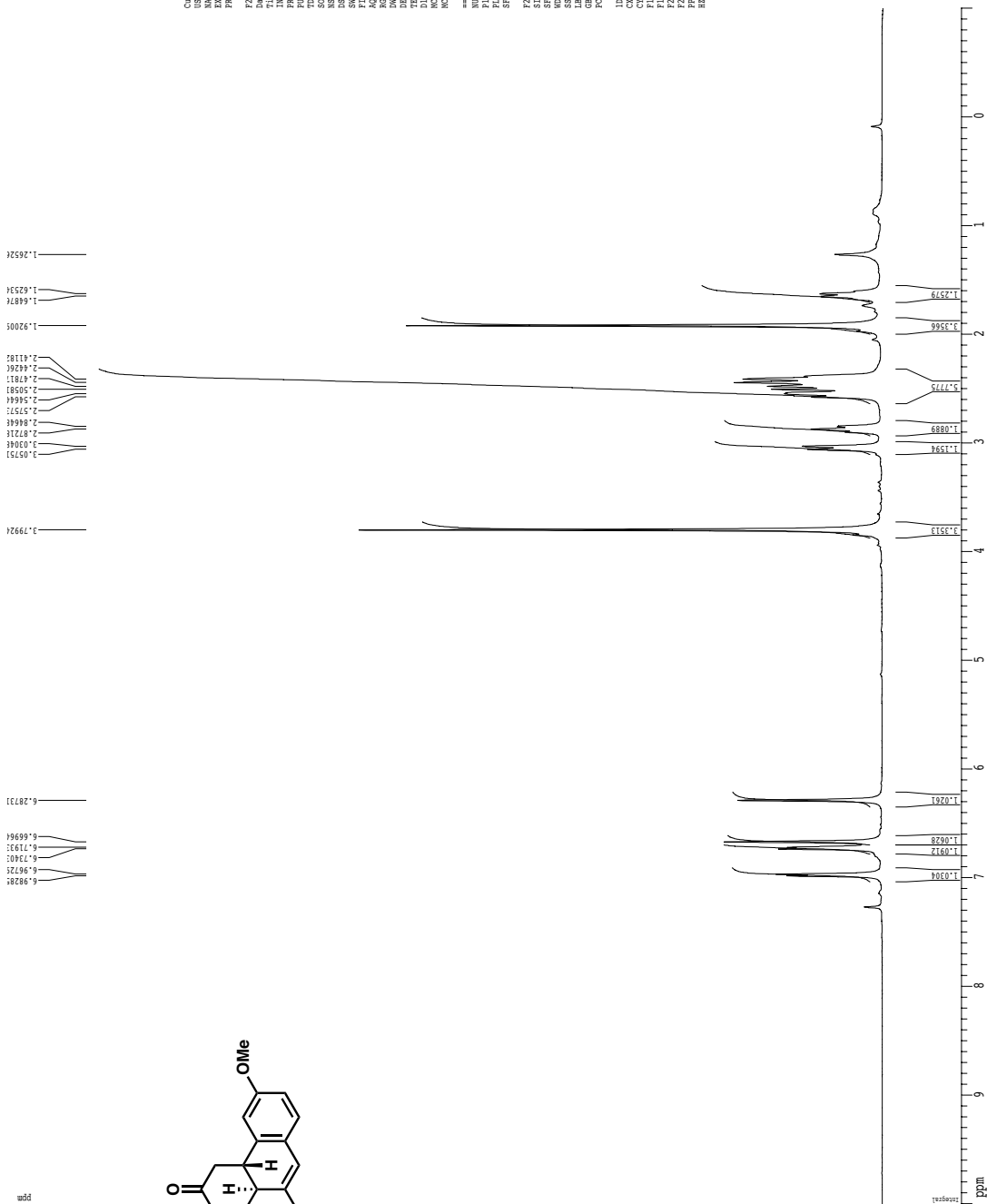


```

Current Data Parameters
NAME          pect-014_56014e
EXPNO         2
PROCNO        1
F2 - Acquisition Parameters
Date_         2011.07.14
Time          11.49
INSTRUM       cryo500
PROBHD        5mmCPYX13
PULPROG       zgpg30
HARDWARE      spect
TD             65536
SOLVENT       CDCl3
NS            2
DS            4
SWH           30303.9338 Hz
FIDRES        1.0813340 Hz
AQ            3849.71
RG            65.00 usec
DE            6.50 usec
TE            0.29830 K
NUC1          13C
NUC2          13C
NUC3          13C
NUC4          13C
NUC5          13C
NUC6          13C
NUC7          13C
NUC8          13C
NUC9          13C
NUC10         13C
NUC11         13C
NUC12         13C
===== CHANNEL F1 =====
NUC1          13C
P1            16.55 usec
PL1           0.00 dB
SFO1          125.7628500 MHz
===== CHANNEL F2 =====
NAME          val123.6
CPDPRG2       waltz16
PCPD2         100.00 usec
PL2           1.00 dB
PL3           1.00 dB
SFO2          500.2253011 MHz
===== CHANNEL CHANNEL =====
GRAMSI        STXZ.100
GRAMP         STXZ.100
GRAMP         0.00 A
GRAPR         0.00 A
GRASQ         0.00 A
GRASH         30.00 A
GRASL         30.00 A
GRASR         50.00 A
GRASE         1000.00 usec
===== Processing parameters =====
SI            65536
SF            125.7604386 MHz
WDW           EM
SSB           0
LB            2.00 Hz
GB            0.00 Hz
PC            2.00
===== ID MS parameters =====
CT           15.45 cm
PI           230.84 ppm
F2           -10.536 ppm
F1           -1.22528 Hz
NUC1         13C
NUC2         13C
NUC3         13C
NUC4         13C
NUC5         13C
NUC6         13C
NUC7         13C
NUC8         13C
NUC9         13C
NUC10        13C

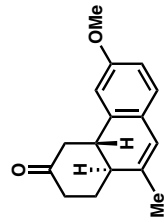
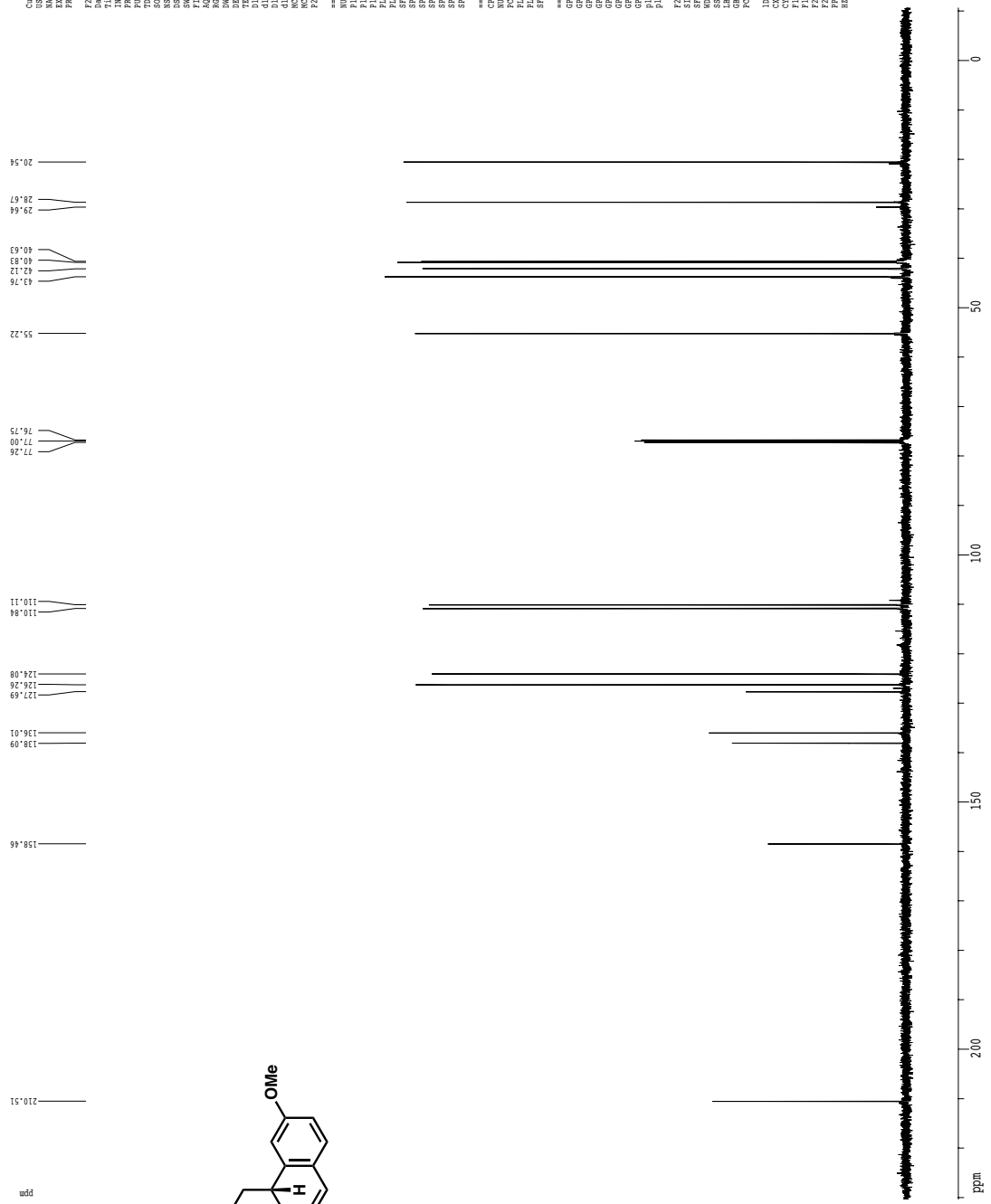
```

1H spectrum



Current Data Parameters  
 Name: Pct4-013\_isolate  
 EXPNO: 1  
 PROCNO: 1  
 F2 - Acquisition Parameters  
 Date\_: 2011220  
 Time: 11.58  
 INSTRUM: cryo500  
 PROBRD: 5 mm CPXI 1H-  
 PULPROG: zgpg30  
 TD: 32768  
 SOLVENT: CDCl3  
 NS: 2  
 DS: 4  
 SWH: 8012.820 Hz  
 FIDRES: 0.250026 Hz  
 AQ: 1.999984 sec  
 RG: 4.5  
 DQ: 62.400 usec  
 DE: 298.0 usec  
 TE: 300.2 K  
 D1: 0.1000000 sec  
 ACQBST: 0.0000000 sec  
 ACQBRK: 0.1000000 sec  
 ===== CHANNEL f1 =====  
 NUC1: 1H  
 P1: 12.00 usec  
 PL1: 0.00 dB  
 F1: 500.1363015 MHz  
 SFO1: 500.1363015 MHz  
 F2 - Processing parameters  
 SI: 65536  
 SF: 500.1363015 MHz  
 WOP: 0.00 usec  
 SSB: 0  
 GB: 0.00 Hz  
 PC: 4.00  
 LO MRB parameters  
 CX: 22.80 cm  
 CY: 10.00 cm  
 CZ: 10.00 cm  
 FFX: 500.1363015 MHz  
 FFY: 500.2010000 MHz  
 FFP: -1.000 ppm  
 FZ: -500.22 Hz  
 GAMMA: 180.0000000 deg/cm  
 HPCON: 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



Current Data Parameters  
 NAME pect-013\_150Date  
 EXPNO 2  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2015.12.08  
 Time 13:14  
 INSTRUM cryo500  
 PULPROG zgpg30  
 PROCPRG zgpg30  
 F2 - Processing parameters  
 SI 65536  
 SF 125.7604351 MHz  
 DS 4  
 LB 1.00 Hz  
 GB 0  
 PC 2.00  
 ID NONE plot parameters  
 CT 10.00 cm  
 P1P 230.32 dB  
 P2P -10.888 dB  
 F2 1331.75 Hz  
 FWHM 1329.8832 Hz/cm

===== CHANNEL f1 =====  
 F1 16.55 uSAC  
 F11 500.00 uSAC  
 F12 100.00 uSAC  
 F13 200.00 dB  
 F14 -1.00 dB  
 F15 125.7604351 MHz  
 SFO1 2.70 dB  
 SFO2 2.70 dB  
 SFO3 CPMAS,6.5,28.1  
 SFO4 CPMAS,6.5,28.1  
 SFO5F1 0.00 Hz  
 SFO5F2 0.00 Hz

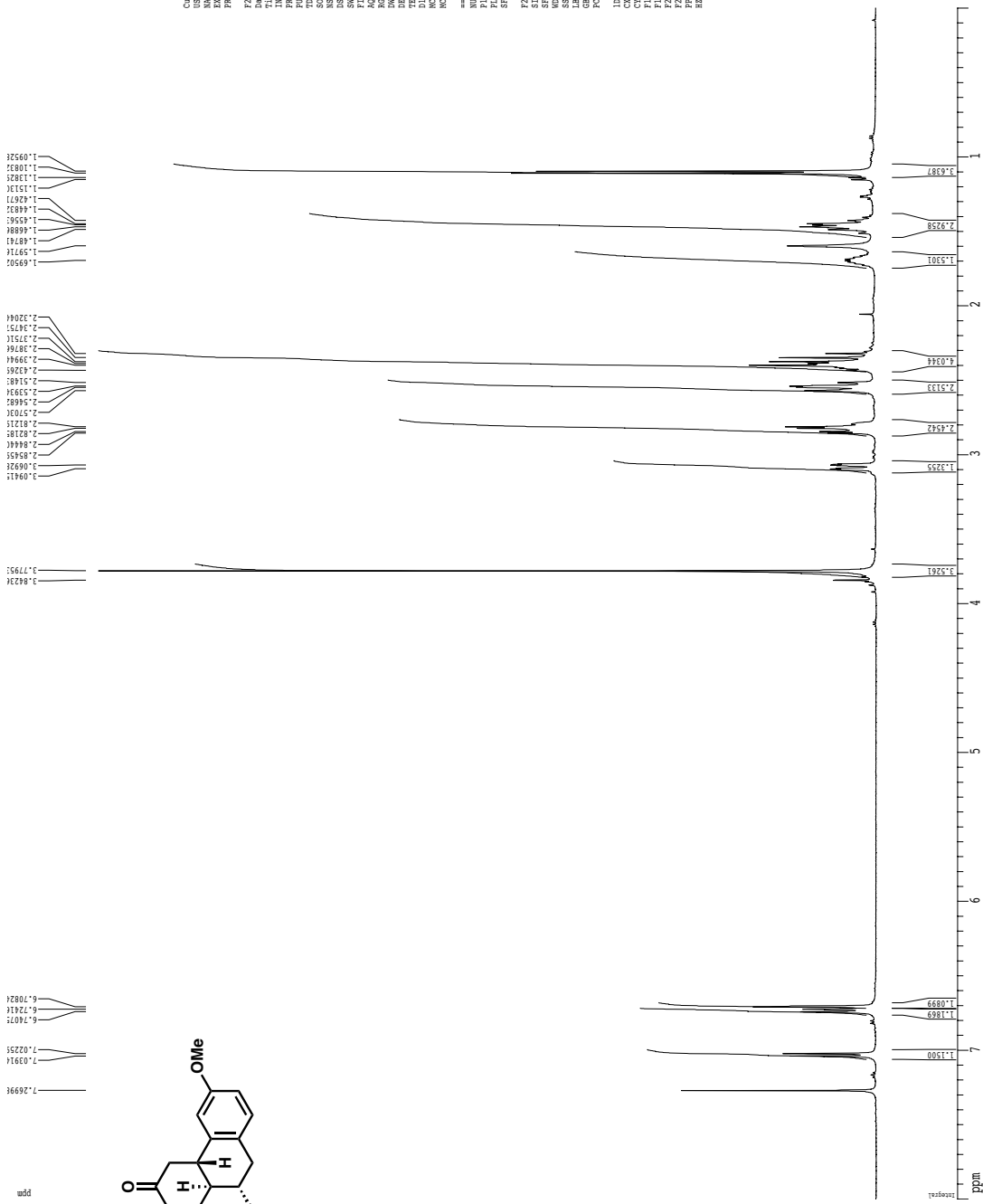
===== CHANNEL f2 =====  
 CPMAS2 waltz16  
 PCPD2 100.00 uSAC  
 P1 1.00 dB  
 P2 1.00 dB  
 SFO2 500.2225411 MHz

===== CHANNEL CHANNEL =====  
 GRANU 518K,100  
 SINE,100  
 GRAN2 518K,100  
 SINE,100  
 GRAN3 518K,100  
 SINE,100  
 GRAN4 518K,100  
 SINE,100  
 GRAN5 518K,100  
 SINE,100  
 GRAN6 518K,100  
 SINE,100  
 GRAN7 518K,100  
 SINE,100  
 GRAN8 518K,100  
 SINE,100  
 GRAN9 518K,100  
 SINE,100  
 GRAN10 518K,100  
 SINE,100  
 GRAN11 518K,100  
 SINE,100  
 GRAN12 518K,100  
 SINE,100  
 GRAN13 518K,100  
 SINE,100  
 GRAN14 518K,100  
 SINE,100  
 GRAN15 518K,100  
 SINE,100

===== CHANNEL CHANNEL =====  
 CPMAS2 waltz16  
 PCPD2 100.00 uSAC  
 P1 1.00 dB  
 P2 1.00 dB  
 SFO2 500.2225411 MHz

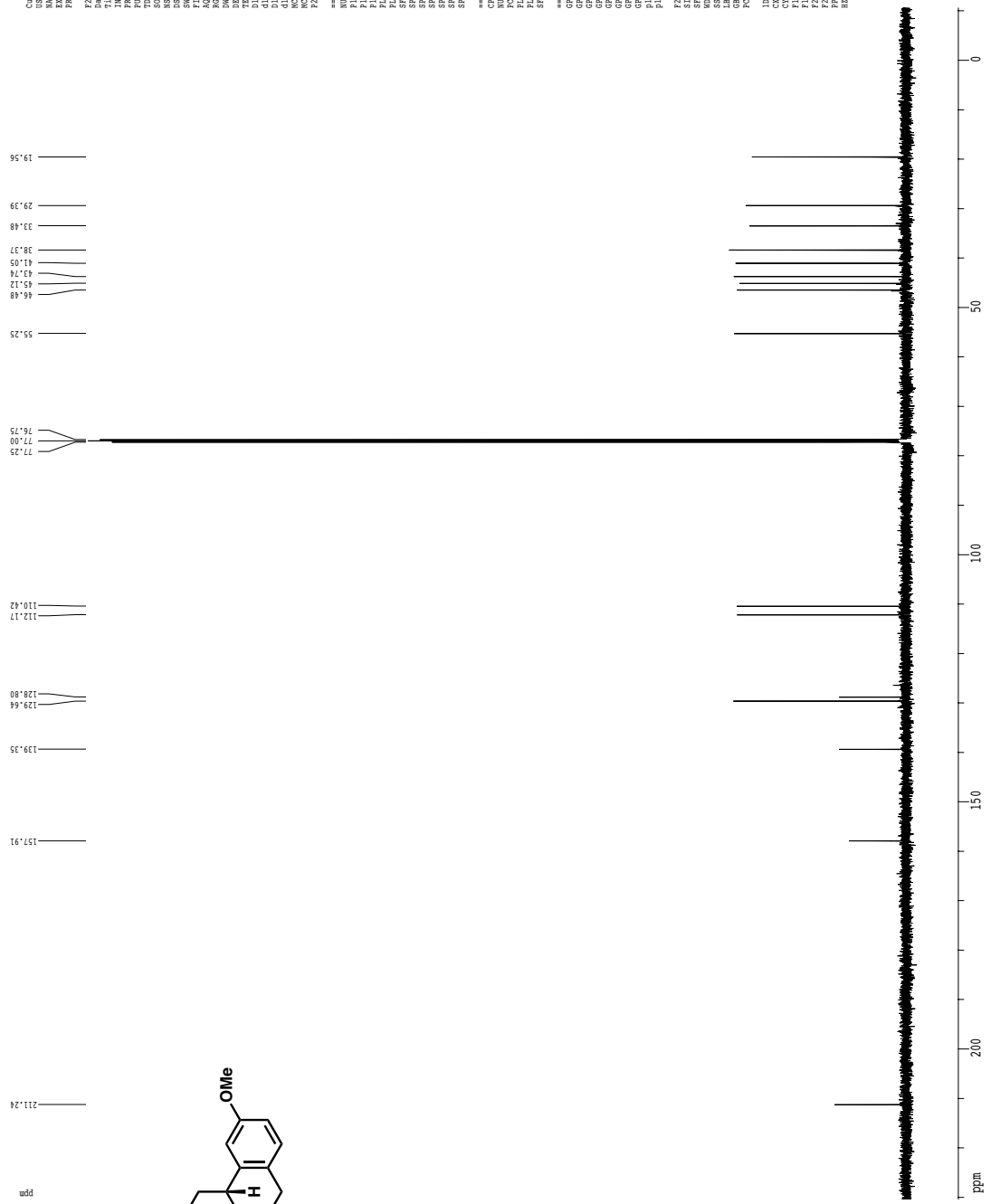
===== CHANNEL CHANNEL =====  
 GRANU 518K,100  
 SINE,100  
 GRAN2 518K,100  
 SINE,100  
 GRAN3 518K,100  
 SINE,100  
 GRAN4 518K,100  
 SINE,100  
 GRAN5 518K,100  
 SINE,100  
 GRAN6 518K,100  
 SINE,100  
 GRAN7 518K,100  
 SINE,100  
 GRAN8 518K,100  
 SINE,100  
 GRAN9 518K,100  
 SINE,100  
 GRAN10 518K,100  
 SINE,100  
 GRAN11 518K,100  
 SINE,100  
 GRAN12 518K,100  
 SINE,100  
 GRAN13 518K,100  
 SINE,100  
 GRAN14 518K,100  
 SINE,100  
 GRAN15 518K,100  
 SINE,100

1H spectrum



Current Data Parameters  
 NAME Pct4.013\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2011215  
 Time 11:00:00  
 INSTRUM ctyes500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250024 Hz  
 AQ 1.199841 sec  
 RG 317  
 EQ 1.7  
 DM 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 DC 0.0000000 sec  
 ACQRES 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.2201000 MHz  
 WOP 30.0000000 MHz  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 FX 400.75 Hz  
 FY 400.75 Hz  
 FZ 0.000 ppm  
 TX 0.00 Hz  
 TY 0.00 Hz  
 TZ 173.5193 Hz/cm  
 HZCN 173.5193 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling

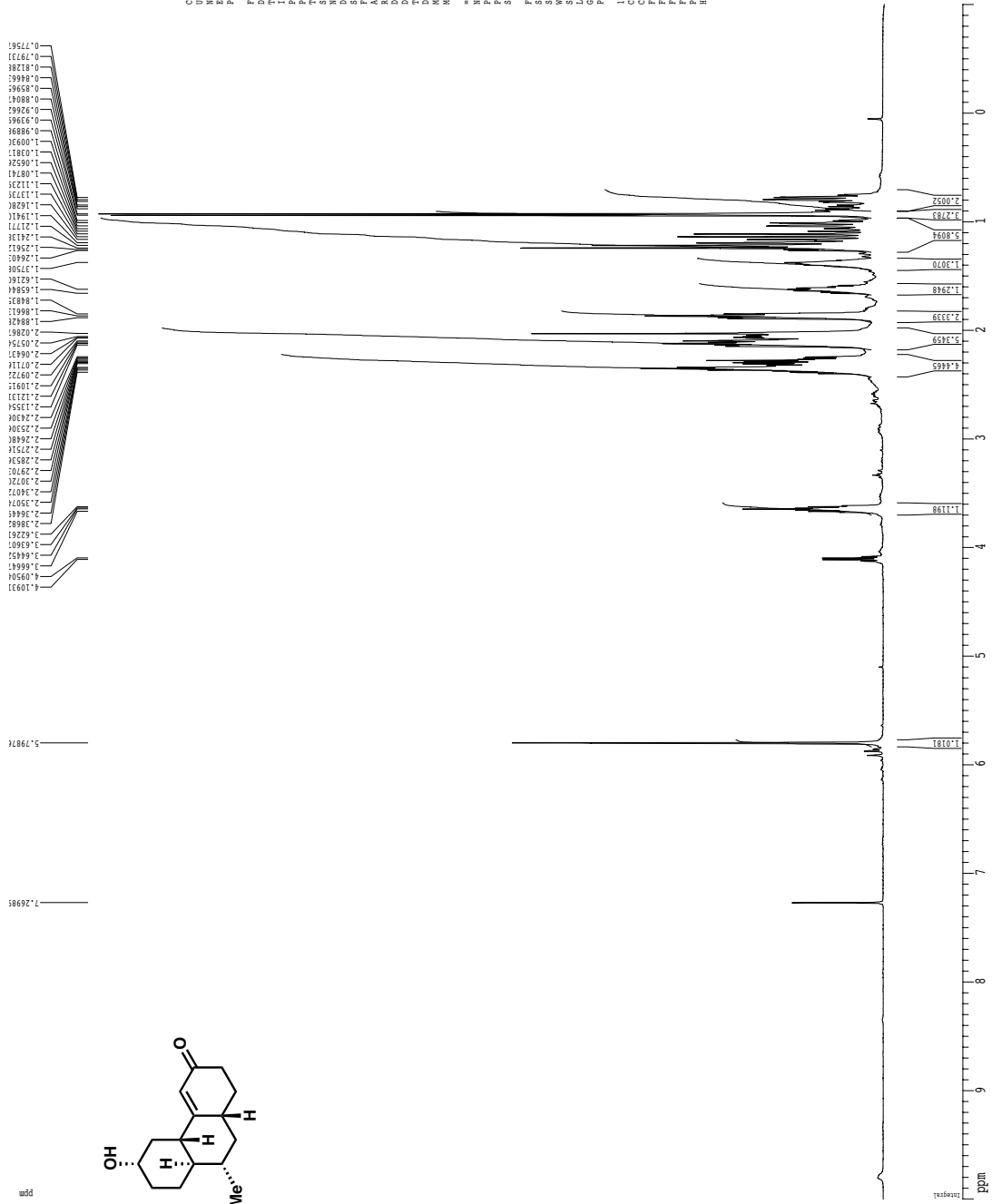
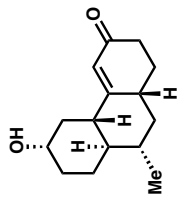


```

Current Data Parameters
NAME          pct4.013_5601a1e
EXPNO         2
PROCNO        1
F2 - Acquisition Parameters
Date_         2011.07.28
Time          12.08
INSTRUM       cryo500
PROBHD        5mm cryo500
PULPROG       zgpg30
TD            65536
SOLVENT       CDCl3
NS           402
DS           2
SWH           30000.000 Hz
FIDRES       1.0813340 Hz
AQ           13.004
RG           652.00
DE           6.500 usec
TE           300.2 K
NUC1          13C
NUC2          13C
NUC3          13C
NUC4          13C
NUC5          13C
NUC6          13C
NUC7          13C
NUC8          13C
NUC9          13C
NUC10         13C
===== CHANNEL f1 =====
NUC1          13C
P1           16.55 usec
PL1          0.00 dB
PC1          100.00 usec
P2           20.00 usec
PL2          0.00 dB
PC2          100.00 usec
P3           20.00 usec
PL3          0.00 dB
PC3          100.00 usec
P4           125.7922282 usec
PL4          -1.00 dB
PC4          100.00 usec
P5           2.70 usec
PL5          0.00 dB
PC5          100.00 usec
===== CHANNEL f2 =====
NAME          v13c2.6
EXPNO         1
PROCNO        1
F2 - Processing parameters
SI           65536
SF           125.7604391 MHz
WDW           EM
SSB           0
GB           0
PC           1.00 usec
AQ           2.00
CT           15.45 cm
C1           230.380 ppm
F1F2         -10.540 ppm
F2          -1225.74 Hz
F2RES        1329.08020 Hz/cm
  
```



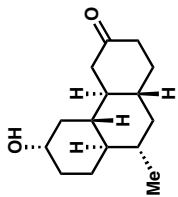
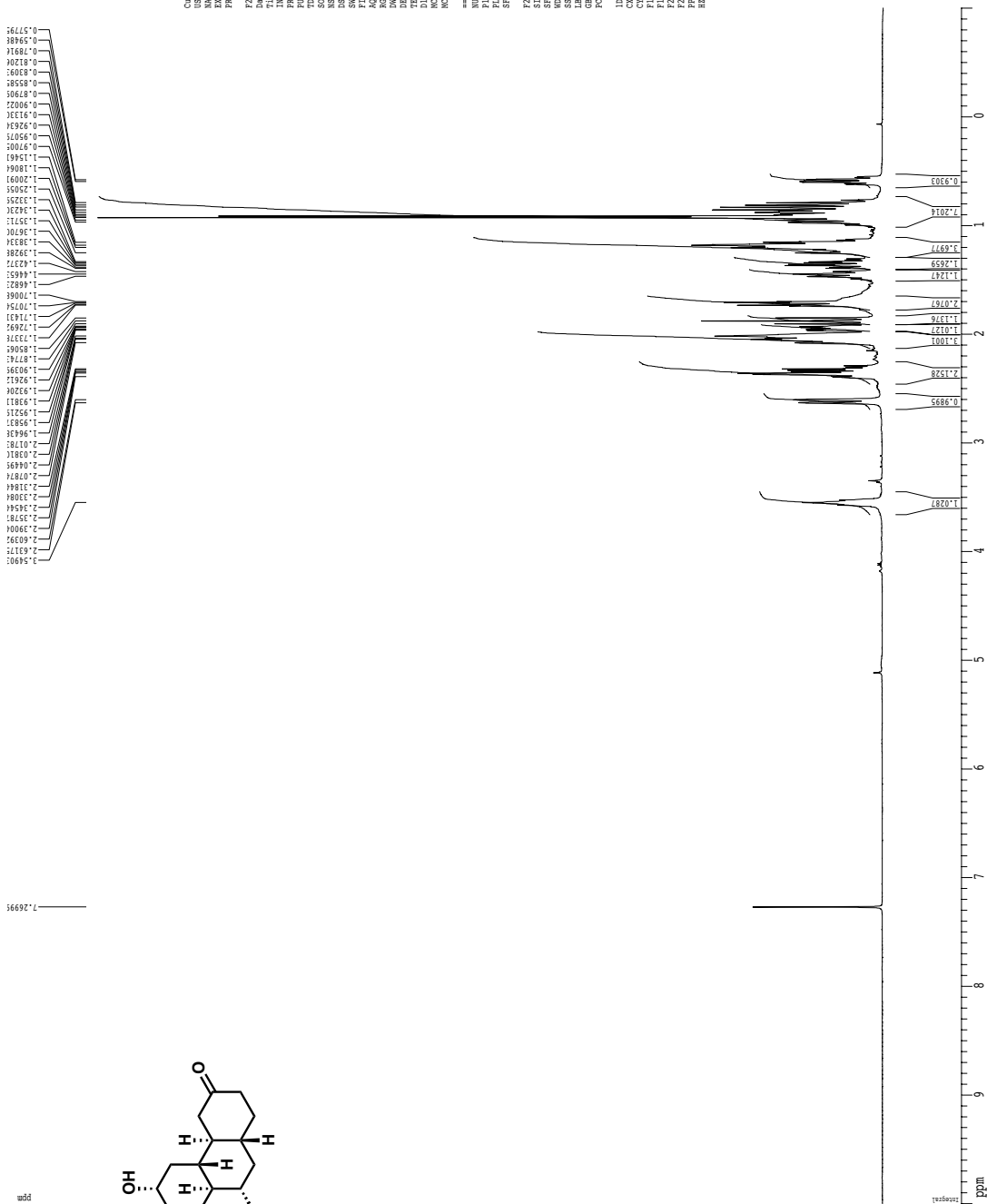
1H spectrum



Current Data Parameters  
 NAME Pct4\_024\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20101112  
 Time 09:11:12  
 INSTRUM cty6500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250026 Hz  
 AQ 1.199846 sec  
 RG 4.5  
 INEPT 0  
 DE 62.400 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 DCOffset 0.0000000 sec  
 ACQNR 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.45 usec  
 PL1 0 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 653.6  
 SF 500.225015 MHz  
 WOP 30.000000 MHz  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MRB p1dr parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 F4 241.33423 Hz/cm  
 HSCN 241.33423 Hz/cm



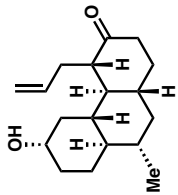
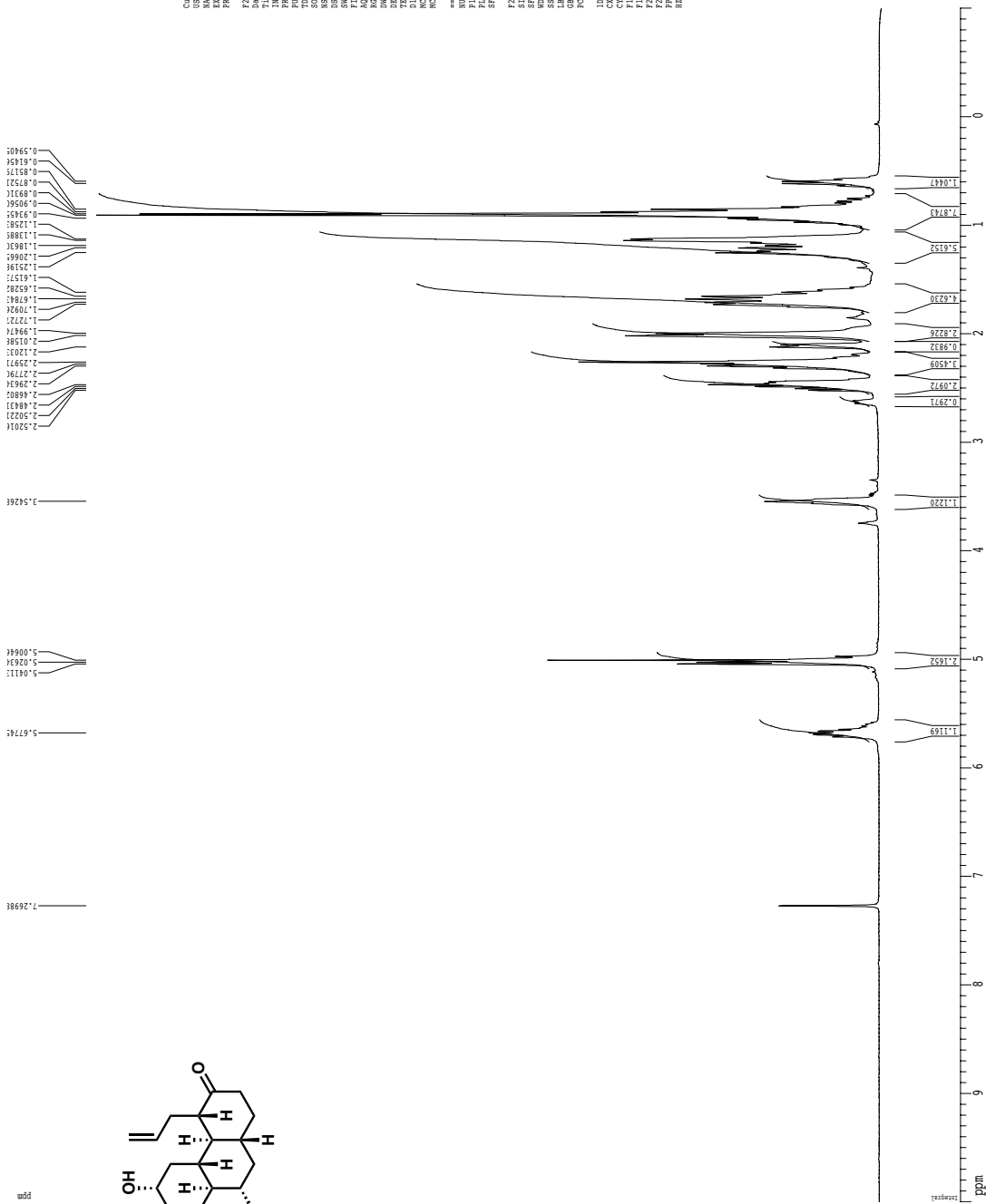
1H spectrum



Current Data Parameters  
 NAME Pct4.083\_Metone  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20101225  
 Time 11:00:00  
 INSTRUM cty6500  
 PROBRD 5 mm CPCL H-  
 PULPROG zgpg30  
 TD 65536  
 SFO1 500.220161 MHz  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250026 Hz  
 AQ 1.1900000 sec  
 RG 111  
 DQ 62.400 usec  
 W 0.640 usec  
 DE 0.0000000 sec  
 DC 0.0000000 sec  
 MCHST 0.11000000 sec  
 ACQPRG 0.11000000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 14.50 usec  
 PL1 0.00 dB  
 SFO1 500.220161 MHz  
 F2 - Processing parameters  
 SI 65536  
 SF 500.220161 MHz  
 WDW RM  
 SSB 0  
 GB 0  
 PC 4  
 CD 0.00 usec  
 SC 22.80 cm  
 CT 15.00 cm  
 F1 500.220161 MHz  
 F2 -1.000 ppm  
 F3 -500.220161 MHz  
 F4 100.620322 MHz  
 GC01 241.33423 Hz/cm

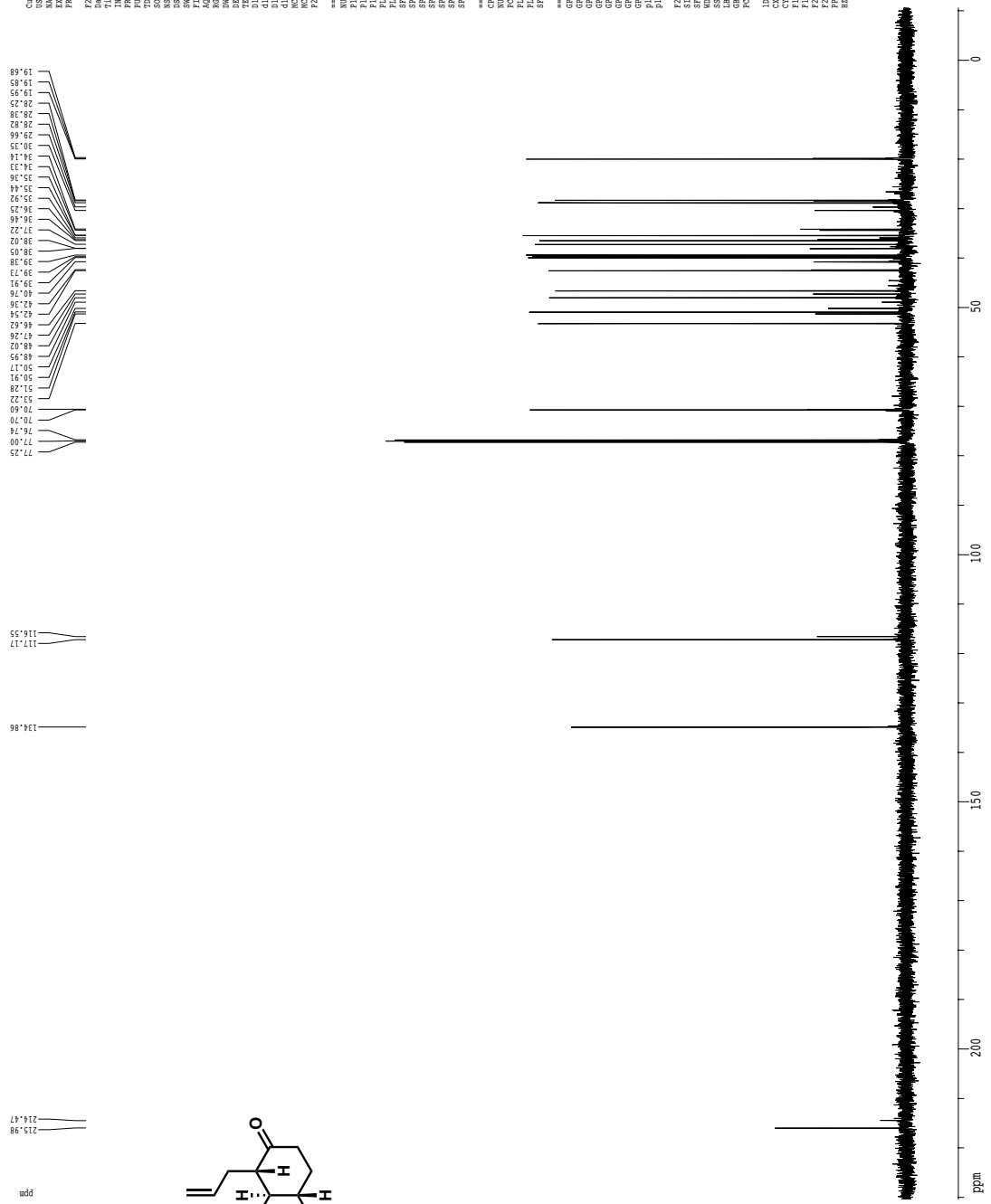


1H spectrum

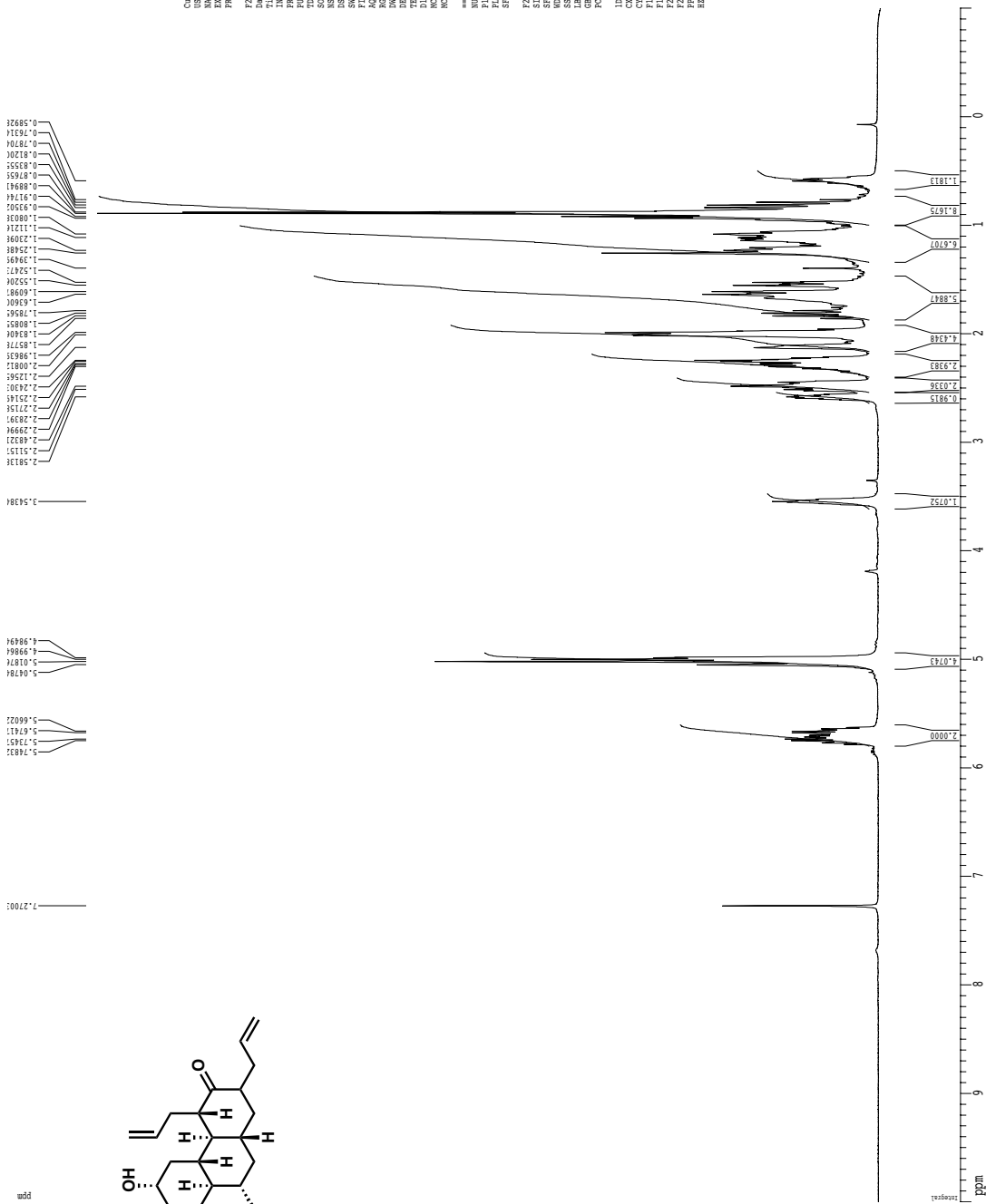


Current Data Parameters  
 NAME pct4.083.monally1  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20080824  
 Time 23:01  
 INSTRUM crys000  
 PULPROG zgpg30  
 TD 32768  
 SFO1 500.136095 MHz  
 F2 500.136095 MHz  
 F1 125.7615375 MHz  
 AQ 0.0150000 sec  
 RG 62.400 usec  
 SWH 7.937500 MHz  
 DE 6.00 usec  
 TE 300.2 K  
 D1 0.1000000 sec  
 D2 0.0000000 sec  
 D3 0.0000000 sec  
 D4 0.0000000 sec  
 D5 0.0000000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.00 usec  
 PL1 0.00 dB  
 SFO1 500.136095 MHz  
 F2 - Processing parameters  
 SI 65536  
 SF 500.136095 MHz  
 WDW EM  
 GB 0  
 CB 0  
 SC 0  
 PC 4.00  
 ID\_NMR plot parameters  
 CT 22.80 cm  
 FID 10.000 usec  
 F1 5002.20 Hz  
 F2 5002.20 Hz  
 F3 5002.20 Hz  
 F4 5002.20 Hz  
 F5 5002.20 Hz  
 F6 5002.20 Hz  
 F7 5002.20 Hz  
 F8 5002.20 Hz  
 F9 5002.20 Hz  
 F10 5002.20 Hz  
 F11 5002.20 Hz  
 F12 5002.20 Hz  
 F13 5002.20 Hz  
 F14 5002.20 Hz  
 F15 5002.20 Hz  
 F16 5002.20 Hz  
 F17 5002.20 Hz  
 F18 5002.20 Hz  
 F19 5002.20 Hz  
 F20 5002.20 Hz  
 F21 5002.20 Hz  
 F22 5002.20 Hz  
 F23 5002.20 Hz  
 F24 5002.20 Hz  
 F25 5002.20 Hz  
 F26 5002.20 Hz  
 F27 5002.20 Hz  
 F28 5002.20 Hz  
 F29 5002.20 Hz  
 F30 5002.20 Hz  
 F31 5002.20 Hz  
 F32 5002.20 Hz  
 F33 5002.20 Hz  
 F34 5002.20 Hz  
 F35 5002.20 Hz  
 F36 5002.20 Hz  
 F37 5002.20 Hz  
 F38 5002.20 Hz  
 F39 5002.20 Hz  
 F40 5002.20 Hz  
 F41 5002.20 Hz  
 F42 5002.20 Hz  
 F43 5002.20 Hz  
 F44 5002.20 Hz  
 F45 5002.20 Hz  
 F46 5002.20 Hz  
 F47 5002.20 Hz  
 F48 5002.20 Hz  
 F49 5002.20 Hz  
 F50 5002.20 Hz  
 F51 5002.20 Hz  
 F52 5002.20 Hz  
 F53 5002.20 Hz  
 F54 5002.20 Hz  
 F55 5002.20 Hz  
 F56 5002.20 Hz  
 F57 5002.20 Hz  
 F58 5002.20 Hz  
 F59 5002.20 Hz  
 F60 5002.20 Hz  
 F61 5002.20 Hz  
 F62 5002.20 Hz  
 F63 5002.20 Hz  
 F64 5002.20 Hz  
 F65 5002.20 Hz  
 F66 5002.20 Hz  
 F67 5002.20 Hz  
 F68 5002.20 Hz  
 F69 5002.20 Hz  
 F70 5002.20 Hz  
 F71 5002.20 Hz  
 F72 5002.20 Hz  
 F73 5002.20 Hz  
 F74 5002.20 Hz  
 F75 5002.20 Hz  
 F76 5002.20 Hz  
 F77 5002.20 Hz  
 F78 5002.20 Hz  
 F79 5002.20 Hz  
 F80 5002.20 Hz  
 F81 5002.20 Hz  
 F82 5002.20 Hz  
 F83 5002.20 Hz  
 F84 5002.20 Hz  
 F85 5002.20 Hz  
 F86 5002.20 Hz  
 F87 5002.20 Hz  
 F88 5002.20 Hz  
 F89 5002.20 Hz  
 F90 5002.20 Hz  
 F91 5002.20 Hz  
 F92 5002.20 Hz  
 F93 5002.20 Hz  
 F94 5002.20 Hz  
 F95 5002.20 Hz  
 F96 5002.20 Hz  
 F97 5002.20 Hz  
 F98 5002.20 Hz  
 F99 5002.20 Hz  
 F100 5002.20 Hz

Z-restored spin-echo 13C spectrum with 1H decoupling



1H spectrum

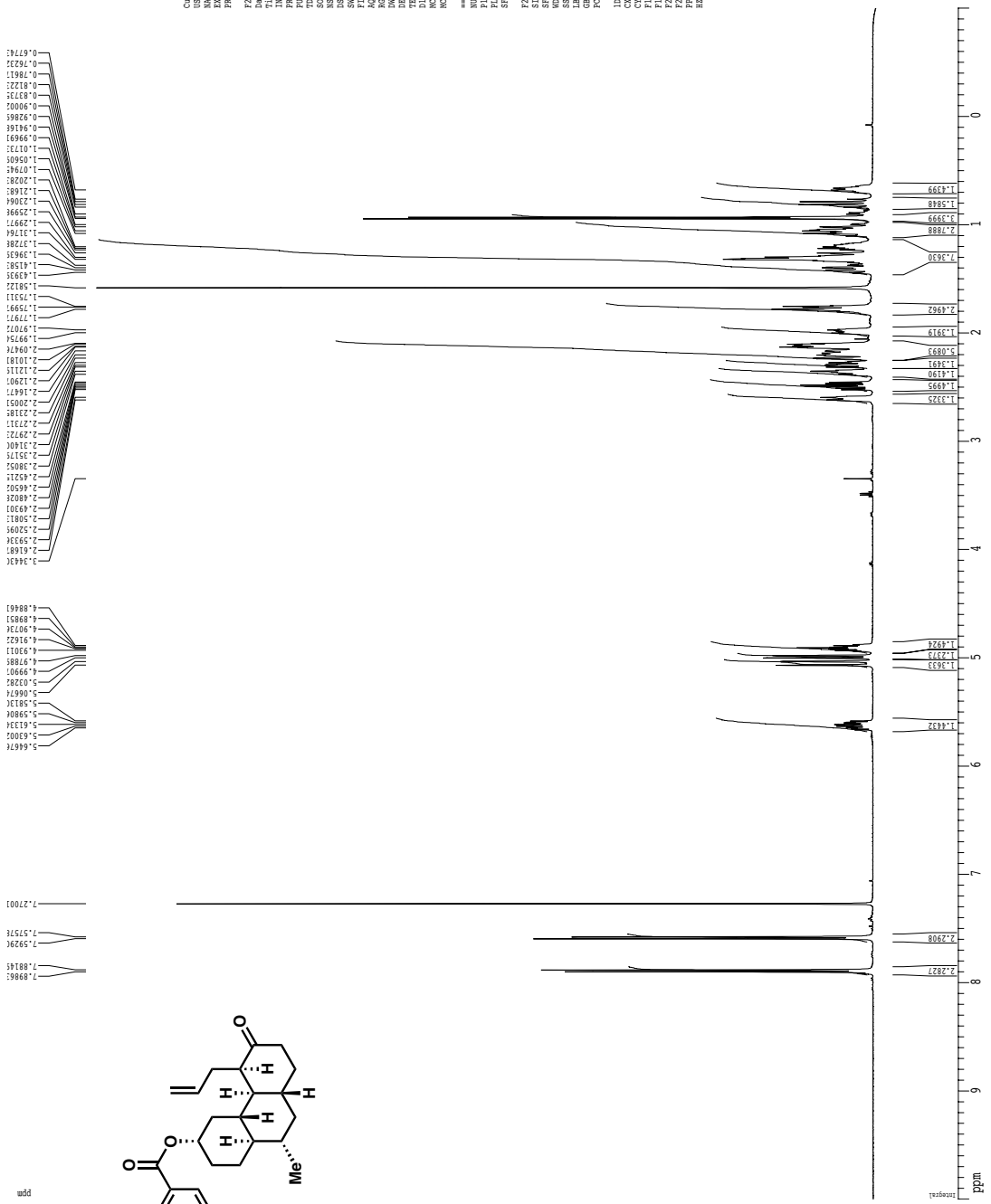


Current Data Parameters  
 NAME pcr4.083\_d(allyl)  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2010225  
 INSTRUM cty6500  
 PROBHD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250204 Hz  
 AQ 1.199181 sec  
 RG 4.5  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 D11 0.0000000 sec  
 ACQRES 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6536  
 SF 500.225015 MHz  
 WOP 20.000000 MHz  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CT 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 HETPC 241.33423 Hz/cm





1H spectrum

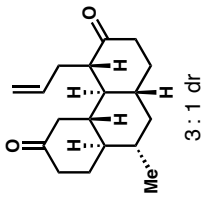
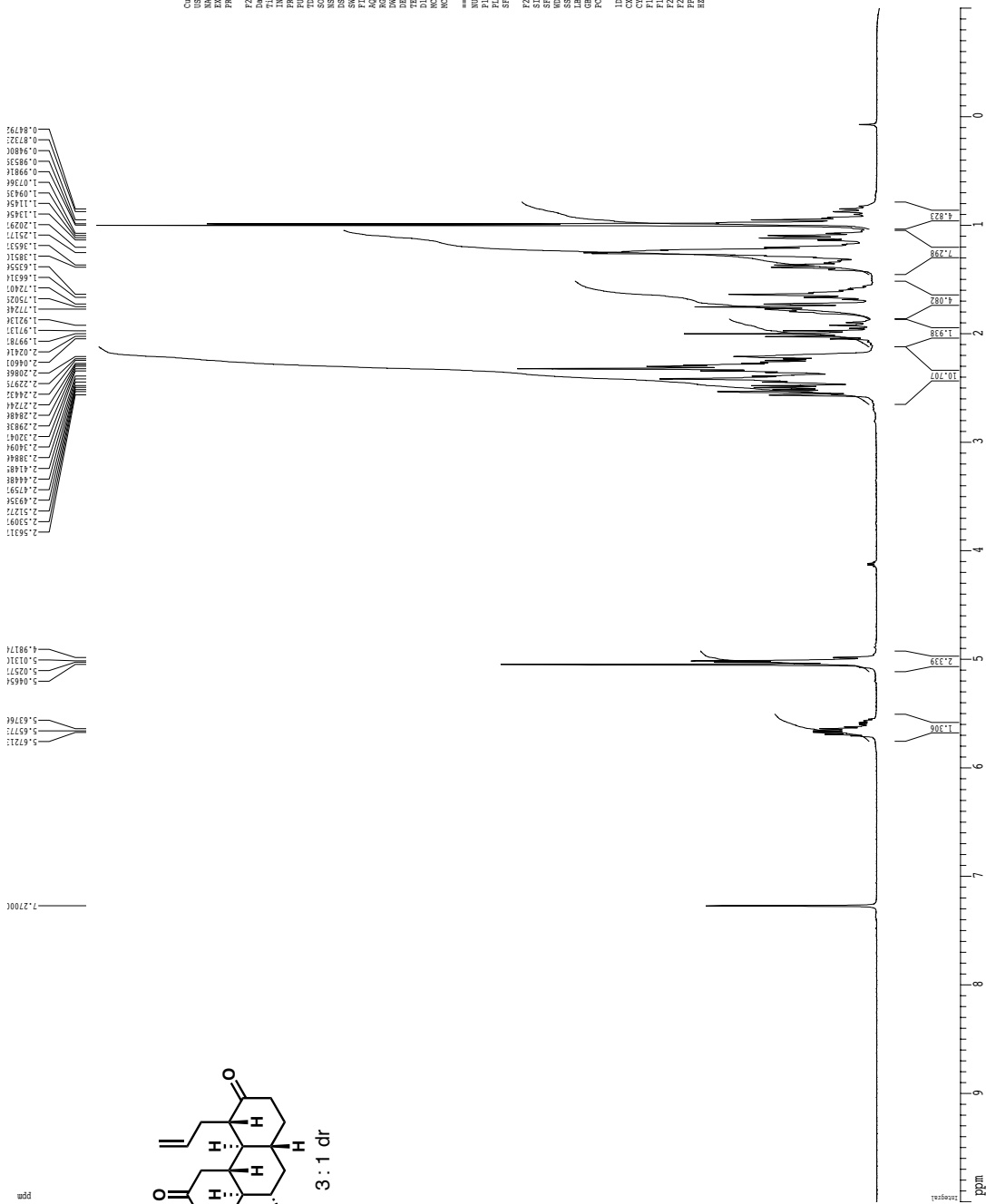


```

Current Data Parameters
NAME      pcr4.07_1solake
EXPNO     1
PROCNO    1
F2 - Acquisition Parameters
Date_     2010226
INSTRUM   crys400
PROBHD    5 mm CPXI 1H-
PULPROG   zgpg30
TD         327448
SOLVENT   CDCl3
NS         2
DS         2
SWH        8012.820 Hz
FIDRES     0.250204 Hz
AQ         1.1991616 sec
RG          8
DQ          0.2400 usec
DE          0.0000000 sec
TE         298.2 K
D1         0.10000000 sec
d11        0.00000000 sec
d12        0.11300000 sec
===== CHANNEL f1 =====
NUC1       13
P1         7.50 usec
PL1        0.00 dB
SFO1       500.225015 MHz
F2 - Processing parameters
SI         6534
WDW         500.2201000 MHz
SSB         0
GB          0
PC          4.00
=====
LO MRB plot parameters
CX         22.80 cm
CY         15.00 cm
CZ         15.00 cm
F1         5002.201 MHz
F2         -1.000 ppm
F3         -500.22 Hz
NUC2       13
NUC3       13
MAGN       241.33423 Hz/cm
  
```



1H spectrum

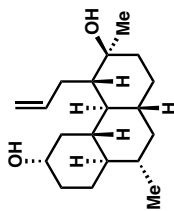
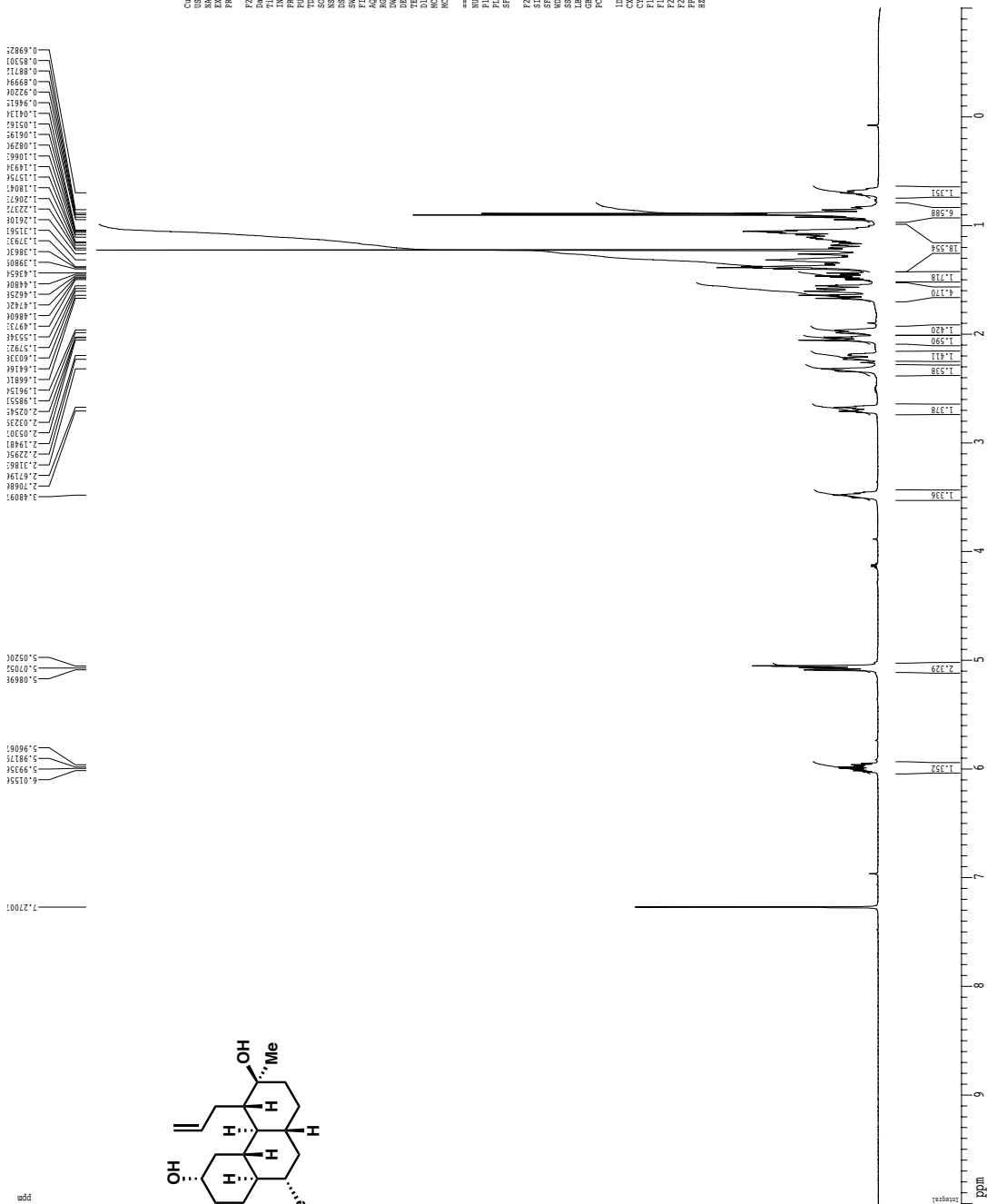


```

Current Data Parameters
NAME      Pct4_048_isolate
EXPNO     1
PROCNO    1
F2 - Acquisition Parameters
Date_     2010129
INSTRUM   cryo500
PROBHD    5 mm CPXI 1H-
PULPROG   zgpg30
TD         32768
SOLVENT   CDCl3
NS         2
DS         2
SWH        8012.820 Hz
FIDRES     0.250024 Hz
AQ         1.1996161 sec
RG          5
DE          62.400 usec
DM          62.400 usec
DI          298.2 K
MAGNET     0.1000000 sec
ACQTIME    0.0000000 sec
SOLVENT    0.1130000 sec
===== CHANNEL f1 =====
NUC1       1H
P1         7.50 usec
PL1        1.60 dB
SFO1       500.225015 MHz
F2 - Processing parameters
SI         6536
SF          500.225015 MHz
WDW         EM
SSB         0
GB          0.0 Hz
PC          4.00
LD MRB plot parameters
CX         22.80 cm
CY         15.00 cm
CZ         10.00 cm
F1         500.225015 MHz
F2         -1.000 ppm
F3         -500.22 Hz
F4         241.33423 Hz/cm
MCON       241.33423 Hz/cm
    
```



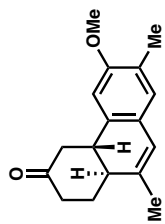
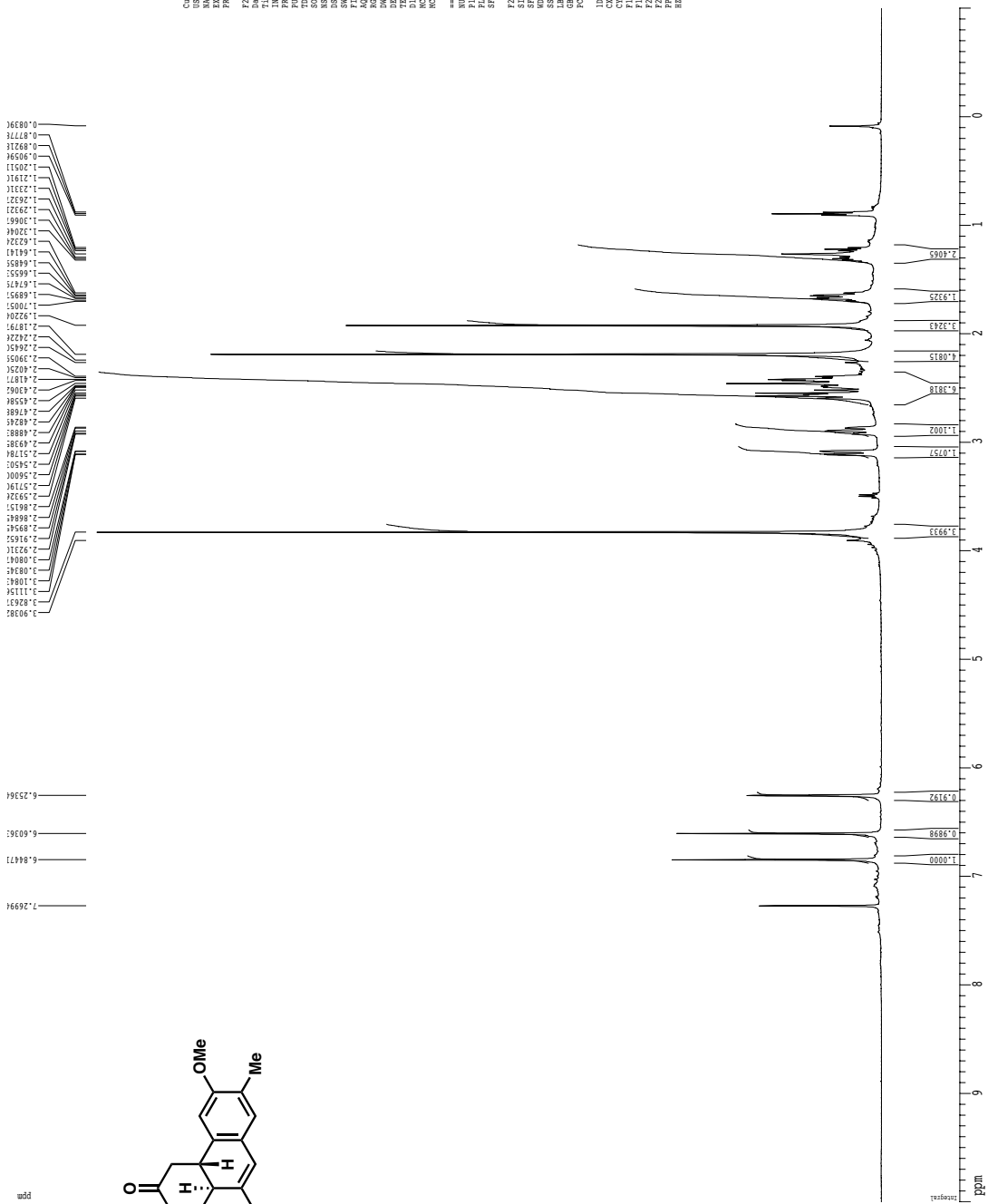
1H spectrum



Current Data Parameters  
 NAME Pct4.086\_m05pc  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20101228  
 Time 11:00:00  
 INSTRUM cty6500  
 PROBRD 5 mm CPCL 1H-  
 PULPROG zgpg30  
 TD 32748  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250024 Hz  
 AQ 1.199161 sec  
 RG 4.43  
 DQ 62.400 usec  
 DE 6.00 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 ACQMSY 0.0000000 sec  
 ACQR 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 0.000000 sec  
 SSB 0  
 GB 0.0 Hz  
 PC 4.00  
 L0 MRB plot parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 GAMMA 241.33423 Hz/cm  
 HSCN 241.33423 Hz/cm



1H spectrum



Current Data Parameters  
 NAME pcat.123.pcat\_4.tit  
 EXPNO 1  
 PROCNO 1

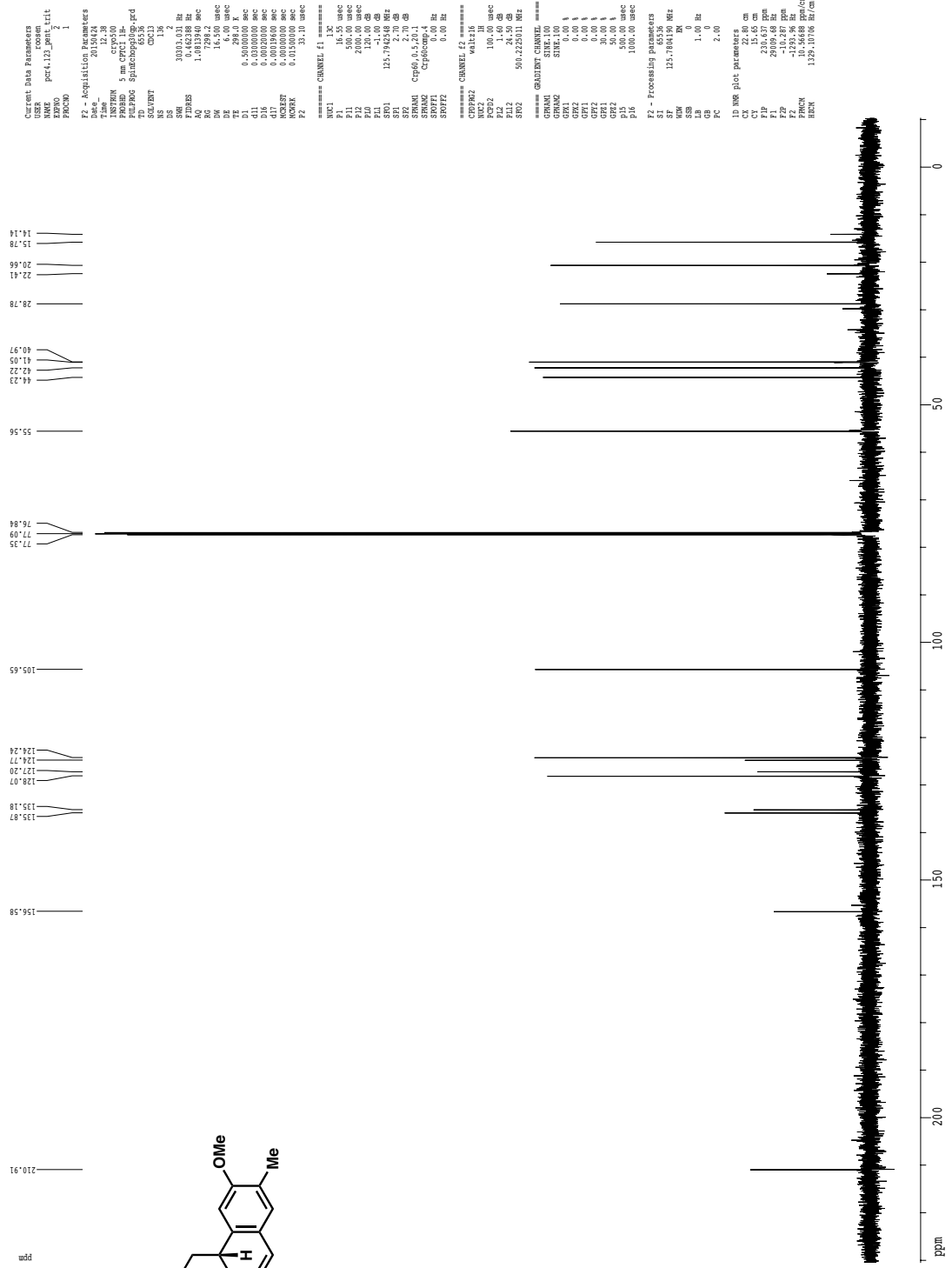
F2 - Acquisition Parameters  
 Date\_ 20080812 12:36  
 Time 20080812 12:36  
 INSTRUM crysov00  
 PULPROG zgpg30  
 TD 32768  
 SFO1 500.136099 MHz  
 CHAN1 cryo1  
 SS 8  
 DS 8  
 SC 8013.47 Hz  
 FIDRES 0.225026 Hz  
 AQ 1.9899451 sec  
 RG 62.400 usec  
 SW 6.00 usec  
 DE 6.00 usec  
 DT 0.1000000 sec  
 NUC1 13C  
 NUC2 13C  
 NUC3 13C  
 CKE1 0.0000000 sec  
 CKE2 0.0150000 sec

===== CHANNEL f1 =====  
 NU1 7.30 usec  
 FL1 1.60 dB  
 SFO1 500.223015 MHz

F2 - Processing parameters  
 SI 65536  
 SF 500.136099 MHz  
 WDW EM  
 GB 0  
 CB 0  
 SC 4.00  
 PC 4.00

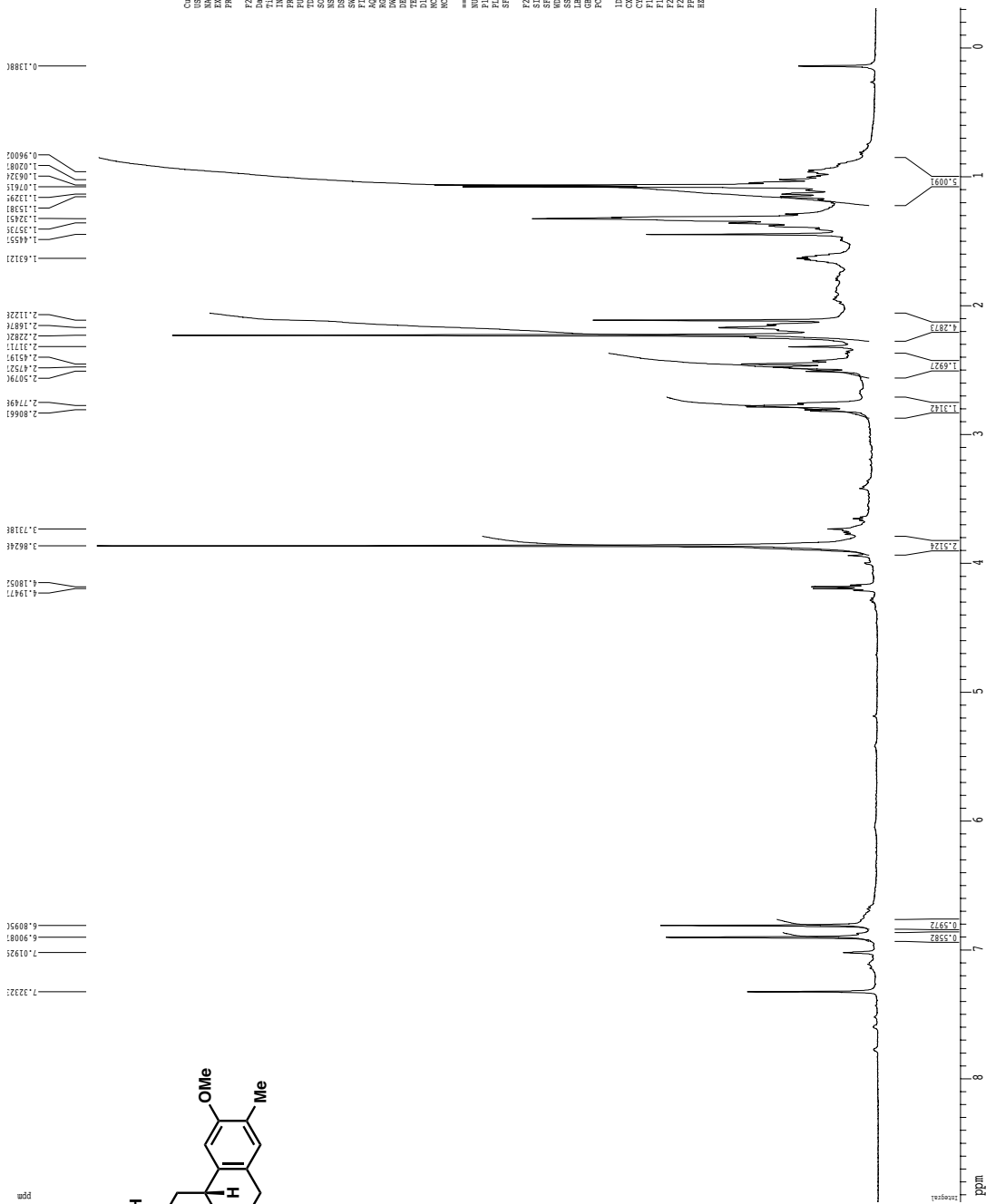
ID: NMR plot parameters  
 CT 22.80 cm  
 C1 10.0000000  
 F1 500.2200000 MHz  
 F2 500.1360990 MHz  
 PR 1.0000000  
 FR 0.48346 ppm/cm  
 BPC 241.33823 Hz/cm

Z-restored spin-echo <sup>13</sup>C spectrum with <sup>1</sup>H decoupling





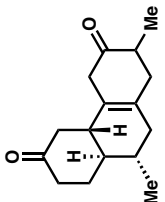
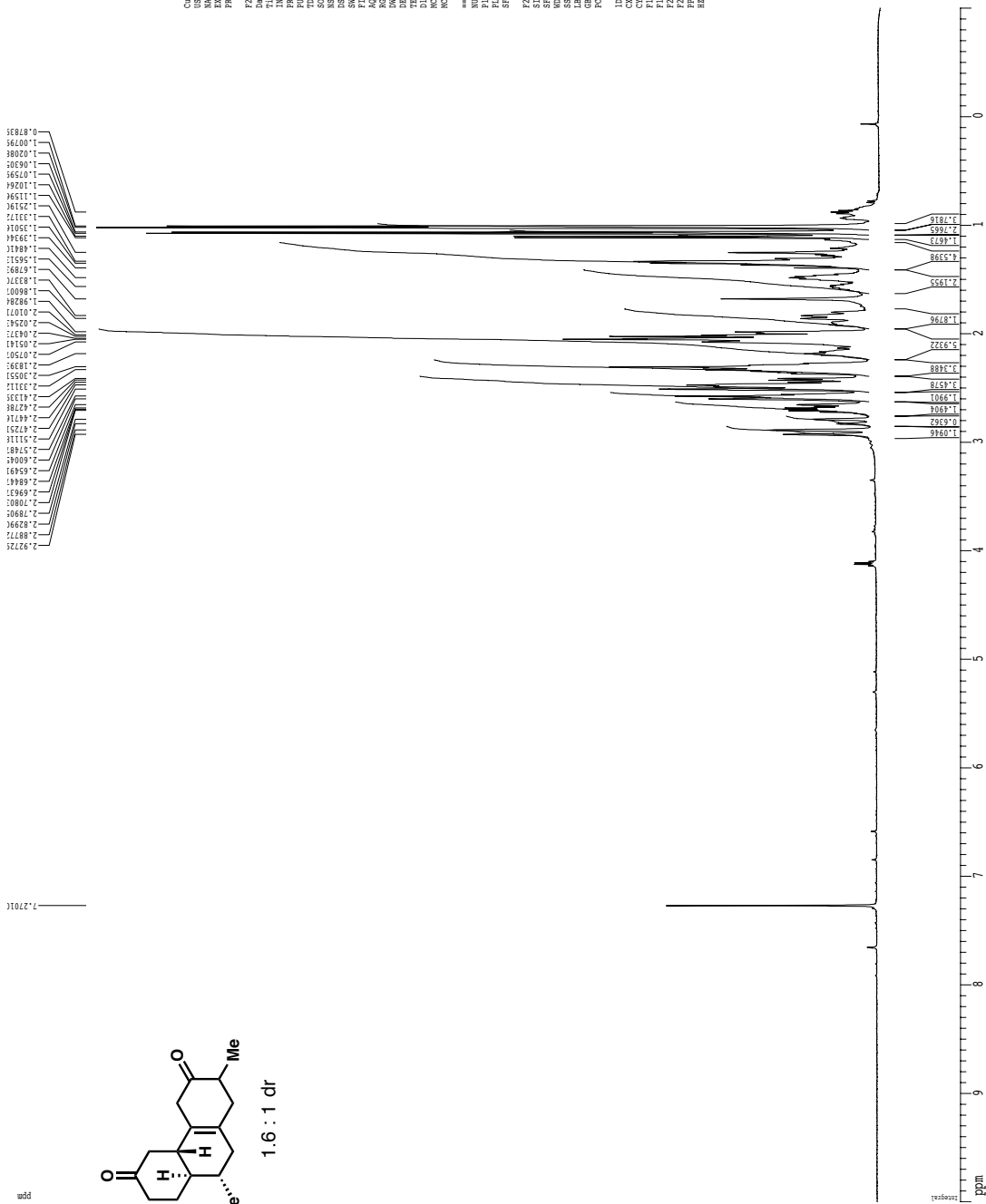
1H spectrum



Current Data Parameters  
 NAME Pct4.128\_crude  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2010127  
 Time 11:22:55  
 INSTRUM ctyes500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 0  
 SWH 802.820 Hz  
 FIDRES 0.250024 Hz  
 AQ 1.199666 sec  
 RG 314  
 EQ 2.14  
 DM 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 D1 0.11000000 sec  
 ACQBST 0.00000000 sec  
 ACQBRK 0.11000000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6536  
 SF 500.225015 MHz  
 WWP 8K  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MRB pulse parameters  
 CK 22.80 cm  
 CT 15.00 cm  
 F1 497.51 MHz  
 F2 -4.309 ppm  
 F3 -15.632 Hz  
 WCN 203.23116 Hz/cm



1H spectrum

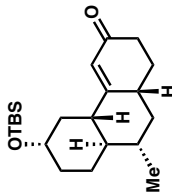
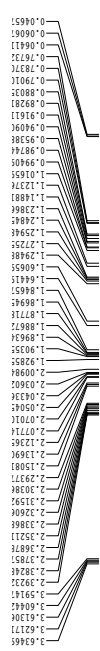


1.6 : 1 dr

Current Data Parameters  
 NAME Pct4.137\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20150315  
 Time 11:05:00  
 INSTRUM cryo500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32748  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 802.820 Hz  
 FIDRES 0.250204 Hz  
 AQ 1.199124 sec  
 RG 3.4  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 ACQSF 0.0000000 sec  
 SOLOR 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6536  
 SF 500.225015 MHz  
 WOP 20.000000 MHz  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MRB pulse parameters  
 CK 22.80 cm  
 CT 15.00 cm  
 CF 100.625000 MHz  
 FI 500.220180 MHz  
 FFP -1.000 ppm  
 FZ -500.22 Hz  
 HPCON 241.33423 Hz/cm

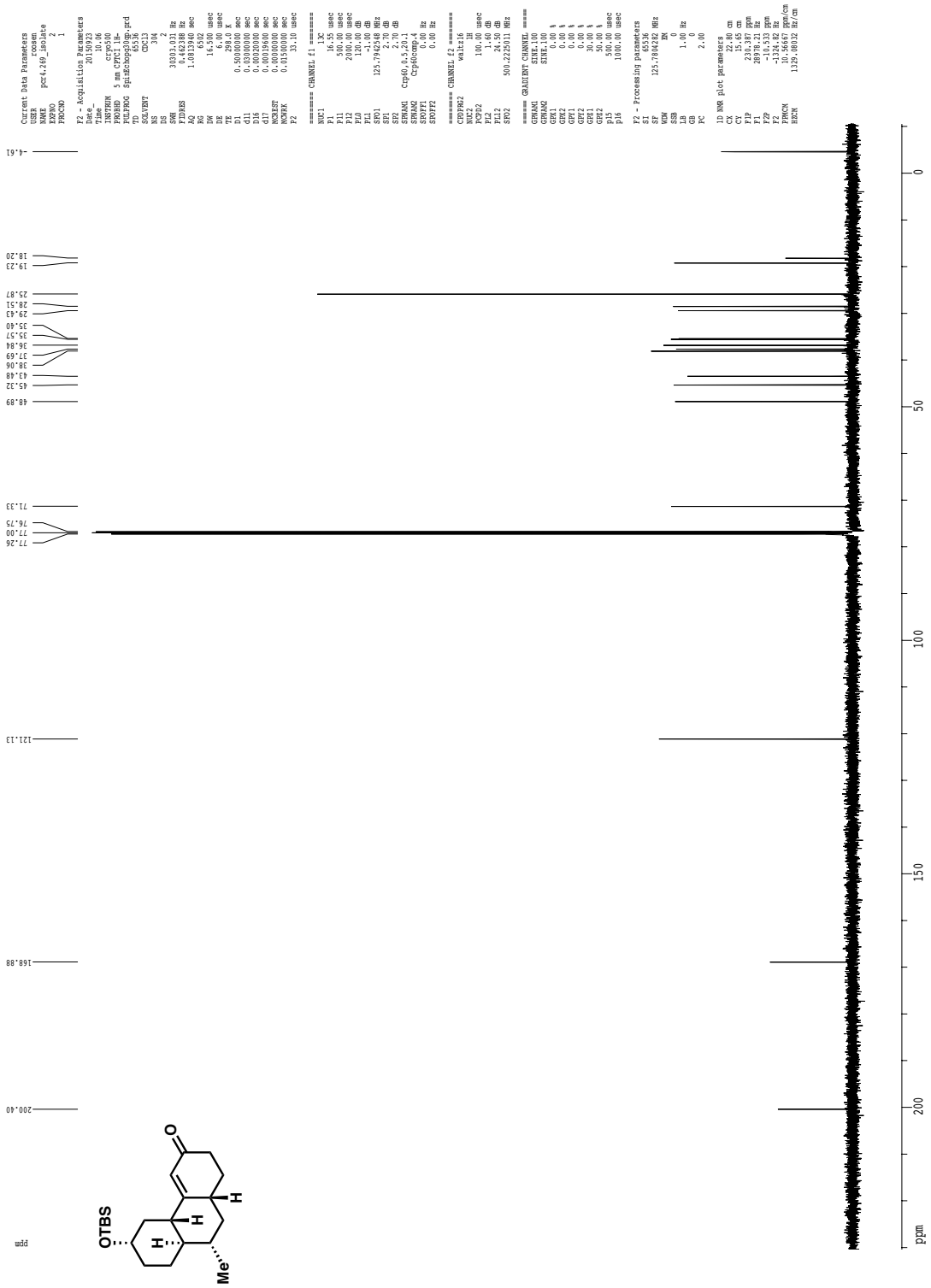


1H spectrum

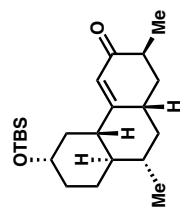
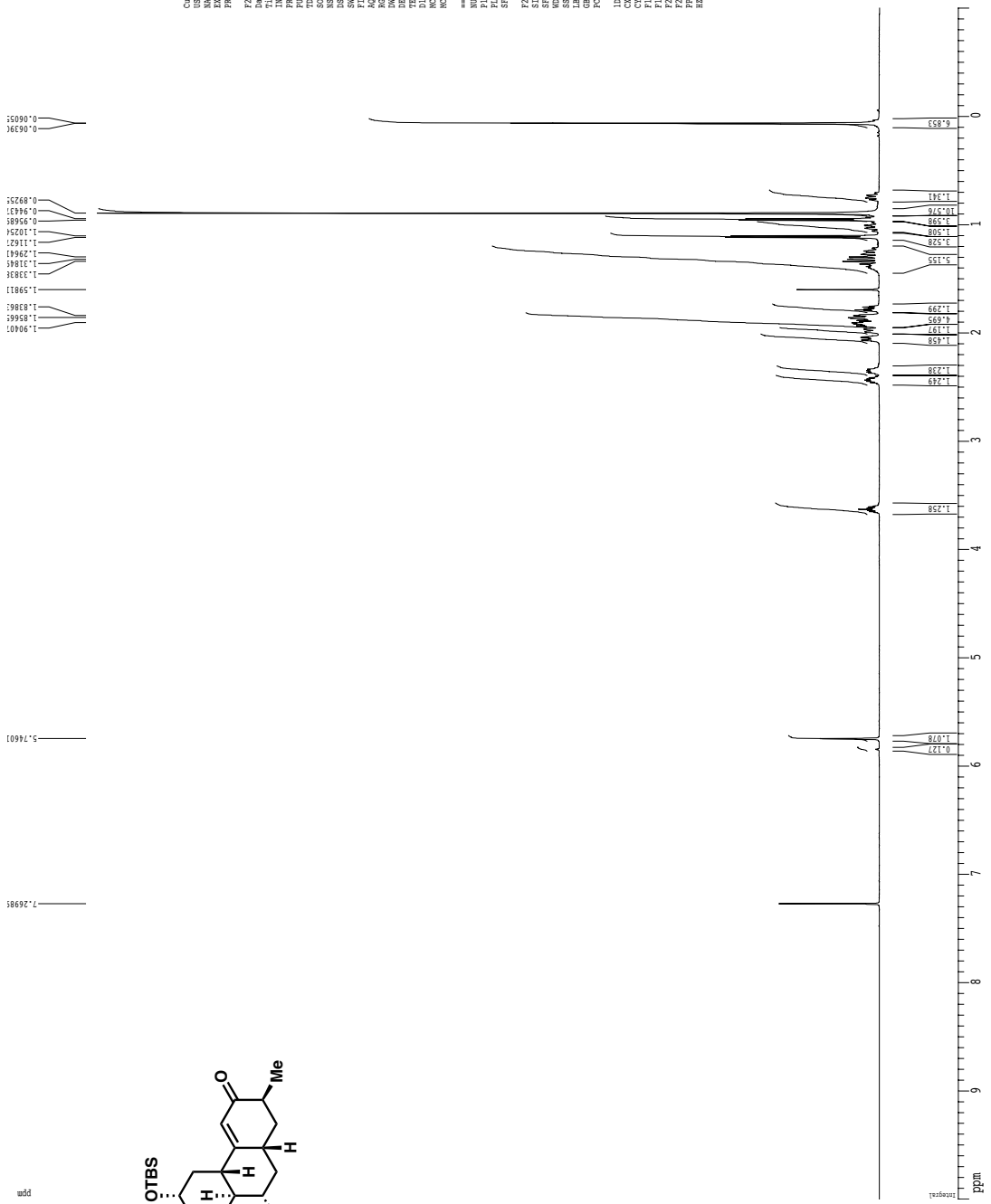


```
Current Data Parameters
NAME      Pct4-261_isolate
EXPNO     1
PROCNO    1
F2 - Acquisition Parameters
Date_     2010023
Time      10.35
INSTRUM   crys000
PROBHD    5 mm CPTCI 1H-
PULPROG   zgpg30
TD         32768
SOLVENT   CDCl3
NS         2
DS         4
SWH        8012.820 Hz
FIDRES     0.252026 Hz
AQ         1.9998412 sec
RG          4.3
DQ          0
ZG          0
ZG2         0
ZG3         0
ZG4         0
TE          298.2 K
DE          0.7000000 sec
AQRESFT    0.00000000 sec
SFOUR      0.11330000 sec
===== CHANNEL f1 =====
NUC1       1H
P1          7.45 usec
PC1         1.60 dB
SFO1       500.225015 MHz
F2 - Processing parameters
SI          6536
SF          500.225015 MHz
WDW         EM
SSB         0
GB          0
PC          1.00
L0 MRB plot parameters
CX          22.80 cm
CY          15.00 cm
CZ          15.00 cm
EL          1.00 degree
FI          500.231 Hz
FP          -1.000 ppm
FQ          -500.22 Hz
RG          241.33423 Hz/cm
HRCOIN      241.33423 Hz/cm
```

Z-restored spin-echo <sup>13</sup>C spectrum with <sup>1</sup>H decoupling

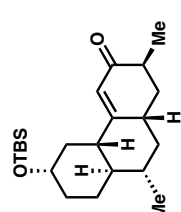
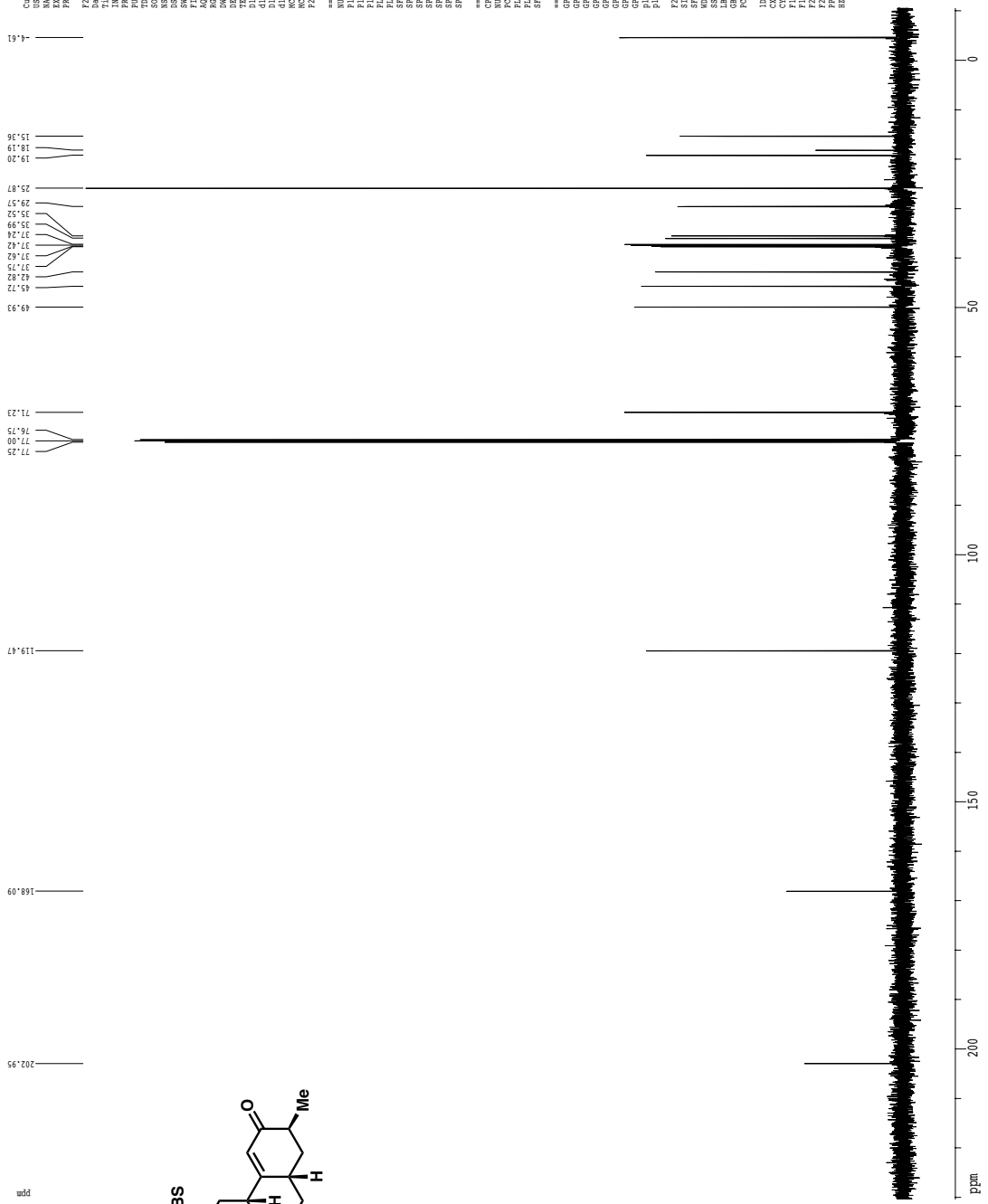


1H spectrum



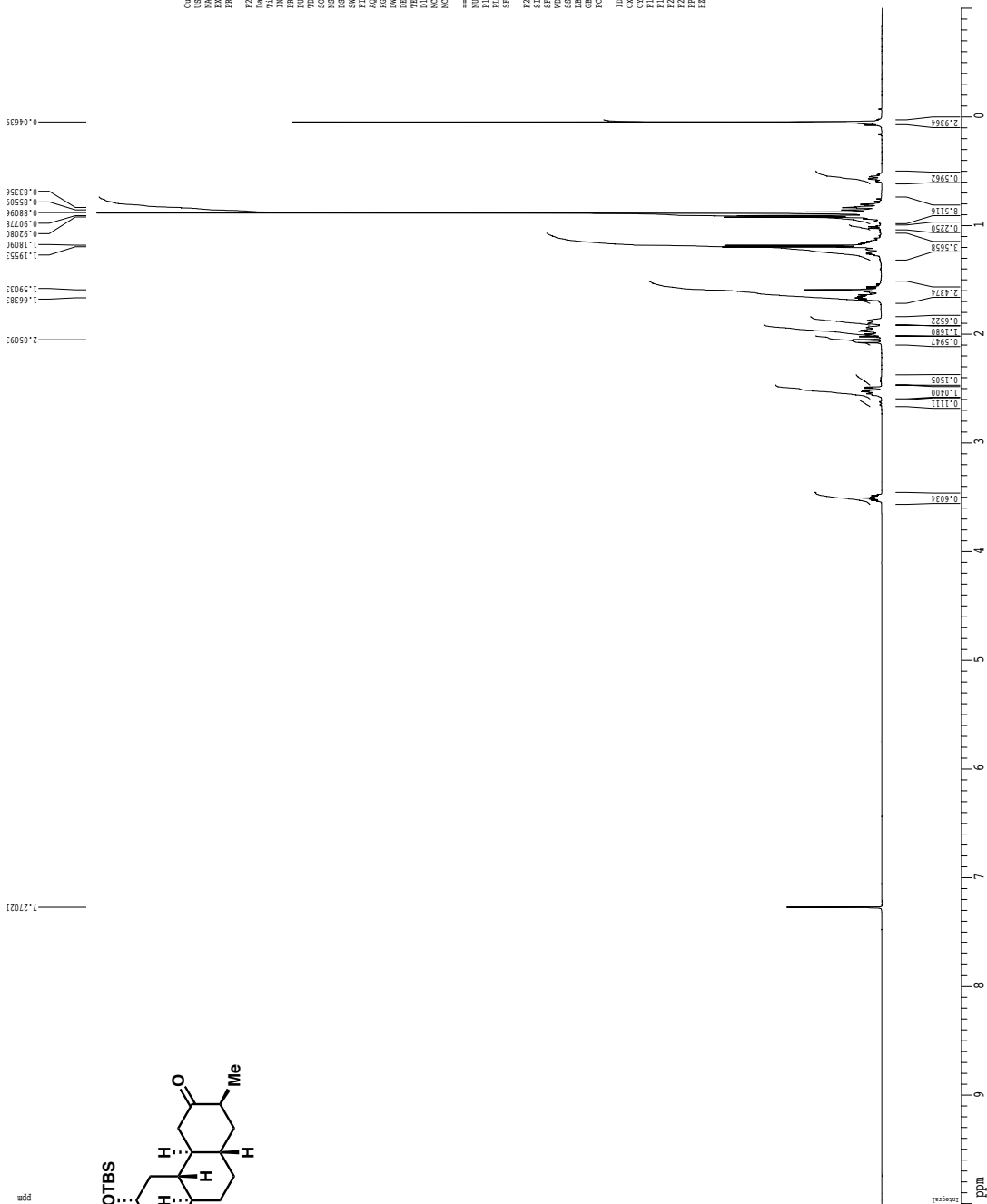
Current Data Parameters  
 NAME Pct4-2B isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 201511  
 INSTRUM ctyco500  
 PROBRD 5 mm CPYC1 H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 4  
 SWH 8012.820 Hz  
 FIDRES 0.250024 Hz  
 RG 1.000000  
 EC 10.1  
 AQ 1.000000 sec  
 DQ 62.400 usec  
 SFO1 500.225015 MHz  
 FE 62.400 usec  
 TE 298.2 K  
 DE 0.000000 sec  
 DI 0.1000000 sec  
 DC 0.0000000 sec  
 ACQST 0.1000000 sec  
 ACQR 0.1000000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 1.000000 sec  
 PL1 0.00 dB  
 F1 125.760 MHz  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 32768  
 SSB 0  
 GB 0  
 CB 0.00 Hz  
 PC 4.00  
 LO MR4 pulse parameters  
 CK 22.80 cm  
 CT 15.00 cm  
 F1 500.225015 MHz  
 F2 500.225015 MHz  
 F3 -1.000 ppm  
 F4 -500.22 Hz  
 HPCON 241.33423 Hz/cm

Z-restored spin-echo <sup>13</sup>C spectrum with <sup>1</sup>H decoupling



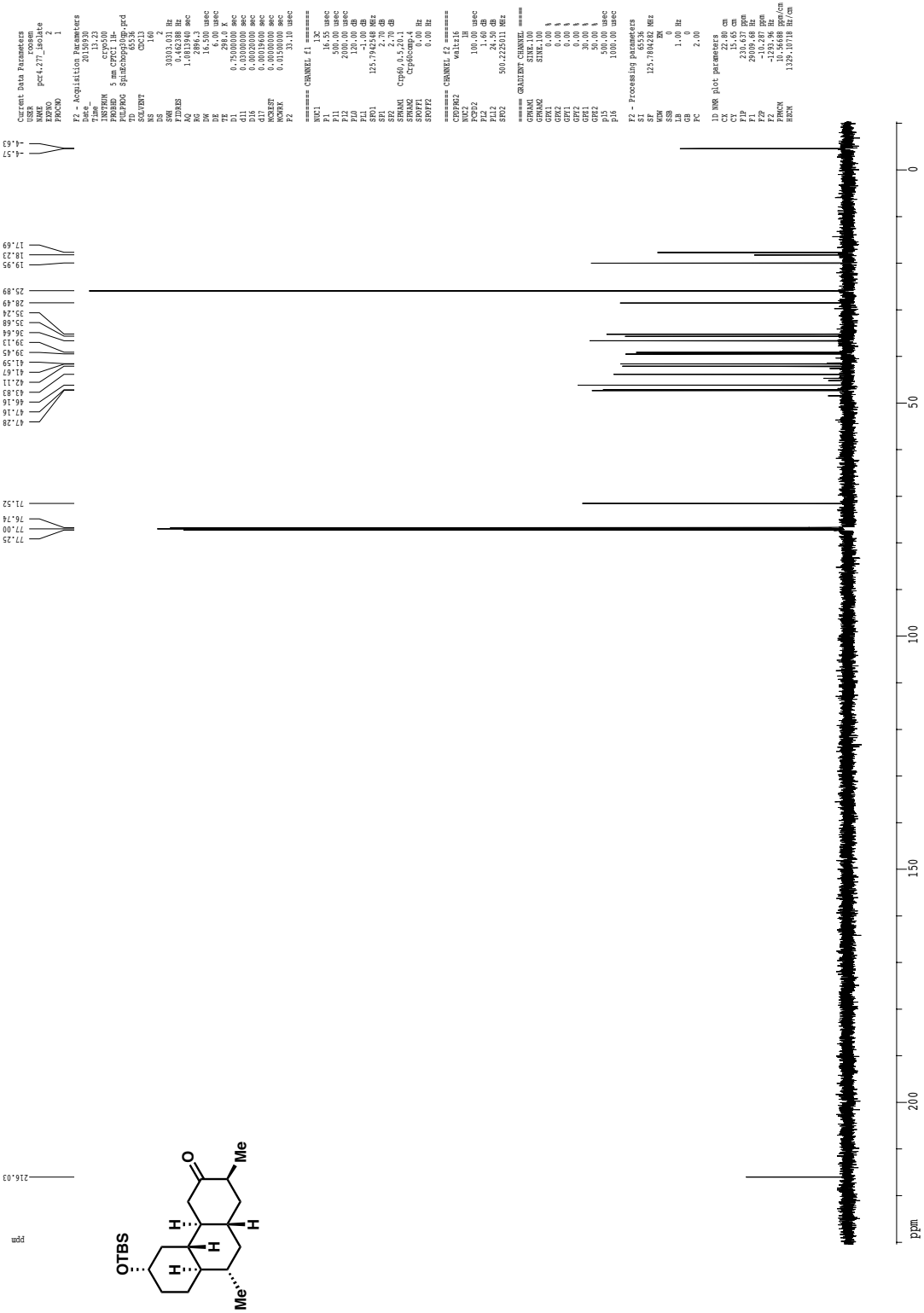


1H spectrum

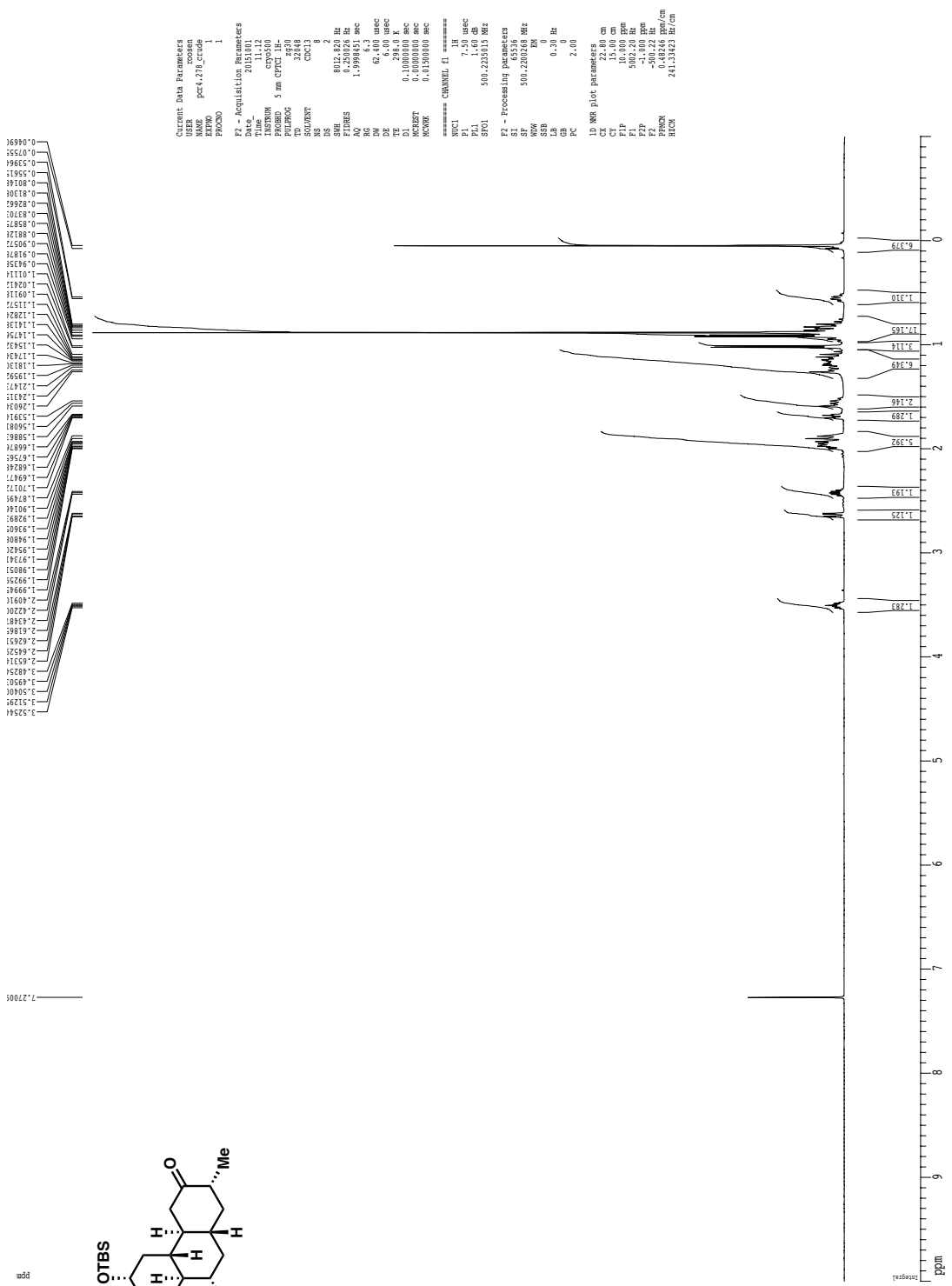


Current Data Parameters  
 NAME Pct4-277\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20100930  
 Time 11:00:00  
 INSTRUM ctyco500  
 PROBRD 5 mm CPCLP 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SFO1 8012.820 MHz  
 FIDRES 0.250024 Hz  
 AQ 1.1991614 sec  
 RG 64.3  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 DI 0.11000000 sec  
 ACQMSF 0.00000000 sec  
 DEPR 0.11300000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 F41 500.225015 MHz  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 0.000000 sec  
 SSB 0  
 CB 0.00 Hz  
 PC 4.00  
 LO MRB pldr parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 HPCON 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling

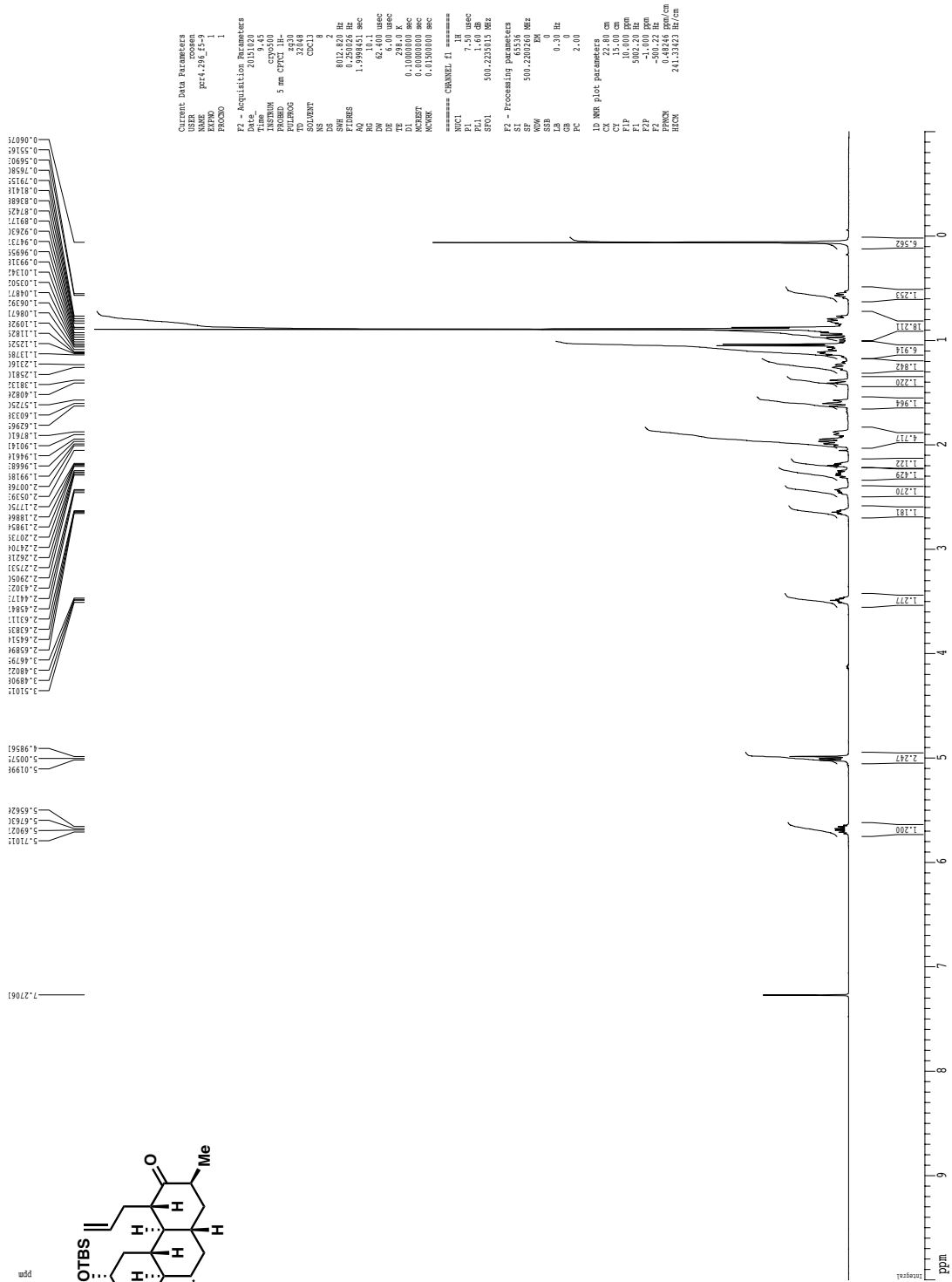
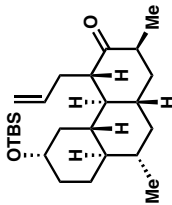


1H spectrum



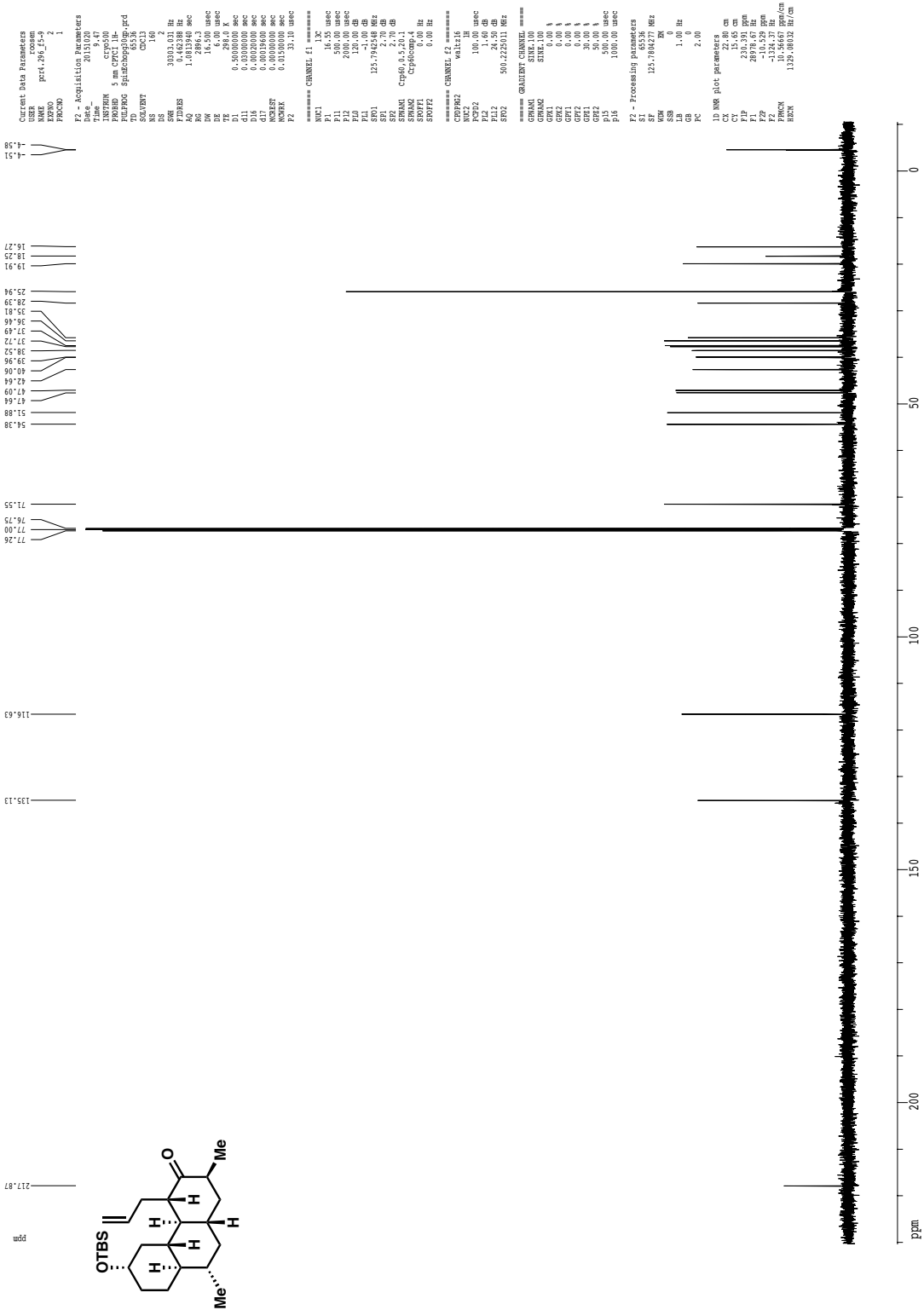


1H spectrum

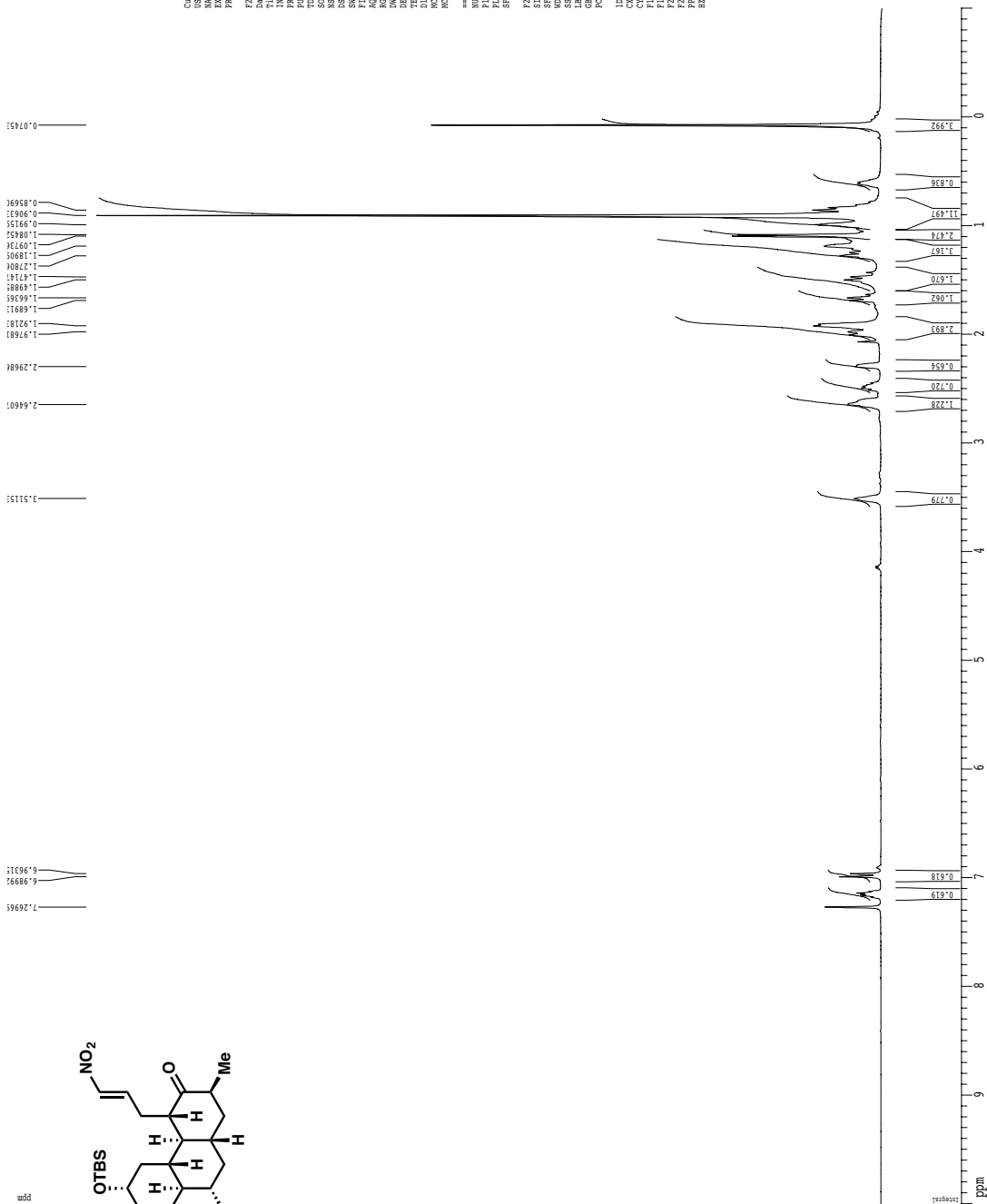


Current Data Parameters  
 NAME pcr4\_286\_61-9  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2015020  
 INSTRUM cryo500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250026 Hz  
 AQ 1.19100000 sec  
 RG 101  
 DQ 62.400 usec  
 DE 6.00 usec  
 TE 298.2 K  
 D1 0.10000000 sec  
 DCOffset 0.00000000 sec  
 ACQBR 0.11300000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.45 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6536  
 SF 500.225015 MHz  
 WWP 32768  
 SSB 0  
 GB 0  
 PC 2.00  
 LO MRB pulse parameters  
 CK 22.80 cm  
 CT 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 HZ 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling

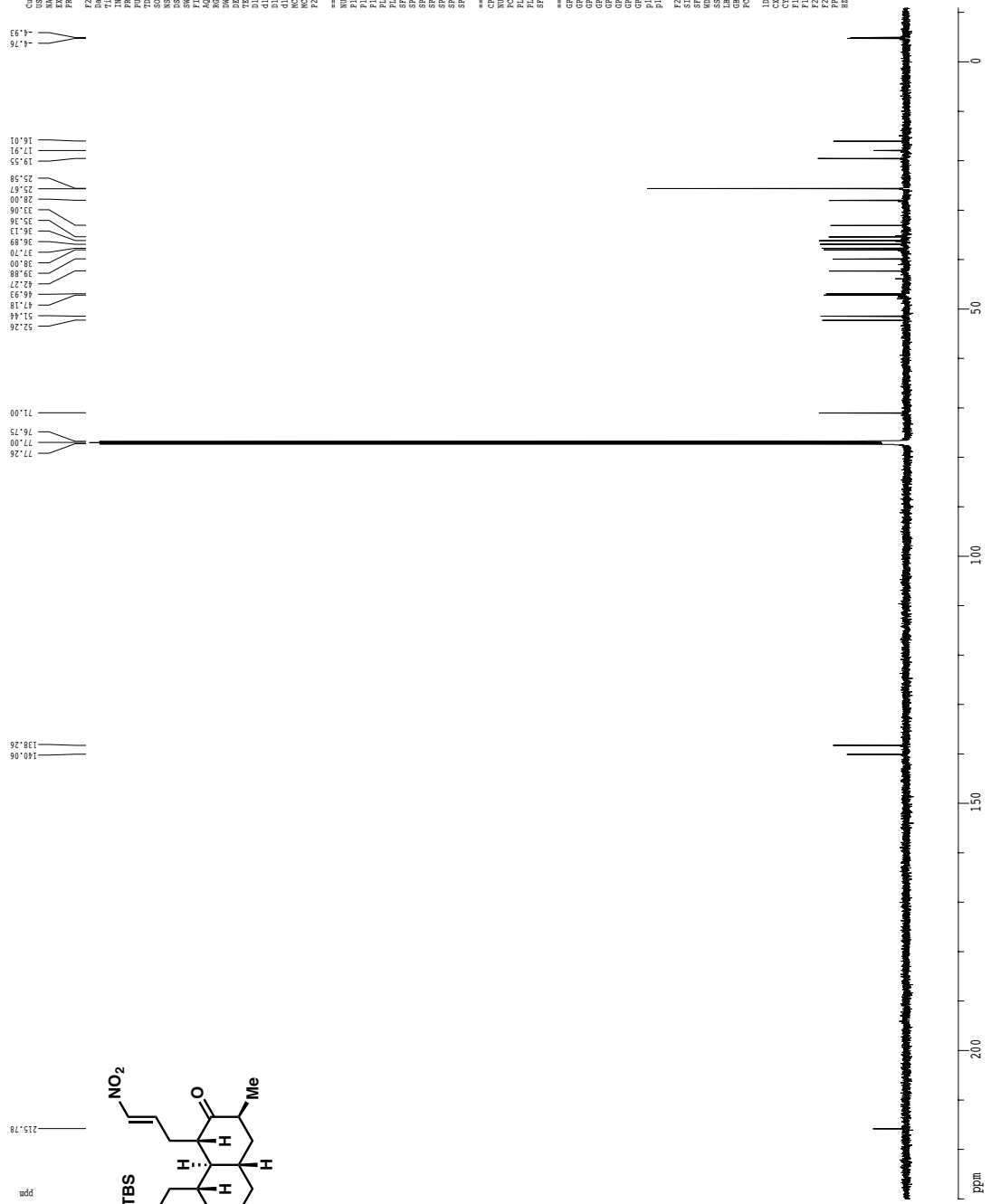


1H spectrum



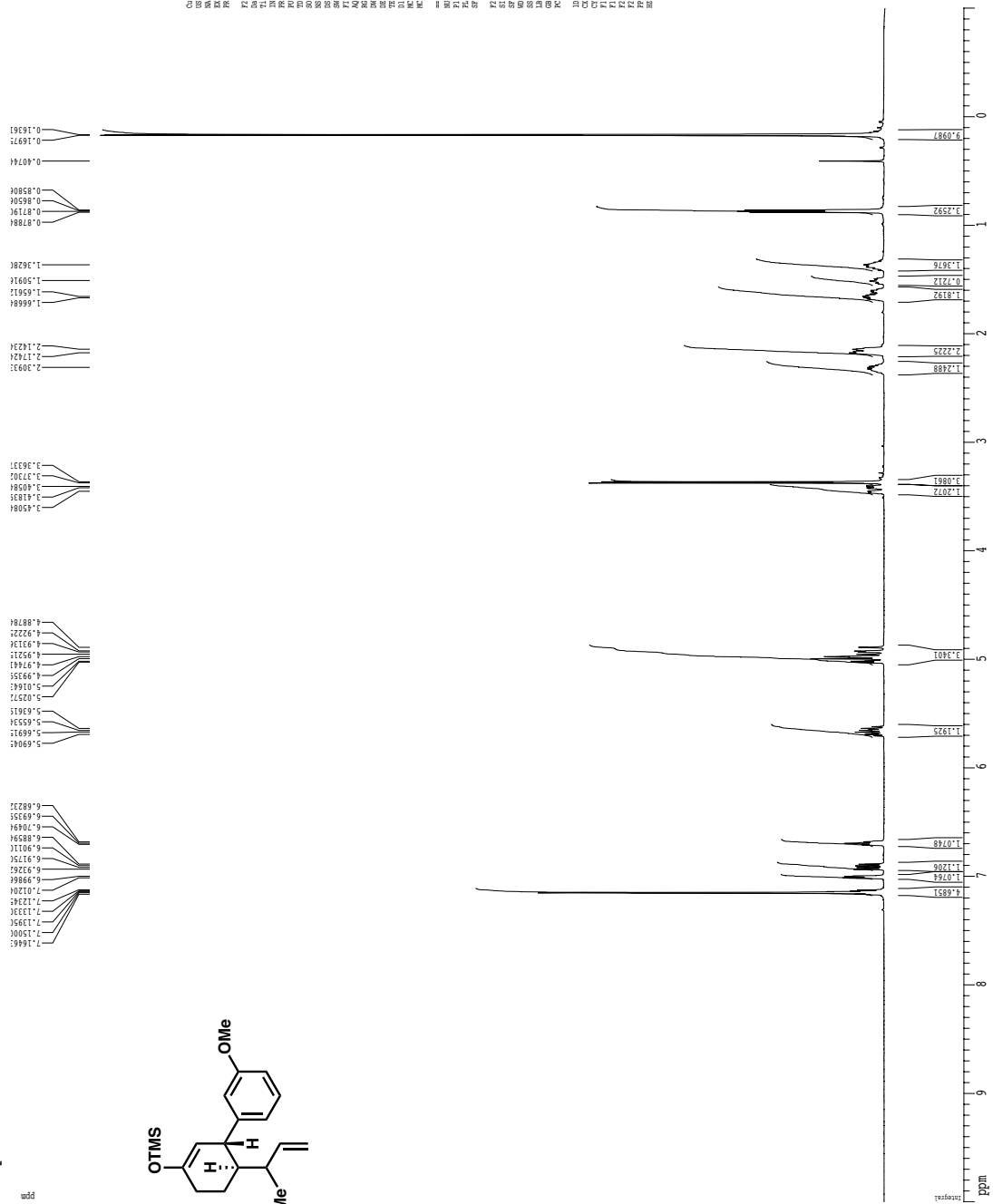
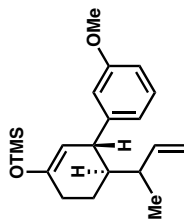
Current Data Parameters  
 NAME pcr3\_011\_01-8  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2015025  
 TIME 11.00  
 INSTRUM ctyco500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250024 Hz  
 AQ 1.1991616 sec  
 RG 317  
 EQ 3.7  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 DI 0.11000000 sec  
 ACQBST 0.00000000 sec  
 ACQBPR 0.11000000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 12.00 usec  
 PL1 0.00 dB  
 F1 500.136350 MHz  
 SFO1 500.136350 MHz  
 F2 - Processing parameters  
 SI 6536  
 SF 500.136350 MHz  
 WOP 30.000000 MHz  
 SSB 0  
 CB 0.00 Hz  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 FFX 3002.201 MHz  
 FFY 3002.201 MHz  
 FFP -1.000 ppm  
 FZ 500.136350 MHz  
 HPCON 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



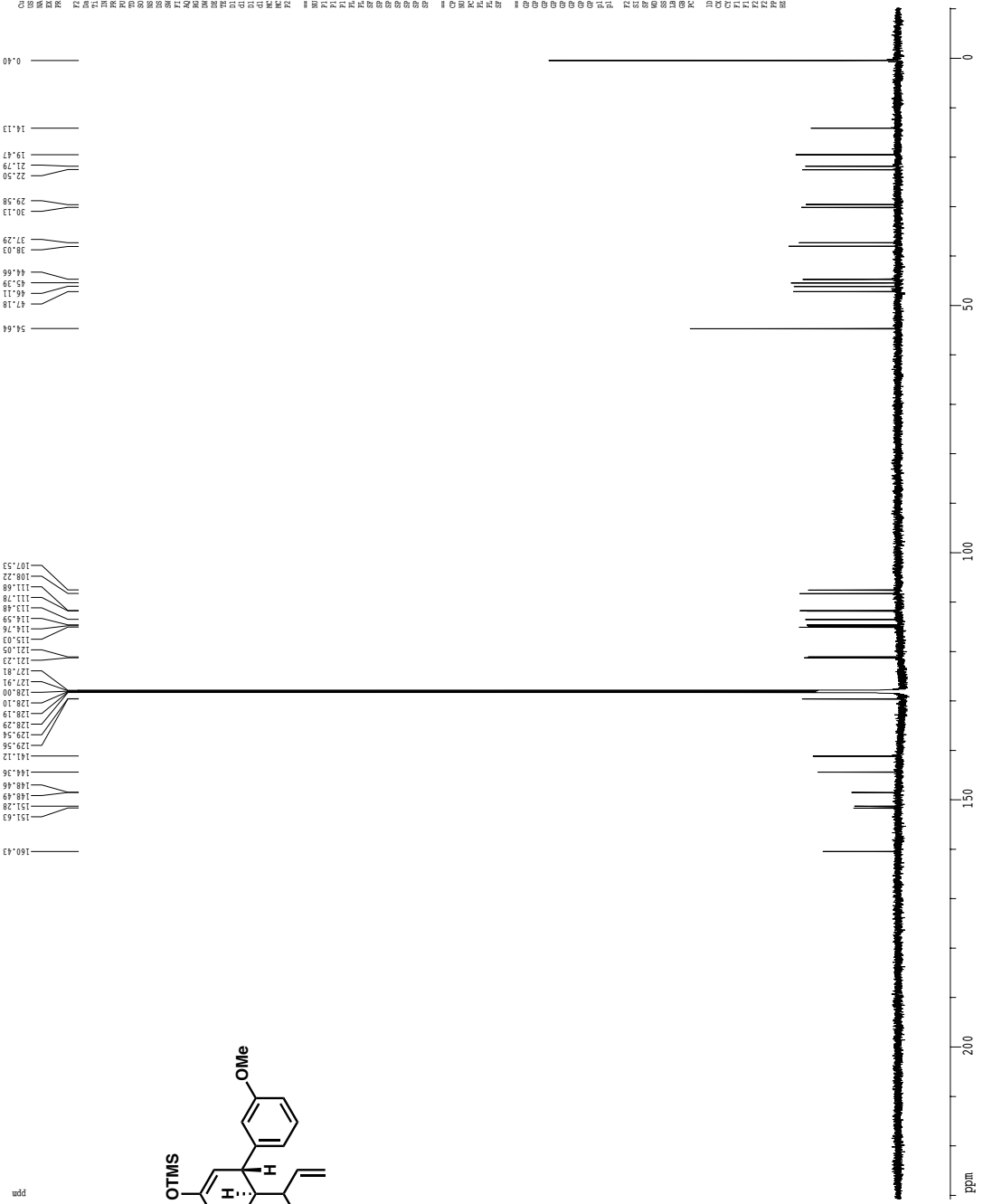


<sup>1</sup>H spectrum

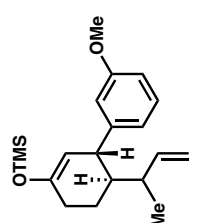


Experiment Data Parameters  
 Name: pcc1\_205\_10a10a\_0516  
 Date: 2015077  
 Time: 12.13  
 C13: 127.13  
 P1: 5.00  
 P2: 5.00  
 P3: 5.00  
 P4: 5.00  
 P5: 5.00  
 P6: 5.00  
 P7: 5.00  
 P8: 5.00  
 P9: 5.00  
 P10: 5.00  
 P11: 5.00  
 P12: 5.00  
 P13: 5.00  
 P14: 5.00  
 P15: 5.00  
 P16: 5.00  
 P17: 5.00  
 P18: 5.00  
 P19: 5.00  
 P20: 5.00  
 P21: 5.00  
 P22: 5.00  
 P23: 5.00  
 P24: 5.00  
 P25: 5.00  
 P26: 5.00  
 P27: 5.00  
 P28: 5.00  
 P29: 5.00  
 P30: 5.00  
 P31: 5.00  
 P32: 5.00  
 P33: 5.00  
 P34: 5.00  
 P35: 5.00  
 P36: 5.00  
 P37: 5.00  
 P38: 5.00  
 P39: 5.00  
 P40: 5.00  
 P41: 5.00  
 P42: 5.00  
 P43: 5.00  
 P44: 5.00  
 P45: 5.00  
 P46: 5.00  
 P47: 5.00  
 P48: 5.00  
 P49: 5.00  
 P50: 5.00  
 P51: 5.00  
 P52: 5.00  
 P53: 5.00  
 P54: 5.00  
 P55: 5.00  
 P56: 5.00  
 P57: 5.00  
 P58: 5.00  
 P59: 5.00  
 P60: 5.00  
 P61: 5.00  
 P62: 5.00  
 P63: 5.00  
 P64: 5.00  
 P65: 5.00  
 P66: 5.00  
 P67: 5.00  
 P68: 5.00  
 P69: 5.00  
 P70: 5.00  
 P71: 5.00  
 P72: 5.00  
 P73: 5.00  
 P74: 5.00  
 P75: 5.00  
 P76: 5.00  
 P77: 5.00  
 P78: 5.00  
 P79: 5.00  
 P80: 5.00  
 P81: 5.00  
 P82: 5.00  
 P83: 5.00  
 P84: 5.00  
 P85: 5.00  
 P86: 5.00  
 P87: 5.00  
 P88: 5.00  
 P89: 5.00  
 P90: 5.00  
 P91: 5.00  
 P92: 5.00  
 P93: 5.00  
 P94: 5.00  
 P95: 5.00  
 P96: 5.00  
 P97: 5.00  
 P98: 5.00  
 P99: 5.00  
 P100: 5.00

Z-restored spin-echo 13C spectrum with 1H decoupling



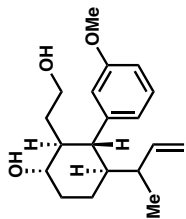
Current Data Parameters  
 NAME pcr1-205\_koelap\_e086  
 EXPNO 2  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20101019  
 Time 12:15  
 Date\_ 20101019  
 PROBRD 5 mm CPXI 1H  
 PULPROG SpinalborgSimpurd  
 SOLVENT CDCl3  
 SOLVENT 242  
 IN 2  
 SHF 300.01 Hz  
 FIDRES 0.001196 Hz  
 AQ 1.001848 sec  
 RG 497.6  
 AC 1.000000 sec  
 DE 6.00 umsec  
 TE 298.00 K  
 D1 0.5000000 sec  
 d11 0.4300000 sec  
 d12 0.7000000 sec  
 d13 0.3000000 sec  
 INCRST 0.0000000 sec  
 ACQPRG 0.0000000 sec  
 PC 0.1500000 umsec  
 ===== CHANNEL f1 =====  
 NUCL1 13C  
 P1 51.5 umsec  
 PL1 0.00 dB  
 F1 125.760342 MHz  
 ===== CHANNEL f2 =====  
 NUCL2 1H  
 P2 10.00 umsec  
 PL2 2.00 dB  
 F2 500.136451 MHz  
 ===== CHANNEL f3 =====  
 CHANM1 13C  
 CHANM2 1H  
 ===== CHANNEL PARAMETERS =====  
 PCP2 10.00 umsec  
 PL2 2.00 dB  
 SF02 500.136451 MHz  
 ===== CHANNEL f4 =====  
 CHANM1 13C  
 CHANM2 1H  
 ===== CHANNEL PARAMETERS =====  
 PCP2 10.00 umsec  
 PL2 2.00 dB  
 SF02 500.136451 MHz  
 ===== CHANNEL f5 =====  
 CHANM1 13C  
 CHANM2 1H  
 ===== CHANNEL PARAMETERS =====  
 PCP2 10.00 umsec  
 PL2 2.00 dB  
 SF02 500.136451 MHz  
 ===== CHANNEL f6 =====  
 CHANM1 13C  
 CHANM2 1H  
 ===== CHANNEL PARAMETERS =====  
 PCP2 10.00 umsec  
 PL2 2.00 dB  
 SF02 500.136451 MHz



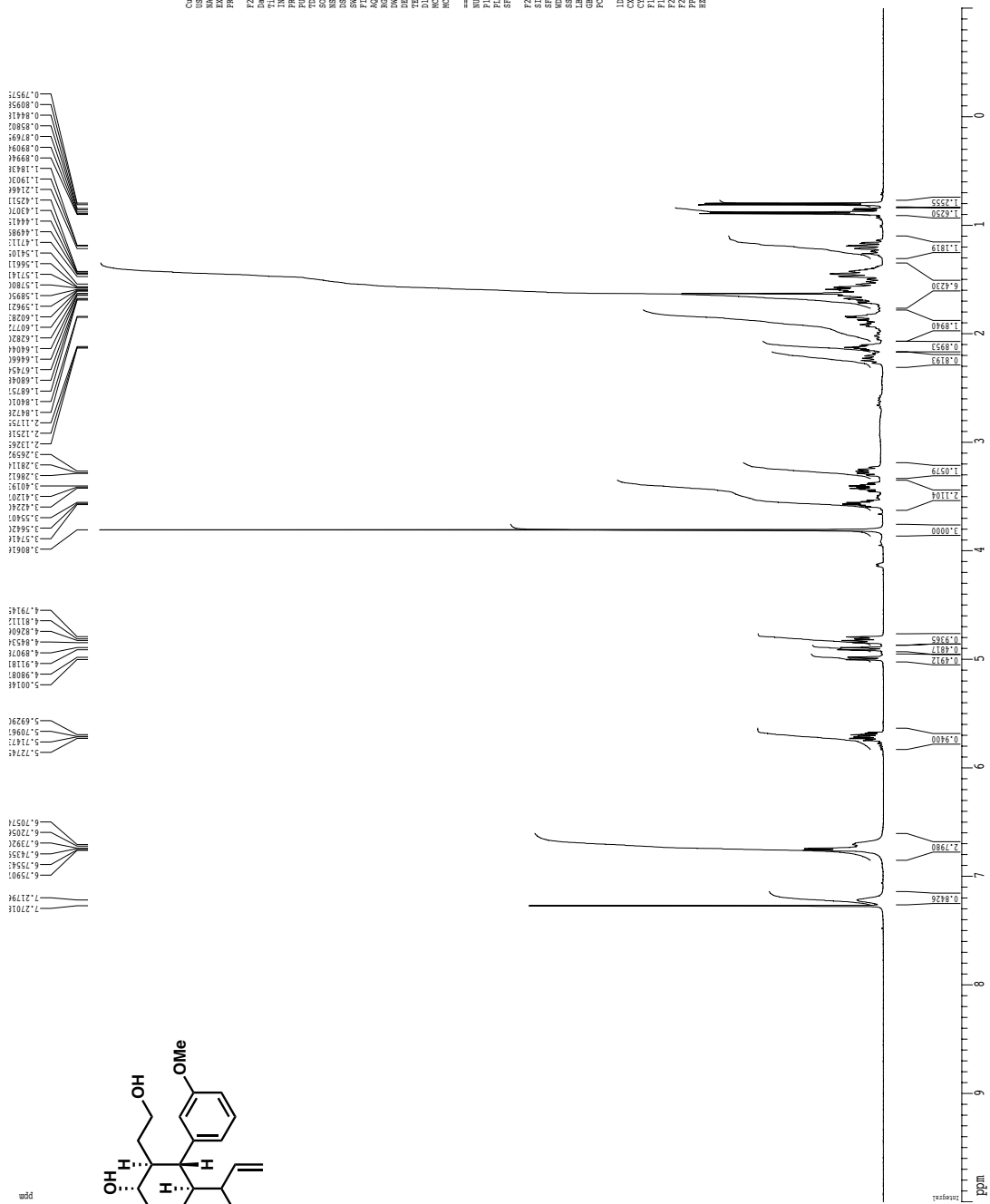




1H spectrum

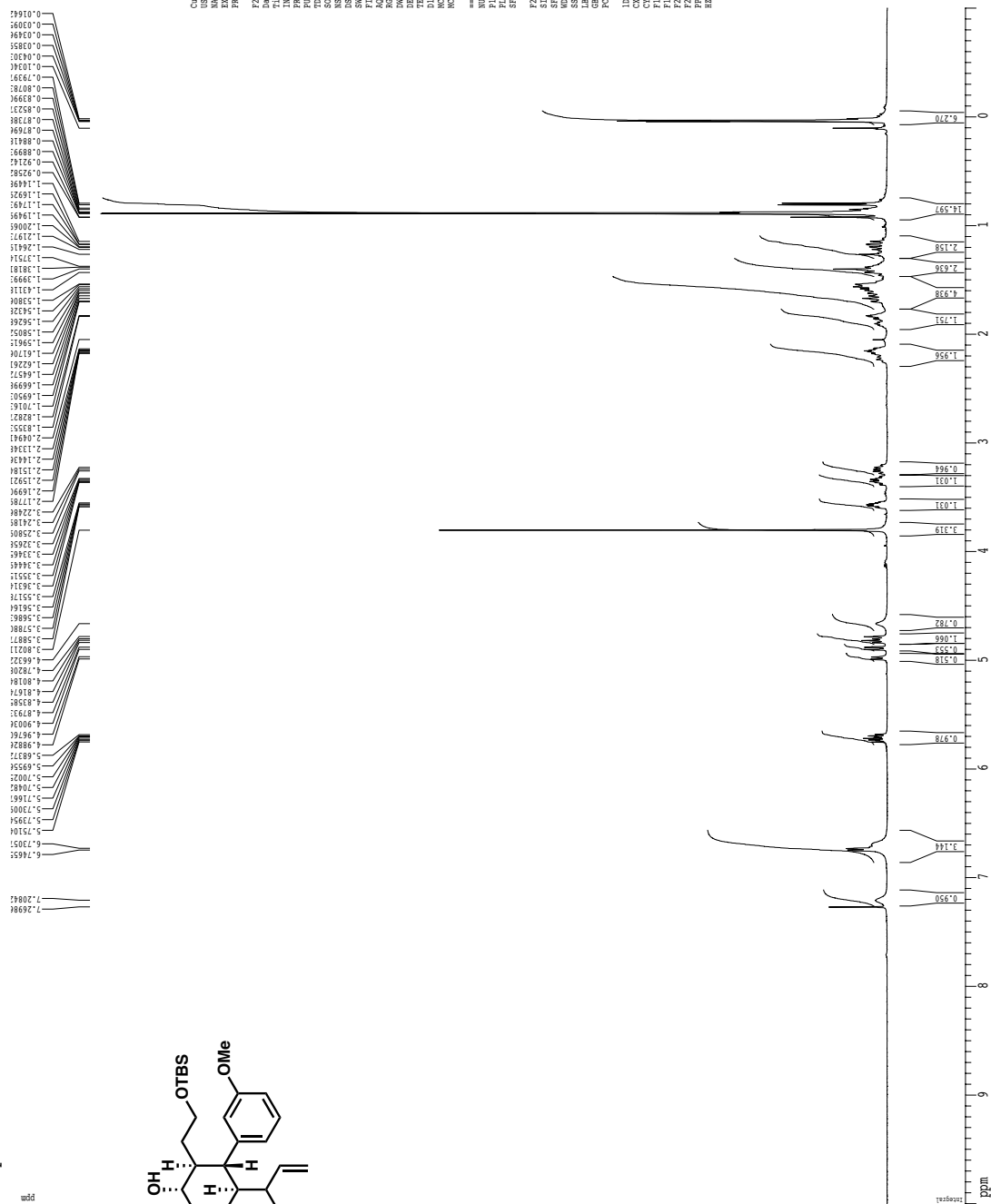


Current Data Parameters  
 NAME Pct4-276\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20151013  
 Time 11:03:05  
 INSTRUM cty6500  
 PROBRD 5 mm CPYU 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl<sub>3</sub>  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250024 Hz  
 AQ 1.1996141 sec  
 RG 9  
 DQ 62.400 usec  
 DE 6.00 usec  
 TE 298.2 K  
 DI 0.11000000 sec  
 ACQST 0.00000000 sec  
 ACQR 0.11000000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.45 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 20.000000 MHz  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 L0 MR pulse parameters  
 CK 22.80 cm  
 CT 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 WOP 20.000000 MHz  
 HZCN 241.33423 Hz/cm

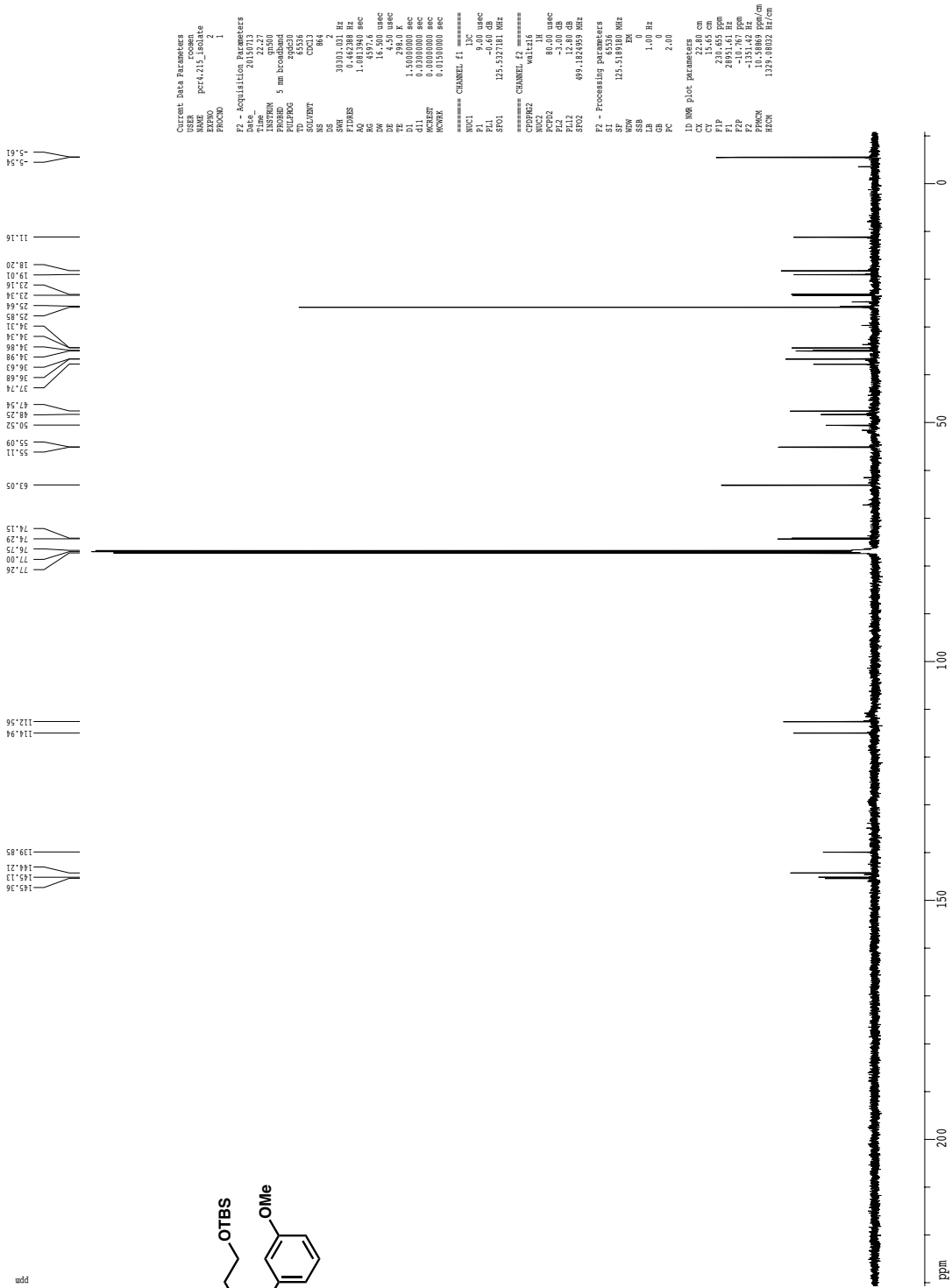




1H spectrum

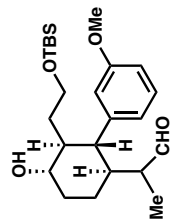
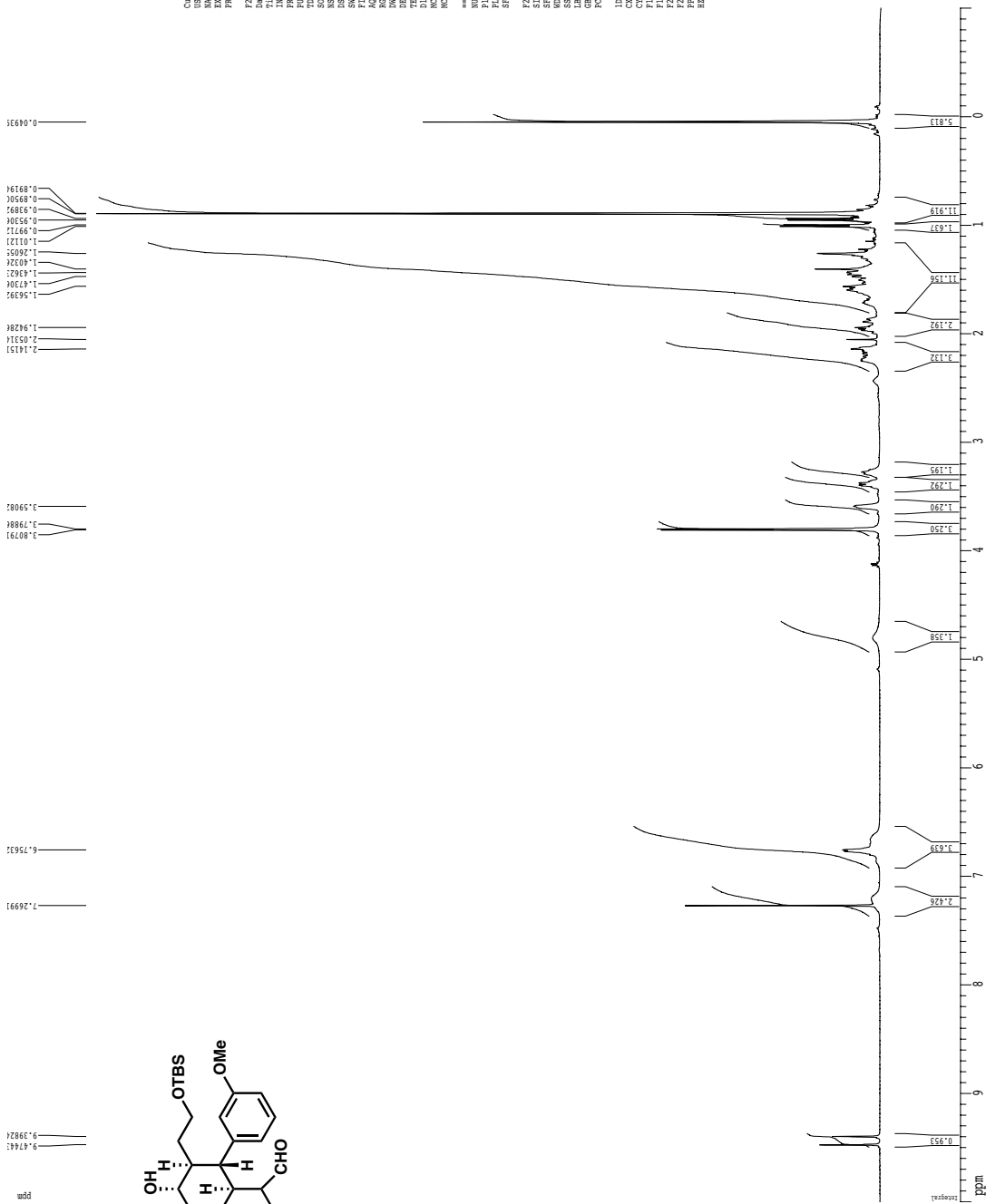


13C spectrum with 1H decoupling



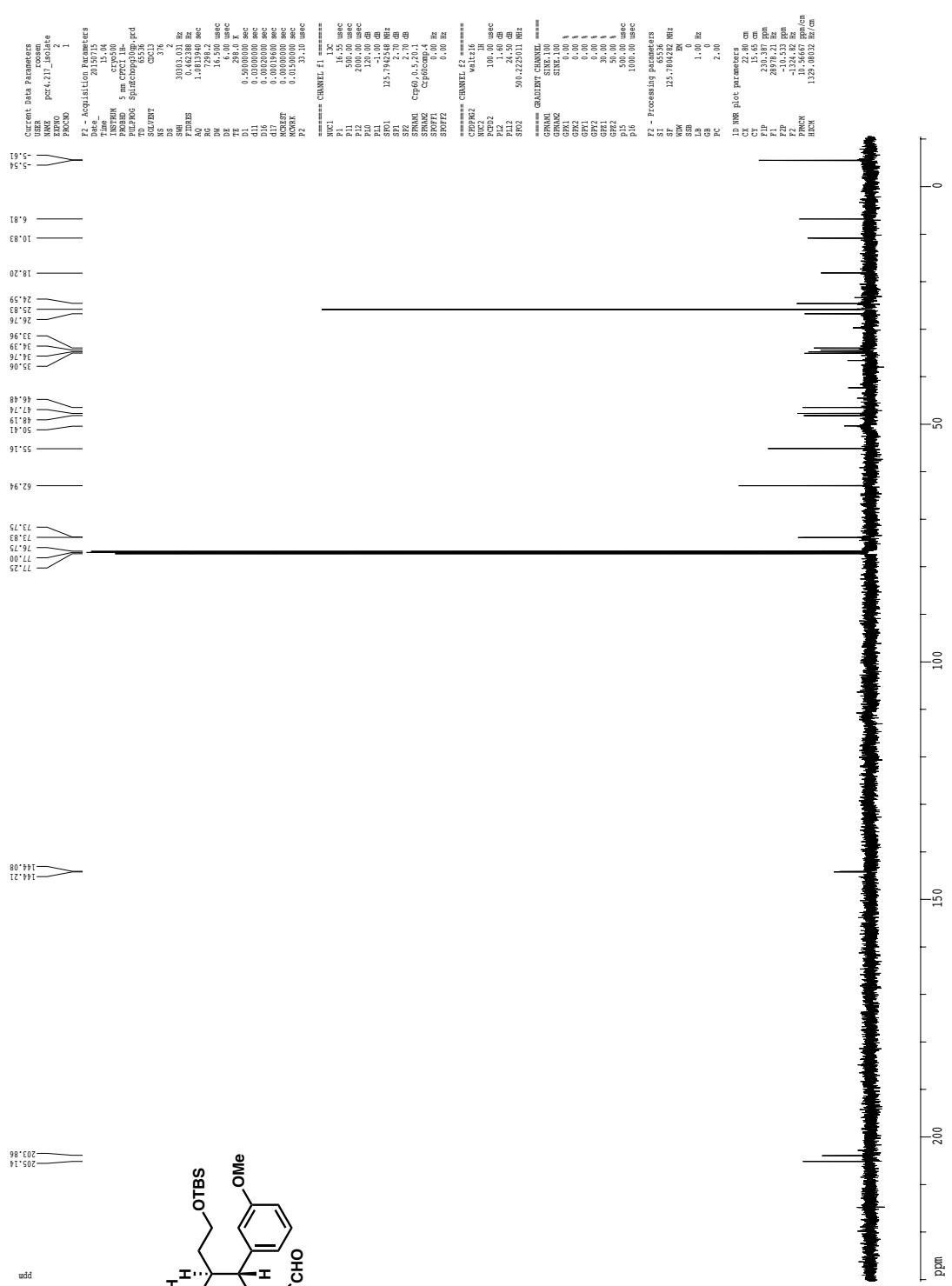


1H spectrum

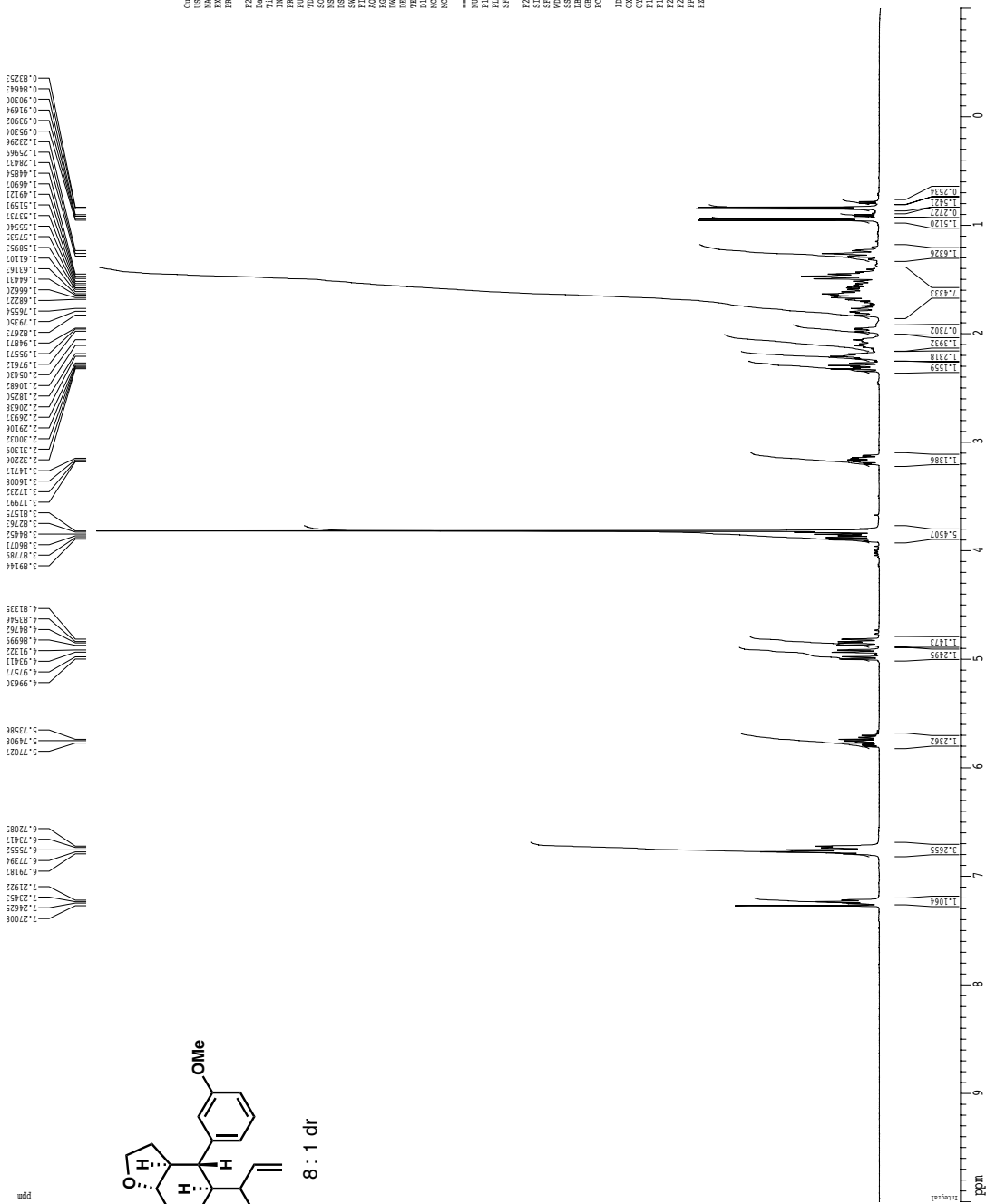


Current Data Parameters  
 NAME Pct4-211\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20101113  
 Time 11:00:00  
 INSTRUM ctyes500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250024 Hz  
 AQ 1.1996100 sec  
 RG 8  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 DCOffset 0.0000000 sec  
 ACQPRG 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 F41 500.225015 MHz  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WWP 8K  
 SSB 0  
 CB 0.0 Hz  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 F4 0.0000000 Hz/cm  
 HZCN 241.33423 Hz/cm

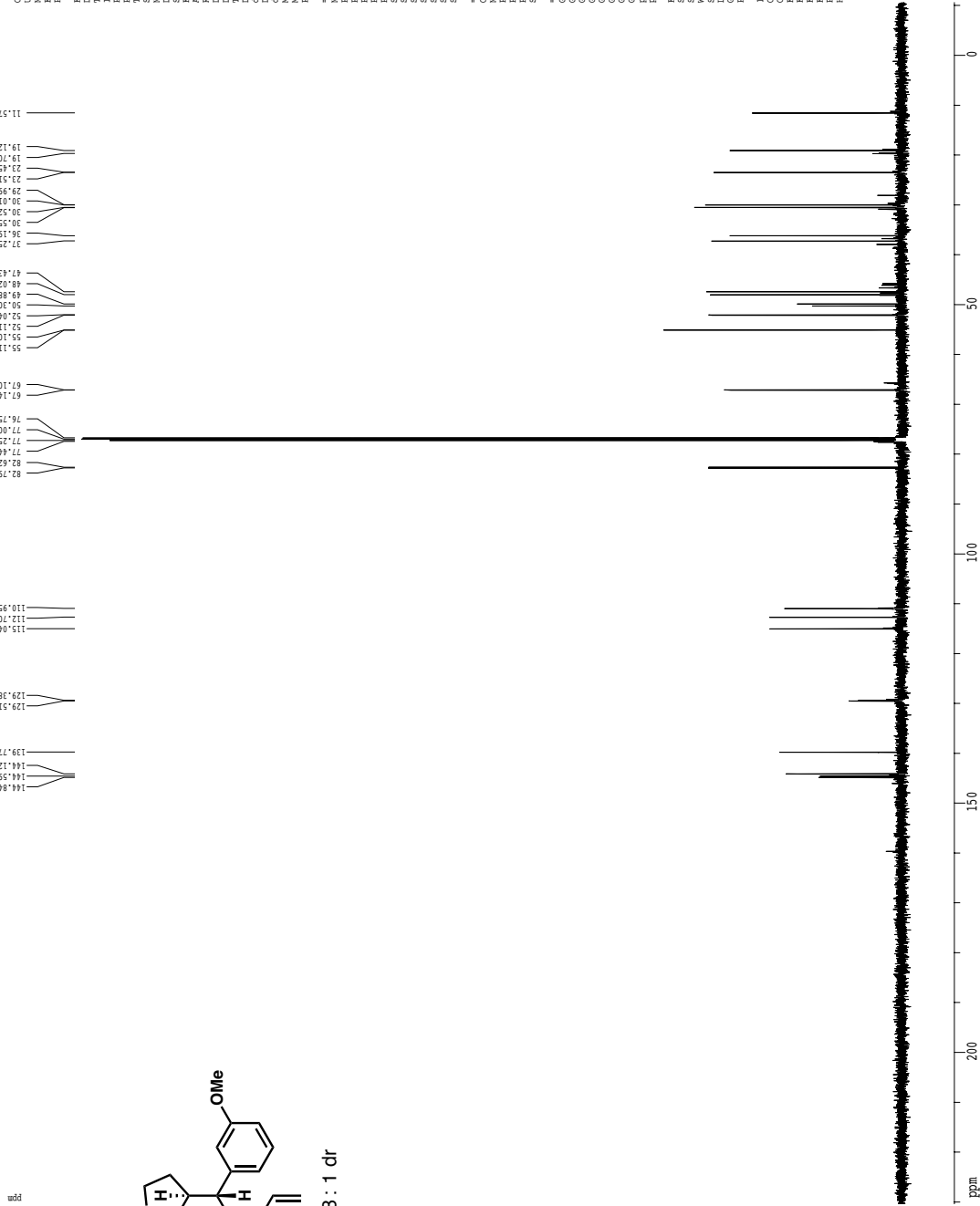
Z-restored spin-echo 13C spectrum with 1H decoupling



1H spectrum



Z-restored spin-echo <sup>13</sup>C spectrum with <sup>1</sup>H decoupling

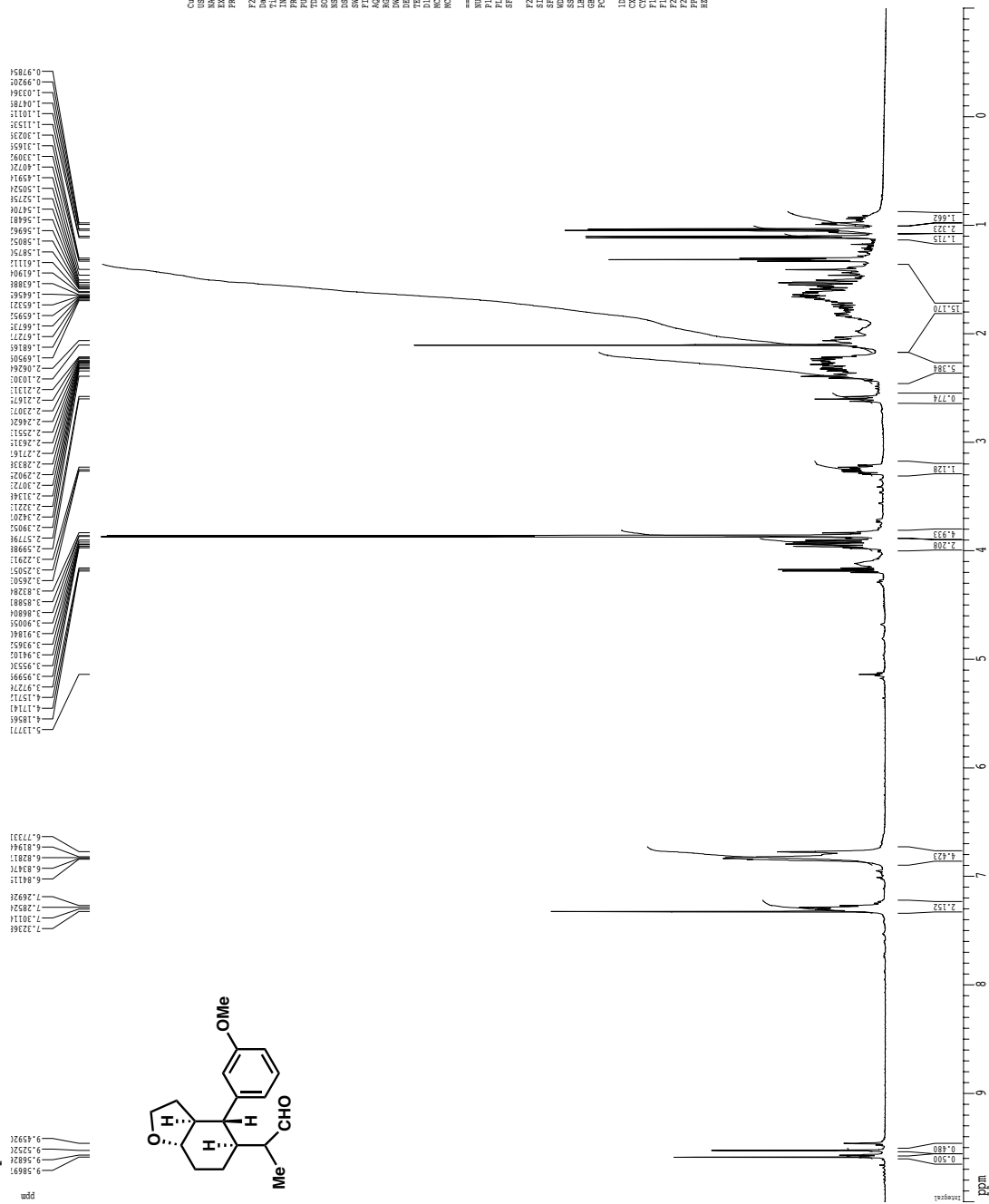


```

Current Data Parameters
=====
NAME          pect-122_150.d1e
EXPNO         2
PROCNO        1
F2 - Acq. Parameters
=====
Date_         20111107
Time          11.17
INSTRUM       cryo500
PROBHD        5mmCPY130
PULPROG       zgpg30
TD             65536
SOLVENT       CDCl3
NS            2
DS            4
SWH           30000.000 Hz
FIDRES        1.0813840 Hz
AQ            2896.3 usec
RG            6.00 usec
DE            6.00 usec
TE            298.2 K
NUC1          13C
NUC2          13C
NUC3          13C
NUC4          13C
NUC5          13C
NUC6          13C
NUC7          13C
NUC8          13C
NUC9          13C
NUC10         13C
===== CHANNEL F1 =====
PC1          16.55 usec
PL1          0.00 dB
PL2          0.00 dB
PL3          0.00 dB
PL4          0.00 dB
PL5          0.00 dB
PL6          0.00 dB
PL7          0.00 dB
PL8          0.00 dB
PL9          0.00 dB
PL10         0.00 dB
PL11         0.00 dB
PL12         0.00 dB
PL13         0.00 dB
PL14         0.00 dB
PL15         0.00 dB
PL16         0.00 dB
PL17         0.00 dB
PL18         0.00 dB
PL19         0.00 dB
PL20         0.00 dB
===== CHANNEL F2 =====
PCPRG2       waltz16
PL21         0.00 dB
PL22         0.00 dB
PL23         0.00 dB
PL24         0.00 dB
PL25         0.00 dB
PL26         0.00 dB
PL27         0.00 dB
PL28         0.00 dB
PL29         0.00 dB
PL30         0.00 dB
PL31         0.00 dB
PL32         0.00 dB
PL33         0.00 dB
PL34         0.00 dB
PL35         0.00 dB
PL36         0.00 dB
PL37         0.00 dB
PL38         0.00 dB
PL39         0.00 dB
PL40         0.00 dB
===== GRABBER CHANNEL =====
GRAB1        SINE_100
GRAB2        SINE_100
GRAB3        SINE_100
GRAB4        SINE_100
GRAB5        SINE_100
GRAB6        SINE_100
GRAB7        SINE_100
GRAB8        SINE_100
GRAB9        SINE_100
GRAB10       SINE_100
===== Processing parameters =====
SI            65536
SF            125.7604866 MHz
WDW           EM
SSB           0
LB            0
GB            0
PC            1.00 Hz
AQ            2.00
RG            2.00
DE            2.00
TE            298.2 K
===== IDMS file parameters =====
CT            15.45 cm
F1P           230.84 ppm
F2P           -10.536 ppm
F3P           -123.28 Hz/cm
F4P           1329.0832 Hz/cm

```

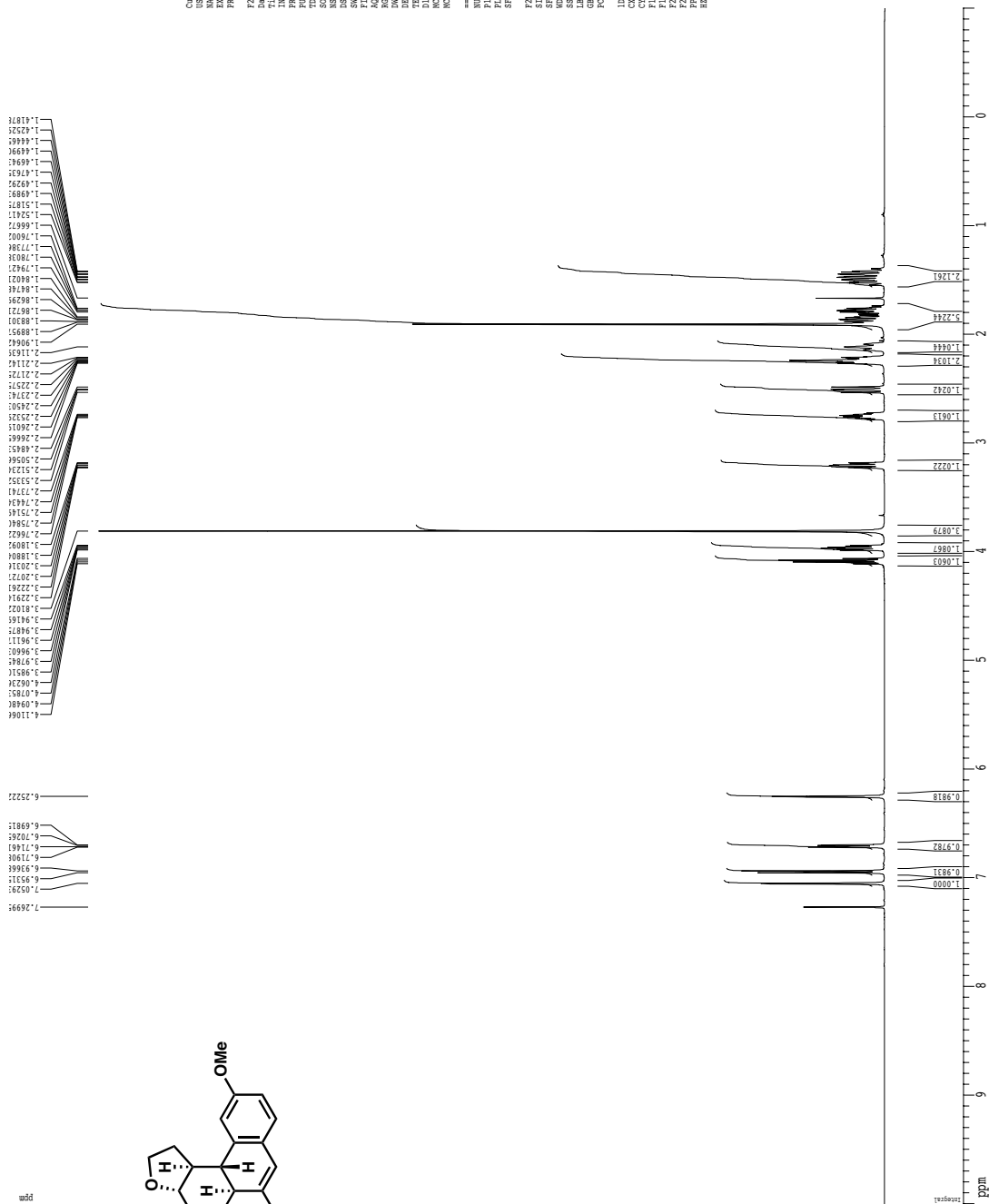
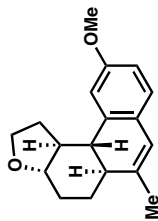
1H spectrum



Current Data Parameters  
 NAME pcr4-224\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2010722  
 TIME 11:52:00  
 INSTRUM cryo500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32748  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250204 Hz  
 AQ 1.1996134 sec  
 RG 64.3  
 DQ 62.400 usec  
 PR 0.00000000 sec  
 DE 294.4 usec  
 TE 300.2 K  
 D1 0.10000000 sec  
 ACQSF 0.00000000 sec  
 DECR 0.11300000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6536  
 WF 500.225015 MHz  
 WOP 0  
 SSB 0  
 GB 0  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 HPCON 241.33421 Hz/cm



1H spectrum

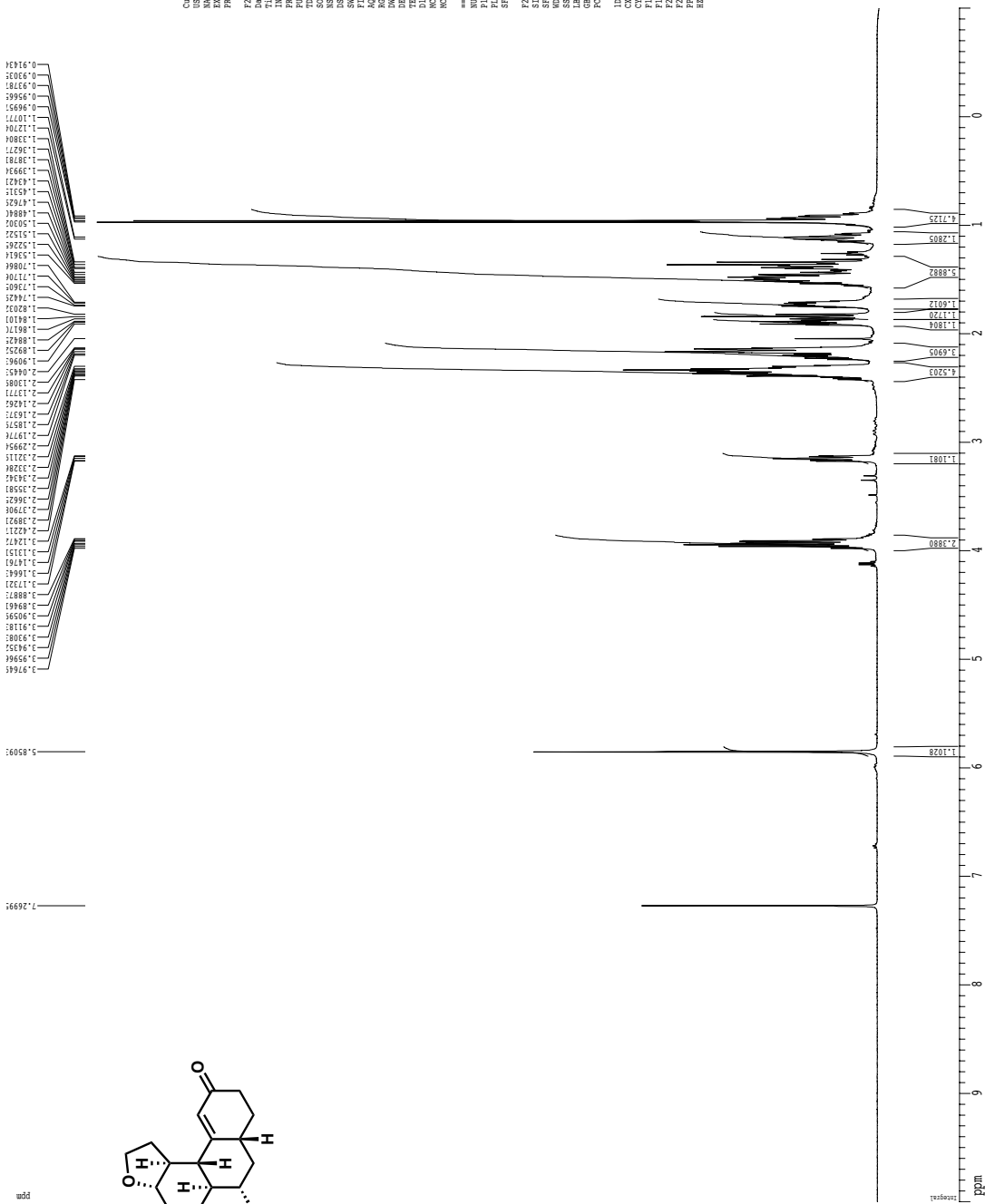


Current Data Parameters  
 NAME Pct4.283\_13-17  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20151012  
 Time 09:00:00  
 INSTRUM cty6500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SFO1 500.136261 MHz  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250024 Hz  
 AQ 1.1991614 sec  
 RG 9  
 DQ 62.400 usec  
 DE 6.000 use

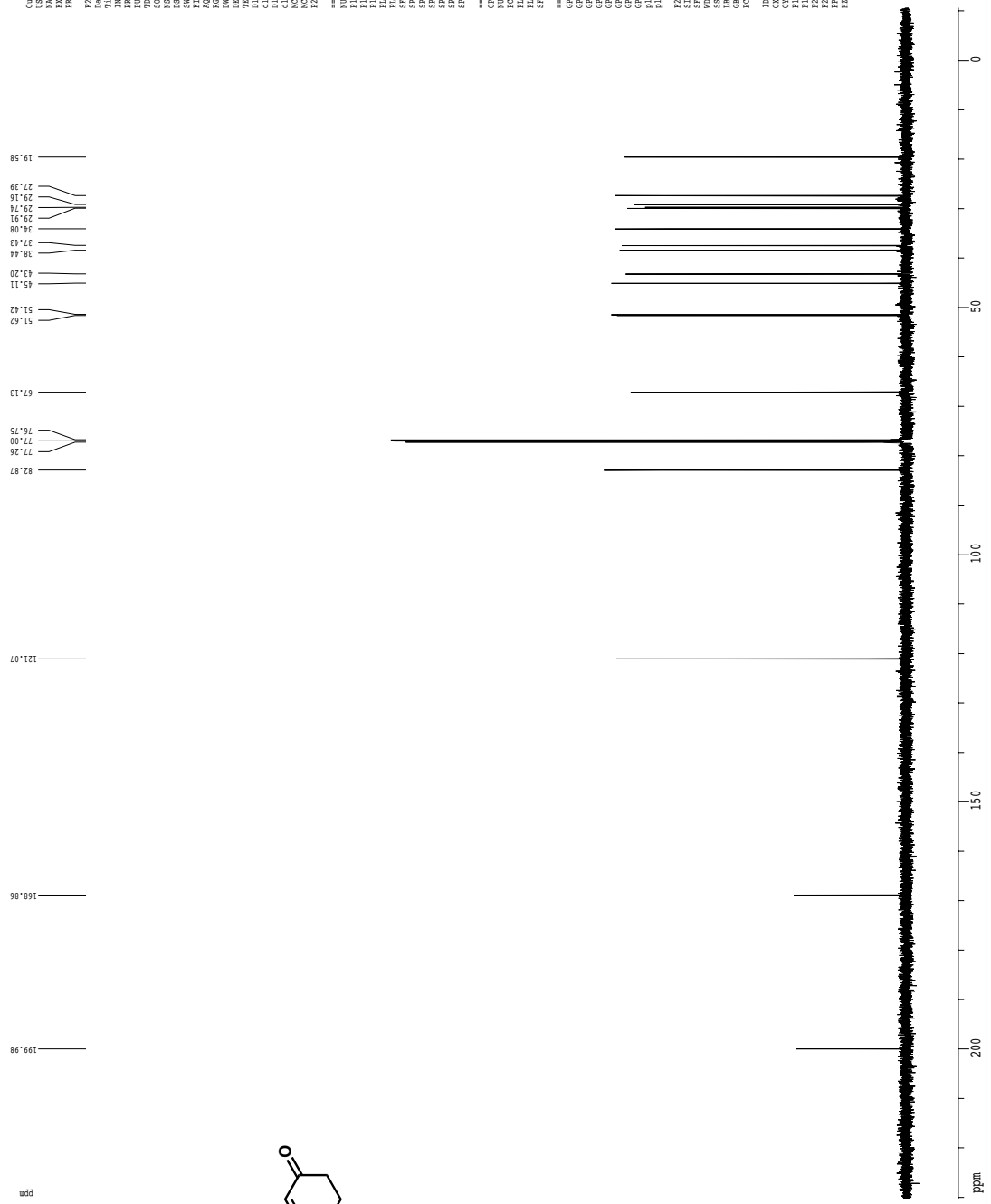




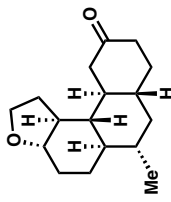
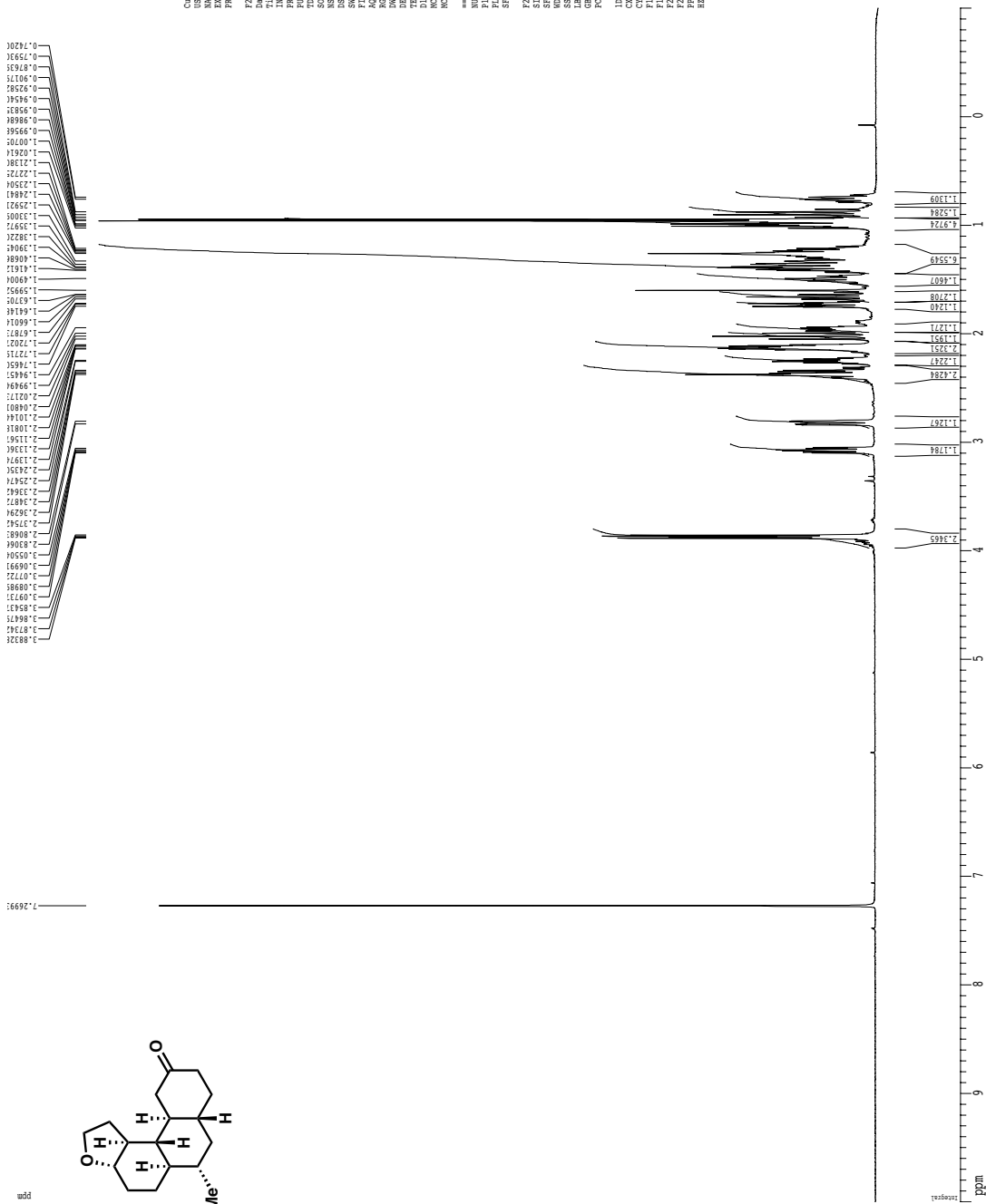
1H spectrum



Z-restored spin-echo <sup>13</sup>C spectrum with <sup>1</sup>H decoupling



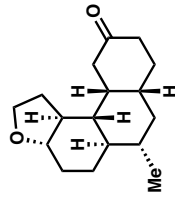
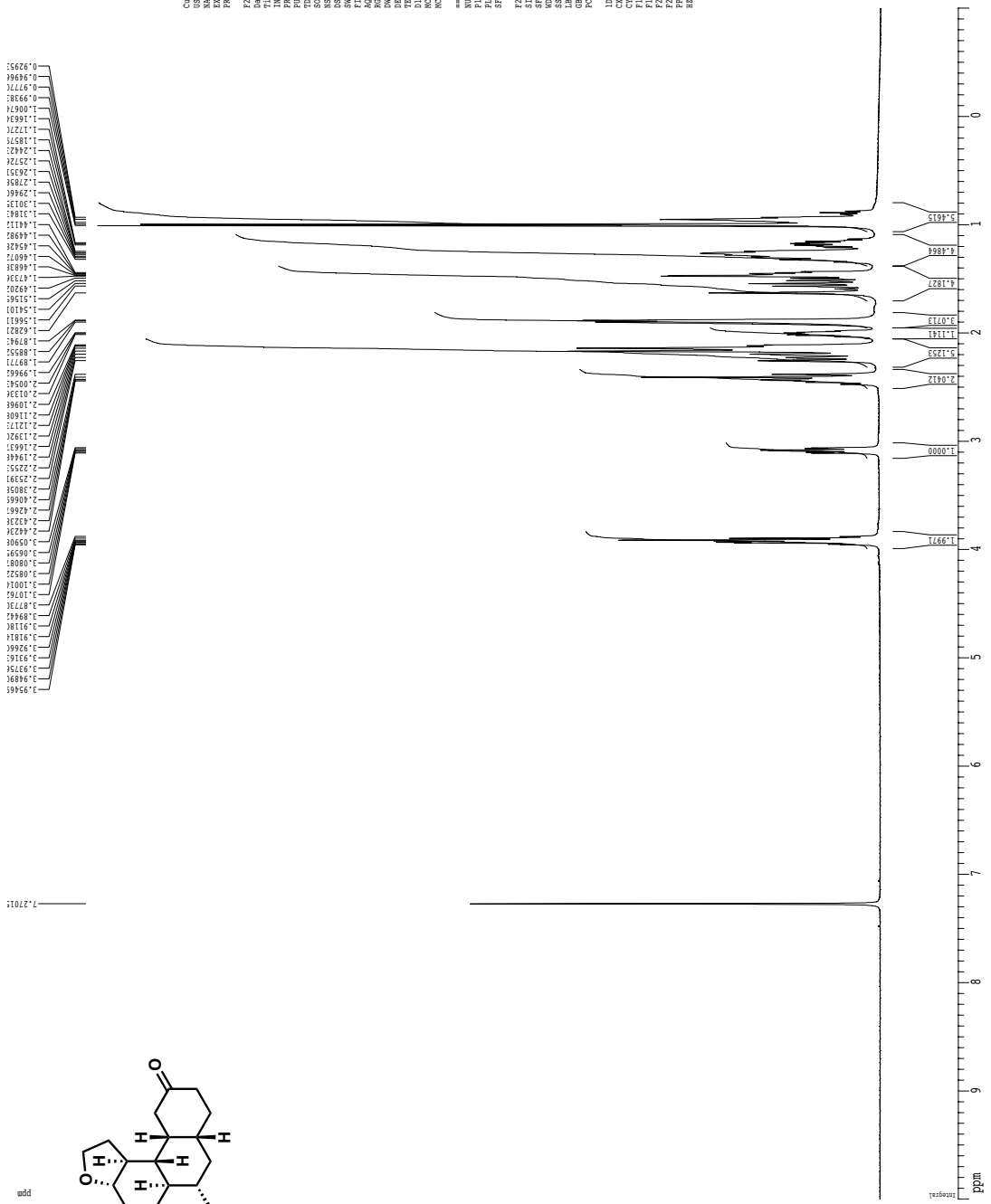
1H spectrum



Current Data Parameters  
 NAME Pct4-271\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20101223  
 Time 11:00:00  
 INSTRUM cryo500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 8012.820 Hz  
 FIDRES 0.250024 Hz  
 AQ 1.1999999 sec  
 RG 1013  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 D1 0.10000000 sec  
 ACQMSF 0.00000000 sec  
 SOLOR 0.11300000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 30.000000 MHz  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CK 22.80 cm  
 CT 15.00 cm  
 CF 100.625000 MHz  
 FI 500.225015 MHz  
 FFP -1.000 ppm  
 FZ -500.22 Hz  
 HPCON 241.33423 Hz/cm



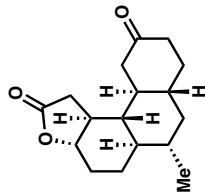
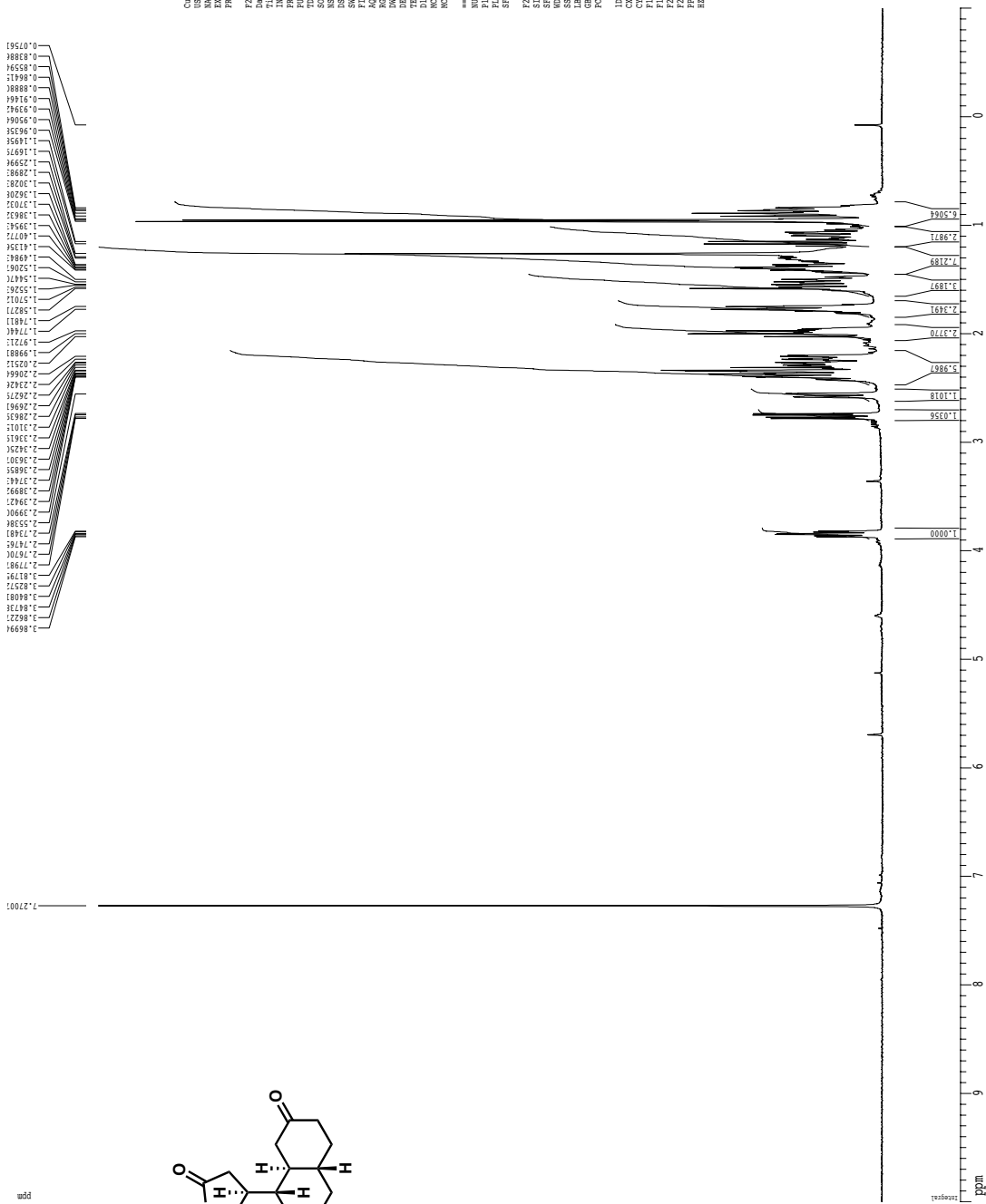
1H spectrum



Current Data Parameters  
 NAME pcc4\_244\_C191ALS  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20090525  
 Time 22:07  
 INSTRUM cryo600  
 PROBP0 5 mm CPCL 13C  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 DS 2  
 SWH 802.820 MHz  
 FWHM 15.998 Hz  
 AQ 1.9198451 sec  
 RG 62.711  
 DE 6.00 uSAC  
 TE 298.2 K  
 D 0.1000000 sec  
 DECTEST 0.1000000 sec  
 ACQRES 0.1500000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.50 uSAC  
 PL 0.00 dB  
 SFO1 500.220013 MHz  
 F2 - Processing parameters  
 SI 32768  
 SF 500.2200048 MHz  
 EQ 1.0000000 sec  
 LB 0.30 Hz  
 GB 0  
 PC 6.00  
 ID NMR plot parameters  
 SI 32768  
 CT 155.00 cm  
 FIP 10.000 ppm  
 FID 500.2200000 Hz  
 F2 -500.22 Hz  
 FWHM 0.48246 ppm/cm  
 BUCK 241.58423 Hz/cm



1H spectrum



Current Data Parameters  
 NAME Pcs3\_031\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20151116  
 Time 11:05:00  
 INSTRUM cty6300  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SFO1 500.136261 MHz  
 FIDRES 0.250026 Hz  
 AQ 1.199161 sec  
 RG 4.3  
 DQ 62.400 usec  
 DE 6.00 usec  
 TE 298.2 K  
 DI 0.1000000 sec  
 ACQST 0.0000000 sec  
 ACQR 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.50 usec  
 PL1 0 dB  
 SFO1 500.136261 MHz  
 F2 - Processing parameters  
 SI 6536  
 SF 500.136261 MHz  
 WOP 0  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 L0 MR phase parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 F1 500.136261 MHz  
 F2 -1.000 ppm  
 F3 -500.136261 MHz  
 F4 500.136261 MHz  
 HSCN 241.33423 Hz/cm

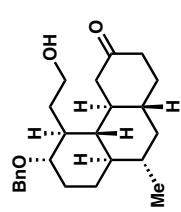
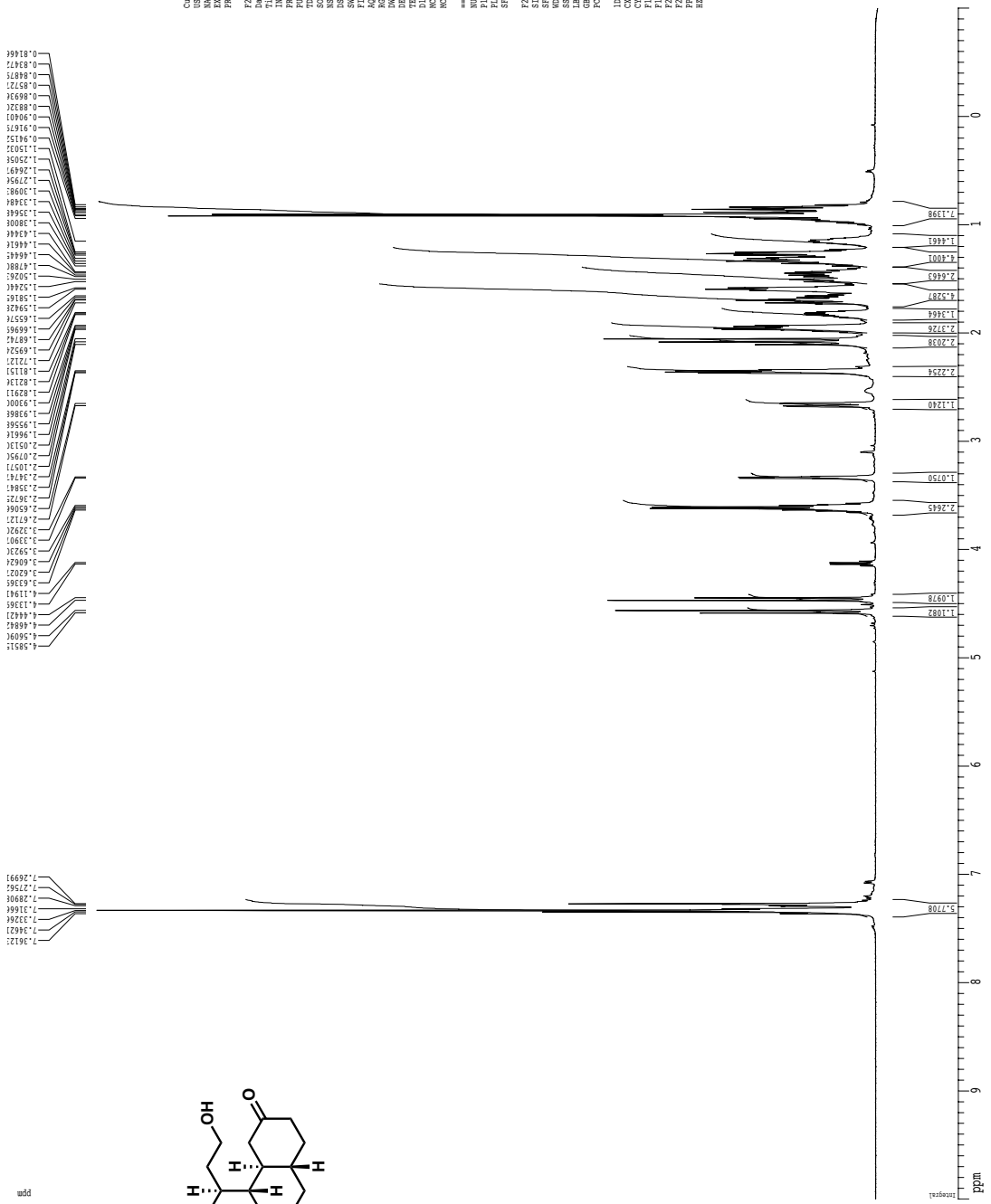






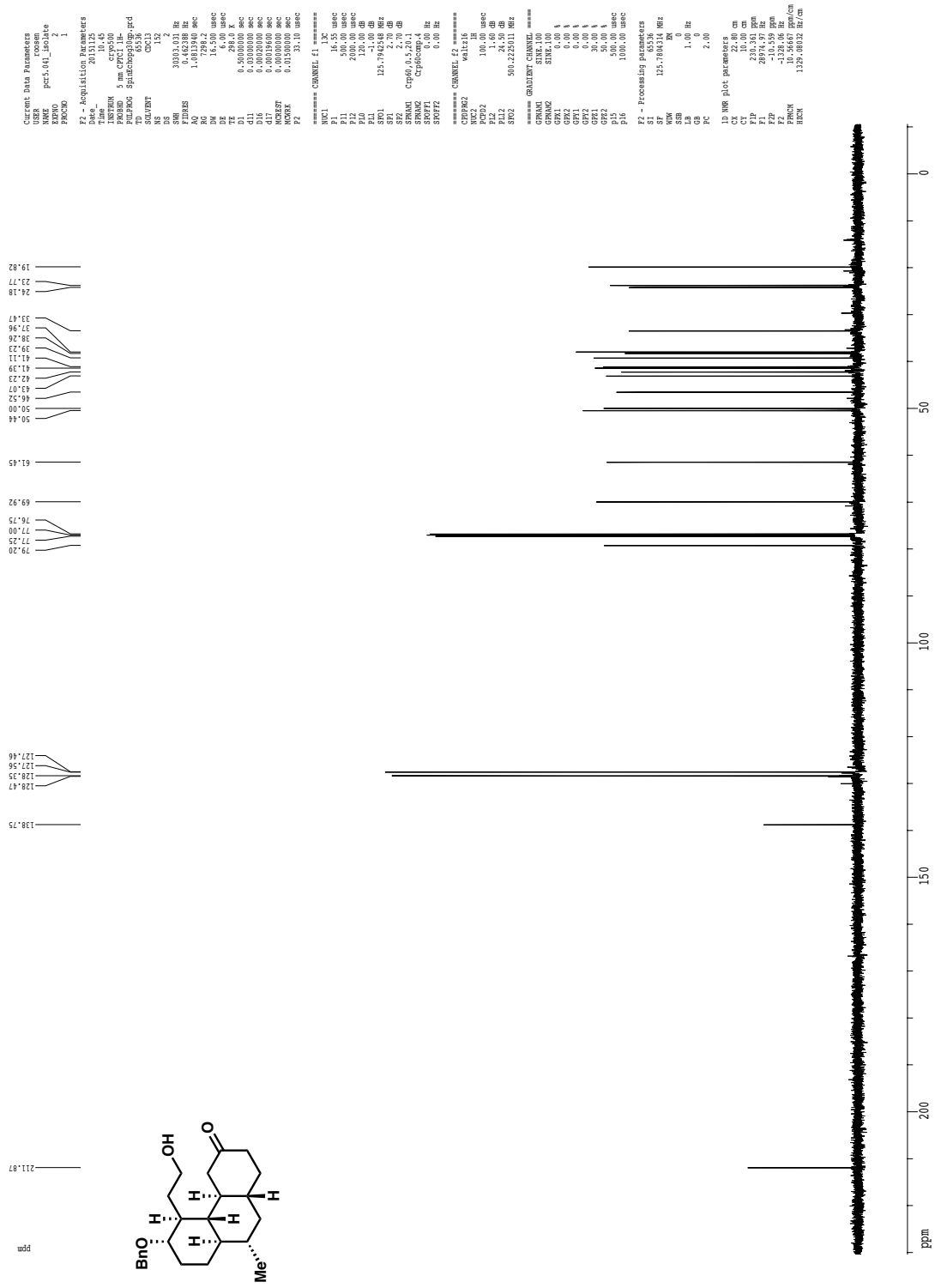


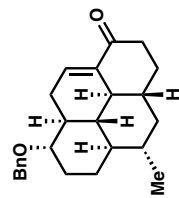
<sup>1</sup>H spectrum



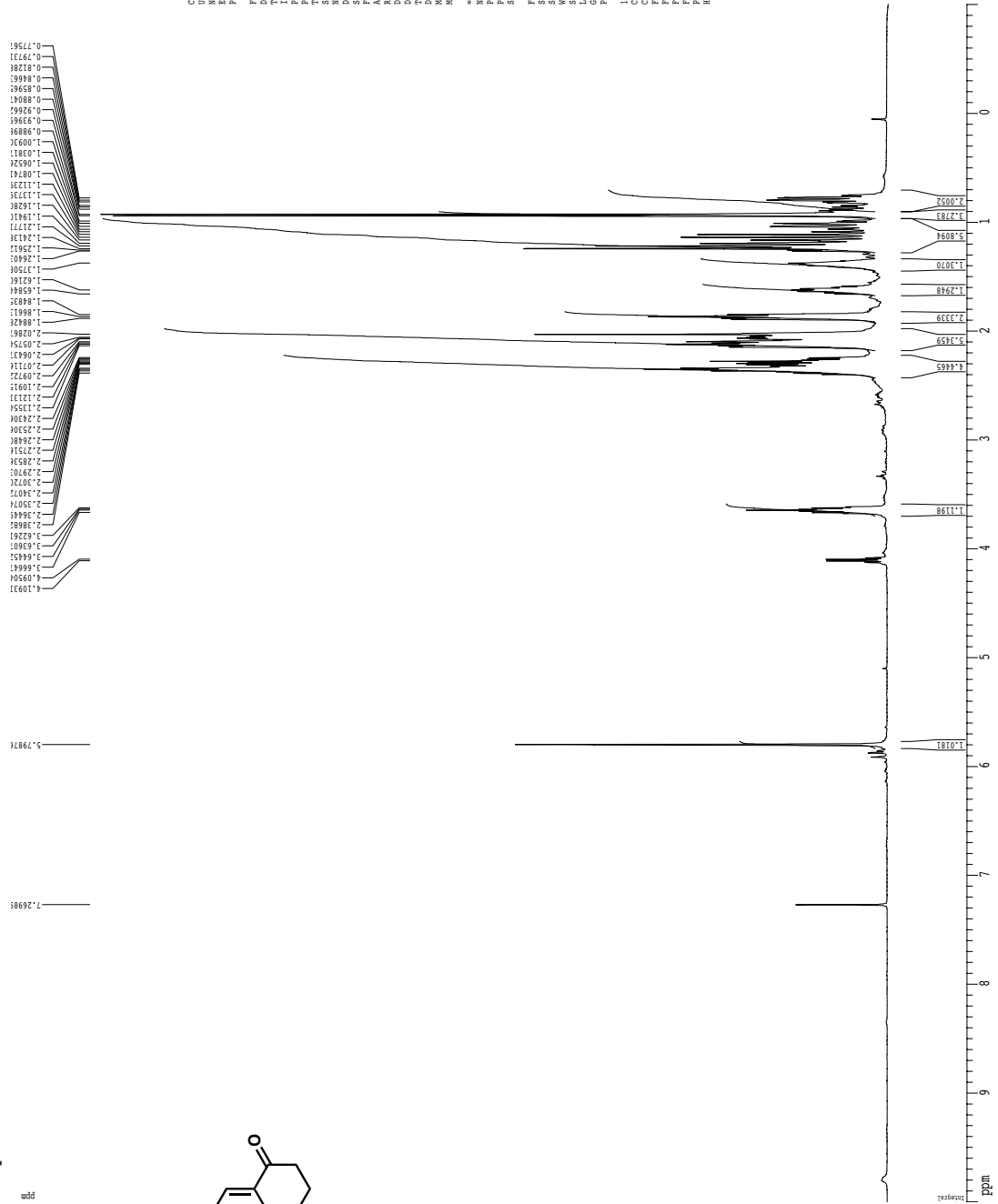
Current Data Parameters  
 NAME Pct3.04\_Isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 201112  
 INSTRUM crysov0  
 PROBHD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TO 3248  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SFO 802.820 MHz  
 FIDRES 0.25026 Hz  
 RG 1.59916 Hz sec  
 SG 5.7  
 DM 62.400 usec  
 DE 6.000 usec  
 TE 298.4 K  
 D1 0.10000000 sec  
 ACQRES 0.00000000 sec  
 NSMR 0.12300000 sec  
 ===== CHANNEL f1 =====  
 NU1 1  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 NS 3248  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MR pulse parameters  
 CX 22.80 cm  
 CT 15.00 cm  
 CP 9.50 cm  
 FI 500.225015 MHz  
 FFP -1.000 ppm  
 FZ -500.22 Hz  
 GPCON 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling





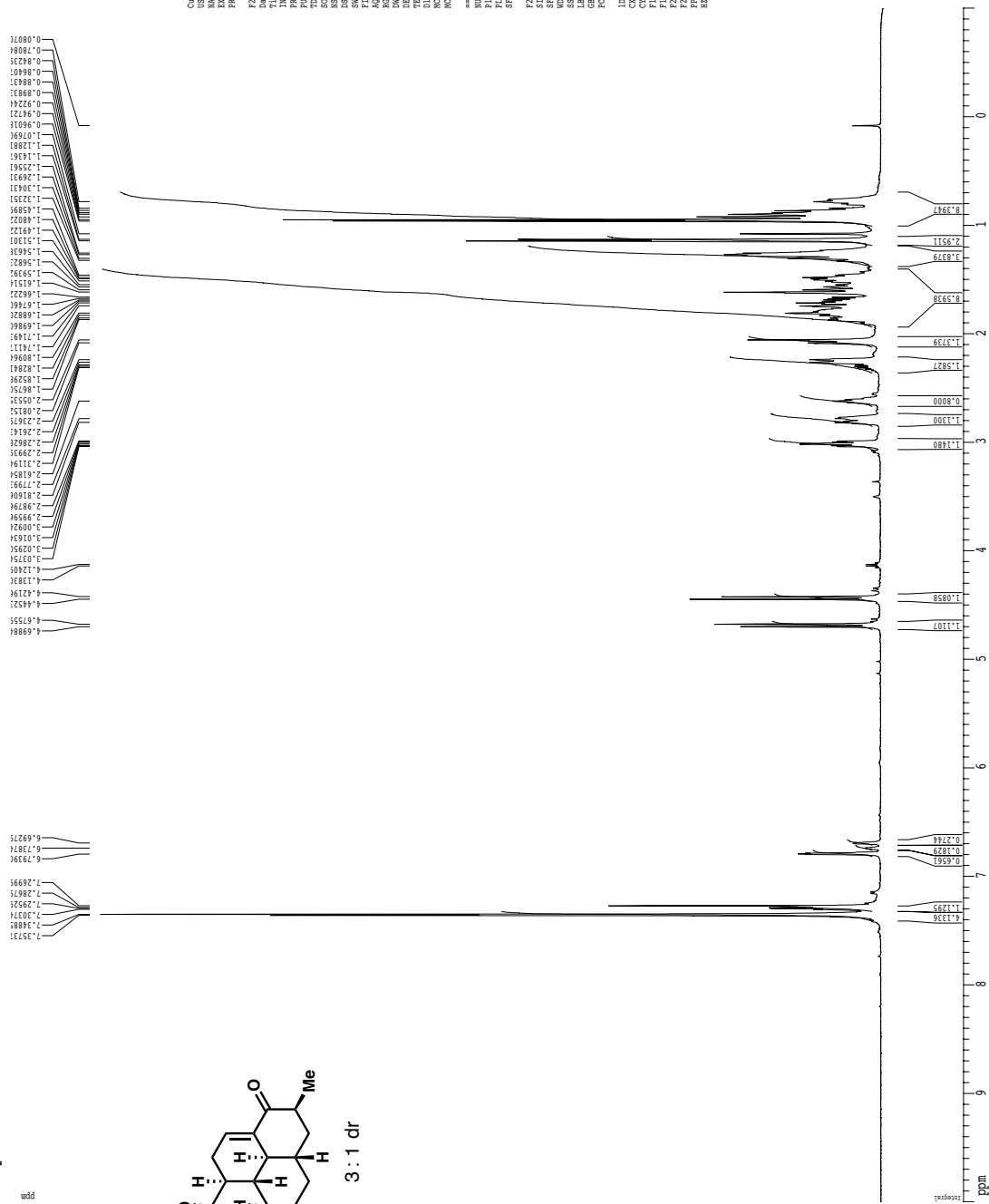
<sup>1</sup>H spectrum



Current Data Parameters  
 NAME pcr4.024\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20150112  
 Time 09:11  
 INSTRUM cryo500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 802.820 Hz  
 FIDRES 0.250204 Hz  
 AQ 1.399648 sec  
 RG 4.5  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 ACQRES 0.0000000 sec  
 NSMR 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.50 usec  
 PL1 0.00 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 30.0000000 sec  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 F4 241.33423 Hz/cm

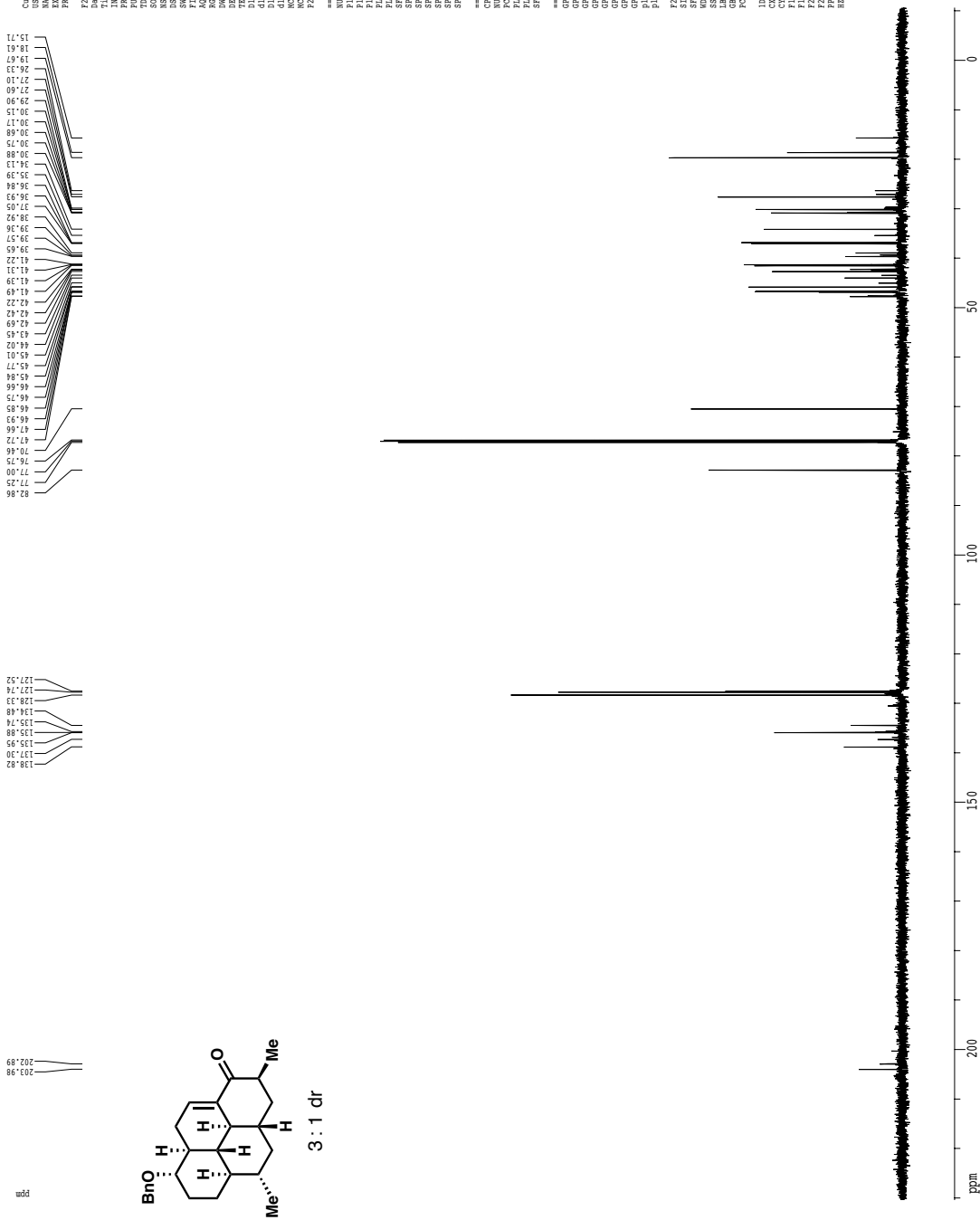


1H spectrum



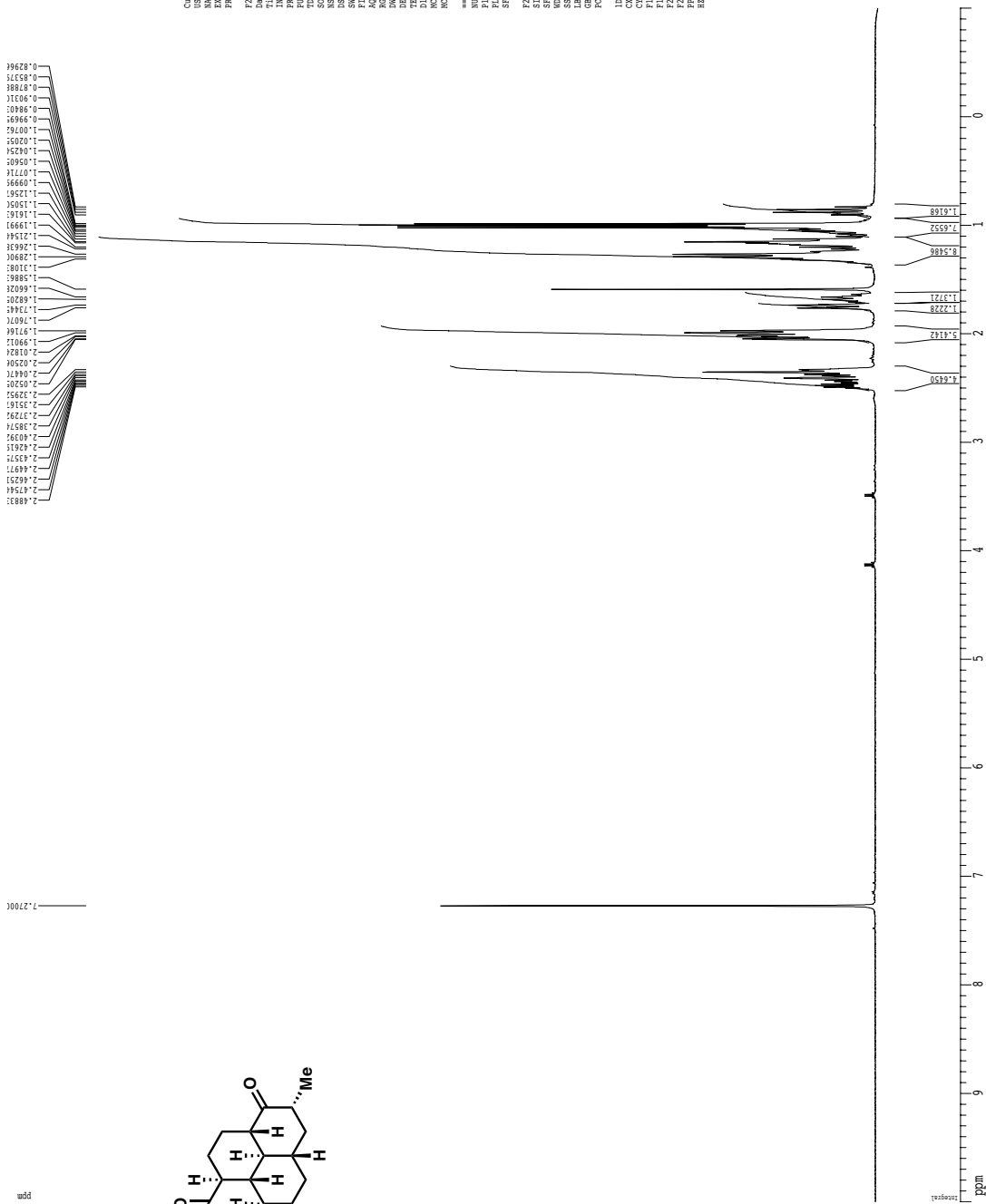
Current Data Parameters  
 NAME Pcs3\_046\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2013129  
 Time 11.17  
 INSTRUM cty6500  
 PROBRD 5 mm CPCL 1H-  
 PULPROG zgpg30  
 TD 32748  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 802.820 Hz  
 FIDRES 0.250204 Hz  
 AQ 1.199161 sec  
 RG 317  
 DQ 1.7  
 ZM 2.00 usec  
 GPC 0.000000 sec  
 PC 294.4 usec  
 DI 0.11000000 sec  
 ACQRES 0.00000000 sec  
 SFO1 500.225015 MHz  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.45 usec  
 PL1 -1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 2.00 usec  
 SSB 0  
 GB 0.00 Hz  
 PC 2.00  
 =====  
 L0 MR parameters  
 CK 22.80 cm  
 CT 15.00 cm  
 CP 1.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 GAMMA 1.0000000000000000  
 HZCN 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



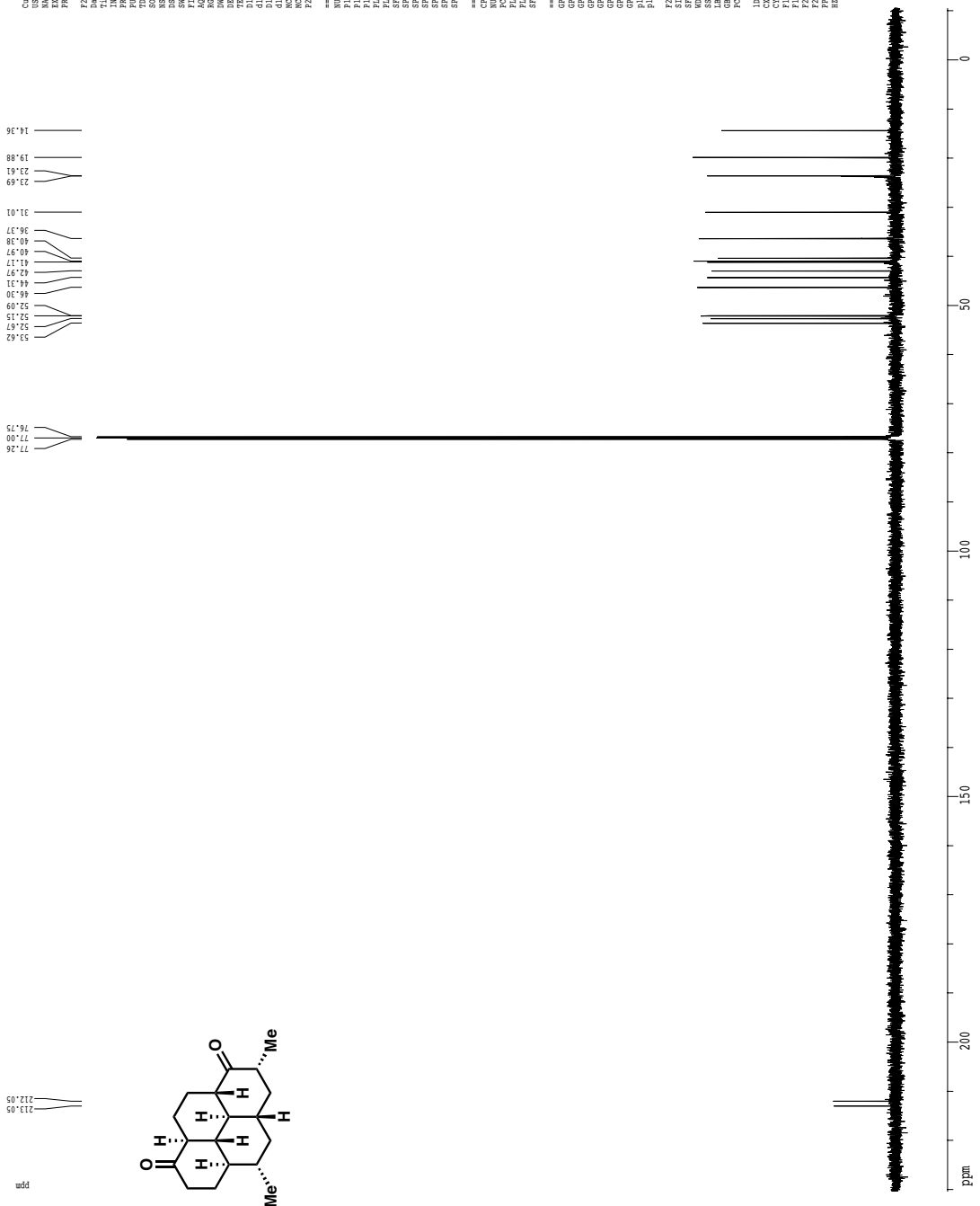


1H spectrum



Current Data Parameters  
 NAME Pct3\_051\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 201109  
 INSTRUM cty6500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32748  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SFO1 500.225015 MHz  
 FIDRES 0.250024 Hz  
 RG 674  
 ACQ 1.391611 sec  
 FWHM 0.43  
 DM 62.400 usec  
 DE 6.00 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 DCOffset 0.0000000 sec  
 ACQPRG 0.1330000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 3000000000 Hz  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CT 10.00 cm  
 CP 10.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 WCON 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



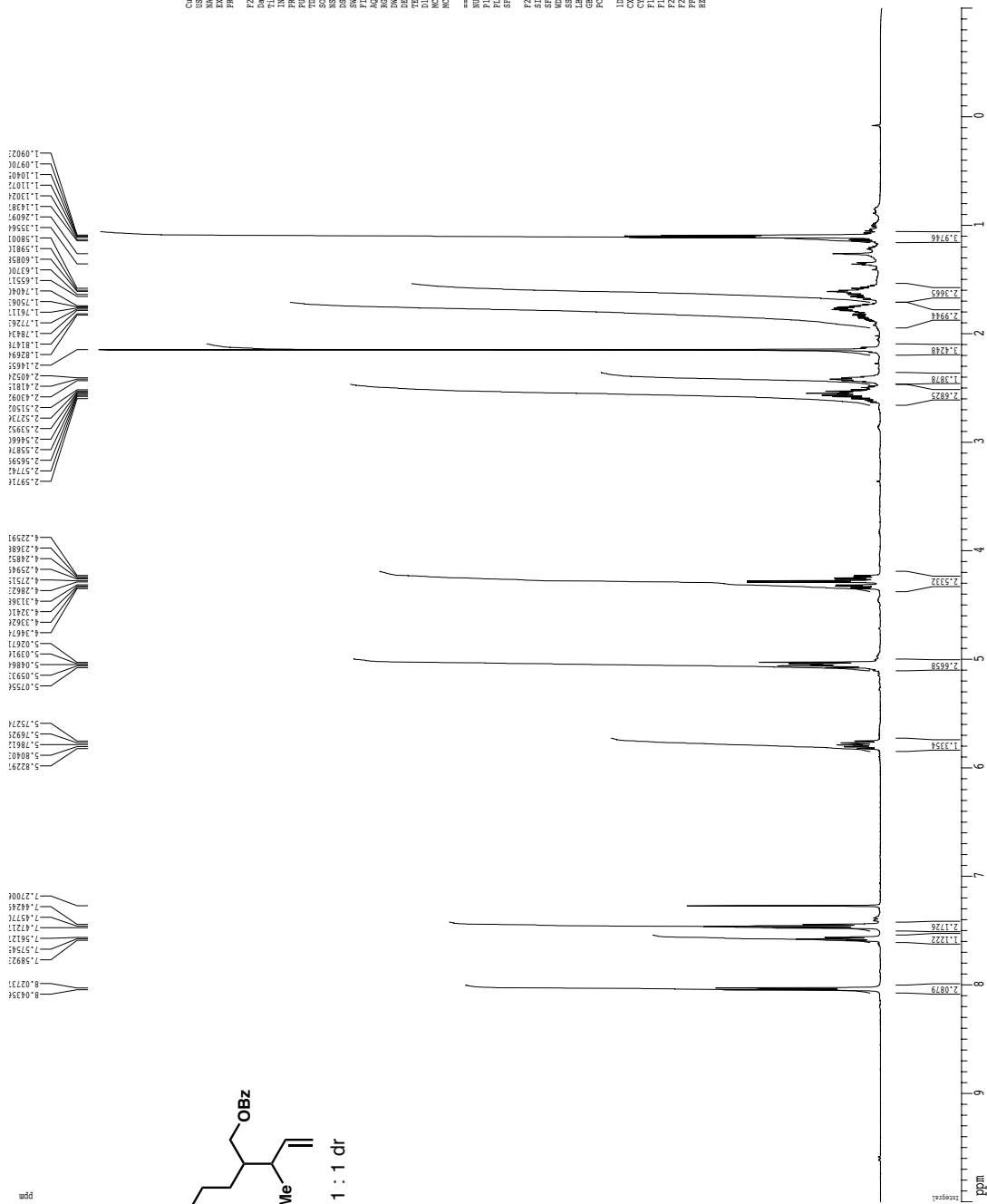
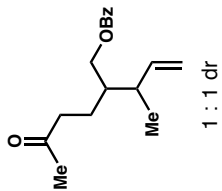
```

Current Data Parameters
NAME          pct-5.051_500Hz
EXPNO         2
PROCNO       1
F2 - Acquisition Parameters
Date_         2011.12.29
Time          11.12
INSTRUM      cryo500
PROBHD       5mm CPY130
PULPROG      zgpg30
TD           65536
SOLVENT      CDCl3
NS           4
DS           2
SWH          30303.938 Hz
FIDRES       1.0813340 Hz
AQ           4597.6
RG           65.00
DE           6.500 uSec
TE           0.2983 K
NUC1         13
NUC2         13
NUC3         13
NUC4         13
NUC5         13
NUC6         13
NUC7         13
NUC8         13
NUC9         13
NUC10        13
NUC11        13
NUC12        13
NUC13        13
===== CHANNEL f1 =====
P1          16.55 uSec
P2          200.00 uSec
P3          200.00 uSec
P4          200.00 dB
P5          -1.00 dB
P6          125.7922562 Hz
P7          2.70 dB
P8          2.70 dB
SFO1        CQ60.632811
SFO2        CQ60.632811
SFO3        CQ60.632811
SFO4        CQ60.632811
SFO5        CQ60.632811
SFO6        CQ60.632811
SFO7        CQ60.632811
SFO8        CQ60.632811
SFO9        CQ60.632811
SFO10       CQ60.632811
SFO11       CQ60.632811
SFO12       CQ60.632811
===== CHANNEL f2 =====
COPROG2    waltz16
PCPD2      100.00 uSec
P42        1.00 dB
P52        500.2225411 MHz
===== CHANNEL CHANNEL =====
GRANU      32768
SINW1      3100
SINW2      3100
SINW3      3100
SINW4      3100
SINW5      3100
SINW6      3100
SINW7      3100
SINW8      3100
SINW9      3100
SINW10     3100
SINW11     3100
SINW12     3100
===== Processing parameters =====
SI          65536
SF          125.7604812 MHz
WDW         EM
SSB         0
LB          1.00 Hz
GB          0
PC          2.00
===== ID MS parameters =====
CT          15.45 cm
PIF         230.87 ppm
F2F         -10.533 ppm
F2          -1224.82 Hz/cm
F2F2        1329.8832 Hz/cm
  
```





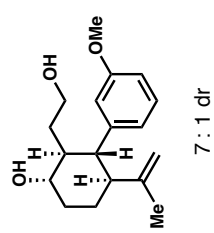
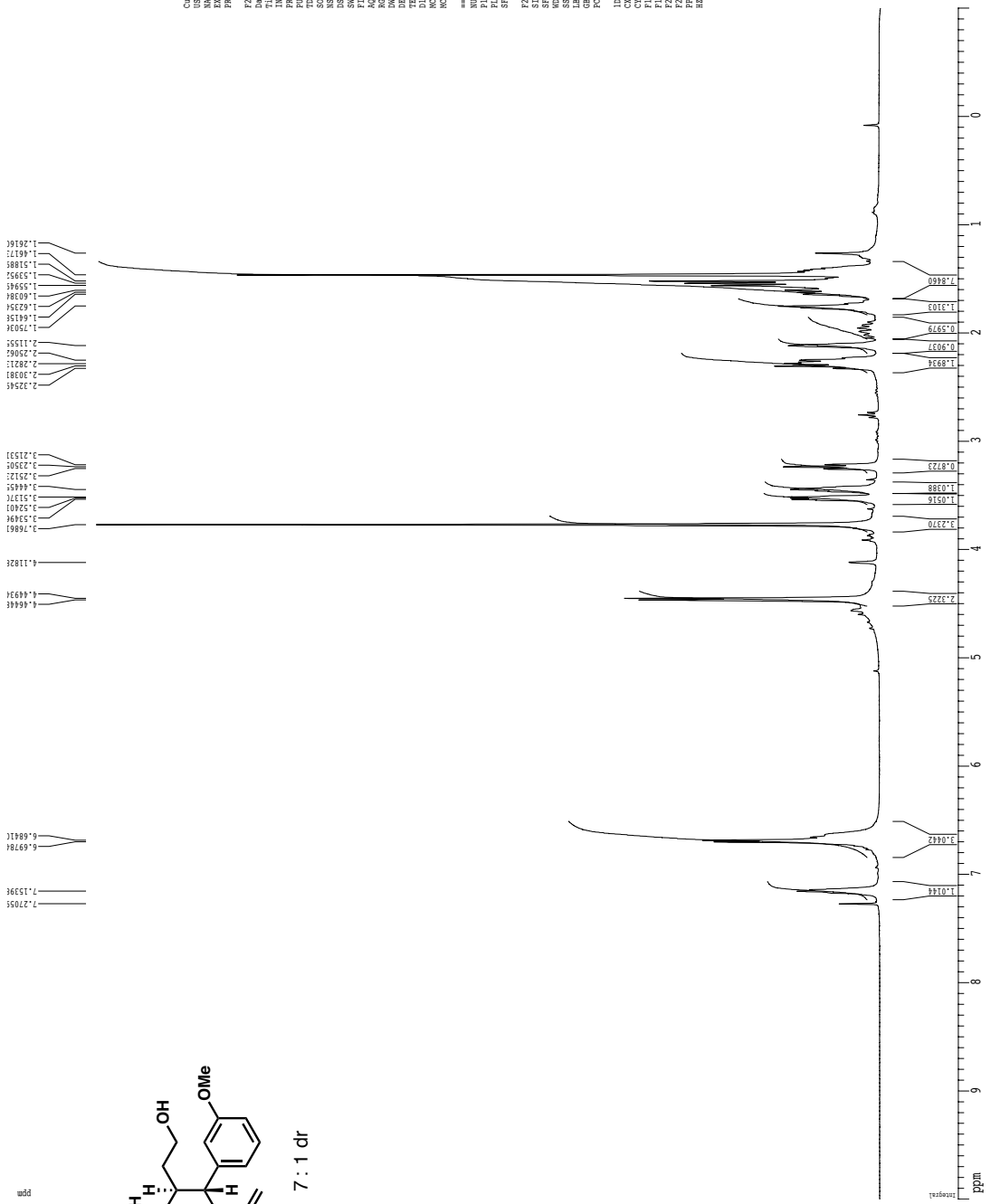
1H spectrum



Current Data Parameters  
 NAME Pct3\_058\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 201313  
 TIME 11:02:13  
 INSTRUM ctyes00  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SFO1 8012.820 MHz  
 FIDRES 0.250024 Hz  
 AQ 1.199161 sec  
 RG 8  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 DCREST 0.0000000 sec  
 ACQBR 0.1130000 sec  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.45 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.2201500 MHz  
 WOP 500.2201500 MHz  
 SSB 0  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 FX 500.2201500 MHz  
 FY 500.2201500 MHz  
 FZ -1.000 ppm  
 TX 500.2201500 MHz  
 HPCON 241.33423 Hz/cm

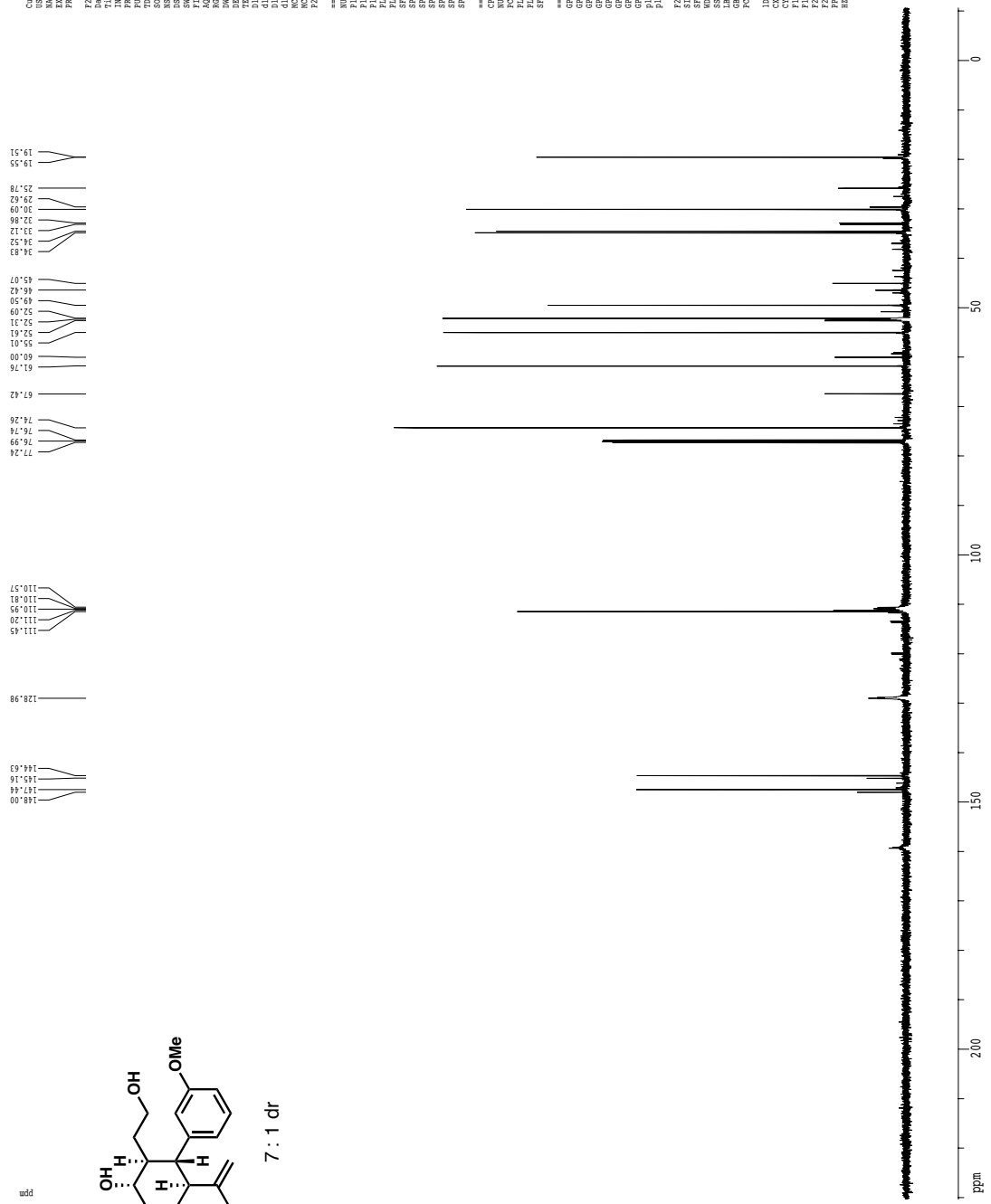


1H spectrum



Current Data Parameters  
 NAME Pc3.123\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 2010039  
 INSTRUM cryo500  
 PROBHD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 SH 802.820 Hz  
 FIDRES 0.250024 Hz  
 AQ 1.1991632 sec  
 RG 312  
 DM 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 D1 0.1000000 sec  
 ACQSF 0.0000000 sec  
 NSMR 0.1000000 sec  
 ===== CHANNEL f1 =====  
 NUC1 1H  
 P1 7.45 usec  
 PL1 0.00 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 NS 32768  
 HSI 0  
 SSF 0.00 Hz  
 GB 0.00 Hz  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CZ 15.00 cm  
 F1 500.225015 MHz  
 F2 500.225015 MHz  
 F3 -1.000 ppm  
 F4 -500.22 Hz  
 F5 0.0000000 Hz/cm  
 HZCN 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling

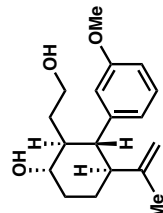


Current Data Parameters  
 NAME test5\_125\_150Date  
 EXPNO 2  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20160805  
 Time 10:45  
 INSTRUM cryo500  
 PULPROG zgpg30  
 PROCNO 8  
 F2 - Processing parameters  
 SI 65536  
 SF 125.760479 MHz  
 DS 4  
 AS 0  
 L8 1.00 Hz  
 SFO 2.00  
 ID TAG parameters  
 CT 10.00 cm  
 F1 230.210 MHz  
 F2 -10.610 ppm  
 F2 -1334.43 Hz  
 FREQ 1329.88132 Hz/cm

===== CHANNEL F1 =====  
 NU1 16.55 uSAC  
 F1 500.00 uSAC  
 F2 500.00 uSAC  
 F3 200.00 dB  
 F4 125.760479 MHz  
 SFO 2.70 dB  
 SFR 0.00 Hz  
 SFOFF 0.00 Hz  
 ===== CHANNEL F2 =====  
 CPDPR2 waltz16  
 PCPD2 100.00 uSAC  
 F1 1.00 dB  
 F2 500.225101 MHz  
 ===== CHANNEL CHANNEL =====  
 GRANU 100  
 STW 100  
 GPC2 0.00 A  
 GPC1 0.00 A  
 GPC3 0.00 A  
 GPC4 0.00 A  
 GPC5 30.00 A  
 GPC6 50.00 A  
 GPC7 1000.00 uSAC  
 GPC8 2.00

F2 - Processing parameters  
 SI 65536  
 SF 125.760479 MHz  
 DS 4  
 AS 0  
 L8 1.00 Hz  
 SFO 2.00

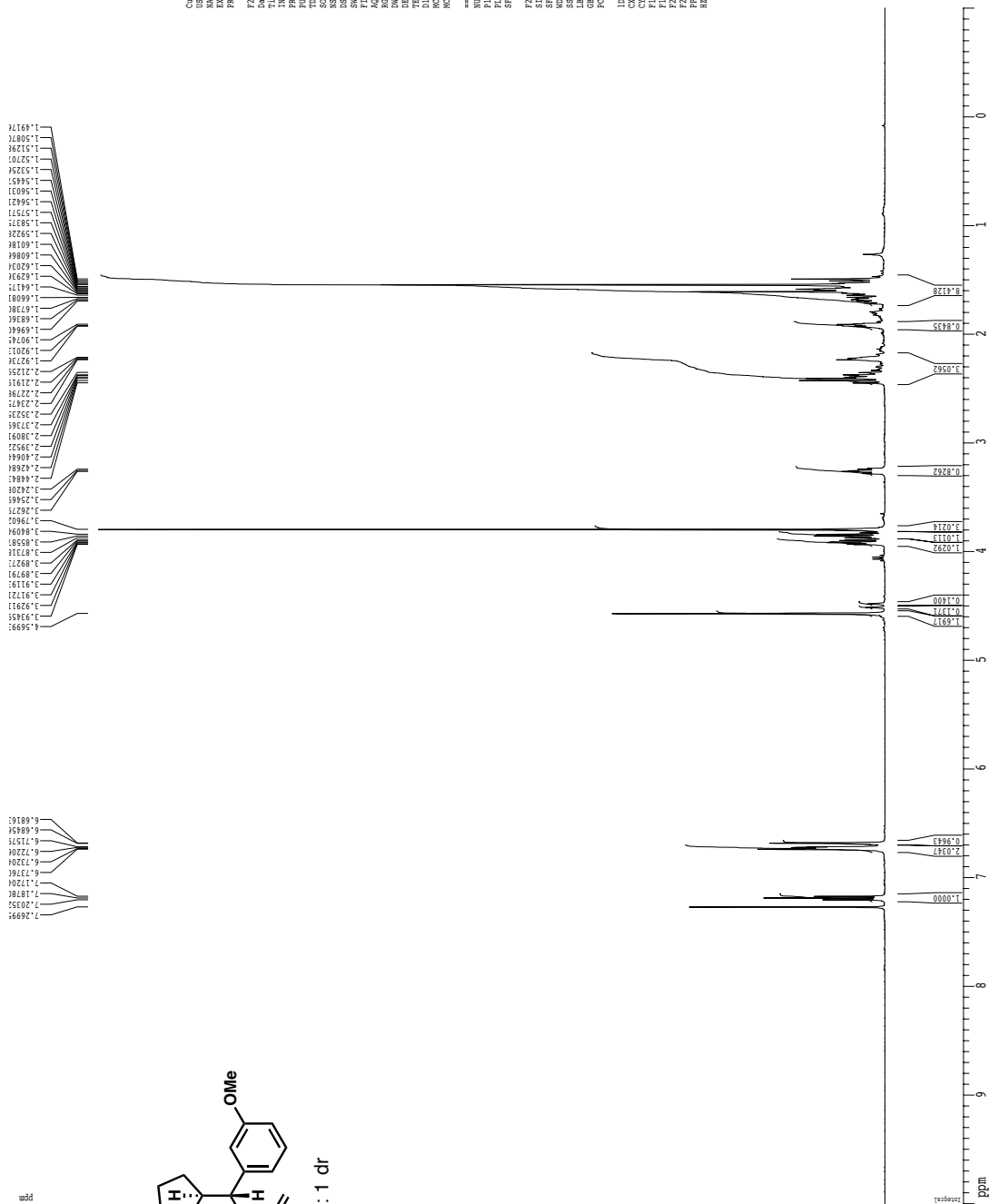
ID TAG parameters  
 CT 10.00 cm  
 F1 230.210 MHz  
 F2 -10.610 ppm  
 F2 -1334.43 Hz  
 FREQ 1329.88132 Hz/cm



7 : 1 dr

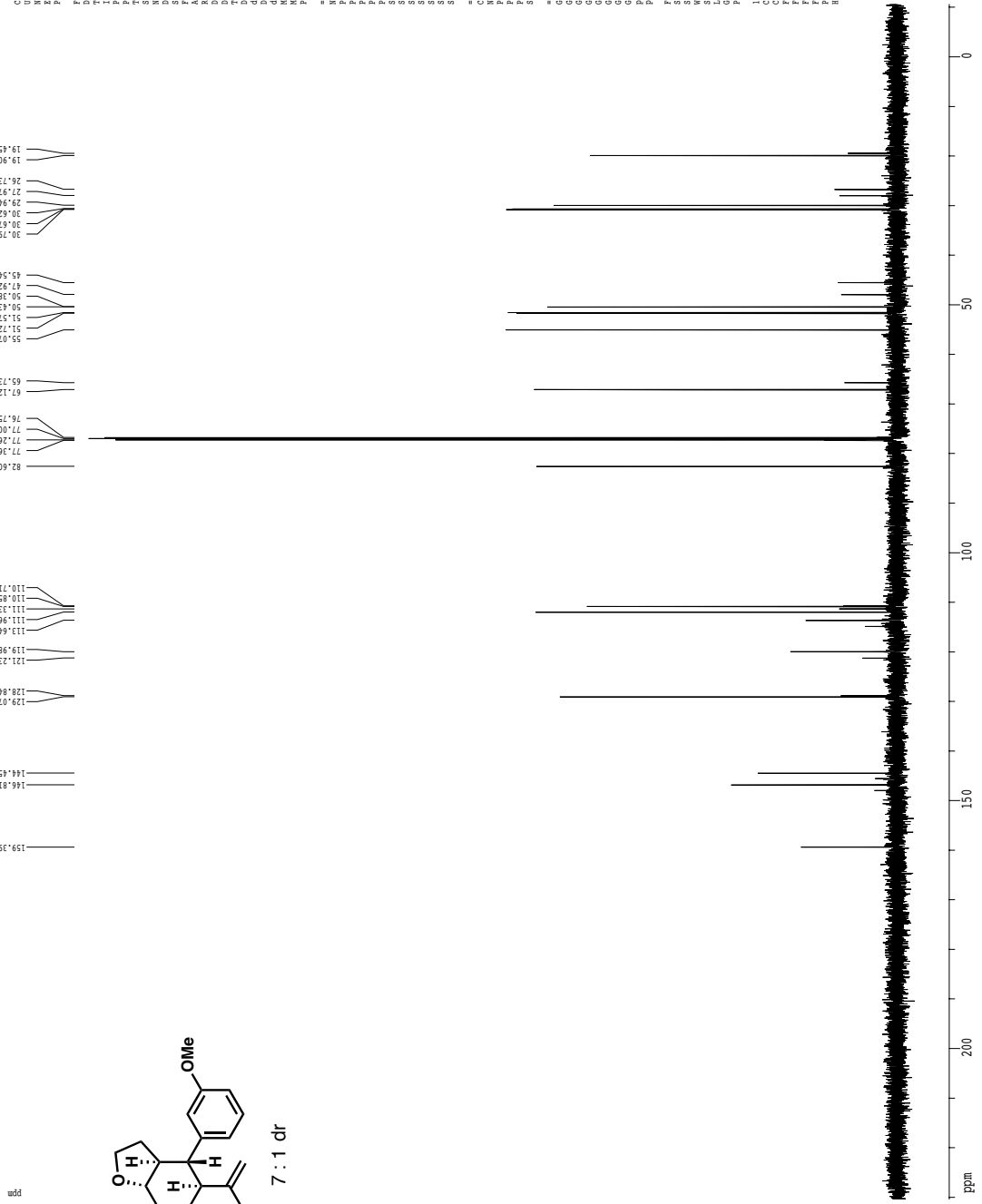


1H spectrum



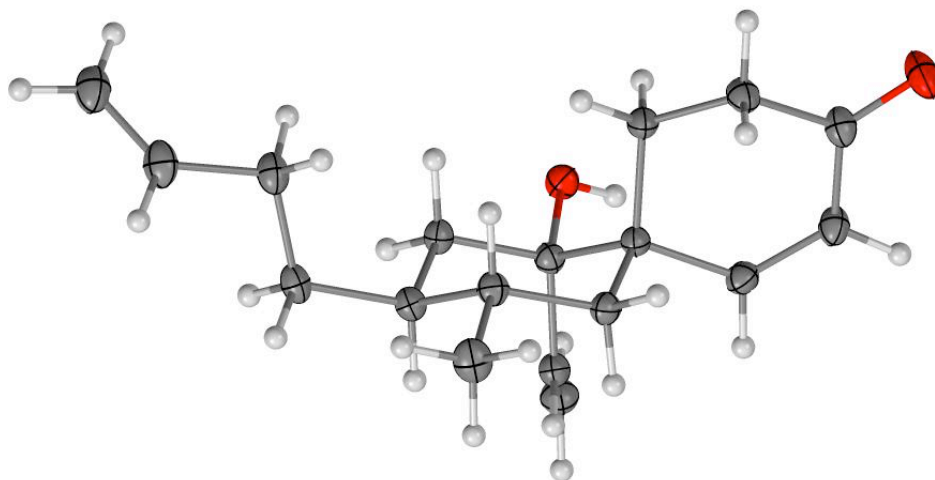
Current Data Parameters  
 NAME Pc35.128\_isolate  
 EXPNO 1  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 200317  
 Time 12.50  
 INSTRUM cty6500  
 PROBRD 5 mm CPXI 1H-  
 PULPROG zgpg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 2  
 DS 2  
 SWH 802.820 Hz  
 FIDRES 0.25026 Hz  
 AQ 1.19961 sec  
 RG 5  
 DQ 62.400 usec  
 DE 6.000 usec  
 TE 298.2 K  
 DI 0.1000000 sec  
 ACQRES 0.0000000 sec  
 SCANS 64  
 ===== CHANNEL f1 =====  
 NUC1 13C  
 P1 7.50 usec  
 PL1 1.60 dB  
 SFO1 500.225015 MHz  
 F2 - Processing parameters  
 SI 6534  
 SF 500.225015 MHz  
 WOP 384  
 SSB 0  
 GB 0  
 PC 4.00  
 LO MR parameters  
 CX 22.80 cm  
 CY 15.00 cm  
 CZ 15.00 cm  
 F1 500.225015 MHz  
 F2 -1.000 ppm  
 F3 -500.22 Hz  
 HETPC 1000000 Hz/cm  
 HRCOR 241.33423 Hz/cm

Z-restored spin-echo 13C spectrum with 1H decoupling



## APPENDIX B: X-ray Crystallographic Data

**Figure B1.** X-Ray Structure for **3.37**.

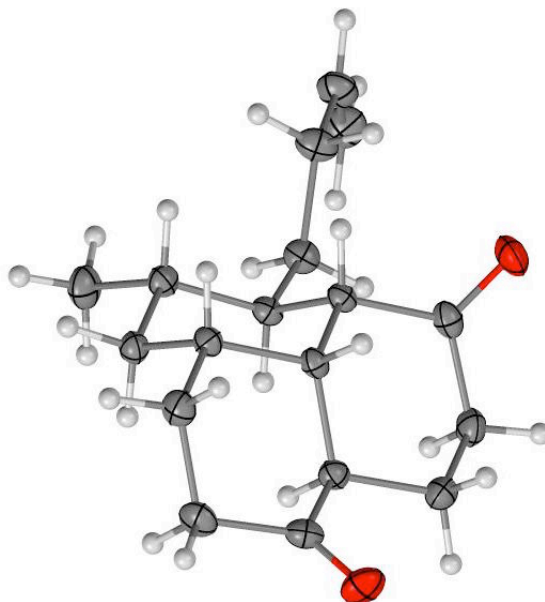


**Table B1.** Crystal Data and Structure Refinement for **3.37**.

Identification code	CCDC #1033631 (cdv24)
Empirical formula	C <sub>18</sub> H <sub>26</sub> O <sub>2</sub>
Formula weight	274.39
Temperature	143(2) K
Wavelength	0.71073 Å
Crystal system	Monoclinic
Space group	<i>P</i> 2 <sub>1</sub> / <i>c</i>
Unit cell dimensions	<i>a</i> = 5.9268(6) Å <i>a</i> = 90°. <i>b</i> = 33.959(4) Å <i>b</i> = 105.1645(12)°. <i>c</i> = 7.8335(8) Å <i>g</i> = 90°.
Volume	1521.7(3) Å <sup>3</sup>
<i>Z</i>	4
Density (calculated)	1.198 Mg/m <sup>3</sup>
Absorption coefficient	0.076 mm <sup>-1</sup>
<i>F</i> (000)	600
Crystal color	colorless
Crystal size	0.284 x 0.265 x 0.152 mm <sup>3</sup>
Theta range for data collection	2.399 to 28.281°
Index ranges	-7 ≤ <i>h</i> ≤ 7, -43 ≤ <i>k</i> ≤ 43, -9 ≤ <i>l</i> ≤ 10
Reflections collected	17594
Independent reflections	3575 [R(int) = 0.0253]
Completeness to theta = 25.500°	99.8 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.8621 and 0.8088
Refinement method	Full-matrix least-squares on <i>F</i> <sup>2</sup>

Data / restraints / parameters	3575 / 0 / 285
Goodness-of-fit on $F^2$	1.040
Final R indices [ $I > 2\sigma(I)$ = 3045 data]	R1 = 0.0390, wR2 = 0.1012
R indices (all data, 0.75 Å)	R1 = 0.0466, wR2 = 0.1064
Extinction coefficient	n/a
Largest diff. peak and hole	0.400 and -0.180 e.Å <sup>-3</sup>

**Figure B2.** X-Ray Structure for **3.58**.

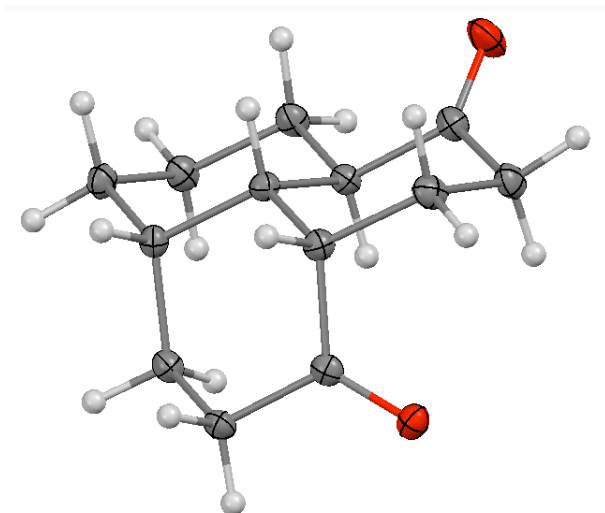


**Table B2.** Crystal Data and Structure Refinement for **3.58**.

Identification code	CCDC #1033632 (cdv27)	
Empirical formula	C <sub>18</sub> H <sub>26</sub> O <sub>2</sub>	
Formula weight	274.39	
Temperature	143(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	<i>P2<sub>1</sub>/c</i>	
Unit cell dimensions	a = 8.5541(7) Å	a = 90°.
	b = 8.9646(8) Å	b = 101.7717(11)°.
	c = 20.1698(17) Å	g = 90°.
Volume	1514.2(2) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.204 Mg/m <sup>3</sup>	
Absorption coefficient	0.076 mm <sup>-1</sup>	
F(000)	600	
Crystal color	colorless	
Crystal size	0.388 x 0.362 x 0.166 mm <sup>3</sup>	

Theta range for data collection	2.063 to 27.094°
Index ranges	-10 ≤ <i>h</i> ≤ 10, -11 ≤ <i>k</i> ≤ 11, -25 ≤ <i>l</i> ≤ 25
Reflections collected	11171
Independent reflections	3309 [R(int) = 0.0255]
Completeness to theta = 25.500°	99.9 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.8621 and 0.8061
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	3309 / 0 / 285
Goodness-of-fit on F <sup>2</sup>	1.023
Final R indices [I > 2σ(I) = 2684 data]	R1 = 0.0401, wR2 = 0.1000
R indices (all data, ? Å)	R1 = 0.0520, wR2 = 0.1072
Largest diff. peak and hole	0.322 and -0.169 e.Å <sup>-3</sup>

**Figure B3.** X-Ray Structure for **3.64**.

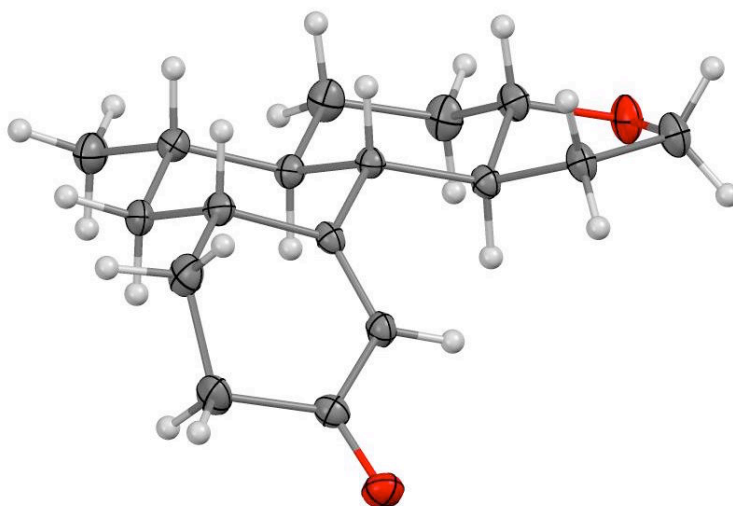


**Table B3.** Crystal Data and Structure Refinement for **3.64**.

Identification code	CCDC #1033633 (cdv25)	
Empirical formula	C <sub>13</sub> H <sub>18</sub> O <sub>2</sub>	
Formula weight	206.27	
Temperature	83(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	<i>P</i> $\bar{1}$	
Unit cell dimensions	a = 7.4167(5) Å	a = 104.2514(8)°.
	b = 8.7830(6) Å	b = 98.7965(8)°.
	c = 9.1296(7) Å	g = 110.1474(7)°.
Volume	522.34(6) Å <sup>3</sup>	
Z	2	

Density (calculated)	1.311 Mg/m <sup>3</sup>
Absorption coefficient	0.086 mm <sup>-1</sup>
F(000)	224
Crystal color	colorless
Crystal size	0.405 x 0.324 x 0.312 mm <sup>3</sup>
Theta range for data collection	2.385 to 28.865°
Index ranges	-9 ≤ h ≤ 10, -11 ≤ k ≤ 11, -12 ≤ l ≤ 11
Reflections collected	6268
Independent reflections	2486 [R(int) = 0.0111]
Completeness to theta = 25.500°	99.1 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.8621 and 0.8158
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	2486 / 0 / 208
Goodness-of-fit on F <sup>2</sup>	1.084
Final R indices [I > 2σ(I) = 2311 data]	R1 = 0.0391, wR2 = 0.1147
R indices (all data, 0.74 Å)	R1 = 0.0411, wR2 = 0.1168
Extinction coefficient	n/a
Largest diff. peak and hole	0.390 and -0.206 e.Å <sup>-3</sup>

**Figure B4.** X-Ray Structure for **5.70**.

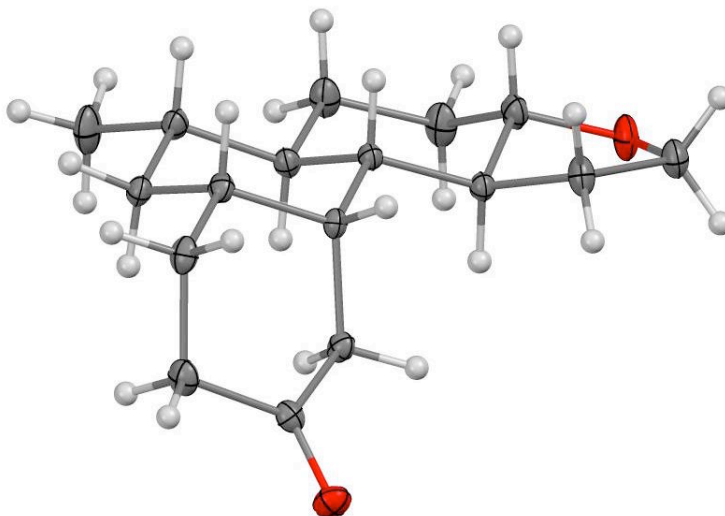


**Table B4.** Crystal Data and Structure Refinement for **5.70**.

Identification code	CCDC #1474440 (cdv39)
Empirical formula	C <sub>17</sub> H <sub>24</sub> O <sub>2</sub>
Formula weight	260.36
Temperature	133(2) K
Wavelength	0.71073 Å
Crystal system	Triclinic

Space group	$P\bar{1}$	
Unit cell dimensions	a = 7.0946(5) Å b = 10.4299(7) Å c = 10.7084(7) Å	a = 91.4370(8)° b = 109.1829(8)° g = 107.5789(8)°
Volume	706.60(8) Å <sup>3</sup>	
Z	2	
Density (calculated)	1.224 Mg/m <sup>3</sup>	
Absorption coefficient	0.078 mm <sup>-1</sup>	
F(000)	284	
Crystal color	colorless	
Crystal size	0.464 x 0.277 x 0.255 mm <sup>3</sup>	
Theta range for data collection	2.033 to 28.669°	
Index ranges	-9 ≤ h ≤ 9, -13 ≤ k ≤ 13, -14 ≤ l ≤ 13	
Reflections collected	8518	
Independent reflections	3351 [R(int) = 0.0142]	
Completeness to theta = 26.000°	99.7 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.8621 and 0.8062	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	3351 / 0 / 172	
Goodness-of-fit on F <sup>2</sup>	1.035	
Final R indices [I > 2σ(I) = 2954 data]	R1 = 0.0397, wR2 = 0.1086	
R indices (all data, 0.74 Å)	R1 = 0.0444, wR2 = 0.1131	
Largest diff. peak and hole	0.351 and -0.224 e.Å <sup>-3</sup>	

**Figure B5.** X-Ray Structure for **5.74**.



**Table B5.** Crystal Data and Structure Refinement for **5.74**.

Identification code	CCDC #1474441 (cdv40)
Empirical formula	C <sub>17</sub> H <sub>26</sub> O <sub>2</sub>
Formula weight	262.38
Temperature	88(2) K
Wavelength	0.71073 Å
Crystal system	Monoclinic
Space group	<i>P</i> 2 <sub>1</sub>
Unit cell dimensions	a = 6.9229(3) Å      a = 90°. b = 7.1568(4) Å      b = 95.9776(7)°. c = 14.6894(7) Å      g = 90°.
Volume	723.84(6) Å <sup>3</sup>
Z	2
Density (calculated)	1.204 Mg/m <sup>3</sup>
Absorption coefficient	0.076 mm <sup>-1</sup>
F(000)	288
Crystal color	colorless
Crystal size	0.433 x 0.297 x 0.156 mm <sup>3</sup>
Theta range for data collection	2.788 to 29.179°
Index ranges	-9 ≤ h ≤ 9, -9 ≤ k ≤ 9, -19 ≤ l ≤ 20
Reflections collected	9092
Independent reflections	3603 [R(int) = 0.0154]
Completeness to theta = 26.000°	99.9 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.7458 and 0.6904
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	3603 / 1 / 173
Goodness-of-fit on F <sup>2</sup>	1.033
Final R indices [I > 2σ(I) = 3427 data]	R1 = 0.0324, wR2 = 0.0864
R indices (all data, 0.73 Å)	R1 = 0.0344, wR2 = 0.0882
Absolute structure parameter	0.0(3)
Largest diff. peak and hole	0.319 and -0.171 e.Å <sup>-3</sup>