

UC Irvine

UC Irvine Previously Published Works

Title

H₂ in Antarctic firm air: Atmospheric reconstructions and implications for anthropogenic emissions

Permalink

<https://escholarship.org/uc/item/1rc9r56f>

Journal

Proceedings of the National Academy of Sciences of the United States of America, 118(36)

ISSN

0027-8424

Authors

Patterson, John D
Aydin, Murat
Crotwell, Andrew M
et al.

Publication Date

2021-09-07

DOI

10.1073/pnas.2103335118

Peer reviewed

H₂ in Antarctic firn air: Atmospheric reconstructions and implications for anthropogenic emissions

John D. Patterson^{a,1} , Murat Aydin^a , Andrew M. Crotwell^{b,c}, Gabrielle Pétron^{b,c}, Jeffrey P. Severinghaus^d , Paul B. Krummel^e , Ray L. Langenfelds^e , and Eric S. Saltzman^{a,f} 

^aDepartment of Earth System Science, University of California, Irvine, CA 92697; ^bGlobal Monitoring Laboratory, National Oceanic and Atmospheric Administration, Boulder, CO 80305; ^cCooperative Institute for Research in Environmental Sciences, University of Colorado, Boulder, CO 80309; ^dScripps Institution of Oceanography, University of California San Diego, La Jolla, CA 92037; ^eOceans and Atmosphere, Climate Science Centre, Commonwealth Scientific and Industrial Research Organisation, Aspendale, VIC 3195, Australia; and ^fDepartment of Chemistry, University of California, Irvine, CA 92697

Edited by Akkihebbal R. Ravishankara, Colorado State University, Fort Collins, CO, and approved July 13, 2021 (received for review February 18, 2021)

The atmospheric history of molecular hydrogen (H₂) from 1852 to 2003 was reconstructed from measurements of firn air collected at Megadunes, Antarctica. The reconstruction shows that H₂ levels in the southern hemisphere were roughly constant near 330 parts per billion (ppb; nmol H₂ mol⁻¹ air) during the mid to late 1800s. Over the twentieth century, H₂ levels rose by about 70% to 550 ppb. The reconstruction shows good agreement with the H₂ atmospheric history based on firn air measurements from the South Pole. The broad trends in atmospheric H₂ over the twentieth century can be explained by increased methane oxidation and anthropogenic emissions. The H₂ rise shows no evidence of deceleration during the last quarter of the twentieth century despite an expected reduction in automotive emissions following more stringent regulations. During the late twentieth century, atmospheric CO levels decreased due to a reduction in automotive emissions. It is surprising that atmospheric H₂ did not respond similarly as automotive exhaust is thought to be the dominant source of anthropogenic H₂. The monotonic late twentieth century rise in H₂ levels is consistent with late twentieth-century flask air measurements from high southern latitudes. An additional unknown source of H₂ is needed to explain twentieth-century trends in atmospheric H₂ and to resolve the discrepancy between bottom-up and top-down estimates of the anthropogenic source term. The firn air-based atmospheric history of H₂ provides a baseline from which to assess human impact on the H₂ cycle over the last 150 y and validate models that will be used to project future trends in atmospheric composition as H₂ becomes a more common energy source.

atmospheric hydrogen | H₂ emissions | firn air | Antarctica

Molecular hydrogen (H₂) is an abundant and reactive constituent of Earth's atmosphere. The utilization of H₂ as an energy source emits no carbon to the atmosphere if produced from water using renewables, and increasing adoption of H₂ as a substitute for fossil fuels is likely (1). As the H₂ energy sector expands, anthropogenic emissions are expected to increase due to leakage. Projecting the effects of increasing anthropogenic emissions requires a comprehensive understanding of the biogeochemical cycle of H₂. Reconstructing the paleoatmospheric levels of H₂ contributes to that understanding by establishing the baseline for quantifying anthropogenic emissions since the industrial revolution.

Presently, the average atmospheric abundance of H₂ is 530 parts per billion (ppb; nmol H₂ mol⁻¹ air), and the atmospheric lifetime is about 2 y (2). The budget of H₂ is complex, including both natural and anthropogenic terms. Globally, the largest source of H₂ is the photolysis of formaldehyde, formed by the atmospheric oxidation of methane and nonmethane hydrocarbons (NMHCs). Other major sources include direct emissions from fossil fuel combustion and biomass burning. N₂ fixation both on land and in the ocean is a small additional source. The major sink for atmospheric H₂ is uptake by soil microbes, with oxidation by OH accounting for the remaining losses. H₂ levels are higher by

about 3% in the southern hemisphere than in the northern hemisphere due to the hemispheric asymmetry of the soil sink (2–5).

Atmospheric H₂ levels impact Earth's radiative budget and air quality. H₂ serves as a sink for the OH radical, increasing the atmospheric lifetime of radiatively important trace gases like methane. Additionally, the reaction of H₂ with OH results in the catalytic production of O₃ in the troposphere (2, 6–8). Oxidation of H₂ by OH in the stratosphere leads to increased concentrations of HO₂ and water vapor. The increase in these species will have indirect radiative effects due to losses of ozone and alterations to the distribution of polar stratospheric clouds. Increased stratospheric concentrations of water vapor will also have direct radiative effects via increased infrared absorption (7, 9–12).

The modern instrumental record of tropospheric H₂ abundance began in the late 1980s (2, 13). Mid-twentieth century studies reported a wide range (400 to 2,000 ppb), which likely reflects analytical issues and/or influence from urban pollutants (14–16). The surface flask air measurements of Khalil and Rasmussen (13), the National Oceanic and Atmospheric Administration Global Monitoring Laboratory (NOAA/GML; refs. 2, 17), and the Commonwealth Scientific and Industrial Research Organisation (CSIRO; ref. 18) show atmospheric levels of H₂ of 510 to 550 ppb during the late 20th and early 21st centuries at background sites around the world. In situ measurements from Cape Grim, Tasmania, and Mace Head, Ireland, made by the Advanced Global

Significance

Atmospheric molecular hydrogen (H₂) levels influence the lifetime of greenhouse gases and the levels of tropospheric ozone. The extent to which human activities have influenced the global H₂ budget is not well established due to our limited knowledge of past variations in atmospheric H₂ levels. Here, we reconstruct atmospheric H₂ over the last 150 y using Antarctic firn air. We find a roughly 70% rise in atmospheric H₂ over the twentieth century that can be attributed to human activities. Surprisingly, there is no evidence that anthropogenic emissions of H₂ decreased during the late twentieth century due to reduced automotive emissions associated with air pollution controls. It is likely that nonautomotive anthropogenic H₂ emissions have been underestimated.

Author contributions: J.D.P. carried out the model simulations and data analysis; M.A. contributed to data analysis; A.M.C., G.P., and P.B.K. made calibration corrections and contributed flask measurements; J.P.S. collected the field samples and assisted with firn air modeling; P.B.K. and R.L.L. contributed to calibration corrections and flask measurements; E.S.S. contributed to modeling and data analysis; and J.D.P. wrote the manuscript. The authors declare no competing interest.

This article is a PNAS Direct Submission.

Published under the PNAS license.

¹To whom correspondence may be addressed. Email: jdpatter@uci.edu.

This article contains supporting information online at <https://www.pnas.org/lookup/suppl/doi:10.1073/pnas.2103335118/-DCSupplemental>.

Published August 23, 2021.

Atmospheric Gases Experiment (AGAGE) show similar levels of atmospheric H_2 during the same time period (19). To date, there have been two published firn air studies of the historical trends of atmospheric H_2 . Petrenko et al. (20) reconstructed northern hemisphere H_2 from Greenland firn air measurements. Their results show an increase in atmospheric H_2 levels from 450 to 520 ppb during the 1960s to a peak near 550 ppb during the late 1980s or early 1990s, then a recent decline to about 515 ppb in 2010. The peak and recent decline are inconsistent with modern flask measurements that show roughly constant H_2 levels during the 1990s (2, 17, 19). The authors note that the firn air model used for the reconstruction does not include pore close-off fractionation of H_2 , and it is possible that the inferred peak is a modeling artifact (20, 21). Firn air measurements of H_2 from the South Pole showed that atmospheric H_2 increased from 350 ppb to 550 ppb over the twentieth century in the high southern latitudes (22). This reconstruction does not include a late twentieth century peak or decline, and the reconstructed levels of H_2 are consistent with available modern flask measurements beginning in the late 1980s. The increase in atmospheric H_2 is primarily attributed to increasing anthropogenic emissions and increasing atmospheric concentrations of methane over the twentieth century.

Here we report H_2 measurements in firn air samples collected from the Megadunes site (80.78 °S, 124.49 °E, Alt: 2,283 m) in central Antarctica during January of 2004 and analyzed at NOAA/GML in April 2004 (23, 24). Megadunes is a complex site, with very low annual accumulation rates and an unusually large amplitude of surface topography. The oldest firn air ever recovered, with a mean CO_2 age of 140 y, was sampled during this campaign (23). A firn air model is used to reconstruct southern-hemisphere atmospheric H_2 levels since the 1850s. The reconstruction is compared to the existing South Pole firn air reconstruction (22), and implications of the reconstructed atmospheric history for the global budget of H_2 are discussed.

Firn Air Model Inversion and Recovering the Atmospheric History

The Megadunes firn air H_2 data were corrected for detector nonlinearity and calibration drift and adjusted from the NOAA96 to the MPI09 calibration scale (SI Appendix; ref. 25). Then the firn air data were corrected for gravitational fractionation and

depth-averaged (SI Appendix). To recover an atmospheric history of H_2 , the measurements are inverted using depth-dependent age distributions or Green's functions ($G(z,t)$; ref. 26). The Megadunes firn air model is initialized with no H_2 in the firn air column, then forced with a 1-y pulse of H_2 at the surface, after which the surface concentration is set to 0. The model is integrated for 300 y, and the evolution of the pulse is tracked as a function of depth and time to produce the Green's functions, which allow us to determine the modeled levels of H_2 in the firn ($H_2(z)_{model}$) from a given atmospheric history ($H_{2(atm)}(t)$) according to Eq. 1:

$$H_2(z)_{model} = \sum_{t=0}^{300} H_{2(atm)}(t) * G(z,t), \quad [1]$$

This technique is computationally efficient and allows for rapid iteration over many possible atmospheric histories in order to optimize the fit between the modeled and measured depth profiles. This inversion problem is under-constrained, and there are an infinite number of atmospheric histories that give good fits to the measured depth profile. To prevent unrealistic extreme fluctuations in the reconstructions, we employ a cost function which includes a smoothing term (26):

$$f = \chi_{red}^2 + S = \sum_{i=1}^n \frac{\left(\frac{H_2(z)_{model} - H_2(z)_{obs}}{\sigma} \right)^2}{n-1} + \alpha \sum_{t=0}^{300} \left(\frac{\partial^2 H_{2(atm)}(t)}{\partial t^2} \right)^2, \quad [2]$$

where f is the cost, χ_{red}^2 is the reduced χ^2 statistic, S is the smoothing term, σ is the analytical uncertainty (Fig. 1B), n is the number of unique depths, and α is a prescribed, dimensionless weighting parameter for the smoothing term. The MATLAB solver "fmincon" is used to minimize f .

The initial guess for the atmospheric H_2 history is a linear ramp from some "preindustrial" level of H_2 in 1704 to 540 ppb in 2003. The 1704 level is fixed at its initial value, and the H_2 levels for 1705 to 2003 are allowed to vary to minimize f . In order to assess the sensitivity of the inversion to the initial guess, the H_2 level in 1704 is initialized at four different values: 270 ppb, 310 ppb, 350 ppb, and 390 ppb based on the early twentieth century levels from Patterson et al. (22). The inversion is run for each

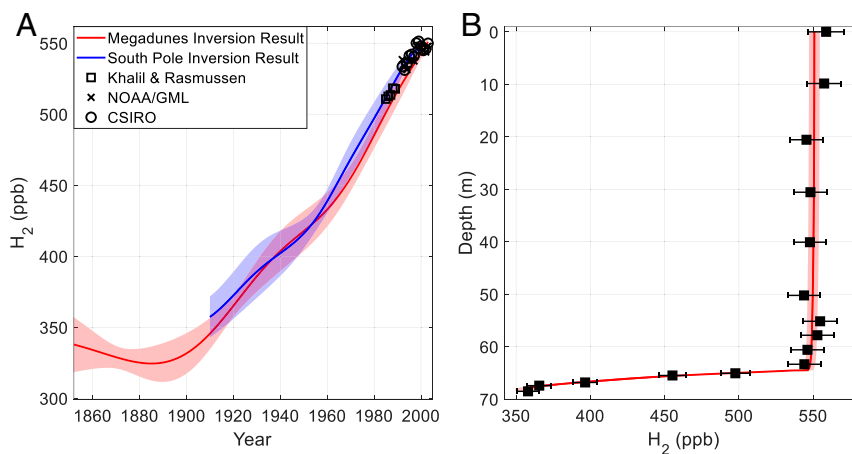


Fig. 1. Reconstructions of atmospheric H_2 and associated Megadunes firn air depth profiles. (A) Red line and shading—mean inferred atmospheric history and $\pm 1\sigma$ uncertainty from 4,000 Monte Carlo inversions on the Megadunes firn air measurements (see *Firn Air Model Inversion and Recovering the Atmospheric History*) using $\alpha = 8 \times 10^{-2}$; blue line and shading— inferred atmospheric history and $\pm 1\sigma$ uncertainty from a South Pole firn air reconstruction (22); and black squares, black x's, and black circles—observed atmospheric H_2 annual means from high southern latitude sites from 1985 to 1989 and 1992 to 2003 (2, 13, 17, 18); (B) Black squares and error bars—depth-averaged Megadunes firn air H_2 measurements corrected for detector nonlinearity, calibration drift, and gravitational fractionation as described in SI Appendix. Error bars are the propagated $\pm 2\%$ measurement uncertainty; red line and shading show the mean modeled H_2 depth profile and $\pm 1\sigma$ uncertainty resulting from the inferred atmospheric history plotted in red in A.

initial guess and the results are compared. To assess sensitivity to smoothing, we vary α across an order of magnitude (8×10^{-3} and 8×10^{-2}). To determine the impact of analytical uncertainty, we use a Monte Carlo method in which we repeatedly sample the normal distributions defined by the depth-averaged measurements and the analytical uncertainty at each depth (Fig. 1B). We then invert the synthetic depth profiles that are generated by the repeated sampling. We performed 1,000 Monte Carlo samplings at each combination of initial condition and α for a total of 8,000 inversion runs, accounting for uncertainties due to initial conditions, smoothing, and analytical uncertainties. The Monte Carlo runs do not assess bias in the modeling parameterizations or Green's functions. Model uncertainties and bias are addressed in a later section.

Inversion Results and the Atmospheric History

The Monte Carlo inversion results show that the reconstruction after 1852 is insensitive to the assumed "preindustrial" level of H_2 (SI Appendix, Fig. S4). Here, only results after 1852 for $\alpha = 8 \times 10^{-2}$ are discussed. The choice of α does not affect the main conclusions of this work. Results using $\alpha = 8 \times 10^{-3}$ are given in SI Appendix (SI Appendix, Fig. S5). The inversion result is the average of the 4,000 Monte Carlo runs (Fig. 1A). H_2 levels were 330 ± 15 ppb from 1852 to 1890. In 1890, atmospheric H_2 begins to increase at a rate of 1.5 ppb y^{-1} until 1960. After 1960, the rate of atmospheric increase rises to 2.5 ppb y^{-1} . Atmospheric H_2 levels reach 550 ppb in 2002. The inversion result is broadly consistent with the previous South Pole firm air reconstruction of H_2 (Fig. 1A; ref. 22). Notably, the firm air history includes a late twentieth century rise in southern-hemisphere H_2 levels.

Modern flask air measurements are in good agreement with the inversion result and support the late twentieth century rise in H_2 (Fig. 1A). These include flask air measurements from 1985 to 1989 at Palmer Station, Antarctica, and Cape Grim Observatory, Tasmania, made by Khalil and Rasmussen (13), from 1992 to 2003 at Palmer, Syowa, South Pole, and Halley Stations, Antarctica, and Cape Grim Observatory, Tasmania, made by NOAA/GML (2, 17), and from Casey, South Pole, and Mawson Stations, Antarctica, and Cape Grim and Macquarie Island, Tasmania, made by CSIRO (18). The NOAA/GML flask measurements were adjusted from the NOAA96 to the MPI09 calibration scale (SI Appendix). One feature in the flask air measurements that is not evident in the firm air history is the abrupt increase in atmospheric H_2 from 1997 to 1998 likely due to El Niño–Southern Oscillation-related biomass burning. This H_2 pulse is not evident in the firm air history because of diffusive smoothing in the firm air column (see Fig. 3, Insets; refs. 18, 19, 27). The surface flask air measurements were not used to constrain the inversion.

Bias and Uncertainty in the Inversion Technique

The Monte Carlo method described above assesses uncertainty in the inversion resulting from measurement uncertainty, but it does not assess bias or uncertainty in the model parameterizations (e.g., molecular diffusion) or the Green's functions used in the inversion. To assess bias or uncertainty in the inversion technique, we use the same methods to invert the measured depth profiles of CH_4 and CO_2 at Megadunes. The inversion results are then compared to the established atmospheric histories of CH_4 and CO_2 . Atmospheric CH_4 and CO_2 levels over the last several centuries are well constrained by high-resolution ice core measurements from Law Dome and modern flask measurements from NOAA/GML (28–31). When the firm air model is forced with these atmospheric histories, modeled depth profiles show a small bias in the lock-in zone relative to measured depth profiles (SI Appendix, Fig. S3).

For CH_4 , the bias between the inversion result and the Law Dome/modern flask record increases from near 0 early in the record to about 15 y in the mid-twentieth century, then decreases

to 0 around 1980 (Fig. 2A). The R^2 between the firm air history and the established history is 0.954 with an average error of 4.8% (52 ppb). The CO_2 firm air history shows less bias than the CH_4 history with an R^2 of 0.978 and average error of 1.0% (2.9 ppm; Fig. 2C). The smaller bias for CO_2 is expected because the model diffusivities were tuned using CO_2 (Methods).

We assume that the differences between the reconstructions and the Law Dome/modern flask atmospheric histories reflects uncertainty in the physical parameterizations or tuning of the diffusivities in the firm air model. We can use the results from the CH_4 and CO_2 reconstructions to further refine the inversion. This was done by calculating the age correction required to optimize the fit between the firm air CH_4 and CO_2 reconstructions to the Law Dome/modern flask atmospheric histories. The correction is formulated as a time-dependent age offset ($B(t)$) for the CH_4 and CO_2 mixing ratios at each year of the reconstruction. We use the MATLAB solver "fmincon" to minimize a new cost function (j) and optimize $B(t)$:

$$j = R^2 + S$$

$$= \sum_{t=0}^{300} [X_{(CH_4_inv)}(t + B(t)) - X_{(CH_4_hist)}(t)]^2$$

$$+ \sum_{t=0}^{300} [X_{(CO_2_inv)}(t + B(t)) - X_{(CO_2_hist)}(t)]^2 \quad [3]$$

$$+ \beta \sum_{t=0}^{300} \left(\frac{\partial^2 B(t)}{\partial t^2} \right)^2,$$

where R^2 is the sum of the squared residuals for both CO_2 and CH_4 , S is a smoothing term, $X_{(CH_4_inv)}(t)$ and $X_{(CO_2_inv)}(t)$ are the reconstructed firm air history of CH_4 and CO_2 respectively, $X_{(CH_4_hist)}(t)$ and $X_{(CO_2_hist)}(t)$ are the well-constrained atmospheric histories of CH_4 and CO_2 , respectively, and β is an arbitrary weighing parameter for the smoothing term. As expected, applying this correction to the CH_4 and CO_2 reconstructions yields better agreement with the established atmospheric histories. For CH_4 , R^2 increases from 0.954 to 0.998, and the average error decreases from 4.8 to 1.2% (52 to 11 ppb). For CO_2 , R^2 increases from 0.977 to 0.982 and the average error decreases from 1.0 to 0.8% (2.9 to 2.6 ppm).

This correction has little effect on the Megadunes H_2 reconstruction before 1900 and after 1980 (Fig. 3). Between 1900 and 1980, the corrected result is on average 4.6% (18.9 ppb) lower than the uncorrected inversion result. We conducted a similar exercise for the South Pole firm air H_2 reconstruction from Patterson et al. (22). CH_4 and CO_2 depth profiles from that firm air campaign were inverted, and Eq. 3 was used to calculate an empirical correction. As with Megadunes, this correction has a relatively minor effect on the H_2 reconstruction, and there is good agreement between the corrected reconstructions from the two sites (Fig. 3B).

Modeling the H_2 Budget

A steady-state box model of the troposphere was used to develop a simulation of atmospheric H_2 based on estimated changes in various H_2 emissions over the past century. The model is comprised of six 30° -wide zonal boxes. Transport between boxes is parameterized from the modern atmospheric distribution of SF_6 (32). First-order losses of H_2 to soils, OH radicals, oceans, and the stratosphere are included. H_2 sources consist of the following: anthropogenic H_2 emissions (including direct emissions from fossil fuel burning and oxidation of anthropogenic NMHCs), biomass-burning emissions, oxidation of CH_4 , oxidation of biogenic NMHCs, and terrestrial and marine N_2 fixation. Annual H_2 emissions are prescribed for 1852 through 2003, and the steady-state H_2

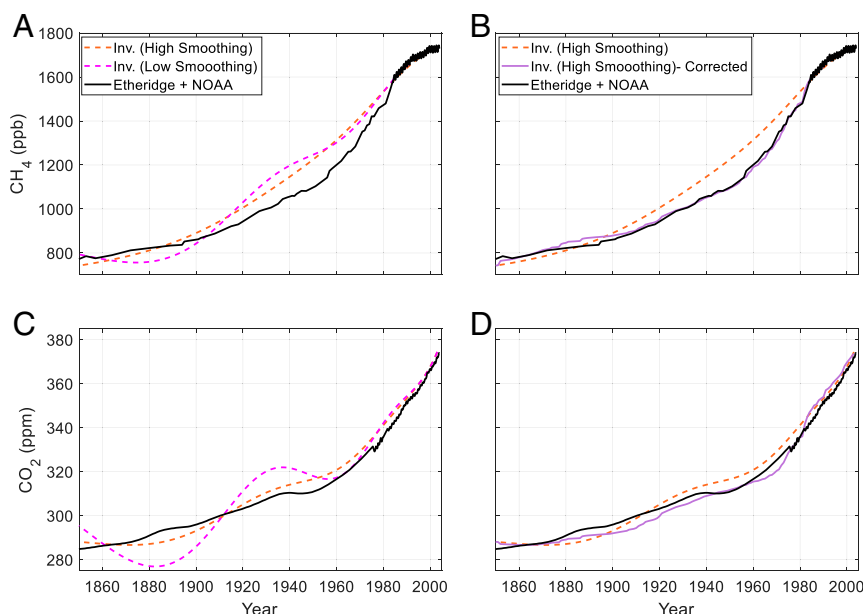


Fig. 2. Inversion results and empirically corrected inversion results for CO_2 and CH_4 compared to the Law Dome/modern flask atmospheric histories. (A) Dashed orange line: inversion result for CH_4 using high smoothing; dashed magenta line: inversion result for CH_4 using 10% of the smoothing used for the “high smoothing” inversion, included for comparison; and solid black line: well-established atmospheric history of CH_4 from Etheridge et al. (29) and Dlugokencky (30). (B) Dashed orange line and solid black line: as in A; solid purple line: inversion result with high smoothing with the empirical correction applied (see *Bias and Uncertainty in the Inversion Technique*). (C and D) As in A and B, respectively, for CO_2 ; the established atmospheric history is from Etheridge et al. (28) and Dlugokencky (31).

abundance is calculated for each year. The geographic distribution of sources and sinks is assumed to be constant in time, and first-order loss rates for all sinks are fixed.

H_2 emissions change over time due to changes in atmospheric CH_4 levels, biomass burning, and anthropogenic H_2 (Fig. 4A and B). CH_4 oxidation was scaled with the atmospheric history of CH_4 based on the ice core and instrumental records (29, 30). Biomass burning and anthropogenic emissions were scaled to

estimates of historical CO emissions. Sources of atmospheric H_2 and CO are thought to be tightly coupled because incomplete combustion and photochemical oxidation of formaldehyde are the most important sources of both gases (SI Appendix, Table S2; refs. 2–4, 20). We followed the approach of Verhulst (33), who used the biomass-burning CO emissions inventory of Lamarque et al. (34) and an anthropogenic CO inventory based on a top-down analysis of firm air-derived CO atmospheric histories. This ap-

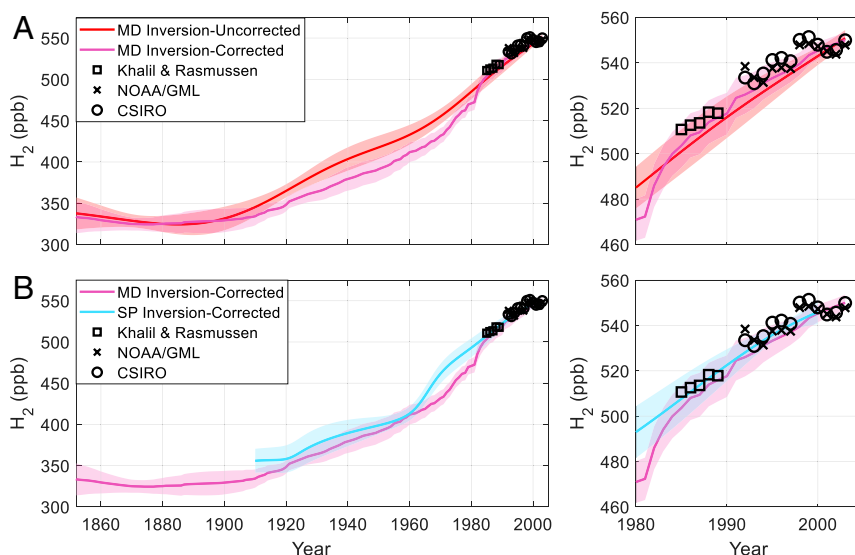


Fig. 3. Atmospheric reconstructions of H_2 with empirical corrections applied. (A) Red line and shading: Megadunes inversion result and $\pm 1\sigma$ uncertainty as in Fig. 1A; purple line and shading: Megadunes Inversion result and $\pm 1\sigma$ uncertainty using the empirical correction; black squares, black x’s, and black circles: atmospheric H_2 annual means from high southern latitude sites from 1985 to 1989 and 1992 to 2003 as in Fig. 1A. (B) Purple line and shading, black squares, black x’s, and black circles: as in A; blue line and shading: corrected South Pole inversion and $\pm 1\sigma$ uncertainty; and right panels: expanded view of the late twentieth century.

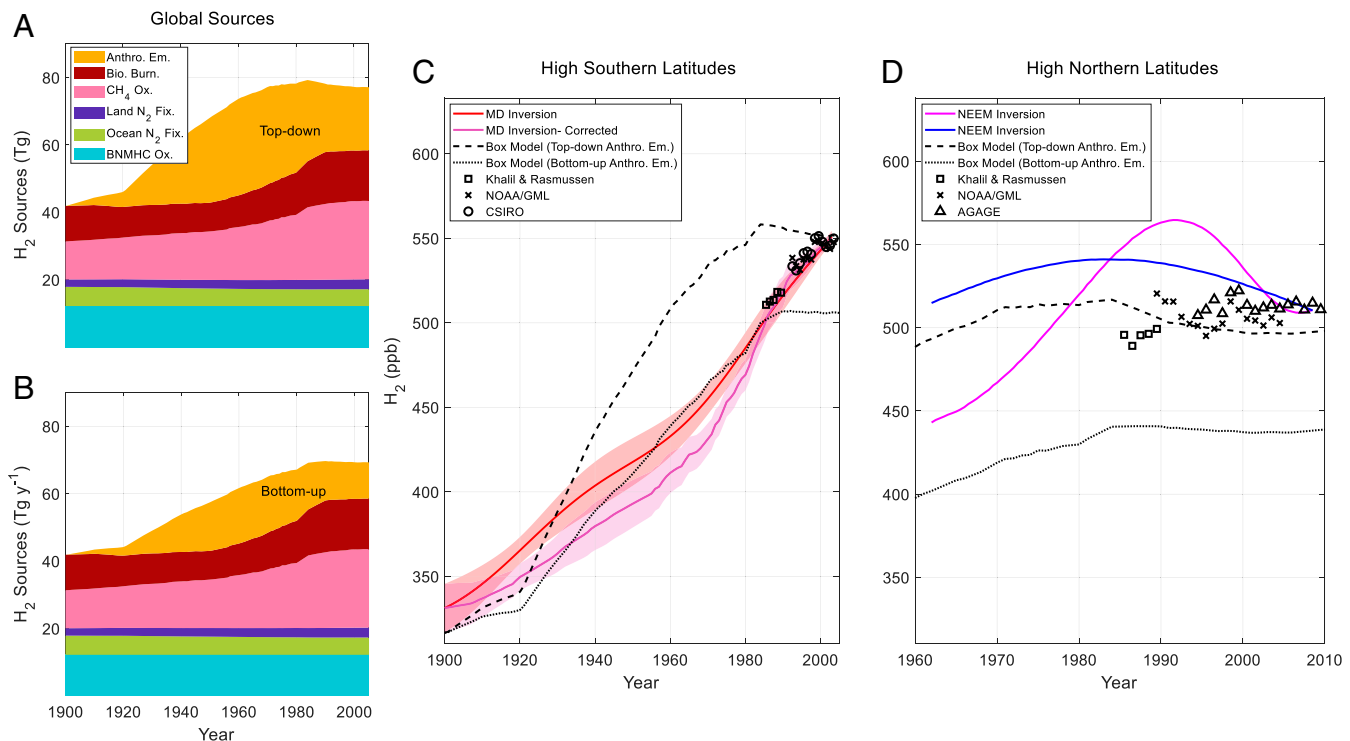


Fig. 4. H_2 global sources and atmospheric history for the twentieth century from firn air and the box model. (A and B) Global sources estimated from Pieterse et al. and Verhulst (*SI Appendix*; refs. 5, 33). In A, anthropogenic emissions are estimated using the top-down inventory of Pieterse et al. (5). In B, anthropogenic emissions are estimated from the bottom-up inventory of Vollmer et al. (50). (C) High southern latitude atmospheric H_2 reconstructed from Antarctic firn air (red, purple lines, and shading: as in Fig. 3A) and 60 to 90° S box model simulations using top-down (dashed black line) and bottom-up (dotted black line) anthropogenic emissions. Observed atmospheric H_2 annual means from high southern latitude sites from 1985 to 1989 and 1992 to 2003 are shown (2, 13, 17, 18). (D) High northern latitude northern hemisphere atmospheric histories of H_2 from Petrenko et al. based on firn air measurements at the Northeast Greenland Eemian drill site (NEEM) using the Institute of Arctic and Alpine Research firn air model (blue line) and using the firn air model of Witrant et al. (magenta line: ref. 62) adjusted from the NOAA96 to the MPI09 calibration scale (20). Observed atmospheric H_2 annual means for high northern latitude sites from 1985 to 2008 are shown (2, 13, 17, 19). Box model simulations are shown for 60 to 90° N using top-down (dashed black line) and bottom-up (dotted black line) anthropogenic emissions.

proach yields substantially larger anthropogenic CO emissions than bottom-up inventories with an earlier peak (~1970) and a larger late twentieth century decline (33–35). Bottom-up inventories of anthropogenic CO have previously been shown to underestimate atmospheric CO levels (34–39). The CO inventories used here agree well with the instrumental record and firn reconstructions from Greenland and Antarctica (*SI Appendix*; refs. 20, 33, 40, 41) and are also consistent with firn air reconstructions of other NMHCs (e.g., refs. 42, 43). We assumed constant H_2/CO molar emissions ratios of 0.46 for biomass burning and 0.23 for anthropogenic emissions. These ratios were chosen to achieve agreement between the box model and the 2003 atmosphere. A small time-dependent change in H_2 emissions from terrestrial N_2 fixation was also assumed based on Fowler et al. (44). A full description of the treatment of each source and sink is given in *SI Appendix*.

Modeled high southern latitude H_2 levels rise slowly from 306 to 341 ppb from 1852 to 1920, then more rapidly to 558 ppb in 1985. From 1985 to 2000, H_2 levels decrease slowly to 550 ppb (Fig. 4C). The general increasing trend in the box model simulation is consistent with the firn air-derived history. The increase is attributable to growing anthropogenic emissions and rising production from CH_4 oxidation. However, the pace of the rise and rollover in modeled H_2 levels after 1980 does not agree with the steady, gradual increase in the firn air history. Forcing the firn air model with the modeled atmospheric history gives a modeled depth profile that is inconsistent with the measured depth profile (*SI Appendix*, Fig. S6). This demonstrates that the

slower, steady rise in H_2 levels from the firn air history is not the result of over-smoothing in the firn air reconstruction but rather reflects a real inconsistency between the CO-based inventory and the firn air data. A historical H_2 budget with anthropogenic emissions that rise monotonically produces a much better fit to the firn air-derived atmospheric history (*SI Appendix*, Fig. S7). Such a history is generated by scaling anthropogenic H_2 emissions to CO_2 emissions rather than CO emissions.

The mismatch between the firn air history and the box model simulation calls into question the assumed relationship between H_2 and CO emissions. The plateau and decrease in anthropogenic CO emissions during the late twentieth and early twenty-first centuries was due to stricter air pollution controls, increased use of automotive catalytic converters, and improvements in the efficiency of catalytic converters (35, 45–47). Automobile exhaust has been estimated to make up >80% of total anthropogenic H_2 emissions (4), and pollution controls are believed to have dramatically reduced H_2 emissions along with CO. Assuming that there were no unexplained compensating changes to the natural budget terms, a plateau or decline in anthropogenic H_2 should be evident in atmospheric levels. By contrast, both the firn air history and flask air measurements show increasing trends in H_2 in the southern hemisphere during the late twentieth century. Furthermore, atmospheric measurements by CSIRO (18) and AGAGE (19) show that H_2 was largely constant from 2000 to 2015 when automotive exhaust emissions of H_2 were decreasing (48–50). This implies that there must be other important sources

of anthropogenic H_2 which are increasing rapidly enough to offset the decreasing emissions from automotive exhaust.

The need for additional anthropogenic H_2 emissions that do not scale directly with CO is also evident in the difference between bottom-up and top-down estimates of anthropogenic H_2 emissions for the modern atmosphere. Top-down estimates require 17 to 18 $Tg\ y^{-1}$ of total anthropogenic emissions in 2008 (5, 51) as prescribed in the box model. By contrast, bottom-up estimates are 11 $Tg\ y^{-1}$ or less (4, 48–50). In their top-down estimate, Pieterse et al. (51) bridged this gap with 9.0 $Tg\ y^{-1}$ of emissions from residential combustion, much larger than the bottom-up estimate of 2.8 $Tg\ y^{-1}$ from Vollmer et al. (50).

This is further illustrated by forcing the box model with an H_2/CO anthropogenic emissions ratio calculated from the bottom-up estimate of anthropogenic H_2 from Vollmer et al. (Fig. 4B; ref. 50). With this historical inventory, the model yields good agreement with the firn air–derived history prior to 1985 but significantly underestimates the firn air history thereafter (Fig. 4C). Closing the gap between the box model result and the firn history would require an additional source that rose from 0 $Tg\ y^{-1}$ in 1985 to $\sim 8\ Tg\ y^{-1}$ in 2003.

Bottom-up estimates of anthropogenic H_2 emissions do not consider leakage from industrial processes. H_2 production for industrial use rose from roughly 40 $Tg\ y^{-1}$ to 80 $Tg\ y^{-1}$ between 1985 and 2005 (52). Estimated leakage rates are substantial, ranging from 0.5 to 12% with most estimates falling in the 1 to 4% range (7–9, 12). To our knowledge, direct measurements of H_2 leakage rates have not been published, and the uncertainties associated with these estimates are large. A 10% leakage rate would result in H_2 leakage increasing from 4 to 8 $Tg\ y^{-1}$ between 1985 and 2005 or roughly half of the late twentieth century disagreement between the box model and the firn air–derived history.

Northern Hemisphere Trends

Anthropogenic H_2 emissions are concentrated in the northern hemisphere, so it is informative to compare northern hemisphere atmospheric and Greenland firn air observations with the box model (Fig. 4D). Petrenko et al. (20) produced two reconstructions of atmospheric H_2 based on NEEM firn air measurements using two different firn air models. The reconstructions show a peak in atmospheric H_2 of 520 to 560 ppb during the 1980s and/or 1990s followed by a decrease to modern levels of about 515 ppb. The H_2/CO emissions ratios that produce agreement with the 2003 atmosphere yield modeled Greenland H_2 levels that increase during the mid-twentieth century and then plateau near 515 ppb from 1975 to 1985. Between 1985 and 2000, the box model shows H_2 levels decreasing to about 500 ppb (Fig. 4D).

Both the firn air–derived atmospheric histories and the box model show a distinct maximum in northern hemisphere H_2 during the late twentieth century. However, there is little evidence of such a maximum in the atmospheric measurements that remain roughly constant near modern levels throughout the late twentieth century (2, 13, 17, 19, 20). In addition, the timing of the maxima in the box model and firn air histories do not agree. The anthropogenic H_2/CO emissions ratio calculated from Vollmer et al. (50) yields modeled Greenland H_2 levels that rise from 400 to 440 ppb between 1960 and 1985. After 1985, H_2 levels remain approximately constant into the modern (Fig. 4D). Modeled H_2 levels using this emissions history are substantially lower than the firn air–derived histories and the flask measurements. It is clear that further work is needed to develop a clear understanding of northern hemispheric atmospheric H_2 trends and the factors determining those trends.

Sources versus Sinks

In this analysis, we attribute the model/data mismatch in the southern hemisphere to uncertainty in sources rather than sinks. It is difficult to construct a scenario whereby the observed trends

in H_2 are dominated by changes in the losses. For example, if the late twentieth century rise in southern hemispheric H_2 was due to a reduction in the soil microbial sink, one would expect an even larger rise in the northern hemisphere where the soil sink is primarily located. Large changes in atmospheric OH would be required to drive the observed trends, whereas photochemical models suggest that global atmospheric OH concentrations were relatively stable over the past century (53, 54). In addition, the reduction in OH required to cause the observed late twentieth century increase would cause an even larger relative increase in CO at a time when CO levels were stabilizing or declining (*SI Appendix*; refs. 33, 40, 54).

Conclusions

Measurements and modeling of firn air at Megadunes, Antarctica, were used to reconstruct the atmospheric history of southern-hemisphere atmospheric levels of H_2 from 1852 to 2003. This reconstruction shows that the atmosphere was relatively constant during the mid-late 1800s around 330 ppb, followed by a steady rise to the modern level of 550 ppb. The magnitude of the rise in atmospheric H_2 is consistent with increasing anthropogenic emissions and production from CH_4 oxidation. The Megadunes reconstruction validates the South Pole firn air record, providing confidence in the atmospheric history of H_2 and modeling of the transport of a highly permeable gas in firn (22). Developing such records for atmospheric H_2 is particularly important because the instrumental record over the twentieth century is limited in duration and subject to challenging calibration issues.

The firn air reconstruction shows a late twentieth century increase in atmospheric H_2 that has not previously been discussed in the literature (Figs. 1 and 3). Rising H_2 levels at this time are also evident in atmospheric measurements from Khalil and Rasmussen (13), CSIRO (18), and AGAGE (19). The same trend is evident in NOAA/GML data after empirical correction to the MPI09 calibration scale (*SI Appendix*). This trend raises questions about the assumption of tight coupling between anthropogenic H_2 and CO emissions during the entire twentieth century. Firn air–derived histories of atmospheric CO clearly show the expected decline due to reduced automotive emissions after 1970s (20, 33, 40). However, a similar trend is not evident in the Antarctic firn air–based H_2 atmospheric histories. Instead, H_2 continues to rise through the end of the twentieth century. This rise cannot be attributed to increasing CH_4 oxidation alone and requires increasing emissions from another anthropogenic source barring unexplained changes in natural sources or sinks. An additional nonautomotive anthropogenic source of H_2 would also help reconcile the significant (7 to 8 Tg) gap between bottom-up and top-down estimates of anthropogenic H_2 emissions.

It is notable that flask measurements in the southern hemisphere display a monotonic rise in atmospheric H_2 throughout the late twentieth century, while they show a plateau in H_2 levels in the northern hemisphere. For a gas with an atmospheric lifetime of $\sim 2\ y$, it is difficult to construct a realistic emissions scenario whereby a strong trend exists in one hemisphere and not the other. Firn air records from Antarctica and Greenland at sites with a wide range of accumulation rates are needed to validate late twentieth century H_2 flask measurements and improve understanding of the global budget and emissions history.

Methods

Firn Air Sampling and Measurements. During January of 2004, firn air was sampled at the Megadunes site in central Antarctica (23). A 3" diameter hole was bored to a depth of 70 m using an ice core drill. Drilling was paused at 15 unique depths so that firn air could be sampled. The sampling method was similar to techniques from previous published firn air studies (21, 55, 56). The hole was sealed above each sampling depth with an inflatable rubber packer to prevent contamination from the modern atmosphere. A waste air intake was positioned directly below the rubber packer. The waste air intake

was separated from the sample air intake by a 2 3/4" diameter baffle. Air was pumped from the waste air intake 3 times faster than from the sample air intake to ensure that no air that had been in contact with the rubber packer was sampled. In total, 34 2-L glass flasks with Teflon seals were filled for analysis by the NOAA/GML Carbon Cycle group.

The firm air samples were analyzed for H₂ at NOAA/GML using gas chromatography with a mercuric oxide reduction gas analyzer. Relative measurement uncertainty is estimated at ±2%. The H₂ measurements are reported as dry air mole fraction. The data were corrected for calibration drift, detector nonlinearity, and gravitational fractionation as described in *SI Appendix*. The raw and corrected H₂ measurements are shown in *SI Appendix*, Fig. S2 and Fig. 1B.

Firn Air Model. A one-dimensional, finite-difference, advective-diffusive model was used to simulate the evolution of H₂ levels in firn air. The model is largely derived from the work of Severinghaus et al. (23), and a complete model description is given in that work. The model also includes the pore close-off fractionation module from Patterson et al. (22). The model is broadly divided into an upper "diffusive zone" and lower "lock-in zone." In the diffusive zone, vertical gas transport occurs via wind-driven convective mixing and molecular diffusion. Diffusive mixing decreases with depth due to the increasing tortuosity of the firn. In the lock-in zone, vertical molecular diffusion ceases due to the presence of impermeable winter layers. Gas transport in the lock-in zone occurs primarily due to advection with a small nonfractionating mixing term. The model uses a forward Euler integration scheme and a time step of 324 s. The primary differences between our model and that of Severinghaus et al. (23) are the omission of thermal diffusion which is unimportant for H₂ and the inclusion of the pore close-off fractionation module. The model tracks the air content and composition in both open pores and closed bubbles as a function of time and depth. The model code is written and executed in MATLAB R2020a (Mathworks Inc.).

Firn physical properties are taken from Severinghaus et al. (23). The annual average temperature is 224 K. The bulk density profile (ρ_{firn}) is calculated from an empirical fit to density measurements of the firn core. Total porosity (s_{total}) is estimated from the density profile:

$$s_{\text{total}} = 1 - \frac{\rho_{\text{firn}}}{\rho_{\text{ice}}}, \quad [4]$$

where ρ_{ice} is the temperature dependent density of ice from Bader (kg m⁻³, ref. 57). Porosity is partitioned between open pores and closed bubbles using an empirical fit to measurements of density and closed porosity from Summit, Greenland (58). The accumulation rate at Megadunes has not been empirically constrained. The model accumulation rate is 2.8 cm y⁻¹ ice equivalent after Severinghaus et al. (23). The onset of the lock-in zone is at 64.5 m, as determined by the sharp gradient in the CO₂ and CH₄ depth profiles (*SI Appendix*, Fig. S3).

In the diffusive zone, model grid spacing is 0.5 m. Gas transport is dominated by convective mixing in the upper part of the diffusive zone and by molecular diffusion in the lower part of the diffusive zone. To simulate nonfractionating convective mixing, an eddy diffusivity term is added to the classical one-dimensional firn air transport equation (23, 58, 59). Additionally, we neglect the typical gravitational term and instead empirically correct the firn air measurements using the $\delta^{15}\text{N}$ depth profile as described in *SI Appendix*. Eq. 5 governs the evolution of the concentration of the gas of interest in the open pores (23):

$$s_o \frac{\partial C}{\partial t} = \frac{\partial}{\partial z} \left(\frac{\partial C}{\partial z} * [s_o D_{\text{mol}}(z, T, P) + s_o D_{\text{eddy}}(z)] \right) - s_o w(z) \frac{\partial C}{\partial z}, \quad [5]$$

where s_o is the open porosity, C is concentration of the gas of interest (mol m⁻³), D_{mol} is the depth, temperature, and pressure-dependent molecular diffusivity constant (m²s⁻¹), D_{eddy} is the depth-dependent eddy diffusivity (m²s⁻¹), and w is the downward velocity of the firn column (ms⁻¹). D_{mol}

and D_{eddy} are tuned using the measured CO₂ and $\delta^{15}\text{N}$ depth profiles. D_{mol} is then scaled using the ratio of the gas phase diffusivity of CO₂ to the gas phase diffusivity of H₂ (or other gas of interest). Temperature- and pressure-dependent gas phase diffusivities are calculated from Reid et al. (60). A detailed description of the tuning of the diffusivity profile is given in Severinghaus et al. (23).

Model grid spacing in the lock-in zone is one annual layer (~3 cm). There is no molecular diffusivity in the lock-in zone, and gas transport occurs via downward advection of annual layers once per year. Small values of eddy diffusivity are prescribed in the upper part of lock-in. This nonfractionating mixing term is included to account for vertical airflow caused by barometric pressure fluctuations. The model parameterizations were validated by forcing the model with the atmospheric history of CH₄ from Law Dome and the NOAA flask air network and comparing the resulting modeled and measured depth profiles (*SI Appendix*, Fig. S3; refs. 28–31).

In addition to diffusion and advection, pore close-off fractionation is parameterized in the firn air model. As closed pores (bubbles) are advected downward in the firn, internal pressure increases, creating a partial pressure gradient which drives a net diffusive flux of highly mobile trace gases out of the bubbles, through the ice lattice, and into the open pores. The open pores are therefore enriched in these gases. This phenomenon affects gases with kinetic diameters (KD) <3.6 Å such as Ne and He due to their large diffusivity in ice (21, 22). As a result of pore close-off fractionation, enrichments of nearly 10% in ²²Ne (KD = 2.75 Å) content were observed in the base of lock-in at Megadunes.

Pore close-off fractionation must also affect H₂ (20, 22). Patterson et al. (22) examined both "kinetic" and "equilibrium" parameterizations for pore close-off fractionation of H₂ (21, 22). In the kinetic parameterization, the time scale for diffusion between closed and open pores is assumed to be slow relative to the advective time scale. A small constant flux out of a given bubble is maintained until the bubble has been compressed by 5% after which the flux out of the bubble decreases to 0. In the equilibrium parameterization, permeation through the ice lattice is assumed to be fast enough to achieve equilibrium between open pores and closed bubbles throughout the model domain. Reconstructions of atmospheric H₂ from South Pole firn air measurements were relatively insensitive to the choice of parameterization (22). Recent laboratory measurements indicate that the permeability of H₂ in ice is sufficiently fast to equalize the partial pressure of H₂ in the open porosity and closed bubbles under the conditions at Megadunes (61). The equilibrium parameterization was used in this study as described by Eqs. 6–8:

$$P_n = (P_{\text{bubble}} x_{n(\text{bubble})} s_c + P x_{n(\text{firn})} s_o) / s_{\text{total}} \quad [6]$$

$$x_{n(\text{bubble})} = P_n / P_{\text{bubble}} \quad [7]$$

$$x_{n(\text{firn})} = P_n / P_{\text{ambient}}, \quad [8]$$

where P is pressure, x is mole fraction, and s_c is closed porosity. The subscripts *bubble* and *firn* distinguish between the closed and open porosity. Eqs. 6–8 are executed at every timestep in each grid cell in the model. Patterson et al. (22) demonstrated that dissolution and vertical diffusion through the ice lattice may have effects on the order of 2 to 3% in the lock-in zone if a permeation constant of 5.0 × 10⁻¹⁵ mol m⁻¹s⁻¹Pa⁻¹ is assumed. Recent measurements of the permeation of H₂ in ice yield an estimate more than 100 times lower (61). The lower permeation constant greatly reduces the magnitude of these effects. For the purposes of this work, dissolution and vertical diffusion through the ice lattice are neglected.

Data Availability. Firn air and atmospheric measurements of H₂ data have been deposited in the DRYAD Data Repository (<https://doi.org/10.7280/D18698>; ref. 24).

ACKNOWLEDGMENTS. This research was supported by the NSF (OPP-1907974).

1. S. van Renssen, The hydrogen solution? *Nat. Clim. Chang.* **10**, 799–801 (2020).
2. P. C. Novelli et al., Molecular hydrogen in the troposphere: Global distribution and budget. *J. Geophys. Res.* **104**, 30427–30444 (1999).
3. H. Price et al., Global budget of molecular hydrogen and its deuterium content: Constraints from ground station, cruise, and aircraft observations. *J. Geophys. Res. Atmos.* **112**, 1–16 (2007).
4. D. H. Ehhalt, F. Rohrer, The tropospheric cycle of H₂: A critical review. *Tellus B Chem. Phys. Meteorol.* **61**, 500–535 (2009).
5. G. Pieterse et al., Global modelling of H₂ mixing ratios and isotopic compositions with the TM5 model. *Atmos. Chem. Phys.* **11**, 7001–7026 (2011).
6. M. J. Prather, An environmental experiment with H₂. *Science* **302**, 581–582 (2003).
7. M. G. Schultz, T. Diehl, G. P. Brasseur, W. Zittel, Air pollution and climate-forcing impacts of a global hydrogen economy. *Science* **302**, 624–627 (2003).

8. D. Wang et al., Impact of a future H₂-based road transportation sector on the composition and chemistry of the atmosphere—Part 1: Tropospheric composition and air quality. *Atmos. Chem. Phys.* **13**, 6117–6137 (2013).
9. T. K. Tromp, R.-L. Shia, M. Allen, J. M. Eiler, Y. L. Yung, Potential environmental impact of a hydrogen economy on the stratosphere. *Science* **300**, 1740–1742 (2003).
10. D. Wang et al., Impact of a future H₂-based road transportation sector on the composition and chemistry of the atmosphere—Part 2: Stratospheric ozone. *Atmos. Chem. Phys.* **13**, 6139–6150 (2013).
11. M. Z. Jacobson, W. G. Colella, D. M. Golden, Atmospheric science: Cleaning the air and improving health with hydrogen fuel-cell vehicles. *Science* **308**, 1901–1905 (2005).
12. W. G. Colella, M. Z. Jacobson, D. M. Golden, Switching to a U.S. hydrogen fuel cell vehicle fleet: The resultant change in emissions, energy use, and greenhouse gases. *J. Power Sources* **150**, 150–181 (2005).

Patterson et al.

H₂ in Antarctic firn air: Atmospheric reconstructions and implications for anthropogenic emissions

PNAS | 7 of 8

<https://doi.org/10.1073/pnas.2103335118>

13. M. A. K. Khalil, R. A. Rasmussen, Global increase of atmospheric molecular hydrogen. *Nature* **347**, 743–745 (1990).
14. D. H. Ehhalt, U. Schmidt, L. E. Heidt, Vertical profiles of molecular hydrogen in the troposphere and stratosphere. *J. Geophys. Res.* **82**, 5907–5911 (1977).
15. F. A. Paneth, The chemical composition of the atmosphere. *Q. J. R. Meteorol. Soc.* **63**, 433–438 (1937).
16. U. Schmidt, Molecular hydrogen in the atmosphere. *Tellus* **26**, 78–90 (1974).
17. P. C. Novelli, *Atmospheric Hydrogen Mixing Ratios from the NOAA GMD Carbon Cycle Cooperative Global Air Sampling Network, 1988–2005* (NOAA/ESRL, 2006).
18. R. L. Langenfelds *et al.*, Interannual growth rate variations of atmospheric CO₂ and its $\delta^{13}\text{C}$, H₂, CH₄, and CO between 1992 and 1999 linked to biomass burning. *Global Biogeochem. Cycles* **16**, 21–1–21–22 (2002).
19. R. G. Prinn *et al.*, History of chemically and radiatively important atmospheric gases from the Advanced Global Atmospheric Gases Experiment (AGAGE). *Earth Syst. Sci. Data* **10**, 985–1018 (2018).
20. V. V. Petrenko *et al.*, A 60 yr record of atmospheric carbon monoxide reconstructed from Greenland firn air. *Atmos. Chem. Phys.* **13**, 7567–7585 (2013).
21. J. P. Severinghaus, M. O. Battle, Fractionation of gases in polar ice during bubble close-off: New constraints from firn air Ne, Kr and Xe observations. *Earth Planet. Sci. Lett.* **244**, 474–500 (2006).
22. J. D. Patterson *et al.*, Atmospheric history of H₂ over the past century reconstructed from South Pole firn air. *Geophys. Res. Lett.* **47**, 1–8 (2020).
23. J. P. Severinghaus *et al.*, Deep air convection in the firn at a zero-accumulation site, central Antarctica. *Earth Planet. Sci. Lett.* **293**, 359–367 (2010).
24. E. Saltzman, Dataset, H₂ in firn air from Megadunes, Antarctica. *Dryad*. <https://doi.org/10.7280/D18698>. Deposited 4 August 2021.
25. A. Jordan, B. Steinberg, Calibration of atmospheric hydrogen measurements. *Atmos. Meas. Tech.* **4**, 509–521 (2011).
26. V. Rommelaere, L. Arnaud, J. Barnola, Reconstructing recent atmospheric trace gas concentrations from polar firn and bubbly ice data by inverse methods. *J. Geophys. Res.* **102**, 30069–30083 (1997).
27. J. T. Randerson, G. R. van der Werf, L. Giglio, G. J. Collatz, P. S. Kasibhatla, “Global Fire Emissions Database, Version 4.1 (GFEDv4)” (ORNL DAAC, Oak Ridge, Tennessee, 2018). <https://doi.org/10.3334/ORNLDAAC/1293>. Accessed 14 October 2020.
28. D. M. Etheridge, L. P. Steele, R. L. Langenfelds, R. J. Francey, Natural and anthropogenic changes in atmospheric CO₂ over the last 1000 years from air in Antarctic ice and firn. *J. Geophys. Res.* **101**, 4115–4128 (1996).
29. D. M. Etheridge, L. P. Steele, R. J. Francey, R. L. Langenfelds, Atmospheric methane between 1000 A.D. and present: Evidence of anthropogenic emissions and climatic variability. *J. Geophys. Res.* **103**, 15,979–15,993 (1998).
30. E. Dlugokencky, X. Lan, *Atmospheric Methane Dry Air Mole Fractions from the NOAA GML Carbon Cycle Cooperative Global Air Sampling Network, 1983–2019* (NOAA/ESRL, 2020).
31. E. Dlugokencky, X. Lan, *Atmospheric Carbon Dioxide Dry Air Mole Fractions from the NOAA GML Carbon Cycle Cooperative Global Air Sampling Network, 1968–2019* (NOAA/ESRL, 2020).
32. T. Marik, *Atmospheric $\delta^{13}\text{C}$ and δD Measurements to Balance the Global Methane Budget* (Ruprecht-Karls-Univ, 1998).
33. K. R. Verhulst, “Atmospheric Histories of Ethane and Carbon Monoxide from Polar Firn Air and Ice Cores,” PhD dissertation, University of California Irvine, Irvine, CA (2014).
34. J. F. Lamarque *et al.*, Historical (1850–2000) gridded anthropogenic and biomass burning emissions of reactive gases and aerosols: Methodology and application. *Atmos. Chem. Phys.* **10**, 7017–7039 (2010).
35. G. Huang *et al.*, Speciation of anthropogenic emissions of nonmethane volatile organic compounds: A global gridded data set for 1970–2012. *Atmos. Chem. Phys.* **17**, 7683–7701 (2017).
36. M. Kopacz *et al.*, Global estimates of CO sources with high resolution by adjoint inversion of multiple satellite datasets (MOPITT, AIRS, SCIAMACHY, TES). *Atmos. Chem. Phys.* **10**, 855–876 (2010).
37. D. T. Shindell *et al.*, Multimodel simulations of carbon monoxide: Comparison with observations and projected near-future changes. *J. Geophys. Res. Atmos.* **111**, D19306 (2006).
38. O. Stein *et al.*, On the wintertime low bias of Northern Hemisphere carbon monoxide found in global model simulations. *Atmos. Chem. Phys.* **14**, 9295–9316 (2014).
39. K. Miyazaki *et al.*, Simultaneous assimilation of satellite NO₂, O₃, CO, and HNO₃ data for the analysis of tropospheric chemical composition and emissions. *Atmos. Chem. Phys.* **12**, 9545–9579 (2012).
40. S. S. Assonov *et al.*, Evidence for a CO increase in the SH during the 20th century based on firn air samples from Berkner Island, Antarctica. *Atmos. Chem. Phys.* **7**, 295–308 (2007).
41. G. Petron *et al.*, Atmospheric carbon monoxide dry air mole fractions from the NOAA GML carbon cycle cooperative global air sampling network, 1988–2020, Version: 2021-07. <https://doi.org/10.15138/33bv-s284>. Accessed 10 August 2021.
42. M. Aydin *et al.*, Recent decreases in fossil-fuel emissions of ethane and methane derived from firn air. *Nature* **476**, 198–201 (2011).
43. D. Helmig *et al.*, Reconstruction of Northern Hemisphere 1950–2010 atmospheric nonmethane hydrocarbons. *Atmos. Chem. Phys.* **14**, 1463–1483 (2014).
44. D. Fowler *et al.*, The global nitrogen cycle in the twenty-first century. *Philos. Trans. R. Soc. Lond. B Biol. Sci.* **368**, 20130164 (2013).
45. B. N. Duncan *et al.*, Global budget of CO, 1988–1997: Source estimates and validation with a global model. *J. Geophys. Res. Atmos.* **112**, 1988–1997 (2007).
46. B. N. Duncan, J. A. Logan, Model analysis of the factors regulating the trends and variability of carbon monoxide between 1988 and 1997. *Atmos. Chem. Phys.* **8**, 7389–7403 (2008).
47. Emissions Database for Global Atmospheric Research (EDGAR), release EDGAR v5.0 (1970–2015). Eur. Comm. Jt. Res. Cent. (EC-JRC)/Netherlands Environ. Assess. Agency (PBL). (2020) https://data.europa.eu/doi/10.2904/JRC_DATASET_EDGAR. Accessed 24 November 2020.
48. S. W. Bond, *Atmospheric Abundance and Anthropogenic Sources of Molecular Hydrogen Status and Outlook Towards A H₂-intensive Economy* (ETH Zurich, 2010).
49. M. K. Vollmer *et al.*, Road vehicle emissions of molecular hydrogen (H₂) from a tunnel study. *Atmos. Environ.* **41**, 8355–8369 (2007).
50. M. K. Vollmer *et al.*, Molecular hydrogen (H₂) combustion emissions and their isotope (D/H) signatures from domestic heaters, diesel vehicle engines, waste incinerator plants, and biomass burning. *Atmos. Chem. Phys.* **12**, 6275–6289 (2012).
51. G. Pieterse *et al.*, Reassessing the variability in atmospheric H₂ using the two-way nested TM5 model. *J. Geophys. Res. Atmos.* **118**, 3764–3780 (2013).
52. International Energy Agency, “The Future of Hydrogen: Seizing today’s opportunities”. OECD. <https://doi.org/10.1787/1e0514c4-en>. Accessed 8 June 2021.
53. V. Naik *et al.*, Preindustrial to present-day changes in tropospheric hydroxyl radical and methane lifetime from the Atmospheric Chemistry and Climate Model Inter-comparison Project (ACCMIP). *Atmos. Chem. Phys.* **13**, 5277–5298 (2013).
54. J. Lelieveld, F. J. Dentener, W. Peters, M. C. Krol, On the role of hydroxyl radicals in the self-cleansing capacity of the troposphere. *Atmos. Chem. Phys.* **4**, 2337–2344 (2004).
55. M. Battle *et al.*, Atmospheric gas concentrations over the past century measured in air from firn at the South Pole. *Nature* **383**, 231–235 (1996).
56. J. P. Severinghaus, A. Grachev, M. Battle, Thermal fractionation of air in polar firn by seasonal temperature gradients. *Geochem. Geophys. Geosyst.* **2** (2001).
57. H. Bader, Density of ice as a function of temperature and stress. *Spec. Rep. U.S. Army Cold Reg. Res. Eng. Lab.* **64**, 1–6 (1964).
58. J. Schwander *et al.*, The age of the air in the firn and the ice at Summit, Greenland. *J. Geophys. Res.* **98**, 2831–2838 (1993).
59. C. M. Trudinger *et al.*, Modeling air movement and bubble trapping in firn. *J. Geophys. Res. Atmos.* **102**, 6747–6763 (1997).
60. R. C. Reid, J. M. Prausnitz, T. K. Sherwood, B. E. Poling, *The Properties of Gases and Liquids* (McGraw-Hill, ed. 4, 1987), pp. 581–631.
61. J. D. Patterson, E. S. Saltzman, Diffusivity and solubility of H₂ in ice Ih: Implications for the behavior of H₂ in polar ice. *J. Geophys. Res. Atmos.* **126**, 1–14 (2021).
62. E. Witrant *et al.*, A new multi-gas constrained model of trace gas non-homogeneous transport in firn: Evaluation and behaviour at eleven polar sites. *Atmos. Chem. Phys.* **12**, 11465–11483 (2012).