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EXTRANEOUS CARBON ASSESSMENTS IN RADIOCARBON MEASUREMENTS OF BLACK CARBON IN ENVIRONMENTAL MATRICES

Alysha I Coppola^{1,2} • Lori A Ziolkowski³ • Ellen R M Druffel¹

ABSTRACT. Extraneous carbon (C_{ex}) added during chemical processing and isolation of black carbon (BC) in environmental matrices was quantified to assess its impact on compound specific radiocarbon analysis (CSRA). Extraneous carbon is added during the multiple steps of BC extraction, such as incomplete removal of solvents, and carbon bleed from the gas chromatographic and cation columns. We use 2 methods to evaluate the size and $\Delta^{14}C$ values of C_{ex} in BC in ocean sediments that require additional pretreatment using a cation column with the benzene polycarboxylic acid (BPCA) method. First, the direct method evaluates the size and $\Delta^{14}C$ value of C_{ex} directly from the process blank, generated by processing initially empty vials through the entire method identically to the treatment of a sample. Second, the indirect method quantifies C_{ex} as the difference between processed and unprocessed (bulk) $\Delta^{14}C$ values in a variety of modern and ^{14}C -free or “dead” BC standards. Considering a suite of hypothetical marine sedimentary samples of various sizes and $\Delta^{14}C$ values and BC Ring Trial standards, we compare both methods of corrections and find agreement between samples that are $>50 \mu\text{g C}$. Because C_{ex} can profoundly influence the measured $\Delta^{14}C$ value of compound specific samples, we strongly advocate the use of multiple types of process standards that match the sample size to assess C_{ex} and investigate corrections throughout extensive sample processing.

INTRODUCTION

Black carbon (BC) is produced from the incomplete combustion of fossil fuels and biomass, ubiquitous in the atmosphere, sediments, soils, and water, and influences a wide range of biogeochemical processes (Schmidt and Noack 2000; Watson et al. 2005). With the new technological developments and smaller accelerator mass spectrometry (AMS) sample size requirements (Santos et al. 2007), the ability to isolate individual compounds using compound specific radiocarbon analysis (CSRA) allows for better understanding of the timescales of individual compounds from a carbon pool (Eglinton et al. 1996; Ingalls and Pearson 2005). The turnover times of BC within these pools are determined by partially oxidizing aromatic BC in environmental matrices using the benzene polycarboxylic acid (BPCA) method to form marker compounds, benzene polycarboxylic acids (BPCAs). These environmental matrices contain non-BC organic matter, fine siliceous dust and heavy metals in a heterogeneous mixture, which can complicate the processing of a BC sample. In turn, for BC to be separated from the matrix, extensive treatment is needed to remove metals that interfere with BC extraction using the BPCA method (Brodowski et al. 2005; Ziolkowski and Druffel 2009). However, extensive processing adds extraneous (C_{ex}) carbon, thereby influencing the size and $\Delta^{14}C$ value of the BC sample (Santos et al. 2007; Ziolkowski and Druffel 2009). The C_{ex} originates from 2 major sources: (1) the chemical processing associated with extracting the BC from the sample (in this case, the BPCA method and pretreatment) and (2) the purification of BPCA marker compounds on a preparative capillary gas chromatograph (PCGC). After correction for graphitization and combustion (Santos et al. 2007), the mass of C_{ex} and the $\Delta^{14}C$ value of the CSRA sample ($C_{measured}$) originates from the contributions of 2 sources (Equation 1):

$$C_{measured} = C_{BPCA} + C_{chemistry} + C_{PCGC} = C_{BPCA} + C_{ex} \quad (1)$$

where C_{BPCA} is the mass of the BC isolated from the sample and C_{ex} is the mass of the extraneous C added due to processing ($C_{chemistry}$) and PCGC collection (C_{PCGC}). In previous studies, ^{14}C analysis of standards of known chemical and isotopic composition have revealed deviations from consensus

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$\Delta^{14}\text{C}$ values, highlighting the need for correction of C_{ex} (Hwang and Druffel 2005; Santos et al. 2007; Ziolkowski and Druffel 2009).

For the purpose of correcting BC $\Delta^{14}\text{C}$ measurements, the size and isotopic composition of C_{ex} can be assessed using 2 different approaches. First, to implement direct method $\Delta^{14}\text{C}$ assessments, we evaluate process blanks, which are initially empty vials that are processed through the entire method, identically to the treatment of a sample or standard. These process blanks serve as direct estimates of the size and isotopic signature of C_{ex} , and are evaluated periodically over time. The mass of C_{ex} added during sample preparation ($C_{\text{chemistry}}$) and purification (C_{PCGC}) can be evaluated by process blanks. To evaluate C_{PCGC} , we use a direct blank, generated by solvent injection onto the PCGC. The difference between the direct blank (C_{PCGC}) and process blank ($C_{\text{chemistry+PCGC}}$) is the $C_{\text{chemistry}}$. It is particularly important for new users to distinguish how much C_{ex} originates from both the chemical preparation of the sample ($C_{\text{chemistry}}$) and PCGC (C_{PCGC}) to determine the quality and total uncertainty of the BC $\Delta^{14}\text{C}$ results.

The second method, the indirect method, assesses the C_{ex} assuming a 2-endmember approach. We assume C_{ex} has a dead ($\Delta^{14}\text{C} = -1000\text{‰}$) component and a modern ($\Delta^{14}\text{C} = 0\text{‰}$) component. Then, using process BC standards of known isotopic values (modern and dead), the size of C_{ex} is estimated by the deviation of the process standard from the consensus $\Delta^{14}\text{C}$ value. After corrections for graphitization and combustion in AMS measurements are made, a carbon mass balance is applied using the mass and isotopic signature of C_{ex} determined by indirect and direct methods to correct samples and standards (Equation 1) (Ziolkowski and Druffel 2009).

The aim of this study is to determine C_{ex} added in the extraction of BPCA marker compounds in marine sediment throughout pretreatment, nitric acid oxidation, derivitization, and PCGC collection. Using direct and indirect assessments of C_{ex} , we evaluate the magnitude and ^{14}C signature of C_{ex} , which allows us to calculate the true BC $\Delta^{14}\text{C}$ value (C_{BPCA}) of the sample. The sum of the different sources of C_{ex} can lead to significant contamination of samples, particularly for samples $<50 \mu\text{g C}$. This results in a size-related bias of the $\Delta^{14}\text{C}$ values reported by AMS laboratories, which usually include corrections only for combustion and graphitization of samples. We compare both direct and indirect C_{ex} assessments by applying corrections of C_{ex} to a suite of hypothetical sedimentary organic carbon (SOC) samples. Based on these assessments, we demonstrate corrections for C_{ex} added during BC extraction in marine sediments with the routine use of processed standards and blanks.

METHODS

Black Carbon Standards

Black-carbon-rich standard reference materials were selected from the multilaboratory method and standard comparison called the BC Ring Trial (Hammes et al. 2007). Two types of dead and modern process standards were used to facilitate comparative analyses of BC: 1) laboratory-produced BC-rich and 2) BC-containing environmental matrices containing fine siliceous clays and heavy metals. Grass char (*Oryza sativa*) and wood char (*Castanea sativa*) BC standards were used as modern standards to estimate the ^{14}C -depleted C_{ex} added during processing (Elmquist et al. 2004; Hammes et al. 2006). A ^{14}C -depleted standard, hexane soot (Akhter et al. 1985; Goldberg 1985; Hammes et al. 2007) was also used to estimate modern C_{ex} . Environmental BC standards that contained a silicate and metal matrix, including urban dust aerosol NIST Standard Reference Material (SRM 1649a) (National Institute of Standards and Technology 2001; Masiello et al. 2002), NIST Standard Reference Material marine sediment (SRM 1941b), and US Geological Survey Green River Shale (Abbey 1983; Gladney and Roelandts 1988; Govindaraju 1994) were used to estimate modern C_{ex} . To

observe the matrix effect in marine sediments samples, wood char was added to SRM 1941b that had previously been baked in a muffle furnace for 2 hr at 550 °C to remove organic carbon. Duplicates of standards were processed to assess total uncertainty of $\Delta^{14}\text{C}$ measurements.

Elimination of Polyvalent Metals, BPCA Oxidation, and Purification

Standards that contained polyvalent metals were treated with trifluoroacetic acid (TFA) to remove metals that interfere with BPCA analysis (Brodowski et al. 2005; Hammes et al. 2007; Ziolkowski and Druffel 2009). First, metals in the environmental matrices standards were removed by high-temperature (104 °C) and high-pressure digestion in TFA for 4 hr (Brodowski et al. 2005). The solution was passed through a 0.8- μm quartz filter into a vacuum filtration flask, and the filter was rinsed with Milli-QTM water. Sample retained on the quartz filter was dried at 30–40 °C for at least 3 hr before high-temperature, high-pressure digestion in 65% nitric acid, at 170 °C for 8 hr for the BPCA method (Brodowski et al. 2005; Ziolkowski and Druffel 2009). The BPCA method partially oxidizes aromatic BC, converting it to BPCA marker compounds (Glaser et al. 1998; Brodowski et al. 2005; Hammes et al. 2007; Ziolkowski and Druffel 2009). The solution was filtered and the filtrate was passed through a cation exchange column (Brodowski et al. 2005) and eluted into Erylermeyer flasks. Briefly, following the method of Ziolkowski et al. (2011), dehydrated BPCAs were dissolved in methanol that contained biphenyl-2,2'-dicarboxylic acid (internal standard) and titrated with (trimethylsilyl)diazomethane (Sigma Aldrich) in 2.0M ethyl ether to derivatize the carboxylic acids to methyl esters. Methanol was evaporated by a stream of nitrogen gas and dichloromethane was used to transfer the samples into freshly baked vial inserts (0.3 mL) for PCGC analysis and separation.

Methylated BPCAs were quantified for BPCA distributions and isolated for ¹⁴C analysis using a Hewlett Packard 6890 Preparative Column Gas Chromograph (PCGC) with an HP 7683B auto-injector, and Gerstel cooled injection system (CIS-4) with a split/splitless inlet. The CIS injector was operated in “solvent vent” mode, with a vent flow adjusted to 60 mL/min and 20 psi. The solvent venting time was 0.3 min, and the split vent time was 1 min. The injection volume was 4 μL for all collections. The temperature of the inlet was 40 °C, then increased to 300 °C at a rate of 10 °C/s, then kept isothermal for 3 min. A megabore fused-silica capillary column (50 m length) coated with 1 μm of DB-XLB was used for all samples in this study. Ultra high-purity hydrogen gas was used as the carrier gas at a flow rate of 8.7 mL/min. The temperature program on the PCGC for separating BPCAs started at 100 °C, 10 °C/min to 250 °C (isothermal for 15 min), 5 °C/min to 280 °C (isothermal for 5 min), then 25 °C/min to 320 °C (isothermal for 3 min). Approximately 1% of the flow was diverted to the FID, while the other 99% was sent to the fraction collector. The fraction collector was computer controlled to collect samples at specific retention times. The fraction collector switch temperature and transfer line was kept at 320 °C, and the traps were chilled at –10 °C. To standardize C_{ex} to samples that have different numbers of injections, time windows of collection, and injection volumes, we normalized the C_{ex} mass determined by a manometer measurement of pressure in a known volume to units of $\mu\text{g C}$ per min collection per 50 1- μL injections, after Ziolkowski and Druffel (2009). In all direct evaluations of C_{PCGC} , the 4-min collection window set for each blank injection reflected the same time window in which BPCAs were collected in a sample run.

Briefly, following the method of Ziolkowski et al. (2011), BPCA marker compounds were identified using commercially available BPCAs (Sigma Aldrich; 1,2,3 benzene tricarboxylic acid, pyromellitic acid, benzene pentacarboxylic acid, mellitic acid) and mass fragmentation patterns when run on a Finnigan Trace MS and GC/MS ESI at UC Irvine. The preparative fraction collector on the PCGC captured BPCA marker compounds with 3 to 6 substituted carboxylic acid groups from the partial oxidation of aromatic BC, including the nitrated BPCAs (about half of the BPCAs were nitrated). The collection windows were set to capture the eluting peaks of interest, for a total of 4 min. The

BPCA marker compounds with only 2-substituted carboxylic acid groups were not collected because they can be derived from recalcitrant lignin or other non-BC material that may survive oxidation (Brodowski et al. 2005). Between sample collections (30 injections), the GC column was baked out twice at 320 °C for 10 min, the injection needle was cleaned with dichloromethane, and a freshly baked (550 °C for 2 hr) injection liner was installed. Also, to remove any contamination or memory from the previous sample, the first 10 injections were discarded for ^{14}C measurements as per Ziolkowski et al. (2011). After combustion, graphitization, and C_{ex} assessments, standards were corrected for the ^{14}C -free derivative C added during the derivitization (Ziolkowski and Druffel 2009).

Radiocarbon Measurements

In preparation for ^{14}C analysis, BPCA marker compound isolates from the PCGC were transferred using dichloromethane to clean quartz tubes, dried, and combusted to CO_2 at 850 °C with cupric oxide and silver. The volume of the CO_2 gas produced from combustion was cryogenically purified, then quantified manometrically and reduced to graphite for ^{14}C analysis (Santos et al. 2007). Measurements were made at UC Irvine in the Keck Carbon Cycle Acceleration Mass Spectrometry Facility and normalized to the AMS $\delta^{13}\text{C}$. ^{14}C results were reported as $\Delta^{14}\text{C}$ without known-age correction (Stuiver and Polach 1977).

Preparation of Standards for Bulk Measurements

In order to facilitate the indirect assessment of C_{ex} , the $\Delta^{14}\text{C}$ values of the unprocessed standards were measured. Inorganic carbon was removed by acidification with 3% phosphoric acid (Hwang et al. 2005). Standards were prepared in small and large sizes to bracket the sizes of the BC samples.

RESULTS AND DISCUSSION

Direct Blank Evaluation of Mass and $\Delta^{14}\text{C}$ Value of C_{ex}

We directly evaluated the C_{ex} mass and $\Delta^{14}\text{C}$ value using 1) process blanks and 2) direct blanks on the PCGC. The process blank contained no sample but was subjected to the same preparatory steps as samples, so it includes both C_{PCGC} and $C_{\text{chemistry}}$ (Equation 1). The direct blank from the PCGC only is determined by injecting solvent onto the PCGC (C_{PCGC}).

We found the process blank ($C_{\text{chemistry+PCGC}}$) was $1.4 \pm 0.7 \mu\text{g C min}^{-1}$ per 50 1- μL injections in 2012 and $0.3 \pm 0.2 \mu\text{g C min}^{-1}$ per 50 1- μL injections in 2011 and the $\Delta^{14}\text{C}$ values were $-957 \pm 46\%$ and $-963 \pm 54\%$, respectively. The difference in the magnitude of C_{ex} between these 2 time periods highlights the imperative need for routine blank assessments.

In order to deduce the relative sizes of $C_{\text{chemistry}}$ and C_{PCGC} , we evaluated C_{PCGC} alone from the injections of clean solvent directly onto the PCGC. We made 230 injections to obtain enough C for an AMS analysis. We report a direct blank C_{PCGC} of $0.1 \pm 0.1 \mu\text{g C min}^{-1}$ per 50 1- μL injections with a $\Delta^{14}\text{C}$ value $-982 \pm 15\%$ (Table 1). Using a mass balance approach from the difference between the total $C_{\text{chemistry+PCGC}}$ determined from the process blank and C_{PCGC} using the direct blank, we calculate that the $C_{\text{chemistry}}$ in 2012 was $1.3 \pm 0.8 \mu\text{g C min}^{-1}$ per 50 1- μL injections. In 2012, we find that $\sim 10\%$ of C_{ex} is C_{PCGC} ($0.1 \pm 0.1 \mu\text{g C min}^{-1}$ per 50 1- μL injections) and $\sim 90\%$ is $C_{\text{chemistry}}$ ($1.3 \pm 0.7 \mu\text{g C min}^{-1}$ per 50 1- μL injections). Additional $C_{\text{chemistry}}$ may originate from the treatment of samples in the cation exchange column following the BPCA method. Interpretations of the $C_{\text{chemistry}}$ suggests that these extra steps add twice the amount of $C_{\text{chemistry}}$ as that found by Ziolkowski and Druffel (2009), who did not use the cation column and pretreatment steps.

Indirect Blank Evaluation of Mass and $\Delta^{14}\text{C}$ Value of C_{ex}

The second method of evaluating C_{ex} , the indirect method, involves processing standards of known consensus $\Delta^{14}\text{C}$ value and measuring the deviation from the unprocessed consensus $\Delta^{14}\text{C}$ value. Differences between the $\Delta^{14}\text{C}$ values of processed and unprocessed standards are used to measure the mass and $\Delta^{14}\text{C}$ value of the C_{ex} to thereby correct sample $\Delta^{14}\text{C}$ values of samples (Hwang and Druffel 2005; Ziolkowski and Druffel 2009; Santos et al. 2010). Incorporation of C_{ex} in standards is assumed to be the same as that in samples that are processed identically. In this study, modern and dead standards were processed throughout the entire pretreatment, chemical extraction, cation exchange column, and PCGC isolation. When using the indirect method, C_{ex} is assessed as 2 end-members, 1 modern ($\Delta^{14}\text{C} = 0\text{‰}$) and 1 dead ($\Delta^{14}\text{C} = -1000\text{‰}$), the mass-weighted sum of which equals C_{ex} .

Modern process standards are used to assess the dead component of C_{ex} , while dead standards are used to assess the modern component of C_{ex} . After standards are corrected for graphitization and combustion, dead C_{ex} is evaluated using a simple mass balance approach (Equation 1) (Ziolkowski and Druffel 2009; Santos et al. 2010). The measured AMS $\Delta^{14}\text{C}$ values of BPCAs produced from wood and grass char were lower than the consensus values due to the presence of low- ^{14}C C_{ex} . The mass of dead C in grass char and wood char standards were $1.8 \pm 0.9 \mu\text{g C min}^{-1}$ per 50 1- μL injections and $1.5 \pm 0.8 \mu\text{g C min}^{-1}$ per 50 1- μL injections, in 2011 and 2012, respectively. Low- ^{14}C standards are used to assess the modern component of C_{ex} (e.g. hexane soot) and samples that contain a silicate matrix (e.g. SRM 1649a aerosol dust and Green River Shale). The masses of modern C_{ex} in these standards were $0.2 \pm 0.1 \mu\text{g C min}^{-1}$ per 50 1- μL injections and $0.1 \pm 0.1 \mu\text{g C min}^{-1}$ per 50 1- μL injections, in 2011 and 2012, respectively.

From these modern and dead components of C_{ex} , the calculated $\Delta^{14}\text{C}$ values of C_{ex} are $-842 \pm 26\text{‰}$ and $-933 \pm 25\text{‰}$ in 2011 and 2012, respectively (Table 1). Variables such as different users and GC column degradation change with time, making it imperative that standards are routinely processed to document the inevitable variability of C_{ex} . To maintain consistency with these C_{ex} variations, standards processed with the same suite of BC samples should be used to correct the $\Delta^{14}\text{C}$ data. In other words, C_{ex} evaluations using the indirect method should be performed with every suite of BC measurements to adequately correct BC sample $\Delta^{14}\text{C}$ values.

Correction of Standards Using C_{ex} Assessments

We report corrected $\Delta^{14}\text{C}$ values for standards using the mass and $\Delta^{14}\text{C}$ value of C_{ex} determined by both methods. More than half of the standard $\Delta^{14}\text{C}$ values corrected using the direct method were outside 2σ from the consensus values (Table 2 and Appendix). In contrast, all but 2 of the standards corrected using the indirect method agreed within 2σ of the consensus values (Table 2). There were greater deviations in corrected $\Delta^{14}\text{C}$ values of modern standards because the majority of C_{ex} is ^{14}C -depleted, thereby affecting the modern $\Delta^{14}\text{C}$ values more substantially. The low- ^{14}C standards (SRM 1649a aerosol dust, hexane soot, and Green River Shale) had corrected $\Delta^{14}\text{C}$ values that were closer to their consensus values (Table 2).

The direct correction applies 1 mass and 1 $\Delta^{14}\text{C}$ value from the process blank, whereas the indirect correction is determined using the average value of dead and modern C_{ex} for a large range of standard types. The indirect method includes the variability of sample processing with multiple standards that mirrors the variability of C_{ex} . The ability of indirect evaluations to correct standards illustrates why we recommend using the indirect method for correcting BC $\Delta^{14}\text{C}$ measurements.

Table 1 Standards subjected to various treatments to evaluate the C_{ex} in the determination of BC in marine sediment. The uncertainty of the mass of C_{ex} was estimated as 50% of the sample mass (but no lower than 0.1 μg of C/min per 50 1- μL injections). The uncertainty of the C_{ex} mass using the indirect method correction was estimated at 50% of the mass value.

Type of standard	Evaluates for	Process standard	n	Cation column and pretreatment	BPCA	PCGC	Extraneous carbon, C_{ex}	
							Average $\mu\text{g C/min}$ per fifty 1- μL injections ^a	$\Delta^{14}\text{C}$ (‰)
Indirect blank								
Dead	Modern C_{ex}	SRM 1649a Hexane soot Green River Shale	4 2 2	Yes Yes Yes	Yes Yes Yes	Yes Yes Yes	0.1 \pm 0.1 (2012) 0.2 \pm 0.1 (2011)	0
Modern	Dead C_{ex}	Grass char Wood char Wood char in muffled SRM1941b	3 4 2	No No Yes	Yes Yes Yes	Yes Yes Yes	1.5 \pm 0.8 (2012) 1.8 \pm 0.9 (2011)	-1000
Total indirect								
							1.6 \pm 0.9 2.0 \pm 1.0	-933 \pm 25 (2012) -842 \pm 26 (2011)
Direct blank								
Process blank (collected in 2012)	Dead and Modern	—	1	Yes	Yes	Yes	1.4 \pm 0.7	-957 \pm 46
Process blank (collected in 2011)	Dead and Modern	—	1	Yes	Yes	Yes	0.3 \pm 0.2	-963 \pm 54
Assessment of C_{PCGC}								
Solvent injection into PCGC (2012)	Dead and Modern	—	1	No	No	Yes	0.1 \pm 0.1	-982 \pm 15

^aThe masses of C_{ex} were normalized to $\mu\text{g C min}^{-1}$ per fifty 1- μL injections.

Extraneous C in ¹⁴C Measurements of Black Carbon

Table 2 Mass and $\Delta^{14}\text{C}$ of the unprocessed and isolated BPCAs in processed standard before and after corrections for C_{ex} .

Standard type (lab code UCID-)	Consensus BC $\Delta^{14}\text{C}$ values (‰)	BPCA compound class		
		$\mu\text{g C}$	Direct method corrected $\Delta^{14}\text{C}$ (‰)	Indirect method corrected $\Delta^{14}\text{C}$ (‰)
Grass char (13206) (13103)	$53 \pm 5^{\text{a}}$ ($n = 4$)	52	-69 ± 19	$+27 \pm 25$
		85	$+53 \pm 11$	$+89 \pm 49$
Wood char (13180) (13179) (16519) (16520)	$165 \pm 5^{\text{a}}$ ($n = 4$)	108	$+149 \pm 14$	$+186 \pm 33$
		178	$+106 \pm 27$	$+139 \pm 34$
		108	$+138 \pm 31$	$+143 \pm 28$
		60	$+147 \pm 21$	$+144 \pm 53$
Wood char in a matrix SRM 1941b (16509) (16510)	$165 \pm 5^{\text{a}}$ ($n = 2$)	40	$+299 \pm 25$	$+129 \pm 28$
		50	$+175 \pm 23$	$+130 \pm 43$
Aerosol SRM 1649a (13183) (13101) (13102) (13184)	$-885 \pm 50^{\text{b}}$	21	-653 ± 28	-884 ± 50
		37	-899 ± 10	-963 ± 29
		25	-742 ± 24	-859 ± 40
		77	-906 ± 8	-924 ± 47
Hexane soot (16511) (16512)	-982 ± 8 ($n = 3$)	102	-986 ± 3	-992 ± 4
		35	-998 ± 2	-992 ± 4
Green River Shale (13207) (13182)	<-976 ($n = 3$)	42	-823 ± 32	-894 ± 38
		73	-796 ± 15	-834 ± 18

^aDetermined by the combustion of unprocessed samples.

^bAerosol 1649a is a mixture of BC and other aerosols containing organic carbon. The unprocessed BC $\Delta^{14}\text{C}$ values are from Ziolkowski and Druffel (2009) and Currie et al. (2002) ($-620 \pm 50\%$; $n = 5$).

Evaluation of the Sediment Matrix on Corrected $\Delta^{14}\text{C}$ Values

We needed to verify the presence or absence of a matrix effect associated with the metals contained in our sediment samples. The effect on C_{ex} of a sediment matrix was evaluated by comparing corrected $\Delta^{14}\text{C}$ values of wood char run with and without a sediment (SRM 1941b) matrix. We processed 2 wood char standards that had added SRM 1941b marine sediment. Results showed that both standards had the same mass of C_{ex} ($1.5 \pm 0.8 \mu\text{g C min}^{-1}$ per 50 $1\text{-}\mu\text{L}$ injections), indicating that C_{ex} is unaffected by the presence of a sediment matrix. When the indirect corrections were applied to wood char standards containing a sediment matrix, the corrected $\Delta^{14}\text{C}$ values were $129 \pm 28\%$ and $130 \pm 43\%$ (Table 2), within 2σ of the consensus value ($165 \pm 5\%$). The corrected $\Delta^{14}\text{C}$ values of the wood char standards without a matrix ($n = 4$) were also equal to the consensus value (average $\Delta^{14}\text{C} = 153 \pm 10\%$).

Hypothetical Marine Sediment BC $\Delta^{14}\text{C}$

To test how the addition of C_{ex} impacts the $\Delta^{14}\text{C}$ values of BC in sediment samples of various sizes, we applied these corrections to a suite of hypothetical samples of different sizes (25 to 150 $\mu\text{g C}$) and $\Delta^{14}\text{C}$ values ($\Delta^{14}\text{C} = -250\text{‰}$ to -750‰). We assumed that the C_{ex} associated with these hypothetical samples was the same as those obtained in 2012 (Table 1). Corrected $\Delta^{14}\text{C}$ values associated with both indirect and direct method corrections are within 2σ of the consensus values (Figure 1). The differences between indirect and direct corrections diverge for samples that are $\leq 25 \mu\text{g C}$, where the “true” $\Delta^{14}\text{C}$ values (C_{BPCA}) (Equation 1) of samples are close to modern ($> -250\text{‰}$) (Figure 1). Deviations in the corrected $\Delta^{14}\text{C}$ values for samples $< 25 \mu\text{g C}$ illustrate the difficulty of assessments of both the mass and $\Delta^{14}\text{C}$ variability of C_{ex} within sample suites. Gaining insights with the use of multiple standards and duplicate samples is necessary to constrain C_{ex} .

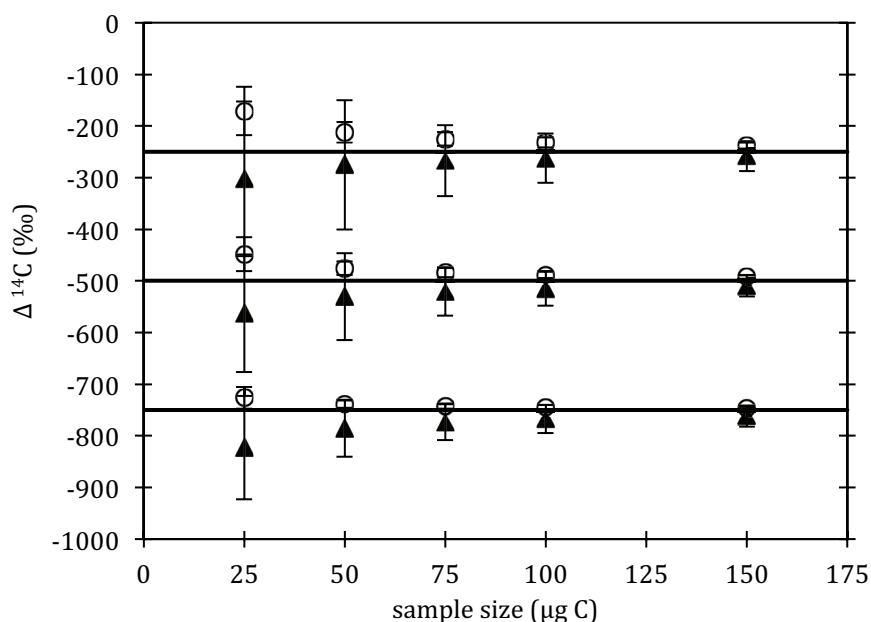


Figure 1 Masses of hypothetical BC in sediment samples plotted versus corrected $\Delta^{14}\text{C}$ values to illustrate the difference between direct (open circles) and indirect (black triangles) corrections. Solid black lines represent the consensus $\Delta^{14}\text{C}$ values for each hypothetical sample.

CONCLUSION

The main challenge for reporting meaningful BC $\Delta^{14}\text{C}$ values in sedimentary matrices involves multiple evaluations of C_{ex} added during extensive chemical processing and PCGC separation. The mass of C_{ex} is significant and variable, thus requiring a correction beyond that made for graphitization and combustion. Correction for C_{ex} is critical, especially for samples $< 25 \mu\text{g C}$.

We were unable to reliably correct standards to their consensus $\Delta^{14}\text{C}$ values using the direct method. We recommend use of the indirect method to capture the variability of sample processing by the use of multiple standards. Standard sizes should match the sample sizes and approximate $\Delta^{14}\text{C}$ values. Although processing dead and modern BC standards is time consuming, it is crucial because C_{ex} is variable over time. To gain the most information about the mass and isotopic signatures of C_{ex} , the indirect method is recommended.

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REFERENCES

- Abbey S. 1983. Studies in "standard samples" of silicate rocks and minerals 1969–1982. *Canadian Geological Survey Paper* 83:85–114.
- Akhter MS, Chughtai AR, Smith DM. 1985. The structure of hexane soot I: spectroscopic studies. *Applied Spectroscopy* 39(1):143–53.
- Brodowski S, Rodionov A, Haumaier L, Glaser B, Amelung W. 2005. Revised black carbon assessment using benzene polycarboxylic acids. *Organic Geochemistry* 36(9):1299–310.
- Currie L, Benner R, Kessler JD, Kliendist DB, Marolf JV, Slater JE, Wise SA, Cachier H, Cary R, Chow JC, Watson J, Druffel ERM, Masiello CA, Eglinton TI, Pearson A, Reddy CM, Gustafsson O, Hartmann PC, Quinn JG, Hedges JI, Prentice KM, Kirchstetter TW, Novakov T, Puxbaum H, Schmid H. 2002. A critical evaluation of interlaboratory data on total, elemental, and isotopic carbon in the carbonaceous particle reference material, NIST SRM 1649a. *Journal of Research of the National Institute of Standards and Technology* (107):279–98.
- Eglinton TI, Aluwihare LI, Bauer JE, Druffel ERM, McNichol AP. 1996. Gas chromatographic isolation of individual compounds from complex matrices for radiocarbon dating. *Analytical Chemistry* 68(5):904–12.
- Elmquist M, O Gustafsson, Andersson P. 2004. Quantification of sedimentary black carbon using the chemothermal oxidation method: an evaluation of *ex situ* pretreatments and standard additions approach. *Limnology and Oceanography: Methods* 2:417–27.
- Gladney ES, Roelands I. 1988. 1987 compilation of elemental concentration data for USGS BHVO-1, MAG-1, QLO-1, RGM-1, SCo-1, SDC-1, SGR-1, and STM-1. *Geostandards Newsletter* 12(2):253–362.
- Glaser B, Haumaier L, Guggenberger G, Zech W. 1998. Black carbon in soils: the use of benzenecarboxylic acids as specific markers. *Organic Geochemistry* 29(4):811–9.
- Goldberg ED. 1985. *Black Carbon in the Environment*. New York: Wiley.
- Govindaraju K. 1994. 1994 compilation of working values and descriptions for 383 geostandards. *Geostandards Newsletter* 18(S1):1–158.
- Hammes K, Smernik RJ, Skjemstad JO, Herzog A, Vogt UF, Schmidt MWI. 2006. Synthesis and characterization of laboratory-charred grass straw (*Oryza sativa*) and chestnut wood (*Castanea sativa*) as reference materials for black carbon quantification. *Organic Geochemistry* 37(11):1629–33.
- Hammes K, Schmidt MWI, Smernik RJ, Currie LA, Ball WP, Nguyen TH, Louchouart P, Houel S, Gustafsson Ö, Elmquist M, Cornelissen G, Skjemstad JO, Masiello CA, Song J, Peng P, Mitra S, Dunn JC, Hatcher PG, Hockaday WC, Smith DM, Hartkopf-Fröder C, Böhmer A, Lür B, Huebert BJ, Amelung W, Brodowski S, Huang L, Zhang W, Gschwend PM, Flores-Cervantes DX, Largeau C, Rouzaud J-N, Rumpel C, Guggenberger G, Kaiser K, Rodionov A, Gonzalez-Vila FJ, Gonzalez-Perez JA, de la Rosa JM, Manning DAC, López-Capel E, Ding L. 2007. Comparison of quantification methods to measure fire-derived (black/elemental) carbon in soils and sediments using reference materials from soil, water, sediment and the atmosphere. *Global Biogeochemical Cycles* 21: GB002914, doi:10.1029/2006GB002914.
- Hwang J, Druffel ERM. 2005. Blank correction for $\Delta^{14}\text{C}$ measurements in organic compound classes of oceanic particulate matter. *Radiocarbon* 47(1):75–87.
- Ingalls AE, Pearson A. 2005. Ten years of compound specific radiocarbon analysis. *Oceanography* 18(3): 18–31.
- Masiello CA, Druffel ERM, Currie LA. 2002. Radiocarbon measurements of black carbon in aerosols and ocean sediments. *Geochimica et Cosmochimica Acta* 66(6):1025–36.
- National Institute of Standards and Technology. 2001. Certificate of analysis, Standard Reference Material (SRM) 1649a: urban dust, report. Gaithersburg: NIST.
- Santos GM, Southon JR, Griffin S, Beaupre SR, Druffel ERM. 2007. Ultra small-mass AMS ^{14}C sample preparation and analyses at KCCAMS/UCI Facility. *Nuclear Instruments and Methods in Physics Research B* 259(1):293–302.
- Santos GM, Southon JR, Drenzek NJ, Ziolkowski LA, Druffel ERM, Xu X, Zhang D, Trumbore SE, Eglinton TI, Hughen KA. 2010. Blank assessment for ultra-small samples: chemical extraction and separation versus AMS. *Radiocarbon* 52(3):1322–35.
- Schmidt MWI, Noack AG. 2000. Black carbon in soils and sediments: analysis, distribution, implications, and current challenges. *Global Biogeochemical Cycles* 14(3):777–94.
- Stuiver M, Polach HA. 1977. Discussion: reporting of ^{14}C data. *Radiocarbon* 19(3):355–63.

Watson JG, Chow JC, Chen LWA. 2005. Summary of organic and elemental carbon/black carbon analysis methods and intercomparisons. *Aerosol Air Quality Research* 5(1):65–102.

Ziolkowski LA, Druffel ERM. 2009. Quantification of extraneous carbon during compound specific radio-carbon analysis of black carbon. *Analytical Chemistry*

81(24):10,156–61.

Ziolkowski LA, Chamberlin AR, Greaves J, Druffel ERM. 2011. Quantification of black carbon in marine systems using the benzene polycarboxylic acid method: a mechanistic and yield study. *Limnology and Oceanography: Methods* 9:140.

APPENDIX

Calculation of Uncertainties in $\Delta^{14}\text{C}$ Corrected Values

To determine the propagated total uncertainty of $\Delta^{14}\text{C}_{BPCA}$ (Equation 2), we applied the following equation:

$$\begin{aligned} \sigma \Delta^{14}\text{C}_{BPCA}^2 = & \left(\frac{\partial \Delta^{14}\text{C}_{BPCA}}{\partial \Delta^{14}\text{C}_{reported}} \right)^2 \sigma \Delta^{14}\text{C}_{reported}^2 + \left(\frac{\partial \Delta^{14}\text{C}_{BPCA}}{\partial \Delta^{14}\text{C}_{ex}} \right)^2 \sigma \Delta^{14}\text{C}_{ex}^2 + \\ & \left(\frac{\partial \Delta^{14}\text{C}_{BPCA}}{\partial C_{reported}} \right)^2 \sigma C_{reported}^2 + \left(\frac{\partial \Delta^{14}\text{C}_{BPCA}}{\partial C_{ex}} \right)^2 \sigma C_{ex}^2 \end{aligned} \quad (2)$$

where $\sigma \Delta^{14}\text{C}_{BPCA}$ is the propagated error of the corrected $\Delta^{14}\text{C}$ value, $\sigma \Delta^{14}\text{C}_{reported}$ is the AMS uncertainty of $\Delta^{14}\text{C}_{reported}$ (machine uncertainty), $\sigma \Delta^{14}\text{C}_{ex}$ is the uncertainty for $\Delta^{14}\text{C}_{ex}$, $\sigma C_{reported}$ is the uncertainty for $C_{reported}$ (uncertainty in graphitization), and σC_{ex} is the uncertainty in C_{ex} (assigned as 50%). The total uncertainty for $\Delta^{14}\text{C}_{ex}$ and C_{ex} was used as the direct process blank.