Lawrence Berkeley National Laboratory

Lawrence Berkeley National Laboratory

Title

Characterization of the effects of Al-substitution in layered oxide cathode materials for Li-ion batteries

Permalink

https://escholarship.org/uc/item/1k88g66v

Author

Conry, Thomas E.

Publication Date

2011-10-10

Characterization of the effects of Al-substitution in layered oxide cathode materials for Li-ion batteries

Thomas E. Conry¹, Apurva Mehta², and Marca Doeff¹

¹Lawrence Berkeley National Lightsource 1 Cyclotron Rd., Berkeley, CA, 94720

²Stanford Synchrotron Radiation Laboratory 2575 Sand Hill Rd., Menlo Park, CA, 94025

Layered mixed transition metal oxide materials have been extensively investigated as replacements for LiCoO2 in consumer electronics and, especially, for potential electric vehicle applications. NMC materials LiNi_xCo_{1-2x}Mn_xO₂ with x=0.33 and x=0.4 have proven commercially successful with improved performance, though a further reduction of the Co-content is desirable for cost and environmental reasons. Al-substitution for Co has been shown to improve both cost and safety aspects of the cathode materials and, notably, enhance the cycling stability for reasons that are not explicitly known (Fig. 1). This work explores the specific effects of Al-substitution in the $LiNi_{0.45}Co_{0.1-x}Al_xMn_{0.45}O_2$ (0 \le x \le 0.1) system, in which Co is decreased both by a relative increase of the amounts of Ni and Mn and direct replacement by Al. The structural and electronic effects of Al-substitution are probed using X-ray absorption spectroscopy (XAS). The near-edge structure (XANES) elucidates the charge compensation mechanisms and electronic structure changes during battery operation. Analysis of the extended fine structure (EXAFS) is used to determine element-specific local structure within the materials during charge, and the changes that occur as a result of prolonged cycling (Fig. 2). Additionally, both in situ and ex situ X-ray diffraction (XRD) are used to further characterize the structural evolution of the materials during the charge and discharge processes and cycle lifetimes.

ACKNOWLEDGEMENTS

This work is supported by the Assistant Secretary for Energy Efficiency and Renewable Energy, Office of Vehicle Technologies of the U.S. Department of Energy under Contract No. DE-AC02-05CH11231.

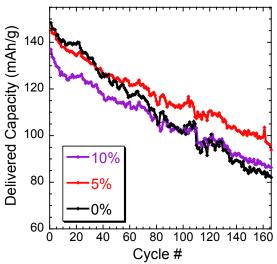


Figure 1: Discharge capacity vs. cycle for LiNi_{0.45}Co_{0.1}. $_xAl_xMn_{0.45}O_2$ ($0 \le x \le 0.1$) materials cycled between 2-4.3 V at 12 mA/g.

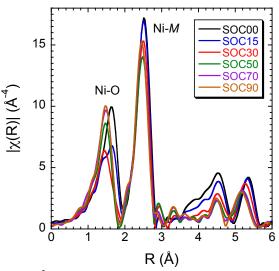


Figure 2: k^3 -weighted Fourier transform of Ni EXAFS data at various states-of-charge (SOC) during Li extraction from $Li_xNi_{0.45}Co_{0.1}Mn_{0.45}O_2$. Marked peaks correspond to scattering between Ni-O and Ni-M where M=Ni, Mn, or Co.

DISCLAIMER

This document was prepared as an account of work sponsored by the United States Government. While this document is believed to contain correct information, neither the United States Government nor any agency thereof, nor the Regents of the University of California, nor any of their employees, makes any warranty, express or implied, or assumes any legal responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by its trade name, trademark, manufacturer, or otherwise, does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof, or the Regents of the University of California. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof or the Regents of the University of California.