

UC Berkeley

UC Berkeley Previously Published Works

Title

Anaerobic 5-Hydroxybenzimidazole Formation from Aminoimidazole Ribotide: An Unanticipated Intersection of Thiamin and Vitamin B12 Biosynthesis

Permalink

<https://escholarship.org/uc/item/1q62f3cw>

Journal

Journal of the American Chemical Society, 137(33)

ISSN

0002-7863

Authors

Mehta, Angad P
Abdelwahed, Sameh H
Fenwick, Michael K
[et al.](#)

Publication Date

2015-08-26

DOI

10.1021/jacs.5b03576

Peer reviewed



Published in final edited form as:

J Am Chem Soc. 2015 August 26; 137(33): 10444–10447. doi:10.1021/jacs.5b03576.

Anaerobic 5-Hydroxybenzimidazole formation from aminoimidazole ribotide: an unanticipated intersection of thiamin and vitamin B₁₂ biosynthesis

Angad P. Mehta¹, Sameh H. Abdelwahed^{1,4}, Michael K. Fenwick², Amrita B. Hazra³, Michiko E. Taga³, Yang Zhang², Steven E. Ealick², and Tadhg P. Begley^{1,*}

¹Department of Chemistry, Texas A&M University, College Station, TX-77843, USA

²Department of Chemistry and Chemical Biology, Cornell University, 120 Baker Lab, Ithaca, New York 14853, USA

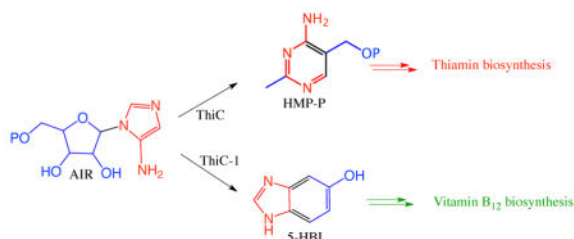
³Department of Plant and Microbial Biology, University of California, Berkeley, USA

⁴therapeutic chemistry department, National research center, Dokki, Cario, Egypt

Abstract

Comparative genomics of the bacterial thiamin pyrimidine synthase (*thiC*) revealed a paralog of *thiC* (*bzaF*) clustered with anaerobic vitamin B₁₂ biosynthetic genes. Here we demonstrate that BzaF is a radical S-adenosylmethionine enzyme that catalyzes the remarkable conversion of aminoimidazole ribotide (AIR) to 5-hydroxybenzimidazole (5-HBI). We identify the origin of key product atoms and propose a reaction mechanism. These studies represent the first step in solving a long-standing problem in anaerobic vitamin B₁₂ assembly and reveal an unanticipated intersection of thiamin and vitamin B₁₂ biosynthesis.

Graphical Abstract



The formation of dimethylbenzimidazole (DMB, 3), the axial ligand of vitamin B₁₂ (cobalamin), 1, (Figure 1A) is the last remaining major unsolved puzzle in B₁₂ biosynthesis. In aerobic organisms, DMB is biosynthesized from flavin mononucleotide (FMN, 2) in a remarkable reaction catalyzed by the BluB gene product (Figure 1B). This oxygen requiring

*Corresponding Author: Prof. Tadhg P. Begley, Department of Chemistry, Texas A&M University, College Station, TX-77843, USA. begley@chem.tamu.edu.

ASSOCIATED CONTENT

Supporting Information Available: Detailed experimental procedures are provided in the supporting information. This material is available free of charge via the Internet at <http://pubs.acs.org>

DMB synthase has been biochemically and structurally characterized, but its mechanism is not yet known.^{1,2} The anaerobic pathway to DMB involves different chemistry and labeling studies demonstrate that FMN is not the precursor (Figure 1C).^{3–6} The anaerobic DMB synthase has not been yet identified.

Our approach to elucidating the anaerobic biosynthesis of DMB was indirect and derived from our studies on the bacterial thiamin pyrimidine synthase (ThiC). This radical SAM enzyme catalyzes a remarkable rearrangement of aminoimidazole ribotide (AIR, 10) to the thiamin pyrimidine (12, Figure 2).^{7–10}

While looking for a ThiC paralog, to facilitate our mechanistic studies on this complex reaction, we noticed that some anaerobes in the SEED database (<http://theseed.uchicago.edu/FIG/>) had a *thiC* paralog (*bzaF*) clustered with vitamin B₁₂ biosynthetic genes (Figure 3). This suggested the possibility that BzaF might catalyze the conversion of AIR to a DMB precursor. Previous *in vivo* labeling studies, using 4-aminoimidazole, eliminated this heterocycle as a DMB precursor.¹² In this communication we report the successful reconstitution of the BzaF-catalyzed conversion of AIR to 5-hydroxybenzimidazole (HBI), we identify the hydrogen atom initially abstracted by the 5'-deoxyadenosyl radical and the origin, in AIR, of key HBI atoms. These results are integrated into a mechanistic hypothesis for the BzaF (HBI synthase) catalyzed reaction.

The *bzaF* gene from *Desulfuromonas. acetoxidans* was synthesized with codon optimization (for gene details - Figure S33), cloned in the THT vector and co-expressed in the presence of a plasmid encoding the *suf* operon for *in vivo* [4Fe-4S] reconstitution in *E.coli* BL21 (DE3) (See supporting information - page 13). The enzyme was then purified under anaerobic conditions using Ni-NTA chromatography (see SI for experimental details). The enzyme yield was 3 mg/L of culture. BzaF has a molecular mass of 48 kDa, and contains 3 irons and 2.6 sulfides/monomer. The UV-Visible spectrum of the purified protein showed the long wavelength absorption characteristic of the [4Fe-4S]⁺² cluster (Figure 4B).

Anaerobic incubation of AIR and SAM with dithionite-reduced BzaF (RT, 60 min) resulted in the formation of 5'-deoxyadenosine (5'-dA) and a new product as determined by HPLC analysis (Figure 4A). This product had an [M+H]⁺ of 135.05 Da and co-migrated with an authentic standard of HBI (Figure 4C and D). This identification was confirmed by scale-up of the reaction followed by NMR analysis of the purified product. The ratios of HBI and 5'-dA to BzaF were 0.7 (HBI:BzaF) and 1.1 (5'-dA:BzaF) respectively. The corresponding ratios for ThiC are 0.4 (HMP-P: *C.c.*ThiC) and 0.6 (5'-dA: *C.c.*ThiC).

As a first step in elucidating the mechanism of HBI synthase, we have determined the origin of key atoms of HBI. Our strategy involved the conversion of a set of ²H- and ¹³C-AIR isotopologs to HBI followed by the identification of the site of labeling by NMR and MS analysis. The results of these experiments are shown in Figure 5 and in Table S1 (See also Figures S20–S29). The key findings, summarized in Figure 5A are as follows: 1) The proS hydrogen at C5 of AIR is abstracted by the adenosyl radical (Table S1, lines 2–4). 2) The C7 hydrogen of HBI is derived from the C5' proR hydrogen of AIR (Table S1, line 3 and Figure 5F). 3) The C6 hydrogen of HBI is derived from the C4' hydrogen of AIR (Table S1, line 5

and Figure 5E). 4) The C1' carbon of AIR is lost as formate (Table S1, lines 6–9 and Figure 5GHI). 5) The C5 amino group nitrogen of AIR is partially retained in the product (Table S1, line 10, 62% retention, see legend to Table S1). 6) The C4 carbon of HBI is derived from the C2' carbon of AIR (Figure 5D). From these experiments, we can deduce the origin of all of the atoms of HBI as shown in Figure 6.

These labeling studies allow us to propose a mechanistic hypothesis for the HBI-synthase-catalyzed reaction (Figure 7A). As expected from the sequence similarity with ThiC, HBI formation is initiated by the abstraction of the C5' proS hydrogen of AIR by the 5'-deoxyadenosyl radical to give 15. C-O bond cleavage followed by N-glycosyl bond cleavage gives 17. At this point, the chemistry diverges from the proposed ThiC pathway (See Figure S32). Electrophilic addition of the ribose C4' to the aminoimidazole leads to the thiamin pyrimidine while electrophilic addition of the ribose C5' gives 19 which leads to DHBI. Loss of phosphate followed by deprotonation gives 21, which is quenched by hydrogen atom donation from an unknown source. A ring opening/closing sequence gives 24. Loss of ammonia from 24 gives imine 25 in which 38% of the imine nitrogen is derived from N1 of AIR and the remainder is derived from the C5 amino group. This is consistent with the partial scrambling of the two amidine nitrogens in 23 by C-C bond rotation. Loss of formate, ring closure and a series of dehydrations and tautomerizations complete the formation of HBI 14.

A homology model for HBI synthase was constructed using the structure of the *Arabidopsis thaliana* ThiC/AIR/SAH/Fe complex as a template.¹³ HBI synthase lacks the chloroplast-targeting sequence specific to the *Arabidopsis* ThiC as well as the N-terminal ThiC domain.^{7,13,14} This N-terminal domain is also missing from ThiCs from anaerobic organisms. HBI synthase is predicted to have the noncanonical active site architecture recently reported for ThiC with an additional iron chelating two conserved histidine residues and the amino and carboxylate groups of SAM.¹³ The homology model was consistent with the observed abstraction of the 5'-ProS hydrogen atom of the substrate and showed remarkable conservation of active site residues with only four differences out of 21 residues in the first shell when compared to ThiC (Figure 7B – 7C). Three of the differences involve contacts with AIR. In ThiC, Asn228 hydrogen bonds to AIR O3' and is replaced by Ser66 in HBI synthase. Thr320 hydrogen bonds to Glu422, which in turn hydrogen bonds to AIR O2', and is replaced by Ala161 in HBI synthase; Glu244 is equivalent to Glu422 and still hydrogen bonds to AIR O2'. Asp383 (if protonated) hydrogen bonds to AIR N3 (2.8 Å) and is replaced by Asn227 in HBI synthase. The fourth difference is near SAM, where Arg343 hydrogen bonds to N7 of SAM and is replaced by Lys187, which is also predicted to hydrogen bond to N7, in HB synthase. BLAST searches¹⁵ starting with either *D. acetoxidans* HBI synthase or *A. thaliana* ThiC suggest that these four differences alone distinguish the two paralogs.

Conserved active sites suggest similar catalytic mechanisms for most enzymatic reactions. This is clearly not the case for ThiC and BzaF. These proteins reveal unique features of enzymatic catalysis of reactions that proceed via high-energy intermediates. For such reactions, it is likely that the high-energy substrate radical 15 is converted to HMP-P 12 or to HBI 14 in a complex multistep sequence in which several steps along the reaction

coordinate do not require catalytic assistance from the enzyme. This allows for similar active sites to produce different products. This catalytic motif has previously been observed in terpene cyclases where similar active sites can catalyze the conversion of high-energy carbocations to different products. For example, the D100E mutant of trichodiene synthase generates six terpenes¹⁶ and native baruol synthase produces a library of 23 terpenes.¹⁷

Our studies on the identification and reconstitution of the vitamin B₁₂ HBI synthase represent the first steps in solving a long-standing problem in anaerobic vitamin B₁₂ biosynthesis. The biosynthesis involves an unanticipated intersection of thiamin and vitamin B₁₂ biosynthesis and uses chemistry that exceeds in complexity any of the enzymatic reactions used in the assembly of the coronoids of vitamin B₁₂. Mechanistic and structural studies are in progress to elucidate the details of this remarkable reaction.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

This research was supported by the Robert A. Welch Foundation (A-0034 to TPB) and the National Institutes of Health (DK44083 to TPB; DK067081 to SEE).

References

1. Taga ME, Larsen NA, Howard-Jones AR, Walsh CT, Walker GC. *Nature (London, U K)*. 2007; 446:449. [PubMed: 17377583]
2. Cooper AJL, Pinto JT. *Chemtracts*. 2006; 19:474.
3. Scott AI. *J Org Chem*. 2003; 68:2529. [PubMed: 12662021]
4. Munder M, Vogt JRA, Vogler B, Renz P. *Eur J Biochem*. 1992; 204:679. [PubMed: 1531794]
5. Vogt JRA, Lamm-Kolonko L, Renz P. *Eur J Biochem*. 1988; 174:637. [PubMed: 2968906]
6. Vogt JRA, Renz P. *Eur J Biochem*. 1988; 171:655. [PubMed: 3345751]
7. Chatterjee A, Li Y, Zhang Y, Grove TL, Lee M, Krebs C, Booker SJ, Begley TP, Ealick SE. *Nat Chem Biol*. 2008; 4:758. [PubMed: 18953358]
8. Chatterjee A, Hazra AB, Abdelwahed S, Hilmey DG, Begley TP. *Angew Chem, Int Ed*. 2010; 49:8653.
9. Lawhorn BG, Mehl RA, Begley TP. *Org Biomol Chem*. 2004; 2:2538. [PubMed: 15326535]
10. Yamada K, Kumaoka H. *J Nutr Sci Vitaminol (Tokyo)*. 1983; 29:389. [PubMed: 6417286]
11. Renz P, Endres B, Kurz B, Marquardt J. *Eur J Biochem*. 1993; 217:1117. [PubMed: 8223635]
12. Endres B, Weurfel A, Volgler B, Renz P. *Biol Chem Hoppe-Seyler*. 1995; 376:595. [PubMed: 8590628]
13. Fenwick MK, Mehta AP, Zhang Y, Abdelwahed SH, Begley TP, Ealick SE. *Nature Commun*. 2015; 6:6480. [PubMed: 25813242]
14. Coquille S, Roux C, Mehta A, Begley TP, Fitzpatrick TB, Thore S. *J Struct Biol*. 184:438. [PubMed: 24161603]
15. Altschul SF, Gish W, Miller W, Myers EW, Lipman DJ. *J Mol Biol*. 1990; 215:403. [PubMed: 2231712]
16. Cane DE, Xue Q, Fitzsimons BC. *Biochemistry*. 1996; 35:12369. [PubMed: 8823172]
17. Lodeiro S, Xiong Q, Wilson WK, Kolesnikova MD, Onak CS, Matsuda SPT. *J Am Chem Soc*. 2007; 129:1121.

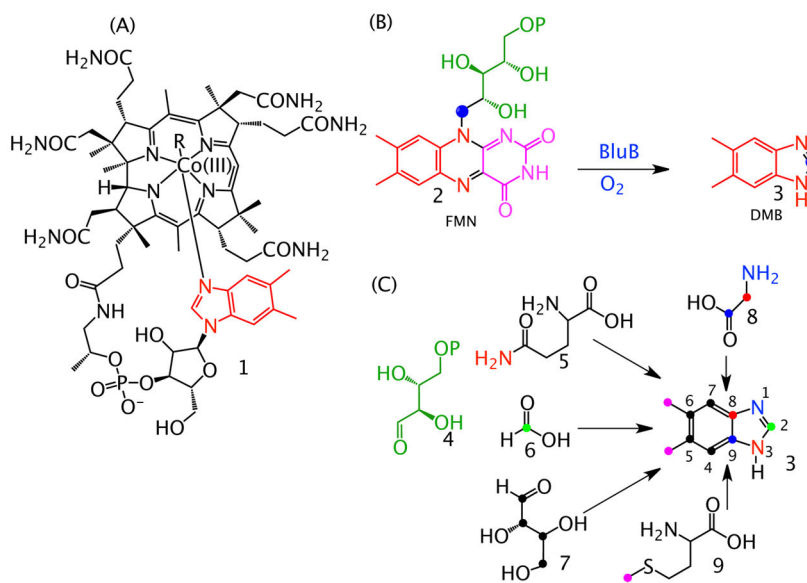


Figure 1. The oxygen-dependent and oxygen-independent biosynthesis of the DMB ligand of vitamin B₁₂. (A) The structure of vitamin B₁₂ 1 showing the DMB ligand in red. (B) Oxygen-dependent DMB formation from FMN 2. (C) Labeling studies showing the oxygen-independent biosynthetic origin of the atoms of DMB in *E. limosum*. 5-Hydroxybenzimidazole (HBI) and 5-hydroxy-6-methylbenzimidazole have been identified as precursors.¹¹

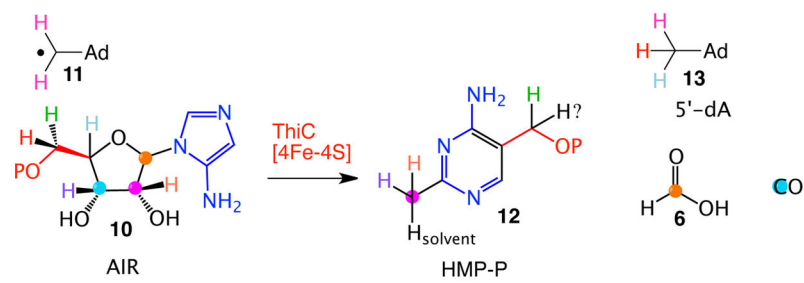


Figure 2.
The remarkable, ThiC-catalyzed, conversion of AIR 10 to the thiamin pyrimidine 12.

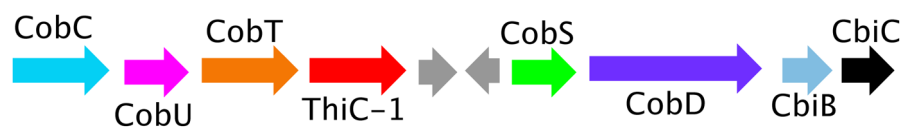


Figure 3. BzaF gene cluster in *D. acetoxidans*. CobCUT, CobSD and CbiBC have established functions in vitamin B₁₂ biosynthesis. BzaF has 38% sequence identity and 53.6% sequence similarity to the *D. acetoxidans* ThiC.

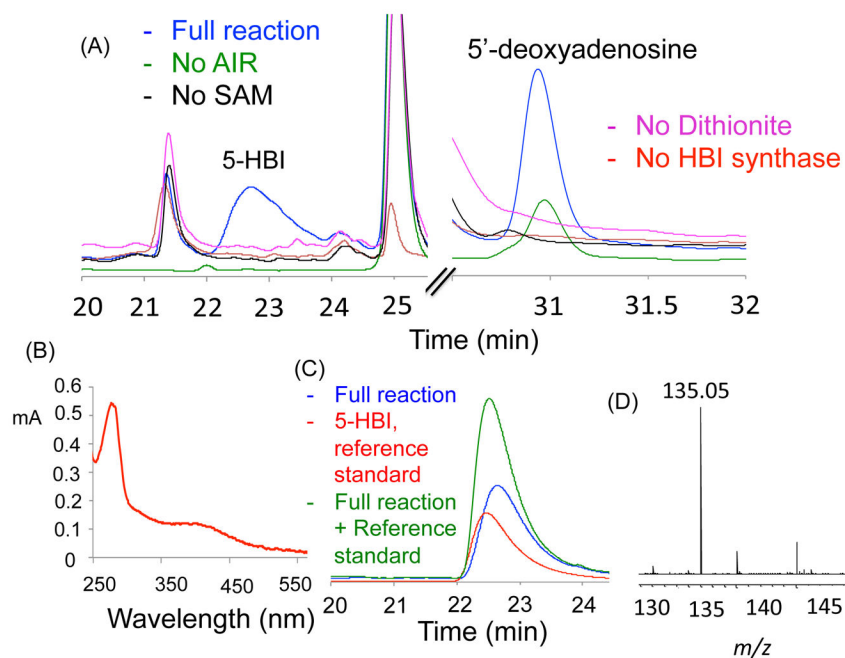


Figure 4. Characterization of the product of the BzaF catalyzed reaction. (A) HPLC (254 nm) analysis of the HBI synthase reaction mixture showing a new product eluting at 22.7 min and 5'-dA eluting at 31 min (blue trace). This product was not formed in reaction mixtures lacking dithionite (red trace). (See Figure S31 for all controls). (B) UV-Vis spectrum of HBI synthase as isolated. (C) Product identification by HPLC analysis (red trace – HBI, blue trace – full enzymatic reaction, green trace – co-injection of HBI and the reaction product). (D) MS analysis of the 22.7 min enzymatic product showing $[M+H]^+ = 135.05$ Da, consistent with HBI formation.

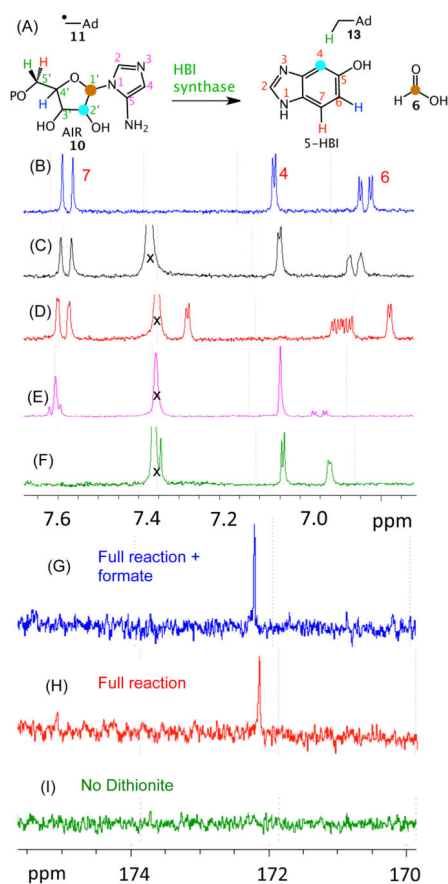


Figure 5. Labeling studies to determine the origin of key HBI atoms. (A) Summary of the labeling studies. (B) Partial ^1H NMR for authentic HBI. (C)–(F): ^1H NMR of HBI purified from enzymatic reactions using (C) unlabeled AIR (D) $2'$ - ^{13}C -AIR (E) 4 - ^2H -AIR and (F) $5'$, $5'$ - $^2\text{H}_2$ -AIR. (G) – (I): ^{13}C DEPT 90 analysis of the reaction mixture. (G) HBI synthase reaction with $1'$, $2'$, $3'$, $4'$, $5'$ - $^{13}\text{C}_5$ -AIR spiked with formate (10 mM) after 2hrs. (H) HBI synthase reaction with $1'$, $2'$, $3'$, $4'$, $5'$ - $^{13}\text{C}_5$ -AIR. (I) No Dithionite – control. X is an unknown impurity. The doublet at 7.6 and the doublet of doublets at 7.05 in spectrum E are due to a trace of unlabeled AIR in the sample (See Figure S30 for ^{13}C -NMR).

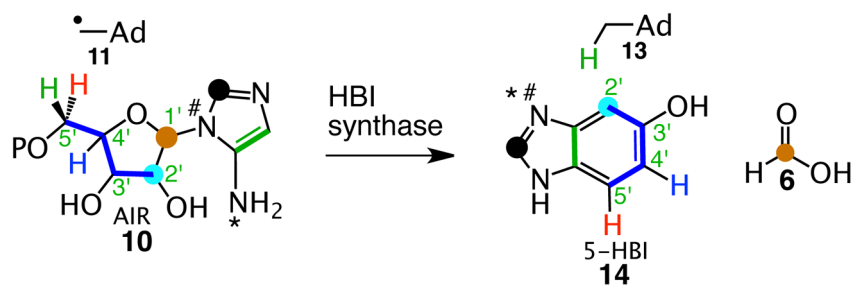


Figure 6. Origin of all of the atoms of HBI based on the labeling studies shown in Figure 5 and in Table S1.

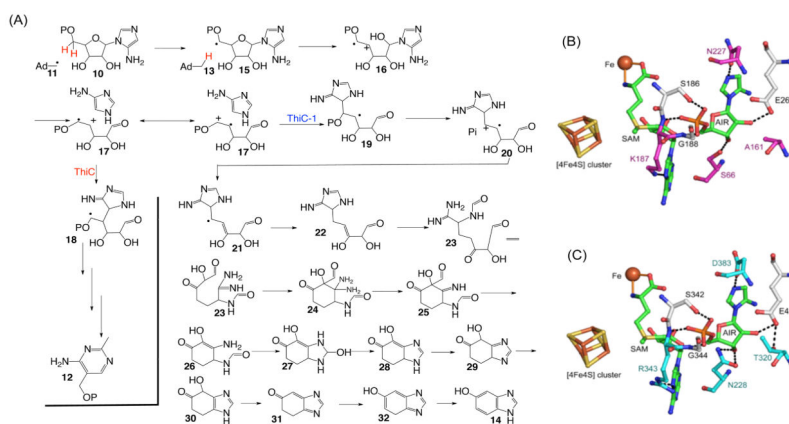


Figure 7.

(A) Mechanistic proposal for the radical-triggered conversion of AIR to HBI catalyzed by HBI synthase. (B) Homology model for HBI synthase. (C) X-ray structure of the ThiC active site. The [4Fe-4S] cluster, additional iron, SAM, AIR, and key active site residues are shown. SAM was generated by adding a methyl group in the *S*-configuration to SAH. The additional iron is also chelated to two conserved side chains (His270 and His334) in HBI synthase and His426 and His490 in ThiC.