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Optimization of pyrazole-containing 1,2,4-triazolo-[3,4b]thiadiazines, a new class of STAT3 pathway inhibitors

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Abstract

Structure-activity relationship studies of a 1,2,4-triazolo-[3,4-*b*]thiadiazine scaffold, identified in an HTS campaign for selective STAT3 pathway inhibitors, determined that a pyrazole group and specific aryl substitution on the thiadiazine were necessary for activity. Improvements in potency and metabolic stability were accomplished by the introduction of an α -methyl group on the thiadiazine. Optimized compounds exhibited anti-proliferative activity, reduction of phosphorylated STAT3 levels and effects on STAT3 target genes. These compounds represent a starting point for further drug discovery efforts targeting the STAT3 pathway.

Graphical Abstract

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Keywords

STAT3 inhibitor; Triazolo-thiadiazines; Anti-cancer agents; STAT1

While advancements in detection and treatment have aided in the longstanding campaign against cancer, new annual U.S. cancer cases and related deaths still exceed 1.5 million and 580,000, respectively.¹ Recently, new therapies have targeted signaling pathways that are aberrantly activated in transformed cells. These pathways promote cancer cell survival and proliferation, but they play a less important role in normal cell survival.² As a part of our interest in the development of mechanism-based anticancer agents,^{3,4} we have been pursuing novel small molecule inhibitors of the signal transducer and activation of transcription 3 (STAT3) pathway.^{5,6,7} STAT3 is a transcription factor that influences many of the acquired capabilities of cancer tumorigenesis, thereby making it an attractive target for the development of oncolytics.^{8–12}

There is mounting evidence for its role in cancer such as increased levels of activated STAT3 (pSTAT3-Y705) observed in many cancers including head and neck squamous cell carcinomas (HNSCC).^{4,5} We previously reported the use of a high content phenotypic screen to identify selective inhibitors of the STAT3 activation pathway compared to STAT1 which served as an important selectivity control since the latter is a tumor-suppressive transcription factor.^{6,7} We identified several scaffolds that met this criteria. Herein, we describe the optimization and structure activity relationship for a series of pyrazole-containing 1,2,4-triazolo-[3,4-*b*]thiadiazines with selective STAT3 pathway inhibition.

The high content phenotypic screen, which utilized an interleukin-6 (IL-6)-induced STAT3 activation assay in Cal33 head and neck tumor cells, identified several triazolo-thiadiazines as selective STAT3 pathway inhibitors (e.g., **1a** and **2b**, Table 1).⁷ The biological activities of these structurally similar analogs were confirmed through resynthesis and re-assay (*vide infra*). These HTS/HCS hits had no effect on interferon- γ (IFN- γ)-induced STAT1 pathway activation at concentrations up to 50 μ M. This selectivity was not observed for many other STAT3 pathway inhibitors reported in the literature including the pan-Janus kinase (JAK) inhibitor, pyridone 6.⁶ Furthermore, this series exhibited acceptable drug-like properties: low molecular weight (<400), clogP values between 3 and 4,¹³ and anti-proliferative activities with several HNSCC cell lines (GI₅₀ 14–45 μ M with 686LN, Cal33, FaDu, and OSC19).

Triazolothiadiazines exhibit an array of pharmacological effects including anti-proliferative activities.¹⁴ However, the HTS library included a number of inactive analogs of **1a** and **2b** where the pyrazole was replaced with an alkyl, aryl, or alternative heterocyclic substituents (Figure 1). This suggested that we were not observing broadly promiscuous effects with this scaffold. These observations together with the desirable biological selectivity profile and

favorable drug-like physical properties encouraged us to pursue a medicinal chemistry optimization effort for this series.

The synthesis of **1a** and **2b** required key amino-triazole intermediates **3a** and **3b** that were readily assembled according to literature procedures.^{15–19} Alkylations with the appropriate α -halo ketone and microwave-assisted cyclodehydrations afforded the original hits **1a** and **2b** (Scheme 1).

In addition to **1a** and **2b**, two sub-libraries of triazolothiadiazines that maintained either the fused cyclopentyl-pyrazole group ("**a**") or the pendant phenyl-pyrazole group ("**b**") with diversified R-group modifications on the triazolothiadiazine were synthesized. These compounds were prepared according to Scheme 1 with yields ranging from 24–90% by using a variety of α -halo ketones in the final microwave-assisted cyclodehydration reaction. A protected aldehyde derivative was used to prepare **9a** and **9b**.

We selected substituents with diverse steric, polarity, and electronic characteristics. Table 2 illustrates a representative subset of these analogs and their activities in STAT3 and STAT1 assays. The R-substitution on the thiadiazine influenced activities by a factor of greater than 10 with many modifications leading to a loss of STAT3 potency. The importance of the chlorine substitution on the arene groups was evident by the significant drop in activity seen with the removal of the halogen regardless of the pyrazole scaffold (**5a** vs. **1a**; **5b** vs. **2b**). As a general trend, hydrogen (**9**), aliphatic (**10**, **11**), and heterocyclic (**7**) R-groups were inactive regardless of the pyrazole substructure. One notable exception was the chlorothiophene analog **8a** that maintained comparable potency to the initial hit. However, when the chlorothiophene was combined with the phenyl-pyrazole ("**b**") scaffold, the loss of activity of analog **8b** was consistent with other heterocyclic derivatives.

A few SAR trends diverged between the two pyrazole scaffolds. For example, exchanging the R-substituents of **1a** and **2b** provided analogs **1b** and **2a** that were 3–4 times less active. The reduced STAT3 potency of compounds **2a** (26μ M) and **4a** (21μ M) established the importance of the *ortho-* and *para*-chlorine atoms on the phenyl ring in **1a**. Interestingly, the phenyl-pyrazole with the same *meta*-chlorophenyl R-substituent (**4b**) retained, or perhaps, improved the STAT3 potency compared to the *para*-chlorophenyl **2a**, suggesting that either lipophilicity or electronic effects were important. However, the potent activity of the *para*-methoxyphenyl analog **6b** did not support a purely electronic contribution. The selective inhibition of STAT3 over STAT1 activation was maintained within this entire subseries.

To evaluate the effect of thiadiazine modifications, hydrazone **6a** was converted to dihydrothiadiazines **12** and **13** (Scheme 2). Both analogs were inactive in the STAT3 assay, which established the requirement for an unsaturated ring system.

The modified pyrazole analogs shown in Figure 2 were prepared according to Scheme 1.¹⁵ The *para*-chloro and *para*-fluoro substitutions on the 3-arylpyrazoles were well tolerated. The *para*-fluoro analog **14** exhibited an IC₅₀ of 7.5 μ M, which was a ~4-fold improvement in STAT3 potency over the unsubstituted analog **1b**.²⁰ In the fused pyrazole "**a**" series, the cyclohexyl homolog **16** was ~2-fold less active than the cyclopentyl analog **1a**.

Despite the favorable biological profiles for some of these compounds, they exhibited poor metabolic stability [**1a**: $t_{1/2}$ =14 min in human liver microsomes (HLM); 4 min in mouse liver microsomes (MLM)]. Predictions of metabolic sites using SMARTCyp²¹ pointed toward the thioether as the most susceptible site of a cytochrome P450-mediated oxidation (see Supporting Information). Subsequent analysis determined that oxidized thioethers were inactive (*vide infra*).

Analysis of the HTS SAR data suggested that carbon and oxygen heterocyclic analogs of the triazolothiadiazines would be inactive despite their potential to be more metabolically stable. Therefore, in an effort to address this possible metabolic liability, we designed, prepared and tested a series of analogs that were modified at the position alpha to the sulfur atom with the aim of sterically blocking oxidation of the thioether (Scheme 3 and Table 3). These compounds were prepared by microwave-mediated cyclodehydration with the corresponding α -halo ketones and amino-triazoles. All of the final products maintained desirable physical properties such as MW and clogP (Table 3 & Supporting Information). The structure of **17** (UPCDC10263) was confirmed by X-ray analysis (Figure 3).

It was apparent from this series that alkyl substitution alpha to the sulfur atom consistently led to potent inhibition of STAT3 activation while maintaining high selectivity over STAT1. The size of the alkyl groups did not seem to influence potency since large groups, such as *i*-Bu (**25** and **26**) and *i*-Pr (**23** and **24**), were as potent or perhaps slightly more potent than the methyl analogs (**17–22**). The benzyl analog **27** was less active, but only one example was prepared and we cannot speculate whether this was a general trend. In contrast, alkyl groups with polar substituents (**28** and **29**) and *gem*-dimethyl substitution (**30**) were either significantly less potent or inactive in the STAT3 assay. Compounds **28** and **29** exhibited potent anti-proliferative activity in the absence of STAT3 inhibition that is presumed to be due to off-target effects. We prepared the sulfoxide analog of compound **21** (not shown) as the putative oxidative metabolite; however, the resultant compound was inactive.

In some cases, combining the α -methyl substitution with the 3-arylpyrazole scaffold (19, 20, and 21) yielded a 2–3 fold improvement in STAT3 potency compared to the original hit 2b (Table 3). Chiral separation (Supporting Information) provided individual enantiomers of 22 (ent-1 and ent-2), and a 3-fold difference in STAT3 potency was observed between these enantiomers. While these differences were not dramatic, the trend suggested the existence of a eutomer [22 (ent-1), configuration not assigned] and a distomer within this series. Interestingly, the eudismic ratio was even larger when considering the anti-proliferative activity, which may reflect additional differences in metabolism or transport between the enantiomers (Table 3). Importantly, effects of α -methyl substitutions on microsomal stability were also evident. By incorporating this substitution, we were indeed able to increase the metabolic stability of compound **19** in HLM and MLM ($t_{1/2} = 27$ min and 10 min, respectively), thereby doubling the stability compared to our initial hits. Some analogs containing alkyl groups larger than methyl (e.g. 23-26) exhibited low micromolar potency but suffered from reduced compound solubility (data not shown) which excluded them from further characterization. Anti-proliferative activity in Cal33 cells was also determined. In most cases, compounds that inhibited STAT3 also inhibited the growth of HNSCC cell lines (Table 4). However, direct correlations between the two activities are likely complicated by

differences between assay conditions, distinct genetic alterations among the HNSCC cell lines, and/or differences between a compound's metabolic stability, residence time, and transport.²²

With these encouraging cell-based results in well-characterized HNSCC models, we evaluated the effects of compounds **19** (UPCDC10131) and **22** (UPCDC10205) on STAT3 expression levels by Western blot analysis (Figures 4 and 5). Consistent with the pSTAT3 HCS assay, pronounced decreases in pSTAT3 levels were observed, and both compounds exhibited significant inhibition of the downstream STAT3 target gene, Cyclin D1.²³

In summary, we identified pyrazole-linked 1,2,4-triazolo-[3,4-b]thiadiazines as a new class of potent and selective STAT3 pathway inhibitors. Optimized compounds exhibited antiproliferative activity, reduction of phosphorylated STAT3 levels, and reduction of downstream effects on STAT3 gene expression targets. Structure-activity relationships established that a pyrazole group and specific aryl substitution on the thiadiazine were required for activity. Significant improvements in potency and metabolic stability were realized through α -substitution of the thiadiazine heterocycle. Further studies of the mechanism of action of these compounds will be reported in due course.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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 $R^1 = CI \text{ or } OMe$



R =





14: R = F; STAT3 IC₅₀ = 7.5 \pm 2.6 μ M **15:** R = CI; STAT3 IC₅₀ = 14.5 μ M*

Figure 2. Modified pyrazole analogs (*mean IC₅₀, n=2).



16: STAT3 IC $_{50}$ = 15.5 \pm 4.7 μM







Figure 4.

Western blot analysis of **19** (UPCDC10131) in interleukin 6 (IL-6, 50 ng/mL)-stimulated Cal33 cells.



Figure 5.

Western blot analysis of **22** (UPCDC10205) in interleukin 6 (IL-6, 50 ng/mL)-stimulated Cal33 cells.

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Scheme 1. Preparation of 1,2,4-triazolo-[3,4-*b*]thiadiazines.



Scheme 2. Synthesis of dihydrothiadiazine analogs.





Table 1

Triazolothiadiazines 1a and 2b and Summary of Biological Activities and Physicochemical Properties.

	CI	
Cmpd #	1a	2b
STAT3 IC50 (µM)	6.8 ± 3.7	9.6 ± 7.8
STAT1 IC ₅₀ (µM)	> 50	> 50
GI ₅₀ HNSCC ^a (µM)	14–26	27–44
MW	391.2	392.8
cLogP	3.2	3.7
HBD/HBA	1/4	1/4
tPSA	71.7	71.7
LE	0.29	0.26

^aCell lines: 686LN, FaDu, Cal33, OSC19

Table 2

STAT3 and STAT1 activities of triazolothiadiazines. (**a**: cyclopentyl-pyrazole series; **b**: 3-phenyl-pyrazole series)

R	Cmpd #	STAT3 IC ₅₀ (μM)	STAT1 IC ₅₀ (µM)
CI ~~*	1a	6.8 ± 3.7	>50
	1b	31.2 ± 20.0	>50
CI*	2a	26.8 ± 22.3	>50
	2b	9.6 ± 7.8	>50
CI	4a	21.2 ± 5.9	>50
~~*	4b	5.6 ± 4.0	45.9 ± 8.3
~~*	5a	21.2 ± 23.5	>50
	5b	27.0 ± 17.4	> 50
MeO —	6a	17.9 ± 22.9	40.6 ± 7.5
	6b	3.7 ± 2.5	> 50
N*	7a	> 50	> 50
	7b	> 50	> 50
~ *	8a	11.0 ± 5.5	> 50
CI S	8b	> 50	> 50
H -*	9a	> 50	> 50
	9b	> 50	> 50
Me _*	10a	> 50	> 50
1vic -	10b	> 50	> 50
~~*	11a	> 30	> 50
	11b	> 50	> 50

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Table 3

STAT3 and STAT1 activities of α -substituted thioether analogs.

Cmpd #	Pyrazole series	Ar	R/R'	$\begin{array}{c} STAT3\\ IC_{50}(\mu M) \end{array}$	STAT1 IC ₅₀ (µM)	Cal33 GI ₅₀ (μM)
17	a	3,4- dichloro-Ph	Me/H	8.2 ± 2.2	>50	1.9 ± 1.2
18	а	4-OMe-Ph	Me/H	9.1^{*}	>50	0.2 *
19	q	Чd	H/əM	3.4 *	15*	$\begin{array}{c} 34.1 \pm \\ 8.7 \end{array}$
20	c	Чd	H/əM	3.8 ± 0.8	>30	36.3 ± 9.7
21	v	4-OMe-Ph	Me/H	4.2 *	>30*	$\begin{array}{c} 29.5 \pm \\ 6.5 \end{array}$
22	q	4-OMe-Ph	Me/H	$\begin{array}{c} 11.4 \pm \\ 10.4 \end{array}$	29.5*	$\begin{array}{c} 2.2 \pm \ 0.3 \end{array}$
22 (ent-1)	p	4-OMe-Ph	H/əM	5.72*	NT	2.7 *
22 (ent-2)	p	4-OMe-Ph	H/əM	17.3*	NT	21.8^{*}
23	a	3,4- dichloro-Ph	H/1d <i>-!</i>	4.4 *	>50	8.2 *
24	a	4-OMe-Ph	<i>i-</i> Pr/H	2.2 ± 1.2	>50	$\begin{array}{c} 0.4 \pm \ 0.1 \end{array}$
25	a	3,4- dichloro-Ph	H∕nB∔	4.1^{*}	>50	3.5 ± 1.5
26	q	4-OMe-Ph	H∕nB∔	5.1*	>50	0.5^{*}
27	a	3,4- dichloro-Ph	Bn/H	7.5*	>50	3.3 *
28	q	4-OMe-Ph	$(CH_2)_2 NH_2 / H$	>20	NT	2.5*
29	p	4-OMe-Ph	(CH ₂) ₂ OMe/H	>50	NT	0.04
30	a	3,4- dichloro-Ph	Me/Me	18.6^{*}	>50	>50*
* mean IC5(), n=2.				-	

Table 4

Growth inhibition on HNSCC cell lines

	18 UPCDC10262	19 UPCDC10131	22 UPCDC10205
686LN GI ₅₀ (µM)	0.88*	32.2*	17.4*
Cal33 GI ₅₀ (µM)	0.18*	34.1*	2.2*
FaDu GI ₅₀ (µM)	1.3 *	32.7*	8.4*
OSC19 GI ₅₀ (µM)	3.1*	31.3*	14.9*

* mean, n=2