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1 **Concurrent measurement of O₂ production and isoprene emission during photosynthesis:**
2 **pros, cons, and metabolic implications of responses to light, CO₂ and temperature**

3

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18

19 **Short running head:** O₂ and isoprene fluxes during leaf photosynthesis

20 **Keywords:** Photosynthesis, oxygen production, ¹⁸O-water labelling, isoprene

21

22 **Abstract**

23 Traditional leaf gas-exchange experiments have focused on net CO₂ exchange (A_{net}). Here, using
24 California poplar (*Populus trichocarpa*), we coupled measurements of net oxygen production (NOP),
25 isoprene emissions and $\delta^{18}\text{O}$ in O₂ to CO₂/H₂O gas exchange with chlorophyll fluorescence, and measured
26 light, CO₂ and temperature response curves. This allowed us to obtain a comprehensive picture of the
27 photosynthetic redox budget including electron transport (ETR) and estimates of the mean assimilatory
28 quotient ($\text{AQ} = A_{\text{net}}/\text{NOP}$). We found that A_{net} and NOP were linearly correlated across environmental
29 gradients with similar observed AQ values during light (1.25 ± 0.05) and CO₂ responses (1.23 ± 0.07). In
30 contrast, AQ was suppressed during leaf temperature responses in the light (0.87 ± 0.28), potentially due
31 to the acceleration of alternative ETR sinks like lipid synthesis. A_{net} and NOP had an optimum temperature
32 (T_{opt}) of 31 °C, while ETR and $\delta^{18}\text{O}$ in O₂ (35 °C) and isoprene emissions (39 °C) had distinctly higher
33 T_{opt} . The results confirm a tight connection between water oxidation and ETR and support a view of light-
34 dependent lipid synthesis primarily driven by photosynthetic ATP/NADPH not consumed by the Calvin-
35 Benson cycle, as an important thermotolerance mechanism linked with high rates of (photo)respiration
36 and CO₂/O₂ recycling.

37 **Keywords:** Photosynthesis, net oxygen production, gross oxygen production, H₂¹⁸O labeling

38 **Summary statement:** Application of a leaf gas-exchange system with net oxygen production and
39 isoprene emission suggests a thermotolerance role of enhanced lipid synthesis and CO₂/O₂ recycling.

40 Introduction

41

42 Terrestrial ecosystems cycle large amounts of carbon dioxide (CO₂) and oxygen (O₂) between the
43 biosphere and atmosphere via photosynthesis, photorespiration, and respiration. However, the majority of
44 gas-exchange observations of photosynthesis and (photo)respiration from individual leaves under
45 controlled environmental conditions have focused on biological and environmental variables impacting
46 net CO₂ assimilation (A_{net}) without the inclusion of gaseous products of photosynthesis such as O₂ and
47 isoprene emissions. The lack of leaf-atmosphere O₂ flux data is largely due to technical difficulty to
48 measure a small change in O₂ mole fraction (e.g. 2-200 ppm O₂) in a high atmospheric O₂ background
49 (21%, i.e., 210,000 ppm), that is, the high measurement precision needed to clearly resolve relatively
50 small atmospheric O₂ concentration changes in gas exchange systems (Kim-Hak, Hoffnagle, Lynch, &
51 Johnson, 2018). The experimental challenge has been partly solved by O₂ measurements under low
52 ambient O₂ concentrations (1-2%) (Laisk et al., 2002). However, low O₂ concentration itself has impacts
53 on leaf gas exchange, suppressing photorespiration and potentially 'mitochondrial' respiration (also
54 referred to as day respiration), but often also inducing feedback-limited photosynthesis (Rasulov, Talts,
55 Bichele, & Niinemets, 2018; Thomas D. Sharkey, 1990; Yang, Preiser, Li, Weise, & Sharkey, 2016).
56 Thus, under physiological conditions, dynamic leaf gas-exchange observations of both A_{net} and net O₂
57 production (NOP) as a function of environmental conditions remain rare across diverse plant functional
58 types and ecosystems, representing a major knowledge gap in terrestrial ecosystem carbon and oxygen
59 cycling. Early studies demonstrated the potential of mass spectrometry to quantify leaf gross production
60 of ¹⁶O₂ in the light simultaneously with gross ¹⁸O₂ uptake under a recirculated leaf headspace atmosphere
61 of 21% ¹⁸O₂ (Canvin, Berry, Badger, Fock, & Osmond, 1980). This technique was used with potted tomato (*Solanum*
62 *lycopersicum*) plants to demonstrate that during leaf water stress, gross oxygen production and
63 consumption declined together with gross CO₂ assimilation and production, suggesting that photosystem
64 II, the Calvin cycle, and mitochondrial respiration were down-regulated (Haupt-Herting & Fock, 2002).
65 Membrane inlet mass spectrometry (MIMS) and ¹⁸O₂ isotope analysis allow differentiation between
66 O₂ produced by photosystem II (PSII) and that consumed by a number of processes including
67 photorespiration, mitochondrial respiration, and the Mehler reaction during the water-water cycle
68 (Allahverdiyeva, Isojärvi, Zhang, & Aro, 2015). More recently, a new method based on measuring $\delta^{18}\text{O}$
69 of O₂ in air of a detached leaf equilibrated with H₂¹⁸O was used to estimate gross oxygen production
70 (GOP) and NOP determined separately from the increase in the O₂/N₂ ratio (Gauthier, Battle, Griffin, &
71 Bender, 2018).

72 While few observations have been reported, the interface of environmentally controlled open-path
73 leaf chambers to high precision real-time oxygen sensors has opened the door to concurrent measurement
74 of A_{net} and NOP, and thus the net assimilatory CO_2/O_2 quotient ($\text{AQ} = A_{\text{net}}/\text{NOP}$) (Cousins & Bloom,
75 2003; LI-COR, 2023). Custom differential O_2 gas analyzers have been developed with two zirconium
76 oxide cells with a precision of ± 2 ppm O_2 against a 21% O_2 background (Bloom, Smart, Nguyen, &
77 Searles, 2002) and using two O_2 fuel cells reaching a precision of ± 1 ppm O_2 (Cen, Turpin, & Layzell, 2001). AQ is
78 expected to be near 1.0 when the Calvin cycle is the dominant sink of photosynthetic energy and reducing
79 equivalents (and when carbohydrates are used as the respiratory substrate). However, AQ values can
80 deviate from 1.0 as a result of alternate sinks not directly coupled to CO_2 metabolism including nitrate
81 photo-assimilation (Cousins & Bloom, 2004; Smart & Bloom, 2001) and potentially lipid and lignin
82 biosynthesis (Cen et al., 2001; Cousins & Bloom, 2003; Searles & Bloom, 2003). Increased activity of
83 alternative NADPH sinks like nitrate reduction can result in reductions in AQ due to a fraction of
84 photosynthetically produced O_2 (and ETR) not directly associated with CO_2 fixation (Bloom, 2015). This
85 effect is particularly pronounced under photorespiratory conditions when low intercellular CO_2 mole
86 fraction (C_i) constrains RuBisCO carboxylation rates, and therefore the demand of the Calvin-Benson
87 cycle for ATP/NADPH. Nitrate assimilation has little effect on net CO_2 assimilation, but enhances NOP
88 leading to a reduction in AQ (Bloom, 2015; Noctor & Foyer, 1998).

89 While little research has studied the impact of plastidic lipid synthesis during photosynthesis on
90 AQ (Tcherkez & Limami, 2019), chloroplastic fatty acid (Tovar-Méndez, Miernyk, & Randall, 2003) and
91 isoprenoid (Eisenreich, Bacher, Arigoni, & Rohdich, 2004) biosynthesis strictly occur in the light,
92 requiring the photosynthetic products NADPH, ATP, and glycerate-3-phosphate produced by RuBisCO
93 catalyzed carboxylation of ribulose-1,5-biophosphate (Rodrigues et al., 2020). Thus, like nitrate
94 assimilation, photosynthetically-linked lipid synthesis represents a potentially significant alternative sink
95 of ATP/NADPH during O_2 production in chloroplasts, especially during photorespiratory conditions like
96 high temperature which greatly enhances rates of isoprene synthesis (K. Jardine et al., 2014; Loreto &
97 Sharkey, 1990; Thomas D Sharkey & Yeh, 2001) due to temperature-dependent changes in substrate pool
98 size, isoprene synthase activity (Rasulov, Hüve, Bichele, Laisk, & Niinemets, 2010). Isoprene is a
99 particularly sensitive measure of chloroplastic ATP status, as isoprene synthase pathway has a high
100 effective K_m for ATP (Rasulov, Bichele, Laisk, & Niinemets, 2014b; Rasulov, Talts, & Niinemets, 2016).

101 Chloroplast membranes contain high amounts of the galactolipid digalactosyldiacylglycerol
102 (DGDG) containing the fatty acid α -linolenic acid identified in early studies as a major fatty acid
103 synthesized within chloroplasts (Bolton & Harwood, 1978). During heat stress, enhanced DGDG
104 synthesis and incorporation into thylakoid membranes plays an important role in 'acquired

105 thermotolerance' of plants (Chen, Burke, Xin, Xu, & Velten, 2006). However, while lipid synthesis and
106 metabolism is widely recognized as a central component of leaf thermotolerance (Wahid, Gelani, Ashraf,
107 & Foolad, 2007), few studies have quantified the temperature sensitivity of lipid synthesis in chloroplasts
108 (Tcherkez & Limami, 2019), with most studies focusing on the composition of lipids present rather than
109 their synthesis rates (Shiva et al., 2020). Isoprene is a volatile light-dependent photosynthetic lipid
110 produced and emitted by leaves of many tree species globally as a function of temperature (Monson et al.,
111 1992). Early pioneering studies combined gas exchange methods with remote sensing methods and
112 quantified leaf isoprene emissions together with CO₂/H₂O gas exchange fluxes and chlorophyll
113 fluorescence (Loreto & Sharkey, 1990). Isoprene emissions from photosynthesizing leaves of red oak
114 (*Quercus rubra* L.) increased with light intensity, were suppressed under CO₂-free and elevated CO₂
115 atmospheres, and strongly enhanced with temperature (Loreto & Sharkey, 1990). While isoprene
116 synthesis depends on carbon skeletons from the Calvin cycle, isoprene production rates are primarily
117 controlled by utilization of products from the light reactions such as ATP and NAPDH (Loreto &
118 Sharkey, 1990). This is consistent with current photosynthesis-based models of isoprene emissions which
119 predict variations in isoprene emissions are primarily driven by changes in the energy status of
120 chloroplasts (Rasulov, Huve, Vålbe, Laisk, & Niinemets, 2009) as well as by the overall isoprenoid
121 synthesis pathway activity (Niinemets, Rasulov, & Talts, 2021; Rasulov, Bichele, Laisk, & Niinemets,
122 2014a).

123 While the majority of carbon in leaf isoprene (C₅H₈) emissions derive from atmospheric CO₂
124 within minutes of photosynthesis in the light (Karl et al., 2002), alternate 'apparent' stored carbon sources
125 for isoprene increase during stress (Funk, Mak, & Lerdau, 2004) such as high temperature (K. Jardine et
126 al., 2014). Externally supplied pyruvate and glucose have been demonstrated as effective isoprene carbon
127 sources (K. J. Jardine et al., 2010; Kreuzwieser et al., 2002) and studies that labeled leaf isoprene
128 with ¹³CO₂ suggested that pyruvate for isoprene synthesis may derive primarily from recent
129 photosynthesis, but also partially from the import of cytosolic pyruvate generated during glycolysis (Karl
130 et al., 2002). Studies using CO₂-free air suggested that re-assimilation of CO₂ under photorespiratory
131 conditions may play important roles as an 'alternative' carbon source for isoprene and become important
132 as a thermotolerance mechanism during stress like high temperature and drought (Garcia et al., 2019).
133 Although they are accounted for in equations describing net photosynthesis and ¹²C/¹³C fractionation,
134 internal CO₂ and O₂ recycling in leaves are difficult to study (Tcherkez et al., 2017), but are known to
135 accelerate under stress when stomata close (Ma, Behboudian, Turner, & Palta, 2001). Thus, isoprene
136 emissions may provide insight into the role of internal CO₂ and O₂ recycling in leaves under
137 photorespiratory stress conditions such as high temperature (Voss, Sunil, Scheibe, & Raghavendra, 2013),

138 with emission rates a potential indicator of de-novo lipid biosynthesis activity in chloroplasts (K. J.
139 Jardine et al., 2020). Thus, we hypothesize that high rates of leaf isoprene emissions correspond to high
140 carbon fluxes through the isoprenoid and fatty acid pathways which are primarily driven by changes in
141 NADPH and ATP availability from the light reactions.

142 Here, we coupled a high precision O₂ cavity ring down spectrometer (CRDS) and a proton
143 transfer reaction-mass spectrometer (PTR-MS) to the sampling port of a commercial leaf gas exchange
144 system with full environmental control and integrated fluorimeter (LI-6800 with 6 cm² leaf chamber). This
145 coupling added O₂ and isoprene fluxes to CO₂/H₂O gas exchange (with chlorophyll fluorescence) for
146 simultaneous, real-time quantification of photosynthetic traits such as electron transport rate (ETR), net
147 CO₂ assimilation (A_{net}), net oxygen production (NOP), stomatal conductance (g_s), and isoprene emissions.
148 Furthermore, it allows the calculation of the assimilatory quotient ($AQ = A_{\text{net}}/NOP$) to obtain additional
149 information on the photosynthetic redox budget in leaves. We measured light, CO₂, and temperature
150 responses of leaf gas-exchange (CO₂, H₂O, O₂, and isoprene), using mature leaves of California poplar
151 (*Populus trichocarpa* Torr. & Gray) as the model tree system. The optimal temperature of gross oxygen
152 production (GOP) was measured using a method derived from Gauthier *et al.* (2018) via ¹⁸O-water
153 labelling.

154 We hypothesize that gross fluxes of photosynthesis, (photo)respiration, and lipid synthesis have
155 distinctly different temperature sensitivities and optimum temperatures. This would imply that as leaf
156 temperature increases beyond the optimal for A_{net} and NOP, an increasing proportion of ATP and NADPH
157 from ‘light’ reactions are used for photorespiration (Long, 1991; Walker, VanLoocke, Bernacchi, & Ort,
158 2016) and lipid synthesis (Rodrigues et al., 2020) instead of CO₂ assimilation. Due to partial stomatal
159 closure leading to reduced C_i , the suppression of atmospheric CO₂ uptake at high leaf temperature is
160 partially compensated for by increased refixation of (photo)respiratory CO₂ (Voss et al., 2013) and thus
161 enhanced CO₂/O₂ recycling (Garcia et al., 2019). We hypothesize that high temperatures will stimulate
162 chloroplastic light-dependent lipid synthesis driven by excess NADPH and ATP not being used by the
163 Calvin cycle (Morfopoulos et al., 2014), leading to a detectable decrease in AQ. To test this hypothesis,
164 we quantified the temperature dependence of ETR, A_{net} , NOP, AQ, and isoprene emissions as well as
165 looked at the temperature dependence of GOP, which we hypothesized would follow the pattern of
166 photosynthetic ETR determined from chlorophyll fluorescence.

167

168 **Material and Methods**

169 **Plant material**

170 We used 15 potted California poplar (*Populus trichocarpa*) saplings (average height of 2 m in 15-gallon
171 pots) obtained from Plants of the Wild (Washington State, USA) and maintained for three years in the
172 South Greenhouse at the Oxford Tract Experimental Facility in Berkeley, CA, USA. The plants were
173 regularly watered using an automated watering system and subject to standard pest control practices. The
174 pots were filled with Supersoil planting media (Scotts Co., Marysville, Ohio, USA) and nitrogen was also
175 added in the form of both nitrate (NO_3^-) and ammonium (NH_4^+) supplied using three fertilizers. Slow
176 release Osmocote plus was added directly to the soil during potting (240 g per pot), whereas Yara Liva
177 $\text{Ca}(\text{NO}_3)_2$ at 90 ppm and Peters Professional at 74 ppm were mixed together in the irrigation water and
178 applied five times per week to soil saturation. Ambient natural light was supplemented with LED lighting
179 for the 16-hour photoperiod (6:00 AM to 10:00 PM) using an Argus Titan environmental control system
180 (Argus Controls, British Columbia, Canada). The LED lamps (10% blue, 90% red) increased light
181 intensities at branch height by 400-1000 $\mu\text{mol m}^{-2} \text{s}^{-1}$ depending on height and position of the top
182 branches, with a controller automatically switching off the supplemental LED lights when the exterior
183 light intensity was above 850 $\mu\text{mol m}^{-2} \text{s}^{-1}$.

184

185 **Leaf gas exchange measurements**

186 Poplar branches were detached from one of the 15 trees in the greenhouse in the morning (9:00-12:00),
187 with stems immediately immersed and recut under water, and then transferred to the nearby laboratory.
188 Harvesting branches for gas exchange studies (only one branch was removed per month per individual)
189 did not have a negative impact on tree growth, as new leaves/branches were continuously generated by
190 the potted trees in the greenhouse. The selected leaf to be studied for gas exchange was placed in the leaf
191 chamber, ensuring complete coverage of the 6 cm^2 or 36 cm^2 chamber window, depending on the leaf
192 chamber used. To hydrate the branch and minimize water loss through transpiration, the branch outside
193 the leaf chamber was immediately covered with a Mylar sheet with wet paper towels placed around the
194 base. Therefore, only the leaf in the chamber was actively transpiring. This was found to be important at
195 high leaf temperatures (e.g., 40 °C) to avoid leaf desiccation in the chamber associated with elevated
196 transpiration rates. After an acclimation period (15 min), light, CO_2 , or temperature response curves were
197 measured (**Figure 1a**). In a separate set of experiments with only the large leaf chamber, following the
198 installation of a leaf in the chamber in darkness, the petiole was cut and placed in a solution of H_2^{18}O for a
199 equilibration period before measurements of a leaf temperature response (**Figure 1b**).

200 For all experiments, CO_2 and H_2O gas exchange was measured under controlled environmental
201 conditions using a portable photosynthesis system (LI-6800, LI-COR Biosciences, USA) coupled to a
202 high precision O_2 CRDS (Picarro Inc., USA) for O_2 and quadrupole PTR-MS (Ionicon, Austria) for

203 isoprene measurements (**Figure 1**). A fraction of air exiting the leaf chamber was diverted from the LI-
204 6800 subsampling gas port to the O₂ CRDS (90 mL min⁻¹) and PTR-MS (75 mL min⁻¹) using a 3.175 mm
205 O.D. Teflon PTFE tube maintained at 50-60 °C with a self-regulating heating tape (SLR10, Omega
206 Engineering, USA) to prevent condensation and gas-tubing wall interactions prior to gas analysis by the
207 CRDS and PTR-MS sensors. The measurement gas source for the LI-6800 was supplied externally by
208 overblowing a T-fitting with high purity zero air (ultra-zero air, CAS: 132259-10-0, Linde Gas) such that
209 at least 200 mL min⁻¹ vented externally while the remaining flow passed through a platinum catalytic
210 converter held at 280 °C (ZA30 catalyst, Aadco instruments, USA) to oxidize any trace volatile organic
211 compounds (VOCs) before entering the air inlet of the LI6800. Therefore, air delivered to the LI6800 air
212 inlet port was CO₂-, H₂O-, and VOC-free while maintaining a constant concentration of O₂, which slightly
213 varied from cylinder to cylinder between 20.09 and 21.03%. Leaf chamber humidity was regulated
214 through automated balancing of air flow through the desiccant (Drierite with 10-20 mesh size CAS:778-
215 18-9, Drierite) and humidifier (1/8" O.D. Nafion tubing immersed in ACS/HPLC water, CAS: 7732-18-5,
216 Honeywell) in order to maintain the absolute humidity of the reference air at the desired setpoint of 0-12
217 mmol mol⁻¹. CO₂ mole fraction inside chamber was controlled by passing all airflow through the CO₂
218 scrubber (soda lime, 4-8 mesh size, CAS: 8006-28-8, Thermo Scientific) while carbon dioxide was
219 supplied by an external cylinder (CAS 124-38-9, 99.9% CO₂, Praxair). When the LED lights inside the
220 leaf chamber was switched on (1000 μmol m⁻² s⁻¹), the spectrum was set to 960 μmol m⁻² s⁻¹ red and 40
221 μmol m⁻² s⁻¹ blue as the manufacturer's recommended color spectrum for the fluorimeter to have just
222 enough blue for stomatal control and to set the actinic and the fluorescence measuring beam as spectrally
223 close as possible.

224 Leaf isoprene emission was measured for all gas-exchange measurements using a real-time high
225 sensitivity quadrupole proton transfer reaction mass spectrometry (PTR-MS, with a QMZ 422
226 quadrupole, Balzers, Switzerland) as previously described (K. Jardine et al., 2014). The PTR-MS was
227 operated with a drift tube voltage of 440 V and pressure of 1.8 mbar. For each measurement cycle lasting
228 24 sec, the following mass to charge (m/z) ratios were monitored: m/z 21 (H₃¹⁸O⁺), m/z 37 (H₃O⁺-H₂O),
229 and m/z 69 (protonated isoprene: H⁺-C₅H₈). To obtain the system background for the PTR-MS signal at
230 m/z 69, measurements were made with no leaf in the chamber both before and after every environmental
231 response curve with a leaf. Once a leaf was installed in the chamber, isoprene concentrations inside the
232 leaf chamber were calculated by subtracting the background m/z 69 and applying the calibration
233 sensitivity of the m/z 69 signal to isoprene determined separately through dynamic dilution of 1.0 ppm
234 isoprene standard. A similar procedure was used to determine the background concentration of O₂ (see
235 section below on NOP calculation).

236 In these experiments, two different leaf chambers were used with distinct advantages and
 237 disadvantages (supplementary **Figure S1**). The smaller leaf chamber (6 cm²) had the added advantage of
 238 including an integrated chlorophyll fluorimeter (6800-01A, LI-COR Biosciences, USA) together with
 239 H₂O and CO₂ gas exchange. Due to the rerouting of a fraction of the outlet air for simultaneous
 240 measurements of O₂ and isoprene concentrations, and small leaks that formed between the gasket and the
 241 leaf/petiole, over-pressurizing the leaf chamber (0.1 KPa) with an optimized flow rate of 323-363 mL
 242 min⁻¹ (240-270 μmol mol⁻¹) ensured high O₂ gradients while maintaining sufficient flow for O₂ (90 mL
 243 min⁻¹), isoprene (75 mL min⁻¹), and CO₂ + H₂O (158-198 mL min⁻¹) measurements (supplementary **Figure**
 244 **S1a**). Chlorophyll fluorescence data were simultaneously recorded with gas exchange data during light,
 245 CO₂ and temperature responses curve measurements with an integrated multiphase flash fluorimeter
 246 system (model 6800-01A, LI-COR Biosciences, USA). To measure the light-adapted maximum
 247 fluorescence yield, F_m' , an actinic light pulse of 1000 μmol m⁻² s⁻¹ was applied for 1 s. The fluorimeter
 248 measurement light frequency was 50 Hz in dark and 1 kHz in light, and 250 kHz during saturating flash.
 249 For steady-state fluorescence measurements (F_s), 15 s chlorophyll fluorescence signal averaging was used
 250 (100 Hz data output rate with a margin of 5 averaged points before and after flash). Photosynthetic
 251 electron transport rate (ETR, μmol e⁻ m⁻² s⁻¹) was calculated according to **Equation 1**, where f is the
 252 fraction of the quantum absorbed and used by Photosystem II, with a value of 0.5 used for C₃ plants (Earl
 253 & Tollenaar, 1998), Photosynthetically Active Radiation (PAR) is the incident photon flux density (μmol
 254 m⁻² s⁻¹), and α_{leaf} is the fraction of light absorbed by the leaf (0.87). Although α_{leaf} was not experimentally
 255 determined, both the blue and red wave lengths are known to be strongly absorbed by green leaves, with
 256 typical values between 0.84-0.90. For example, when leaf light absorption was quantified for four broad
 257 leaf tree species, α_{leaf} values ranged between 0.87 and 0.92 (Kang, Zhu, Yamori, & Tang, 2020).

258 **Equation 1:**
$$ETR = \frac{F_m' - F_s}{F_m'} \times f \times PAR \times \alpha_{leaf}$$

259 A larger leaf chamber with integrated LED light source (model 6800-03 LI-COR Biosciences,
 260 USA) was also used with the advantage of enclosing a much larger area of enclosed leaf (36 cm²). This
 261 allowed for a higher flow rate of air to be delivered to the leaf chamber (538 mL min⁻¹ or 400 μmol s⁻¹).
 262 While lacking a fluorimeter, the large leaf chamber can control leaf temperature and the actinic light
 263 spectra which was set identically to the small chamber (960 μmol m⁻² s⁻¹ red and 40 μmol m⁻² s⁻¹ blue)
 264 (supplementary **Figure S1b**).

265

266 **Photosynthesis, ETR, and isoprene emission responses to environmental drivers**

267 **Light response curves.** Photosynthetic light response curves were measured at a constant leaf temperature
268 (32 °C), leaf chamber CO₂ mole fraction of 400 μmol mol⁻¹, and reference (inlet) air humidity of 12 mmol
269 mol⁻¹. For both large and small chambers, after 30 min dark acclimation (PAR: 0 μmol m⁻² s⁻¹), leaf gas
270 exchange and chlorophyll fluorescence (small chamber only) responses to light intensity were measured.
271 This included a sequence of increasing followed by decreasing PAR (0, 200, 400, 600, 800, 1000, 1200,
272 1400, 1600, 1200, 800, 400, 100, 50, 40, 30, 20, 10 and 0 μmol m⁻² s⁻¹). Total time duration for
273 measurement of a light response curve was 200 min. Three replicate light response curves were collected
274 using the small chamber with gas exchange and chlorophyll fluorescence and two replicate light response
275 curves were collected using the large chamber with gas exchange only. For each replicate, a branch from
276 a different tree (5 out of 15 total) was used.

277 **C_i response curves.** The response of leaf gas exchange to intercellular CO₂ mole fraction (C_i) were
278 measured by varying the reference CO₂ mole fraction entering the leaf chamber while maintaining
279 constant leaf temperature (32 °C), PAR (1000 μmol m⁻² s⁻¹) and reference air humidity (10 mmol mol⁻¹).
280 For both small and large leaf chambers, after 30 min light acclimation (PAR: 1000 μmol m⁻²s⁻¹), leaf gas
281 exchange and chlorophyll fluorescence (small chamber only) response curves to CO₂ were measured. This
282 included a sequence of decreasing followed by increasing reference CO₂ mixing ratios (400, 350, 300,
283 250, 200, 150, 125, 100, 75, 50, 25, 0, 50, 100, 150, 200, 250, 300, 400, 500, 600, 700, 800, 900, 1000
284 ppm). Total time duration for a single C_i response curve was 160 min. Four replicate C_i response curves
285 were collected using the small chamber and three replicate C_i response curves were collected using the
286 large chamber. For each replicate, a branch from a different tree (7 out of 15 total) was used.

287 **Leaf temperature response curves.** Leaf temperature response curves were measured with varying leaf
288 temperature at a constant CO₂ mixing ratio in the leaf chamber (C_a, 400 μmol mol⁻¹) and inlet (reference)
289 air humidity (0-8 mmol mol⁻¹). To prevent condensation in the large leaf chamber, dry inlet air (with 0
290 mmol mol⁻¹ water vapor) was supplied, while in the small leaf chamber, the shorter gas residence time
291 allowed us to use inlet air with a humidity of 8 mmol mol⁻¹ (see also *Discussion*). For both large and small
292 leaf chambers, after 30 min dark acclimation (PAR: 0 μmol m⁻² s⁻¹) at 25 °C leaf temperature, leaf gas
293 exchange (both chambers) and chlorophyll fluorescence (small chamber only) responses to leaf
294 temperature were measured. The sequence started with leaf dark respiration measurements at 25.0 °C
295 (PAR: 0 μmol m⁻² s⁻¹). Following a 20 min period of light acclimation, measurements of the temperature
296 response curve in the light (PAR: 1000 μmol m⁻² s⁻¹) was initiated with increasing leaf temperatures (25,
297 27.5, 30, 32.5, 35, 37.5, 40 °C). After the temperature response curve measurements, incident light was
298 switched off to record leaf dark respiration at 40 °C. Total time duration for a leaf temperature response
299 curve measurement was 120 min. Eight replicate leaf temperature response curves were collected using

300 the small chamber. For each replicate, a branch from a different tree (8 out of 15 total) was utilized. In
301 addition, seven replicate leaf temperature response curves were collected using the large chamber (7 out
302 of 15 total).

303

304 **Leaf H₂¹⁸O labeling**

305 To determine optimal temperature of gross oxygen production (GOP), leaf responses to temperature were
306 monitored using the large leaf chamber with detached poplar leaves equilibrated with a solution of H₂¹⁸O
307 water (seven replicate temperature curves from individual replicate trees). The O₂ CRDS was switched
308 into isotope mode, where δ¹⁸O in O₂ of the leaf headspace air was measured with <1‰ precision using 7-
309 min averages. Water enriched in H₂¹⁸O (δ¹⁸O value of +8,000‰ relative to Vienna Standard Mean Ocean
310 Water, V-SMOW) was prepared by diluting 10 atom % H₂¹⁸O water (CAS:14314-42-2, Sigma-Aldrich)
311 with HPLC grade water. The leaf was detached from the branch and the petiole immediately recut under
312 H₂¹⁸O enriched water, and then placed in the large chamber under constant light (PAR: 1000 μmol m⁻²s⁻¹),
313 leaf temperature (32 °C), and leaf chamber CO₂ mole fraction (C_a, 400 μmol mol⁻¹). δ¹⁸O of O₂ of air
314 inside the leaf chamber was measured before, during, and after gas exchange experiments with a leaf
315 (before and after measured with no leaf in the chamber). Concurrently, continuous measurement of leaf
316 isoprene emission was measured using PTR-MS (**Figure 1b**). Equilibration of the leaf with ¹⁸O-enriched
317 water occurred for 2-3 h during which the δ¹⁸O of O₂ values reached a steady state, indicating the turn-
318 over of all non-static leaf water pools. Following the equilibration period, the leaf temperature response
319 curve was measured with the same protocol as that used for attached leaves (see above).

320

321 **Real-time measurement of leaf net oxygen production (NOP)**

322 CO₂ and H₂O were quantitatively scrubbed from the air exiting the leaf chamber by passing it through
323 indicating soda lime (replaced monthly) followed by indicating dririte (replaced daily) using separate
324 chemical tube assemblies (Licor Inc., part # 9960-093). For all O₂ measurements, H₂O remained below
325 0.1%. Following the scrubbing of CO₂ and H₂O from the diverted air flow exiting the leaf chamber, an
326 infrared laser-based cavity ring-down spectrometer (CRDS, Picarro G2207-i, O₂/H₂O, USA) was used for
327 continuous high precision measurement of O₂ mole fraction or δ¹⁸O values in O₂ (**Figure 1**). In fact, the
328 CRDS could be operated in one of two different modes: high precision concentration and isotopic ratio
329 modes. In concentration mode, O₂ mole fraction was measured with < 2 ppm precision using 7-min
330 averages. O₂ reference mole fraction measurements were made with an empty chamber before and after
331 all leaf environmental response curves and used to calculate the change in O₂ concentrations due to leaf
332 gas exchange (ΔO₂). Small drifts in measured O₂ inlet mole fraction during the response curves, as

333 determined from measurements without a leaf in the chamber before and after the environmental response
334 curves, were < 20 ppm O₂ (see example raw data supplementary **Figure S2**). This drift in reference leaf
335 chamber O₂ concentrations is attributed to the CRDS itself and was subtracted from the headspace O₂
336 concentrations when a leaf was in the chamber. That way, the difference in O₂ mole fractions between leaf
337 chamber and reference air (ΔO_2) could be determined in real-time during the environmental response
338 curves. Leaf NOP ($\mu\text{mol m}^{-2} \text{s}^{-1}$) fluxes were calculated using **Equation 2** where μ is the air flow rate
339 entering the leaf chamber (mol air s^{-1}), ΔO_2 is the difference in oxygen mole fraction between leaf
340 chamber and reference air corrected for CRDS drift ($\mu\text{mol mol}^{-1}$), and S is leaf surface area (0.0006 or
341 0.0036 m²) inside the chamber. Note, that due to quantitative scrubbing of CO₂ and H₂O in the air exiting
342 the leaf chamber just prior to making O₂ measurements by the CRDS, corrections associated with air flow
343 rate due to transpiration and photosynthesis were not necessary.

344 **Equation 2:**
$$NOP = \mu \frac{\Delta O_2}{S}$$

345 To determine the average AQ = A_{net}/NOP for each environmental leaf response curve, a linear
346 regression analysis was performed with A_{net} (y-axis) plotted against NOP (x-axis). Highly linear
347 correlations were observed in all cases, with the slope of the regression representing AQ. In addition to
348 AQ values determined from the slope of the linear correlations from each leaf environmental response
349 curve, mean AQ values were also determined by directly dividing the mean values of A_{net} by NOP for
350 each value of PAR, C_i , and leaf temperature.

351

352 **Results**

353 **Light response**

354 A_{net} and NOP increased as a function of photosynthetically active radiation (PAR) together with ETR and
355 isoprene emissions. Whereas ETR saturated around 1000 $\mu\text{mol m}^{-2} \text{s}^{-1}$ PAR, A_{net} , NOP, and isoprene
356 emissions continued to increase with light up to the highest intensity (1600 $\mu\text{mol m}^{-2} \text{s}^{-1}$ PAR). An
357 example light response curve is shown in **Figure 2** and summarized in **Figures S3** for all replicate
358 experiments. A_{net} showed a higher magnitude relative to NOP (**Figure 2b**). This resulted in an
359 assimilatory quotient AQ = A_{net}/NOP higher than unity (AQ = 1.3, **Figure 2c**). In this example, dark CO₂
360 evolution was 3.8 $\mu\text{mol m}^{-2} \text{s}^{-1}$ while dark oxygen consumption was 2.5 $\mu\text{mol m}^{-2} \text{s}^{-1}$. Similarly, under
361 saturating light (1600 $\mu\text{mol m}^{-2} \text{s}^{-1}$ PAR), A_{net} (18.4 $\mu\text{mol m}^{-2} \text{s}^{-1}$) was higher than NOP (15.5 $\mu\text{mol m}^{-2} \text{s}^{-1}$)
362 (**Figure 2b,c**). At low light intensity (0-200 $\mu\text{mol m}^{-2} \text{s}^{-1}$) a linear response was observed with A_{net} , NOP,
363 and ETR (**Figure 2b** and supplementary **Figures S3**). Stomatal conductance (g_s) and transpiration rate (E)
364 also increased with PAR, reaching a maximum value (g_s : 0.35 mol m⁻² s⁻¹, E : 8.0 mmol m⁻² s⁻¹) at 1200

365 $\mu\text{mol m}^{-2} \text{s}^{-1}$. At high light, both g_s and E decreased slightly. AQ values determined using the small leaf
366 chamber (6 cm^2 , $\text{AQ} = 1.26 \pm 0.06$) were similar to those determined with the large leaf chamber (36 cm^2 ,
367 $\text{AQ} = 1.22 \pm 0.01$) (**Table 1**).

368

369 **CO₂ response**

370 Response curves to intercellular CO₂ mole fraction (C_i) are shown in **Figure 3** and summarized in
371 supplementary **Figure S4** for all of the replicate experiments. Both A_{net} and NOP increased with C_i ,
372 although A_{net} showed a larger magnitude at both low (more negative) and high C_i (more positive) than
373 NOP. Across all C_i response curves, AQ values determined using the small leaf chamber (6 cm^2 , $\text{AQ} =$
374 1.23 ± 0.08) were similar to those determined using the large leaf chamber (36 cm^2 , $\text{AQ} = 1.27 \pm 0.02$)
375 (**Table 1**). For the example shown in **Figure 3**, as reference CO₂ entering the leaf chamber declined from
376 400 to $0 \mu\text{mol mol}^{-1}$, C_i declined from 320 to $31 \mu\text{mol mol}^{-1}$. A_{net} , NOP, and ETR strongly declined by
377 122% , 122% , and 59% , respectively with A_{net} and NOP becoming negative below a C_i of $56 \mu\text{mol mol}^{-1}$
378 (the CO₂ compensation point for A_{net} and NOP), such that the leaf became a net source of CO₂ and sink of
379 O₂ in the light. At the lowest C_i ($31 \mu\text{mol mol}^{-1}$), net CO₂ evolution ($3.8 \mu\text{mol m}^{-2}\text{s}^{-1}$) was slightly higher
380 than net oxygen consumption ($2.2 \mu\text{mol m}^{-2}\text{s}^{-1}$). In contrast, isoprene emissions were stimulated as C_i
381 declined from 320 to $56 \mu\text{mol mol}^{-1}$ (67% increase), followed by a decline as C_i reached the lowest value,
382 $31 \mu\text{mol mol}^{-1}$ (19% decrease). As reference CO₂ entering the leaf chamber increased above 400 ppm , A_{net} ,
383 NOP, and ETR strongly increased reaching a photosynthetic plateau for C_i above $588 \mu\text{mol mol}^{-1}$. In
384 contrast, isoprene emissions were suppressed at elevated C_i , decreasing by 87% from 207 to $868 \mu\text{mol}$
385 mol^{-1} . Taken as a whole, A_{net} , NOP, and ETR were much more sensitive to changes in C_i than isoprene
386 emissions.

387

388 **Temperature response**

389 An example leaf temperature response curves is shown in **Figure 4** with a summary of replicates
390 presented in supplementary **Figures S5**. In the light at $25 \text{ }^\circ\text{C}$, A_{net} , NOP, and ETR showed high values
391 while isoprene emissions remained low, but detectable ($< 0.5 \text{ nmol m}^{-2} \text{ s}^{-1}$). As leaf temperature increased
392 in the light, A_{net} and NOP reached maximum values near $31 \text{ }^\circ\text{C}$ and then decreased slightly at higher
393 temperature. In contrast, ETR continued to increase in the light to a maximum near $36 \text{ }^\circ\text{C}$ while isoprene
394 emissions continued to increase up to the highest leaf temperature used ($40 \text{ }^\circ\text{C}$). Upon switching off the
395 light at the highest leaf temperature ($40 \text{ }^\circ\text{C}$), A_{net} and NOP rapidly declined and became negative while
396 isoprene emission was nearly suppressed. In all leaves studied, an increase in leaf dark respiration (A_{net}
397 and NOP) was observed at $40 \text{ }^\circ\text{C}$ relative to $25 \text{ }^\circ\text{C}$. In contrast to what was observed with light and C_i

398 response curves, leaf temperature response curves in the light showed AQ values less than unity with
399 relatively higher variability ($AQ = 0.865 \pm 0.275$, **Table 1**).

400

401 **Temperature response curves with ^{18}O -water**

402 The temperature dependence of gross oxygen production (GOP) was assessed using detached mature
403 poplar leaves placed in the large (36 cm^2) chamber in the light, with the petiole immersed in ^{18}O -enriched
404 water (**Figure 5** and summarized in supplementary **Figures S6**). In the example shown in **Figure 5**,
405 during the equilibration period (incubation in ^{18}O -enriched water) in the light ($1000 \mu\text{mol m}^{-2} \text{ s}^{-1}$) at $30 \text{ }^\circ\text{C}$,
406 the $\delta^{18}\text{O}$ of outlet O_2 increased from the background value (ca. $+11\text{‰}$) and reached $+21\text{‰}$ within two
407 hours in the steady state. During this equilibration period, leaf isoprene emissions also increased, but
408 reached a steady state much faster (within 15 min). Upon switching off the light and reducing leaf
409 temperature to $25 \text{ }^\circ\text{C}$, $\delta^{18}\text{O}$ of outlet O_2 values quickly returned to background values of $+12\text{‰}$. When the
410 light was switched on again at $25 \text{ }^\circ\text{C}$, $\delta^{18}\text{O}$ values reached $+20\text{‰}$ and increased with leaf temperature up
411 to a maximum of $+23\text{‰}$ at $37.5 \text{ }^\circ\text{C}$, and then decreased slightly at the highest leaf temperature ($40 \text{ }^\circ\text{C}$).
412 When the light was switched off at $40 \text{ }^\circ\text{C}$ (end of the temperature response curve), $\delta^{18}\text{O}$ of outlet O_2
413 rapidly returned to the background value of $+12\text{‰}$. Although A_{net} showed a similar optimum leaf
414 temperature of ($30 \text{ }^\circ\text{C}$) to that of non-detached leaves (**Figure 4b** versus **5b**), the optimum temperature of
415 $\delta^{18}\text{O}$ was much higher ($37.5 \text{ }^\circ\text{C}$).

416

417 **Optimal temperature of photosynthetic parameters**

418 Data on optimal temperature (T_{opt}) of A_{net} , NOP, GOP, ETR, and isoprene emissions were compiled from
419 the temperature response curves using the small ($n = 8$) and large ($n = 7$) leaf chambers as well as the
420 large leaf chamber during ^{18}O -water labeling ($n = 7$). As summarized in **Table 2**, A_{net} and NOP showed
421 mean \pm SD optimal temperatures of $31.0 \pm 3.1 \text{ }^\circ\text{C}$ and $31.0 \pm 3.4 \text{ }^\circ\text{C}$, respectively. ETR and GOP showed
422 distinctively higher temperature optima of $35.0 \pm 1.8 \text{ }^\circ\text{C}$ and $34.9 \pm 1.8 \text{ }^\circ\text{C}$, respectively. Isoprene
423 emission had the highest temperature optimum at $38.9 \pm 1.0 \text{ }^\circ\text{C}$. Despite a suppression in stomatal
424 conductance at high temperature (g_s temperature optima of $33.0 \pm 5.7 \text{ }^\circ\text{C}$), transpiration continued to
425 increase with leaf temperatures (T_{opt} of 38.9 ± 2.6).

426

427 **Discussion**

428 Using California poplar (*P. trichocarpa*) as a model tree species, we added net oxygen production (NOP)
429 and isoprene fluxes as well as $\delta^{18}\text{O}$ in O_2 to $\text{CO}_2/\text{H}_2\text{O}$ gas exchange with chlorophyll fluorescence and
430 measured light, CO_2 and temperature response curves. For a detailed discussion of the specific leaf

431 chambers, flow rates, and O₂ concentration gradients (see Supplementary Discussion section, ‘Pros and
432 cons of coupling gas exchange to O₂ and isoprene flux measurements’). It should be noted that we
433 focused here on short-term leaf responses to changes in environmental variables (including temperature)
434 using controlled leaf chambers, and thus our study does not include potential longer-term acclimation
435 effects to growth temperature (Hikosaka, Ishikawa, Borjigidai, Muller, & Onoda, 2006), light (Kull,
436 2002) and CO₂ (Wolfe, Gifford, Hilbert, & Luo, 1998). For example, while we determined that the optimal temperature
437 that maximizes both A_{net} and NOP (T_{opt}) is 31 °C (**Table 2**), previous studies have found that T_{opt} may
438 increase with increasing growth temperature (Hikosaka et al., 2006). *P. trichocarpa* has a very broad and
439 extensive natural distribution in western North America in the foothills of the Sierra Nevada range,
440 Northern California, and throughout much of Oregon and Washington including both sides of the Cascade
441 range, extending into western Canada and north to Alaska as well as east to Alberta, Montana, Utah, and
442 Wyoming (USDA, 2024). Maximum daytime air temperature in Poplar forests in the western United
443 States and Oregon has been reported to vary between 15.5-47.2 °C (Niemic, 1995). Although leaf
444 temperature is not always measured, leaves can be 1-7 °C warmer than air temperature during the day
445 (Gimenez et al., 2019; Monson et al., 2020). Therefore, depending on the local climate and acclimation
446 processes, T_{opt} may be regularly surpassed during the summer growing season at some sites, especially
447 during summer heat waves such as the one in the US Pacific Northwest in June 2021 which broke all-time
448 maximum temperature records by more than 5 °C, and set a new record high temperature of 49.6 °C in
449 Canada (White et al., 2023). Monthly average climatological data maintained by the Western Regional
450 Climate Center at The Poplars site in Oregon (site 358420), USA showed a monthly average high air
451 temperature in August of 29.5 °C, but with daily maximum temperatures reaching up to 40 °C (WRCC,
452 1941-2012). At a poplar plantation in a semi-arid site in Arizona during the summer growing season
453 (June-September), continuous canopy temperature observations during the summer months of 2014
454 showed that leaf temperatures surpassed 31°C almost every day and reached up to ~40 °C on many days
455 (Monson et al., 2020).

456

457 **Apparent assimilatory quotient (AQ)**

458 Across light, CO₂, and temperature response curves, A_{net} and NOP were tightly coupled and highly
459 positively correlated, enabling the determination of apparent AQ values (AQ = A_{net}/NOP) for each leaf
460 experiment. For the light and CO₂ response curves, A_{net} displayed a ~30% higher magnitude compared
461 with NOP during conditions of low or negative net photosynthesis rates (darkness and low C_i) and during
462 conditions of high net photosynthesis rates (e.g. saturating light and CO₂) (**Figures 2-3** and
463 supplementary **Figures S3-4**). This caused the values of AQ to be higher than 1.0. When all leaf response

464 curves were analyzed for AQ values (**Table 1**) with means compared using a t-test, no statistically
465 significant difference was observed between AQ values determined from the light and C_i response curves
466 (two-tailed P value of 0.5524). In contrast, statistically significant differences were observed between AQ
467 values from the light and temperature response curves (two-tailed P value of 0.0067) and C_i and
468 temperature response curves (two-tailed P value of 0.0029). This suggests that the mean leaf assimilatory
469 quotient AQ, as determined here by the regression of all A_{net} versus NOP fluxes obtained during each
470 environmental response curve, did not depend on light or C_i (driven by changes in leaf headspace CO_2
471 concentrations), but appeared to be suppressed as leaf temperature increase in the light. This is consistent
472 with AQ values determined by directly dividing the mean values of A_{net} by NOP for each value of PAR,
473 C_i , and leaf temperature, respectively. That is, while mean AQ values determined as a function of
474 environmental variables remained relatively constant as a function of PAR (supplementary **Figure S3**) or
475 C_i (supplementary **Figure S4**), they appeared to decline as a function of leaf temperature (supplementary
476 **Figure S5**).

477 AQ may deviate from 1.0 when significant activity of alternative electron transport processes
478 occurs, not involved in CO_2 fixation including nitrate photo-assimilation (Bloom, Caldwell, Finazzo,
479 Warner, & Weissbart, 1989) and potentially lipid biosynthesis (Stumpf, Bove, & Goffeau, 1963). For
480 example, when wheat (*Triticum aestivum*) seedlings were grown with NH_4^+ , leaf AQ values were $1.21 \pm$
481 0.06 . Seedlings grown with NO_3^- showed suppressed AQ values of 1.13 ± 0.05 (Smart & Bloom, 2001).
482 Given that both NH_4^+ and NO_3^- were nitrogen sources in both the soil and daily watering in the current
483 study, a major reduction in AQ due to nitrate photo-assimilation in poplar leaves in this study is not
484 expected. Thus, the average AQ values determined here for California poplar leaves during PAR ($1.25 \pm$
485 0.05) and C_i (1.23 ± 0.07) response curves compare well with AQ values (1.21 ± 0.06) determined for
486 wheat leaves supplied with NH_4^+ (Smart & Bloom, 2001). Additional studies with wheat and maize (*Zea mays*)
487 using NH_4^+ as the nitrogen source also observed similar leaf AQ values in the light (e.g. 1.0-1.3), but
488 observed some values less than 1.0 (e.g. 0.8) (Cousins & Bloom, 2004), more comparable to the mean
489 AQ value determined here from the leaf temperature response curves (0.87 ± 0.28 , see **Table 1**). These
490 results support our hypothesis that lipid synthesis in chloroplasts may influence AQ as a function of
491 temperature.

492 Thus, as has been shown for nitrate-photo-assimilation (Bloom et al., 1989), the results are
493 consistent with an increasing fraction of photosynthetic electron transport (and resulting ATP and
494 NADPH) allocated to chloroplastic lipid synthesis (e.g. isoprenoids and fatty acids) at high leaf
495 temperature, resulting in a significant suppression of AQ (**Table 1** and supplementary **Figure S5**). In
496 effect, AQ declines as temperature increases because NOP is more resilient than A_{net} with respect to

497 temperature; ATP and NADPH generated by the light-dependent reactions are increasingly allocated to
498 functions other than CO₂ assimilation, such as lipid synthesis.

499 Isoprene emission rates can increase to very high leaf temperatures up to 45 °C (Monson et al.,
500 1992) and although by itself represents a minor fraction of total ETR (e.g. maximum 1-4%) (Ü.
501 Niinemets, J. Tenhunen, P. C. Harley, & R. Steinbrecher, 1999; Rodrigues et al., 2020), we suggest that
502 its emissions may be an indicator of overall isoprenoid and fatty acid synthesis rates in chloroplasts,
503 which could be expected to accelerate with temperature driven by increased ATP/NADPH availability
504 (see section below, 'Isoprene emission and its potential relationship with NOP and GOP). However,
505 quantitative studies on light-dependent fatty acid and isoprenoid synthesis rates as a function of
506 temperature are rare with most studies quantifying leaf lipid composition profiles rather than synthesis
507 rates. Earlier studies estimated fatty acids synthesis rates using ¹⁴C-acetate labelling of isolated
508 chloroplasts (Heinz & Roughan, 1983; Roughan & Ohlrogge, 1996). More recently, ¹³CO₂ labeling
509 studies have shown rapid ¹³C-incorporation into fatty acids (Ohlrogge et al., 2000) and isoprenoids
510 (Karl et al., 2002) within minutes of photosynthesis. When ¹³CO₂ labeling was used to quantify the
511 absolute rate of fatty acid synthesis of Arabidopsis plants, synthesis was halted in the dark, but proceeded
512 at high rates in the light (12-24 µg hr⁻¹ mg chl⁻¹). Assuming synthesis of α-linolenic acid (C₁₈H₃₀O₂) and a
513 leaf chlorophyll content of 0.5 mg cm⁻², this corresponds to a fatty acid synthesis rate of 0.06-0.12 µmol
514 α-linolenic acid s⁻¹ m⁻², requiring a photosynthesis flux of 1.1-2.2 µmol CO₂ s⁻¹ m⁻² (18 moles of CO₂
515 assimilated/mole of α-linolenic acid synthesized). Presuming a light-saturated A_{net} flux of 7.0 µmol CO₂ s⁻¹
516 m⁻² for *A. thaliana* leaves (Tanaka, Sugano, Shimada, & Hara-Nishimura, 2013), this suggest that a
517 substantial fraction (15-31%) of net CO₂ assimilation can be allocated to fatty acid synthesis. Studies
518 conducting 2-min pulse-chase ¹⁴CO₂ labeling of *A. thaliana* leaves confirmed that a considerable portion
519 of the assimilated ¹⁴CO₂ (10.4 ± 1.1%) can be allocated to lipid synthesis (ethanol-soluble compounds) in
520 the light (Kölling, Thalmann, Müller, Jenny, & Zeeman, 2015). Thus, studies quantifying total volatile
521 and non-volatile isoprenoid and fatty acid synthesis rates as a function of leaf temperature are needed to
522 quantitatively compare lipid synthesis fluxes with A_{net} and NOP in order to evaluate the potential
523 temperature dependency of chloroplast lipid synthesis rates and AQ in the light (see additional discussion
524 on AQ in the supplementary discussion section, 'Plant CO₂/O₂ metabolism and transport and the net
525 assimilatory quotient (AQ)'.
526

526 Nonetheless, it is important to note that AQ (assimilation quotient) and RQ (respiration quotient:
527 CO₂ produced/O₂ consumed) values of plant tissues based on gas-exchange methods are well known to be
528 difficult to accurately obtain with a high degree of confidence because of the separate analytical
529 techniques required (Scafaro et al., 2017). Systematic errors in either CO₂ flux or O₂ flux measurements

530 will propagate into AQ and RQ values. In studies of leaf RQ values in the dark, when different methods to
531 measure CO₂ and O₂ fluxes were compared, statistically significant differences were found between them.
532 For example, when three methods to determine leaf dark respiration by fluorophore, O₂-electrode, IRGA,
533 and membrane inlet mass spectrometry techniques were compared, substantially different RQ values were
534 obtained (Scafaro et al., 2017). Using leaf dark respiration based on IRGA observations of net CO₂ fluxes
535 in the dark, RQ values equal to 1.0 as well as substantially less and greater than 1.0 could be obtained,
536 depending on the method used to measure O₂ fluxes. Calibration of CO₂ and/or O₂ sensors can improve
537 the accuracy of flux measurements but requires highly accurate and precise gas standards spanning the
538 range in observed concentrations. In our study, we lacked a suite of high precision CO₂ and O₂ standards
539 and relied on recent factory calibrations for the CO₂ (IRGA) and/or O₂ (CRDS). Our method for AQ
540 determination depends on the slope of the instrument response to small changes (e.g. 0-200 ppm) in O₂
541 concentrations (the sensitivity). Although we don't have any evidence pointing to this possibility, a slight
542 underestimate of the actual CRDS sensitivity to changes in O₂ concentrations relative to the recent factory
543 calibration would lead to underestimating NOP, and therefore overestimating AQ. Future studies should
544 therefore attempt to address this issue by calibrating CO₂ and O₂ sensors with high accuracy and precision
545 gas standards that span the range of concentrations encountered in dynamic plant enclosures. Calibration
546 of high precision CO₂ and O₂ sensors has been achieved using high pressure cylinders of ambient air with
547 known CO₂ and O₂ mole fractions certified by a specialized laboratory using an LI-6252 for CO₂ (IRGA)
548 and an Oxzilla II (lead fuel cell O₂) (Pickers, 2016). However, regardless of absolute AQ values, our data
549 allows for a relative comparison of AQ changes in response to light, CO₂, and temperature variations.

550 Given that ETR measured by chlorophyll fluorescence is based on PSII electron flow, and gross
551 rates of oxygen production also reflect PSII activity, a tight correlation is expected between GOP and
552 ETR as leaf temperatures vary in the light. Consistent with this prediction, ETR, determined using the
553 fluorimeter, and δ¹⁸O of O₂ during H₂¹⁸O leaf labeling as a proxy for GOP, both increased to a similar
554 optimal temperature of 35 °C before declining at higher temperatures (see supplementary discussion: Pros
555 and cons of the ¹⁸O-labeling method). This is distinctly higher than the optimal temperature of NOP and
556 A_{net} (31 °C). This suggests that the suppression of A_{net} and NOP at high temperature is mainly due to
557 higher (photo)respiratory CO₂ production/O₂ consumption. This would be consistent with a model where
558 at the optimal temperature for A_{net} and NOP of 31 °C in the light, relatively low rates of photorespiration,
559 respiration, lipid biosynthesis (fatty acids and isoprenoids), and CO₂/O₂ recycling occur (**Figure 7**). In
560 contrast, at the optimal temperature for GOP and ETR (35 °C), a reduction in g_s leads to a decrease in
561 gross atmospheric CO₂ uptake, which is partially compensated for by increased re-assimilation of internal

562 CO₂. In effect, there is an increase in CO₂ liberation by photorespiration (due to a decline of RuBisCO
563 specificity and thus an increase in Γ^*) and mitochondrial respiration when temperature increases (Voss et
564 al., 2013). This mechanism is illustrated in **Figure 7**, where the suppression of A_{net} and NOP at 35 °C
565 versus 31 °C is not due to high temperature stress on photosynthesis per se, but rather a concurrent change
566 in gross photosynthesis, (photo)respiration as well as CO₂ and O₂ recycling (Eckert, Jensen, & Gu, 2020;
567 Eckert, Martens, Gu, & Jensen, 2021; Garcia et al., 2019). In contrast, temperatures higher than 35 °C
568 negatively impacted on GOP and ETR, while (photo)respiration and isoprene emissions increased (see
569 T_{opt} values in **Table 2** and **Figure 6**). Due to the concurrent decrease in O₂ production (GOP) and
570 increased O₂ sinks like (photo)respiration, this resulted in further decline in A_{net} and NOP up to the highest
571 leaf temperature used here (40 °C).

572

573 **Isoprene emission and its potential relationship with NOP and GOP**

574 The response of leaf isoprene emissions to light (PAR), intercellular CO₂ mole fraction (C_i), and leaf
575 temperature was broadly consistent with the common assumption of isoprene energetics models
576 (commonly referred to as Niinemets et al. (1999) model) that isoprene emission relies on available
577 reducing power in chloroplasts (Morfopoulos et al., 2013). Here, we propose that this model could be
578 extended to represent all lipids (isoprenoids and fatty acids) synthesized de novo in chloroplasts. This
579 assumption reflects situations where the demand by the Calvin-Benson cycle for CO₂ assimilation
580 outcompetes other pathways (e.g. the MEP pathway for isoprenoid biosynthesis) for ATP/NADPH
581 (Rodrigues et al., 2020). In fact, CO₂ assimilation and photorespiration are the greatest sinks for
582 ATP/NADPH, and control on lipid synthesis can occur when the effective Michaelis-Menten constant for
583 ATP or/and NADPH is high for the MEP and fatty acid pathways (Rasulov et al., 2016). Thus, one may
584 anticipate that at elevated C_i (e.g. due to elevated atmospheric CO₂), a suppression of isoprene emissions
585 together with a stimulation of A_{net} and NOP occurs, due to the increased demand for photosynthetic ATP/
586 NADPH by the Calvin cycle (Morfopoulos et al., 2014; Niinemets et al., 2021; Rasulov et al., 2018), and
587 this is what we observed (**Figure 4** and supplementary **Figure S3**). Similarly, at low light, isoprene
588 emissions were barely detectable despite significant NOP and A_{net} fluxes, probably due to limited excess
589 ATP/NADPH. Conversely, under light saturating conditions, further increases in PAR did not
590 significantly enhance A_{net} or NOP, but stimulated increased isoprene emissions, likely due to increased
591 ATP/NADPH availability (**Figure 3** and supplementary **Figure S2**) associated with a decline in C_i
592 (supplementary **Figure S5**). Consequently, as predicted from isoprene photosynthesis energetic models
593 and previous experimental observations (K. J. Jardine et al., 2016), the fraction of carbon (in % of A_{net} or
594 NOP) emitted as isoprene increased with light intensity (see example Supplementary **Figure S2c**).

595 Also, we observed a progressive increase in isoprene emission with temperature (**Figures 4-5** and
596 supplementary **Figures S5-S6**), consistent with previous studies where isoprene emissions increased up to
597 40 °C in many species (Harley, Monson, & Lerdau, 1999; Rasulov et al., 2010). As observed here in
598 poplar, ETR is frequently reported to have a higher leaf temperature optimum than A_{net} (Sage & Kubien, 2007).
599 Also, isoprene energetic models predict a temperature optimum of isoprene emission that is strongly
600 influenced by the optimal temperature of ETR and isoprene synthase activity, the later which has been
601 reported to be 45 °C or higher (Monson et al., 1992; Ülo Niinemets et al., 1999; Rasulov et al., 2010).
602 Therefore, at leaf temperatures higher than the ETR and GOP optimum (35 °C), the increase in isoprene
603 emissions up to 40 °C could be explained by a high temperature optimum for isoprene synthase (e.g. 45
604 °C) (**Figure 6**). However, we note that C_i decreased at leaf temperatures higher than the optimal for
605 stomatal conductance (i.e., 33 °C) (Supplementary **Figure S7** and **Table 2**). We thus suggest that in
606 addition to the effect of isoprene synthase thermal optimum, the increase in isoprene emission is also
607 driven by lower ATP/NADPH utilization for carboxylation in the Calvin cycle, resulting from stomatal
608 closure and the decline in C_i . Although O_2 fixation (photorespiration) increasingly consumes
609 photosynthetic ATP/NADPH as temperature increases in the light (Voss et al., 2013), it is believed that
610 lipid biosynthesis also consumes excess ATP/NADPH not utilized by the Calvin and photorespiratory
611 cycles (Rasulov, Hüve, Vålbe, Laisk, & Niinemets, 2009). These processes occur in parallel with other
612 known processes that help relax the chloroplast redox poise at high temperatures including the
613 malate/oxaloacetate shuttle (Selinski & Scheibe, 2019) and assimilatory nitrate (Bloom, 2015) and sulfate
614 (Abadie & Tcherkez, 2019) reduction. Consistent with experimental and modeling studies demonstrating
615 a lack of direct stomatal control over isoprene emissions (Niinemets & Reichstein, 2003), we observed
616 isoprene emissions increase with transpiration as a function of temperature despite partial stomatal
617 closure (**Figure 4** and supplementary **Figure S5**). At high leaf temperatures, the continued increase in leaf
618 transpiration can be explained by a dominant effect of increasing leaf-to-atmosphere water vapor
619 concentration gradients (vapor pressure deficit, VPD). Likewise, reduced stomatal conductance does not
620 suppress light-dependent isoprene emissions due to high production rates quickly generating a larger leaf
621 to atmosphere isoprene concentration gradients to overcome stomatal limitations on emissions (Thomas D
622 Sharkey & Yeh, 2001).

623 Taken as a whole, our results agree with the availability of ATP/NADPH in the chloroplast being
624 rate-limiting for isoprene synthesis (Rasulov, Huve, et al., 2009; Rasulov et al., 2018) and suggest that
625 global carbon-chemistry-climate models that predict isoprene emissions from the terrestrial biosphere
626 using a photosynthesis-based energetics model are valid (Unger et al., 2013). While carbon limitations for
627 isoprene biosynthesis have been generally considered negligible, previous studies using CO_2 -free air

628 demonstrated that light-dependent isoprene emissions can occur at surprisingly high rates, are light and
629 temperature stimulated, depend on electron transport, and associated with refixation of (photo)respiratory
630 CO₂ (Garcia et al., 2019). Our observations are consistent with the idea that carbon limitation for isoprene
631 synthesis occurs only at very low C_i (Lantz et al., 2019) and suggests that CO₂ refixation in leaves is a
632 carbon source for isoprene synthesis under photorespiratory conditions (i.e. high light and temperature)
633 and thus could be a potentially important thermotolerance mechanism (especially if generalizable to light-
634 dependent plastidic lipid synthesis). When photorespiration is high, such as during heat stress, increased
635 lipid synthesis probably contributes to regulating chloroplast redox poise by consuming excess
636 photosynthetic ATP/NADPH. This would in turn help mitigate excessive reactive oxygen species
637 formation and oxidative damage to the photosynthetic machinery and thereby provide resilience to
638 photosynthetic parameters such as maximum carboxylation velocity and membrane stability (Loreto et
639 al., 2001; Loreto & Velikova, 2001). In other words, the present results support the notion that plastidic
640 lipid synthesis plays a role in protecting photosynthesis against damage during high light, heat, and
641 drought stress and therefore plays an important, but poorly quantified indirect role in terrestrial carbon
642 cycling under climate extremes (Velikova, Loreto, Tsonev, Brillì, & Edreva, 2006).

643

644 **Conclusions and perspectives**

645 Our study shows that coupling O₂ and isoprene exchange to traditional CO₂/H₂O gas exchange is possible,
646 using CRDS-based oxygen and PTR-MS-based isoprene measurements. This configuration allows a more
647 complete picture of the photosynthetic redox budget via photosynthetic production of O₂, electron
648 transport rate (ETR), and isoprene biosynthesis. This opens avenues for useful measurements during
649 photosynthesis, such as the temperature sensitivity of gross oxygen production (GOP) using ¹⁸O-water
650 labeling, and the assimilatory quotient (AQ) which appears to be suppressed at high leaf temperature.
651 However, as accurate measurements of both A_{net} and NOP are needed to calculate AQ, great care in
652 calibrating the separate analytical sensors is needed with a suite of high accuracy standards spanning the
653 observed concentration range. Also, our findings may help resolve some confusion in the literature as to
654 whether isoprene emissions, and perhaps lipid synthesis in chloroplasts in general, may or may not be
655 directly linked to net photosynthesis. In agreement with numerous previous studies, we found that
656 isoprene emission can be uncoupled from A_{net}, i.e., at low C_i and high temperature (**Figures. 3-4**, and
657 supplementary **Figures S4-S5**), and thus it is unlikely that lipid biosynthesis in chloroplasts strictly
658 depends on photosynthesis rate or carbon provision by photosynthates. Therefore, our results suggest that
659 (i) isoprene synthesis (and potentially lipid synthesis in general) in chloroplasts is related to electron
660 generation by photolysis and thus probably via excess photosynthetic ATP/NADPH (not consumed by the

661 Calvin cycle, the photorespiratory cycle, and other pathways acting in parallel like the
662 malate/oxaloacetate shuttle), and (ii) is carbon-limited only when gross photosynthesis declines
663 considerably. We nevertheless recognize that dual isotopic labelling with $^{13}\text{CO}_2$ and ^{18}O -water together
664 with total isoprenoid and fatty acid synthesis rates would be useful to ascertain this and quantify precisely
665 the temperature dependencies between ^{13}C -lipid appearance and $^{18}\text{O}_2$ evolution. This will be addressed in
666 another study.

667 **Supplemental data section**

668 The following supplemental materials are available in the online version of this article. All raw and
669 derived leaf gas exchange and chlorophyll fluorescence data presented in **Figures 2-6** and supplementary
670 **Figures S2-S7** with this manuscript are available to download free of charge as a supplementary data file.

671 **Supplementary Discussion:** Pros and cons of coupling gas exchange to O₂ and isoprene flux
672 measurements

673 **Supplementary Discussion:** Pros and cons of the ¹⁸O-labeling method

674 **Supplementary Discussion:** Plant CO₂/O₂ metabolism and transport and the net assimilatory quotient
675 (AQ)

676 **Supplementary Figures**

677 **Figure S1:** Pros and Cons of quantifying leaf net O₂ production (NOP) fluxes and δ¹⁸O of leaf headspace
678 O₂ using a small dynamic leaf chamber (6 cm²) with integrated chlorophyll fluorimeter and large dynamic
679 leaf chamber with actinic light source (36 cm²).

680 **Figure S2:** Example raw and 1-min averaged O₂ concentrations exiting the dynamics leaf chamber versus
681 time during leaf gas response curves to (a.) CO₂, (b.) light, and (c.) temperature.

682 **Figure S3:** Scatter and linear correlation plots between average gas exchange and photochemical
683 parameters during leaf light response curves (*n* = 5).

684 **Figure S4:** Scatter and linear correlation plots between average gas exchange and photochemical
685 parameters during leaf internal CO₂ (C_i) response curves (*n* = 7).

686 **Figure S5:** Scatter and linear correlation plots between average gas exchange and photochemical
687 parameters during leaf temperature response curves (*n* = 15).

688 **Figure S6:** Scatter plot of average leaf isoprene emissions in the light and □¹⁸O of headspace O₂ plotted as
689 a function of leaf temperature following leaf equilibration with ¹⁸O-water (*n* = 6).

690 **Figure S7:** Example dependence of C_i and g_s on (a) PAR during a leaf light response curve and (b) leaf
691 temperature during a leaf temperature response curve.

692 **Supplementary References**

693

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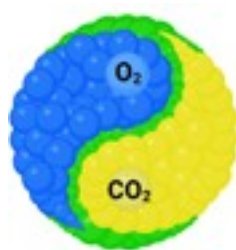
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702

703 **Conflict of Interest Statement**

704 The authors have no conflict of interest to declare.

705 **Tables and Figures**



706
707 **Graphical Abstract:** Integration of CO₂ (yellow), O₂ (blue) and isoprene (green) leaf gas exchange.

708

Leaf Chamber	AQ (Light)	AQ (C _i)	AQ (Temp)
Large chamber	1.223 ± 0.003 (n = 2)	1.272 ± 0.016 (n = 3)	0.975 ± 0.330 (n = 7)
Small chamber	1.268 ± 0.057 (n = 3)	1.192 ± 0.082 (n = 4)	0.793 ± 0.226 (n = 8)
Large + Small chamber	1.250 ± 0.047 (n = 5)	1.227 ± 0.073 (n = 7)	0.865 ± 0.275 (n = 15)

709

710 **Table 1:** Assimilatory quotient (AQ = A_{net}/NOP) determined as the slope from linear regressions between
711 net CO₂ exchange (A_{net}) and net oxygen production (NOP) during leaf gas-exchange response curves to
712 light, leaf internal CO₂ concentrations (C_i) and leaf temperature (under constant light) for both large and
713 small dynamic leaf chambers. AQ values shown for are mean ± 1 standard deviation with n indicating the
714 number of replicate leaf response curves.

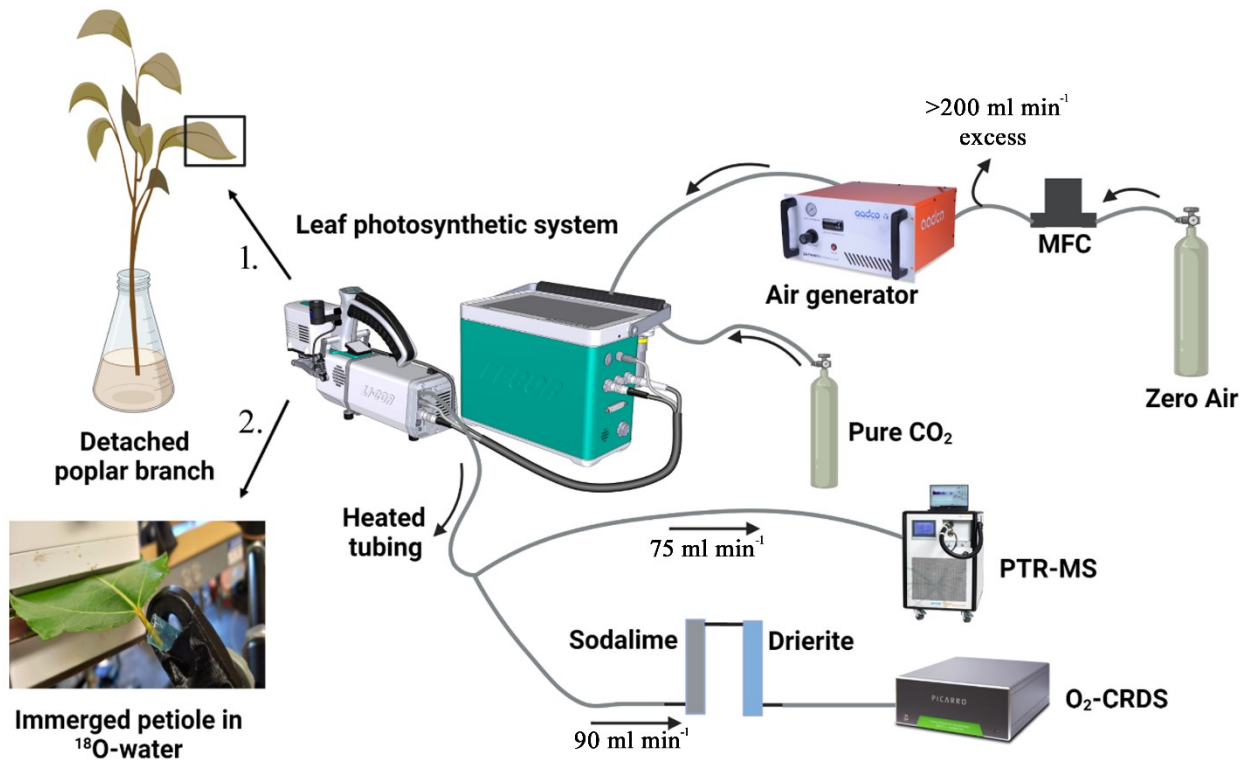
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Parameter	T_{opt} (°C)	Symbol (unit)	Instrument
Net photosynthesis	31.0 ± 3.1	A_{net} ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	Li6800
Net oxygen production	31.0 ± 3.4	NOP ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	CRDS in O ₂ conc. mode
Stomatal conductance	33.0 ± 5.7	g_s ($\text{mol m}^{-2} \text{s}^{-1}$)	Li6800
Photosynthetic Electron Transport Rate	35.0 ± 1.8	ETR ($\mu\text{mol e}^{-} \text{m}^{-2} \text{s}^{-1}$)	Li6800
Gross oxygen production	34.9 ± 1.8	$\delta^{18}\text{O}$ in O ₂ (‰)	CRDS in $\delta^{18}\text{O}$ mode
Transpiration	38.9 ± 2.6	E ($\text{mmol m}^{-2} \text{s}^{-1}$)	Li6800
Leaf isoprene emissions	38.9 ± 1.0	Isoprene ($\text{nmol m}^{-2} \text{s}^{-1}$)	PTR-MS

716

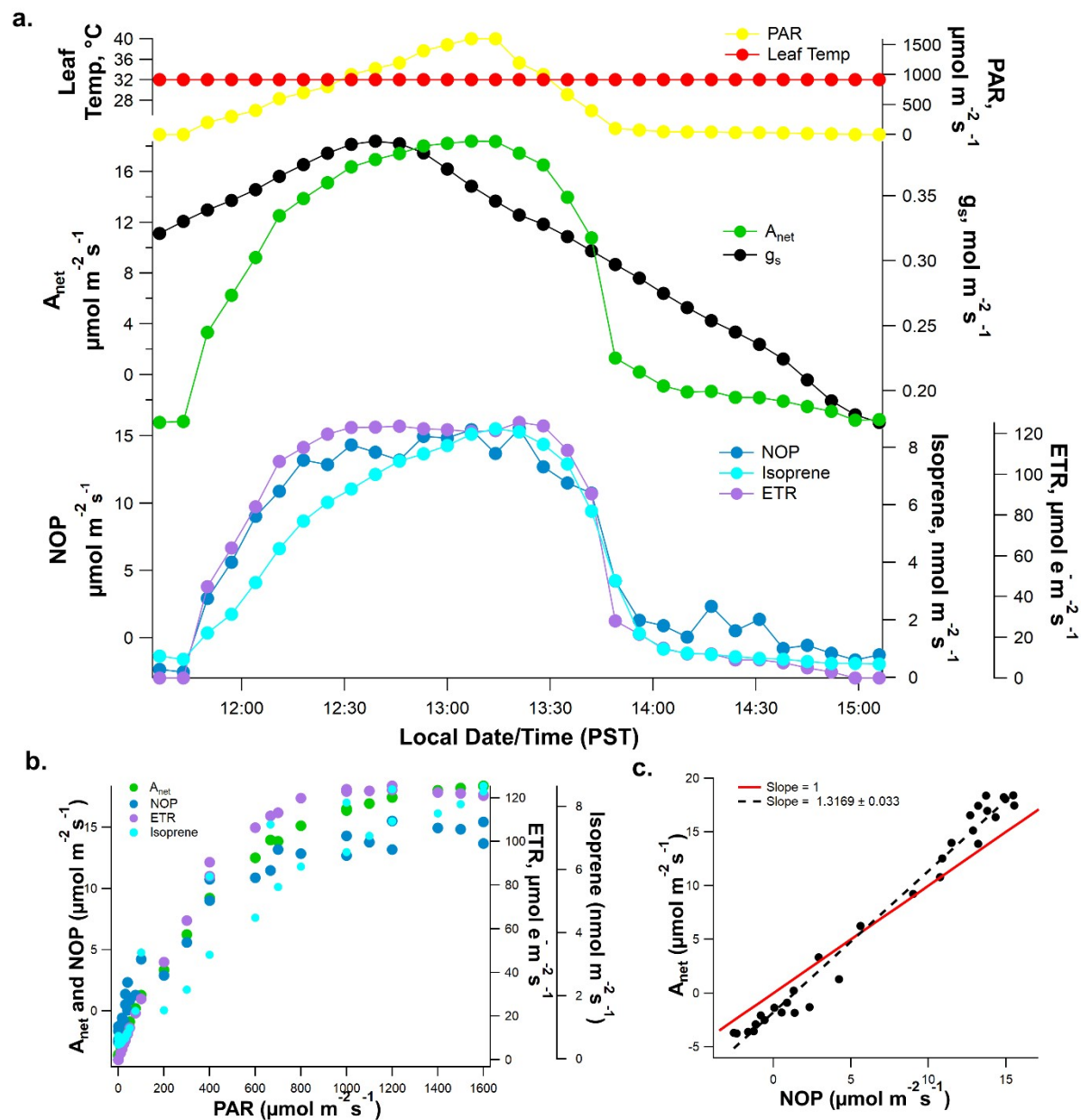
717 **Table 2:** Optimal temperature (T_{opt}) of leaf gas exchange characteristic and electron transport and
718 isoprene emission determined from the leaf temperature response curves.

719

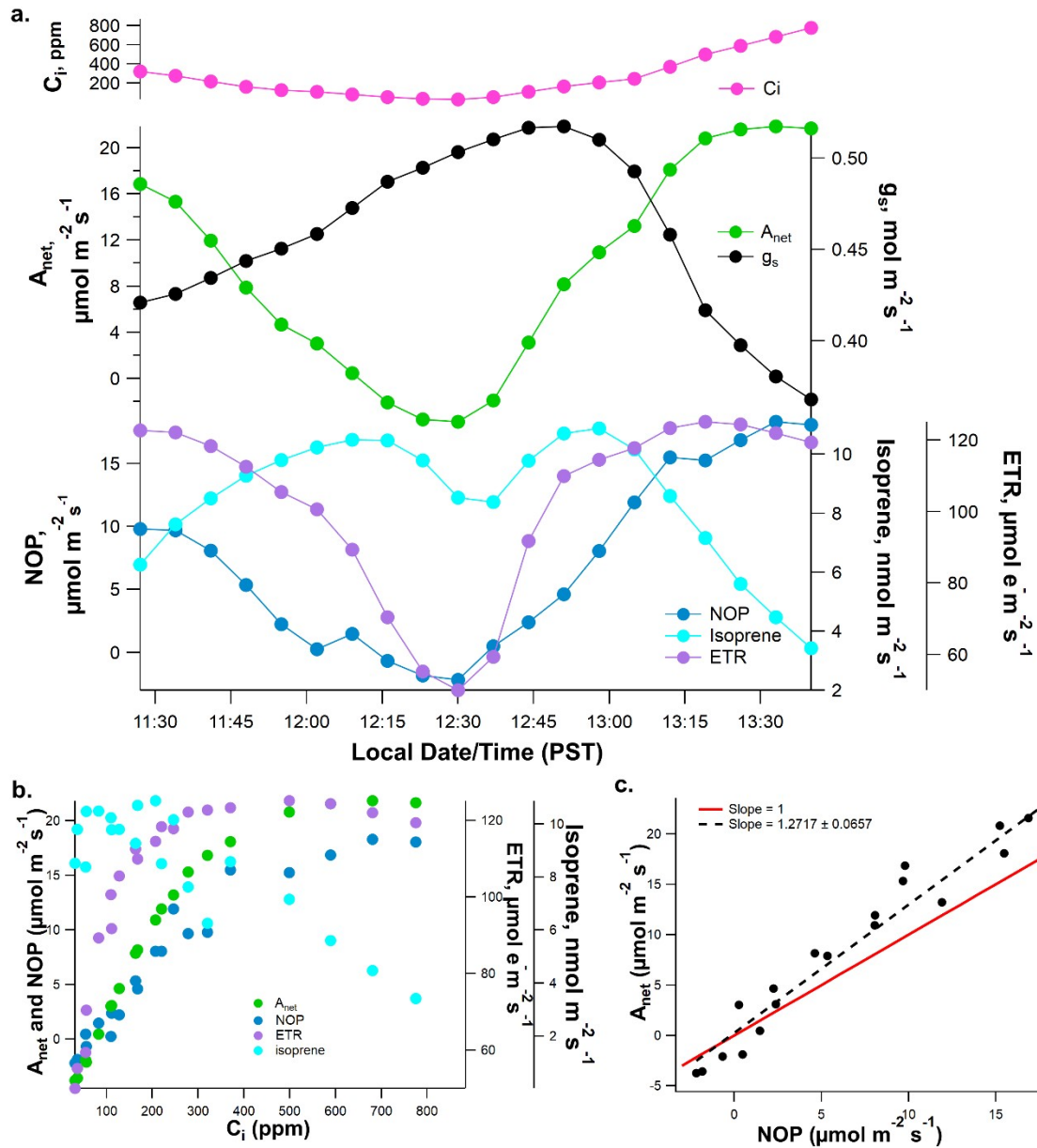


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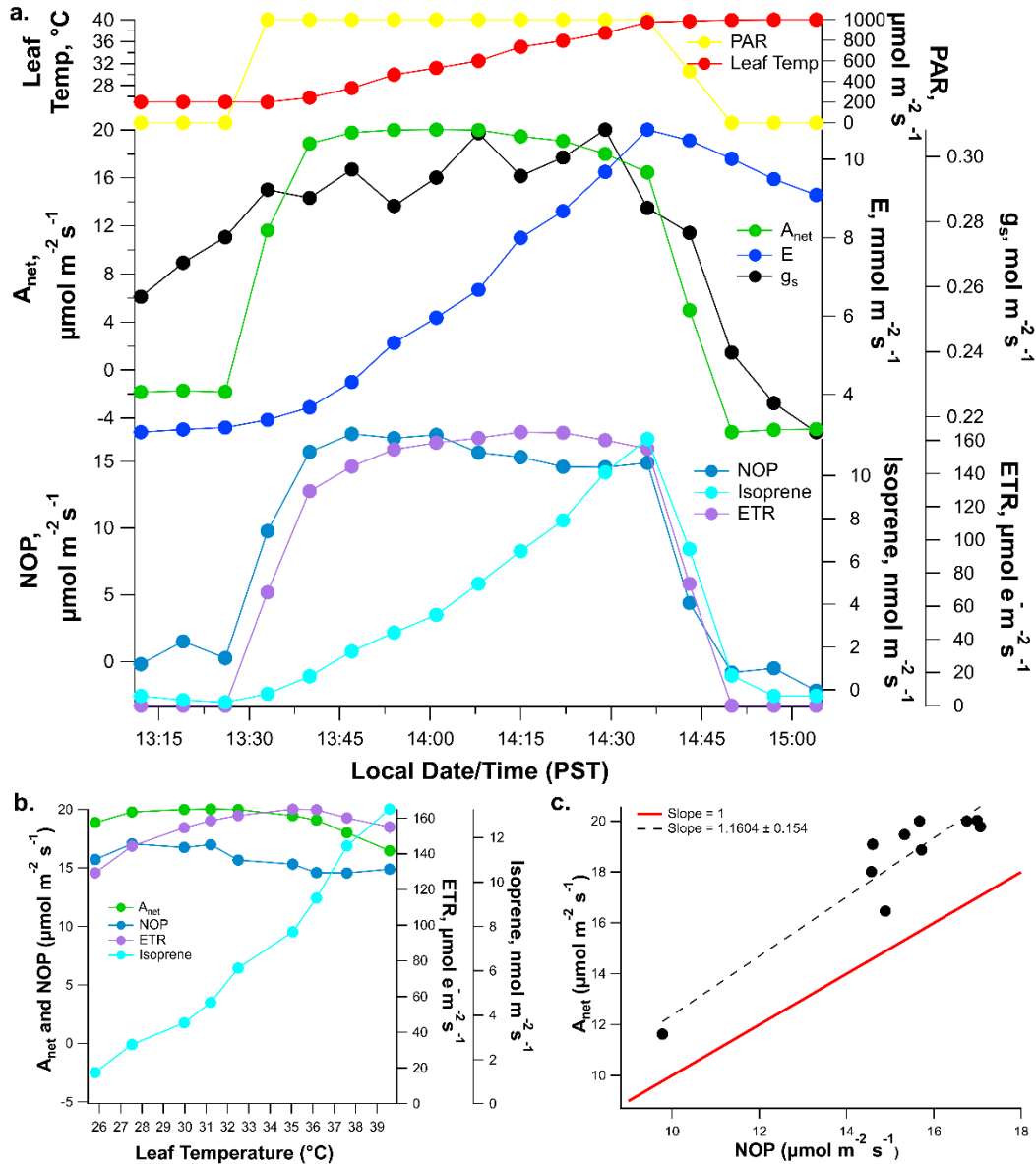
721 **Figure 1.** Schematic diagram of experimental setup for (1) real-time leaf to atmosphere fluxes of CO₂,
 722 H₂O, O₂, and isoprene together with *chlorophyll fluorescence across environmental leaf response curves
 723 of PAR, C_i, and leaf temperature using both small and large leaf chambers (2) real-time leaf to
 724 atmosphere fluxes of CO₂, H₂O, and isoprene together with δ¹⁸O of leaf chamber O₂ during leaf
 725 temperature response curves using the large leaf chamber. *Chlorophyll fluorescence was only quantified
 726 using the small leaf chamber. Note, the air flow rate through the small leaf chamber (6 cm²) varied
 727 between 323-363 mL min⁻¹ (depending on the leaf) while the large chamber (36 cm²) maintained the same
 728 air flow rate (538 mL min⁻¹) for all leaves studied (see discussion).



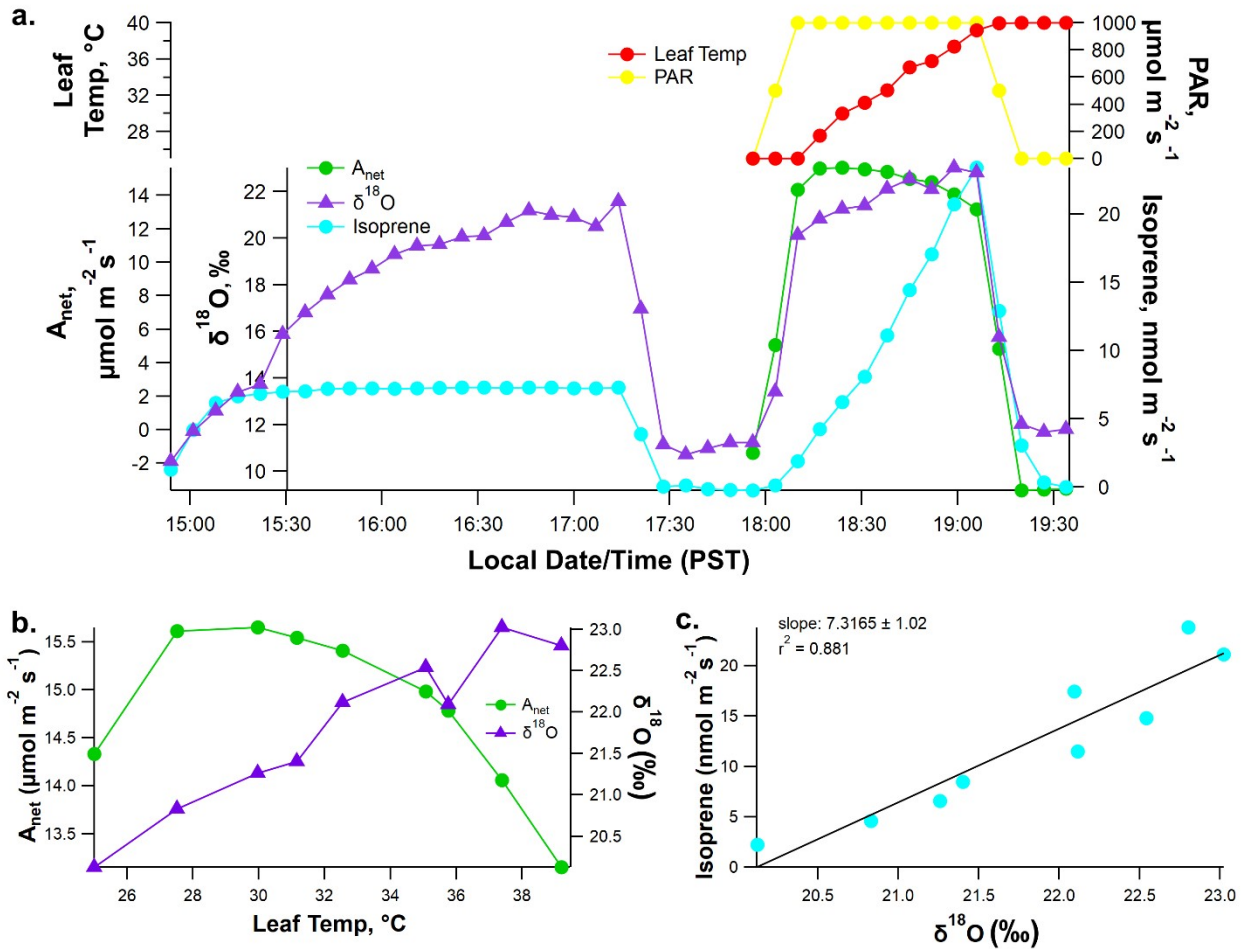
729
 730 **Figure 2. a.** Example real-time leaf-gas exchange fluxes of A_{net} , NOP, and isoprene emissions together
 731 with chlorophyll fluorescence-derived ETR during controlled light response curves (photosynthetically
 732 active radiation, PAR) under constant leaf temperature (32 °C) and leaf chamber headspace CO_2
 733 concentrations (400 ppm) collected using the 6 cm^2 leaf chamber with integrated chlorophyll fluorimeter.
 734 **b.** A_{net} and NOP and ETR and isoprene emissions plotted as a function of PAR. **c.** Linear regression
 735 between A_{net} and NOP. Note the slope of the regression as well as the 1:1 line.



736
 737 **Figure 3** (a). Example real-time leaf-gas exchange fluxes of A_{net}, NOP, and isoprene emissions together
 738 with ETR during controlled C_i response curves under constant leaf temperature (32 °C) and PAR (1000
 739 μmol m⁻² s⁻¹) collected using the 6 cm² leaf chamber with integrated chlorophyll fluorimeter. (b). A_{net} and
 740 NOP together with ETR and isoprene emissions plotted as a function of C_i. (c) Linear regression between
 741 A_{net} and NOP. Note the slope of the regression as well as the 1:1 line.

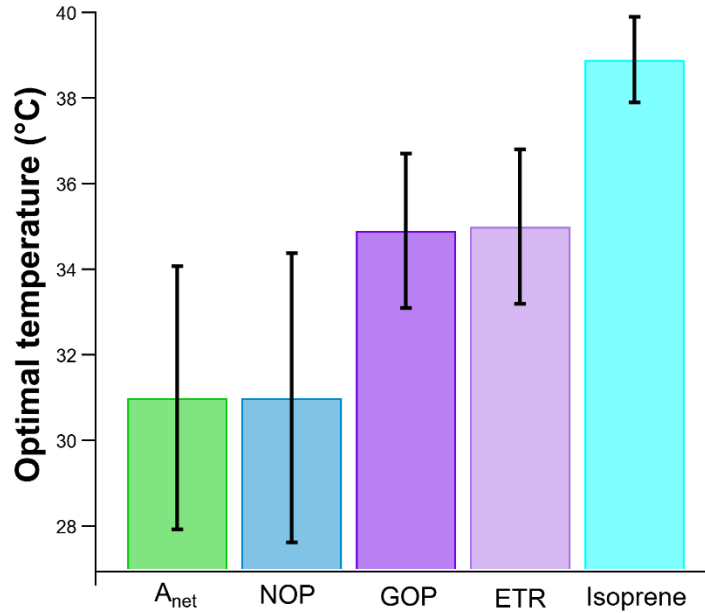


742
 743 **Figure 4** (a). Example, real-time leaf-gas exchange fluxes of A_{net} , NOP, and isoprene emissions together
 744 with ETR during controlled leaf temperature response curves under constant leaf headspace enclosure
 745 CO_2 (400 ppm) and PAR ($1000 \mu\text{mol m}^{-2} \text{s}^{-1}$) collected using the 6 cm^2 leaf chamber with integrated
 746 chlorophyll fluorimeter. (b). A_{net} and NOP together with ETR and isoprene emissions plotted as a function
 747 of leaf temperature. (c). Linear regression between A_{net} and NOP in the light. Note the slope of the
 748 regression as well as the 1:1 line shown.

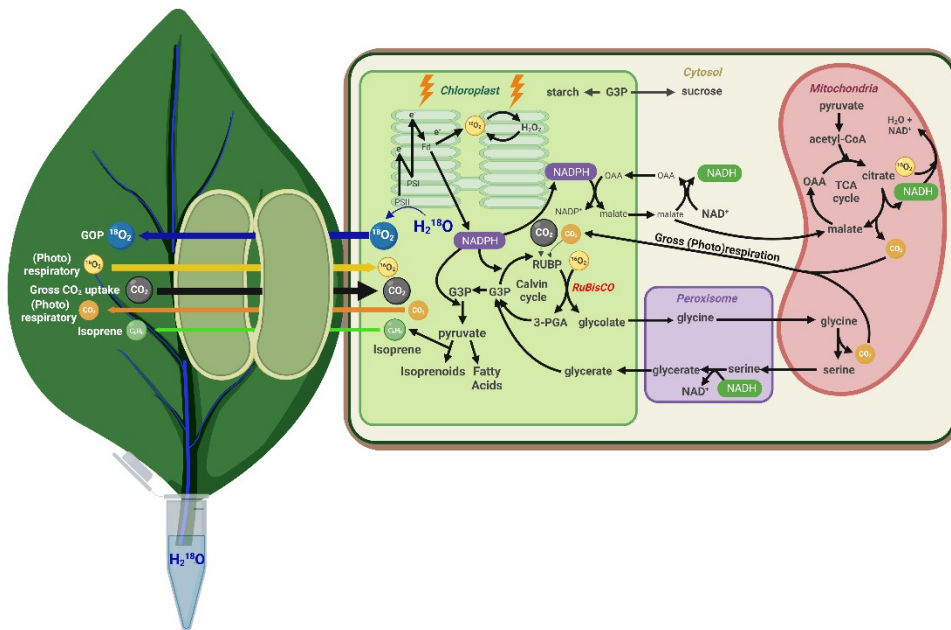


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Figure 5. Example dynamics of ¹⁸O-labeled O₂ evolution in the light as a function of leaf temperature from a detached poplar leaf equilibrated with ¹⁸O-water (□¹⁸O 8000 ‰) using the 36 cm² leaf chamber. Pretreatment occurred under constant PAR (1000 μmol m⁻²s⁻¹), leaf temperature (32 °C), and leaf enclosure headspace CO₂ (400 ppm). **a.** Example, real-time leaf-gas exchange fluxes of A_{net} and isoprene emissions together □¹⁸O in headspace O₂ during a controlled leaf temperature response curves under constant leaf headspace enclosure CO₂ (400 ppm). Following equilibration, the light was switched off (PAR 0 μmol m⁻²s⁻¹) and the leaf temperature reduced to 25 °C. Following measurements of dark gas parameters, PAR was returned to 1000 μmol m⁻² s⁻¹ and the leaf temperature response curve was initiated (25-40 °C). Finally, the light was switched off to determine the dark gas exchange rates at 40 °C leaf temperature, **b.** A_{net} and □¹⁸O of O₂ plotted as a function of leaf temperature, **c.** Linear regression between isoprene emissions and □¹⁸O of O₂ across leaf temperature in the light (PAR, 1000 μmol m⁻² s⁻¹). Note the slope of the regression as well as the 1:1 line shown.



762
 763 **Figure 6.** Average optimum temperature (T_{opt}) of net CO_2 assimilation (A_{net}), net oxygen production
 764 (NOP), gross oxygen production (GOP), photosynthetic electron transport rates (ETR), and isoprene
 765 emission during controlled leaf temperature response curves ($n = 15$) using the small ($n = 8$) and large (n
 766 $= 7$) leaf chambers. Vertical error bars represent ± 1 standard deviation. T_{opt} for GOP was only determined
 767 with the large chamber (36 cm^2) and T_{opt} for ETR was only determined with the small chamber (6 cm^2).
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 769



770
 771 **Figure 7:** Simplified metabolic model of primary CO_2 and O_2 metabolism at elevated leaf temperatures
 772 (e.g. $35\text{ }^\circ\text{C}$) in poplar leaves (accelerated metabolism). Elevated temperature leads to a suppression of
 773 stomatal conductance (g_s), net oxygen production (NOP), and net atmospheric CO_2 uptake (A_{net}) and a
 774 stimulation of photosynthesis, (photo)respiration, and internal CO_2/O_2 recycling and isoprenoid synthesis
 775 consuming ATP/NADPH. Note the activity of the water-water cycle is depicted as the cycling between O_2
 776 and H_2O_2 .

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