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Author Shackley, M. Steven

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GEOARCHAEOLOGICAL XRF LAB

GEOARCHAEOLOGICAL X-RAY FLUORESCENCE SPECTROMETRY LABORATORY 8100 WYOMING BLVD., SUITE M4-158 ALBUQUERQUE, NM 87113 USA

X-RAY FLUORESCENCE (XRF) ANALYSIS MAJOR AND MINOR OXIDE CONCENTRATIONS OF TWO ROCK SAMPLES FROM THE BIG HOLE SITE (41TV2161), TRAVIS COUNTY, TEXAS

by

M. Steven Shackley Ph.D., Director Geoarchaeological XRF Laboratory Albuquerque, New Mexico

Report Prepared for

Michael Quigg TRC Environmental Austin, Texas

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INTRODUCTION

The qualitative and quantitative analysis here of two sandstone samples suggests that an iron oxide was used in some way on one surface. Qualitative analysis of both surfaces indicate iron oxide on a single surface.

LABORATORY SAMPLING, ANALYSIS AND INSTRUMENTATION

 All archaeological samples are analyzed whole. The results presented here are quantitative in that they are derived from "filtered" intensity values ratioed to the appropriate xray continuum regions through a least squares fitting formula rather than plotting the proportions of the net intensities in a ternary system (McCarthy and Schamber 1981; Schamber 1977). Or more essentially, these data through the analysis of international rock standards, allow for interinstrument comparison with a predictable degree of certainty (Hampel 1984; Shackley 2011).

Major and Minor Oxide Analysis

Analysis of the major oxides of Na, Mg, Al, Si, P, K, Ca, Ti, Mn, and Fe is performed under the multiple conditions elucidated below. The composition of alkalis Na₂O and K₂O, and silica $(SiO₂)$ in these rocks allows for elemental determination of rock type (Table 1 and here).

The fundamental parameter analysis (theoretical with standards), while not as accurate as destructive analyses (pressed powder and fusion disks) is usually within a few percent of actual, based on the analysis of USGS RGM-1 obsidian standard (see also Shackley 2011). The fundamental parameters (theoretical) method is run under conditions commensurate with the elements of interest and calibrated with four USGS standards (RGM-1, rhyolite; AGV-2, andesite; BHVO-1, hawaiite; BIR-1, basalt), and one Japanese Geological Survey rhyolite standard (JR-1).

Conditions of Fundamental Parameter Analysis¹

Low Za (Na, Mg, Al, Si, P)

 1 Multiple conditions designed to ameliorate peak overlap identified with digital filter background removal, least squares empirical peak deconvolution, gross peak intensities and net peak intensities above background.

 2^2 Current is set automatically based on the mass absorption coefficient.

In order to evaluate these quantitative determinations, machine data were compared to measurements of known standards during each run. SARM-69, a South African Bureau of Standards Neolithic ceramic standard was analyzed during each sample run to check machine calibration (Table 1).

Analytic Trajectory

 A quantitative determination of the two rock samples and standard was the first step. Table 1 indicates that SiO_2 , CaO, and $Fe₂O₃$ comprise the bulk of the composition of the rock, consistent with a sandstone, although the iron content is high on one side. The qualitative analysis is more enlightening (Figure 1 a-c). When both sides of the rocks were analyzed qualitatively, it is apparent that the side with reddish hue exhibited a higher concentration of iron. This does suggest that iron oxide was applied to one surface of this sandstone.

REFERENCES CITED

Hampel, Joachim H.

1984 Technical Considerations in X-ray Fluorescence Analysis of Obsidian. In *Obsidian Studies in the Great Basin*, edited by R.E. Hughes, pp. 21-25. Contributions of the University of California Archaeological Research Facility 45. Berkeley.

Hildreth, W.

 1981 Gradients in Silicic Magma Chambers: Implications for Lithospheric Magmatism. *Journal of Geophysical Research* 86:10153-10192.

Hughes, Richard E., and Robert L. Smith

1993 Archaeology, Geology, and Geochemistry in Obsidian Provenance Studies. *In Scale on Archaeological and Geoscientific Perspectives*, edited by J.K. Stein and A.R. Linse, pp. 79-91. Geological Society of America Special Paper 283.

Mahood, Gail A., and James A. Stimac

 1990 Trace-Element Partitioning in Pantellerites and Trachytes. *Geochemica et Cosmochimica Acta* 54:2257-2276.

McCarthy, J.J., and F.H. Schamber

1981 Least-Squares Fit with Digital Filter: A Status Report. In *Energy Dispersive X-ray Spectrometry*, edited by K.F.J. Heinrich, D.E. Newbury, R.L. Myklebust, and C.E. Fiori, pp. 273-296. National Bureau of Standards Special Publication 604, Washington, D.C.

Schamber, F.H.

1977 A Modification of the Linear Least-Squares Fitting Method which Provides Continuum Suppression. In *X-ray Fluorescence Analysis of Environmental Samples*, edited by T.G. Dzubay, pp. 241-257. Ann Arbor Science Publishers.

Shackley, M. Steven

 2011 An Introduction to X-Ray Fluorescence (XRF) Analysis in Archaeology. In *X-Ray Fluorescence Spectrometry (XRF) in Geoarchaeology*, edited by M.S. Shackley, pp. 7- 44. Springer, New York.

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Figure 1. a = analysis of reddish side of sample 2396-0; b = analysis of non-reddish side of sample 2396-0; c = analysis of reddish side of sample 2396-11.