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ENZYMIC REDUCTION OF CODEINONE IN VITRO: CELL-FREE SYSTEMS FROM PAPAVER SOMNIFERUM AND P. BRACTEATUM

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ENZYMIC REDUCTION OF CODEINONE IN VITRO:

CELL-FREE SYSTEMS FROM PAPAVER SOMNIFERUM AND P. BRACTEATUM 2 CRAIG C. HODGES and HENRY RAPOPORT Department of Chemistry and Lawrence Berkeley Laboratory, University of California, Berkeley, CA 94720; USA 10 Key Word Index - Papaver somniferum; Papaver bracteatum; 11 Papaveracae; poppy; cell-free conversion; codeinone; codeine. 12 1 3 Abstract - Cell-free extracts have been prepared from Papaver 14 somniferum which perform the reduction of [16-³H]codeinone to 1.5 .16 codeine. Methodology for quantitating this conversion has allowed 17 conditions for preparing the enzyme extracts to be explored. From 18 P. bracteatum, a cell-free system was also prepared which reduced 1, codeinone to codeine, both of which are foreign alkaloids to this 20 species. 21 22 2 3

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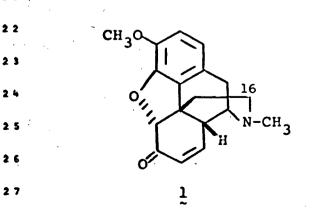
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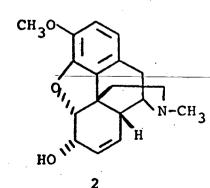
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ENERGIA ENDUCTION

CILL Through identification of the intermediates in the pathway of morphine biosynthesis in Papaver somniferum seems nearly comз plete [1-3], little progress has been made in the study of the fn 1,2,3 enzymic systems which perform the alkaloid transformations. In 3 other than living plants, even detection of these enzymes has 4 been difficult. Successful in vitro conversions of the morphinan 7 alkaloids have been claimed in cases where a radioactive pre-. cursor has undergone a number of enzymic transformations before the product has been isolated [3-6]. Most studies have been fn 4,5,6 1.0 performed on plant latex, which is tedious to obtain and difficult 11 to handle in large quantities. A single study has detected the 1 2 reduction of codeinone after a 16 hour incubation in a crude 13 extract from cultured poppy cells [7]. fn 7 1 4

> To provide a more direct basis for the ultimate goal of isolating the enzymes responsible for alkaloid biosynthesis, our studies have utilized cell-free extracts from entire opium plants to demonstrate the specific <u>in vitro</u> conversion of codeinone (1) to codeine (2). Methods of quantitating nmole conversions were developed to allow the possibility of optimizing the preparation of active enzyme extracts.





Recent studies have indicated that, although codeinone is
 not a natural product in <u>P. bracteatum</u>, the intact plants can
 reduce injected codeinone to codeine [8,9]. Therefore, an enzyme
 extract of <u>P. bracteatum</u> was made and tested to see if indeed such
 enzymic activity could be detected.

RESULTS AND DISCUSSION

in 8,9

To achieve high sensitivity and reaction specificity,
 our assay monitored the conversion of radio-labelled substrate
 to product. [16-³H]Codeinone was synthesized and purified free of
 codeine by TLC and HPLC. Its specific activity was determined by
 standard techniques of liquid scintillation counting and quanti tative mass determinations by HPLC.

14 [16-³H]Codeinone in aqueous solution was mixed with samples 15 of the enzyme extracts containing added NADH. After 15 min 16 incubation periods, the enzymic conversions were halted by acidifi-17 cation and cooling. Unlabelled carrier codeinone and codeine 18 were added and the alkaloids recovered by organic solvent extraction. 19 Codeine was purified by TLC and its specific activity determined 20 by liquid scintillation and GLC.

To quantitate the enzymic conversion, the concentrations of endogenous substrate and product also had to be determined, and for this purpose separate samples of the enzyme extract were used. Endogenous codeinone was determined by reduction with NaBT₄ of known specific activity for this reduction. Endogenous codeine was determined by an isotope dilution technique. Control experiments for each enzyme extract determined the contamination of codeine

with radioactivity from the substrate resulting from non-enzymic
processes. These assays were performed on enzyme preparations
previously heat-inactivated. Contamination of codeine ranged from
0.5 to 1.5% of the total radioactivity initially present as [16-³H]codeinone.

As a test to demonstrate in vitro enzyme activity and assay precision, an 83 g bolting plant was homogenized in 150 mM Tris HCl buffer, pH 7.5, to yield 115 ml of enzyme extract. Using 3 ml of this extract, the endogenous codeine concentration was determined to be 48.0 μ g/ml, whereas a 1 ml sample of the 10 extract was found to contain 0.68 µg (2.15 nmoles) of endogenous 11 codeinone. A four ml sample of the extract was heat-inactivated 12 for the control experiment. This sample was treated with NADH 13 and [16-³H]codeinone (2.41 µg, 8.13 nmoles). Recovered codeine 14 from the assay was contaminated with only 1.47% of the $[16-{}^{5}H]$ -15 codeinone radioactivity. When 4 ml of the live enzyme preparation 16 was treated with the same quantities of NADH and $[16-^{3}H]$ codeinone, 17 26.1% (4.42 nmoles) of the total codeinone (4x2.15 + 8.13; 16.73 18 nmoles) was converted to codeine. Smaller volumes of the enzyme 19 preparation were assayed and gave reasonably proportional conversions. 20 These results thus clearly demonstrate quantitative and reproducible 21 evidence for the in vitro enzymic reduction of [16-3H]codeinone 22 and are presented in Table 1. 23

Table l here

> The same assay procedure was utilized to develop improved conditions for preparing the enzyme extract. Activity was detectable if the plant was homogenized in either 150 mM Tris·HCl or 150 mM MES [2-(N-morpholino)ethanesulfonic acid]/NaOH with pH

ranging from 6.3 to 8.0. Optimum pH for the assay was found to be
7.5 over the pH range 6 to 8 in 0.5 unit increments. This may
represent the combined effects of pH on both enzyme turnover rate
and enzyme stability in these crude extracts.

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Suggestions in the literature [10,11] were followed in
addressing problems of the loss of enzyme activity caused by
endogenous polyphenols and phenol oxidases. Attempts to remove
polyphenols by absorption on Amberlite resins XAD-2 and XAD-4 and
insoluble polyvinylpyrrolidone (PVP) did not significantly increase
the yield of enzyme activity. Attempts to chelate polyphenols
with soluble polyvinylpyrrolidone slightly decreased activity.
These results are summarized in Table 2.

To reduce the production of enzyme inactivating products from 13 phenol oxidases, the plant extracts were prepared by homogenization 1.4 in a nitrogen atmosphere with nitrogen-purged buffers. Despite these 15 precautions, the extracts developed discoloration as evidence 16 of phenol oxidase activity. Recommended thiol and sulfite inhibitors 17 reacted with codeinone and were thus unsuitable. Although codeinone 18 was stable in the presence of ascorbic acid, additions of this 19 reagent showed no promoting or stabilizing effect on codeinone 20 reduction activity. Even with storage at 0° under nitrogen, these 21 crude extracts lost detectable ability to reduce codeinone within 22 3 hr after plant homogenization. 23

Both plant age and plant tissue were examined as variables. A bolting plant was simply dissected into roots, stems, and leaves, and these tissues were assayed separately. All tissues showed similar concentrations of codeinone reducing activity, based on

fresh weight of tissue. Thus, our crude extracts were later prepared from whole plants.

Plants of various ages were assayed. Codeinone-reducing
activity per gram tissue increased with plant age. Maximum activity
was obtained from bolting plants with buds on flowers.

Eight-month old, non-bolting specimens of <u>P. bracteatum</u> were
used to prepare an enzyme extract. Codeinone-reducing activity
per gram tissue was definite, although low compared to bolting
<u>P. somniferum</u>. This result corroborates our previous studies in
which radioactivity from labelled codeinone injected into intact
<u>P. bracteatum</u> plants was recovered in codeine [8]. The summary of
these enzymic and control experiments is given in Table 3.

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fresh weight of tissue. EXPERIMENTAL

Materials: The plant materials, conditions of GLC analyses, and the synthesis of $[16-^{3}H]$ codeinone have been described previously [8,12]. All alkaloids were purified on analytical silica gel TLC plates containing UV fluor. TLC plates for $[16-^{3}H]$ codeinone purification were developed in CHCl₃/CH₃OH/Et₃N (100/20/1). The recovered $[16-^{3}H]$ codeinone was further purified by HPLC using a Spectra Physics SP-8000 instrument equipped with a UV detector set at 280 nm. Purification was achieved on a 3.2x250 mm column packed with 10μ LiChrosorb RP-18 using an eluting solvent of CH₃OH/H₂O (10/1) containing 0.01% conc NH₃ flowing at 0.4 ml/min.

Enzyme extracts. All extract preparation was performed in a nitrogen-filled glove box at 3°. Buffers were deoxygenated by bubbling N₂ through for several hr. Fresh plants were homogenized in a glass blender with std buffer (150 mM Tris·HCl, pH 7.5) using 1.3 ml buffer per l g tissue. The homogenate was filtered through Miracloth to give the enzyme extract.

Enzymic conversion of [16-³H]codeinone to [16-³H]codeine. 18 In the nitrogen box, a 4 ml portion of the enzyme extract was 19 placed in a septum-capped vial. The closed vial was removed from 2 0 the box and placed in a water bath at 24°. Rapidly, 100 µl of std 21 buffer containing 2 µmoles of NADH was injected by syringe. By a 22 second injection, 50 µl of 1 mM HCl containing 8.13 nmoles of 23 $[16-{}^{3}H]$ codeinone $(1.56 \times 10^{5} \text{ dpm})$ was added. (The amount of ${}^{3}H-$ 24 codeinone varied from assay to assay but was determined exactly 25 by scintillation analysis of an aliquot of the alkaloid soln.). 26 The vial was shaken for 15 min and the enzymic reaction was then 27

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stopped by adding 50 μ l of 85% H₃PO₄ and cooling the vial in ice-1 water. To the soln was added 150 µg each of non-radioactive 2 codeinone and codeine in 100 µl CH₃OH. The soln was then extracted 3 2 times, each with 2 ml of CHCl₃ containing 0.1% cyclohexanone [13]. Phase separation was hastened by 1 min spins in a clinical centri-5 fuge. The organic phases were discarded and the remaining aqueous 2 fraction was cooled in ice-water. Basification with 300 µl conc 5 NH40H was followed by 3 extractions as above. The organic phases were removed, combined, and evaporated under N2. Traces of water were removed under vacuum. The residue was resuspended in 75-100 10 μl of acetone and spotted on a TLC plate at room temp which was 11 developed 3 times in acetone containing 0.5% Et₃N. The codeine 12 band was scraped from the plate and the [16-3H]codeine recovered 1 3 by elution with $CH_2Cl_2/CH_3OH/cyclohexanone (4/1/0.1%)$. To the 14 eluted [16- 3 H]codeine was added 150 µg of non-radioactive 15 codeinone in the eluting solvent, and the solution was evaporated 16 under N_2 , dissolved in acetone, and spotted on another TLC plate. 17 This plate was developed once in Et₃N/CHCl₃/CH₃OH (10/1/1). The 18 codeine band was scraped off and the [16-³H]codeine freed from the 19 silica by elution with CH_2Cl_2/CH_3OH (4/1). The soln was evaporated 20 under N₂ and dissolved in exactly 500 μ l CH₃OH. Codeine concen-21

traction was determined by GLC and the rest of the sample was used for liquid scintillation analysis.

To assay less than 4 ml of extract, the desired volume of extract was diluted with std buffer to give a total volume of 4 ml. Control assays were performed on extract samples inactivated by heating the sample in boiling water for 15 min.

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To calculate the total codeinone reduced, T (μ g): 1 Vater. Provide zila 2 concernent $T = (\frac{u + EC}{v})(\frac{r}{v})(z + 4 ECO)$ à where u = carrier codeine (µg); EC = endogenous codeine (µg see below); q = recovered codeine (µg); r = dpm of recovered codeine; v = dpm added as ³H-codeinone; z = amount of ³H codeinone (μ g); ECO = endogenous codeinone (μ g - see below). Endogenous codeine. To 4 ml of enzyme extract was added 16.8 µg (6.30x10⁴ dpm) of carrier $[16-{}^{3}H]$ codeine in 2 ml CH₃OH. Codeine was recovered and analyzed for specific activity by the 11 methods described above. Using the radioactivity recovery to 12 determine a normalization factor, the endogenous codeine, EC (μg) 1 3 in the extract before alkaloid purification could be calculated. 14 15 Thus: $EC = \left(\frac{bc}{2}\right) - d$ 16 17 where a = recovered dpm; b = total added dpm; 18 $c = recovered codeine (\mu g); d = carrier codeine (\mu g)$ 19 Endogenous codeinone. NaBT4, previously titrated as described 20 21 below, was suspended in 1 ml CH₂OH. To this was added 1.00 ml 22 enzyme extract and the mixture shaken for 10 min. The excess 23 NaBT, was destroyed by adding 1 ml of acetone and the mixture was 24 concentrated under a stream of N2. To the remaining solution were 25 added 2 ml std buffer then 150 μ g non-radioactive codeine in 100 μ l ²⁶ CH₃OH, and the codeine was then isolated and purified as above. The total production of [16-³H]codeine was calculated from

the specific activity of the purified codeine and the known masses
of endogenous and added carried codeine. Then the specific
activity of the NaBT₄ allowed determination of the concentration
of endogenous codeinone, ECO (µg).

Thus: See ball to be bar it

ECO = $\left(\frac{e}{s \cdot f}\right) \left(g + \frac{EC}{4}\right)$

where e = recovered dpm; s = specific activity of NaBT₄ reduction (see below); f = recovered codeine (µg); g = added carrier codeine.

Specific activity of NaBT₄. NaBT₄ (approx. 1.5 Ci/mmole)
was dissolved in an aqueous solution of 0.2% NaOH. Equal portions
of the soln, each containing about 1 µmole of NaBT₄, were placed
in vials convenient for the determination of endogenous codeinone.
The water was removed by evaporation <u>in vacuo</u> to leave stable
NaBT₄ residues in the vials.

To determine the specific activity by codeinone reduction, S, (as dpm incorporated into 1 μ g codeinone by reduction to codeine) the endogenous codeinone determination was performed on two fresh plant extracts; one neat and one to which was added 1.24 μ g of codeinone/ml plant extract. The difference between the radioactivity yields of total [16-³H]codeine from these two determinations represented the incorporation of tritium by reduction of the added codeinone. The results agreed with the determination performed on 1.24 μ g codeinone in buffer only.

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Thus: solid activity of the putties and the second sec 1 of endogenous $a:\left(\frac{hm}{n}\right) = \left(\frac{hm}{n}\right) = \left(\frac$ ż equivity of the NaDDy plitter detroped of the proŝ where h = recovered dpm; m = carrier plus endogenous codeine; **n** = recovered codeine; **x** = from extract with exogenous codeinone; y = from neat extract; p = amount of exogenous codeinone (µg). where e = recovered dpt : e = specific claim log of NaEl.

Effects of pH on extract activity. A bolting plant was ground in 150 mM MES, pH 7.0. From the filtered extract, 10 ml 10 portions were adjusted to desired pHs with NaOH or HCl solns. The 11 adjustments were performed in the N2 box at 3° with a calibrated 12 pH meter. The portions were then assayed. Similar tests were 1 3 performed with 150 mM tris as the grinding buffer. 14

Attempts to remove or deactivate polyphenols. Amberlite 15 XAD-2 and XAD-4 were washed several times with distilled water, 16 equilibrated with buffer, and drained. Insoluble (360,000 MW) PVP 17 was boiled in 1N HCl for 1/2 hr, filtered, washed with distilled 18 water, neutralized with NaOH, washed with distilled water then 19 buffer, and drained. Buffer solutions of 10,000 MW PVP (500 mg/ml 20 and 50 mg/ml) were prepared. The reagents were placed in the 21 nitrogen box. Enzyme extract aliquots (8 ml) were combined with 22 l g (wet wt) of adsorbent or 2 ml of reagent soln, or, for the 23 control experiment, 2 ml of std. buffer. The resulting mixtures 2 4 were shaken for 3 min and filtered through glass wool to give 2 5 treated enzyme extracts which were assayed in normal fashion. 2 6 Into

Stabilization of enzyme activity with ascorbic acid.

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separate flasks were placed 60 mg of sodium ascorbate, dissolved
in 2 ml of buffer, and 600 mg of sodium ascorbate in 2 ml buffer.
The control flask contained 2 ml of buffer only. In the nitrogen
box, 28 ml portions of enzyme extract were added to each flask
to give solns of 0, 10 mM, and 100 mM ascorbic acid. The solns
were kept in the cold nitrogen box and aliquots assaysed at 15,
40, 75, 135 and 195 min.

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in part by the National Institute on Drug Abuse and the Division
of Biomedical and Environmental Research of DOE.

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25	13.	To reduce the possibility of Oppenauer equilibration of
26		3 H-codeinone to 3 H-codeine, ketonic solvents and TLC plates
27		at room temperature were used in codeine purification.

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				of codeinone.
° C .:#1.:51117		-		
Assay a	s a fund	tion of sa	mple volume	from a

single extract of <u>P. somniferum</u>. : · · · · ·

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	ract vol.	Codeinone reduced*
2	4, heat inact. [†]	0.25 ‡
	4	4.42
· ·	3	3.73
	2	2.36
	1	0.50
	0.5	0.29

Amount reduced during entire 15 min assay period.

Heat inactivated for control experiment. †

‡ Codeine radioactivity was 1.47% of total added as ³H-codeinone.

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Table 2. Effects on enzyme activity by

treating extract aliquots with phenol adsorbents.

Adsorbent added	3H-Codeinone reduced,
None, heat inact.	1.1
None	20.2
Amberlite XAD-2	22.8
Amberlite XAD-2	25.6
PVP, insoluble*	17.5
PVP, soluble, 10% [†]	15.6
PVP, soluble, 1%	16.2

Polyvinylpyrrolidone; 360,000 MW.

+ Grams of polyvinylpyrrolidone (10,000 MW)
 per ml of extract.

•				1.c. 7.2 7.2		tl	•
Species	Tissue	Age of plant, days	Extract A or HI*	³ H-Codeinone reduced, %		none:∍ ti	nmoles ssue
P. somniferum	.	· • •				·	. <u>.</u>
	stems	125	IH	1.4		: 2213	-
	stems	125	A	18.0	*	6. 6	Ξ
	leaves	125	IH	1.0			E E S
	leaves	125	A	25.0		5.2	: : :
	roots	125	ΗI	1.2	•		3 - 0
	roots	125	A	12.0		2: • ©	•
	whole plant	27 [†]	HI	1,4			e;:
	whole plant	27	A	14.0	: -	5.8	- ::
	whole plant	65 [†]	TH ~	1.6	••••	•••	• <u>-</u>
•	whole plant	65	A	18.0		9.6	212
	whole plant	130 ⁵	IH	- - - -	4 1 1 	:	_ •••
·	whole plant	130	A	20.1		13.3	:
P. bracteatum		ļ				ير. • • •	
	whole plant	265	ΙH	1.0		 	. •
	whole plant	265	Å	5.2		6.0	

Reduction of codeinone to codeine by various extracts of

Table 3.

Active (A) or heat inactivated (HI) extracts.

† Non-bolting.

Beginning to bolt, no bud.

§ Flowering ¶ Non-bolting.

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